

# Measurement of ambient NO<sub>3</sub> reactivity: Design, characterization and first deployment of a new instrument

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**Abstract.** We describe the first instrument for measurement of the rate constant ( $s^{-1}$ ) for reactive loss (i.e. the total reactivity) of NO<sub>3</sub> in ambient air. Cavity-ring-down spectroscopy is used to monitor the mixing ratio of synthetically generated NO<sub>3</sub> ( $\approx$  30-50 pptv) after passing through a flow-tube reactor with variable residence time (generally 10.5 s). The change in concentration of NO<sub>3</sub> upon modulation of the bath gas between zero-air and ambient air is used to derive its loss rate constant, which is then corrected for formation and decomposition of N<sub>2</sub>O<sub>5</sub> via numerical simulation. The instrument is calibrated and characterized using known amounts of NO and NO<sub>2</sub> and tested in the laboratory with an isoprene standard. The lowest reactivity that can be detected (defined by the stability of the NO<sub>3</sub> source, instrumental parameters and NO<sub>2</sub> mixing ratios) is 0.005 s<sup>-1</sup>. An automated dilution procedure enables measurement of NO<sub>3</sub> reactivities up to 45 s<sup>-1</sup>, this upper limit being defined mainly by the dilution accuracy. The typical total uncertainty associated with the reactivity measurement at the centre of its dynamic range is 16 %, though this is dependent on ambient NO<sub>2</sub> levels. Results from the first successful deployment of the instrument at a forested mountain site with urban influence are shown and future developments outlined.