

We would like to thank the reviewer for the helpful comments and suggestions. We answer each of them here after, with the original comment in blue and our response in black. We added the modifications done in the revised version in italics.

Specific Comments:

[Referee #1]

One clear omission is the C_2H_6 / CH_4 signatures from the individual plume measurements. A great deal of important work is done in calibrating the C_2H_6 channel from the instrument, but then the field data from this channel is not presented. Reporting these ratios would serve to validate that the instrument is reporting reasonable values; conversely, if the ratios do not make sense (e.g., if the landfill C_2/C_1 ratio is not zero within uncertainties), then that may point to an additional unknown systematic error in the system. For example, these data could be easily added to Table 2.

Similarly, for the bag samples of the local natural gas source, the measurements of the ethane content would be an interesting addition to the manuscript.

Furthermore, the ethane content in the natural gas infrastructure is generally recorded by the natural gas distribution company; the values obtained from the analysis could be compared to the known values.

[Hoheisel et al.]

Thank you for pointing this out, we have added an additional section (see below) and another figure.

“ C_2H_6 to CH_4 ratio of direct samples and mobile measurements

The C_2H_6 to CH_4 ratio of gas samples from the natural gas distribution network in Heidelberg varies between 0.027 and 0.072 with lower values in winter, 0.04 ± 0.01 , and higher ones in summer, 0.06 ± 0.01 . This result can indicate that the percentage of Russian gas is higher in winter than in summer taking into account, that Nitschke-Kowsky et al., (2012) reported a C_2H_6 to CH_4 ratio for Russian natural gas to be 0.014 while for North Sea gas it is 0.078. Also the isotopic signatures of our natural gas samples support this trend with more depleted values in winter than in summer.

Gas emitted by other CH_4 sources like landfills, biogas plants and wastewater treatment plants do not contain C_2H_6 . The C_2H_6 to CH_4 ratio of the gas samples taken directly at the gas collecting systems of these sources are zero within the errors and can be clearly separated from the natural gas samples (see Fig.7). The separation due to C_2H_6 to CH_4 ratio works well with direct gas samples, but unfortunately not yet for mobile AirCore measurements. In contrast to the direct sample measurement, in mobile AirCore measurements CH_4 and C_2H_6 emitted by the source are diluted in the background. To determine the ratio, a linear fit of the measured C_2H_6 concentration to the measured CH_4 concentration is used.

However, due to the high uncertainty of the C_2H_6 measurement without averaging as it is possible for the direct sample measurement, in combination with the very small or not existing changes in C_2H_6 the fit does not provide reasonable data.”

[Referee #1]

A second aspect of the manuscript that could use more attention is the calibration of $\delta^{13}C-CH_4$. Hoheisel (2017) is cited, but at least a brief description should be provided here (and/or a more detailed description in the supplemental material). For example, there are terms in Fig 3 (e.g., $\delta^{13}CH_4$ _Nominal) which are not even described in the text. This is critical information if these data are to be used in partitioning of global or regional methane emissions. It is not clear from the text or the figure whether standards with a range of isotope ratios was used in the calibration; please clarify.

[Hoheisel et al.]

We understand the point of the referee. For a better understanding, we deleted “and calibration” on page 4 line 22, since in this section we focus on corrections. In addition, we added some explanations in section 2.3.3 and in the caption of Figure 3. This section also provides more information about the used standard.

“All data measured with the CRDS analyser in the laboratory or during mobile campaigns was corrected using the factors from Table 1 and following Fig.3 prior to the one point calibration calculation.”

“The gas cylinder used for calibration was chosen according to the experiment to ensure a similar composition and similar concentrations for sample and standard.”

“Scheme to correct and calibrate C_2H_6 and $\delta^{13}CH_4$. $\delta^{13}CH_4$ _Nominal is the nominal (known value of the standard) and $\delta^{13}CH_4$ _Standard the measured (or interpolated) value of the calibration standard.”

Detailed Comments and Questions:

[Referee #1] P1L3: "Therefore..." poor word usage. Rewrite.
[Hoheisel et al.] Agreed, we have changed it.

[Referee #1] P3L17: polypropylene
[Hoheisel et al.] Yes, corrected.

[Referee #1] P3L27: 260 amp-hr. check units
[Hoheisel et al.] Done.

[Referee #1] P3L28: Supplier & Model Number for the inverter
[Hoheisel et al.] Done, we now have included the Supplier and model number:
"e-ast HighPowerSinus HPLS 1000-12".

[Referee #1] P4L3: How were the valves triggered for "replay" mode? Add a description.
[Hoheisel et al.] We switch the valves manually after passing a methane plume. For clarification, we now have added the following phrase.
"by switching the valves manually".

[Referee #1] P6L24-25: Relative increase of which uncertainty? This statement is unclear.
Please clarify.
[Hoheisel et al.] Ok, we have changed it.
"the uncertainty of $\delta^{13}\text{CH}_4$ "

[Referee #1] P7L29: Please provide a brief description of the York fit.
[Hoheisel et al.] Thank you for pointing this out. We have now included a short description.
"York's solution is the general LSE (least-squares estimation) solution: his equations provide the best possible, unbiased estimates of the true intercept a and slope b in all cases where the points are independent and the errors are normally distributed" (Wehr and Saleska, 2017)

[Referee #1] P8L12: Why do the Keeling and MT methods produce identical source signatures and uncertainties (as shown in the figure)? I assume that this is an error in the preparation of the figure. If not, please explain.
[Hoheisel et al.] We appreciate the recommendation for clarity and we looked again into the matter. However, with both methods (KP and MT) the isotopic signature is determined identical within the first few decimal places when using the measured data and also for simulated datasets.

[Referee #1] P10L13: It's surprising to me that north sea gas is high in ethane content and heavy in isotope ratio. Generally heavy isotope ratios are associated with low C2+ content. Please confirm these values.
[Hoheisel et al.] We have checked it again and the values are correct.

[Referee #1] P11L33 (for example). The plume peak heights should be reported above the local baseline value, rather than absolute methane values. Please clarify in the description (e.g., "Peak height above baseline")
[Hoheisel et al.] We generally agree with the reviewer. However, in the text we wanted to give an expression of the CH_4 concentrations measured around the different sources. To clarify it, we change the phrase "peak height", when we report maximum CH_4 concentrations measured in the plume. Since it is correct, that the CH_4 peak heights above the background are mostly driving the fit error, we decided, to change the maximum values reported in Table 2 and in the supplements to peak heights above baseline.

- [Referee #1] P12 Landfill section: The isotope signature of landfills are complicated by a) the isotope signature of the source material in the landfill, this fractionation that occurs by the anaerobic bacteria, and the further fractionation that can occur in the landfill cap where the methane is consumed aerobically.
- [Hoheisel et al.] We now have added an additional sentence at the beginning of the section.
“In addition to the source material and fractionation during the production the isotopic signature of gas emitted by landfills depends also on fractionation processes in the upper soil layers of the landfill. Due to the presence of oxygen in the upper soil layers, aerobic bacteria oxidate parts of CH₄ which diffuses through the soil cover and shift the isotopic composition to more depleted values. Bergamaschi et al. (1998) measured these different isotopic signatures.”
- [Referee #1] P13: WWTP: The methane emissions from WWTPs can come from various locations and processes within the facility, and these processes may have different isotope signatures. Further, different WWTP plants employ different techniques to digest the waste; some discussion of these processes and their likely impact on your measurements would be helpful.
- [Hoheisel et al.] We understand the concern of the referee and added a few minor explanations (see below). However, more details about the specific processes lead too far for this rather technical paper. Indeed, the study will be continued and another paper is planned that will describe the different processes in more detail.
“There, the sludge treatment inside the digestion towers takes place in three septic tanks with a volume of 2500m³ each under anaerobic mesophilic conditions, that means without oxygen at 37°C.”

We would like to thank the reviewer for the helpful comments and suggestions. We answer each of them here after, with the original comment in blue and our response in black. We added the modifications done in the revised version in italics.

- [Referee #2] I suggest altering the title to something like “Improved method for mobile characterisation of $^{13}\text{CH}_4$ source signatures and its application in Germany.” The current title sells short of a major contribution of the paper.
- [Hoheisel et al.] Thank you for pointing this out. We have changed the title.
- Detailed comments follow:
- [Referee #2] 1. P 1, In 3: Remove “Therefore”; This sentence doesn’t logically follow from the previous sentence. It is detail in addition to the previous sentence.
- [Hoheisel et al.] Yes, corrected.
- [Referee #2] 2. P 1, In 5: Explain gas matrix or replace with less jargon.
- [Hoheisel et al.] Ok, we have changed it.
“To achieve precise results a CRDS analyser, ..., was characterised especially with regard to cross sensitivities of composition differences of the gas matrix in air samples or calibration tanks”.
- [Referee #2] 3. P 1, In 7: Abundant in many natural gases, but not all. Dry gas regions can contain only very small traces of ethane.
- [Hoheisel et al.] Yes, we have changed it.
“ C_2H_6 is typically abundant in natural gases”.
- [Referee #2] 4. P 1, In 16: A 2.8 per mil difference begs the question whether the above mentioned up to 3 per mil ethane bias could have played a role here.
- [Hoheisel et al.] In the 1990s Levin et al. (1999) measured $\delta^{13}\text{CH}_4$ with a mass spectrometer after separation of CH_4 from other components of air and so there is no cross-sensitivity with C_2H_6 .
- [Referee #2] 5. Switching between “ CH_4 ” and “methane” throughout the MS. Check for consistency.
- [Hoheisel et al.] Thanks, “methane” has been replaced in the manuscript by “ CH_4 ”.
- [Referee #2] 6. P 1, In 20: Use original data references instead of reviews, e.g., <https://agupubs.onlinelibrary.wiley.com/doi/full/10.1029/2009GL039780>
- [Hoheisel et al.] Yes, corrected.
- [Referee #2] 7. P 2, In 1: You probably mean biomass burning.
- [Hoheisel et al.] Thanks, corrected.
- [Referee #2] 8. P 2, In 2: More accurate to say “e.g. a sub-category of fossil fuel extraction”. Some gases extracted by the fossil fuel industry is biogenic.
- [Hoheisel et al.] Clarified.
- [Referee #2] 9. P 2, In 7: Do you mean to say that the isotopic signal was used to diagnose methane emission reductions? Not clear as written.
- [Hoheisel et al.] Agreed, we have clarified it.
- [Referee #2] 10. P 2, In 13: What do you mean with “all of these seasonal variations”?
- [Hoheisel et al.] Yes, this was unclear. We wanted to write “*seasonal variations*”.
- [Referee #2] 11. P 2, In 26: Briefly describe the importance/use of a storage tube (first mention).
- [Hoheisel et al.] Done.
- [Referee #2] 12. P 2, In 26: Nine facilities in 21 campaigns? Do you mean 21 measurement days?
- [Hoheisel et al.] Yes, we have changed it to “*21 mobile measurement days*”.

- [Referee #2] 13. P 3, In 15: Perhaps that's explained later, but how do you get so close to the source that you're able to sample between 50 and 90% CH₄?
- [Hoheisel et al.] We take gas samples for example directly from the gas collecting system of the landfill and the WWTP. We now have clarified it and changed the phrase.
"Gas samples taken directly from different installations (e.g. natural gas pipelines, biogas plants, gas collecting systems of landfills and wastewater treatment plants) need to be diluted before the measurement with the CRDS analyser, because such samples usually consist of between 50 and 90% CH₄."
- [Referee #2] 14. P 3, In 17: Specify the composition of synthetic air.
- [Hoheisel et al.] We now have added the composition ($20.5 \pm 0.5\%$ O₂ in N₂).
- [Referee #2] 15. P 3, In 33: Do you mean a 20-25 sec lag time between air sampling at inlet and fully arriving in the cavity? Sounds like it, but not fully clear.
- [Hoheisel et al.] Yes, we clarified it.
"Due to the length of the intake line, the volume of the cavity, and a flow rate of 160ml/min the time lag between air sampling at inlet and measurement in the cavity of the CRDS analyser is 20 to 25sec."
- [Referee #2] 16. P 4, In 11: What do you mean with representations have the same width? They don't. It's a bit confusing. The next sentence already states that both peaks represent the same emission plume.
- [Hoheisel et al.] What we meant is that when measuring a CH₄ plume first without and subsequently with the AirCore the two measured CH₄ peaks have the same height, but since we measured with a slower flow in monitoring mode not the same width. For better comparison we stretched the peak measured without the AirCore in x direction, to make it easier to compare the measurements done with and without the AirCore and to note that both peaks have the same shape and height. We also changed the text to:
"For comparison the peak measured in 'monitoring mode' (blue dots/line on the left side) is stretched by a factor of 12.5 in x-direction (blue line on the right side) so that the peak measured with the AirCore and the stretched one measured without it have the same width. The peak measured in 'replay mode' precisely corresponds to the stretched one measured in 'monitoring mode', because both peaks reproduce the same emission plume."
- [Referee #2] 17. P 4, In 13: Any hypotheses why this may be the case? This seems to be an important observation.
- [Hoheisel et al.] As described above we measured the air stored in the AirCore directly after we measured a CH₄ peak in monitoring mode. So the storage time is relatively short and we expect that the peaks measured with and without the AirCore have the same shape and the same height.
- [Referee #2] 18. Table 1: Is the unit for ethane sensitivity to water (ppm) a typo? Units of ppb would make more sense. For reference, atmospheric ethane background concentrations are in the order of < 1 ppb.
- [Hoheisel et al.] No, the unit ppm is correct.
- [Referee #2] 19. P 5, In 5: Natural gases on a continuum between 0 to 40% ethane have been measured. See Sherwood et al., 2017 (already in your refs). Hence, the importance of ethane correction varies: very important for wet gas basins (meaning lots of associated gas) and less important for very dry gas regions (mature thermogenic dry gas).
- [Hoheisel et al.] Thank you for pointing this out. In this study we measured natural gas samples from the natural gas network in Heidelberg. Therefore, we wanted to give an overview of C₂H₆ contents of natural gases in the pipeline network in Germany. We have now added:
"As typical natural gases in the pipeline network in Germany contain..."

- [Referee #2]
[Hoheisel et al.] 20. P 5, In 10: where does the 3 per mil value come from?
The value of 3‰, is the bias $\delta^{13}\text{CH}_4$ has in presence of a C_2H_6 to CH_4 ratio of 0.073, which was the highest value we measured for our natural gas samples. To make the text more understandable, we changed it.
“A correction is necessary because for typical C_2H_6 to CH_4 ratios between 0.027 and 0.073 measured for our natural gas samples, $\delta^{13}\text{CH}_4$ shows a bias between 1 and 3‰ to more enriched values. We must also keep in mind that similar shifts in, $\delta^{13}\text{CH}_4$ to less enriched values can occur when using a calibration cylinder which contains C_2H_6 .”
- [Referee #2]
[Hoheisel et al.] 21. P 7, In 3: What is the sign of the drift? Is the drift due to fractionation in the bag or due to leakage of background air into the bag?
The sign is + and the major cause for the drift is fractionation by the leakage of sample air out of the bag. So for a better understanding we added the phrase “to more enriched values” and the following sentence.
“The drift occurs especially due to fractionation by diffusion of air through the sample bag”.
- [Referee #2]
[Hoheisel et al.] 22. P 7, In 28: What are simulated data?
How we simulated data is explained a few lines below. For better understanding we changes this section:
“Similar to the method described by Wehr and Saleska (2017) for CO_2 and $\delta^{13}\text{CO}_2$, we create several typical emission plume crossings. We generated synthetic CH_4 peaks using a background concentration of 1.95 ppm CH_4 and a Gaussian curve with 10-280 equidistant data points every 3.7 s and an enhancement of 100-10000 ppb. The corresponding $\delta^{13}\text{CH}_4$ values were calculated with CH_4 source signatures between -35‰ and -65‰ and a background of -48‰. To reproduce the statistical uncertainties of a real measurement, we add a normally distributed scattering around zero to the synthetic CH_4 concentrations and the corresponding isotope ratios. The standard deviation of the normal distributed scattering depends on the CH_4 concentrations and was chosen as the Allan standard deviation measured for raw data of the analyser. However, when simulating possible improved analysers, we reduced the scattering by a factor 2 to 10. Such sets of data were generated 5000 times for each condition. To study the influence of the averaging time, we calculate mean data sets with varying averaging periods (up to 1 min). For each dataset the $\delta^{13}\text{CH}_4$ source signature was calculated with the Miller-Tans and the Keeling method using the York or the OLS fit.”
- [Referee #2]
[Hoheisel et al.] 23. P 10, In 4: I'm confused. Here it says no significant seasonal cycle has been observed. In the next paragraph, it says the values are more depleted in winter than in summer. Are there enough samples to determine this correctly?
We changed this paragraph.
- [Referee #2]
[Hoheisel et al.] 24. P 10, In 13: What are the uncertainties for the ethane-to-methane ratios?
The uncertainties for the ethane-to-methane ratios is in both cases 0.01. We now have included it.
- [Referee #2]
[Hoheisel et al.] 25. P 12, In 2: Why were the plumes expected to be smaller than on the other farm?
Both farms have a different number of dairy cows. We now added the comment “due to lower animal number”.
- [Referee #2]
[Hoheisel et al.] 26. P 12, In 11: Why are seasonal differences for the biogas plant expected?
Thanks, we have removed “seasonal”.
- [Referee #2]
[Hoheisel et al.] 27. P 12, par. 1 & 3: Is C3/C4 diet information available for the Ladenburg and Kleve farms?
No, unfortunately not.

- [Referee #2] 28. P 14, In 11: How small the fluxes were actually depends on the size of the facility (is it possible to detect all plumes at once?) and the wind conditions in addition to the measured methane concentrations.
- [Hoheisel et al.] The facility is in a small forest and so in difficult terrain. It is unlikely that we detect all plumes at once. So we weakened our statement.
- [Referee #2] 29. P 14, In 20: Did your measurements include incomplete combustion from the compressor turbines? This could be detected by the associated CO₂ or CO signal. It is still an open question whether high heat leads to isotopic fractionation of the source.
- [Hoheisel et al.] To determine the origin of a CH₄ plume measured around the natural gas facility between Hähnlein und Gernsheim is difficult, because this facility contains a natural gas storage and several compressor stations even operated by different gas providers. At this site further work and more measuring campaigns are planned to receive more detailed results.
- [Referee #2] 30. P 14, In 25: Do you use “open” and “in service” interchangeably? Not clear. If yes, it’s not a surprise that a mine currently in service produces more emissions than a closed mine.
- [Hoheisel et al.] Yes, we changed “open” to “in service”.
- [Referee #2] 31. P 14, In 30: The Bottrop mine shaft measurements do not match coal bed gas samples except for the closed mine with a large error bar and only one AirCore measurement.
- [Hoheisel et al.] Done.
- [Referee #2] 32. Conclusions: In the first paragraph, it’s important to highlight again that these results (including the ethane bias) are specific to the CRDS instrument used.
- [Hoheisel et al.] OK, we have added the following phrase to highlight your recommendation.
“characterisation of each individual analyser”

We would like to thank the reviewer for the helpful comments and suggestions. We answer each of them here after, with the original comment in blue and our response in black. We added the modifications done in the revised version in italics.

General Comments

[Referee #3] It is nevertheless regrettable that the C_2H_6 to CH_4 ratios are not presented along with the isotopic signatures. I strongly suggest to add it in order to give the paper more visibility.

[Hoheisel et al.] We appreciate the recommendation and we added the short section *C_2H_6 to CH_4 ratio of direct samples and mobile measurements* as well as one Figure.

C_2H_6 to CH_4 ratio of direct samples and mobile measurements

The C_2H_6 to CH_4 ratio of gas samples from the natural gas distribution network in Heidelberg varies between 0.027 and 0.072 with lower values in winter, 0.04 ± 0.01 , and higher ones in summer, 0.06 ± 0.01 . This result can indicate that the percentage of Russian gas is higher in winter than in summer taking into account, that Nitschke-Kowsky et al., (2012) reported a C_2H_6 to CH_4 ratio for Russian natural gas to be 0.014 while for North Sea gas it is 0.078. Also the isotopic signatures of our natural gas samples support this trend with more depleted values in winter than in summer.

Gas emitted by other CH_4 sources like landfills, biogas plants and wastewater treatment plants do not contain C_2H_6 . The C_2H_6 to CH_4 ratio of the gas samples taken directly at the gas collecting systems of these sources are zero within the errors and can be clearly separated from the natural gas samples (see Fig.7).

The separation due to C_2H_6 to CH_4 ratio works well with direct gas samples, but unfortunately not yet for mobile AirCore measurements. In contrast to the direct sample measurement, in mobile AirCore measurements CH_4 and C_2H_6 emitted by the source are diluted in the background. To determine the ratio, a linear fit of the measured C_2H_6 concentration to the measured CH_4 concentration is used.

However, due to the high uncertainty of the C_2H_6 measurement without averaging as it is possible for the direct sample measurement, in combination with the very small or not existing changes in C_2H_6 the fit does not provide reasonable data."

Specific comments

[Referee #3] The language used in the manuscript is not always precise. For instance, the terms " ^{13}C values" and " ^{13}C signatures" are not the same and sometimes confusing as used in the MS. Also, the ^{13}C signatures are not directly measured by the CRDS but determined after data treatment from the measurements. Please correct through the all manuscript.

[Hoheisel et al.] Yes, we clarified it.

[Referee #3] Most of the time, the uncertainties are presented, but without any explanation on how they are calculated and which parameters are used to computed them. It is also not clear what is the difference between uncertainties and precision. For instance, page 9 line 5/6, it is stated that increasing the number of data point improves the uncertainties on signatures. The next sentence, "uncertainties" has been replaced by "precision". It is confusing. Please clarify through the all manuscript.

[Hoheisel et al.] Yes, we corrected it.

[Referee #3] The G2201-i instrument can be used in different mode, which drives the measurement frequency of each species. Also, the methane is reported by the instrument in high range (HR) and high precision (HP) mode depending on the level of the measured mole fractions. This drives the instrumental precision. Please, add some details on these two points.

[Hoheisel et al.] We agree that this is an important point and added the following text.
"All measurements with the CRDS analyser were done in the combined CO_2/CH_4 mode to measure CH_4 and CO_2 parallel. In addition, the High Precision (HP) mode for CH_4 is chosen to provide the most precise CH_4 measurements for CH_4 concentrations up to 12 ppm."

[Referee #3] Part2.1.2 and figure1: here is a list of questions/comments which should be addressed:

[Referee #3] - how are the valves switched?

[Hoheisel et al.] We switch the valves manually, after passing a methane plume. For clarification, we now have added the phrase “*by switching the valves manually*”.

[Referee #3] - how is the flow measured and/or controlled?

[Hoheisel et al.] The flow is adjusted by needle valves and measured by a flowmeter from time to time.

[Referee #3] -why the flow presented here differs so much from the laboratory setup (160ml/min vs 20 to 80ml/min)?

[Hoheisel et al.] In the laboratory we use a rotary valve to switch between cylinder and ambient air measurements and we will have a small flow rate. In the mobile setup, we would like to have a higher flow rate for a better time resolution and a shorter lag time between air sampling at inlet and measurement in the cavity of the CRDS analyser. Therefore, we bypass the rotary valve.

[Referee #3] - what is the typical air flow going through the AC in monitoring mode?

[Hoheisel et al.] The typical air flow through the AirCore in monitoring mode is 0.8 l/min.

[Referee #3] - what is the typical vehicle speed while in monitoring mode?

[Hoheisel et al.] The speed depends strongly on the road. We tried to drive as slowly as possible, but without constraining other vehicles. Typical vehicle speeds are 10-50 km/h.

[Referee #3] - in figure1, the blue and green arrows are difficult to differentiate. Please, make them thicker.

[Hoheisel et al.] Done.

[Referee #3] Part3: how do you make sure that the direct samples are not mixed with ambient air? This could bias the isotopic signatures.

[Hoheisel et al.] We take gas samples for example directly from the gas collecting system of the landfill and the WWTP. Therefore, the CH₄ concentration of these samples is between 50-90% and a potential mixture with small amounts of ambient air would be negligible. We now have clarified it and changed the text.
“Gas samples taken directly from different installations (e.g. natural gas pipelines, biogas plants, gas collecting systems of landfills and wastewater treatment plants) need to be diluted before the measurement with the CRDS analyser, because such samples usually consist of between 50 and 90% CH₄.”

[Referee #3] Part 3.2: it has been shown in figure5 that the CH₄ peak height above the background is mostly driving the fit error. It would make much more sense to present all the peak heights above the background instead of the absolute CH₄ value.

[Hoheisel et al.] We generally agree with the reviewer. However, in the text we wanted to give an expression of the CH₄ concentrations measured around the different sources. To clarify it, we changed the phrase “peak height”, when we report maximum CH₄ concentrations measured in the plume. Since it is correct, that the CH₄ peak height above the background is mostly driving the fit error, we decided, to change the maximum values reported in Table 2 and in the supplements to peak heights above baseline.

[Referee #3] Conclusion: the Miller-Tans and Keeling approaches give the same results. Why do you suggest using one instead of the other one?

[Hoheisel et al.] We agree that this is misleading. In our study we have seen that when using the York fit, it does not matter if we use the Miller-Tans or the Keeling approach. So we changed it in the conclusion.

[Referee #3] P1, line 6: C₂H₆ only affects the 13C measurements.

[Hoheisel et al.] Yes, we changed it.

[Referee #3] P1, line 11: 13CH₄ signatures instead of values.

[Hoheisel et al.] Done.

[Referee #3] [Hoheisel et al.]	P2, line 1: "biogas burning" is not a CH ₄ source. Yes, corrected to "biomass burning".
[Referee #3] [Hoheisel et al.]	P2, line 11: do you mean "due to its origin,"? Yes, we have changed it.
[Referee #3] [Hoheisel et al.]	P2, line 15: introduce IRMS here and not line 17, and add what GC stands for. Done.
[Referee #3] [Hoheisel et al.]	P2, line 16: I suggest to delete "has been"! as shown by Röckmann et al. Done.
[Referee #3] [Hoheisel et al.]	P3, line 1: replace signature by ratio. Signatures are not directly measured by the CRDS. Corrected.
[Referee #3] [Hoheisel et al.]	P3, line 13: it is surprising to observe such a flow range (factor 4). Could you explain why and give more details? Is the flow controlled somehow? Ok, we have made several changes: <i>"The gasflow to the analyser is typically 25 to 35ml/min for calibration gas, target gas and sample bag measurements. For some applications like ambient air measurements the flow is higher with values around 80ml/min to resolve shorter temporal variabilities. The flow is measured by an electronic flow meter (model: 5067-0223, Agilent Technologies, Inc., Santa Clara, CA) before entering the analyser."</i> Tests have shown that in the flow regime of 25-80ml/min the measurement did not depend on the flow. An electronic flow meter measures the flow but the flow is not controlled.
[Referee #3] [Hoheisel et al.]	P3, line 17: what do you mean by synthetic air? Have you checked it is CH ₄ free? We now have added the composition ($20.5 \pm 0.5\%$ O ₂ in N ₂). We had also checked that it is CH ₄ free.
[Referee #3] [Hoheisel et al.]	P3, line 20: do you keep the 15min or is there a stabilization time? Yes, there is a stabilisation time, so we cut off the first 5 minutes. We have included the following sentence. <i>"However, only the last 10min were used to take into account the stabilisation time."</i>
[Referee #3] [Hoheisel et al.]	P3, line 25: decabon. Done.
[Referee #3] [Hoheisel et al.]	P3, line 33: 160ml/min to be consistent with the previous part. Changed as suggested.
[Referee #3] [Hoheisel et al.]	P4, line 5: How the measurement precision can be better in replay mode than in monitoring mode with the same instrument? Please clarify? The measurement precision of the analyser is the same in replay and in monitoring mode. What we wanted to point out is that in replay mode we have a higher time resolution and so more data points to describe the peak. The higher amount of data points also leads to a higher precision of the determined isotopic source signature. Since the sentence is misleading, we removed the phrase " <i>and a better precision</i> ".
[Referee #3] [Hoheisel et al.]	P4, line 30: have you tested the Nafion for potential fractionation? Yes, our tests did not show a fractionation.
[Referee #3] [Hoheisel et al.]	P5, line 29: Assan et al. 2017 showed that the intercept changes in time due to baseline drift. Have you regularly checked it? During the testing phase the intercept stayed constant.

[Referee #3] [Hoheisel et al.]	P5, line 33: please, check the unit. Corrected.
[Referee #3] [Hoheisel et al.]	P6, line 1: what do you mean by fully? We replaced the phrase “fully” with “yet”.
[Referee #3] [Hoheisel et al.]	P6, line 6: you cannot get a concentration range with a single cylinder. Please reformulate. Yes, we changed it. <i>“The gas cylinder used for calibration was chosen according to the experiment to ensure a similar composition and similar concentrations for sample and standard”.</i>
[Referee #3] [Hoheisel et al.]	P6, line 8: as I understood, a one point calibration strategy is used, meaning that you assume that the instrument has a linear response through the all measurement scale (mobile and sample measurements). Then why two different cylinders are used as calibration gases? We only used one calibration cylinder for each calibration. We use a cylinders for samples with atmospheric CH ₄ concentrations and a different one for samples around 10ppm CH ₄ since we have noticed that the instrument drift in $\delta^{13}\text{CH}_4$ is stronger for lower CH ₄ concentrations.
[Referee #3] [Hoheisel et al.]	P6, line 10: have you seen some changes in the CRDS regime before and after each experiments? What was the maximal drift observed and how are they taking into account? The maximal drift of the CH ₄ concentration was around 0.3ppb. We now added the phrase: <i>“To take into account possible drifts during the measurement we determined the time function of the standard ($\delta^{13}\text{CH}_4$ Standard), used in the one point calibration, for each measuring point with a linear interpolation between the two calibration measurements.”</i>
[Referee #3] [Hoheisel et al.]	P6, line 4 to 11: it is not clear to me how the CH ₄ is calibrated. There is no CH ₄ calibration factor in Table1. Yes, we have changed the text for better understanding and we added a short explanation according the one point calibration. <i>“All data measured with the CRDS analyser in the laboratory or during mobile campaigns was corrected using the factors from Table 1 and following Fig.3 prior to the one point calibration calculation.”</i> <i>“The gas cylinder used for calibration was chosen according to the experiment to ensure a similar composition and similar concentrations for sample and standard.”</i>
[Referee #3] [Hoheisel et al.]	P6, line 25: Please, detail how these uncertainties are calculated. We calculated these relative increase by comparing the error of $\delta^{13}\text{CH}_4$ before and after the correction and calibration. We changed the text to: <i>“Due to the correction and calibration $\delta^{13}\text{CH}_4$ there is a relative increase in the uncertainty of $\delta^{13}\text{CH}_4$ of approximately 3 to 12% for H₂O concentrations below 1.3% and atmospheric CH₄ concentrations.”</i>
[Referee #3] [Hoheisel et al.]	P6, line 32: is it the last 10min over the 15min measurements? Yes, corrected.
[Referee #3] [Hoheisel et al.]	P7, line 5: is it a linear drift? The uncertainties is larger than the drift itself. Yes, because the uncertainties are larger than the drift itself we did not make a drift correction. We only use it to have an estimate how long a sample can be stored in the sample bag and in addition to quantify if a bag is leaky.
[Referee #3] [Hoheisel et al.]	P7, line 19: please clarify which uncertainties you are talking about. Ok, we have corrected it to “fit uncertainties”.
[Referee #3]	P7, line 23/24: were you driving while analyzing the AC? Micro-vibrations due to vehicle motion degrade the CRDS performances.

[Hoheisel et al.] No, the vehicle stands while analysing the AC. We also noticed that especially the measurement of C_2H_6 is not as good as in the laboratory.

[Referee #3] P7, line 25: do you mean uncertainties? Is that calculated only from the fits?
[Hoheisel et al.] Yes, we changed it to "*fit uncertainty*".

[Referee #3] P8, line 17: isotopic signatures are not directly measured.
[Hoheisel et al.] Yes, we have changed it to: "*the isotopic signatures of CH_4 from the AirCore measurements*".

[Referee #3] P8, line 33: do you mean the fit error? Or is there more parameters used to derived the uncertainties?
[Hoheisel et al.] Yes, we meant fit error. Corrected.

[Referee #3] P9, line 10: what precision?
[Hoheisel et al.] Clarified. We added "*the precision of the determined $\delta^{13}CH_4$ signature*".

[Referee #3] P10, line 17/18: these criteria are already described earlier, no need to state it again here.
[Hoheisel et al.] Ok. We still keep the criteria, because we find them importance enough to remind them again in this context.

[Referee #3] P10, line 20: why is the daily mean used and not the single signatures? Same p13 line 6.
[Hoheisel et al.] We discuss the single signatures as well as the averaged daily mean values for each source. Because we had taken a different number of AirCore measurements per day and per site, the simple average over all samples can be biased compared to the average over the daily mean values.
By changing the phrase in P10 line 20 (see below) and including the phrase "*averaged daily mean*", we try to make it easier to understand in the.
"*During each measurement day one to five AirCore measurements were carried out at selected CH_4 sources. In addition to the single signatures the averaged daily mean isotopic signatures of each CH_4 source were calculated (see Fig.9, Table 2, and Supplement TableS1)*".

[Referee #3] P11, line 30: what is the peak height of the third AC? Please replaced value with signature.
[Hoheisel et al.] Done. The three peak heights are 8.3, 8.5 and 8.9ppm.

[Referee #3] P14, line 10; replace the dot with and.
[Hoheisel et al.] Done.

[Referee #3] P14, line 11: I suggest to replace monitor by sample.
[Hoheisel et al.] Yes, corrected.

[Referee #3] P14, line 15: choose between "always" and "mostly".
[Hoheisel et al.] OK, I removed both.

[Referee #3] P14, line 16: only peak heights are measured, not fluxes. I would then delete "therefore" and add "from these natural gas facilities".
[Hoheisel et al.] Yes, corrected.

[Referee #3] P15, line 3: what is the detection limit of the system? Are you sure there is plumes coming out?
[Hoheisel et al.] Ok, we changed the text.
"*Around the opencast mine Hambach, the CH_4 concentration varied only slightly between 1.94 and 1.98 ppm when we measured upwind as well as downwind of the pit. Therefore, it was not possible to identify an emission peak.*"

[Referee #3] P15, line 6: check for typo.
[Hoheisel et al.] Done.

Characterisation ~~Improved method for mobile characterisation of~~ $\delta^{13}\text{CH}_4$ source signatures ~~from methane sources and its application~~ in Germany ~~using mobile measurements~~

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Abstract.

The carbon isotopic signature ($\delta^{13}\text{CH}_4$) of several methane sources in Germany (around Heidelberg and in North Rhine-Westphalia) were characterised. ~~Therefore, mobile~~ Mobile measurements of the plume of CH_4 sources are carried out using a cavity ring-down spectrometer (CRDS). To achieve precise results a CRDS analyser, which measures methane (CH_4), carbon dioxide (CO_2) and their ^{13}C to ^{12}C ratios, was characterised especially with regard to cross sensitivities of composition differences of the gas matrix in air samples or calibration tanks. The two most important gases which affect ~~the measurements~~ $\delta^{13}\text{CH}_4$ are water vapour (H_2O) and ethane (C_2H_6). To avoid the cross sensitivity with H_2O , the air is dried with a nafion dryer during mobile measurements. C_2H_6 is typically abundant in natural gas-gases and thus in methane plumes or samples originating from natural gas. A C_2H_6 correction and calibration are essential to obtain accurate $\delta^{13}\text{CH}_4$ results, which can deviate up to 3‰ depending on whether ~~an ethane a~~ C_2H_6 correction is applied.

The isotopic signature is determined with the Miller-Tans approach and the York fitting method. During 21 field campaigns the mean $\delta^{13}\text{CH}_4$ ~~values-signatures~~ of three dairy farms ($-63.9 \pm 0.9\text{‰}$), a biogas plant ($-62.4 \pm 1.2\text{‰}$), a landfill ($-58.7 \pm 3.3\text{‰}$), a wastewater treatment plant ($-52.5 \pm 1.4\text{‰}$), an active deep coal mine ($-56.0 \pm 2.3\text{‰}$) and two natural gas storage and gas compressor stations ($-46.1 \pm 0.8\text{‰}$) were recorded.

In addition, between December 2016 and ~~June-November~~ 2018 gas samples from the Heidelberg natural gas distribution network ~~were-measured~~ were measured with a mean $\delta^{13}\text{CH}_4$ value of $-43.3 \pm 0.8\text{‰}$. Contrary to former measurements between 1991 and 1996 (~~Levin et al., 1999~~) by Levin et al. (1999) no strong seasonal cycle is shown. ~~The mean $\delta^{13}\text{CH}_4$ value of this study is $-43.1 \pm 0.8\text{‰}$ which is 2.8‰ more depleted than in former years.~~

1 Introduction

Methane (CH_4) is the second most important anthropogenic greenhouse gas. The atmospheric growth rate of ~~methane~~ CH_4 has changed significantly during the last decades, stabilising at zero growth from 1999 to 2006 before beginning to increase again

after 2007 (~~Kirschke et al., 2013~~)(Dlugokencky et al., 2009). Several studies have focused on the recent CH₄ growth caused by changes in sources and sinks (Rigby et al., 2017; Turner et al., 2017).

Recent studies by Schaefer et al. (2016), Rice et al. (2016) and Nisbet et al. (2016) have shown how the $\delta^{13}\text{CH}_4$ measurements can help to understand the changes in global CH₄ increase rates, and to assign the related source types. The stable carbon isotope ratio ($^{13}\text{C}/^{12}\text{C}$) of CH₄ sources varies due to the initial source material, and the fractionation during production and release to the atmosphere. The source categories can be classified as pyrogenic (e.g. ~~biogas-biomass~~ burning), biogenic (e.g. wetlands and livestock) or thermogenic (e.g. a sub-category of fossil fuel extraction), which show different but also overlapping isotope ratio ranges. Various studies have shown, that the assignment of isotopic signatures from different CH₄ sources remains uncertain due to large temporal variabilities and also regional specificities (e.g. Sherwood et al., 2017). This missing knowledge may result in large uncertainties when the CH₄ budget is determined on global or regional scales using isotope based estimates. In addition to global studies, the use of $\delta^{13}\text{CH}_4$ was already successfully applied by Levin et al. (1999) in Heidelberg or Lowry et al. (2001) in London. The study by Levin et al. (1999) showed the CH₄ emission reduction in the catchment area of Heidelberg which was accompanied by a significant change in the $\delta^{13}\text{CH}_4$ source mixture from -47.4‰ in 1992 to -52.9‰ in 1995/1996. ~~This was illustrated~~ Both was explained by decreasing contributions from fossil sources (~~mainly coal mining~~).

In order to apply $\delta^{13}\text{CH}_4$ in regional models, a better knowledge of the regional source signature of each CH₄ source type is needed, taking into account the temporal variations of these sources. For instance, due to ~~the its~~ origin the source signature of natural gas in Germany varies between -55‰ and -30‰ for Russia or North Sea respectively (Levin et al., 1999). In addition to ~~all of these~~ seasonal variations, changes in landfill managements like gas collector systems, and implementation of biogas plants at many farms need to be taken into account for a new study of the global and regional source signature of CH₄.

Traditionally, the isotopic ratio of CH₄ has been measured with isotope ratio mass spectrometry ~~coupled with GC (IRMS)~~ coupled with gas chromatography (GC) (Fisher et al., 2006) and this technique is still the most precise, as ~~has been~~ shown by Röckmann et al. (2016) by a comparison of dual isotope mass spectrometry (~~IRMS~~), quantum cascade laser absorption spectroscopy (QCLAS), and cavity ring down spectroscopy (CRDS). Instrumental development in measurement technique now allows isotope analysis of $\delta^{13}\text{CH}_4$ by CRDS analyser and even its use on a mobile platform (Rella et al., 2015; Lopez et al., 2017). This is a further improvement to the study of Zazzeri et al. (2015), which involved collecting air samples in bags and analysing them later in the laboratory by IRMS. The studies of Rella et al. (2015) and Assan et al. (2017) demonstrated the importance of a careful determination of cross sensitivities and a good calibration strategy for precise isotope measurements with a CRDS analyser.

In this paper, a strategy to monitor and determine the isotopic carbon source signature of major CH₄ sources in Germany using mobile measurements is presented. One major aspect is a careful characterisation of the CRDS analyser to take into account the cross sensitivity between $\delta^{13}\text{CH}_4$ and other components like water vapour and ethane (C₂H₆), and to improve the use of a storage tube, for later reanalysis, described by Rella et al. (2015). During 21 mobile measurement ~~campaigns~~ days, emission plumes from a biogas plant, three dairy farms, a landfill, a wastewater treatment plant, two natural gas facilities and a bituminous deep coal mine were ~~able to be~~ measured with our setup.

2 Methods

2.1 Experimental setup

The core component of our experimental setup is the commercially available cavity ring-down spectrometer (CRDS) G2201-i (Picarro, Inc., Santa Clara, CA) which measures routinely the mole fraction of $^{12}\text{CO}_2$, $^{13}\text{CO}_2$, $^{12}\text{CH}_4$, $^{13}\text{CH}_4$, and H_2O in an air sample. In addition to the raw spectroscopic measurements, the analyser automatically calculates and outputs the carbon isotopic ~~signature ratio~~ $\delta^{13}\text{CH}_4$ and $\delta^{13}\text{CO}_2$. Furthermore, the mole fraction of C_2H_6 is measured as an additional feature, which had to be investigated and calibrated for our analyser. All measurements with the CRDS analyser were done in the combined CO_2/CH_4 mode to measure CH_4 and CO_2 parallel. In addition, the High Precision (HP) mode for CH_4 is chosen to provide the most precise CH_4 and $\delta^{13}\text{CH}_4$ measurements for CH_4 concentrations up to 12 ppm. A more detailed description of this type of CRDS analyser can be found in Rella et al. (2015). Two different setups are used in this study: a laboratory setup for sample bag analysis and test series and a mobile configuration in a vehicle.

2.1.1 Laboratory setup

In the laboratory in Heidelberg the analyser continuously measures ambient air alternating with regular calibration gas and quality control gas injections (Dinger, 2014). In addition, diluted samples from different CH_4 sources and gas cylinders can be measured and calibrated. The mobile measurements ~~, using the analyser in a vehicle,~~ are also calibrated using the immediate calibration runs in the laboratory before and after a mobile campaign.

The schematic of the laboratory setup is shown in Figure 1 (a). A 16-port rotary-valve (model: EMT2CSD16UWE, Valco Vici, Switzerland) can be switched automatically by the analyser, to change between different measurements. Ambient air is measured at port 1. Port 3, 7 and 15 are reserved for calibration and quality control measurements. Sample bags are measured on port 11 or 13. The gasflow to the analyser ~~of typically 20 to~~ is typically 25 to 35 ml/min for calibration gas, target gas and sample bag measurements. For some applications like ambient air measurements the flow is higher with values around 80 ml/min to resolve shorter temporal variabilities. The flow is measured by an electronic flow meter (model: 5067-0223, Agilent Technologies, Inc., Santa Clara, CA) before entering the analyser.

Gas samples ~~from different CH_4 sources taken directly from different installations~~ (e.g. natural gas ~~, biogas, landfill gas pipelines, biogas plants, gas collecting systems of landfills and wastewater treatment plants~~) need to be diluted before the measurement with the CRDS analyser, because such samples usually consist of between 50 and 90 % CH_4 . Therefore, approximately 40 μl of the sample was injected into a three litre bag (Tedlar[®] with ~~Polypropylen~~ polypropylene valve with septum, Restek GmbH, Bad Homburg, Germany) filled with synthetic air ($20.5 \pm 0.5\%$ O_2 in N_2) to dilute the CH_4 concentration to approximately 10 ppm in the new sample bag. Due to cross sensitivity with water vapour, these gas samples were dried using a cooling trap below a mole fraction of 0.0015 % water vapour previous to analysis. Each diluted sample was measured for 15 min. However, only the last 10 min were used to take into account 5 min stabilisation time.

2.1.2 Mobile measurement setup

For mobile measurements the CRDS analyser is installed inside a vehicle and measures air while driving. The system consists of the CRDS analyser, a nafion dryer and a storage tube, the so called AirCore (Karion et al., 2010), which enable us to remeasure the stored air from the last 2 minutes of continuous measurement (Fig.1 (b)). The AirCore was built after Rella et al. (2015) using a 25 m ~~Decarbon~~ decabon tubing with an inner diameter of 9.5 mm and a volume of 1.77 l (Yeman, 2015). The setup of the nafion dryer is similar to the one built by Welp et al. (2013) using a Perma Pure MD-070-96-S nafion dryer and a vacuum pump. The CRDS device and the vacuum pumps are powered by a portable power source (260 ~~h~~ Ah deep cycle battery, Winnerbatterien Germany) and a 1000 W inverter (e-ast HighPowerSinus HPLS 1000-12) which offers 230 V output) which allows for over 12 hours of measurement time.

The ambient air enters the air intake line 20 cm above the vehicle roof. It can follow two different paths to the analyser depending on the valve positions. In the ‘monitoring mode’, indicated by blue arrows in Fig.1 (b), the ambient air enters the CRDS analyser after the air is dried with the nafion dryer to a mole fraction of less than 0.1 % water vapour. Simultaneously, a second split-off flow leads the ambient air through the AirCore. Due to the length of the intake line, the volume of the cavity, and a flow rate of ~~0.16 l~~ 160 ml/min the ~~air needs approximately~~ time lag between air sampling at inlet and measurement in the cavity of the CRDS analyser is 20 to 25 sec ~~to be measured in the CRDS analyser.~~

The vehicle usually passes an emission plume of a CH₄ source within 40 sec and the analyser records approximately 10 data points per CH₄ peak. To achieve higher time resolution and accuracy for $\delta^{13}\text{CH}_4$ analysis, it is possible to remeasure CH₄ peaks by analysing the air stored in the AirCore with the ‘replay mode’ by switching the valves manually. This enables us to remeasure the stored air that contains the sampled CH₄ peak. The average analysis time is then 4.5 min corresponding to approximately 70 data points and thus the measurement in ‘replay mode’ has a higher time resolution ~~and a better precision~~ than the one in ‘monitoring mode’.

In Figure 2 (left) a typical mobile measurement of a plume from a biogas plant close to Heidelberg (Germany) is shown for CH₄ concentrations and $\delta^{13}\text{CH}_4$ values. The vertical black line indicates the switching from ‘monitoring mode’ to ‘replay mode’. The small dots represent the reported data in monitoring (blue) and replay (black) mode, logged approximately every 3.7 sec, while the red lines show the 15 sec averages in ‘replay mode’. For comparison the peak measured in ‘monitoring mode’ (blue dots/line on the left side) is stretched by a factor of 12.5 in x-direction (blue line on the right side) so that the ~~representations of both peaks~~ peak measured with the AirCore and the stretched one measured without it have the same width. The peak measured in ‘replay mode’ precisely corresponds to the stretched one measured in ‘monitoring mode’, because both peaks reproduce the same emission plume. This differs from the AirCore measurements performed by Lopez et al. (2017) which show higher CH₄ ~~values~~ concentrations in replay than in ‘monitoring mode’.

During the mobile measurements the vehicle position was recorded by a GPS mouse (Navilock 602u) with an accuracy of 2 m CEP (circular error probable). A weather station (Vantage Pro2 TM, Davis Instruments) was set up near the measurement site to record the wind speed and direction, the temperature and the incident solar radiation.

2.2 Characterisation of the CRDS analyser G2201-i

2.2.1 Correcting the measured $\delta^{13}\text{CH}_4$ values

With regard to the publications of Rella et al. (2015) and Assan et al. (2017) our main focus during the instrumental characterisation was on $\delta^{13}\text{CH}_4$. The cross sensitivities of H_2O , CH_4 , CO_2 and C_2H_6 concentrations on $\delta^{13}\text{CH}_4$ were investigated to determine correction factors. The correction factors subsequently applied in this study are summarised in Table 1. The correction and calibration scheme is sketched in Fig. 3 and described in more detail in Hoheisel (2017). The H_2O interference on $\delta^{13}\text{CH}_4$ was tested by carrying out several humidity tests (Fig. S1). For this purpose, two dry compressed air gases with gas mixtures of 2.3 ppm and 10.1 ppm CH_4 were humidified by flushing them through a reversed glass condensation trap kept at room temperature and filled with one droplet of deionised water. Due to evaporation of the water droplet the humidity of the gas passing the condensation trap changed with time between 1.5 to 0 % water vapour. Rella et al. (2015) recommended a reduction of the humidity below a mole fraction of 0.1 % water vapour for accurate $\delta^{13}\text{CH}_4$ results. Our tests confirm this recommendation for humidity levels below 0.15 % but observed a significant cross sensitivity of $0.54 \pm 0.29 (\text{‰}^{13}\delta\text{CH}_4)(\text{‰}\text{H}_2\text{O})^{-1}$ for humidity levels above 0.15 %. To reduce possible uncertainties due to humidity correction, the air was dried with a nafion dryer below a mole fraction of 0.1 % water vapour during mobile measurement. However, the nafion drying unit was not installed until September 2016, so the measurements before this date were corrected.

Additionally, the cross sensitivities of CH_4 and CO_2 on $\delta^{13}\text{CH}_4$ were tested (Fig. S2 and S3). Two dilution tests were carried out, generating different gas mixtures. No significant cross sensitivities of CH_4 and CO_2 on $\delta^{13}\text{CH}_4$ were detected up to concentrations of 10 ppm CH_4 or rather 450 ppm CO_2 .

Previous studies from Rella et al. (2015) and Assan et al. (2017) have reported higher $\delta^{13}\text{CH}_4$ results when the gas sample contains C_2H_6 . ~~As natural gas contains~~ As typical natural gases in the pipeline network in Germany contain between 1.4 to 7 Mol% of C_2H_6 (Nitschke-Kowsky et al., 2012), the C_2H_6 interference is especially important when analysing CH_4 emissions from natural gas facilities or the isotopic composition of natural gas. The C_2H_6 interference on $\delta^{13}\text{CH}_4$ measurements was carefully tested with our analyser by carrying out three dilution tests, to determine a correction (Fig. S4). $\delta^{13}\text{CH}_4$ increases linearly with increasing C_2H_6 to CH_4 ratio. The slope of the regression line and thus the correction factor was found to be $40.87 \pm 0.49 \text{‰} (\text{ppm}\text{CH}_4)/(\text{ppm}\text{C}_2\text{H}_6)$. ~~The A~~ correction is necessary due to because for typical C_2H_6 to CH_4 ratios between 0.027 and 0.073 measured for our natural gas samples, $\delta^{13}\text{CH}_4$ showing a bias ~~of up to between 1 and 3‰ in our study depending on the to more enriched values. We must also keep in mind that similar shifts in $\delta^{13}\text{CH}_4$ to less enriched values can occur when using a calibration cylinder which contains C_2H_6 ratio of the sample and the calibration cylinder.~~

2.2.2 Correcting the measured C_2H_6 concentration

To correct for the strong cross sensitivity between C_2H_6 and $\delta^{13}\text{CH}_4$ measurements, an accurate determination of the C_2H_6 concentration is required. Because the measurement of C_2H_6 is an additional feature of the instrument a correction and calibration of the C_2H_6 concentration were performed.

The C₂H₆ concentration decreases strongly with increasing humidity, even for H₂O concentrations below 0.15 % (Fig. S1). For humidity below 0.15 % a correction factor of 0.43 ± 0.51 (ppm C₂H₆)/(% H₂O) was determined and for humidity higher than 0.16 % the correction factor is 0.70 ± 0.10 (ppm C₂H₆)/(% H₂O). There is no correction for H₂O mole fractions between 0.15 and 0.16 %, because in this range the behaviour of C₂H₆ in the presence of H₂O changes. However, no discontinuity, such
5 that observed by Assan et al. (2017), was seen.

Besides H₂O also the concentrations of CH₄ and CO₂ interfere with the measured C₂H₆. To determine the cross sensitivities of CH₄ and CO₂ on C₂H₆ two dilution series and three injection tests were performed and produced gas mixtures with concentration ranges of 1.8 to 10 ppm CH₄ or 2 to 600 ppm CO₂. All dilution and injection tests with C₂H₆ concentrations between 0 to 1.3 ppm show similar results with an average of 0.0077 ± 0.0007 (ppm C₂H₆)/(ppm CH₄) and $(1.25 \pm 0.94) \cdot 10^{-4}$ (ppm
10 C₂H₆)/(ppm CO₂) (Fig. S5).

To calibrate the C₂H₆ measurement two dilution tests with C₂H₆ concentrations ranging from 0 to 3 ppm were performed (Fig. S6). The measured C₂H₆ concentrations were nearly twice as large as expected. After correcting the measured C₂H₆ concentrations due to H₂O, CH₄ and CO₂ a calibration factor (slope of the regression line) of 0.538 ± 0.002 ppm/ppm and a calibration intercept of 0.070 ± 0.005 ppm was determined.

15 2.2.3 Calibration to international scales

All calibration gases used in this study are compressed air filled in aluminium cylinders. The CH₄ and CO₂ concentrations were calibrated against the WMO scale (Dlugokencky et al., 2005) using a GC system (Levin et al., 1999). To determine the $\delta^{13}\text{CH}_4$ values, flasks filled from our calibration gases were ~~sent to analysed at~~ MPI Jena ($\delta^{13}\text{CH}_4$: ± 0.05 ~~ppm‰~~). These analyses connect our Heidelberg measurements to the VPDB (Vienna Pee Dee Belemnite) isotope scale (Sperlich et al., 2016).
20 C₂H₆ is not ~~fully yet~~ calibrated to an international scale. One calibration cylinder filled by Deuste-Steininger (Mühlhausen, Germany) with 4.98 ppm C₂H₆ is certified by this company with an uncertainty of ± 2 %.

All data measured with the CRDS analyser in the laboratory or during mobile campaigns was corrected ~~prior to the one point calibration calculation~~ using the factors from Table 1 and following Fig. 3 ~~:-prior to the one point calibration calculation.~~

The gas cylinder used for calibration was chosen according to the experiment to ensure a similar composition and ~~concentration~~
25 ~~range similar concentrations~~ for sample and standard. For ambient air measurements in the laboratory and for mobile measurements a gas cylinder filled with compressed air is used to calibrate the data. For diluted gas samples from CH₄ sources a gas cylinder with atmospheric concentrations spiked with natural gas to 10 ppm CH₄ is used. The calibration gas is measured before and after every experiment/field campaign in the laboratory or in the vehicle. Tests at the beginning of this study showed that measurements of the calibration gas inside the vehicle do not increase the precision and are therefore not necessary for mo-
30 bile measurements of less than 10 hours. To take into account possible drifts during the measurement we determined the time function of the standard ($\delta^{13}\text{CH}_4^{\text{Standard}}$), used in the one point calibration, for each measuring point with a linear interpolation between the two calibration measurements.

2.2.4 Instrument performance and uncertainties

The repeatability of the analyser as a function of the CH_4 concentration was determined by the measurement of three different gas cylinders for 120 min each. The Allan variance (Werle et al., 1993) was calculated with the raw data for averaging times of up to 11 min (Fig. 4). The Allan standard deviation σ (the square root of the Allan variance σ^2) for the raw (3.7 sec) CH_4 data is between 0.34 to 2.69 ppb for gases with a CH_4 concentration of 1900 to 10000 ppb. For the corresponding $\delta^{13}\text{CH}_4$ data, an improvement of the Allan standard deviation with higher CH_4 concentration from 3.76 to 0.77‰ can be seen. The Allan standard deviation of C_2H_6 is approximately 0.09 ppm for gases with C_2H_6 concentrations up to 5 ppm.

During mobile measurements especially CH_4 and $\delta^{13}\text{CH}_4$ show rapid changes when driving through the emission plume of a CH_4 source and thus do not allow us to average the data over long time periods. However, for sample measurements in the laboratory (e.g. natural gas samples) longer averaging times of up to 10 or 15 min significantly decrease the Allan standard deviation (see Fig. 4). For a 10 min averaging period the Allan standard deviation of 1900 ppb or 10000 ppb CH_4 decreases to values of 0.09 ppb and 0.47 ppb, and for $\delta^{13}\text{CH}_4$ to values of 0.40‰ and 0.06‰. The Allan standard deviation of C_2H_6 decreases to 0.006 ppm. Due to the correction and calibration of $\delta^{13}\text{CH}_4$ there is a relative increase in the uncertainty of approximately 5‰ $\delta^{13}\text{CH}_4$ of approximately 3 to 12 ‰ for H_2O concentrations below 1.3 % and atmospheric CH_4 concentrations.

2.3 Analysis of $\delta^{13}\text{CH}_4$

2.3.1 Gas samples from natural gas distribution network

Between December 2016 and June–November 2018, gas samples from the Heidelberg natural gas distribution network were collected two to three times a month from the gas-glass blowing workshop at the university campus in one litre sample bags (Tedlar® with Polypropylen-polypropylene valve with septum, Restek GmbH, Bad Homburg, Germany).

The gas samples were measured as described in Sect. 2.1.1, corrected by the factors given in Table 1 and calibrated as described above. For each gas sample the average and standard deviation of the last 10 min over the 15 min measurement were calculated.

To determine the repeatability of a measurement as well as the storage effect, pair samples were taken and storage tests carried out, with storage times of the bags up to 226 days and two to five measurements taken from each sample bag. Duplicate samples taken on the same day and measured one after another show a mean difference in $\delta^{13}\text{CH}_4$ of 0.12 ± 0.08 ‰ with a maximal difference of 0.30‰. Storage tests of 12 natural gas samples stored on average for 104 days (41 to 226 days) in Tedlar® bags show an average drift of 0.0023 ± 0.0028 ‰/day to more enriched values. The drift occurs especially due to fractionation by diffusion of air through the sample bag.

Since the samples are measured for the first time on average 26 days (0 to 88 days) after the sample day, the $\delta^{13}\text{CH}_4$ signature of the samples-value of a sample will change by approximately 0.06‰ due to this storage in Tedlar® bags. Even after 100 days the average drift is only 0.23‰ and therefore for each sample the $\delta^{13}\text{CH}_4$ values measured within 100 days after sampling were averaged. To quantify the short-term variations of $\delta^{13}\text{CH}_4$ from the local gas supply network within one week, two samples per

~~day-daily gas samples~~ were taken over 5 days at the end of November 2017 and averaged the $\delta^{13}\text{CH}_4$ values for the duplicate ~~samples-2017~~. The maximal difference between the ~~five averaged values samples~~ was $0.7 \pm 0.2\text{‰}$.

2.3.2 Determination of $\delta^{13}\text{CH}_4$ source signatures from mobile plume measurements

For mobile measurements the CRDS analyser is installed inside a vehicle and measurements are carried out as described in Sect. 2.1.2. The $\delta^{13}\text{CH}_4$ signature of the CH_4 sources were determined by the Miller-Tans approach (Miller and Tans, 2003) using the unaveraged data measured in ‘replay mode’ with the AirCore. To fit a linear regression line to the data the York fit (York et al., 2004) was used as recommended also by Wehr and Saleska (2017). York’s solution is the general least-squares estimation solution, providing the best possible, unbiased estimates of the true intercept and slope in all cases where the points are independent and the errors are normally distributed (Wehr and Saleska, 2017). Because the York fit allows errors in x and y, it also account for the finding that the analyser can measure $\delta^{13}\text{CH}_4$ more accurately at higher CH_4 concentrations. The errors for CH_4 and $\delta^{13}\text{CH}_4$ for different concentrations were determined with the Allan standard deviation.

For accurate results the following criteria are used to select 79 AirCore measurements out of 135. Only $\delta^{13}\text{CH}_4$ signatures with fit uncertainties lower than 5‰ are used. The number of data points and especially the peak height above background concentration control the ~~precision-uncertainty~~ of the determined isotopic signature when applying a Miller-Tans Plot, therefore only plume measurements with peak heights above background concentration higher than 0.45ppm and more than 25 data points fulfil this criterion. Furthermore, in some cases the reported C_2H_6 concentration jumps while driving although there cannot be a change in the C_2H_6 concentration of the ambient air. These jumps in C_2H_6 also results in $\delta^{13}\text{CH}_4$ jumps. Therefore, all AirCore measurements with a sudden change in C_2H_6 larger than 1ppm were neglected. With these criteria the isotopic signature of a CH_4 source determined from one AirCore plume measurement has an average ~~precision-fit uncertainty~~ of $1.8 \pm 1.3\text{‰}$.

2.3.3 Comparison of different methods to determine $\delta^{13}\text{CH}_4$ source signatures

In order to define the optimal method for the determination of the source signature the 135 AirCore measurements as well as simulated data were used. In the following the differences in the $\delta^{13}\text{CH}_4$ source signature when using the Keeling method or the Miller-Tans approach (Keeling, 1958; Miller and Tans, 2003) will be discussed and the York fit will be compared to the ordinary least squares (OLS) fit (here the `lm()` fit function from GNU R is used).

Similar to the method described by Wehr and Saleska (2017) for CO_2 and $\delta^{13}\text{CO}_2$, we ~~simulated-create~~ several typical emission plume crossings ~~with CH_4 source signatures of -35‰ to -65‰ and a background of -48‰ . In addition, the CH_4 concentration enhancements in the plume $\Delta\text{e}_{\text{source}}$ ($100-10000\text{ppb}$), the number of measured data points during plume crossing n ($10-280$) and the averaging times (up to 1min) were varied. For each set of conditions ($\delta^{13}\text{CH}_{4\text{source}}$, $\Delta\text{e}_{\text{source}}$, n), we generated synthetic CH_4 concentrations and calculated the corresponding $\delta^{13}\text{CH}_4$ values peaks using a background concentration of 1.95ppm CH_4 and a Gaussian curve with $n=10-280$ equidistant data points every 3.7s and a peak height of $\Delta\text{e}_{\text{source}}$ an enhancement of $100-10000\text{ppb}$. The corresponding $\delta^{13}\text{CH}_4$ values were calculated with CH_4 source signatures between -35‰ and -65‰ and a background of -48‰ .~~ To reproduce the statistical uncertainties of a real measurement, we

add a normally distributed scattering around zero to the synthetic CH₄ concentrations and the corresponding isotope ratios. The standard deviation of the normal distributed scattering depends on the CH₄ concentrations and was chosen as the Allan standard deviation measured for raw data of the analyser. However, when simulating possible improved analysers, we reduced the scattering by a factor 2 to 10. Such sets of data were generated 5000 times for each condition. [To study the influence of the](#)

5 [averaging time, we calculate mean data sets with varying averaging periods \(up to 1 min\).](#) For each dataset the $\delta^{13}\text{CH}_4$ source signature was calculated with the Miller-Tans and the Keeling method using the York or the OLS fit.

For the York fit the $\delta^{13}\text{CH}_4$ source signature determined using the Miller-Tans approach is identical within the relevant order of magnitude to the one calculated using the Keeling method. This can be shown for the AirCore measurements and is confirmed by our simulations. Figure 2 (right) shows an example of the Keeling Plot (upper panel) and the Miller-Tans Plot
10 (lower panel) used to calculate the isotopic signature of the corresponding CH₄ source.

The $\delta^{13}\text{CH}_4$ signature calculated with the simple OLS fit out of the AirCore measurements differ between -2 and 2‰ depending on the method which is used (Miller-Tans or Keeling method). This finding is in agreement with the simulated results.

~~Comparing the measured isotopic~~ [Comparing the isotopic source](#) signatures of CH₄ [from the AirCore measurements](#) result-
15 ing from York and OLS fit in approximately 90 % of the measurements the result of the York fit lies in between the results from the OLS with the Miller-Tans and the Keeling method. This agrees well with our simulated results, where the value of the $\delta^{13}\text{CH}_4$ signature determined with the York fit for peak enhancements between 0.1 and 3 ppm lies between the values calculated for the OLS fit with Miller-Tans and Keeling method in more than 98.5 % of the results.

The average values for the 5000 determined isotopic signatures for the York and the OLS (Keeling and Miller-Tans) fit in
20 this study are nearly the same ($<0.05\text{‰}$ for CH₄ ~~ranges-peaks~~ higher than 0.2 ppm) and have in all three cases significant larger differences to the true value ($<0.2\text{‰}$ for CH₄ ~~ranges-peaks~~ higher than 0.2 ppm and $<0.1\text{‰}$ for CH₄ ranges higher than 0.6 ppm) than between each other. However, the 5000 individual simulated values for the $\delta^{13}\text{CH}_4$ signatures for one condition vary widely around the average and the true value.

Due to the above described comparisons, the York fit and the Miller-Tans approach were chosen to determine the $\delta^{13}\text{CH}_4$
25 source signature in our study. A further characterisation of this method showed that the uncertainty of a single source signature determination depends mainly on three criteria: the CH₄ range, the number of data points used for the fit and the precision of the analyser.

The first large limitation for a precise determination of the isotopic source signature is the CH₄ concentration of the plume above background. The higher the CH₄ peak the more accurately the $\delta^{13}\text{CH}_4$ signature can be determined. Especially for small
30 CH₄ sources, it is important to drive as close as possible to the source to increase the peak height. In Fig. 5 the ~~uncertainty~~ [fit error](#) of the isotopic signature of every AirCore measurement (black dots) is given as a function of CH₄ peak height above background. For CH₄ enhancements lower than 1 ppm the uncertainty increases strongly to values higher than 20‰. The coloured lines show the standard deviation of the 5000 synthetic data with different numbers of data points used for the Miller-Tans approach. The synthetic data agrees well with the measured values which were calculated out of 25 to 280 data points.

The second parameter which influences the accuracy of the determined $\delta^{13}\text{CH}_4$ signature is the amount of data points. During measurements significantly different isotopic signatures were ~~measured in~~ determined using the ‘monitoring’ (approximately 10 data points) ~~and in or~~ ‘replay mode’ with the AirCore (on average approximately 70 data points) (see Fig. 2). The synthetic data confirms that with increasing amount of data points the uncertainty of the $\delta^{13}\text{CH}_4$ signature improves (Fig. 5). The precision can be more than doubled by increasing the number of points from 10 to 70 and more than quadrupled by an increase from 10 to 280. In ‘monitoring mode’ the amount of data points per peak is constrained by the small width of the plume and the driving speed. Therefore, it is important to remeasure the plume using the AirCore to increase the number of data points and thus the precision of the determined $\delta^{13}\text{CH}_4$ signature.

The third limitation of the accuracy of the determined source signature is the measurement precision of the instrument for raw (3.7 sec) data, especially for $\delta^{13}\text{CH}_4$. The measuring intervals of the plume are short and thus the CH_4 concentration and isotopic ~~signature~~ composition change rapidly, making it impossible to increase the precision through averaging over time periods longer than one minute. The value as well as the uncertainty of the isotopic signatures determined from the original and the 15 sec averaged data from AirCore measurements do not show significant differences using the Keeling or the Miller-Tans approach for the real measurements (see Fig. 2). Moreover, additional tests with synthetic data show that averaging over 7 to 60 sec improves the precision of the measurement, but not the source signature determination due to a smaller amount of data points. Therefore, the raw unaveraged data from the analyser measured in ‘replay mode’ was used instead of the averaged one. The Allan standard variance without averaging for $\delta^{13}\text{CH}_4$ is up to 3.76‰ (1.9 ppm CH_4).

An increase of the precision to a standard deviation of 1‰ would lead to a nearly four times better precision of the determined isotopic source signature. For future measurements more precise instruments are important. Finally, simulated results for different isotopic source signatures were compared and no dependence on the determined ~~methane~~ CH_4 source signature was noticed.

3 Results

3.1 $\delta^{13}\text{CH}_4$ from Heidelberg gas distribution network

Between 1991 and 1996 ~~measurements of the natural gas~~ $\delta^{13}\text{CH}_4$ measurements of natural gas from the distribution network in Heidelberg were carried out by Glatzel-Mattheier (1997). The measured $\delta^{13}\text{CH}_4$ ~~signatures~~ values underlied a strong seasonal variation with -30‰ in winter and up to -50‰ in summer. The annual average was $-40.3 \pm 3.0\text{‰}$ (Glatzel-Mattheier, 1997; Levin et al., 1999). The seasonal cycle in the isotopic ~~signature~~ composition of natural gas in the 1990s was explained by seasonal changes in gas imports with a larger contribution from Russian gas in summer months and mainly from northern Germany and Scandinavia during winter, because the isotopic signature of natural gas differs depending on its formation process and therefore its origin. Natural gas from Siberia has an isotopic signature between -48 to -54‰ (Cramer et al., 1998) and is thus less enriched than North Sea gas with $\delta^{13}\text{CH}_4$ ~~values~~ signatures of approximately $-34 \pm 3\text{‰}$ (Lowry et al., 2001). In the late 1990s the percentage of natural gas from import and domestic production in Germany (BAFA, 2017) varies

with the seasons. While in summer 1998 and 1999 approximately 44 % of the natural gas imports in Germany originate from Russia, in winter it was only 25 to 30 %.

Between December 2016 and ~~June-November~~ 2018 the measured $\delta^{13}\text{CH}_4$ ~~signatures-values~~ vary between -44.7‰ and -41.4‰ with an average value of $-43.143.3 \pm 0.8\text{‰}$ (Fig.6). No ~~significant seasonal cycle~~ strong seasonal cycle as in the 1990s has been observed during these 19 months. The measurements in our recent study show that natural gas in Heidelberg is nowadays on average approximately 2.83‰ more depleted than in the 1990s. The percentage of natural gas from import and domestic production in Germany (BAFA, 2017) affirm our findings of no ~~significant~~ strong seasonal cycle, with reporting a mixture of natural gas which is nearly the same over the year. It should be noted that the statistics are for Germany as a whole, while no information for the Heidelberg region is available from the local gas network.

~~A closer look at the measured isotopic signature of natural gas of the last year (2017) (Fig.6) shows that the isotopic signature of natural gas-~~

3.2 C₂H₆ to CH₄ ratio of direct samples and mobile measurements

The C₂H₆ to CH₄ ratio of gas samples from the natural gas distribution network in Heidelberg varies between 0.027 and 0.072 with lower ratios in winter, 0.04 ± 0.01 , and higher ones in ~~Heidelberg is more depleted in winter than in summer, which is~~ opposite to the results found in the 1990s. Our results 0.06 ± 0.01 . This finding can indicate that the percentage of Russian gas is higher in winter than in summer ~~.The taking into account, that Nitschke-Kowsky et al. (2012) reported a C₂H₆ to CH₄ ratio seems to for Russian natural gas to be 0.014 while for North Sea gas it is 0.078. Also the isotopic signatures of our natural gas samples support this trend with lower slightly more depleted values in winter ,0.04, and higher ones in summer, 0.06. Nitschke-Kowsky et al. (2012) reported the~~ than in summer.

Gas emitted by other CH₄ sources like landfills, biogas plants and wastewater treatment plants do not contain C₂H₆. The C₂H₆ to CH₄ ratio for Russian natural gas to be 0.014 while for North Sea gas it is 0.078 of the gas samples taken directly at the gas collecting systems of these sources are zero within the errors and can be clearly separated from the natural gas samples (see Fig.7).

The separation due to C₂H₆ to CH₄ ratio works well with direct gas samples, but unfortunately not yet for mobile AirCore measurements. In contrast to the direct sample measurement, in mobile AirCore measurements CH₄ and C₂H₆ emitted by the source are diluted in the background. To determine the ratio, a linear fit of the measured C₂H₆ concentration to the measured CH₄ concentration is used. However, due to the high uncertainty of the C₂H₆ measurement without averaging as it is possible for the direct sample measurement, in combination with the very small or not existing changes in C₂H₆ the fit does not provide reasonable data.

3.3 $\delta^{13}\text{CH}_4$ source signatures from mobile measurements

The $\delta^{13}\text{CH}_4$ signature for different ~~methane~~ CH₄ sources (see Fig.78) are determined out of 135 plumes measured over 21 days while using the AirCore. For the evaluation only 79 AirCore measurements with peak heights of more than 0.45 ppm above background and more than 25 data points were selected (see Sect. 2.3.2). During each measurement day one to five AirCore

measurements were carried out at selected CH₄ sources and the determined isotopic ~~signatures of each CH₄ source~~ source signatures were averaged to ~~a daily mean~~ daily means. For each source the daily mean values calculated for each visit were averaged (see Fig. 89, Table 2, and Supplement Table S1).

In the following the determined isotopic signatures of CH₄ sources will be discussed for every measuring site and be
5 compared with values from other studies and $\delta^{13}\text{CH}_4$ signatures measured from gas samples taken at selected measuring sites.

3.3.1 Biogas plant

In biogas plants, microbial organisms produce CH₄ under anaerobic conditions. The isotopic signature of CH₄ in biogas can vary widely due to the substrate, the microbial producers of CH₄ and kinetic values like temperature and frequency of feeding
10 (Polag et al., 2015; personal communication with D. Polag, 2017).

The biogas plant Pfistererhof in Heidelberg has two fermenter tanks. One is fed with a substrate mainly consisting of maize silage and the other predominantly of food waste. Gas samples from both fermenter tanks were taken and measured. The $\delta^{13}\text{CH}_4$ signature of the produced biogas was $-61.5 \pm 0.1\text{‰}$ for the maize-silage tank and $-64.1 \pm 0.3\text{‰}$ for the food-waste tank. Therefore, the isotopic source signature determined out of the measurement of the CH₄ plume is expected to lie between
15 the above mentioned values, because CH₄ from both fermenter tanks is mixed downwind of the biogas plant.

Over 10 days, mobile measurements were carried out downwind of the biogas plant between August and December 2016 and in February and March 2017. The maximum CH₄ ~~peak height~~ concentrations of the measured plumes varied between 2.5 and 17 ppm. Often multiple peaks were measured while driving through the plume, caused by several sources on the biogas plant. The isotopic signatures of CH₄ emitted by the biogas plant were determined out of 17 measured plumes. The values
20 varied between -59.0 and -64.2‰ with one exception of -67.4‰ and the ~~overall~~ average of $\delta^{13}\text{CH}_4$ was $-62.4 \pm 1.2\text{‰}$. The ~~overall average and also the daily averages~~ determined isotopic signatures agree well with the isotopic signatures of the direct samples.

3.3.2 Dairy farms

The $\delta^{13}\text{CH}_4$ source signature emitted at three dairy farms (in Ladenburg, Weinheim and Kleve) were characterised. The dairy
25 farm in Weinheim holds 320 to 340 dairy cows and the one in Ladenburg holds 80 dairy cows. Haus Riswick in Kleve is an education and research centre of the Agricultural Chamber of North-Rhine Westphalia with 230 dairy cows in conventional livestock farming, 45 dairy cows in organic livestock farming and more than 200 sheep and calves each. In the largest dairy cowshed in Kleve (conventional dairy cowshed) feeding experiments and emission measurements have been carried out (Schiefler, 2013; Schmithausen et al., 2016).

30 All three dairy farms have an associated biogas plant. This is not representative for Germany because most dairy farms do not have such a facility. In 2013 there were 285 000 agricultural holdings in Germany, 45.8 % of them were cattle farms including dairy cow farms. But only 2.2 % (6300) of all agricultural holdings had a biogas plant and thus much less than 5 % of all cattle farms (including dairy cow farms) could have a biogas plant (Agrarstrukturerhebung, 2013).

Levin et al. (1993) showed that the isotopic signature of CH₄ produced by cows strongly depends on the diet. Cows with a 100 % C3 ($-65.1 \pm 1.7\text{‰}$) diet emit less enriched CH₄ than cows with a 60 to 80 % C4 diet ($-55.6 \pm 1.4\text{‰}$). In addition, CH₄ emitted by liquid manure has a more depleted isotope ratio of $-73.9 \pm 0.7\text{‰}$.

The dairy cows in Weinheim are full-time in the cowsheds and were fed nearly identically throughout the year with 36 % C4 plants (maize) and 64 % C3 plants. Therefore, no strong variations in the determined $\delta^{13}\text{CH}_4$ signature of CH₄ would be expected. However, the values vary between -40 to -66‰ . A more detailed inspection of the origin of the peaks showed a possible influence of the biogas plants placed on the farms. In Ladenburg and Weinheim most wind conditions made it impossible to separate between CH₄ produced from the cows and from the biogas plant. To determine the CH₄ emissions from the dairy cows and the cowshed only AirCore measurements with distinct wind directions were used. These measurements were carried out directly next to the cowshed on the farm, where an influence of the biogas plant could be excluded.

In Weinheim only 3 out of 15 plume measurements were used (Sep 16 to Feb 17), because during all other samplings an influence of the biogas plant cannot be excluded. These AirCore measurements were taken when driving directly over the farm. Therefore, the ~~peak heights~~ maximum CH₄ concentration measured were relatively high with 8.3 ~~and to~~ 8.9 ppm. The $\delta^{13}\text{CH}_4$ ~~values~~ signatures varied between -62.6 and -66.0‰ with an average of $-64.9 \pm 1.6\text{‰}$. For the ten other AirCore measurements CH₄ emitted from cowshed and biogas plant cannot be separated. The resulting mean isotopic signature is $-54.0 \pm 8.0\text{‰}$ spanning a range between -43.1 and -62.6‰ . The plumes measured downwind of the dairy farm had ~~peak heights~~ CH₄ concentrations between 2.6 and 9 ppm with an average of 4.3 ppm.

Next to the dairy farm in Ladenburg the plumes measured over 6 days between October 2016 and February 2017 had most of the time very small ~~peak heights~~ maximum CH₄ concentrations of 2.1 to 2.8 ppm (on average 2.4 ppm). As expected, ~~due to lower animal number~~, the plumes were smaller than the ones measured near the dairy farm in Weinheim although the measurements were carried out closer to the source. Only on one day in November 2016 a CH₄ concentration of up to 8 ppm was measured in the plume. The $\delta^{13}\text{CH}_4$ signatures determined out of three AirCore measurements taken when driving on the road next to the farm have values around $-44.4 \pm 0.8\text{‰}$. For these measurements it was not possible to separate between CH₄ emitted by the cows and by the biogas plant. To determine the isotopic signature of CH₄ from the dairy cows and the cowshed alone, three AirCore measurements of the plume directly on the farm next to the cowshed were taken in October 2016, which had concentrations up to 4.1 to 7.3 ppm. The determined $\delta^{13}\text{CH}_4$ ~~values~~ source signatures varied between -61.6 and -64.0‰ with an average of $-63.2 \pm 1.4\text{‰}$.

In Weinheim as well as in Ladenburg the $\delta^{13}\text{CH}_4$ signature of the whole farm (cowshed and biogas plant) is less depleted than the isotopic signature of the cowshed alone. Further experiments are needed to determine the ~~seasonal~~ isotopic signature of the biogas plants on dairy farms and the influence on the plume of the farm in total.

On 24 March 2017 five AirCore plume measurements were taken on the dairy farm in Kleve with maximal CH₄ concentrations between 4.7 and 13.6 ppm. The determined $\delta^{13}\text{CH}_4$ signatures vary between -61.7 and -65.1‰ and the average is $-63.5 \pm 1.6\text{‰}$. The weather conditions made it possible to exclude an influence from the biogas plant. Two measurements were taken directly next to both the large cowshed with dairy cows of conventional farming and next to the cowsheds of organic keeping. The average isotopic signatures of CH₄ emitted by the cowsheds of conventional and organic livestock farming do

not differ significantly. For conventional livestock the determined $\delta^{13}\text{CH}_4$ signature is $-64.3 \pm 1.5\text{‰}$ and for organic livestock $-64.4 \pm 0.9\text{‰}$. The fifth AirCore measurement was done on the downwind side of the farm ($-61.7 \pm 1.7\text{‰}$).

The average $\delta^{13}\text{CH}_4$ ~~signature~~ signatures of all three dairy farms match each other and the isotopic signature expected from the results from Levin et al. (1993). It is important to note that the measured CH_4 from the plume of cowsheds is a mixture of CH_4 emitted by cows and manure.

3.3.3 Landfill

~~Bergamaschi et al. (1998)~~ In addition to the source material and fractionation during the production the isotopic signature of gas emitted by landfills depends also on fractionation processes in the upper soil layers of the landfill. Due to the presence of oxygen in the upper soil layers, aerobic bacteria oxidate parts of CH_4 which diffuses through the soil cover and shift the isotopic composition to more depleted values. Bergamaschi et al. (1998) measured these different isotopic signatures. They determined $\delta^{13}\text{CH}_4$ signatures of different sample types from four German and Dutch landfills. For direct gas samples from the gas collecting system they measured an isotopic signature of $-59.0 \pm 2.2\text{‰}$. Emission samples taken with static chambers at covered areas of the landfill showed, however, more enriched isotopic signatures of $-45.9 \pm 8.0\text{‰}$. ~~Due to the presence of oxygen in the upper soil layers, aerobic bacteria oxidate parts of CH_4 which diffuses through the soil cover and shift the isotopic signature to higher values.~~ Upwind-downwind measurements of CH_4 around the landfill lead to an isotopic signature of $-55.4 \pm 1.4\text{‰}$.

In this study, the isotopic signature of CH_4 emitted from a landfill with a disposal area of approximately 1.45km^2 which is located near Sinsheim, south-east of Heidelberg, was characterised. From 1978 to 1998 biodegradable domestic waste was deposited there. A degassing system collects the produced biogas which is used to generate electricity (AVR, 2016). The landfill is covered in large parts by a final surface sealing and during the measuring period construction works were done to cover further parts.

Over 10 days from July to November 2016 and in March and July 2017, 26 plume measurements were performed. During this period the CH_4 plume was measured twice on the landfill and the other times while driving on a public road next to it. The measured CH_4 concentrations of the plumes downwind of the landfill were relatively small with 2.1 to 2.7 ppm. Therefore, the $\delta^{13}\text{CH}_4$ signature cannot be determined to a high accuracy. From 18 measured plumes only four can be used to determine the isotopic source signature precisely. The resulting values vary between -54.2 and -62.2‰ . No seasonal variations have been observed. The average ~~daily mean~~ is $-58.7 \pm 3.3\text{‰}$. This result is comparable to the upwind-downwind measurements of CH_4 by Bergamaschi et al. (1998) and to the study of Zazzeri et al. (2015) in the UK with values between $-55.2 \pm 0.6\text{‰}$ and $-60.2 \pm 1.4\text{‰}$, with an average of $-58.0 \pm 3.0(2\text{SD})\text{‰}$.

In July 2016 the CH_4 concentration was measured directly on the landfill. The maximum measured concentration was, with values up to 6 ppm, higher than the ones measured downwind of the landfill. The average $\delta^{13}\text{CH}_4$ signature is $-66.5 \pm 2.5\text{‰}$ (-64.0 to -69.3‰). Nearly one year later measurements were carried out on the landfill again. The average $\delta^{13}\text{CH}_4$ signature out of two AirCore measurements is, with $-59.5 \pm 0.5\text{‰}$ (-59.9 and -59.1‰), much more enriched and in good agreement

with the measurements next to the landfill. Again the maximum CH₄ peaks-concentrations of the plumes were, with values between 2.6 to 7.2 ppm, higher than the measurements downwind of the landfill.

Direct gas samples from the gas collecting system taken on the same day in July 2017 have an average isotopic signature of CH₄ of $-59.5 \pm 0.1\text{‰}$. This value matches the isotope ratio of $-59.0 \pm 2.2\text{‰}$ reported by Bergamaschi et al. (1998) for direct samples from the gas collecting system. Like Bergamaschi et al. (1998) the isotopic signature of CH₄ in the gas collecting system is less enriched than the isotope ratio measured in the plume next to the landfill. The isotopic signature of CH₄ determined out of the plume on the landfill in July 2017 is the same as for the direct gas sample. The large CH₄ peaks measured on the landfill seem to originate from the gas collecting system.

As previously mentioned, less enriched $\delta^{13}\text{CH}_4$ values-signatures of -66‰ were determined out of measurements carried out on the landfill in July 2016. Bergamaschi et al. (1998) measured such depleted $\delta^{13}\text{CH}_4$ signatures of approximately -69‰ , too, once for a gas sample from the gas collecting system and in one depth profile measurement. Our measurement may have been influenced by constructions work which were done on the landfill during the whole measurement period.

3.3.4 Wastewater treatment plant

Every year approximately 23 million m³ of wastewater is cleaned in the wastewater treatment plant (WWTP) in Heidelberg. During our field campaigns mobile measurements were carried out next to the southern part. There, the sludge treatment inside the digestion towers takes place in three septic tanks with a volume of 2500 m³ each, under anaerobic mesophilic conditions, that means without oxygen at 37 °C. The produced sewage gas consists predominantly of CH₄ and is collected to be utilised in a block heating station (Abwasserzweckverband Heidelberg, 2017).

In February 2017 two gas samples of the collected gas were taken from the WWTP and were analysed in the laboratory. The average $\delta^{13}\text{CH}_4$ signature of the gas produced in the WWTP is $-51.3 \pm 0.2\text{‰}$. 13 plume measurements next to the WWTP were taken over 5 days in October 2016 to February 2017. The maximum CH₄ peak-heights-concentrations of the plumes varied between 2.4 and 8.5 ppm. The isotopic signature for the seven used plume measurements are within the range of -49.4 to -56.3‰ with an average daily-mean of $-52.5 \pm 1.4\text{‰}$. This agrees well with the results of Zazzeri (2016) who reported isotopic signatures of CH₄ between -48.1 to -59.2‰ for wastewater treatment emissions.

3.3.5 Natural gas facilities

Besides the direct sampling of natural gas in Heidelberg (see Sect. 3.1) the plumes at two natural gas facilities were measured to determine the isotopic signature of CH₄ from natural gas in the region of Heidelberg. Between July 2016 and March 2017 the CH₄ concentration around the natural gas storage site in Sandhausen was measured over 10 days. Except for 2 days, the CH₄ concentration of the plumes was lower than 2.15 ppm and four times no significant changes of CH₄ could be measured at all. On these 2 days the maximal CH₄ concentration of the plume was higher with values between 2.3 and 10 ppm, so that the isotopic source signature could be determined with the Miller-Tans approach. The resulting $\delta^{13}\text{CH}_4$ signature was on average $-45.5 \pm 5.2\text{‰}$ -and $-41.8 \pm 0.4\text{‰}$ (two AirCores) on one day and $-49.2 \pm 4.6\text{‰}$ (one AirCore) on the other day. The natural

gas storage in Sandhausen emitted probably only small amounts of CH₄ except during some events making it difficult to monitorsample.

Between Hähnlein and Gernsheim a natural gas storage, compressor stations and other natural gas facilities were placed together on one site. Over 5 days between September 2016 and February 2017 mobile measurements were carried out next to this site and showed that natural gas escaped at different locations. Contrary to the natural gas storage in Sandhausen, the measured CH₄ plumes had ~~always~~ maximum CH₄ concentrations ~~mostly~~ between 2.2 ppm and 6 ppm, but some plumes even reached 6 to 25 ppm. ~~Therefore, emissions from~~ The emissions from these natural gas facilities are not negligible and seem to be highly heterogeneous. The determined $\delta^{13}\text{CH}_4$ signature of the CH₄ plumes was between -41.1 and -57.4‰ . The average ~~daily mean~~ was $-46.6 \pm 6.8\text{‰}$ and thus a little bit less enriched than the isotopic signatures of CH₄ measured in Sandhausen and than the natural gas samples taken in Heidelberg (-43.1 $-43.3 \pm 0.8\text{‰}$). The location of this natural gas facility can be the explanation for more depleted values, because the gas pipeline MEGAL passes this site directly and has a compressor station there. MEGAL runs from the border of the Czech Republic to France and mainly transports Russian natural gas, which has a more depleted isotopic signature.

3.3.6 Coal mines

On 25 March 2017 the emitted CH₄ concentration from bituminous deep coal mines in Bottrop were measured. In particular, the plume of one closed mine shaft and two ones that are still in service were measured. In the plume of the closed mine shaft the maximum CH₄ concentration measured was between 2.2 and 2.6 ppm while for the ~~open mine shafts~~ mine shafts in service concentrations between 3 and 7.5 ppm were detected, although the mobile measurements were carried out much closer to the closed mine shaft. It seems that the CH₄ emissions from ~~open mine shafts~~ mine shafts in service are larger than from closed ones. $\delta^{13}\text{CH}_4$ of the closed mine shaft is $-50.0 \pm 6.3\text{‰}$ while for the active mine shafts the average $\delta^{13}\text{CH}_4$ signature is $-56.0 \pm 2.3\text{‰}$ (-54.7 to -59.5‰). However, only one AirCore was measured for the closed mine shaft and the error of the isotopic source signature is larger than our criterion of 5‰ . The determined isotope ratios of CH₄ in Bottrop ~~match the are less depleted than the values of the~~ coal bed gas samples from Bottrop (-47.1 to -52.4‰) measured by Thielemann et al. (2004). ~~In addition~~ However, the values are similar to the average isotope ratio of CH₄ of -55‰ measured for CH₄ from bituminous coal in deep mines by Zazzeri et al. (2016).

On 23 March 2017 mobile measurements of CH₄ were carried out in the area around the lignite opencast mines Hambach and Garzweiler. However, no CH₄ emitted by the opencast mines directly could be detected. ~~On the roads where the measurements were performed, the emission plume of~~ Around the opencast mine Hambach, the CH₄ ~~from these mines is apparently below the detection limit of the mobile system. concentration varied only slightly between 1.94 and 1.98 ppm when we measured~~ upwind as well as downwind of the pit. Therefore, it was not possible to identify an emission peak. High CH₄ concentrations were only measured at two locations next to the opencast pit. However, the two detected CH₄ plumes were measured upwind of the opencast mine and thus did not originate from the pit itself, but from the drainage system. The ~~measured peak heights of the plumes~~ maximum CH₄ concentrations measured were between 3 and 7.5 ppm. The $\delta^{13}\text{CH}_4$ e-signature of the measured CH₄ is between -79.7 and -84.8‰ with an average of $-82.0 \pm 2.6\text{‰}$. These extremely depleted values indicate that the

measured CH_4 is of microbial origin and thus is probably produced by CO_2 -reduction similarly to one gas sample measured by Thielemann et al. (2004) with values of -85.1 to -85.9‰ .

4 Conclusions

We have developed and tested a mobile instrument setup to determine the $\delta^{13}\text{CH}_4$ signature by measuring the plume of different CH_4 sources. The advantage of such a mobile application is that measurements can be performed downwind of the emission source and therefore outside of any industrial installation such as a gas compressor station or landfill without the consent of the owners. For accurate results, a carefully characterisation of the each individual analyser, especially the cross sensitivities of C_2H_6 , and the drying of air previous to the measurement is required. To reduce the H_2O concentration below 0.1% a nafion dryer was installed in the mobile setup and the cross sensitivity between C_2H_6 and the measurement of $\delta^{13}\text{CH}_4$ was corrected as shown in Fig. 3. Especially for natural gas samples, the precise determination and correction of C_2H_6 is important as in our study C_2H_6 can bias $\delta^{13}\text{CH}_4$ up to 3‰ depending on the CH_4 to C_2H_6 ratio of the sample and the calibration cylinder.

For the precise determination of the isotopic signature of different CH_4 sources we ~~suggest to~~ use the Miller-Tans approach using together with a York fit for most accurate results. There are three major limitations to the precise determination of the $\delta^{13}\text{CH}_4$ source signature: the number of data points during plume crossing, the measured concentration enhancement and the precision of the analyser for isotope analysis. The amount of data points limits the accuracy as the uncertainty decreases with increasing number of data points. To enlarge the amount of points the measurement should be carried out while driving as slowly as possible through the plume and then the plume should be remeasured using the AirCore. It is important to use the AirCore because it is a simple option to reduce the uncertainty by more than half. The most important limitation of the $\delta^{13}\text{CH}_4$ source signature is the plume concentration above background. Measured plumes with a peak height above background smaller than 0.45 ppm have uncertainties larger than 5‰ and thus are not used in this study. Driving as close as possible to the source increases the CH_4 concentration. However, where it is not possible, or the increase is not enough, the isotopic signature of the source cannot be determined with a sufficient precision with this method. To get better results even for smaller enhancements, more precise instruments are required in the future.

In this study, the $\delta^{13}\text{CH}_4$ signature of CH_4 emitted from a biogas plant, a landfill, dairy farms, a wastewater treatment plant, natural gas storage and compressor stations and bituminous deep mines were determined. The $\delta^{13}\text{CH}_4$ signatures measured during mobile campaigns are in good agreement with the measured isotope ratios from direct samples taken at some of the CH_4 sources and with values from other studies. Thus this method provides an opportunity to characterise the CH_4 emissions from a source where it is not possible or difficult to take direct samples; for example from an industrial site without the authorisation of the operating company, or from a large area where CH_4 emits heterogeneously at multiple unknown positions. Gas samples from Heidelberg city gas supply from December 2016 to ~~June~~ November 2018 confirm a change in the natural gas mixture, especially of Russian and North Sea gas. While in former years (1991 to 1996) strong seasonal variations of $\delta^{13}\text{CH}_4$ were measured, whereas recently the isotopic signature is nearly constant during the year. In addition, the average is approximately ~~2.83~~ ‰ more depleted than in the 1990s.

Data availability. Data collected are presently available upon request.

Competing interests. The authors declare that they have no conflict of interest.

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References

- Abwasserzweckverband Heidelberg, http://www.azv-heidelberg.de/2_02_01_kwsued.htm, last access: 8 March 2018.
- Agrarstrukturhebung 2013 <https://www.destatis.de/DE/Publikationen/Thematisch/LandForstwirtschaft/Betriebe/BetriebeAnlagen.html> and <https://www.destatis.de/DE/Publikationen/Thematisch/LandForstwirtschaft/ViehbestandTierischeErzeugung/Viehhaltung.html>
- 5 Assan, S., Baudic, A., Guemri, A., Ciais, P., Gros, V., and Vogel, F. R.: Characterization of interferences to in situ observations of $\delta^{13}\text{CH}_4$ and C_2H_6 when using a cavity ring-down spectrometer at industrial sites, *Atmos. Meas. Tech.*, 10, 15, 10.5194/amt-10-2077-2017, 2017.
- AVR: Deponien Sinsheim und Wiesloch, available at: https://www.avr-kommunal.de/data/mediapool/2016_deponien_sinsheim_und_wiesloch.pdf, last access: 8 March 2018, 2016.
- BAFA, Bundesamt für Wirtschaft und Ausfuhrkontrolle, Entwicklung des deutschen Gasmarktes (monatliche Bilanz 1998 – 2016, Einfuhr seit 1960), available at: http://www.bafa.de/DE/Energie/Rohstoffe/Erdgas/erdgas_node.html, last access: 29 June 2017.
- 10 Bergamaschi, P., Lubina, C., Königstedt, R., Fischer, H., Veltkamp, A. C., and Zwaagstra, O.: Stable isotopic signatures ($\delta^{13}\text{C}$, δD) of methane from European landfill sites, *J. Geophys. Res.-Atmos.*, 103, 8251-8265, 10.1029/98jd00105, 1998.
- Cramer, B., Poelchau, H. S., Gerling, P., Lopatin, N. V., and Littke, R.: Methane released from groundwater: the source of natural gas accumulations in northern West Siberia, *Mar. Pet. Geol.*, 16, 225-244, 10.1016/s0264-8172(98)00085-3, 1999.
- 15 Dinger, F.: Characterisation of a Cavity Ring-Down Spectrometer for measuring CO_2 , CH_4 , $\delta^{13}\text{CO}_2$, and $\delta^{13}\text{CH}_4$ in ambient air, M.S. thesis, Heidelberg University, Germany, 2014.
- Dlugokencky, E. J., Myers, R. C., Lang, P. M., Masarie, K. A., Crotwell, A. M., Thoning, K. W., Hall, B. D., Elkins, J. W., and Steele, L. P.: Conversion of NOAA atmospheric dry air CH_4 mole fractions to a gravimetrically prepared standard scale, *J. Geophys. Res.-Atmos.*, 110, 8, 10.1029/2005jd006035, 2005.
- 20 [Dlugokencky, E. J., Bruhwiler, L., White, J. W. C., Emmons, L. K., Novelli, P. C., Montzka, S. A., Masarie, K. A., Lang, P. M., Crotwell, A. M., Miller, J. B., and Gatti, L. V.: Observational constraints on recent increases in the atmospheric \$\text{CH}_4\$ burden, *Geophys. Res. Lett.*, 36, 5, 10.1029/2009gl013978, 2009.](#)
- Fisher, R., Lowry, D., Wilkin, O., Srisankharajah, S., and Nisbet, E. G.: High-precision, automated stable isotope analysis of atmospheric methane and carbon dioxide using continuous-flow isotope-ratio mass spectrometry, *Rapid Commun. Mass Spectrom.*, 20, 200-208, 10.1002/rcm.2300, 2006.
- 25 Glatzel-Mattheier, H.: Bilanzierung von CH_4 -Emissionen in Deutschland anhand atmosphärischer Messungen in Heidelberg, Ph.D. thesis, University of Heidelberg, Germany, 1997.
- Hoheisel, A.: Characterization of a CRDS Analyzer and Determination of Isotopic Signatures of Anthropogenic CH_4 Sources Using Mobile Measurements, M.S. thesis, Heidelberg University, Germany, 2017.
- 30 Karion, A., Sweeney, C., Tans, P., and Newberger, T.: AirCore: An Innovative Atmospheric Sampling System, *J. Atmos. Ocean. Technol.*, 27, 1839-1853, 10.1175/2010jtecha1448.1, 2010.
- Keeling, C. D.: The concentration and isotopic abundances of atmospheric carbon dioxide in rural areas, *Geochim. Cosmochim. Acta*, 13, 322-334, 10.1016/0016-7037(58)90033-4, 1958.
- [Kirschke, S., Bousquet, P., Ciais, P., Saunoy, M., Canadell, J. G., Dlugokencky, E. J., Bergamaschi, P., Bergmann, D., Blake, D. R., Bruhwiler, L., Cameron-Smith, P., Castaldi, S., Chevallier, F., Feng, L., Fraser, A., Heimann, M., Hodson, E. L., Houweling, S., Josse, B., Fraser, P. J., Krummel, P. B., Lamarque, J. F., Langenfelds, R. L., Le Quere, C., Naik, V., O'Doherty, S., Palmer, P. I., Pison, I., Plummer, D., Poulter, B., Prinn, R. G., Rigby, M., Ringeval, B., Santini, M., Schmidt, M., Shindell, D. T., Simpson, I. J., Spahni, R., Steele, L. P., Strode, S. A.,](#)

- ~~Sudo, K., Szopa, S., van der Werf, G. R., Voulgarakis, A., van Weele, M., Weiss, R. F., Williams, J. E., and Zeng, G.: Three decades of global methane sources and sinks, Nat. Geosci., 6, 813-823, 10.1038/ngeo1955, 2013.~~
- Levin, I., Bergamaschi, P., Dörr, H., and Trapp, D.: Stable isotopic signature of methane from major sources in Germany, Chemosphere, 26, 161-177, 10.1016/0045-6535(93)90419-6, 1993.
- 5 Levin, I., Glatzel-Mattheier, H., Marik, T., Cuntz, M., Schmidt, M., and Worthy, D. E.: Verification of German methane emission inventories and their recent changes based on atmospheric observations, J. Geophys. Res.-Atmos., 104, 3447-3456, 10.1029/1998jd100064, 1999.
- Lopez, M., Sherwood, O. A., Dlugokencky, E. J., Kessler, R., Giroux, L., and Worthy, D. E. J.: Isotopic signatures of anthropogenic CH₄ sources in Alberta, Canada, Atmos. Environ., 164, 280-288, 10.1016/j.atmosenv.2017.06.021, 2017.
- Lowry, D., Holmes, C. W., Rata, N. D., O'Brien, P., and Nisbet, E. G.: London methane emissions: Use of diurnal changes in concentration and $\delta^{13}\text{C}$ to identify urban sources and verify inventories, J. Geophys. Res.-Atmos., 106, 7427-7448, 10.1029/2000jd900601, 2001.
- 10 Miller, J. B., and Tans, P. P.: Calculating isotopic fractionation from atmospheric measurements at various scales, Tellus Ser. B-Chem. Phys. Meteorol., 55, 207-214, 10.1034/j.1600-0889.2003.00020.x, 2003.
- Nisbet, E. G., Dlugokencky, E. J., Manning, M. R., Lowry, D., Fisher, R. E., France, J. L., Michel, S. E., Miller, J. B., White, J. W. C., Vaughn, B., Bousquet, P., Pyle, J. A., Warwick, N. J., Cain, M., Brownlow, R., Zazzeri, G., Lanoiselle, M., Manning, A. C., Gloor, E.,
- 15 Worthy, D. E. J., Brunke, E. G., Labuschagne, C., Wolff, E. W., and Ganesan, A. L.: Rising atmospheric methane: 2007-2014 growth and isotopic shift, Glob. Biogeochem. Cycle, 30, 1356-1370, 10.1002/2016gb005406, 2016.
- Nitschke-Kowsky, P., Schenk, J., Schley, P. and Altfeld, K.: Gasbeschaffenheit in Deutschland: Was zum Wobbe-Index gesagt werden muss, https://www.di-verlag.de/media/content/GWI/GWI_06_2012/gwi_06_2012_FB_Altfeld.pdf?xaf26a=100ffbcca198e6f72d93c1dbab99e957, (Last visit: 2017-07-26).
- 20 Phillips, N. G., Ackley, R., Crosson, E. R., Down, A., Hutrya, L. R., Brondfield, M., Karr, J. D., Zhao, K., and Jackson, R. B.: Mapping urban pipeline leaks: Methane leaks across Boston, Environ. Pollut., 173, 1-4, 10.1016/j.envpol.2012.11.003, 2013.
- Polag, D., May, T., Müller, L., König, H., Jacobi, F., Laukenmann, S., and Keppler, F.: Online monitoring of stable carbon isotopes of methane in anaerobic digestion as a new tool for early warning of process instability, Bioresour. Technol., 197, 161-170, 10.1016/j.biortech.2015.08.058, 2015.
- 25 Rice, A. L., Butenhoff, C. L., Teama, D. G., Roger, F. H., Khalil, M. A. K., and Rasmussen, R. A.: Atmospheric methane isotopic record favors fossil sources flat in 1980s and 1990s with recent increase, Proc. Natl. Acad. Sci. U. S. A., 113, 10791-10796, 10.1073/pnas.1522923113, 2016.
- Rigby, M., Montzka, S. A., Prinn, R. G., White, J. W. C., Young, D., O'Doherty, S., Lunt, M. F., Ganesan, A. L., Manning, A. J., Simmonds, P. G., Salameh, P. K., Harth, C. M., Muhle, J., Weiss, R. F., Fraser, P. J., Steele, L. P., Krummel, P. B., McCulloch, A., and Park, S.: Role
- 30 of atmospheric oxidation in recent methane growth, Proc. Natl. Acad. Sci. U. S. A., 114, 5373-5377, 10.1073/pnas.1616426114, 2017.
- Röckmann, T., Eyer, S., van der Veen, C., Poppa, M. E., Tuzson, B., Monteil, G., Houweling, S., Harris, E., Brunner, D., Fischer, H., Zazzeri, G., Lowry, D., Nisbet, E. G., Brand, W. A., Necki, J. M., Emmenegger, L., and Mohn, J.: In situ observations of the isotopic composition of methane at the Cabauw tall tower site, Atmos. Chem. Phys., 16, 10469-10487, 10.5194/acp-16-10469-2016, 2016.
- Rella, C. W., Hoffnagle, J., He, Y., and Tajima, S.: Local- and regional-scale measurements of CH₄, $\delta^{13}\text{CH}_4$, and C₂H₆ in the Uintah Basin using a mobile stable isotope analyzer, Atmos. Meas. Tech., 8, 4539-4559, 10.5194/amt-8-4539-2015, 2015.
- 35 Schaefer, H., Fletcher, S. E. M., Veidt, C., Lassey, K. R., Brailsford, G. W., Bromley, T. M., Dlugokencky, E. J., Michel, S. E., Miller, J. B., Levin, I., Lowe, D. C., Martin, R. J., Vaughn, B. H., and White, J. W. C.: A 21st-century shift from fossil-fuel to biogenic methane emissions indicated by (CH₄)-C-13, Science, 352, 80-84, 10.1126/science.aad2705, 2016.

- Schiefler, I.: Greenhouse gas and ammonia emissions from dairy barns, Ph.D. thesis, Landwirtschaftlichen Fakultät der Rheinischen Friedrich-Wilhelms-Universität Bonn, Germany, 2013.
- Schmithausen, A. J., Trimborn, M., and Buscher, W.: Methodological Comparison between a Novel Automatic Sampling System for Gas Chromatography versus Photoacoustic Spectroscopy for Measuring Greenhouse Gas Emissions under Field Conditions, *Sensors*, 16, 10, 10.3390/s16101638, 2016.
- Sherwood, O. A., Schwietzke, S., Arling, V. A., and Etiope, G.: Global Inventory of Gas Geochemistry Data from Fossil Fuel, Microbial and Burning Sources, version 2017, *Earth Syst. Sci. Data*, 9, 639-656, 10.5194/essd-9-639-2017, 2017.
- Sperlich, P., Uitslag, N. A. M., Richter, J. M., Rothe, M., Geilmann, H., van der Veen, C., Röckmann, T., Blunier, T., and Brand, W. A.: Development and evaluation of a suite of isotope reference gases for methane in air, *Atmos. Meas. Tech.*, 9, 3717-3737, 10.5194/amt-9-3717-2016, 2016.
- Thielemann, T., Cramer, B., and Schippers, A.: Coalbed methane in the Ruhr Basin, Germany: a renewable energy resource?, *Org. Geochem.*, 35, 1537-1549, 10.1016/j.orggeochem.2004.05.004, 2004.
- Turner, A. J., Frankenberg, C., Wennberg, P. O., and Jacob, D. J.: Ambiguity in the causes for decadal trends in atmospheric methane and hydroxyl, *Proc. Natl. Acad. Sci. U. S. A.*, 114, 5367-5372, 10.1073/pnas.1616020114, 2017.
- Wehr, R., and Saleska, S. R.: The long-solved problem of the best-fit straight line: application to isotopic mixing lines, *Biogeosciences*, 14, 17-29, 10.5194/bg-14-17-2017, 2017.
- Welp, L. R., Keeling, R. F., Weiss, R. F., Paplawsky, W., and Heckman, S.: Design and performance of a Nafion dryer for continuous operation at CO₂ and CH₄ air monitoring sites, *Atmos. Meas. Tech.*, 6, 1217-1226, 10.5194/amt-6-1217-2013, 2013.
- Werle, P., Mücke, R., and Slemr, F.: The limits of signal averaging in atmospheric trace-gas monitoring by tunable diode-laser absorption spectroscopy (TDLAS), *Applied Physics B-Photophysics and Laser Chemistry*, 57, 131-139, 10.1007/bf00425997, 1993.
- Yeman, C.: Quantifying emission rates and isotopic signatures of methane sources using a mobile CRDS analyzer, M.S. thesis, Heidelberg University, Germany, 2015.
- York, D., Evensen, N. M., Martinez, M. L., and Delgado, J. D.: Unified equations for the slope, intercept, and standard errors of the best straight line, *Am. J. Phys.*, 72, 367-375, 10.1119/1.1632486, 2004.
- Zazzeri, G.: Methane Emissions in UK: Deciphering Regional Sources with Mobile Measurements and Isotopic Characterisation, Ph.D. thesis, Royal Holloway, University of London, 2016.
- Zazzeri, G., Lowry, D., Fisher, R. E., France, J. L., Lanoisellé, M., and Nisbet, E. G.: Plume mapping and isotopic characterisation of anthropogenic methane sources, *Atmos. Environ.*, 110, 151-162, 10.1016/j.atmosenv.2015.03.029, 2015.
- Zazzeri, G., Lowry, D., Fisher, R. E., France, J. L., Lanoisellé, M., Kelly, B. F. J., Necki, J. M., Iverach, C. P., Ginty, E., Zimnoch, M., Jasek, A., and Nisbet, E. G.: Carbon isotopic signature of coal-derived methane emissions to the atmosphere: from coalification to alteration, *Atmos. Chem. Phys.*, 16, 13669-13680, 10.5194/acp-16-13669-2016, 2016.

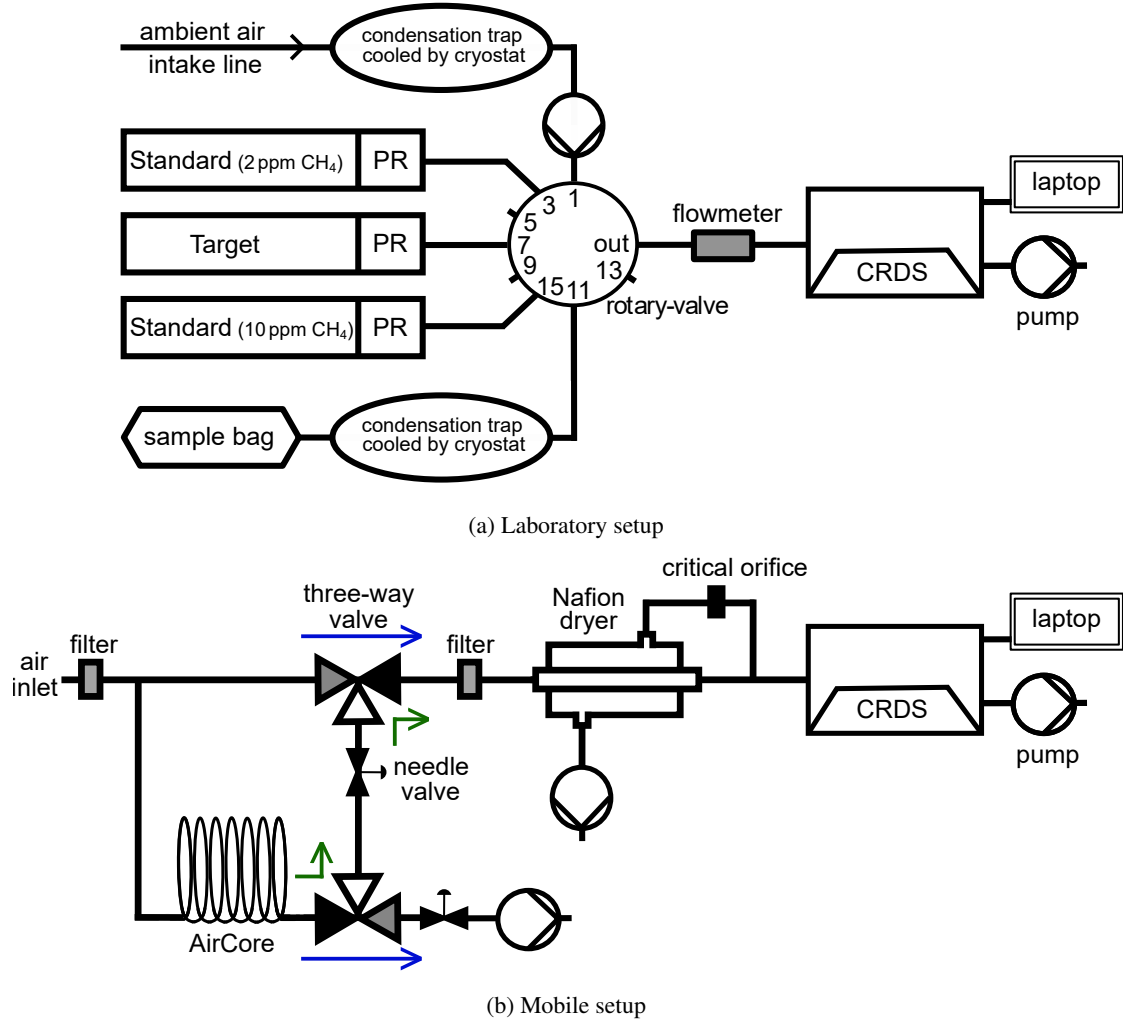


Figure 1. Setup for measurements in the laboratory and with the mobile platform.

Figure (a) shows the measurement setup in the laboratory. Over port 1 ambient air measurements are performed. Port 11 is used to measure sample bags. Standard gas and target cylinders are measured on port 3, 7 and 15 to calibrate the above mentioned measurements and also the mobile ones.

Figure (b) shows the mobile measuring setup installed inside a van. The blue arrows indicate the flow of air in ‘monitoring mode’ and the green ones in ‘replay mode’.

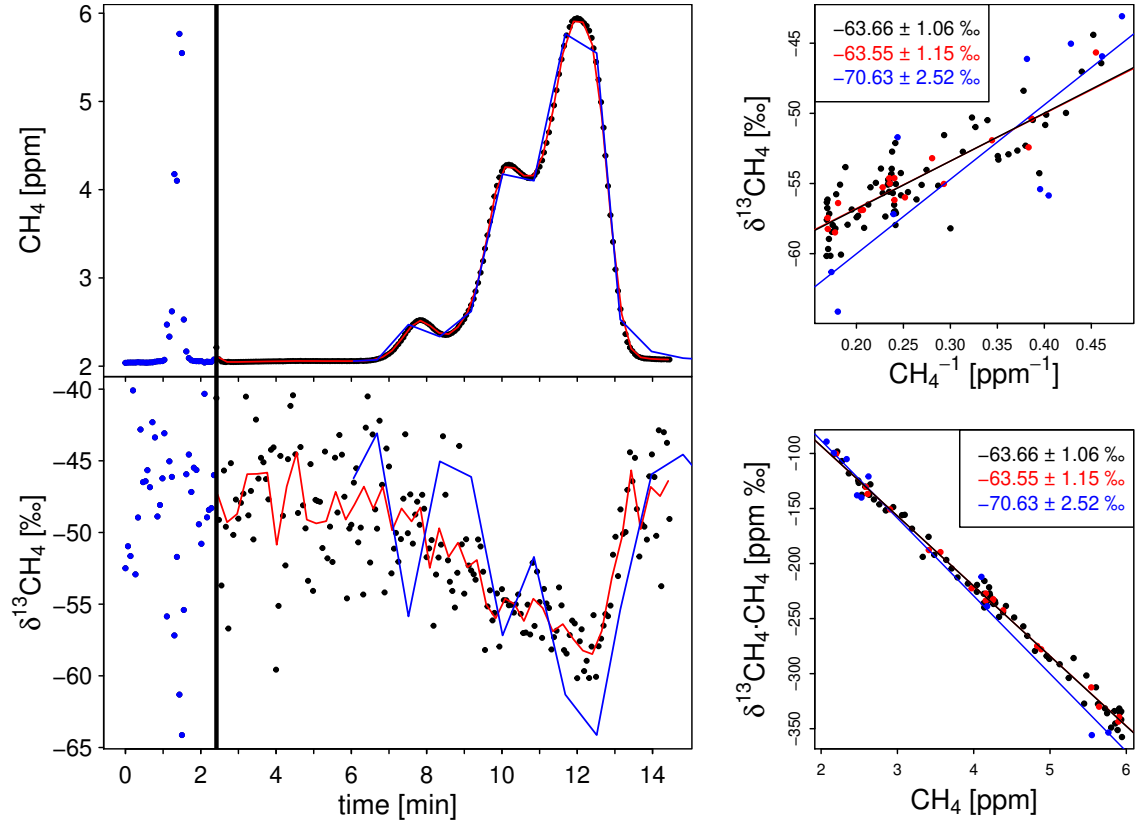


Figure 2. Measurement of a typical plume passing a biogas plant as well as Keeling Plot and Miller-Tans Plot to calculate the $\delta^{13}\text{CH}_4$ signature of the biogas plant.

Left: Typical CH_4 and $\delta^{13}\text{CH}_4$ peaks in the biogas plant plume. The vertical black line shows the switch from ‘monitoring’ to ‘replay mode’. The red data are 15 sec average and the blue line is the in situ peak (first peak) stretched by a factor of 12.5.

Right: Keeling Plot (upper panel) and Miller-Tans Plot (lower panel) to calculate the $\delta^{13}\text{CH}_4$ source signature (insets). The blue colour represents the ‘monitoring mode’, the black and red (15 sec mean) ones the ‘replay mode’. For better visibility the errorbars are not displayed.

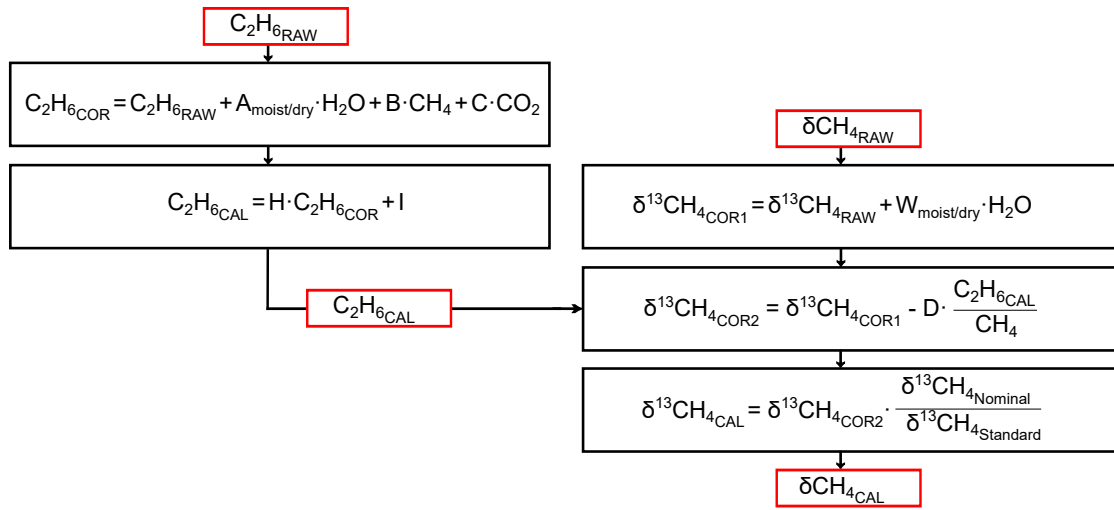


Figure 3. Scheme to correct and calibrate C_2H_6 and $\delta^{13}CH_4$. $\delta^{13}CH_{4Nominal}$ is the nominal (known value of the standard) and $\delta^{13}CH_{4Standard}$ the measured (or interpolated) value of the calibration standard.

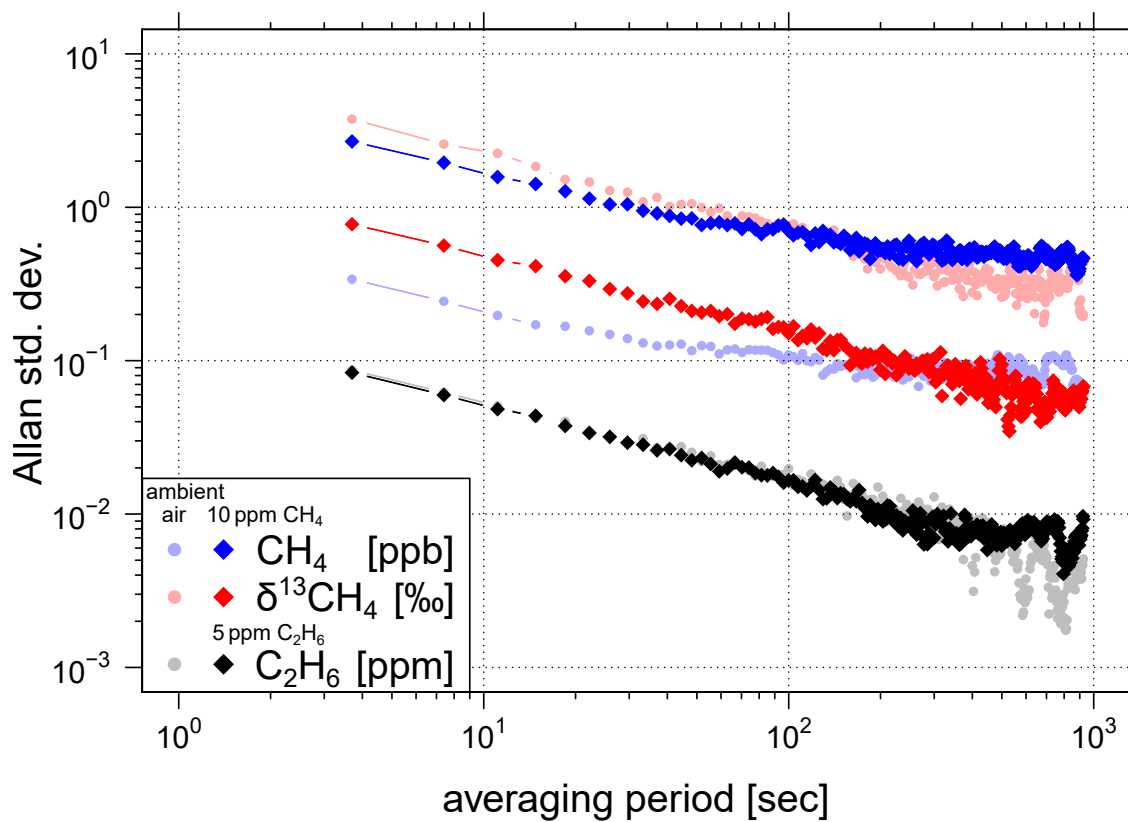


Figure 4. Allan standard deviations of CH₄, δ¹³CH₄ and C₂H₆ for two different sample gases each. Results for the first sample gas at atmospheric concentrations are shown in light (red, blue and grey) colours. Results for the second sample gas with 10 ppm CH₄ are shown in bright red and blue and for the third sample gas with 5 ppm C₂H₆ in black.

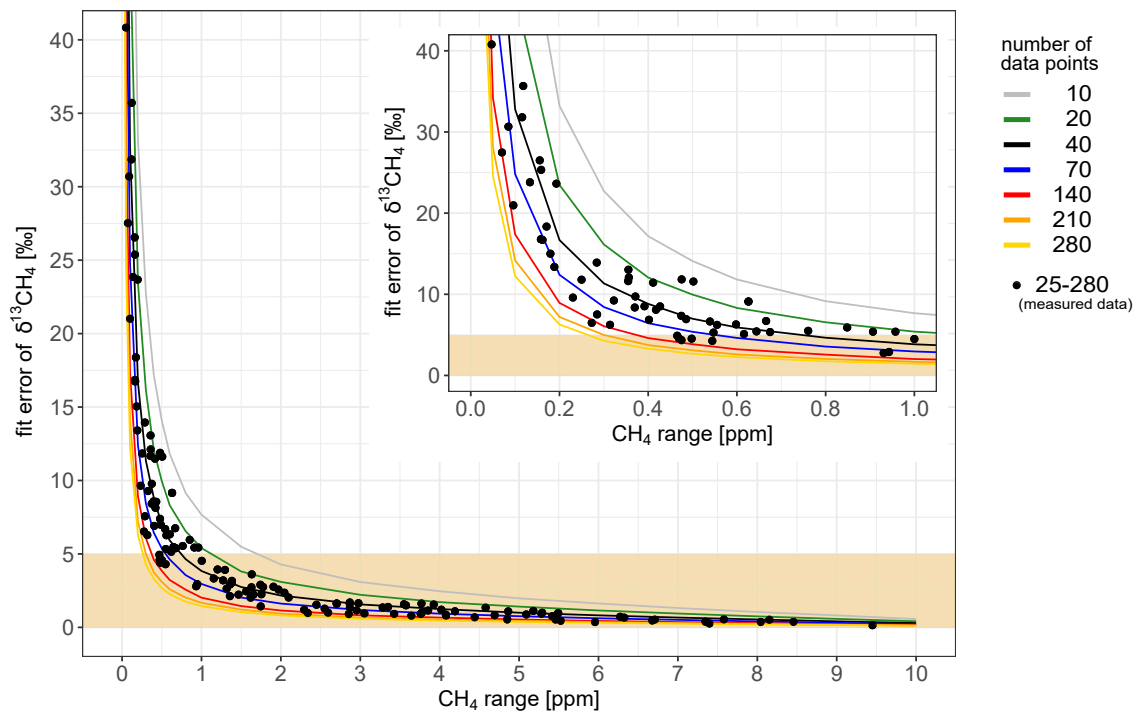


Figure 5. Dependency between peak height above background and error of the $\delta^{13}\text{CH}_4$ signature from the according measured peaks. The inserted figure shows an enlarged section with CH₄ ranges up to 1 ppm. The measured $\delta^{13}\text{CH}_4$ signatures with errors below 5‰ (data points within yellow shaded area) are used in this study. The lines show simulated data with different numbers of data points used in the Miller-Tans plot.

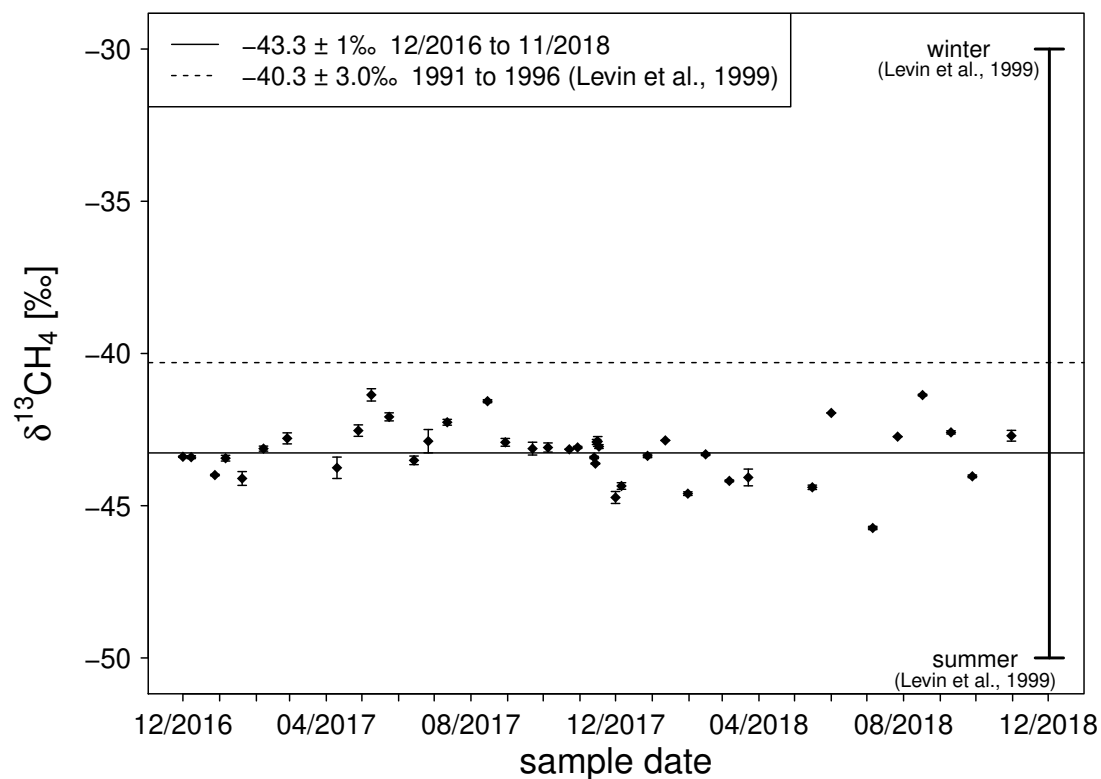


Figure 6. Isotopic signature of natural gas in Heidelberg measured between end of 2016 and ~~June~~November 2018. The horizontal solid line is the average monthly mean $\delta^{13}\text{CH}_4$ value. The horizontal dashed line is the average $\delta^{13}\text{CH}_4$ value measured from 1991 to 1996 with data ranging from -50‰ in summer to -30‰ in winter (Levin et al., 1999).

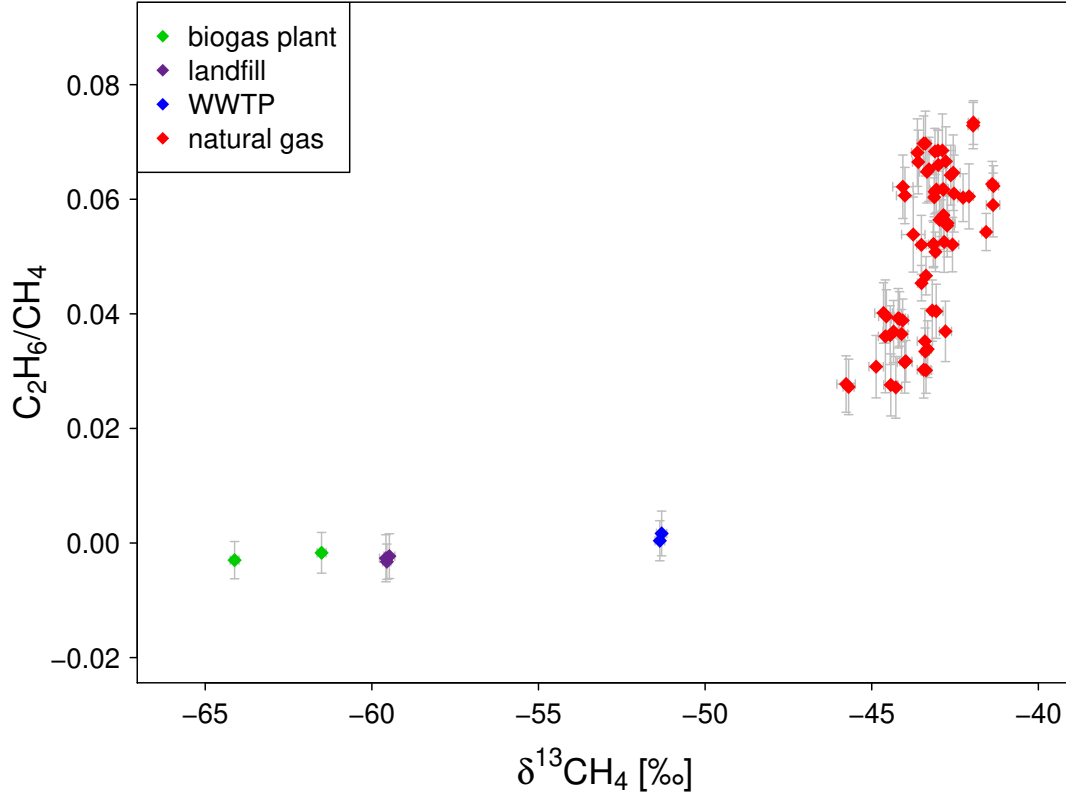
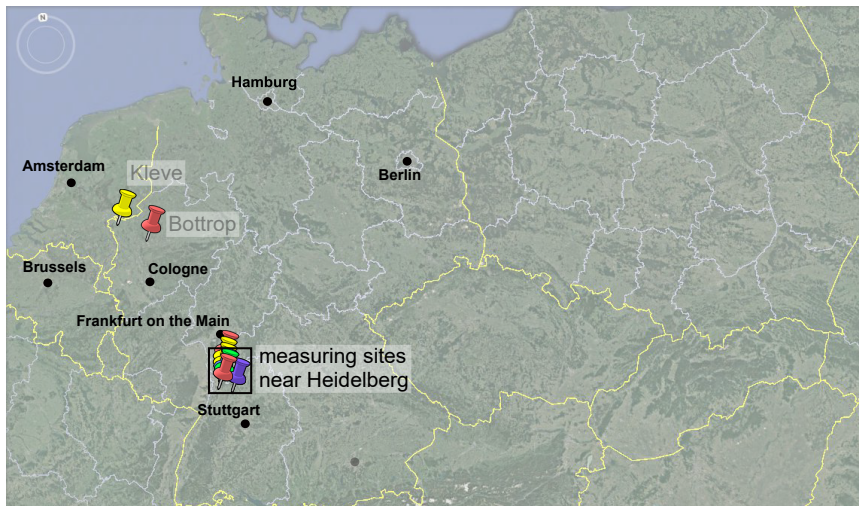


Figure 7. Locations of the measuring sites. $\delta^{13}\text{CH}_4$ isotopic signature of and C_2H_6 to CH_4 sources. Each determined $\delta^{13}\text{CH}_4$ signature is shown as grey dot. The black diamond-shaped points show the averaged daily mean $\delta^{13}\text{CH}_4$ signature with the standard deviation. The $\delta^{13}\text{CH}_4$ values measured from ratio of gas samples taken at from a biogas plant, a landfill, a WWTP and the different sites are plotted as red points. For both natural gas facilities it was not possible to take direct samples. Here the red points indicate the mean $\delta^{13}\text{CH}_4$ signature of natural gas distribution system in Heidelberg measured between end of 2016 and June 2018 as described in this study November 2018.



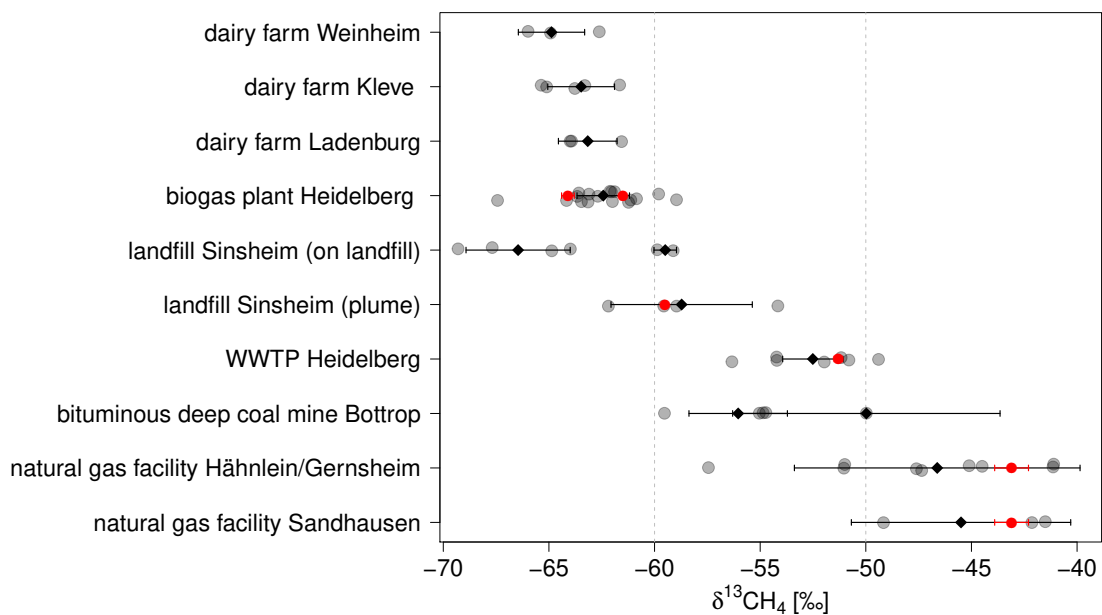


Figure 9. $\delta^{13}\text{CH}_4$ signature of CH_4 sources. Each determined $\delta^{13}\text{CH}_4$ signature is shown as grey dot. The black diamond shaped points show the average $\delta^{13}\text{CH}_4$ signature with the standard deviation. The $\delta^{13}\text{CH}_4$ signatures measured from gas samples taken at the different sites are plotted as red points. For both natural gas facilities it was not possible to take direct samples. Here the red points indicate the mean $\delta^{13}\text{CH}_4$ signature of natural gas in Heidelberg measured between end of 2016 and November 2018 as described in this study.

Table 1. Correction and calibration factors for C₂H₆ and $\delta^{13}\text{CH}_4$.

influence of		correction/ calibration factor	unit	method	tested range
H ₂ O on $\delta^{13}\text{CH}_4$	W _{moist}	-0.54 ± 0.29	$(\text{‰ } \delta^{13}\text{CH}_4) (\% \text{ H}_2\text{O})^{-1}$	humidity tests	0.16 to 1.5 % H ₂ O
	W _{dry}	—			up to 0.15 % H ₂ O
H ₂ O on C ₂ H ₆	A _{moist}	0.70 ± 0.10	$(\text{ppm C}_2\text{H}_6) (\% \text{ H}_2\text{O})^{-1}$	humidity tests	0.16 to 1.5 % H ₂ O
	A _{dry}	0.43 ± 0.51	$(\text{ppm C}_2\text{H}_6) (\% \text{ H}_2\text{O})^{-1}$		up to 0.15 % H ₂ O
CH ₄ on C ₂ H ₆	B	0.0077 ± 0.0007	$(\text{ppm C}_2\text{H}_6) (\text{ppm CH}_4)^{-1}$	dilution & injection tests	2 to 10ppm CH ₄ 0 to 1 ppm C ₂ H ₆
CO ₂ on C ₂ H ₆	C	$(1.25 \pm 0.94) \cdot 10^{-4}$	$(\text{ppm C}_2\text{H}_6) (\text{ppm CO}_2)^{-1}$	dilution & injection tests	10 to 600ppm CO ₂ 0 to 1.3 ppm C ₂ H ₆
C ₂ H ₆ calibration	H	0.538 ± 0.002	ppm	dilution tests	0 to 3 ppm C ₂ H ₆
	I	0.070 ± 0.005			
C ₂ H ₆ on $\delta^{13}\text{CH}_4$	D	40.87 ± 0.49	$(\text{‰ } \delta^{13}\text{CH}_4) \left(\frac{\text{ppm C}_2\text{H}_6}{\text{ppm CH}_4} \right)^{-1}$	dilution tests	up to 0.7 (ppm C ₂ H ₆)(ppm CH ₄) ⁻¹

Table 2. Determined $\delta^{13}\text{CH}_4$ signatures of CH_4 sources.

location	$\delta^{13}\text{CH}_4$ signature from mobile measurements		$\delta^{13}\text{CH}_4$ signature of direct gas samples	peak height* <u>above baseline</u>	number of AirCores**	number of visits**	mobile measuring period/dates
	average [‰]	range [‰]	[‰]	[ppm]			[MM,YY]
biogas plant							
Heidelberg	-62.4 ± 1.2	-67.4 to -59.0	-61.5 ± 0.1 -64.1 ± 0.3	3.4 to 14.1 <u>1.3 to 9.4</u>	17 (25)	7 (10)	Aug,16 to Mar,17
dairy farm							
Weinheim (on farm)	-64.9 ± 1.6	-66.0 to -62.6		8.3 to 8.9 <u>5.3 to 6.7</u>	3 (3)	2 (2)	Oct,16 and Nov,16
Weinheim (plume with biogas plant)	-54.0 ± 8.0	-62.6 to -43.1		3.9 to 13.1 <u>1.6 to 10.9</u>	10 (12)	5 (5)	Sep,16 to Feb,17
Ladenburg (on farm)	-63.2 ± 1.4	-64.0 to -61.6		4.1 to 7.3 <u>1.6 to 4.7</u>	3 (3)	1 (1)	Oct,16
Ladenburg (plume with biogas plant)	-44.4 ± 7.2	-55.1 to -40.3		3.9 to 8.2 <u>1.7 to 6.0</u>	3 (8)	1 (3)	Nov,16 to Feb,17
Kleve	-63.5 ± 1.6	-65.1 to -61.7		4.7 to 13.6 <u>2.7 to 11.2</u>	5 (5)	1 (1)	Mar,17
landfill							
Sinsheim (plume)	-58.7 ± 3.3	-62.2 to -54.2	-59.5 ± 0.1	2.4 to 2.6 <u>0.5 to 0.6</u>	4 (18)	4 (8)	Jul,16 to Mar,17
Sinsheim (on landfill)	-59.5 ± 0.5	-59.9 to -59.1		3.9 to 7.2 <u>1.8 to 4.9</u>	2 (4)	1 (1)	Jul,17
	-66.5 ± 2.5	-69.3 to -64.0		2.6 to 6.0 <u>0.5 to 3.4</u>	4 (4)	1 (1)	Jul,16
WWTP							
Heidelberg	-52.5 ± 1.4	-56.3 to -49.4	-51.3 ± 0.2	3.5 to 6.0 <u>1.3 to 3.8</u>	7 (13)	5 (5)	Oct,16 to Feb,17
natural gas facilities							
Sandhausen	-45.5 ± 5.2	-49.2 to -41.5		3.0 and 10.0 <u>1.0 and 7.6</u>	3 (9)	2 (10)	Jul,16 and Mar,17
Hähnlein/Gernsheim	-46.6 ± 6.8	-57.4 to -41.1		3.3 to 8.2 <u>1.2 to 5.5</u>	9 (21)	5 (5)	Sep,16 to Feb,17
bituminous deep coal mine							
Bottrop (<u>active in service</u>)	-56.0 ± 2.3	-59.5 to -54.7		3.4 to 7.6 <u>1.4 to 5.5</u>	4 (4)	1 (1)	Mar,17
Bottrop (closed)	-50.0 ± 6.3	-50.0		2.6 <u>0.6</u>	1 (1)	1 (1)	Mar,17 5pt

* The range of peak heights above baseline of the applied peaks measured with the AirCore.

** Instead of the used AirCore measurements and the corresponding visits, the number in brackets refer to all AirCore measurements and visits.