



# **Instrumentation and Measurement Strategy for the NOAA SENEX Aircraft Campaign as Part of the Southeast Atmosphere Study 2013**

Warneke C.<sup>1,2</sup>, M. Trainer<sup>2</sup>, J.A. de Gouw<sup>1,2</sup>, D.D. Parrish<sup>1,2</sup>, D.W. Fahey<sup>2</sup>, A.R. Ravishankara<sup>2,9</sup>,  
A.M. Middlebrook<sup>2</sup>, C.A. Brock<sup>2</sup>, J.M. Roberts<sup>2</sup>, S.S. Brown<sup>2</sup>, J.A. Neuman<sup>1,2</sup>, B.M. Lerner<sup>1,2</sup>, D.  
Lack<sup>1,2</sup>, D. Law<sup>1,2</sup>, G. Hübler<sup>1,2</sup>, I. Pollack<sup>1,2,9</sup>, S. Sjostedt<sup>1,2</sup>, T.B. Ryerson<sup>2</sup>, J.B. Gilman<sup>1,2</sup>, J.  
Liao<sup>1,2</sup>, J. Holloway<sup>1,2</sup>, J. Peischl<sup>1,2</sup>, J.B. Nowak<sup>1,2,10</sup>, K. Aikin<sup>1,2</sup>, K.-E. Min<sup>1,2,11</sup>, R.A.  
Washenfelder<sup>1,2</sup>, M.G. Graus<sup>1,2,12</sup>, M. Richardson<sup>1,2</sup>, M.Z. Markovic<sup>1,2,13</sup>, N.L. Wagner<sup>1,2</sup>, A.  
Welti<sup>1,2,14,14</sup>, P.R. Veres<sup>1,2</sup>, P. Edwards<sup>1,2,15</sup>, J.P. Schwarz<sup>2</sup>, T. Gordon<sup>1,2</sup>, W.P. Dube<sup>1,2</sup>, S.  
McKeen<sup>1,2</sup>, J. Brioude<sup>1,2</sup>, R. Ahmadov<sup>1,2</sup>, A. Bougiatioti<sup>3</sup>, J. Lin<sup>3</sup>, A. Nenes<sup>3</sup>, G.M. Wolfe<sup>4,16</sup>, T.F.  
Hanisco<sup>4</sup>, B.H. Lee<sup>5</sup>, F.D. Lopez-Hilfiker<sup>5</sup>, J.A. Thornton<sup>5,19</sup>, F.N. Keutsch<sup>6,17</sup>, J. Kaiser<sup>6</sup>, J.  
Mao<sup>7,18</sup>, C. Hatch<sup>8</sup>

<sup>1</sup> Cooperative Institute for Research in Environmental Sciences, Univ. of Colorado, Boulder

<sup>2</sup> Chemical Sciences Division, NOAA Earth System Research Laboratory, Boulder, CO

<sup>3</sup> Georgia Institute of Technology, Atlanta, GA

<sup>4</sup> NASA Goddard Space Flight Center, Greenbelt, MD

<sup>5</sup> University of Washington

<sup>6</sup> University of Wisconsin-Madison, Madison, WI

<sup>7</sup> Geophysical Fluid Dynamics Laboratory, NOAA, Princeton, NJ

<sup>8</sup> Department of Chemistry, Hendrix College, 1600 Washington Ave., Conway, AR, USA

<sup>9</sup> now at Department of Atmospheric Science, Colorado State University, Ft Collins, CO, USA

<sup>10</sup> now at Aerodyne Research Inc., Billerica, MA

<sup>11</sup> now Gwangju Institute of Science and Technology, Gwangju, Korea

<sup>12</sup> now at Institute of Atmospheric and Cryospheric Sciences University of Innsbruck,  
Austria

<sup>13</sup> now at Air Quality Processes Research Section, Environment Canada, Toronto, Ontario,  
Canada

<sup>14</sup> now at Leibniz Institute for Tropospheric Research

<sup>15</sup> now at University of York, UK

<sup>16</sup> University of Maryland Baltimore County

<sup>17</sup> now at Harvard University, Cambridge, MA

<sup>18</sup> Princeton University

<sup>19</sup> now at Laboratory of Atmospheric Chemistry, Paul Scherrer Institute, Villigen,  
Switzerland



38 **Abstract:**

39 The Southeast United States (US) might not have warmed as much as the rest of the  
40 country over the past 50-100 years. Providing an improved understanding of this  
41 potential anomaly, and specifically the roles played by aerosols, was one of the main goals  
42 for the Southeast Atmosphere Study (SAS). Natural emissions of ozone-and-aerosol-  
43 precursor gases such as isoprene and monoterpenes are high in the southeast of the US. In  
44 addition, anthropogenic emissions are significant in the Southeast US and summertime  
45 photochemistry is rapid. The NOAA-led SENEX (Southeast Nexus) aircraft campaign was  
46 one of the major components of the SAS study and was focused on studying the  
47 interactions between biogenic and anthropogenic emissions to form secondary pollutants.  
48 During SENEX, the NOAA WP-3D aircraft conducted 20 research flights between 27 May  
49 and 10 July 2013 based out of Smyrna, TN.

50 Here we describe the experimental approach, the science goals and early results of the  
51 NOAA SENEX campaign. The aircraft, its capabilities and standard measurements are  
52 described. The instrument payload is summarized including detection limits, accuracy,  
53 precision and time resolutions for all gas-and-aerosol phase instruments. The inter-  
54 comparisons of compounds measured with multiple instruments on the NOAA WP-3D are  
55 presented and were almost all within the stated uncertainties.

56 The SENEX flights included day- and nighttime flights in the Southeast as well as flights  
57 over areas with intense shale gas extraction (Marcellus, Fayetteville and Haynesville  
58 shale). We present one example flight on 16 June 2013, which was a daytime flight over  
59 the Atlanta region, where several crosswind transects of plumes from the city and nearby  
60 point sources, such as power plants, paper mills and landfills, were flown. The area around  
61 Atlanta has large biogenic isoprene emissions, which provided an excellent case for  
62 studying the interactions between biogenic and anthropogenic emissions. In this example  
63 flight, chemistry in and outside the Atlanta plumes was observed for several hours after  
64 emission. The analysis of this flight showcases the strategies implemented to answer some  
65 of the main SENEX science questions.

66  
67



## 68 1. Introduction

69 The SENEX campaign (Southeast Nexus-Studying the Interactions between Natural  
70 and Anthropogenic Emissions at the Nexus of Climate Change and Air Quality) was a large-  
71 scale National Oceanic and Atmospheric Administration (NOAA) led field study in the  
72 Southeastern United States (U.S.) in summer 2013. The SENEX measurement platform was  
73 the NOAA WP-3D aircraft operated out of Smyrna, Tennessee. SENEX was part of a large,  
74 comprehensive and coordinated research effort to understand the emission sources,  
75 chemistry, and meteorology of the summertime atmosphere in the Southeast U.S.: the  
76 Southeast Atmosphere Study (SAS) (Xu et al., 2015), which included the other field  
77 campaigns: Southern Oxidant and Aerosol Study (SOAS), Tropospheric HONO  
78 (TropHONO), and the North American Airborne Mercury Experiment (NAAMEX). Besides  
79 the NOAA WP-3D, measurements during SAS were made on the following platforms and  
80 locations: the National Science Foundation (NSF) National Center for Atmospheric  
81 Research (NCAR) C-130 aircraft, the Purdue University Duchess aircraft, the State  
82 University of New York-Stony Brook Long-EZ aircraft, the Centreville and Alabama Aquatic  
83 Biodiversity Centre (AABC) flux ground site located in Alabama, the Look Rock, Tennessee  
84 ground site, the Research Triangle Park (RTP) ground site in North Carolina and Caltech  
85 chamber studies (FIXIT).

86 The detailed science goals for SENEX can be found in the SENEX white paper  
87 (<http://esrl.noaa.gov/csd/projects/senex/>) and are briefly listed here:

88 (1) Understanding the emissions of aerosol, aerosol and ozone ( $O_3$ ) precursors, and  
89 greenhouse gases in the Southeast U.S. Special focus was aimed at evaluating available  
90 emission inventories for organic aerosol, black carbon,  $NO_x$  ( $NO+NO_2$ ), volatile organic  
91 compounds (VOCs), sulfur dioxide ( $SO_2$ ), greenhouse gases, and aerosol precursors from  
92 point sources such as coal-fired power plants, urban areas as well as biogenic VOC  
93 emissions. Another focus was to understand the importance of emissions from biomass  
94 burning in the region.

95 (2) Understanding the formation mechanisms of secondary species such as ozone,  
96 sulfate and organic aerosols in the Southeast U.S. The main focus here was to determine  
97 the influence of biogenic emissions, nighttime chemistry, aqueous-phase processes, and  
98 organic nitrates on the formation of the secondary species.



99 (3) Determining the composition and distribution of aerosol in the Southeast U.S.  
 100 by looking at the relative abundance of sulfate, organics and other chemical components  
 101 over the whole study region and at accessible altitude levels.

102 (4) Quantifying deposition and loss processes critical for determining atmospheric  
 103 concentrations of aerosol, ozone and NO<sub>y</sub> (sum of nitrogen oxides).

104 (5) Determining the climate-relevant properties of aerosol in the Southeast U.S. by  
 105 looking at the extinction, absorption and CCN properties of aerosol from primary and  
 106 secondary sources and their dependence on the high humidity in the Southeast U.S. Special  
 107 focus was given on determining the fraction of organic aerosol that occurs naturally  
 108 versus the fraction that is controlled by anthropogenic emissions and how each may  
 109 change in the future as a result of warming and changes in anthropogenic emissions.  
 110 Additional focus was on black carbon and its co-emitted species to understand whether  
 111 controlling specific BC sources has a net warming or cooling effect.

112 (6) Quantifying methane (CH<sub>4</sub>) and VOC emissions from selected shale gas  
 113 extraction regions (Marcellus, Haynesville and Fayetteville).

114 In this paper we describe the payload of the NOAA WP-3D, describe the locations of  
 115 the SENEX flights, show inter-comparisons used to evaluate the measurements and  
 116 describe an example flight to showcase the measurement strategies that were used during  
 117 SENEX.

118

## 119 **2. NOAA WP-3D aircraft**

120 The two NOAA WP-3D aircraft have been used in air quality and climate related  
 121 airborne field campaigns since 1994. The NOAA WP-3D carried its maximum payload of  
 122 3600 kg of scientific equipment during SENEX and 4-6 scientists. The aircraft has a range  
 123 of 3000 km and a ceiling of about 7600 m. During SENEX the highest altitude was about  
 124 6400 m due to the heavy payload. Flight duration was typically around 7 hr, and the  
 125 majority of the flights were conducted in the daytime boundary layer approx. 0.5 km  
 126 above ground level. A picture of the aircraft taken during SENEX is shown in Figure 1.

127 The WP-3D was equipped by the NOAA Aircraft Operations Center (AOC) flight  
 128 facility with instruments detailing the position and motion of the aircraft as well as many  
 129 meteorological parameters such as 3D wind speed and direction, ambient, potential and



dew point temperatures, water vapor mixing ratios, pressure and sea surface temperature.  
 A list of the most commonly used aircraft-provided parameters and their uncertainties is  
 given in Table 1.

133

### 134 **3. NOAA WP-3D SENEX flight summaries**

During SENEX a total of 20 research flights were conducted; of those, two were test  
 flights from Tampa, Florida and two were the transfer flights between Tampa, FL and  
 Smyrna, TN. All of the flights, including the test and transfer flights, addressed multiple  
 science goals. All the SENEX flight tracks are shown in Figure 2 on a map of the Southeast  
 US that also shows most of the larger point sources in the region. Twelve daytime, three  
 nighttime and five shale gas region flights (Marcellus, Haynesville and Fayetteville shale)  
 were conducted to answer the major SENEX science questions. The flight tracks in Figure  
 2 are color-coded by those three categories and details about each flight can be found in  
 Tables 2, 3, and 4, where a short description of the flight, the investigated emission  
 sources, and the coordinating activities are listed.

145

### 146 **4. NOAA WP-3D SENEX chemical and aerosol instrumentation**

The WP-3D instrumentation payload on the WP-3D was specifically designed to  
 provide the necessary measurements to answer the SENEX science questions. The  
 instrumentation included a wide variety of gas and aerosol-phase measurements. A  
 schematic drawing of the payload of the WP-3D is shown in Figure 1b. All the instruments  
 for aerosol phase measurements are listed in Table 5 and for gas phase measurements in  
 Table 6 together with their measurement technique, accuracy and precision, sample  
 interval, and a reference to a publication describing the respective instrument in detail.  
 Overall 22 different instruments were installed on the NOAA WP-3D with a total power  
 consumption of 40 A (110V, 400 Hz 3 phase), 130 A (110V, 400 Hz), 40A (110V 60 Hz),  
 and 42 A (28 V DC). Most instruments were mounted inside the fuselage, but two  
 instrumented wingpods added significant scientific payload capacity including 72 whole-  
 air canister samples, a carbon monoxide (CO) analyzer and the fine particle counter to add  
 significant scientific payload capacity. Four to six scientists were on board during each  
 flight to monitor all the instruments and adjust the flight plans to current meteorological



161 conditions as needed. During the flights, selected aircraft and instrument data were  
162 streamed to the ground and could be monitored in near real time on a website for  
163 situational awareness for all SONGNEX scientists.

164 A detailed description for each instrument can be found in the SI; in the following  
165 two paragraphs the instrument name and measurement technique are given and in Tables  
166 5 and 6, accuracy, precision, sample interval and literature reference are listed in addition.

167 Aerosol-phase measured parameters were: (1) the particle (0.004-8.3 $\mu$ m) number,  
168 size and volume with parallel condensation particle counters (CPCs) and white and laser  
169 light scattering, (2) sub-micrometer extinction and absorption of dry, humidified, and  
170 thermodenuded aerosol at three wavelengths spanning the visible with a cavity ringdown  
171 aerosol extinction spectrometer (CRD) and a photoacoustic aerosol absorption  
172 spectrometer (PAS), (3) the non-refractory submicron aerosol composition of organics,  
173 sulfate, nitrate, ammonium and chloride with an aerosol mass spectrometer (AMS), (4)  
174 cloud condensation nuclei (CCN) and supersaturation, (5) accumulation-mode refractory  
175 black carbon (rBC) mass content of single particles with an SP2. Most of the aerosol  
176 instrumentation was connected to a low turbulence inlet (LTI) (Wilson et al., 2004), which  
177 slows down the sample flow from aircraft speeds to 5 m/s generating minimal turbulence  
178 and improving particle transmission.

179 Gas-phase measurements were: (1) the greenhouse gases carbon dioxide (CO<sub>2</sub>) and  
180 methane (CH<sub>4</sub>) with wavelength scanned cavity ringdown spectroscopy, (2) two  
181 measurements of nitric oxide (NO) and O<sub>3</sub>, each measured by gas-phase  
182 chemiluminescence (CL) and by cavity ringdown absorption spectroscopy (CRDS), three  
183 measurements of nitrogen dioxide (NO<sub>2</sub>), by UV photolysis and gas-phase  
184 chemiluminescence (P-CL) and by CRDS and by airborne cavity enhanced absorption  
185 spectroscopy (ACES), NO<sub>y</sub> by gold-catalyzed thermal conversion and gas-phase CL, (3)  
186 carbon monoxide (CO) with vacuum UV resonance fluorescence, (4) SO<sub>2</sub> with pulsed UV  
187 fluorescence, (5) ammonia (NH<sub>3</sub>), nitric acid (HNO<sub>3</sub>), and two measurements of nitrous  
188 acid (HONO), and formic acid (HCOOH) with chemical ionization mass spectrometry  
189 (CIMS), and (6) the nighttime oxidants NO<sub>3</sub> and N<sub>2</sub>O<sub>5</sub> with CRDS and CIMS. Various volatile  
190 organic compounds (VOCs) were measured with several different techniques: (7)



191 oxygenates, aromatics, isoprene, monoterpenes and acetonitrile with Proton-Transfer-  
 192 Reaction Mass Spectrometry (PTR-MS); (8) hydrocarbons, halocarbons and a few selected  
 193 oxygenates from canister samples and post-flight GC-MS analysis (iWAS/GCMS); (9)  
 194 formaldehyde with the In Situ Airborne Formaldehyde (ISAF) using laser induced  
 195 fluorescence (LIF); (10) glyoxal with ACES; (11) organic and inorganic acids by UW-TOF-  
 196 CIMS; and (12) peroxyacyl nitrates PANs and nitryl chloride ( $\text{ClNO}_2$ ) with a separate CIMS.  
 197 In addition up and down welling photolysis rates ( $j_{\text{NO}_2}$  and  $j_{\text{O}_3}$ ) were measure with filter  
 198 radiometers.

199

## 200 **5. Inter-comparison of Duplicate Measurements on the WP-3D**

201 Some parameters were measured by more than one instrument on the WP-3D,  
 202 giving opportunities for inter-comparisons and the results are described in the following.

203 Three instruments measured  $\text{NO}_2$ : P-CL, CRDS, and ACES. The agreement between  
 204 CRDS and ACES with the standard P-CL technique, as shown in Figure 3, was on average  
 205 6% and 10% and the measurements were correlated with a linear correlation coefficient  
 206 ( $R^2$ ) of 0.99 and 0.93, respectively. The agreement is within the combined uncertainties,  
 207 given in Table 6, for CRDS and just outside for ACES and P-CL. Two instruments measured  
 208 ozone: P-CL and CRDS and the inter-comparison is also shown in Figure 3. The ozone  
 209 measurements correlated with  $R^2$  of 0.96 and agreed on average within 8%, which is  
 210 within the combined measurement uncertainties of the two instruments as given in Table  
 211 6. All the data for the whole campaign were included for this inter-comparison using 1-  
 212 second ozone data;  $\text{NO}_2$  data were averaged to the 5-second ACES time resolution. Two  
 213 instruments measured NO: CL and CRDS, with the CRDS data subject to an optical  
 214 instability that degraded the detection limit during this campaign. The large majority of  
 215 the data were below this degraded detection limit, and therefore the inter-comparison  
 216 was not included here.

217 Several VOCs were measured on the WP-3D with both the PTR-MS and with  
 218 iWAS/GCMS. As an example the isoprene time series for the flight on June 29, 2013 is  
 219 shown for both instruments in Figure 4. For the purpose of this comparison, the PTR-MS  
 220 data are averaged over an interval that starts 10 s before and stops 10 s after the canister  
 221 filling time to ensure at least one PTR-MS data point was used in the comparison. Isoprene





222 has a very high variability in the boundary layer, due to its short lifetime and high  
223 emissions. This variability and imperfect time alignment causes a large part of the scatter  
224 observed in Figure 4. The scatter plots for the inter-comparison of isoprene and other  
225 VOCs are shown in Figure 4 as well. The comparison had slopes between 0.64-1.45, which  
226 is just within the combined uncertainties of the two instruments given in Table 6, and  $R^2$   
227 of 0.5 or higher. The iWAS/GCMS was deployed during SENEX for the first time and some  
228 instrument issues occurred, causing some degradation of the data quality compared to  
229 previous inter-comparisons (de Gouw and Warneke, 2007; Warneke et al., 2011). More  
230 details on the instrument performance during SENEX, the inter-comparison and the  
231 stability of VOCs, especially oxygenates, in canisters can be found in Lerner et al. (2015).

232 Two instruments measured formic acid ( $\text{HCOOH}$ ): the  $\text{HNO}_3$ -CIMS and the  
233 University of Washington high-resolution time-of-flight chemical ionization mass  
234 spectrometer (UW HR-ToF-CIMS) and their comparison is shown in Figure 5. The time  
235 series shows results from one individual flight and the scatter plot shows all data from the  
236 campaign, where the color code indicates the individual flights. The comparison using all  
237 the data has a slope of 1.03 and  $R^2$  of 0.80, while the slopes of individual flights ranged  
238 from 1.40 to 0.66 with  $R^2$  always higher than 0.91. The reason for the flight-to-flight  
239 variability in their agreement is yet unknown. The output of the continuously added  $^{13}\text{C}$   
240 formic acid permeation device – to which the UW HR-ToF-CIMS instrument sensitivity was  
241 referenced (see SI) – may have contributed to the variability of the reported formic acid  
242 mixing ratio between flights, because an independent method of quantification of its  
243 output was not available (Veres et al., 2010). Cross calibrations were not conducted  
244 between the two instruments during the campaign and therefore do not allow direct  
245 comparisons of instrument sensitivity on a flight-to-flight basis. Nevertheless, the  
246 variability between the two measurements is within the combined uncertainties of the  
247 two instruments ( $\pm 20\%$  for  $\text{HNO}_3$ -CIMS and  $\pm 50\%$  for UW HR-ToF-CIMS).

248 During the night flights two instruments measured  $\text{ClNO}_2$ : the UW HR-ToF-CIMS  
249 and the PAN-CIMS and  $\text{N}_2\text{O}_5$  was measured with the UW HR-ToF-CIMS and CRDS. The  
250 comparison is shown in Figure 6 as time series and scatter plots for the flight on  
251 07/03/2013. The slopes are 1.19 and 0.91 and the  $R^2$  0.74 and 0.92, respectively. For  
252 small signals such as  $\text{ClNO}_2$ , the signal to noise of the UW HR-ToF-CIMS is aided by its





253 ability to distinguish isobaric contaminants from halogen containing molecules, which  
 254 have a distinct mass defect (Kercher et al., 2009; Lee et al., 2014). The scatter plot displays  
 255 some non-linearity and the  $\text{N}_2\text{O}_5$  is just outside the range of a previous comparison (Chang  
 256 et al., 2011), but the results are within the combined uncertainties of the instruments  
 257 given in Table 6.

258 Figure 7 shows the  $\text{NO}_y$  budget for all the individually measured  $\text{NO}_y$  species  
 259 compared to the measured total  $\text{NO}_y$  for the NOAA WP-3D flight on June 16, 2013. Aerosol  
 260 nitrate might contribute about 2% to the sum. This assumes a quantitative sampling and  
 261 conversion of aerosol nitrate. This is likely not the case and  $\text{NO}_y$  from aerosol nitrate is  
 262 likely an upper limit and the data are shown with and without the potential aerosol  
 263 contribution. The highest mixing ratios of  $\text{NO}_y$  are observed in power plant plumes, where  
 264 most  $\text{NO}_y$  consists of  $\text{NO}_x$ . For a more detailed comparison the  $\text{NO}_z (= \text{NO}_y - \text{NO}_x)$  budget is  
 265 shown in Figure 7 as well. The power plant plumes were removed for this comparison,  
 266 because the time resolution and the accuracy of  $\text{NO}_y$  and  $\text{NO}_x$  are not high enough to  
 267 calculate small differences in  $\text{NO}_z$  during these periods with very high  $\text{NO}_x$  mixing ratios.  
 268 On this flight the sum of individually measured  $\text{NO}_y$  constituents was roughly 90% of the  
 269 total measured as  $\text{NO}_y$ , similar to the whole campaign  $\text{NO}_y$  budget. The unmeasured  $\text{NO}_y$   
 270 outside power plants was about 25% (or 15%, when including aerosol nitrate). This  
 271 missing fraction may be comprised largely of organic nitrates derived from the oxidation  
 272 of isoprene and monoterpene (Lee et al., 2014).

273 The aerosol volume derived from the chemical composition data (AMS and SP2)  
 274 was compared to the volume derived from the measured size distributions, following  
 275 Middlebrook et al. (2012). All of these measurements sampled aerosol downstream of a 1  
 276 micron impactor. For each 10-s AMS measurement, the composition-derived volume was  
 277 calculated by adding the average rBC mass from the SP2 instrument to the AMS total  
 278 aerosol mass and dividing it by the density estimated from the AMS and BC composition.  
 279 The mass-weighted density ( $\rho$ ) was calculated using  $\rho_{\text{org}} = 1.25 \text{ g cm}^{-3}$  (Cross et al., 2007;  
 280 Kiendler-Scharr et al., 2009; Zelenyuk et al., 2008),  $\rho_{\text{inorg}} = 1.75 \text{ g cm}^{-3}$  (primarily dry  
 281 ammonium sulfate, (Perry and Green, 1997)), and  $\rho_{\text{BC}} = 1.8 \text{ g cm}^{-3}$  (Park et al., 2004), for  
 282 organic mass, inorganic mass, and BC, respectively. The measured AMS lens transmission  
 283 curve (Bahreini et al., 2008) was applied to the particle number distributions to account



for particle transmission losses in the AMS lens before calculating the volume from the size distributions, which were also averaged over the AMS sampling time. For this field project, the fraction of aerosol volume behind the 1 micron impactor that was transmitted into the AMS instrument by the lens was on average 99% with a minimum of 92%.

The composition-derived volume was then plotted against the volume calculated from the size distributions for each flight with available data. Resulting slopes are depicted in Figure 8 as a function of flight date color coded with the linear correlation coefficient  $R^2$ . The grey bands indicate the overall combined  $2\sigma$  uncertainty of  $\pm 60\%$  (Bahreini et al., 2009; Brock et al., 2011; Schwarz et al., 2006). The volumes from most of the flights agree within this combined uncertainty and with  $R^2$  values between 0.62 to 0.98, indicating that most of the aerosol in the AMS lens transmission size-range was composed of non-refractory material and black carbon. Only the slopes for flights on 29 June 2013 was outside the uncertainty band. We note that rBC only contributed 1% on average to the total accumulation mode mass, and in 1-min averages only exceeded 3% less than 1% of the time during SENEX.

On June 29, 2013 the NOAA WP-3D and the NSF NCAR C-130 did coordinated wing-to-wing flight legs in the free troposphere and the boundary layer for an inter-comparison in southern Tennessee and northern Alabama with a duration of just over one hour. Several over-flights over the SOAS ground site in Centreville were performed during SENEX. Results of the platform inter-comparisons will not be presented here.

## 6. Example Flight on 16 June 2013 near Atlanta, GA

Results from the SENEX research flight on 16 June 2013 are presented here to demonstrate the strategy used to address many of the SENEX science questions such as the determination of anthropogenic and biogenic emissions, and the subsequent atmospheric chemistry, transformation, and production of secondary species. Flights over the shale gas regions will not be discussed here, but calculations of the methane emission fluxes from the three shale gas regions can be found elsewhere (Peischl et al., 2015; Yuan et al., 2015). The major goal of the 16 June 2013 flight was to investigate the Atlanta urban plume and the Scherer and Harllee Branch power plant plumes as they were transported over heavily forested areas in Georgia with strong biogenic emissions.



315

## 316 6.1 Anthropogenic, biogenic and point source emissions

317 Figure 9a shows the WP-3D flight track over Atlanta and surrounding areas color-  
 318 coded by  $\text{NO}_y$  on top of a map showing anthropogenic emission sources, which are the  
 319 urban areas and point sources: power plants, landfills, paper mills and coal mines. Other  
 320 point sources studied that are not shown on this map include bio-fuel refineries (de Gouw  
 321 et al., 2015a). The point sources are sized by their respective emission strengths or  
 322 capacity. The flight included eight tracks perpendicular to the wind direction (numbered  
 323 0-7 in Figure 9a): one upwind of Atlanta, three over the metro area and four downwind.  
 324 The flight tracks were set such that the distance between each leg represents about 1 hour  
 325 of transport at the prevailing wind speed and also such that many of the point source  
 326 plumes were intercepted.

327 Figure 10 shows results for the intercepts of such point source plumes. In Figure  
 328 10a the methane measurements along transect 4 downwind of the Pine Bluff landfill in  
 329 Georgia are shown. Landfills are an important source of methane in the US, but they do not  
 330 emit many other compounds and indeed methane was the only species measured aboard  
 331 the WP-3D payload that showed a detectable enhancement in the plume. The forested  
 332 Southeast US is heavily managed for large-scale wood and wood products and therefore  
 333 has a large density of pulp and paper mills. Pulp and paper mills use a significant amount  
 334 of energy, which they often produce partially on site. For example the investigated facility  
 335 has four steam producing boilers at close to 80 MWh that mainly burn coal, natural gas, oil  
 336 and wood/bark waste biomass. The power production results in emissions of the  
 337 combustion species  $\text{NO}$ ,  $\text{NO}_2$ ,  $\text{CO}$ ,  $\text{SO}_2$  and  $\text{CO}_2$  (only  $\text{NO}$  is shown in Figure 10b). The paper  
 338 mill plumes were intercepted on transect 0 during this flight. High mixing ratios of  
 339 monoterpenes, methanol and acetaldehyde were also observed downwind of those  
 340 facilities (Figure 10b).

341 U.S. urban emissions, and therefore urban mixing ratios of many air pollutants have  
 342 decreased significantly over the last few decades (Dallmann and Harley, 2010; Emmons et  
 343 al., 2015; von Schneidemesser et al., 2010; Warneke et al., 2012). For example, Warneke et  
 344 al. (2012) analyzed 50 years of ambient measurements and found that VOCs and  $\text{CO}$  have  
 345 decreased at an annual rate of about 7.5% in Los Angeles, CA. Blanchard et al. (2015)



analyzed Southeastern Aerosol Research and Characterization (SEARCH) network data and found downward trends in ambient carbon monoxide (CO), sulfur dioxide (SO<sub>2</sub>), and oxidized nitrogen species (NO<sub>y</sub>) concentrations averaged  $1.2 \pm 0.4$  to  $9.7 \pm 1.8\%$  per year from 1999 to 2010. The NOAA WP-3D flew over Atlanta, GA during SOS (Southern Oxidant Study 1999) on 6 July 1999 and the results are shown in Figure 11 and are compared to the SENEX 16 June 2013 data. These two days were comparable in meteorological conditions with wind speeds around 4 m/s, temperatures around 26 degrees C in the boundary layer, and boundary layer heights of about 1.6 km on 6 July 1999 and 1-1.2 km on 16 June 2013. The flight track on top of the map color coded with 1999 NO<sub>y</sub> shows qualitatively that the pollution was more intense and widespread. The time series of CO and NO<sub>y</sub> for the two flights in Figure 11 are consistent with significant emissions decreases between 1999 and 2013. It is expected that the comparison between the 1999 and 2013 airborne data sets will provide important insights and evidence to answer the main science questions from SENEX.

360

## 6.2 Coal and natural gas fired power plant plumes

During SENEX several power plant plumes were sampled. Figure 12 shows the flight track from the 22 June 2013 over Atlanta that included transects downwind of the coal fired Bowen and the natural gas combined cycle McDonough power plants. The emission intensities of these two different kinds of power plants are very different; combined cycle natural gas power plant have much lower CO<sub>2</sub>, SO<sub>2</sub> and NO<sub>x</sub> emissions per unit energy produced than coal fired power plants (de Gouw et al., 2014). The Bowen power plant produced 3.3 TWh and McDonough 4.7 TWh in the 1<sup>st</sup> quarter of 2013. According to the continuous emissions monitoring systems (CEMS) monitoring data, during the 1<sup>st</sup> quarter of 2013 the Bowen power plant emitted 930 g/kWh CO<sub>2</sub>, 0.20 g/kWh SO<sub>2</sub> and 0.56 g/kWh NO<sub>x</sub>, while McDonough emitted 360 g/kWh CO<sub>2</sub>, 0.0019 g/kWh SO<sub>2</sub> and 0.018 g/kWh NO<sub>x</sub>. These large differences in emission intensities are clearly reflected in the enhancements measured in the downwind transects shown in Figure 12. In the Bowen power plant plume about 20 ppmv CO<sub>2</sub>, 5 ppbv NO<sub>y</sub> and 4 ppbv SO<sub>2</sub> enhancements were observed, while the McDonough plume had only about 5 ppmv of CO<sub>2</sub> enhancement and SO<sub>2</sub> and NO<sub>y</sub> were not measurably enhanced above background. To



account for the different dilutions during transport (5km distance for Bowen and 10 km for McDonough at about 3m/s average wind speed) enhancement ratios need to be considered. In the Bowen plume 0.24 ppb/ppm of  $\text{NO}_y/\text{CO}_2$  and 0.13 ppb/ppm of  $\text{SO}_2/\text{CO}_2$  were measured. Because no enhancements in the McDonough plume were seen, enhancement ratios cannot be determined, but using a  $S/N=2$  the upper limit for enhancement ratios in the McDonough plume are 0.06 ppb/ppm for  $\text{NO}_y/\text{CO}_2$  and 0.11 ppb/ppm for  $\text{SO}_2/\text{CO}_2$  are determined. This shows that the  $\text{NO}_y$  and  $\text{SO}_2$  enhancements in the gas fired McDonough plant are clearly smaller than in the coal fired Bowen plant. In addition to investigating emissions from the power plant plumes as was shown here, the emissions of those power plants mix with the large emissions of isoprene in this area as can be seen in Figure 9. This provides an ideal case for studying the interactions between natural and anthropogenic emissions. The chemistry of isoprene, OH, formaldehyde and  $\text{NO}_x$  in power plant plumes and other areas during SENEX will be described in detail elsewhere (de Gouw et al., 2015b; Kaiser et al., 2015; Wolfe et al., 2015).

391

### 392 6.3 Modeling Support for SENEX

Figure 13 shows example results from one of the modeling tools that is available for analysis of the SENEX results, the Lagrangian particle dispersion model FLEXPART (Stohl et al., 2005). From locations along the flight tracks, 20,000 particles were released in the model and tracked for 10 days backward in time. The model outputs the residence time of the particles in a volume such as the surface layer. By multiplying the footprint with gridded emission fluxes the model calculates the mixing ratio of the emitted species at the location of the aircraft. All species are considered as conserved tracers; the model does not contain chemical transformations, but it does keep track of the time since emission. As an example, Figure 13 a and b show the time series of FLEXPART  $\text{NO}_y$  (accumulating emissions from the previous 48 hours) together with the flight track color coded with  $\text{NO}_y$ . Comparing the modeled and measured  $\text{NO}_y$  in Figure 13a and Figure 9, it can be seen that the model reproduces the time series qualitatively, including the broader features and the power plant plume encounters. The very high mixing ratios in the narrow power plant plumes are underestimated in the model (the plumes are too narrow for the model resolution). The footprint map for a point along the last flight track downwind of



the Harllee Branch power plant plume is shown in Figure 13c showing that the mixing ratios at this point along the flight track will have the highest contribution from the immediate upwind area that includes the Harllee Branch power plant, just as expected. But there was also a significant contribution to the mixing ratios from long-range transport from the Northeast US. Other available FLEXPART model outputs include CO, biomass burning CO, SO<sub>2</sub>, isoprene and monoterpenes. Besides FLEXPART other models are (or will be) also available including the NOAA AM3 model (<http://esrl.noaa.gov/csd/projects/senex/>), an MCM-based 0-D box model (Wolfe et al., 2015) and WRF-Chem (Weather Research and Forecasting with Chemistry) simulations.

417

## 418 7. Summary

The Southeast Atmosphere Study (SAS) was a large collaborative and community effort to understand the air quality and climate issues in the Southeast United States. This paper provides a summary of the experimental setup for the NOAA-led SENEX study, which was an important component of the SAS. The NOAA WP-3D aircraft capabilities, the payload, instrument descriptions, inter-comparisons and flight locations and goals are described in detail in this paper. The flight on 16 June 2013 in the Atlanta area was described in some detail to demonstrate the strategies used during SENEX to study the air quality and climate relevant interactions of biogenic and anthropogenic emissions in the Southeast, which was one of the main foci of the SAS study.

428

## 429 ACKNOWLEDGMENT

The US Weather Research Program within NOAA/OAR Office of Weather and Air Quality supported S. McKeen and R. Ahmadov. We are grateful M. Dumas (NOAA Holling's Scholar), D. Hughes, and A. Jaksich from Hendrix College for their help with the iWAS2 measurements. Participation of ISAF was enabled by US EPA Science to Achieve Results (STAR) program grant 83540601.

435



## 436 Tables

437

438 **Table 1:** Standard NOAA WP-3D provided parameters

439

| Aircraft Parameters    | Technique                      | Units | Uncertainty  |
|------------------------|--------------------------------|-------|--------------|
| aircraft position      | GPS latitude                   | deg   | ±16m         |
|                        | GPS longitude                  | deg   | ±16m         |
|                        | GPS altitude                   | m     | ±16m         |
|                        | pressure altitude              | m     | ±10m         |
|                        | radar altitude above ground    | m     | ±15m or 1-2% |
| aircraft meteorology   | ambient temperature            | deg C | ±0.5C        |
|                        | dew point temperature          | deg C | ±0.5C        |
|                        | TDL dew point temperature      | deg C | 5%           |
|                        | H <sub>2</sub> O mixing ratio* | g/kg  | 5%           |
|                        | potential temperature          | deg K | ±0.5K        |
|                        | relative humidity*             | %     | ±5%          |
|                        | static pressure                | mb    | ±2.2mb       |
|                        | vertical wind speed            | m/s   | ±0.5m/s      |
|                        | wind direction                 | deg   | 5 deg        |
|                        | wind speed                     | m/s   | 1 m/s        |
| aircraft miscellaneous | attack angle                   | deg   | ±0.2 deg     |
|                        | cabin pressure                 | mb    | N/A          |
|                        | ground speed                   | m/s   | ±3.4m/s      |
|                        | heading                        | deg   | ±0.5 deg     |
|                        | pitch angle                    | deg   | ±0.05 deg    |
|                        | roll angle                     | deg   | ±0.05 deg    |
|                        | slip angle                     | deg   | ±0.2 deg     |
|                        | true air speed                 | m/s   | ±0.5 m/s     |

440 \* H<sub>2</sub>O mixing ratio and relative humidity are derived from dew point temperature

441





442 **Table 2:** Flight descriptions for the NOAA WP-3D daytime flights in the SE US  
 443

| Flight Date in 2013 | Day of the week | Description                                                                                                                             | Investigated Emission Source                         |
|---------------------|-----------------|-----------------------------------------------------------------------------------------------------------------------------------------|------------------------------------------------------|
| 5/29                | Wednesday       | Testflight in Florida<br>Jacksonville<br>St John's River                                                                                | biogenic<br>urban<br>power plant                     |
| 5/31                | Friday          | Testflight in Florida<br>Jacksonville<br>St John's River                                                                                | biogenic<br>urban<br>power plant                     |
| 6/03                | Monday          | Transfer Tampa to Smyrna<br>Birmingham<br>EC Gaston, Johnsonville, Cumberland, Colbert<br>Centreville spiral                            | urban<br>power plant<br>coal mine                    |
| 6/11                | Tuesday         | Centreville<br>Birmingham west to east<br>EC Gaston                                                                                     | urban<br>power plant                                 |
| 6/12                | Wednesday       | Atlanta west to east<br>Scherer, Bowen, Yates, Wansley, Harlee Branch                                                                   | urban<br>power plant                                 |
| 6/16                | Sunday          | Atlanta southwest to northeast on weekend<br>Scherer, Bowen, Yates, Wansley, Harlee Branch<br>paper mills, landfills<br>poultry farming | urban<br>power plant<br>point sources<br>agriculture |
| 6/18                | Tuesday         | Aborted flight, circled over Franklin                                                                                                   |                                                      |
| 6/22                | Saturday        | Birmingham and Atlanta west to east<br>Centreville<br>EC Gaston<br>coal mines, land fills, paper mills                                  | urban<br>power plant<br>point sources                |
| 6/23                | Sunday          | Indianapolis<br>biogenic/landscape emission change<br>Johnsonville, Cumberland                                                          | urban<br>biogenic<br>power plant                     |
| 6/29                | Saturday        | Centreville<br>C-130 inter-comparison<br>Birmingham<br>James H Miller Jr, EC Gaston                                                     | urban<br>power plant                                 |
| 7/05                | Friday          | Ozarks<br>St Louis<br>Archer Daniels Midland biofuel refinery                                                                           | biogenic<br>urban<br>point source                    |
| 7/10                | Wednesday       | Transfer flight Smyrna to Tampa<br>coal mines, paper mill<br>hog farming                                                                | point sources<br>agriculture                         |

444  
 445  
 446



**Table 3:** Flight descriptions for the NOAA WP-3D nighttime flights in the SE US

| Flight Date in 2013 | Day of the week | Description                                                                                       | Investigated Emission Sources   |
|---------------------|-----------------|---------------------------------------------------------------------------------------------------|---------------------------------|
| 6/19                | Wednesday       | Atlanta day into night<br>Missed approaches<br>step profile in aged Atlanta plume                 | urban                           |
| 7/02                | Tuesday         | Birmingham north to south<br>Centreville<br>JH Miller, EC Gaston, Gorgas, US Steel, Greene County | urban<br>power plants           |
| 7/03                | Wednesday       | New Madrid, White Bluff<br>agricultural fire                                                      | power plants<br>biomass burning |

**Table 4:** Flight descriptions for the NOAA WP-3D flights in shale gas regions

| Flight Date in 2013 | Day of the week | Shale Play   | Additional Investigated Emission Sources        |
|---------------------|-----------------|--------------|-------------------------------------------------|
| 6/10                | Monday          | Haynesville  |                                                 |
| 6/25                | Tuesday         | Haynesville  |                                                 |
| 6/26                | Wednesday       | Fayetteville | Biogenics in Ozarks<br>Independence power plant |
| 7/06                | Saturday        | Marcellus    |                                                 |
| 7/08                | Monday          | Fayetteville | New Madrid power plant                          |



**Table 5:** Aerosol instrumentation on the NOAA WP-3D during SENEX

| Measurement                                                                                        | Name/Technique                                                                                                   | Accuracy                      | Precision                                                                 | Sample Interval | Reference                                     |
|----------------------------------------------------------------------------------------------------|------------------------------------------------------------------------------------------------------------------|-------------------------------|---------------------------------------------------------------------------|-----------------|-----------------------------------------------|
| Low turbulence inlet                                                                               | LTI: decelerating inlet to provide sample air to aerosol instruments in fuselage                                 | N/A                           | N/A                                                                       | N/A             | (Wilson et al., 2004)                         |
| Size distributions fine (0.004-1 $\mu$ m) and coarse (1-8.3 $\mu$ m)                               | parallel CPCs, and white and laser light scattering                                                              |                               |                                                                           | 1s              | (Brock et al., 2011; Brock et al., 2000)      |
| Cloud condensation nuclei (CCN) and supersaturation                                                | CCN: Continuous-flow streamwise thermal-gradient CCN counter with scanning flow CCN analysis                     | 10%<br>0.04% super-saturation | 10 CCN cm <sup>-3</sup>                                                   | 30-60s          | (Lance et al., 2006; Roberts and Nenes, 2005) |
| 8 cell optical extinction (dry 405, 532, 662nm, 70% and 90% RH 532nm, thermodenuded 405 and 662nm) | CRD: Cavity ringdown aerosol extinction spectrometer                                                             | <2%                           | 10% 0.1 Mm <sup>-1</sup>                                                  | 1s              | (Langridge et al., 2011)                      |
| 5 cell optical absorption (dry 405, 532, 662nm, thermodenuded 405nm and 662nm)                     | PAS: Photoacoustic Absorption Spectrometer                                                                       | 10 %                          | 1 Mm <sup>-1</sup>                                                        | 1s              | (Lack et al., 2012)                           |
| Refractory BC mass content of individual particles                                                 | SP2: Single-Particle Soot Photometer with laser-induced incandescence                                            | 30%                           | 0.5 fg (0.08 $\mu$ m mass-equiv. diameter with 2 g/cc density)            | 1s              | (Schwarz et al., 2008; Schwarz et al., 2010)  |
| Non-refractory, submicron sulfate, nitrate, ammonium, organic and chloride mass concentrations     | AMS: Aerosol Mass Spectrometer                                                                                   | 50%                           | 0.05, 0.07, 0.24, 0.36, and 0.05 $\mu$ g sm <sup>-3</sup> (study average) | 10s             | (Bahreini et al., 2009)                       |
| Cloud particle size distribution (0.6-50 $\mu$ m) (3-50 $\mu$ m) (50-6000 $\mu$ m)                 | Cloud probes: Laser light forward and back scattering<br>Laser light forward scattering<br>Droplet imaging probe |                               |                                                                           | 1s              | (Lance et al., 2010)                          |

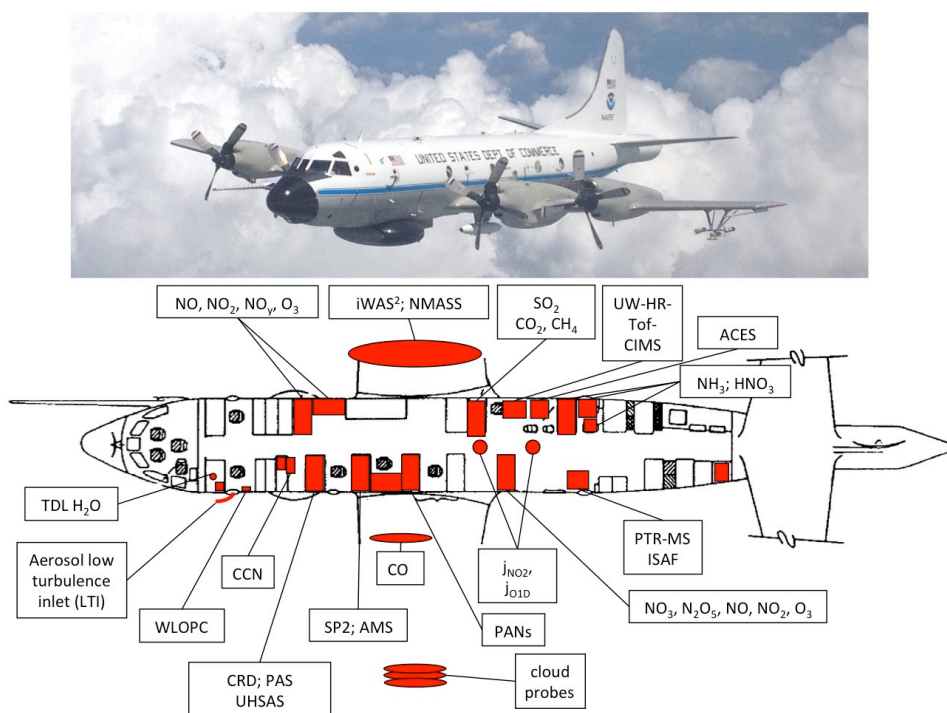


**Table 6:** Gas-phase instrumentation on the NOAA WP-3D during SENEX

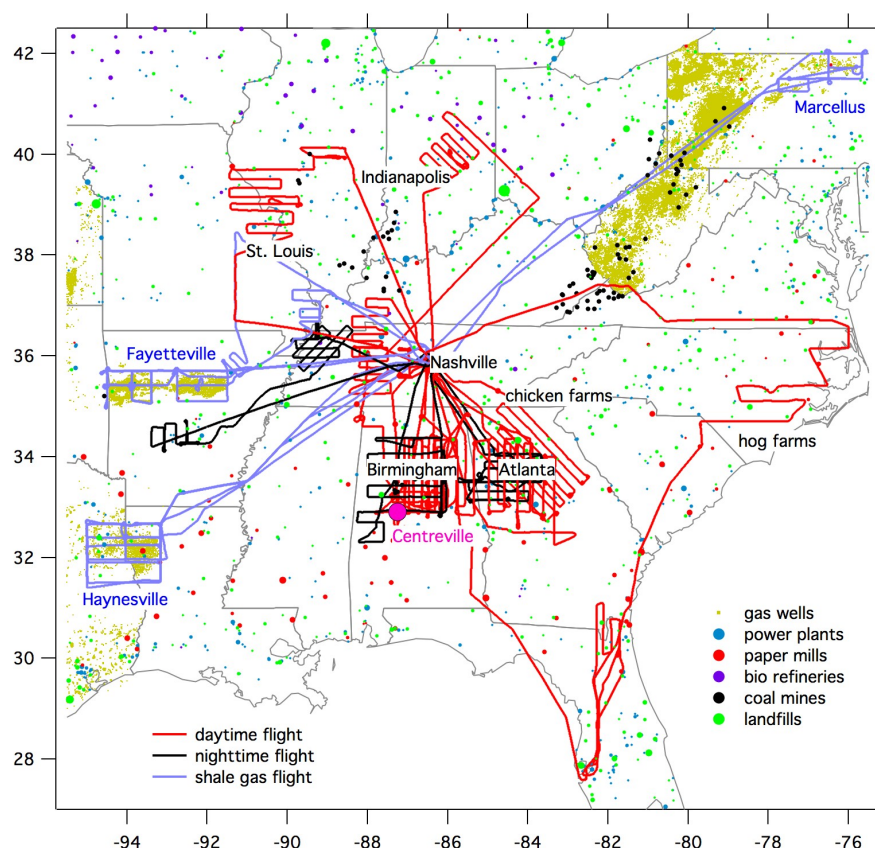
| Measurement                                                                                     | Technique                                                                                                  | Accuracy                                                    | Precision or Detec. Limit                          | Sample Interval  | Reference                                                          |
|-------------------------------------------------------------------------------------------------|------------------------------------------------------------------------------------------------------------|-------------------------------------------------------------|----------------------------------------------------|------------------|--------------------------------------------------------------------|
| CH <sub>4</sub><br>CO <sub>2</sub>                                                              | wavelength-scanned cavity ring-down absorption spectroscopy                                                | 0.07 ppm<br>1 ppb                                           | 0.11 ppm<br>0.4 ppb                                | 1 s              | (Peischl et al., 2012)                                             |
| CO                                                                                              | vacuum UV resonance fluorescence                                                                           | 5%                                                          | 0.5ppb                                             | 1 s              | (Holloway et al., 2000)                                            |
| SO <sub>2</sub>                                                                                 | pulsed UV fluorescence                                                                                     | 20%                                                         | 250ppt                                             | 1 s              | (Ryerson et al., 1998)                                             |
| NO<br>NO <sub>2</sub><br>NO <sub>y</sub><br>O <sub>3</sub>                                      | Gas phase chemiluminescence                                                                                | 3%<br>4%<br>12%<br>2%                                       | 10ppt<br>30ppt<br>40ppt<br>15ppt                   | 1 s              | (Pollack et al., 2010; Ryerson et al., 1998; Ryerson et al., 1999) |
| various VOCs                                                                                    | PTR-MS: proton transfer reaction mass spectrometer using H <sub>3</sub> O <sup>+</sup> as reagent ion      | 25%                                                         | depending on signal and species                    | 1 s every 17 s   | (de Gouw and Warneke, 2007)                                        |
| hydrocarbons, oxygenated VOCs                                                                   | iWAS: whole air sampler with immediate GC-MS analysis                                                      | 12-20%<br>%<br>%                                            | 4-7ppt<br>ppt<br>ppt                               | 72/flight (3-8s) | (Gilman et al., 2009; Lerner et al., 2015)                         |
| HNO <sub>3</sub><br>HCOOH<br>HONO                                                               | HNO <sub>3</sub> -CIMS: chemical ionization mass spectrometer with I <sup>-</sup> as reagent ion           | 20%+50ppt<br>20%+120ppt<br>40%+30 ppt                       | 25 ppt<br>40 ppt<br>25 ppt                         | 1 s              | (Neuman et al., 2002; Neuman et al., 2003)                         |
| NH <sub>3</sub>                                                                                 | NH <sub>3</sub> -CIMS: chemical ionization mass spectrometer with protonated acetone dimers as reagent ion | 25%+(0.02-0.5) ppb (depending on flight)                    | 0.02-0.07 ppb (depending on flight)                | 1 s              | (Neuman et al., 2003; Nowak et al., 2007)                          |
| PAN<br>PPN<br>APAN<br>ClNO <sub>2</sub>                                                         | PAN-CIMS: chemical ionization mass spectrometry with I <sup>-</sup> as reagent ion                         | 0.04-0.05ppb<br>0.04-0.1ppb<br>0.01-0.02ppb<br>0.01-0.02ppb | 0.01ppb<br>0.003ppb<br>0.006ppb<br>0.02ppb         | 2 s              | (Osthoff et al., 2008; Slusher et al., 2004; Zheng et al., 2011)   |
| various oxygenated VOCs<br>ClNO <sub>2</sub><br>N <sub>2</sub> O <sub>5</sub><br>alkyl nitrates | UW HR-ToF-CIMS: chemical ionization mass spectrometer with I <sup>-</sup> as reagent ion                   | 50%                                                         | depending on signal and species                    | 1 s              | (Lee et al., 2014)                                                 |
| glyoxal<br>NO <sub>2</sub>                                                                      | ACES: cavity enhanced absorption spectroscopy                                                              | 5.8%<br>5%                                                  | 34 pptv<br>80 ppt                                  | 10s<br>5s        | (Min et al., 2015; Washenfelter et al., 2011)                      |
| NO<br>NO <sub>2</sub><br>O <sub>3</sub><br>NO <sub>3</sub><br>N <sub>2</sub> O <sub>5</sub>     | CRDS: cavity ring-down absorption spectrometer                                                             | 5%<br>5%<br>10%<br>20%<br>12%                               | 1 ppbv<br>0.2 ppbv<br>0.2 ppbv<br>3 pptv<br>3 pptv | 1 s              | (Dube et al., 2006; Wagner et al., 2011)                           |
| HCHO                                                                                            | In Situ Airborne Formaldehyde (ISAF): laser induced fluorescence                                           | 10%                                                         | 36ppt                                              | 1 s              | (Cazorla et al., 2015; DiGangi et al., 2011; Hottle et al., 2009)  |
| j <sub>NO2</sub> and j <sub>O1D</sub>                                                           | j-heads: filter radiometers                                                                                | 10%                                                         |                                                    | 1 s              |                                                                    |



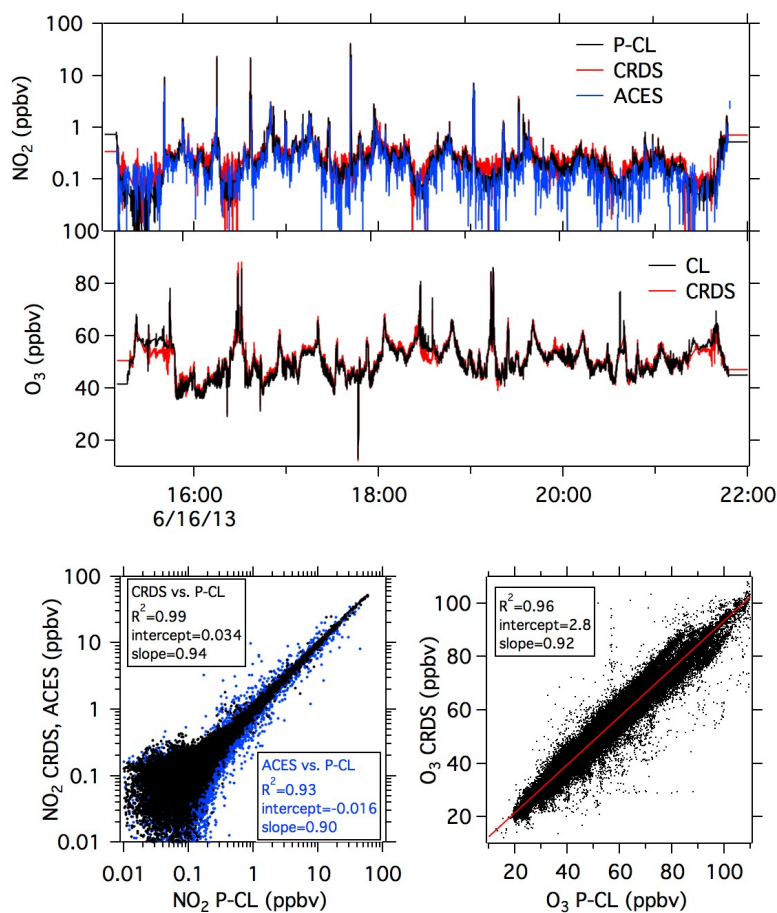
464 **Figures:**  
 465  
 466



467  
 468  
 469 **Figure 1:** NOAA WP-3D aircraft picture, payload and layout. The photo was taken during  
 470 the inter-comparison flight with the NCAR C-130 by Lynne Gratz.  
 471  
 472

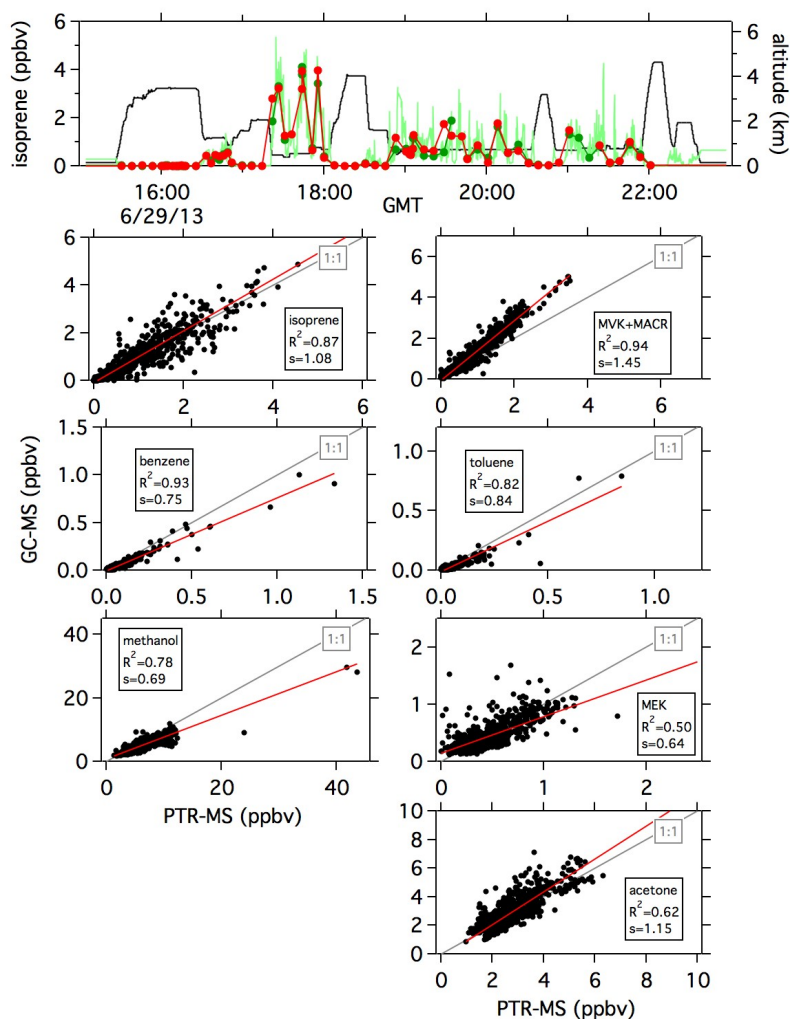


**Figure 2:** NOAA WP-3D flight tracks for daytime, nighttime and shale gas flights during SENEX. The marker size for the power plants is the annual gross load, for the paper mills the capacity, for the bio refineries the biofuel production, for the coal mines the methane emissions, and for the landfills the methane emissions.

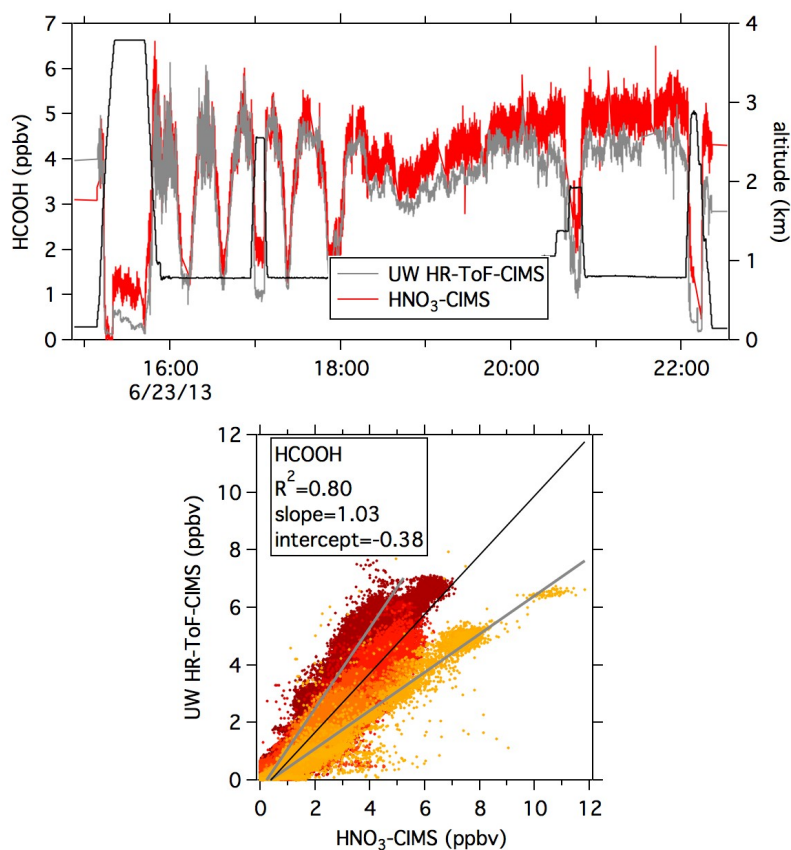


**Figure 3:**  $\text{NO}_2$  inter-comparison between P-CL, CRDS and ACES instruments and ozone inter-comparison between P-CL and CRDS.

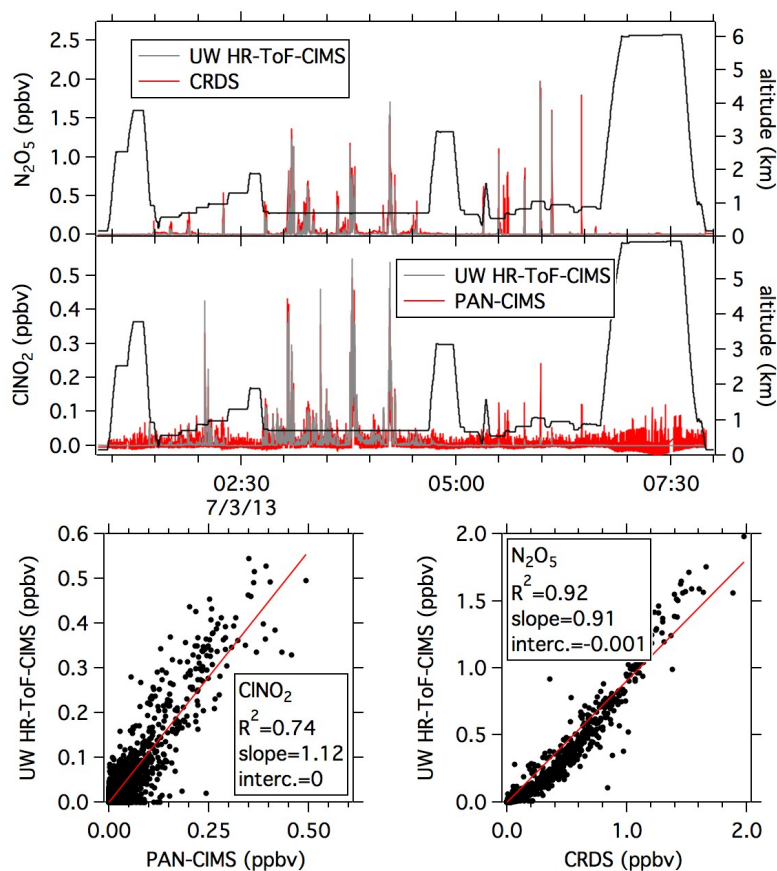




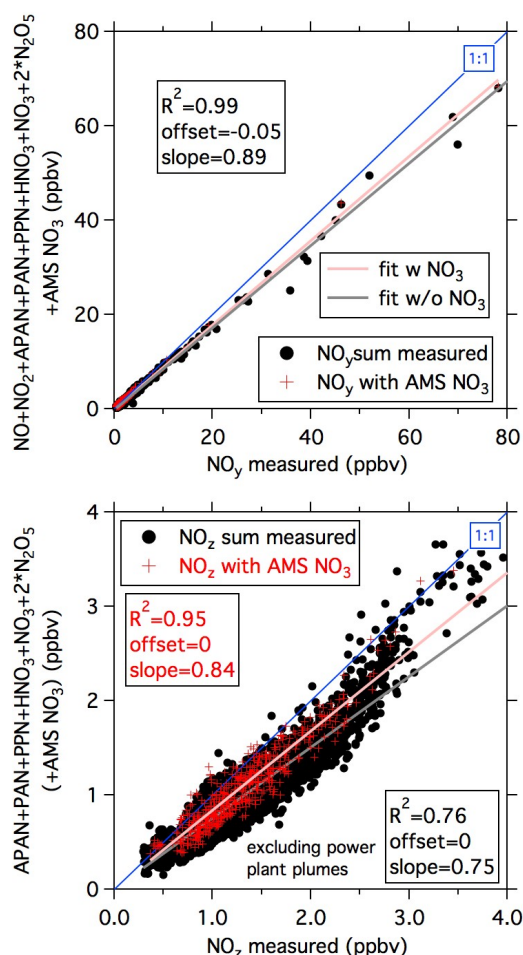
**Figure 4:** Inter-comparison between PTR-MS and iWAS/GCMS.



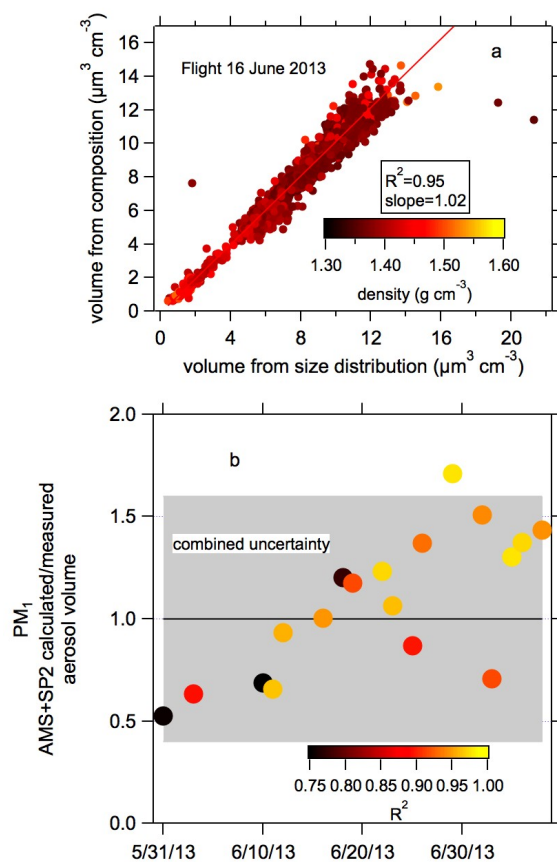
**Figure 5:** HCOOH inter-comparison between the HNO<sub>3</sub>-CIMS and the UW HR-ToF-CIMS as a time series for a selected flight and a scatter plot. The color code in the scatter plot indicates all the individual flights. The black line is a fit using all the data the grey lines fits for individual flights with the highest or lowest slope, respectively.



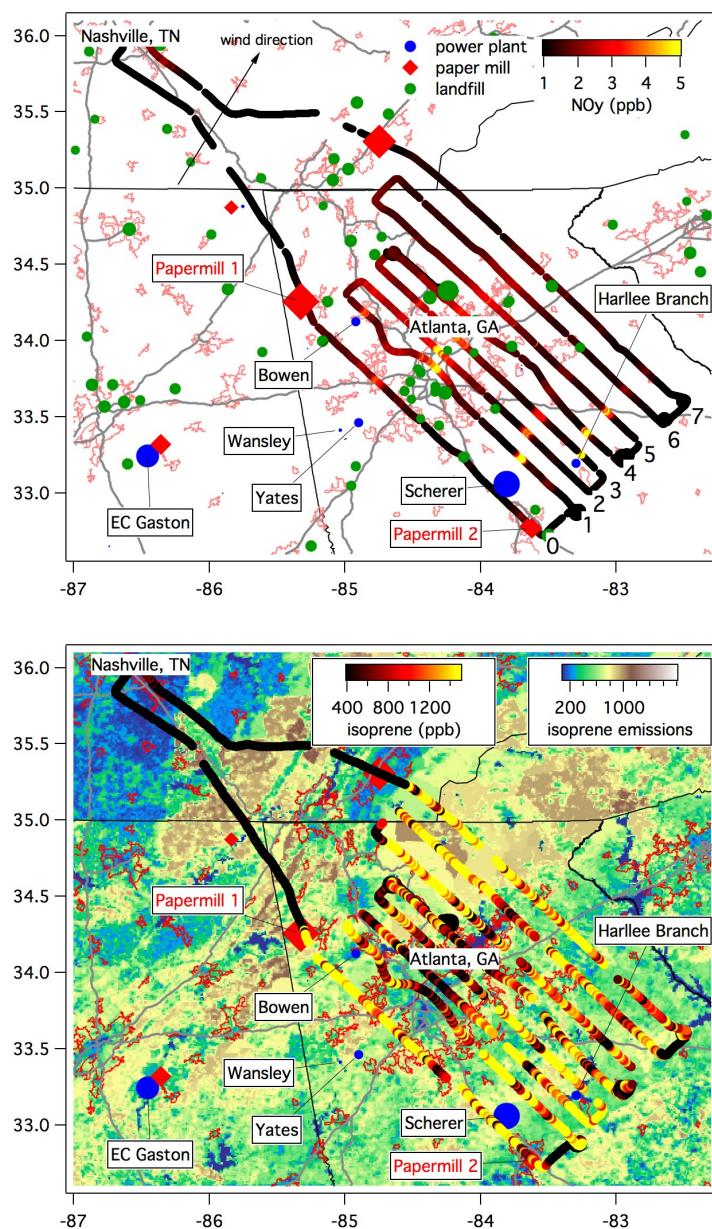
**Figure 6:** Inter-comparison between the UW HR-ToF-CIMS of  $\text{N}_2\text{O}_5$  with CRDS and  $\text{ClNO}_2$  with the PAN-CIMS as time series and scatter plots for the nighttime flight on 3 July 2013.



**Figure 7:**  $\text{NO}_y$  and  $\text{NO}_z$  ( $=\text{NO}_y-\text{NO}_x$ ) budgets for the NOAA WP-3D flight on 16 June 2013 with and without aerosol nitrate.

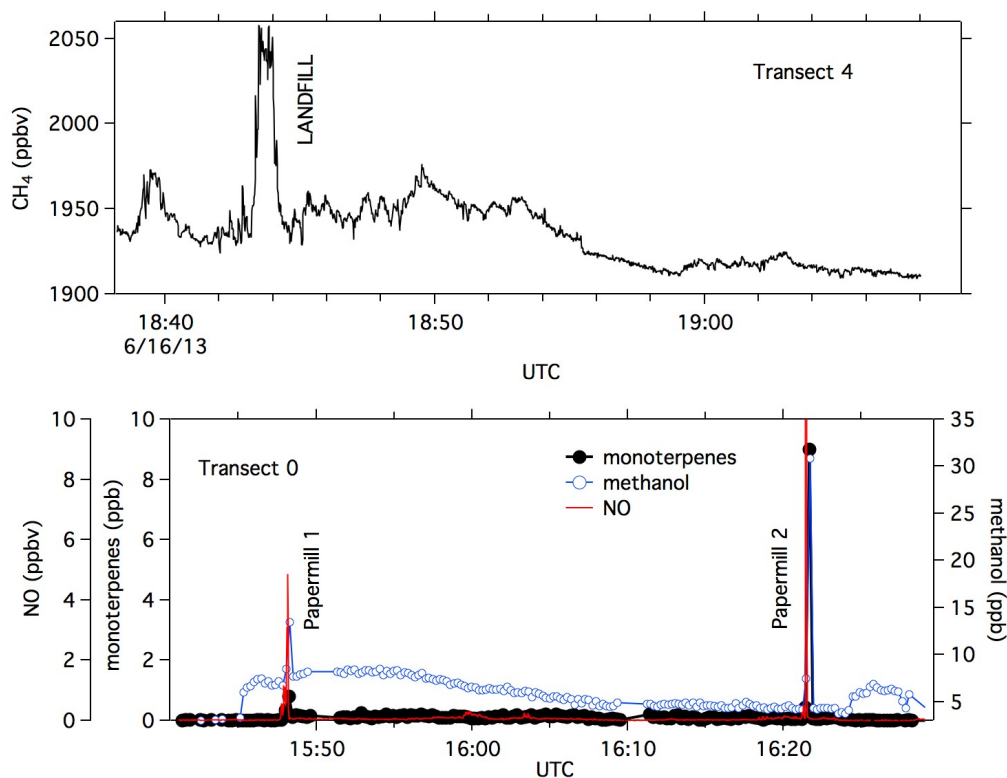


**Figure 8:** The aerosol volume derived from the chemical composition data (AMS and SP2) was compared to the volume from the size distribution data (NMASS and UHSAS). (a) The correlation for the flight on 16 June 2013 color-coded by the density. (b) The slopes for all the flights color-coded by the respective correlation coefficient determined as shown in (a).



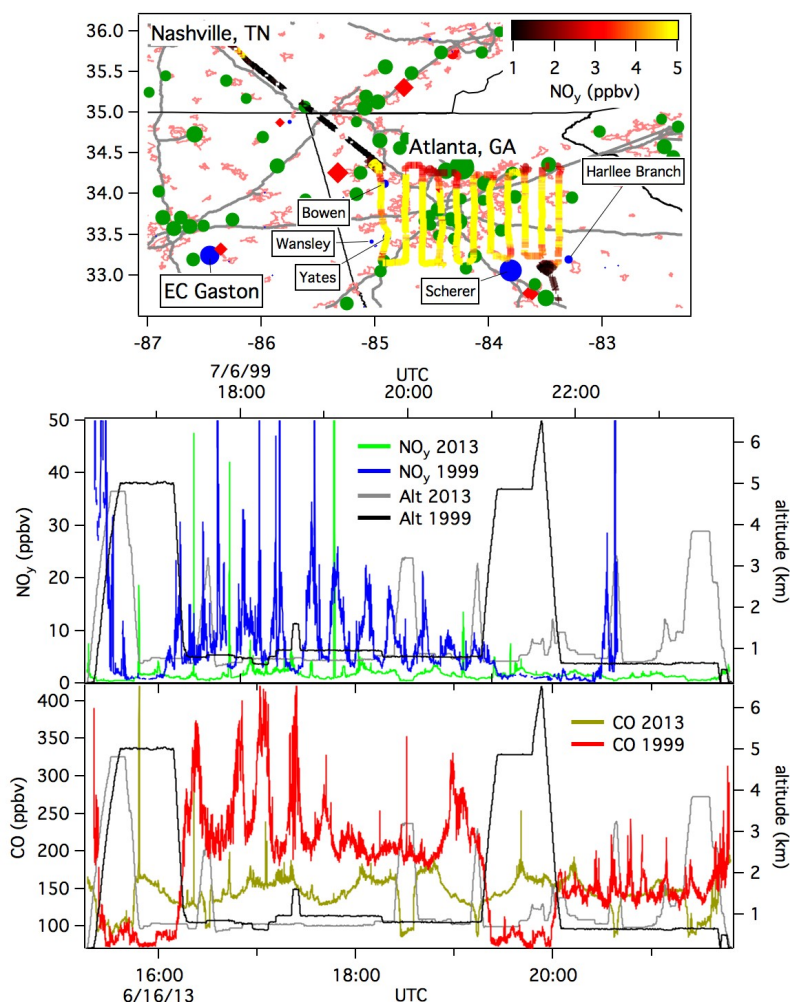
**Figure 9:** The flight track of the NOAA WP-3D on June 16, 2013 over Atlanta, GA color coded with  $\text{NO}_y$  in the top panel and with isoprene on the bottom panel. The underlying maps show the point source emissions (power plants, paper mills and landfills) in the top panel and the isoprene emissions potential in the bottom panel.



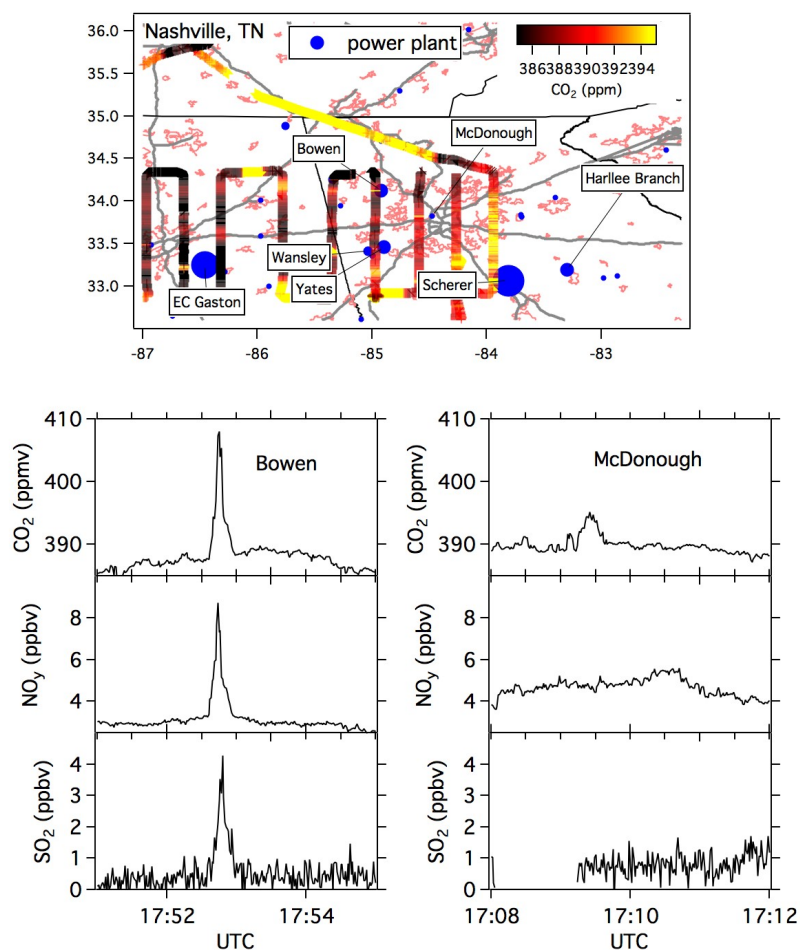


**Figure 10:** Time series of two transects during the 16 June 2013 flight downwind of a landfill and two paper mills.





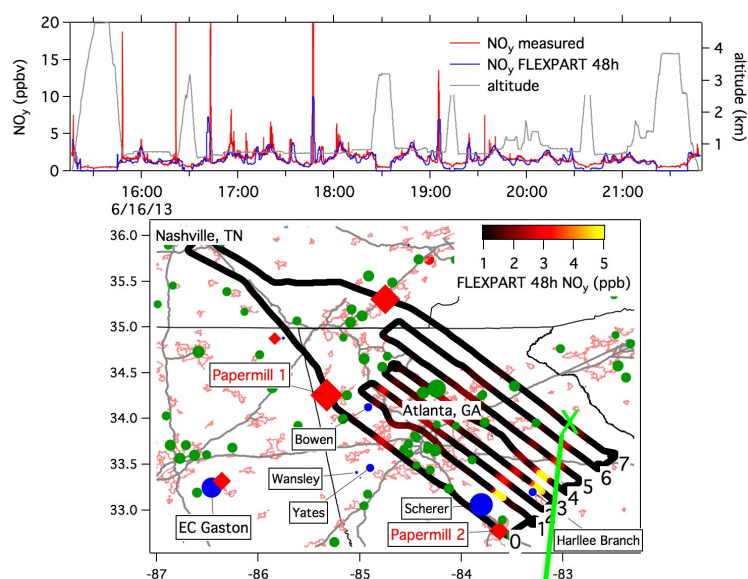
**Figure 11:** The track of a flight on 6 July 1999 over Atlanta during the SOS99 campaign color-coded with the  $\text{NO}_y$  mixing ratio. Time series of the 16 June 2013 and the 6 July 1999 flights for  $\text{NO}_y$  and CO show that the mixing ratios over Atlanta have decreased significantly over the past 14 years.



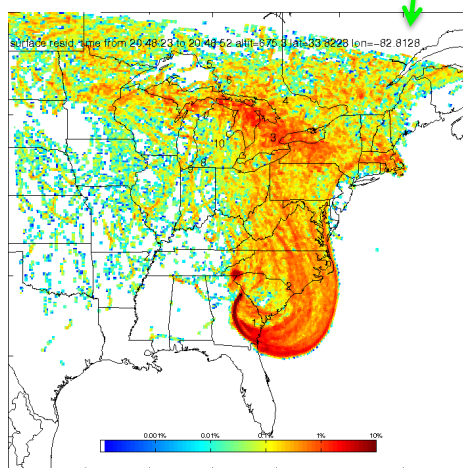
**Figure 12:** The track of a flight on 6 July 1999 over Atlanta during the SOS99 campaign color-coded with the  $\text{NO}_y$  mixing ratio. Time series of the 16 June 2013 and the 6 July 1999 flights for  $\text{NO}_y$  and  $\text{CO}$  show that the mixing ratios over Atlanta have decreased significantly over the past 14 years.



550



551



552

553

554

555

556

557

558

**Figure 13:** FLEXPART model results: time series of  $\text{NO}_y$  with 48 hours of accumulation time, the flight track color-coded by modeled  $\text{NO}_y$  and the surface residence time for a point on the last transect downwind of the Harlee Branch power plant.



## References:

- Bahreini, R., Dunlea, E. J., Matthew, B. M., Simons, C., Docherty, K. S., DeCarlo, P. F., Jimenez, J. L., Brock, C. A., and Middlebrook, A. M.: Design and operation of a pressure-controlled inlet for airborne sampling with an aerodynamic aerosol lens, *Aerosol Sci. Technol.*, 42, 465-471, 2008.
- Bahreini, R., Ervens, B., Middlebrook, A. M., Warneke, C., de Gouw, J. A., DeCarlo, P. F., Jimenez, J. L., Brock, C. A., Neuman, J. A., Ryerson, T. B., Stark, H., Atlas, E., Brioude, J., Fried, A., Holloway, J. S., Peischl, J., Richter, D., Walega, J., Weibring, P., Wollny, A. G., and Fehsenfeld, F. C.: Organic aerosol formation in urban and industrial plumes near Houston and Dallas, Texas, *J. Geophys. Res.-Atmos.*, 114, 2009.
- Brock, C. A., Cozic, J., Bahreini, R., Froyd, K. D., Middlebrook, A. M., McComiskey, A., Brioude, J., Cooper, O. R., Stohl, A., Aikin, K. C., de Gouw, J. A., Fahey, D. W., Ferrare, R. A., Gao, R. S., Gore, W., Holloway, J. S., Hubler, G., Jefferson, A., Lack, D. A., Lance, S., Moore, R. H., Murphy, D. M., Nenes, A., Novelli, P. C., Nowak, J. B., Ogren, J. A., Peischl, J., Pierce, R. B., Pilewskie, P., Quinn, P. K., Ryerson, T. B., Schmidt, K. S., Schwarz, J. P., Sodemann, H., Spackman, J. R., Stark, H., Thomson, D. S., Thornberry, T., Veres, P., Watts, L. A., Warneke, C., and Wollny, A. G.: Characteristics, sources, and transport of aerosols measured in spring 2008 during the aerosol, radiation, and cloud processes affecting Arctic Climate (ARCPAC) Project, *Atmos. Chem. Phys.*, 11, 2423-2453, 2011.
- Brock, C. A., Schroder, F., Bernd, K., Petzold, A., Busen, R., and Fiebig, M.: Ultrafine particle size distributions measured in aircraft exhaust plumes, *J. Geophys. Res.*, 105, 26,555-526,567, 2000.
- Cazorla, M., Wolfe, G. M., Bailey, S. A., Swanson, A. K., Arkinson, H. L., and Hanisco, T. F.: A new airborne laser-induced fluorescence instrument for in situ detection of formaldehyde throughout the troposphere and lower stratosphere, *Atmospheric Measurement Techniques*, 8, 541-552, 2015.
- Chang, W. L., Bhawe, P. V., Brown, S. S., Riemer, N., Stutz, J., and Dabdub, D.: Heterogeneous Atmospheric Chemistry, Ambient Measurements, and Model Calculations of N<sub>2</sub>O<sub>5</sub>: A Review, *Aerosol Science and Technology*, 45, 665-695, 2011.
- Cross, E. S., Slowik, J. G., Davidovits, P., Allan, J. D., Worsnop, D. R., Jayne, J. T., Lewis, D. K., Canagaratna, M., and Onasch, T. B.: Laboratory and ambient particle density determinations using light scattering in conjunction with aerosol mass spectrometry, *Aerosol Sci. Technol.*, 41, 343-359, 2007.
- Dallmann, T. R. and Harley, R. A.: Evaluation of mobile source emission trends in the United States, *J. Geophys. Res.-Atmos.*, 115, 2010.
- de Gouw, J. A., McKeen, S. A., Aikin, K. C., Brock, C. A., Brown, S. S., Gilman, J. B., Graus, M., Hanisco, T., Holloway, J. S., Kaiser, J., Keutsch, F. N., Lerner, B. M., Liao, J., Markovic, M. Z., Middlebrook, A. M., Min, K. E., Neuman, J. A., Nowak, J. B., Peischl, J., Pollack, I. B., Roberts, J.



- 598 M., Ryerson, T. B., Trainer, M., Veres, P. R., Warneke, C., Welti, A., and Wolfe, G. M.: Airborne  
599 measurements of the atmospheric emissions from a fuel ethanol refinery, *J. Geophys. Res.-*  
600 *Atmos.*, 120, 4385-4397, 2015a.
- 601 de Gouw, J. A., Parrish, D. D., Frost, G. J., and Trainer, M.: Reduced emissions of CO<sub>2</sub>,  
602 NO<sub>x</sub>, and SO<sub>2</sub> from US power plants owing to switch from coal to natural gas with  
603 combined cycle technology, *Earth's Future*, 2, 75-82, 2014.
- 604 de Gouw, J. A., Trainer, M., Brown, S. S., Edwards, P., Gilman, J. B., Graus, M., Hanisco,  
605 T., Kaiser, J., Keutsch, F. N., Kim, S. W., Lerner, B. M., Neuman, J. A., Parrish, D. D., Pollack, I.  
606 B., Roberts, J. M., Ryerson, T. R., Veres, P. R., Warneke, C., and Wolfe, G. M.: Enhanced  
607 Removal of Biogenic Hydrocarbons in Power Plant Plumes Constrains the Dependence of  
608 Atmospheric Hydroxyl Concentrations on Nitrogen Oxides, *Geophys. Res. Lett.*, in  
609 preparation, 2015b.
- 610 de Gouw, J. A. and Warneke, C.: Measurements of volatile organic compounds in the  
611 earths atmosphere using proton-transfer-reaction mass spectrometry, *Mass Spectrometry*  
612 *Reviews*, 26, 223-257, 2007.
- 613 DiGangi, J. P., Boyle, E. S., Karl, T., Harley, P., Turnipseed, A., Kim, S., Cantrell, C.,  
614 Maudlin, R. L., III, Zheng, W., Flocke, F., Hall, S. R., Ullmann, K., Nakashima, Y., Paul, J. B.,  
615 Wolfe, G. M., Desai, A. R., Kajii, Y., Guenther, A., and Keutsch, F. N.: First direct  
616 measurements of formaldehyde flux via eddy covariance: implications for missing in-  
617 canopy formaldehyde sources, *Atmos. Chem. Phys.*, 11, 10565-10578, 2011.
- 618 Dube, W. P., Brown, S. S., Osthoff, H. D., Nunley, M. R., Ciciora, S. J., Paris, M. W.,  
619 McLaughlin, R. J., and Ravishankara, A. R.: Aircraft instrument for simultaneous, in situ  
620 measurement of NO<sub>3</sub> and N<sub>2</sub>O<sub>5</sub> via pulsed cavity ring-down spectroscopy, *Rev. Sci. Instr.*,  
621 77, 2006.
- 622 Emmons, L. K., Arnold, S. R., Monks, S. A., Huijnen, V., Tilmes, S., Law, K. S., Thomas, J.  
623 L., Raut, J. C., Bouarar, I., Turquety, S., Long, Y., Duncan, B., Steenrod, S., Strode, S.,  
624 Flemming, J., Mao, J., Langner, J., Thompson, A. M., Tarasick, D., Apel, E. C., Blake, D. R.,  
625 Cohen, R. C., Dibb, J., Diskin, G. S., Fried, A., Hall, S. R., Huey, L. G., Weinheimer, A. J.,  
626 Wisthaler, A., Mikoviny, T., Nowak, J., Peischl, J., Roberts, J. M., Ryerson, T., Warneke, C., and  
627 Helmig, D.: The POLARCAT Model Intercomparison Project (POLMIP): overview and  
628 evaluation with observations, *Atmos. Chem. Phys.*, 15, 6721-6744, 2015.
- 629 Gilman, J. B., Kuster, W. C., Goldan, P. D., Herndon, S. C., Zahniser, M. S., Tucker, S. C.,  
630 Brewer, W. A., Lerner, B. M., Williams, E. J., Harley, R. A., Fehsenfeld, F. C., Warneke, C., and  
631 de Gouw, J. A.: Measurements of volatile organic compounds during the 2006  
632 TexAQS/GoMACCS campaign: Industrial influences, regional characteristics, and diurnal  
633 dependencies of the OH reactivity, *J. Geophys. Res.-Atmos.*, 114, 2009.
- 634 Holloway, J. S., Jakoubek, R. O., Parrish, D. D., Gerbig, C., Volz-Thomas, A., Schmitgen,  
635 S., Fried, A., Wert, B., Henry, B., and Drummond, J. R.: Airborne intercomparison of vacuum



- 636 ultraviolet fluorescence and tunable diode laser absorption measurements of tropospheric  
637 carbon monoxide, *J. Geophys. Res.*, 105, 24,251-224,261, 2000.
- 638 Hottle, J. R., Huisman, A. J., Digangi, J. P., Kammrath, A., Galloway, M. M., Coens, K. L.,  
639 and Keutsch, F. N.: A Laser Induced Fluorescence-Based Instrument for In-Situ  
640 Measurements of Atmospheric Formaldehyde, *Environ. Sci. Technol.*, 43, 790-795, 2009.
- 641 Kaiser, J., Wolfe, G. M., Min, K. E., Brown, S. S., Miller, C. C., Jacob, D. J., deGouw, J. A.,  
642 Graus, M., Hanisco, T. F., Holloway, J., Peischl, J., Pollack, I. B., Ryerson, T. B., Warneke, C.,  
643 Washenfelder, R. A., and Keutsch, F. N.: Reassessing the ratio of glyoxal to formaldehyde as  
644 an indicator of hydrocarbon precursor speciation, *Atmos. Chem. Phys.*, 15, 7571-7583,  
645 2015.
- 646 Kercher, J. P., Riedel, T. P., and Thornton, J. A.: Chlorine activation by N<sub>2</sub>O<sub>5</sub>:  
647 simultaneous, in situ detection of ClNO<sub>2</sub> and N<sub>2</sub>O<sub>5</sub> by chemical ionization mass  
648 spectrometry, *Atmospheric Measurement Techniques*, 2, 193-204, 2009.
- 649 Kiendler-Scharr, A., Zhang, Q., Hohaus, T., Kleist, E., Mensah, A., Mentel, T. F.,  
650 Spindler, C., Uerlings, R., Tillmann, R., and Wildt, J.: Aerosol mass spectrometric features of  
651 biogenic SOA: Observations from a plant chamber and in rural atmospheric environments,  
652 *Environ. Sci. Technol.*, 43, 8166-8172, 2009.
- 653 Lack, D. A., Richardson, M. S., Law, D., Langridge, J. M., Cappa, C. D., McLaughlin, R. J.,  
654 and Murphy, D. M.: Aircraft Instrument for Comprehensive Characterization of Aerosol  
655 Optical Properties, Part 2: Black and Brown Carbon Absorption and Absorption  
656 Enhancement Measured with Photo Acoustic Spectroscopy, *Aerosol Science and  
657 Technology*, 46, 555-568, 2012.
- 658 Lance, S., Brock, C. A., Rogers, D., and Gordon, J. A.: Water droplet calibration of the  
659 Cloud Droplet Probe (CDP) and in-flight performance in liquid, ice and mixed-phase clouds  
660 during ARCPAC, *Atmospheric Measurement Techniques*, 3, 1683-1706, 2010.
- 661 Lance, S., Medina, J., Smith, J. N., and Nenes, A.: Mapping the operation of the DMT  
662 Continuous Flow CCN counter, *Aerosol Science and Technology*, 40, 242-254, 2006.
- 663 Langridge, J. M., Richardson, M. S., Lack, D., Law, D., and Murphy, D. M.: Aircraft  
664 Instrument for Comprehensive Characterization of Aerosol Optical Properties, Part I:  
665 Wavelength-Dependent Optical Extinction and Its Relative Humidity Dependence  
666 Measured Using Cavity Ringdown Spectroscopy, *Aerosol Science and Technology*, 45,  
667 1305-1318, 2011.
- 668 Lee, B. H., Lopez-Hilfiker, F. D., Mohr, C., Kurten, T., Worsnop, D. R., and Thornton, J.  
669 A.: An Iodide-Adduct High-Resolution Time-of-Flight Chemical-Ionization Mass  
670 Spectrometer: Application to Atmospheric Inorganic and Organic Compounds, *Environ. Sci.  
671 Technol.*, 48, 6309-6317, 2014.





- 672 Lerner, B., Gilman, J. B., Dumas, M., Hughes, D. D., Jaksich, A., Hatch, C. D., Graus, M.,  
673 Tokarek, T. W., Peischl, J., Koss, A., Yuan, B., Warneke, C., Isaacman-Van Wertz, G., Sueper,  
674 D., and de Gouw, J. A.: An improved, automated whole-air sampler and VOC GC-MS analysis  
675 system, *Atmospheric measurement Techniques Discussions*, in preparation, 2015.
- 676 Middlebrook, A. M., Bahreini, R., Jimenez, J. L., and Canagaratna, M. R.: Evaluation of  
677 composition-dependent collection efficiencies for the Aerodyne aerosol mass  
678 spectrometer using field data, *Aerosol Sci. Technol.*, 46, 258-271, 2012.
- 679 Min, K. E., Washenfelder, R. A., Dube, W. P., Langford, A. O., Edwards, P. M., Zarzana,  
680 K. J., Stutz, J., Lu, K., Zhang, Y., and Brown, S. S.: A broadband cavity enhanced absorption  
681 spectrometer for aircraft measurements of glyoxal, methyl glyoxal, nitrous acid, nitrogen  
682 dioxide, and water vapor, *Atmospheric measurement Techniques Discussions*, submitted,  
683 2015.
- 684 Neuman, J. A., Huey, L. G., Dissly, R. W., Fehsenfeld, F. C., Flocke, F., Holecek, J. C.,  
685 Holloway, J. S., Hubler, G., Jakoubek, R., Nicks Jr., D. K., Parrish, D. D., Ryerson, T. B., Sueper,  
686 D. T., and Weinheimer, A. J.: Fast-response airborne in situ measurements of HNO<sub>3</sub> during  
687 the Texas 2000 Air Quality Study, *J. Geophys. Res.*, 107, 4436,  
688 doi:4410.1029/2001JD001437, 2002.
- 689 Neuman, J. A., Ryerson, T. B., Huey, L. G., Jakoubek, R., Nowak, J. B., Simons, C., and  
690 Fehsenfeld, F. C.: Calibration and evaluation of nitric acid and ammonia permeation tubes  
691 by UV optical absorption, *Environ. Sci. Technol.*, 37, 2975-2981, 2003.
- 692 Nowak, J. B., Neuman, J. A., Kozai, K., Huey, L. G., Tanner, D. J., Holloway, J. S.,  
693 Ryerson, T. B., Frost, G. J., McKeen, S. A., and Fehsenfeld, F. C.: A chemical ionization mass  
694 spectrometry technique for airborne measurements of ammonia, *J. Geophys. Res.-Atmos.*,  
695 112, 2007.
- 696 Osthoff, H. D., Roberts, J. M., Ravishankara, A. R., Williams, E. J., Lerner, B. M.,  
697 Sommariva, R., Bates, T. S., Coffman, D., Quinn, P. K., Dibb, J. E., Stark, H., Burkholder, J. B.,  
698 Talukdar, R. K., Meagher, J., Fehsenfeld, F. C., and Brown, S. S.: High levels of nitryl chloride  
699 in the polluted subtropical marine boundary layer, *Nature Geoscience*, 1, 324-328, 2008.
- 700 Park, K., Kittelson, D. B., Zachariah, M. R., and McMurry, P. H.: Measurement of  
701 inherent material density of nanoparticle agglomerates, *J. Nanoparticle Res.*, 6, 267-272,  
702 2004.
- 703 Peischl, J., Ryerson, T. B., Aikin, K. C., de Gouw, J. A., Gilman, J. B., Holloway, J. S.,  
704 Lerner, B. M., Nadkarni, R., Neuman, J. A., Nowak, J. B., Trainer, M., Warneke, C., and Parrish,  
705 D. D.: Quantifying atmospheric methane emissions from the Haynesville, Fayetteville, and  
706 northeastern Marcellus shale gas production regions, *J. Geophys. Res.-Atmos.*, 120, 2119-  
707 2139, 2015.
- 708 Peischl, J., Ryerson, T. B., Holloway, J. S., Trainer, M., Andrews, A. E., Atlas, E. L.,  
709 Blake, D. R., Daube, B. C., Dlugokencky, E. J., Fischer, M. L., Goldstein, A. H., Guha, A., Karl, T.,





- 710 Kofler, J., Kosciuch, E., Misztal, P. K., Perring, A. E., Pollack, I. B., Santoni, G. W., Schwarz, J. P.,  
711 Spackman, J. R., Wofsy, S. C., and Parrish, D. D.: Airborne observations of methane  
712 emissions from rice cultivation in the Sacramento Valley of California, *J. Geophys. Res.-*  
713 *Atmos.*, 117, 2012.
- 714 Perry, R. H. and Green, D. W. (Eds.): *Perry's Chemical Engineers' Handbook*,  
715 McGraw-Hill, New York, NY, 1997.
- 716 Pollack, I. B., Lerner, B. M., and Ryerson, T. B.: Evaluation of ultraviolet light-  
717 emitting diodes for detection of atmospheric NO<sub>2</sub> by photolysis - chemiluminescence,  
718 *Journal of Atmospheric Chemistry*, 65, 111-125, 2010.
- 719 Roberts, G. C. and Nenes, A.: A continuous-flow streamwise thermal-gradient CCN  
720 chamber for atmospheric measurements, *Aerosol Science and Technology*, 39, 206-221,  
721 2005.
- 722 Ryerson, T. B., Buhr, M. P., Frost, G. J., Goldan, P. D., Holloway, J. S., Hübler, G., Jobson,  
723 B. T., Kuster, W. C., McKeen, S. A., Parrish, D. D., Roberts, J. M., Sueper, D. T., Trainer, M.,  
724 Williams, J., and Fehsenfeld, F. C.: Emissions lifetimes and ozone formation in power plant  
725 plumes, *J. Geophys. Res.*, 103, 22,569-522,583, 1998.
- 726 Ryerson, T. B., Huey, L. G., Knapp, K., Neuman, J. A., Parrish, D. D., Sueper, D. T., and  
727 Fehsenfeld, F. C.: Design and initial characterization of an inlet for gas-phase NO<sub>y</sub>  
728 measurements from aircraft, *J. Geophys. Res.*, 104, 5483-5492, 1999.
- 729 Schwarz, J. P., Gao, R. S., Fahey, D. W., Thomson, D. S., Watts, L. A., Wilson, J. C.,  
730 Reeves, J. M., Darbeheshti, M., Baumgardner, D. G., Kok, G. L., Chung, S. H., Schulz, M.,  
731 Hendricks, J., Lauer, A., Kaercher, B., Slowik, J. G., Rosenlof, K. H., Thompson, T. L., Langford,  
732 A. O., Loewenstein, M., and Aikin, K. C.: Single-particle measurements of midlatitude black  
733 carbon and light-scattering aerosols from the boundary layer to the lower stratosphere, *J.*  
734 *Geophys. Res.-Atmos.*, 111, D16207, 2006.
- 735 Schwarz, J. P., Spackman, J. R., Fahey, D. W., Gao, R. S., Lohmann, U., Stier, P., Watts, L.  
736 A., Thomson, D. S., Lack, D. A., Pfister, L., Mahoney, M. J., Baumgardner, D., Wilson, J. C., and  
737 Reeves, J. M.: Coatings and their enhancement of black carbon light absorption in the  
738 tropical atmosphere, *J. Geophys. Res.-Atmos.*, 113, 2008.
- 739 Schwarz, J. P., Spackman, J. R., Gao, R. S., Watts, L. A., Stier, P., Schulz, M., Davis, S. M.,  
740 Wofsy, S. C., and Fahey, D. W.: Global-scale black carbon profiles observed in the remote  
741 atmosphere and compared to models (vol 37, art L18812 , 2010), *Geophys. Res. Lett.*, 37,  
742 2010.
- 743 Slusher, D. L., Huey, L. G., Tanner, D. J., Flocke, F. M., and Roberts, J. M.: A thermal  
744 dissociation-chemical ionization mass spectrometry (TD-CIMS) technique for the  
745 simultaneous measurement of peroxyacyl nitrates and dinitrogen pentoxide, *J. Geophys.*  
746 *Res.-Atmos.*, 109, 2004.



- 747 Stohl, A., Forster, C., Frank, A., Seibert, P., and Wotawa, G.: Technical Note : The  
748 Lagrangian particle dispersion model FLEXPART version 6.2, Atmos. Chem. Phys., 5, 2461-  
749 2474, 2005.
- 750 Veres, P., Gilman, J. B., Roberts, J. M., Kuster, W. C., Warneke, C., Burling, I. R., and de  
751 Gouw, J.: Development and validation of a portable gas phase standard generation and  
752 calibration system for volatile organic compounds, Atmospheric Measurement Techniques,  
753 3, 683-691, 2010.
- 754 von Schneidemesser, E., Monks, P. S., and Plass-Duelmer, C.: Global comparison of  
755 VOC and CO observations in urban areas, Atmos. Environ., 44, 5053-5064, 2010.
- 756 Wagner, N. L., Dube, W. P., Washenfelder, R. A., Young, C. J., Pollack, I. B., Ryerson, T.  
757 B., and Brown, S. S.: Diode laser-based cavity ring-down instrument for NO<sub>3</sub>, N<sub>2</sub>O<sub>5</sub>, NO,  
758 NO<sub>2</sub> and O<sub>3</sub> from aircraft, Atmospheric Measurement Techniques, 4, 1227-1240, 2011.
- 759 Warneke, C., de Gouw, J. A., Holloway, J. S., Peischl, J., Ryerson, T. B., Atlas, E., Blake,  
760 D., Trainer, M., and Parrish, D. D.: Multiyear trends in volatile organic compounds in Los  
761 Angeles, California: Five decades of decreasing emissions, J. Geophys. Res.-Atmos., 117,  
762 2012.
- 763 Warneke, C., Roberts, J. M., Veres, P., Gilman, J., Kuster, W. C., Burling, I., Yokelson, R.,  
764 and de Gouw, J. A.: VOC identification and inter-comparison from laboratory biomass  
765 burning using PTR-MS and PIT-MS, International Journal of Mass Spectrometry, 303, 6-14,  
766 2011.
- 767 Washenfelder, R. A., Young, C. J., Brown, S. S., Angevine, W. M., Atlas, E. L., Blake, D.  
768 R., Bon, D. M., Cubison, M. J., de Gouw, J. A., Dusanter, S., Flynn, J., Gilman, J. B., Graus, M.,  
769 Griffith, S., Grossberg, N., Hayes, P. L., Jimenez, J. L., Kuster, W. C., Lefer, B. L., Pollack, I. B.,  
770 Ryerson, T. B., Stark, H., Stevens, P. S., and Trainer, M. K.: The glyoxal budget and its  
771 contribution to organic aerosol for Los Angeles, California, during CalNex 2010, J. Geophys.  
772 Res.-Atmos., 116, 2011.
- 773 Wilson, J. C., Lafleur, B. G., Hilbert, H., Seebaugh, W. R., Fox, J., Gesler, D. W., Brock, C.  
774 A., Huebert, B. J., and Mullen, J.: Function and performance of a low turbulence inlet for  
775 sampling supermicron particles from aircraft platforms, Aerosol Science and Technology,  
776 38, 790-802, 2004.
- 777 Wolfe, G. M., Kaiser, J., Hanisco, T. F., Keutsch, F. N., de Gouw, J. A., Gilman, J. B.,  
778 Graus, M., Hatch, C. D., Holloway, J., Horowitz, L., Lee, B. H., Lerner, B., Lopez-Hilfiker, F. D.,  
779 Mao, J., Marvin, M., Peischl, J., Pollack, I. B., Roberts, J. M., Ryerson, T. B., Thornton, J. A.,  
780 Veres, P., and Warneke, C.: Formaldehyde production from isoprene oxidation across NO<sub>x</sub>  
781 regimes, in preparation, 2015. 2015.
- 782 Xu, L., Guo, H., Boyd, C. M., Klein, M., Bougiatioti, A., Cerully, K. M., Hite, J. R.,  
783 Isaacman-VanWertz, G., Kreisberg, N. M., Knote, C., Olson, K., Koss, A., Goldstein, A. H.,  
784 Hering, S. V., de Gouw, J., Baumann, K., Lee, S.-H., Nenes, A., Weber, R. J., and Ng, N. L.:



785 Effects of anthropogenic emissions on aerosol formation from isoprene and monoterpenes  
786 in the southeastern United States, Proceedings of the National Academy of Sciences of the  
787 United States of America, 112, 37-42, 2015.

788 Yu, H.: Development of landcover and emission factors for isoprene and  
789 monoterpene emission modeling and evaluation in the southern United States using  
790 airborne direct and indirect flux measurements, 2015. 2015.

791 Yuan, B., Kaser, L., Karl, T., Graus, M., Peischl, J., Campos, T. L., Shertz, S., Apel, E. C.,  
792 Hornbrook, R. S., Hills, A., Gilman, J. B., Lerner, B. M., Warneke, C., Flocke, F. M., Ryerson, T.  
793 B., Guenther, A. B., and de Gouw, J. A.: Airborne flux measurements of methane and volatile  
794 organic compounds over the Haynesville and Marcellus shale gas production regions,  
795 Journal of Geophysical Research: Atmospheres, 120, 6271-6289, 2015.

796 Zelenyuk, A., Imre, D., Han, J. H., and Oatis, S.: Simultaneous measurements of  
797 individual ambient particle size, composition, effective density, and hygroscopicity, Anal.  
798 Chem., 80, 1401-1407, 2008.

799 Zheng, W., Flocke, F. M., Tyndall, G. S., Swanson, A., Orlando, J. J., Roberts, J. M., Huey,  
800 L. G., and Tanner, D. J.: Characterization of a thermal decomposition chemical ionization  
801 mass spectrometer for the measurement of peroxy acyl nitrates (PANs) in the atmosphere,  
802 Atmos. Chem. Phys., 11, 6529-6547, 2011.

803

804