Profiling of CH4 background mixing ratio in the lower troposphere with Raman lidar: a feasibility experiment.

Igor Veselovskii¹, Philippe Goloub², Qiaoyun Hu², Thierry Podvin², David .N. Whiteman³, Mikhael Korenskiy¹, Eduardo Landulfo⁴

¹Physics Instrumentation Center of General Physics Institute, Troitsk, Moscow, Russia.
 ²Laboratoire d'Optique Atmosphérie, Université de Lille-CNRS, Villeneuve d'Ascq, France
 ³NASA Goddard Space Flight Center, Greenbelt, USA
 ⁴Instituto de Pesquisas Energeticas e Nucleares, Sao Paulo, Brazil

10 Abstract

We present the results of methane profiling in the lower troposphere using LILAS Raman lidar from Lille University observatory platform (France). The lidar is based on a tripled Nd:YAG laser and nighttime profiling up to 4000 m with 100 m height resolution is possible for methane. Agreement between measured the photon counting rate in the CH₄ Raman channel in 15 the free troposphere and numerical simulations for a typical CH₄ background mixing ratio (2 ppm) confirms that CH₄ Raman scattering is detected. The mixing ratio is calculated from the ratio of methane (395.7 nm) and nitrogen (386.7 nm) Raman backscatters, and within the planetary boundary layer, an increase of the CH₄ mixing ratio, up to a factor of 2, is observed. Different possible interfering factors, such as leakage of the elastic signal and aerosol 20 fluorescence have been taken into consideration. Tests using backscattering from clouds confirmed that the filters in the Raman channel provide sufficient rejection of elastic scattering. The measured methane profiles do not correlate with aerosol backscattering, which corroborates the hypothesis that, in the PBL, not aerosol fluorescence but CH₄ is observed. However, the

25 to install an additional control channel close to 393 nm where no strong Raman lines exist and only fluorescence can be observed.

fluorescence contribution cannot be completely excluded and, for future measurements, we plan

1. Introduction.

Raman spectroscopy is a powerful technique for identification of different gases in the atmosphere and for the estimation of their concentration (Weber, 1979), which can be used in

conjunction with lidar technology (Inaba and Kobayasi, 1972). An example of such synergy is the Raman lidar for water vapor monitoring (Whiteman et al., 1992). For optimum application of the Raman technique, the gas of interest should be abundant in the atmosphere, possess a large scattering cross section and have a Raman spectrum that is isolated from potential interfering

- 35 species. Detection of water vapor with Raman spectroscopy satisfies all of these conditions and has become a very popular application of lidar (e.g. Whiteman et al, 2007 and references therein). Besides water vapor, Raman lidar profiling of carbon dioxide (Ansmann et al., 1992; Whiteman et al., 2007; Zhao et al., 2008) as well as quartz crystals in dust layers (Müller et al., 2010) has been reported.
- 40 Methane is currently the second (after carbon dioxide) most important greenhouse gas of anthropogenic origin (IPCC, 2013). Methane is emitted from a variety of natural and anthropogenic sources (e.g. Baray et al., 2018; Kavitha et al., 2016 and references therein) and on a per-molecule basis, methane is about 30 times more effective a greenhouse gas than carbon dioxide (Etminan et al., 2016). Global information about the CH₄ column concentration is available from satellite observations with, for example, the SCIAMACHY sensor on board ENVISAT (Bovensmann et al., 1999) or the TANSO-FTS sensor on board GOSAT (Kuze et al., 2009). Passive observations, however, can be influenced by the presence of aerosol layers or thin ice clouds. The lack of precise global measurements of atmospheric methane initiated, in particular, the upcoming MERLIN mission, offering integrated path differential absorption space
 50 borne lidar (https://earth.esa.int/web/eoportal/satellite-missions/m/merlin), which should provide unprecedented accuracy of CH₄ column values. Still, the integrated path technique doesn't
- Today, it is well established that, in the free troposphere the CH₄ mixing ratio is about 2 ppm, while inside the planetary boundary layer (PBL), the mixing ratio can be increased in the vicinity of methane sources (Baray et al., 2018). Such enhancement up to 4 ppm was observed, for example, in the airborne measurements over oil sands (Baray et al., 2018). At low altitudes, the methane concentration depends on the PBL dynamics, so it is important to profile the methane mixing ratio simultaneously with the PBL parameters such as PBL height and aerosol backscattering. Profiling of the PBL is commonly done by aerosol lidars (Kovalev and

troposphere.

provide the profile of the methane mixing ratio, which can vary significantly in the lower

Eichinger, 2004), while for methane profiling either the differential absorption (DIAL) or Raman lidars can be used.

Existing DIAL systems for measuring methane are based on tunable parametric laser sources and operate in the shortwave infrared (SWIR) spectral range between 1.65 and 2.3 µm, 65 where methane has strong absorption lines (Refaat et al., 2013; Riris et al., 2017). Due to low Raleigh scattering in the SWIR region, methane profiling using the DIAL technique is possible only inside regions containing significant amounts of aerosol. Raman lidars, by contrast, use standard off-the-shelf tripled Nd: YAG lasers, are relatively simple in design and can be operated in aerosol free atmosphere. The methane molecule is guite suitable for Raman detection. The vibrational Raman line at 2914 cm⁻¹ is well isolated and has the scattering cross section about 70 eight times higher than that of nitrogen (Weber, 1979). The main difficulties of CH₄ Raman detection are related to its low background atmospheric concentration. The first attempts to implement CH₄ Raman spectroscopy in lidars go back to eighties. Raman lidar was used for monitoring of methane plumes with relative CH₄ volume concentration of about 2% (Houston et 75 al., 1986). Monitoring of the background CH₄ concentrations in the troposphere with airborne Raman lidar was reported by Heaps and Burris (1996). In both cases powerful excimer lasers

(XeCl and XeF, respectively) were used. However, the wideband radiation of excimer lasers requires the use of wideband interference filters in Raman channel, which, in turn, increases the sky background noise and possible contribution of aerosol fluorescence. Wideband detection also

80 creates an additional complication related to interference from the oxygen Raman overtone (second Stokes shift) (Heaps and Burris, 1996). Significant progress in the development of the interference filters, detectors and laser sources during the last two decades provides, now the opportunity to develop the CH₄ Raman lidar based on a relatively compact tripled Nd:YAG laser. For narrowband 354.7 nm laser radiation the vibrational Raman line of methane is at 395.7 nm, while the oxygen Raman overtone (3089 cm⁻¹) is at 398.4 nm which can be rejected by the 85

interference filter.

90

In our paper we present the first results of methane profiling in the lower troposphere using LILAS Raman lidar from Lille University observatory platform (Hauts-de-France region, France). The observations demonstrate that inside the PBL, CH_4 mixing ratio may exceed the background concentration levels by up to a factor of 2. Enhancement of the CH₄ mixing ratio in weak elevated aerosol layers was also detected.

2. Experimental setup.

The experiments described here were performed using LILAS - multiwavelength Mie-Raman lidar from the Lille University (Veselovskii et al., 2016). The lidar is based on a tripled 95 Nd:YAG laser with a 20 Hz repetition rate, and pulse energy of 70 mJ at 355 nm. The backscattered light is collected by a 40-cm aperture Newtonian telescope. The outputs of the detectors are recorded at 7.5 m range resolution using Licel transient recorders that incorporate both analog and photon-counting electronics. The full geometrical overlap of the laser beam and 100 the telescope field of view (FOV) is achieved at approximately 1000 m height using a 0.75 mrad field of view. In its usual configuration, LILAS allows detection of three elastic backscattered signals (355, 532, 1064 nm), rotational Raman signal from N_2 and O_2 molecules at approximately 530 nm (Veselovskii et al., 2015), vibrational nitrogen and water vapor Raman signals at 387 nm and 408 nm respectively. To perform CH₄ measurements shown in this paper, 105 we modified the water vapor channel (408 nm interference filter was replaced by the methane filter centered at 395.7 nm). The dichroic mirror in the receiver did not provide efficient selection of the methane (395.7 nm) Raman component, so it was replaced by the mirror with high reflectance at 395.7 nm. Only a small portion of the 387 nm component was transmitted to the nitrogen channel, while more than 95% of the 395.7 nm signal was reflected to the methane 110 channel. Comparing nitrogen Raman backscatter intensity before and after the modification we find that the sensitivity of the 387 nm channel was degraded by a factor of 185. The strength of the nitrogen Raman signal, though being low, was sufficient for the purpose of our experiment.

The Alluxa interference filter in the methane Raman channel has a bandwidth of 0.3 nm with peak transmission greater than 80%. The corresponding transmission curve, simulated by the manufacturer in the 300 nm – 550 nm range, is shown in Fig.1. Suppression of 355 and 532 nm radiation is specified to be greater than 12 orders of magnitude. The Raman scattering cross section of methane is about 8 times higher than that of nitrogen, so for a 2 ppm methane concentration the ratio of intensities of Raman nitrogen to methane scattering is about 6*10⁴. Thus not only should the elastic backscatter components of the signal be suppressed sufficiently, but also the nitrogen and oxygen Stokes and anti-Stokes Raman lines should be as well. To eliminate the possible contribution of rotational anti-Stokes lines, the interference filter was

combined with a notch filter, providing an additional OD4 blocking in the 348 – 360 nm range.

The contribution of the vibrational anti-Stokes lines is the most significant for oxygen (336.2 nm), where the corresponding intensity is about 20 times the intensity of CH_4 line. The filter suppression at 336.2 nm is about OD9, which is sufficient to block the anti-Stokes line contribution.

The intensity of the oxygen overtone (398.4 nm) is approximately three times the intensity of the methane line (Heaps and Burris, 1996), while the filter manufacturer specifies the suppression at 398.4 nm to be above OD10, hence the contribution of the oxygen overtone is negligible. To verify that no 532 nm backscatter or corresponding Raman lines leak into the methane channel, during the initial test phase a UV glass filter with transmission of less than 5% in the 500 – 750 nm range was also added to the methane channel. No noticeable changes in CH₄ Raman signals were discovered with this additional blocking filter in place. The transmission curve in Fig.1 shows an increase in the 450 nm – 480 nm range, where aerosol fluorescence may occur (Sugimoto et al., 2012; Reichardt et al., 2017), with maximal transmission of 4*10⁻⁴% at 455 nm. The dichroic mirrors in the detection module provide additional suppression of factor 20 at this wavelength, so total suppression of fluorescence signal at 455 nm is greater than 4*10⁶. All presented Raman methane measurements were performed in the photon counting mode and at night only.

140

150

3. Numerical simulation

Numerical simulation was performed to estimate the power of the Raman backscatter for the background methane mixing ratio. The lidar equation describing the number of detected photons N_x^{ph} , scattered by molecule "x" at distance z due to a single laser pulse can be written as:

145
$$N_x^{ph}(z) = O(z)A_x \frac{E}{\hbar\nu} \Delta z \frac{S}{z^2} N_x \sigma_x \exp\left\{-\int_0^z (\alpha_L^a + \alpha_L^m + \alpha_x^a + \alpha_x^m) dz'\right\}$$
(1)

Here O(z) is the geometrical overlap factor, A_x is an efficiency factor, including the transmission of the optics and the quantum efficiency of the detectors. E and hv are the laser pulse and the photon energies, Δz - range resolution, S - receiving telescope area, N_x – number concentration of molecule "x" and σ_x is the differential Raman scattering cross section of molecule "x", α – is the extinction coefficient, where superscripts "a" and "m" indicate aerosol and molecular

contributions, respectively. Subscripts "L" and "x" correspond to the laser wavelength λ_L and to the wavelength of Raman backscatter λ_x .

Table 1 shows the parameters of H_2O , CO_2 and CH_4 molecules, such as Raman frequency shift and Raman differential scattering cross section σ_x , normalized to the cross section of nitrogen σ_{N_2} . Results are presented for an excitation wavelength of 337 nm basing on Weber 155 (1979). The table also provides typical concentrations of gases in the troposphere. The efficiency of detection of molecule "x" is determined by the factor $n_x \times \frac{\sigma_x}{\sigma_{N_x}}$ (n_x is the molecule "x" mixing

ratio), which is approximately 10^4 for the H₂O molecule and about 320 for CO₂. However for CH₄ this factor is about 20 times lower than for CO₂, so detection of the methane background concentrations demands a powerful Raman lidar and significant signal accumulation time.

The lidar derived mixing ratio of methane can be calculated from the ratio of CH₄ and N₂ lidar Raman signals (P_{CH_4} and P_{N_2}), corrected for the aerosol and molecular differential extinction:

$$n_{CH4}(z) = K \frac{P_{CH_4}}{P_{N_2}} \exp\left\{-\int_{0}^{z} \left[\alpha_{N_2}^{a} \left(1 - \left(\frac{\lambda_{CH_4}}{\lambda_{N_2}}\right)^{-\gamma}\right) + \alpha_{N_2}^{m} \left(1 - \left(\frac{\lambda_{CH_4}}{\lambda_{N_2}}\right)^{-4}\right)\right] dz'\right\}$$
(2)

Here λ_{N_2} and λ_{CH_4} are the wavelengths of nitrogen and methane Raman components; $\alpha_{N_2}^a$, $\alpha_{N_2}^m$ 165 are the aerosol and molecular extinctions at λ_{N_2} ; γ is the Ångstrom exponent and K – is the calibration constant. In our measurements, we assume that the CH₄ mixing ratio above the boundary layer is 2 ppm and this value was used for calibration purpose. The calibration, in principle, can be performed from first principles by using a calibration lamp with known 170 spectrum, as it has been done for Raman water vapor lidars (Venable et al., 2011). The methane mixing ratio in (2) is calculated from the ratio of the lidar signals, so the geometrical overlap factors are at least partially compensated and thus measurements below the height of the full overlap are possible. We still need to extrapolate the extinction coefficient to the region of incomplete overlap, however the influence of the aerosol differential extinction term in (2) is 175 lower than in the water vapor measurements due to the lower wavelength separation between

nitrogen and methane Raman components.

aerosol layer extends from the ground up to 2 km height. In modeling the aerosol extinction coefficients at 355 nm, extinction values of 0.05, 0.1, and 0.2 km⁻¹ were considered. The number of detected photons was calculated from (1) for Δz =100 m, A_X=0.1 and n_{CH4} =2 ppm. The assumed laser pulse energy was 70 mJ at 354.7 nm, which corresponds with the LILAS laser energy during the observations reported. The nitrogen Raman scattering cross section of 5.4*10⁻³¹ cm²/sr at 488 nm is taken from (Penney et al., 1974) and recalculated for 355 nm. Finally,

To estimate the statistical uncertainties of methane detection, we assume that a uniform

assuming that
$$\frac{\sigma_{CH_4}}{\sigma_{N_2}} = 8.2$$
 (Weber, 1979), the value $\sigma_{CH_4} = 1.9 \times 10^{-29}$ cm²/sr at 355 nm was used.

185 Statistical uncertainties of the measurements are determined mainly by the weak CH₄ Raman backscatter and, in the absence of background noise, the uncertainty can be estimated as

$$\varepsilon \approx \frac{1}{\sqrt{N_{CH_4}^{ph}}}.$$

Fig.2 shows vertical profiles of statistical uncertainties for three values of aerosol extinction coefficient: 0.05, 0.1, 0.2 km⁻¹ and a signal averaging time of 4 hours. The figure

190 shows also the photon counting rate in the methane Raman channel $v_{CH_4} = N_{CH_4}^{ph} \frac{2\Delta z}{c}$, where c is the speed of the light. For the clean atmosphere ($\alpha_{355}=0.05 \text{ km}^{-1}$) the measurements with uncertainty below 10% are possible up to 4 km, while for $\alpha_{355}=0.2 \text{ km}^{-1}$ the corresponding range decreases to 3 km. The simulation results confirm the necessity of long-term (several hours) signal accumulation in methane measurements using Raman lidar.

195

4. Results of measurements.

Measurements were performed at Lille University observatory platform, France, during the period May-June, 2018. In total, 20 nighttime observation sessions were accomplished. Fig.3 shows CH₄ and N₂ Raman lidar signals together with the backscattered signal at 1064 nm on the night of 14-15 June 2018. The results are averaged over the temporal interval τ_{av} =4.0 hours. Aerosols are mainly located below 1700 m (maximal value of aerosol extinction α_{355} inside the PBL is about 0.1 km⁻¹), though a weak aerosol layer is also visible in the 1064 nm lidar signal in the 2.5 – 4.0 km height interval. HYSPLIT back trajectory analysis (Stein et al., 2015; Rolph et

al., 2017) shows that the air mass in this layer originates from Canada, being transported over the

- 205 Atlantic. The photon counting rate in the methane channel at 2000 m is about 1.8 KHz, which agrees with simulation results in Fig.2 for $\alpha_{355}=0.1$ km⁻¹. As mentioned, for a 2 ppm CH₄ mixing ratio, the nitrogen Raman backscatter should be approximately 6*10⁴ stronger than that from methane. The value estimated from our measurements of $5*10^4$ agrees well with this prediction, which confirms that we observed methane Raman scattering. The profile of the methane mixing
- 210 ratio calculated from the measurements in Fig.3 and averaged over 100 m height bins is shown in Fig.4. The same figure provides the profile of the backscattering coefficient at 532 nm. The mixing ratio is given in arbitrary units, assuming that the value of 1.0 corresponds to $n_{CH_4} = 2$ ppm. The mixing ratio inside the PBL exceeds the corresponding values above 4000 m by approximately a factor of 2. The profiles of n_{CH_4} and β_{532} are not correlated: inside the PBL the
- maximum of β_{532} is at a height of 1500 m, while the maximum of n_{CH_4} is at 1100 m. The 215 backscattering coefficient β_{532} of the weak aerosol layer at 3500 m is about 7.6*10⁻⁵ km⁻¹sr⁻¹, which is almost a factor of 50 lower than the maximum value of β_{532} inside the PBL. In this elevated layer, the CH₄ mixing ratio also increases, however n_{CH_4} at 3500 m is 1.5, which is close to the values in the PBL. Thus, the enhancement of n_{CH_4} at 3500 nm is very unlikely to be an artifact related to aerosol interference.
- 220

The derived methane profiles exhibited strong night to night variation. Fig.5 shows the results of six measurement sessions, representing nights with different aerosol loading., On the night of 20-21 May (Fig.5a) a scattering layer with peak value $\beta_{532}=0.09 \text{ km}^{-1}\text{sr}^{-1}$ occurs in the 2500 - 3100 m height range. Low lidar ratio (below 20 sr) and low depolarization ratio (below 225 5%) indicate that this layer is likely a water cloud. Strong cloud scattering demonstrates no influence on the mixing ratio, which is about 1.0 in the center of the cloud, proving that the interference filters provide sufficient rejection of elastic scattering. It should be mentioned also that the Raman band of the liquid water extends from 395 nm to 409 nm (Reichardt, 2014), so potentially it can be an interfering factor in the methane measurements. However, Fig.5a does 230 not reveal a noticeable effect of liquid water Raman scattering on the methane profile due to the narrowband filter in the CH₄ channel. Cloud layers occurred also on 26-27 May and 27-28 May (Fig.5b,c) at a height of approximately 4000 m with maximum values of backscattering coefficients of $\beta_{532}=0.006 \text{ km}^{-1}\text{sr}^{-1}$ and 0.02 km⁻¹sr⁻¹, respectively. As in Fig.5a, the presence of

clouds does not influence the methane measurements. By contrast, the night of 12-13 June 235 (Fig.5f) was characterized by a low aerosol backscattering coefficient in the 500 - 4000 m range $(\beta_{532} \text{ is below } 2^*10^{-4} \text{ km}^{-1} \text{sr}^{-1})$ and the mixing ratio shows no significant deviation from the 1.0 value in the whole height range.

The vertical variation of methane content was related to the PBL height, as can be concluded from a comparison of Fig.5d and Fig.5e. On 30-31 May the aerosol is confined below 2000 m, while on 2-3 June it is below 750 m. Respectively, the n_{CH_4} decreases from 2.4 at 500 m 240 to 1.0 at 2000 m in the first case, while in the second case the background level of 1.0 is observed for the heights above 750 m. On 2-3 June, the increase of CH4 mixing ratio at 3400 m correlates with a weak aerosol layer ($\beta_{532} < 10^{-4} \text{ km}^{-1} \text{sr}^{-1}$) at the same height. It is interesting that a stronger aerosol layer at 2300 m is not accompanied by an increase in n_{CH_4} . The air masses in both layers were transported over the Atlantic from Canada and corresponding backward 245 trajectories are close, so we are not able to make conclusions about difference in nature of these layers.

Enhancement of n_{CH_4} in weak elevated aerosol layers was observed several times during the campaign, in all cases air mass was transported over the Atlantic. One such observation session 250 was on the night 13-14 June 2018. Fig.6 shows the temporo-spatial distribution of the range corrected lidar signal at 1064 nm and the particle depolarization ratio at 532 nm for this session. Most of the aerosols are below 2000 m, but there is an elevated layer in the 3000 - 5000 m height range. The depolarization ratio inside the PBL is low (δ_{532} < 5%) while in the elevated layer δ_{532} increases up to approximately 18%. The available radiosonde data from Paris (France) 255 and Essen (Belgium) show that the relative humidity in the elevated layer is below 40%. The profiles of aerosol backscattering coefficient β_{532} , particle depolarization δ_{532} and CH₄ mixing ratio on 13-14 June for the temporal interval of 22:00 - 02:00 UTC are given in Fig.7. In the PBL the mixing ratio is about 2.0, and in the elevated layer the $n_{CH_{L}}$ demonstrates also an increase up to approximately 1.5.

260 To understand the origin of this elevated layer, a ten-days back-trajectory analysis, for the air mass over Lille, at 4000 m, on 14 June 2018 at 00:00 UTC, was performed using the HYSPLIT model. According to the analysis, the air mass was transported from eastern Asia (Russia and China) to North America and then over the Atlantic Ocean to Europe. Large-scale

boreal fire activities were detected near the border of Russia and China in early June, thus the air

- 265 mass at 4000 m could have contained fire emissions. Fig.8 (a)–(d) plot the transport pathway of the air mass overlaid with the CO columnar concentration maps on 03, 06, 09 and 12 June, respectively. The CO concentration is derived from AIRS Level 3 CO products (Texeira, 2013). The propagation of the air mass is clearly coincident with the transport of CO plumes. Studies have shown that CO₂, CO and CH₄ are among the main products of boreal forest fires (Hao et al.,
- 270 1993; Kasischke et al., 2002; Worden et al., 2013). CO originating from boreal fires is positively correlated with CH₄ concentration, however, the CH₄ product of AIRS is not as mature as the CO product due to the low sensitivity to CH₄ in the lower troposphere, so CO is a favorable tracer of fire emissions. In Fig.8 (a), intense CO plumes are detected at the origin of the trajectory, which is close to the fire activities. Hence, it is possible that the observed methane plume comes from fire emissions in eastern Asia. Aged smoke particles mixed with Asian dust particles could be the
- reason for the high particle depolarization ratio observed in the elevated layer.

5. Discussions and conclusion

- The results presented here demonstrate the feasibility of profiling the background mixing ratios of methane in the lower troposphere using Raman lidar. The photon counting rate in the methane Raman channel agrees with numerical simulation for typical aerosol loading and a background CH₄ mixing ratio of 2 ppm, which confirms that we observe the methane Raman scattering. In our measurements we always observed enhanced concentrations of the methane inside the PBL, compared to aerosol free regions, thus analysis of methane ground sources in Northern France is in our upcoming plans.
- Raman measurements of CH₄ mixing ratio close to 2 ppm is a challenging task due to different potential interfering factors, such as leakage of the elastic signal into the Raman channel, contribution of liquid water Raman scattering and aerosol fluorescence. Measurements performed inside the clouds revealed no interfering of elastic signal or Raman liquid water spectra. Estimation of aerosol fluorescence contributions is more difficult. The aerosol fluorescence at wavelengths above 440 nm was reported recently by Reichardt et al. (2017). For profiling, the authors had to integrate the fluorescence signal over the spectral range of approximately 80 nm. In our system the filter bandwidth is only 0.3 nm, so we expect that the fluorescence contribution is suppressed. The CH₄ profiles are not always correlated with aerosol

295 backscattering, which corroborates the hypothesis that, in the PBL, not aerosol fluorescence but methane is measured. However, we cannot completely exclude fluorescence contribution. To measure and correct for it, if necessary, in future measurements we plan to introduce an additional control channel close to 393 nm where no strong Raman lines exist and only fluorescence can be detected (Reichardt, 2014). Furthermore, an additional filter will be used to 300 increase the rejection in 450-480 nm spectral range.

One of the main problems in the measurements presented is the long signal accumulation time, which was about 4 hours in our case. A more powerful laser is needed to improve the temporal resolution of the measurements. Today, compact diode pumped lasers, with pulse energy of 60 mJ at 355 nm and 200 Hz repetition rate have become widely available (e.g. 305 www.quantel-laser.com/en/products/item/q-smart-dpss-650-mj.html), so it is possible to decrease the measurement time to less than 30 minutes. Numerous lidar technologies developed previously for H₂O Raman systems can be used for the methane Raman lidar. In particular, the calibration technique based on the tungsten lamp spectrum, can provide absolute values of methane mixing ratio from first principles (Venable et al., 2011).

- 310 Raman lidars for CH₄ monitoring cannot, of course, compete with airborne DIAL systems in sensitivity and accuracy, especially when column concentrations are considered. However, when one needs to evaluate the vertical profile of methane concentrations through the boundary layer, the Raman lidar may have some advantages. In particular, the IR DIALs can profile methane only in the region loaded with aerosol, while Raman lidar is capable to profile in the aerosol free atmosphere also. Our results demonstrate that conventional Mie-Raman lidars designed for aerosol and the water vapor observations can be relatively easy modified for methane observations. Such ground-based lidars can be also used for ground validation activities during the upcoming MERLIN mission.
- 320 Acknowledgments: The authors are very grateful to the labex CaPPA for supporting this study. The CaPPA project (Chemical and Physical Properties of the Atmosphere) is funded by the French National Research Agency (ANR) through the PIA (Programme d'Investissement d'Avenir) under contract "ANR-11-LABX-0005-01" and by the Regional Council "Nord-Pas de Calais" and the "European Funds for Regional Economic Development" (FEDER).

References

335

- Ansmann, A., Riebesell, M., Wandinger, U., Weitkamp, C., Voss, E., Lahmann, W., and Michaelis, W.: Combined Raman elastic-backscatter lidar for vertical profiling of moisture, aerosols extinction, backscatter, and lidar ratio, Appl.Phys.B, 55, 18-28, 1992.
- 330 Avila, G., Fernidndez, J. M., Tejeda, G., and Montero, S.: The Raman spectra and cross-sections of H20, D20, and HDO in the OH/OD stretching regions. J. Mol. Spectr., 228, 38-65, 2004.
 - Baray, S., Darlington, A., Gordon, M., Hayden, K., L., Leithead, A., Li, S.-M., Liu, P. S. K.,Mittermeier, R. L., Moussa, S. G, O'Brien, J, Staeble, R., Wolde, M., Worthy, D., andMcLaren, R.: Quantification of methane sources in the Athabasca Oil Sands Region of
 - Bovensmann, H., Burrows, J. P., Buchwitz, M., Frerick, J., Noël, S., Rozanov, V. V., Chance, K. V., and Goede, A. H. P.: SCIAMACHY Mission objectives and measurement modes, J.

Alberta by aircraft mass balance, Atmos. Chem. Phys., 18, 7361–7378, 2018

Atmos. Sci., 56, 127–150, 1999.

- Etminan, M., Myhre, G., Highwood, E., and Shine, K.: Radiative forcing of carbon dioxide,
- 340 methane, and nitrous oxide: A significant revision of the methane radiative forcing, Geophysical Research Letters, 43, doi:10.1002/2016GL071930, 12.614-12.623, 2016.
 - Hao, W. M., & Ward, D. E.: Methane production from global biomass burning. Journal of Geophysical Research: Atmospheres, 98 (D11), 20657-20661, 1993.

Heaps, Wm. S. and Burris, J.: Airborne Raman lidar, Appl. Opt. 35, 7128-7135, 1996.

- 345 Houston, J.D., Sizgoric, S., Ulitsky, A., Banic, J.: Raman lidar system for methane gas concentration measurements, Appl. Opt., 25, 2115-2121, 1986.
 - Inaba H. and Kobayasi T.: Laser-Raman Radar Laser-Raman scattering methods for remote detection and analysis of atmospheric pollution, Opto-electronics, 4, 101-123, 1972.

IPCC: Climate Change 2013: The Physical Science Basis. Contribution of Working Group I to

- the Fifth Assessment Report of the Intergovernmental Panel on Climate Change, Cambridge University Press, Cambridge, United Kingdom and New York, NY, USA, 2013.
 - Kavitha, M., Nair, P. R.: Non-homogeneous vertical distribution of methane over Indian region using surface, aircraft and satellite based data, Atmospheric Environment 141, 174 185, 2016.
- 355 Kovalev, V.A., Eichinger, W. E.: Elastic lidar: theory, practice, and analysis methods. New York: Wiley-Interscience; 2004

- Kuze, A., Suto, H., Nakajima, M., and Hamazaki, T.: Thermal and near infrared sensor for carbon observation Fourier-transform spectrometer on the Greenhouse Gases Observing Satellite for greenhouse gases monitoring, Appl. Opt., 48, 6716–6733, 2009.
- 360 Müller, D., Mattis, I., Tatarov, B., Noh, Y. M., Shin, D. H., Shin, S. K., Lee, K. H., Kim, Y. J., and Sugimoto, N.: Mineral quartz concentration measurements of mixed mineral dust/urban haze pollution plumes over Korea with multiwavelength aerosol Raman - quartz lidar, Geophys. Res. Let. 37, L20810, doi:10.1029/2010GL044633, 2010.
 - Penney, C. M., Peters, R. L. St., and Lapp, M. : Absolute rotational Raman cross sections for N₂, 0₂, and CO₂, JOSA 64, 712-716, 1974.
 - Reichardt, J.: Cloud and aerosol spectroscopy with Raman lidar, J. Atm. Ocean. Tech., 31, 1946-1963, 2014.
 - Reichardt, J., Leinweber, R., Schwebe, A.: Fluorescing aerosols and clouds: investigations of coexistance, Proceedings of the 28th ILRC, Bucharest, Romania, 25-30 June, 2017.
- 370 Refaat, T. F., Ismail, S., Nehrir, A. R., Hair, J. W., Crawford, J. H., Leifer, I., Shuman, T.: Performance evaluation of a 1.6-µm methane DIAL system from ground, aircraft and UAV platforms, Opt. Expr., 21, 30415-30432, 2013.
 - Riris, H., Numata, K., Wu, S., Gonzalez, B., Rodriguez, M., Scott, S., Kawa, S., Mao, J.: Methane optical density measurements with an integrated path differential absorption lidar
- from an airborne platform, J. of Appl. Remote Sensing, 11, 034001-19, 2017.
 Rolph, G., Stein, A., and Stunder, B.: Real-time Environmental Applications and Display system: READY. Environmental Modelling & Software, 95, 210-228, 2017.
 - Stein, A.F., Draxler, R.R, Rolph, G.D., Stunder, B.J.B., Cohen, M.D., and Ngan, F.: NOAA's HYSPLIT atmospheric transport and dispersion modeling system, Bull. Amer. Meteor. Soc.,
- 380 96, 2059-2077, 2015.

- Sugimoto, N., Huang, Z., Nishizawa, T., Matsui, I., Tatarov, B.: Fluorescence from atmospheric aerosols observed with a multichannel lidar spectrometer," Opt. Expr. 20, 20800-20807, 2012.
- Texeira, A. S. T.: AIRS/Aqua L3 Daily Standard Physical Retrieval (AIRS-only) 1 degree x 1 degree V006, Greenbelt, MD, USA, Goddard Earth Sciences Data and Information Services
- Center (GES DISC), https://doi.org/10.5067/Aqua/AIRS/DATA303, accessed: 2018-04-28, 2013.

- Venable, D. D., Whiteman, D. N., Calhoun, M. N., Dirisu, A.O., Connell, R. M., Landulfo, E.: Lamp mapping technique for independent determination of the water vapor mixing ratio calibration factor for a Raman lidar system, Appl. Opt., 50, 4622 - 4632, 2011.
- Veselovskii, I., Goloub, P., Podvin, T., Bovchaliuk, V., Derimian, Y., Augustin, P., Fourmentin, M., Tanre, D., Korenskiy, M., Whiteman, D., Diallo, A., Ndiaye, T., Kolgotin, A., Dubovik, O.: Study of African dust with multi-wavelength Raman lidar during the "SHADOW" campaign in Senegal, Atm. Chem. Phys. 16, 7013–7028, 2016.
- Veselovskii, I., Whiteman, D. N., Korenskiy, M., Suvorina, A., Perez-Ramirez, D.: Use of
 rotational Raman measurements in multiwavelength aerosol lidar for evaluation of particle
 backscattering and extinction, Atmos. Meas. Tech., 8, 4111-4122, 2015.
 - Weber, A.: Raman spectroscopy of gases and liquids", Springer, New York, 1979.
 - Whiteman, D. N., S. H. Melfi, R. A. Ferrare, "Raman Lidar System for Measurement of Water Vapor and Aerosols in the Earth's Atmosphere", Appl. Opt. Vol. 31 No. 16 (1992)
- 400 Whiteman, D.N., Veselovskii, I., Cadirola, M., Rush, K., Comer, J., Potter, J.R., Tola, R.: Demonstrations measurements of water vapor, cirrus clouds, and carbon dioxide using a highperformance Raman lidar", J. Atmos. Oceanic Technol. 24, 1377-1388 (2007).
 - Worden, J., Wecht, K., Frankenberg, C., Alvarado, M., Bowman, K., Kort, E., Kulawik, S., Lee,M., Payne, V., Worden, H.,: CH₄ and CO distributions over tropical fires during October 2006
- 405 as observed by the Aura TES satellite instrument and modeled by GEOS-Chem. Atmos. Chem. Phys. 13, 3679 – 3692, 2013.
 - Zhao, P., Zhang, Y., Wang, L Cao, K Su, J., (€W)1, Hu, S., Hu, H.: Measurement of tropospheric CO₂ and aerosol extinction profiles with Raman lidar, Chinese Optics Letters, 6, 157-160, 2008.

Table.1 Raman frequency shift for H_2O , CO_2 , CH_4 molecules and their scattering cross sections normalized to the cross section of nitrogen. Results are given for an excitation wavelength of 337 nm (Weber, 1979). Typical background gases mixing ratios n_x in the low troposphere and the

$$n_x \times \frac{\sigma_x}{\sigma}$$

415

product

 σ_{N_2} are also given.

Molecule	Frequency, cm ⁻¹	σ_x	Typical values of n _x , ppm	$n \times \frac{\sigma_x}{1}$
		$\sigma_{_{N_2}}$		σ_{N_2} , ppm
H ₂ O	3657	3.1	$3*10^{3}$	~10 ⁴
CO ₂	1285	0.8	400	320
CH ₄	2914	8.2	2	16.4

Figures



Fig.1. Simulated transmission curve of the methane filter provided by Alluxa.



Fig.2. Modeled photon counting rate (lines) and statistical uncertainties of the methane mixing ratio measurements (lines + symbols) for a 2 ppm methane concentration and three values of aerosol extinction coefficient α_{355} =0.05, 0.1, 0.2 km⁻¹. Aerosol extends from z=0 to z=2000 m. Signal averaging time is 4 hours.





Fig.3. Lidar signals corresponding to elastic scattering at 1064 nm, nitrogen Raman scattering at 387 nm and methane Raman scattering at 396 nm on the night 14-15 June 2018. The units are MHz for 387 and 396 nm, and mV for 1064 nm. Black lines show the profiles of molecular scattering. Measurements were performed from 22:00 to 02:00 UTC, signals averaging time t_{aver} =4.0 hours.



Fig.4. Vertical profiles of aerosol backscattering coefficient at 532 nm and methane mixing ratio
calculated from measurements on 14-15 June for the same temporal interval as in Fig.1. Mixing ratio is uncalibrated and the value 1.0 corresponds to approximately 2 ppm.



Fig.5. Vertical profiles of the methane mixing ratio and aerosol backscattering at 532 nm for six
night measurement sessions: (a) 20-21 May, (b) 26-27 May, (c) 27-28 May, (d) 30-31 May, (e)
2-3 June, (f) 12-13 June 2018. Mixing ratios are not calibrated and the value 1.0 corresponds to approximately 2 ppm. Signals averaging time τ_{av} is given in hours.



455 Fig.6. Range corrected lidar signal at 1064 nm and the particle depolarization ratio at 532 nm for the night 13-14 June 2018.



465 Fig.7. Aerosol backscattering coefficient β_{532} , particle depolarization δ_{532} and CH₄ mixing ratio on 13-14 June 2018 for temporal interval 22:00 – 02:00 UTC. Values of δ_{532} are multiplied by factor 0.1. Mixing ratios are not calibrated and the value 1.0 corresponds to approximately 2 ppm.



Fig.8. Ten-day backward trajectories for the air mass in Lille at altitude 4000 m on 14 June 2018 at 00:00 UTC. Plots (a)-(d) show the trajectory pathways overlaid with CO columnar concentration maps retrieved from AIRS data. The triangles represent the location of the traced air mass on corresponding dates. Plots (e) and (f) show the trajectory and vertical propagation of air mass.