

Figure S1. Recorded mass spectrum of AlkC3 (black bars) at the lowest extent of fragmentation of the molecular ion at $m/z = 106$ ($E/N = 78$ Td) in the H_3O^+ ionization mode. The green thin line represents the expected molecular ion of the analyte and the intense signals depicted by the red thin bars represent the matrix analytes at $m/z = 30$ (NO^+) and 46 (NO_2^+) and their isotopic abundance signals at $m/z = 31$ and $47, 48$ respectively. The signals corresponding to the water clusters ($H_3O^+(H_2O)_n$) at $m/z = 37$ and 55 are systematically erased for simplification.

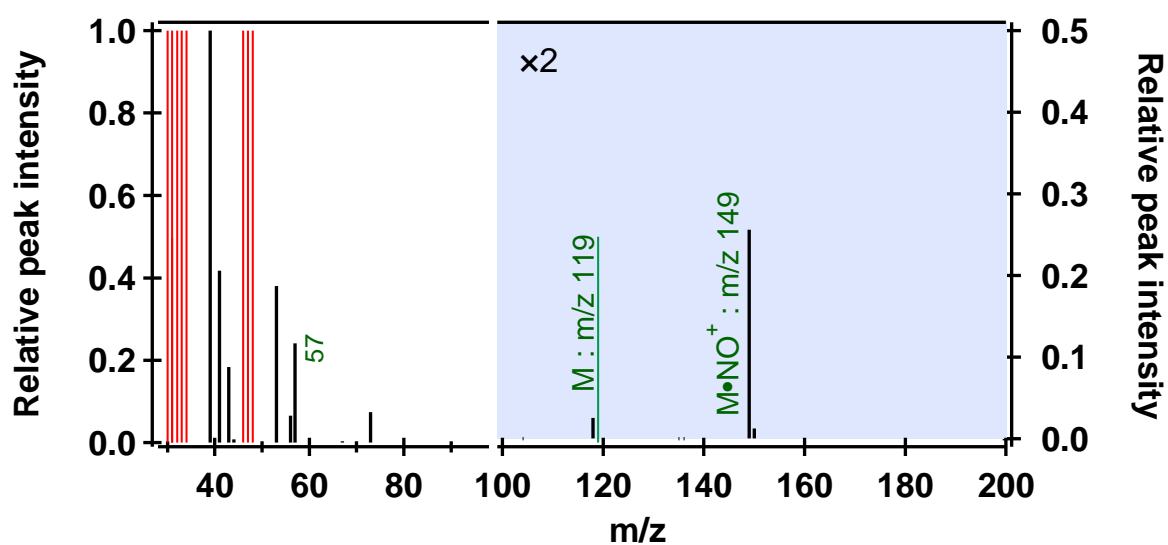


Figure S2. Recorded mass spectra of AlkiC4 (black bars) at the lowest extent of fragmentation ($E/N^* = 34$ Td) in the NO^+ ionization mode.

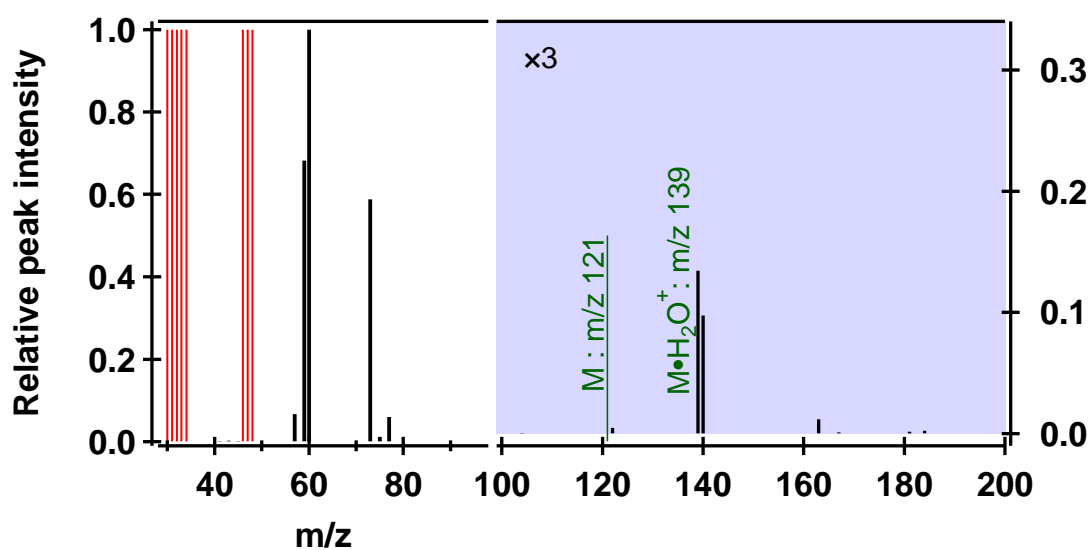


Figure S3. Recorded mass spectrum of 1OH3C3 (black bars) at the lowest extent of fragmentation ($E/N = 62$ Td) in the H_3O^+ ionization mode in absence of the RF device.

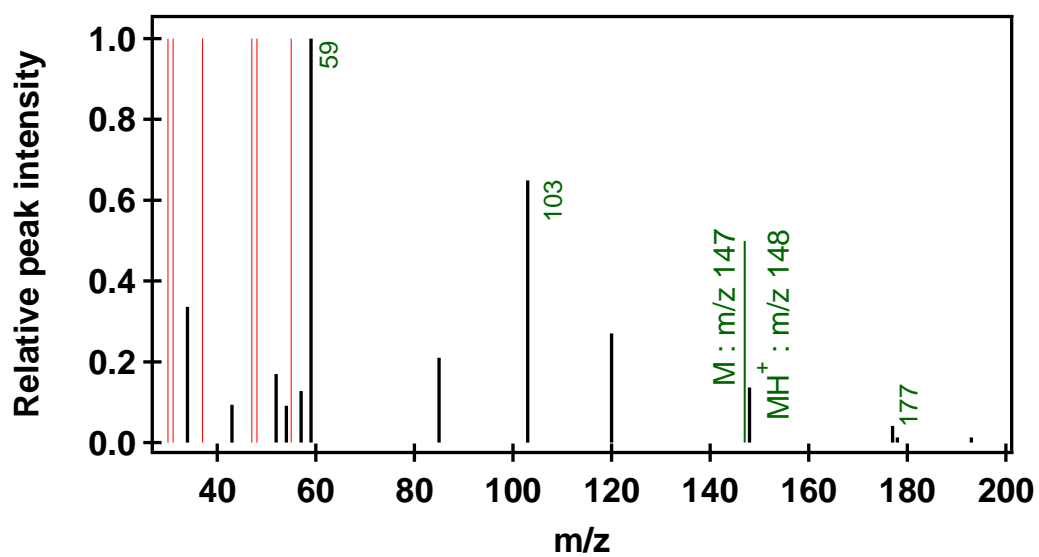


Figure S4. Recorded mass spectra of protonated KnC5 (black bars) for $E/N = 75$ Td, corresponding to the highest sensibility for the protonated analyte signal detection (m/z 148).

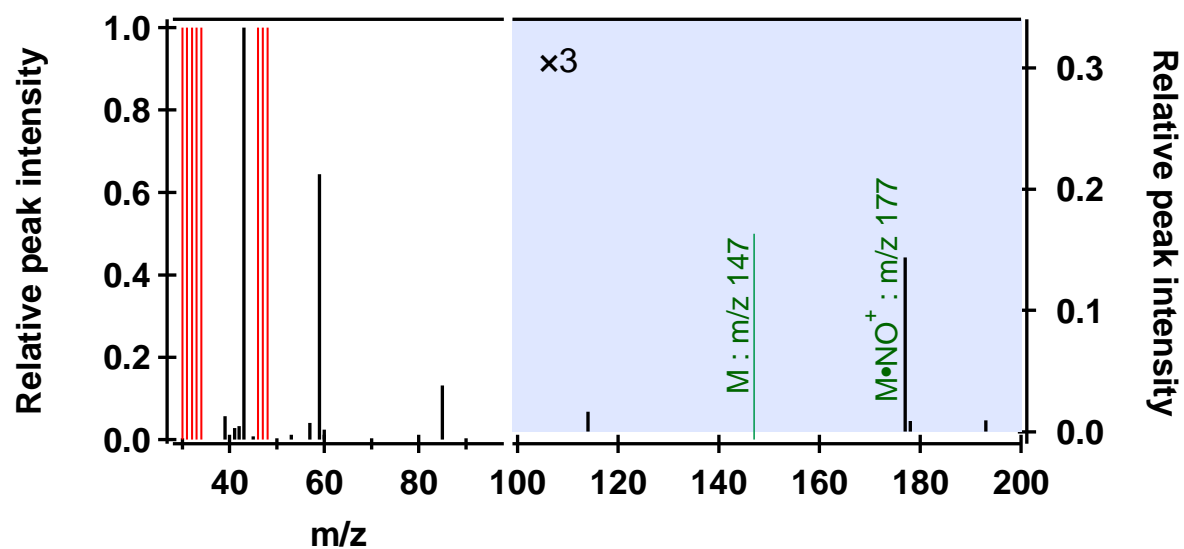


Figure S5. Recorded mass spectra of KnC5 adduct (black bars) at the lowest extent of fragmentation ($E/N^* = 36$ Td) in the NO^+ ionization mode.

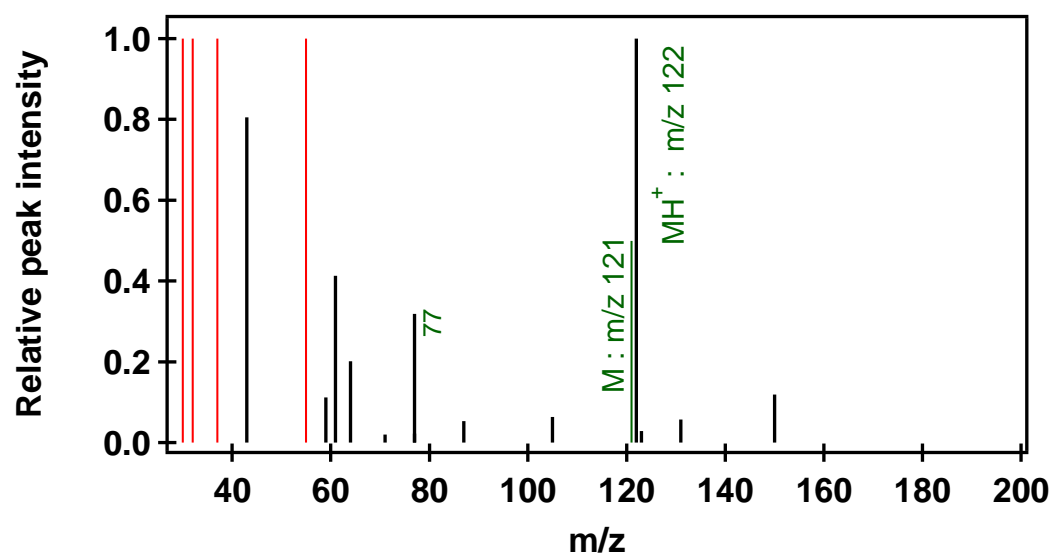


Figure S6. Recorded mass spectrum of protonated PAN (black bars) for E/N = 85 Td, corresponding to the highest sensibility of the m/z 122 signal detection.