

Supplementary information for Evaluation of NO⁺ reagent ion chemistry for on-line measurements of atmospheric volatile organic compounds

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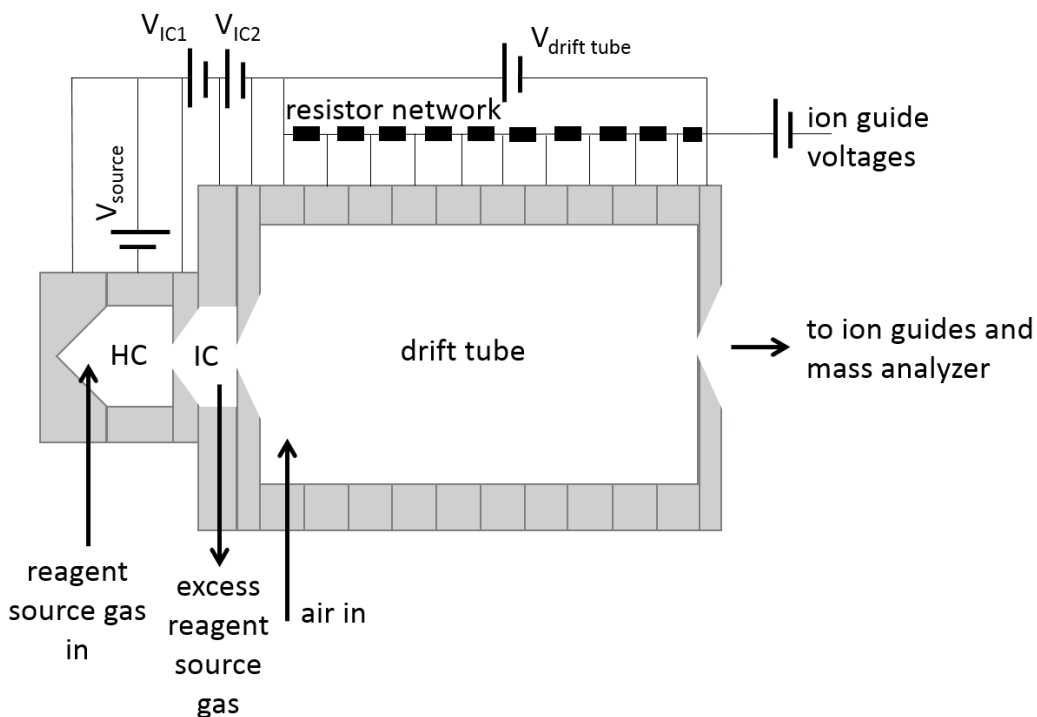


Figure S1. Ion source and drift tube schematic.

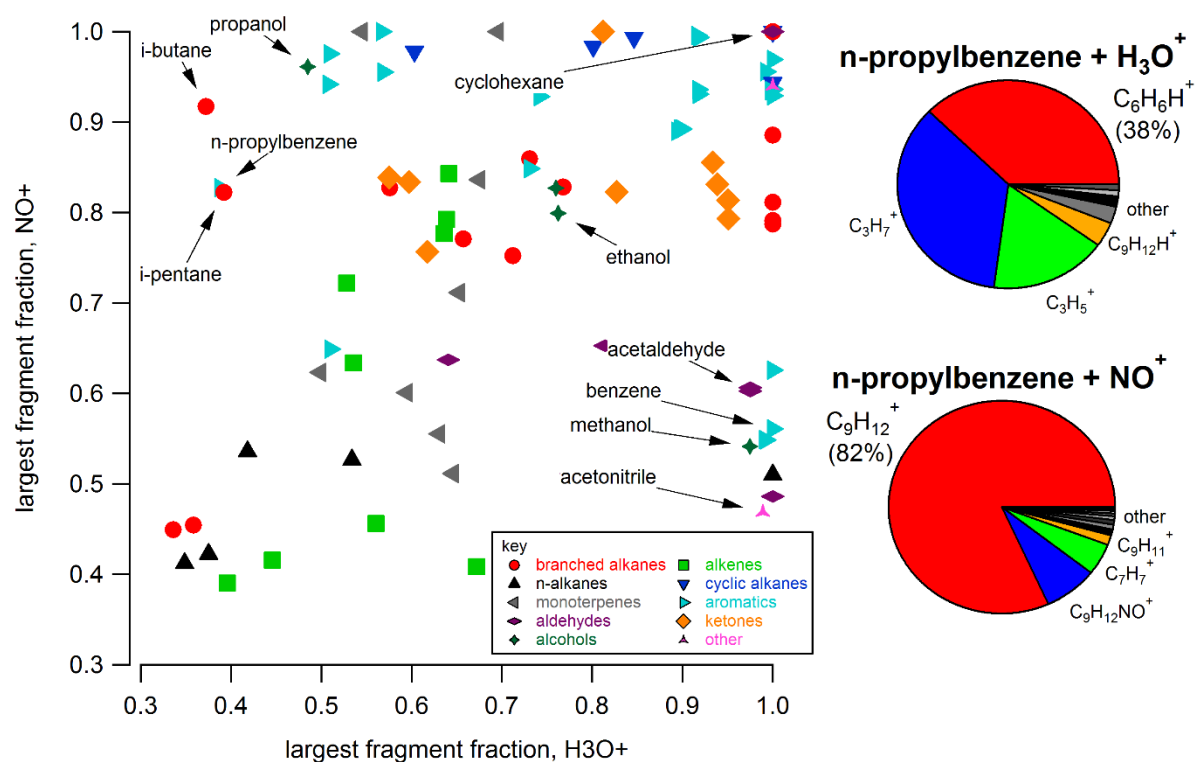


Figure S2. Comparison of product ion distributions between NO^+ CIMS and H_3O^+ CIMS. The complete product ion distribution of n-propylbenzene is shown as an example of a compound with a complex mass spectrum resulting from H_3O^+ chemistry, and a simple mass spectrum resulting from NO^+ chemistry.

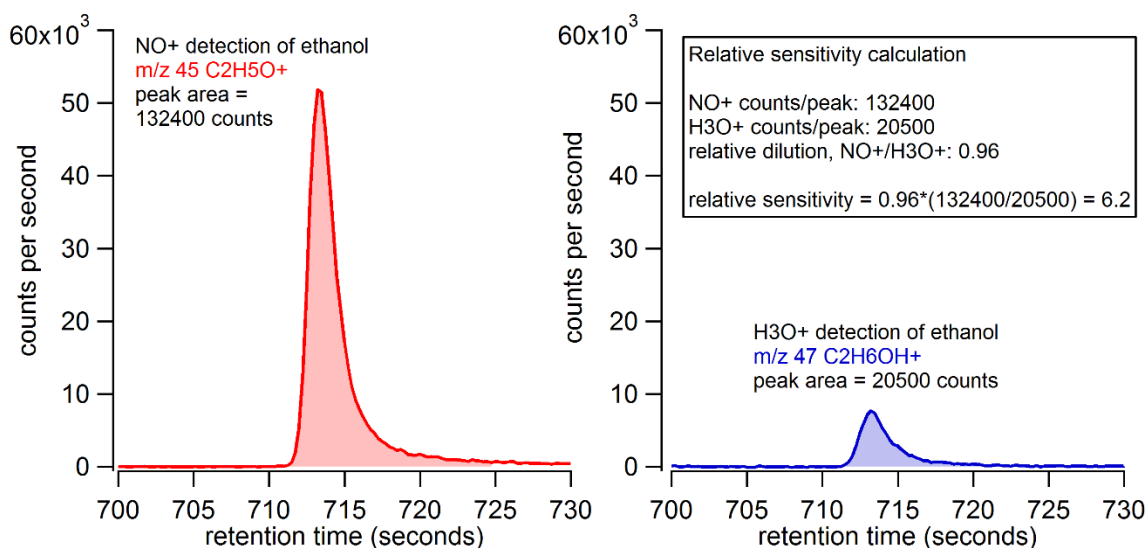


Figure S3. Example chromatograms and relative sensitivity calculation of ethanol for Table 2.

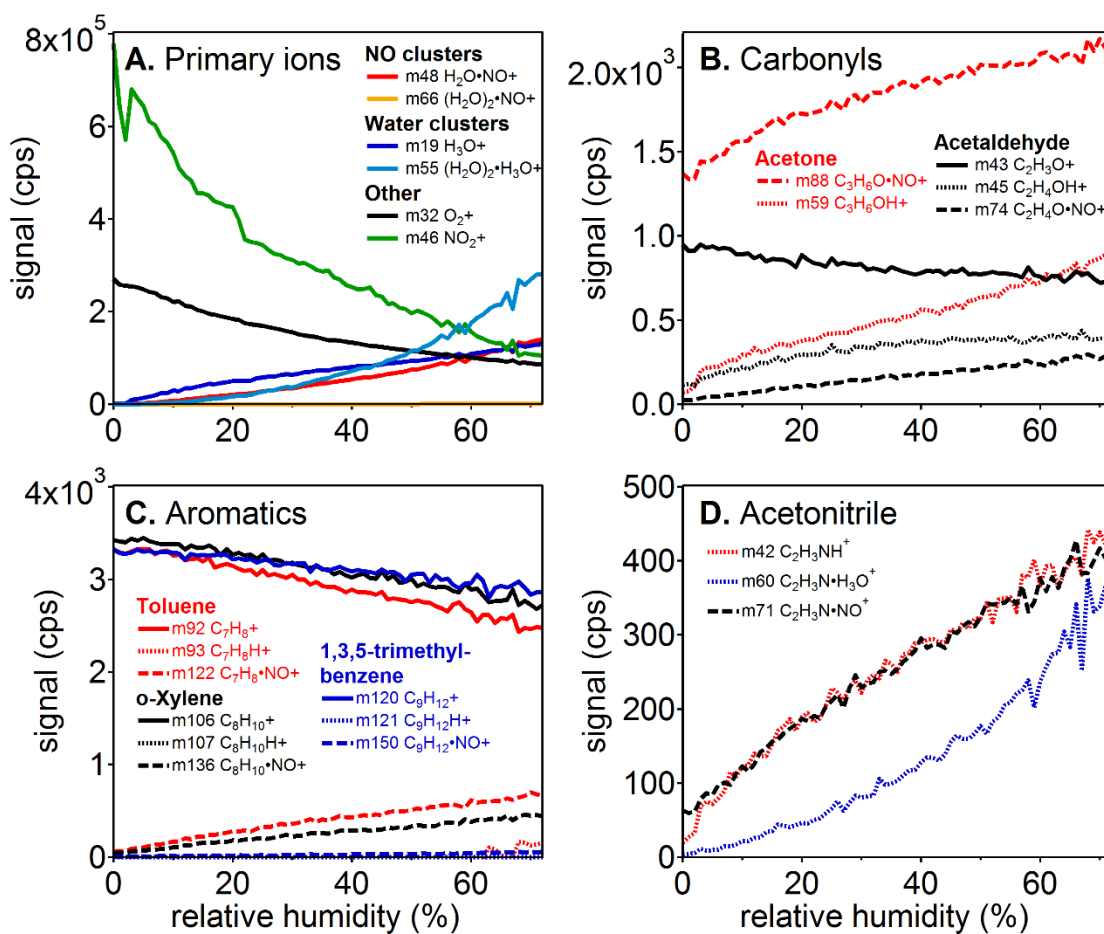


Figure S4. Humidity dependence of primary ions and VOCs. (a) Impurity ions and water clusters. (b) Carbonyls. (c) Aromatics. (d) Acetonitrile. Acetonitrile is detected with poor sensitivity using NO⁺; the NO⁺ and H₃O⁺ products are approximately equal in magnitude.

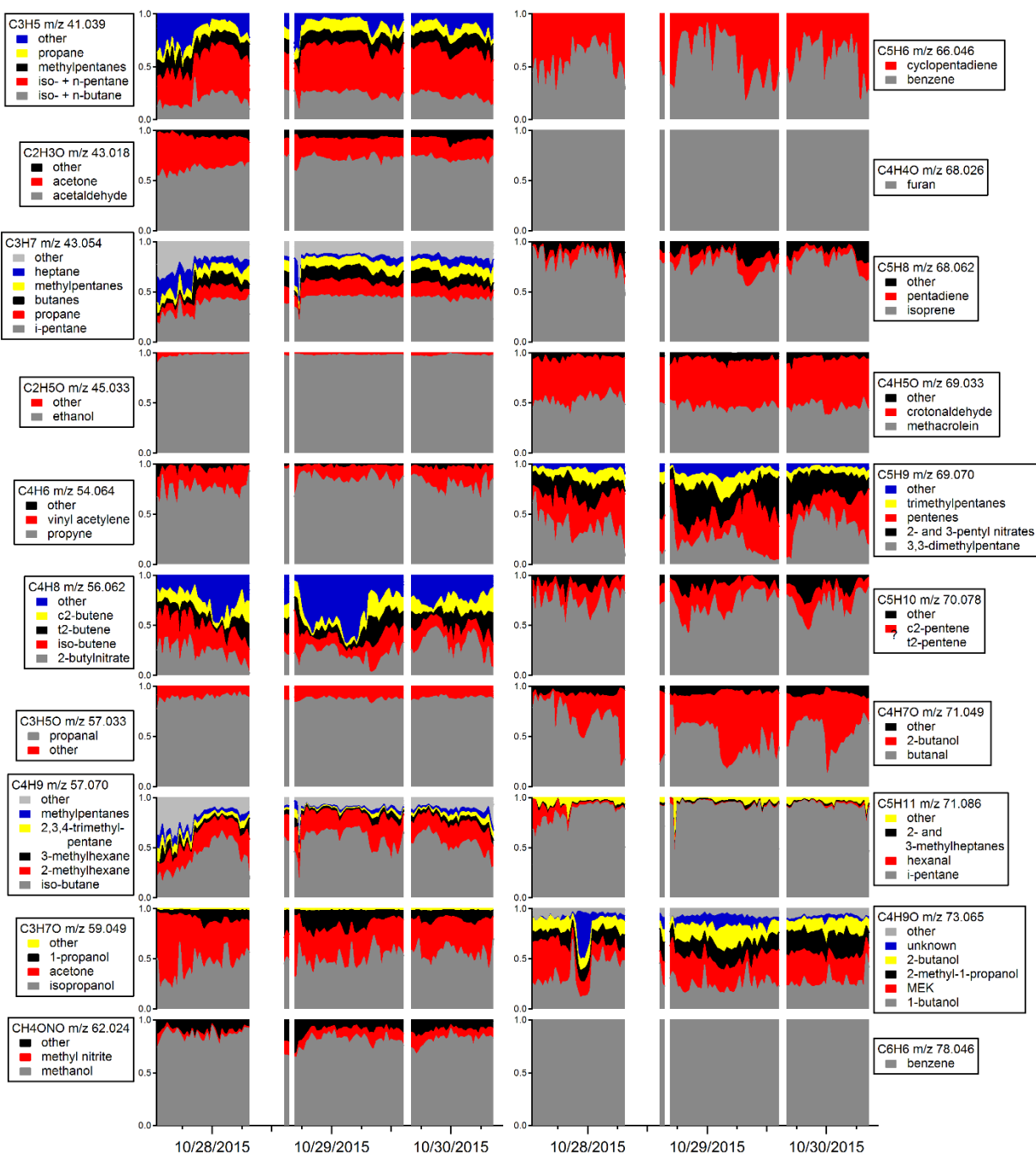


Figure S5. Speciated contributions to various NO^+ CIMS masses, in urban air. Values on Y-axes are the fractional contribution of each VOC to total signal. Includes m/z 41- m/z 78.

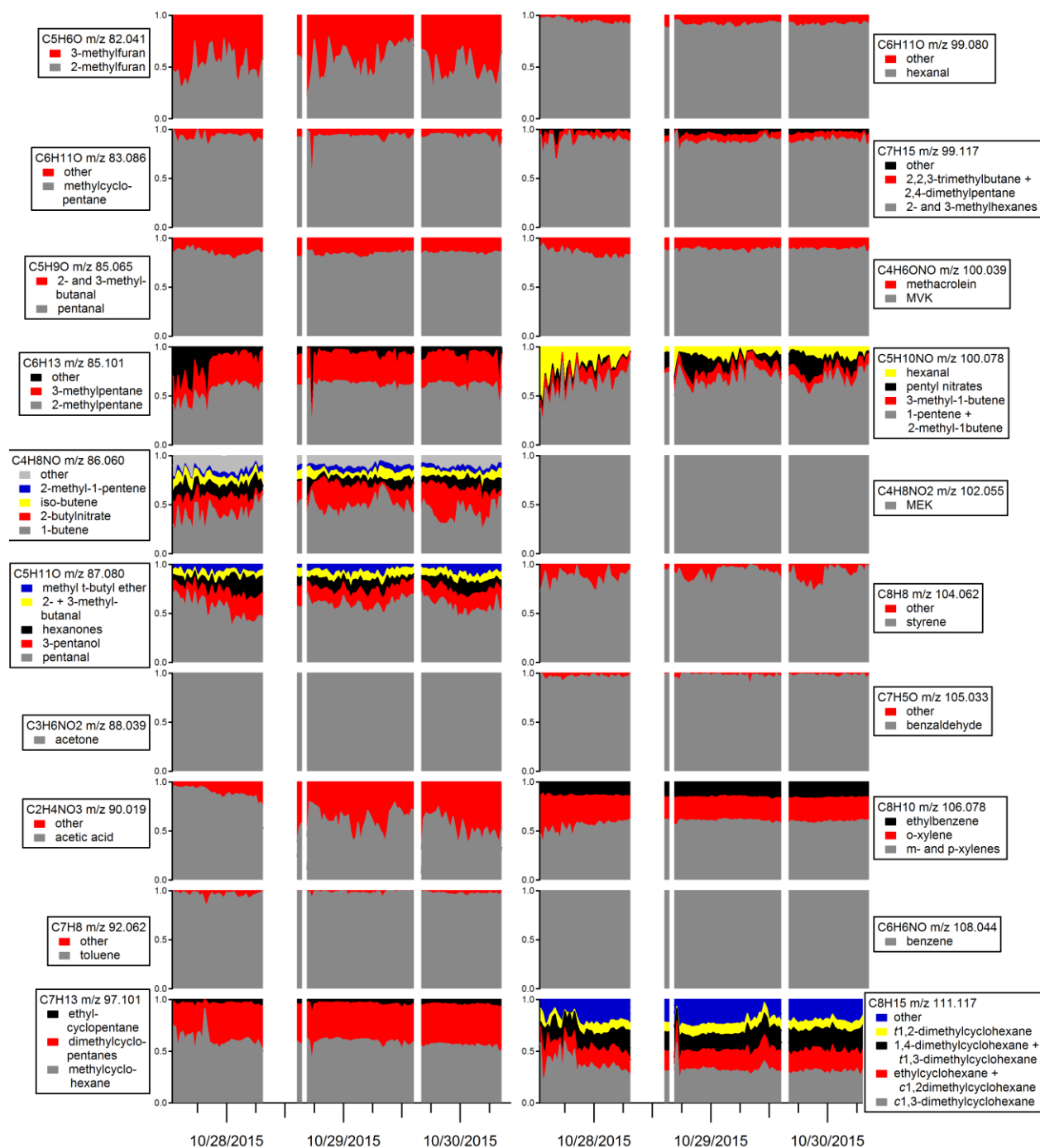


Figure S6. Figure S5, continued. Includes m/z 82-m/z 111.

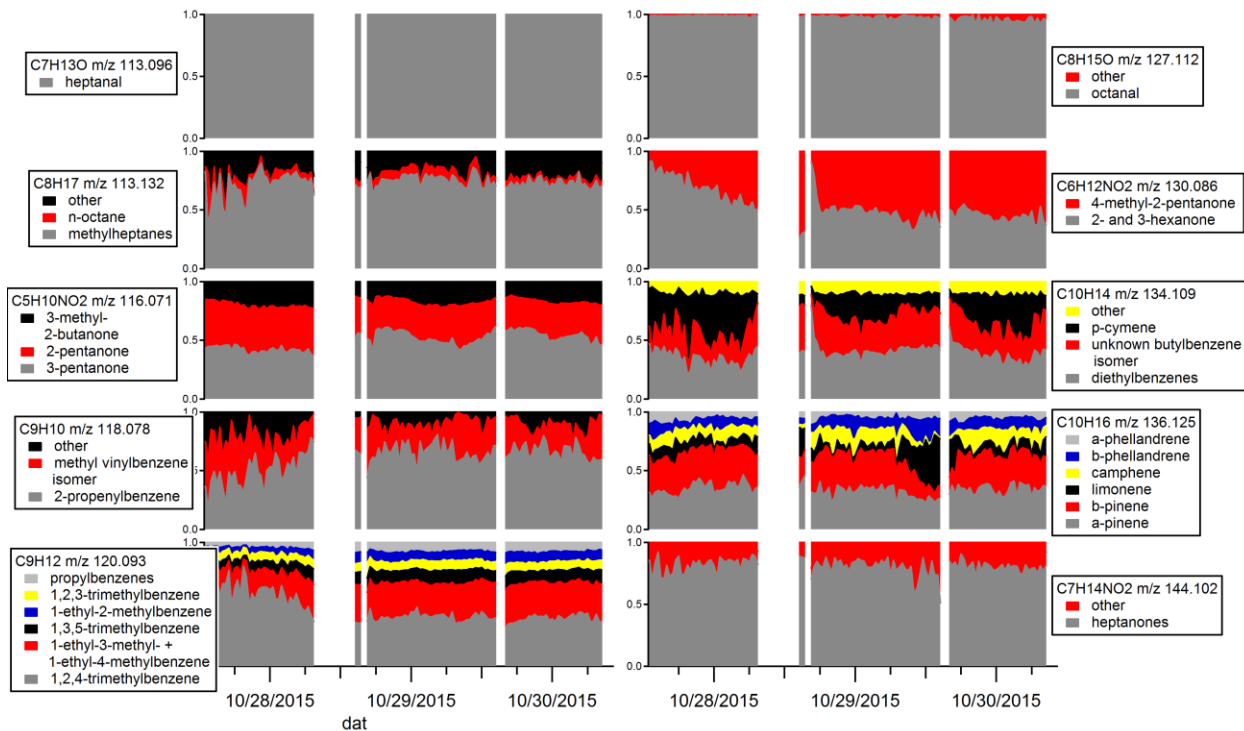


Figure S7. Figure S5, continued. Includes m/z 113-m/z 144.

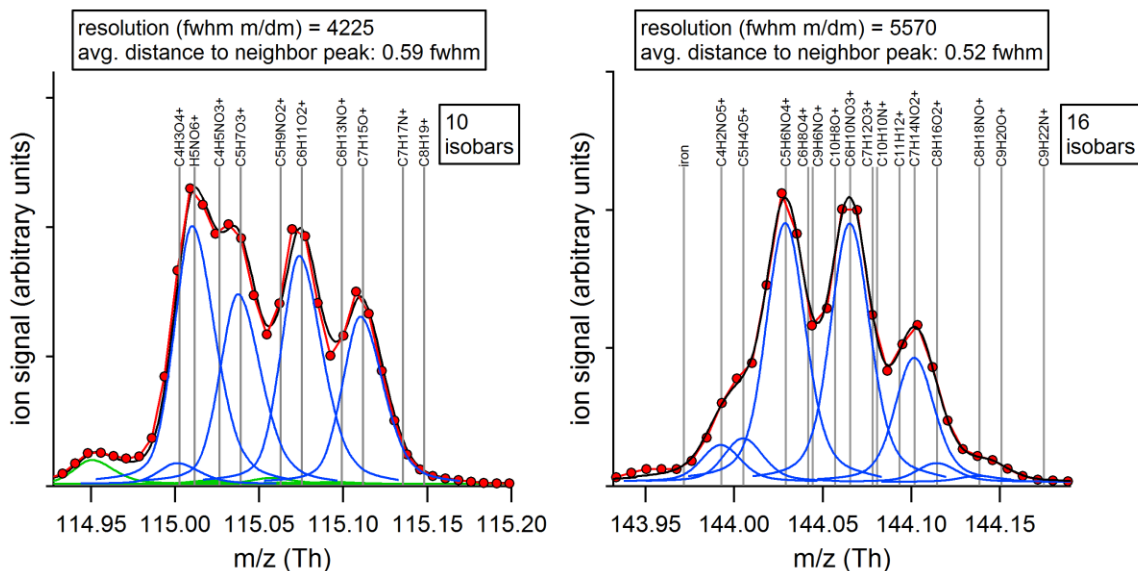


Figure S8. Example isobaric interferences for heptanone measured with H_3O^+ CIMS, at m/z 115 $\text{C}_7\text{H}_{14}\text{OH}^+$, and with NO^+ CIMS, at m/z 144 $\text{C}_7\text{H}_{14}\text{ONO}^+$. Although the resolution m/dm is better at m/z 144, there are more possible isobaric interferences and the average distance to neighboring peaks is smaller. The m/z range of each window is 10 FWHM. H_3O^+ ToF-CIMS mass spectrum courtesy of M. Coggon, collected in Boulder, CO in Dec. 2015.

Table S1. VOCs sampled in series GC-ToFCIMS laboratory experiments.

VOC name	Formula		
alkanes		β -pinene	C ₁₀ H ₁₆
ethane	C ₂ H ₆	limonene	C ₁₀ H ₁₆
propane	C ₃ H ₈	camphene	C ₁₀ H ₁₆
n-butane	C ₄ H ₁₀	γ -terpinene	C ₁₀ H ₁₆
n-pentane	C ₅ H ₁₂	α -phellandrene	C ₁₀ H ₁₆
n-hexane	C ₆ H ₁₄	1,8-cineol	C ₁₀ H ₁₆
n-octane	C ₈ H ₁₈	3-carene + myrcene	C ₁₀ H ₁₆
n-decane	C ₁₀ H ₂₂	aromatics	
n-undecane	C ₁₂ H ₂₆	benzene	C ₆ H ₆
i-butane (2-methylpropane)	C ₄ H ₁₀	toluene	C ₇ H ₈
i-pentane (2-methylbutane)	C ₅ H ₁₂	ethylbenzene	C ₈ H ₁₀
2,2-dimethylbutane	C ₆ H ₁₄	m-xylene + p-xylene	C ₈ H ₁₀
2-methylpentane	C ₆ H ₁₄	o-xylene	C ₈ H ₁₀
2,3-dimethylbutane	C ₆ H ₁₄	vinylbenzene (styrene)	C ₈ H ₈
3-methylpentane	C ₆ H ₁₄	isopropylbenzene	C ₉ H ₁₂
2,4-dimethylpentane	C ₇ H ₁₆	n-propylbenzene	C ₉ H ₁₂
2-methylhexane	C ₇ H ₁₆	1-ethyl,3-methylbenzene + 1-ethyl,4-methylbenzene	C ₉ H ₁₂
2,3-dimethylpentane	C ₇ H ₁₆	1,3,5-trimethylbenzene	C ₉ H ₁₂
3,3-dimethylpentane	C ₇ H ₁₆	1-ethyl,2-methylbenzene	C ₉ H ₁₂
3-methylhexane	C ₇ H ₁₆	1,2,4-trimethylbenzene	C ₉ H ₁₂
2,2,4-trimethylpentane	C ₈ H ₁₈	1,2,3-trimethylbenzene	C ₉ H ₁₂
2,3,4-trimethylpentane	C ₈ H ₁₈	1,3-diethylbenzene	C ₁₀ H ₁₄
2-methylheptane	C ₈ H ₁₈	1,4-diethylbenzene	C ₁₀ H ₁₄
3-methylheptane	C ₈ H ₁₈	aldehydes	
4-methylheptane	C ₈ H ₁₈	acetaldehyde	C ₂ H ₄ O
alkenes		propanal	C ₃ H ₆ O
ethene	C ₂ H ₄	butanal	C ₄ H ₈ O
propene	C ₃ H ₆	pentanal	C ₅ H ₁₀ O
ethyne	C ₂ H ₂	hexanal	C ₆ H ₁₂ O
trans-2-butene	C ₄ H ₈	heptanal	C ₇ H ₁₄ O
1-butene	C ₄ H ₈	octanal	C ₈ H ₁₆ O
iso-butene (2-methylpropene)	C ₄ H ₈	methacrolein	C ₄ H ₆ O
cis-2-butene	C ₄ H ₈	ketones	
1-pentene	C ₅ H ₁₀	acetone	C ₃ H ₆ O
trans-2-pentene	C ₅ H ₁₀	2-butanone (MEK)	C ₄ H ₈ O
cis-2-pentene	C ₅ H ₁₀	3-methyl-2-butanone	C ₅ H ₁₀ O
1-hexene	C ₆ H ₁₂	2-pentanone	C ₅ H ₁₀ O
isoprene	C ₅ H ₈	3-pentanone	C ₅ H ₁₀ O
cycloalkanes		3-hexanone	C ₆ H ₁₂ O
cyclopentane	C ₅ H ₁₀	methyl vinyl ketone (MVK)	C ₄ H ₆ O
methylcyclopentane	C ₆ H ₁₂	other	
cyclohexane	C ₆ H ₁₂	methanol	CH ₄ O
methylcyclohexane	C ₇ H ₁₄	ethanol	C ₂ H ₆ O
ethylcyclohexane	C ₈ H ₁₆	2-propanol	C ₃ H ₈ O
1,1-dimethylcyclopentane	C ₇ H ₁₄	methyl- <i>t</i> -butyl ether (MTBE)	C ₅ H ₁₂ O
ethylcyclopentane	C ₇ H ₁₄	acetonitrile	C ₂ H ₃ N
monoterpenes		3-methylfuran	C ₅ H ₆ O
α -pinene	C ₁₀ H ₁₆		