

Reply to the comments of Reviewer #1 on the manuscript “Greenhouse gas measurements from a UK network of tall towers: technical description and first results”

Kieran M. Stanley¹, Aoife Grant¹, Simon O’Doherty¹, Dickon Young¹, Alistair J. Manning^{1,2}, Ann R.
5 Stavert¹, T. Gerard Spain³, Peter K. Salameh⁴, Christina M. Harth⁴, Peter G. Simmonds¹, William T.
Sturges⁵, David E. Oram⁵, Richard G. Derwent⁶

¹School of Chemistry, University of Bristol, Bristol, United Kingdom

²Met Office, Exeter, Devon, United Kingdom

10 ³Department of Experimental Physics, National University of Ireland, Galway, Ireland

⁴Scripps Institution of Oceanography, University of California San Diego, La Jolla, California, USA

⁵School of Environmental Sciences, University of East Anglia, Norwich, United Kingdom

⁶rdscientific, Newbury, Berkshire, United Kingdom

15

We thank the reviewer #1 for their time and effort in evaluating this manuscript and for their suggestions
for improvements. All points made by the reviewer are addressed on the following pages.

20 **The manuscript presents a comprehensive overview of the recently set-up network of greenhouse
gas measurements at tall towers in the UK. The paper is clearly structured and well written. It
provides much useful information for readers that also aim at establishing monitoring capabilities
at tall towers. Therefore, the manuscript merits publication in Atmospheric Measurement
Techniques. However, the paper summarizes several things already published in the literature.**

Thus, I feel that some paragraphs of the manuscript could be shortened by minimizing repetition of already published technical description.

Response: It is understood that the paper summarises a number of things already published in the literature and that some paragraphs could be shortened.

5 Modification: A number of sections have been modified, as outlined below:

- Page 7, line 4: “This tubing is made from high-density polyethylene bonded to overlapped aluminium tape and has a total wall thickness of 1.57 mm (Andrews et al., 2014). The outer polyethylene coating makes it resistant to water (H₂O) vapour condensation on the inner aluminium core tube.” was removed.
- 10 - Page 7, line 13: “For each inlet at RGL and TAC, an inverted stainless steel cup covers the inlet, acting as a shield to prevent H₂O entering the line. A monel mesh screen is inserted within the cup to help prevent large particles from entering the inlet lines.” was changed to “For each inlet at RGL and TAC, an inverted stainless steel cup with a monel mesh screen inserted within the cup covers the inlet, acting as a shield to prevent H₂O entering the line.”
- 15 - Page 7 line 23-24: “The filters are present to protect instruments and pumps from particles (Andrews et al., 2014)” was removed.
- Page 10, line 3-10: “H₂O can damage system components and interfere with measurements of GHGs, even at low levels (Andrews et al., 2014). H₂O influences mole fractions of GHGs measured via CRDS through a dilution effect, whereby a difference of 100 $\mu\text{mol mol}^{-1}$ (or parts per million; ppm) H₂O can cause a “dilution offset” of 0.04 $\mu\text{mol mol}^{-1}$ in CO₂ (Andrews et al., 2014). Also, H₂O vapour differences between calibration gases and air samples can cause spectral issues within optical instruments. H₂O causes these spectral artefacts through line broadening effects on the spectroscopic line shapes, specifically Lorentzian broadening and Dicke line narrowing (Chen et al., 2010; Rella et al., 2013). The extent of these effects is dependent on the atmospheric mixing ratio of H₂O (Chen et al., 2010)” changed to “H₂O can damage system components and interfere with measurements of GHGs, even at low levels, through a dilution effect (Andrews et al., 2014) and pressure broadening effects (Chen et al., 2010; Rella et al., 2013)”.
- 25 - Page 10, line 13: “The effectiveness of permeation Nafion dryers is dependent on the H₂O partial pressure gradient between the sample and counter purge flows (Andrews et al., 2014)” was removed.
- 30 - Page 11, line 7 – page 12, line 10: two paragraphs changed to “The SF₆ and N₂O analysis method used at RGL and TAC was similar to that described in detail in Ganesan et al. (2013), except P-5 carrier gas (a mixture of 5 % CH₄ in 95 % Ar; Air Products, UK) is used. Briefly, calibration gas and air samples are flushed through an 8 mL sample loop at 40 mL min⁻¹ for 60 seconds at a fixed exhaust pressure (~ 20 psi; Fig. 3 Backflush) before decaying down to ambient pressure. Flow through the loop is controlled by a ‘RED-y’ smart series mass flow controller (GSC-A4TA-BB22, Voeglin Instruments AG, Switzerland) and pressure in the loop
- 35

is measured using an ‘All Sensor’ pressure sensor (100PSI-A-DO, All Sensors, BS-Rep GmbH, Germany). Once equilibrated to ambient pressure, samples are then injected through an 8-port, 2-position valve (V3 in Fig. 2; EUDAC8UWEPH, VICI Valco AG International, Switzerland) onto a pre- (1.0 m Porapak Q, 80/100 mesh, 3/16” O.D.) and main column (2.0 m Porapak Q, 80/100 mesh, 3/16” O.D.) held at 90 °C, where N₂O and SF₆ are separated from air. Oxygen is “heart-cut” to vent (V3, Fig. 3) after it has eluted from the two columns, whilst the pre-column is back flushed with P5. The remaining, O₂ minimised, sample flows through the post-column (0.9 m of 1/8” O.D. stainless steel packed with molecular sieve 5 Å, 45/60 mesh), housed in a thermostatically controlled heated inlet port of the GC at 180 °C. The post column reverses the elution order of SF₆ and N₂O to prevent the larger N₂O peak from tailing into the small SF₆ peak, improving sample reproducibility and precision. Detection occurs in the ECD which is held at 350 °C. The ECDs at TAC and RGL measure at a rate of 10 and 20 Hz respectively. Samples are dried using a Nafion Dryer (MD-050-72S-1, Perma Pure, USA) with a dry zero air counter purge (as outlined in Sect. 3.3)”.

- Page 12, line 12: “The use of P5 carrier gas enables the omission of CO₂ doping” was removed.
- Page 12, line 13: “N₂O co-elutes with CO₂ on the column combination used within the UK DECC network, saturating the MS 5 Å and providing a constant doping effect, thus care is taken to make sure the N₂O response is not affected by this” was changed to “N₂O co-elutes with CO₂ on the post-column, saturating the MS 5 Å and providing a constant doping effect and reducing precision”.
- Page 12, line 23 – page 13, line 11: paragraph changed to “CO and H₂ are measured at two sites, MHD and TAC, using a RGA (RGA3 (MHD) and Peak Performer 1 (TAC), Trace Analytical Inc., USA). Table 4 outlines RGA instrumental setup at TAC and MHD. The MHD RGA setup is different to TAC and is outlined in (Prinn et al., 2000). The TAC sample selection system is integrated within the GC-ECD system (Sect. 3.4) (Grant et al., 2010a; Grant et al., 2010b). The GC-ECD has a 10-port, 2-position valve (VICI Valco AG International, Switzerland) for V2 (Fig. 2), instead of an 8-port 2-position valve, as at RGL. This allows for a 1 mL RGA sample loop to be put in sequence before the ECD sample loop (Fig. 3 TAC). After samples have been dried using the Nafion Dryer (MD-050-72S-1, Perma Pure, USA), passed through the sample loops and decayed to ambient pressure, they are injected onto two isothermal packed columns held at 105 °C: a 0.768 m pre-column (1/8” O.D. stainless steel packed with 60/80 mesh Unibeads 1S) and a 0.768 m main column (1/8” O.D. stainless steel packed with MS 5 Å, 60/80 mesh). After separation, gases are injected into the RGA for analysis using zero air plus (Air Products, UK) carrier gas, where the samples pass over a heated bed of mercuric oxide before being quantitatively determined using UV photometry (Grant et al., 2010a; Grant et al., 2010b)”.

At the same time, the paper somehow lacks novelty or the author missed to emphasize the new approaches applied here. I suggest to highlight the new approaches and, in addition, to provide

more information likely of interest for the reader. Issues to be addressed in this respect are (a) the required maintenance of the measurements (e.g. how did the regular maintenance look like, how many maintenance (regular as well as “emergency”) visits were needed over the years); (b) how did the troubleshooting look like; (c) which major technical problems were the authors facing; and (d)

5 **add a paragraph on lessons learnt in the conclusions**

Response: Two extra sections have been added to include information on maintenance (section 3.8; Table 5) and troubleshooting data issues (section 5.4; Table 8) as suggested.

How does this exercise refer to European efforts like ICOS, in terms of instrumentation, quality control and data processing?

10 Response: Instrumentation used within the UK DECC network is very similar to those used for CO₂ and CH₄ measurements in ICOS (Yver Kwok et al., 2015; Hazan et al., 2016) and other non-European measurement programmes, such as the Los Angeles Mega City Project (Verhulst et al., 2017). However, the data processing and quality control is different within ICOS to the UK DECC network. The

15 ICOS method calibrates data by linear interpolation between a linear fit of a suite of calibration gases (n = 4) spanning above and below ambient concentrations (Hazan et al., 2016), whereas the method used in the UK DECC network uses a daily standard to calibrate out daily instrumental drift and then suite of calibration cylinders to correct for effects above and below ambient mole fractions.

20 **Specific comments:**

Page 1, line 17: I don’t agree with the wording “automated custom-built instrumentation” here as a large part of the instruments is commercially available.

Response: “custom-built” removed.

25

Page 2, lines 11-13: “. . .independent emission estimates for comparison with the UK national inventory . . .”: if this is mentioned in the abstract, I expected to see some related results in the manuscript but I couldn’t find it.

Response: This sentence has been removed as these modelling results are outside of the scope of
5 this paper.

Page 4, line 6: “. . .except MHD which samples every 20 minutes . . .”: this statement is in contradiction to Table 1 where a Picarro G2301 is listed for MHD.

Response: “Except MHD which samples every 20 minutes” has been removed.
10

Page 5, line 19 – 20: Here for RGL and also further below for TAC: what is the rationale for measuring at several heights?

Response: CO₂ and CH₄ are measured at several heights on the tall towers to try and assess for boundary layer stratification. Measurements at several inlet heights with instruments that are able to
15 measure at high-frequency has the added benefit of aiding troubleshooting of data.

Page 7, lines 3 – 10: why Synflex tubing is used at RGL and TAC while stainless steel tubing is used at MHD. What are the pros and cons for using one of them? What is the difference between Synflex 1300 and Synflex 3000?

Response: Historically, stainless steel (SS) has been used at MHD when it was set up as a sampling site for AGAGE in 1994, as well as for the predecessor research programmes, as SS does not outgas any of the halocarbons measured in the research programme. The relatively short run of tubing from the inlet to the line pump at MHD meant that ¼” i.d. SS tubing gave the desired flow rates with the line pumps used. However, for the tall towers, a wider i.d. tubing (½” i.d.) was necessary to get the necessary flow
20 rates to flush the tubing on the tall towers well enough. The weight of ½” i.d. stainless steel tubing on the tall towers, as well as the impracticality and financial cost of installing a continuous length of ½” i.d. SS tubing, Synflex tubing was used instead. Double bonded aluminium core tubing, such as Synflex, is used
25

routinely on tall tower sites (it is recommended in the ICOS specifications: <http://icos-atc.lscce.ipsl.fr/filebrowser/download/27251>) due to its flexibility, light weight and cost.

The difference between Synflex 1300 and Synflex 3000 is the material used for the inner core of the tubing. The former has a polyethylene inner core, whilst the latter has a nylon inner core. Both plastics outgas; hence, the flow rates at the sites were kept high ($\sim 20 \text{ L min}^{-1}$) to ensure that no build-up of trace gases being measured at the sites occurred.

Page 7, line 17: I suppose that only the bowl is made out of Perspex. What is the material used for the other wetted parts like the body, the seals and the gaskets?

Response: Correct, only the bowl is made of Perspex on the H_2O decanting bowl. The other wetted parts of the system are made of the following: the housing is aluminium, the filter is sintered polypropylene and the seals are either nitrile or neoprene. The sentence on page 7, line 17, has been reworded to make it clearer what the wetted parts of the system are.

Page 9, line 22 to page 10, line 1: If the temperature is maintained at $318 \pm 0.004 \text{ K}$, line 1 must read a few / thousandth of a K . . ., correct?

Response: Correct, changed to thousandths.

Page 10, line 11: 0.25 % volume ratio of water: which dew point is that?

Response: 0.25 % H_2O is -10.3°C dew point. This has been added to the sentence in parentheses.

Page 10, lines 12 – 14: did you test for potential CO_2 losses in the Nafion?

Response: No tests were undertaken by the authors themselves before setting up the network but we had spoken to collaborators at AGAGE and Scripps Institute of Oceanography about the use of Nafion dryers and their tests into cross membrane leakage of CO_2 and CH_4 ; work presented in Welp et al. (2013). CO_2 and CH_4 bias from cross-membrane permeation was observed by the authors; however, the procedure

of passing both the sample air and calibration gases through the Nafion cancels out the bias (Welp et al., 2013).

Page 10, lines 17 – 21: Correction for water vapour interferences: are the correction coefficients listed in Rella (2010) the ones that are implemented in the Picarro software? If so, the direct Picarro output internally corrected for H₂O cross-talk can be used right away, correct? Did you also use the Rella (2010) factors for TTA when running the measurements w/o dryer? If so, why didn't you use individually determined correction factors as suggested by Rella et al. (2013) (<https://www.atmos-meastech.net/6/837/2013/>).

Response: The correction coefficients listed in (Rella, 2010) are implemented in the Picarro software to derive CO₂ and CH₄ dry mole fractions and were used at TTA to correct data to dry mole fractions. No individually determined correction factor were assessed or implemented at TTA due to time constraints and site access issues.

Page 12, lines 11 – 12: It reads like it was a new idea of the authors to use 5% CH₄ in Ar (i.e. P5) instead of CO₂-doped N₂ as carrier gas. However, this is already done by many groups for many years. You may refer to Schmidt et al. (2001) (JGR; <http://onlinelibrary.wiley.com/doi/10.1029/2000JD900701/epdf>).

Response: The sentence was reworded as part of the first comment on minimising repetition of previously published work. The above reference has also been added to the sentence so that it now reads: "The SF₆ and N₂O analysis method used at RGL and TAC was similar to that described in detail in Ganesan et al. (2013), except P-5 carrier gas (a mixture of 5 % CH₄ in 95 % Ar; Air Products, UK) is used (Schmidt et al., 2001)."

Page 16, line 13: what is a "significant change"? Was the change detected automatically or manually (visually)? Which criterion was applied?

Response: Any changes in the coefficients from the curve of the nonlinearity correction would be classed as a significant change. The word significant was removed from the sentence to make the sentence clearer. Changes were detected manually when comparing the previous coefficients with the new coefficients. The sentence was altered to make this clearer.

5

Page 16, lines 14 – 15: “A second order non-linear curve is fit to the data . . .”: Does that mean that the Picarro has a non-linear response? If so, please state it clearly and elaborate on it and quantify the effect when erroneously neglecting the non-linear response.

Response: For all of the UK DECC network sites there is a small non-linear response on the
10 CRDSs. The non-linear response is not thought to be due to calibration gases used within the network as the non-linear effect has been seen when NOAA standards have been used to calibrate instruments that use GCWerks. An extra figure has been added to show an example of the non-linear fits used within the network (Figure 4) and histograms to show the offsets between data with and without the non-linear corrections (Figure 5). Additional information has been put in the paragraph to exemplify what the effect
15 would be on the data if the non-linear correction wasn't applied. There was a small difference between the two steps in data correction, with a median value of $< -0.01 \mu\text{mol mol}^{-1} \text{CO}_2$ and $< -0.002 \text{ nmol mol}^{-1} \text{CH}_4$ (linear corrected - non-linear corrected data).

Page 17, lines 9 – 11: If I understand correctly, only a one point calibration approach is applied. This only works when assuming no detector signal at zero concentration, right. Was that tested and how was the one point calibration approach applied when a GC-MS signal > 0 was detected for species-free air?

Response: Yes, the one point calibration approach does assume that no detector signal at zero concentration for the Medusa GC-MS and was tested when the Medusa system was being developed at
25 Scripps Institute of Oceanography (Miller et al., 2008; Arnold et al., 2012). Weekly, a system blank is analysed on the Medusa GC-MS, whereby He is passed through the pre-concentration system instead of a sample gas to the GC-MS. The system blank allows the detection of signals > 0 in species-free gas and

GCWerks can then integrate the contamination peak. A blank correction can then be implemented to cancel out the contamination.

5 **Page 17, lines 25 – 26: how did you make sure that there were no traces of N₂O and SF₆ in the zero air? The zero air, was it real air having N₂O and SF₆ trapped or was it a N₂/O₂ mixture? How about the Argon content in the zero air?**

10 Response: The zero air used for the non-linearity tests was analysed for traces of N₂O and SF₆ during the non-linearity test sequence. Any cylinders with detectable N₂O or SF₆ contamination were not used. A sentence has been added to clarify this. The zero air used for the tests was a synthetic blend of N₂/O₂, with Ar contents < 0.01 μmol mol⁻¹.

Page 18, 8: troubleshooting is mentioned here but it remains too vague if and how and how often troubleshooting was required. See one of my general comments above.

15 Response: Created section 5.4 to discuss troubleshooting of data within the network.

Page 18, line 18: I suggest skipping the explanation of the naming convention of the stripchart files. This is irrelevant.

Response: Removed.

20 **Page 18, line 23: remove “with time stamps corresponding to the beginning of the measurement, and stored“ It isn’t of importance here.**

Response: Removed

Page 19, line 5: add a table with the thresholds for the maximum allowed standard deviations?

25 Response: Added in Table 7 with parameter thresholds for all CRDS filters.

Page 19, line 24: remove explanation of the naming convention.

Response: Removed

Page 20, line 1: is it important that there was a 4:1 compression ratio?

5 Response: Removed

Page 21, lines 3 – 5: how often did it happen that the processing routines filtered false negatives? Is that a time consuming (and important) task to review the automatically filtered data?

10 Response: When the data filters were introduced to the software and the parameterisation of the filters had not been finely tuned, there were a number of occasions when the air filter was not set high enough and real pollution events were filtered out. Through reviewing the stripcharts, this was spotted and the standard deviation filter was increased to 10. It is important to check all of the data, including filtered data, to ensure that no instrumental problems are missed due to the filtering of data. Within GCWerks, all data is shown within the stripcharts, so the viewing of the stripcharts is not that time
15 consuming (a day's worth of data can usually be checked in about 5 minutes).

Page 21, lines 9 – 10: Is the flagging of spurious data a manual process? If so, I suggest clarifying it by saying "Spurious data are manually flagged and a justification . . . can be added and logged."

20 Response: "manually" added.

Page 21, line 11: what is "GCcompare"?

Response: GCcompare is data visualisation software, where data within a tab delimited format can be imported to the visualisation software and the time series data can be overlain with other time series data from other sites for the same compound to look and investigate errors.

25 **Page 22, lines 18 – 19: where are the 21 nmol mol⁻¹ coming from? Add a reference.**

Response: The amplitude data is from Mace Head. This has been clarified in the sentence.

Page 23, lines 2 -3: Add a statement that the trend of 0.8 nmol mol⁻¹ per year is lower than the global trend. You may refer to the latest WMO GHG bulletin #12
5 (https://library.wmo.int/opac/doc_num.php?explnum_id=3084). What do you mean by seasonal trend, isn't it the annual growth rate?

Response: There is a 0.8 nmol mol⁻¹ amplitude in the seasonal trend at MHD, not the annual growth rate. The sentence has been altered to clarify this.

10 Page 23, chapter 6.2: this is all largely text-book knowledge and can be considerably shortened.

Response: The section has been shortened.

Page 25, Summary and conclusions: It only summarizes what was said before. I would like to read some kind of outlook and some recommendations that go beyond a simple description of the setup
15 as given above. Topics to be potentially addressed could be: are there any modifications planned (based on some lessons –learnt); are there any major flaws in the setup which cannot be easily changed anymore; with the experience gained during the few years of operation, would the setup again look the same when you may be able to once more start from scratch?

Response: A new recommendations section has been added (section 7) to discuss
20 recommendations based on experiences within the network. Additionally, future work planned in the network has been added the conclusion.

Page 26, lines 11 – 13: The reference given here to underline the benefit for such measurements for GHG inventory verification was published in 2011, i.e. before the presented measurements were
25 implemented. Either remove the reference to the emission verification on page 26 (and in the abstract) or elaborate on the benefit of additional observations for the GHG inventory assessment based on tall-tower measurements and inverse modelling.

Page 31, footnote a to Table 1: I cannot find the data on the ICOS carbon portal as indicated.

Response: When accessed on 12/01/18, data was found to be available in the ICOS data portal

5 search (<https://data.icos-cp.eu/portal/#search?station=%5B%22Mace%20Head%20%22%5D>).**References:**

- Andrews, A. E., Kofler, J. D., Trudeau, M. E., Williams, J. C., Neff, D. H., Masarie, K. A., Chao, D. Y., Kitzis, D. R., Novelli, P. C., Zhao, C. L., Dlugokencky, E. J., Lang, P. M., Crotwell, M. J., Fischer, M. L., Parker, M. J., Lee, J. T., Baumann, D. D., Desai, A. R., Stanier, C. O., De Wekker, S. F. J., Wolfe, D. E., Munger, J. W., and Tans, P. P.: CO₂, CO, and CH₄ measurements from tall towers in the NOAA Earth System Research Laboratory's Global Greenhouse Gas Reference Network: instrumentation, uncertainty analysis, and recommendations for future high-accuracy greenhouse gas monitoring efforts, *Atmos. Meas. Tech.*, 7, 647-687, 10.5194/amt-7-647-2014, 2014.
- 10 Arnold, T., Mühle, J., Salameh, P. K., Harth, C. M., Ivy, D. J., and Weiss, R. F.: Automated Measurement of Nitrogen Trifluoride in Ambient Air, *Analytical Chemistry*, 84, 4798-4804, 10.1021/ac300373e, 2012.
- 15 Chen, H., Winderlich, J., Gerbig, C., Hofer, A., Rella, C. W., Crosson, E. R., Van Pelt, A. D., Steinbach, J., Kolle, O., Beck, V., Daube, B. C., Gottlieb, E. W., Chow, V. Y., Santoni, G. W., and Wofsy, S. C.: High-accuracy continuous airborne measurements of greenhouse gases (CO₂ and CH₄) using the cavity ring-down spectroscopy (CRDS) technique, *Atmospheric Measurement Techniques*, 3, 375-386, doi:10.5194/amt-3-375-2010, 2010, 2010.
- 20 Ganesan, A. L., Chatterjee, A., Prinn, R. G., Harth, C. M., Salameh, P. K., Manning, A. J., Hall, B. D., Muhle, J., Meredith, L. K., Weiss, R. F., O'Doherty, S., and Young, D.: The variability of methane, nitrous oxide and sulfur hexafluoride in Northeast India, *Atmos. Chem. Phys.*, 13, 10633-10644, doi: 10.5194/acp-13-10633-10644-2013, 2013.
- 25 Grant, A., Stanley, K. F., Henshaw, S. J., Shallcross, D. E., and O'Doherty, S.: High-frequency urban measurements of molecular hydrogen and carbon monoxide in the United Kingdom, *Atmos. Chem. Phys.*, 10, 4715-4724, 10.5194/acp-10-4715-2010, 2010a.
- Grant, A., Witham, C. S., Simmonds, P. G., Manning, A. J., and O'Doherty, S.: A 15 year record of high-frequency, in situ measurements of hydrogen at Mace Head, Ireland, *Atmos. Chem. Phys.*, 10, 1203-1214, 10.5194/acp-10-1203-2010, 2010b.
- Hazan, L., Tarniewicz, J., Ramonet, M., Laurent, O., and Abbaris, A.: Automatic processing of atmospheric CO₂ and CH₄ mole fractions at the ICOS Atmospheric Thematic Center, *Atmos. Meas. Tech.*, 9, 1-34, 10.5194/amt-9-4719-2016, 2016.
- 30 Miller, B. R., Weiss, R. F., Salameh, P. K., Tanhua, T., Grealley, B. R., Muhle, J., and Simmonds, P. G.: Medusa: A Sample Preconcentration and GC/MS Detector System for in Situ Measurements of Atmospheric Trace Halocarbons, Hydrocarbons, and Sulfur Compounds, *Anal. Chem.*, 80, 1536-1545, 2008.
- Prinn, R. G., Weiss, R. F., Fraser, P. J., Simmonds, P. G., Cunnold, D. M., Alyea, F. N., O'Doherty, S., Salameh, P., Miller, B. R., Huang, J., Wang, R. H. J., Hartley, D. E., Harth, C., Steele, L. P., Sturrock, G., Midgley, P. M., and McCulloch, A.: A history of chemically and radiatively important gases in air deduced from ALE/GAGE/AGAGE, *J. Geophys. Res.*, 105, doi:10.1029/2000JD900141, 2000.
- 35 Rella, C.: Accurate greenhouse gas measurements in humid gas streams using the Picarro G1301 carbon dioxide/methane/water vapor gas analyzer, White paper, Picarro Inc, Sunnyvale, CA, USA, 2010.
- 40 Rella, C., Chen, H., Andrews, A. E., Filges, A., Gerbig, C., Hatakka, J., Karion, A., Miles, N. L., Richardson, S. J., Steinbacher, M., Sweeney, C., Wastine, B., and Zellweger, C.: High accuracy measurements of dry mole fractions of carbon dioxide and methane in humid air, *Atmos. Meas. Tech.*, 6, 837-860, 10.5194/amt-6-837-2013, 2013.

- Schmidt, M., Glatzel-Mattheier, H., Sartorius, H., Worthy, D. E., and Levin, I.: Western European N₂O emissions: A top-down approach based on atmospheric observations, *Journal of Geophysical Research: Atmospheres*, 106, 5507-5516, 10.1029/2000JD900701, 2001.
- 5 Verhulst, K. R., Karion, A., Kim, J., Salameh, P. K., Keeling, R. F., Newman, S., Miller, J., Sloop, C., Pongetti, T., Rao, P., Wong, C., Hopkins, F. M., Yadav, V., Weiss, R. F., Duren, R. M., and Miller, C. E.: Carbon dioxide and methane measurements from the Los Angeles Megacity Carbon Project – Part I: calibration, urban enhancements, and uncertainty estimates, *Atmos. Chem. Phys.*, 17, 8313-8341, 10.5194/acp-17-8313-2017, 2017.
- 10 Welp, L. R., Keeling, R. F., Weiss, R. F., Paplawsky, W., and Heckman, S.: Design and performance of a Nafion dryer for continuous operation at CO₂ and CH₄ air monitoring sites, *Atmos. Meas. Tech.*, 6, 1217-1226, 10.5194/amt-6-1217-2013, 2013.
- Yver Kwok, C., Laurent, O., GueMRI, A., Philippon, C., Wastine, B., Rella, C. W., Vuillemin, C., Truong, F., Delmotte, M., Kazan, V., Darding, M., Lebègue, B., Kaiser, C., Xueref-Rémy, I., and Ramonet, M.: Comprehensive laboratory and field testing of cavity ring-down spectroscopy analyzers measuring H₂O, CO₂, CH₄ and CO, *Atmos. Meas. Tech.*, 8, 3867-3892, 10.5194/amt-8-3867-2015, 2015.

Reply to the comments of Reviewer #2 on the manuscript

“Greenhouse gas measurements from a UK network of tall towers: technical description and first results”

Kieran M. Stanley¹, Aoife Grant¹, Simon O'Doherty¹, Dickon Young¹, Alistair J. Manning^{1,2}, Ann R. Staver¹, T. Gerard Spain³, Peter K. Salameh⁴, Christina M. Harth⁴, Peter G. Simmonds¹, William T. Sturges⁵, David E. Oram⁵, Richard G. Derwent⁶

¹School of Chemistry, University of Bristol, Bristol, United Kingdom

²Met Office, Exeter, Devon, United Kingdom

³Department of Experimental Physics, National University of Ireland, Galway, Ireland

⁴Scripps Institution of Oceanography, University of California San Diego, La Jolla, California, USA

⁵School of Environmental Sciences, University of East Anglia, Norwich, United Kingdom

⁶rdscientific, Newbury, Berkshire, United Kingdom

We thank the reviewer #2 for their time and effort in evaluating this manuscript and for their suggestions for improvements. All points made by the reviewer are addressed on the following pages.

The paper is describing the setup for greenhouse gases measurements at four sites in UK. The authors are providing a very detailed description of the inlet parts, analyzers, and calibration protocols. In the last section of the paper atmospheric signals (diurnal and seasonal cycles, trend) are very briefly discussed. Considering the purpose of the manuscript I am not fully convinced by the need for this very general discussion on the observed variabilities, coming after very detailed technical descriptions of the setup and protocols. From my point of view, the most problematic point of this manuscript is the lack of analysis of the measurements in terms of uncertainties and quality control. There are very few quantitative indicators which could help to justify some choices

in the protocols. For CRDS measurements only precisions estimates are provided, whereas repeatability's are provided for GCs. Even more problematic is the use of the precision (based on standard deviations calculated over one minute intervals) as a 'demonstration' that measurements are compliant with WMO recommendation for compatibility of monitoring sites. But it seems that
5 few efforts were put to characterize the possible biases (e.g. no measurement of a target gas as recommended by WMO/GAW). I would expect at least some analysis of the existing information (e.g. variability of the calibration residuals), description of the troubleshooting, etc. . . Also I am very surprised some suspicious signals are not discussed at all, even though the protocol described in the manuscript claims a significant emphasis of the data review.

10 Response: The aim of this paper was to predominantly describe the setup of the UK DECC network, including the instrumentation used, sampling and calibration protocols and data processing methods to help guide the setup of other future sites and networks. The addition of the data since the network starts help show what data there is available and the potential uses for it. No in depth analysis was presented in this paper of the data as the authors felt that this would be better done in a paper where
15 interesting signals and patterns can be looked at in conjunction with inverse modelling results.

No quantification of uncertainty within the data has been presented in this paper as this is something that the authors want to work on to include with the data. This has been outlined in the final paragraph of the summary and conclusions, where we have added in the future improvements for the network. We realise the added benefit of using target tanks as an independent quality control measure;
20 however, when the network was set up, we used protocols based on GC measurements within AGAGE, which has not historically measured target tanks. The inclusion of target tanks within the network is another aspect that we are looking into to include within the network.

We recognise the issues related to the use of precision information for CRDS measurements and repeatability values for GC measurements. We have changed our precision data to short-term precision
25 and included repeatability data for the CRDS instruments, which is outlined in the specific comments below. In addition, we have removed the comparison with the WMO recommendations as we realise that our precision data does not fully encapsulate the error from measurements in the network.

Three extra sections have been added to this manuscript to improve information on the reader, including a troubleshooting section (section 5.4) and a section on maintenance (section 3.8) and a recommendations section for readers wishing to create future sites or stations.

5 **Page 1, line 14: “A network of . . .three. . . tall tower”**

Response: Done.

Page 1, line 24: “The long-term 1-minute mean precisions (1s)” You should show an indicator comparable to GC

10 Response: This has been changed to short-term precision (1σ of 1 minute mean data per standard injection) and a repeatability of standard injections (1σ of 20-minute mean injections).

Page 3, line 2: “the accuracy of the inversion is limited by the number and distribution of measurement locations available”: I would suggest to mention also the capacity of models to properly represent the observed time series, which lead to favor tall towers over flat terrains.

15 Response: Done.

Page 3, line 9: “Measurements. . .constrained global or hemispheric scale fluxes”: I would not say that atmospheric measurements constrained the fluxes (also on the next line of the same paragraph). Hopefully they can give some constraints to the estimation of the fluxes by inverse method.

20 Response: Text altered to give clarity for both sentences.

Page 4, line 3: I do not think the list of species measured by Medusa is needed in Table 1. Please refer to a publication.

25 Response: Medusa species removed, apart from SF₆, and Miller et al. (2008) and Arnold et al. (2012) papers were referred to.

Page 5, line 4: 51% of marine air at MHD: please give a reference

Response: Reference added

Page 6, line 2: Is there a reason why you chose to sample N₂O, SF₆, CO at 100m a.g.l. and not at the highest point of the tower (185m). Also, why did not you adopt the same sapling strategy at TTA where you have a single inlet for CO₂, CH₄, whereas you set up 2 or inlets at the other tall towers ? Could you please explain your choices ?

Response: The reason for sampling N₂O, SF₆ and CO at the 100 m.a.g.l. inlet is purely historical. When the site was setup in 2012, the 185 m.a.g.l. inlet was not installed. A sentence has been added about the later installation of the 185m inlet. Multiple inlets were chosen at sites except TTA to try and assess boundary layer stratification and is a protocol used by a number of different networks with tall tower sites, including ICOS. The added advantage of having multiple inlets is that if a line pump fails or a filter blocks on an inlet, the affected line can be taken out of the sampling strategy. Additionally, data from other inlets can be used to help diagnose issues in data by looking at patterns between each inlet (see section 5.4 for more details). The sampling line set up at TTA was different as the site was inherited by the University of Bristol and the cost of adding additional sampling lines was too much for that site.

Page 7, line 10: What do you mean by: 'Horizontal sections of tubing at the base of the tower were kept to a minimum' ? Please clarify.

Response: From the base of the towers, the tubing runs horizontally along trunking until it reaches the laboratory. We tried to minimise these long horizontal stretches to prevent water from accumulating and the potential for contaminants to alter the composition of the air drawn down from the inlet.

Page 7, line 17: Have you tested the Perspex H₂O decanting bowls to ensure the noncontamination of the measurements ? Why did you changed from the previous system used at TTA ?

Response: Yes, the Perspex H₂O decanting bowls were tested in the laboratory before being installed at. A steady gas stream of ambient air (whole air compressed into a cylinder) was passed through

Synflex 1300 tubing and analysed on a CRDS with and without the H₂O decanting bowls in line to observe for differences in CO₂ and CH₄ mole fractions. No effect was found. A sentence has been added to state that the bowls were tested before being installed. Perspex H₂O decanting bowls were used in preference over the stainless steel bowls to enable site operators to see if H₂O had accumulated within the bowl.

5

Figure 2: Can you precise the meaning of TOC on the figures.

Response: TOC (zero air generator) has been added to the figure legend.

10 **Page 8, line 20: “This has the advantage of eliminating sample contamination from the pump, reducing the likelihood of a torn diaphragm introducing laboratory air into the sample and improving the performance of Nafion dryers.” Can you please elaborate on this sentence ? The first point is clear, but I understand that the inlet is slightly under pressure which is not an advantage to avoid the contamination from lab air.**

Response: The Nafion section was removed as it was found to be incorrect.

15

Page 10, line 20: “The correction applied is minimized due to the removal of most H₂O using the Nafion dryer.” In return you introduced a possible CO₂ bias due to the nafion. How have you quantified this bias ?

20 Response: The passing of calibration and standard gases through the Nafion dryer makes bias associated with CO₂ permeation across the membrane negligible (Welp et al., 2013; Andrews et al., 2014). Laboratory tests by Andrews et al. (2014) showed that losses of CO₂ from sample air and calibration gases were the same.

25 **Page 12, line 17: “Each cylinder is now individually analysed as a sample to check for contamination prior to use”. Can you precise how frequently you get problem with the purity of the carrier gas ?**

Response: The frequency of SF₆ contamination within P5 varies greatly as it depends on the cylinder age. SF₆ was used to pressure test cylinders nearing the end of their inspection date to check for

leaks and cylinder stability. On average, about one in every six cylinders were contaminated with trace amounts of SF₆. A section was also added in the recommendations about testing carrier gases before use (section 7.4).

- 5 **Page 15: “4.1 Sampling sequence” : Is there any justification of the different configurations at the sites. For example why 20 or 30 min sampling time ? Is it because of the number of sampling levels ? Also could you clarify if the GCs are measuring at only one level ?**

Response: As stated within the text, the justification for different configurations in CRDS sampling times is to ensure that all inlet heights are measured within one hour and is purely based on the number of sampling heights at the site. A sentence has been added to clarify that the GC are only sampled from one height.

- 15 **Page 16, line 5: Can you precise the lifetimes of the standard and calibration gases like you have done for the GCs standards ? Are the standards mixing ratios reevaluated at the end ?**

Response: A sentence has been added to the previous paragraph (first paragraph of section 4.2.1) stating the lifetimes of CRDS calibration and standard gases. A sentence has also been added stating that cylinders are recalibrated once removed from site.

- 20 **Page 16, line 9: “line flushing”: are you choices in duration, number of cycle and frequency of the calibration gases measurements based on specific tests ?**

Response: Choices were made according to suggestions made by ICOS when setting up the network rather than specific tests. Specifications used by ICOS are available in Hazan et al. (2016).

- 25 **Page 16, line 16: “long-term precision”: what about the measurement repeatability ?**

Response: This has been changed to long-term repeatability based on standard injections on the CRDS. A short-term precision has also been included.

Page 16, line 19: “CRDS precision within the UK DECC network is within the WMO compatibility guidelines for CO₂ ($\pm 0.1 \mu\text{mol mol}^{-1}$) and CH₄ ($\pm 2 \text{ nmol mol}^{-1}$) (WMOGAW, 2014)”: Indeed the precision is lower than the WMO recommendation for compatibility, but this is comparing apples and oranges. The WMO compatibility has to take into account the measurement repeatability, non-linearity, calibration uncertainties, H₂O correction, dryer and inlet biases. Please remove or rephrase correctly this sentence. Regarding WMO recommendations it should be noted that you are not following the recommendation of measuring a target tank (“Each analysis system must include at least one ‘target tank’ which is a very important quality control tool “, WMO/GAW report n° 229, 2015)

Response: Sentence removed.

Page 17, line 16: “Calibration scales vary depending on the gas species”: the CH₄ scale for GC-FID measurements is not given. Is it a different scale compared to CRDS measurements ? If so what about the compatibility of the two scales ? Also the CH₄ is not mentioned at all in paragraph 5.2 (GC data processing).

Response: CH₄ measured on the MHD GC-FID is on the Tohoku University scale. Based on in situ measurements made at MHD by AGAGE and NOAA flask measurements since 1993, there is an average difference of $1.01 \text{ nmol mol}^{-1}$ (NOAA-AGAGE) (Krummel, 2018). CH₄ has been added into section 5.2.

Page 17, line 24: “periodically in the field”: Can you precise the frequency of the nonlinearity tests for N₂O and SF₆ ?

Response: Tests were conducted approximately every year. The sentence has been altered to reflect this.

Page 19, line 4: “H₂O level too high”: what is the reason of rejecting values with high H₂O level ? what is the typical threshold values ?

Response: Data was filtered out when H₂O values were > 6% and was rejected as this was the highest mixing ratio used in the H₂O correction in Rella (2010) . Additionally, data with such a high H₂O concentration was indicative of liquid water passing the H₂O decanting bowl. Table 7 was added to show filter parameters.

Page 19, line 14: “A second order function can then be fitted to the data to provide a non-linearity correction”: CRDS instruments have the reputation to be quite linear. Why do you apply a non-linear correction? Have you estimated the importance of this term ?

Response: For all of the UK DECC network sites there is a small non-linear response on the CRDSs. The non-linear response is not thought to be due to calibration gases used within the network as the non-linear effect has been seen when NOAA standards have been used to calibrate instruments that use GCWerks. An extra figure has been added to shown an example of the non-linear fits used within the network (Figure 4) and histograms to show the offsets between data with and without the non-linear corrections (Figure 5). Additional information has been put in the paragraph to exemplify what the effect would be on the data if the non-linear correction wasn't applied.

Page 20, line 11: “using sample integrated height or area”: can you precise which one you are using for the different species ?

Response: Done

Page 23, line 3: “There is an approximate 0.8 nmol mol⁻¹ northern hemisphere midlatitude seasonal trend”: if you give an estimate of the trend for N₂O, I would suggest to do the same for all species, and gather this information in a table.

Response: This sentence has been altered to show the amplitude of the seasonal cycle.

Figure 5: How do you explain the low N₂O and SF₆ concentrations measured at RGL in the first year (2012) ? Those data look suspicious considering that such event do not appear later on. According to your description of the final data processing (5.3) you are evaluating the time series by comparing the stations with MHD. From my understanding this is typically a case where such comparison could lead to flagging the first year of measurements. Can you provide a possible explanation of this atmospheric signal ?

Response: Within the UK DECC network, data is not manually flagged out unless there is a specific reason for the spurious data, such as instrumental issues or leaks, even if the data looks odd against MHD. Instead, if no reason can be found, the data is left in the time series. The N₂O and SF₆ data shown in Figure 5 (now Figure 7) from the start of the time series to 11/07/2012 is noisier due to poorer instrument precision, which improved in July 2012 when more insulation was added to around the inlet of the post-column. Hence why both gases have periods with lower mole fractions than MHD.

Figure 7: Please precise if you are using local or UTC time on this figure.

Response: Done, UTC.

Figure 7c: CH₄ diurnal cycle at RGL shows a very high standard deviation in midday/summer. Outliers which could cause such anomaly on the 2012/2015 average signal are not visible in the time series on figure 4.

Response: Plot altered to show all of data.

Page 25, line 5: “Fig. 8(b) demonstrates a regionally polluted period at RGL for CH₄ on 30/11/2014, where air has passed over Europe and the south of the UK before arriving at the site”: in such a case we could expect higher CH₄ concentrations at MHD compared to TAC, due to the additional contribution from south UK. Do you such ‘reverse’ gradients sometimes ?

Response: If both sites are receiving the same air masses, then an increasing gradient of CH₄ mole fractions would be expected between RGL and MHD. However, in the instance shown in Figure 8(b), now figure 10(b), the air mass that RGL received was from over the midlands and south of the UK and the Benelux region and Germany; however, at this time, MHD was received air that had passed over the south west of the UK, the west of France and Spain as shown in Figure 1 below. The reverse gradient from that shown in Figure 10(b) – an increasing concentration from west to east over the UK, is frequently observed due to the UK predominant wind direction. Figure 10(a) demonstrates the increasing CH₄ mole fraction with longitude on the 4th and 5th December. A sentence has been added to section 6.3 to reflect this.

10

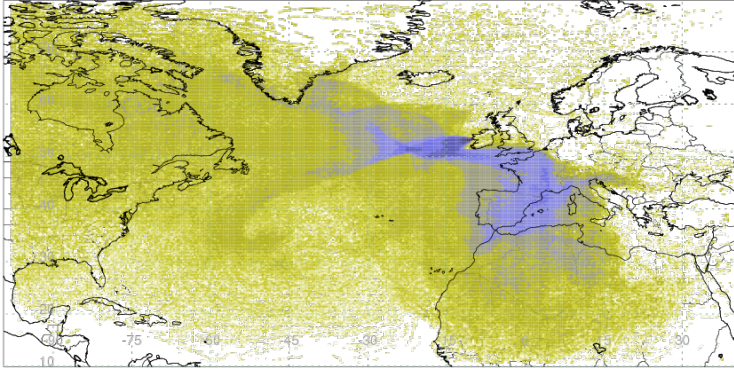


Figure 1: 2-hour air history maps derived from NAME for MHD on 29th November 2014. The air-history maps describe which surface areas (0–40m) in the previous 30-days impact the observation point within a particular 2-hour period.

15 **Page 26, line 6: “Results from the network give good spatial and temporal coverage”: I fully agree for the temporal resolution, but I would remove the comment on the spatial coverage, the relevance of which depends on the intended scientific purpose.**

Response: Spatial has been removed.

References

- Andrews, A. E., Kofler, J. D., Trudeau, M. E., Williams, J. C., Neff, D. H., Masarie, K. A., Chao, D. Y., Kitzis, D. R., Novelli, P. C., Zhao, C. L., Dlugokencky, E. J., Lang, P. M., Crotwell, M. J., Fischer, M. L., Parker, M. J., Lee, J. T., Baumann, D. D.,
5 Desai, A. R., Stanier, C. O., De Wekker, S. F. J., Wolfe, D. E., Munger, J. W., and Tans, P. P.: CO₂, CO, and CH₄ measurements from tall towers in the NOAA Earth System Research Laboratory's Global Greenhouse Gas Reference Network: instrumentation, uncertainty analysis, and recommendations for future high-accuracy greenhouse gas monitoring efforts, *Atmos. Meas. Tech.*, 7, 647-687, 10.5194/amt-7-647-2014, 2014.
- Arnold, T., Mühle, J., Salameh, P. K., Harth, C. M., Ivy, D. J., and Weiss, R. F.: Automated Measurement of Nitrogen
10 Trifluoride in Ambient Air, *Analytical Chemistry*, 84, 4798-4804, 10.1021/ac300373e, 2012.
- Hazan, L., Tarniewicz, J., Ramonet, M., Laurent, O., and Abbaris, A.: Automatic processing of atmospheric CO₂ and CH₄ mole fractions at the ICOS Atmospheric Thematic Center, *Atmos. Meas. Tech.*, 9, 1-34, 10.5194/amt-9-4719-2016, 2016.
- Krummel, P.: RE: MHD NOAA-AGAGE difference for CH₄. Personal communication to Stanley, K. M. on 15/01/2018.
- 15 Miller, B. R., Weiss, R. F., Salameh, P. K., Tanhua, T., Grealley, B. R., Mühle, J., and Simmonds, P. G.: Medusa: A Sample Preconcentration and GC/MS Detector System for in Situ Measurements of Atmospheric Trace Halocarbons, Hydrocarbons, and Sulfur Compounds, *Anal. Chem.*, 80, 1536-1545, 2008.
- Rella, C.: Accurate greenhouse gas measurements in humid gas streams using the Picarro G1301 carbon dioxide/methane/water vapor gas analyzer, White paper, Picarro Inc, Sunnyvale, CA, USA, 2010.
- 20 Welp, L. R., Keeling, R. F., Weiss, R. F., Paplawsky, W., and Heckman, S.: Design and performance of a Nafion dryer for continuous operation at CO₂ and CH₄ air monitoring sites, *Atmos. Meas. Tech.*, 6, 1217-1226, 10.5194/amt-6-1217-2013, 2013.

Greenhouse gas measurements from a UK network of tall towers:
technical description and first results

Kieran M. Stanley^{1*}, Aoife Grant¹, Simon O'Doherty¹, Dickon Young¹, Alistair J. Manning^{1,2}, Ann R. Stavert¹, T. Gerard Spain³, Peter K. Salameh⁴, Christina M. Harth⁴, Peter G. Simmonds¹, William T. Sturges⁵, David E. Oram⁵, Richard G. Derwent⁶

¹School of Chemistry, University of Bristol, Bristol, United Kingdom

²Met Office, Exeter, Devon, United Kingdom

³Department of Experimental Physics, National University of Ireland, Galway, Ireland

⁴Scripps Institution of Oceanography, University of California San Diego, La Jolla, California, USA

⁵School of Environmental Sciences, University of East Anglia, Norwich, United Kingdom

⁶rdscientific, Newbury, Berkshire, United Kingdom

Correspondence to: Kieran M. Stanley (k.m.stanley@bristol.ac.uk)

Abstract. A network of **three** tall tower measurement stations was set up in 2012 across the United Kingdom to expand measurements made at the long-term background northern hemispheric site, Mace Head, Ireland. Reliable and precise in situ greenhouse gas (GHG) analysis systems were developed and deployed at three sites in the UK with automated instrumentation measuring a suite of GHGs. The UK Deriving Emissions linked to Climate Change (UK DECC) network uses tall (165 - 230 m) open lattice telecommunications towers, which provide a convenient platform for boundary layer trace gas sampling. In this paper we describe the automated measurement system and first results from the UK DECC network for CO₂, CH₄, N₂O, SF₆, CO and H₂.

CO₂ and CH₄ are measured at all of the UK DECC sites by cavity ring-down spectroscopy (CRDS) with multiple inlet heights at two of the three tall tower sites to assess for boundary layer stratification. The **short-term precisions (1σ on 1-minute means)** of CRDS measurements at background mole fractions for January 2012 to September 2015 is <0.05 μmol mol⁻¹ for CO₂ and <0.3 nmol mol⁻¹ for CH₄. **Repeatability of standard injections (1σ) is <0.03 μmol mol⁻¹ for CO₂ and <0.3 nmol mol⁻¹ for CH₄ for the same time period.** N₂O and SF₆ are measured at three of the sites, and CO and H₂ measurements are made at two of the sites, from a single inlet height using gas chromatography (GC) with an electron capture detector

Deleted: custom-built

Deleted: long

Deleted: 1-minute mean

Deleted: 03

Deleted: 2

(ECD), flame ionisation detector (FID) or reduction gas analyser (RGA). Repeatability of individual injections (1σ) on GC and RGA instruments between January 2012 and September 2015 for CH_4 , N_2O , SF_6 , CO and H_2 measurements made using GC-ECD/FID or -RGA were $< 2.8 \text{ nmol mol}^{-1}$, $< 0.4 \text{ nmol mol}^{-1}$, $< 0.07 \text{ pmol mol}^{-1}$, $< 2 \text{ nmol mol}^{-1}$ and $< 3 \text{ nmol mol}^{-1}$, respectively.

- 5 Instrumentation in the network is fully automated and includes sensors for measuring a variety of instrumental parameters such as flow, pressures, and sampling temperatures. Automated alerts are generated and emailed to site operators when instrumental parameters are not within defined set ranges. Automated instrument shutdowns occur for critical errors such as carrier gas flow rate deviations. Results from the network give good spatial and temporal coverage of atmospheric mixing ratios within
- 10 the UK since early 2012. Results also show that all measured GHGs are increasing in mole fraction over the selected reporting period and, except for SF_6 , exhibit a seasonal trend. CO_2 and CH_4 also show strong diurnal cycles, with night-time maxima and daytime minima in mole fractions. ▼

Deleted: These data are used to produce independent emission estimates for comparison with the UK national inventory based on independent methods.

1 Introduction

- 15 Carbon dioxide (CO_2), methane (CH_4), nitrous oxide (N_2O), sulfur hexafluoride (SF_6), and carbon monoxide (CO) are potent greenhouse gases (GHGs), which have a significant influence on the earth's climate system (Stocker et al., 2013). H_2 is an important indirect GHG, due to its photochemical reaction with hydroxyl radicals (OH) in the troposphere reducing OH mole fractions, thus increasing the lifetime of CH_4 and affecting ozone production (Grant et al., 2010a; Grant et al., 2010b; Luan et al., 2016).
- 20 Atmospheric mole fractions of CO_2 , CH_4 , N_2O , SF_6 and CO have all exceeded pre-industrial levels due to anthropogenic activities (Kirschke et al., 2013; Stocker et al., 2013; Le Quéré et al., 2015). The increased concern about rising GHG emissions has already caused many nations to regulate their emissions. Inversion modelling techniques using data from atmospheric measurements can be used to derive emissions (Manning et al., 2011) and verify the national GHG inventories created using bottom up
- 25 approaches; however, the accuracy of the inversion is limited by the number and distribution of measurement locations available, as well as the capacity to properly represent observed time series data.

Remote measurements of GHGs first started in the 1950s at the Mauna Loa Observatory, Hawaii, USA. Remote background locations were chosen as to avoid strong anthropogenic sources encountered at stations close to populated regions which made data interpretation more difficult at the time (Keeling et al., 1976; Popa et al., 2010). Other background stations followed in the decades after Mauna Loa was set up, such as at Baring Head, New Zealand in 1970 (Brailsford et al., 2012) and the Atmospheric Life Experiment (ALE; a predecessor to the current Advanced Global Atmospheric Gases Experiment; AGAGE) in 1978 (Prinn et al., 2000). Measurements from these background stations only constrained estimations of global or hemispheric scale fluxes within inverse models and were not able to capture local to regional scales (Gloor et al., 2001). Tall tower measurements in conjunction with transport models were proposed as a means to estimate local to regional scale GHG fluxes (Tans, 1993). GHG measurements from tall towers began in the 1990s (Haszpra et al., 2001; Popa et al., 2010) and have been expanded in the 2000s as part of a number of national and international measurement campaigns (Vermeulen, 2007; Kozlova et al., 2008; Thompson et al., 2009; Popa et al., 2010). Measurements made from ground level at terrestrial sites often display complex atmospheric signals with source and sink interactions visible. Sampling from tall towers reduces the influence of these local effects (Gerbig et al., 2003; Gerbig et al., 2009).

For over 30 years, high-frequency measurements of GHGs have been made at Mace Head (MHD), a global background measurement station on the west coast of Ireland. MHD predominantly receives well-mixed air masses, which have travelled across the Northern Atlantic, in the prevailing south-westerly winds, providing a good mid-latitude Northern Hemisphere background signal. These in situ, high-frequency, high-precision measurements have been used to estimate emissions of GHGs from the UK using the Inversion Technique for Emission Modelling (InTEM) methodology (Manning et al., 2011). In 2011, the UK government funded the establishment and integration of three new tall tower measurements stations in the UK. The UK Deriving Emissions linked to Climate Change (UK DECC) network was established to monitor the atmospheric mole fractions of GHGs, improve the spatial and temporal distribution of measurements across the UK and improve GHG emission estimates for comparison with the national inventory, see Manning et al. (2011) for more details. The new network became operational

Deleted: constrain

in 2012. Of the four atmospheric monitoring stations, two main stations (MHD and Tacolneston: TAC) measure a suite of ~ 50 GHGs and ozone depleting substances (ODS; Table 1), while the two other stations (Ridge Hill: RGL; and Angus: TTA) measure the key GHGs. CO₂, CH₄, N₂O, SF₆, CO and H₂ are the main focus of this paper. CO₂ and CH₄ are measured at all stations at high frequency (~ 3 seconds),

Deleted: , except MHD which samples every 20 minutes

whilst N₂O, SF₆, CO and H₂ are measured at a lower frequency (detailed in Sect. 3).

The main objective of this paper is to describe an automated, reliable and high-precision analysis system for routine unattended monitoring of atmospheric CO₂, CH₄, N₂O, SF₆, CO, and H₂ within the UK. We focus on the technical details of the network, review the performance of and present first results from the network.

2 Site location

The location of the three tall tower UK DECC stations was designed to provide good spatial measurement coverage across the UK utilising open lattice tall towers. Good spatial coverage was necessary to provide information on emissions from the UK's devolved administrative regions of Scotland, Wales, England and Northern Ireland. The network consists of four sites all measuring key GHGs (Table 1). Instruments at the Irish coastal site at MHD take whole air samples from 10 meters above ground level (m.a.g.l.), whilst the three UK sites sample from differing heights on tall telecommunications towers (45 – 222 m.a.g.l.). The site locations and descriptions are given in Fig. 1 and Table 2, respectively. Minor instrumental changes have occurred within the network lifetime; however, the described instrumentation at the sites is correct as of September 2015.

2.1 Mace Head (MHD)

The MHD atmospheric research station is one of only a few western European stations that for significant periods of time is representative of mid-latitude Northern Hemispheric background air and provides an essential baseline input for the UK DECC network. At the station (Fig. 1), numerous ambient air measurements are made as part of the AGAGE (Cunnold et al., 1997; Prinn et al., 2000), Integrated Carbon Observation System (ICOS) (Vardag et al., 2014) and the Global Atmospheric Watch (GAW)

networks. Prevailing winds from the west to southwest sector bring well mixed background Atlantic air to the site on average 51 % of the time (Jennings et al., 2003). Polluted European air masses, as well as tropical maritime air masses, cross the site periodically. MHD is uniquely positioned to observe these different air masses. Galway, the closest city, has a population of ~ 75,000 and lies 55 km to the east.

5 The area immediately surrounding MHD is very sparsely populated, providing very low local anthropogenic emissions. The area surrounding MHD is generally wet and boggy with areas of exposed rock (Dimmer et al., 2001). The sample inlet is located 90 m inland from the shoreline (5 meters above sea level; m.a.s.l.) and samples air from 10 m.a.g.l.. CH₄ and N₂O measurements started at MHD on 23 January 1987. CO and H₂ measurements were added on 17 February 1994, and SF₆ measurements were

10 included on 15 November 2003. A fully synoptic weather station operated by Met Eireann is located ~ 300 m from shore at 21 m.a.s.l..

2.2 Ridge Hill (RGL)

RGL is a rural UK site located 30 km east from the border of England and Wales (Fig. 1). It is 16 km south-east of Hereford (population 55,800), and 30 km south-west of Worcester (population 98,800), in

15 Herefordshire, UK (ONS, 2012). The land surrounding the tower is primarily used for agricultural purposes and there are 25 waste water treatment plants within a 40 km radius of the site, the majority of which are in the northeast to south-easterly wind sector (DEFRA, 2012). Air samples are taken from inlet lines located at 45 and 90 m.a.g.l. from a tall open lattice telecommunications tower at 204 m.a.s.l.. N₂O and SF₆ measurements started on 1 March 2012 and are measured from 90 m.a.g.l. only, whilst CO₂ and

20 CH₄ are measured from both heights sequentially and started on 23 February 2012.

2.3 Tacolneston (TAC)

TAC is a rural UK site located towards the east coast of England (Fig. 1). It is 16 km south-west of Norwich (population 200,000), and 28 km east of Thetford (population 20,000), in Norfolk, UK (ONS, 2012). Lines sample air at 54, 100, and 185 m.a.g.l. from a tall open lattice telecommunications tower at

25 56 m.a.s.l.. CO₂ and CH₄ measurements started on 26 July 2012 and are measured from all three heights sequentially, whilst all other GHGs and ODSs (Table 1) are measured from the 100 m.a.g.l. inlet only.

This inlet was chosen as when the site was set up in 2012, the 185 m.a.g.l. inlet had not yet been installed and came on line in January 2013. N₂O, SF₆, CO and H₂ measurements started on 26 July 2012. Land surrounding the tower is primarily used for agriculture, which is dominated by arable farming. Out of a total farmed area of over 400,000 hectares, 79 % of this is used in arable farming (DEFRA, 2010). There are three landfill sites between 30 and 50 km from the site, the closest being 30 km to the east (NCC, 2013). There is also a poultry litter power station in Eye, 20 km south of the site (EPRL, 2013).

2.4 Angus (TTA)

TTA is a rural UK site located near the east coast of Scotland (Fig. 1). It is 10 km north of Dundee (population 148,000; GRO, 2013). A single line samples air at 222 m.a.g.l. from the tall open lattice tower at 400 m.a.s.l., which measures CO₂ and CH₄. Land surrounding the tower is predominantly under agricultural use, primarily livestock farming due to its hilly terrain. A Picarro G2301 was installed on 29 May 2013 and all TTA data reported in this paper is from 29 May 2013 to 30 September 2015 only.

3 Instrumentation

GHG measurement systems were developed in 2011 and then deployed in 2012 to enable measurements of GHGs from telecommunication towers within the UK. The system designs are similar to sampling equipment already deployed at Mace Head (Prinn et al., 2000) and at other tall tower sites (Popa et al., 2010; Winderlich et al., 2010). The systems are designed to utilise easily obtainable parts, so that rapid replacement is possible on component failure, thus minimising system downtime and data gaps. This section outlines the instrumental setup used within the UK DECC network to measure GHGs. Table 1 summarises the trace gas species measured at each of the sites and the instrumentation used. Fig. 2 shows a schematic diagram for TTA, RGL and TAC, whereas the MHD setup is outlined in Prinn et al. (2000).

3.1 Sample tubing

At all UK DECC sites, instrumentation is located at the base of the towers in a building or a modified shipping container. At RGL and TAC, air is sampled through ½" O.D. 'Synflex 1300' or 'Dekabon'

tubing (Hose Tech Ltd., UK), whilst at TTA, it is sampled through 3/8" 'Synflex 3000' tubing (Andrews et al., 2014). Air at MHD is sampled through 1/4" stainless steel tubing (304 stainless steel, 1/4" O.D., 0.209" I.D., Supelco, Sigma-Aldrich, UK). For the number of inlets at each site, please refer to Sect. 2. Tubing is held in place using UV-resistant plastic clips or cable ties and runs down vertical metal tubes

5 on the tower. Horizontal sections of tubing at the base of the tower were kept to a minimum and low points were avoided to prevent the accumulation of water (H₂O). For each inlet at RGL and TAC, an inverted stainless steel cup with a monel mesh screen inserted within the cup covers the inlet, acting as a shield to prevent H₂O entering the line. The mesh screen was removed from the inlet cups at RGL in September 2013 as it was thought that H₂O was accumulating on the mesh and then being sucked into the

10 inlet lines. This effect of the mesh promoting H₂O entering the inlet lines has not been observed at TAC and the mesh is still in place. H₂O decanting bowls with coalescing filter (Norgren, model F74G-4GN-QP3; wetted parts include: aluminium housing, Perspex bowl, sintered 40 µm polypropylene filter and nitrile and neoprene o-rings) are fitted to each sample line at its lowest point at the base of the tower to ensure that liquid H₂O does not enter the laboratory and instrumentation. Perspex H₂O decanting bowls

15 were first tested in the UoB laboratory to check for non-contamination of measurements. The H₂O traps at site are checked on a weekly to monthly basis and emptied manually using a toggle valve at the base of the decanting bowl.

Once the sample lines enter the laboratory, whole air samples pass through an inline 40 µm filter (SS-8TF-40, Swagelok, UK) to trap larger particles and then a 7 µm filter (SS-4F-7, Swagelok, UK) on the

20 branched secondary instrument lines, according to the details in Fig. 2. Filters were not installed on the tower inlets to prevent blockages from ice and subsequent system downtime. Unless stated otherwise, tubing within the laboratories is 1/4" O.D., 0.209" I.D. 304 stainless steel (Supelco, Sigma-Aldrich, UK). The sample line setup at TTA was different to the two other UK sites (RGL and TAC) as this site had previously been managed by the University of Edinburgh before being transferred to the University of

25 Bristol (UoB) from January 2013. Specific differences at TTA include the sample inlet does not have a protective cup covering it (the air sampling line is cut on a bias so H₂O has a drip point); and the H₂O trap at the base of the tower is stainless steel rather than Perspex.

Deleted: This tubing is made from high-density polyethylene bonded to overlapped aluminium tape and has a total wall thickness of 1.57 mm

Deleted: The outer polyethylene coating makes it resistant to water (H₂O) vapour condensation on the inner aluminium core tube.

Deleted:

Deleted: A monel mesh screen is inserted within the cup to help prevent large particles from entering the inlet lines.

Deleted: Perspex

Deleted: se

Deleted: The filters are present to protect instruments and pumps from particles (Andrews et al., 2014).

3.2 Pumps

Each sample line has its own dedicated oil-less linear pump (DBM20-801, GAST Group LTD., UK; TTA: Capex L2, Charles Austin, UK), continuously flushing at a flow rate of $\sim 20 \text{ L min}^{-1}$, located downstream of all sampling equipment. Flow is measured downstream of the sample line pump using flow meters (VFB-68, 3-30 L min^{-1} , Dwyer, UK) and vented into the laboratory. The continuous flushing of the inlet lines results in residence times within the tubing of between 7 and 35 seconds, depending on sampling height, from air intake to the instrumentation. At MHD, air is flushed at a flow rate of 5 L min^{-1} , with instruments sampling at rates between 100 to 110 mL min^{-1} . At TTA the sample line pump has a flow rate of 8.5 L min^{-1} . Branched secondary lines subsample from the main samples lines for all instruments

(Fig. 2).

Cavity ring-down spectrometers (CRDS) subsample from the main sample lines, passing through the sample selection system and a Nafion dryer (described in Sect. 3.3). CRDS instrument pumps (MD1 pump, Vacuubrand GMBH + Co KG, UK) are located downstream of the analyser. This has the advantage of eliminating sample contamination from the pump, reducing the likelihood of a torn diaphragm introducing laboratory air into the sample. The CRDS outlet valve pressure is monitored as a diagnostic for instrument pump failure. The CRDS instrument pump uses Polytetrafluoroethylene (PTFE)/Viton® (also known as FKM) diaphragms. CRDS MD1 instrument pumps are located in the ambient internal laboratory air to allow efficient cooling from the fitted heat-sink.

All gas chromatograph (GC) systems and reduction gas analysers (RGA) use a similar line pump setup (described above), housed within a custom-built GC instrument sample module (Fig. 2). A KNF pump (N86 STE, KNF Neuberger UK Ltd, Oxfordshire, UK) is located downstream of the diaphragm pump, which subsamples at a flow rate of 6 L min^{-1} . A Circor backpressure regulator (GO model LB1-2A01DCE171, Boiswood Ltd, UK) is used to control the KNF pump output pressure, which is viewed using a pressure gauge (0-60 psi Wika, Cole Palmer Instrument Co, UK) and a flowmeter ($0.2\text{-}4 \text{ L min}^{-1}$, VFB-65, Dwyer, UK). This design enables the supply pressure of the air and standard to be matched and prevent pressure artefacts on the columns. Pressure and flow into the sample selection system is monitored

Deleted: and improving the performance of Nafion dryers

by electronic pressure transducers and flow meters (details in section 3.4), meaning that the GC sample module performance can be monitored remotely and failures can be easily diagnosed.

Deleted: -

3.3 Cavity ring-down spectrometer

CO₂ and CH₄ measurements at RGL, TAC and TTA are made using G2301 (Picarro Inc., USA) CRDS analysers (Tremblay et al., 2004; Crosson, 2008). Custom-made sample selection systems made at the UoB are used to switch between air inlets, calibration and standard gases for CRDS analysis, in addition to drying the samples (Fig. 2). The sample selection system consists of a 10 port multi-position valve (EUTACSD10MWEPP, VICI Valco AG International, Switzerland) to direct samples through the Nafion permeation dryer and to the CRDS. All air inlets and calibration gases are plumbed into the multi-position valve for ease of use and sample selection. Automatic sampling is achieved by controlling the multi-position valve using Linux based software (GCWerks™, www.gcwerks.com). An inline 2 µm filter (SS-4F-2, Swagelok, UK) is in place between the outlet of the front end system and the inlet to the analyser (Fig. 2) to further remove any particles that may negatively affect the CRDS.

The CRDS systems measure the decay time of the pulse of laser light inside a 35 cm³ cavity at two wavelengths for ¹²C¹⁶O₂ (1651 nm), ¹²CH₄ (1603 nm) and H₂¹⁶O (1603 nm) (Winderlich et al., 2010). Mole fraction measurements of each gas are provided at a frequency of ~ 3 seconds. Sample flow, temperature (318 ± 0.004 K) and pressure (140 ± 0.05 Torr) are maintained at specific set points as the size and shape of the spectral lines are sensitive to both temperature and pressure. The analyser is thus designed to control temperature to a few thousandths of a K from 10-35°C and sample pressure to 0.05 Torr.

Deleted: hundredths

H₂O can damage system components and interfere with measurements of GHGs, even at low levels, through a dilution effect (Andrews et al., 2014), and pressure broadening effects (Chen et al., 2010; Rella et al., 2013). To minimise these effects, samples were dried to < 0.25 % (< -10.3 °C dew point) using permeation Nafion dryers (MD-050-72S-1, Perma Pure, USA) housed within the sample selection systems. Dried zero air is used as the counter purge at 20 psi, supplied by a compressor (JUN-AIR, model

Deleted: H₂O influences mole fractions of GHGs measured via CRDS through a dilution effect, whereby a difference of 100 µmol mol⁻¹ (or parts per million; ppm) H₂O can cause a “dilution offset” of 0.04 µmol mol⁻¹ in CO₂ (Andrews et al., 2014). Also, H₂O vapour differences between calibration gases and air samples can cause spectral issues within optical instruments. H₂O causes these spectral artefacts through line broadening effects on the spectroscopic line shapes, specifically Lorentzian broadening and Dicke line narrowing (Chen et al., 2010; Rella et al., 2013). The extent of these effects is dependent on the atmospheric mixing ratio of H₂O (Chen et al., 2010).

Deleted: The effectiveness of permeation Nafion dryers is dependent on the H₂O partial pressure gradient between the sample and counter purge flows (Andrews et al., 2014). Therefore, dried

2000, Norgren, Denmark at TAC and RGL, or DK50 plus, Ekom, Slovak Republic at TTA) connected to a zero air generator (TOC-1250, Parker Balston, USA).

In addition to using permeation Nafion dryers, a data correction can also be applied to remove spectral effects caused by H₂O for CRDS systems. All CRDSs in the network produce data corrected for the H₂O effects, using the correction coefficients listed in Rella (2010). The correction applied is minimised due to the removal of most H₂O using the Nafion dryer. TTA air samples were not dried prior to measurement until a Nafion drying system was installed in September 2014.

The CRDS instruments at TAC and RGL were fully operational for > 98 % for the time-period reported here. The TTA CRDS was operational for > 93 %. The inlet at Angus was not shielded from rainwater by an inverted cup, as was the case at other sites, and access to the site was more difficult due to its remote location. As a result, the H₂O trap filled up more frequently resulting in more ambient air data being rejected.

3.4 Gas chromatograph-electron capture detector

N₂O and SF₆ were measured using gas chromatography coupled with micro electron capture detectors (GC-ECD) at RGL and TAC with a similar instrumental set-up (specific setup outlined in Table 3). A simplified schematic diagram for the GC-ECD systems at RGL and TAC is shown in Fig. 2. The GC-ECD at MHD measured N₂O using a different experimental setup, details of which can be found in (Prinn et al., 2000), alongside details of the GC-FID instrument which measures CH₄ at Mace Head.

The SF₆ and N₂O analysis method used at RGL and TAC was similar to that described in detail in Ganesan et al. (2013), except P-5 carrier gas (a mixture of 5 % CH₄ in 95 % Ar; Air Products, UK) is used (Schmidt et al., 2001). Briefly, calibration gas and air samples are flushed through an 8 mL sample loop at 40 mL min⁻¹ for 60 seconds at a fixed exhaust pressure (~ 20 psi; Fig. 3 Backflush) before decaying down to ambient pressure. Flow through the loop is controlled by a 'RED-y' smart series mass flow controller (GSC-A4TA-BB22, Voeglin Instruments AG, Switzerland) and pressure in the loop is measured using an 'All Sensor' pressure sensor (100PSI-A-DO, All Sensors, BS-Rep GmbH, Germany). Once equilibrated to ambient pressure, samples are then injected through an 8-port, 2-position valve (V3 in Fig.

Deleted:

Deleted: . A multi-position valve (EUTACSD8MWEPH, VICI Valco AG International, Switzerland; V1 in Fig. 2) directs calibration gas or sample air through a Nafion dryer (MD-050-72S-1, Perma Pure, USA) with a dry zero air counter purge (as outlined in Sect. 3.3), a second valve (V2 in Fig. 2; 8-port, 2-position valve; EUDAC8UWEPH, VICI Valco AG International, Switzerland) and into an 8 mL sample loop. The loop is flushed

Deleted:), volumetrically flushing the loop ~ 5 times. This second valve (V2 in Fig. 2) is used to control the injection of sample gas onto the columns prior to analysis. The inlet

Deleted: for ambient air and calibration standards are matched to ensure reproducible sample introduction and equal pressure decay time to ambient pressure.

Deleted: The loop is allowed to decay

Deleted: before injection

Deleted: to

2; EUDAC8UWEPH, VICI Valco AG International, Switzerland) onto a pre- (1.0 m Porapak Q, 80/100 mesh, 3/16" O.D.) and main column (2.0 m Porapak Q, 80/100 mesh, 3/16" O.D.) held at 90 °C, where N₂O and SF₆ are separated from air. Oxygen is "heart-cut" to vent (V3, Fig. 3) after it has eluted from the two columns, whilst the pre-column is back flushed with P5. The remaining, O₂ minimised, sample flows through the post-column (0.9 m of 1/8" O.D. stainless steel packed with molecular sieve 5Å, 45/60 mesh), housed in a thermostatically controlled heated inlet port of the GC at 180 °C. The post column reverses the elution order of SF₆ and N₂O to prevent the larger N₂O peak from tailing into the small SF₆ peak improving sample reproducibility and precision. Detection occurs in the ECD which is held at 350 °C. The ECDs at TAC and RGL measure at a rate of 10 and 20 Hz respectively. Samples are dried using a Nafion Dryer (MD-050-72S-1, Perma Pure, USA) with a dry zero air counter purge (as outlined in Sect. 3.3).

The main difference between the method used in our systems and those described by Hall et al. (2011) is the use of P5 carrier gas instead of CO₂ doped N₂. N₂O co-elutes with CO₂ on the post-column, saturating the MS 5Å and providing a constant doping effect and reducing precision. Purity of the P5 carrier gas has previously been an issue where certain cylinders were found to be contaminated with SF₆ in varying amounts (5-80 pmol mol⁻¹, or parts per trillion; ppt). On average, one cylinder in every 6 has been contaminated with SF₆ across the network but is dependent on cylinder age as the gas was used to leak test cylinders. Each cylinder is now individually analysed as a sample to check for contamination prior to use.

The three valves (Valco universally actuated, RS-232 communication, purged housing) are controlled remotely using GCWerks, enabling automatic sampling (see Sect. 3.8).

3.5 Reduction gas analyser

CO and H₂ are measured at two sites, MHD and TAC, using a RGA (RGA3 (MHD) and Peak Performer 1 (TAC), Trace Analytical Inc., USA). Table 4 outlines RGA instrumental setup at TAC and MHD. The MHD RGA setup is different to TAC and is outlined in (Prinn et al., 2000). The TAC sample selection system is integrated within the GC-ECD system (Sect. 3.4) (Grant et al., 2010a; Grant et al., 2010b). The

Deleted:), which controls the sequential injection
Deleted: the
Deleted: -,
Deleted: -

Deleted: post-columns

Deleted: P-5 carrier gas (a mixture of 5 % CH₄ in 95 % Ar; Air Products, UK).

Deleted: The pre- and main columns, 1.0 m and 2.0 m, respectively (3/16" O.D. stainless steel packed with 80/100 mesh, Porapak Q), are contained in the main GC oven (Agilent GC-7890 at RGL and GC-6890 at TAC) at 90 °C, the post-column (0.9 m of 1/8" O.D. stainless steel packed with molecular sieve 5Å, 45/60 mesh) is housed in a thermostatically controlled heated inlet port of the GC at 180 °C. Separation of N₂O and SF₆ from air occurs on the pre- and main columns after which a "heart-cut" is taken from the flow to remove the majority of O₂ from the sample (Fig. 3). The O₂ is vented by switching V3 (Fig. 3) to backflush the pre-column with P5. This prevents the build-up of O₂ which may result in increased column bleed into the ECD resulting in increased baseline noise and reduced ECD sensitivity. The remaining, O₂ minimised, sample flows from the main to post-column where N₂O and SF₆ are separated further. The post column reverses the elution order of SF₆ and N₂O, preferentially retaining N₂O as the 5Å pore size is too small for the SF₆ to physically enter. SF₆ molecules travel around the bulk molecular size with its retention time determined by the mesh size (Moore et al., 2003). This order of separation prevents the larger N₂O peak from tailing into the small SF₆ peak, improving sample reproducibility and precision. Detection occurs in the micro-ECD which is held at 350 °C. The micro-ECDs at TAC and RGL measure at a rate of 10 and 20 Hz respectively.

Deleted: The use of P5 carrier gas enables the omission of CO₂ doping.

Deleted: combination used within the UK DECC network

Deleted: , thus care is taken to make sure the N₂O response is not affected by this

Deleted: -

Deleted: The

GC-ECD has a 10-port, 2-position valve (VICI Valco AG International, Switzerland) for V2 (Fig. 2), instead of an 8-port 2-position valve, as at RGL. This allows for a 1 mL RGA sample loop to be put in sequence before the ECD sample loop (Fig. 3 TAC). After samples have been dried using the Nafion Dryer (MD-050-72S-1, Perma Pure, USA), passed through the sample loops and decayed to ambient pressure, they are injected onto two isothermal packed columns held at 105 °C: a 0.768 m pre-column (1/8" O.D. stainless steel packed with 60/80 mesh Unibeads 1S) and a 0.768 m main column (1/8" O.D. stainless steel packed with MS 5Å, 60/80 mesh). After separation, gases are injected into the RGA for analysis using zero air plus (Air Products, UK) carrier gas, where the samples pass over a heated bed of mercuric oxide before being quantitatively determined using UV photometry (Grant et al., 2010a; Grant et al., 2010b).

Deleted: front-end system at MHD and TAC have

Deleted: is used to protect the 0.768 m main column (1/8" O.D. stainless steel packed with MS 5Å, 60/80 mesh) from contamination by gases which adsorb to the surface of the main column and the main column separates H₂ and CO (Grant et al., 2010a).

3.6 Medusa gas chromatograph-mass spectrometer

The Medusa is a custom-built pre-concentration unit coupled to a gas chromatograph-mass spectrometer (GC-MS, the entire system is hereafter referred to as a Medusa GC-MS), which measures a wide range of GHGs and ODSs. The Medusa GC-MS system is used at both MHD and TAC to measure SF₆, amongst other compounds. A detailed description of the Medusa setup is presented in Miller et al. (2008) for TAC and Arnold et al. (2012) for the NF₃ conversion (MHD setup). Briefly, a 2 L whole air sample is collected by the Medusa pre-concentration unit from the same sample pump as the GC-ECD (outlined in Sect. 3.6; Fig. 2) wherein the sample is dried using two Nafion dryers (MD-050-72S-1, Perma Pure, USA) before being sequentially passed through two adsorbent traps cooled to -165 °C using a Cryotiger (Brooks Automation, Massachusetts, USA). More abundant gases (e.g. N₂, O₂, CO₂ and CH₄) are removed using temperature programming of the traps, allowing the trace species of interest to be isolated on the second refocusing trap after thermal desorption from the first trap. Trace gas species adsorbed on the second trap are released by heating the trap to 100 °C and passed through two columns (three columns for NF₃ method at MHD) temperature programmed between 40 and 200 °C (Agilent 6890 GC, Agilent Technologies, UK) using helium carrier gas (MHD: BIP grade, Air Products, UK; TAC: 6.0 grade, BOC, UK), separating

out trace species chromatographically. Analytes are then detected via a quadrupole mass selective detector (Agilent 5973, Agilent Technologies, UK) in Selected Ion Monitoring (SIM) mode to increase sensitivity.

3.7 Logging, control and ancillary equipment

All instruments within the UK DECC network are controlled by GCWerks, installed on a local site computer running Ubuntu 12.04 LTS. GCWerks automates all instrument parameters (valves, trap and column temperatures, MSD, etc.), regulates switching processes, controls calibration cycles, displays chromatograms, performs peak integration and gives graphical and tabulated displays of all results. The automation of all instrumental processes helps to reduce problems and data loss associated with connection problems between independent sample modules to instruments.

GCWerks generates automated user-specified alarms when instrument parameter conditions are not met. These alarms can also initiate instrument shutdown when specified to prevent instrumental damage.

The local site computer is connected to CRDS and GC analysers via Ethernet and the sample selection systems communicate through serial (RS232) connections. Each site has a broadband internet connection which is utilised for remote access and control, automated data backup and maintaining system time synchronisation for each computer using the network time protocol. Data from instrumentation and ancillary equipment is logged and archived at all sites at a frequency of 0.3-20 Hz.

Uninterruptible power systems (UPS) are used at MHD (SG5K-6K, Falcon Electric Inc., USA), RGL and TAC (Sentinel Dual SDL8000, Aiello UPS Ltd., UK) to prevent power surges and temporary power outages affecting instrumentation. The UPS provides up to 20 minutes of power to instrumentation in the event of a power outage. Additionally, an onsite generator provides continuous backup power at MHD with the UPS providing power for long enough to enable a seamless transition of power from line to generator.

3.8 Maintenance

Maintenance schedules of each of the sites varies greatly depending on instrumentation; sites with Medusa GC-MSs (TAC and MHD) were visited more frequently due to the instrumentation being complicated and having greater maintenance needs and sites with only CRDSs were visited the least. Table 5 outlines

Formatted: Heading 2

routine site visits, as well as emergency visits when issues with the instrumentation arose that couldn't be rectified remotely. Scheduled site visits included checking calibration cylinder pressures and line pump flow rates, changing of carrier gases, updating software on instruments and computers at site, emptying water decanting bowls of any liquid water, and changing line and equipment (compressor and zero air generator) filters.

4 Sampling and calibration

4.1 Sampling sequence

Sampling sequences within the network varies between instruments. CRDS instruments within the network are continuously measuring, with RGL and TAC measuring each sampling height sequentially for 30 and 20 minutes, respectively, to ensure each sampling height is measured within each hour. The CRDS at TTA measures continuously from the single 222 m inlet. To ensure a good stabilisation period when sampling between different heights, the first 2 minutes of data after the valve switches to a new sample intake is automatically flagged out. The air sampling sequence is interrupted to analyse a daily standard gas and a monthly calibration sequence, outlined in Sect. 4.2.1.

The GC-ECD, RGA and Medusa GC-MS all have a lower sampling frequency than CRDSs and therefore only sample from one inlet. Sampling frequencies within the network are 10 minutes for the GC-ECD at RGL and TAC, 20 minutes for GC-FID and GC-ECD at MHD, 10 and 20 minutes for the RGA at TAC and MHD respectively, and 65 minutes for the Medusa GC-MS. Measurements alternate between ambient air and calibration gas, as outlined in Sect. 4.2.2.

4.2 Calibration

To guarantee the reliability and stability of measurements, automated calibrations are carried out periodically. Two separate calibration schemes are used, one for the CRDS and another for all other instruments. All tubing used for calibrant gases are 1/16" O.D., 0.03" I.D. 304 stainless steel (Supelco, Sigma-Aldrich, UK) to minimise dead volumes and wasted gas.

4.2.1 Cavity ring-down spectrometer calibration

CRDS instruments have two types of calibration standards, a standard of approximately ambient mole fraction and a set of calibration standards that span from below ambient up to elevated mole fractions. High-pressure aluminium tanks (Luxfer Gas Cylinders, UK) are used rather than steel to ensure long-term stability of CO₂ in the calibration gases. Regulator components may also have effects on the stability of calibrations gases (Winderlich et al., 2010). Within the UK DECC network, regulators (64-2640KA411, Tescom Europe) with polychlorotrifluoroethylene (PCTFE) seals were used to prevent gas permeation. Calibrant and standard gases used in the CRDS instruments at all sites were filled and calibrated at GasLab MPI-BGC Jena and are of natural composition. CO₂ is on the WMO-X2007 scale (Zhao and Tans, 2006) and CH₄ is reported on the WMO-X2004A scale (Dlugokencky et al., 2005). Standard gases have a working lifetime of between 2 to 3 years, whilst calibration gases last for approximately 5 years. Once cylinders are removed from site, they are sent for recalibration.

The standard gas is measured once a day for 20 minutes to assess for linear instrumental drift and the suite of calibration gases with varying mole fractions are measured once a month in quintuplicate to assess for instrument non-linearity and non-linear drift. The first five minutes of standard and calibration data and the first entire suite of calibration runs are removed to compensate for variability caused by regulator and line flushing. Table 6 details the standard and calibrant CO₂ and CH₄ mole fractions currently used at RGL, TAC, and TTA. Linear interpolation between each daily standard gas analysis is used to remove instrumental drift and is performed automatically by GCWerks (see Sect. 5.1).

Instrument non-linearity is assessed on a monthly basis by manually viewing the curve of the calibration gases and adjusted accordingly if there is a difference between the previous and current curve coefficients. Instrument nonlinearity is also reassessed after changes in instrumental soft- and hardware. Despite other studies showing CRDSs to be linear (Yver Kwok et al., 2015; Hazan et al., 2016) over the concentration ranges used in the network, all instruments were found to have a small non-linear response (example shown in Fig. 4). Therefore, a second order non-linear curve is fit to the data and implemented in GCWerks manually (see Sect. 5.1). There is a possibility that this is an artefact from the calibration cylinders; however, the non-linearity effect has also been seen when NOAA (National Oceanic and

Deleted: 5

Deleted: significant change

Deleted: in

Deleted: A

Atmospheric Administration) cylinders have been used to calibrate instruments. There is a small difference (median \pm s.d.) between data corrected only for the linear instrumental drift and the non-linear corrected data (-0.012 ± 0.007 , -0.002 ± 0.01 and $0.07 \pm 0.03 \mu\text{mol mol}^{-1} \text{CO}_2$ for RGL, TTA and TAC, respectively, and a -0.002 ± 0.001 , -0.17 ± 0.28 and $0.19 \pm 0.3 \text{ nmol mol}^{-1}$ - or parts per billion, ppb - CH_4).

The long-term repeatability of daily of daily standard measurements (standard deviation, 1σ , from January 2012 to September 2015) on CRDS instruments within the UK DECC network of $< 0.03 \mu\text{mol mol}^{-1}$, and $< 0.3 \text{ nmol mol}^{-1}$ for CO_2 and CH_4 , respectively. Short-term precision (1σ of 1-minute means) was $< 0.05 \mu\text{mol mol}^{-1}$ and $< 0.3 \text{ nmol mol}^{-1}$.

4.2.2 Gas chromatograph calibration

The MHD GC-ECD, GC-FID, RGA, and Medusa GC-MS instruments and the TAC Medusa GC-MS are calibrated using tertiary standards. Working standards (also known as quaternary standards) are used to calibrate the Medusa GC-MS systems within the network and the GC-ECD and RGA at TAC and RGL. Tertiary and quaternary standards are prepared by compressing background ambient air into 34 L electropolished stainless steel cylinders (Essex Cryogenics, Missouri, USA) using a modified oil-free compressor (SA-3, RIX California, USA). Tertiary standards are filled at La Jolla, California, USA and calibrated at Scripps Institution of Oceanography (SIO) against their primary calibration scales via secondary working standards before being sent to MHD or TAC. Tertiary standards are also re-calibrated on return from site to assess each standard for sample stability over its working lifetime. Quaternary standards are filled at MHD and are calibrated/re-calibrated against the SIO calibrated tertiary standards at MHD on the GC-ECD and RGA before and after use at the tall towers. Mole fractions within the tertiary and quaternary standards are close to ambient background air sample values, minimising possible sample matrix non-linearities.

The quaternary standards are used to bracket air measurements on the GC-ECD, GC-FID, RGA and Medusa GC-MS. Tertiary standards are used to bracket air measurements on the MHD GC-ECD/FID/RGA. In addition, for the Medusa GC-MS, tertiary standards are analysed weekly and are used

Deleted: Measurements of daily standard gas showed a
Deleted: precision
Deleted: of 1 minute means
Deleted: of
Deleted: 2
Deleted: (or parts per billion; ppb)
Deleted: CRDS precision within the UK DECC network is within the WMO compatibility guidelines for CO_2 ($\pm 0.1 \mu\text{mol mol}^{-1}$) and CH_4 ($\pm 2 \text{ nmol mol}^{-1}$) (WMO-GAW, 2014).

- to calibrate the quaternary standards over the course of their use in the field. Quaternary standards last for two months to two years, depending on which instrument they are being used to calibrate, and tertiary standards last approximately eight months to two years. Studies have shown that no significant drift of species contained in these standards occur over this time period (Hall et al., 2007; 2011). Calibration
- 5 scales vary depending on the gas species, with N₂O on SIO-98, SF₆ on SIO-05, CH₄ on Tohoku University, H₂ on MPI-2009, and CO on the CSIRO04 calibration scales. A concentration difference of $1.01 \pm 4.14 \text{ nmol mol}^{-1}$ CH₄ ($0.6 \pm 0.22 \%$) between the WMO-x2004A and Tohoku University scales has been observed at MHD based on NOAA Earth System Research Laboratory Global GHG Reference Network flask sampling and in situ measurements (Krummel, 2018).
- 10 Repeatability of 20 minute injections of tertiary/quaternary standards (1σ) between January 2012 and September 2015 was $< 2.8 \text{ nmol mol}^{-1}$ for GC-FID measurements of CH₄ (only at MHD), $< 0.4 \text{ nmol mol}^{-1}$ for N₂O (GC-ECD), $< 0.07 \text{ pmol mol}^{-1}$ for GC-ECD measurements of SF₆ and $< 0.34 \text{ pmol mol}^{-1}$ for Medusa GC-MS SF₆ measurements, $< 2 \text{ nmol mol}^{-1}$ for CO (RGA) and $< 3 \text{ nmol mol}^{-1}$ for H₂ (RGA). Due to the non-linear response of the GC-ECD to N₂O, non-linearity testing was carried out
- 15 approximately yearly in the field. Non-linearity testing was undertaken using a high mole fraction reference gas (20 $\mu\text{mol mol}^{-1}$ N₂O and 1 nmol mol^{-1} SF₆ gas mix, BOC, Surrey, UK), which was dynamically diluted with zero air (Zero Air Plus, Air Liquide, Cheshire, UK) to the range of atmospheric mole fractions (N₂O: 240 – 400 nmol mol^{-1} ; SF₆: 6 – 14 pmol mol^{-1}) using a custom-made dynamic dilution unit made up of two RED-y mass flow controllers (GSC-A3TA-BB21, 100 mL min^{-1} ; GSC-
- 20 A4TA-BB22, 200 mL min^{-1} ; Vögtlin Instruments AG, Switzerland). During the nonlinearity test, the zero air cylinder was analysed to check for traces of N₂O and SF₆. Cylinders with detectable contamination were not used. Results were used to create a second order non-linearity curve and a correction was implemented in GCWerks (see Sect. 5.2).

Deleted: CO measurements are within the WMO compatibility guidelines of $\pm 2 \text{ nmol mol}^{-1}$.

Deleted: periodically

5 Data processing

- 25 GCWerksTM is used to process all of the CRDS, RGA and GC data. Raw measurement data and ancillary parameters stored on the local site computers are processed on site in near real time (NRT) for calibration

and H₂O corrections. Processing of data on site has the added advantage of aiding troubleshooting of instruments for site technicians.

Raw and processed data are mirrored daily from the local site computer to data processing servers at the UoB or at the University of East Anglia (UEA) for TAC GC-ECD/RGA and Medusa GC-MS. Post
5 processed UEA data is also mirrored to the UoB servers for archiving. All raw and processed data (calibrated and H₂O corrected; Sect. 5.1-5.2) are subjected to QA/QC (Sect. 5.3), ensuring comparison with physical instrument parameters, such as CRDS cavity temperature and pressure, or flow rates, to check for spurious data. Data corrections outlined in Sect. 5.1 – 5.2 are investigated and implemented on the processing servers and then mirrored back to the sites.

10 5.1 Algorithms for calculating CO₂ and CH₄

Raw CRDS data and ancillary parameters are acquired by GCWerks and are stored in binary strip-charts. These strip-charts contain all relevant data from the CRDS, i.e. wet and dry mole fractions, H₂O mole fractions, cavity temperature and pressure. Metadata stored within log files in GCWerks describes each sample type (air, std, cal), inlet height, how much data is to be rejected and data averaging frequencies
15 for each Valco valve port within the sample selection system. Data within the network is averaged over 1, 5, 20 and 60-minute intervals. The first 2 minutes of air and 5 minutes of standard/calibration data after the Valco selector valve switches within the sample system are automatically rejected to allow for stabilisation time and the tubing to condition with the new sample.

A number of data filters are automatically applied to the CRDS data before means are calculated. Data
20 filters include cavity pressure and temperature out of normal operating range, H₂O level too high, cycle time too slow and the standard deviation in sample values being too great (Table 7). Parameterisation of filters is generic for the type of analyser but can be user defined within GCWerks. For each single data point, these filter parameter values are verified and the data point discarded from the final dataset if not. CRDS measurements are then corrected for linear instrumental drift and instrumental response over a
25 span of different mole fractions, referred to here as non-linearity. As in Verhulst et al. (2017), linear instrumental drift, monitored by repeated measurements of a calibrated standard gas measured daily for

Deleted: , named by UTC date, time, sample type (air, std, cal, tank) and inlet port number

Deleted: , with time stamps corresponding to the beginning of the measurement, and stored

Deleted: 6

Deleted: Linear

20 minutes, is corrected for by ratio of a measurement to the linear interpolation between bracketing standard measurements, as outlined in Eq. (1). ~~However, unlike in~~ Verhulst et al. (2017), ~~instrumental~~ non-linearity is assessed ~~and implemented~~ using a function of the sample/standard ratio, outlined in Eq. (2). A second order function can then be fitted to the data to provide a non-linearity correction in Eq. (1).

$$5 \quad C_{smp} = \frac{\left(\frac{R_{smp}}{R_{std}} C_{std}\right)}{NonLin} \quad (1)$$

$$NonLin = f\left(\frac{R_{cal}}{R_{std}}\right) = \frac{\left[\frac{R_{cal}}{C_{cal}}\right]}{\left[\frac{R_{std}}{C_{std}}\right]} \quad (2)$$

where C_{smp} is the calibrated CO₂ or CH₄ mole fraction, R_{smp} and R_{cal} are the sample and calibrant raw dry-air mole fraction from the CRDS, respectively, R_{std} is the linear interpolation between the raw dry-air mole fraction of the two bracketing standards, C_{std} and C_{cal} are the calibrated mole fraction assigned at GASLAB MPI-BGC Jena, and $NonLin$ is the non-linearity correction coefficient assigned by the user from the second order fit of calibration data.

5.2 Algorithms for calculating N₂O, SF₆, CH₄ (from GC-FID), CO and H₂

GC data and ancillary parameters are acquired by GCWerks and stored in chromatograms and strip-charts, ~~as well as being~~ displayed in real time. Temperature (ambient and sample selection module), loop flow

15 rates and pressures at the time of sample injection onto the columns are also stored in a sample log file with the corresponding date and time.

User defined integration parameters allow for automatic integration of peaks. Chromatograms can be reprocessed for selected periods when peak integration parameters need to be altered due to changes in baseline and retention times. Integrated peak heights and areas are stored and used along with pressure and temperature data stored in the sample log file to calculate mixing ratios.

20 N₂O, SF₆, CH₄, CO and H₂ are calibrated for linear instrumental drift and non-linearity in a similar way to measurements from the CRDS instrument. A variation of Eq. (1) is used to calibrate data for linear instrumental drift using sample integrated height (SF₆, CH₄, CO and H₂) or area (N₂O; Eq. (3)). The non-

Deleted: Instrumental

Deleted: , named by UTC date, time, sample type (air, std, cal, tank) and inlet port number. GC chromatograms are acquired

Deleted: and

Deleted: , and are stored with a 4:1 compression ratio

linearity fit is defined using the dynamic dilution of a high concentration cylinder, as described in sect. 4.2.2 and implemented using Eq. (4).

$$C_{smp} = \frac{RL \cdot C_{std}}{NonLin} \quad (3)$$

$$RL = \frac{R_{smp}}{R_{std}} \cdot \frac{\frac{P_{smp} \cdot V}{R \cdot T_{smp}}}{\frac{P_{std} \cdot V}{R \cdot T_{std}}} \quad (4)$$

- 5 where C_{smp} is the calibrated N_2O , SF_6 , CO and H_2 mole fraction, RL is the sample/standard ratio, R_{smp} is the sample raw dry mole fraction, R_{std} is the linear interpolation between the raw dry-air mole fraction of the two bracketing standards, C_{std} is the calibrated standard mole fraction, $NonLin$ is the nonlinearity correction coefficient assigned by the user from the second order fit of calibration data, P_{smp} and P_{std} are the sample and standard loop pressures, respectively, at the time of sample injection, T_{smp} and T_{std} are the sample and standard gas temperatures, respectively, at the time of injection, V is the loop volume and R is a gas constant.

5.3 Final data processing

In the first phase, chromatograms and strip-charts are reviewed daily, on a site-by-site basis to check for good integration and systematic biases not detected by automatic data processing routines. Filtered data

- 15 is also reviewed in the strip-charts, shown in conjunction with the unfiltered data, to ensure good filter parameterisation, ensure non-spurious data isn't unnecessarily filtered out and help diagnose instrumental issues. Data and ancillary measurements are reviewed in parallel to help observe potential errors and diagnose issues within the data. Instrument precision is reviewed by monitoring the standard gas concentrations for anomalies. 1, 5, 20 and 60-minute mean (CRDS) or discrete (GC-ECD/FID/RGA and Medusa GC-MS) air data can also be plotted against instrumental and ancillary parameters to further investigate data issues. Spurious data are manually flagged and a justification for the flagging of data is given and logged.

In the second phase, data from the entire network are imported and reviewed simultaneously in GCcompare, custom built data visualisation software for time series data from multiple sites. Flagged

- 25 GCWerks time series data from the network are overlain to compare sites with the background station

Deleted: can

Deleted: be

Deleted: and

Deleted: .

(MHD) and to look for differences between sites for each compound measured in the network. Potential issues not previously noted are investigated using ancillary and instrumental parameters, which are also imported into GCcompare, as well as air history maps produced on an hourly basis using the Numerical Atmospheric dispersion Modelling Environment (NAME) Lagrangian dispersion model outlined in Manning et al. (2011).

5.4 Troubleshooting data issues

Collection, storage and visualisation of ancillary data in GCWerks in parallel with mole fraction data has made troubleshooting data issues easier. When potential issues are observed in the time series data, site operators check the ancillary data recorded to try and help identify potential issues.

Within the network, one of the greatest issues observed so far is laboratory temperature stability, which can affect the performance of the instrumentation. At sites with and without air conditioning, rapid fluctuations in temperature have resulted in poorer precision in the data, as observed in the ambient temperature data recorded on the GC-MDs. In laboratories with air conditioning (TAC and MHD), economy modes have often been used to reduce the frequency of the unit being switched on and off, thus smoothing the temperature swings in the laboratory. At RGL, the ventilation system is either set to being constantly on in the summer or off during the winter months to smooth temperature fluctuations in the laboratory. A number of other issues observed within the network are shown in Table 8.

6 Results

Measurements of GHGs from the UK DECC network are presented from January 2012 through to September 2015 (Fig. 6 to 8; see Sect. 2 for details of start dates of data acquisition). Results shown in this paper are limited to qualitative analyses of the most prominent features of the data; utilisation of this large and comprehensive dataset to its full potential lies in the use of high resolution inverse atmospheric transport models (Manning et al., 2011; Vermeulen et al., 2011; Ganesan et al., 2015). All CO₂ and CH₄ data are publicly available as hourly means, whilst N₂O, SF₆, CO and H₂ are available as discrete samples, at EBAS, as database infrastructure operated by the Norwegian Institute for Air Research

Formatted: Heading 2

Deleted: .

Deleted: 4

Deleted: 6

(<http://ebas.nilu.no/>) and the World Data Centre for Greenhouse Gases (<http://ds.data.jma.go.jp/gmd/wdcgg/>). All MHD data, except CO mole fractions, are available from the Carbon Dioxide Information Analysis Centre (CDIAC) at <http://cdiac.ornl.gov/>.

6.1 Seasonal cycles

- 5 CO₂ shows the most marked seasonal cycle of all the GHGs measured in the UK DECC network, due to its major biogenic uptake via photosynthesis and production from respiration, as well as anthropogenic sources. The approximate amplitude for mid-latitude northern hemisphere seasonal cycle is 15 $\mu\text{mol mol}^{-1}$ and has an upwards trend. Fig. 6(a) shows CO₂ maxima in January/February and minima in August. Sites show a CO₂ signal that varies in a “noise” band of approximately 20 $\mu\text{mol mol}^{-1}$ (~ 5 %), alongside
- 10 a strong seasonal cycle. Large differences between the sites can also be observed from CO₂ data in Fig. 6(a). TTA shows the lowest frequency and magnitude of above baseline events. This is thought to be a combination of the tower inlet height, which is at 222 m.a.g.l. compared to lower inlets at RGL (45/90 m.a.g.l.) and TAC (54/100/185 m.a.g.l.), and the location of TTA in the north of the UK in a much more sparsely populated region than the other sites. TAC CO₂ mole fractions have greater excursions from
- 15 baseline compared with the other sites due to its eastern location, downstream of the predominant south-westerly wind direction, and location near to a number of large sources.
- CH₄ also shows seasonal variation, see Fig. 6(b), with a winter maxima and a summer minima, driven by greater oxidation by hydroxyl radicals in strong sunlight and greater uptake from the troposphere by methanogenic bacteria in soils (Dlugokencky et al., 2011). The approximate amplitude for mid-latitude
- 20 northern hemisphere seasonal cycle at MHD is 21 nmol mol^{-1} and has an upwards trend. Differences between site baseline (unpolluted) mole fractions can be observed in Fig. 6(b). The relative variability in CH₄ atmospheric signal within the UK DECC network varies roughly between 1800 – 2300 nmol mol^{-1} . MHD and TTA generally agree well over the observation period and have the lowest frequency and magnitude of pollution events. This is thought to be due to fewer pollution sources within the prevailing
- 25 wind direction at both MHD and TTA. As per CO₂, TAC has the greatest excursions in CH₄ mole fraction

Deleted: 4

Deleted: 4

Deleted: 4

Deleted: 4

compared with the other sites in the network due to its location downwind of major urban areas such as London and Birmingham in prevailing wind directions.

A seasonal cycle is also observed in N₂O mole fractions, see Fig. 7(a); however, this seasonality is less well defined than for CO₂, especially in 2014. There is an approximate 0.8 nmol mol⁻¹ amplitude in the northern hemisphere mid-latitude seasonal trend at MHD. Summer minima in N₂O are thought to be caused by the descent of stratospheric air bringing N₂O-depleted air into the troposphere across the polar tropopause (Nevison et al., 2011). The atmospheric signal has an upward trend and varies between 322 – 338 nmol mol⁻¹. Like CO₂ and CH₄, magnitude and frequency of N₂O pollution events are greater at RGL and TAC than MHD. This is thought to be due to the surrounding land being used predominately for agriculture and fertiliser application causing nitrification and denitrification to occur.

Tropospheric SF₆ mole fractions do not show a seasonal cycle but a clear increase over time, see Fig. 7(b). The atmospheric variability within the network is between 7.5 – 13 pmol mol⁻¹. The magnitude and frequency of SF₆ pollution events at RGL and TAC are greater than at MHD as emissions are predominantly from anthropogenic sources. Natural sources of SF₆ are considered to be so low they can be ignored (Levin et al., 2010).

A seasonal cycle is also observed in CO and H₂ at MHD and TAC (Fig. 8). There is an approximate 37 and 36 nmol mol⁻¹, for CO and H₂ respectively, northern hemisphere mid-latitude seasonal trend. CO has winter maxima and summer minima, driven predominantly by anthropogenic emissions and the strength of the summertime OH sink (Grant et al., 2010b; Satar et al., 2016). H₂ has delayed spring maxima and autumn minima due to maximum summertime loss by OH oxidation and greatest rates of soil uptake in the summer and early autumn when soils are driest (Grant et al., 2010b).

6.2 Diurnal cycles and vertical gradients

In principal, measurements of GHGs at different heights at a station allow observations of sources and sinks from different spatial footprints (Vermeulen et al., 2011). The average diurnal profiles for CO₂ and CH₄ at RGL and TAC (Fig. 9) are from 1 January 2012 to 30 September 2015.

Deleted: 5

Deleted: 5

Deleted: 6

Deleted: The influence of local source of GHGs is greatest at night and the early morning, as observed in Fig. 7, with higher mixing ratios at lower inlet heights caused by the low night-time planetary boundary layer heights.

Deleted: 7

Deleted: The summer data shows a strong diurnal cycle with CO₂ maxima at night-time when respiration (emissions) dominates and CO₂ minima at mid-day when photosynthesis (uptake) dominates. Night-time CO₂ is greatest during the winter and spring months, with maxima around 06:00 UTC. Winter months have a smaller diurnal cycle than other seasons and a delayed uptake of CO₂ (approximately 09:00 UTC) in comparison to summertime uptake (06:00 UTC), caused by reduced rates of biosphere assimilation and later sunrises.

The greatest difference in CO₂ mole fraction between the lowest and highest inlets were 3 and 8 μmol mol⁻¹ at RGL and TAC respectively. Daytime vertical differences were very small for all seasons (< 1 μmol mol⁻¹) for both TAC and RGL. Spring, summer and autumn daytime concentrations are generally lower at the lowest inlet heights for RGL and TAC due to the net biospheric CO₂ uptake within the footprint area. CO₂ uptake during the daytime was also observed during winter months, although this is less pronounced than other seasons and wintertime average daytime CO₂ concentrations are always greater at the lowest inlet height.

Diurnal variation in CH₄ show similar patterns to CO₂, with early morning maxima and early afternoon minima; however, the daytime increase in mole fraction with height as in Fig. 9(a)&(b) are not observed in CH₄ (Fig. 9(c)&(d)) as mixing ratios are not dominated by biospheric photosynthesis. Average summer time CH₄ concentrations are ~ 20 nmol mol⁻¹ less than other seasons due to greater oxidation by hydroxyl radicals. In winter, vertical CH₄ concentration gradients are maintained throughout the day and night due to the persistence of a low planetary boundary layer during daytime. Maximum gradients between the lowest and highest inlets show variation of ~ 10 and 20 nmol mol⁻¹ for RGL and TAC respectively, a similar percentage difference to CO₂ concentration gradients.

Deleted: . The highest mole fractions were observed at the lowest inlet heights, due to local net CO₂ emission in the area of the inlet footprint during night-time and a lower boundary layer heights

Deleted: These seasonal patterns in vertical concentration gradients can be explained by the diurnal evolution of the planetary boundary layer. Usually gases accumulate at night when the planetary boundary layer is low due to cooler conditions, they then get diluted in greater volumes of the atmosphere within a well-mixed convective boundary layer that develops during daylight hours.

Deleted: 7

Deleted: 7

Deleted: , caused by cooler atmospheric temperatures

6.3 Discerning pollution events

Air history maps, showing the previous 30 days of surface influence at the station in a 1-hour period, were produced using the Met Office NAME Lagrangian atmospheric dispersion model (Jones et al., 2007) for each of the sites within the UK DECC network in order to discern and explain pollution signals in the mole fraction measurements. Increasing mole fractions with longitude across the UK from the baseline station (MHD) are frequently seen within the data, as demonstrated in Fig 10(a) on 5th December 2014. Fig. 10(b) also demonstrates a regionally polluted period at RGL for CH₄ on 30/11/2014, where air has passed over Europe and the south of the UK before arriving at the site. Fig. 10(c) shows an example of baseline conditions for CH₄ on 01/12/2014, where air passes over the North Atlantic Ocean, resulting in low mole fractions at MHD and variation between the other UK DECC sites.

Formatted: Superscript

Deleted: 8

Deleted: 8

7 Recommendations

Many lessons have been learnt with setting up and running the UK DECC network and we have tried to summarise the main points for future stations or networks.

7.1 Instrumentation modularity and automation

Formatted: Heading 2

Monitoring stations are often located in remote areas that are not easily accessible. The need for designing instrumentation that can be fully automated and controlled on site with minimal on site human attention is extremely important. Additionally, it is crucial that the software used to control the instrumentation can be accessed remotely to make changes to sampling regimes when issues arise. As suggested by Andrews et al. (2014), modularity in the analytical systems helps simplify maintenance and repairs. We aim to have spare modules, such as line pump modules, based in the UoB laboratory that can be sent or taken to site as soon as there is a sign of an imminent problem. This is not always feasible for larger and more expensive items, such as instrument boards; however, these items do not fail as frequently as line pumps or inlet filters.

7.2 Software

Formatted: Heading 2

A number of software packages are now commercially available and are able to control instruments and log data. However, there are fewer packages available that are able to control instruments, log data and visualise data rapidly. Being able to visualise all data, including ancillary data and even after the data has been post-processed, at site has the added advantage of being able to look back through the time series for when an issue may have previously happened and then check the operations log to see how the problem was rectified. This is especially important when a number of site operators make visits. Comprehensive measurement and logging of critical pressures, temperatures and flow rates are necessary for detecting instrumental problems. Automated alarm emails based on the data can notify site operators of failures and help to reduce instrument downtime. Prior to the alarms being integrated within the network, problems with the data and instruments went unnoticed for several days. Customisation of alarm parameters can also help to reduce false alarms.

7.3 Calibration

High precision data requires frequent field calibration, even for modern CO₂ and CH₄ spectrometers that are extremely stable. Any calibration gases used within a network should be traceable to a scale to ensure that measurements are comparable between sites. We recommend using scales that are already well defined and widely used, such as the WMO scales (WMO-GAW, 2016). There are a number of different ways to calibrate data (Andrews et al., 2014; Hazan et al., 2016; Verhulst et al., 2017); however, we recommend using a method that calibrates out short-term instrumental drift as this is not fully captured when linearly interpolating between monthly calibrations and can introduce error into data.

It is recommended to have more than the minimum number of calibration cylinders required to generate a calibration curve depending on the instrument needs; however, space and financial constraints can reduce the number of cylinders available. As a guide for CRDS instruments, we have a minimum of three calibration cylinders and one standard cylinder per analyser. We also recommend having spare standard and calibration cylinders, which can be kept off site, in case cylinders need urgently replacing following failures. Standard cylinders should be at ambient mole fractions and calibration cylinders should span the expected ambient range. Standards and calibrants should have the same matrix, as the sample air and a similar isotopic composition to ambient air.

7.4 Carrier gas purity

On a number of occasions, P5 and He carrier gases have been contaminated with SF₆ due to being pressure tested at the end of their inspection life cycle. We recommend testing all carrier gases as an unknown sample on instruments to double check for contamination of gases being measured. Before testing cylinders within the network for contamination, contaminated carrier gases were connected to GCs and resulted in increasing baselines and sample signal suppression.

8 Summary and conclusions

The UK DECC network was established in January 2012 to monitor atmospheric GHG and ODS mole fractions and verify the UK emission inventories submitted to the UNFCCC. The network was expanded

from MHD, where GHG and ODS measurements have been made since 1987, to include RGL, Herefordshire, England; TAC, Norfolk, England; and TTA, Dundee, Scotland.

We have designed a network with robust systems for unattended continuous measurement of atmospheric CO₂, CH₄, N₂O, SF₆, CO and H₂ mixing ratios using a suite of instruments at tall open lattice telecommunications towers. Results from the network give good temporal coverage of atmospheric mixing ratios since January 2012. Results from the network show that all GHGs are increasing in concentration over the selected reporting period and, except for SF₆, exhibit seasonal trends. Discrete sample and hourly mean data are freely available from EBAS at <http://ebas.nilu.no/> for GC-ECD/RGA and CRDS, respectively, for all three UK sites. All MHD data, except CO mole fractions, are available from CDIAC (<http://cdiac.ornl.gov/>). The instrumentation, software used and post-processing methods described in this paper are one model that can be used to inform future national or international monitoring networks.

Future improvements for the network include instrument specific H₂O corrections for the CRDSs, target tanks as an independent quality control measure, the inclusion of more optical measurements into the network to obtain data at higher frequency data and capture episodic interesting signals, as well as developing uncertainty algorithms that represent the main sources of error within the data.

Acknowledgements

We specifically acknowledge the cooperation and efforts of the station operators Gerard Spain and Duncan Brown at Mace Head monitoring station, and Mr Stephen Humphrey and Mr Andy MacDonald at the Tacolneston tall tower station. We also thank the Physics Department, National University of Ireland, Galway, for making the research facilities at Mace Head available. The operation of all stations was funded by the UK Department of Business, Energy and Industrial Strategy (formerly the Department of Energy and Climate Change) through contract TRN1028/06/2015 with additional funding at Mace Head under NASA contract NNX11AF17G through MIT with a sub award 5710002970 to the UoB.

Deleted: Each of the UK DECC sites, excluding MHD, were equipped with CRDS instruments to measure CO₂ and CH₄ at ultra-high frequency (~ 3 seconds). RGL and TAC also had GC-ECD systems installed to measure N₂O and SF₆ at high frequency (10 minutes). TAC and MHD also had a RGA integrated with the GC-ECD system to measure CO and H₂. At MHD, a GC-ECD and –FID measure N₂O and CH₄. At two selected sites (MHD and TAC), a Medusa GC-MS system was set up to measure SF₆ and a wide range of GHGs and ODSs. All instruments are controlled by Linux based software, GCWerks, and are accessible remotely. ... [1]

Deleted: spatial and

Deleted: The high-frequency data can be used for inversion modelling of GHG mole fractions and to discern GHG emissions for GHG inventory verification, as outlined in Manning et al. (2011). .

References

- Andrews, A. E., Kofler, J. D., Trudeau, M. E., Williams, J. C., Neff, D. H., Masarie, K. A., Chao, D. Y., Kitzis, D. R., Novelli, P. C., Zhao, C. L., Dlugokencky, E. J., Lang, P. M., Crotwell, M. J., Fischer, M. L., Parker, M. J., Lee, J. T., Baumann, D. D., Desai, A. R., Stanier, C. O., De Wekker, S. F. J., Wolfe, D. E., Munger, J. W., and Tans, P. P.: CO₂, CO, and CH₄ measurements from tall towers in the NOAA Earth System Research Laboratory's Global Greenhouse Gas Reference Network: instrumentation, uncertainty analysis, and recommendations for future high-accuracy greenhouse gas monitoring efforts, *Atmos. Meas. Tech.*, 7, 647-687, 10.5194/amt-7-647-2014, 2014.
- Arnold, T., Mühle, J., Salameh, P. K., Harth, C. M., Ivy, D. J., and Weiss, R. F.: Automated Measurement of Nitrogen Trifluoride in Ambient Air, *Analytical Chemistry*, 84, 4798-4804, 10.1021/ac300373e, 2012.
- Brailsford, G. W., Stephens, B. B., Gomez, A. J., Riedel, K., Mikaloff Fletcher, S. E., Nichol, S. E., and Manning, M. R.: Long-term continuous atmospheric CO₂ measurements at Baring Head, New Zealand, *Atmospheric Measurement Techniques*, 5, 3109-3117, 10.5194/amt-5-3109-2012, 2012.
- Chen, H., Winderlich, J., Gerbig, C., Hofer, A., Rella, C. W., Crosson, E. R., Van Pelt, A. D., Steinbach, J., Kolle, O., Beck, V., Daube, B. C., Gottlieb, E. W., Chow, V. Y., Santoni, G. W., and Wofsy, S. C.: High-accuracy continuous airborne measurements of greenhouse gases (CO₂ and CH₄) using the cavity ring-down spectroscopy (CRDS) technique, *Atmospheric Measurement Techniques*, 3, 375-386, doi:10.5194/amt-3-375-2010, 2010, 2010.
- Crosson, E. R.: A cavity ring-down analyser for measuring atmospheric levels of methane, carbon dioxide, and water vapor, *Applied Physics B*, 92, 403-408, doi: 10.1007/s00340-008-3135-y, 2008.
- Cunnold, D. M., Weiss, R. F., Prinn, R. G., Hartley, D., Simmonds, P. G., Fraser, P. J., Miller, B., Alyea, F. N., and Porter, L.: GAGE/AGAGE measurements indicating reductions in global emissions of CCl₃F and CCl₂F₂ in 1992-1994, *Journal of Geophysical Research*, 102, 1259, doi: 10.1029/1296JD02973, 1997.
- DEFRA: Waste water treatment in the United Kingdom-2012: Implementation of the European Union Urban Waste Water treatment directive 91/271/EEC, PB13811, Department for Environment, Food and Rural Affairs, London, 2012.
- Dimmer, C. H., Simmonds, P. G., Nickless, G., and Bassford, M. R.: Biogenic fluxes of halomethanes from Irish peatland ecosystems, *Atmos. Environ.*, 35, 321-330, 2001.
- Dlugokencky, E. J., Myers, R. C., Lang, P. M., Masarie, K. A., Crotwell, A. M., Thoning, K. W., Hall, B. D., Elkins, J. W., and Steele, L. P.: Conversion of NOAA atmospheric dry air CH₄ mole fractions to a gravimetrically prepared standard scale, *J. Geophys. Res.*, 110, doi:10.1029/2005JD006035, 2005.
- Dlugokencky, E. J., Nisbet, E. G., Fisher, R., and Lowry, D.: Global atmospheric methane: budget, changes and dangers, *Philosophical Transactions of the Royal Society of London A: Mathematical, Physical and Engineering Sciences*, 369, 2058-2072, 10.1098/rsta.2010.0341, 2011.
- EPRL: Energy Power Resources Ltd., location of power generation sites, <http://www.epri.co.uk/assets/eye/overview.html>, 2013.
- Ganesan, A. L., Chatterjee, A., Prinn, R. G., Harth, C. M., Salameh, P. K., Manning, A. J., Hall, B. D., Muhle, J., Meredith, L. K., Weiss, R. F., O'Doherty, S., and Young, D.: The variability of methane, nitrous oxide and sulfur hexafluoride in Northeast India, *Atmos. Chem. Phys.*, 13, 10633-10644, doi: 10.5194/acp-13-10633-2013, 2013.
- Ganesan, A. L., Manning, A. J., Grant, A., Young, D., Oram, D. E., Sturges, W. T., Moncrieff, J. B., and O'Doherty, S.: Quantifying methane and nitrous oxide emissions from the UK and Ireland using a national-scale monitoring network, *Atmos. Chem. Phys.*, 15, 6393-6406, 10.5194/acp-15-6393-2015, 2015.
- Gerbig, C., Lin, J. C., Wofsy, S. C., Daube, B. C., Andrews, A. E., Stephens, B. B., Bakwin, P. S., and Grainger, C. A.: Toward constraining regional-scale fluxes of CO₂ with atmospheric observations over a continent: 1. Observed spatial variability from airborne platforms *J. Geophys. Res.*, 108, 4756, doi:10.1029/2002JD003018, 2003.
- Gerbig, C., Dolman, A. J., and Heimann, M.: On observational and modelling strategies targeted at regional carbon exchange over continents, *Biogeosciences*, 6, 1949-1959, doi:10.5194/bg-6-1949-2009, 2009.
- Gloor, M., Bakwin, P., Hurst, D., Lock, L., Draxler, R., and Tans, P.: What is the concentration footprint of a tall tower?, *Journal of Geophysical Research: Atmospheres*, 106, 17831-17840, 10.1029/2001JD900021, 2001.

Deleted: Andrews, A. E., Kofler, J. D., Trudeau, M. E., Williams, J. C., Neff, D. H., Masarie, K. A., Chao, D. Y., Kitzis, D. R., Novelli, P. C., Zhao, C. L., Dlugokencky, E. J., Lang, P. M., Crotwell, M. J., Fischer, M. L., Parker, M. J., Lee, J. T., Baumann, D. D., Desai, A. R., Stanier, C. O., De Wekker, S. F. J., Wolfe, D. E., Munger, J. W., and Tans, P. P.: CO₂, CO, and CH₄ measurements from tall towers in the NOAA Earth System Research Laboratory's Global Greenhouse Gas Reference Network: instrumentation, uncertainty analysis, and recommendations for future high-accuracy greenhouse gas monitoring efforts, *Atmos. Meas. Tech.*, 7, 647-687, 10.5194/amt-7-647-2014, 2014. ... [2]

- Grant, A., Stanley, K. F., Henshaw, S. J., Shallcross, D. E., and O'Doherty, S.: High-frequency urban measurements of molecular hydrogen and carbon monoxide in the United Kingdom, *Atmos. Chem. Phys.*, 10, 4715-4724, 10.5194/acp-10-4715-2010, 2010a.
- 5 Grant, A., Witham, C. S., Simmonds, P. G., Manning, A. J., and O'Doherty, S.: A 15 year record of high-frequency, in situ measurements of hydrogen at Mace Head, Ireland, *Atmos. Chem. Phys.*, 10, 1203-1214, 10.5194/acp-10-1203-2010, 2010b.
- GRO: General Register Office for Scotland, <http://www.gro-scotland.gov.uk/>, 2013.
- Hall, B. D., Dutton, G. S., and Elkins, J. W.: The NOAA nitrous oxide standard scale for atmospheric observations, *J. Geophys. Res.*, 112, doi:10.1029/2006JD007954, 2007.
- 10 Hall, B. D., Dutton, G. S., Mondeel, D. J., Nance, J. D., Rigby, M., Butler, J. H., Moore, F. L., Hurst, D. F., and Elkins, J. W.: Improving measurements of SF₆ for the study of atmospheric transport and emissions, *atmos. Meas. Tech.*, 4, 2441-2451, 2011.
- Haszpra, L., Barcza, Z., Bakwin, P. S., Berger, B. W., Davis, K. J., and Weidinger, T.: Measuring system for the long-term monitoring of biosphere/atmosphere exchange of carbon dioxide, *J. Geophys. Res.*, 106, 3057-3069, doi:10.1029/2000jd900600, 2001.
- 15 Hazan, L., Tarniewicz, J., Ramonet, M., Laurent, O., and Abbaris, A.: Automatic processing of atmospheric CO₂ and CH₄ mole fractions at the ICOS Atmospheric Thematic Center, *Atmos. Meas. Tech.*, 9, 1-34, 10.5194/amt-9-4719-2016, 2016.
- Jennings, S. G., Kleefeld, C., O Dowd, C. D., Junker, C., Spain, T. G., O'Brien, P., Roddy, A. F., and O'Connor, T. C.: Mace Head Atmospheric Research Station-characterization of aerosol radiative parameters, *Boreal environment research*, 8, 303-314, 2003.
- 20 Jones, A. R., Thomson, D. J., Hort, M., and Devenish, B.: The U.K. Met Office's next-generation atmospheric dispersion model, NAME III, in Borrego C. and Norman A.-L. (Eds), *Air Pollution Modeling and its Application XVII Proceedings of the 27th NATO/CCMS International Technical Meeting on Air Pollution Modelling and its Application*, 580-589, 2007.
- Keeling, C. D., Bacastow, R. B., Bainbridge, A. E., Ekdahl, C. A., Guenther, P. R., Waterman, L. S., and Chin, J. F. S.: Atmospheric carbon dioxide variations at Mauna Loa Observatory, Hawaii, *Tellus*, 28, 538-551, 10.1111/j.2153-3490.1976.tb00701.x, 1976.
- 25 Kirschke, S., Bousquet, P., Ciais, P., Saunoy, M., Canadell, J. G., Dlugokencky, E. J., Bergamaschi, P., Bergmann, D., Blake, D. R., Bruhwiler, L., Cameron-Smith, P., Castaldi, S., Chevallier, F., Feng, L., Fraser, A., Heimann, M., Hodson, E. L., Houweling, S., Josse, B., Fraser, P. J., Krummel, P. B., Lamarque, J.-F., Langenfelds, R. L., Le Quere, C., Naik, V., O'Doherty, S., Palmer, P. I., Pison, I., Plummer, D., Poulter, B., Prinn, R. G., Rigby, M., Ringeval, B., Santini, M., Schmidt, M., Shindell, D. T., Simpson, I. J., Spahni, R., Steele, L. P., Strode, S. A., Sudo, K., Szopa, S., van der Werf, G. R., Voulgarakis, A., van Weele, M., Weiss, R. F., Williams, J. E., and Zeng, G.: Three decades of global methane sources and sinks, *Nature Geosci.*, 6, 813-823, 10.1038/ngeo1955, 2013.
- Kozlova, E. A., Manning, A. C., Kisilyakhov, Y., Seifert, T., and Heimann, M.: Seasonal, synoptic, and diurnal-scale variability of biogeochemical trace gases and O₂ from a 300-m tall tower in central Siberia, *Global Biogeochem. Cycles*, 22, doi:10.1029/2008GB003209, 2008.
- 35 Krummel, P.: RE: MHD NOAA-AGAGE difference for CH₄. Personal communication to Stanley, K. M. on 15/01/2018.
- Le Quéré, C., Moriarty, R., Andrew, R. M., Canadell, J. G., Sitch, S., Korsbakken, J. I., Friedlingstein, P., Peters, G. P., Andres, R. J., Boden, T. A., Houghton, R. A., House, J. I., Keeling, R. F., Tans, P., Arneeth, A., Bakker, D. C. E., Barbero, L., Bopp, L., Chang, J., Chevallier, F., Chini, L. P., Ciais, P., Fader, M., Feely, R. A., Gkritzalis, T., Harris, I., Hauck, J., Ilyina, T., Jain, A. K., Kato, E., Kitidis, V., Klein Goldewijk, K., Koven, C., Landschützer, P., Lauvset, S. K., Lefèvre, N., Lenton, A., Lima, I. D., Metzl, N., Millero, F., Munro, D. R., Murata, A., Nabel, J. E. M. S., Nakaoka, S., Nojiri, Y., O'Brien, K., Olsen, A., Ono, T., Pérez, F. F., Pfeil, B., Pierrot, D., Poulter, B., Rehder, G., Rödenbeck, C., Saito, S., Schuster, U., Schwinger, J., Séférian, R., Steinhoff, T., Stocker, B. D., Sutton, A. J., Takahashi, T., Tilbrook, B., van der Laan-Luijkx, I. T., van der Werf, G. R., van Heuven, S., Vandemark, D., Viovy, N., Wiltshire, A., Zaehle, S., and Zeng, N.: Global Carbon Budget 2015, *Earth Syst. Sci. Data*, 7, 349-396, 10.5194/essd-7-349-2015, 2015.
- 45 Levin, I., Naegler, T., Heinz, R., Osusko, D., Cuevas, E., Engel, A., Ilmberger, J., Langenfelds, R. L., Neininger, B., Rohden, C. v., Steele, L. P., Weller, R., Worthy, D. E., and Zimov, S. A.: The global SF₆ source inferred from long-term high precision atmospheric measurements and its comparison with emission inventories, *Atmos. Chem. Phys.*, 10, 2655-2662, doi: 10.5194/acp-10-2655-2010, 2010.

- Luan, T., Fang, S., Yao, B., Wang, H., Dong, F., Shi, Q., Zhou, L., and Zhou, H.: Atmospheric molecular hydrogen (H₂) at the Shangdianzi regional background station in China, *Atmospheric Environment*, 145, 318-325, doi: 10.1016/j.atmosenv.2016.09.044, 2016.
- 5 Manning, A. J., O'Doherty, S., Jones, A. R., Simmonds, P. G., and Derwent, R. G.: Estimating UK methane and nitrous oxide emissions from 1990 to 2007 using an inversion modeling approach, *J. Geophys. Res.*, 116, D02305, 10.1029/2010jd014763, 2011.
- Miller, B. R., Weiss, R. F., Salameh, P. K., Tanhua, T., Grealley, B. R., Muhle, J., and Simmonds, P. G.: Medusa: A Sample Preconcentration and GC/MS Detector System for in Situ Measurements of Atmospheric Trace Halocarbons, Hydrocarbons, and Sulfur Compounds, *Anal. Chem.*, 80, 1536-1545, 2008.
- 10 NCC: Norfolk County Council, Landfill Sites location,, http://www.norfolk.gov.uk/Environment/Waste_and_Recycling/Waste_collection_and_disposal/Landfill_sites/index.htm, 2013.
- Nevison, C. D., Dlugokencky, E., Dutton, G., Elkins, J. W., Fraser, P., Hall, B., Krummel, P. B., Langenfelds, R. L., O'Doherty, S., Prinn, R. G., Steele, L. P., and Weiss, R. F.: Exploring causes of interannual variability in the seasonal cycles of tropospheric nitrous oxide, *Atmos. Chem. Phys.*, 11, 3713-3730, 10.5194/acp-11-3713-2011, 2011.
- 15 ONS: Office for National Statistics; 2011 Census of England and Wales, 2012.
- Popa, M. E., Gloor, M., Manning, A. C., Jordan, A., Schultz, U., Haensel, F., Seifert, T., and Heimann, M.: Measurements of greenhouse gases and related tracers at Bialystok tall tower station in Poland, *Atmos. Meas. Tech.*, 3, 407-427, doi:10.5194/amt-3-407-2010, 2010.
- 20 Prinn, R. G., Weiss, R. F., Fraser, P. J., Simmonds, P. G., Cunnold, D. M., Alyea, F. N., O'Doherty, S., Salameh, P., Miller, B. R., Huang, J., Wang, R. H. J., Hartley, D. E., Harth, C., Steele, L. P., Sturrock, G., Midgley, P. M., and McCulloch, A.: A history of chemically and radiatively important gases in air deduced from ALE/GAGE/AGAGE, *J. Geophys. Res.*, 105, doi:10.1029/2000JD900141, 2000.
- Rella, C.: Accurate greenhouse gas measurements in humid gas streams using the Picarro G1301 carbon dioxide/methane/water vapor gas analyzer, White paper, Picarro Inc, Sunnyvale, CA, USA, 2010.
- 25 Rella, C., Chen, H., Andrews, A. E., Filges, A., Gerbig, C., Hatakka, J., Karion, A., Miles, N. L., Richardson, S. J., Steinbacher, M., Sweeney, C., Wastine, B., and Zellweger, C.: High accuracy measurements of dry mole fractions of carbon dioxide and methane in humid air, *Atmos. Meas. Tech.*, 6, 837-860, 10.5194/amt-6-837-2013, 2013.
- Satar, E., Berhanu, T. A., Brunner, D., Henne, S., and Leuenberger, M.: Continuous CO₂/CH₄/CO measurements (2012–2014) at Beromünster tall tower station in Switzerland, *Biogeosciences*, 13, 2623-2635, 10.5194/bg-13-2623-2016, 2016.
- 30 Schmidt, M., Glatzel-Mattheier, H., Sartorius, H., Worthy, D. E., and Levin, I.: Western European N₂O emissions: A top-down approach based on atmospheric observations, *Journal of Geophysical Research: Atmospheres*, 106, 5507-5516, 10.1029/2000JD900701, 2001.
- Stocker, T. F., Qin, D., Plattner, G. K., Tignor, M., Allen, S. K., Boschung, J., Nauels, A., Xia, Y., Bex, V., and Midgley, P. M.: *Climate Change 2013: The Physical Science Basis. Contribution of Working Group I to the Fifth Assessment Report of the Intergovernmental Panel on Climate Change*, Cambridge University Press, Cambridge, United Kingdom and New York, NY, USA, 1535, doi:10.1017/CBO9781107415324, 2013.
- Tans, P. P.: Observational strategy for assessing the role of terrestrial ecosystems in the global carbon cycle: scaling down to regional levels, in: *Scaling physiological processes: leaf to globe*, edited by: Ebleringer, J. R., and Field, C. B., Academic Press, San Diego, California, USA, 179-190, 1993.
- 40 Thompson, R. L., Manning, A. C., Gloor, E., Schultz, U., Seifert, T., Hansel, F., and Jordan, A.: In-situ measurements of oxygen, carbon monoxide and greenhouse gases from Ochsenkopf tall tower in Germany, *Atmos. Meas. Tech.*, 2, 573-591, doi:10.5194/amt-2-573-2009, 2009.
- Tremblay, A., Lambert, M., and Gagnon, L.: Do Hydroelectric Reservoirs Emit Greenhouse Gases?, *Environmental Management*, 33, S509-S517, 10.1007/s00267-003-9158-6, 2004.
- 45 Vardag, S. N., Hammer, S., O'Doherty, S., Spain, T. G., Wastine, B., Jordan, A., and Levin, I.: Comparisons of continuous atmospheric CH₄, CO₂ and N₂O measurements - results from a travelling instrument campaign at Mace Head, *Atmos. Chem. Phys.*, 14, 8403-8418, 10.5194/acp-14-8403-2014, 2014.

- Verhulst, K. R., Karion, A., Kim, J., Salameh, P. K., Keeling, R. F., Newman, S., Miller, J., Sloop, C., Pongetti, T., Rao, P., Wong, C., Hopkins, F. M., Yadav, V., Weiss, R. F., Duren, R. M., and Miller, C. E.: Carbon dioxide and methane measurements from the Los Angeles Megacity Carbon Project – Part I: calibration, urban enhancements, and uncertainty estimates, *Atmos. Chem. Phys.*, 17, 8313-8341, 10.5194/acp-17-8313-2017, 2017.
- 5 Vermeulen, A. T.: CHITTO Final Report, ECN energy research Centre of the Netherlands, 2007.
- Vermeulen, A. T., Hensen, A., Popa, M. E., van den Bulk, W. C. M., and Jongejan, P. A. C.: Greenhouse gas observations from Cabauw Tall Tower (1992–2010), *Atmos. Meas. Tech.*, 4, 617-644, 10.5194/amt-4-617-2011, 2011.
- Winderlich, J., Chen, H., Gerbig, C., Seifert, T., Kolle, O., Lavric, J. V., Kaiser, C., Hofer, A., and Heimann, M.: Continuous low-maintenance CO₂/CH₄/H₂O measurements at the Zotino Tall Tower Observatory (ZOTTO) in Central Siberia, *Atmos. Meas. Tech.*, 3, 1113-1128, doi: 10.5194/amt-3-1113-2010, 2010.
- 10 WMO-GAW: 18th WMO/IAEA Meeting on Carbon Dioxide, Other Greenhouse Gases and Related Tracers Measurement Techniques (GGMT-2015), La Jolla, CA, USA, 13 - 17 September 2015, WMO, Geneva, Switzerland, 2016.
- Yver Kwok, C., Laurent, O., Guemri, A., Philippon, C., Wastine, B., Rella, C. W., Vuillemin, C., Truong, F., Delmotte, M., Kazan, V., Darding, M., Lebégue, B., Kaiser, C., Xueref-Rémy, I., and Ramonet, M.: Comprehensive laboratory and field testing of cavity ring-down spectroscopy analyzers measuring H₂O, CO₂, CH₄ and CO, *Atmos. Meas. Tech.*, 8, 3867-3892, 10.5194/amt-8-3867-2015, 2015.
- 15 Zhao, C. L., and Tans, P. P.: Estimating uncertainty of the WMO mole fraction scale for carbon dioxide in air, *J. Geophys. Res.*, 111, doi: 10.1029/2005JD006003, 2006.

Table 1: Greenhouse gas and ozone depleting substance species and instrumentation at each UK DECC site.

Species	Sites			
	Mace Head (MHD)	Tacolneston (TAC)	Ridge Hill (RGL)	Angus (TTA)
CO ₂	Picarro 2301 ^a	Picarro G2301	Picarro G2301	Picarro G2301
CH ₄	GC-FID ^b	Picarro G2301	Picarro G2301	Picarro G2301
N ₂ O	GC-ECD	GC-ECD	GC-ECD	-
SF ₆	Medusa ^c	GC-ECD/Medusa ^c	GC-ECD	-
H ₂	GC-RGA	GC-RGA	-	-
CO	GC-RGA	GC-RGA	-	-

^a Picarro G2301 instruments on site is owned and managed by Laboratoire des Sciences du Climat et de l'Environnement (LSCE), France. Data is available through the ICOS Carbon Portal (<https://www.icos-cp.eu/>).

^b CH₄ is also analysed on the Picarro G2301 instrument maintained by LSCE. Data is available through the ICOS Carbon Portal (<https://www.icos-cp.eu/>).

^c Other ozone depleting species measured on the Medusa GC-MS can be found in Miller et al. (2008) and Arnold et al. (2012).

Formatted Table

Deleted: CF₄

... [3]

Formatted: Normal

Table 2: Site names, locations and inlet heights

Site Name	Acronym	Location	Altitude ^a (m.a.s.l.)	Inlet Heights (m.a.g.l.)
Mace Head	MHD	53.327 ° N 9.904 ° W	8	10
Ridge Hill tower	RGL	51.998 ° N 2.540 ° W	204	45, 90
Tacolneston tower	TAC	52.518 ° N 1.139 ° E	56	54, 100, 185
Angus tower	TTA	56.555 ° N 2.986 ° W	400	222

^a Altitude measured at base of tower.

Table 3: Gas chromatograph-flame ionisation and electron capture detector equipment and setup at UK DECC stations. * indicates N₂O channel only on the MHD GC-ECD.

	FID	MHD ECD*	TAC ECD	RGL ECD
Instrument	Carle AGC-211	Hewlett-Packard 5890	Agilent 6890N	Agilent 7890A
Detector	FID	ECD	μECD	μECD
Sample volume	10 mL	8 mL	8 mL	8 mL
Oven temperature	60° C	55° C	90° C	90° C
Column temperature	60° C	185° C (pre), 55° C (main)	90° C (pre and main), 180° C (post)	90° C (pre and main), 180° C (post)
Detector temperature	N/A	325° C	350° C	350° C
Pre-column	Silica gel	Molecular sieve 5Å, 60/80 mesh	Porapack Q, 80/100 mesh, 1.0 m x 3/16" SS	Porapack Q, 80/100 mesh, 1.0 m x 3/16" SS
Main-column	Molecular sieve 5Å, 60/80 mesh	Porasil C	Porapack Q, 80/100 mesh, 2.0 m x 3/16" SS	Porapack Q, 80/100 mesh, 2.0 m x 3/16" SS
Post-column	N/A	N/A	Molecular sieve 5Å, 45/60 mesh, 0.9 m x 1/8"	Molecular sieve 5Å, 45/60 mesh, 0.9 m x 1/8"
Carrier gas supply	N ₂ cylinder (5.0)	AR/CH ₄ cylinder (95 %/5 %) (5.0)	AR/CH ₄ cylinder (95 %/5 %) (5.0)	AR/CH ₄ cylinder (95 %/5 %) (5.0)
H ₂ supply	Cylinder (5.0)	N/A	N/A	N/A
Zero air supply	TOC generator (Parker Balston TOC-1250)	TOC generator (Parker Balston TOC-1250)	TOC generator (Parker Balston TOC-1250)	TOC generator (Parker Balston TOC-1250)

Table 4: Reduction gas analyser equipment and setup at UK DECC stations.

	MHD	TAC
Instrument	Trace Analytical RGA3	Trace Analytical PP1
Detector	RGA	RGA
Sample volume	1 mL	1 mL
Column temperature	105° C	105° C
Pre-column	Unibeads 1S, 60/80 mesh, 0.768 m x 1.8”	Unibeads 1S, 60/80 mesh, 0.768 m x 1.8”
Main-column	Molecular sieve 5Å, 60/80 mesh, 0.768 m x 1.8”	Molecular sieve 5Å, 60/80 mesh, 0.768 m x 1.8”
Carrier gas supply	Zero air cylinder (5.5) + purifier	Zero air cylinder (5.5) + purifier
Zero air supply	TOC generator (Parker Balston TOC-1250)	TOC generator (Parker Balston TOC-1250)

Table 5: Maintenance schedule for all UK DECC sites from January 2012 to September 2015.

Site	Scheduled visits	Emergency visits			
		2012	2013	2014	2015
MHD	Biweekly	1	1	1	N/A
RGL	6 weekly	2	2	3	4
TAC	Biweekly	2	2	2	3
TTA	8 weekly	N/A	2	3	3

Formatted: Normal

Table 6. Cavity Ring-Down Spectrometry calibrant (cal) and standard gas mole fractions for CO₂ (μmol mol⁻¹) and CH₄ (nmol mol⁻¹) assigned by GASLAB at MPI-BGC Jena for the UK DECC network.

Species	Gas	Site		
		Tacolneston (TAC)	Ridge Hill (RGL)	Angus (TTA)
CO ₂ (μmol mol ⁻¹)	Standard	386.70	385.44	401.29
	Cal 1	338.85	338.52	346.93
	Cal 2	380.23	380.11	374.75
	Cal 3	419.91	419.61	449.51
	Cal 4	469.55	469.22	-
CH ₄ (μmol mol ⁻¹)	Standard	1900.1	1953.7	1947.4
	Cal 1	1598.3	1598.2	1742.9
	Cal 2	1797.3	17989.8	1851.5
	Cal 3	1994.5	1992.0	2145.0
	Cal 4	2189.2	2188.7	-

Deleted: 5

Table 7: Parameters used in GCWerks for automatic CRDS data filtering until September 2015.

Filter	Threshold
Cavity pressure (Torr)	< 139.9, > 140.1
Cavity temperature (°C)	< 44.98, > 45.02
Water value (%)	6
Cycle time (seconds)	> 8
Standard deviation ^{a,b}	10

^a for measured compounds (pmol/nmol/μmol mol⁻¹) and water (%).

^b Points outside of the set number of standard deviations are filtered recursively until all points are lower than the set parameter. For air data, a 2 minute moving window is used to filter out extreme outliers. The moving windows overlap by 1 minute.

Table 8: Troubleshooting data issues observed in the UK DECC network from it starting in 2012 to September 2015.

Problem	Observation	Comment
Increasing laboratory temperature at TAC	Automatic email triggered when maximum ambient temperature setpoint reached	Automatic shutdown of Medusa GC-MS and email sent to site operators, who logged on and shutdown instruments to reduce heat load into laboratory and protect instruments.
Line filter blockage	Decreasing outlet valve value in CRDS time series	Change relative to other sample inlet values. Once detected, site operator can replace the inlet filter.
Line pump failure	Increasing outlet valve value in CRDS time series	A steadily increasing outlet valve value can indicate a tear forming in the diaphragm of the line pumps. Eventually there will be a step change in the valve value when the diaphragm completely fails. Site operators can logon remotely to remove the affected inlet line from the sampling sequence.
Router problems	No data transfer	A number of routers have failed at sites due to the constant use. Usually if no access to site can be gained, a site operator will make an emergency visit.
Cylinders vented	Spurious data for cylinder and if on CRDS, lower than normal outlet valve value	Cylinders have vented during the time period reported. Safe positions have been introduced so that when the instrument stops, the valve in the sample module goes to the safe position. If mismatches in valve positions occur, site operators are warned by automatically generated emails.
Contaminated carrier gas cylinder	Increasing baseline, smaller peaks or increasing blanks concentrations.	Test all carrier gases for contamination on instruments as an unknown before connection to any instrument.

Formatted Table

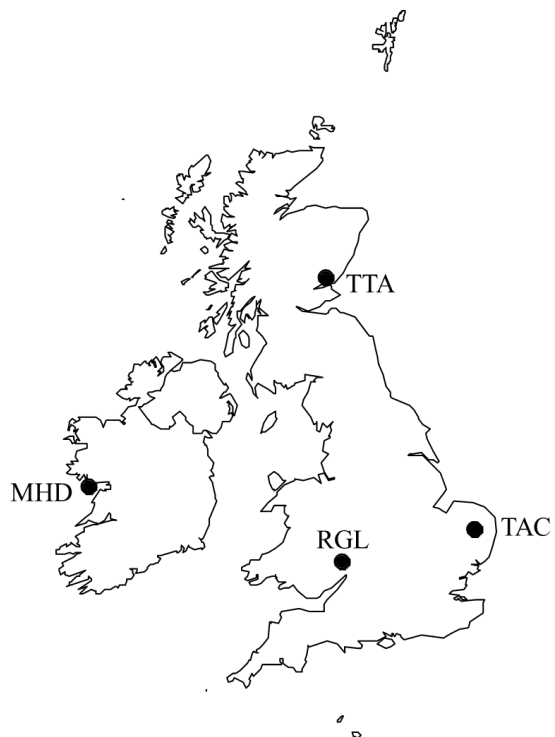


Figure 1: Location of UK DECC network stations, showing from north to south: TTA, Angus, UK; MHD, Mace Head, Ireland; TAC, Tacolneston, UK; and RGL, Ridge Hill, UK.

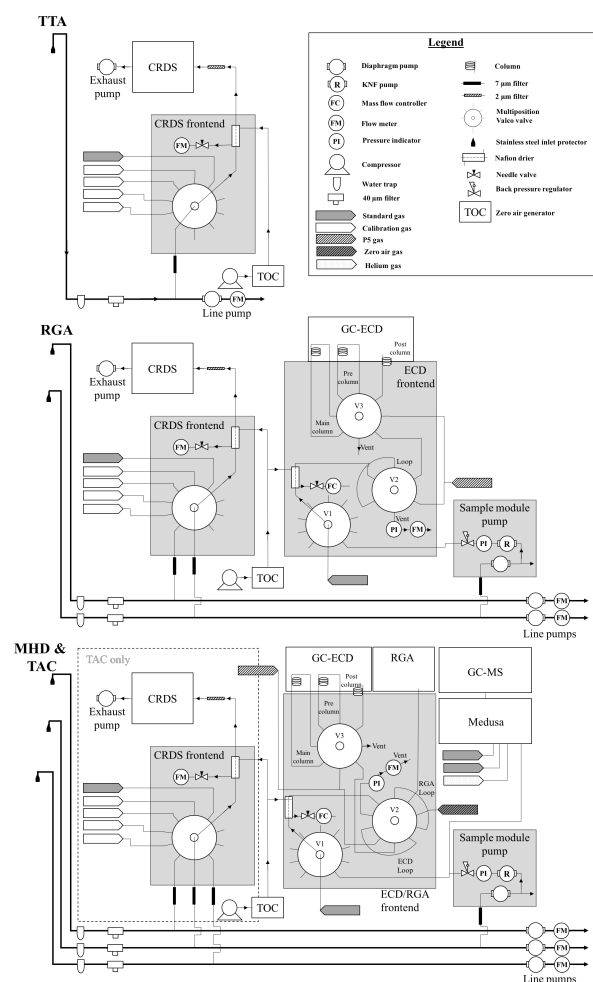
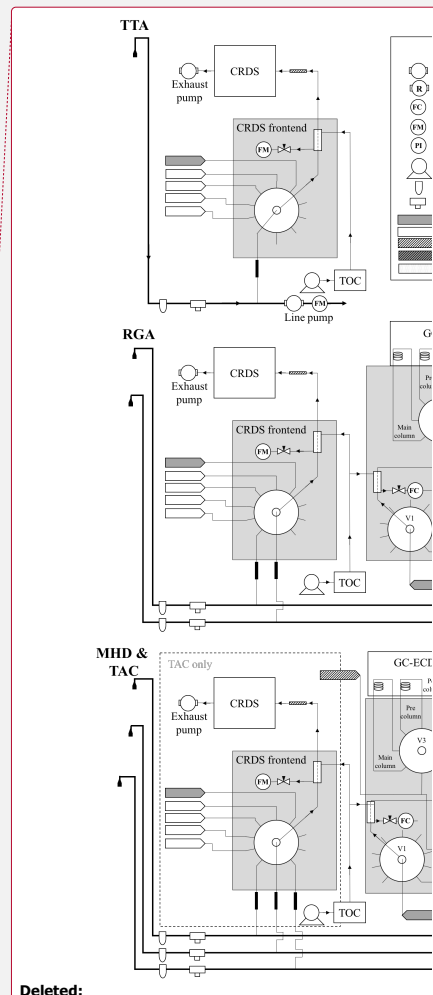


Figure 2: Schematic diagram of the UK DECC network (Angus, TTA; Ridge Hill, RGL; and Tacolneston, TAC) CO₂, CH₄, N₂O, SF₆, CO and H₂ analysis system. The MHD setup is outlined in Prinn et al. (2000).



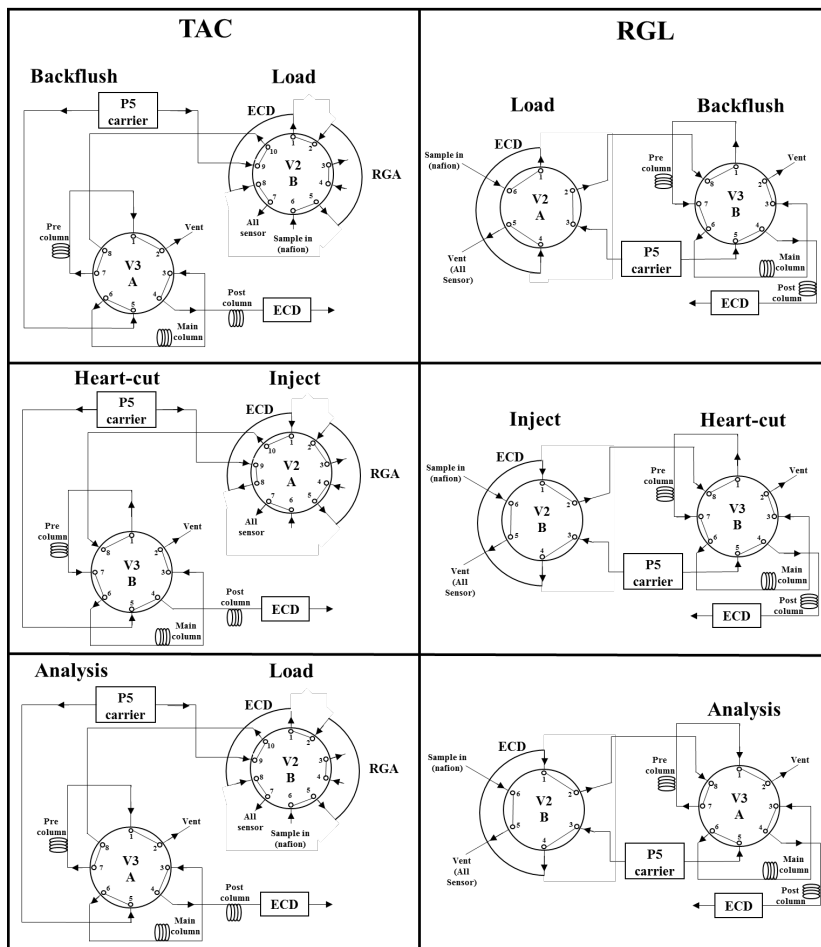
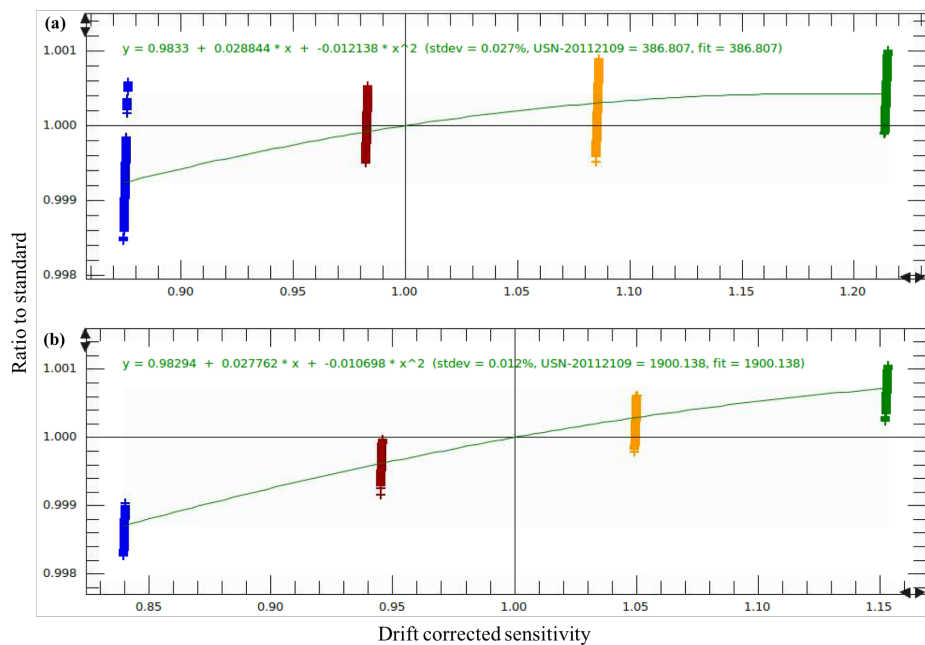


Figure 3: ECD frontend valve configuration for sample backflush, heart-cut and analysis at Tacolneston (TAC) and Ridge Hill (RGL). The MHD setup is outlined in Prinn et al. (2000).



Formatted: Caption

Figure 4: An example of a non-linear fit used for (a) CO₂ and (b) CH₄ at TAC from 02/04/2012 to 01/10/2015. Curve coefficients are shown within the plots, along with the standard used (USN-20112109), the assigned values and the adjusted values (fit =). Drift corrected sensitivity is a function of the measured cylinder dry mole fraction divided by its assigned value, over the measured standard dry mole fraction divided by the assigned/adjusted standard value, as outlined in Eq. 2.

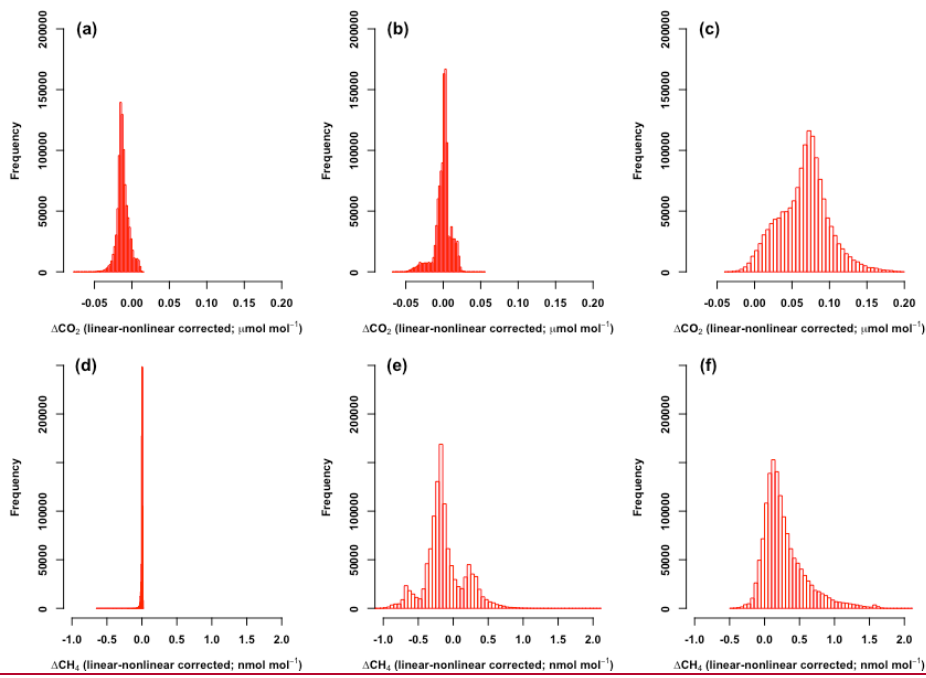


Figure 5: Histograms of the difference between linear corrected data (Eq. 1) and non-linear corrected data (Eq. 2) for CO₂ at (a) RGL, (b) TTA, and (c) TAC, and CH₄ at (d) RGL, (e) TTA, and (f) TAC.

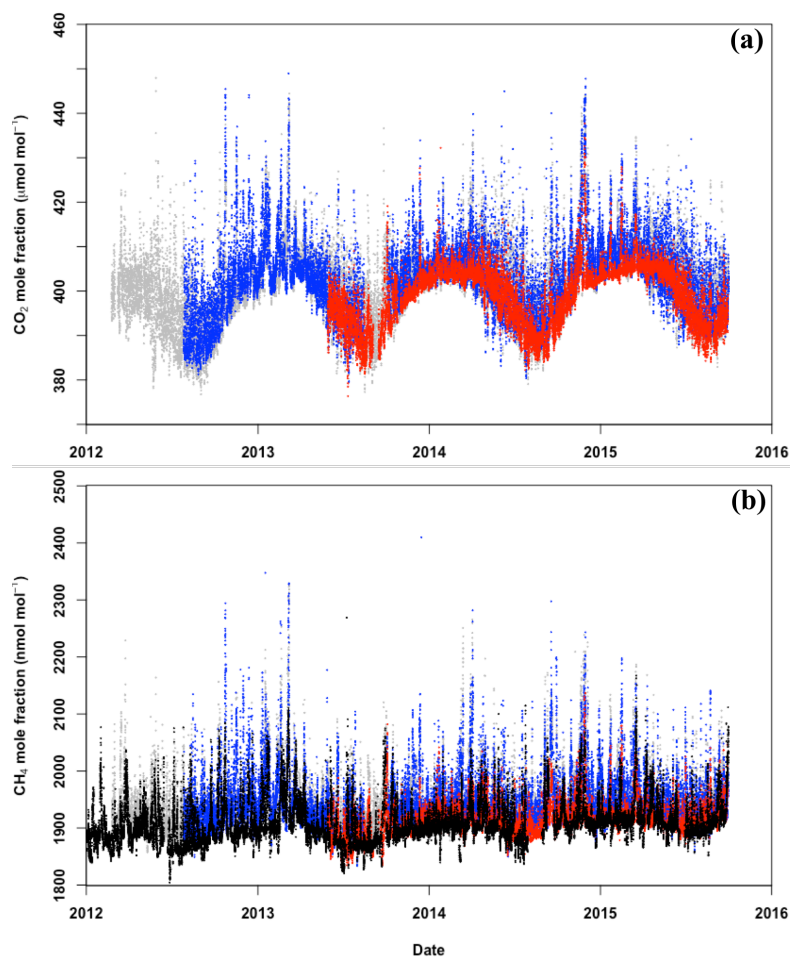


Figure 6: Time series of (a) CO₂ and (b) CH₄ mole fractions at MHD (black; 10 m inlet) RGL (grey; 90 m inlet), TTA (red; 222 m inlet) and TAC (blue; 100 m inlet) from 1 January 2012 to 30 September 2015. Results shown are hourly averages. CO₂ results shown are in $\mu\text{mol mol}^{-1}$ and are on the WMO-X2007 scale. CH₄ results are shown in nmol mol^{-1} and are on the WMO-X2004A scale.

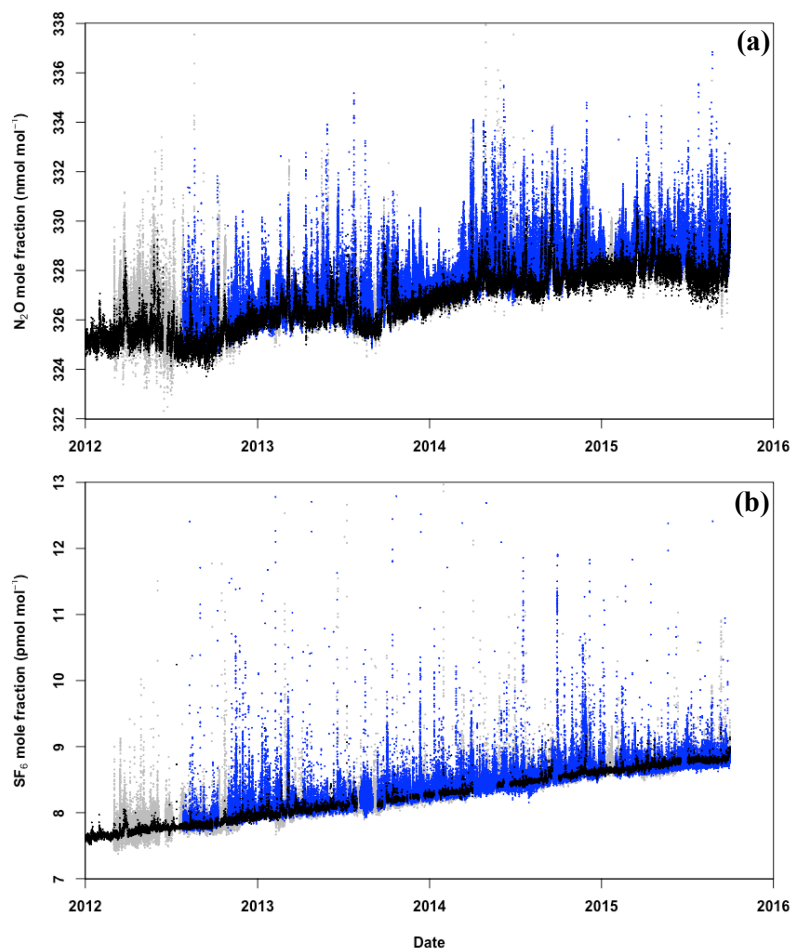


Figure 7: Time series of (a) N_2O and (b) SF_6 mole fractions at MHD (black; 10 m inlet), RGL (grey; 90 m inlet) and TAC (blue; 100 m inlet) from 1 January 2012 to 30 September 2015. All results shown are hourly averages. N_2O results are shown in nmol mol^{-1} and are on the SIO-98 scale, whilst SF_6 results are shown in pmol mol^{-1} and are on SIO-05 scale. SF_6 results from MHD are made using the Medusa GC-MS.

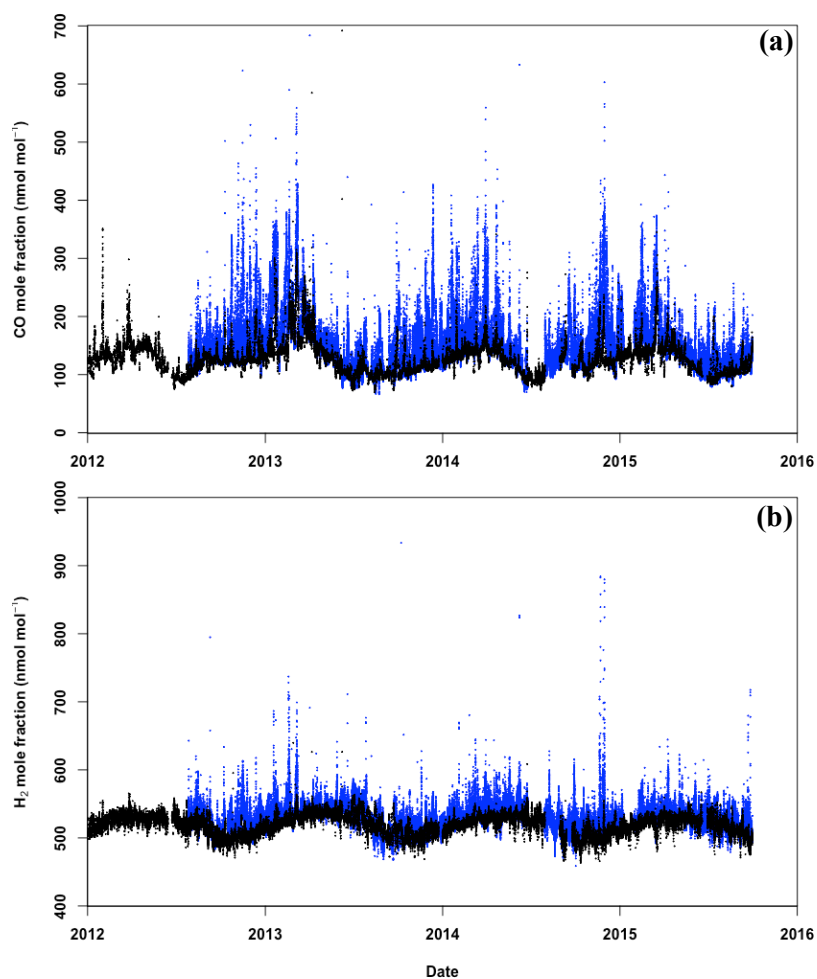


Figure 8: Time series of (a) CO and (b) H_2 mole fractions at MHD (black; 10 m inlet) and TAC (blue; 100 m inlet) from 1 January 2012 to 30 September 2015. All results shown are hourly averages, shown in nmol mol^{-1} and are on CSIRO04 and MPI-2009 scales for CO and H_2 , respectively.

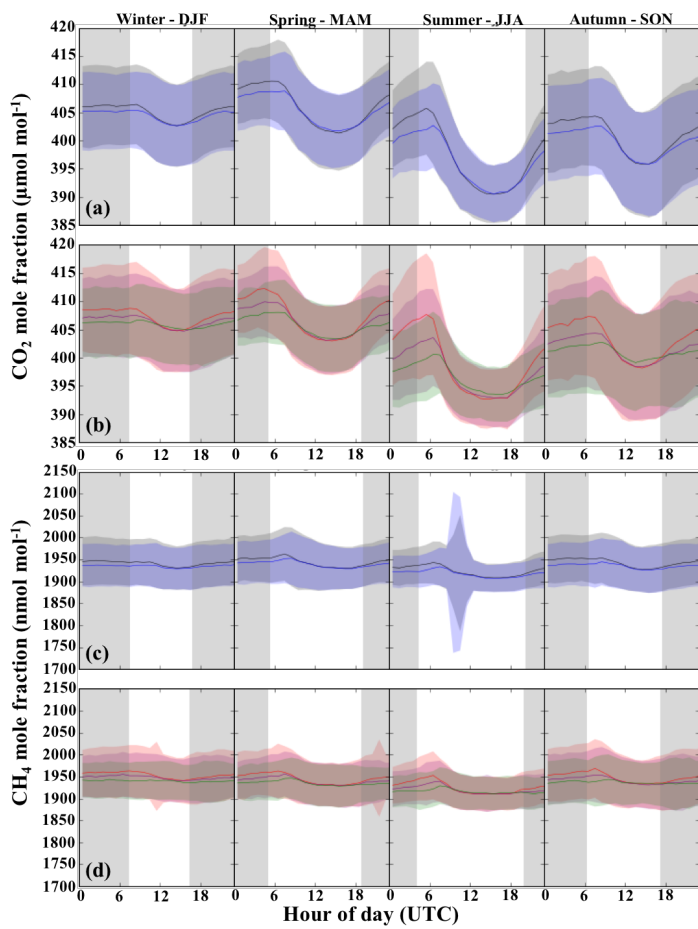
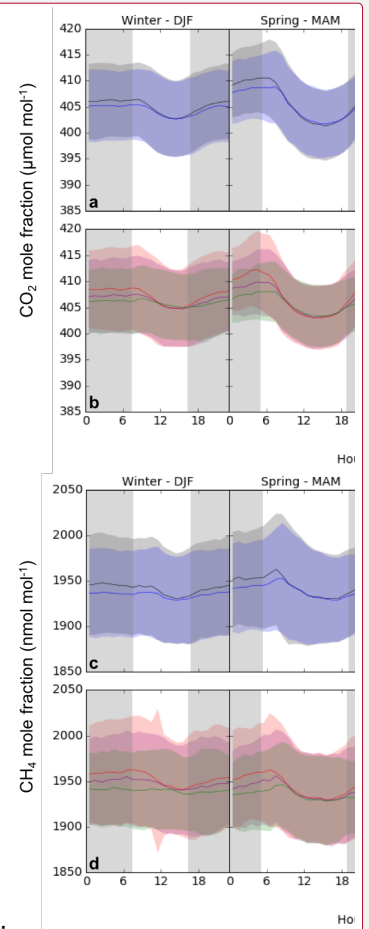


Figure 9: Overview of average diurnal concentration gradients in CO₂ at (a) RGL and (b) TAC, and CH₄ at (c) RGL and (d) TAC from 23 January 2012 to 1 October 2015. Lines are the median of the entire data period and shaded areas represent the standard deviation (1 σ). Black and blue data correspond to the 45 and 90 m.a.g.l inlet at RGL, respectively, and red, purple and green correspond to the 54, 100 and 185 m.a.g.l inlets at TAC. Shaded grey areas represent mean seasonal night-time based on the sites latitude and longitude. Data shown is in UTC.



Deleted:

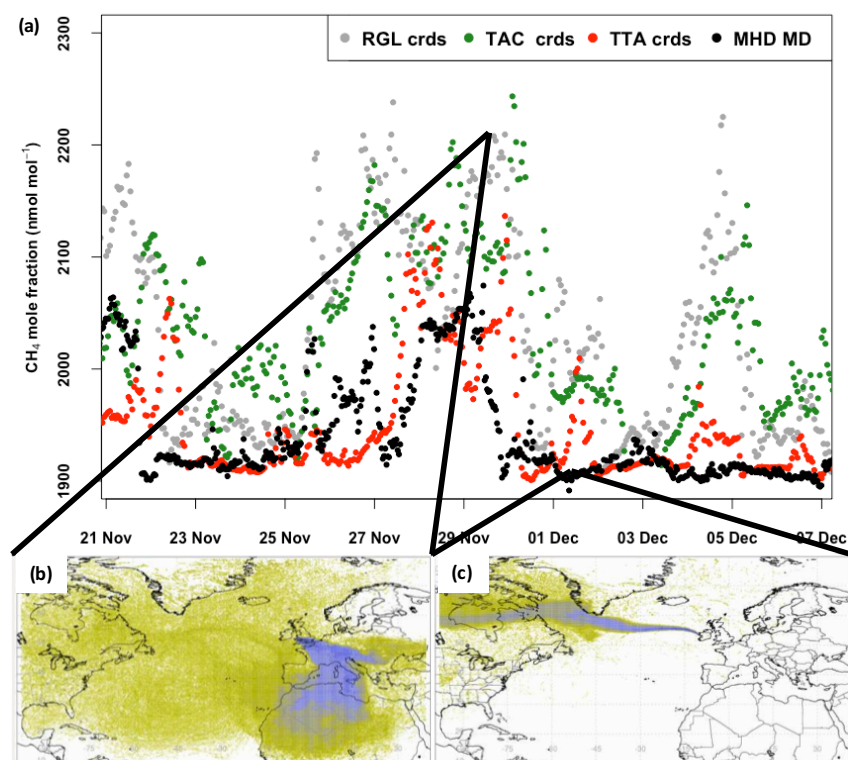


Figure 10: Examples of (a) CH₄ mole fractions from 21st November to 7th December 2014 (MHD MD discrete samples, CRDS values 20 minute averages), and 2-hour air history maps derived from NAME (b) for RGL, regionally polluted period, and (c) for MHD, baseline period. The air-history maps describe which surface areas (0-40m) in the previous 30-days impact the observation point within a particular 2-hour period.

Each of the UK DECC sites, excluding MHD, were equipped with CRDS instruments to measure CO₂ and CH₄ at ultra-high frequency (~ 3 seconds). RGL and TAC also had GC-ECD systems installed to measure N₂O and SF₆ at high frequency (10 minutes). TAC and MHD also had a RGA integrated with the GC-ECD system to measure CO and H₂. At MHD, a GC-ECD and –FID measure N₂O and CH₄. At two selected sites (MHD and TAC), a Medusa GC-MS system was set up to measure SF₆ and a wide range of GHGs and ODSs. All instruments are controlled by Linux based software, GCWerks, and are accessible remotely.

Regular standard tank analysis shows CRDS precisions (1 σ of 1-minute mean data of standard gas at ambient mole fractions) of < 0.03 $\mu\text{mol mol}^{-1}$ and < 0.2 nmol mol^{-1} for CO₂ and CH₄ respectively. GC-ECD instruments were also installed at RGL and TAC in January 2012 to measure N₂O and SF₆. Repeatability of 20 minute injections (1 σ) between January 2012 and September 2015 for N₂O and SF₆ measurements made using GC-ECD < 0.4 nmol mol^{-1} and < 0.07 pmol mol^{-1} respectively. Repeatability of 40/40/130 minute injections (1 σ) between January 2012 and September 2015 for CH₄, SF₆, CO and H₂ measurements made using GC-FID/RGA/Medusa GC-MS were < 2.8 nmol mol^{-1} , < 2 nmol mol^{-1} and < 3 nmol mol^{-1} and 0.34 pmol mol^{-1} , respectively.

Andrews, A. E., Kofler, J. D., Trudeau, M. E., Williams, J. C., Neff, D. H., Masarie, K. A., Chao, D. Y., Kitzis, D. R., Novelli, P. C., Zhao, C. L., Dlugokencky, E. J., Lang, P. M., Crotwell, M. J., Fischer, M. L., Parker, M. J., Lee, J. T., Baumann, D. D., Desai, A. R., Stanier, C. O., De Wekker, S. F. J., Wolfe, D. E., Munger, J. W., and Tans, P. P.: CO₂, CO, and CH₄ measurements from tall towers in the NOAA Earth System Research Laboratory's Global Greenhouse Gas Reference Network: instrumentation, uncertainty analysis, and recommendations for future high-accuracy greenhouse gas monitoring efforts, *Atmos. Meas. Tech.*, 7, 647-687, 10.5194/amt-7-647-2014, 2014.

Arnold, T., Mühle, J., Salameh, P. K., Harth, C. M., Ivy, D. J., and Weiss, R. F.: Automated Measurement of Nitrogen Trifluoride in Ambient Air, *Analytical Chemistry*, 84, 4798-4804, 10.1021/ac300373e, 2012.

Brailsford, G. W., Stephens, B. B., Gomez, A. J., Riedel, K., Mikaloff Fletcher, S. E., Nichol, S. E., and Manning, M. R.: Long-term continuous atmospheric CO₂ measurements at Baring Head, New Zealand, *Atmospheric Measurement Techniques*, 5, 3109-3117, 10.5194/amt-5-3109-2012, 2012.

Chen, H., Winderlich, J., Gerbig, C., Hofer, A., Rella, C. W., Crosson, E. R., Van Pelt, A. D., Steinbach, J., Kolle, O., Beck, V., Daube, B. C., Gottlieb, E. W., Chow, V. Y., Santoni, G. W., and Wofsy, S. C.: High-accuracy continuous airborne measurements of greenhouse gases (CO₂ and CH₄) using the cavity ring-down spectroscopy (CRDS) technique, *Atmospheric Measurement Techniques*, 3, 375-386, doi:10.5194/amt-3-375-2010, 2010, 2010.

Crosson, E. R.: A cavity ring-down analyser for measuring atmospheric levels of methane, carbon dioxide, and water vapor, *Applied Physics B*, 92, 403-408, doi: 10.1007/s00340-008-3135-y, 2008.

Cunnold, D. M., Weiss, R. F., Prinn, R. G., Hartley, D., Simmonds, P. G., Fraser, P. J., Miller, B., Alyea, F. N., and Porter, L.: GAGE/AGAGE measurements indicating reductions in global emissions of CCl₃F and CCl₂F₂ in 1992-1994, *Journal of Geophysical Research*, 102, 1259, doi: 10.1029/1296JD02973, 1997.

DEFRA: Waste water treatment in the United Kingdom-2012: Implementation of the European Union Urban Waste Water treatment directive 91/271/EEC, PB13811, Department for Environment, Food and Rural Affairs, London, 2012.

Dimmer, C. H., Simmonds, P. G., Nickless, G., and Bassford, M. R.: Biogenic fluxes of halomethanes from Irish peatland ecosystems, *Atmos. Environ.*, 35, 321-330, 2001.

Dlugokencky, E. J., Myers, R. C., Lang, P. M., Masarie, K. A., Crotwell, A. M., Thoning, K. W., Hall, B. D., Elkins, J. W., and Steele, L. P.: Conversion of NOAA atmospheric dry air CH₄ mole fractions to a gravimetrically prepared standard scale, *J. Geophys. Res.*, 110, doi:10.1029/2005JD006035, 2005.

Dlugokencky, E. J., Nisbet, E. G., Fisher, R., and Lowry, D.: Global atmospheric methane: budget, changes and dangers, *Philosophical Transactions of the Royal Society of London A: Mathematical, Physical and Engineering Sciences*, 369, 2058-2072, 10.1098/rsta.2010.0341, 2011.

EPRL: Energy Power Resources Ltd., location of power generation sites, <http://www.eprl.co.uk/assets/eye/overview.html>, 2013.

Ganesan, A. L., Chatterjee, A., Prinn, R. G., Harth, C. M., Salameh, P. K., Manning, A. J., Hall, B. D., Muhle, J., Meredith, L. K., Weiss, R. F., O'Doherty, S., and Young, D.: The variability of methane, nitrous oxide and sulfur hexafluoride in Northeast India, *Atmos. Chem. Phys.*, 13, 10633-10644, doi: 10.5194/acp-13-10633-2013, 2013.

Ganesan, A. L., Manning, A. J., Grant, A., Young, D., Oram, D. E., Sturges, W. T., Moncrieff, J. B., and O'Doherty, S.: Quantifying methane and nitrous oxide emissions from the UK and Ireland using a national-scale monitoring network, *Atmos. Chem. Phys.*, 15, 6393-6406, 10.5194/acp-15-6393-2015, 2015.

Gerbig, C., Lin, J. C., Wofsy, S. C., Daube, B. C., Andrews, A. E., Stephens, B. B., Bakwin, P. S., and Grainger, C. A.: Toward constraining regional-scale fluxes of CO₂ with atmospheric observations over a continent: 1. Observed spatial variability from airborne platforms *J. Geophys. Res.*, 108, 4756, doi:10.1029/2002JD003018, 2003.

Gerbig, C., Dolman, A. J., and Heimann, M.: On observational and modelling strategies targeted at regional carbon exchange over continents, *Biogeosciences*, 6, 1949-1959, doi:10.5194/bg-6-1949-2009, 2009.

Gloor, M., Bakwin, P., Hurst, D., Lock, L., Draxler, R., and Tans, P.: What is the concentration footprint of a tall tower?, *Journal of Geophysical Research: Atmospheres*, 106, 17831-17840, 10.1029/2001JD900021, 2001.

Grant, A., Stanley, K. F., Henshaw, S. J., Shallcross, D. E., and O'Doherty, S.: High-frequency urban measurements of molecular hydrogen and carbon monoxide in the United Kingdom, *Atmos. Chem. Phys.*, 10, 4715-4724, 10.5194/acp-10-4715-2010, 2010a.

Grant, A., Witham, C. S., Simmonds, P. G., Manning, A. J., and O'Doherty, S.: A 15 year record of high-frequency, in situ measurements of hydrogen at Mace Head, Ireland, *Atmos. Chem. Phys.*, 10, 1203-1214, 10.5194/acp-10-1203-2010, 2010b.

GRO: General Register Office for Scotland, <http://www.gro-scotland.gov.uk/>, 2013.

Hall, B. D., Dutton, G. S., and Elkins, J. W.: The NOAA nitrous oxide standard scale for atmospheric observations, *J. Geophys. Res.*, 112, doi:10.1029/2006JD007954, 2007.

Hall, B. D., Dutton, G. S., Mondeel, D. J., Nance, J. D., Rigby, M., Butler, J. H., Moore, F. L., Hurst, D. F., and Elkins, J. W.: Improving measurements of SF₆ for the study of atmospheric transport and emissions, *atmos. Meas. Tech.*, 4, 2441-2451, 2011.

Haszpra, L., Barcza, Z., Bakwin, P. S., Berger, B. W., Davis, K. J., and Weidinger, T.: Measuring system for the long-term monitoring of biosphere/atmosphere exchange of carbon dioxide, *J. Geophys. Res.*, 106, 3057-3069, doi:10.1029/2000jd900600, 2001.

Jones, A. R., Thomson, D. J., Hort, M., and Devenish, B.: The U.K. Met Office's next-generation atmospheric dispersion model, NAME III, in Borrego C. and Norman A.-L. (Eds), *Air Pollution Modeling and its Application XVII Proceedings of the 27th NATO/CCMS International Technical Meeting on Air Pollution Modelling and its Application*, 580-589, 2007.

Keeling, C. D., Bacastow, R. B., Bainbridge, A. E., Ekdahl, C. A., Guenther, P. R., Waterman, L. S., and Chin, J. F. S.: Atmospheric carbon dioxide variations at Mauna Loa Observatory, Hawaii, *Tellus*, 28, 538-551, 10.1111/j.2153-3490.1976.tb00701.x, 1976.

Kirschke, S., Bousquet, P., Ciais, P., Saunio, M., Canadell, J. G., Dlugokencky, E. J., Bergamaschi, P., Bergmann, D., Blake, D. R., Bruhwiler, L., Cameron-Smith, P., Castaldi, S., Chevallier, F., Feng, L., Fraser, A., Heimann, M., Hodson, E. L., Houweling, S., Josse, B., Fraser, P. J., Krummel, P. B., Lamarque, J.-F., Langenfelds, R. L., Le Quere, C., Naik, V., O'Doherty, S., Palmer, P. I., Pison, I., Plummer, D., Poulter, B., Prinn, R. G., Rigby, M., Ringeval, B., Santini, M., Schmidt, M., Shindell, D. T., Simpson, I. J., Spahni, R., Steele, L. P., Strode, S. A., Sudo, K., Szopa, S., van der Werf, G. R., Voulgarakis, A., van Weele, M., Weiss, R. F., Williams, J. E., and Zeng, G.: Three decades of global methane sources and sinks, *Nature Geosci.*, 6, 813-823, 10.1038/ngeo1955, 2013.

Kozlova, E. A., Manning, A. C., Kisilyakhov, Y., Seifert, T., and Heimann, M.: Seasonal, synoptic, and diurnal-scale variability of biogeochemical trace gases and O₂ from a 300-m tall tower in central Siberia, *Global Biogeochem. Cycles*, 22, doi:10.1029/2008GB003209, 2008.

Le Quéré, C., Moriarty, R., Andrew, R. M., Canadell, J. G., Sitch, S., Korsbakken, J. I., Friedlingstein, P., Peters, G. P., Andres, R. J., Boden, T. A., Houghton, R. A., House, J. I., Keeling, R. F., Tans, P., Arneeth, A., Bakker, D. C. E., Barbero, L., Bopp, L., Chang, J., Chevallier, F., Chini, L. P., Ciais, P., Fader, M., Feely, R. A., Gkritzalis, T., Harris, I., Hauck, J., Ilyina, T., Jain, A. K., Kato, E., Kitidis, V., Klein Goldewijk, K., Koven, C., Landschützer, P., Lauvset, S. K., Lefèvre, N., Lenton, A., Lima, I. D., Metz, N., Millero, F., Munro, D. R., Murata, A., Nabel,

J. E. M. S., Nakaoka, S., Nojiri, Y., O'Brien, K., Olsen, A., Ono, T., Pérez, F. F., Pfeil, B., Pierrot, D., Poulter, B., Rehder, G., Rödenbeck, C., Saito, S., Schuster, U., Schwinger, J., Séférian, R., Steinhoff, T., Stocker, B. D., Sutton, A. J., Takahashi, T., Tilbrook, B., van der Laan-Luijkx, I. T., van der Werf, G. R., van Heuven, S., Vandemark, D., Viovy, N., Wiltshire, A., Zaehle, S., and Zeng, N.: Global Carbon Budget 2015, *Earth Syst. Sci. Data*, 7, 349-396, 10.5194/essd-7-349-2015, 2015.

Levin, I., Naegler, T., Heinz, R., Osusko, D., Cuevas, E., Engel, A., Ilmberger, J., Langenfelds, R. L., Neininger, B., Rohden, C. v., Steele, L. P., Weller, R., Worthy, D. E., and Zimov, S. A.: The global SF₆ source inferred from long-term high precision atmospheric measurements and its comparison with emission inventories, *Atmos. Chem. Phys.*, 10, 2655-2662, doi: 10.5194/acp-10-2655-2010, 2010.

Luan, T., Fang, S., Yao, B., Wang, H., Dong, F., Shi, Q., Zhou, L., and Zhou, H.: Atmospheric molecular hydrogen (H₂) at the Shangdianzi regional background station in China, *Atmospheric Environment*, 145, 318-325, doi: 10.1016/j.atmosenv.2016.09.044, 2016.

Manning, A. J., O'Doherty, S., Jones, A. R., Simmonds, P. G., and Derwent, R. G.: Estimating UK methane and nitrous oxide emissions from 1990 to 2007 using an inversion modeling approach, *J. Geophys. Res.*, 116, D02305, 10.1029/2010jd014763, 2011.

Miller, B. R., Weiss, R. F., Salameh, P. K., Tanhua, T., Greally, B. R., Muhle, J., and Simmonds, P. G.: Medusa: A Sample Preconcentration and GC/MS Detector System for in Situ Measurements of Atmospheric Trace Halocarbons, Hydrocarbons, and Sulfur Compounds, *Anal. Chem.*, 80, 1536-1545, 2008.

Moore, F. L., Elkins, J. W., Ray, E. A., Dutton, G. S., Dunn, R. E., Fahey, D. W., McLaughlin, R. J., Thompson, T. L., Romashkin, P. A., Hurst, D. F., and Wamsley, P. R.: Balloonborne in situ gas chromatograph for measurements in the troposphere and stratosphere, *J. Geophys. Res.*, 108, 8330, 10.1029/2001jd000891, 2003.

NCC: Norfolk County Council, Landfill Sites location,, http://www.norfolk.gov.uk/Environment/Waste_and_Recycling/Waste_collection_and_disposal/Landfill_sites/index.htm, 2013.

Nevison, C. D., Dlugokencky, E., Dutton, G., Elkins, J. W., Fraser, P., Hall, B., Krummel, P. B., Langenfelds, R. L., O'Doherty, S., Prinn, R. G., Steele, L. P., and Weiss, R. F.: Exploring causes of interannual variability in the seasonal cycles of tropospheric nitrous oxide, *Atmos. Chem. Phys.*, 11, 3713-3730, 10.5194/acp-11-3713-2011, 2011.

ONS: Office for National Statistics; 2011 Census of England and Wales, 2012.

Popa, M. E., Gloor, M., Manning, A. C., Jordan, A., Schultz, U., Haensel, F., Seifert, T., and Heimann, M.: Measurements of greenhouse gases and related tracers at Bialystok tall tower station in Poland, *Atmos. Meas. Tech.*, 3, 407-427, doi:10.5194/amt-3-407-2010, 2010.

Prinn, R. G., Weiss, R. F., Fraser, P. J., Simmonds, P. G., Cunnold, D. M., Alyea, F. N., O'Doherty, S., Salameh, P., Miller, B. R., Huang, J., Wang, R. H. J., Hartley, D. E., Harth, C., Steele, L. P., Sturrock, G., Midgley, P. M., and McCulloch, A.: A history of chemically and radiatively important gases in air deduced from ALE/GAGE/AGAGE, *J. Geophys. Res.*, 105, doi:10.1029/2000JD900141, 2000.

Rella, C.: Accurate greenhouse gas measurements in humid gas streams using the Picarro G1301 carbon dioxide/methane/water vapor gas analyzer, White paper, Picarro Inc, Sunnyvale, CA, USA, 2010.

Rella, C., Chen, H., Andrews, A. E., Filges, A., Gerbig, C., Hatakka, J., Karion, A., Miles, N. L., Richardson, S. J., Steinbacher, M., Sweeney, C., Wastine, B., and Zellweger, C.: High accuracy measurements of dry mole fractions of carbon dioxide and methane in humid air, *Atmos. Meas. Tech.*, 6, 837-860, 10.5194/amt-6-837-2013, 2013.

Satar, E., Berhanu, T. A., Brunner, D., Henne, S., and Leuenberger, M.: Continuous CO₂/CH₄/CO measurements (2012–2014) at Beromünster tall tower station in Switzerland, *Biogeosciences*, 13, 2623-2635, 10.5194/bg-13-2623-2016, 2016.

Stocker, T. F., Qin, D., Plattner, G. K., Tignor, M., Allen, S. K., Boschung, J., Nauels, A., Xia, Y., Bex, V., and Midgley, P. M.: Climate Change 2013: The Physical Science Basis. Contribution of Working Group I to the Fifth Assessment Report of the Intergovernmental Panel on Climate Change, Cambridge University Press, Cambridge, United Kingdom and New York, NY, USA, 1535, doi:10.1017/CBO9781107415324, 2013.

Tans, P. P.: Observational strategy for assessing the role of terrestrial ecosystems in the global carbon cycle: scaling down to regional levels, in: *Scaling physiological processes: leaf to globe.*, edited by: Ebleringer, J. R., and Field, C. B., Academic Press, San Diego, California, USA, 179-190, 1993.

Thompson, R. L., Manning, A. C., Gloor, E., Schultz, U., Seifert, T., Hansel, F., and Jordan, A.: In-situ measurements of oxygen, carbon monoxide and greenhouse gases from Ochsenkopf tall tower in Germany, *Atmos. Meas. Tech.*, 2, 573-591, doi:10.5194/amt-2-573-2009, 2009.

Tremblay, A., Lambert, M., and Gagnon, L.: Do Hydroelectric Reservoirs Emit Greenhouse Gases?, *Environmental Management*, 33, S509-S517, 10.1007/s00267-003-9158-6, 2004.

Vardag, S. N., Hammer, S., O'Doherty, S., Spain, T. G., Wastine, B., Jordan, A., and Levin, I.: Comparisons of continuous atmospheric CH₄, CO₂ and N₂O measurements - results from a travelling instrument campaign at Mace Head, *Atmos. Chem. Phys.*, 14, 8403-8418, 10.5194/acp-14-8403-2014, 2014.

Vermeulen, A. T.: CHITTO Final Report, ECN energy research Centre of the Netherlands, 2007.

Vermeulen, A. T., Hensen, A., Popa, M. E., van den Bulk, W. C. M., and Jongejan, P. A. C.: Greenhouse gas observations from Cabauw Tall Tower (1992–2010), Atmos. Meas. Tech., 4, 617–644, 10.5194/amt-4-617-2011, 2011.

Winderlich, J., Chen, H., Gerbig, C., Seifert, T., Kolle, O., Lavric, J. V., Kaiser, C., Hofer, A., and Heimann, M.: Continuous low-maintenance CO₂/CH₄/H₂O measurements at the Zotino Tall Tower Observatory (ZOTTO) in Central Siberia, Atmos. Meas. Tech., 3, 1113–1128, doi: 10.5194/amt-3-1113-2010, 2010.

WMO-GAW: 17th WMO/IAEA Meeting on Carbon Dioxide, Other Greenhouse Gases and Related Tracers Measurement Techniques (GGMT-2013), Beijing, China, 10 - 13 June 2013, WMO, Geneva, Switzerland, 2014.

Zhao, C. L., and Tans, P. P.: Estimating uncertainty of the WMO mole fraction scale for carbon dioxide in air, J. Geophys. Res., 111, doi: 10.1029/2005JD006003, 2006.

Page 56: [3] Deleted		Kieran Stanley	13/01/2018 18:50:00	
CF ₄	Medusa	Medusa	-	-
NF ₃	Medusa	-	-	-
PFC-116	Medusa	Medusa	-	-
PFC-218	Medusa	Medusa	-	-
PFC-318	Medusa	Medusa	-	-
HFC-23	Medusa	Medusa	-	-
HFC-32	Medusa	Medusa	-	-
HFC-134a	Medusa	Medusa	-	-
HFC-152a	Medusa	Medusa	-	-
HFC-125	Medusa	Medusa	-	-
HFC-143a	Medusa	Medusa	-	-
HFC-227ea	Medusa	Medusa	-	-
HFC-236fa	Medusa	Medusa	-	-
HFC-43-10mee	Medusa	Medusa	-	-
HFC-365mfc	Medusa	Medusa	-	-
HCFC-22	Medusa	Medusa	-	-
HCFC-141b	Medusa	Medusa	-	-
HCFC-142b	Medusa	Medusa	-	-
HCFC-124	Medusa	Medusa	-	-
CFC-11	Medusa	Medusa	-	-
CFC-12	Medusa	Medusa	-	-
CFC-13	Medusa	Medusa	-	-
CFC-113	Medusa	Medusa	-	-
CFC-114	Medusa	Medusa	-	-
CFC-115	Medusa	Medusa	-	-
H-1211	Medusa	Medusa	-	-
H-1301	Medusa	Medusa	-	-
H-2402	Medusa	Medusa	-	-
CH ₃ Cl	Medusa	Medusa	-	-
CH ₃ Br	Medusa	Medusa	-	-
CH ₃ I	Medusa	Medusa	-	-
CH ₂ Br ₂	Medusa	Medusa	-	-
CHCl ₃	Medusa	Medusa	-	-
CHBr ₃	Medusa	Medusa	-	-
CCl ₄	Medusa	Medusa	-	-
CH ₃ CCl ₃	Medusa	Medusa	-	-
CHCl=CCl ₂	Medusa	Medusa	-	-
CCl ₂ =CCl ₂	Medusa	Medusa	-	-