

Reaction	Rate coefficient or absorption cross section	Reference
$O_2 + h\nu (184.9 \text{ nm}) \rightarrow 2 O \rightarrow 2 O_3$	$1.0 \times 10^{-20}$	Yoshino et al. (1992) Creasey et al. (2000)
$H_2O + h\nu (184.9 \text{ nm}) \rightarrow OH + H \rightarrow OH + HO_2$	$7.14 \times 10^{-20}$	Cantrell et al. (1997)
$O_3 + h\nu (253.7 \text{ nm}) \rightarrow O_2 + O(^1D_2)$	$1.15 \times 10^{-17}$	Burkholder et al. (2015)
$OH + HO_2 \rightarrow H_2O + O_2$	$1.01 \times 10^{-10}$	Burkholder et al. (2015)
$OH + O_3 \rightarrow HO_2 + O_2$	$8.45 \times 10^{-14}$	Burkholder et al. (2015)
$HO_2 + O_3 \rightarrow OH + 2 O_2$	$2.09 \times 10^{-15}$	Burkholder et al. (2015)
$OH + OH \rightarrow H_2O + O \rightarrow H_2O + O_3$	$1.8 \times 10^{-12}$	Burkholder et al. (2015)
$OH + OH (+M) \rightarrow H_2O_2 (+M)$	$1.59 \times 10^{-11}$	Burkholder et al. (2015)
$HO_2 + HO_2 \rightarrow H_2O_2 + O_2$	$1.30 \times 10^{-12}$	Burkholder et al. (2015)
$HO_2 + HO_2 + M \rightarrow H_2O_2 + O_2$	$3.96 \times 10^{-32}$	Burkholder et al. (2015)
$O(^1D_2) + O_2 \rightarrow O + O_2 \rightarrow O_3 + O_2$	$3.93 \times 10^{-11}$	Burkholder et al. (2015)
$O(^1D_2) + N_2 \rightarrow O + N_2 \rightarrow O_3 + N_2$	$3.05 \times 10^{-11}$	Burkholder et al. (2015)
$O(^1D_2) + H_2O \rightarrow 2 OH$	$1.97 \times 10^{-10}$	Burkholder et al. (2015)