



# 1 Improved source apportionment of organic aerosols in complex 2 urban air pollution using the multilinear engine (ME-2)

3 Qiao Zhu<sup>1</sup>, Xiao-Feng Huang<sup>1,\*</sup>, Li-Ming Cao<sup>1</sup>, Lin-Tong Wei<sup>1</sup>, Bin Zhang<sup>1</sup>, Lin-Yan He<sup>1</sup>, Miriam Elser<sup>2</sup>, Francesco  
4 Canonaco<sup>2</sup>, Jay G. Slowik<sup>2</sup>, Carlo Bozzetti<sup>2</sup>, Imad El-Haddad<sup>2</sup>, and André S.H. Prévôt<sup>2</sup>

5 <sup>1</sup>Key Laboratory for Urban Habitat Environmental Science and Technology, School of Environment and Energy, Peking  
6 University Shenzhen Graduate School, Shenzhen, 518055, China.

7 <sup>2</sup>Paul Scherrer Institute (PSI), 5232 Villigen-PSI, Switzerland

8 **Abstract** Organic aerosols (OAs), which consist of thousands of complex compounds emitted from various sources,  
9 constitute one of the major components of fine particulate matter. The traditional positive matrix factorization (PMF) method  
10 often apportions aerosol mass spectrometer (AMS) organic datasets into less meaningful or mixed factors, especially in  
11 complex urban cases. In this study, an improved source apportionment method using a bilinear model of the multilinear  
12 engine (ME-2) was applied to OAs collected during the heavily polluted season from two Chinese megacities located in the  
13 north and south with an Aerodyne high-resolution aerosol mass spectrometer (HR-ToF-AMS). We applied a rather novel  
14 procedure for utilization of prior information and selecting optimal solutions. Ultimately, six reasonable factors were clearly  
15 resolved and quantified for both sites by constraining one or more factors: hydrocarbon-like OA (HOA), cooking-related OA  
16 (COA), biomass burning OA (BBOA), coal combustion (CCOA), less-oxidized oxygenated OA (LO-OOA) and more-  
17 oxidized oxygenated OA (MO-OOA). In comparison, the traditional PMF method could not effectively resolve the  
18 appropriate factors, e.g., BBOA and CCOA, in the solutions. Moreover, coal combustion and traffic emissions were  
19 determined to be primarily responsible for the concentrations of PAHs and BC, respectively, through the regression analyses  
20 of the ME-2 results.

## 21 1 Introduction

22 Atmospheric aerosols are generating increasing interest due to their adverse effects on human health, visibility and the  
23 climate (IPCC, 2013; Pope and Dockery, 2006). Among different particulate compositions, many studies focus on organic  
24 aerosols (OAs) because they contribute 20-90% to the total submicron mass (Jimenez et al., 2009; Zhang et al., 2007). OAs  
25 can be either directly emitted by various sources, including anthropogenic (i.e., traffic and combustion activities) and  
26 biogenic sources, or produced via secondary formation after the oxidation of volatile organic compounds (VOCs) (Hallquist

---

<sup>1</sup> Correspondence to: X.-F. Huang ([huangxf@pku.edu.cn](mailto:huangxf@pku.edu.cn))



27 et al., 2009). Therefore, the reliable source identification and quantification of OAs are essential before developing effective  
28 political abatement strategies.

29 Aerodyne aerosol mass spectrometer (AMS) systems are the most widely adopted on-line aerosol measurement systems  
30 for acquiring aerosol chemical compositions (Canagaratna et al., 2007; Pratt and Prather, 2012). An AMS provides on-line  
31 quantitative mass spectra of non-refractory components from the submicron aerosol fraction with a high temporal resolution  
32 (i.e., seconds to minutes) (Canagaratna et al., 2007). The total mass spectra can be assigned to both several inorganic  
33 compounds and the organic fraction through mass spectral fragmentation tables (Allan et al., 2004). To further investigate the  
34 different types of organic fractions, numerous studies have exploited the positive matrix factorization (PMF) algorithm and  
35 apportioned the AMS organic mass spectra in terms of their source emissions or formation processes (Zhang et al., 2011).  
36 PMF is a standard multivariate factor analysis tool (Paatero, 1999; Paatero and Tapper, 1994) that models the time series of  
37 measured organic mass spectra as a linear combination of positive factor profiles and their respective time series. Most of the  
38 earlier PMF studies were conducted on unit-mass resolution (UMR) mass spectrometers (Lanz et al., 2007; Lanz et al., 2010;  
39 Ulbrich et al., 2009), although more have recently focused on high-resolution (HR) mass spectra PMF (Aiken et al., 2009;  
40 Docherty et al., 2008; Huang et al., 2010). The use of HR mass spectra data to constrain PMF solutions can reduce their  
41 rotational ambiguity and result in more interpretable OA factors. For example, Aiken et al. (2009) found that hydrocarbon-  
42 like OA (HOA) and biomass burning OA (BBOA) were better separated using HR-AMS data than with UMR data. However,  
43 even HR-AMS-PMF can also yield mixed factors (especially in heavily polluted areas) due to their complex emission  
44 patterns.

45 The abundant characteristic fragments for cooking-related OA (COA) (e.g.,  $m/z$  55 and 57) and coal combustion OA  
46 (CCOA) (e.g.,  $m/z$  51, 53, and 65) can be observed in the mass spectrum of the HOA factor (He et al., 2010; Hu et al., 2013).  
47 Elser et al., 2016 analyzed two urban HR-AMS datasets in China, and their PMF results showed an HOA profile that  
48 contained a high concentration of  $C_2H_4O_2^+$  ( $m/z$  60), which is a BBOA tracer ion. In addition,  $CO_2^+$  ( $m/z$  44) contributed  
49 more to COA compared to oxygenated OA (OOA). To solve this “mixed factor” problem in PMF analysis, some researchers  
50 attempted to use the multilinear engine algorithm (ME-2) with user-provided constraints (Canonaco et al., 2013; Crippa et al.,  
51 2014; Elser et al., 2016; Reyes-Villegas et al., 2016). However, several key issues with the ME-2 in these studies, such as  
52 reliability of the user-input constraints and the criteria used to determine an optimal result, still require further investigation.  
53 Most ME-2 studies (Crippa et al., 2014; Elser et al., 2016; Reyes-Villegas et al., 2016) were based on HR-AMS datasets and  
54 utilized mass profiles of PMF results from Paris (mostly due to the lack of other reliable source profiles) and did not consider  
55 the specific sampling sites, which could result in uncertainties.

56 In this study, a novel source apportionment technique using the multi-linear engine tool (ME-2) was successfully  
57 applied to organic mass spectra obtained with an HR-ToF-AMS at two urban sites during pollution-heavy periods during the  
58 same year. The improved OA source apportionment results are discussed and compared with an unconstrained PMF analysis.

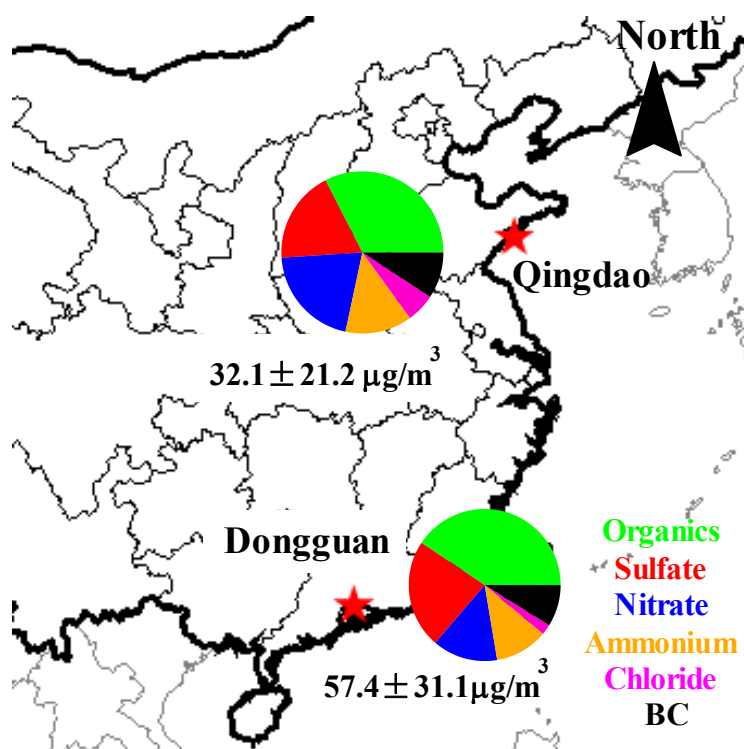


## 59 2 Materials and methods

### 60 2.1 Sampling sites and period

61 Measurements at Qingdao (36.10°N, 120.47°E, 10 m above ground level, a.g.l.) were performed from 1 to 31  
62 November 2013, while those in Dongguan were conducted from 12 December 2013 to 1 January 2014 (33.03°N, 113.75°E,  
63 100 m a.g.l.). Qingdao is a coastal city with over 9 million inhabitants in northern China, while Dongguan has over 8 million  
64 inhabitants and is located in southern China (shown in Figure 1). Both of the sampling sites are on the tops of buildings in  
65 urban central areas. The surroundings include some agricultural counties, and thus, the sites are influenced by not only local  
66 urban emissions but also biomass burning from nearby farmlands.

67



68

69 **Figure 1.** The locations and the average PM<sub>1</sub> chemical compositions of the Qingdao and Dongguan sampling sites.

### 70 2.2 Instrumentation

71 An HR-ToF-AMS was deployed for the on-line measurement of non-refractory PM<sub>1</sub> (Canagaratna et al., 2007). The  
72 setup and operation of the HR-ToF-AMS was similar to that in our previous studies (Huang et al., 2015; Huang et al., 2010).  
73 A PM<sub>2.5</sub> cyclone inlet was briefly placed on the roof of a building to remove coarse particles and to introduce an air stream  
74 containing the remaining particles into a room through a copper tube with a flow rate of 10 l min<sup>-1</sup>. A nafion dryer (MD-070-



75 12S-4, Perma Pure Inc.) was positioned upstream of the HR–ToF–AMS to eliminate the potential influence of relative  
76 humidity on the particle collection (Matthew et al., 2008), after which the HR–ToF–AMS isokinetically sampled from the  
77 center of the copper tube at a flow rate of 80 ml min<sup>-1</sup>. The instrument was operated at two ion optical modes with a cycle of  
78 4 min, including 2 min for the mass-sensitive V-mode and 2 min for the high mass resolution W-mode. An aethalometer  
79 (AE-31, Magee), which also has a PM<sub>2.5</sub> inlet, was simultaneously used for measurements of refractory black carbon (BC)  
80 with a temporal resolution of 5 min.

81 A routine analysis of the HR–ToF–AMS data was performed using the software SQUIRREL (version 1.57) and PIKA  
82 (version 1.16) written in Igor Pro 6.37 (Wave Metrics  
83 Inc.)(<http://cires1.colorado.edu/jimenezgroup/ToFAMSResources/ToFSoftware/index.html>). The ionization efficiency (IE)  
84 was calibrated using pure ammonium nitrate particles following standard protocols (Drewnick et al., 2005; Jayne et al.,  
85 2000). The relative IEs (RIEs) for organics, nitrate and chloride were assumed to be 1.4, 1.1 and 1.3, respectively. A  
86 composition-dependent collection efficiency (CE) was applied to the data based on the method of Middlebrook et al. (2012)  
87 and an organic elemental analysis was performed using the latest approach recommended by Canagaratna et al. (2015). The  
88 mass concentrations of polycyclic aromatic hydrocarbons (PAHs) were quantitatively determined from the HR-AMS data  
89 using the method of Brunts et al. (2015).

### 90 **2.3 PMF and ME-2 methods for OA source apportionment**

91 PMF is a mathematical technique used to solve bilinear unmixing problems (Paatero and Tapper, 1994) that enables a  
92 description of the variability of a multivariate database as the linear combination of static factor profiles and their  
93 corresponding time series. The bilinear factor analytic model in matrix notation is defined in Eq. (1), where the measured  
94 matrix X (consisting of i rows and j columns) is approximated by the product of G (containing the factor time series) and F  
95 (the factor profiles). E denotes the model residuals. The entries in G and F are fitted using a least-squares algorithm that  
96 iteratively minimizes the quantity Q (Eq. 2), which is defined as the sum of the squared residuals ( $e_{ij}$ ) weighted by their  
97 respective uncertainties ( $\sigma_{ij}$ ).

$$X = G \times F + E \quad (1)$$

$$Q = \sum_{i=1}^m \sum_{j=1}^n \left( \frac{e_{ij}}{\sigma_{ij}} \right)^2 \quad (2)$$

98 In this study, we adopted SoFi (Canonaco et al., 2013), which is an implementation of the multilinear engine (ME-2)  
99 (Paatero, 1999), to perform the organic HR-AMS data analysis. In contrast to an unconstrained PMF analysis, ME-2 enables  
100 a more complete exploration of the rotational ambiguity of the solution space. In our case, this is achieved by directing the  
101 solution towards environmentally meaningful rotations using the *a* value approach. This method uses prior input profiles and  
102 the scalar *a* to constrain one or more output factor profiles such that they fall within a predetermined range. The *a* value  
103 determines the extent to which the output profiles are allowed to vary from the input profiles according to Eq. (3), where *f*  
104 represents the factor profile and *j* indicates the *m/z* of the ions.



$$f_{j,\text{solution}} = f_j \pm a \times f_j \quad (3)$$

The number of output factors, which is selected by the user, is a key consideration for PMF analysis. Most unconstrained PMF results were chosen following the procedures detailed in Zhang et al. (2007). However, additional outputs in ME-2 can be generated to explore more of the solution space, and more criteria should be developed to support the factor identification, which will be discussed in section 3.

### 3 Results and discussion

In this section, a conventional PMF without any prior information is performed to analyze the OA sources. Then, we use the ME-2 method to optimize the OA source apportionment based on the information obtained from the PMF method. Finally, the improved source apportionment results derived using ME-2 are further discussed and analyzed.

#### 3.1 OA source apportionment using an unconstrained PMF method

We performed unconstrained runs with a range from two to ten factors. Generally, PMF solutions with large numbers of factors are not considered due to possible mathematical splits of the factor profiles. However, some factors that have small contributions or that have similar mass profiles as other factors (but different time series) may only be found in solutions with large numbers of factors. We observe that most of the solutions provided via PMF include either multiply split factors or mixed factors that are not properly separated from one another. In other words, PMF does not produce an appropriate solution. The 6-factor solutions for Qingdao and Dongguan are shown in Figure S1 and S2, and three types of primary OAs (POAs) were identified for each sampling site, including HOA, coal combustion OA (CCOA) and cooking OA (COA) for Qingdao and HOA, biomass burning OA (BBOA) and COA for Dongguan. Oxygenated OA (OOA) seems to be excessively split in the 6-factor solutions for both of the sites. HOA is distinguished by alkyl fragment signatures with prominent contributions of  $m/z$  55 ( $C_4H_7^+$ ) and  $m/z$  57 ( $C_4H_9^+$ ) (Ng et al., 2011). The COA profile is similar to that of HOA but has a higher contribution from oxygenated ions at  $m/z$  55 ( $C_3H_3O^+$ ) and  $m/z$  57 ( $C_3H_5O^+$ ) (Mohr et al., 2012). BBOA is characterized by the presence of signals at  $m/z$  60 ( $C_2H_4O_2^+$ ) and  $m/z$  73 ( $C_3H_5O_2^+$ ), which are identified as fragments from anhydrous sugars present in biomass smoke (Alfarra et al., 2007). The OOA profile is characterized by a high signal at  $m/z$  44 ( $CO_2^+$ ). Note that some POA profiles in this solution indicate mixing; for example, CCOAs in Qingdao contain a high concentration of the biomass burning tracer ion ( $m/z$  60,  $C_2H_4O_2^+$ ), and HOAs in Dongguan have a higher-than-expected contribution of  $m/z$  44 ( $CO_2^+$ ) with a high O/C ratio (0.26). In addition, CCOA seems to be mixed with BBOA. We then further verified the solutions with additional factors. The results show that BBOA and CCOA are separated from each other in the 7- and 8-factor solutions for Qingdao (see Figure S1) and that better signals for unmixed and stable HOA with low O/C ratios of 0.17 or 0.18 emerged in the 7- to 10-factor solutions for Dongguan (see Figure S2).

#### 3.2 Improved OA source apportionment using the ME-2 method

Before operating ME-2, feasible and reasonable prior input profiles must be determined. To the best of our knowledge, this is the first HR-OA data set that employs anchor profiles extracted from an unconstrained PMF solution with a higher

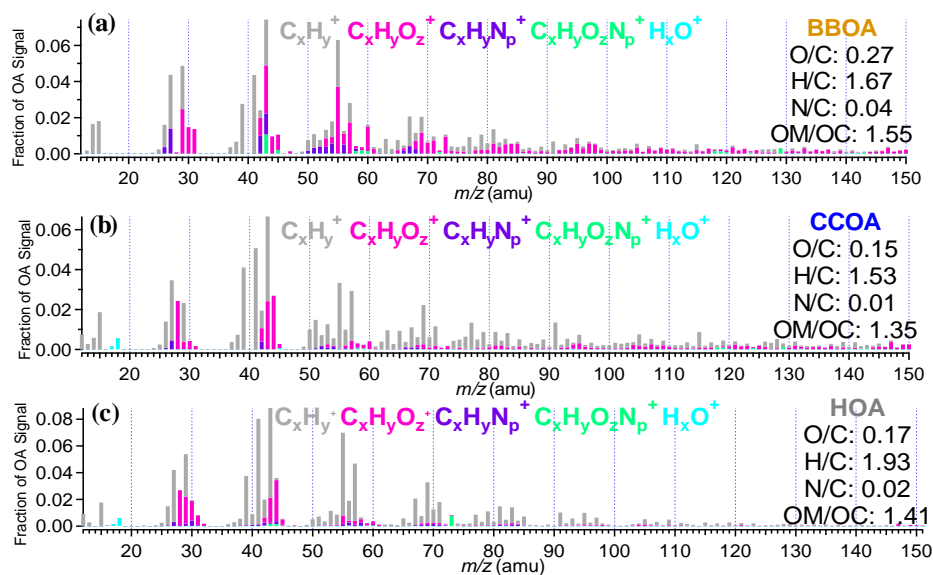


136 number of factors, and the same approach has been successfully applied to source apportionment efforts using UMR ME-2  
137 (Fröhlich et al., 2015). In our case for Qingdao, the BBOA factors from the 7- and 8-factor solutions may be used as anchor  
138 profiles. Although these two BBOA factors are quite similar, the BBOA from the 8-factor solution is better suited to be a  
139 constraining profile due to its smaller  $m/z$  44 ( $\text{CO}_2^+$ ) signal and higher  $m/z$  60 ( $\text{C}_2\text{H}_4\text{O}_2^+$ ) signal (see Figure S3). In addition,  
140 the BBOA from the 8-factor solution also correlates better with the BBOA from a Chinese biomass burning simulation  
141 ( $R^2=0.81$ ) than the 7-factor solution ( $R^2=0.79$ ) (He et al., 2010). Moreover, the best interpretable results from the data set are  
142 the 6-factor solutions with factors that include HOA, COA, BBOA, CCOA, less-oxidized oxygenated OA (LO-OOA) and  
143 more-oxidized oxygenated OA (MO-OOA) (Figure 3a). Considering  $a$  values between 0 and 1 with a step of 0.1 for BBOA  
144 yields 11 possible solutions, and some criteria were established to obtain a better environmental OA source apportionment.  
145 In this study, we used two simple and reasonable criteria: the reasonability of the O/C ratio and the correlation between the  
146 factors and the tracers. The O/C ratios for six resolved factors and the correlations between CCOA and PAHs, HOA and BC  
147 for 11 solutions with different  $a$  values are shown in Table S2. These results indicate that all of the O/C ratios for each factor  
148 and each factor-tracer correlation are quite similar to one another and that they agree with the range of values in the literature  
149 (Canagaratna et al., 2015). Therefore, the solutions averaged over the 11 outputs were considered the final results for  
150 Qingdao.

151 The anchor profile for HOA for Dongguan can be obtained from unconstrained PMF solutions. The averaged HOA  
152 profile from the 7- to 10-factor solutions was used as the anchor profile for ME-2 due to the small differences among the  
153 different solutions. Additionally, the constraining CCOA profile for Dongguan is still under consideration because the mass  
154 spectrum of BBOA was found to be very similar to that of CCOA, raising the concern that coal combustion particles might  
155 have been incorrectly apportioned to biomass burning sources (Wang et al., 2013). Furthermore, an appropriate CCOA  
156 anchor profile could not be obtained due to an increase in the unconstrained PMF factor number (see Figure S2) and because  
157 few studies have reported a comparison of the CCOA profile to other PMF factors. The best approach is to employ the  
158 CCOA profile from Qingdao as the constraining profile for Dongguan in ME-2, as these two campaigns were conducted  
159 using the same HR-ToF-AMS in the same year and because coal combustion is a significant source of OAs in Qingdao due  
160 to domestic heating during the wintertime. In addition, the CCOA from Qingdao has a very good correlation ( $R^2=0.97$ ) with  
161 CCOA profiles reported at other Chinese urban sites (Elser et al., 2016) (see Figure S4). The input profiles for HOA and  
162 CCOA in Dongguan prior to operating ME-2 are shown in Figure 2. The  $a$  values were set from 0 to 1 with an increment of  
163 0.1 for both HOA and CCOA. All of the O/C ratios for HOA, CCOA, COA and BBOA among the 121 possible solutions are  
164 listed in Table S3. The O/C ratio of HOA in the unconstrained PMF results remained between approximately 0.17 and 0.18,  
165 providing a filter criterion with which to assess reasonable ME-2 solutions, and only solutions with  $a$  values between 0 and  
166 0.1 fell into this range (Table S1). The O/C ratios of other factors for  $a$  values between 0 and 0.1 are shown in Table S3. The  
167 solutions using  $a$  values between 0 and 0.1 for the HOA profile and an  $a$  value of 0.9 for the CCOA profile are considered  
168 ideal results for three reasons. First, unlike the HOA mass spectra, CCOAs from different sites show higher variability and  
169 the CCOA anchor profile is not derived from itself, and therefore, it is reasonable to restrict the constraint with small  $a$



170 values for HOA and a looser constraint should be applied for CCOA, which is consistent with the  $a$  values selecting rules in  
 171 London ME-2 study (Reyes-Villegas et al., 2016). Second, the POA factors in Dongguan, including HOA and CCOA, have  
 172 higher O/C ratios likely as a result of a higher atmospheric oxidizing capacity and a stronger photochemical formation in  
 173 Southern China (Hofzumahaus et al., 2009). Moreover, some studies reported that BBOAs undergo substantial chemical  
 174 processing immediately after emission and that aged BBOAs had significant concentrations in fresh plumes (Zhou et al.,  
 175 2017). Thus, CCOAs in Dongguan are very likely to demonstrate relatively higher ages than those in Qingdao (0.15) with  
 176 higher O/C ratios (but with an O/C ratio of up to 1.25 when the  $a$  value is 1, which is unacceptable). Third, with an increase  
 177 in the  $a$  value for CCOA, two types of OOAs become more distinctive, and the factor correlates better with the tracer (Table  
 178 S1 and Table S4).  
 179



180  
 181 **Figure 2.** The anchor mass spectra for (a) BBOA, (b) CCOA and (c) HOA in the ME-2 analysis.

### 182 3.3 Variations in the OA factors

183 Figure 1 shows the chemical compounds of  $PM_{10}$ , including the non-refractory (NR) components measured via HR-  
 184 AMS (i.e., OA,  $SO_4$ ,  $NO_3$ ,  $NH_4$  and Cl) and BC concentrations measured via the AE-31, during the sampling period in both  
 185 Qingdao and Dongguan. The average  $PM_{10}$  mass concentration was  $32.1 \pm 21.2 \mu\text{g}/\text{m}^3$  (mean  $\pm$  standard deviation) in Qingdao  
 186 and  $57.4 \pm 31.1 \mu\text{g}/\text{m}^3$  in Dongguan. The temporal variations in the  $PM_{10}$  species in conjunction with meteorological  
 187 parameters are shown in Figure S5. Although Dongguan is located in southern China with relatively less air pollution  
 188 (Huang et al., 2012), the  $PM_{10}$  mass concentration was higher. This is mainly because of stagnant meteorological conditions  
 189 with low average wind speeds (i.e., 2.3 m/s) and a maximum wind speed of less than 6 m/s. Among the  $PM_{10}$  compounds,



190 OAs accounted for 32.5% of  $PM_{10}$  in Qingdao and 40.6% in Dongguan. This suggests that OA constitutes a very important  
191 fraction at both urban sites. Thus, the final and detailed results of the OA source apportionment are presented in this section.  
192 For Qingdao, the final result is the average of all of the ME-2 runs with constraints including  $a$  values from 0 to 1 fulfilling  
193 the criteria described in section 3.2. The mass spectra and time series of the resolved OA sources are shown in Figure 3a.  
194 The characteristics of each factor were distinct. The BBOA profile contained the highest  $m/z$  60 fraction  $f_{60}$  (1.5%)  
195 compared to the other factors, and the concentrations were highly correlated with  $C_2H_4O_2^+$  ( $R^2=0.81$ ). The mass spectra of  
196 COA was characterized by a high  $m/z$  55/57 ratio, which is consistent with previous results (He et al., 2010; Mohr et al.,  
197 2012; Sun et al., 2016). In addition, the time series of COA showed a good correlation with its tracer ion  $C_6H_{10}O^+$  in  
198 accordance with (Sun et al., 2016). HOAs were correlated well with BC ( $R^2=0.65$ ), and CCOAs were highly correlated with  
199 PAHs ( $R^2=0.94$ ). Among the two types of OOAs, the less-oxidized OOA (LO-OOA) had a lower  $CO_2^+$  fraction and O/C  
200 ratio (0.62) compared with the more oxidized OOA (MO-OOA), which had a higher  $CO_2^+$  fraction and O/C (0.91) ratio. The  
201 sum of LO-OOA and MO-OOA showed a high correlation with the sum of sulfate and nitrate ( $R^2=0.76$ ). The POAs  
202 (including HOA, COA, BBOA and CCOA) contributed 53.4% to the OA concentration (Figure 3a), which was almost equal  
203 to the SOA fraction. In terms of the diurnal trends of the OA factors shown in Figure 3a, they are all partially driven both by  
204 PBL dynamics (demonstrating an increased dilution during the daytime and an accumulation of particulate matter overnight)  
205 and by the diurnal emission profile. The diurnal trend of HOA showed pronounced peaks during the morning and evening  
206 rush hours (8:00-9:00 and 19:00-21:00), which is typically the case for traffic-related pollutants. COA shows a very distinct  
207 daily trend with strong peaks during the lunch (approximately 12:00) and dinner (19:00-20:00) periods. CCOAs constituted  
208 an important and dominant source of pollutants during the wintertime in northern Chinese areas (Elser et al., 2016) due to  
209 heating activities, especially with regard to the central-heating supply that began on November 13 and continued until the  
210 end of the campaign. The diurnal variations of the four POA factors before and during the central-heating period are shown  
211 in Figure S6. In comparison with the other three POAs, the diurnal pattern of CCOA showed a clear increase during the  
212 central-heating period with concentration peaks during the morning (at approximately 9:00) and at night (starting to rise at  
213 18:00), which seems consistent with heating emissions and atmospheric dilution. The diurnal trends of BBOA were similar  
214 to those of CCOA. The dilution of these particles within a deeper PBL during the daytime resulted in a decreasing trend in  
215 the BBOA concentration, while peaks related to residential heating were observed during the morning (between 09:00 to  
216 10:00) and at night (starting to rise at 17:00). The main difference between the LO-OOA and MO-OOA diurnal patterns is  
217 that an increase in the MO-OOA mass concentration was observed during the daytime, implying that the formation of  
218 secondary organic aerosols was greatly enhanced during the afternoon. In addition, the diurnal cycle for LO-OOA showed a  
219 relatively smaller decrease during the daytime compared with the POA factors. These characteristics of the OOA diurnal  
220 trend confirm their secondary nature.

221 For Dongguan, similar to the OA source apportionment using ME-2 in Qingdao, the final result is the average of two  
222 accepted  $a$ -value solutions with six identified factors, including HOA, CCOA, COA, BBOA, LO-OOA and MO-OOA. All of  
223 the information regarding the final source results is shown in Figure 3b. Good correlations between each OA factor and their

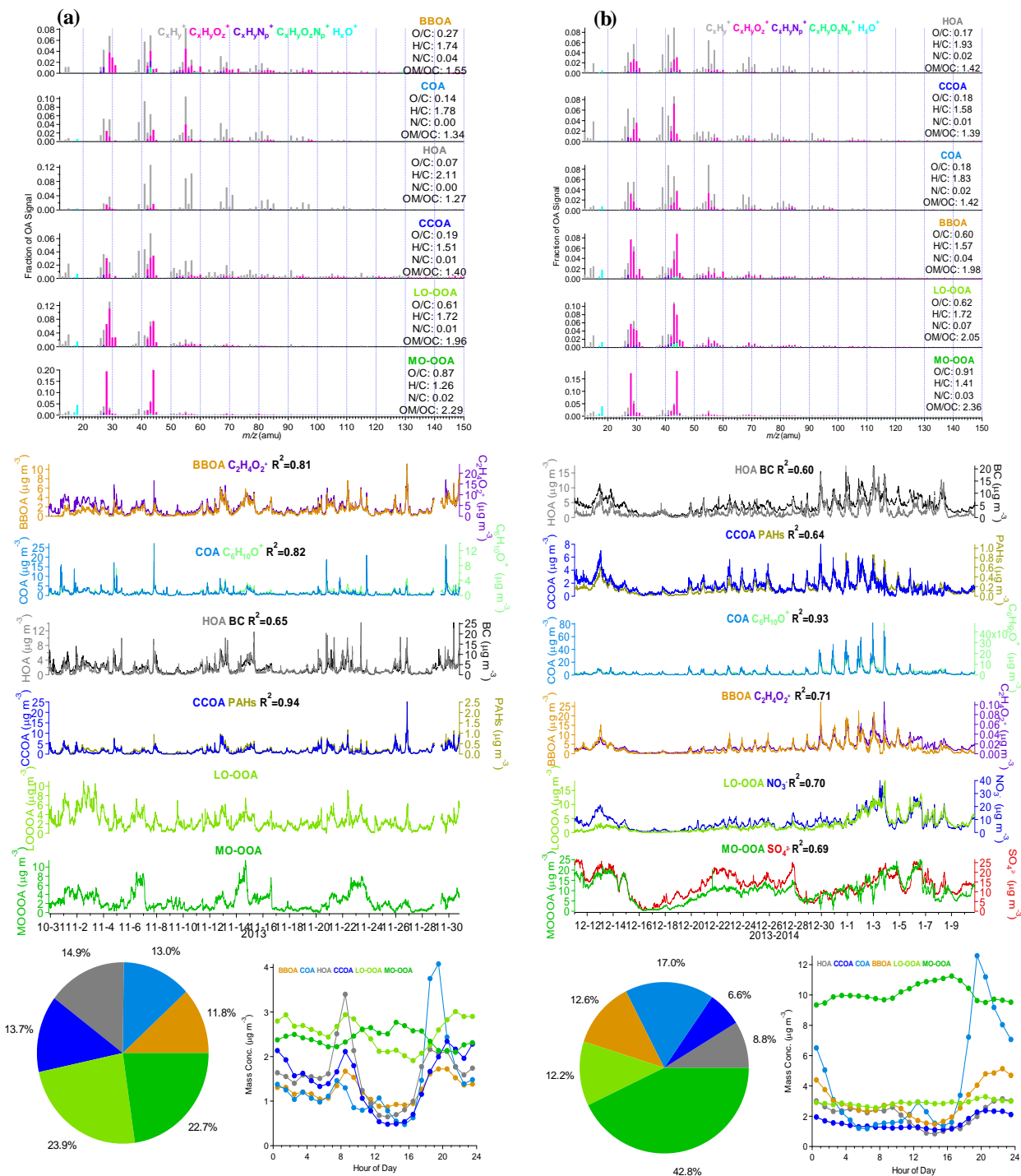




224 tracers indicate that the resolved ME-2 results are reasonable. Note that a few sharp drops (which always occurred at  
225 approximately 20:00) were observed in the MO-OOA time series ranging from December 29 to January 5, which coincides  
226 with extreme organic aerosol pollution (Figure S5). The inherent mechanisms for these drops remain unexplained, although  
227 we have tried a number of reasonable approaches (e.g., splitting the period into sub-periods to identify the sources,  
228 constraining more factors before running ME-2, and examining more factors) to address this issue. A similar problem in the  
229 MO-OOA time series was also found in a recent ME-2 application (Qin et al., 2017). In our case, we presume this might be  
230 the result of relatively worse meteorological conditions at night during the sampling period, thereby increasing the  
231 contribution of late supper emissions and leading to the overestimation of COAs offset by drops in the MO-OOA  
232 concentration. Also note that the O/C ratios of the POAs in Dongguan were higher than those in Qingdao, suggesting that  
233 POA emissions in Dongguan underwent faster chemical processing. In addition, the relatively smaller contributions of POAs  
234 further support this inference. Freshly emitted POAs may get mixed with aged OAs more easily, while ME-2 may still  
235 consider them unmixed. MO-OOAs accounted for an average of 42.8% of the total OA mass (which is much greater than the  
236 contribution of LO-OOAs), which is probably because some POA species could have been rapidly converted to very aged  
237 OOs (Bougiatioti et al., 2014; Xu et al., 2015). As mentioned above, the characteristics of the diurnal trends of the POA  
238 factors in Dongguan were similar to those in Qingdao, and thus, we focused on the OOA factors. MO-OOAs still showed  
239 higher concentrations during the daytime but, unlike LO-OOAs in Qingdao, the diurnal patterns of LO-OOAs in Dongguan  
240 were flat, implying that secondary OA formation in the LO-OOAs basically offset the influences of PBL variations.

241 Meteorological conditions (especially wind) play a crucial role in the dilution and transport of air pollution. We used  
242 the relationships between the component concentrations and wind to profoundly understand the origins of the OA factors and  
243 their nature. The distributions of the OA factor concentrations versus the wind direction and speed are plotted in Figure S7.  
244 For both of the urban sites, higher mass concentrations of the POA factors were mostly accompanied by low wind speeds,  
245 denoting their local emission characteristics. Additionally, for the OOA factors, a large proportion of their higher  
246 concentrations were maintained at higher wind speeds, indicating that the OOs were formed by transport processes.  
247 However, the small fraction of high-level OOs that was concentrated within the low wind-speed region represents the fast  
248 formation of OOs from some local POA.

249  
250



**Figure 3.** Mass spectra of the OA factors, average fractions of the OA factors, diurnal variations of the OA factors and time series of the OA factors identified by the ME-2 method for (a)Qingdao and (b) Dongguan.

251  
 252  
 253



254

255

### 3.4 Regression analysis for POA tracers

256

257

258

259

260

261

262

BC and PAHs are mainly derived from incomplete combustion processes (Schmidt and Noack, 2000; White, 1985), and thus, they were used as tracers for the POAs. In this study, the BC was directly measured by the AE-31, and the PAHs were quantified using the method developed by Bruns et al. (2015) based on AMS data. Both the BC and PAHs showed pronounced diurnal cycles similar to those of the POAs (see Figure S8). In addition, POAs are properly split into different subtypes via the ME-2 method, thereby providing the possibility to better understand the contributions of different POAs to BC and PAHs and to verify the POA source identification. In this section, we use a multi-linear regression method to analyze the POA factors for BC and PAHs.

263

264

265

266

267

268

269

270

271

272

273

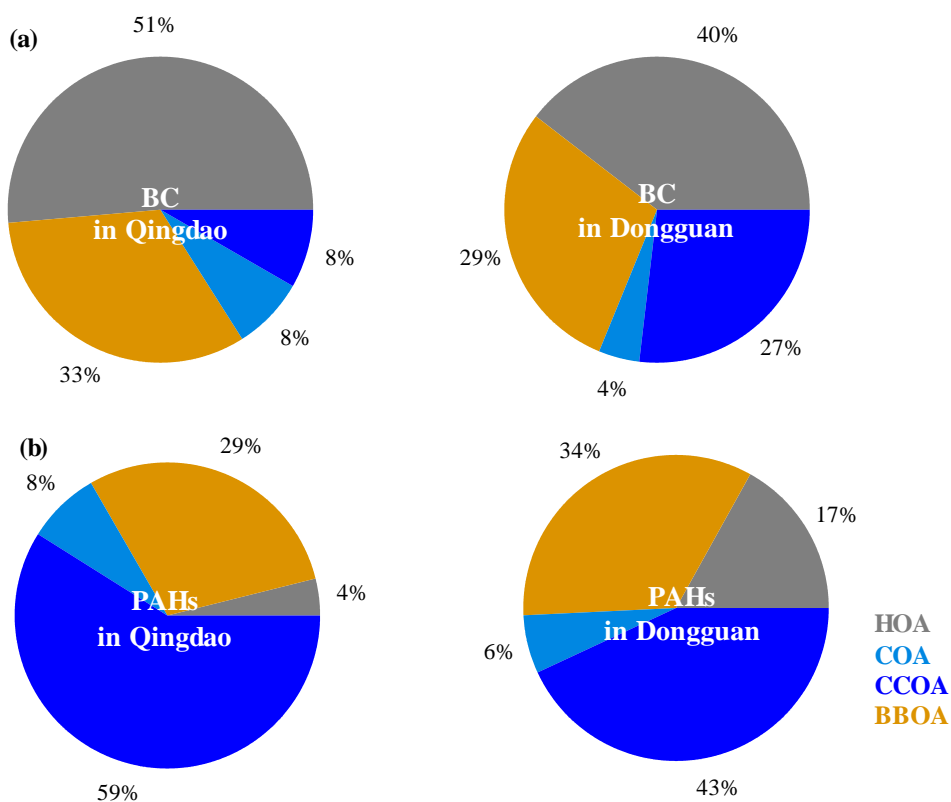
274

275

276

277

Figure 4 shows the average contributions of OA sources to BC and PAHs in Qingdao and Dongguan. At both sites, HOAs were the dominant attribute of BC (51% for Qingdao and 40% for Dongguan) and CCOAs contributed the most to the PAHs (59% for Qingdao and 43% for Dongguan), indicating that BC mainly originates from traffic emissions and that PAHs in the Chinese urban polluted atmosphere are dominated by coal combustion during the wintertime. These findings are consistent with results reported in similar studies (Elser et al., 2016; Huang et al., 2015; Huang et al., 2010; Sun et al., 2016; Xu et al., 2014; Zhang et al., 2008). Moreover, the ratio of PAHs to OAs (1.8%) in Qingdao was similar to that in the northern Chinese urban site of Xi'an (1.9%) (Elser et al., 2016) but was higher than that in Dongguan (0.9%). This is likely because a larger fraction of coal combustion to the total OA concentration would enhance the ratio of PAHs to OAs (Elser et al., 2016). Biomass burning was the second-most important source for both BC and PAHs; it was responsible for 33% and 29% of the BC at Qingdao and Dongguan, respectively, and for 29% and 34% of the PAHs at Qingdao and Dongguan, respectively. Cooking emissions were a minor source of BC and PAHs, accounting for less than 10%. These results also correspond with published findings. For example, biomass burning is an important source for BC (Kondo et al., 2011; Reddy et al., 2002) and, in some regions with fewer traffic emissions, BC has the best correlation with BBOAs (Schwarz et al., 2008). In addition, in Beijing and California, PAHs are correlated well with BBOAs but are much more weakly correlated with COAs (Ge et al., 2012; Hu et al., 2016; Sun et al., 2016).



**Figure 4.** (a) Average contributions of POA factors to BC; (b) average contributions of POA factors to PAHs.

#### 4 Conclusions

In this study, we used PMF to interpret the pollutants of two heavily polluted urban cities, and we found that PMF does not work properly (i.e., it does not allow for the separation of several primary sources of OAs). Therefore, we adopted the ME-2 approach, which yields more reliable solutions. Technically, there are three important steps when using the ME-2 method to interpret the sources of OAs. The first step is to investigate the mixed and unidentified factors that are constrained according to issues in the unconstrained PMF results. Generally, we constrained one or more POA factors (i.e., HOA, COA, BBOA and CCOA) for the polluted urban sites. The second step is to search for a reasonable anchor profile for each constrained factor. Two approaches were used: searching for anchor profiles via an increase in the number of unconstrained PMF factors from the same data set and using mass profiles derived from other similar studies. The third step is to choose the criteria for obtaining the optimal results. The choice of a reasonable range of O/C ratios may represent a good criterion for HR-OA apportionment since the O/C ratio is a significant and distinctive characteristic for different OA factors. In addition, correlations between the resolved OA factors and their tracers were also suggested.



292 **Acknowledgments**

293 This work was supported by the National Natural Science Foundation of China (U1301234, 41622304), the Ministry of  
294 Science and Technology of China (2017YFC0210004), and the Science and Technology Plan of Shenzhen Municipality.

295 **References**

- 296 Aiken, A. C., D. Salcedo, M. J. Cubison, J. A. Huffman, P. F. DeCarlo, I. M. Ulbrich, K. S. Docherty, D. Sueper, J. R.  
297 Kimmel, D. R. Worsnop, A. Trimborn, M. Northway, E. A. Stone, J. J. Schauer, R. M. Volkamer, E. Fortner, B. de Foy,  
298 J. Wang, A. Laskin, V. Shutthanandan, J. Zheng, R. Zhang, J. Gaffney, N. A. Marley, G. Paredes-Miranda, W. P. Arnott,  
299 L. T. Molina, G. Sosa, and J. L. Jimenez : Mexico City aerosol analysis during MILAGRO using high resolution aerosol  
300 mass spectrometry at the urban supersite (T0) – Part 1: Fine particle composition and organic source apportionment,  
301 Atmos. Chem. Phys., 9(17), 6633-6653, doi: 10.5194/acp-9-6633-2009,2009.
- 302 Alfarra, M. R., A. S. H. Prevot, S. Szidat, J. Sandradewi, S. Weimer, V. A. Lanz, D. Schreiber, M. Mohr, and U.  
303 Baltensperger : Identification of the Mass Spectral Signature of Organic Aerosols from Wood Burning Emissions,  
304 Environ. Sci. Technol., 41(16), 5770-5777, doi: 10.1021/es062289b,2007.
- 305 Allan, J. D., A. E. Delia, H. Coe, K. N. Bower, M. R. Alfarra, J. L. Jimenez, A. M. Middlebrook, F. Drewnick, T. B. Onasch,  
306 M. R. Canagaratna, J. T. Jayne, and D. R. Worsnop: A generalised method for the extraction of chemically resolved mass  
307 spectra from Aerodyne aerosol mass spectrometer data, J. Aerosol Sci., 35(7), 909-922, doi:  
308 <https://doi.org/10.1016/j.jaerosci.2004.02.007>,2004.
- 309 Bougiatioti, A., I. Stavroulas, E. Kostenidou, P. Zampas, C. Theodosi, G. Kouvarakis, F. Canonaco, A. S. H. Prévôt, A.  
310 Nenes, S. N. Pandis, and N. Mihalopoulos : Processing of biomass-burning aerosol in the eastern Mediterranean during  
311 summertime, Atmos. Chem. Phys., 14(9), 4793-4807, doi: 10.5194/acp-14-4793-2014,2014.
- 312 Bruns, E. A., M. Krapf, J. Orasche, Y. Huang, R. Zimmermann, L. Drinovec, G. Močnik, I. El-Haddad, J. G. Slowik, J.  
313 Dommen, U. Baltensperger, and A. S. H. Prévôt : Characterization of primary and secondary wood combustion products  
314 generated under different burner loads, Atmos. Chem. Phys., 15(5), 2825-2841, doi: 10.5194/acp-15-2825-2015,2015.
- 315 Canagaratna, M. R., J. L. Jimenez, J. H. Kroll, Q. Chen, S. H. Kessler, P. Massoli, L. Hildebrandt Ruiz, E. Fortner, L. R.  
316 Williams, K. R. Wilson, J. D. Surratt, N. M. Donahue, J. T. Jayne, and D. R. Worsnop : Elemental ratio measurements of  
317 organic compounds using aerosol mass spectrometry: characterization, improved calibration, and implications, Atmos.  
318 Chem. Phys., 15(1), 253-272, doi: 10.5194/acp-15-253-2015,2015.
- 319 Canagaratna, M. R., J. T. Jayne, J. L. Jimenez, J. D. Allan, M. R. Alfarra, Q. Zhang, T. B. Onasch, F. Drewnick, H. Coe, A.  
320 Middlebrook, A. Delia, L. R. Williams, A. M. Trimborn, M. J. Northway, P. F. DeCarlo, C. E. Kolb, P. Davidovits, and  
321 D. R. Worsnop : Chemical and microphysical characterization of ambient aerosols with the aerodyne aerosol mass  
322 spectrometer, Mass Spectrom. Rev., 26(2), 185-222, doi: 10.1002/mas.20115,2007.



- 323 Canonaco, F., M. Crippa, J. G. Slowik, U. Baltensperger, and A. S. H. Prévôt : SoFi, an IGOR-based interface for the  
324 efficient use of the generalized multilinear engine (ME-2) for the source apportionment: ME-2 application to aerosol  
325 mass spectrometer data, *Atmos. Meas. Tech.*, 6(12), 3649-3661, doi: 10.5194/amt-6-3649-2013, 2013.
- 326 Crippa, M., F. Canonaco, V. a. Lanz, M. Äijälä, J. D. Allan, S. Carbone, G. Capes, D. Ceburnis, M. Dall'Osto, D. A. Day, P.  
327 F. DeCarlo, M. Ehn, a. Eriksson, E. Freney, L. Hildebrandt Ruiz, R. Hillamo, J. L. Jimenez, H. Junninen, A. Kiendler-  
328 Scharr, A. M. Kortelainen, M. Kulmala, A. Laaksonen, A. A. Mensah, C. Mohr, E. Nemitz, C. O'Dowd, J. Ovadnevaite,  
329 S. N. Pandis, T. Petäjä, L. Poulain, S. Saarikoski, K. Sellegri, E. Swietlicki, P. Tiitta, D. R. Worsnop, U. Baltensperger,  
330 and A. S. H. Prévôt: Organic aerosol components derived from 25 AMS data sets across Europe using a consistent ME-2  
331 based source apportionment approach, *Atmos. Chem. Phys.*, 14(12), 6159-6176, doi: 10.5194/acp-14-6159-2014, 2014.
- 332 Docherty, K. S., E. A. Stone, I. M. Ulbrich, P. F. DeCarlo, D. C. Snyder, J. J. Schauer, R. E. Peltier, R. J. Weber, S. M.  
333 Murphy, J. H. Seinfeld, B. D. Grover, D. J. Eatough, and J. L. Jimenez : Apportionment of Primary and Secondary  
334 Organic Aerosols in Southern California during the 2005 Study of Organic Aerosols in Riverside (SOAR-1), *Environ Sci.*  
335 *Technol.*, 42(20), 7655-7662, doi: 10.1021/es8008166, 2008.
- 336 Drewnick, F., S. S. Hings, P. DeCarlo, J. T. Jayne, M. Gonin, K. Fuhrer, S. Weimer, J. L. Jimenez, K. L. Demerjian, S.  
337 Borrmann, and D. R. Worsnop : A New Time-of-Flight Aerosol Mass Spectrometer (TOF-AMS)—Instrument  
338 Description and First Field Deployment, *Aerosol Sci. Tech.*, 39(7), 637-658, doi: 10.1080/02786820500182040, 2005.
- 339 Elser, M., R. J. Huang, R. Wolf, J. G. Slowik, Q. Wang, F. Canonaco, G. Li, C. Bozzetti, K. R. Daellenbach, Y. Huang, R.  
340 Zhang, Z. Li, J. Cao, U. Baltensperger, I. El-Haddad, and A. S. H. Prévôt : New insights into PM<sub>2.5</sub> chemical  
341 composition and sources in two major cities in China during extreme haze events using aerosol mass spectrometry,  
342 *Atmos. Chem. Phys.*, 16(5), 3207-3225, doi: 10.5194/acp-16-3207-2016, 2016.
- 343 Fröhlich, R., V. Crenn, A. Setyan, C. A. Belis, F. Canonaco, O. Favez, V. Riffault, J. G. Slowik, W. Aas, M. Aijälä, A.  
344 Alastuey, B. Artiñano, N. Bonnaire, C. Bozzetti, M. Bressi, C. Carbone, E. Coz, P. L. Croteau, M. J. Cubison, J. K.  
345 Esser-Gietl, D. C. Green, V. Gros, L. Heikkinen, H. Herrmann, J. T. Jayne, C. R. Lunder, M. C. Mingüillón, G. Močnik,  
346 C. D. O'Dowd, J. Ovadnevaite, E. Petralia, L. Poulain, M. Priestman, A. Ripoll, R. Sarda-Estève, A. Wiedensohler, U.  
347 Baltensperger, J. Sciare, and A. S. H. Prévôt : ACTRIS ACSM intercomparison – Part 2: Intercomparison of ME-2  
348 organic source apportionment results from 15 individual, co-located aerosol mass spectrometers, *Atmos. Meas. Tech.*,  
349 8(6), 2555-2576, doi: 10.5194/amt-8-2555-2015, 2015.
- 350 Ge, X., A. Setyan, Y. Sun, and Q. Zhang : Primary and secondary organic aerosols in Fresno, California during wintertime:  
351 Results from high resolution aerosol mass spectrometry, *J. Geophys. Res.-Atmos.*, 117(D19), 161-169, doi:  
352 10.1029/2012JD018026, 2012.
- 353 Hallquist, M., J. C. Wenger, U. Baltensperger, Y. Rudich, D. Simpson, M. Claeys, J. Dommen, N. M. Donahue, C. George,  
354 A. H. Goldstein, J. F. Hamilton, H. Herrmann, T. Hoffmann, Y. Iinuma, M. Jang, M. E. Jenkin, J. L. Jimenez, A.  
355 Kiendler-Scharr, W. Maenhaut, G. McFiggans, T. F. Mentel, A. Monod, A. S. H. Prévôt, J. H. Seinfeld, J. D. Surratt, R.



- 356 Szmigielski, and J. Wildt : The formation, properties and impact of secondary organic aerosol: current and emerging  
357 issues, *Atmos. Chem. Phys.*, 9(14), 5155-5236, doi: 10.5194/acp-9-5155-2009,2009.
- 358 He, L. Y., Y. Lin, X. F. Huang, S. Guo, L. Xue, Q. Su, M. Hu, S. J. Luan, and Y. H. Zhang : Characterization of high-  
359 resolution aerosol mass spectra of primary organic aerosol emissions from Chinese cooking and biomass burning, *Atmos.*  
360 *Chem. Phys.*, 10(23), 11535-11543, doi: 10.5194/acp-10-11535-2010,2010.
- 361 Hofzumahaus, A., F. Rohrer, K. Lu, B. Bohn, T. Brauers, C.-C. Chang, H. Fuchs, F. Holland, K. Kita, Y. Kondo, X. Li, S.  
362 Lou, M. Shao, L. Zeng, A. Wahner, and Y. Zhang : Amplified Trace Gas Removal in the Troposphere, *Science*,  
363 324(5935), 1702, doi: 10.1126/science.1164566,2009.
- 364 Hu, W., M. Hu, W. Hu, J. L. Jimenez, B. Yuan, W. Chen, M. Wang, Y. Wu, C. Chen, Z. Wang, J. Peng, L. Zeng, and M.  
365 Shao : Chemical composition, sources, and aging process of submicron aerosols in Beijing: Contrast between summer  
366 and winter, *J. Geophys. Res.-Atmos.*, 121(4), 1955-1977, doi: 10.1002/2015JD024020,2016.
- 367 Hu, W. W., M. Hu, B. Yuan, J. L. Jimenez, Q. Tang, J. F. Peng, W. Hu, M. Shao, M. Wang, L. M. Zeng, Y. S. Wu, Z. H.  
368 Gong, X. F. Huang, and L. Y. He: Insights on organic aerosol aging and the influence of coal combustion at a regional  
369 receptor site of central eastern China, *Atmos. Chem. Phys.*, 13(19), 10095-10112, doi: 10.5194/acp-13-10095-2013,2013.
- 370 Huang, R. J., Y. Zhang, C. Bozzetti, K. F. Ho, J. J. Cao, Y. Han, K. R. Daellenbach, J. G. Slowik, S. M. Platt, F. Canonaco,  
371 P. Zotter, R. Wolf, S. M. Pieber, E. A. Bruns, M. Crippa, G. Ciarelli, A. Piazzalunga, M. Schwikowski, G. Abbaszade, J.  
372 Schnelle-Kreis, R. Zimmermann, Z. An, S. Szidat, U. Baltensperger, I. El Haddad, and A. S. H. Prévôt : High secondary  
373 aerosol contribution to particulate pollution during haze events in China, *Nature*, 514(7521), 218-222, doi:  
374 10.1038/nature13774,2015.
- 375 Huang, X. F., L. Y. He, L. Xue, T. L. Sun, L. W. Zeng, Z. H. Gong, M. Hu, and T. Zhu: Highly time-resolved chemical  
376 characterization of atmospheric fine particles during 2010 Shanghai World Expo, *Atmos. Chem. Phys.*, 12(11), 4897-  
377 4907, doi: 10.5194/acp-12-4897-2012,2012.
- 378 Huang, X. F., L. Y. He, M. Hu, M. R. Canagaratna, Y. Sun, Q. Zhang, T. Zhu, L. Xue, L. W. Zeng, X. G. Liu, Y. H. Zhang,  
379 J. T. Jayne, N. L. Ng, and D. R. Worsnop : Highly time-resolved chemical characterization of atmospheric submicron  
380 particles during 2008 Beijing Olympic Games using an Aerodyne High-Resolution Aerosol Mass Spectrometer, *Atmos.*  
381 *Chem. Phys.*, 10(18), 8933-8945, doi: 10.5194/acp-10-8933-2010,2010.
- 382 IPCC , *Climate Change 2013: The Physical Science Basis. Contribution of Working Group I to the Fifth Assessment Report*  
383 *of the Intergovernmental Panel on Climate Change*, 1535 pp., Cambridge University Press, Cambridge, United Kingdom  
384 and New York, NY, USA,2013.
- 385 Jayne, J. T., D. C. Leard, X. Zhang, P. Davidovits, K. A. Smith, C. E. Kolb, and D. R. Worsnop : Development of an Aerosol  
386 Mass Spectrometer for Size and Composition Analysis of Submicron Particles, *Aerosol Sci. Tech.*, 33(1-2), 49-70, doi:  
387 10.1080/027868200410840,2000.
- 388 Jimenez, J. L., M. R. Canagaratna, N. M. Donahue, A. S. H. Prevot, Q. Zhang, J. H. Kroll, P. F. DeCarlo, J. D. Allan, H. Coe,  
389 N. L. Ng, A. C. Aiken, K. S. Docherty, I. M. Ulbrich, A. P. Grieshop, A. L. Robinson, J. Duplissy, J. D. Smith, K. R.



- 390 Wilson, V. A. Lanz, C. Hueglin, Y. L. Sun, J. Tian, A. Laaksonen, T. Raatikainen, J. Rautiainen, P. Vaattovaara, M. Ehn,  
391 M. Kulmala, J. M. Tomlinson, D. R. Collins, M. J. Cubison, J. Dunlea, J. A. Huffman, T. B. Onasch, M. R. Alfarra, P. I.  
392 Williams, K. Bower, Y. Kondo, J. Schneider, F. Drewnick, S. Borrmann, S. Weimer, K. Demerjian, D. Salcedo, L.  
393 Cottrell, R. Griffin, A. Takami, T. Miyoshi, S. Hatakeyama, A. Shimono, J. Y. Sun, Y. M. Zhang, K. Dzepina, J. R.  
394 Kimmel, D. Sueper, J. T. Jayne, S. C. Herndon, A. M. Trimborn, L. R. Williams, E. C. Wood, A. M. Middlebrook, C. E.  
395 Kolb, U. Baltensperger, and D. R. Worsnop : Evolution of Organic Aerosols in the Atmosphere, *Science*, 326(5959),  
396 1525-1529, doi: 10.1126/science.1180353,2009.
- 397 Kondo, Y., H. Matsui, N. Moteki, L. Sahu, N. Takegawa, M. Kajino, Y. Zhao, M. J. Cubison, J. L. Jimenez, S. Vay, G. S.  
398 Diskin, B. Anderson, A. Wisthaler, T. Mikoviny, H. E. Fuelberg, D. R. Blake, G. Huey, A. J. Weinheimer, D. J. Knapp,  
399 and W. H. Brune: Emissions of black carbon, organic, and inorganic aerosols from biomass burning in North America  
400 and Asia in 2008, *J. Geophys. Res.-Atmos.*, 116(D8), 353-366, doi: 10.1029/2010JD015152,2011.
- 401 Lanz, V. A., M. R. Alfarra, U. Baltensperger, B. Buchmann, C. Hueglin, and A. S. H. Prévôt : Source apportionment of  
402 submicron organic aerosols at an urban site by factor analytical modelling of aerosol mass spectra, *Atmos. Chem. Phys.*,  
403 7(6), 1503-1522, doi: 10.5194/acp-7-1503-2007,2007.
- 404 Lanz, V. A., A. S. H. Prévôt, M. R. Alfarra, S. Weimer, C. Mohr, P. F. DeCarlo, M. F. D. Gianini, C. Hueglin, J. Schneider,  
405 O. Favez, B. D'Anna, C. George, and U. Baltensperger: Characterization of aerosol chemical composition with aerosol  
406 mass spectrometry in Central Europe: an overview, *Atmos. Chem. Phys.*, 10(21), 10453-10471, doi: 10.5194/acp-10-  
407 10453-2010,2010.
- 408 Matthew, B. M., A. M. Middlebrook, and T. B. Onasch : Collection Efficiencies in an Aerodyne Aerosol Mass Spectrometer  
409 as a Function of Particle Phase for Laboratory Generated Aerosols, *Aerosol Sci. Tech.*, 42(11), 884-898, doi:  
410 10.1080/02786820802356797,2008.
- 411 Middlebrook, A. M., R. Bahreini, J. L. Jimenez, and M. R. Canagaratna : Evaluation of Composition-Dependent Collection  
412 Efficiencies for the Aerodyne Aerosol Mass Spectrometer using Field Data, *Aerosol Sci. Tech.*, 46(3), 258-271, doi:  
413 10.1080/02786826.2011.620041,2012.
- 414 Mohr, C., P. F. DeCarlo, M. F. Heringa, R. Chirico, J. G. Slowik, R. Richter, C. Reche, A. Alastuey, X. Querol, R. Seco, J.  
415 Peñuelas, J. L. Jiménez, M. Crippa, R. Zimmermann, U. Baltensperger, and A. S. H. Prévôt : Identification and  
416 quantification of organic aerosol from cooking and other sources in Barcelona using aerosol mass spectrometer data,  
417 *Atmos. Chem. Phys.*, 12(4), 1649-1665, doi: 10.5194/acp-12-1649-2012,2012.
- 418 Ng, N. L., M. R. Canagaratna, J. L. Jimenez, Q. Zhang, I. M. Ulbrich, and D. R. Worsnop : Real-Time Methods for  
419 Estimating Organic Component Mass Concentrations from Aerosol Mass Spectrometer Data, *Environ. Sci. Technol.*,  
420 45(3), 910-916, doi: 10.1021/es102951k,2011.
- 421 Paatero, P. : The Multilinear Engine: A Table-Driven, Least Squares Program for Solving Multilinear Problems, including  
422 the n-Way Parallel Factor Analysis Model, *J. Comput. Graph. Stat.*, 8(4), 854-888, doi: 10.2307/1390831,1999.





- 423 Paatero, P., and U. Tapper : Positive matrix factorization: A non-negative factor model with optimal utilization of error  
424 estimates of data values, *Environmetrics*, 5(2), 111-126, doi: 10.1002/env.3170050203,1994.
- 425 Pope, C. A., and D. W. Dockery : Health Effects of Fine Particulate Air Pollution: Lines that Connect, *J. Air Waste*  
426 *Manage.*,56(6), 709-742, doi: 10.1080/10473289.2006.10464485,2006.
- 427 Pratt, K. A., and K. A. Prather : Mass spectrometry of atmospheric aerosols—Recent developments and applications. Part II:  
428 On-line mass spectrometry techniques, *Mass Spectrom. Rev.*, 31(1), 17-48, doi: 10.1002/mas.20330,2012.
- 429 Qin, Y. M., H. B. Tan, Y. J. Li, M. I. Schurman, F. Li, F. Canonaco, A. S. H. Prévôt, and C. K. Chan : Impacts of traffic  
430 emissions on atmospheric particulate nitrate and organics at a downwind site on the periphery of Guangzhou, China,  
431 *Atmos. Chem. Phys.*, 17(17), 10245-10258, doi: 10.5194/acp-17-10245-2017,2017.
- 432 Reddy, C. M., A. Pearson, L. Xu, A. P. McNichol, B. A. Benner, S. A. Wise, G. A. Klouda, L. A. Currie, and T. I.  
433 Eglinton :Radiocarbon as a Tool To Apportion the Sources of Polycyclic Aromatic Hydrocarbons and Black Carbon in  
434 Environmental Samples, *Environ Sci. Technol.*, 36(8), 1774-1782, doi: 10.1021/es011343f,2002.
- 435 Reyes-Villegas, E., D. C. Green, M. Priestman, F. Canonaco, H. Coe, A. S. H. Prévôt, and J. D. Allan : Organic aerosol  
436 source apportionment in London 2013 with ME-2: exploring the solution space with annual and seasonal analysis, *Atmos.*  
437 *Chem. Phys.*, 16(24), 15545-15559, doi: 10.5194/acp-16-15545-2016,2016.
- 438 Schmidt, M. W. I., and A. G. Noack : Black carbon in soils and sediments: Analysis, distribution, implications, and current  
439 challenges, *Global Biogeochem. Cy.*, 14(3), 777-793, doi: 10.1029/1999GB001208,2000.
- 440 Schwarz, J. P., R. S. Gao, J. R. Spackman, L. A. Watts, D. S. Thomson, D. W. Fahey, T. B. Ryerson, J. Peischl, J. S.  
441 Holloway, M. Trainer, G. J. Frost, T. Baynard, D. A. Lack, J. A. de Gouw, C. Warneke, and L. A. Del  
442 Negro :Measurement of the mixing state, mass, and optical size of individual black carbon particles in urban and biomass  
443 burning emissions, *Geophys Res. Lett.*, 35(13), L13810, doi: 10.1029/2008GL033968,2008.
- 444 Sun, Y., W. Du, P. Fu, Q. Wang, J. Li, X. Ge, Q. Zhang, C. Zhu, L. Ren, W. Xu, J. Zhao, T. Han, D. R. Worsnop, and Z.  
445 Wang : Primary and secondary aerosols in Beijing in winter: sources, variations and processes, *Atmos. Chem. Phys.*,  
446 16(13), 8309-8329, doi: 10.5194/acp-16-8309-2016,2016.
- 447 Ulbrich, I. M., M. R. Canagaratna, Q. Zhang, D. R. Worsnop, and J. L. Jimenez : Interpretation of organic components from  
448 Positive Matrix Factorization of aerosol mass spectrometric data, *Atmos. Chem. Phys.*, 9(9), 2891-2918, doi:  
449 10.5194/acp-9-2891-2009,2009.
- 450 Wang, X., B. J. Williams, X. Wang, Y. Tang, Y. Huang, L. Kong, X. Yang, and P. Biswas : Characterization of organic  
451 aerosol produced during pulverized coal combustion in a drop tube furnace, *Atmos. Chem. Phys.*, 13(21), 10919-10932,  
452 doi: 10.5194/acp-13-10919-2013,2013.
- 453 White, H. :Black carbon in the environment, J. Wiley,1985.
- 454 Xu, J., Q. Zhang, M. Chen, X. Ge, J. Ren, and D. Qin : Chemical composition, sources, and processes of urban aerosols  
455 during summertime in northwest China: insights from high-resolution aerosol mass spectrometry, *Atmos. Chem. Phys.*,  
456 14(23), 12593-12611, doi: 10.5194/acp-14-12593-2014,2014.



- 457 Xu, L., S. Suresh, H. Guo, R. J. Weber, and N. L. Ng : Aerosol characterization over the southeastern United States using  
458 high-resolution aerosol mass spectrometry: spatial and seasonal variation of aerosol composition and sources with a  
459 focus on organic nitrates, *Atmos. Chem. Phys.*, 15(13), 7307-7336, doi: 10.5194/acp-15-7307-2015,2015.
- 460 Zhang, Q., J. L. Jimenez, M. R. Canagaratna, I. M. Ulbrich, N. L. Ng, D. R. Worsnop, and Y. Sun : Understanding  
461 atmospheric organic aerosols via factor analysis of aerosol mass spectrometry: a review, *Anal. Bioanal. Chem.*, 401(10),  
462 3045-3067, doi: 10.1007/s00216-011-5355-y,2011.
- 463 Zhang, Q., J. L. Jimenez, M. R. Canagaratna, J. D. Allan, H. Coe, I. Ulbrich, M. R. Alfarra, A. Takami, A. M. Middlebrook,  
464 Y. L. Sun, K. Dzepina, E. Dunlea, K. Docherty, P. F. DeCarlo, D. Salcedo, T. Onasch, J. T. Jayne, T. Miyoshi, A.  
465 Shimono, S. Hatakeyama, N. Takegawa, Y. Kondo, J. Schneider, F. Drewnick, S. Borrmann, S. Weimer, K. Demerjian,  
466 P. Williams, K. Bower, R. Bahreini, L. Cottrell, R. J. Griffin, J. Rautiainen, J. Y. Sun, Y. M. Zhang, and D. R.  
467 Worsnop : Ubiquity and dominance of oxygenated species in organic aerosols in anthropogenically-influenced Northern  
468 Hemisphere midlatitudes, *Geophys. Res. Lett.*, 34(13), L13801, doi: 10.1029/2007GL029979,2007.
- 469 Zhang, Y., J. J. Schauer, Y. Zhang, L. Zeng, Y. Wei, Y. Liu, and M. Shao : Characteristics of Particulate Carbon Emissions  
470 from Real-World Chinese Coal Combustion, *Environ. Sci. Technol.*, 42(14), 5068-5073, doi: 10.1021/es702257g,2008.
- 471 Zhou, S., S. Collier, D. A. Jaffe, N. L. Briggs, J. Hee, A. J. Sedlacek Iii, L. Kleinman, T. B. Onasch, and Q. Zhang :  
472 Regional influence of wildfires on aerosol chemistry in the western US and insights into atmospheric aging of biomass  
473 burning organic aerosol, *Atmos. Chem. Phys.*, 17(3), 2477-2493, doi: 10.5194/acp-17-2477-2017,2017.
- 474