



## *Supplement of*

# **Characterization of trace metals with the SP-AMS: detection and quantification**

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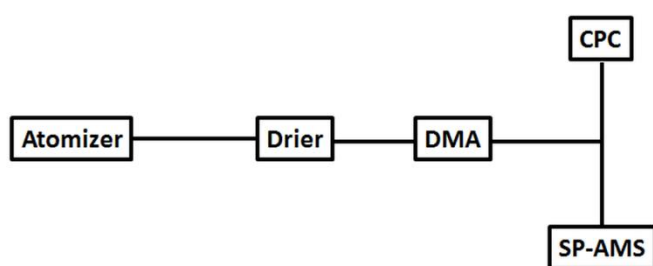


Figure S1 – Instrument setup to measure trace elements in the laboratory.

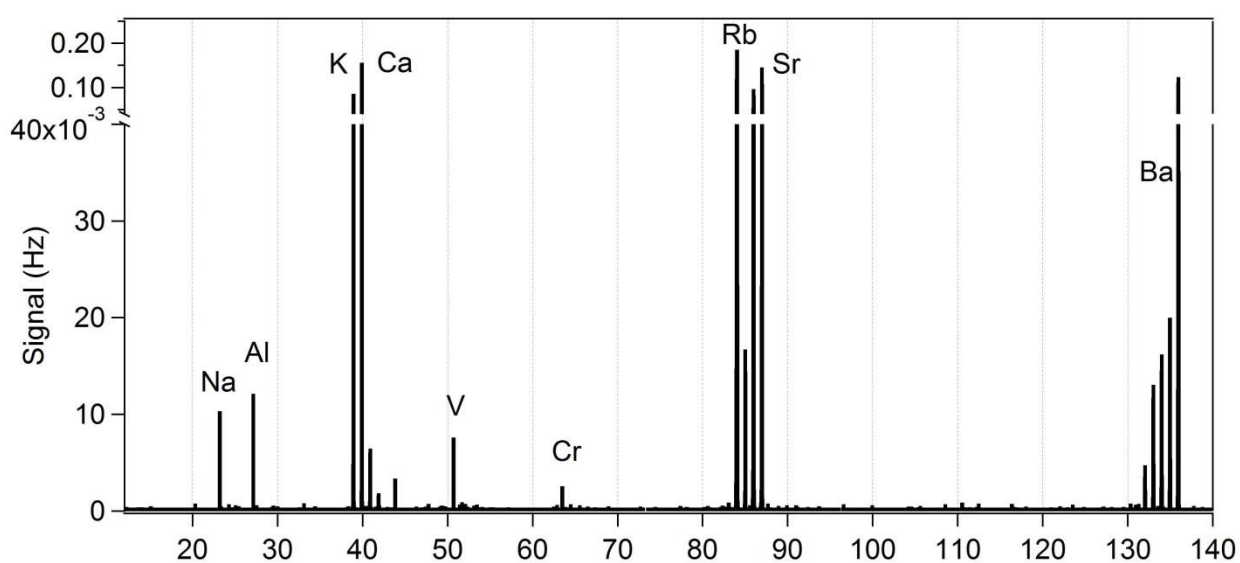


Figure S2 – Mass spectrum obtained with Tungsten and filament switched off and laser vaporizer on.

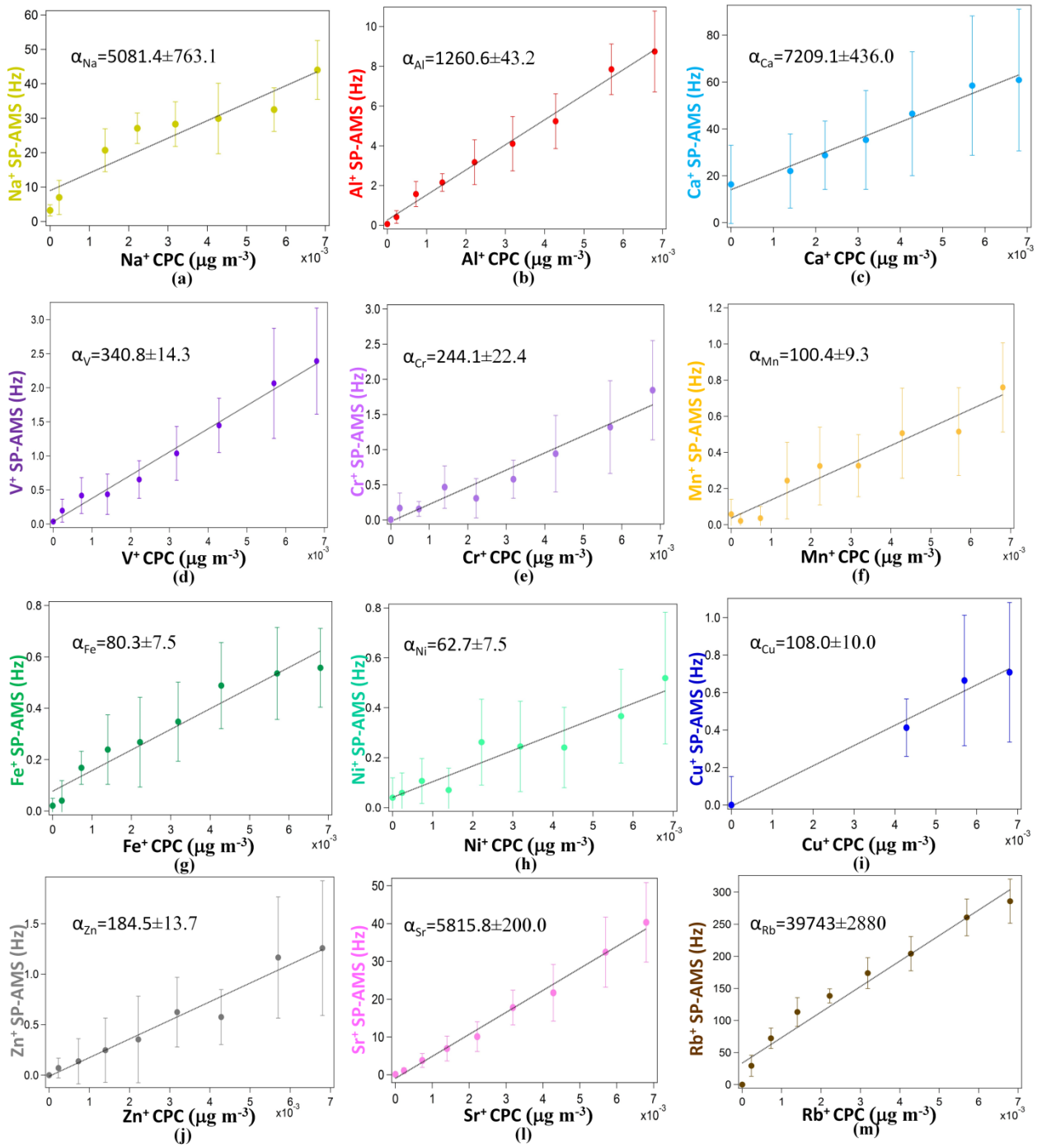


Figure S3- Signal measured by the SP-AMS (Hz) versus the mass concentration obtained by the CPC ( $\mu\text{g m}^{-3}$ ) for each trace element and respective slope values ( $\alpha$ ).

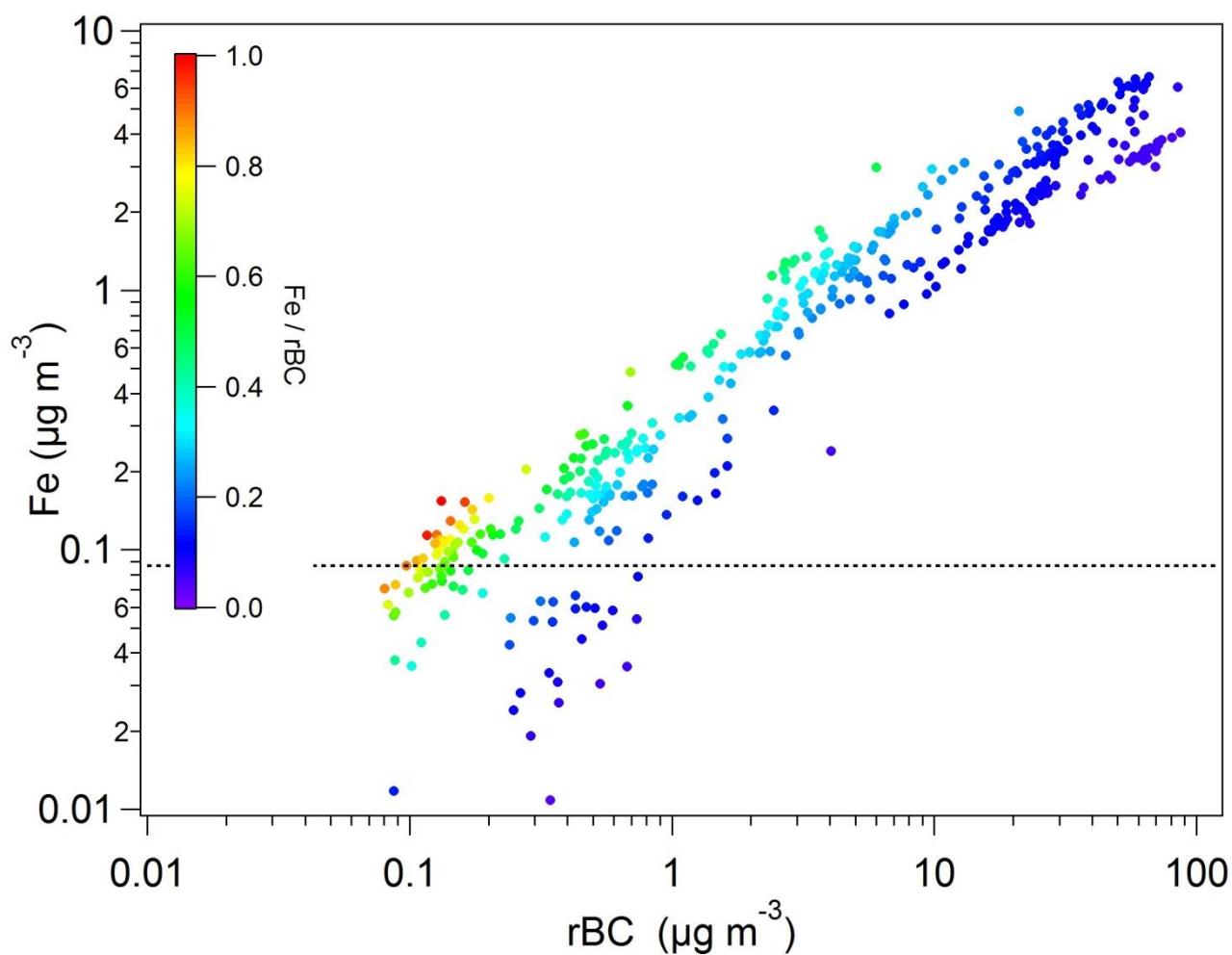


Figure S4 – Fe mass concentration as a function of rBC concentration measured at the heating station with the SP-AMS. The dashed black line represents the Fe limit of detection for this instrument.

Table S1- Relative atomic mass, isotopic composition, standard concentration, nitric acid concentration and identified isotopes of the trace elements investigated in this study.

Element	Relative atomic mass <sup>a</sup>	Isotopic composition (%) <sup>a</sup>	Standard concentration (mg L <sup>-1</sup> ) <sup>b</sup>	HNO <sub>3</sub> concentration (% w/w) <sup>b</sup>	Identified isotopes with the SP-AMS
<b>Na</b>	22.9897	100.00	1000	2	<sup>23</sup> Na <sup>+</sup>
<b>Al</b>	26.9815	100.00	1000	2	<sup>27</sup> Al <sup>+</sup>
<b>Ca</b>	39.9625	96.94	1000	2	<sup>40</sup> Ca <sup>+</sup> , <sup>42</sup> Ca <sup>+</sup> , <sup>44</sup> Ca <sup>+</sup>
<b>V</b>	50.9439	99.75	1000	3	<sup>51</sup> V <sup>+</sup>
<b>Cr</b>	51.9405	83.78	1000	2	<sup>52</sup> Cr <sup>+</sup> , <sup>53</sup> Cr <sup>+</sup> , <sup>54</sup> Cr <sup>+</sup>
<b>Mn</b>	54.9380	100.00	998	2	<sup>55</sup> Mn <sup>+</sup>
<b>Fe</b>	55.9349	91.75	1000	2	<sup>54</sup> Fe <sup>+</sup> , <sup>55</sup> Fe <sup>+</sup> , <sup>56</sup> Fe <sup>+</sup> , <sup>57</sup> Fe <sup>+</sup> ,
<b>Ni</b>	57.9353	68.07	1000	2	<sup>58</sup> Ni <sup>+</sup> , <sup>60</sup> Ni <sup>+</sup> , <sup>61</sup> Ni <sup>+</sup> , <sup>62</sup> Ni <sup>+</sup> , <sup>64</sup> Ni <sup>+</sup>
<b>Cu</b>	62.9295	69.15	1000	2	<sup>63</sup> Cu <sup>+</sup> , <sup>64</sup> Cu <sup>+</sup> ,
<b>Zn</b>	63.9291	48.26	1000	2	<sup>64</sup> Zn <sup>+</sup> , <sup>66</sup> Zn <sup>+</sup> , <sup>67</sup> Zn <sup>+</sup> , <sup>68</sup> Zn <sup>+</sup> , <sup>70</sup> Zn <sup>+</sup>
<b>Rb</b>	84.9117	72.17	1000	2	<sup>85</sup> Rb <sup>+</sup> , <sup>87</sup> Rb <sup>+</sup>
<b>Sr</b>	87.9056	82.58	1000	2	<sup>84</sup> Sr <sup>+</sup> , <sup>86</sup> Sr <sup>+</sup> , <sup>87</sup> Sr <sup>+</sup> , <sup>88</sup> Sr <sup>+</sup> ,
<b>Ba</b>	137.9052	71.69	1000	2	<sup>130</sup> Ba <sup>+</sup> , <sup>132</sup> Ba <sup>+</sup> , <sup>134</sup> Ba <sup>+</sup> , <sup>135</sup> Ba <sup>+</sup> , <sup>136</sup> Ba <sup>+</sup> , <sup>137</sup> Ba <sup>+</sup> , <sup>138</sup> Ba <sup>+</sup>

<sup>a</sup> Watson et al., 2004.

<sup>b</sup> Sigma-Aldrich standards in nitric acid.