

**A new AMS single
particle classification
algorithm**

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**Quantitative single particle analysis with
the Aerodyne aerosol mass spectrometer:
development of a new classification
algorithm and its application to field data**

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Abstract

Single particle mass spectrometry has proven a valuable tool for gaining information on the mixing state of aerosol particles. With the Aerodyne aerosol mass spectrometer (AMS) equipped with a light scattering probe, non-refractory components of sub-micron particles with diameters larger than about 300 nm can even be quantified on a single particle basis. Here, we present a new method for the analysis of AMS single particle mass spectra. The developed algorithm classifies the particles according to their components (e.g., sulphate, nitrate, different types of organics) and simultaneously provides quantitative information about the composition of the single particles. This classification algorithm was validated by applying it to data acquired in laboratory experiments with particles of known composition, and applied to field data acquired during the MEGAPOLI summer campaign (July 2009) in Paris. As shown, it is not only possible to directly measure the mixing state of atmospheric particles, but also to directly observe repartitioning of semi-volatile species between gas and particle phase during the course of the day.

1 Introduction

A mixture of atmospheric aerosol particles can comprise single particles of vastly different chemical composition, size and properties, depending on the particles' individual sources and atmospheric processing (Pöschl, 2005). Especially in order to gain direct information on the mixing state of these particles, the chemical composition of the individual particles needs to be analysed. Also particle types present only in low number might not be detected by analysis of the whole particle ensemble, but only by single particle analysis. Though rare, they nonetheless could be of interest, e.g. because they are characteristic for a specific source.

In order to obtain information on the composition of individual particles with high time resolution, currently especially laser desorption/ionization (LDI) single particle

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mass spectrometers are being used (Johnston and Wexler, 1995; Noble and Prather, 2000; Hinz and Spengler, 2007). These are able to analyse both refractory and non-refractory particle components. The drawback of these instruments is that they are typically not able to provide quantitative information about the particle composition (Noble and Prather, 2000). Quantitative measurements of ensembles of non-refractory particles, on the other hand, can be obtained with the Aerodyne aerosol mass spectrometer (AMS), which applies a thermal desorption/electron impact ionization measurement scheme. It was originally designed especially for the on-line analysis of ensembles of submicron particles (Jayne et al., 2000). In conjunction with a time-of-flight (ToF) mass spectrometer, also with the AMS single particle measurements have become possible, though only with very low analysis duty cycle (Drewnick et al., 2005). This duty cycle was greatly improved with the introduction of the light scattering probe (LSP) into the ToF-AMS (Cross et al., 2009), which enables the quantitative measurement of the chemical composition of non-refractory, submicron particles on a single particle basis with the AMS. The lower 50 % cut-off for optical detection typically here is in the order of about 300 nm particle diameter (Liu et al., 2013).

Like for single particle mass spectra acquired with LDI instruments (Hinz and Spengler, 2007), the data analysis of single particle mass spectra acquired with the AMS is not straightforward. Few attempts have been made so far to analyse AMS single particle mass spectra from ambient air sampling (Cross et al., 2009; Liu et al., 2013). In both cases, the standard fragmentation pattern table (Allan et al., 2004), which was developed to deconvolute the mass spectral contribution of nitrate, sulphate, ammonium, chloride and organics to ensemble AMS mass spectra, was applied to the single particle mass spectra. By this means, the mass spectra of only the organic content of the individual particles were separated. This approach has certain drawbacks: the fragmentation pattern table was developed especially for an ensemble of ambient particles, i.e. it assumes the measured ensemble to always consist of the five aforementioned components, with time-varying relative contributions. While this is applicable in most cases to ensemble data, this cannot be expected per se from single particles, which

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may consist only of one or some of these constituents. Furthermore, with the fragmentation pattern table the relative signal contribution of different species at various mass to charge ratios m/z are calculated from signal intensities at other m/z (Allan et al., 2004). This is appropriate for ensemble mass spectra with good signal-to-noise-ratio, but adds further uncertainty in the mass contributions obtained for the single particle mass spectra. Also the organics fragmentation pattern obtained by this means might be more or less disturbed. This can be problematic especially for particle mass spectra with very unusual fragmentation patterns, which thereby might be easily missed or misinterpreted.

To gain additional information on the organic mass fraction of individual particles, different methods were adopted in the aforementioned publications. Cross et al. (2009) applied a method derived from principal component analysis (Zhang et al., 2005), Liu et al. (2013) used k means clustering (Bishop, 2006) to classify different types of organic aerosol mass spectra. Both methods provided similar results, in that they separated OOA-like (oxygenated organic aerosol) and HOA-like (hydrocarbon-like organic aerosol) mass spectra. Liu et al. (2013) furthermore separated two subtypes of OOA that were interpreted to be more and less volatile due to slight differences in the cluster centroid mass spectra. Both methods, however, have the inherent problem of not yielding comparable results between different datasets. For example, an “HOA-like” cluster mass spectrum of one dataset might look different from an “HOA-like” cluster mass spectrum retrieved for another dataset, and it is within the subjective decision of the user to decide whether they are similar enough to comprise the same type of organic matter. In addition, both methods were not able to identify particles present only in low number fraction. The method used by Cross et al. (2009) is only capable of identifying HOA- and OOA-content, and will not provide information on any other particle type. With k means clustering as applied by Liu et al. (2013), in principle it should be possible to identify also new, so far unknown particle types, as the clustering is not constrained to any specific mass spectral patterns. However, if these particles are occurring only in low number fraction (depending on the specific dataset), they might easily be mixed

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into the clusters dominated by other particle types, and thereby be missed. Furthermore, for our datasets presented here we found that this method is typically not able to separate particles with the same components but in different degrees of mixing, and therefore can only provide limited information about the particles' mixing state.

5 Here, we present a new method to reliably identify and also quantify components within single particle mass spectra acquired with the AMS. The algorithm yields unambiguous results, which are comparable between different datasets, and gives direct information on the quantitative composition of the individual particles and their mixing state. Furthermore, the algorithm is able to recognize individual particles of specific
10 types within the dataset, and it can be easily adapted to identify new, so far unknown particle types. The algorithm was validated using mass spectra of laboratory-generated particles of known composition, and applied to single particle mass spectra acquired during ambient measurements in Paris, France, in summer 2009.¹

2 Materials and methods

15 2.1 Instrument details

A C-ToF-AMS (Compact ToF-AMS, Aerodyne, Inc.; Drewnick et al., 2005) equipped with an LSP as described in detail in (Cross et al., 2007; Cross et al., 2009; Liu et al., 2013) was used in this work. In short, the particle beam entering the instrument is chopped at its front end by a spinning wheel with two slits to determine the starting point
20 of the measurement. The light scattering probe is mounted at a distance of 0.265 m from the chopper. It consists of an unfocused continuous wave laser (Crystalaser, model BCL-050-405, 405 nm, 50 mW) and an elliptic mirror, which focuses scattered light from particles passing the laser beam onto a photomultiplier tube. At a distance of 0.395 m from the chopper, the vaporizer is situated. Here, non-refractory material

¹ The content of this paper has, in different form, been published previously as part of a PhD thesis (Freutel, 2012; in German).

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is flash-vaporized and consecutively ionized by electron impact, and the resulting ions are orthogonally extracted into the ToF mass spectrometer. During measurement in light scattering mode, time-resolved mass spectra are recorded (here: 250 per chopper cycle), and if a particle is detected during one chopper cycle at the photomultiplier tube of the LSP, the time-resolved mass spectra of the whole chopper cycle are saved along with the light scattering signal.

To retrieve single particle mass spectra from the measured raw data, the AMS single particle data analysis software Sparrow, version 1.04 A (Sparrow, 2013) was used, modified and extended within this work. Sparrow as well as the classification algorithm described in this work have been developed within the framework of the scientific calculation software tool IGOR Pro, version 6.22 A (Wavemetrics, Inc.).

For all measurements, single ion area (Jayne et al., 2000) was determined for the mass spectrometer on a regular basis to be able to convert the raw measured signal into number of ions. On-board AP240 thresholding was used during all measurements with a threshold of 4–5 bits. During all measurements, the AMS was operated with a vaporizer temperature of $\sim 600^\circ\text{C}$; only during the first two weeks of the MEGAPOLI summer campaign (see below), it was set to $\sim 800^\circ\text{C}$. No indication was found from comparison to co-located instruments that this difference in vaporizer temperature caused any differences in fragmentation patterns or mass concentration measurements (Freutel et al., 2013).

In all measurements the standard AMS aerodynamic lens was used, which has its lower and upper 50% cut-off at particle sizes of about 70 and 750 nm vacuum-aerodynamic diameter, respectively (Liu et al., 2007). For LSP detection efficiency, the lower 50% cut-off particle size was at ~ 380 nm volume-equivalent diameter for the measurements within this work. The detection efficiency reached about 100% for particles ≥ 400 nm. The measured number distribution in this size range agreed well with that from co-located measurements of an optical particle counter during the MEGAPOLI 2009 campaign (see below). All conclusions drawn from AMS single

particle data in this work (e.g. on number concentrations or average mass per particle) are based on this size range.

2.2 Retrieval of single particle mass spectra

Each single particle dataset consists of the time-resolved light scattering signal and the time-resolved mass spectra, both acquired during one individual chopper cycle. In order to extract the single particle mass spectrum from these mass spectra, the method described by Cross et al. (2009) was slightly modified: first, for each chopper cycle the temporal evolution of the total ion current (TIC), calculated as the sum of all significant signals but m/z 18, 28, and 32, was determined. (m/z 18, 28 and 32 were excluded due to high background signal from water, N_2 and O_2 .) “Significant” here means a maximum intensity of more than three times the standard deviation of the background regions, i.e. the mass spectra at the beginning and the end of the chopper cycles. The time of detection of the particle in the mass spectrometer is identified from the maximum in the TIC signal. As ions from this single particle evaporation event are distributed over several mass spectra around this maximum, all these mass spectra need to be summed to obtain the overall mass spectrum of the single particle. To do so, the summation region is determined around the TIC maximum as all mass spectra that contain significant signal, including at most five data points in each direction. If this summation region contains at least two mass spectra, all m/z but 18, 28 and 32 are summed over the corresponding consecutive mass spectra, and the corresponding background (average of mass spectra from the background regions) is subtracted. Ion duty cycle correction (Drewnick et al., 2005) and single ion area are applied to convert the signal of the resulting mass spectrum into ion duty cycle corrected number of ions.

It was found that these integrated mass spectra on average contain about 80 % of the total measured mass from the respective single particle event. The remaining 20 % are contained within the tailing of the TIC peak (slow vaporization) and are not regarded in the determination of the summation region due to their insignificant signal.

Of all retrieved mass spectra, only those are used for the further analysis for which:

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- the flight time measured between chopper and detection in the mass spectrometer fits within 20 % to the expected arrival time calculated from the flight time between chopper and detection of the light scattering signal,
- the sum of (positive and negative) signals in the single particle mass spectrum is at least 5 ions, and a minimum signal of 5 ions is found at least at one m/z between m/z 18 and 200 (both without regarding signal at m/z 18, 28, 32),
- in the light scattering signal, no indication for coincident particles is found (Liu et al., 2013).

2.3 Particle generation

For the *laboratory measurements* (Table 1), an aqueous solution (emulsion in the case of oleic acid) of the respective substance or mixture of substances was nebulized (nebulizer model 3075, TSI Inc.). The aerosol was dried using two consecutive silica gel diffusion driers, and led through 1/4" stainless steel tubing to a differential mobility analyser (model 3080, TSI Inc.) for size selection. From there, the aerosol flow was split and led in parallel to a condensation particle counter (Grimm, model 5.401) and the C-ToF-AMS.

Source measurement. For the measurement of engine exhaust particles, exhaust of an electric power generator (SDMO, SH 15 000 TE) fuelled with gasoline was probed by measuring directly in the exhaust plume without any further inlet lines in ambient air. The inlet of the AMS was in about 1 m distance to the generator. To exclude mass spectra of the ambient aerosol background, all mass spectra were manually inspected. Only those which clearly resembled literature mass spectra of engine exhaust (Canagaratna et al., 2004; Schneider et al., 2006) were used for further analysis.

Field measurements. Particles from ambient air were measured in a north-eastern suburb of Paris, France, during the MEGAPOLI summer campaign (1–31 July 2009). Details about this campaign, the sampling of the C-ToF-AMS, and comparison to co-located measurements can be found in (Freutel et al., 2013). For this dataset, from

107 734 single particle events, 26 123 single particle mass spectra could be retrieved after applying the criteria described above. The major reason (in about 90 % of all cases) for particle events to being excluded was low ion signal.

It was found that the shape of the number size distribution of those particles for which mass spectra were retrieved was very similar to that of all particles detected by the LSP. Only the largest particles were slightly underestimated, likely due to increasing bounce-off from the vaporizer, as also found e.g. for impactors (Hinds, 1999). For a comparison of mass size distributions, the time-resolved TIC of all single particle events was used. The summed time-resolved TIC of all single particle events, after applying the necessary conversion factors, was in agreement with ensemble size distribution measurements in the range of optimum LSP detection efficiency. Also the sum of the time-resolved TIC of all single particles was compared to that of only the particles for which mass spectra were retrieved. This comparison showed that only about 20 % of the total measured mass was not included within the particle events used for the further analysis. All in all, the time-resolved TIC from all particles was comparable to that of only the particles used in the analysis. Only for the largest particle sizes, a slight underestimation of the retrieved particle mass compared to the whole measured mass was found, due to delayed vaporization events which are not included in the analysis (see above). From these comparisons, it therefore is assumed that the analysed subset of data is representative of the whole particle population, and that the majority of particles not included in the analysis due to missing mass spectra are of comparable composition to the ones investigated.

In order to obtain typical single particle mass spectra of oxygenated organic aerosol (OOA), from all single particle mass spectra sampled those resembling internal mixtures of OOA, nitrate and/or sulphate were extracted as follows: for each single particle mass spectrum, the signal at m/z 30 and 46 was summed as a proxy for the nitrate content, and the signal at m/z 48, 64, 80, 81, and 98 as a proxy for the sulphate content. The signal at the remaining m/z (without m/z 16, 17, 18, 28, and 32) was summed as a proxy for the content of organics. A potential interference of chloride with

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this calculated organic content was considered insignificant as its average mass concentration was negligible during this campaign (Freutel et al., 2013). From these values, for each single particle mass spectrum the relative contribution of organics to the total ion signal was calculated. All single particle mass spectra containing similar organic contributions were averaged. Inspection of these average single particle mass spectra showed that for particles with less than 80 % organic content (by ion number), the average mass spectrum resembled that of an internal mixture of OOA and inorganics. For particles with higher organic content, the average mass spectrum resembled pure hydrocarbon-like organic aerosol (HOA; examples of mass spectra from both types of organic aerosol can be found e.g. in Zhang et al., 2011). Therefore, all particles with less than 80 % of organic content were classified as “OOA-containing”. Out of those, 10 % were randomly chosen for the development and testing of the classification algorithm described in Sect. 3.1. To ensure sufficient signal-to-noise-ratio, only mass spectra with absolute organics signal of at least 50 ions were used, which resulted in a total of 822 single particle mass spectra used for the further analysis described in Sect. 3.1.

3 Results and discussion

3.1 Development and validation of a classification algorithm

3.1.1 Working principle of the classification algorithm

From the single particle measurements listed in Table 1, average reference single particle mass spectra of ammonium nitrate and sulphate, glucose, oleic acid, engine exhaust, and OOA (internally mixed with ammonium nitrate and sulphate) were determined. Ammonium nitrate and sulphate are commonly found in atmospheric aerosol, as well as engine exhaust organic aerosol and OOA (Zhang et al., 2007). Glucose, the building element of cellulose, is used here as a proxy for cellulose-derived sugars

with each subtraction, an arbitrary value was chosen. These R^2 values might still be optimized by validation with more laboratory datasets.

For each component identified, the corresponding number of ions subtracted from the mass spectrum (excluding m/z 14–17 to avoid influence of ammonium and water) is saved as quantitative information regarding the contribution of this substance to the total mass spectrum measured (Fig. 2). Compounds are only considered detected if their quantitative contribution exceeds a number of at least five ions, such that the relative counting statistics uncertainty is below 50%. At the end of this procedure, if no further component is detected, the remaining signal is saved as residual mass spectrum.

Calculation of mass per particle

Similar to the calculation of AMS ensemble mass concentrations from the measured number of ions as described e.g. in (Jimenez et al., 2003), the retrieved number of ions I_{ions} of the different components can be converted into the mass per particle of these species (M_{species}) using the equation

$$M_{\text{species}} = \frac{I_{\text{ions}} \cdot S_{\text{ions, species}} \cdot \text{MW}_{\text{Nitrate}}}{N_A \cdot \text{RIE}_{\text{species}} \cdot \text{IE}} \quad (1)$$

$\text{MW}_{\text{Nitrate}}$ is the molecular weight of nitrate (62 g mol^{-1}), N_A Avogadro's number ($6 \times 10^{23} \text{ mol}^{-1}$), $\text{RIE}_{\text{species}}$ the relative ionization efficiency of the respective species, and IE the ionization efficiency of nitrate which is determined routinely in laboratory calibrations using ammonium nitrate. $S_{\text{ions, species}}$ is a scaling factor for the respective species which accounts for the fact that m/z 14–18, 28, and 32 are not included for the calculation of I_{ions} from the single particle mass spectra due to the influence of high background signal at these m/z . Using the average reference mass spectra, the fractional contribution of mass spectral signal at the neglected m/z can be calculated. From this,

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the resulting scaling factors $S_{\text{ions, species}}$ were found to be 1 for nitrate, and 1.1 for sulphate and for organic species. The latter was calculated as average scaling factor from those derived for oleic acid and glucose. For the calculation of $S_{\text{ions, glucose}}$, the signal at m/z 16, 17 and 18 in the glucose mass spectrum was calculated from the mass spectral signal at m/z 44 (analogous to the ensemble fragmentation pattern table) to account for the presence of water of crystallization. A conservative estimate of the uncertainty of $S_{\text{ions, species}}$ (Δ_{scaling}) is 10 %, determined from the variation observed for the organic substances investigated. To further reduce this uncertainty, $S_{\text{ions, organics}}$ should be determined for a larger variety of different organic substances.

$\text{RIE}_{\text{species}}$ are the relative ionization efficiencies also used for the calculation of ensemble mass concentrations of nitrate, sulphate and organics (standard $\text{RIE}_{\text{species}}$ values: 1.1, 1.2, and 1.4, respectively; Canagaratna et al., 2007). For the specific instrument employed in this work, an $\text{RIE}_{\text{sulphate}}$ of 0.76 and an $\text{RIE}_{\text{organics}}$ of 1 was used. These values were derived from calibration measurements using NH_4NO_3 and $(\text{NH}_4)_2\text{SO}_4$ in the case of sulphate, and from the comparison to a co-located AMS during two measurement campaigns in the case of organics (Freutel et al., 2013; Crippa et al., 2013). $\text{RIE}_{\text{organics}}$ therefore in this case is associated with a higher uncertainty (Δ_{RIE}). This uncertainty is estimated to 20 % from the range of values of $\text{RIE}_{\text{organics}}$ observed, instead of 10 % as for $\text{RIE}_{\text{nitrate}}$ and $\text{RIE}_{\text{sulphate}}$. If the $\text{RIE}_{\text{species}}$ value of a certain species is not known, $\text{RIE}_{\text{species}}$ is set to 1, and a nitrate-equivalent mass of this species is obtained.

Uncertainty estimate of calculated mass per particle

The uncertainty Δ_{mass} of the different substances' relative mass contributions to the single particle mass spectrum can be estimated as

$$\Delta_{\text{mass}} = \sqrt{\Delta_{\text{scaling}}^2 + \Delta_{\text{MS-retrieval}}^2 + \Delta_{\text{RIE}}^2} \quad (2)$$

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with $\Delta_{\text{MS-retrieval}}$ the relative uncertainty for the subtraction of the reference mass spectra from the single particle mass spectra (i.e., the uncertainty in the determination of $S_{\text{MS-retrieval}}$). This uncertainty is dominated by counting statistics. $\Delta_{\text{MS-retrieval}}$ can be estimated from the standard deviation of average residuals after the subtraction of the reference mass spectra from the single particle mass spectra of pure substances (ammonium nitrate, ammonium sulphate, glucose, and oleic acid) to about 20 % for a single particle mass spectrum. For a larger number of single particle mass spectra n , $\Delta_{\text{MS-retrieval}}$ is scaling with $1/\sqrt{n}$.

Note that Δ_{mass} is the overall uncertainty of the *relative* mass contribution. Δ_{mass} can be estimated using Eq. (2) to 25 % for inorganics and 30 % for organics. For a larger number of mass spectra, the uncertainty is reduced to about 15 % for inorganics and 25 % for organics. The absolute mass concentration is subject to additional uncertainties (e.g. the uncertainty of IE).

3.1.2 Application to laboratory generated mass spectra

Validation of classification

For validation purposes, the classification algorithm was applied to the single particle mass spectra of known composition detailed in Table 1. The resulting classification is shown in Fig. 3. For each particle type, the classifications assigned by the algorithm are divided into three separate bars: one shows the fraction of correctly assigned particles (left bar), one the fraction of particles for which one or several components could not be identified (false negatives, middle bar), and one the fraction of particles for which one or several components were erroneously identified (false positives, right bar).

For pure substances (particles with 400 nm mobility diameter) and particles from a single source, generally more than 90 % of all particle mass spectra were correctly assigned; the number of false positive detections typically was very low (< 1 %). The only exception were the engine exhaust mass spectra, for which only about 80 % of all mass spectra were correctly assigned; the remaining mass spectra were to a large

part incorrectly assigned as “oleic acid”. This is due to the fact that both oleic acid and engine exhaust mass spectra exhibit typical patterns of hydrocarbon-like organic aerosol (R^2 for correlation of reference mass spectra is 0.76), so a distinction between these two species is difficult.

For mixtures of different substances, the fraction of completely correctly assigned particle mass spectra was decreasing with increasing number of components: for mixtures of nitrate and sulphate, about 80 %, and for mixtures of nitrate, sulphate and glucose, about 60 % of all mass spectra were correctly assigned to all components. The remaining fraction of particles for the most part consisted of false negatives. For all laboratory measurements, particles of the same size (400 nm mobility diameter) have been used. Therefore, an increasing number of components in the particles led to a decreasing absolute signal for the individual components, which complicates their detection. This demonstrates the limitation of detection due to counting statistics, since the retrieved number of ions in AMS measurements scales linearly with the number of molecules per particle (Jayne et al., 2000). Further work is needed to determine estimates of the limits of detection for the different compounds.

For the particles consisting of a mixture of OOA, ammonium nitrate and sulphate, the exact composition is not known as they are retrieved from ambient aerosol sampling (see Sect. 2). Therefore, all particles were classified as assigned correctly for which at least one of these components was identified by the algorithm, which is the case for about 90 % of all particles (Fig. 3). In about 60 % of all particle mass spectra, all three components were detected. Only for a small fraction of particles (< 8 %), other components than OOA, nitrate and sulphate were found; these are (short of knowing the exact composition) considered false positives in Fig. 3.

Pure succinic acid particles were also tested. This particle type was not integrated in the classification algorithm, and is used here as an example of an “unknown” particle type which should not be classified by the algorithm. Indeed, no false positives are found for this “alien” particle type, demonstrating the ability of the algorithm to also separate hitherto unknown particle types.

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Validation of quantification

The correctness of quantification was investigated through the comparison of single particle and ensemble measurements. For this, the measurements of laboratory generated, internally mixed particles of ammonium nitrate and sulphate, and of ammonium nitrate, ammonium sulphate and glucose were used. At the investigated particle size of 400 nm mobility diameter, all particles of the same type can reasonably be assumed to have the same relative composition. For both measurements, the average relative composition was retrieved from the single particle mass spectra using the described algorithm and compared to the composition measured simultaneously in ensemble mode (Table 2). Since ensemble and single particle measurements are not directly comparable, different methods for the calculation of the relative composition obtained from single particle measurements were evaluated:

- ratio of averages,
- average ratios, and
- Gaussian fit to the distribution of ratios observed.

For the ratio of averages, averages were determined

- for only a subset of particles sampled – namely only for particles in which both substances were identified whose ratio is calculated (termed “identified” in Table 2), and
- for all particles of the respective dataset (termed “all” in Table 2). In this case, if a substance was not detected within a particle, the content is assumed to be zero.

These two methods give slightly different results. Compared to the results for only the “identified” particles, when including “all” particles, the average content is underestimated strongly for sulphate and slightly for glucose. This is due to the fact that all particles with content below a threshold of 5 ions are assumed to contain none of the

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respective substance (compare Figs. 1 and 3). This lowers the calculated ratio of sulphate to nitrate and of sulphate to glucose, and increases the ratio of nitrate to glucose (Table 2). This demonstrates the limitation of quantification on a single particle basis due to the threshold of minimum number of ions for positive detection. Further work is needed to determine the limits of quantification for the different substances in detail.

Also the average ratio and Gaussian fit to the distribution of ratios yield slightly different results (for both methods, only particles were used in which both respective substances were identified): due to the fact that the distribution of ratios is not symmetrical, the value obtained from the Gaussian fit is typically smaller than the average ratio (Table 2).

The deviation from the average value obtained from simultaneous ensemble measurements (as a measure of accuracy) varies between about 5 and 20 % for the different methods (Table 2). In this comparison, the uncertainty of the single particle measurements is dominated by Δ_{scaling} (see Sect. 3.1.1 and footnote c in Table 2). Thus, a relative uncertainty of about 15 % is expected for the ratios obtained from single particle measurements, which therefore on average seem to be in agreement with ensemble mode measurements. For the different ratios, from all three methods (regarding only identified particles) it seems like the sulphate to nitrate ratio is well represented within the uncertainties, while the sulphate to glucose and nitrate to glucose ratios show some larger, systematic deviations. These seem to indicate a slight overestimation of the glucose content within the particles. This could possibly be due to an additional systematic error in the determination of the scaling factor for the reference mass spectrum, $S_{\text{MS-retrieval}}$ (see Sect. 3.1.1), which is not included in the statistical error calculated for $\Delta_{\text{MS-retrieval}}$. In further work, this potential source of error should be investigated in more detail and, if possible, quantified.

Also the average standard deviation (an indication for the scatter of values observed for the single particles around the average value; as a measure of precision) shows systematic differences between the different methods. While the standard deviation obtained from the Gaussian fit is smallest, the standard deviation of the average ratios

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is slightly larger, since also the largest ratios which do not influence the standard deviation from the Gaussian fit are included here. These standard deviations of about 45 % reflect the uncertainty for the ratio obtained from one single particle mass spectrum. This uncertainty is somewhat larger than what can be estimated from Eq. (2), and therefore likewise points to the fact that not all uncertainties (especially of $\Delta_{\text{MS-retrieval}}$) might be identified yet. The standard deviation from the ratio of averages is even larger, since here not only the distribution of the ratios plays a role, but also the distribution of the particle sizes (even though monodisperse particles were investigated, these show a distribution around the average value). Finally, the average standard deviation is largest when including all values for the ratio of averages, since here also the most extreme values (0 ions detected) are included.

This broadness of distribution shows that the quantification of substances within an individual particle is subject to high uncertainty. However, the reasonable agreement for average values from single particle with ensemble measurements presented in Table 2 shows that quantitative information regarding the average content of the individual particles can be obtained using the described algorithm. The uncertainty of this average content is mostly due to Δ_{RIE} , Δ_{scaling} and likely an additional systematic error in $\Delta_{\text{MS-retrieval}}$, which needs to be investigated in more detail.

3.2 Single particle measurements in the field

3.2.1 Application of the algorithm to field data

In Sect. 3.1.2, we have shown that the developed classification algorithm is capable of providing qualitative and quantitative information about the composition of single particles. For an application to single particle mass spectra acquired from measurements of ambient air, it has also to be accounted for the fact that particle types might exist in the atmosphere which the algorithm is not able to recognize. Therefore, the following approach for the analysis of single particle mass spectra acquired in field measurements is suggested: first, the classification algorithm is applied to the whole dataset of

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single particle mass spectra, after which ideally all spectra of unknown particle types remain unclassified. In a second step, all mass spectra which could not be classified by the algorithm are clustered either manually, or by using clustering algorithms like k means (Bishop, 2006) or fuzzy c means clustering (Bezdek et al., 1984). If by this procedure new particle types or components are found, the classification algorithm can be adapted accordingly, in order to detect these as well. The modified algorithm then is applied again to the whole dataset, and this is repeated until no new particle types can be found in the particle mass spectra not classified by the algorithm. Using this approach it is possible to detect not only known particle components, but also new, foreign particle types, even if they are present only in low number.

This approach was tested on a dataset acquired during measurements of ambient air in Paris during the MEGAPOLI summer 2009 campaign (see Sect. 2). When applying the original classification algorithm on the whole dataset of 26 123 single particle mass spectra, already 86.4 % of all mass spectra were classified into one of the following groups:

- “cooking” (only oleic acid detected),
- “traffic” (only engine exhaust detected),
- “biomass (burning)” (only glucose detected),
- “OOA and inorganics” (OOA and/or sulphate and/or nitrate detected), and
- “mixed organics” (organics other than OOA mixed with inorganics and/or another type of organics detected).

The mass spectra not identified by the algorithm were clustered using the fuzzy c means algorithm. Two new particle types were found by this means: particles containing potassium (i.e., mass spectra dominated by m/z 39; these are present during the fireworks around 14 July), and particles dominated by contaminations of silicone (m/z 73, 147; Schneider et al., 2006). The algorithm was adapted to identify these particle

types as well. Furthermore, 535 mass spectra were identified which were dominated by electronic noise within the instrument. These mass spectra were removed from the dataset.

On the remaining 25 588 mass spectra, the modified algorithm was applied again.

The resulting classification is shown in Fig. 4a. The 2945 unidentified particle mass spectra were manually sorted (Fig. 4b). About two thirds of these particle mass spectra contained too little signal to allow any identification also by manual sorting. The manually identified mass spectra mostly were classified as “OOA and inorganics” or “HOA” (which could be both, due to cooking or engine exhaust). Several more mass spectra with unusual fragmentation pattern (high signal at m/z 45; HOA with high signal at m/z 44) were found. Also several mass spectra with high signal at m/z 60 and 73 were found; these were classified as biomass (burning) organic aerosol (Alfarra et al., 2007; Schneider et al., 2006). Furthermore several more potassium containing mass spectra and additionally ammonium chloride containing mass spectra were found, which both occurred during the fireworks around 14 July.

Calculation of particle number concentrations

To calculate hourly particle number concentrations of a particle type i ($C_{\text{number},i}$), Eq. (3) was applied:

$$C_{\text{number},i} = \frac{\#\text{MS}_i}{Q \cdot t \cdot \text{DC} \cdot S_{\text{eff}} \cdot f_{\#\text{MS},i}}. \quad (3)$$

$\#\text{MS}_i$ is the number of particle mass spectra from type i for the respective averaging period, Q the inlet flow rate (here $1.29 \text{ cm}^3 \text{ s}^{-1}$), t the duration of the averaging period (one hour), and DC the chopper duty cycle (2% in this work). S_{eff} is the saving efficiency, and $f_{\#\text{MS},i}$ a scaling factor to account for the fact that not all particles for which mass spectral data were saved are used in the analysis (see below). The saving efficiency S_{eff} is defined as the number of particles for which mass spectral data were

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saved ($\#P_{\text{saved}}$) divided by the total number of particles detected by the LSP ($\#P_{\text{detected}}$) during a one hour saving cycle:

$$S_{\text{eff}} = \frac{\#P_{\text{saved}}}{\#P_{\text{detected}}}. \quad (4)$$

While Q , t and DC remain constant, both $\#MS_i$ as well as S_{eff} vary with time, as both depend on the actual number concentration of particles present.

$f_{\#MS,i}$ is defined as the ratio of total number of mass spectra of particle type i used in the analysis ($\#MS_i$) to the number of all particles of this type for which mass spectral data were saved ($\#P_{\text{saved},i}$):

$$f_{\#MS,i} = \frac{\#MS_i}{\#P_{\text{saved},i}} \cong \frac{\#MS}{\#P_{\text{saved}}}. \quad (5)$$

Since $\#P_{\text{saved},i}$ is not known, it is assumed here that $f_{\#MS,i}$ is comparable to the ratio of all mass spectra used in the analysis ($\#MS$, the sum of all $\#MS_i$) to the total number of particles for which mass spectral data were saved ($\#P_{\text{saved}}$, Eq. 5). This yields an estimate for $f_{\#MS,i}$ of 0.22 for this dataset. The underlying assumption is only valid if refractory particles contribute only little to the whole number of particles for which data were saved, and if different particle types i do not show significant differences in their degree of bouncing off the vaporizer. Especially the latter is only a rough approximation (see e.g. Matthew et al., 2008). However, as $f_{\#MS,i}$ is a constant factor, the uncertainty of its determination will only affect the absolute number concentration obtained, not the relative courses of diurnal cycles as discussed in Sect. 3.2.2.

3.2.2 Results from single particle analysis

Cooking- and traffic-related HOA

Average diurnal cycles of the measured number concentrations of particles classified as “cooking” and “traffic” are shown in Fig. 5. Also shown is the diurnal cycle of “HOA

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ensemble”, which was retrieved from ensemble organics mass spectra using positive matrix factorization (Freutel et al., 2013). This “HOA ensemble” is assumed to contain both traffic- and cooking-related HOA, as suggested by its mass spectral pattern and diurnal cycle.

5 For the number concentrations of particles classified as “cooking” and “traffic”, two very different diurnal cycles are observed, demonstrating the ability of the algorithm to sufficiently reliably distinguish between these two types of particles despite the similarity of their mass spectra (see Sect. 3.1.2). For the “cooking” particle type, a peak in the evening and a smaller one around noon are observed, consistent with typical
10 time periods for cooking activities. The diurnal cycle of number concentrations of particles classified as “traffic”, on the other hand, shows a distinct peak only during the evening rush hour, while in the morning barely any enhanced number concentrations are observed. This is consistent with the diurnal cycle of “HOA ensemble”, which also shows only a very broad and weak peak during the morning rush hour. Likely, evening
15 peaks of both “traffic” and “cooking” related particles are much more distinct due to the superimposed modulation of the boundary layer height: boundary layer height is lower during the night than during the day, leading to a night-time increase of primary pollutant concentrations.

20 These results for the first time give direct evidence of the presence of externally mixed particles consisting of different types of HOA in an urban surrounding from measurements with the AMS. Although from AMS ensemble measurements it is possible to distinguish between cooking- and traffic-related HOA by positive matrix factorization of ensemble mass spectra (e.g., Allan et al., 2010), those results are not completely unambiguous, and depend partially on subjective decisions by the operator. From single
25 particle analysis with the algorithm presented here, much more unambiguous results are obtained.

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Mixed nitrate/sulphate/OOA

Two very distinct air masses were sampled during the field campaign: during the first three days, air masses from the north-east were advected (“Central Europe” air masses), while for the remaining time, air masses originated from the Atlantic Ocean (“Atlantic” air masses) (Freutel et al., 2013). For these two types of air masses, average diurnal cycles of number concentrations of particles classified as “OOA and inorganics” were calculated individually (Fig. 6a). Furthermore, for both types of air masses, the diurnal cycles of average mass per particle of OOA, nitrate and sulphate within the particles classified as “OOA and inorganics” are shown (Fig. 6b and c).

For both types of air masses, maximum number concentrations are observed during the later morning hours. Average number concentrations are much higher for “Central Europe” air masses than for “Atlantic” air masses, consistent with enhanced mass concentrations of OOA, nitrate and sulphate observed from ensemble mode AMS measurements during the sampling of “Central Europe” air masses (Freutel et al., 2013).

The diurnal cycles of mass per particle of OOA, nitrate and sulphate within the particles classified as “OOA and inorganics” for both air masses show an increase of nitrate and a decrease of OOA and sulphate in the morning hours. During “Atlantic” air masses (Fig. 6c), the average nitrate content of the particles is comparably low during the whole day, with an increase during the morning hours (nitrate content varies from less than 5 to about 20 % during the day). In contrast, during the “Central Europe” air masses, a large increase of nitrate content during the morning hours is observed, leading to a diurnal variation of nitrate content from less than 5 % to about 60 %. At such nitrate mass fractions of more than 50 %, an increased AMS collection efficiency should be expected (Matthew et al., 2008), which however could not be confirmed from ensemble mode measurements (Freutel et al., 2013). From ensemble mode measurements, a maximum nitrate content of about 35 % was observed during “Central Europe” air masses (for all particles, not just “OOA and inorganics”), so possibly the absolute

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fraction of these nitrate-dominated particles was not large enough to alter the overall AMS collection efficiency significantly.

The aforementioned relative compositions retrieved from the results of the classification algorithm were validated by comparison with the relative compositions derived using an independent method. For all particles classified as “OOA and inorganics”, the diurnal cycles of average mass spectra were calculated separately for “Atlantic” and “Central Europe” air masses. The diurnal cycles of nitrate, sulphate and organic content were calculated from these average mass spectra by using only typical marker m/z for nitrate (m/z 30 and 46) and sulphate (m/z 48, 64, 80, 81, and 98) and by calculating organics as sum of all m/z without nitrate, sulphate, and m/z 14–18, 28, and 32. The fractions obtained by this means were found to agree within 10 % with the results given above, which further validates the procedure for retrieval of quantitative information using the described algorithm.

From ensemble mode measurements, also a night-time increase of nitrate mass concentration was found. However, from these measurements it is not possible to distinguish between influence of lower boundary layer height during the night, and of nitrate partitioning into the particle phase. The latter can be directly observed from the single particle measurements: the increase of nitrate mass per particle gives direct evidence of nitrate partitioning to the particle phase. Also the number of measured particles increases during night-time (Fig. 6a), which could be an indication for the additional influence of the lowering of the boundary layer height. However, also another fact can contribute to this increase in observed number concentration: as nitrate condenses onto pre-existing particles, small particles which up to then were below the detection limit of the light scattering probe can grow to sizes where they are detected, increasing the number of detected particles. This effect could also explain the decreasing mass per particle of OOA and sulphate with increasing nitrate mass per particle (Fig. 6b and c). This smaller mass per particle of OOA and sulphate reflects the contributions of these species to the previously smaller particles which did not change by the condensation of nitrate. The diurnal cycle of OOA and sulphate mass per particle therefore

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can also be explained by the diurnal cycle of nitrate partitioning. However, the rate of increase of OOA exceeds that of sulphate mass per particle in the hours after noon. This can be explained by additional condensation of OOA, formed from photooxidation of gaseous precursors, onto the pre-existing particles. The relative fraction of this additional, freshly generated OOA however is apparently too low to alter the observed single particle mass spectral patterns significantly.

Fireworks around 14 July

From the application of the revised algorithm and the manual clustering of the unidentified particles, in total 82 particles containing potassium and 25 particles containing chloride were found, all during the time period of the fireworks around 14 July, the French National Day. Potassium and chloride were detected in separate particles, and not in mixtures with each other. Both particle types furthermore contain small amounts of sulphate, organics and ammonium and/or water. The average mass spectrum of the potassium containing particles furthermore shows a large signal at m/z 30, which likely corresponds to nitrate. However, due to the small m/z 46 to 30 signal ratio we assume that this nitrate derives from potassium nitrate instead of ammonium nitrate. An enhanced ratio of m/z 30 to m/z 46 (both likely from nitrate) was also observed in the average mass spectrum of the chloride containing particles, where however no other potential cations than ammonium were measured.

Although the fireworks aerosol was found to be acidic from ensemble mode measurements (Freutel et al., 2013), no single particles dominated by sulphuric acid (instead of ammonium sulphate) were identified. This could be due to the fact that pure sulphuric acid particles were too small to be detected by the light scattering probe. Furthermore, sulphuric acid might be easily missed when condensing onto pre-existing, ammonium sulphate containing particles, as the accuracy of quantification of ammonium is limited due to large background values. Further laboratory work is needed here in order to investigate into the ability of the LSP-AMS to differentiate between sulphuric acid and ammonium sulphate in different mixing states (i.e., different degrees of neutralization).

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4 Summary

A new classification algorithm was developed for the analysis of single particle mass spectra acquired with the AMS. It retrieves both, information on particle types and quantitative information about the chemical composition of the individual particles by comparison to reference mass spectra of compounds and proxies of compound classes typically found in ambient aerosol. Both qualitative and quantitative retrieval were tested and validated in the laboratory using mass spectra from particles of known composition. The uncertainty of the relative chemical composition derived from one single particle mass spectrum was estimated to be about 25 % for inorganic, and about 30 % for organic substances. The uncertainty decreases slightly (to about 15 and 25 %, respectively) when averaging over a larger number (≥ 25) of mass spectra. Incorporation of additional reference mass spectra like from ammonium chloride, sulphuric acid, ammonium bisulphate, or additional organic species and determination of detection limits for the various substances will extend the applicability of the algorithm further.

The algorithm was applied to a dataset of single particle mass spectra acquired during ambient aerosol measurements in a suburb of Paris during July 2009. The algorithm identified about 90 % of all particle mass spectra successfully; the remaining mass spectra to the most part contained too little signal to be identified even by manual inspection. New particle types (potassium, silicone contaminations, ammonium chloride) were found in the unidentified mass spectra. This possibility to identify new particle types and incorporate their detection easily into the existing algorithm is an important feature in order to adapt to new, foreign particle types found in ambient aerosol samples.

The analysis showed that not only information on mixing state (like in the case of fireworks aerosol sampled around 14 July) can be retrieved from the single particle analysis. It was also possible to unambiguously differentiate between hydrocarbon-like organic aerosol of different sources (cooking- and traffic-related emissions). This is only under favourable conditions possible by positive matrix factorization of organic

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mass spectra from ensemble AMS measurements. The separation based on a constant set of reference mass spectra also ensures the direct comparability of results between different datasets.

Analysis of mass per particle within the particles containing OOA and/or nitrate and/or sulphate showed a clear diurnal cycle with nitrate mass per particle increasing during night-time, giving evidence of nitrate partitioning into the particle phase. Such direct information is not available from AMS ensemble measurements. From the single particle measurements, it was also found that number concentration of particles classified as “OOA and inorganics” increases during night-time, which however can have two causes: the increase of particle number concentrations due to the decrease of boundary layer height, and/or hitherto smaller particles growing into the detectable size range of the LSP-AMS. This lower cut-off of the measurable size range due to decreasing detection efficiency is an important limitation which has always to be kept in mind when interpreting single particle data acquired with the LSP-AMS. Furthermore, in the afternoon, an increase of OOA mass per particle is observed, likely due to condensation of newly formed OOA after photooxidation of gaseous precursors.

Especially in combination with complementary information from ensemble AMS measurements, the analysis of the chemical composition of individual particles can give valuable new and more substantiated information about various processes in the atmosphere than when relying on one of the methods alone. Therefore, a combined analysis of both AMS single particle and ensemble measurements is a promising approach to gain more complete information from the data acquired with both methods from one single instrument.

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Table 1. Measurements of reference single particle mass spectra.

Substance	Reagent grade and manufacturer	Particle mobility diameter/nm	Number of single particle mass spectra retrieved	Number of ions in single particle mass spectrum ^a (average and standard deviation)	Marker <i>m/z</i>	<i>m/z</i> in marker-MS
Laboratory measurements (10 s/20 s/30 s in MS ^b /PToF ^c /LS ^d mode)						
NH ₄ NO ₃	p.a., Labor-Service GmbH	400	180	nitrate: 123 ± 46	30, 46 ^e	30, 46
(NH ₄) ₂ SO ₄	p.a., Merck	400	80	sulphate: 67 ± 35	48, 64, 80, 81, 98 ^e	48, 64, 80, 81, 98
internally mixed NH ₄ NO ₃ /(NH ₄) ₂ SO ₄	see pure substances	400	106	nitrate: 68 ± 33 sulphate: 25 ± 17	–	–
glucose	α -D-(+)-glucose monohydrate, ≥ 99.5 %, Roth	400	313	glucose: 347 ± 181	31, 60 ^f	29, 31, 43, 60, 73
internally mixed NH ₄ NO ₃ /(NH ₄) ₂ SO ₄ /glucose	see pure substances	400	864	nitrate: 42 ± 21 sulphate: 17 ± 13 glucose: 96 ± 43	–	–
oleic acid	vegetable, extra pure, Merck	400	476	oleic acid: 295 ± 245	55 ^f	41, 43, 55, 69, 83
succinic acid	≥ 99 %, Alfa Aesar	400	188	succinic acid: 101 ± 72	–	–
Source measurement (10 s/20 s/30 s in MS/PToF/LS mode)						
gasoline engine exhaust	–	polydisperse, (300 ± 100) ^g	53	exhaust: 884 ± 1785	57 ^f	41, 55, 69, 83
Ambient measurement (20 s/20 s/20 s in MS/PToF/LS mode)						
OOA internally mixed with nitrate, sulphate	–	polydisperse, (560 ± 90) ^g	822	OOA: 104 ± 56 nitrate: 43 ± 58 sulphate: 36 ± 33	44 ^f	44, 55

^a *m/z* 14–18, 28, 32 not regarded; sulphate, nitrate calculated as sum of all respective marker *m/z*.

^bMS: mass spectra mode. Yields the ensemble mass concentrations of organics, nitrate, sulphate, ammonium, and chloride. See e.g. Canagaratna et al. (2007).

^cPToF: particle time of flight mode. Yields the ensemble mass size distributions for the species listed under ^b. See e.g. Canagaratna et al. (2007).

^dLS: light scattering mode. Yields the single particle mass spectral data as described e.g. in Cross et al. (2009).

^eSum of signal at all *m/z* needs to be ≥ 5 ions for positive detection (see Fig. 1).

^fSignal at each *m/z* needs to be ≥ 1 ion for positive detection (see Fig. 1).

^gVacuum-aerodynamic diameter; maximum and standard deviation σ of Gaussian fit to single particle number size distribution.

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Table 2. Comparison of relative composition from single particle and ensemble measurements of internally mixed particles of nitrate and sulphate, and of nitrate, sulphate and glucose. In case of ensemble measurements, ratios of average mass concentrations are given. For the single particle measurements, the average ratios and standard deviations (σ) were determined using three different methods: ratio of the averages, average ratios, and a Gaussian fit to the distribution of ratios observed. All: all particles of this type; identified: only for particles of this type in which both respective substances were identified; avg.: average; unc.: uncertainty.

Mass fraction	Ratio of averages		Average ratios	Gaussian fit	Ensemble ^a
	all	identified	identified	identified	avg. \pm unc. ^b
Nitrate & sulphate					
Sulphate/nitrate:					
Avg. \pm unc. ^c :	0.62 \pm 0.10	0.67 \pm 0.11	0.68 \pm 0.10	0.68 \pm 0.10	0.76 \pm 0.05
σ :	0.56	0.44	0.33	0.28	–
Nitrate, sulphate & glucose					
Sulphate/nitrate:					
Avg. \pm unc.:	0.65 \pm 0.09	0.78 \pm 0.11	0.82 \pm 0.12	0.76 \pm 0.11	0.84 \pm 0.03
σ :	0.66	0.51	0.40	0.34	–
Sulphate/glucose:					
Avg. \pm unc.:	0.25 \pm 0.04	0.29 \pm 0.04	0.36 \pm 0.05	0.29 \pm 0.04	0.38 \pm 0.02
σ :	0.26	0.18	0.15	0.16	–
Nitrate/glucose:					
Avg. \pm unc.:	0.39 \pm 0.06	0.36 \pm 0.05	0.45 \pm 0.06	0.36 \pm 0.05	0.45 \pm 0.03
σ :	0.29	0.23	0.17	0.12	–
Deviation ^d	22.5 %	15.7 %	4.8 %	15.9 %	–
Scatter ^e	92.8 %	64.7 %	44.4 %	43.2 %	–

^aEnsemble mass concentrations are calculated using a slightly modified organics fragmentation pattern table (frag_organics[30] = 1.49-frag_organics[44]; frag_organics[39] = 39; frag_organics[41] = 41) inferred from the average ensemble mass spectrum of pure glucose.

^bThe uncertainty range is estimated from the standard deviation of the average, and from additional uncertainties of 1 % and 4 % for the mass concentrations of nitrate and glucose, respectively, conservatively estimated from the range of values derived with and without changes in the fragmentation pattern table (changes in the organics pattern described under ^a, and changes due to corrections from measurements of particulate-free air).

^cThe statistical uncertainty of the average (Gaussian fit: obtained from the fitting procedure; else: standard deviation of the average) and Δ_{scaling} (10 %) of sulphate, nitrate and glucose are considered in the uncertainty range given. Δ_{RIE} was not regarded as it affects results from single particle and ensemble measurements in the same way.

^dAverage deviation from ensemble average relative to the ensemble average.

^eAverage relative standard deviation.

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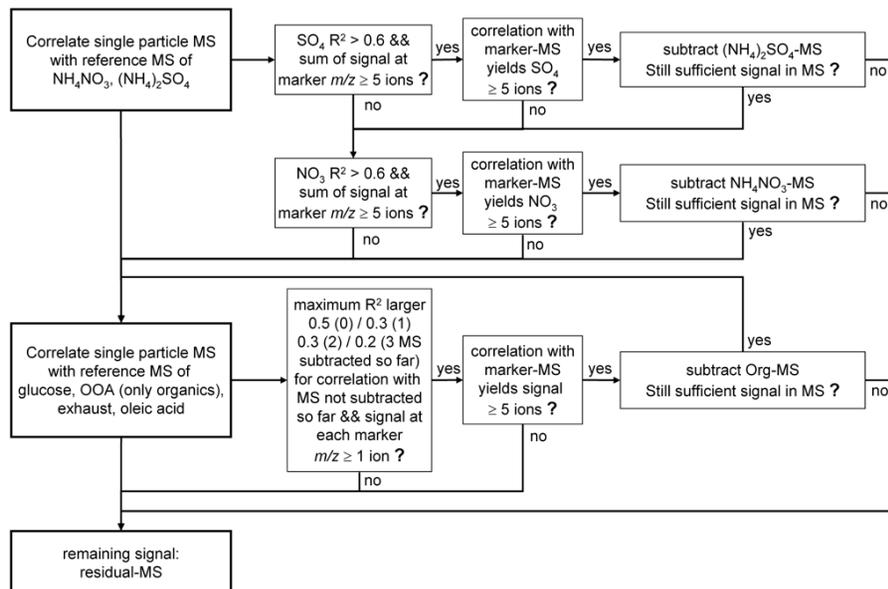


Fig. 1. Schematics of the developed classification algorithm. SO_4 : sulphate; NO_3 : nitrate; Org: organics; MS: mass spectrum; &&: logical AND.

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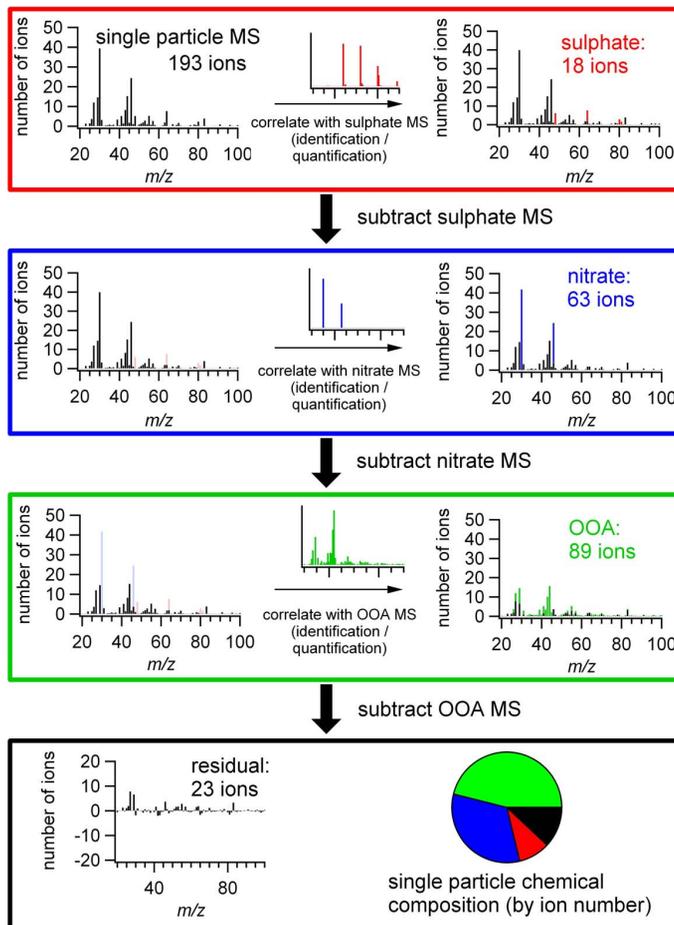


Fig. 2. Example for the application of the algorithm on one single particle mass spectrum containing OOA, sulphate and nitrate. MS: mass spectrum.

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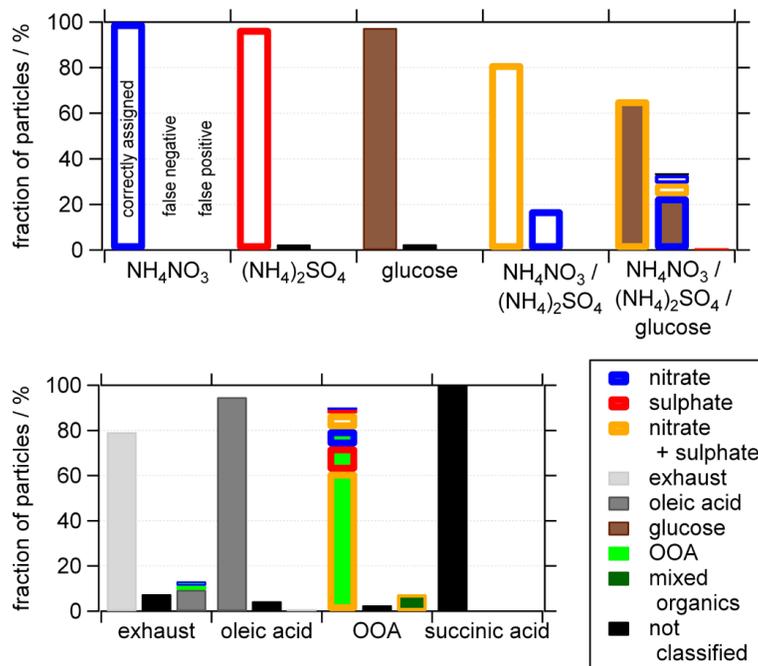


Fig. 3. Qualitative validation of the algorithm. On the x-axis the different particle types of known composition (compare Table 1) are shown. For each particle type, the left bar shows the fraction of correctly assigned particle mass spectra, the middle bar the fraction of false negatives, and the bar on the right hand side the fraction of false positives. The border colour of the bars represents the identified inorganic content, the fill colour the identified organic content. Particles in which no component was identified are shown in black. For the particles of the OOA type, the exact content of nitrate, sulphate and OOA is not known, therefore, all particle mass spectra have been regarded as correctly assigned where at least one of these components was detected.

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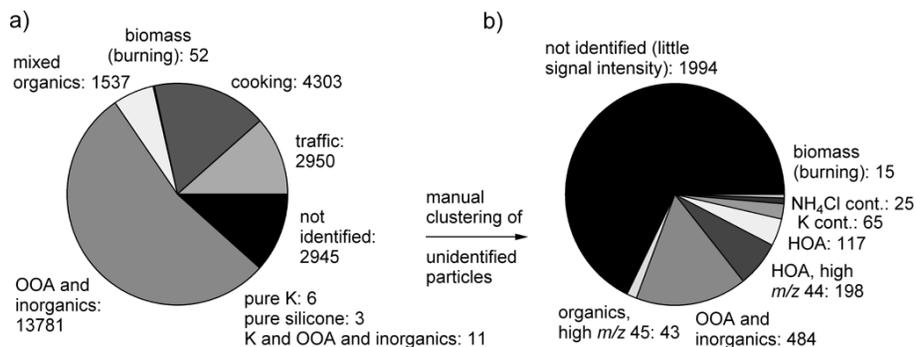


Fig. 4. Application of the classification algorithm on an ambient dataset. Cont.: containing.

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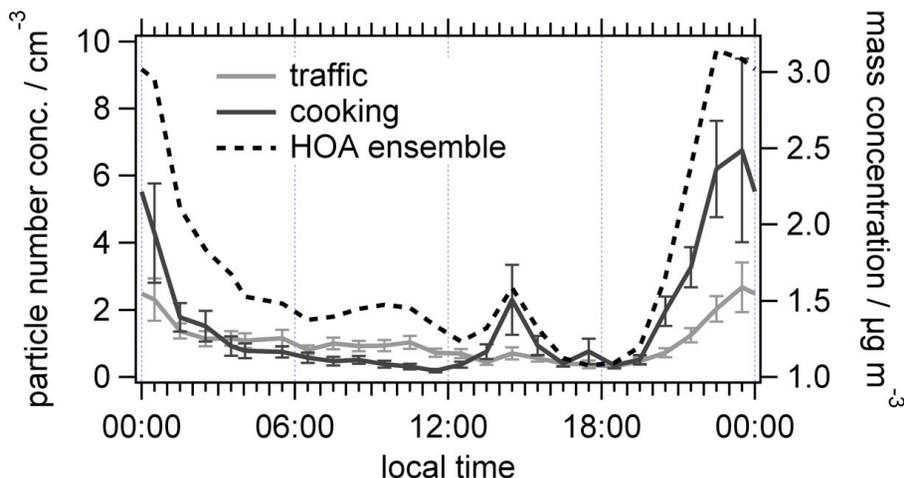


Fig. 5. Average diurnal cycles of number concentrations of particles classified as “cooking” and “traffic” (left axis), and of the mass concentration of the HOA factor (right axis) retrieved via positive matrix factorization of ensemble organics mass spectra (Freutel et al., 2013). Error bars represent the standard deviations of the averages; error bars for “HOA ensemble” (relative uncertainty of 20 % from PMF analysis; Freutel et al., 2013) are omitted for clarity.

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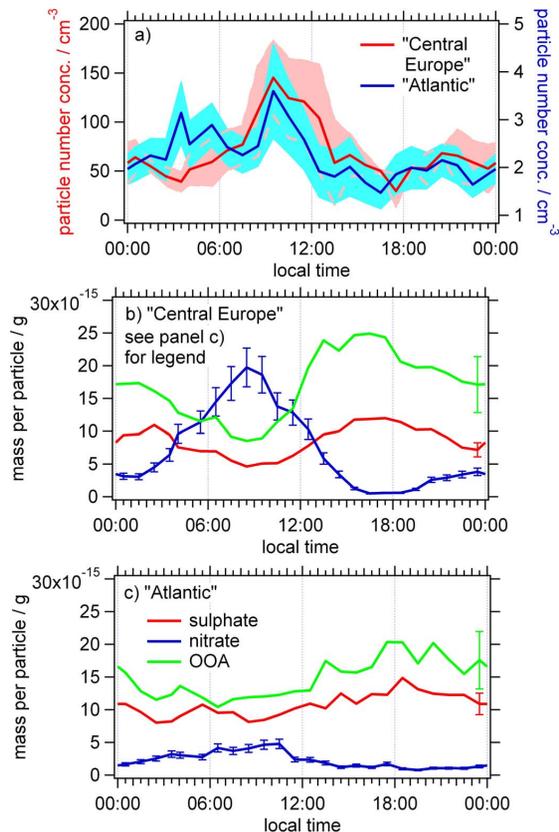


Fig. 6. Average diurnal cycle of the number concentration of particles classified as “OOA and inorganics”, separated for “Central Europe” and “Atlantic” air masses (a). Standard deviations of the average values are shown as light shaded areas. Average diurnal cycles of mass per particle of nitrate, sulphate, and OOA within the particles classified as “OOA and inorganics” for “Central Europe” (b) and “Atlantic” air masses (c). Error bars as calculated from Eq. (2) are shown for nitrate, and only exemplarily for clarity for OOA and sulphate.