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Five years of CO, HCN, C₂H₆, C₂H₂, CH₃OH, HCOOH, and H₂CO total columns measured in the Canadian High Arctic

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Abstract

We present a five-year timeseries of seven tropospheric species measured using a ground-based Fourier Transform InfraRed (FTIR) spectrometer at the Polar Environment Atmospheric Research Laboratory (PEARL, Eureka, Nunavut, Canada, 80°05' N, 86°42' W) from 2007 to 2011. Total columns and temporal variabilities of carbon monoxide (CO), hydrogen cyanide (HCN), and ethane (C₂H₆), as well as the first derived total columns at Eureka of acetylene (C₂H₂), methanol (CH₃OH), formic acid (HCOOH), and formaldehyde (H₂CO) are investigated, providing a new dataset in the sparsely sampled high latitudes.

Total columns are obtained using the SFIT2 retrieval algorithm based on the Optimal Estimation Method. The microwindows, as well as the a priori profiles and variabilities are selected to optimize the information content of the retrievals, and error analyses are performed for all seven species. Our retrievals show good sensitivities in the troposphere. The seasonal amplitudes of the timeseries, ranging from 34 to 104 %, are captured while using a single a priori profile for each species. The timeseries of the CO, C₂H₆ and C₂H₂ total columns at PEARL exhibit strong seasonal cycles with maxima in winter and minima in summer, in opposite phase to the HCN, CH₃OH, HCOOH and H₂CO timeseries. These cycles result from the relative contributions of the photochemistry, oxidation, and transport, as well as biogenic and biomass burning emissions.

Comparisons of the FTIR partial columns with coincident satellite measurements by the Atmospheric Chemistry Experiment Fourier Transform Spectrometer (ACE-FTS) show good agreement. The correlation coefficients and the slopes range from 0.56 to 0.97, and 0.50 to 3.35, respectively, for the seven target species.

Our new dataset is compared with previous measurements found in the literature to assess atmospheric budgets of these tropospheric species in the high Arctic. The CO and C₂H₆ concentrations are consistent with negative trends observed over the Northern Hemisphere, attributed to fossil fuel emission decrease. The importance of poleward transport on the atmospheric budgets of HCN and C₂H₂ is highlighted. Columns

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and variabilities of CH₃OH, and HCOOH at PEARL are comparable to previous measurements performed at other remote sites. However, the small columns of H₂CO in early May might reflect its large atmospheric variability, and/or the effect of the updated spectroscopic parameters used in our retrievals. Overall, emissions from biomass burning contribute to the day-to-day variabilities of the seven tropospheric species observed at Eureka.

1 Introduction

Future climate change may cause significant air quality reduction by changing the outflow of pollutants as well as the strength of emissions from the biosphere, fires and dust. The sign and magnitude of these effects are highly uncertain and will vary regionally (IPCC, 2007). A key challenge for scientists is to evaluate changes in atmospheric composition induced by human activities. This is especially true in sensitive areas, such as the Arctic, which has been warming rapidly over the past century with an accentuated heating in the past decades (Trenberth et al., 2007; Lesins et al., 2010). The chemical environment of the Arctic is unique and characterized by low temperatures, extremes in radiation with long periods of light and darkness, and chemical processes involving snow and ice at the surface. The Arctic is a major receptor for mid-latitude pollution (Shaw, 1995; Quinn et al., 2007, 2008) and changes in chemistry and influx of pollution may disrupt this sensitive system (Rinke et al., 2004). Several studies have identified pollution transport to the Arctic based on model simulations and meteorological analyses (Eckhardt et al., 2003; Klonecki et al., 2003; Koch and Hansen, 2005; Stohl et al., 2006; Shindell et al., 2008), but our ability to verify these pathways through chemical observations has been limited.

Ground-based instruments can use solar absorption spectroscopy to measure the chemical composition of the atmosphere. These measurements provide a key dataset for the validation of satellite remote-sensing instruments and model data. The atmospheric variabilities of pollutants and natural gases in a remote area can be quantified

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using long-term monitoring of tropospheric molecules released by both natural sources and human activities. This contributes to a better understanding of Arctic chemistry, as well as the factors driving current changes in Arctic atmospheric composition and climate. In this study, we investigate the atmospheric concentrations and variabilities of seven tropospheric trace gases using ground-based Fourier Transform InfraRed (FTIR) spectra, recorded at the Polar Environment Atmospheric Research Laboratory (PEARL, Eureka, Nunavut, Canada, 80°05′ N, 86°42′ W) from 2007 to 2011.

These molecules (listed in Table 1) are carbon monoxide (CO), hydrogen cyanide (HCN), ethane (C₂H₆), acetylene (C₂H₂), methanol (CH₃OH), formic acid (HCOOH), and formaldehyde (H₂CO). They present different source and sink mechanisms in the atmosphere and their different lifetimes play a role in their observed seasonal variabilities (Notholt et al., 1997a). All these atmospheric species absorb incoming sunlight in the infrared region and therefore exhibit spectral features in the FTIR measurements used to derive their atmospheric concentrations.

The ground-based FTIR technique can be used to sample the chemical composition for long periods, providing vertically integrated measurements from ground to the top of the atmosphere. This allows the different temporal variabilities of various tropospheric constituents to be assessed. This technique has previously been employed to monitor atmospheric species in various part of the world. CO, HCN, C₂H₆ and C₂H₂ total and partial columns have been measured by ground-based FTIR spectroscopy at several locations in the Northern Hemisphere (Mahieu et al., 1997; Rinsland et al., 1998, 2000; Zhao et al., 2002) and in the Southern Hemisphere (Rinsland et al., 1999, 2001, 2002; Paton-Walsh et al., 2010; Vigouroux et al., 2012).

CH₃OH and HCOOH have been measured in a limited number of ground-based FTIR studies. CH₃OH measurements have been performed at Wollongong (Australia, Paton-Walsh et al., 2008) and at Kitt Peak (United States, Rinsland et al., 2009), and HCOOH has been measured mainly in the Northern Hemisphere (Rinsland et al., 2004; Zander et al., 2010; Paulot et al., 2011), and also in the Southern Hemisphere (Paton-Walsh et al., 2005).

of the target species, listed in Table 1, have been reported in biomass burning plumes (Goode et al., 2000; Simpson et al., 2011; Le Breton et al., 2013).

In contrast to these measurements, the ground-based FTIR technique can provide total columns, with good sensitivity in the lower troposphere (compared to satellite measurements), as well as long-term spectral acquisition, in clear-sky conditions, thereby enabling an assessment of the temporal variabilities of the target species in the high Arctic (compared to campaign-based measurements).

We focused on these species because there remain numerous gaps in the available observational datasets, especially at high latitudes. Furthermore, the transport and the degradation mechanisms of non-methane hydrocarbons (NMHCs) are poorly understood and should be better quantified in order to increase our ability to predict trace gas concentrations and variability in models (Stavrakou et al., 2009).

For instance, simulated CO concentrations disagreed by a factor of two to three at all altitudes in the Arctic in a comparison of eleven chemical transport models (Shindell et al., 2008). This has been attributed to model differences in emissions, transport, and OH concentrations. Thus, there is a need to better understand the sources and the transport of CO to the Arctic, as an indicator of pollution effects (Fisher et al., 2010). Concerning C₂H₂, large uncertainties remain with regard to the magnitude of its sources and sinks, as well as its spatial distribution and seasonality in the atmosphere (Parker et al., 2011). In addition, large uncertainties remain in the CH₃OH atmospheric budget, in terms of its source magnitudes, seasonality, and spatial distribution in the atmosphere (Millet et al., 2008). Concerning HCOOH, several studies have highlighted the underestimation of emissions in the models, by a factor of nine in the marine boundary layer (Baboukas et al., 2000) and an order of magnitude for the free troposphere (Von Kuhlmann et al., 2003). Recently, Paulot et al. (2011) has investigated an underestimation in the model by a factor two to five compared to polar FTIR measurements at Thule (Greenland, 76° N, 69° W), confirming the missing local sources in the HCOOH budget simulated in the model. Finally, HCOOH and H₂CO spectroscopic parameters have been refined recently (HITRAN

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atmosphere, \mathbf{S}_e is the measurement covariance matrix and \mathbf{S}_a is the a priori covariance matrix used in the OEM.

The spectroscopic parameters are from the HITRAN 2008 database (Rothman et al., 2009) for all the species derived in this study. As noted in Sect. 1, it contains significant improvements concerning HCOOH and H₂CO line intensities in the spectral regions of interest.

The retrieval parameters (microwindows, interfering species, a priori VMR sources, diagonal values of the a priori covariance matrix, and Signal-to-Noise Ratios (SNR)) used for all the target species are summarized in Table 2 and discussed below.

2.3 Microwindows

The choice of microwindows (MWs) is crucial because the majority of the target species have relatively weak absorptions in the infrared region compared to the main interfering species, such as CH₄, H₂O, and O₃. All MWs have been selected in order to increase the information content and minimize the errors in the retrievals. These MWs are shown in the Fig. 1, with examples of spectra recorded in clear and polluted conditions in blue and red, respectively. The MWs of CO and C₂H₆ were selected based on the recommendations of the NDACC/IRWG in their harmonization effort strategy.

The CO total columns are retrieved in three widely used MWs, including a weak line of CO at 2057.858 cm⁻¹, another weak line of CO at 2069.656 cm⁻¹ and a strong line of CO at 2158.300 cm⁻¹ (Notholt et al., 2000; Zhao et al., 2002). The use of a combination between weak and strong absorption lines increases the vertical sensitivity of the retrievals (Barret et al., 2003). Interfering species (CO₂, O₃, OCS, N₂O and H₂O) are simultaneously scaled from their a priori profiles during the inversions. In order to obtain a good estimate of the measurement noise covariance matrix, we constructed trade-off curves of the RMS residual (i.e. spectral fit residuals of the retrieval) vs. SNR (Batchelor et al., 2009). The resulting SNR of 85 minimized the RMS residuals and maximized the DOFS of the CO retrievals.

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For HCN, two MWs, around 3287.248 cm^{-1} (Mahieu et al., 1997; Notholt et al., 2000; Zhao et al., 2002), and around 3268.200 cm^{-1} , are preferred over the three IRWG recommendations (which are $3268.05\text{--}3268.40\text{ cm}^{-1}$, $3287.10\text{--}3287.35\text{ cm}^{-1}$, and $3299.40\text{--}3299.60\text{ cm}^{-1}$). This choice increases the information content and decreases the errors in the Eureka retrievals. The profiles of the only interfering species, H₂O, are scaled from the a priori profile and a SNR of 200 is used in the retrievals.

The C₂H₆ retrievals are performed in three MWs around 2976.800 cm^{-1} (Mahieu et al., 1997; Rinsland et al., 2002; Paton-Walsh et al., 2010), 2983.300 cm^{-1} (Meier et al., 2004) and 2986.700 cm^{-1} (Notholt et al., 1997a) using a SNR of 250. Single scaling parameters are used for each of the interfering gases (H₂O and O₃).

For C₂H₂, we used three MWs, around 3250.500 cm^{-1} (Petersen et al., 2008), 3255.200 cm^{-1} (Mahieu et al., 2008), and 3305.400 cm^{-1} (Mahieu et al., 2008; Paton-Walsh et al., 2010). H₂O and its main isotopologue (HDO) are scaled simultaneously from their a priori profiles. Because the infrared absorptions of the target gas are weak compared to the H₂O lines, a reduced SNR is employed in the spectral region with no C₂H₂ features. While a SNR of 200 is used overall the MWs, we apply a SNR of 50 between 3250.430 and 3250.550 cm^{-1} and between 3255.180 and 3255.455 cm^{-1} and a SNR of 75 from 3305.065 to 3305.350 cm^{-1} .

For CH₃OH, we use the two MWs employed by Bader et al. (2013). In this region, the CH₃OH band at 1033 cm^{-1} represents around 1.5 % of the absorption, whereas the O₃ lines represent about 94 %. Thus, O₃ and its isotopologues (¹⁶O¹⁸O¹⁶O labeled O₃⁶⁸⁶, ¹⁶O¹⁶O¹⁸O labeled O₃⁶⁶⁸, ¹⁶O¹⁷O¹⁶O labeled O₃⁶⁷⁶, and ¹⁶O¹⁶O¹⁷O labeled O₃⁶⁶⁷, Table 2), as well as the other interfering species (CO₂ and H₂O) are retrieved simultaneously by scaling their a priori profiles. We use a SNR of 200. For more clarity in Fig. 1, we have shown an example of spectral fitting with the contributions of all the species in the second MW. The significant broad absorption feature of CH₃OH can be seen from around 1032 to 1035 cm^{-1} (Fig. 1, right panel, cyan line).

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For HCOOH, we use one MW between 1104.650 and 1105.600 cm⁻¹, which correspond to the band used in Zander et al. (2010), Paulot et al. (2011) and Vigouroux et al. (2012). The spectroscopic parameters of this band have been updated (Perrin and Vander Auwera, 2007). Zander et al. (2010) have shown that these improvements reduce the retrieved total columns by a factor of two. In this MW, a global SNR of 800 is used and interfering species (HDO, O₃, O₃⁶⁶⁸, O₃⁶⁷⁶, H₂O, NH₃, CCl₂F₂, CHF₂Cl, CH₄) are scaled from their a priori profiles, simultaneously.

For H₂CO, MWs are from Paton-Walsh et al. (2005) where the lines belong to two bands, centered at 2782.000 cm⁻¹ and 2843.000 cm⁻¹, respectively. The interfering species, such as CH₄, O₃, CO₂ and N₂O, are simultaneously scaled from their a priori profiles. The SNR assumed in the MWs is 500, and is reduced to 100 in the spectral regions where systematic residual features were caused by various inaccuracies in the CH₄ line parameters (Sussmann et al., 2011).

2.4 A priori information

The location of the instrument at high latitudes offers a relatively dry atmosphere (total precipitable water ranges between 0 and 1.8 g cm⁻², Wagner et al., 2006, their Fig. 3). Compared to tropical FTIR sites, such as Reunion Island (Vigouroux et al., 2012) or Darwin in Australia (Paton-Walsh et al., 2010), our spectra are less affected by strong features due to water vapor abundance. Thus a pre-fitting of water vapour (or “two-step retrieval”) is not necessary here.

Instead, the difficulty resides in the lack of continuous tropospheric measurements in the Arctic, needed to build our a priori knowledge of the polar atmospheric states. Our a priori VMR profiles are from the Whole Atmosphere Community Climate Model (WACCM, <http://www2.cesm.ucar.edu/working-groups>) as recommended by the NDACC/IRWG, except for the C₂H₂ retrievals (see above). WACCM is a numerical model developed at the National Center for Atmospheric Research (Sassi et al., 2002),

which was used to compute 40 yr average of the modelled profiles at Eureka, which served as a priori VMRs for the target species.

In Figs. 2 to 8 (upper panels), the a priori profiles adopted for the FTIR retrievals are shown in black, with the mean of all retrieved profiles in red. The black and red error bars correspond to the standard deviation of the a priori covariance matrix used in the retrievals and the standard deviations around the mean retrieved profiles, respectively, at each layer. All the a priori VMR profiles exhibit large contributions in the boundary layer and the troposphere, and decrease to almost zero at 50 km, except for CO, for which the sources are dominated by CH₄ oxidation at this altitude (Clerbaux et al., 2008).

The CO a priori VMR is about 92 ppbv (1 ppbv = 10⁻⁹ per unit volume) through the boundary layer, decreases to about 20 ppbv at the tropopause region and increases again with altitude to 0.6 ppmv (1 ppmv = 10⁻⁶ per unit volume) at 50 km. The HCN a priori VMR slightly increases from 25 ppbv to 32 ppbv in the troposphere and decreases in the stratosphere to reach 11 ppbv in the stratosphere. The C₂H₆, C₂H₂ and CH₃OH a priori VMR profiles have similar shapes as a function of altitude. Note that the C₂H₆ and C₂H₂ profiles are plotted on a log scale whereas the CH₃OH profile is plotted on a linear scale for clarity (Figs. 2 to 8, upper panel). Their VMRs are almost constant in the boundary layer and the troposphere, with values of about 1 ppbv for C₂H₆, and 0.5 ppbv for both C₂H₂ and CH₃OH. At the upper altitudes, the a priori profiles of C₂H₆, C₂H₂ and CH₃OH decrease until reaching ~ 1 pptv (1 pptv = 10⁻¹² per unit volume) at 22 km, 17 km and 21 km, respectively. For HCOOH and H₂CO, the a priori VMRs in the boundary layer are about 8 pptv and 30 pptv, and then decrease to 1 pptv and 10 pptv around 30 km for HCOOH and H₂CO, respectively. However, the sinks and sources are not as well understood (Paulot et al., 2011; Jones et al., 2009), especially in the polar region, so the interpretation of the a priori profiles has to be done carefully.

The a priori VMR profiles are from WACCM version 6 for CO, HCN, CH₃OH, HCOOH, and divided by two at all altitudes for H₂CO (see explanation below). The low stratospheric values of C₂H₆ VMRs in WACCM version 6 were beyond the retrieval software

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precision, therefore the a priori VMR profiles of C₂H₆ are from WACCM version 4.5. For C₂H₂, the a priori profile assumed here is not derived from WACCM because the VMR values were extremely low in the upper atmosphere and not representative, causing the retrievals to rarely converge. To avoid these instabilities in the retrievals, we used a profile derived from MkIV balloon measurements made at the high-latitude NDACC site of Kiruna (Sweden, Toon et al., 1999), between 6 and 34 km altitude, combined with spit-prim.set (Ft. Sumner MkIV flights, 1990s, <http://mark4sun.jpl.nasa.gov/m4.html>). This a priori profile has been divided by two so that the tropospheric VMR values are comparable to the ones derived from the WACCM runs.

Instabilities in the retrieval also appear if the measured columns are substantially smaller than the a priori columns. As a result, the retrieved profile returns non-physical negative mixing ratios at the altitudes where the information content is low. This means that the most practical approach is often to use a priori profiles with a significantly lower total column than the average expected from the dataset, and to compensate by using larger values in the covariance matrix **S_a** (Paton-Walsh, 2009). Thus, the a priori profile of H₂CO has been divided by two in order to reduce the RMS residual and avoid oscillations in the retrieved profiles. We also verified that the retrieved total columns did not change significantly (within error bars of the measurements) when using both profiles (divided by two and not). This confirms that the majority of the information content is coming from the measurements and not from the a priori states. It is important to note that a single a priori profile for each species is used for all spectra recorded from 2007 to 2011. This ensures that our observed atmospheric variabilities mainly come from the information content of the measurements.

In the retrieval process, the measured spectrum is weighted by the empirically defined a priori covariance matrix **S_a**, with diagonal elements reflecting the one-sigma uncertainties used in the OEM. For CO, C₂H₂, CH₃OH, HCOOH, and H₂CO, the off-diagonal elements of **S_a** correspond to an exponential inter-layer correlation with a correlation length of 4 km. For the others species, no interlayer correlation is applied.

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The relative uncertainties of the a priori VMR profiles (or the standard deviations used in the covariance matrix) are assumed to be 20 % for CO and CH₃OH, 50 % for HCN and C₂H₂, 30 % for C₂H₆, and 100 % for HCOOH and H₂CO, at all altitude layers. These values have been scaled to account for the different thicknesses of the 47 layers in our retrieval scheme. These values, summarized in Table 2, are representative of the tropospheric variabilities derived from the WACCM model, except for CH₃OH and H₂CO. Those are, on average over the troposphere, 17 % for CO, 28 % for HCN, 30 % for C₂H₆, and 66 % for C₂H₂. For CH₃OH, the WACCM model suggests a high tropospheric variability of about 71 %. Since this species has broad absorption features, we prefer to use a lower value of SNR in the retrievals, and then decrease the one-sigma uncertainties to 20 % in the OEM, to avoid oscillations in the retrieved profiles. In contrast, the high SNR employed in the HCOOH and H₂CO retrievals, commensurate with the SNR of most analyzed spectra, has been compensated for by using large values in the diagonal elements of the a priori covariance matrix. For those two species, the average tropospheric variabilities given by the model are 66 % for HCOOH and 39 % for H₂CO. These variabilities are rather small and not consistent with the idea of an additional large source of HCOOH from snow photochemistry (Dibb and Arsenault, 2002) and the large H₂CO variability of 30 pptv to 700 pptv observed at the Arctic surface (De Serves, 1994).

Finally, the total columns of all the target species are representative of the tropospheric columns. The numbers in blue in Figs. 2 to 8 (upper panels) are the percentages of the tropospheric column contributions (between 0.6 and 10.7 km) relative to the total column (between 0.6 and 120 km). As can be seen, the tropospheric columns contribute to more than 90 % of the total columns.

2.5 Averaging kernels

The rows of **A** (Eq. 1) correspond to the averaging kernels for a certain altitude layer and characterize the information content of the retrievals. An ideal observation has an averaging kernel of one in the region of interest and zero outside (Connor et al., 1996).

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smoothing error (expressing the limited vertical resolution of the retrieval), and the forward model parameter error (including error on the temperature profiles, on spectroscopic and retrieval parameters, interfering species uncertainties, and error on solar zenith angles). The error analyses have been performed on a representative dataset of around 10 spectra per species, selected with different values of Solar Zenith Angle (SZA), SNR, hour and season of measurements. The averages of all the calculated errors are shown for each target species in Table 3.

The measurement error covariance matrix \mathbf{S}_m is calculated as:

$$\mathbf{S}_m = \mathbf{G}_y \mathbf{S}_\varepsilon \mathbf{G}_y^T \quad (2)$$

where \mathbf{G}_y is the gain matrix representing the sensitivity of the retrieved parameter to the measurement, and \mathbf{S}_ε is the measurement covariance matrix as seen in Eq. (1). The diagonal elements of \mathbf{S}_ε are the squares of the spectral noise, which is determined by the RMS residuals derived from the spectra.

The smoothing error covariance matrix \mathbf{S}_s is calculated as:

$$\mathbf{S}_s = (\mathbf{I} - \mathbf{A}) \mathbf{S}_e (\mathbf{I} - \mathbf{A})^T \quad (3)$$

where \mathbf{I} is the identity matrix, \mathbf{A} is the averaging kernel matrix, and \mathbf{S}_e is the climatology matrix representing the natural variability of the target species. The lack of continuous measurements of tropospheric species in the Arctic makes it difficult to build this matrix. Thus, the values derived from WACCM have been used as diagonal elements, to express the natural variabilities of each trace gas. For CO, C₂H₂, CH₃OH, HCOOH, and H₂CO, the off-diagonal elements of \mathbf{S}_e correspond to an exponential inter-layer correlation with a correlation length of 4 km. For the others species, no interlayer correlation is applied, consistent with the retrievals.

Finally the forward parameter error covariance matrix \mathbf{S}_f is calculated as:

$$\mathbf{S}_f = (\mathbf{G}_y \mathbf{K}_b) \mathbf{S}_b (\mathbf{G}_y \mathbf{K}_b)^T \quad (4)$$

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where \mathbf{G}_y is the gain matrix, and \mathbf{K}_b is the Jacobian matrix obtained by perturbation methods. \mathbf{K}_b represents the sensitivity of the measurements to the model parameter b . \mathbf{S}_b is the covariance error matrix on the parameter b . These parameters b can have systematic (spectroscopic parameters) and random (uncertainties on temperature and SZA, for instance) components. To generate the Jacobians, parameters are perturbed by 0.1° for the error on SZA, by 2 K for the temperature profiles, and by a factor of 1.05 for the line intensity and air-broadened parameters (Batchelor et al., 2009).

In addition to these errors, we have examined the errors on the other retrieved parameters, called interference errors as described in Rodgers and Connor (2003) and explained in detail in Sussmann and Borsdorff (2007) and in Batchelor et al. (2009, Sect. 4). The interference errors combine the errors on the interfering species in the spectral region of interest, and the uncertainty due to the retrieval parameters, such as the wavelength shift, the ILS, the background slope and curvature, the zero-level shift, and the phase error.

The total random errors are calculated by adding all the uncertainties in quadrature, except the uncertainties on line intensity and line width. The predominant contribution to the total random errors is the measurement error for HCN and HCOOH. For CO, the uncertainty on temperature is slightly higher than the uncertainty on the measurement error, but still remains small. For C₂H₆ and C₂H₂, the measurement error is almost as large as the uncertainties on retrieval parameters. This might be an effect of the H₂O continuum, which affects the background slope and curvature of retrievals. For CH₃OH and H₂CO, the temperature uncertainty is high, which is consistent with what has been reported in Vigouroux et al. (2012) for CH₃OH. Concerning H₂CO, the largest random uncertainties come from the uncertainty on the interfering species. This might be explained by the fact that H₂CO absorption lines are close to CH₄ lines (in both MWs), which are difficult to fit because of spectroscopic parameter uncertainties (Sussmann et al., 2011). This can also be an effect of the deweighting function used to reduce the SNR of the interfering species in the H₂CO microwindows.

of correlation is 0.56 with a slope of 3.35 ± 0.49 . The mean altitude ranges for partial column calculations are 8.0–48.6 km for CO, 7.9–33.2 km for HCN, 8.0–17.0 km for C₂H₂, and 8.1–18.5 km for HCOOH.

The FTIR C₂H₆ partial columns, calculated on average between 8.0 and 19.3 km, agree well with the ACE-FTS data. The coefficient of correlation is 0.97, with a slope of 0.71 ± 0.04 over 17 coincident measurements, within a mean distance to PEARL of 436 km.

For CH₃OH and H₂CO comparisons, the small number of coincidences ($N = 6$) makes it difficult to draw significant conclusions. The coefficients of correlation between ACE-FTS and FTIR partial columns, calculated on average from 9.4 to 17.5 km for CH₃OH and from 9.1 to 38.7 km, are 0.91 and 0.75, with slopes of 0.74 ± 0.14 and 0.50 ± 0.22 , respectively.

It is worth noting that 21 observations were rejected for the H₂CO comparison because they were measured after October 2010. The ACE-FTS algorithm version 3.5 will improve our H₂CO comparisons in the future.

Given the values of the coefficients of correlation and the slopes, the Eureka FTIR measurements of CO, HCN, C₂H₆ and C₂H₂ and HCOOH are well correlated with ACE-FTS data. For CH₃OH and H₂CO, the small number of coincident profiles measurements does not allow us to draw significant conclusions.

3.3 Discussion

Comparisons between our dataset and previous measurements reported in the literature lead to a discussion of the atmospheric budget of the different target species observed in the high Arctic.

Our retrieved CO total columns are on five-years average smaller by a factor of 1.3 compared to CO measured at Ny Ålesund (Norway, 79° N, 12° E) from 1992 to 1995 (Notholt et al., 1997b). This is consistent with the decrease of tropospheric CO of $-0.61 \pm 0.16 \text{ \% yr}^{-1}$ observed at high-latitude sites between 1996 and 2006 (Angelbratt et al., 2011). This trend has been explained with the combination of a 20 % decrease in

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anthropogenic CO emissions in Europe and North America and 20 % increase in CO anthropogenic emission in East Asia.

Concerning C₂H₆, our total columns are smaller by a factor 1.1 compared to measurements performed in the Arctic from 1992 to 1995 (Notholt et al., 1997b), suggesting that the Arctic C₂H₆ budget has been decreasing from this period to 2007–2011. This confirms the significant negative trend observed in the Northern Hemisphere at Kiruna (Sweden, 67° N, 20° E) and Harestua (Norway, 60° N, 10° E) from 1996 to 2006 (Angelbratt et al., 2011). Aydin et al. (2011) attributed this trend to the decrease of C₂H₆ based fossil-fuel use in the Northern Hemisphere with a potential role of increased chlorine atoms which could play a role in the C₂H₆ decline. Recent studies also evaluated a negative trend in the Southern Hemisphere at Lauder (New Zealand, 45° S, 170° E) and Arrival Height (Antarctica, 78° S, 167° E) from 1997 to 2009, as well as at Wollongong (Australia, 34° S, 150° E) (Zeng et al., 2012).

For C₂H₂, our total column values and seasonal variability are in agreement with values reported at Reunion Island (France, 21° S, 55° E) from 2004 to 2011 (Vigouroux et al., 2012), and at Jungfraujoch station (Switzerland, 46° N, 8° E) from 1995 to 2008 (Mahieu et al., 2008), as well as at Ny Ålesund from 1992 to 1999 (Albrecht et al., 2002). The similarity between high-latitude and mid-latitude C₂H₂ concentrations has already been reported (Albrecht et al., 2002) and highlights the importance of transport on the C₂H₂ budget in the Arctic. However, the primary source of C₂H₂ is from transportation at mid-latitudes (Xiao et al., 2007). So the fact that our C₂H₂ total columns are comparable to those measured in the Arctic between 1992 and 1999 (i.e. no decrease is seen from previous years compared to the observed decline of CO or C₂H₆) is unclear. One hypothesis might be that C₂H₂ emissions from cars have not changed compared to the technological advances in the catalytic converters employed to reduce CO and C₂H₆ emissions.

Our HCN total columns are of the same order of magnitude, in term of absolute values and variabilities, with those reported by Notholt et al. (1997b) in the Arctic from 1992 to 1995. They also agree well with HCN columns observed at mid-latitude

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molecules in the Arctic. Concentrations and seasonal cycles of CH₃OH, and HCOOH at PEARL are comparable with previous measurements performed at remote sites. Excellent agreement is found between our HCOOH total columns and measurements from Thule (Paulot et al., 2011). However, the atmospheric concentrations of H₂CO at PEARL in the early spring are smaller than those reported in the literature. This might reflect the large atmospheric variability of H₂CO, given its small lifetime, with no local sources in winter at Eureka, and/or the effect of the updated spectroscopic parameters used in our retrievals.

To conclude, our measurements of tropospheric species at Eureka constitute a new dataset which can be used as a constraint to improve the model simulations of chemical and dynamical processes in the high Arctic. In particular, since all seven molecules are biomass burning products, they can be used to identify fire events, in order to study the chemical composition of the plumes above PEARL, and to derive emission ratios, which are key parameters needed to simulate fire emissions in atmospheric models.

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Table 1. Sources, sinks and average lifetimes of all the target species. BB, VOC and NMHCs are the acronyms of Biomass Burning, Volatile Organic Compound and Non Methane Hydro-Carbons, respectively.

Target species	Name	Sources	Sinks	Lifetimes
CO	Carbon monoxide	BB, transport, steel industry, methane and VOC oxidation	reaction with OH	2 months
HCN	Hydrogen cyanide	BB, industry, fungi and plant emission	reaction with OH and ocean uptake	2–6 months
C ₂ H ₆	Ethane	BB, biofuel use, natural emission	reaction with OH	1.5 months
C ₂ H ₂	Acetylene	BB, combustion product, natural emission	reaction with OH	2 weeks
CH ₃ OH	Methanol	BB, biogenic emission	reaction with OH	5–10 days
HCOOH	Formic acid	BB, biogenic emission, photo-oxidation of NMVOCs	reaction with OH, dry and wet deposition	3–4 days
H ₂ CO	Formaldehyde	BB, oxidation of methane and NMVOCs	reaction with OH	< 2 days

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Table 2. Parameters (microwindows, interfering species, a priori VMR sources, standard deviations (STD) of the a priori covariance matrix, and signal-to-noise ratios – SNR) used in the retrievals of the seven target gases.

Target species		Microwindows	Interfering species	a priori VMR	STD for S_a (%)	SNR
Carbon monoxide	CO	2057.684–2058.000, 2069.560–2069.760, 2157.507–2159.144	O ₃ , CO ₂ , OCS, H ₂ O, N ₂ O	WACCM V6	20	85
Hydrogen cyanide	HCN	3268.000–3268.380, 3287.000–3287.480	H ₂ O	WACCM V6	50	200
Ethane	C ₂ H ₆	2976.660–2976.950, 2983.200–2983.550, 2986.500–2986.950	H ₂ O, O ₃	WACCM V4.5	30	250
Acetylene	C ₂ H ₂	3250.430–3250.770, 3255.180–3255.725, 3304.825–3305.350	H ₂ O, HDO	GC Toon Kiruna991203 Mk4-flight 6–34 km, outside spitprim.set, divided by 2	50	200
Methanol	CH ₃ OH	992.000–998.700, 1029.000–1037.000	O ₃ , O ₃ ⁶⁸⁶ , O ₃ ⁶⁸⁸ , O ₃ ⁶⁷⁶ , O ₃ ⁶⁶⁷ , H ₂ O, CO ₂	WACCM V6	20	200
Formic acid	HCOOH	1104.650–1105.600	HDO, O ₃ , O ₃ ⁶⁸⁸ , O ₃ ⁶⁷⁶ , H ₂ O, NH ₃ , CCL ₂ F ₂ , CHF ₂ Cl, CH ₄	WACCM V6	100	800
Formaldehyde	H ₂ CO	2778.120–2778.800, 2780.600–2781.170	CH ₄ , CO ₂ , O ₃ , N ₂ O	WACCM V6 divided by 2	100	500

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Five years of CO, HCN, C₂H₆, C₂H₂, CH₃OH, HCOOH, and H₂CO total columns

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Table 3. Error budgets, in percentage, as a function of the different sources of random and systematic uncertainties on typical CO, HCN, C₂H₆, C₂H₂, CH₃OH, HCOOH, and H₂CO total columns retrieved at PEARL. DOFS and SZA are the acronyms of Degrees of Freedom for Signal and Solar Zenith Angle, respectively.

Error budget (%)	CO	HCN	C ₂ H ₆	C ₂ H ₂	CH ₃ OH	HCOOH	H ₂ CO
<i>N</i> spectra retrieved in timeseries	3980	1815	1819	1269	1095	1973	1242
DOFS	2.5	1.6	2.0	1.1	1.2	1.1	0.8
Random errors:							
Measurement error	0.2	4.3	1.2	7	4.6	16.3	10
Uncertainty on temperature	0.7	0.4	0.3	2.9	4.4	2.5	5.5
Uncertainty on retrieval parameters	0.3	4.3	2.6	8.2	0	0	1.1
Uncertainty on interfering species	0	0.2	0	0.1	0	0.3	24.1
Uncertainty on SZA	0.3	0.3	0.3	0.7	0.2	0	0.2
Total random error	0.8	6.1	2.9	11.6	6.3	16.5	26.7
Systematic errors:							
Uncertainty on line intensity	2.9	4.9	6.4	5.7	10	3.6	2.4
Uncertainty on line width	0.4	3.7	2.4	7.7	1.1	0.9	2.1
Total random and systematic error	3.1	8.7	7.4	15.1	11.9	17	26.9
Smoothing error	0.2	5.9	12.2	16.7	3	1.5	5.7
Total error (random + systematic + smoothing)	3.1	10.5	14.3	22.5	12.3	17	27.5

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Table 5. Comparison of FTIR and ACE-FTS partial columns for all trace gases derived in this study. N represents the number of coincidences involved in the comparison, R is the correlation coefficient, and the fifth column gives the values of the slope of the regression plot between the FTIR partial columns and the coincident ACE-FTS measurements along with the 1σ uncertainties of the slopes, for each target species. The second column shows the mean distance of ACE-FTS occultations to PEARL, and the last column shows the mean altitude ranges used for the partial column calculations.

	N distances (km) to PEARL	Mean	R (FTIR vs ACE-FTS)	Slope	Mean altitude range for partial columns (km)
CO	106	317	0.66	0.97 ± 0.11	9.4–48.6
HCN	93	313	0.96	0.69 ± 0.02	7.9–33.2
C ₂ H ₆	17	436	0.97	0.71 ± 0.04	8.0–19.3
C ₂ H ₂	93	319	0.78	1.21 ± 0.10	8.0–17.0
CH ₃ OH	6	316	0.91	0.74 ± 0.14	9.4–17.5
HCOOH	103	317	0.56	3.35 ± 0.49	8.1–18.5
H ₂ CO	6	317	0.75	0.50 ± 0.22	9.1–38.7

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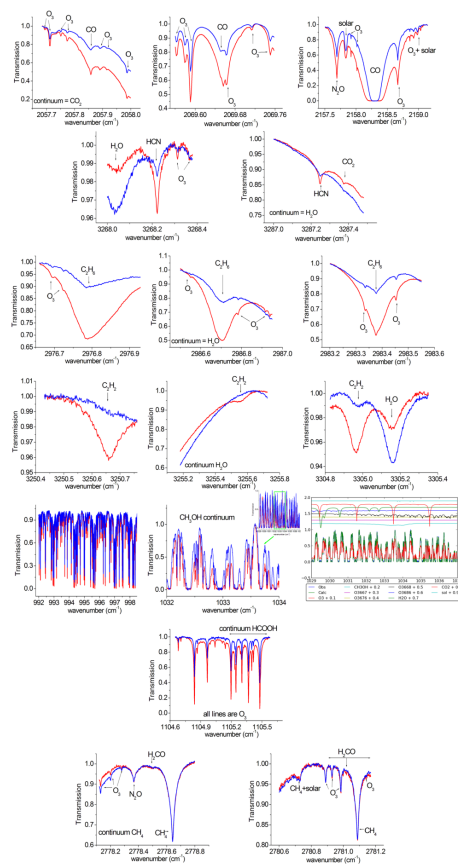


Fig. 1. Microwindows used in the retrievals of CO, HCN, C₂H₆, C₂H₂, CH₃OH, HCOOH, and H₂CO, with examples of clear and polluted conditions in blue and red, respectively.

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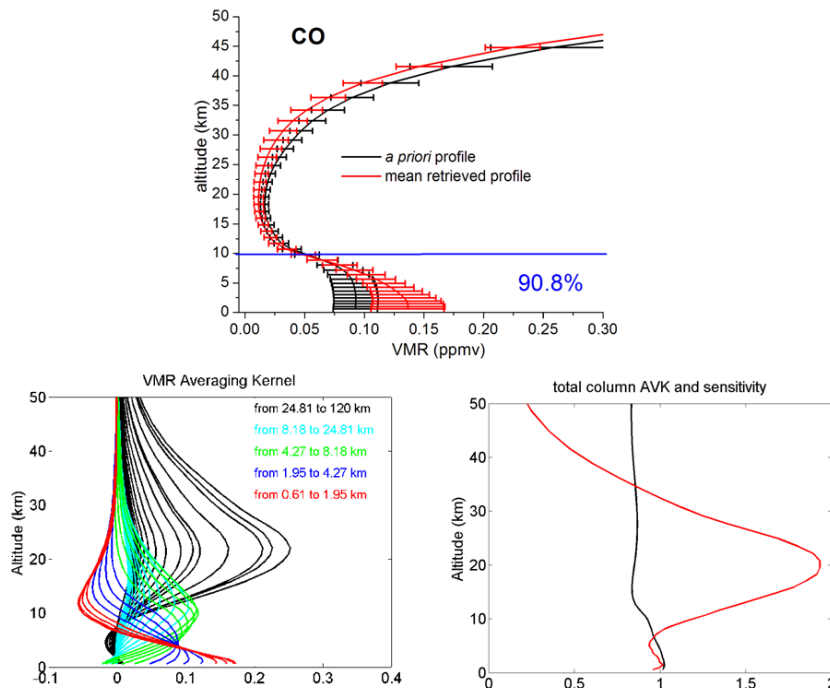


Fig. 2. The CO a priori VMR profile from WACCM v6 in black (upper panel). The black error bars represent the standard deviation of the a priori covariance matrix used in the retrievals, at each layer. The mean retrieved profiles is in red, with error bars corresponding to the 1σ standard deviation around the mean. The number in blue is the percentages of the tropospheric column contributions (between 0.6 and 10.7 km) relative to the total column (between 0.6 and 120 km). Typical VMR averaging kernels in VMR/VMR (lower left panel), total column averaging kernels in $(\text{molecule cm}^{-2})/(\text{molecule cm}^{-2})$ (lower right panels, black line), and sensitivity profiles (right panel, red line) as a function of altitude. The colours correspond to averaging kernels at altitudes lying in a partial column for which the DOFS is about 0.5.

Five years of CO, HCN, C₂H₆, C₂H₂, CH₃OH, HCOOH, and H₂CO total columns

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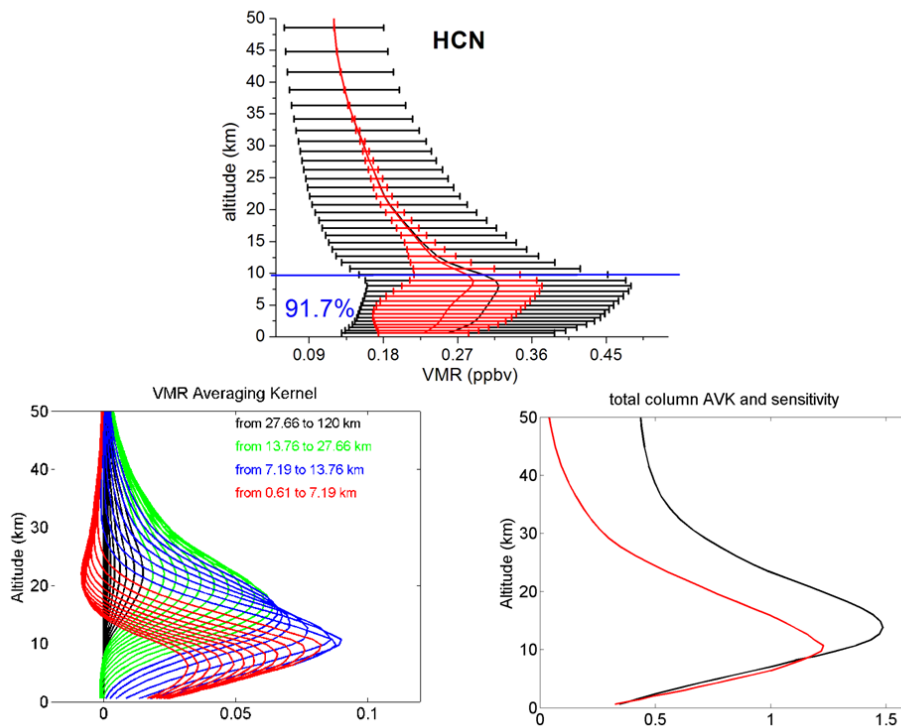


Fig. 3. Same as Fig. 2 but for HCN.

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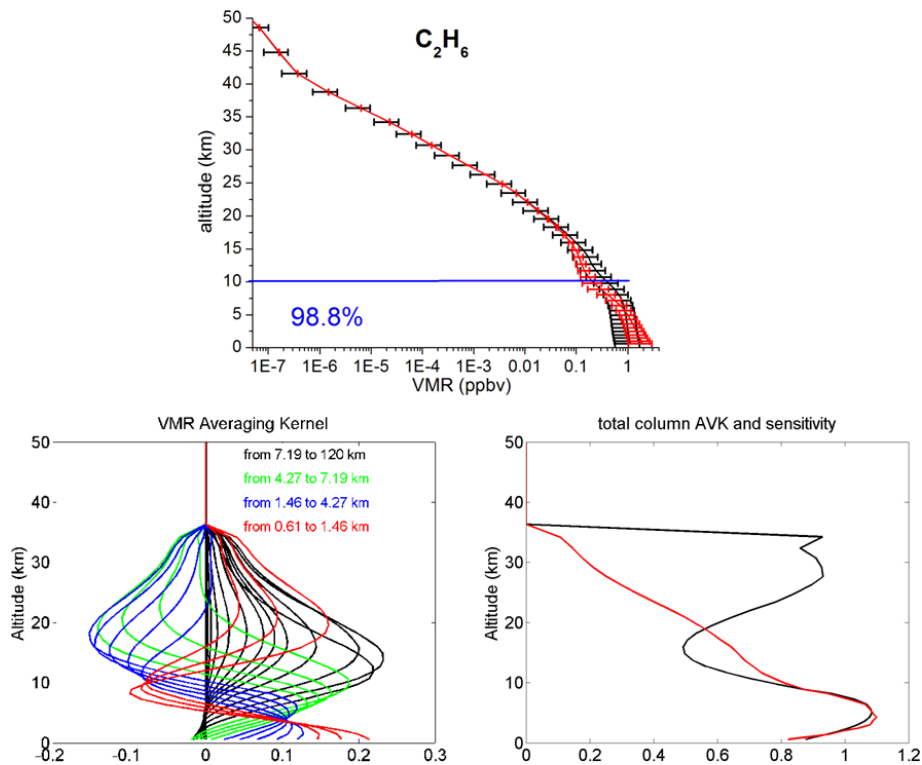


Fig. 4. Same as Fig. 2 but for C₂H₆. The C₂H₆ a priori VMR profile is from WACCM v4.5.

Five years of CO, HCN, C₂H₆, C₂H₂, CH₃OH, HCOOH, and H₂CO total columns

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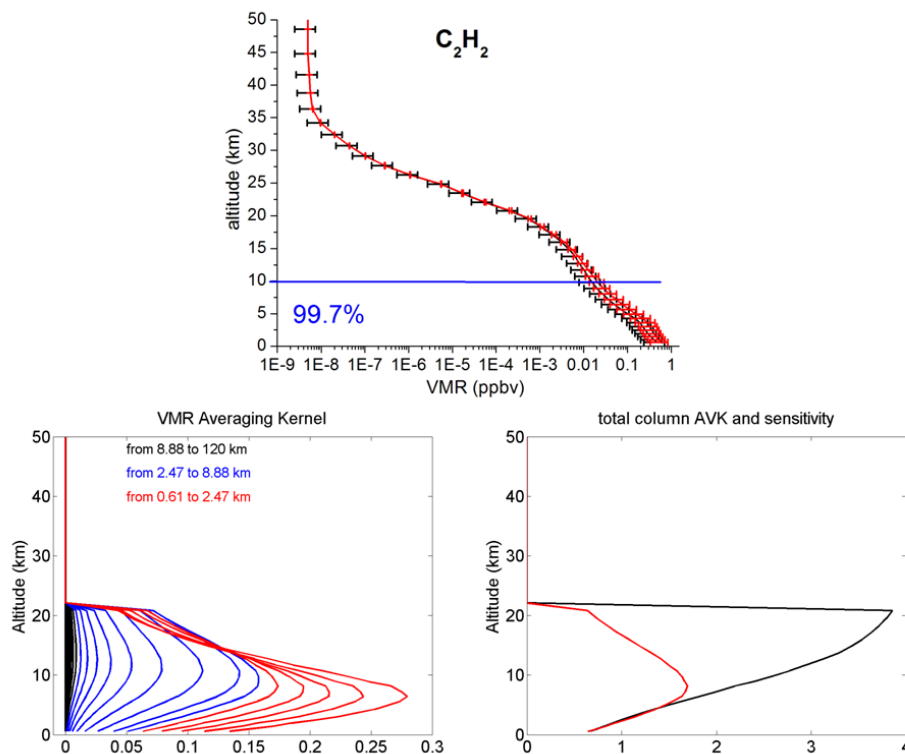


Fig. 5. Same as Fig. 2 but for C₂H₂. The C₂H₂ a priori VMR profile is from MkIV balloon measurements made at the high-latitude NDACC site of Kiruna, between 6 and 34 km altitude, combined with spitprim.set (Ft. Sumner MkIV flights, 1990s, <http://mark4sun.jpl.nasa.gov/m4.html>).

Five years of CO, HCN, C₂H₆, C₂H₂, CH₃OH, HCOOH, and H₂CO total columns

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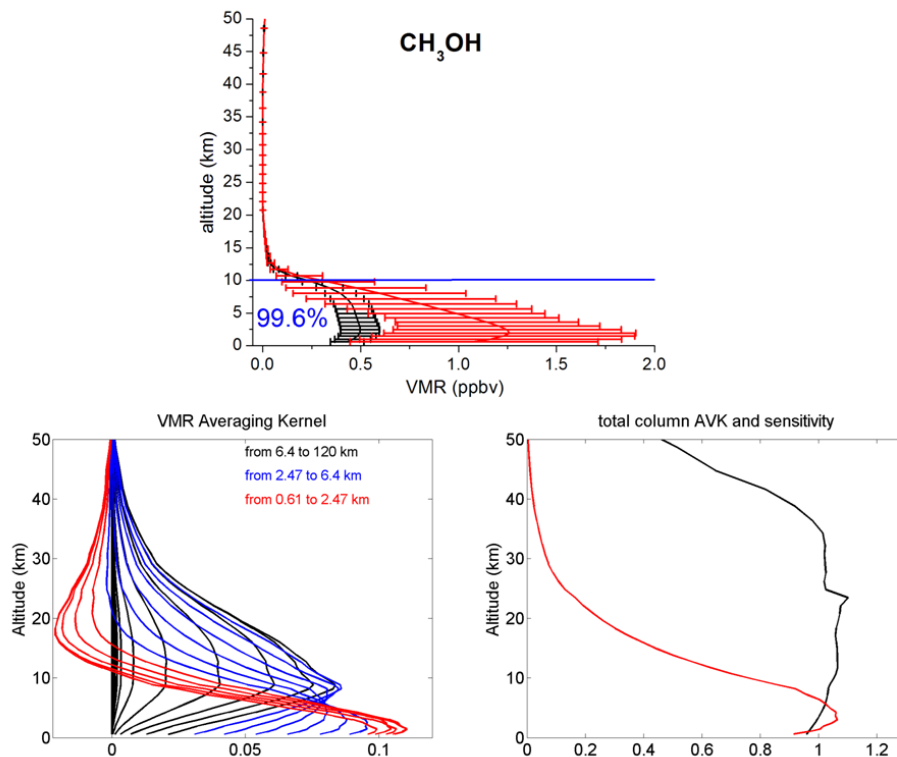


Fig. 6. Same as Fig. 2 but for CH₃OH.

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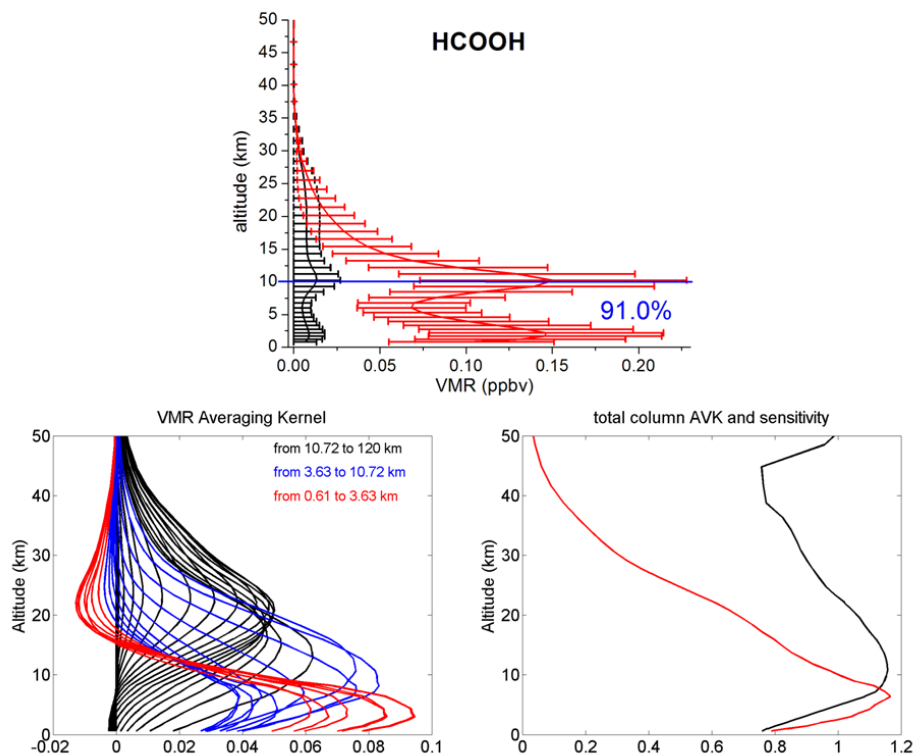
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**Five years of CO,
HCN, C₂H₆, C₂H₂,
CH₃OH, HCOOH, and
H₂CO total columns**

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**Fig. 7.** Same as Fig. 2 but for HCOOH.

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CH₃OH, HCOOH, and
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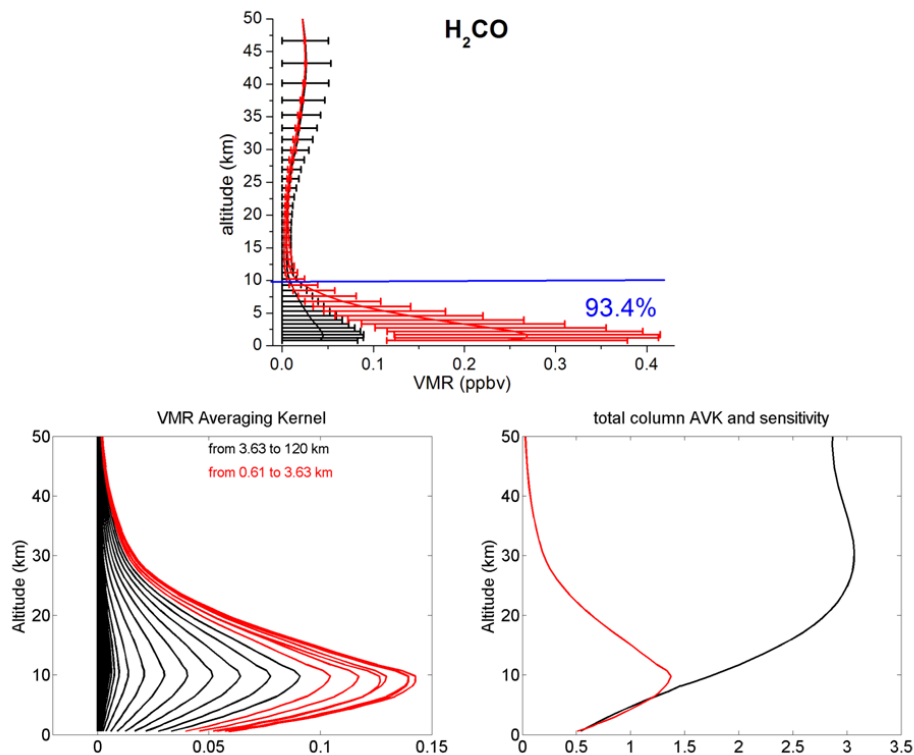


Fig. 8. Same as Fig. 2 but for H_2CO . The H_2CO a priori VMR profile is from WACCM v6 divided by 2.

**Five years of CO,
HCN, C₂H₆, C₂H₂,
CH₃OH, HCOOH, and
H₂CO total columns**

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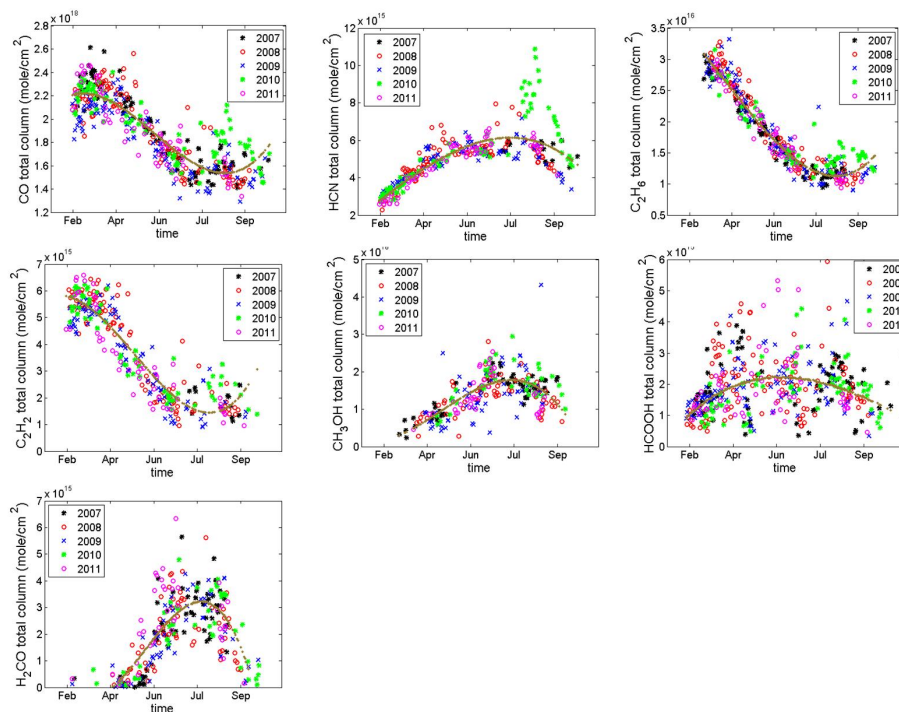


Fig. 9. Daily mean total columns (in molecules cm^{-2}) of CO, HCN, C₂H₆, C₂H₂, CH₃OH, HCOOH, and H₂CO, plotted in different colours for the different years of measurements. The brown line shows the polynomial fits to the data.

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