Atmos. Meas. Tech. Discuss., 6, 11345–11403, 2013 www.atmos-meas-tech-discuss.net/6/11345/2013/ doi:10.5194/amtd-6-11345-2013 © Author(s) 2013. CC Attribution 3.0 License.



This discussion paper is/has been under review for the journal Atmospheric Measurement Techniques (AMT). Please refer to the corresponding final paper in AMT if available.

Five years of CO, HCN, C₂H₆, C₂H₂, CH₃OH, HCOOH, and H₂CO total columns measured in the Canadian High Arctic

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Received: 9 October 2013 – Accepted: 3 December 2013 – Published: 20 December 2013

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Published by Copernicus Publications on behalf of the European Geosciences Union.





Abstract

We present a five-year timeseries of seven tropospheric species measured using a ground-based Fourier Transform InfraRed (FTIR) spectrometer at the Polar Environment Atmospheric Research Laboratory (PEARL, Eureka, Nunavut, Canada, 80°05' N,

- $_{5}$ 86°42′ W) from 2007 to 2011. Total columns and temporal variabilities of carbon monoxide (CO), hydrogen cyanide (HCN), and ethane (C₂H₆), as well as the first derived total columns at Eureka of acetylene (C₂H₂), methanol (CH₃OH), formic acid (HCOOH), and formaldehyde (H₂CO) are investigated, providing a new dataset in the sparsely sampled high latitudes.
- Total columns are obtained using the SFIT2 retrieval algorithm based on the Optimal Estimation Method. The microwindows, as well as the a priori profiles and variabilities are selected to optimize the information content of the retrievals, and error analyses are performed for all seven species. Our retrievals show good sensitivities in the troposphere. The seasonal amplitudes of the timeseries, ranging from 34 to 104 %, are
- ¹⁵ captured while using a single a priori profile for each species. The timeseries of the CO, C_2H_6 and C_2H_2 total columns at PEARL exhibit strong seasonal cycles with maxima in winter and minima in summer, in opposite phase to the HCN, CH_3OH , HCOOH and H_2CO timeseries. These cycles result from the relative contributions of the photochemistry, oxidation, and transport, as well as biogenic and biomass burning emissions.
- ²⁰ Comparisons of the FTIR partial columns with coincident satellite measurements by the Atmospheric Chemistry Experiment Fourier Transform Spectrometer (ACE-FTS) show good agreement. The correlation coefficients and the slopes range from 0.56 to 0.97, and 0.50 to 3.35, respectively, for the seven target species.

Our new dataset is compared with previous measurements found in the literature to assess atmospheric budgets of these tropospheric species in the high Arctic. The CO and C_2H_6 concentrations are consistent with negative trends observed over the Northern Hemisphere, attributed to fossil fuel emission decrease. The importance of poleward transport on the atmospheric budgets of HCN and C_2H_2 is highlighted. Columns





and variabilities of CH₃OH, and HCOOH at PEARL are comparable to previous measurements performed at other remote sites. However, the small columns of H₂CO in early May might reflect its large atmospheric variability, and/or the effect of the updated spectroscopic parameters used in our retrievals. Overall, emissions from biomass burning contribute to the day-to-day variabilities of the seven tropospheric species observed at Eureka.

1 Introduction

Future climate change may cause significant air quality reduction by changing the outflow of pollutants as well as the strength of emissions from the biosphere, fires and dust. The sign and magnitude of these effects are highly uncertain and will vary regionally (IPCC, 2007). A key challenge for scientists is to evaluate changes in atmospheric composition induced by human activities. This is especially true in sensitive areas, such as the Arctic, which has been warming rapidly over the past century with an accentuated heating in the past decades (Trenberth et al., 2007; Lesins et al., 2010). The

- chemical environment of the Arctic is unique and characterized by low temperatures, extremes in radiation with long periods of light and darkness, and chemical processes involving snow and ice at the surface. The Arctic is a major receptor for mid-latitude pollution (Shaw, 1995; Quinn et al., 2007, 2008) and changes in chemistry and influx of pollution may disrupt this sensitive system (Rinke et al., 2004). Several studies have
- identified pollution transport to the Arctic based on model simulations and meteorological analyses (Eckhardt et al., 2003; Klonecki et al., 2003; Koch and Hansen, 2005; Stohl et al., 2006; Shindell et al., 2008), but our ability to verify these pathways through chemical observations has been limited.

Ground-based instruments can use solar absorption spectroscopy to measure the chemical composition of the atmosphere. These measurements provide a key dataset for the validation of satellite remote-sensing instruments and model data. The atmospheric variabilities of pollutants and natural gases in a remote area can be quantified





using long-term monitoring of tropospheric molecules released by both natural sources and human activities. This contributes to a better understanding of Arctic chemistry, as well as the factors driving current changes in Arctic atmospheric composition and climate. In this study, we investigate the atmospheric concentrations and variabilities of seven tropospheric trace gases using ground-based Fourier Transform InfraRed

(FTIR) spectra, recorded at the Polar Environment Atmospheric Research Laboratory (PEARL, Eureka, Nunavut, Canada, 80°05' N, 86°42' W) from 2007 to 2011.

These molecules (listed in Table 1) are carbon monoxide (CO), hydrogen cyanide (HCN), ethane (C₂H₆), acetylene (C₂H₂), methanol (CH₃OH), formic acid (HCOOH),
and formaldehyde (H₂CO). They present different source and sink mechanisms in the atmosphere and their different lifetimes play a role in their observed seasonal variabilities (Notholt et al., 1997a). All these atmospheric species absorb incoming sunlight in the infrared region and therefore exhibit spectral features in the FTIR measurements used to derive their atmospheric concentrations.

- ¹⁵ The ground-based FTIR technique can be used to sample the chemical composition for long periods, providing vertically integrated measurements from ground to the top of the atmosphere. This allows the different temporal variabilities of various tropospheric constituents to be assessed. This technique has previously been employed to monitor atmospheric species in various part of the world. CO, HCN, C₂H₆ and C₂H₂ total and
- ²⁰ partial columns have been measured by ground-based FTIR spectroscopy at several locations in the Northern Hemisphere (Mahieu et al., 1997; Rinsland et al., 1998, 2000; Zhao et al., 2002) and in the Southern Hemisphere (Rinsland et al., 1999, 2001, 2002; Paton-Walsh et al., 2010; Vigouroux et al., 2012).

CH₃OH and HCOOH have been measured in a limited number of ground-based FTIR studies. CH₃OH measurements have been performed at Wollongong (Australia, Paton-Walsh et al., 2008) and at Kitt Peak (United States, Rinsland et al., 2009), and HCOOH has been measured mainly in the Northern Hemisphere (Rinsland et al., 2004; Zander et al., 2010; Paulot et al., 2011), and also in the Southern Hemisphere (Paton-Walsh et al., 2005).



The first FTIR datasets of H₂CO have been obtained in remote areas from Ny Ålesund (Norway, 79° N, 12° E), Jungfraujoch (Switzerland, 46° N, 8° E) and McMurdo (Antarctica, 78° S, 167° E) stations (Mahieu et al., 1997; Notholt et al., 1997a, 1997b). Later, FTIR measurements were extended to urban areas in Italy (Hak et al., 2005) and in the Southern Hemisphere at Reunion Island (France, Vigouroux et al., 2009), Darwin (Australia, Paton-Walsh et al., 2010), and Lauder (New Zealand, Jones et al.,

In the Arctic, remote measurements of these tropospheric species have been performed using different instrumental platforms (ground-based, aircraft and satellites).

For instance, the Infrared Atmospheric Sounding Interferometer (IASI) has measured CH₃OH and HCOOH globally from space, but no observations are available above 45° N in winter and 65° N in summer, because of the thermal emission sensitivity of this instrument (Razavi et al., 2011). The Michelson Interferometer for Passive Atmospheric Sounding (MIPAS) limb emission instrument mapped upper-tropospheric distributions

2009).

- of several species, such as HCN and C₂H₆, between September 2003 and March 2004 (Glatthor et al., 2009). It also measured HCOOH at 10 km altitude between 70° N and 90° N, from 2003 to 2008 (Grutter et al., 2010). The Atmospheric Chemistry Experiment (ACE) is able to monitor all the seven species derived in this study, with a periodic sampling over the Arctic region. Indeed, it has measured CO, C₂H₆, HCN, C₂H₂,
- ²⁰ CH₃OH and HCOOH in the upper troposphere, over Alaska and Canada, during June– July 2004 (Rinsland et al., 2007), as well as H_2 CO over North America (50–80° N) from March 2004 to November 2006 (Dufour et al., 2009).

Furthermore, several campaigns have investigated the atmospheric composition of the Arctic based on the synergy of measurements using different platforms at different time periods. These campaigns include the Polar Study using Aircraft, Remote

ent time periods. These campaigns include the Polar Study using Aircraft, Remote Sensing, Surface Measurements and Models, of Climate, Chemistry, Aerosols, and Transport (POLARCAT, http://www.atmos-chem-phys.net/special_issue182.html), and the Arctic Research of the Composition of the Troposphere from Aircraft and Satellites (ARCTAS, Jacob et al., 2010). During these campaigns, several aircraft measurements





at all altitudes in the Arctic in a comparison of eleven chemical transport models

- (Shindell et al., 2008). This has been attributed to model differences in emissions, 15 transport, and OH concentrations. Thus, there is a need to better understand the sources and the transport of CO to the Arctic, as an indicator of pollution effects (Fisher et al., 2010). Concerning C₂H₂, large uncertainties remain with regard to the magnitude of its sources and sinks, as well as its spatial distribution and seasonal-
- ity in the atmosphere (Parker et al., 2011). In addition, large uncertainties remain 20 in the CH₃OH atmospheric budget, in terms of its source magnitudes, seasonality, and spatial distribution in the atmosphere (Millet et al., 2008). Concerning HCOOH, several studies have highlighted the underestimation of emissions in the models, by a factor of nine in the marine boundary layer (Baboukas et al., 2000) and an order
- of magnitude for the free troposphere (Von Kuhlmann et al., 2003). Recently, Paulot 25 et al. (2011) has investigated an underestimation in the model by a factor two to five compared to polar FTIR measurements at Thule (Greenland, 76° N, 69° W), confirming the missing local sources in the HCOOH budget simulated in the model. Finally, HCOOH and H₂CO spectroscopic parameters have been refined recently (HITRAN

of the target species, listed in Table 1, have been reported in biomass burning plumes Discussion (Goode et al., 2000; Simpson et al., 2011; Le Breton et al., 2013). In contrast to these measurements, the ground-based FTIR technique can provide total columns, with good sensitivity in the lower troposphere (compared to satellite mea-Paper ⁵ surements), as well as long-term spectral acquisition, in clear-sky conditions, thereby enabling an assessment of the temporal variabilities of the target species in the high

We focused on these species because there remain numerous gaps in the available observational datasets, especially at high latitudes. Furthermore, the transport and the degradation mechanisms of non-methane hydrocarbons (NMHCs) are poorly under-10 stood and should be better quantified in order to increase our ability to predict trace gas concentrations and variability in models (Stavrakou et al., 2009).

For instance, simulated CO concentrations disagreed by a factor of two to three

Arctic (compared to campaign-based measurements).





Discussion

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2008, http://www.cfa.harvard.edu/hitran/updates.html, Rothman et al., 2009). These improvements increase the HCOOH infrared band intensity by a factor of two (Perrin and Vander Auwera, 2007) and increase the H_2CO line intensities by about 30 % in the spectral region of 3.6 microns (Perrin et al., 2009), which is used in this study. It is thus important to generate new datasets with improved spectroscopic parameters and

thus important to generate new datasets with improved spectroscopic parameters and optimized retrievals, to help improve atmospheric model simulations, and expand our knowledge about the chemical and dynamical processes of the high Arctic.

This paper first describes the methodology employed to obtain our dataset. The FTIR measurements and retrievals are presented for each of the seven tropospheric

- ¹⁰ species, CO, HCN, C₂H₆, C₂H₂, CH₃OH, HCOOH, and H₂CO, observed at Eureka from 2007 to 2011. Details about the selected microwindows, a priori information, information content of the retrievals, and error analyses are described in Sect. 2. This section also describes the procedure employed to compare our dataset with ACE-FTS satellite measurements. In Sect. 3, the timeseries are discussed in terms of their sea-
- ¹⁵ sonal variabilities in connection with the different origins and atmospheric lifetimes of the molecules. Results of the comparisons between our dataset and ACE-FTS and previous measurements reported in the literature are presented. This leads to a discussion of the atmospheric budget of the different target species observed in the high Arctic.

20 2 Methodology

2.1 FTIR measurements at PEARL

We present five years of observations of seven tropospheric species, listed in Table 1, from 2007 to 2011. These timeseries are obtained from ground-based FTIR measurements made in the high Arctic at PEARL (80°05' N, 86°42' W, 0.61 km a.s.l., Eureka, Nunput Canada, Eagel et al. 2012). This high resolution color observation spectrum

²⁵ Nunavut, Canada, Fogal et al., 2013). This high-resolution solar absorption spectrometer (Bruker IFS 125HR, operated at a maximum optical path difference = 257 cm) is





part of the international Network for the Detection of Atmospheric Composition Change (NDACC, http://www.ndsc.ncep.noaa.gov/). It was installed at PEARL by the Canadian Network for the Detection of Atmospheric Change (CANDAC) in July 2006 (Batchelor et al., 2009; Lindenmaier, 2012). The spectrometer measures spectra using two detec-

- tors (Indium Antimonide InSb or Mercury Cadmium Telluride MCT –), a potassium bromide (KBr) beamsplitter, and a sequence of seven narrow-band interference filters covering the 600–4300 cm⁻¹ spectral range. No apodization is applied to these measurements. A reference low-pressure hydrogen bromide (HBr) cell spectrum is recorded regularly with the internal globar to characterize the Instrumental Line Shape
 (ILS) and monitor the alignment of the instrument. By using the LINEFIT software anal-
- ¹⁰ (ILS) and monitor the alignment of the instrument. By using the LINEFTI software analysis (Hase et al., 1999), the modulation efficiency and the phase error are retrieved and included in the forward model.

As an NDACC instrument, CO, HCN and C_2H_6 are retrieved following the InfraRed Working Group (IRWG, http://www.acd.ucar.edu/irwg/) recommendations. For the other species, C_2H_2 , CH_3OH , HCOOH, and H_2CO , the full retrieval methodology

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- was developed in this work. The CH_3OH total columns shown here are the first retrieved from ground-based FTIR measurements in the high Arctic. Total columns of CO, HCN, and C_2H_6 were recently used to investigate the composition of an intense biomass burning plume transported from Russia to Eureka in August 2010 (Viatte et al., 2013).
- Total columns of CO, HCN, C₂H₆, C₂H₂, CH₃OH, HCOOH, and H₂CO result from the analysis of 3980, 1815, 1819, 1269, 1095, 1973, 1242 spectra, respectively, recorded between 2007 and 2011. Because solar absorption FTIR measurements require the sun as the source, there are no data from mid-October to mid-February (polar night). Thus, the timeseries are from February to October for CO, HCN, C₂H₂ and HCOOH,
- from March to October for C₂H₆ and H₂CO, and from March to September for CH₃OH. The differences in the number of observations and the duration of the timeseries are the results of the quality filter used to reject inconsistent retrievals. They are based on negative Volume Mixing Ratio (VMR) values (due to small oscillations in the retrieval), Root Mean Square (RMS) residuals exceeding a threshold value (of two average standard





2.2 Retrieval methodology: optimal estimation method

iterations in the inversion procedure.

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deviations from the median of all RMS residuals), and failure to converge after fifteen

Atmospheric trace gas concentrations are retrieved using inversion procedures performed on atmospheric spectra, within suitable spectral regions, called microwindows. We use the SFIT2 algorithm (Pougatchev et al., 1995; Rinsland et al., 1998) version 3.94, which is based on a semi-empirical implementation of the Optimal Estimation Method (OEM) of Rodgers (2000).

The inversion procedure is an ill-posed problem and requires the use of constraints to stabilize the solution, usually provided by the a priori information. This a priori knowledge is defined by the states of the atmosphere in terms of VMR vertical profiles and variabilities, for each molecule involved in the analysis (target and interfering species). The retrieval and the forward model, which involves a radiative transfer calculation, also require appropriate meteorological parameters, as well as a discretized vertical model

 of the atmosphere. Daily pressure and temperature profiles are from the National Prediction for Environment Prediction (NCEP, http://www.ncep.noaa.gov/) and combined with daily averages of morning (11:00 UTC) and afternoon (23:00 UTC) radiosonde measurements performed at Environment Canada's Eureka weather station. Our retrieved profiles are obtained on a 48-level altitude grid (from 0.61 to 120 km) and total
 columns are then derived by vertically integrating these profiles.

The vertical information content of the retrieved profiles is quantified by the Degrees Of Freedom for Signal (DOFS), which corresponds to the trace of the averaging kernel matrix A (Rodgers, 2000), defined as:

$$\mathbf{A} = \frac{\partial \hat{\mathbf{x}}}{\partial \mathbf{x}} = \left(\mathbf{K}^{\mathsf{T}} \mathbf{S}_{\varepsilon}^{-1} \mathbf{K} + \mathbf{S}_{a}^{-1}\right)^{-1} \mathbf{K}^{\mathsf{T}} \mathbf{S}_{\varepsilon}^{-1} \mathbf{K}$$
(1)

where \hat{x} and x represent the estimate and the true state vectors, respectively. **K** is the weighting function matrix that relates the measurement state to the true state of the



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atmosphere, S_{ε} is the measurement covariance matrix and S_{a} is the a priori covariance matrix used in the OEM.

The spectroscopic parameters are from the HITRAN 2008 database (Rothman et al., 2009) for all the species derived in this study. As noted in Sect. 1, it contains significant improvements concerning HCOOH and H₂CO line intensities in the spectral regions of 5 interest.

The retrieval parameters (microwindows, interfering species, a priori VMR sources, diagonal values of the a priori covariance matrix, and Signal-to-Noise Ratios (SNR)) used for all the target species are summarized in Table 2 and discussed below.

Microwindows 2.3 10

The choice of microwindows (MWs) is crucial because the majority of the target species have relatively weak absorptions in the infrared region compared to the main interfering species, such as CH_4 , H_2O , and O_2 . All MWs have been selected in order to increase the information content and minimize the errors in the retrievals. These MWs are shown

in the Fig. 1, with examples of spectra recorded in clear and polluted conditions in 15 blue and red, respectively. The MWs of CO and C_2H_6 were selected based on the recommendations of the NDACC/IRWG in their harmonization effort strategy.

The CO total columns are retrieved in three widely used MWs, including a weak line of CO at 2057.858 cm⁻¹, another weak line of CO at 2069.656 cm⁻¹ and a strong line

- of CO at 2158.300 cm⁻¹ (Notholt et al., 2000; Zhao et al., 2002). The use of a combi-20 nation between weak and strong absorption lines increases the vertical sensitivity of the retrievals (Barret et al., 2003). Interfering species (CO₂, O₃, OCS, N₂O and H₂O) are simultaneously scaled from their a priori profiles during the inversions. In order to obtain a good estimate of the measurement noise covariance matrix, we constructed
- trade-off curves of the RMS residual (i.e. spectral fit residuals of the retrieval) vs. SNR 25 (Batchelor et al., 2009). The resulting SNR of 85 minimized the RMS residuals and maximized the DOFS of the CO retrievals.



For HCN, two MWs, around 3287.248 cm⁻¹ (Mahieu et al., 1997; Notholt et al., 2000; Zhao et al., 2002), and around 3268.200 cm⁻¹, are preferred over the three IRWG recommendations (which are 3268.05–3268.40 cm⁻¹, 3287.10–3287.35 cm⁻¹, and 3299.40–3299.60 cm⁻¹). This choice increases the information content and decreases the errors in the Eureka retrievals. The profiles of the only interfering species, H₂O, are scaled from the a priori profile and a SNR of 200 is used in the retrievals.

The C_2H_6 retrievals are performed in three MWs around 2976.800 cm⁻¹ (Mahieu et al., 1997; Rinsland et al., 2002; Paton-Walsh et al., 2010), 2983.300 cm⁻¹ (Meier et al., 2004) and 2986.700 cm⁻¹ (Notholt et al., 1997a) using a SNR of 250. Single scaling parameters are used for each of the interfering gases (H₂O and O₃).

For C_2H_2 , we used three MWs, around 3250.500 cm^{-1} (Petersen et al., 2008), 3255.200 cm^{-1} (Mahieu et al., 2008), and 3305.400 cm^{-1} (Mahieu et al., 2008; Paton-Walsh et al., 2010). H_2O and its main isotopologue (HDO) are scaled simultaneously from their a priori profiles. Because the infrared absorptions of the target gas are weak compared to the H_2O lines, a reduced SNR is employed in the spectral region with no C_2H_2 features. While a SNR of 200 is used overall the MWs, we apply a SNR of 50 between 3250.430 and 3250.550 cm⁻¹ and between 3255.180 and 3255.455 cm⁻¹ and a SNR of 75 from 3305.065 to 3305.350 cm⁻¹.

For CH₃OH, we use the two MWs employed by Bader et al. (2013). In this region,
the CH₃OH band at 1033 cm⁻¹ represents around 1.5% of the absorption, whereas the O₃ lines represent about 94%. Thus, O₃ and its isotopologues (¹⁶O¹⁸O¹⁶O labeled O₃⁶⁸⁶, ¹⁶O¹⁶O¹⁸O labeled O₃⁶⁶⁸, ¹⁶O¹⁷O¹⁶O labeled O₃⁶⁷⁶, and ¹⁶O¹⁶O¹⁷O labeled O₃⁶⁶⁷, Table 2), as well as the other interfering species (CO₂ and H₂O) are retrieved simultaneously by scaling their a priori profiles. We use a SNR of 200. For more clarity in Fig. 1, we have shown an example of spectral fitting with the contributions of all the species in the second MW. The significant broad absorption feature of CH₃OH can be seen from around 1032 to 1035 cm⁻¹ (Fig. 1, right panel, cyan line).





For HCOOH, we use one MW between 1104.650 and 1105.600 cm⁻¹, which correspond to the band used in Zander et al. (2010), Paulot et al. (2011) and Vigouroux et al. (2012). The spectroscopic parameters of this band have been updated (Perrin and Vander Auwera, 2007). Zander et al. (2010) have shown that these improvements reduce the retrieved total columns by a factor of two. In this MW, a global SNR of 800 is used and interfering species (HDO, O₃, O₃⁶⁶⁸, O₃⁶⁷⁶, H₂O, NH₃, CCl₂F₂, CHF₂Cl, CH₄) are scaled from their a priori profiles, simultaneously.

For H₂CO, MWs are from Paton-Walsh et al. (2005) where the lines belong to two bands, centered at 2782.000 cm⁻¹ and 2843.000 cm⁻¹, respectively. The interfering species, such as CH₄, O₃, CO₂ and N₂O, are simultaneously scaled from their a priori profiles. The SNR assumed in the MWs is 500, and is reduced to 100 in the spectral regions where systematic residual features were caused by various inaccuracies in the CH₄ line parameters (Sussmann et al., 2011).

2.4 A priori information

The location of the instrument at high latitudes offers a relatively dry atmosphere (total precipitable water ranges between 0 and 1.8 gcm⁻², Wagner et al., 2006, their Fig. 3). Compared to tropical FTIR sites, such as Reunion Island (Vigouroux et al., 2012) or Darwin in Australia (Paton-Walsh et al., 2010), our spectra are less affected by strong features due to water vapor abundance. Thus a pre-fitting of water vapour (or "two-step retrieval") is not necessary here.

Instead, the difficulty resides in the lack of continuous tropospheric measurements in the Arctic, needed to build our a priori knowledge of the polar atmospheric states. Our a priori VMR profiles are from the Whole Atmosphere Community Climate Model (WACCM, http://www2.cesm.ucar.edu/working-groups) as recommended by the

²⁵ NDACC/IRWG, except for the C₂H₂ retrievals (see above). WACCM is a numerical model developed at the National Center for Atmospheric Research (Sassi et al., 2002),





which was used to compute 40 yr average of the modelled profiles at Eureka, which served as a priori VMRs for the target species.

In Figs. 2 to 8 (upper panels), the a priori profiles adopted for the FTIR retrievals are shown in black, with the mean of all retrieved profiles in red. The black and red error bars correspond to the standard deviation of the a priori covariance matrix used in the retrievals and the standard deviations around the mean retrieved profiles, respectively, at each layer. All the a priori VMR profiles exhibit large contributions in the boundary layer and the troposphere, and decrease to almost zero at 50 km, except for CO, for which the sources are dominated by CH₄ oxidation at this altitude (Clerbaux et al., 2008).

The CO a priori VMR is about 92 ppbv (1 ppbv = 10^{-9} per unit volume) through the boundary layer, decreases to about 20 ppbv at the tropopause region and increases again with altitude to 0.6 ppmv (1 ppmv = 10^{-6} per unit volume) at 50 km. The HCN a priori VMR slightly increases from 25 ppbv to 32 ppbv in the troposphere and decreases in the stratosphere to reach 11 ppbv in the stratosphere. The C_2H_6 , C_2H_2 and 15 CH₃OH a priori VMR profiles have similar shapes as a function of altitude. Note that the C_2H_6 and C_2H_2 profiles are plotted on a log scale whereas the CH₃OH profile is plotted on a linear scale for clarity (Figs. 2 to 8, upper panel). Their VMRs are almost constant in the boundary layer and the troposphere, with values of about 1 ppbv for C_2H_6 , and 0.5 ppbv for both C_2H_2 and CH_3OH . At the upper altitudes, the a priori profiles of C_2H_6 , 20 C_2H_2 and CH_3OH decrease until reaching ~ 1 pptv (1 pptv = 10^{-12} per unit volume) at 22 km, 17 km and 21 km, respectively. For HCOOH and H₂CO, the a priori VMRs in the boundary layer are about 8 pptv and 30 pptv, and then decrease to 1 pptv and 10 pptv around 30 km for HCOOH and H₂CO, respectively. However, the sinks and sources are not as well understood (Paulot et al., 2011; Jones et al., 2009), especially in the polar 25

region, so the interpretation of the a priori profiles has to be done carefully. The a priori VMR profiles are from WACCM version 6 for CO, HCN, CH₃OH, HCOOH,

and divided by two at all altitudes for H_2CO (see explanation below). The low stratospheric values of C_2H_6 VMRs in WACCM version 6 were beyond the retrieval software





precision, therefore the a priori VMR profiles of C_2H_6 are from WACCM version 4.5. For C_2H_2 , the a priori profile assumed here is not derived from WACCM because the VMR values were extremely low in the upper atmosphere and not representative, causing the retrievals to rarely converge. To avoid these instabilities in the retrievals, we used a pro-

- file derived from MkIV balloon measurements made at the high-latitude NDACC site of Kiruna (Sweden, Toon et al., 1999), between 6 and 34 km altitude, combined with spitprim.set (Ft. Sumner MkIV flights, 1990s, http://mark4sun.jpl.nasa.gov/m4.html). This a priori profile has been divided by two so that the tropospheric VMR values are comparable to the ones derived from the WACCM runs.
- ¹⁰ Instabilities in the retrieval also appear if the measured columns are substantially smaller than the a priori columns. As a result, the retrieved profile returns non-physical negative mixing ratios at the altitudes where the information content is low. This means that the most practical approach is often to use a priori profiles with a significantly lower total column than the average expected from the dataset, and to compensate
- ¹⁵ by using larger values in the covariance matrix S_a (Paton-Walsh, 2009). Thus, the a priori profile of H₂CO has been divided by two in order to reduce the RMS residual and avoid oscillations in the retrieved profiles. We also verified that the retrieved total columns did not change significantly (within error bars of the measurements) when using both profiles (divided by two and not). This confirms that the majority of the information content is coming from the measurements and not from the a priori states
- information content is coming from the measurements and not from the a priori states. It is important to note that a single a priori profile for each species is used for all spectra recorded from 2007 to 2011. This ensures that our observed atmospheric variabilities mainly come from the information content of the measurements.

In the retrieval process, the measured spectrum is weighted by the empirically de-²⁵ fined a priori covariance matrix \mathbf{S}_a , with diagonal elements reflecting the one-sigma uncertainties used in the OEM. For CO, C_2H_2 , CH_3OH , HCOOH, and H_2CO , the offdiagonal elements of \mathbf{S}_a correspond to an exponential inter-layer correlation with a correlation length of 4 km. For the others species, no interlayer correlation is applied.





The relative uncertainties of the a priori VMR profiles (or the standard deviations used in the covariance matrix) are assumed to be 20% for CO and CH₃OH, 50% for HCN and C_2H_2 , 30 % for C_2H_6 , and 100 % for HCOOH and H_2CO , at all altitude layers. These values have been scaled to account for the different thicknesses of the 47 layers 5 in our retrieval scheme. These values, summarized in Table 2, are representative of the tropospheric variabilities derived from the WACCM model, except for CH₃OH and H₂CO. Those are, on average over the troposphere, 17% for CO, 28% for HCN, 30% for C₂H₆, and 66 % for C₂H₂. For CH₃OH, the WACCM model suggests a high tropospheric variability of about 71 %. Since this species has broad absorption features, we prefer to use a lower value of SNR in the retrievals, and then decrease the one-sigma 10 uncertainties to 20% in the OEM, to avoid oscillations in the retrieved profiles. In contrast, the high SNR employed in the HCOOH and H_2CO retrievals, commensurate with the SNR of most analyzed spectra, has been compensated for by using large values in the diagonal elements of the a priori covariance matrix. For those two species, the average tropospheric variabilities given by the model are 66% for HCOOH and 39% 15 for H₂CO. These variabilities are rather small and not consistent with the idea of an ad-

ditional large source of HCOOH from snow photochemistry (Dibb and Arsenault, 2002) and the large H_2CO variability of 30 pptv to 700 pptv observed at the Arctic surface (De Serves, 1994).

Finally, the total columns of all the target species are representative of the tropospheric columns. The numbers in blue in Figs. 2 to 8 (upper panels) are the percentages of the tropospheric column contributions (between 0.6 and 10.7 km) relative to the total column (between 0.6 and 120 km). As can be seen, the tropospheric columns contribute to more than 90 % of the total columns.

25 2.5 Averaging kernels

The rows of **A** (Eq. 1) correspond to the averaging kernels for a certain altitude layer and characterize the information content of the retrievals. An ideal observation has an averaging kernel of one in the region of interest and zero outside (Connor et al., 1996).





Figures 2 to 8 (lower panels) present the typical VMR averaging kernels, in VMR/VMR (left panels), of the seven retrieved species. The different colours correspond to averaging kernels at altitudes lying in a partial column for which the DOFS is 0.5.

- For CO, HCN and C_2H_6 , the averaging kernels are more vertically resolved than for the other species, i.e., quasi-independent tropospheric partial columns can be considered between 0.6 and 4 km for CO, between 0.6 and 14 km for HCN, and between 0.6 and 5 km for C_2H_6 . Their mean DOFS over 3980, 1815 and 1819 retrievals are 2.5 (±0.3), 1.6 (±0.6), and 2.0 (±0.2), respectively. The numbers in parenthesis correspond to the one-sigma standard deviations around the means. For C_2H_2 , CH_3OH , HCOOH and H_2CO , the mean DOFS over 1269, 1095, 1973 and 1242 retrievals, are 1.1 (±0.2),
- 1.2 (\pm 0.1), 1.1 (\pm 0.4), and 0.8 (\pm 0.2), respectively. Total column averaging kernels, in (molecule cm⁻²)/(molecule cm⁻²), are shown in black lines on the right panels of Figs. 2 to 8 (right lower panels). We have also added the sensitivity profiles of the retrievals to the measurements (Figs. 2 to 8, right lower panels, red lines). The sensitivity
- at a certain altitude is the sum of the elements of the corresponding averaging kernels. It indicates the fraction of the retrievals that comes from the measurements rather than from the a priori information. A sensitivity value of one means that the information content of the retrievals is 100 % from the measurement at this altitude.

All total column averaging kernels are around one between 0.6 and 10 km, which ²⁰ confirms the good sensitivity of our retrievals in the troposphere. Furthermore, the sensitivity profiles reach one in the troposphere, confirming that the majority of the information content comes from the measurement in this altitude region. For C_2H_6 and C_2H_2 , the total column averaging kernels and sensitivities reach zero at 36 km and 22 km, respectively. As a consequence, the retrieved profiles revert to the a priori values above these altitudes (Figs. 2 to 8, upper panels).

2.6 Error budgets

Full error analyses have been performed following the formalism of Rodgers (2000). The error budget is calculated by separating the measurement noise error, the





smoothing error (expressing the limited vertical resolution of the retrieval), and the forward model parameter error (including error on the temperature profiles, on spectroscopic and retrieval parameters, interfering species uncertainties, and error on solar zenith angles). The error analyses have been performed on a representative dataset
 of around 10 spectra per species, selected with different values of Solar Zenith Angle (SZA), SNR, hour and season of measurements. The averages of all the calculated errors are shown for each target species in Table 3.

The measurement error covariance matrix \mathbf{S}_{m} is calculated as:

 $\mathbf{S}_{m} = \mathbf{G}_{y}\mathbf{S}_{\varepsilon}\mathbf{G}_{y}^{\mathsf{T}}$

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where \mathbf{G}_{y} is the gain matrix representing the sensitivity of the retrieved parameter to the measurement, and \mathbf{S}_{ε} is the measurement covariance matrix as seen in Eq. (1). The diagonal elements of \mathbf{S}_{ε} are the squares of the spectral noise, which is determined by the RMS residuals derived from the spectra.

The smoothing error covariance matrix \mathbf{S}_{s} is calculated as:

 $\mathbf{S}_{s} = (\mathbf{I} - \mathbf{A}) \mathbf{S}_{e} (\mathbf{I} - \mathbf{A})^{\mathsf{T}}$

where **I** is the identity matrix, **A** is the averaging kernel matrix, and S_e is the climatology matrix representing the natural variability of the target species. The lack of continuous ²⁰ measurements of tropospheric species in the Arctic makes it difficult to build this matrix. Thus, the values derived from WACCM have been used as diagonal elements, to express the natural variabilities of each trace gas. For CO, C_2H_2 , CH_3OH , HCOOH, and H_2CO , the off-diagonal elements of S_e correspond to an exponential inter-layer correlation with a correlation length of 4 km. For the others species, no interlayer correlation ²⁵ is applied, consistent with the retrievals.

Finally the forward parameter error covariance matrix \mathbf{S}_{f} is calculated as:

 $\mathbf{S}_{\mathsf{f}} = \left(\mathbf{G}_{y}\mathbf{K}_{b}\right)\mathbf{S}_{b}\left(\mathbf{G}_{y}\mathbf{K}_{b}\right)^{\mathsf{T}}$

(2)

(3)

(4)

where \mathbf{G}_{y} is the gain matrix, and \mathbf{K}_{b} is the Jacobian matrix obtained by perturbation methods. \mathbf{K}_{b} represents the sensitivity of the measurements to the model parameter *b*. \mathbf{S}_{b} is the covariance error matrix on the parameter *b*. These parameters *b* can have systematic (spectroscopic parameters) and random (uncertainties on temperature and

SZA, for instance) components. To generate the Jacobians, parameters are perturbed by 0.1° for the error on SZA, by 2 K for the temperature profiles, and by a factor of 1.05 for the line intensity and air-broadened parameters (Batchelor et al., 2009).

In addition to these errors, we have examined the errors on the other retrieved parameters, called interference errors as described in Rodgers and Connor (2003) and

- explained in detail in Sussmann and Borsdorff (2007) and in Batchelor et al. (2009, Sect. 4). The interference errors combine the errors on the interfering species in the spectral region of interest, and the uncertainty due to the retrieval parameters, such as the wavelength shift, the ILS, the background slope and curvature, the zero-level shift, and the phase error.
- ¹⁵ The total random errors are calculated by adding all the uncertainties in quadrature, except the uncertainties on line intensity and line width. The predominant contribution to the total random errors is the measurement error for HCN and HCOOH. For CO, the uncertainty on temperature is slightly higher than the uncertainty on the measurement error, but still remains small. For C_2H_6 and C_2H_2 , the measurement error is almost as
- ²⁰ large as the uncertainties on retrieval parameters. This might be an effect of the H_2O continuum, which affects the background slope and curvature of retrievals. For CH_3OH and H_2CO , the temperature uncertainty is high, which is consistent with what has been reported in Vigouroux et al. (2012) for CH_3OH . Concerning H_2CO , the largest random uncertainties come from the uncertainty on the interfering species. This might be ex-
- ²⁵ plained by the fact that H_2CO absorption lines are close to CH_4 lines (in both MWs), which are difficult to fit because of spectroscopic parameter uncertainties (Sussmann et al., 2011). This can also be an effect of the deweighting function used to reduce the SNR of the interfering species in the H_2CO microwindows.





The dominant contribution to the systematic error is the uncertainty on the line intensity for all species except C_2H_2 , for which the line width uncertainty is larger. For H_2CO , the uncertainties on the line width and line intensity are of the same order of magnitude.

⁵ As expected, the smoothing error of CO is smaller than for the other species, given the greater DOFS and the tighter averaging kernels.

Unlike the uncertainties on the measurement and the spectroscopic parameters, which are considered as truly random and systematic, respectively, some parameters have both components, making the division between those uncertainties difficult.

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Therefore the total errors shown in Table 3 have been determined by adding all these errors in quadrature.

Overall, the total errors of the CO, HCN, C_2H_2 , C_2H_6 , CH_3OH , HCOOH, and H_2CO total columns are 3.1, 10.5, 14.3, 22.5, 12.3, 17.0, and 27.5 %, respectively.

2.7 ACE-FTS measurements

- ¹⁵ The Atmospheric Chemistry Experiment (ACE) (Bernath et al., 2005) was launched in August 2003 to investigate atmospheric composition mainly in the upper troposphere and the stratosphere. ACE is equipped with a high-resolution (0.02 cm⁻¹) Fourier Transform Spectrometer (FTS), operating in the 750 to 4400 cm⁻¹ spectral range. It performs solar occultation measurements with limb geometry, with a vertical resolution of
- ²⁰ about 3–4 km (Boone et al., 2005). VMR profiles of the atmospheric species are retrieved with the version 3.0 algorithm within different altitude ranges for each molecule (http://www.ace.uwaterloo.ca/molecules.html; Boone et al., 2013). We chose ACE-FTS measurements for comparison with our new dataset because ACE has good sampling coverage at high latitudes (Bernath, 2006) and measures all seven species of interest in this study.

As shown previously, ACE-FTS measures VMR profiles of CO (Clerbaux et al., 2008), hydrocarbons, such as HCN, C_2H_6 , and C_2H_2 (Park et al., 2013) and HCOOH (Gonzales Abad et al., 2009) in the upper troposphere and lower stratosphere. In



addition to these species, CH_3OH and H_2CO have been measured by ACE-FTS globally (Dufour et al., 2007; Dufour et al., 2009) and in biomass burning plumes over Northern high latitudes (Rinsland et al., 2007; Tereszchuk et al., 2011, 2013) and in the Southern Hemisphere (Coheur et al., 2007).

- For comparisons with the FTIR dataset, all ACE-FTS occultations within 500 km of PEARL and within 24 h of each FTIR measurement are selected. ACE-FTS data recorded after October 2010 are not used because of an algorithm problem in the version 3.0, which will be fixed in the future version 3.5 (Boone et al., 2013). For the calculations, each ACE-FTS profile was interpolated to the FTIR altitude grid and the
 ¹⁰ FTIR a priori profiles were used to fill the ACE-FTS profiles from the ground to the
 - lowest available ACE-FTS altitudes.

The FTIR and ACE-FTS trace gas profiles are retrieved over different altitude ranges, with different vertical resolutions and sensitivities. To properly account for the different vertical sensitivities of the correlative observations, the ACE-FTS profiles are

smoothed by convolution with the FTIR averaging kernel functions, following this equation (Rodgers and Connors, 2003):

 $\boldsymbol{x}_{\rm s} = \boldsymbol{\mathsf{A}} \left(\boldsymbol{x} - \boldsymbol{x}_{\rm a} \right) + \boldsymbol{x}_{\rm a}$

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where x_s is the smoothed ACE-FTS profile, **A** is the FTIR averaging kernel matrix and x_a is the FTIR a priori profile.

Each partial column is obtained by integration of the trace gas VMR from the lowest to the highest available ACE-FTS level for both the FTIR and ACE-FTS data. Density profiles are obtained by taking the temperature and pressure profiles derived from ACE-FTS measurements.

25 3 Results and discussion

Seasonal variations of atmospheric concentrations at high latitudes are the largest on Earth. Mixing of mid-latitude air into the Arctic occurs typically between mid to late



(5)



March after the final stratospheric warming (Schoeberl et al., 1992). Since we use a single a priori profile for all the retrievals for a gas, we ensure that the observed variabilities come from the information content of the measurement. The timeseries of the daily mean total columns of CO, HCN, C₂H₆, C₂H₂, CH₃OH, HCOOH, and H₂CO are plotted in Fig. 9, using different colours for the different years of measurements from 2007 to 2011. The brown lines represent the polynomial fits to the data. The timeseries are obtained during the polar day periods since the FTIR measurements require the sun as the source.

3.1 Seasonal variabilities

The timeseries of the CO, C₂H₆ and C₂H₂ total columns show strong seasonal cycles with maxima in winter and minima in summer. These compounds are produced mainly at mid-latitudes. CO is a product of incomplete combustion and atmospheric oxidation of Volatile Organic Compounds (VOCs) and CH₄. The sources of C₂H₆ are natural gas and fossil fuel emissions (Singh and Zimmerman, 1992). C₂H₂ main sources include natural gas, biofuel combustion products and biomass burning emissions (Gupta et al., 1998; Logan et al., 1981; Rudolph, 1995; Xiao et al., 2007; Zhao et al., 2002). CO, C₂H₆ and C₂H₂ are removed by oxidation via OH reaction (Logan et al., 1981), leading to atmospheric lifetimes of approximately fifty-two days (Daniel and Solomon, 1998), eighty days (Xiao et al., 2008), and two weeks in the atmosphere (Xiao et al., 2007), respectively.

The total columns of these three gases are greater in winter because their common sources, which are fossil fuel emissions, are usually enhanced in the dry and cold period (Zhao et al., 2002). In addition, these gases accumulate during this period because their main removal pathway is the reaction with OH, which disappears in the dark.

²⁵ The monthly mean total columns for each species are given in Table 5. The percentages of the seasonal variability (or amplitude) are obtained by taking the difference between the maximum and minimum monthly means, divided by the annual average, for each gas (as described below).





The CO total columns at Eureka have a maximum in March of 2.20 × 10¹⁸ molecules cm⁻² (Table 4) and a minimum in September of 1.56 × 10¹⁸ molecules cm⁻². The seasonal amplitude of CO total columns is thus estimated to be 34 %. The one-sigma standard deviation around the mean (number in parentheses in Table 4) is slightly higher for the summer months (July and August), reflecting the higher atmospheric variability of CO due the contribution of boreal forest fire plumes transported from lower latitudes during this period.

The C_2H_6 total columns have a maximum in March of 2.85×10^{16} molecules cm⁻² and a minimum in August of 1.20×10^{16} molecules cm⁻², with a seasonal amplitude of 93%. The standard deviations around the C_2H_6 monthly means are larger for April and July (Table 4), certainly due to transport of emissions from extreme fire events during these periods. This is also confirmed by the rather high standard deviation of the well-known biomass burning tracer HCN, at this time (as described below).

For C₂H₂, the total columns present a maximum in March of 5.50×10^{15} molecules cm⁻² and a minimum in September–October of 1.61×10^{15} molecules cm⁻². Its seasonal cycle of 104 % is larger than for C₂H₆, in agreement with the fact C₂H₂ is more quickly removed by OH reaction (Logan et al., 1981; Singh and Zimmerman, 1992). As seen for C₂H₆, the standard deviations for C₂H₂ are larger for April and July, again due to greater transport of biomass burning plumes.

Since the lifetimes of these three molecules are rather long (from two weeks to two months, Table 1), the day-to-day variability is attributed to the long-range transport through the Arctic. For instance, the enhanced concentrations in August 2010, represented by the green dots (Fig. 9), have been attributed to an extreme fire event occurring nine days earlier in Russia and transported to Eureka (Viatte et al., 2013). To conclude, we clearly see the combined effect of the chemistry via OH reaction (sea-

²⁵ conclude, we clearly see the combined effect of the chemistry via OH reaction (seasonal variability) and transport (day-to-day variability) in the annual cycles of these three molecules in the high Arctic region.

In contrast to CO, C_2H_6 and C_2H_2 , the timeseries of the HCN, CH_3OH , HCOOH and H_2CO total columns show seasonal cycles with maxima in summer and minima





in winter. HCN is a relatively inactive species and is a good tracer of biomass burning emission (Rinsland et al., 2001) with a lifetime of about five months in the troposphere (Li et al., 2003). The reaction with OH and $O(^{1}D)$ (Cicerone and Zellner, 1983) and ocean uptake (Li et al., 2003) have been shown to be the principal pathways for HCN loss, but its sinks are not well quantified yet (Zeng et al., 2012).

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The HCN total columns measured at PEARL have a broad maximum in summer with a sharp increase in August $(6.63 \times 10^{15} \text{ molecules cm}^{-2})$ due to the plumes from extreme fire events transported to Eureka in summer, and especially in August 2010, when the HCN total columns are enhanced by more than 50% compared to the four-

- ¹⁰ year average, including all years except 2010 (Viatte et al., 2013). This summer maximum is a result of the combination of emissions by plants, bacteria and fungi due to enhanced vegetative activity in springtime (Cicerone and Zellner, 1983) and biomass burning emissions from boreal forest in summer (Rinsland et al., 2000). The HCN concentrations present a minimum in February of 2.78×10^{15} molecules cm⁻² (Table 4),
- ¹⁵ with a seasonal amplitude of 78%. In addition, the higher standard deviation for July, August and September, is consistent with severe fire emissions transported from midlatitudes and the fact that HCN stays longer in the atmosphere, given its long atmospheric lifetime (Table 1).

CH₃OH is the second most abundant volatile organic compound in the atmosphere
after CH₄ (Jacob et al., 2005), representing 20% of the total global VOC emissions (Guenther et al., 2006). Sources include plant growths, ocean, decomposition of plant matter, as well as biomass burning emission (Rinsland et al., 2009; Andreae and Merlet, 2001). The principal sink of CH₃OH is chemical loss due to OH reaction (Heikes et al., 2002) leading to the formation of CO and H₂CO (Millet et al., 2006; Rinsland et al., 2009; Stavrakou et al., 2011). The lifetime of CH₃OH in the surface boundary layer is three to six days (Heikes et al., 2002) and between five and ten days at a global scale (Jacob et al., 2005; Stavrakou et al., 2011).

Our CH₃OH total columns have a seasonal amplitude of 94 % with maxima in July of 1.82×10^{16} molecules cm⁻² and minima in March of 0.42×10^{16} molecules cm⁻². This





is consistent with ACE-FTS measurements over North America and the role of vegetation growth in this region in driving its seasonal cycle (Dufour et al., 2006). However, biomass burning is a significant source of CH_3OH during summer and the higher standard deviations in June and August confirm the key role of enhanced plant growth and biomass burning at this time.

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HCOOH is the second most abundant global organic carboxylic acid in the atmosphere (Zander et al., 2010 and references therein). Direct sources of HCOOH include human activities, biomass burning and plant leaves, and the largest source is the photooxidation of NMHCs. HCOOH is removed through oxidation by OH, as well as dry and wet depositions (Stavrakou et al., 2012). However, a recent study suggested a missing source in the HCOOH budget of the Northern latitudes and inferred its lifetime of about three to four days (Paulot et al., 2011).

The HCOOH total columns measured at Eureka show a broad maximum around June of 2.20×10^{15} molecules cm⁻², with two maxima in April and August caused by the activation of biogenia amiasiana and the presence of intense fire events in these

- the activation of biogenic emissions and the presence of intense fire events in those months, respectively. It presents a minimum in February of 1.05 × 10¹⁵ molecules cm⁻². This is consistent with the study of Zander et al. (2010) in which the seasonal modulation is in phase with biogenic emissions, which depend on solar isolation, local temperature, atmospheric transport and soil coverage. However, the HCOOH seasonal
- amplitude of 61 % is smaller at Eureka than the other species, certainly due to the smaller lifetime of this molecule (three to four days, Table 1), as well as the importance of wet deposition as a sink. Indeed, the smaller standard deviations in February and March might be attributed to the lower winter atmospheric water content, decreasing the effect of wet deposition, and resulting in fewer fluctuations in the HCOOH time-
- ²⁵ series. During winter, total columns are affected by the short-range transport with no biogenic emission sources.

Finally, H_2CO is produced by the oxidation of CH_4 and NMHCs, which are emitted into the atmosphere by plants, animals, industrial processes and incomplete combustion of biomass and fossil fuel. Isoprene has been suggested to as an additional





significant source of H₂CO (Jones et al., 2009 and references therein). In addition, secondary H₂CO formation in biomass burning plumes has been proposed (Paton-Walsh et al., 2010). It is primarily removed via photo-dissociation and reaction with OH radical with a half-life of approximately three hours in daylight (Warneck, 2000), or a lifetime of less than two days (Coheur et al., 2007).

Our H₂CO total columns show a seasonal amplitude of 93% with summer maxima (in July of 3.14×10^{15} molecules cm⁻²) and late winter minima (in April of 0.27×10^{15} molecules cm⁻²). This is consistent with photochemical control by OH and its formation by CH₄ oxidation in the winter seasons, as well as isoprene emissions from plants and forest during the growing season. Most of the seasonal variability corresponds to the biogenic emission and biomass burning events occurring in the boreal summer months.

The seasonal cycles of CH_3OH , HCOOH, and H_2CO in the high Arctic are driven by biogenic emissions and short-range transport from lower latitudes, whereas the biomass burning emission and long-range transport affect the seasonal variability of HCN in the Arctic.

Overall, emissions from biomass burning seem to play a significant role in the dayto-day variabilities of the seven tropospheric species observed in the high Arctic.

3.2 Comparisons with ACE-FTS

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²⁰ The results of the comparisons of the FTIR measurements with these from ACE-FTS, from 2007 to 2010, for each molecule are given in Table 5.

The CO, HCN, C_2H_2 and HCOOH partial columns are compared with all ACE-FTS observations satisfying the coincidence criteria defined in Sect. 2.7; this gives 106, 93, 93 and 103 pairs, respectively. For these species, the mean distance of the ACE-

FTS occultations to PEARL are between 313 and 319 km. CO, HCN and C_2H_2 partial columns measured by the FTIR are in good agreement with ACE-FTS, with coefficients of correlation (*R*) of 0.66, 0.96, and 0.78 and slopes of 0.97 ± 0.11, 0.69 ± 0.02 and 1.21 ± 0.10, respectively. For HCOOH, the agreement is less clear since the coefficient





of correlation is 0.56 with a slope of 3.35 ± 0.49 . The mean altitude ranges for partial column calculations are 8.0–48.6 km for CO, 7.9–33.2 km for HCN, 8.0–17.0 km for C₂H₂, and 8.1–18.5 km for HCOOH.

The FTIR C_2H_6 partial columns, calculated on average between 8.0 and 19.3 km, agree well with the ACE-FTS data. The coefficient of correlation is 0.97, with a slope of 0.71 ± 0.04 over 17 coincident measurements, within a mean distance to PEARL of 436 km.

For CH₃OH and H₂CO comparisons, the small number of coincidences (N = 6) makes it difficult to draw significant conclusions. The coefficients of correlation between ACE-FTS and FTIR partial columns, calculated on average from 9.4 to 17.5 km for CH₃OH and from 9.1 to 38.7 km, are 0.91 and 0.75, with slopes of 0.74 ± 0.14 and 0.50 ± 0.22, respectively.

It is worth noting that 21 observations were rejected for the H₂CO comparison because they were measured after October 2010. The ACE-FTS algorithm version 3.5 will improve our H₂CO comparisons in the future.

Given the values of the coefficients of correlation and the slopes, the Eureka FTIR measurements of CO, HCN, C_2H_6 and C_2H_2 and HCOOH are well correlated with ACE-FTS data. For CH₃OH and H₂CO, the small number of coincident profiles measurements does not allow us to draw significant conclusions.

20 3.3 Discussion

Comparisons between our dataset and previous measurements reported in the literature lead to a discussion of the atmospheric budget of the different target species observed in the high Arctic.

Our retrieved CO total columns are on five-years average smaller by a factor of 1.3 compared to CO measured at Ny Ålesund (Norway, 79° N, 12° E) from 1992 to 1995 (Notholt et al., 1997b). This is consistent with the decrease of tropospheric CO of -0.61 ± 0.16 % yr⁻¹ observed at high-latitude sites between 1996 and 2006 (Angelbratt et al., 2011). This trend has been explained with the combination of a 20% decrease in





anthropogenic CO emissions in Europe and North America and 20% increase in CO anthropogenic emission in East Asia.

Concerning C_2H_6 , our total columns are smaller by a factor 1.1 compared to measurements performed in the Arctic from 1992 to 1995 (Notholt et al., 1997b), suggesting that the Arctic C_2H_6 budget has been decreasing from this period to 2007–2011. This confirms the significative negative trend observed in the Northern Hemisphere at Kiruna (Sweden, 67° N, 20° E) and Harestua (Norway, 60° N, 10° E) from 1996 to 2006 (Angelbratt et al., 2011). Aydin et al. (2011) attributed this trend to the decrease of C_2H_6 based fossil-fuel use in the Northern Hemisphere with a potential role of increased chlorine atoms which could play a role in the C_2H_6 decline. Recent studies also evaluated a negative trend in the Southern Hemisphere at Lauder (New Zealand, 45° S, 170° E) and Arrival Height (Antarctica, 78° S, 167° E) from 1997 to 2009, as well as at Wollongong (Australia, 34° S, 150° E) (Zeng et al., 2012).

For C_2H_2 , our total column values and seasonal variability are in agreement with values reported at Reunion Island (France, 21° S, 55° E) from 2004 to 2011 (Vigouroux et al., 2012), and at Jungfraujoch station (Switzerland, 46° N, 8° E) from 1995 to 2008 (Mahieu et al., 2008), as well as at Ny Ålesund from 1992 to 1999 (Albrecht et al., 2002). The similarity between high-latitude and mid-latitude C_2H_2 concentrations has already been reported (Albrecht et al., 2002) and highlights the importance of transport

²⁰ on the C_2H_2 budget in the Arctic. However, the primary source of C_2H_2 is from transportation at mid-latitudes (Xiao et al., 2007). So the fact that our C_2H_2 total columns are comparable to those measured in the Arctic between 1992 and 1999 (i.e. no decrease is seen from previous years compared to the observed decline of CO or C_2H_6) is unclear. One hypothesis might be that C_2H_2 emissions from cars have not changed compared to the technological advances in the catalytic converters employed to reduce CO and C_2H_6 emissions.

Our HCN total columns are of the same order of magnitude, in term of absolute values and variabilities, with those reported by Notholt et al. (1997b) in the Arctic from 1992 to 1995. They also agree well with HCN columns observed at mid-latitude





regions, as in Northern Japan at 44° N in 1995 (Zhao et al., 2000), as well as at Jungfraujoch from 2001 to 2009 (Li et al., 2009). In addition, our extreme values, exceeding 10×10^{15} molecules cm⁻² in summer 2010, are comparable to values found in the tropics at Reunion Island during the biomass burning seasons from 2004 to 2011 (Vigouroux et al., 2012). This confirms the key role of the transport from mid-latitudes in the HCN budget at high-latitudes. However, a negative trend has been highlighted in the Southern Hemisphere for the tropospheric HCN columns observed at Lauder (New Zealand, 45° S, 170° E) from 1997 to 2009 (Zeng et al., 2012), which is not seen in our data. This may be due to the fact that during the five years of HCN measurements at Eureka, several extreme biomass burning events have been detected, especially a persistent and intense one in August 2010 (Viatte et al., 2013).

Concerning CH₃OH, our total columns are of the same order of magnitude as those of Vigouroux et al. (2012) and Bader et al. (2013) obtained at Reunion Island and Jungfraujoch station, respectively. In the latter study, extreme enhancements

- ¹⁵ have been shown to reach 2.5×10^{16} molecules cm⁻², in agreement with our high total columns in the June months. Furthermore, the difference between the largest and the smallest individual columns at Jungfraujoch station exceed a factor of 14, which is in excellent agreement with the observed variability at Eureka. Finally, our monthly mean total columns show a factor of four in the seasonal amplitude, comparable to a fac-²⁰ tor three of variation found for free tropospheric CH₃OH columns at Kitt Peak (United
- States, 32° N, 111° W) over 22 yr of observations (Rinsland et al., 2009).

Our HCOOH retrieved columns are in agreement in term of mean and extreme values, as well as seasonal amplitude, compared to those measured at Jungfraujoch (Zander et al., 2010). They reported a mean value of $1.70\pm0.50 \times 10^{15}$ molecules cm⁻² for lung lube August from 1085 to 2007, which is comparable to our every of the second sec

for June-July-August from 1985 to 2007, which is comparable to our average of $2.10\pm0.20 \times 10^{15}$ molecule cm⁻² for the same summer months from 2007 to 2011. Also, our outliers are enhanced by a factor of four relative to the monthly means, which is in agreement with the large variability of HCOOH columns measured at Jungfraujoch. However, our seasonal variability is less clear than at mid-latitudes, certainly due





to the contribution of the short-range transport and the importance of wet deposition as a sink of HCOOH. Finally, HCOOH total columns have been recently obtained with ground-based FTIR measurements in the Arctic at Thule (Greenland, 76° N, 69° W) from 2004 to 2009 (Paulot et al., 2011, their Fig. 3). Excellent agreement, in term of seasonal variability and extreme values, is seen with this dataset, with higher values in 2008 (red circles, our Fig. 9) due to exceptional biomass burning in Asian boreal forest (Giglio et al., 2010).

Finally, our H₂CO total columns, which vary from 1.90×10^{13} to 6.3×10^{15} , molecules/cm² with an annual average of 2.3×10^{15} molecules cm⁻² (between February and October), are lower than those found in the literature, especially in the spring and autumn. For instance, Notholt et al. (1997b) reported Arctic total columns varying from around 2 to 8×10^{15} molecules cm⁻² between 1992 and 1998 at Ny Ålesund. Part of this difference might be attributed to the use of updated line strengths in HITRAN

- 2008. Indeed, the new spectroscopic intensities have been increased by around 30 %
 ¹⁵ in the spectral region employed here (Perrin et al., 2009), leading to a significant decrease of the retrieved total columns. In addition, Jones et al. (2009) showed columns from around 2.5 to 8 × 10¹⁵ molecules cm⁻² between 1992 and 2004 at Lauder (New Zealand, 45° S, 170° E), and Vigouroux et al. (2009) measured columns from 1.6 to 7 × 10¹⁵ molecules cm⁻² at Reunion Island (France, 21° S, 55° E) between 2004 and 2011. Our low H₂CO total columns measured in late-winter might also reflect the large atmospheric variability of H₂CO, given its small lifetime, with no local sources in winter at Eureka. However, our data show a strong seasonal cycle consistent with the FTIR measurements at Ny Ålesund (Albrecht et al., 2002) and in situ measurements at Alert
 - (Canada, 82° N, 63° W) from 30 to 700 pptv (De Serves, 1994).

25 4 Conclusions

The ground-based FTIR technique is a powerful tool for deriving long-term measurements of various atmospheric species having both natural and anthropogenic





sources. These measurements in remote areas, such as the polar regions, contribute to quantifying atmospheric chemical composition and variability, especially in the troposphere where vertically integrated measurements are sparse. Monitoring tropospheric molecules with different atmospheric lifetimes and origins, provides useful information

- ⁵ about chemical and physical processes in the Arctic, such as transport (Shindell et al., 2008) and the degradation mechanisms of non-methane hydrocarbons (Stavrakou et al., 2009), which need to be better understood in model simulations (Millet et al., 2008; Parker et al., 2011; Paulot et al., 2011). Our ground-based FTIR measurements at PEARL constitute a new Arctic dataset of seven tropospheric species, CO, HCN,
- $_{10}$ C₂H₆, C₂H₂, CH₃OH, HCOOH, and H₂CO, observed from 2007 to 2011. The total columns of CH₃OH are the first to be reported from ground-based FTIR measurements in the high Arctic.

These timeseries are obtained using the SFIT2 version 3.94 retrieval algorithm based on the OEM. The choices of the microwindows, a priori profiles and variabilities, as well as the signal-to-noise ratios have been described and selected to optimize

- the information content of the retrievals. As recommended by the NDACC/IRWG, a priori VMR profiles are from the WACCM model for all the species, except for C_2H_2 (which is from a combination of measurements). We also ensured that the observed seasonal variabilities are derived from the information content of the measurement, since a sin-
- $_{20}$ gle a priori profile for each species is used in the retrievals. The mean DOFS of the retrievals of the seven species range from 2.5 (CO) to 0.8 (H₂CO), allowing the derivation of total columns, which are representative of the tropospheric columns since they contribute to more than 90 % of the total columns. Averaging kernels and sensitivities of the target species show that the major contributions to the retrieved information in
- the troposphere come from the measurements. A full error analyses has been performed for all of the seven species derived from the FTIR measurements, with total errors ranging from 3.1% (CO) to 27.5% (H₂CO).

The different lifetimes of those tropospheric molecules, from less than two days to six months, play a role in their observed seasonal variabilities at PEARL. The seasonal





variability provides additional evidence of the interplay between chemistry and transport which will help to constrain global atmospheric chemical transport models. The timeseries of the CO, C₂H₆ and C₂H₂ total columns show strong seasonal amplitudes of 34, 93 and 104 %, respectively, with maxima in winter and minima in summer. These seasonal cycles highlight the combined effect of the chemistry via OH reaction (seasonal variability) and long-range transport (day-to-day variability) in the Arctic budget of CO, C₂H₆ and C₂H₂. In contrast to these molecules, the timeseries of the HCN, CH₃OH, HCOOH and H₂CO total columns show seasonal amplitudes of 78, 94, 61 and 93 %, respectively, with maxima in summer and minima in winter. These seasonal cycles are driven by biogenic emissions and short-range transport for CH₃OH, HCOOH

- and H₂CO, and biomass burning emissions and long-range transport for HCN. Overall, emissions from biomass burning seem to play a significant role in the day-to-day variability of the seven tropospheric species observed in the high Arctic. This dataset highlights the importance of the transport of pollutants from lower latitudes and can be used to assess the influx of pollution into the sensitive area of the Arctic.
- In order to assess our new dataset, the ACE-FTS satellite instrument was selected since it has good sampling in the Arctic and measures all seven species studied here. Given the values of the coefficients of correlation and the slopes, ranging from 0.56 to 0.97, and from 0.71 to 3.35, respectively, the FTIR measurements at Eureka of CO, HCN, C₂H₆ and C₂H₂ and HCOOH are well correlated with the ACE-FTS data. For CH₃OH and H₂CO, the small number of coincidences (6 pairs) does not allow us to draw significant conclusions. Given the good correlation with ACE-FTS measurements, our FTIR datasets provide an excellent means of ground validation for the ACE satellite over the high Arctic.
- Finally, our new dataset has been compared to previous published measurements. Our measurements are consistent with the negative trends of CO and C_2H_6 observed over the Northern Hemisphere, which has been attributed to decreasing fossil fuel emissions. The similarity of the HCN and C_2H_2 concentrations between mid- and high latitudes highlights the importance of transport on the atmospheric budget of these





molecules in the Arctic. Concentrations and seasonal cycles of CH_3OH , and HCOOH at PEARL are comparable with previous measurements performed at remote sites. Excellent agreement is found between our HCOOH total columns and measurements from Thule (Paulot et al., 2011). However, the atmospheric concentrations of H_2CO at

⁵ PEARL in the early spring are smaller than those reported in the literature. This might reflect the large atmospheric variability of H₂CO, given its small lifetime, with no local sources in winter at Eureka, and/or the effect of the updated spectroscopic parameters used in our retrievals.

To conclude, our measurements of tropospheric species at Eureka constitute a new dataset which can be used as a constraint to improve the model simulations of chemical and dynamical processes in the high Arctic. In particular, since all seven molecules are biomass burning products, they can be used to identify fire events, in order to study the chemical composition of the plumes above PEARL, and to derive emission ratios, which are key parameters needed to simulate fire emissions in atmospheric models.

- Acknowledgements. The PEARL Bruker 125HR measurements at Eureka were made by the Canadian Network for the Detection of Atmospheric Change (CANDAC), which was supported by the Atlantic Innovation Fund/Nova Scotia Research and Innovation Trust, the Canada Foundation for Innovation, the Canadian Foundation for Climate and Atmospheric Sciences, the Canadian Space Agency (CSA), Environment Canada, Government of Canada International Detection for Innovation Part (CSA), Environment Canada, Government of Canada International
- Polar Year funding, the Natural Sciences and Engineering Research Council, the Ontario Innovation Trust, the Ontario Research Fund, and the Polar Continental Shelf Program. The authors wish to thank the staff at the Eureka weather station and CANDAC for logistical and on-site support. Thanks to Rodica Lindenmaier, Rebecca Batchelor, PEARL Site Manager Pierre Fogal, and CANDAC/PEARL operators Ashley Harrett, Alexei Khmel, Paul Loewen, Keith MacQuarrie,
- ²⁵ Oleg Mikhailov, and Matt Okraszewski, for their invaluable assistance in maintaining and operating the Bruker 125HR. The Atmospheric Chemistry Experiment is a Canadian-led mission mainly supported by the CSA.





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Table 1. Sources, sinks and average lifetimes of all the target species. BB, VOC and NMHCs are the acronyms of Biomass Burning, Volatile Organic Compound and Non Methane Hydro-Carbons, respectively.

Target species	Name	Sources	Sinks	Lifetimes
CO	Carbon monoxide	BB, transport, steel industry, methane and VOC oxidation	reaction with OH	2 months
HCN	Hydrogen cyanide	BB, industry, fungi and plant emission	reaction with OH and ocean uptake	2–6 months
C_2H_6	Ethane	BB, biofuel use, natural emission	reaction with OH	1.5 months
C_2H_2	Acetylene	BB, combustion product, natural emission	reaction with OH	2 weeks
CH ₃ OH	Methanol	BB, biogenic emission	reaction with OH	5–10 days
НСООН	Formic acid	BB, biogenic emission, photo-oxidation of NMVOCs	reaction with OH, dry and wet deposition	3–4 days
H ₂ CO	Formaldehyde	BB, oxidation of methane and NMVOCs	reaction with OH	< 2 days





Table 2. Parameters (microwindows, interfering species, a priori VMR sources, standard deviations (STD) of the a priori covariance matrix, and signal-to-noise ratios – SNR) used in the retrievals of the seven target gases.

Target species		Microwindows	Interfering species	a priori VMR	STD for \mathbf{S}_{a} (%)	SNR
Carbon monoxide	CO	2057.684–2058.000, 2069.560–2069.760, 2157.507–2159.144	$O_3, CO_2, OCS, H_2O, N_2O$	WACCM V6	20	85
Hydrogen cyanide	HCN	3268.000–3268.380, 3287.000–3287.480	H ₂ O	WACCM V6	50	200
Ethane	C ₂ H ₆	2976.660–2976.950, 2983.200–2983.550, 2986.500–2986.950	H ₂ O, O ₃	WACCM V4.5	30	250
Acetylene	C_2H_2	3250.430–3250.770, 3255.180–3255.725, 3304.825–3305.350	H ₂ O, HDO	GC Toon Kiruna991203 Mk4-flight 6–34 km, outside spitprim.set, divided by 2	50	200
Methanol	CH₃OH	992.000–998.700, 1029.000–1037.000	O ₃ , O ₃ ⁶⁸⁶ , O ₃ ⁶⁶⁸ , O ₃ ⁶⁷⁶ , O ₃ ⁶⁶⁷ , H ₂ O, CO ₂	WACCM V6	20	200
Formic acid	НСООН	1104.650–1105.600	$\begin{array}{l} \text{HDO}, \text{ O}_3,\\ \text{O}_3^{668}, \text{O}_3^{676},\\ \text{H}_2\text{O}, \text{NH}_3,\\ \text{CCL}_2\text{F}_2, \text{CHF}_2\text{CL},\\ \text{CH}_4 \end{array}$	WACCM V6	100	800
Formaldehyde	H ₂ CO	2778.120–2778.800, 2780.600–2781.170	CH ₄ , CO ₂ , O ₃ , N ₂ O	WACCM V6 divided by 2	100	500



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Table 3. Error budgets, in percentage, as a function of the different sources of random and systematic uncertainties on typical CO, HCN, C_2H_6 , C_2H_2 , CH_3OH , HCOOH, and H_2CO total columns retrieved at PEARL. DOFS and SZA are the acronyms of Degrees of Freedom for Signal and Solar Zenith Angle, respectively.

Error budget (%)	CO	HCN	C_2H_6	C_2H_2	CH_3OH	HCOOH	H_2CO
N spectra retrieved in timeseries	3980	1815	1819	1269	1095	1973	1242
DOFS	2.5	1.6	2.0	1.1	1.2	1.1	0.8
Random errors:							
Measurement error	0.2	4.3	1.2	7	4.6	16.3	10
Uncertainty on temperature	0.7	0.4	0.3	2.9	4.4	2.5	5.5
Uncertainty on retrieval parameters	0.3	4.3	2.6	8.2	0	0	1.1
Uncertainty on interfering species	0	0.2	0	0.1	0	0.3	24.1
Uncertainty on SZA	0.3	0.3	0.3	0.7	0.2	0	0.2
Total random error	0.8	6.1	2.9	11.6	6.3	16.5	26.7
Systematic errors:							
Uncertainty on line intensity	2.9	4.9	6.4	5.7	10	3.6	2.4
Uncertainty on line width	0.4	3.7	2.4	7.7	1.1	0.9	2.1
Total random and systematic error	3.1	8.7	7.4	15.1	11.9	17	26.9
Smoothing error	0.2	5.9	12.2	16.7	3	1.5	5.7
Total error (random + systematic + smoothing)	3.1	10.5	14.3	22.5	12.3	17	27.5



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Table 4. Monthly mean total columns (in molecules cm ⁻²) for all trace gases derived in this study
from 2007 to 2011. Numbers in parenthesis correspond to the one-sigma standard deviation
around the monthly mean. The three last rows show the annual means, the one-sigma standard
deviation (STD), and the seasonal amplitude in percentage.

Monthly mean total columns	CO × 10 ¹⁸	HCN × 10 ¹⁵	$C_2H_6 \times 10^{16}$	$C_2H_2 \times 10^{15}$	CH ₃ OH × 10 ¹⁶	HCOOH × 10 ¹⁵	$H_2CO \times 10^{15}$
Feb	2.02(0.11)	2.78(0.15)		4.57		1.05(0.25)	
Mar	2.20(0.14)	3.37(0.48)	2.85(0.14)	5.50(0.53)	0.42(0.20)	1.43(0.64)	0.36(0.22)
Apr	2.17(0.14)	4.58(0.59)	2.66(0.28)	5.00(0.79)	0.74(0.19)	2.23(0.93)	0.27(0.20)
May	1.97(0.15)	5.39(0.54)	2.07(0.21)	3.31(0.74)	1.08(0.45)	1.99(1.05)	0.59(0.63)
Jun	1.69(0.10)	5.58(0.34)	1.55(0.15)	2.12(0.58)	1.57(0.80)	2.20(0.86)	2.42(1.00)
Jul	1.59(0.17)	5.76(0.72)	1.29(0.24)	1.87(0.71)	1.82(0.39)	1.87(1.05)	3.14(1.02)
Aug	1.63(0.17)	6.63(1.60)	1.20(0.17)	1.94(0.59)	1.53(0.49)	2.22(0.88)	2.81(0.95)
Sep	1.57(0.14)	5.28(1.10)	1.23(0.18)	1.61(0.04)	1.36(0.33)	1.58(0.72)	1.13(0.77)
Oct	1.56(0.08)	4.69(0.58)	1.24(0.03)	1.38		1.14(0.47)	0.55(0.38)
Mean	1.89	4.97	1.77	3.95	1.49	1.94	2.30
STD	0.30	1.29	0.65	1.69	0.59	0.90	1.28
Seasonal amplitude (%)	34	78	93	104	94	61	93





Table 5. Comparison of FTIR and ACE-FTS partial columns for all trace gases derived in this study. *N* represents the number of coincidences involved in the comparison, *R* is the correlation coefficient, and the fifth column gives the values of the slope of the regression plot between the FTIR partial columns and the coincident ACE-FTS measurements along with the 1 σ uncertainties of the slopes, for each target species. The second column shows the mean distance of ACE-FTS occultations to PEARL, and the last column shows the mean altitude ranges used for the partial column calculations.

	N distances (km) to PEARL	Mean	<i>R</i> (FTIR vs ACE-FTS)	Slope	Mean altitude range for partial columns (km)
CO	106	317	0.66	0.97 ± 0.11	9.4-48.6
HCN	93	313	0.96	0.69 ± 0.02	7.9–33.2
C_2H_6	17	436	0.97	0.71 ± 0.04	8.0–19.3
C_2H_2	93	319	0.78	1.21 ± 0.10	8.0-17.0
CH ₃ OH	6	316	0.91	0.74 ± 0.14	9.4–17.5
HCOOH	103	317	0.56	3.35 ± 0.49	8.1–18.5
H ₂ CO	6	317	0.75	0.50 ± 0.22	9.1–38.7







Fig. 1. Microwindows used in the retrievals of CO, HCN, C_2H_6 , C_2H_2 , CH_3OH , HCOOH, and H_2CO , with examples of clear and polluted conditions in blue and red, respectively.





Fig. 2. The CO a priori VMR profile from WACCM v6 in black (upper panel). The black error bars represent the standard deviation of the a priori covariance matrix used in the retrievals, at each layer. The mean retrieved profiles is in red, with error bars corresponding to the 1σ standard deviation around the mean. The number in blue is the percentages of the tropospheric column contributions (between 0.6 and 10.7 km) relative to the total column (between 0.6 and 120 km). Typical VMR averaging kernels in VMR/VMR (lower left panel), total column averaging kernels in (molecule cm⁻²)/(molecule cm⁻²) (lower right panels, black line), and sensitivity profiles (right panel, red line) as a function of altitude. The colours correspond to averaging kernels at altitudes lying in a partial column for which the DOFS is about 0.5.







Fig. 3. Same as Fig. 2 but for HCN.

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Fig. 4. Same as Fig. 2 but for C_2H_6 . The C_2H_6 a priori VMR profile is from WACCM v4.5.



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Fig. 5. Same as Fig. 2 but for C_2H_2 . The C_2H_2 a priori VMR profile is from MkIV balloon measurements made at the high-latitude NDACC site of Kiruna, between 6 and 34 km altitude, combined with spitprim.set (Ft. Sumner MkIV flights, 1990s, http://mark4sun.jpl.nasa.gov/m4. html).





Fig. 6. Same as Fig. 2 but for CH₃OH.

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Fig. 7. Same as Fig. 2 but for HCOOH.



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Fig. 8. Same as Fig. 2 but for H₂CO. The H₂CO a priori VMR profile is from WACCM v6 divided by 2.

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Fig. 9. Daily mean total columns (in molecules cm⁻²) of CO, HCN, C₂H₆, C₂H₂, CH₃OH, HCOOH, and H₂CO, plotted in different colours for the different years of measurements. The brown line shows the polynomial fits to the data.



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