

Interactive comment on “Calibration of sealed HCl cells used for TCCON instrumental line shape monitoring” by F. Hase et al.

Anonymous Referee #1

Received and published: 21 September 2013

This is a very clear and concise description of a very solid piece of research. The authors exhibit a thorough understanding of their subject and the skill to convey their research and technique for use by others. It is timely and important and should find application by others working in the same area.

I include here only a few minor comments that might improve the presentation or content.

Page 7187 Line 8 What is residual here? Is it not just variation?

P 7190 L 26 This is the first introduction of the XCO₂ nomenclature, it would be good to define it here.

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P 7192 L 20 Regarding the wave number ratio: in Figure 1 the ratio of CO₂/O₂ = 6300/7900 = .80.

P 7193 L 16 "fictive" is strictly correct but "imaginary" or "notional" are more widely used

P 7194 L 3 CaF(subscript 2)

P 7194 L 9 Figure 2 is not very helpful The radiation path is not clear. All the components are not shown. Perhaps an optical diagram could be included. P 7195 L 17 Not clear what is meant by 2,.....2.5%

P 7195 L 20 Why is the ME curve not sensitive to information at path lengths greater than 45 cm? Is the max OPD in these tests 45cm or greater (as Fig 4 would imply)?

P 7196 L 4 add: ...3.00 mbar C₂H₂, with the addition of an independent measure of the cell pressure.

P 7196 L 15 Is this the standard deviation of the difference in the temperature retrieved from C₂H₂ and HCl?

P 7198 L 12 This calibration can identify some misalignment that may influence the ILS of a specific instrument. Does it necessarily eliminate a bias?

Table 1

The number density of HCl in a cell is: $n = Ap/RT$ and the column amount is nl

Where A is Avogadro's number, R is the universal gas constant, p is pressure and T is temperature For cell #1 H₃₅Cl T=296K, p= 473.2 Pa, and l= .1m

so $nl = (6.02E23 \text{ molecules/mole } 473.2 \text{ Pa } .1 \text{ m}) / (8.314 \text{ Pa m}^3 / \text{mole K } 296 \text{ K}) = 1.16 \text{ E } 22 \text{ molecules m}^{-2}$

Which is not the value in the table.

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