



1 Intercomparison of Open-Path Trace Gas Measurements with Two Dual Frequency Comb Spectrometers

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- 11 Abstract

12 We present the first quantitative intercomparison between two open-path dual comb spectroscopy

13 (DCS) instruments which were operated across adjacent 2-km open-air paths over a two-week period.

14 We used DCS to measure the atmospheric absorption spectrum in the near infrared from 6021 to 6388

 15 cm^{-1} (1565 to 1661 nm), corresponding to a 367 cm⁻¹ bandwidth, at 0.0067 cm⁻¹ sample spacing. The

measured absorption spectra agree with each other to within 5×10^{-4} without any external calibration of either instrument. The absorption spectra are fit to retrieve concentrations for carbon dioxide (CO₂),

methane (CH₄), water (H₂O), and deuterated water (HDO). The retrieved dry mole fractions agree to

- 19 0.14% (0.57 ppm) for CO₂, 0.35% (7 ppb) for CH₄, and 0.40% (36 ppm) for H₂O over the two-week
- 20 measurement campaign, which included 23 °C outdoor temperature variations and periods of strong

21 atmospheric turbulence. This agreement is at least an order of magnitude better than conventional

active-source open-path instrument intercomparisons and is particularly relevant to future regional flux

23 measurements as it allows accurate comparisons of open-path DCS data across locations and time. We

additionally compare the open-path DCS retrievals to a WMO-calibrated cavity ringdown point sensor
 located along the path with good agreement. Short-term and long-term differences between the two

systems are attributed, respectively, to spatial sampling discrepancies and to inaccuracies in the current

27 spectral database used to fit the DCS data. Finally, the two-week measurement campaign yields diurnal

cycles of CO_2 and CH_4 that are consistent with the presence of local sources of CO_2 and absence of local sources of CH_4 .

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33 1. Introduction

34 Quantitative determination of greenhouse gas fluxes over a variety of temporal and spatial 35 scales is necessary for characterizing source strength and intermittency and for future emissions 36 monitoring, reporting, and verification. To this end, techniques exist to measure greenhouse gas 37 concentrations on a variety of length scales, each of which has advantages and disadvantages. Point 38 sensors provide valuable information about local sources, but their use for continuous regional 39 measurements on sampling towers is complicated by local wind patterns, local sources, and mixing 40 within the planetary boundary layer (PBL), especially at night (Lauvaux et al., 2008; Ciais et al., 2010; 41 Lauvaux et al., 2012). Similarly, total-column measurements are particularly useful for sub-continental to 42 global scale measurements; however they are sensitive to atmospheric transport errors within the PBL 43 (Lauvaux and Davis, 2014), are affected by clouds and aerosols, are primarily limited to daytime 44 measurements, and lack either the revisit rates or mobility for regional flux measurements. Horizontal 45 integrated path measurements are complementary to point sensors and satellites: they cover spatial 46 scales from 1-10s of kilometers and provide measurements on the second to minute time scales with 47 portable instruments and are thus appropriate for regional studies. Active-source open-path sensors such as open-path Fourier Transform Infrared spectroscopy (FTIR), differential optical absorption 48





49 spectroscopy (DOAS), differential LIDAR (DIAL), or tunable diode laser absorption spectroscopy (TDLAS) 50 are often used for these measurements and can retrieve path-averaged concentrations but typically 51 with 10% or greater uncertainties (EPA Handbook, and references therein). Recently, open-path dual-52 comb spectroscopy (DCS) has emerged as a new technique that could potentially provide precise, 53 accurate continuous regional measurements of the mole fractions of CO₂, CH₄, H₂O, and HDO over 54 kilometer-scale open paths (Rieker et al., 2014), thereby providing a new open-path sensing capability 55 that falls between point sensing and total-column measurements. 56 Here we demonstrate that open-path DCS can indeed yield dry mole fractions over open-air 57 paths with a high level of intercomparibility, over long periods of time, and with sufficient precision to 58 track variations in the ambient levels from local sources and sinks. Two completely independent open-59 path DCS instruments are operated over neighboring open-air paths during a two-week measurement 60 campaign. Although both DCS instruments use fully stabilized frequency combs, they are portable 61 (Truong et al., 2016) and are operated nearly continuously during both day and night through laboratory 62 temperature variations from 17 to 25°C, strong atmospheric turbulence, and outdoor air temperature 63 variations from 4.6 to 28.9°C. The retrieved dry mole fractions for the two DCS instruments agree to 64 better than 0.57 ppm¹ (0.14 %) for CO₂ and 7.0 ppb (0.37 %) for CH₄. This agreement is achieved without 65 any "bias correction" or calibration of either instrument for absolute wavelength or for absolute 66 concentration. Instead, it is a direct consequence of the negligible instrument lineshape and precise 67 frequency calibration of the DCS instruments, which leads to measured atmospheric absorption spectra 68 that are identical to below 10^{-3} (limited by the instrument noise level). The measured path-averaged CO₂ 69 precision over a 2-km path is 0.90 ppm in 30 seconds, improving to 0.24 ppm in 5 minutes. For CH₄, the 70 precision is 9.6 ppb in 30 seconds, improving to 2.1 ppb in 5 minutes. We also compare the DCS 71 retrievals to a WMO-calibrated, cavity ringdown point sensor located near the path. The agreement is 72 within 3.4 ppm and 17 ppb for CO₂ and CH₄ respectively, limited by differences in the sampling volume 73 and by the spectral database used to analyze the DCS transmission spectra. 74 Similar intercomparison measurements between conventional active, open-path sensors are

75 rare but have shown agreement of typically 1-20% (Thoma et al., 2005; Hak et al., 2005; Smith et al., 76 2011; Shao et al., 2013; Conde et al., 2014; Reiche et al., 2014; Thalman et al., 2015). Here, we find 77 agreement between two DCS instruments that is an order of magnitude better and is comparable to 78 that achieved with highly-calibrated, state-of-the-art, solar-looking FTIRs that retrieve vertical column 79 measurements (Messerschmidt et al., 2011; Frey et al., 2015; Hedelius et al., 2016); however, open-path 80 DCS does not require instrument-specific calibrations (e.g. of the instrument line shape) and provides a 81 very different capability by retrieving the dry mole fractions across regional, kilometer-scale paths over 82 day and night in a mobile platform. Moreover, as the agreement between open-path DCS instruments is 83 below the level of natural background fluctuations, future measurements can facilitate accurate inverse 84 modeling to identify sources and sinks of carbon emission over regions. As an initial demonstration, we 85 discuss the observed diurnal variations from this two-week measurement campaign in the final section 86 of the paper.

- 87
- 88 2 Technique
- 89 2.1 Overview

90DCS is based on a frequency comb laser source, which is a pulsed laser that outputs a spectrum91consisting of evenly-spaced, narrow modes ("comb teeth") underneath a broad spectral envelope92(Cundiff and Ye, 2003; Hall, 2006; Hänsch, 2006). In DCS, two such frequency combs are used to

93 measure the atmospheric absorption on a comb-tooth-by-tooth basis across broad bandwidths

¹ In this work we use dry mole fraction. ppm is defined as micromoles of CO_2 per mole of dry air and ppb is defined as nanomoles of CH_4 per mole of dry air.





94 (Coddington et al., 2016; Ideguchi, 2017). As shown in Figure 1, two combs with nominal repetition 95 rates of f_r and offset by Δf_r are phase locked together, transmitted through a sample, and their 96 heterodyne signal measured on a photodetector. The resulting rf frequency comb can be mapped back 97 to the optical domain to generate an overall spectrum, as shown in Figure 1(b), that is the product of the 98 comb spectra and any atmospheric absorption. One important difference between DCS and other 99 broad-band laser techniques is that here all wavelengths are measured at once rather than sequentially 100 as would be the case for a swept laser system; as a result, DCS is much more immune to spectral 101 distortions from turbulence effects. Moreover, for a fully phase-locked comb, as is used here, the optical 102 frequency axis is stable and known to high accuracy, and the instrument lineshape is effectively the sum 103 of two delta-functions, as shown in the spectrum in Figure 1(b). Alternatively, DCS can be thought of as 104 high-resolution Fourier-transform spectroscopy with diffraction-limited light sources, no moving parts, 105 negligible instrument line shape, and a rapid scanning rate of $1/\Delta f_r$, which we tune to be faster than 106 turbulence-induced intensity variations. Here, both combs are transmitted over the open path yielding 107 the atmospheric absorption spectrum, but it is also possible to transmit only a single comb through the 108 air to measure both dispersion and absorbance (Giorgetta et al., 2015: Coddington et al., 2016). 109 Figure 2 provides an overview of our experiment. Two DCS instruments measured the 110 atmospheric absorption across a 2-km-roundtrip open path that extended from the top of a building at 111 the National Institute of Standards and Technology (NIST) Boulder campus to a pair of retroreflectors 112 located on a nearby hill. Both DCS instruments were based on a similar overall design and used self-113 referenced, stabilized frequency combs (Sinclair et al., 2015), but one was built by a team at NIST and 114 the other by a team at the University of Colorado; they are hereafter referred to as DCS A and DCS B,

115 respectively. As outlined below, the two instruments differed in their exact design and physical 116 parameters. Nevertheless, no instrument-specific calibration or bias offset was applied to either system. The acquired atmospheric absorption spectra were fit to retrieve the column density of CO₂, CH₄, and 117 118 H_2O (as well as HDO and $^{13}CO_2$ at lower precision) along with the path-averaged temperature from the 119 CO₂ spectrum. From these data, combined with the measured atmospheric pressure and the path length 120 (measured via time-of-flight laser ranging), we retrieved the path-averaged dry mole fractions as a 121 function of time, which are compared between DCS instruments and to a nearby cavity ringdown (CRDS) 122 point sensor.

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124 2.2 Dual comb spectrometer

125 Figure 3(a) shows a simplified schematic of both DCS setups. Briefly, each DCS system used two 126 mutually coherent self-referenced erbium-doped fiber frequency combs based on the design of (Sinclair 127 et al., 2015) with nominal repetitions rates and Δf_r given in Table 1. Mutual optical coherence between 128 the combs is enforced by phase-locking an optical tooth of each to a common cw laser and the carrier-129 envelope offset frequency of each to a common quartz microwave oscillator. Absolute frequency 130 accuracy is then enforced by a bootstrapped approach that effectively locks the common cw laser to the 131 same quartz microwave oscillator (Truong et al., 2016). The result is sub-Hz mutual coherence, ~120-kHz 132 absolute linewidths, and 3.6 x 10⁻⁵ cm⁻¹ absolute frequency accuracy (Truong et al., 2016). This linewidth is orders of magnitude lower than the ~5 GHz or ~0.2 cm⁻¹ width of pressure-broadened absorption 133 134 lines. The direct output of the combs is spectrally broadened in highly nonlinear fiber to cover 7140-135 5710 cm⁻¹ (1.4-1.75 μm) and then filtered to isolate the spectral region of interest from 6021 to 6388 136 cm⁻¹ (1565 to 1661 nm).

137The combined light from both combs is transmitted via single-mode fiber to a telescope, where138it is launched to a retroreflector. The returning signal is collected onto an amplified, 100-MHz-139bandwidth InGaAs photodetector and digitized at f_r . We acquire a single interferogram at a period of140 $1/\Delta f_r$ or 1.6 ms for DCS A; 100 such interferograms are directly summed in real time on a field-141programmable gate array (FPGA). These are transferred to a computer where they are carrier-phase





142 corrected and further summed over an acquisition time of ~30 seconds. These summed interferograms 143 are then Fourier transformed and scaled, using the known optical frequency comb tooth positions, to 144 generate a transmission spectrum (e.g. Figure 4a) spanning 367 cm⁻¹ (>10 THz) with a point spacing of 145 0.0067 cm⁻¹.

146The exact optical layout of DCS A is given in (Truong et al., 2016). While following the same147basic design, DCS B differs in several technical details. These include a slightly different output148spectrum, as well as slight different comb tooth spacings and offset frequency, minor differences in the149reference cw laser and its locking scheme, and different amplifier design, launched and received powers,150and telescope design. Some of these differences are laid out in Figure 3, Table 1, and Section 2.3 below.

151 We have found that the use of stabilized, phase coherent frequency combs is a necessary but 152 not sufficient prerequisite to reaching sub-percent agreement in retrieved gas concentrations. It is 153 critical that the spectrally-filtered comb output does not include stray unfiltered light. Similarly, any 154 stray reflections from the telescope that can "short circuit" the atmospheric path must be avoided. As 155 with FTIR systems, nonlinearities are problematic. In the optical domain, nonlinearities can arise when 156 the combs are combined in fiber with high optical power. These are minimized for DCS A by filtering the 157 light, which decreases the peak powers, before combining the combs. For DCS B the combs do not have 158 a booster amplifier and thus have significantly lower power. Nonlinearities in the photodetection can 159 also occur (Zolot et al., 2013); in laboratory tests with a CO reference cell, we verified no bias in 160 retrieved concentration as a function of received power up to 300 µW, which is a factor of two higher 161 than the maximum power for the open path data. It was also critical to match the interferogram 162 amplitude to the full dynamic range of the analog-to-digital converters (ADCs) to avoid effective 163 nonlinearities in the digitization process.

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165 2.3 Launch/Receive telescope

166 The two telescope systems are shown in Figure 3(a). Due to the large spectral bandwidth, 167 reflective optics are preferred to minimize chromatic dispersion. For DCS A, the launch/receive system 168 was based on a bi-directional off-axis parabolic telescope with a 3" aperture while for DCS B, it was 169 based on a 6"-aperture Ritchey-Chretien (RC) telescope with the light launched separately from behind 170 the secondary mirror. In both cases, the launched beam diameter was ~ 40 mm and the light was 171 directed to a hollow corner-cube retroreflector of 2.5" (DCS A) or 5" (DCS B) diameter. A slow servo was 172 implemented for long-term pointing of the telescope to the retroreflectors. For this servo, a low-173 divergence 850 nm LED is co-aligned with the telescope and its retro-reflected light is detected by a co-174 aligned CMOS camera with a long focal-length lens and an 850-nm-bandpass optical filter. We then 175 servo the overall telescope pointing via its gimbal using the LED spot location on the camera. Further 176 servo details are described in Cossel et al. (2017).

177 Figure 3(b) shows the return power for both systems as a function of time. For reference, the 178 minimum return power required to obtain useful spectra was ~15 µW (horizontal black line). At lower 179 powers, the acquired individual spectra are excluded. Turbulence-induced intensity variations are lower 180 for the RC-telescope than the off-axis parabolic telescope because of its larger aperture; however, the 181 long-term stability of the off-axis parabolic telescope was better due to a higher-quality gimbal system. 182 The collection efficiency of the 6" RC telescope system was about 10-20% in low to moderate turbulence $(C_n^2 \text{ of } 10^{-14})$. The collection efficiency of the off-axis parabolic telescope system was lower, at ~ 2-4% in 183 184 similar conditions, due to 1) the smaller collection aperture and 2) the 50:50 beam splitter, which causes 185 a factor of 4 loss. Attempts to replace the 50:50 splitter with a polarizing beam splitter and guarter-186 wave plate combination increased the collection efficiency but introduced additional etalons across the 187 spectrum and for this reason was not used.

188

189 2.4 Data processing





190 The acquired transmission spectra are the product $S(v) = I_0(v) \times e^{-A(v)}$, where I_0 is the 191 geometric mean of the two individual comb spectra, A(v) is the desired atmospheric absorbance, and v 192 is the average optical frequency of the two participating comb teeth, (e.g. Fig. 1(b)). We fit the natural 193 logarithm of the transmission spectra, $-\ln[S(v)] = -\ln[I_0(v)] + A(v)$, where the first term is 194 represented by a piecewise polynomial and the second by an absorption spectrum calculated from a spectral database with floated concentrations of ¹²CO₂, ¹³CO₂, ¹²CH₄, ¹³CH₄, H₂O, and HDO. For a spectral 195 196 database we use HITRAN 2008 (Rothman et al., 2009) and Voigt lineshapes as this generates a consistent 197 set of line parameters across our conditions and gases. The fit is performed in three steps: first, we fit 198 the polynomial (typically 7th order) over small windows (typically 100 GHz or 3.33 cm⁻¹) and include the 199 expected absorbance from relevant gas absorption lines. These polynomials are then stitched together 200 to generate the overall polynomial baseline, which is removed from the measured spectrum to 201 find A(v). We then fit only the 30013 \leftarrow 00001 CO₂ band in order to retrieve the path-averaged 202 temperature. Finally, A(v) is then re-fit over the entire spectral window by floating the gas 203 concentrations at the retrieved path-averaged temperature. The retrieved path-averaged 204 concentrations are converted to wet mole fractions by normalizing to the total number density of air 205 molecules, which is calculated from the fitted (or separately measured) air temperature combined with 206 the atmospheric pressure, as measured by a sensor co-located with the CRDS sensor and corrected for 207 the altitude difference. Finally, wet CO₂, ¹³CO₂, and CH₄ are converted to dry values (XCO₂, X¹³CO₂, XCH₄) 208 using XS = $S/(1-c_{H2O})$ where XS is the dry species concentration, S is the retrieved wet species 209 concentration and c_{H2O} is the retrieved H_2O volume mole fraction. 210 211 3 Intercomparison Results and Discussion

212 3.1 Atmospheric spectrum comparison

213 Figure 4(a) shows the overall raw DCS transmission spectra from the two instruments averaged for a 214 three-hour period. They differ significantly because of the different comb intensity profiles, $I_0(v)$. 215 However, after the polynomial baseline fit discussed above is applied, the resulting 3-hour averaged 216 absorption spectra are nearly identical as shown in Figure 4(b). The inset of Figure 4(b) shows the data 217 sampling points (spaced at ~200 MHz) across several absorption lines with width of 5 GHz 0.2 cm⁻¹) 218 indicating we have sufficient optical resolution to over-sample the lines. The difference of the 219 absorption spectra, shown as the black line in Figure 4(c), has a standard deviation of 9×10^{-4} with no 220 observable structure at absorption lines. This difference is dominated by an etalon on the off-axis 221 telescope used with DCS A. After manually fitting out the etalon structure, DCS A and DCS B agree to 222 better than 5×10⁻⁴ (limited by the instrumental noise level) over the full spectral region (with the 223 exception of a 7 cm⁻¹ section at 6290 cm⁻¹), and better than 2.5×10⁻⁴ over the region near 6100 cm⁻¹ 224 where both DCS systems have significant returned optical power. This very high level of agreement 225 between the two spectra shows that there are no instrumental line shapes or detector nonlinearity 226 effects distorting the observed spectral line shapes; otherwise, structure would be observed in the 227 difference. Thus, the two DCS instruments measure the same comb-tooth-resolved atmospheric 228 absorbance spectrum.

Figure 4(d) shows the residuals after fitting the absorption lines in the DCS A spectrum to HITRAN 2008 and removing the etalon. The higher SNR of the DCS A yields an even lower broadband noise than the difference spectrum, but there are clear residuals near spectral lines attributable to incorrect line shapes/parameters in the HITRAN 2008 database. Nevertheless, the overall magnitude of the residuals is very small in comparison to the spectral absorption.

234

235 3.2 Comparison of retrieved mole fractions from DCS A and DCS B

From the fitted concentrations, we retrieve the mole fractions as outlined in Section 2.4. The
 retrieved time series for XCO₂, X¹³CO₂, XCH₄, H₂O, and HDO are given in Figure 5 at ~30 second intervals.





238 Gaps in the data are due to either telescope misalignment (primarily on the 6" RC telescope due to the 239 lower-quality gimbal system) or, more rarely, a loss of phase lock of one of the four frequency combs. 240 Excellent agreement is observed between both systems for all retrieved concentrations. Figure 6 shows 241 the concentration differences, which exhibit a high-frequency white noise consistent with the 242 quadrature sum of the DCS precisions given in Section 3.3. In addition, the differences show a slow 243 wander about zero indicating slowly changing, small offsets between the two DCS instruments. CH4 also 244 shows a small negative offset for the second week of the campaign. A Gaussian curve approximates the 245 distribution of the differences over the full two weeks reasonably well and is shown in Figure 7. At 32-246 second averaging times, the mean and width of the distributions are Δ XCO₂ = 0.57 \pm 2.4 ppm, Δ XCH₄ 247 $= -7.0 \pm 16$ ppb, $\Delta c_{H20} = 36 \pm 90$ ppm, and $\Delta c_{HD0} = 390 \pm 860$ ppm. These widths decrease to 1.5 248 ppm, 12 ppb, 66 ppm, and 480 ppm, respectively, for 5-minute averaging times. These mean values 249 correspond to a relative offset of 0.14 % CO₂, -0.35 % CH₄, and 0.4 % H₂O and are close to the WMO 250 compatibility standards of 0.1 ppm for CO_2 and 2 ppb for CH_4 (Tans and Zellweger, 2015). We emphasize 251 the agreement here is achieved over a two-week period despite outdoor temperature variations of 4.6 252 to 28.9 °C. DCS instrument ambient temperature variations from 17 to 25 °C. 10% to 90% relative 253 humidity fluctuations, and large turbulence-induced return power fluctuations.

254 Table 2 summarizes the systematic uncertainties of the DCS systems. The choice of spectral 255 model effectively sets the calibration that converts the measured absorbance spectrum to path-256 averaged concentrations. The temperature primarily affects the conversion of the path-averaged 257 concentration to mole fractions (through the calculation of the overall air concentration). For the direct 258 intercomparison, both DCS data were analyzed with a common spectral model (HITRAN 2008) and 259 temperature in order to separate out instrument-specific systematics from the more fundamental 260 connection between absorption and concentration. Below we discuss these instrument-specific 261 systematics (given in the top part of Table 2). A discussion of the uncertainties from the spectral model 262 and temperature (given in the bottom part of Table 2) is given in Section 3.5.

263 To explore the source of the small systematic offsets between the DCS retrievals, we have 264 performed a number of control comparisons. In the processing, we have varied the initial concentration 265 guess in the fit with negligible effect. We have also varied the polynomial baseline fit by adjusting the window size from 100 to 150 GHz and polynomial order from 7th to 9th order and again found negligible 266 267 variations of 0.02 % for CO_2 (<0.07 ppm), 0.07% for CH_4 (<1.4 ppb), and 0.05% (~4 ppm) for H_2O . In 268 laboratory tests, we verified that the two DCS instruments retrieve the same CO₂ concentrations to 269 within 0.04% for 8450 ppm of CO_2 in a 30-meter multipass cell (roughly mimicking the total absorption 270 over the open path). In open-path tests, we have separated effects of the detection/acquisition system 271 and optical system. First, the detected DCS A return signal was split to the two separate data acquisition 272 systems. The two processed signals yielded small differences of 0.16 ppm CO₂, 0.34 ppb CH₄, and 1.0 273 ppm H₂O, presumably due to residual nonlinearities and reflections in the rf system and digitization. 274 Second, the outgoing DCS A comb light was split and directed to the two different telescopes and 275 acquisition systems. These two processed signals yielded larger differences of 0.45 ppm CO₂, 1.5 ppb 276 CH₄, and 56 ppm H₂O, possibly due to scattered light or polarization dependences in the launch and 277 receive optical systems. Finally, residual phase noise between the two combs in a single DCS system can 278 cause small biases in the retrieved concentrations, but these should be well below 0.1% in this 279 configuration (Truong et al. 2017, in prep). All these instrument uncertainties are summarized in Table 2. 280

281 3.3 DCS precision

282Figure 8 shows the precision versus averaging time (determined using the modified Allan283deviation) based on the scatter across a 6-hour period over which the CO2 and CH4 concentrations are284reasonably flat, shown as the highlighted part of Figure 5. (The Allan deviation for H2O is not calculated285because the atmospheric H2O concentration varies significantly over this time period.) Under perfectly





stable concentrations and white instrument noise, the precision should decrease as the square root of
 averaging time, indicated as a grey line in Figure 8. Initially, the Allan deviations do follow this slope, but
 the atmospheric concentrations, especially of CO₂, vary over this 6-hour period and the Allan deviations
 reach a floor at ~ 1000 s.

290 The precision at 30-second and 5-minute averaging time is given at the bottom of Table 1. DCS 291 A has superior CO₂ precision because it has higher received optical comb power in that spectral region, 292 whereas the DCS instruments have similar received power in the CH₄ spectral region and therefore 293 similar CH₄ precisions. Regardless, the precision of either instrument is sufficiently high to measure the 294 characteristic atmospheric fluctuations of these gases on tens-of-seconds timescales.

295

296 3.4 Comparison of open-path DCS to a cavity ringdown point sensor (CRDS)

A commercial cavity-ringdown point sensor, Picarro Model 1301² (Crosson, 2008), was also located along the path as shown in Figure 2. Its inlet was 30 m above ground on a radio tower, approximately 160 m perpendicular to the DCS beam path. Figure 9 compares the DCS A and CRDS (smoothed to 32-s resolution) time series. In general, their overall shapes agree well with both systems tracking ~40 ppm variations in XCO₂, 200 ppb variations in XCH₄, and 1% variations in H₂O over days. Nevertheless, there are clear discrepancies in terms of both short-duration spikes and a long-term overall offset between the DCS and CRDS time series.

304 The short-duration spikes are present in the CRDS time series and presumably arise from the very 305 different spatial sampling of the two instruments. The DCS system measures the integrated column over 306 one kilometer (one way), while the CRDS is a point sensor and therefore much more sensitive to local 307 sources. For example, a 1 m³ volume of air containing 500 ppm of CO₂ from a vehicle driving under the 308 sampling line will result in a sharp spike in the CRDS data as the air mass passes the sampling inlet. 309 However, that same air mass will result in only a 0.025% or 0.1 ppm increase in the DCS path-averaged 310 concentration (assuming a 400 ppm background). These spikes in the CRDS time series are damped here 311 by the 32-second smoothing but are occasionally evident especially during the second week. The general 312 scarcity of such events does suggest that the air over the open path is usually fairly well mixed.

313 The long-term overall offset between the CRDS and DCS data is a consequence of their very different 314 calibrations. The CRDS is tied to the WMO scale for CO₂ and CH₄ by directly injecting known dry WMO-315 calibrated CO₂/CH₄ mixtures at different trace gas concentrations and different water vapor 316 concentrations into its temperature- and pressure-controlled sampling cavity. This instrument was 317 calibrated shortly after the measurement campaign and should thus have an absolute uncertainty close 318 to that of the WMO-scale uncertainties of ~ 0.07 ppm for CO₂ (Zhao and Tans, 2006) and ~1.5 ppb for 319 CH₄ (Dlugokencky et al., 2005). In contrast, the DCS has no instrument-specific calibration but relies 320 completely on a fit to a spectral database to extract the gas concentrations from the measured 321 absorbance across a wide range of ambient pressures and temperatures. Here, we use HITRAN 2008 which has ¹²CO₂ linestrength uncertainties of 1-2 %, ¹²CH₄ linestrength uncertainties of 10-20%, and 322 323 H₂¹⁶O linestrength uncertainties of 5-10% (Rothman et al., 2009), leading to a poorer absolute 324 calibration than the WMO-calibrated point sensor. From the data in Figure 9, the differences between 325 the CRDS and DCS data across the two-week period are -3.4 ± 3.4 ppm CO₂, 17 ± 15 ppb CH₄, and $580 \pm$ 326 462 ppm H_2O at 5-minute averaging. These correspond to relative offsets of -0.85% for CO_2 , 0.94% for 327 CH₄, and 6.9% for H₂O, well within the stated uncertainties of HITRAN 2008. In previous DCS 328 measurements, we found slightly different offsets, specifically 1.78% for CO₂, 0.20% for CH₄, and 1.74% 329 for H₂O in (Rieker et al., 2014) and ~1 % for CO₂ in (Giorgetta et al., 2015). However, these previous data

² The use of trade names is necessary to specify the experimental results and does not imply endorsement by the National Institute of Standards and Technology.





330 covered much shorter timespans, used an older CRDS point sensor calibration, and may have included 331 small systematic offsets in the DCS systems due to technical issues discussed in Section 3.2. 332 This basic discrepancy between retrievals based on lineshape parameters from a spectral database 333 and manometric calibrations (WMO standard) is not unique to DCS. Several studies have calibrated the 334 Total Carbon Column Observing Network (TCCON) retrievals against WMO-based instruments (Wunch et 335 al., 2010; Messerschmidt et al., 2011; Geibel et al., 2012; Tanaka et al., 2012). Although TCCON is not a 336 solely HITRAN-based analysis (Wunch et al., 2011), a correction factor of 0.9898 for CO₂, 0.9765 for CH₄, 337 and 1.0183 for H₂O (Wunch et al., 2010) is needed to bring the overall TCCON retrievals into agreement 338 with the WMO-based data. Additionally, theoretical calculations by (Zak et al., 2016) found an 339 approximately 0.5% difference between CO₂ line parameters from HITRAN 2012 and their density 340 functional theory calculations and an additional 0.5% difference between the calculations and new 341 measurements by (Devi et al., 2016) in the 1.6-micron region. Certainly this discrepancy between 342 retrievals from HITRAN and WMO-calibrated instruments is not fundamental and further experimental 343 work should lead to improved spectral database parameters and much better agreement. As noted in 344 earlier work on CO₂, it will be important to establish both the correct linestrengths as well as account for 345 complex lineshapes and line mixing (e.g. Devi et al., 2007; Thompson et al., 2012; Bui et al., 2014; Long 346 et al., 2015; Devi et al., 2016). A direct comparison of the open-path DCS spectra acquired here and 347 laboratory DCS spectra acquired for WMO-calibrated gas samples can contribute to these future 348 improvements and is planned. Finally, we emphasize that because the DCS instruments record the 349 atmospheric absorption without instrument distortions, as spectral models improve, past open-path 350 spectra can be refit with reduced uncertainty.

351 An accurate path-averaged air temperature is also important to avoid systematic offsets. Unlike 352 vertical total-column measurements through the entire atmosphere, km-scale open horizontal paths 353 should have relatively low temperature inhomogeneities of around a few degrees C, and thus the use of 354 a single "path averaged" temperature in the fit is sufficient for accurate retrievals. We verified this 355 through a sensitivity study comparing retrievals for simulated spectra with temperature gradients up to 356 10 °C over the path; the resulting bias was below 0.03 ppm CO_2 (0.007%) and 0.4 ppb CH_4 (0.022%), as 357 shown in Table 2. On the other hand, any error in the path-averaged temperature can bias the mole 358 fractions through two effects. First, the retrieved path-averaged concentration will vary weakly with 359 temperature because of temperature-dependent line parameters. Second and dominantly, the final 360 mole fraction calculation requires normalization by the air density. Here, this density is calculated from 361 the ideal gas law using the measured air pressure and path-averaged temperature. Therefore, a fractional error in temperature leads to a corresponding fractional error in mole fraction. For example, a 362 363 0.15% uncertainty in mole fraction requires 0.5 °C uncertainty in the path-averaged air temperature. 364 (See Table 2.) We verified that this simple linear relationship is valid up to a temperature uncertainty of 365 10 °C in a sensitivity study. From the discussion in Appendix A, the use of a point temperature sensor 366 near the end of the open path is clearly insufficient to achieve <0.5 °C uncertainty at many times of the 367 day. Instead, for the data here, we have used the fitted path-averaged temperature, as discussed in 368 Section 2.4. The approach effectively relies on the spectral database but, in this case, on the variation in 369 the Boltzmann distribution of the J-level population with temperature. In Table 2, we have taken a 370 hopefully conservative uncertainty of 0.5 °C for the path-averaged temperature, but more work is 371 needed to establish the true uncertainty from these retrieved values. Finally, we note the fractional 372 uncertainty in the measured atmospheric pressure from the sensor or altitude-based pressure changes 373 across the optical path was below 0.36%.

374

375 4. Diurnal cycles and source analysis

The two weeks of open path data are analyzed for diurnal cycles, as shown in Figure 10 with the intent of an initial understanding of CO₂ and CH₄ sources. For this analysis, the wind speed and wind direction





were taken from the NCAR Mesa weather data (<u>ftp://ftp.eol.ucar.edu/pub/archive/weather/mesa/</u>),
while the gas concentrations are from DCS A.

380

381 4.1 Carbon dioxide

382 As expected, the median of the diurnal cycle for CO_2 shows a peak in the early to mid-morning 383 from commuter traffic after which the CO₂ concentration decreases as the boundary layer rises. It 384 remains approximately steady throughout the afternoon, decreases to a minimum between 19:00 and 385 20:00, and then increases slightly overnight as the boundary layer collapses. We hypothesize that the 386 afternoon behavior is due to the change in wind direction. Often overnight and through early morning 387 the wind blows from the west to southwest, which brings in cleaner background air from the mountains 388 bordering Boulder. However, in late morning the predominant wind direction shifts to the east and 389 southeast, possibly bringing in higher CO₂ concentrations from the Denver metropolitan area – which 390 lies approximately 30 km to the southeast of Boulder – over the course of the afternoon. Typically, the 391 evening wind shifts back to out of the west, once again bringing in the cleaner mountain air and with it a 392 decrease in CO₂ concentration.

393 394 4.2 Methane

Methane has a significantly weaker diurnal cycle than carbon dioxide, which is consistent with a species that lacks significant diurnally-varying local sources. Rather, its concentration follows expected variations in the boundary layer height; the concentration increases overnight into the early morning as the boundary layer collapses, and then decreases during the late morning through afternoon as the boundary layer rises again. The largest likely methane source near Boulder is local oil/gas fields, but these typically lie to the northeast, while the wind directions are generally out of the west to southeast. It is also possible that the methane comes from leaking natural gas infrastructure within the city.

402

403 5 Conclusions

404 Here we provide the first quantitative comparison of open-path dual comb spectroscopy 405 instruments. The dual-comb spectrometers were based on fully phase-coherent and stabilized fiber 406 frequency combs and operated nearly continuously over a two-week period. We performed these 407 measurements over adjacent 2-kilometer round-trip paths to measure concentrations of dry CO₂, dry 408 CH₄, H₂O, and HDO. The measured atmospheric absorbance spectra agree to better than 10^{-3} . 409 Correspondingly, we find excellent agreement between the retrieved concentrations from the two 410 instruments without the need for instrument calibration: over two weeks of near-continuous 411 measurements, the retrieved CO₂ concentrations agree to better than 0.14% (0.57 ppm), CH₄ 412 concentrations agrees to better than 0.35% (7.0 ppb), and H₂O concentrations agrees to better than 413 0.4% (36 ppm). These values are very close to the WMO compatibility goals. The remaining 414 disagreement is likely due to scattered stray light, polarization dependencies, and residual comb phase 415 noise. We further compare the DCS measurements to a cavity ringdown point sensor located along our 416 path. The measured dry CO_2 mole fraction agrees to within 1%, the CH₄ dry mole fraction to within 417 1.2%, and H_2O mole fraction to within 6.2%. However, this CRDS point sensor is directly calibrated to 418 the WMO scale for CO₂ and CH₄ while the DCS results are based on HITRAN 2008; we attribute the 419 disagreement in CO₂ and CH₄ to inaccurate line parameters in the HITRAN database. (Most of the water 420 discrepancy is attributed to the imperfect absolute water calibration of the CRDS point sensor.) Further 421 improvements to the spectral database should reduce these discrepancies. Finally, this open-path DCS 422 can exploit even broader spectrum combs up to 2.3 μm and down to 1.1 μm (Zolot et al., 2012; Okubo et 423 al., 2015), which would enable measurements of similar quality for ${}^{13}CO_2$, NH₃, N₂O, and O₂. These 424 results make open-path DCS a promising new system for greenhouse gas flux measurements from 425 distributed sources.





426	
427	The authors declare that they have no conflict of interest.

428

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	DCS A	DCS B				
Design Details						
Comb 1 repetition rate (<i>f</i> _r)	~200 MHz	~204 MHz				
Difference in repetition rate (Δf_r)	624 Hz	870 Hz				
Spectral filtering	Before combining combs	After combining combs				
Booster amplifier	Yes	No				
Average power launched	4 mW	1.5 mW				
Filtered spectral output	6376 to 6023 cm ⁻¹	6359 to 6003 cm ⁻¹				
Telescope design	Home-built 3"-diameter off-	Modified commercial 6"-				
	axis telescope	diameter Ritchey-Chretien				
		telescope				
Retroreflector	2.5" HCC, 5 arc seconds	5" HCC, 5 arc seconds				
Round-trip path length	1950.17 m	1963.67 m				
Typical averaging time	32 s	28 s				
Performance Metrics						
30-second precision	0.90 ppm CO ₂ , 9.6 ppb CH ₄	2.15 ppm CO ₂ , 11.5 ppb CH ₄				
5-minute precision	0.24 ppm CO ₂ , 2.1 ppb CH ₄	0.60 ppm CO ₂ , 3.2 ppb CH ₄				

578

579 Table 1: Specifications of the two DCS systems. HCC: hollow corner cube





Systematic source [effect]	Effect on retrieved CO ₂	Effect on retrieved CH ₄	Effect on retrieved H ₂ O
Fitting procedure [initial guess, baseline polynomial order and window size]	0.07 ppm	1.4 ppb	4 ppm
Rf detection and processing [rf reflections, ADC nonlinearities]	0.16 ppm	0.34 ppb	1.0 ppm
Telescope system [Scattered light, polarization effects]	0.45 ppm	1.5 ppb	56 ppm
Spectral database [linestrengths in HITRAN 2008]	1-2%	10-20%	5-10%
Temperature path inhomogeneities [if <10 C across path]	0.024 ppm	0.36 ppb	3.52 ppm
Path-averaged temperature [for 0.5 C uncertainty]	0.64 ppm	2.9 ppb	8.6 ppm

581

582 **Table 2:** List of systematic uncertainties. See discussion in Section 3.2 and 3.4 for more details. Upper

half of table: systematics due to hardware and software. Lower half of the table: systematics due tospectral model and temperature uncertainties.







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588 Figure 1. (a) Time and frequency domain overview of dual comb spectroscopy. Two frequency combs are phase-locked together 589 with pulse repetition rates of f_r and $f_r - \Delta f_r$. Their detected heterodyne signal is a series of interferograms in the time domain, or 590 equivalently a comb in the radio frequency (rf) domain. Provided the combs are sufficiently coherent and Nyquist sampling 591 conditions are met, each rf comb tooth maps to a particular, known pair of optical frequency comb teeth (Coddington et al., 592 2008). As a result, the optical spectrum can be obtained from the magnitude of each rf comb tooth versus the average optical 593 frequency of the relevant comb tooth pair. (b) Actual DCS spectrum acquired over 1.15 seconds for DCS A after transmission over 594 2-km air path. The overall shape is governed by the comb spectrum but there are narrow absorption dips present from 595 atmospheric gases, as shown in the first expanded view. The second expanded view shows the fully resolved rf comb teeth with 596 time-bandwidth limited widths. The highly resolved nature of these spectral elements translates to a negligible instrument 597 lineshape, set by the narrow comb linewidths. The sample points are separated by 0.0067 cm⁻¹ (or f_r =200 MHz). For long-term 598 averaging, we implement coherent co-adding of interferograms that effectively measures the power at the individual rf comb 599 teeth (Coddington et al., 2008)

600





602

 Combs, telescopes

 B path

 DCS B path

 DCS A path

609 space DCS paths.







Figure 3. (a) Configuration of DCS A and DCS B, both of which are based on fully self-referenced fiber-laser frequency combs. See
text and Table I for details. DCS A includes a booster amplifier for higher launched optical power than DCS B. (b) Average optical
return power for DCS A (blue) and DCS B (red) measured at the detector over about 2.5 days. The horizontal black line shows the
approximate minimum power for useable SNR (15 μW). Inset: The normalized power fluctuations for DCS A over 100 ms. The

615 acquisition time for a single DCS spectrum is shown by the thickness of the vertical grey bar. RC: Ritchey-Chretien; HNLF: highly

⁶¹⁶ nonlinear fiber; HCC: hollow corner cube retroreflector; BS: beam splitter.







618

619Figure 4. Raw spectra from DCS A (blue) and DCS B (red). (b) Corresponding baseline-corrected absorption spectra averaged for620a three-hour period. The spectra overlap completely on this scale so the DCS A absorbance has been flipped about zero. Inset:621expanded view of several CO_2 lines. (c) Difference between the absorption spectra from DCS A and DCS B. The difference is622shown both before (black trace) and after (grey trace) removing an etalon structure and agree to better than 5×10^{-4} after the623etalon is removed. Inset: Expanded view. (d) Residuals from a fit of the DCS A spectrum to HITRAN 2008. In general, the residuals624are lower noise than the difference spectrum because of the higher signal-to-noise ratio of the DCS A than DCS B, but there are625clear structures present near absorption lines due to imperfect line shapes of the spectral database.







 626
 Local Time

 627
 Figure 5. Concentration retrievals from DCS A (blue dots) and DCS B (red lines) for HDO (% by volume) H₂O (% by volume), dry

628 CH₄, dry CO₂ and dry ¹³CO₂ over two weeks at 30-second intervals. Excellent agreement is observed between both systems for all

629 species, though it is clear that over this path length ${}^{13}CO_2$ does not provide a strong enough signal to retrieve reliably.

630 Highlighted section: 6-hour, well-mixed period over which Allan deviations (Figure 8) are calculated. Missing data is primarily

631 *due to telescope misalignment and less frequently, phase lock by one of the combs.*















640







642 Averaging Time (seconds)
 643 Figure 8. Precision (Allan deviation) versus averaging time, τ, for CH₄ and CO₂ calculated for DCS A (blue) and DCS B (red) over a
 644 2 km path for the time period highlighted in Figure 3. The previously-published precisions from Rieker et al. (2014) are also

645 shown (gold). The grey line illustrates the slope expected for white noise. For DCS A, at averaging times from 30s to 1000s, the

646 precision roughly follows ~40 $ppbv/\sqrt{\tau}$ for CH4 and ~4 $ppmv/\sqrt{\tau}$ for CO₂ (gray lines).







647 648 Figure 9. Comparison between the open-path DCS A data (blue) and the point CRDS data (gold) for H₂O, dry CH₄, and dry CO₂ at

649 32-second intervals over two weeks. The lower three panels directly plot the corresponding difference between the two.

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 652
 Time of Day

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 Figure 10. Diurnal cycles for wind speed, wind direction, XCH₄, and XCO₂. Data from each day in Figure 5 is over-plotted in grey

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 along with the hourly mean (blue) and median (red) values. Uncertainty bars on the median values span the 75th quantile and

 655
 25th quantile.





657 Appendix A: Temperature studies

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As described in Section 2.4, we extract the path-averaged temperature directly from a fit to the 30013
← 00001 overtone band of CO₂. We perform this fit on 5-minute averages, rather than 32-second
averages, under the assumption the temperature changes are still slow at that timescale. This pathaveraged temperature is then used in a subsequent fit over the full spectral region to extract the column
densities, and finally the mole fractions. We use a common temperature (from DCS A fit) to analyze
both data sets in order to separate out instrument effects from the temperature, but the fitted
temperatures between instruments show less than 0.25 °C bias.

666 Figure A1 compares this fitted path-averaged temperature from DCS A to three point sensors, 667 two of which are located on the rooftop near the telescope launch point and one that is located ~ 2.2 668 km away at an altitude ~200 m above the overall open path. As shown in Figure A1, the two rooftop 669 temperature sensors located near the telescope agree well with each other, but do not agree with the 670 fitted path-averaged temperature. Moreover, that disagreement has a distinct diurnal character, 671 supporting the argument it arises from a real temperature gradient. In contrast, the path-averaged 672 temperature does often agree well with the temperature measured by the third temperature point 673 sensor located at similar or higher altitude as the open path on the NCAR Mesa building 674 (ftp://ftp.eol.ucar.edu/pub/archive/weather/mesa/). These data indicate that the point sensor located 675 at the telescope site is not a good proxy for the path-averaged temperature; instead, the fitted path-676 averaged temperature should be used for the concentration fits because of temperature gradients. Note that the temperature gradients themselves do not lead to appreciable errors in the retrieved mole 677

- 678 fractions if the correct path-averaged temperature is used (see Table 2 and Section 3.4).
- 679
- 680





682 Figure A1: The fitted path-averaged temperature over two-weeks at 5 minute intervals (red) compared 683 to the measured air temperature from a roof-top anemometer located near the telescope (blue), a 684 second thermistor temperature sensor also located on the roof but 100-m distant (black), and a third 685 rooftop temperature ~2.2 km distant at the NCAR Mesa facility (gold). Top panel: The difference 686 between the two rooftop temperature (black) agree to within 1°C, but the difference between these 687 rooftop sensors and fitted path-averaged temperature (red) shows larger 2-4°C diurnal differences, 688 indicating it is not sufficient to measure the temperature at one "end point" of the open path. In fact, 689 the path-averaged temperature agrees better with the more distant, but higher elevation temperature 690 sensor located at the NCAR Mesa facility.

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