Authors' response to comments on "Instrument Artifacts Lead to Uncertainties in Parameterizations of Cloud Condensation Nucleation", Revised Submission, Referee #1

Authors' response: We thank the Reviewer for her/his detailed feedback. In particular, we agree with the Reviewer that our assumptions regarding volume weighted components of kappa warranted revision and we have now revised that section accordingly. Specific modifications are discussed below.

General Comments

1) **Reviewer comment:** The authors responded to my comment. For Part B, I suggest that this response regarding to what extent the artifacts investigated here can explain the discrepancies in the kappa in the literature should be also incorporated into the revised manuscript.

   Authors' response: Agree. This was an oversight on our part. We have now added this discussion to the manuscript.

   Authors' changes to the manuscript: Page 18 Lines 320-325 the text now reads: "This analysis was also applied to the range of apparent hygroscopicity values Svenningsson et al., 2006 found for ammonium nitrate $0.577 \leq \kappa_{app} \leq 0.753$, with a mean value of 0.670. If 0.670 is assumed to be the true κ_{app} for ammonium nitrate, then the sample/sheath ratio used to determine κ_{app} (1.2-2 L min^{-1}) could lead an experimental kappa as low as 0.665 or as high as 0.674, which would not fully explain the actual experimental range. This assessment ignores possibility of under/over counting which could introduce additional errors."

2) **Reviewer comment:** Regarding my formed general comment 2

   A. I think that the assumption of the "volume-weighted approach accepted as a standard convention of kappa theory" is that different components are internally mixed on particles! Particles at different sizes are apparently not internally mixed. Taking an extreme example, if an aerosol population consists of black carbon particles for all particles <100 nm and (NH4)2SO4 particles for all particle >100 nm. If one measure D50 at 0.1% one would get a D50 of ~140 nm since the D50 of (NH4)2SO4 is ~140 nm. And from that D50 vs. SS, one would get a kappa of ~0.6 (the kappa value of (NH4)2SO4). Of course, the D50 vs. SS data sets obtained in this Case will not fall on the lines of constant kappa.

   B. This again leads to my concern about the approach used to derive artifacts from DMA in this study because it is different from the real CCN measurement. To clarify my comment, in the real CCN measurement, at first activated fractions of particles are

obtained, either activated fraction vs. supersaturation(SS) for particles of a given size or activated fraction vs. particle size at a given SS. The activation curve is then fitted to derive a D50 or SS50 and from D50 vs. SS or D vs. SS50, kappa is obtained. In order to investigate the uncertainties of kappa due to instrument artifacts, one would need simulate the data acquisition process of CCN activation measurement by simulating the number of particles and number of activated particles in each size bin (in the Case of D50 vs. SS) and then activation fraction and D50. I am not sure whether the artifacts in this study can reflect the real uncertainties in CCN measurement and is useful to get an idea of the uncertainties in CCN measurement.

A. Authors' response: The reviewer is correct, and we thank the reviewer for identifying this flaw in our previous analysis. Specifically, equation 12, read

$$\kappa_{app,theory} = \sum_i \epsilon_i \kappa_i \quad (12)$$

where ϵ_i is the volume fraction of aerosol of each diameter i, and κ_i is the perceived κ_{app} for each diameter (adapted from Petters and Kreidenweis [2007]). In this equation, there is an underlying assumption that all particles of a certain size have a certain composition, and that variations in composition (and thus kappa) occur only with corresponding variations in size. Since this assumption doesn't hold true for all aerosol populations, we have revised the analysis to include the active fraction as described below.

B**. Authors' response:** The conceptual changes detailed in 2A have been applied to the assessment of apparent hygroscopicity artifacts derived from DMA measurements.

Authors' changes to the manuscript to address the issues raised in 2A&B: In addition to the text below, Figures 4 and 10 have been updated, and Supplemental Figure S2 has been added.

Pages 16-17 Lines 272-317 the text now reads: "To test how uncertainties in DMA diameter translate to uncertainties in κ_{app}, the activation of particles downstream of the DMA was assessed. First, for each case and diameter (25, 50, 100, and 200 nm) the critical saturation ratio s_{crit} was calculated for each particle diameter range downstream from the DMA using Eq. 3a. These critical saturation ratios were converted to critical percent supersaturation SS_{crit} and used to calculate the activated fraction AF for the aerosol particles downstream from the DMA for percent supersaturations $0.01 < SS < 1.5$, using the equation:

$$AF = \frac{1}{2}\left(1 + \mathrm{erf}\left(\frac{SS - SS_{crit}}{\sigma\sqrt{2}}\right)\right) \quad (12)$$

Where the standard deviation σ was equal to one-hundredth of SS_{crit}. The small σ/SS_{crit} ratio was chosen in order to generate accurate activated fraction curves for each particle diameter.

The activated fraction curve for each selected diameter (25, 50, 100, and 200 nm) was then calculated as the sum of the number-weighted activated fractions of each particle diameter downstream from the DMA. For example, for a selected diameter of 25 nm, the downstream diameters ranged from 23 nm to 27 nm for DMA Case 1 and from 20 nm to 36 nm in DMA Case 4. The equation used for this calculation is:

$$AF_{weighted} = \sum_i \frac{n_i}{n_{total}} AF_i \quad (13)$$

where AF_i is the activated fraction calculated using Eq. 12 and $\frac{n_i}{n_{total}}$ is the fraction of particle downstream from the DMA of diameter i.

This calculation was repeated for each selected diameter (25, 50, 100, and 200 nm), each DMA Case (1-7), and percent supersaturation (0.01-1.5) in order to construct activation curves for each selected diameter and DMA Case. As an example, in Fig. S2, the shape and position of each activated fraction curve vary with the DMA flow ratios. As the aerosol/sheath ratio increases, the activated fraction curve flattens out (DMA Case 4). The critical percent supersaturation SS_{crit} was then determined for each activation curve as the percent supersaturation where $AF = 0.50$. These results are shown in Fig. 4a for ammonium sulfate and sodium chloride. Eq. 4 was then used to calculate $\kappa_{app,theory}$ for each DMA Case and selected diameter, as shown in Fig. 4b. Discrepancies between $\kappa_{app,theory}$ calculated in this study and literature values (hereon referred to as "κ_{app} artifacts") are shown for both compounds in Fig. 4c-d.

The largest κ_{app} artifact was found in DMA case 4 (where the aerosol/sheath ratio was the highest) for both ammonium sulfate and sodium chloride aerosols. The artifacts for 25 nm ammonium sulfate aerosol in DMA case 4 was 0.08 , or ~13% of the literature value used for $\kappa_{app}^{(NH_4)_2SO_4}$, while the artifacts for 25 nm sodium chloride in DMA case 4 was 0.16, or ~13% of the literature value used for κ_{app}^{NaCl}. Artifacts were also high for DMA case 6 ($-0.041 \leq \kappa_{app,artifact}^{(NH_4)_2SO_4} \leq -0.048$) and DMA case 7 ($0.014 \leq \kappa_{app,artifact}^{(NH_4)_2SO_4} \leq 0.024$), where sheath and excess flow were unequal. This result demonstrates that artifacts may still occur when low aerosol/sheath flow ratios are chosen (0.15 and 0.08 for DMA cases 6 and 7, respectively) due to small differences between sheath and excess flow rates (5% and 2% for DMA cases 6 and 7, respectively).

κ_{app} artifacts were larger for sodium chloride ($-0.10 \leq \kappa_{app,artifact}^{NaCl} \leq 0.16$) than for ammonium sulfate ($-0.05 \leq \kappa_{app,artifact}^{(NH_4)_2SO_4} \leq 0.08$) across the DMA cases. As our results show, when two or more compounds are compared, the more hygroscopic compound will have larger κ_{app} artifacts."

The edited Fig. 4 is shown below.

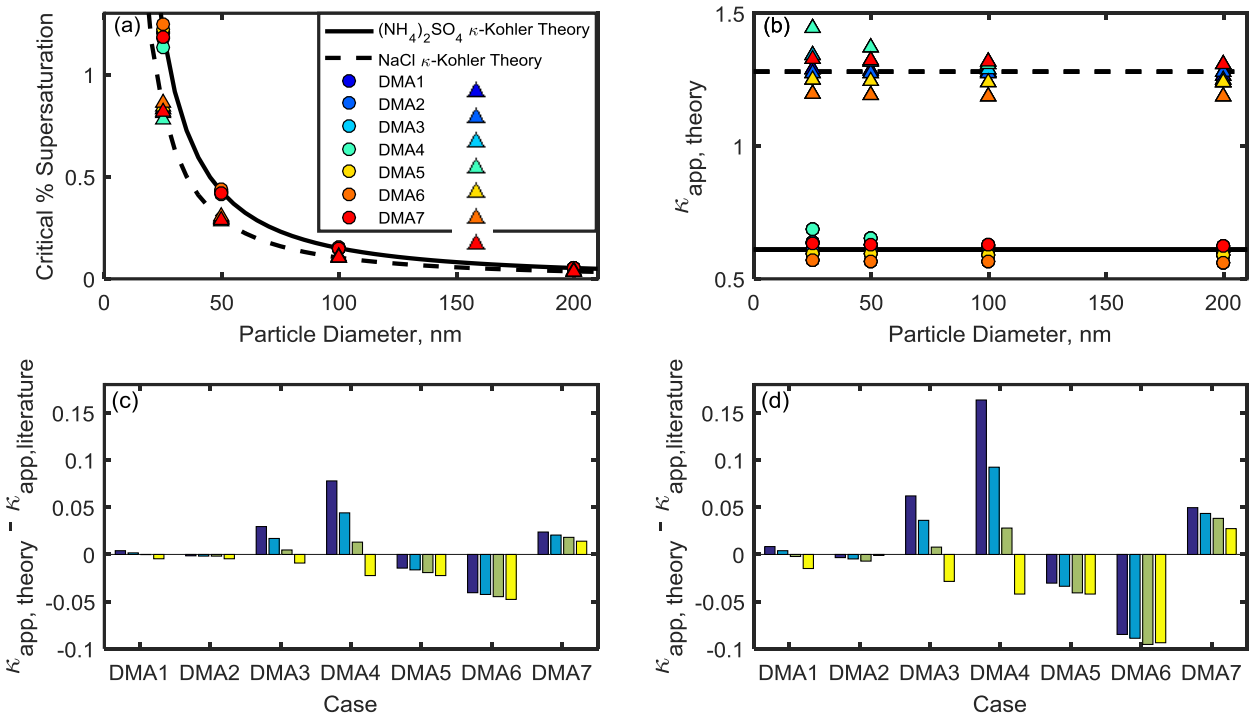


Figure 4 (a) Critical supersaturation of ammonium sulfate and sodium chloride particles calculated for DMA Cases 1-7 for sodium chloride (triangles) and ammonium sulfate (circles). Ammonium sulfate and sodium chloride curves from κ-Köhler theory are shown for comparison. (b) Apparent hygroscopicity κ_{app} for DMA Cases 1-7. (c) DMA-flow-derived artifacts in ammonium sulfate κ_{app} are shown for each DMA case. (d) DMA-flow-derived artifacts in sodium chloride κ_{app} are shown for each DMA case.

Fig. 10 has also been edited in order to include these new results.

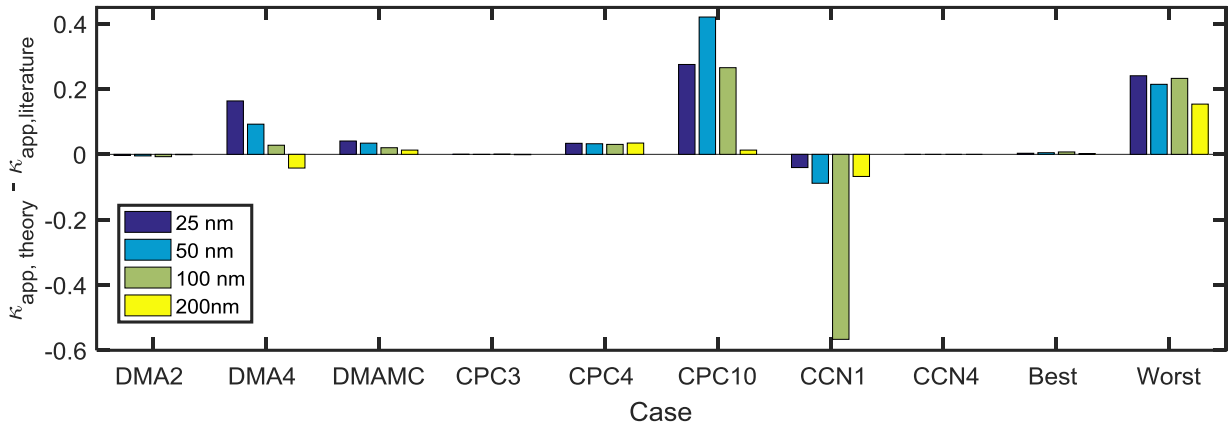
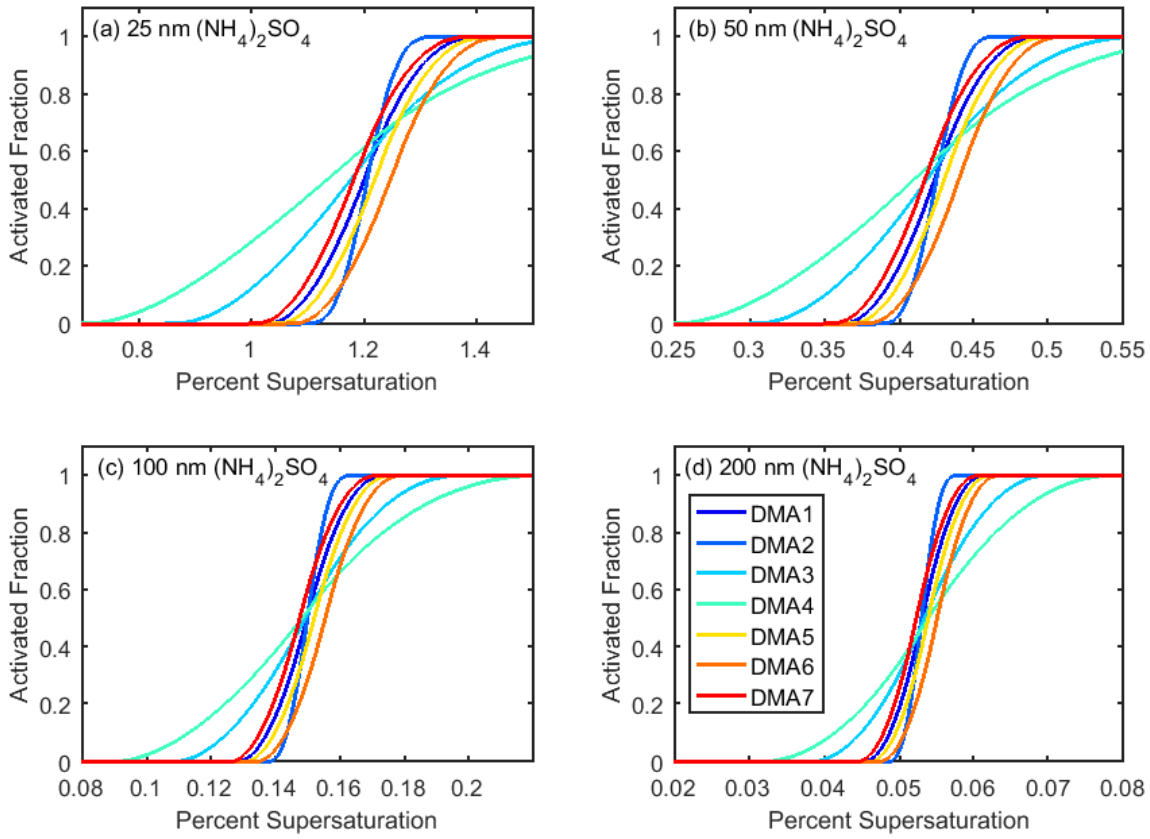


Figure 10 Comparison of κ_{app} artifacts derived from best and worst Case scenarios for instrumental measurements for sodium chloride. Combined artifacts for the lowest-artifact Cases (Best: DMA Case 2, CPC Case 3, and CCN Case 4) and the highest-artifact Cases (Worst: DMA Case 4, multiple charging, CPC Case 4, CPC Case 8, and CCN Case 1).



Added Figure S2 Exemplary $(NH_4)_2SO_4$ CCN activation curves for DMA Cases 1-7.

3) **Reviewer comment:** Regarding my formed general comment 3: I think one would like to see a more quantitative analysis of the uncertainties of kappa after taking the influence of instrument calibration by a standard compound into account. This is most relevant to real CCN measurement and is most interesting to those who do the measurement and who use these data. I also suggest that the discussion should be somewhat included in the conclusion because if after the calibration using (NH4)2SO4 or NaCl, the discrepancy caused by instrument artifacts would be much smaller than values shown in the manuscript.

Authors' response: As the Reviewer points out, accuracy in measurements depend on accurate calibrations. In this case, accurate determination of the supersaturation setpoints within the CCN instrument are dependent on accurate sizing of aerosols entering the CCN, and therefore are dependent on the DMA sizing during CCN calibration. CCN calibrations during two standard compounds, ammonium sulfate and sodium chloride, as described in detail in Rose (2008). Fortunately, if the calibration procedure described by Rose is followed and an optimal DMA aerosol to sheath ratios employed, the uncertainties will be minimal. Specifically, our analysis shows that an aerosol to sheath ratio of 1:10 or 1:20 (Case 1 or 2, respectively) is recommended for all CCN calibrations. This will result in kappa uncertainties of less than 1% for all dry sizes (25 to 200 nm). However, if CCN calibrations are performed using a DMA operated with less than ideal aerosol to sheath ratios, substantial errors will be introduced. Analysis of the impact of DMA uncertainties on CCN calibrations are discussed in detail in the Supplemental Materials. In the worst case scenario amongst the cases evaluated here (Case 4), the resulting uncertainty in apparent kappa is 15%.

Details of the assessment of impacts of DMA sizing on CCN calibration follow:

We have now assessed the effects of calibration with a standard compound on subsequent CCN measurements, given that the DMA flow settings used in the calibration are the same as those used for subsequent measurements. Calibration with a standard will yield new parameters, A and B, for a linear equation that relates percent supersaturation, $\%SS$, to the change in temperature set by the instrument, ΔT_{set}:

$$\Delta T_{set} = (A \times \%SS) + B$$

If the slope A and/or y-intercept B are inaccurate, the instrument will choose ΔT_{set} incorrectly for input percent supersaturation values. In order to model this error, the κ-Köhler theory $\%SS_{crit}$ for 25, 50, 100, and 200 nm ammonium sulfate and sodium chloride was used to determine the "correct" ΔT_{set} for the CCNC, using parameters A and B from a previous CCNC calibration in our lab. These ΔT_{set} were then paired with the $\%SS_{crit}$ for DMA Cases 1-7 determined in Section 3.1.2 κ_{app} artifacts arising from DMA flow ratios, as shown in Figure S3. A linear regression was run for each composition and DMA Case to

find new parameters A and B. For clarity, the original ΔT_{set} equation with the original parameters A_0 and B_0 will hereon be referred to as Eq. S1, and the new ΔT_{set} equations for each DMA Case C with new parameters A_C and B_C will be referred to as Eq. S2.

$$\Delta T_{set} = (A_0 \times \%SS) + B_0 \quad (S1)$$

$$\Delta T_{set} = (A_C \times \%SS) + B_C \quad (S2)$$

For each composition (sodium chloride or ammonium sulfate) and DMA Case, Eq. S2 was used to determine the ΔT_{set} that the CCNC would set to achieve a series of percent supersaturations (0.01-1.5%). Then, Eq. S1 was used to determine the actual percent supersaturation that would result from each ΔT_{set}. A few assumptions have been made so far: first, that the DMA aerosol/sheath ratio that was used during the calibration is also used in order to collect CCN activation data later, using the same compound; and second, that the original A_0 and B_0 used Eq. S1 were correct.

Two activated fraction curves were then plotted for each DMA Case (an example with DMA Cases 1-4 is shown in Figure S4). The activated fraction values for both curves were taken from the results in Section 3.1.2 κ_{app} artifacts arising from DMA flow ratios. The accurate percent supersaturation values (dashed lines) were obtained from the original equation, and the observed percent supersaturation values (solid lines) are the values that would be reported by the CCNC according to the new equations.

Then, $\%SS_{crit}$ was determined for each observed activated fraction curve, as shown in Fig. S5a. The apparent hygroscopicity κ_{app} was calculated for each DMA Case using Eq. 4, as shown in Fig. S5b. Apparent hygroscopicity artifacts are shown in Fig. S5c-d.

Authors' changes to the manuscript: Minor changes have been made to the text, and further details have been added to the supplemental information.

Pages 18-19 Lines 327-339 the text now reads: "In addition to the errors discussed above, accuracy in CCN measurements depend on the accuracy of the instrument calibration. Specifically, accurate determination of the percent supersaturation set points within the CCN instrument are dependent on accurate sizing of aerosols entering the CCN, and therefore are dependent on the DMA sizing during CCN calibration. CCN calibrations during two standard compounds, ammonium sulfate and sodium chloride, as described in detail in Rose (2008). Fortunately, if the calibration procedure described by Rose is followed and an optimal DMA aerosol to sheath ratios employed, the uncertainties will be minimal. Specifically, this analysis shows that an aerosol/sheath ratio of 1:10 or 1:20 (Case 1 or 2, respectively) is recommended for all CCN calibrations. This will result in κ_{app} uncertainties of less than 1%

for all dry sizes (25 to 200 nm). However, if CCN calibrations are performed using a DMA operated with less than ideal aerosol to sheath ratios, substantial errors will be introduced. Analysis of the impact of DMA uncertainties on CCN calibrations are discussed in detail in the Supplemental Materials. In the worst case scenario amongst the cases evaluated here (Case 4), the resulting uncertainty in κ_{app} is 15%."

Pages 36 Lines 700-703 the text now reads: "By extension, the issue of uncertain sizing by the DMA leads to added uncertainties in the CCN instrument calibrations which are strongly dependent on the chosen aerosol to sheath ration within the DMA. We recommend conducting all CCN calibrations with DMA aerosol to sheath ratio of 1:10 or 1:20 which will reduce kappa uncertainties to less than 1% for all dry sizes (25 to 200 nm)."

In addition, Figures S3-S5 and the associated text have now been added to the Supplemental Materials.

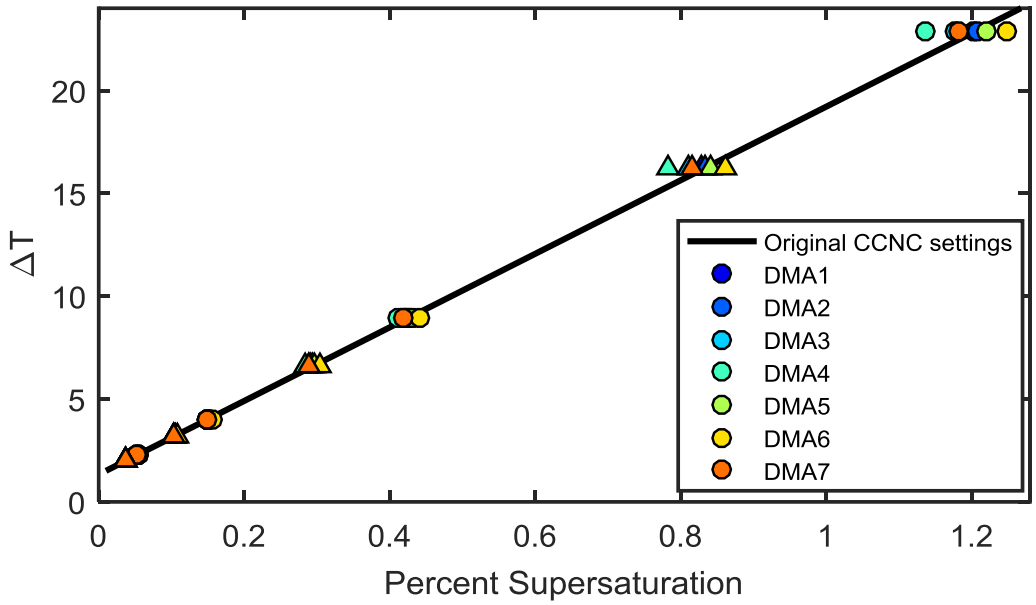


Figure S3 CCN data used to determine new parameters during calibration.

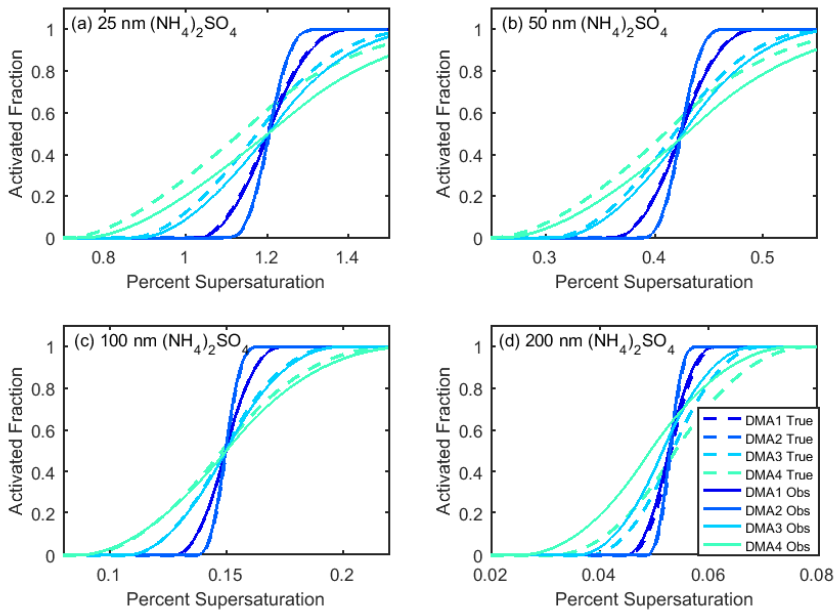


Figure S4 Exemplary $(NH_4)_2SO_4$ CCN activation curves for DMA Cases 1-7, resulting from inaccurate CCN instrument calibration. The true activation curve is shown with dashed lines, and the observed activation (inaccurate supersaturation reported by the instrument) is shown with solid lines.

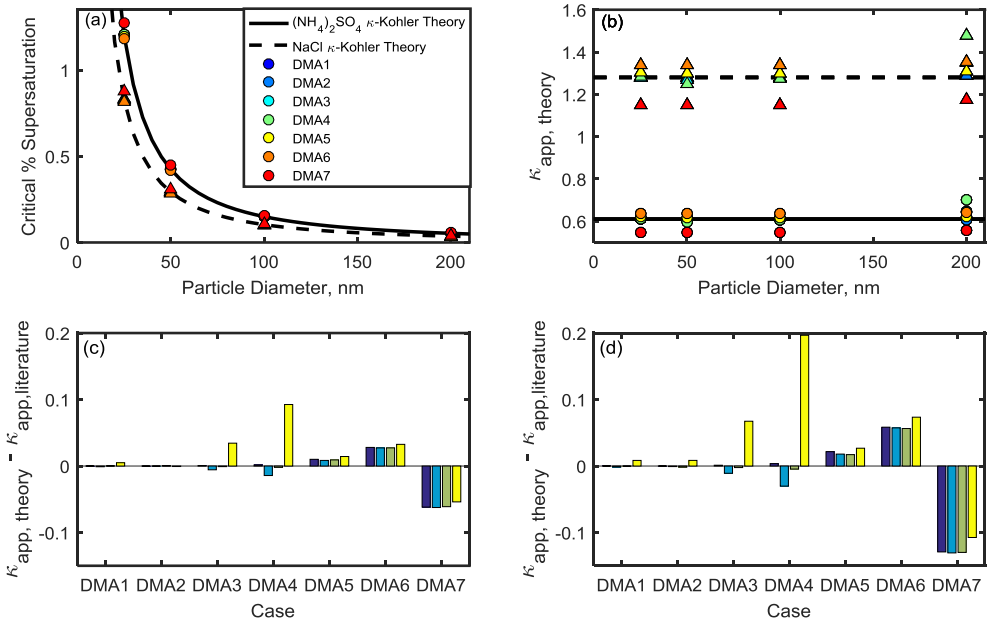


Figure S5 (a) Critical supersaturation of ammonium sulfate and sodium chloride particles calculated for DMA Cases 1-7 for sodium chloride (triangles) and ammonium sulfate (circles), following calibration with the same DMA settings. Ammonium sulfate and sodium chloride curves from κ-Köhler theory are shown for comparison. (b) Apparent hygroscopicity κ_{app} for DMA Cases 1-7. (c) DMA-flow-derived artifacts in ammonium sulfate κ_{app} are shown for each DMA case. (d) DMA-flow-derived artifacts in sodium chloride κ_{app} are shown for each DMA case.

Technical Comments

1. Reviewer comment: In figure 3a, in the lognormal distribution it is dN/dlogDp that follows the shape of the curve rather than N (number concentration, y-axis) or dN/dDp.

   Authors' response: Our distribution represents the number concentration.

Authors' response to comments on "Instrument Artifacts Lead to Uncertainties in Parameterizations of Cloud Condensation Nucleation", Revised Submission, Referee #3

Authors' response: We thank the Reviewer for her/his feedback, which we feel have improved the manuscript.
Additional specific modifications are discussed below.

1) Reviewer comment: Line 98-Differences in reported kapp values… 'ed' added

   Authors' response: We thank the reviewer for finding this error.

   Authors' changes to the manuscript: Page 6 Line 95 the text now reads: "Differences in reported κ_{app} values…"

2) Reviewer comment: Line 96-Differences in an aerosol's ability… 'an' added

   Authors' response: We thank the reviewer for finding this error.

   Authors' changes to the manuscript: Page 6 Line 99 the text now reads: "…rather any actual differences in an aerosol's ability…"

3) Reviewer comment: Line 643-Operating conditions: lowest DMA/sheath… 'colon and space issue'

   Authors' response: We thank the reviewer for finding this error.

   Authors' changes to the manuscript: Page 35 Lines 664-668 the text now reads: "The lowest combined artifacts ($0.0021 < \kappa_{app,artifact} < 0.0074$, NaCl) occurred as a result of ideal operating conditions: lowest DMA/sheath ratio, corrected multiple particle charging, and little to no undercounting."

4) Reviewer comment: The references are not spaced.

   Authors' response: We thank the reviewer for finding this error.

   Authors' changes to the manuscript: Spaces have been inserted between references.

5) Reviewer comment: Finally, the DMT CCN-100 can be operated under various flows from 200-900 cc/min although it is typically operated at 500 cc/min. The flow ratio (aerosol to

sheath) of the CCNC may also be altered but is typically set at 1:10. The paper should specify these details of operation.

Authors' response: We have added the recommended details to ***Section 5. Artifacts derived from cloud condensation nuclei instruments.***

Authors' changes to the manuscript: Page 30 Lines 559-562 the text now reads: "The total flow through the DMT CCN-100 is 0.20-0.90 L min^{-1}, though the instrument is typically operated with a total flow of 0.50 L min^{-1}. The aerosol/sheath ratio in the DMT CCN-100 is set by the user, and a ratio of 1:10 is commonly chosen."

Rose, D., Gunthe, S. S., Mikhailov, E., Frank, G. P., Dusek, U., Andreae, M. O., and Poschl, U.: Calibration and measurement uncertainties of a continuous-flow cloud condensation nuclei counter (DMT-CCNC): CCN activation of ammonium sulfate and sodium chloride aerosol particles in theory and experiment, Atmos. Chem. Phys., 8, 1153-1179, 10.5194/acp-8-1153-2008, 2008.

Instrument Artifacts Lead to Uncertainties in Parameterizations of Cloud Condensation Nucleation

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Abstract

The concentrations of cloud condensation nuclei (CCN) modulate cloud properties, rainfall location and intensity, and climate forcings. This work assesses uncertainties in CCN measurements and the apparent hygroscopicity parameter (κ_{app}) which is widely used to represent CCN populations in climate models. CCN measurements require accurate operation of three instruments: the CCN instrument, the differential mobility analyzer (DMA), and the condensation particle counter (CPC). Assessment of DMA operation showed that varying the ratio of aerosol to sheath flow from 0.05 to 0.30 resulted in discrepancies between the κ_{app} values calculated from CCN measurements and the literature value. Discrepancies were found to increase from < 1% to 13% for both sodium chloride and ammonium sulfate. The ratio of excess to sheath flow was also varied, which shifted the downstream aerosol distribution towards smaller particle diameters (for excess flow < sheath flow) or larger particle diameters (for excess flow > sheath flow) than predicted. For the CPC instrument, undercounting occurred at high concentrations, resulting in calculated κ_{app} lower than the literature values. Lastly, undercounting by CCN instruments at high concentration was also assessed, taking the effect of supersaturation on counting efficiency into account. Under recommended operating conditions, the combined DMA, CPC, and CCN uncertainties in κ_{app} are 1.2 % or less for 25 to 200 nm diameter aerosols.

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1. Introduction

Aerosol-cloud interactions represent a major uncertainty in current predictions of the Earth's climate (IPCC, 2013). According to well-known Köhler theory, an aerosol's potential to catalyze cloud droplet formation by activating as a cloud condensation nucleus (CCN) depends on its physical and chemical properties. For any given composition, the CCN activation potential of an aerosol increases as its diameter decreases. While the relationship between aerosol diameter and CCN activation is straightforward, the effect of composition on an aerosol's ability to participate in cloud formation is more complex (Petters and Kreidenweis, 2013; Ovadnevaite et al., 2011). Predicting the cloud forming capacity of various air masses based on the properties of the aerosol they contain is essential for evaluating relative contributions from pollution, continental background and marine aerosol sources (Brooks and Thornton, 2018; Carslaw et al., 2013). Long-term CCN measurements are available from numerous locations globally (Schmale et al., 2018). However, understanding regional and temporal variability in CCN populations requires the ability to assess whether observed differences reflect true physical differences or simply variations in CCN sampling strategies.

Parameterizations of CCN activity which accurately prescribe CCN measurements are needed for climate models, cloud resolving models, and air quality predictions (Betancourt and Nenes, 2014; Betancourt et al., 2013; Chang et al., 2017; Crosbie et al., 2015; Karydis et al., 2012; Kawecki and Steiner, 2018). One parameterization was designed to represent the cloud droplet activation potential ambient aerosol masses of unknown composition with a single variable, kappa (κ) based on the dry aerosol's hygroscopicity, or ability to uptake water and form a solution droplet (Petters and Kreidenweis, 2007). Various names and abbreviations have been given to κ throughout the literature: "hygroscopicity parameter", "single hygroscopicity parameter", κ (Petters and Kreidenweis, 2007;

Carrico et al., 2008; Asa-Awuku et al., 2010; Moore et al., 2012b); "CCN-derived κ", κ_{CCN} (Carrico et al., 2008; Petters and Kreidenweis, 2007); and the "apparent hygroscopicity parameter" κ_{app} (Sullivan et al., 2009; Collins et al., 2016; Petters and Kreidenweis, 2013). The term *apparent* hygroscopicity is favored by many because it emphasizes that fact that while CCN activation can often be predicted accurately by hygroscopic water uptake, they are different physical processes. It is possible for a compound to have high intrinsic hygroscopicity and low apparent hygroscopicity if it is poorly soluble in water (Sullivan et al., 2009).

Parameterizations of hygroscopicity that pre-date Petters and Kreidenweiss 2007 exist as well. Winkler 1973 developed an equation for approximating the growth of an aerosol particle with relative humidity, based on the quantity and physical characteristics of the soluble species in the particle. Another approximation for the relationship between the equilibrium size of a particle and relative humidity was derived by Fitzgerald in 1975, in which the soluble fraction and composition of the soluble component(s) are taken into account. Fitzgerald et al., 1982 derived a particle composition parameter using the mass fraction and physical properties of soluble material in a particle. Kreidenweis et al., 2005 determined that the critical activation diameter of dry aerosol particles can be calculated from simplified Köhler theory using the physical properties of water and the solute in a solution droplet. This parameterization has been used in CCN closure studies (Bougiatioti et al., 2009; Moore et al., 2011; Moore et al., 2012a). The earliest prediction of CCN concentrations for specific particle diameters and hygroscopicity used this parameterization as well (Mochida et al., 2006).

Once calculated, hygroscopicity parameters are useful tools for comparing CCN field measurements conducted in various regions and seasons and for making predictions about cloud formation, aerosol-cloud interactions in weather, and climate models. Values of κ_{app} can be used to compare the CCN

results in field and laboratory studies, including sea spray aerosol. For example, aggregation of results from several mesocosm experiments and marine field studies found submicron (30-100 nm) κ_{app} for sea spray aerosol as low as 0.4 and as high 1.3 (Collins et al., 2016). Another study, which included a survey of observational CCN data, reported that marine and continental aerosols could be described by κ_{app} values of 0.7 ± 0.2 and 0.3 ± 0.1 respectively (Andreae and Rosenfeld, 2008).

Several studies have examined the sensitivity of models to κ values derived from HTDMA measurements. An analysis of the NASA Global Modeling Initiative Chemical Transport Model and the GEOS-Chem CTM (Karydis et al., 2012) found that cloud droplet number concentration is sensitive to κ in Arctic and remote regions, where background aerosol loadings are low. Another study (Betancourt and Nenes, 2014) found that a ±50 % uncertainty range in the κ of secondary organic aerosols and particulate organic matter resulted in a cloud droplet number concentration uncertainty of up to 15 % and 16 %, respectively. Updating precipitation models with lab-derived κ values for specific inorganic and organic species may increase the accuracy of storm forecasts by providing better predictions of intense precipitation (Kawecki and Steiner, 2018). In terms of climate, (Liu and Wang, 2010) found that increasing the κ of primary organic aerosols from 0 to 0.1, and decreasing the κ of secondary organics aerosols from 0.14 to 0.07, resulted in an uncertainty in global secondary aerosol indirect forcing of 0.4 Wm^{-2} from pre-industrial times to present day.

The sensitivity of weather and climate models to hygroscopicity parameters demonstrates the need for accurate measurements. In this study, we examine experimental uncertainties in CCN measurements and the resulting uncertainties in determination of κ_{app}. Differences in reported κ_{app} values may result from experimental artifacts rather any actual differences in an aerosol's ability to facilitate cloud formation. By systematically quantifying sources of experimental error, this study provides a

framework for determining the significance of variations in CCN properties reported in multiple studies and defining the operating conditions which minimize instrumental artifacts.

2. Background

The Köhler equation relates water vapor saturation ratio at the surface of a wet droplet, s, to its radius at equilibrium (Rogers and Yau, 1989):

$$s = \left(1 - \frac{b}{r^3}\right) \exp\left(\frac{a}{r}\right) \tag{1a}$$

$$a = \frac{2\sigma_w M_w}{\rho_w R T} \tag{1b}$$

$$b = \frac{3 i m_s M_w}{4\pi \rho_w M_s} \tag{1c}$$

where s is the equilibrium saturation ratio of a solution droplet with radius r, σ_w is the surface tension of water, M_w is the molecular weight of water, R is the ideal gas constant, T is temperature in Kelvin, ρ_w is the density of water, and M_s is the molecular weight of the solute. The minimum saturation ratio that is required for spontaneous droplet growth, s_{act}, is therefore:

$$s_{crit} = 1 + \sqrt{\frac{4a^3}{27b}} \tag{2}$$

Petters and Kreidenweis [2007] reformulated the Köhler equation as κ-Köhler theory:

$$s_{crit} = exp\left(\sqrt{\frac{4A^3}{27 D_{act}^3 \kappa_{app}}}\right) \tag{3a}$$

and

$$A = \frac{4\sigma_{lv}M_w}{RT\rho_w} \quad (3b)$$

Where s_{crit} is the critical water vapor saturation ratio, D_{act} is the dry particle activation diameter and κ_{app} is the apparent hygroscopicity parameter. Solving for κ_{app} yields:

$$\kappa_{app} = \frac{4A^3\sigma_{lv}^3}{27T^3D_{act}^3 ln^2(s_{crit})} \quad (4)$$

The apparent hygroscopicity parameter can be calculated from experimental CCN results, where the dry diameter and water vapor saturation ratio are known. For a chosen aerosol diameter, the activated fraction is the ratio of the concentration aerosols that activate as CCN to the total aerosol concentration:

$$Activated\ fraction = \frac{CCN\ Concentration}{Aerosol\ Concentration} \quad (5)$$

Activated fraction data is fit with a sigmoid error function to determine the percent supersaturation at which 50 % of the particles have activated as CCN (activated fraction = 0.50), which is considered the operationally defined critical percent supersaturation SS_{crit} (Rose et al., 2008). The critical saturation ratio s_{crit} can then be determined and entered into Eq. (4) in order to calculate κ_{app} for the near-monodisperse aerosol:

$$s_{crit} = 1 + \frac{SS_{crit}}{100} \quad (6)$$

Reporting κ_{app} as a function of diameter allows for the comparison of the cloud condensation nucleation abilities of multimodal aerosol populations, without overlooking differences which arise due to aerosol composition.

The apparent hygroscopicity parameter is related to chemical composition; therefore, the calculated κ_{app} of a pure substance should be constant across CCN experiments. However, discrepancies between κ_{app} for a single chemical species have been observed. Experimental results for ammonium nitrate are inconsistent with reported values ranging from $0.577 \leq \kappa_{app} \leq 0.753$ (Svenningsson et al., 2006). Also, large ranges are often observed for organic compounds, such as glutaric acid ($0.054 \leq \kappa_{app} \leq 0.16$) and malonic acid ($0.199 \leq \kappa_{app} \leq 0.255$) (Koehler et al., 2006; Kumar et al., 2003; Hartz et al., 2006). Below we evaluate potential sources of uncertainties in CCN measurements and the resulting uncertainties in κ_{app}.

3. Artifacts derived from sized CCN measurements

CCN measurements used for calculating apparent hygroscopicity from monodisperse aerosol require accurate operation of three instruments: the CCN, the differential mobility analyzer (DMA), and the condensational particle counter (CPC). The setup for laboratory CCN experiments is shown in Fig. 1. First, a polydisperse population of aerosols is generated by an atomizer and dried using a desiccant tube packed with silica gel. A near-monodisperse flow is obtained through size-selection in the DMA. The flow is then split between a CPC (which measures aerosol concentration) and a CCN counter (which measures the concentration of particles that activate as cloud condensation nuclei at a given percent supersaturation). Instrument artifacts will first be assessed separately for the DMA, CPC, and CCN counter. In the concluding section of the paper (and Fig. 10), the overall uncertainty due to the combination of these is presented and discussed.

We note that this study considers sized CCN measurements which may be used for the determination of κ_{app}. In contrast, a number of earlier CCN studies conducted on the full ambient aerosol population without sizing the aerosol (Jennings et al., 1996; Hudson and Xie, 1998; Modini et al., 2015; Duan et al., 2017; Schmale et al., 2018; Leng et al., 2013). While useful, such studies do not produce the data required for accurate determination of κ_{app} from the CCN measurements.

3.1 Artifacts derived from differential mobility analyzers

3.1.1 DMA operation and electrical mobility

Differential mobility analyzers used in atmospheric science include commercially available instruments from Grimm Aerosol Technik, TSI Incorporated, and MSP Corporation. They have also been custom built by a number of research groups (Mei et al., 2011;Barmpounis et al., 2016;Jokinen and Makela, 1997;Seol et al., 2000). All models allow for the selection of particles through electrical mobility, the ability of a particle to move through a medium (such as air) while acted upon by an electrical field. The DMA size-selects near-monodisperse aerosol from a polydisperse aerosol source, as shown in Fig. 2 (modeled after the Vienna-type long Differential Mobility Analyzer from Grimm Technologies). The electrical mobility Z_p of a particle with mobility diameter d_m can be calculated according to:

$$Z_p = \frac{neC_C(d_m)}{3\pi\eta d_m} \tag{7}$$

where n is the number of charges on the particle (assumed to be one in this study), e is the elementary unit of charge, η is the gas dynamic viscosity, and $C_C(d_m)$ is the Cunningham slip correction factor:

$$C_C(d_m) = 1 + \frac{2\lambda}{d_m}\left(\alpha_{CC} + \beta_{CC}\exp\left[-\frac{\gamma_{CC}}{2\lambda/d_m}\right]\right) \tag{8}$$

where λ is the mean free path (DeCarlo et al., 2004). For the Vienna-type long Differential Mobility Analyzer from Grimm Technologies, Inc. considered here, $\alpha_{CC} = 1.246$, $\beta_{CC} = 0.42$, and $\gamma_{CC} = 0.86$ (Grimm Aerosol Technik, 2009).

Particle-laden flow enters the differential mobility analyzer through the aerosol inlet (flow Q_a), and travels down the DMA column (inner radius r_1, outer radius r_2) with the clean air sheath flow Q_{sh}. Positively-charged particles are attracted by the negatively-charged inner electrode, to which voltage V_0 has been

applied. Ideally, selection of a voltage allows only particles of a specific mobility diameter to exit the DMA through the sample flow Q_s. All particles with larger diameter (lower Z_p) or smaller diameter (higher Z_p) will exit the DMA through the excess flow Q_e. In other words, Q_s would ideally consist only of aerosols with diameters equal to, or very nearly equal to, the selected diameter.

In reality, the aerosol flow that leaves the DMA through Q_s is polydisperse with a mobility distribution determined by instrumental parameters. A triangular approximation has been chosen as a model for this distribution, as particle inertia is negligible for the diameters considered in this study (Stratmann et al., 1997; Mamakos et al., 2007). The probability that a particle at the aerosol inlet will exit with the sampling flow is defined by transfer function $f\left(Z_p, Z_{p,mid}\right)$:

$$f\left(Z_p, Z_{p,mid}\right) = \frac{\alpha_{TF}}{2\beta_{TF}}\left(\left|\frac{Z_p}{Z_{p,mid}} - (1+\beta_{TF})\right| + \left|\frac{Z_p}{Z_{p,mid}} - (1-\beta_{TF})\right| - 2\left|\frac{Z_p}{Z_{p,mid}} - 1\right|\right) \quad (9)$$

where $Z_{p,mid}$ is the midpoint mobility of the transfer function, and α_{TF} and β_{TF} are flow-derived constants, defined as:

$$\alpha_{TF} = \frac{Q_s + Q_a}{2Q_a} \quad (10a)$$

and

$$\beta_{TF} = \frac{Q_s}{Q_{sh}} \quad (10b)$$

The midpoint and half-width of the transfer function are respectively calculated according to: (Knutson and Whitby, 1975)

$$Z_{p,mid} = \frac{Q_e + Q_{sh}}{4\pi L V_0} \ln\left(\frac{r_2}{r_1}\right) \quad (11a)$$

and

$$\Delta Z_p = \frac{Q_a}{2\pi L V_0} \ln\left(\frac{r_2}{r_1}\right) \quad (11b)$$

where L is the distance between the DMA inlet and outlet.

3.1.2 κ_{app} artifacts arising from DMA flow ratios

Next we assess the ramifications of the DMA transfer function for the derived κ_{app}. A lognormal theoretical aerosol number distribution was used to represent a polydisperse ambient aerosol population (Fig. 3a). This distribution was converted to an electrical mobility distribution using Eq. (7) and Eq. (8), assuming that the aerosols in the distribution were spherical and singly charged. From the distribution, a series of single aerosol sizes were selected (25, 50, 100, and 200 nm diameter). For each aerosol size, the resulting DMA transfer functions were calculated for seven cases using Eq. (9) and the various parameters for DMA sheath, excess, aerosol, and sample flow listed in Table 1. These seven cases were chosen to represent possible measurements scenarios that may be encountered in a CCN experiment. The aerosol/sheath ratio is varied in Cases 1-4 in order to study the effects of chosen experimental parameters. Sheath flow is predetermined in some DMAs (for example, the Grimm Vienna DMA considered in this study), but can be varied in other instruments. The aerosol flow rate may also be selected in an experiment. Cases 5-7 vary the excess/sheath ratio in order to take proper instrument operation into account. The excess and sheath flow should be identical, but small discrepancies may occur.

For example, the resulting DMA transfer functions for a 100 nm aerosol conditions constrained by Cases 1-4 are shown in Fig. 3b, where an increase in Q_a/Q_{sh} from 0.1 (black line) to 0.3 (green line) tripled the width of the number distribution, and decreasing Q_a/Q_{sh} to 0.05 (blue line) from 0.10 halved the width of the number distribution. The result of applying the transfer functions shown in Fig. 3b to the distribution in Fig. 3a is shown in Fig. 3c.

All downstream distributions for all seven DMA cases and all aerosol sizes are shown in Fig. S1 in the Supplement. DMA Cases 1-4 represent experimental conditions in which the sheath and excess air flows are equal and the aerosol/sheath flow ratio is varied. As Q_a/Q_{sh} increases, the width of the number distribution measured downstream of the DMA increases, while the midpoint diameter remains constant. It was found that doubling the aerosol to sheath ratio doubled the width of the downstream number distribution for 25, 50, 100, and 200 nm particles. For example, when selecting 200 nm particles, increasing Q_a/Q_{sh} from 0.10 to 0.20 increased the downstream diameter range from 181-222 nm (a spread of 41 nm) to 167-250 nm (a spread of 87 nm). The particle diameter ranges that would be observed downstream of the DMA are summarized in Table 2.

To assess the variations in CCN properties resulting from DMA uncertainties the critical percent supersaturation were calculated for representative atmospheric aerosols. The value of SS_{crit} was calculated for each particle diameter using Eq. (3a), using literature values for apparent hygroscopicity of 0.61 for ammonium sulfate and 1.28 for sodium chloride (Clegg et al., 1998). It should be noted that this analysis considers two homogeneous aerosol distributions of hygroscopic salts. Real aerosol distributions tend to be mixtures of many species, and the shape of the number distribution can vary between species.

To test how uncertainties in DMA diameter translate to uncertainties in κ_{app}, the activation of particles downstream of the DMA was assessed. First, for each case and diameter (25, 50, 100, and 200 nm) the critical saturation ratio s_{crit} was calculated for each particle diameter range downstream from the DMA using Eq. 3a. These critical saturation ratios were converted to critical percent supersaturation SS_{crit} and used to calculate the activated fraction AF for the aerosol particles downstream from the DMA for percent supersaturations $0.01 < SS < 1.5$, using the equation:

$$AF = \frac{1}{2}\left(1 + \text{erf}\left(\frac{SS - SS_{crit}}{\sigma\sqrt{2}}\right)\right) \quad (12)$$

where the standard deviation σ was equal to one-hundredth of SS_{crit}. The small σ/SS_{crit} ratio was chosen in order to generate accurate activated fraction curves for each particle diameter.

The activated fraction curve for each selected diameter (25, 50, 100, and 200 nm) was then calculated as the sum of the number-weighted activated fractions of each particle diameter downstream from the DMA. For example, for a selected diameter of 25 nm, the downstream diameters ranged from 23 nm to 27 nm for DMA Case 1 and from 20 nm to 36 nm in DMA Case 4. The equation used for this calculation is:

$$AF_{weighted} = \sum_i \frac{n_i}{n_{total}} AF_i \quad (13)$$

where AF_i is the activated fraction calculated using Eq. 12 and $\frac{n_i}{n_{total}}$ is the fraction of particle downstream from the DMA of diameter i.

This calculation was repeated for each selected diameter (25, 50, 100, and 200 nm), each DMA Case (1-7), and percent supersaturation (0.01-1.5) in order to construct activation curves for each selected diameter and DMA Case. As an example, in Fig. S2, the shape and position of each activated fraction curve vary with the DMA flow ratios. As the aerosol/sheath ratio increases, the activated fraction curve flattens out (DMA Case 4). The critical percent supersaturation SS_{crit} was then determined for each activation curve as the percent supersaturation where $AF = 0.50$. These results are shown in Fig. 4a for ammonium sulfate and sodium chloride. Eq. 4 was then used to calculate $\kappa_{app,theory}$ for each DMA Case and selected diameter, as shown in Fig. 4b. Discrepancies between $\kappa_{app,theory}$ calculated in this study and literature values (hereon referred to as "κ_{app} artifacts") are shown for both compounds in Fig. 4c-d.

The largest κ_{app} artifact was found in DMA case 4 (where the aerosol/sheath ratio was the highest) for both ammonium sulfate and sodium chloride aerosols. The artifacts for 25 nm ammonium sulfate aerosol in DMA case 4 was 0.08 , or ~13% of the literature value used for $\kappa_{app}^{(NH_4)_2SO_4}$, while the artifacts for 25 nm sodium chloride in DMA case 4 was 0.16, or ~13% of the literature value used for κ_{app}^{NaCl}. Artifacts were also high for DMA case 6 ($-0.041 \leq \kappa_{app,artifact}^{(NH_4)_2SO_4} \leq -0.048$) and DMA case 7 ($0.014 \leq \kappa_{app,artifact}^{(NH_4)_2SO_4} \leq 0.024$), where sheath and excess flow were unequal. This result demonstrates that artifacts may still occur when low aerosol/sheath flow ratios are chosen (0.15 and 0.08 for DMA cases 6 and 7, respectively) due to small differences between sheath and excess flow rates (5% and 2% for DMA cases 6 and 7, respectively).

κ_{app} artifacts were larger for sodium chloride ($-0.10 \leq \kappa_{app,artifact}^{NaCl} \leq 0.16$) than for ammonium sulfate ($-0.05 \leq \kappa_{app,artifact}^{(NH_4)_2SO_4} \leq 0.08$) across the DMA cases. As our results show, when two or more compounds are compared, the more hygroscopic compound will have larger κ_{app} artifacts.

This analysis was also applied to the range of apparent hygroscopicity values Svenningsson et al., 2006 reported for ammonium nitrate $0.577 \leq \kappa_{app} \leq 0.753$, with a mean value of 0.670. If 0.670 is assumed to be the true κ_{app} for ammonium nitrate, then the sample/sheath ratio used to determine κ_{app} (1.2-2.0 L min^{-1}) could lead an experimental kappa as low as 0.665 or as high as 0.674, which would not fully explain the actual experimental range. This assessment ignores possibility of under/over counting which could introduce additional errors.

In addition to the errors discussed above, accuracy in CCN measurements depend on the accuracy of the instrument calibration. Specifically, accurate determination of the percent supersaturation set points within the CCN instrument are dependent on accurate sizing of aerosols entering the CCN, and therefore are dependent on the DMA sizing during CCN calibration. CCN calibrations during two standard compounds, ammonium sulfate and sodium chloride, as described in detail in Rose (2008). Fortunately, if the calibration procedure described by Rose is followed and an optimal DMA aerosol to sheath ratios employed, the uncertainties will be minimal. Specifically, this analysis shows that an aerosol/sheath ratio of 1:10 or 1:20 (Case 1 or 2, respectively) is recommended for all CCN calibrations. This will result in κ_{app} uncertainties of less than 1% for all dry sizes (25 to 200 nm). However, if CCN calibrations are performed using a DMA operated with less than ideal aerosol to sheath ratios, substantial errors will be introduced. Analysis of the impact of DMA uncertainties on CCN calibrations

are discussed in detail in the Supplemental Materials. In the worst case scenario amongst the cases evaluated here (Case 4), the resulting uncertainty in κ_{app} is 15%.

3.1.3 Effect of double and triple charges on particles

During normal operation, the Grimm DMA employs a bipolar charger (also known as a neutralizer) to charge aerosol particles through the capture of gaseous ions. The analysis in Section 3.1.2 assumes that each particle carries a single (+1) charge. In reality, the methods used to charge particles prior to entering a DMA may impart two, three, or more charges to individual particles (Fuchs, 1963). The charge distribution resulting from a bipolar charger is roughly approximated using the Boltzmann law (Keefe et al., 1959). However, the Boltzmann law assumes symmetric aerosol particle charging (equal concentrations of negatively and positively charged particles). Deviation from symmetric charging is observed in regions of high ionizations, and this deviation becomes more pronounced as particle size increases (Hoppel and Frick, 1990). A more accurate estimation of stationary charge distribution has been calculated using an approximation formula for the charge distribution produced by a bipolar charger:

$$f(k) = 10^{\left[\sum_{i=0}^{i=5} a_i(k)(\log_{10} D_{nm})^i\right]} \quad (14)$$

where $f(k)$ is the fraction of particles carrying k charges, $a_i(k)$ are approximation coefficients determined using a least-squares regression analysis, and D_{nm} is the particle diameter in nanometers (Wiedensohler, 1988). The approximation coefficients only apply to particles with 0, ±1, and ±2 charges. In a separate study, Maricq et al., 2008 determined approximation coefficients for poly (α-olefin) oligomer oil droplets with ±1, ±2, and ±3 charges. The approximation coefficients reported by

these two studies were in excellent agreement for particles with ±1 and in weak agreement for ±2 charges (+2 and -2 charging efficiencies were overestimated by 50% and 100%, respectively). Therefore, this analysis will use the approximation coefficients from Wiedensolher, 1988 for particles with +1 and +2 charges, and the approximation coefficient for particles with +3 charge from Maricq et al., 2008.

In order to assess the impact of multiple charges on κ_{app}, Eq. (14) and the approximation coefficients from Wiedensolher, 1988 and Maricq et al., 2008 were used to calculate the charge distribution of the representative aerosol population shown in Fig. 3a. The resulting charge distribution is shown in Fig. S6a. An increase in multiple charging is observed as particle diameter increases, though this is offset somewhat by the decrease in concentration with particle size above 50 nm.

It follows that aerosols incorrectly sized due to double and triple changing will be passed from the DMA to the CCN and result in an additional uncertainty in the CCN measurements. To illustrate this, activated fraction curves, were generated for 25, 50, 100, and 200 nm sodium chloride particle selection by the DMA (Fig. 5). The activation of sodium chloride is represented by sigmoid curves, where the midpoint of each activation curve is the κ-Köhler-derived critical supersaturation of sodium chloride, and the standard deviation of each curve is one-tenth of this value (consistent with the standard deviation/midpoint ratio observed from our instrument's ammonium sulfate CCN calibration data). For each particle diameter, D, the observed activated fraction, $AF_{D,weighted}^{SS}$, for each percent supersaturation SS was determined by weighting the activated fraction $AF_{D,i}^{SS}$ of each particle diameter/charge at that percent supersaturation, by the fraction of particles of that diameter:

$$AF_{D,weighted}^{SS} = \sum_{i=1}^{i=3} \frac{concentration\ of\ particles\ with\ charge\ i\ and\ diameter\ D}{concentration\ of\ particles\ with\ charge{+}1{,+}2{,+}3{,and\ diameter\ D}} AF_{D,i}^{SS} \qquad (15)$$

386

387 The raw data shown in Fig. 5 (green curves) can be corrected for multiple charging by determining the
388 fraction of particles with > +1 charge from the lower plateau in each plot (dashed lines). The adjusted
389 activated fraction for each percent supersaturation, $AF_{adjusted}$, is calculated using the equation:

390

391 $$AF_{adjusted} = \frac{AF_{raw} - AF_{plateau}}{1 - AF_{plateau}} \quad (16)$$

392

393 where AF_{raw} is the raw activated fraction at that percent supersaturation, and $AF_{plateau}$ is the activated
394 fraction corresponding to the lower plateau (Rose, 2008). The adjusted activated fraction curves are
395 shown in Fig. 5 (blue curves). These are in good agreement with the theoretical κ-Köhler-derived
396 activation curves for sodium chloride (not shown).

397

398 Critical supersaturation was determined for each diameter by calculating the percent supersaturation at
399 which the raw $AF_{D,weighted}^{SS} = 0.5$. These critical supersaturations are shown in Fig. 6a, and the
400 theoretical critical supersaturations calculated from κ-Kohler theory are shown for comparison. Eq. 4
401 was used to calculate apparent hygroscopicity for each particle diameter, shown in Fig. 6b. A dashed
402 line in Fig. 6b indicates the literature value for κ_{app}^{NaCl}. It is apparent that failing to account for multiply-
403 charged particle in the activated fraction curves shown in Fig. 5 leads to an overestimation of κ_{app} .
404 Artifacts in κ_{app} are shown in Fig. 6c.

405

406 For the theoretical aerosol distribution used in this analysis (Fig. 3a), small, positive deviations from κ-
407 Köhler theory and the literature value for κ_{app}^{NaCl} were observed ($0.01 \leq \kappa_{app,artifact}^{NaCl} \leq 0.04, 1 -$
408 3 % of κ_{app}^{NaCl}). As shown in the figure, κ_{app} artifacts resulting from unaccounted-for multiple charges

409 decrease with particle diameter for this theoretical aerosol population. Greater κ_{app} artifacts would be
410 expected for aerosol populations with more prevalent accumulation modes.
411
412 The aerosol/sheath ratio within the DMA also modulates the effect of multiple charges on κ_{app}. As the
413 aerosol/sheath ratio increases, the transfer function broadens, allowing particles that are both larger and
414 smaller than the selected diameter to exit the DMA. This in turn broadens the CCN activated fraction
415 curve (Rose et al., 2008). The larger particles will activate as CCN at lower supersaturations than
416 particles with the selected diameter, resulting in an increase in the activated fraction plateau due to
417 multiple-charged particles and a further decrease in the determined SS_{crit}. Petters et al. 2007b showed
418 that CCN activated fraction curves are significantly skewed by multiply-charged particles when the
419 mode diameter of the aerosol population upstream of the DMA exceeds the critical diameter of the size-
420 selected particles. In an example CCN activated fraction curve, Rose et al. 2008 demonstrated that a 1:6
421 ratio of doubly-to-singly charged particles resulted in an underestimation of the critical activation
422 diameter by 2%. Zhao-Ze and Liang, 2014 also showed that multiply-charged particles can introduce
423 significant uncertainty into hygroscopicity calculations.
424
425 **3.1.4 Additional artifacts resulting from DMA measurements**
426
427 Several additional factors that may impact experimental κ_{app} are beyond the scope of this study, but are
428 worth mentioning as they represent additional potential sources of error in some cases. First, volatile
429 aerosols may partially evaporate inside the DMA, resulting in a decrease in particle size exiting the
430 DMA. DMA sizing error due to aerosol volatility (defined as the ratio of sampled diameter to the
431 selected diameter) increases with volatility, though sizing error can be decreased by increasing the
432 sheath flow rate in the DMA. Conversely, hygroscopic aerosols may grow inside the DMA, resulting in

larger particles existing the DMA. Operationally, errors in DMA sizing due to hygroscopic growth can be mitigated if aerosols entering the DMA inlet are in wet metastable states (higher aerosol RH at DMA inlet), and if DMA sheath flow rates are kept low (Khlystov, 2014).

Voltage shifts within the DMA (differences between the selected voltage and the actual voltage inside the DMA) can lead to discrepancies between selected and sampled particle diameters. Voltage shifts may result from a space-charge field generated by the motion of charges within the DMA. Particles charged by the bi-polar neutralizer will either be attracted towards or repelled away from the inner column of the DMA, depending on whether they are positively or negatively charged. This charge separation creates a space-charge field which shifts the actual voltage within the DMA from the selected voltage. The impact of the space-charge field on the midpoint and spread of the DMA transfer function increases as particle mobility increases (as particle size decreases), and as particle concentration increases (Alonso and Kousaka, 1996; Alonso et al., 2000; Alonso et al., 2001).

4. Artifacts derived from condensation particle counters

4.1 CPC operation at low concentration

The second instrument which must function accurately during CCN experiments is the condensation particle counter. CPC performance is characterized by the maximum counting efficiency (which may be influenced by the working fluid in the instrument) and the 50 %-cut-off diameter (d_{50}), the particle diameter at which 50 % counting efficiency is observed, both of which can vary between commercially available models and even between individual CPCs (Heim et al., 2004). One study found that n-butanol CPCs (TSI, Inc. Models 3772, 3775, and 3776) exhibited smaller d_{50} for silver particles than sodium chloride ($3.3\ nm \leq d_{50}^{Ag} \leq 7.8\ nm$ and $4.1\ nm \leq d_{50}^{NaCl} \leq 14.7\ nm$), due to the more effective condensation of n-butanol on silver particles (Hermann et al., 2007).

Maximum counting efficiencies in that study varied from 88.9 % to 100.3 %. Another comparison of n-butanol CPCs (TSI Inc. Models 3010 and 3022, Grimm Tech. Inc. Model 5.403) found $3.1\ nm \leq d_{50} \leq 11.9\ nm$ for sodium chloride aerosols (Heim et al., 2004). In another study, the counting efficiencies observed in measurements of tungsten oxide particles by different instruments of the same model (TSI 3025) varied from 88.9 % to 138.9 %, while $d_{50}^{WO_x}$ varied from 3.2 nm to 11.0 nm (Hameri et al., 2002).

While some issues can cause undercounting at all concentrations, the additional issue of uncounted particles due to the arrival of more than one particle in the detector's field of view at any time arises only at higher concentrations. The cut-off between "low" and "high" concentration is not exact and varies between instruments. CPC undercounting issues which arise even at relatively low concentrations (which one would expect to encounter under standard experimental conditions) will be discussed in this

section. Concentration-dependent effects encountered at higher concentrations will be explored in Sect. 4.2.

Six counting efficiency curves were generated using sigmoidal distributions and the 50 % cut-off diameters and maximum counting efficiencies listed in Table 3. Chosen values represent d_{50} values and maximum counting efficiencies reported in the literature under relatively low concentrations of 1000-4000 cm^{-3} (Hermann et al., 2007). The resulting sigmoidal distributions (Fig. 7a) were used to determine the counting efficiency of 25, 50, 100, and 200 nm particles.

Next, κ_{app} was calculated from theoretical critical percent supersaturations for each chosen diameter. To do so, four sigmoid curves representing sodium chloride CCN activation (hereon referred to as "activation curves") for 25, 50, 100, and 200 nm were generated. The κ-Köhler-SS_{crit} of sodium chloride was used as the midpoint of each activation curve, and one-tenth of this value was used as the standard deviation (100 % CE, Fig. 7b-e). These values are consistent with the standard deviation/midpoint ratio observed from our instrument's ammonium sulfate CCN calibration data.

Activation curves were then generated for CPC Cases 1-6 by dividing the activated fraction for each dry particle diameter by the counting efficiency for that diameter. SS_{crit} was determined for each CPC case by finding the percent supersaturation at which activated fraction = 0.50. Results are summarized in Fig. 7f. Next, critical supersaturation was converted to saturation, and $\kappa_{app,theory}$ was calculated for each diameter in each CPC Case using Eq. (4) (see Fig. 7g). As above, κ_{app} artifacts were calculated by finding the difference between these results and the literature value of κ_{app} for sodium chloride (see Fig. 7h).

For the diameters studied, the effect of maximum counting efficiency on CPC concentration (and activated fraction) is greater than the effect of 50 %-cutoff diameter. However, neither characteristic resulted in large κ_{app} artifacts. The largest κ_{app} artifact observed at "low" concentrations was 0.035 for CPC Case 4, 2.4 % of the literature value for the apparent hygroscopicity factor for sodium chloride.

4.2 CPC operation at high concentration

Operation at high concentrations introduces an additional source of undercounting through particle coincidence at the CPC optical counter. For the TSI 3010 CPC, undercounting is observed is for particle concentrations above $1 \times 10^4\ cm^{-3}$. At $5 \times 10^4\ cm^{-3}$, the detector saturates and cannot detect higher concentrations. By comparison, the TSI 3025 is effective at counting higher particle concentrations, of up to $2.5 \times 10^4\ cm^{-3}$ (Hameri et al., 2002;Sem, 2002).

To model undercounting due to particle coincidence, four CPC counting curves (Fig. 8a) were generated using the equations in Table 4. Case 7 represents a CPC where counting efficiency decreases with particle concentration, without reaching saturation. Cases 8-10 represent CPCs where saturation is reached at $4 \times 10^4\ cm^{-3}$, $2 \times 10^4\ cm^{-3}$, and $1 \times 10^4\ cm^{-3}$, respectively. These saturation concentrations are of similar magnitude to those observed from TSI 3010 concentration data. It should be noted that the CPC concentration in Cases 7-10 levels off at the saturation concentration for each case.

In order to assess the importance of undercounting in CPC Cases 7-10, four theoretical aerosol distributions with a peak concentration at 50 nm were employed (Table 5, Fig. 8b). CPC Distribution 1 represents a worst-case scenario of similar magnitude to the highest particle concentrations measured during a coastal nucleation event (Hameri et al., 2002; Sem, 2002), while CPC Distributions 2, 3, and 4 are lower in concentration (due to the lack of undercounting in CPC Distributions 2, 3, and 4 as demonstrated in Figure 6b, the remaining analysis for CPC operation at high concentration considers only CPC Distribution 1.) CPC Cases 8-10 were applied to CPC Distribution 1 in order to determine the

concentration measured by the CPC for 25, 50, 100 and 200 nm aerosols. The counting efficiency was then calculated for each case and aerosol diameter in CPC Distribution 1.

Sigmoidal activated fraction curves were generated for 25, 50, 100 and 200 nm sodium chloride aerosols. As in the low concentration cases, the midpoint of each 100 % CE curve was chosen to be equal to the κ-Köhler-derived SS_{crit} of sodium chloride at each dry diameter, and the standard deviation of each curve is equal to one-tenth of the SS_{crit}. These activated fraction curves were adjusted using the counting efficiencies calculated in the previous step. In cases where the activated fraction has increased due to undercounting by the CPC, the theoretical sigmoidal curve shifts to the left relative to the 100 % CE case (Fig. 8c-f). Thus, undercounting by the CPC effectively increases the reported activated fraction. As before, SS_{crit} was determined from each of these curves, and $\kappa_{app,theory}$ was subsequently calculated using Eq. (4) (Fig. 8g-h).

$\kappa_{app,theory}$ fell over a much wider range for 25, 50, and 100 nm particles (1.30-1.56, 1.32-1.70, and 1.30-1.55, respectively) than for 200 nm particles (1.28-1.29) due to the lower concentration of 200 nm particles in the chosen aerosol distribution, which resulted in a higher counting efficiency for these aerosols. In comparison, the largest range in $\kappa_{app,theory}$ was observed for 50 nm aerosols, the peak diameter in this aerosol distribution.

A wider range in $\kappa_{app,theory}$ was observed for the high-concentration CPC Cases (7-10) compared to the low-concentration CPC Cases (1-6). The lowest counting efficiency observed across the low-concentration cases was 89.9 % for 25 nm aerosol in Case 4, while the lowest counting efficiency observed in the high-concentration cases was 18.0 % for 50 nm aerosol in Case 10.

Artifacts in the apparent hygroscopicity parameter are shown in Fig. 8i. κ_{app} artifacts were the greatest for a CPC that becomes saturated at 20,000 particles/cm^3 ($0.0131 \leq \kappa_{app} \leq 0.4206$). The lower the concentration at which a CPC becomes saturated, the more quickly its counting efficiency will drop as concentration increases, resulting in increased activated fraction and increased apparent hygroscopicity. The magnitude of artifacts due to CPC undercounting depends on the saturation concentration of the CPC and the distribution of the aerosol population being studied.

5. Artifacts derived from cloud condensation nuclei instruments

Finally, the third instrument whose performance accuracy contributes to the overall certainty in CCN assessment in the CCN instrument itself. Several instruments have been implemented for measuring CCN concentrations over the last few decades. Older models include the Continuous Flow Parallel Plate Diffusion Chamber (Sinnarwalla, 1973) and the Hudson CCN spectrometer (Hudson, 1989) which both employ an applied temperature gradient perpendicular to the aerosol flow. Newer models, such as the widely-used Droplet Measurement Technology Cloud Condensation Nuclei Counter (DMT CCN-100), operate with a streamwise temperature gradient and continuous, laminar flow (Lance et al., 2006). The total flow through the DMT CCN-100 is 0.20-0.90 L min^{-1}, though the instrument is typically operated with a total flow of 0.50 L min^{-1}. The aerosol/sheath ratio in the DMT CCN-100 is set by the user, and a ratio of 1:10 is commonly chosen. The following analysis considers the DMT CCN-100. According to the CCN-100 manual, the counting efficiency for this CCN instrument depends on concentration and supersaturation (Fig. 9a). The counting efficiency decreases rapidly with concentration at < 0.2 % SS due to rapid water vapor depletion at these low supersaturations, and falls off more slowly for > 0.2 % SS (DMT CCN-100 manual).

The counting efficiency of the DMT CCN-100 was tested for four lognormal aerosol distributions with peak concentrations at 50 nm and varying total concentrations (Table 5, Fig. 9b). Note that CCN Cases 1-4 are identical to the aerosol distributions CPC Distributions 1-4 used for the high-concentration CPC cases.

The counting efficiencies for each case were applied to theoretical sodium chloride sigmoidal activated fraction curves to produce normalized activated fraction curves (Fig. 9c-f). As above, the midpoint is

575 set to the SS_{crit} of sodium chloride at each dry diameter, and the standard deviation is assumed to be
576 one-tenth of SS_{crit}. CCN undercounting effectively decreases activated fraction, therefore shifting the
577 activated fraction curve downwards and towards higher percent supersaturations. The opposite effect is
578 observed when CPC undercounting occurs. Critical supersaturation was determined for each CCN case,
579 as above (Fig. 9g). Values of SS_{crit} were then converted to saturation, and $\kappa_{app,theory}$ was calculated
580 using Eq. (4) (Fig. 9h).
581
582 Significant deviations from κ-Köhler theory were only observed in CCN Case 1, with total aerosol
583 concentration 5×10^6 particles/cm^3 (Fig. 9g-i). The largest deviation for CCN Case 1 was observed in
584 100 nm particles ($\kappa_{app,artifact} = -0.57$), due to the higher concentration of 100 nm particles compared
585 to 25 and 200 nm particles, and the lower percent supersaturation necessary for activation. The largest
586 artifacts across CCN Cases 2 and 3 were also observed for 100 nm particles, though no artifacts were
587 observed for any particle diameter in CCN Case 4 due to the much lower concentrations.
588
589 Sodium chloride is very hygroscopic. It should be noted that aerosols consisting of less hygroscopic
590 compounds will activate at higher percent supersaturations (> 0.2 % SS regime) which will lead to
591 smaller κ_{app} artifacts when the same aerosol distribution and total aerosol concentration is considered.
592 If a mixture was considered (for example, sodium chloride with a non-hygroscopic species such as soot)
593 the results may also be different. The shape of the aerosol distribution must also be taken into account.
594 A distribution with a narrower peak than the one generated for this analysis would be at risk for larger
595 κ_{app} artifacts for any total aerosol concentration, and these artifacts would be greater at the peak
596 diameter, while a broader distribution would result in less variation in κ_{app} artifacts for each particle
597 diameter.

598 6. **Counting statistics in CCN and CPC measurements**

599

600 Though it is beyond the scope of this analysis, it should be mentioned that sampling at very low particle
601 concentrations (< 200 cm^{-3} total particle concentration) can introduce additional error into CCN and
602 CPC measurements. This error can be mitigated by increasing scan times (Moore et al., 2010). For
603 example, Moore et al., 2010 averaged CCN and particle concentrations over 5-second intervals for
604 monodisperse particle concentrations < 10 cm^{-3}, and increased averaging time to 20-second intervals
605 when the monodisperse particle concentration reached < 6 cm^{-3}.

606

607 **7. Discussion**

608

609 A comparison of the major instrument sources of error in CCN-derived κ_{app} is shown in Fig. 10. In
610 addition, the best and worst case combination of errors, determined by additive error propagation, are
611 also shown. DMA Case 4, CPC Case 4, CPC Case 10, and CCN Case 1 represent the operating
612 conditions that resulted in the largest κ_{app} artifacts in this study. In DMA Case 4, the aerosol/sheath
613 ratio of 0.30 resulted in a broadened aerosol distribution downstream of the DMA. Compared to DMA
614 Case 1, where $Q_a/Q_{sh} = 0.10$, the downstream diameter range in DMA Case 4 was 300 % higher for 25
615 nm particles, resulting in a spread of 20-36 nm. Similarly, the diameter ranges for 50, 100, and 200 nm
616 diameter were 220 %, 230 %, and 220 % wider than in Case 1, respectively. Compared to the most ideal
617 DMA case presented in this study (DMA Case 2), where $Q_a/Q_{sh} = 0.05$, the downstream diameter
618 range in DMA Case 4 was 700 % higher for 25 nm particles; the diameter ranges for 50, 100, and 200
619 nm diameter were 540 %, 560 %, and 520 % wider than in Case 2, respectively. The results
620 demonstrate that limiting Q_a/Q_{sh} to ≤ 0.10 will result in a narrow particle size distribution downstream
621 of the DMA. Other studies have recommended employing DMA sample/sheath ratios of 0.2 (Petters et

622 al., 2007; Carrico et al., 2008; Moore et al., 2010) or 0.1 (Moore et al., 2010; Zhao-Ze and Liang, 2014)
623 in order to minimize measurement aerosols due to transfer function broadening.
624
625 The effects of multiply-charged particles on κ_{app} calculations were also quantified, as shown in Fig. 10.
626 Small, positive κ_{app} artifacts ($1-3$ % of κ_{app}^{NaCl}) were observed when particles with +2 and +3 charges
627 were not accounted for. This analysis considered a theoretical aerosol distribution in which most of the
628 particles measure less than 100 nm in diameter. Actual aerosol distributions vary temporally and
629 spatially, and often include accumulation and coarse modes that would result in larger κ_{app} artifacts.
630
631 CPC Case 4 represents κ_{app} artifacts (0.031-0.035) due to undercounting that arises from poor
632 maximum CPC counting efficiency (90 %), which may be observed when using butanol as a working
633 fluid while measuring the concentration of inorganic aerosols. In contrast, κ_{app} artifacts are negligible
634 (< 0.10 % of κ_{app}^{NaCl}) in CPC Case 3, where maximum counting efficiency = 100 %. CPC Cases 8 and 10
635 (applied to the highest-concentration case, CPC Distribution 1) represent undercounting at high
636 concentration with CPCs where saturation is observed at $4 \times 10^4\ cm^{-3}$ and $1 \times 10^4\ cm^{-3}$, respectively.
637 Counting efficiency drops off more rapidly with concentration in the latter case, resulting in κ_{app}
638 artifacts that are highest at the peak of the aerosol distribution (0.1190 and 0.4206 for 50 nm aerosols in
639 CPC Cases 8 and 10, respectively). It should be noted that undercounting was only observed for one of
640 the four upstream distributions studied, CPC Distribution 1. No undercounting was observed when CPC
641 Cases 7-10 were applied to CPC Distributions 2-4.
642
643 CCN Case 1 represents CCN undercounting at high concentration (total aerosol concentration = $5 \times$
644 $10^6 cm^{-3}$). CCN undercounting is greatest for low supersaturation (< 0.2 %) and high concentration,

resulting in the lowest counting efficiency and highest κ_{app} artifacts (- 0.57) for 100 nm aerosols in CCN Case 1. The largest CCN-derived κ_{app} artifact observed outside of CCN Case 1 was – 0.01 for 100 nm aerosols in CCN Case 2.

The combined artifacts for the cases where the highest artifacts were observed (DMA Case 4, multiple particle charging, CPC Case 4, CPC Case 10, CCN Case 1) are 0.24, 0.21, 0.23, and 0.15 for 25, 50, 100, and 200 nm particles respectively, as shown in Fig. 10. The combined artifacts for the lowest-artifact cases (DMA Case 2, CPC Case 3, and CCN Case 4) are < 0.008 except for all four particle diameters.

Conclusions

The sensitivity of weather and climate models to accuracy in CCN activation predictions has been demonstrated in other works. Possible sources of apparent hygroscopicity artifacts calculated from CCN measurements have been presented in this study. This analysis has focused on sodium chloride and ammonium sulfate aerosols, but it can be extended to other aerosol populations, including mixtures and field samples.

The greatest combined artifacts ($0.15 < \kappa_{app,artifact} < 0.24$, NaCl) occurred as a result of the combined issues of the highest DMA aerosol/sheath ratio, uncorrected multiple particle charging, and undercounting by both CPC and CCN instrument. The lowest combined artifacts ($0.0021 < \kappa_{app,artifact} < 0.0074$, NaCl) occurred as a result of ideal operating conditions: lowest DMA/sheath ratio, corrected multiple particle charging, and little to no undercounting.

The largest single-instrument artifacts ($-0.57 < \kappa_{app,artifact} < 0.42$ for sodium chloride) in this study arise from undercounting by either the CPC or CCN counter at high concentration. This problem occurs during attempts to measure aerosol concentrations of ~10^4 cm^{-3} which is much higher than the recommended concentration ranges for either instrument, (CPC Cases 7-10 and CCN Case 4). Corrective action should be taken to dilute aerosol samples in order to avoid undercounting. It should be noted that these artifacts are for individual instruments and do not take combined operation of the CPC and CCN into account; when both instruments undercount, artifacts in $\kappa_{app,artifact}$ are reduced.

Smaller single-instrument artifacts ($\kappa_{app,artifact} < 0.04$) were observed for the CPC cases where 50 %-cut-off diameter and maximum counting efficiency were varied. Given the chosen particle diameters

(25, 50, 100, 200 nm), κ_{app} artifacts due to d_{50} were minimal. The largest κ_{app} artifacts for a CPC counting at low concentration (0.031-0.035) were observed where the maximum counting efficiency was equal 0.90. This may represent a compositional mismatch between n-butanol as the working fluid and sodium chloride as the aerosol, due to the poor solubility of the latter in the former. Individual n-butanol CPCs may exhibit higher maximum counting efficiencies for sodium chloride.

Uncertainty arising from the DMA depended greatly on the chosen aerosol and sheath settings. One set of DMA cases (Cases 2-4) examined the effect of aerosol/sheath ratio. By decreasing this ratio, a narrower near-monodisperse flow can be produced, which increases the accuracy of calculated κ_{app}. The κ_{app} artifacts for an aerosol/sheath ratio of 0.10 were 0.65 % of $\kappa_{literature}$ for 25 nm sodium chloride aerosols, 0.31 % for 50 nm, -0.17 % for 100 nm, and -1.2 % for 200 nm.

The second set of DMA cases (5-7) were designed to evaluate the effects of holding the sheath flow constant while varying the excess air flow by -2 %, +2 %, and +5 %. These resulted in shifts of ≤ 2 nm for 25 nm and 50 nm particles, ≤ 4 nm for 100 nm particles, and ≤ 7 nm for 200 nm particles. The downstream aerosol distribution was shifted towards larger particle diameters when sheath flow exceeded excess flow, and towards smaller particle diameter when sheath flow was less than excess flow. When taking field measurements, the composition of the sample may vary with particle diameter, thereby introducing another source of error from a broader DMA distribution.

By extension, the issue of uncertain sizing by the DMA leads to added uncertainties in the CCN instrument calibrations which are strongly dependent on the chosen aerosol to sheath ration within the DMA. We recommend conducting all CCN calibrations with DMA aerosol to sheath ratio of 1:10 or 1:20 which will reduce kappa uncertainties to less than 1% for all dry sizes (25 to 200 nm).

Overall, under optimal operating conditions, where the DMA aerosol/sheath ratio is 0.10 and excess/sheath ratio is 1.0, and in the absence of undercounting by the CPC or CCN, uncertainties in κ_{app} are less than ±1.2 % for 25 to 200 nm particles. During sampling, when the DMA sample/sheath ratio is reduced to 0.05, κ_{app} uncertainties decrease to ±0.58 %. Additionally, errors in activated fraction (and therefore κ_{app}) resulting from the bipolar charge distribution can be corrected by determining the fraction of particles with multiple charges.

In this study, apparent hygroscopicity parameter artifacts were calculated for two pure, inorganic species in this study. This analysis could be used to estimate κ_{app} artifacts for ambient aerosol populations, which may result in a better understanding of the "real' differences between these populations. As discussed in the introduction, Collins et al. 2016 aggregated κ_{app} from several mesocosm and field studies for 30-100 nm sea spray aerosol ($0.4 < \kappa_{app}^{SSA} < 1.3$). The wide range of κ_{app} in these studies may be attributed to differences in composition, experimental artifacts, or a combination of the two. Quantification of experimental artifacts would facilitate interpretation of κ_{app} in aerosol populations and constrain the importance of composition in CCN activation. There has been a recent proliferation of CCN data availability from multiple researchers and multiple experimental setups. To maximize the utility of these studies and to compare cloud-activating properties of various ambient aerosol masses, it is essential that artifacts are considered in both CCN data collection and in reporting of the data.

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Author Contribution

Sarah D. Brooks provided the conceptual framework and contributed to the writing of the manuscript. Jessica A. Mirrielees performed the analysis and lead the writing of the manuscript.

Competing Interests

The authors declare that they have no conflict of interest.

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Notation	
$\alpha_{CC}, \beta_{CC}, \gamma_{CC}$	Empirically-determined constants used to calculate Cunningham slip correction factor
Z_p	Aerosol particle electrical mobility
C_C	Cunningham slip correction factor
d_m	Electrical mobility diameter
n	Number of charges on particle
e	Elementary unit of charge
η	Gas dynamic viscosity
λ	Mean free path
Q_{sh}	Sheath flow
Q_e	Excess air flow
Q_a	Aerosol flow
Q_s	Sample flow
κ_{app}	Apparent hygroscopicity parameter
$\kappa_{app,artifact}$	Apparent hygroscopicity parameter artifact
s	Equilibrium water vapor saturation
s_{crit}	Critical saturation (50 % of aerosols active as cloud condensation nuclei)
A	Constant used in calculating κ_{app}
σ_{lv}	Surface tension of water
T	Temperature
D_{act}	Activation diameter
SS_{crit}	Critical percent supersaturation
α_{TF}	Height of DMA transfer function
β_{TF}	Half-width of DMA transfer function
Z_p'	Mobility of particle at DMA inlet
$Z_{p,mid}$	Midpoint of transfer function
ΔZ_p	Half-width of transfer function
V_0	Voltage selected at DMA
r_1	DMA inner radius
r_2	DMA outer radius
L	DMA length
d_{50}	50 %-cut-off diameter

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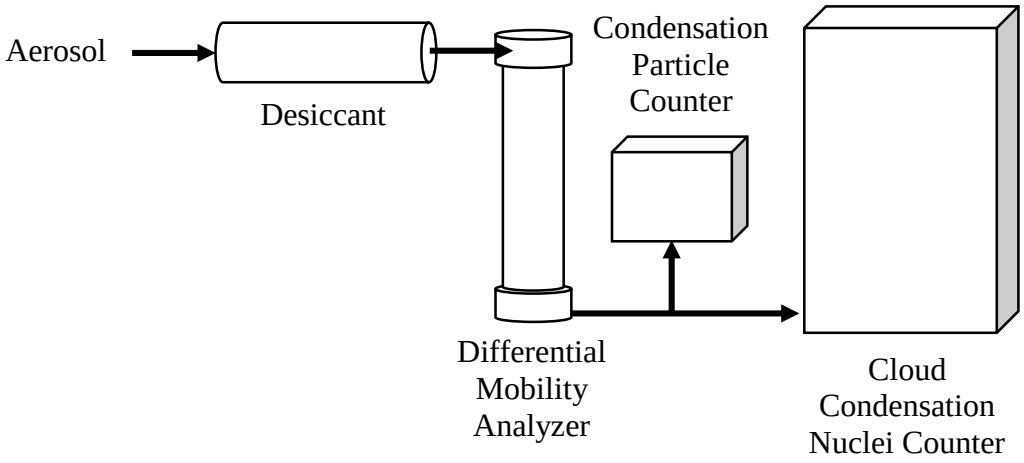

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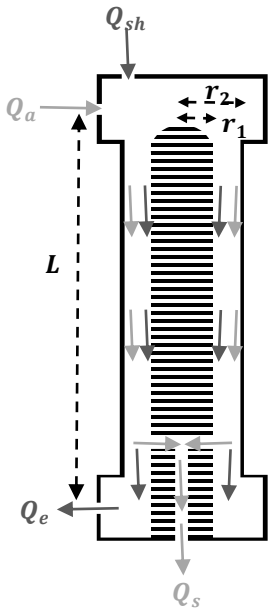
1005 **Figure 1** Experimental setup used for obtaining sized CCN and particle concentration measurements

1006 from an aerosol sample.



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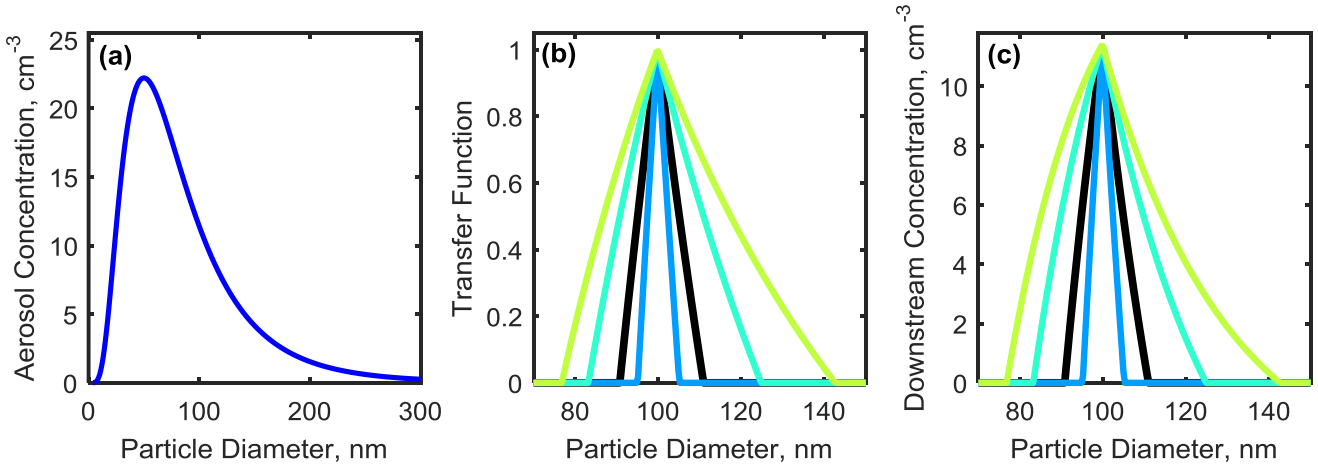
1008 **Figure 2** Simplified flow diagram of a DMA with an inner electrode radius r_1, outer electrode radius r_2,
1009 distance between aerosol inlet and sample outlet L, clean sheath air flow Q_{sh}, aerosol flow Q_a, excess air
1010 flow Q_e, and sample air flow Q_s.



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1013 **Figure 3** (a) A theoretical aerosol distribution generated using a lognormal function centered at 50 nm. .

1014 (b) The transfer function calculated using Eq. (7). (c) Multiplying the distribution by the transfer function

1015 gives the downstream aerosol concentration (cm^{-3}).

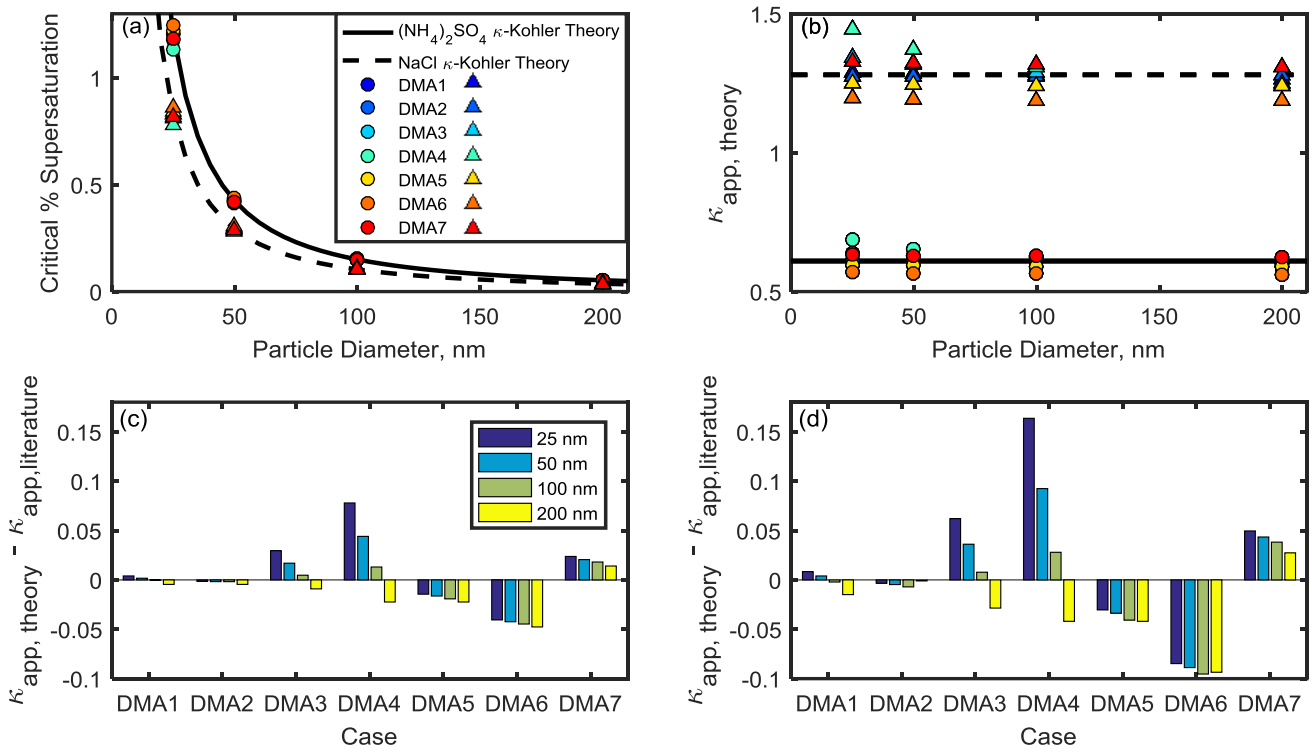


Figure 4 (a) Critical supersaturation of ammonium sulfate and sodium chloride particles calculated for DMA Cases 1-7 for sodium chloride (triangles) and ammonium sulfate (circles). Ammonium sulfate and sodium chloride curves from κ-Köhler theory are shown for comparison. (b) Apparent hygroscopicity κ_{app} for DMA cases 1-7. (c) DMA-flow-derived artifacts in ammonium sulfate κ_{app} are shown for each DMA case. (d) DMA-flow-derived artifacts in sodium chloride κ_{app} are shown for each DMA case.

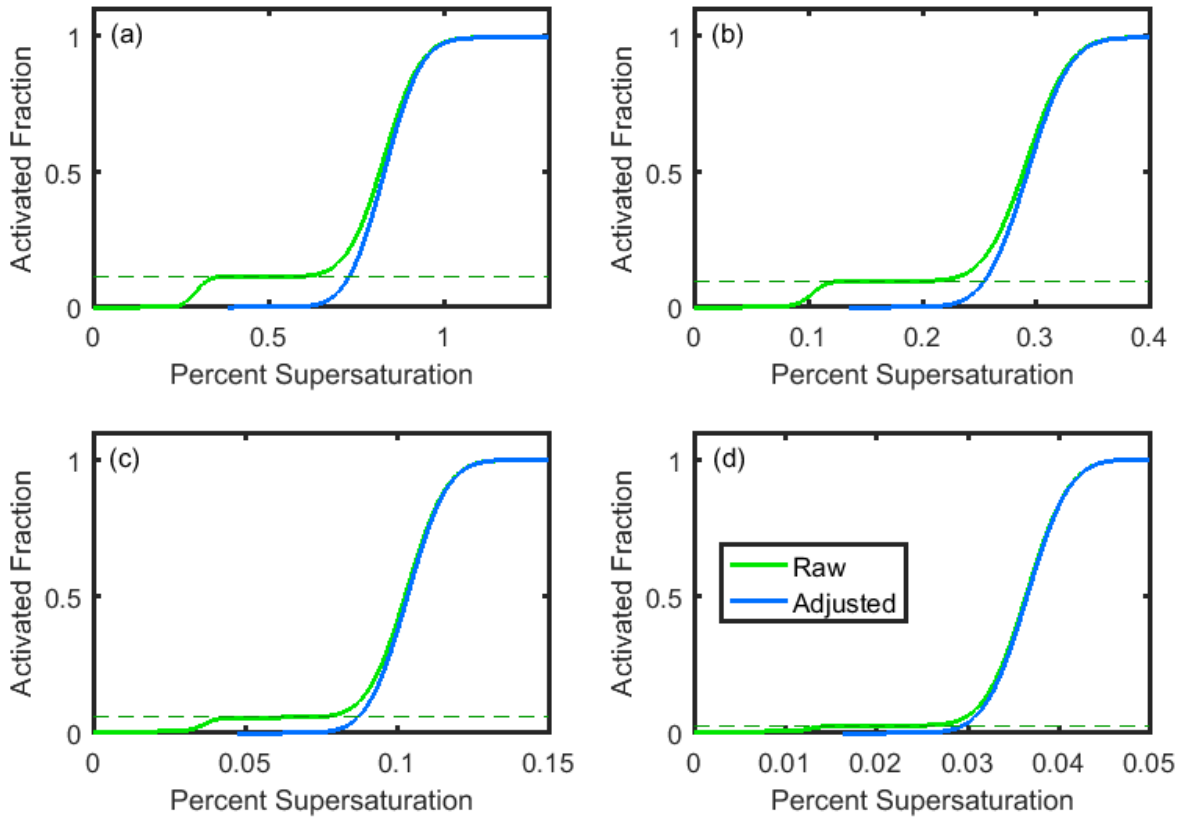
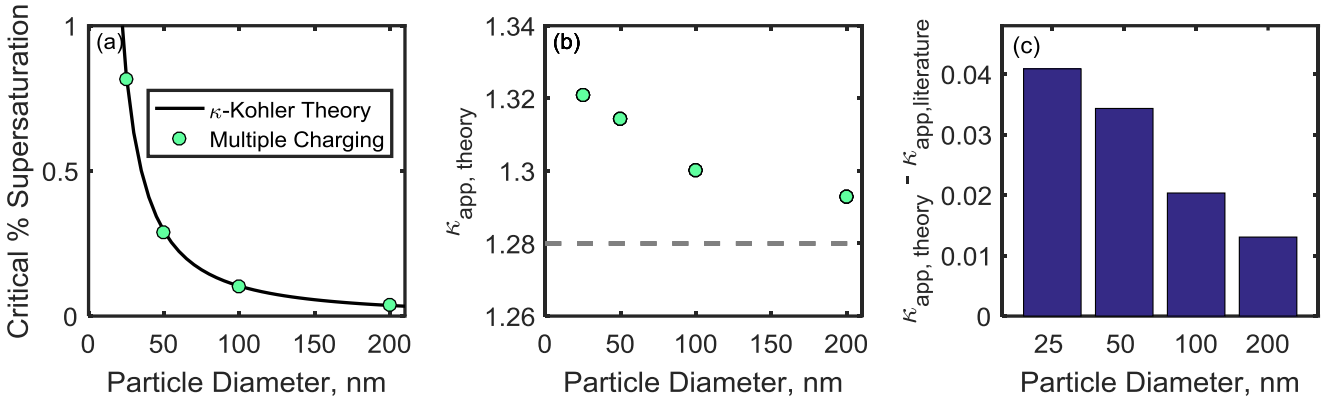


Figure 5 Theoretical raw (green) and adjusted (blue) activated fraction curves for (a) 25 nm (+1), 50 nm (+2), and 75 nm (+3) particles; (b) 50 nm (+1), 100 nm (+2), and 150 nm (+3) particles; (c) 100 nm (+1), 200 nm (+2), and 300 nm (+3) particles; (d) 200 nm (+1), 400 nm (+2), and 600 nm (+3) particles. All particles are pure sodium chloride.



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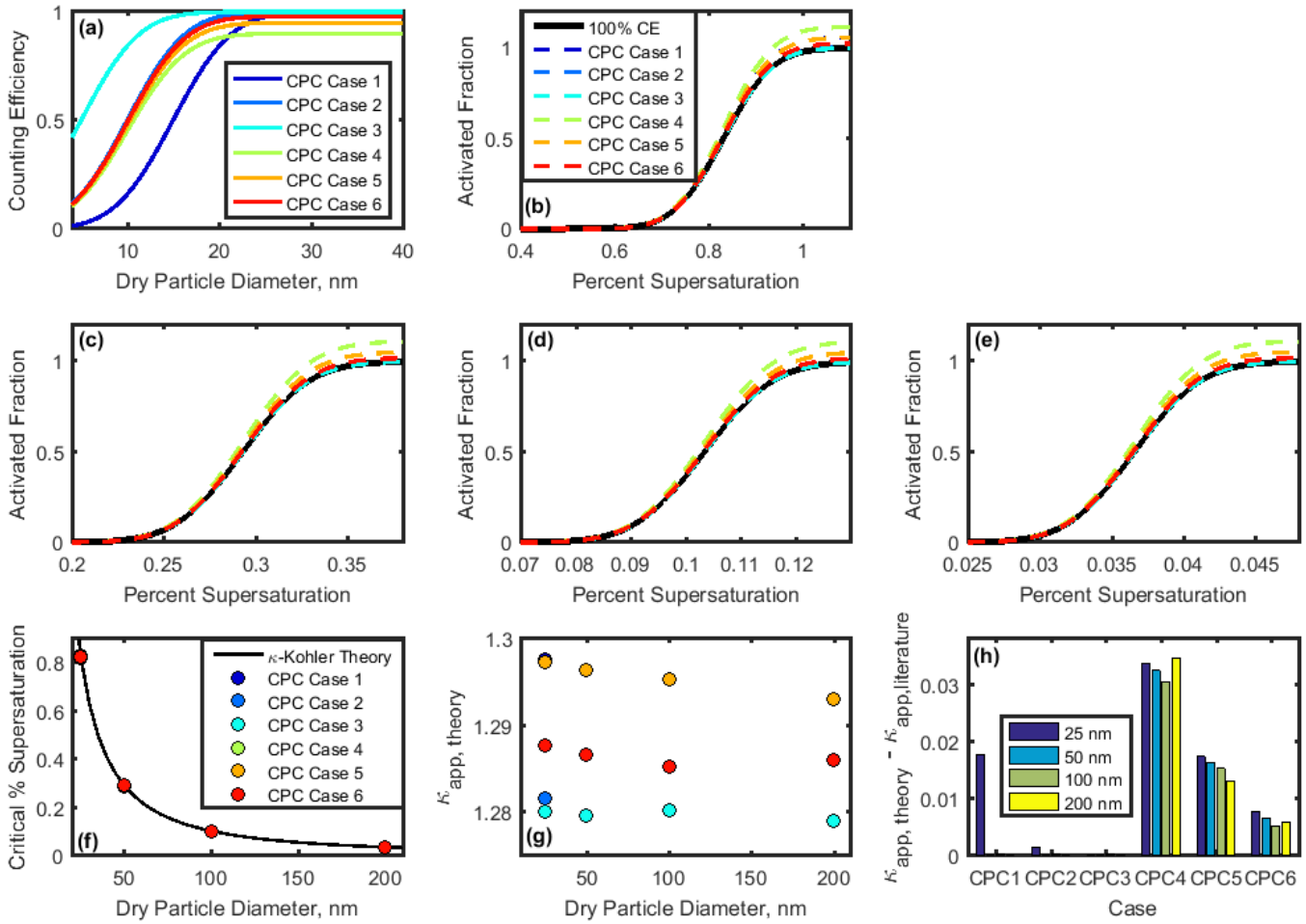
1029 **Figure 6** (a) Critical percent supersaturation of sodium chloride particles determined from activated

1030 fraction curves shown in Fig. 5. A κ-Köhler curve for sodium chloride is shown for comparison. (b)

1031 Theoretical κ_{app} for each particle diameter (gray dashed line indicates literature value for κ_{app}^{NaCl}). (c)

1032 Artifacts in κ_{app} resulting from multiple particle charges.



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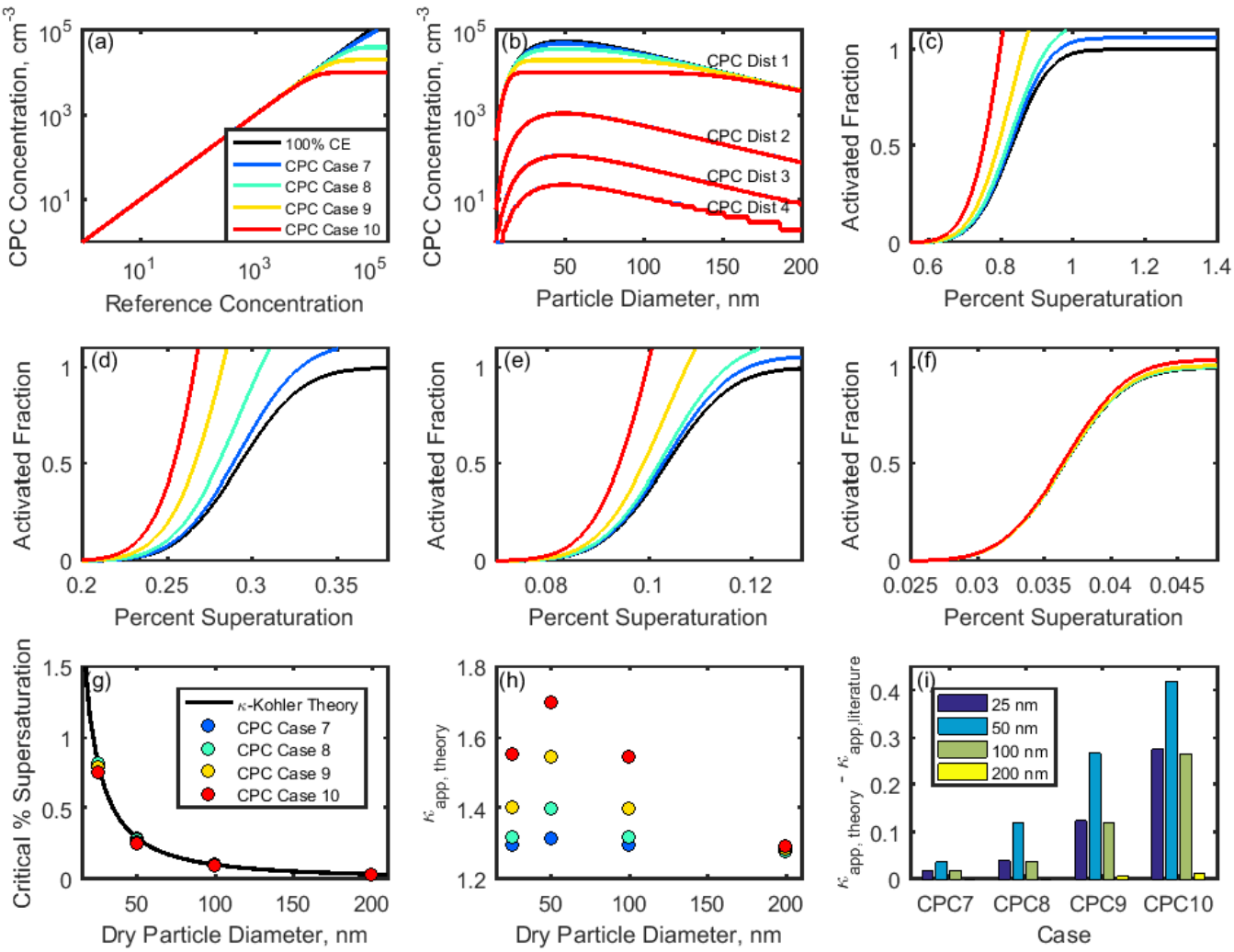
1034 **Figure 7** (a) Counting efficiency curves for CPC Cases 1-6 (shown in Table 3).

1035 (b-e) CCN activated fraction curves for 25, 50, 100, and 200 nm NaCl, respectively. (f) Critical

1036 supersaturation calculated for each particle diameter. (g) Theoretical κ_{app} for each CPC case and

1037 particle diameter. (h) Artifacts in κ_{app} for each CPC case and particle diameter.



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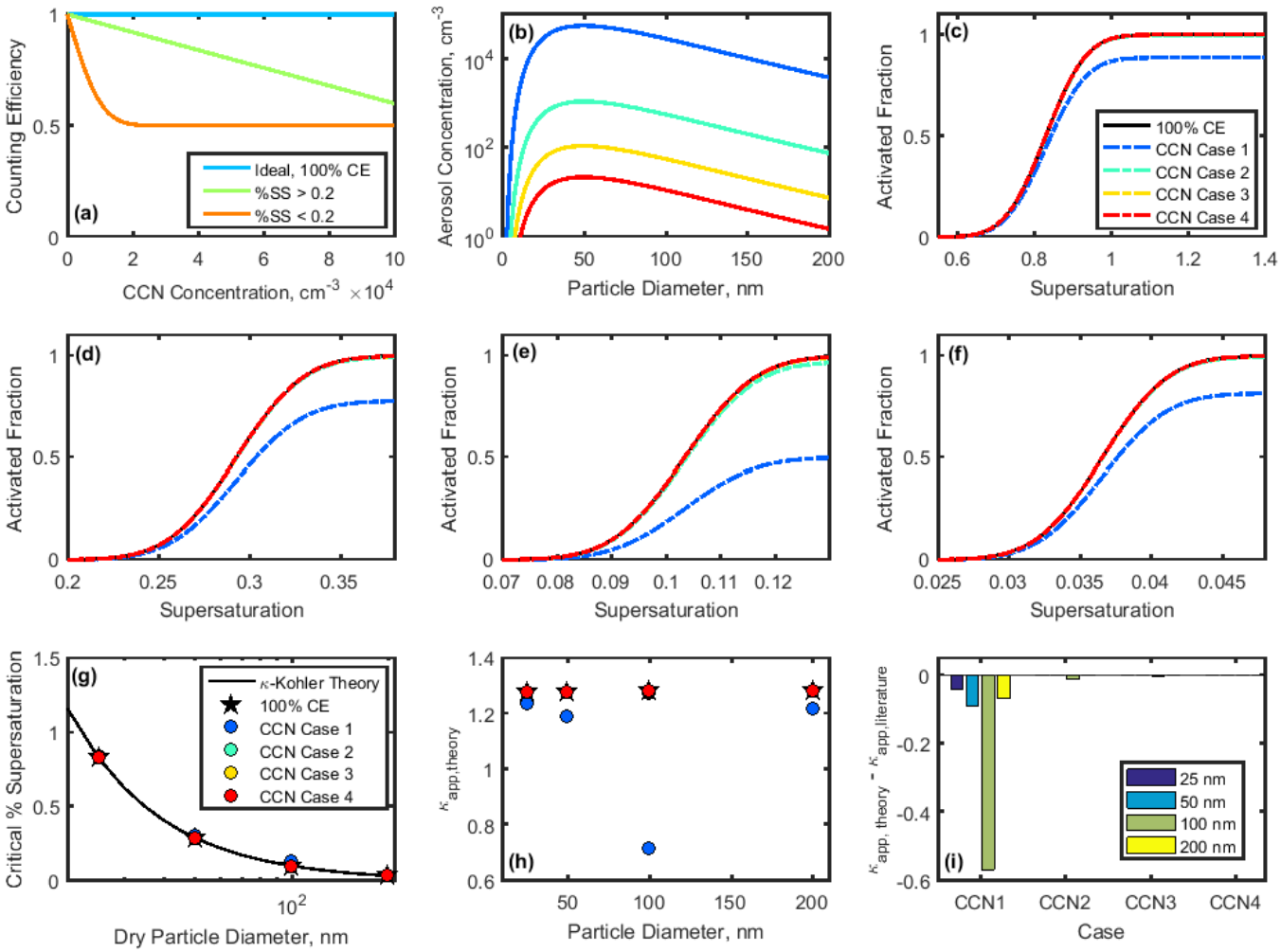
1039 **Figure 8** (a) Theoretical relationships between the reference aerosol concentration and CPC

1040 concentration. (b) Concentration-dependent counting efficiencies from (a) were applied to four

1041 theoretical aerosol distributions. (c-f) Activated fraction curves for CPC Distribution 1 and particle

1042 diameters 25, 50, 100, and 200 nm NaCl aerosol, respectively. (g,h) Critical supersaturation and κ_{app} for

1043 each case. (i) Artifacts in κ_{app} for each case.



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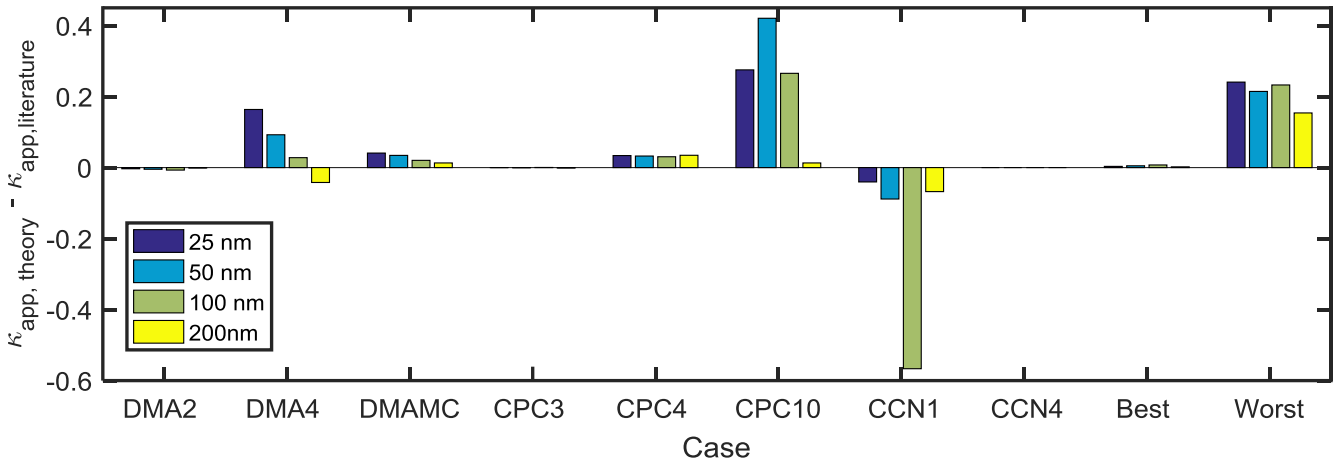
1045 **Figure 9** (a) Counting efficiencies of the DMT CCN-100 for specific supersaturations. (b) Lognormal
1046 aerosol distributions used to study CCN undercounting at high concentrations. (c-f) Activated fraction
1047 curves for 25, 50, 100, and 200 nm NaCl particles. Supersaturation-specific counting efficiencies from
1048 (a) applied to theoretical sigmoid curves for NaCl CCN activation. Activated fraction in the case of 100
1049 % counting efficiency is shown for comparison. (g) Critical supersaturation for each case. (h)
1050 Theoretical κ_{app} calculated for each case. (i) Artifacts in κ_{app} artifacts for each case.



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1052 **Figure 10** Comparison of κ_{app} artifacts derived from best and worst case scenarios for instrumental

1053 measurements for sodium chloride. Combined artifacts for the lowest-artifact cases (Best: DMA Case

1054 2, CPC Case 3, and CCN Case 4) and the highest-artifact cases (Worst: DMA Case 4, multiple charging,

1055 CPC Case 4, CPC Case 8, and CCN Case 1).

1056

Table 1 Theoretical DMA Flow Test Cases

Case	Q_{sh} (L min^{-1})	Q_e (L min^{-1})	Q_a (L min^{-1})	Q_s (L min^{-1})	Q_a/Q_{sh}	Q_e/Q_{sh}
DMA 1	3.00	3.00	0.30	0.30	0.10	1.00
DMA 2	3.00	3.00	0.15	0.15	0.05	1.00
DMA 3	3.00	3.00	0.60	0.60	0.20	1.00
DMA 4	3.00	3.00	0.90	0.90	0.30	1.00
DMA 5	3.00	3.06	0.36	0.30	0.12	1.02
DMA 6	3.00	3.15	0.45	0.30	0.15	1.05
DMA 7	3.00	2.94	0.24	0.30	0.08	0.98

1057

Table 2 Predicted downstream particle diameter range for each DMA case.				
Case	25 nm	50 nm	100 nm	200 nm
DMA 1	23-27	46-56	91-111	181-222
DMA 2	24-26	48-53	95-105	190-211
DMA 3	21-31	42-62	83-125	167-250
DMA 4	20-36	39-71	77-143	154-285
DMA 5	23-27	45-55	90-110	181-220
DMA 6	22-27	45-54	89-107	178-215
DMA 7	23-28	46-56	92-112	183-225

1058

Table 3 Values of 50%-cutoff diameter and maximum counting efficiency used in investigating κ_{app} artifacts for low particle concentrations measured by a CPC.

Case	d_{50}, nm	Maximum Counting Efficiency
CPC 1	15	100 %
CPC 2	10	100 %
CPC 3	5	100 %
CPC 4	10	90 %
CPC 5	10	95 %
CPC 6	10	98 %

1059

Table 4 Equations used to model the relationship between a reference or "true" aerosol concentration x (particles cm^{-3}), and the concentration measured by a condensation particle counter y (particles cm^{-3}).

Case	Equation
CPC 7	$y = x - 2 \times 10^{-6} x^2$
CPC 8	$y = 40000 \operatorname{erf}\left(\frac{x}{32000\sqrt{2}}\right)$
CPC 9	$y = 20000 \operatorname{erf}\left(\frac{x}{16000\sqrt{2}}\right)$
CPC 10	$y = 10000 \operatorname{erf}\left(\frac{x}{8000\sqrt{2}}\right)$

1060

Table 5 Total concentrations used in theoretical aerosol distribution for CPC operation at high concentration and CCN-derived κ_{app} artifacts.

CPC Distribution	CCN Case	Total Concentration (particles cm^{-3})
CPC Distribution 1	CCN 1	5×10^6
CPC Distribution 2	CCN 2	1×10^5
CPC Distribution 3	CCN 3	1×10^4
CPC Distribution 4	CCN 4	2×10^3