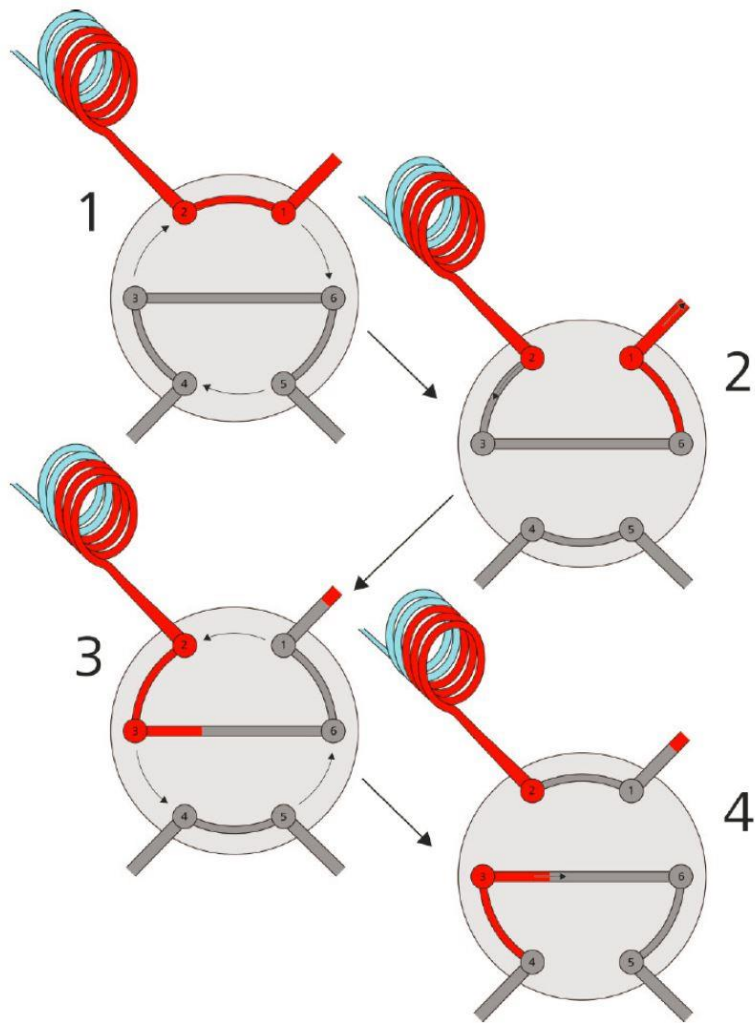


1 Electronic Supplementary Material

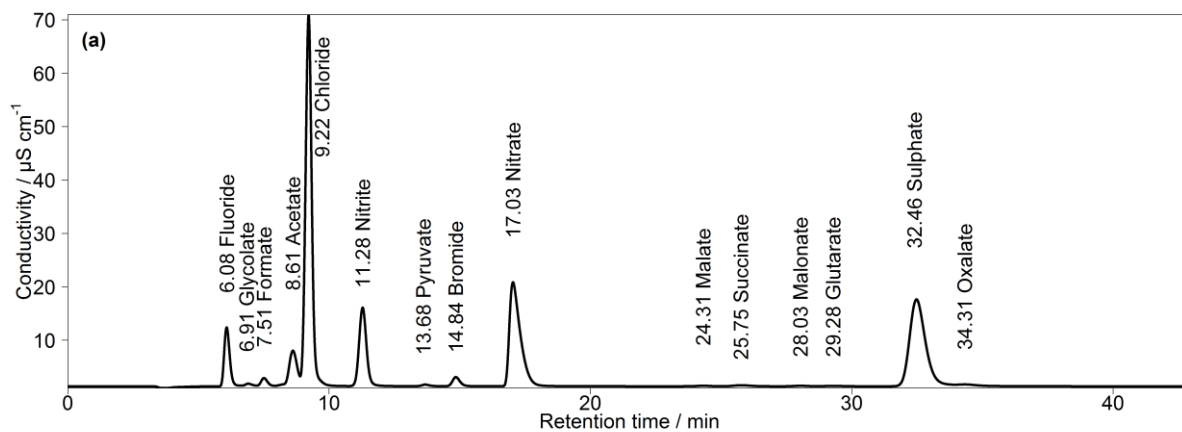
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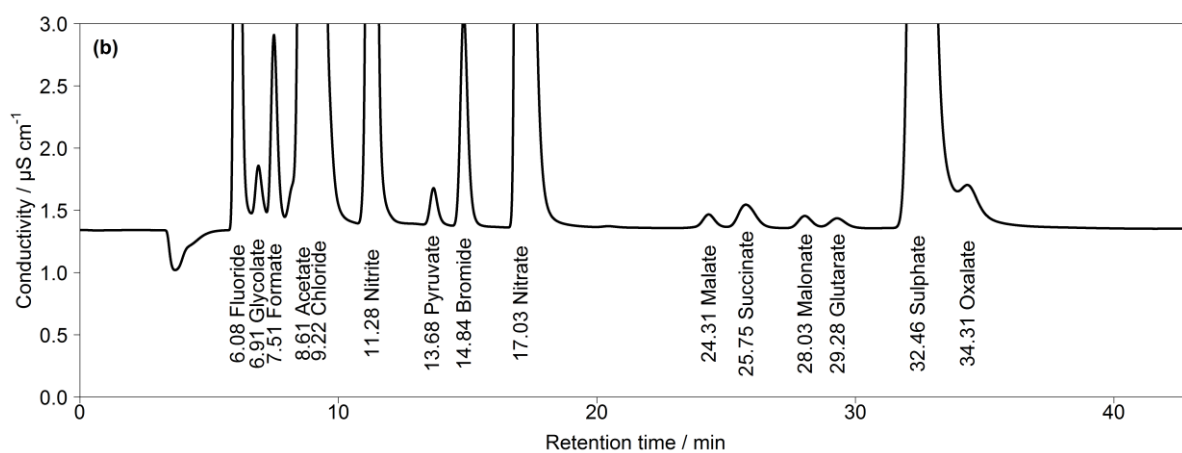
3

4 **Figure S1.** Switch modes of the external 6-way-valve. (1) Sample loop is filled with the sample
5 solutions. In (2) and (3) valve is switched to “fill mode” to pre-concentrate the sample. During
6 the “injection mode” in (4), the eluent dissolved the trapped ions and flows to the
7 separation columns. ([https://www.metrohm.com/de-de/produkte/ionenchromatographie/ionen](https://www.metrohm.com/de-de/produkte/ionenchromatographie/ionenchromatographie-inline-probenvorbereitung/)
8 [chromatographie-inline-probenvorbereitung/](https://www.metrohm.com/de-de/produkte/ionenchromatographie/ionenchromatographie-inline-probenvorbereitung/), 10th August 2018)

9



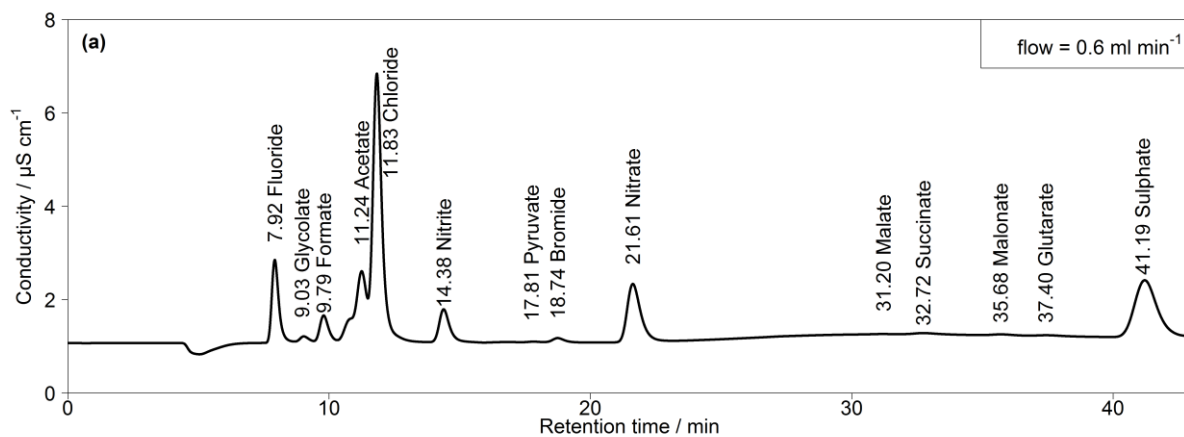
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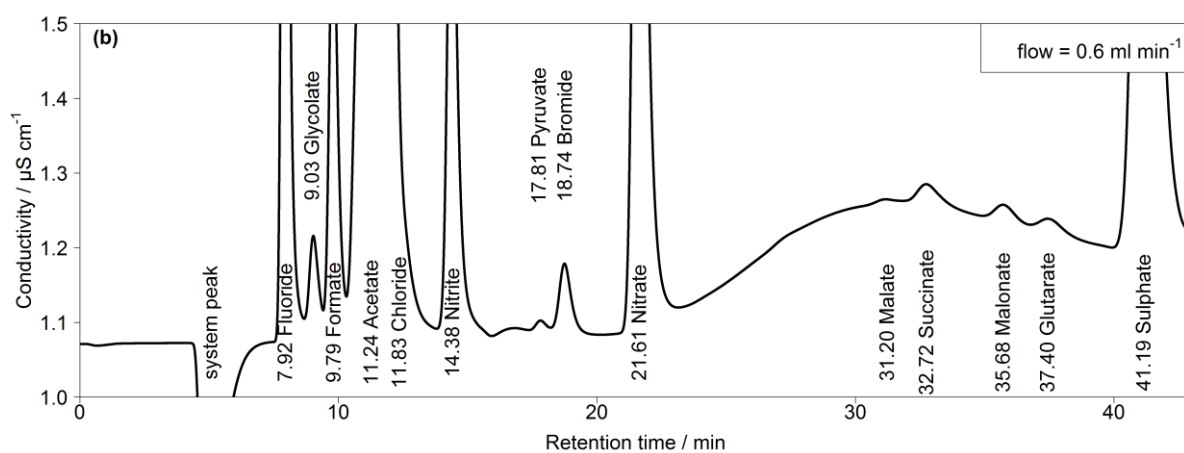
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12 **Figure S2.** (a) Chromatogram of a standard solution with aqueous concentrations of $150 \mu\text{g l}^{-1}$
 13 for Cl^- , NO_3^- , SO_4^{2-} , $75 \mu\text{g l}^{-1}$ for NO_2^- , $15 \mu\text{g l}^{-1}$ for F^- , Br^- and $3 \mu\text{g l}^{-1}$ for all organic acids.
 14 Numbers in front of the ion names are the retention times. $T = 65^\circ\text{C}$ and eluent flow of
 15 0.8 ml min^{-1} . (b) Zoom in of (a).

16



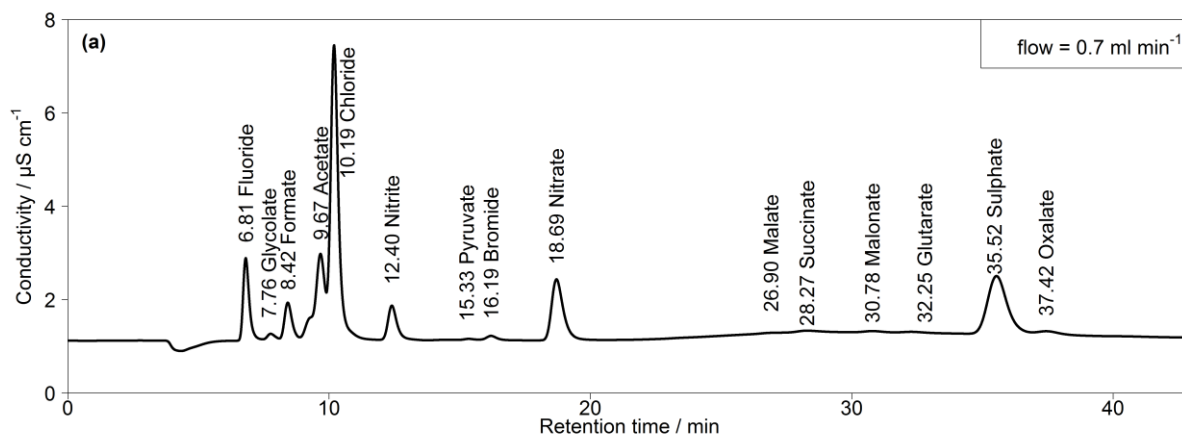
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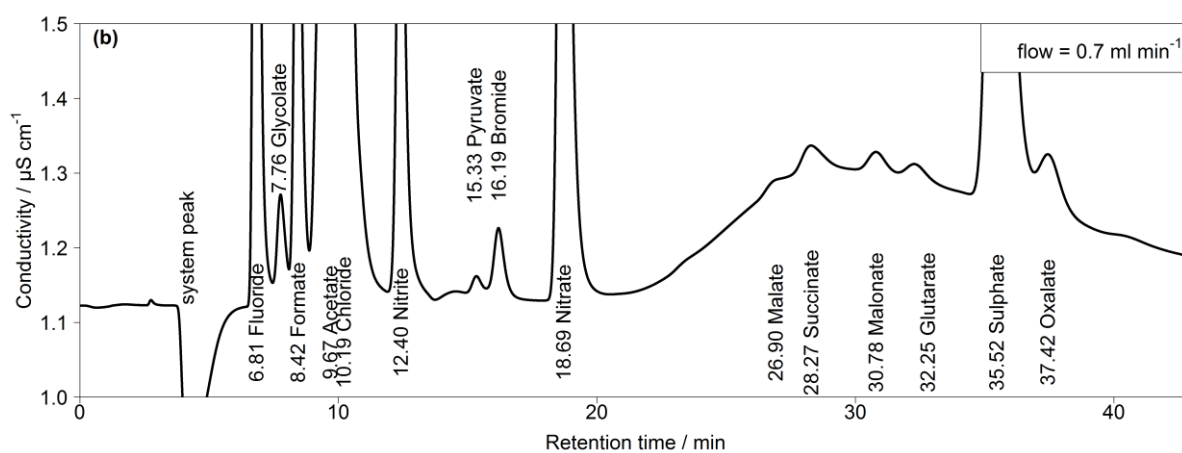
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19 **Figure S3.** (a) Chromatogram of a standard solution with aqueous concentrations of $10 \mu\text{g l}^{-1}$
 20 for Cl^- , NO_3^- , SO_4^{2-} , $5 \mu\text{g l}^{-1}$ for NO_2^- and $1 \mu\text{g l}^{-1}$ for F^- , Br^- as well as all organic acids.
 21 Numbers in front of the ion names are the retention times. $T = 65^\circ\text{C}$ and eluent flow of
 22 0.6 ml min^{-1} . (b) Zoom in of chromatogram in (a).

23



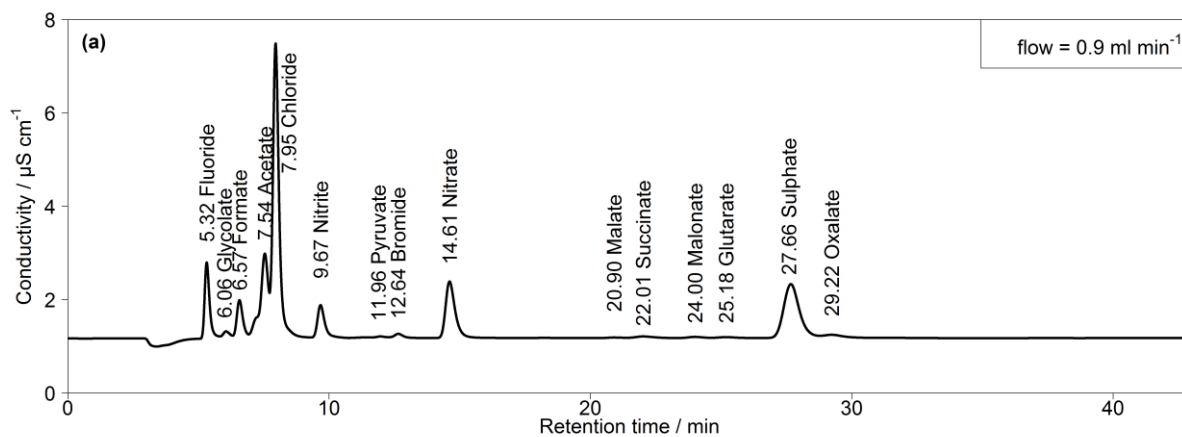
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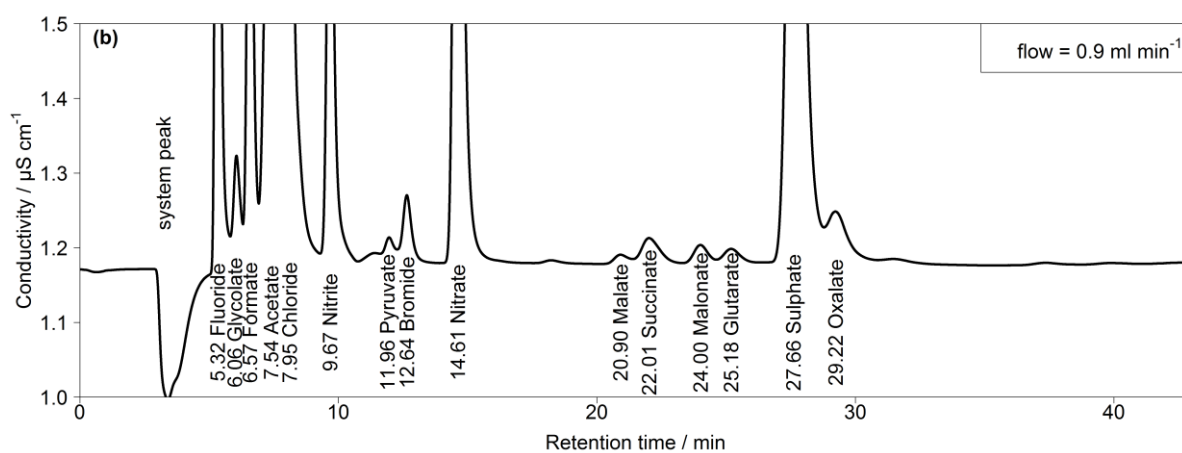
25

26 **Figure S4.** (a) Chromatogram of a standard solution with aqueous concentrations of $10 \mu\text{g l}^{-1}$
 27 for Cl^- , NO_3^- , SO_4^{2-} , $5 \mu\text{g l}^{-1}$ for NO_2^- and $1 \mu\text{g l}^{-1}$ for F^- , Br^- as well as all organic acids.
 28 Numbers in front of the ion names are the retention times. $T = 65 \text{ }^\circ\text{C}$ and eluent flow of
 29 0.7 ml min^{-1} . (b) Zoom in of chromatogram in (a).

30



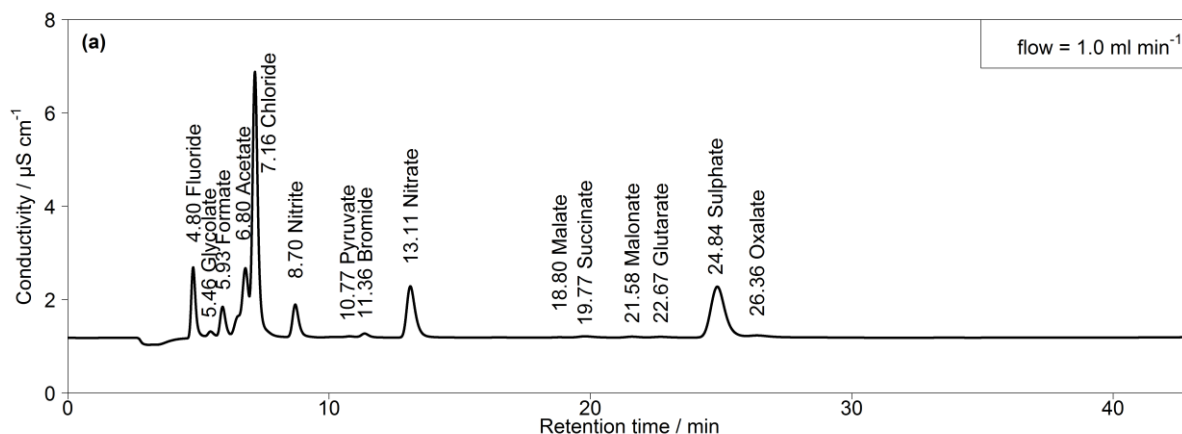
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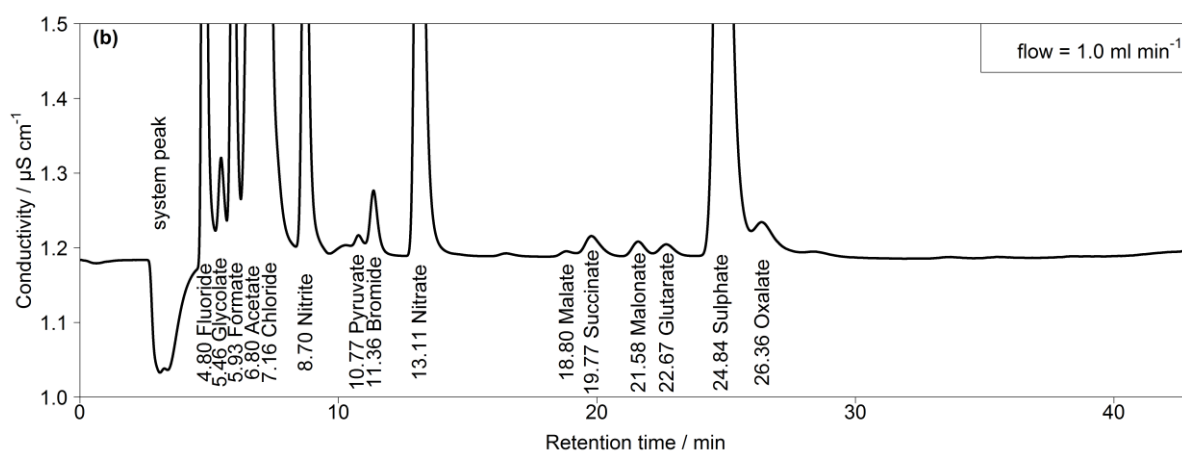
32

33 **Figure S5.** (a) Chromatogram of a standard solution with aqueous concentrations of $10 \mu\text{g l}^{-1}$
 34 for Cl^- , NO_3^- , SO_4^{2-} , $5 \mu\text{g l}^{-1}$ for NO_2^- and $1 \mu\text{g l}^{-1}$ for F^- , Br^- as well as all organic acids.
 35 Numbers in front of the ion names are the retention times. $T = 65^\circ\text{C}$ and eluent flow of
 36 0.9 ml min^{-1} . (b) Zoom in of chromatogram in (a).

37



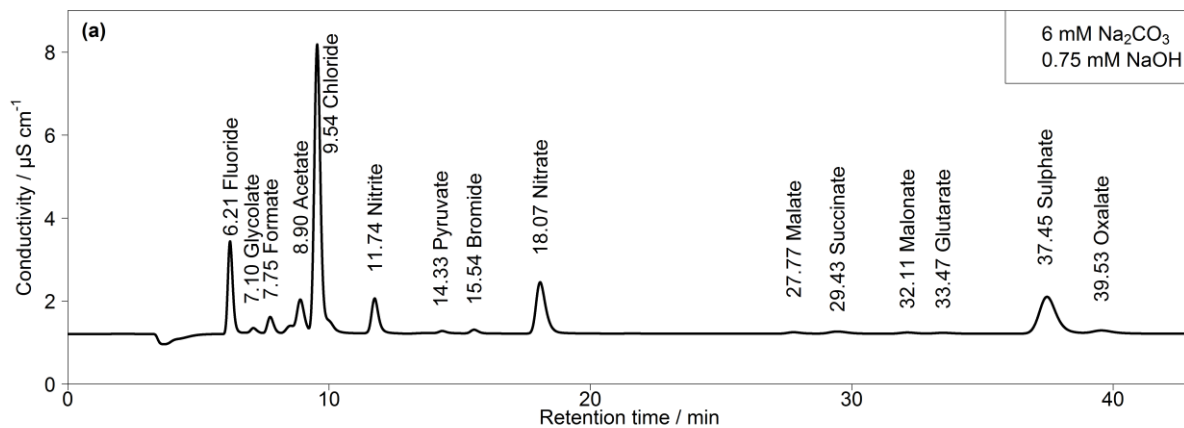
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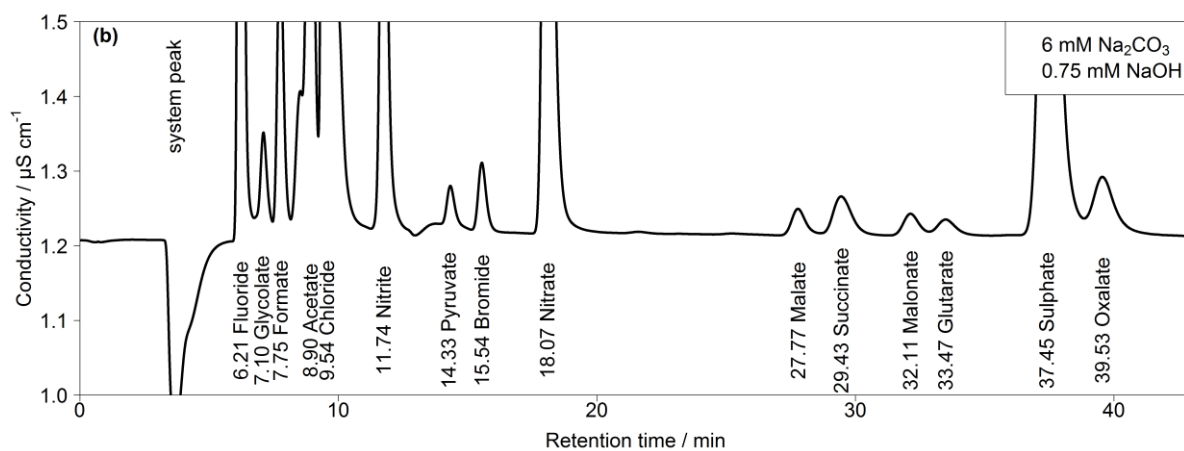
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40 **Figure S6.** (a) Chromatogram of a standard solution with aqueous concentrations of $10 \mu\text{g l}^{-1}$
 41 for Cl^- , NO_3^- , SO_4^{2-} , $5 \mu\text{g l}^{-1}$ for NO_2^- and $1 \mu\text{g l}^{-1}$ for F^- , Br^- as well as all organic acids.
 42 Numbers in front of the ion names are the retention times. $T = 65^\circ\text{C}$ and eluent flow of
 43 1.0 ml min^{-1} . (b) Zoom in of chromatogram in (a).

44



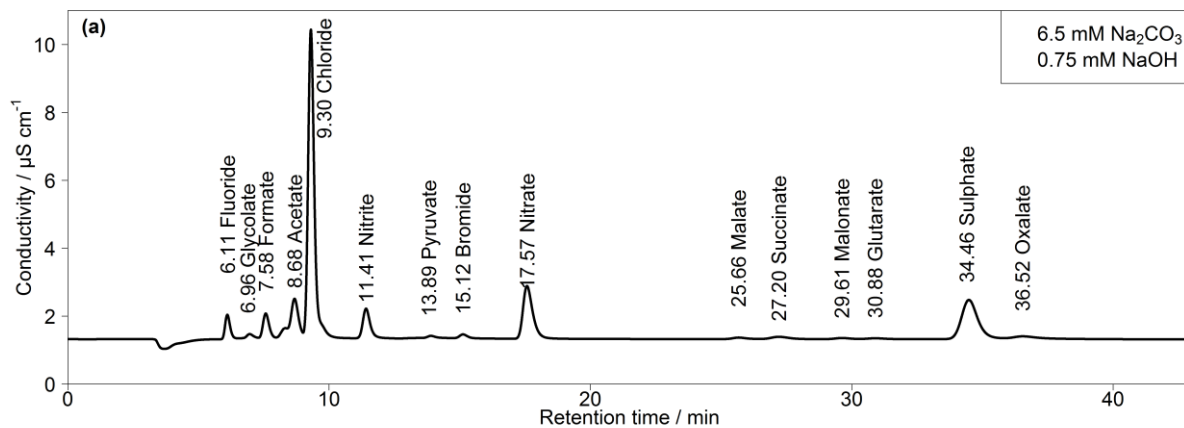
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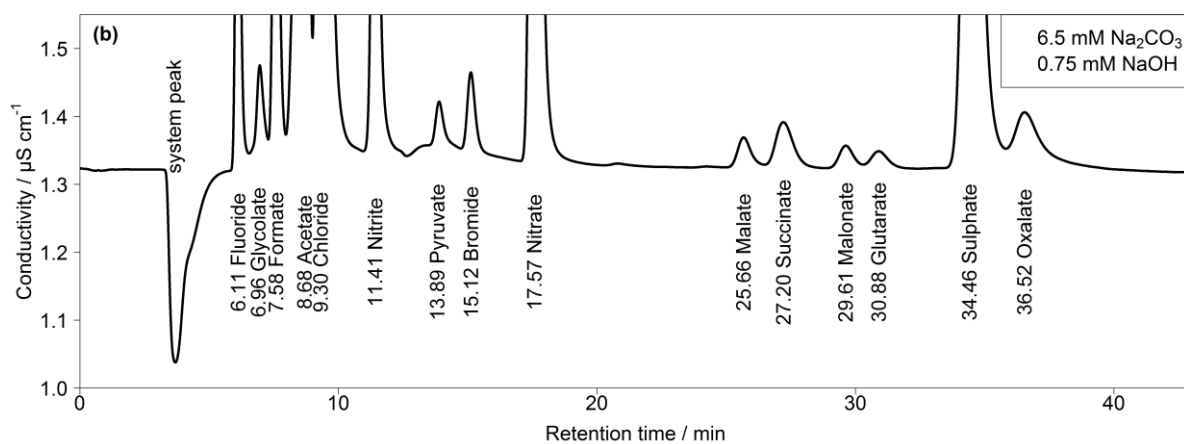
46

47 **Figure S7.** (a) Different eluent concentration of 6 mM Na_2CO_3 and 0.75 mM NaOH.
 48 Chromatogram of a standard solution with aqueous concentrations of $10 \mu\text{g l}^{-1}$ for Cl^- , NO_3^- ,
 49 SO_4^{2-} , $5 \mu\text{g l}^{-1}$ for NO_2^- and $1 \mu\text{g l}^{-1}$ for F^- , Br^- as well as all organic acids. Numbers in front of
 50 the ion names are the retention times. $T = 65 \text{ }^\circ\text{C}$ and eluent flow of 0.8 ml min^{-1} . (b) Zoom in
 51 of chromatogram in (a).

52



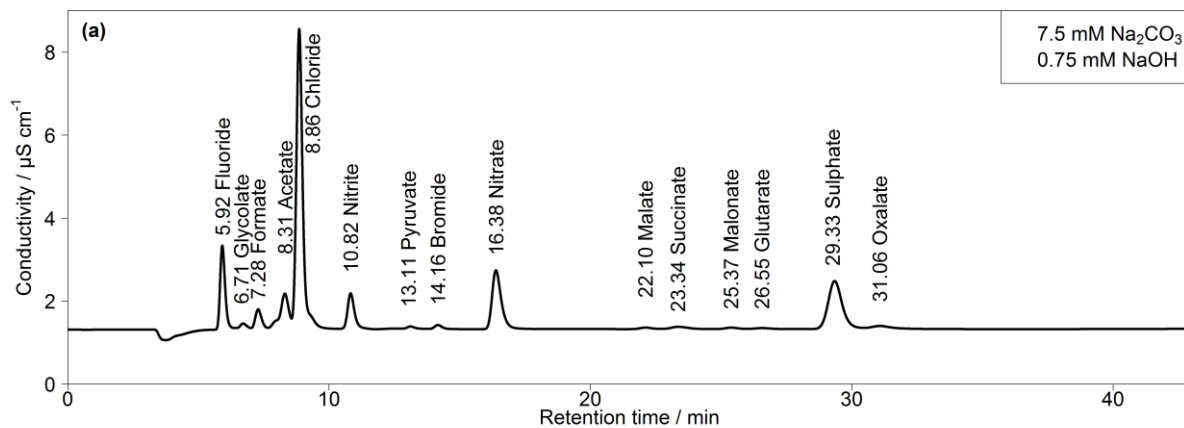
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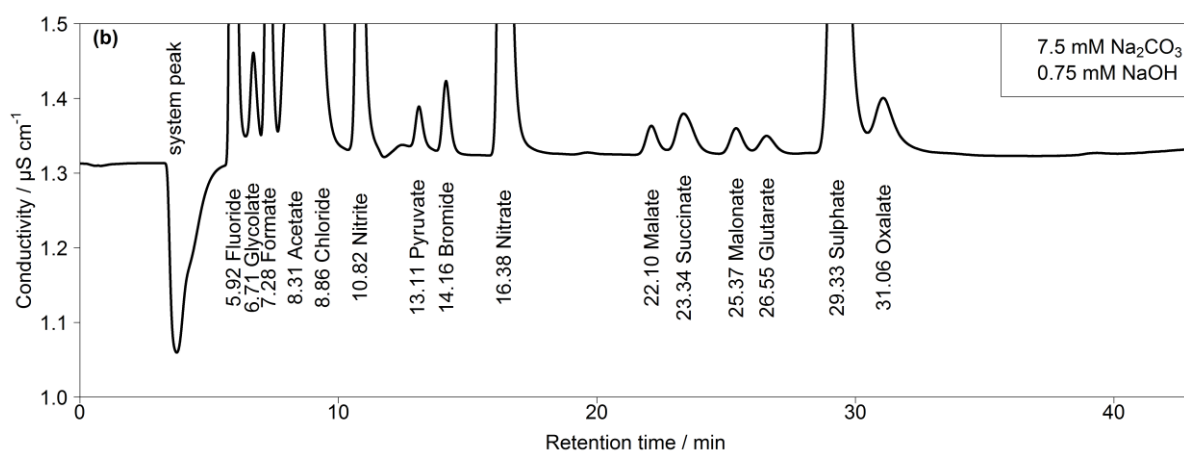
54

55 **Figure S8.** (a) Different eluent concentration of 6.5 mM Na_2CO_3 and 0.75 mM NaOH.
 56 Chromatogram of a standard solution with aqueous concentrations of $10 \mu\text{g l}^{-1}$ for Cl^- , NO_3^- ,
 57 SO_4^{2-} , $5 \mu\text{g l}^{-1}$ for NO_2^- and $1 \mu\text{g l}^{-1}$ for F^- , Br^- as well as all organic acids. Numbers in front of
 58 the ion names are the retention times. $T = 65 \text{ }^\circ\text{C}$ and eluent flow of 0.8 ml min^{-1} . (b) Zoom in
 59 of chromatogram in (a).

60



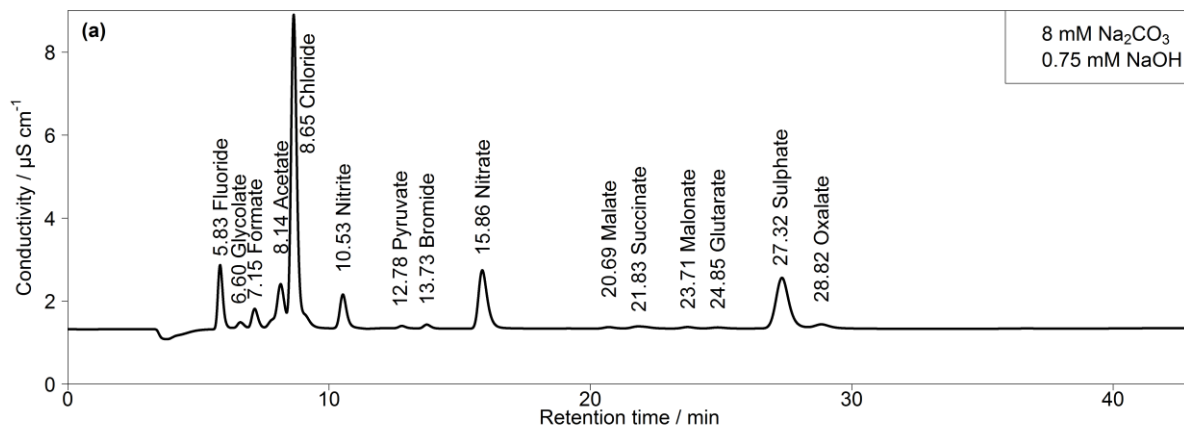
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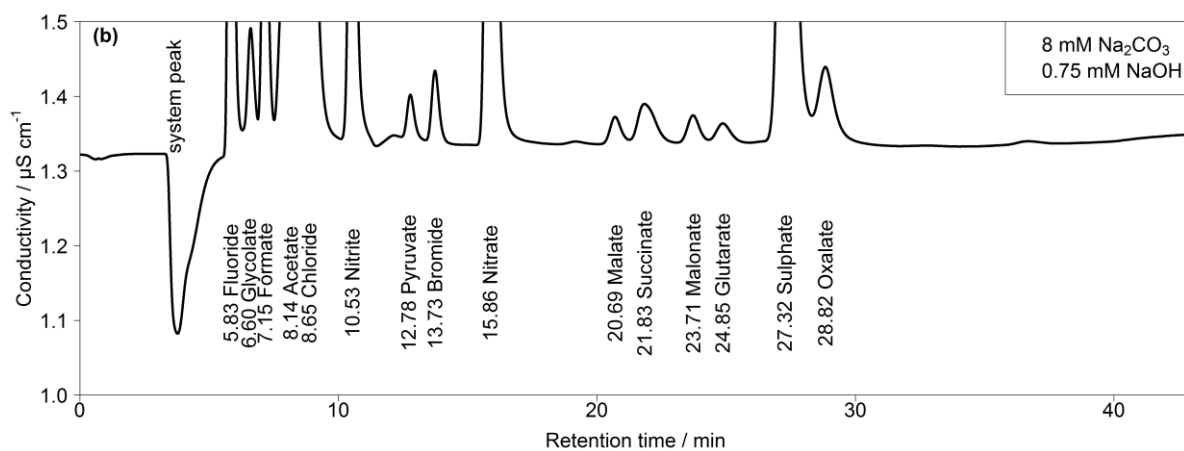
62

63 **Figure S9.** (a) Different eluent concentration of 7.5 mM Na₂CO₃ and 0.75 mM NaOH.
 64 Chromatogram of a standard solution with aqueous concentrations of 10 μg l⁻¹ for Cl⁻, NO₃⁻,
 65 SO₄²⁻, 5 μg l⁻¹ for NO₂⁻ and 1 μg l⁻¹ for F⁻, Br⁻ as well as all organic acids. Numbers in front of
 66 the ion names are the retention times. T = 65 °C and eluent flow of 0.8 ml min⁻¹. (b) Zoom in
 67 of chromatogram in (a).

68



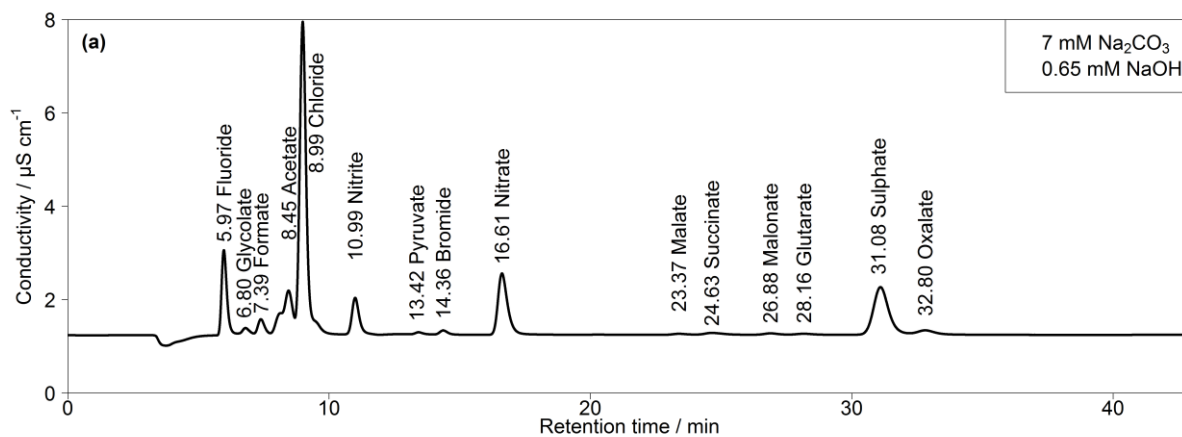
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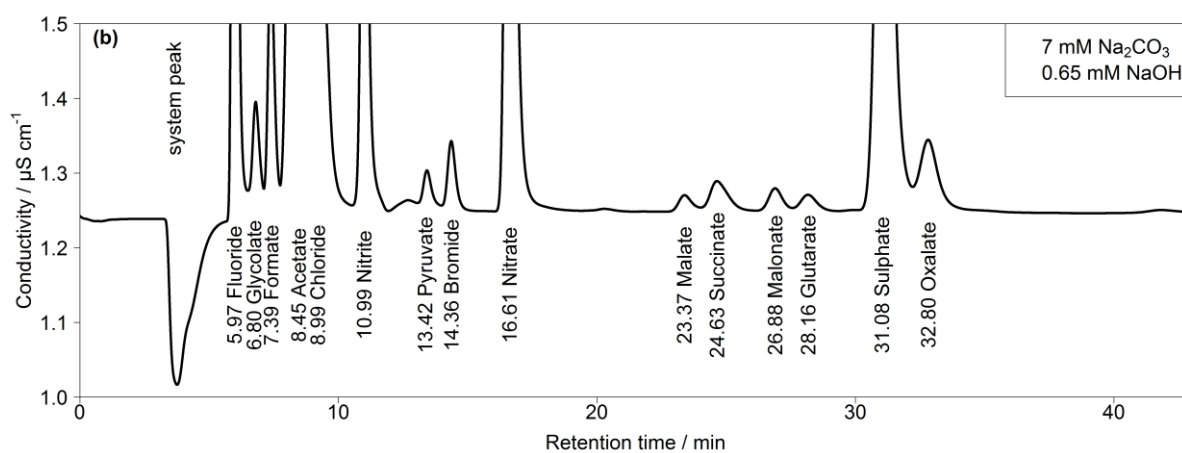
70

71 **Figure S10.** (a) Different eluent concentration of 8 mM Na_2CO_3 and 0.75 mM NaOH.
 72 Chromatogram of a standard solution with aqueous concentrations of $10 \mu\text{g l}^{-1}$ for Cl^- , NO_3^- ,
 73 SO_4^{2-} , $5 \mu\text{g l}^{-1}$ for NO_2^- and $1 \mu\text{g l}^{-1}$ for F^- , Br^- as well as all organic acids. Numbers in front of
 74 the ion names are the retention times. $T = 65 \text{ }^\circ\text{C}$ and eluent flow of 0.8 ml min^{-1} . (b) Zoom in
 75 of chromatogram in (a).

76



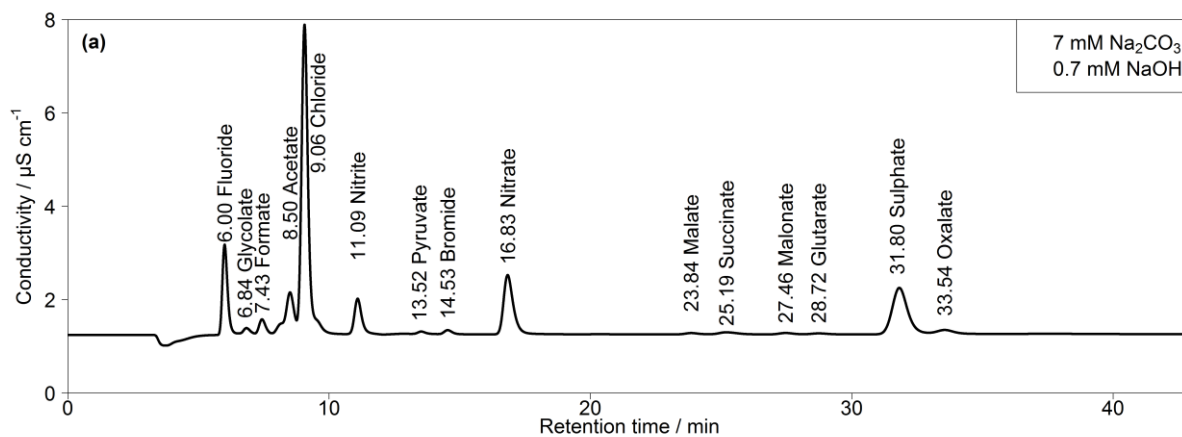
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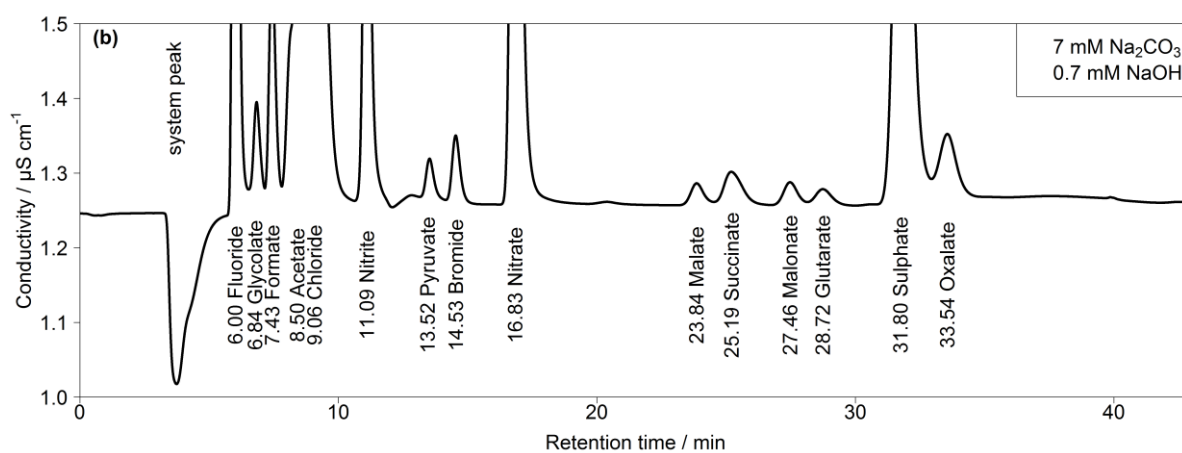
78

79 **Figure S11.** (a) Different eluent concentration of 7 mM Na₂CO₃ and 0.65 mM NaOH.
 80 Chromatogram of a standard solution with aqueous concentrations of 10 μg l⁻¹ for Cl⁻, NO₃⁻,
 81 SO₄²⁻, 5 μg l⁻¹ for NO₂⁻ and 1 μg l⁻¹ for F⁻, Br⁻ as well as all organic acids. Numbers in front of
 82 the ion names are the retention times. T = 65 °C and eluent flow of 0.8 ml min⁻¹. (b) Zoom in
 83 of chromatogram in (a).

84



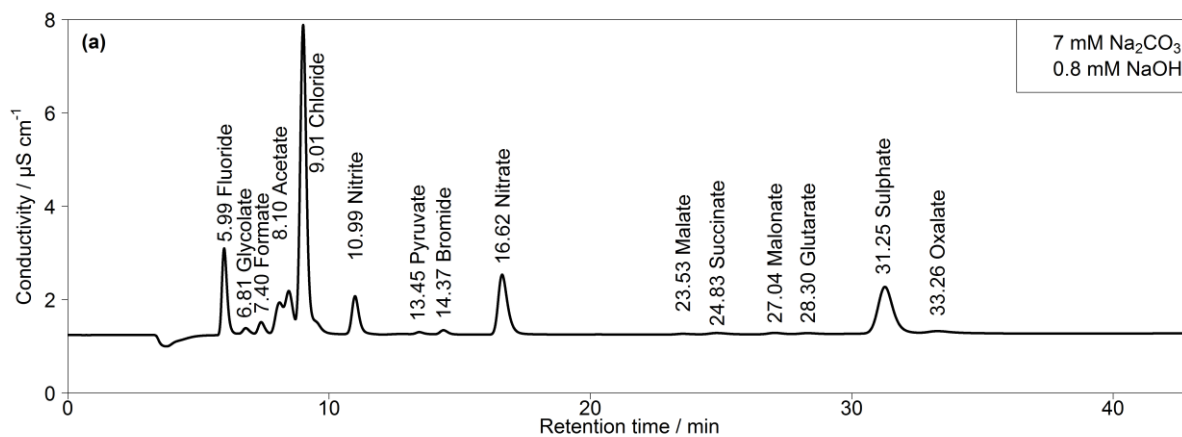
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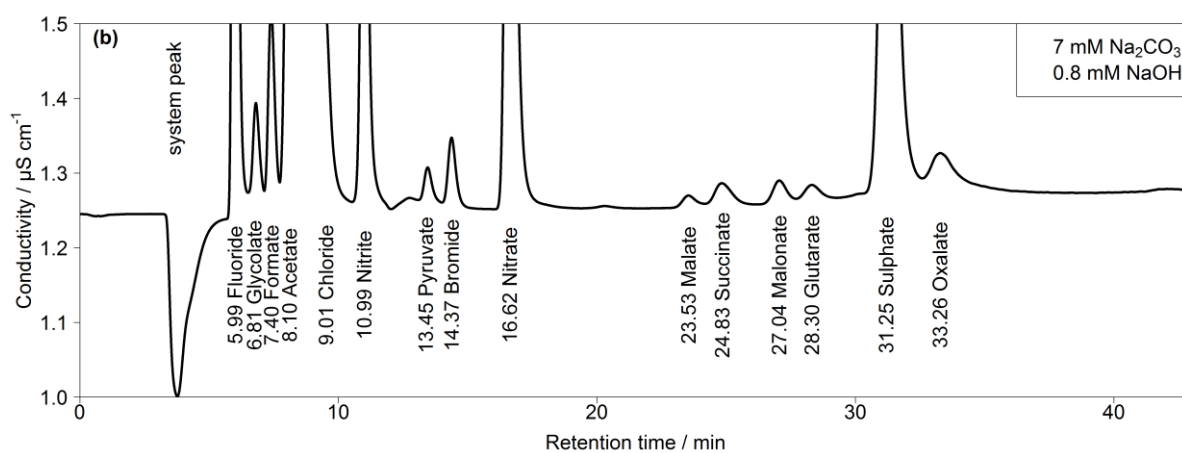
86

87 **Figure S12.** (a) Different eluent concentration of 7 mM Na_2CO_3 and 0.7 mM NaOH .
 88 Chromatogram of a standard solution with aqueous concentrations of $10 \mu\text{g l}^{-1}$ for Cl^- , NO_3^- ,
 89 SO_4^{2-} , $5 \mu\text{g l}^{-1}$ for NO_2^- and $1 \mu\text{g l}^{-1}$ for F^- , Br^- as well as all organic acids. Numbers in front of
 90 the ion names are the retention times. $T = 65 \text{ }^\circ\text{C}$ and eluent flow of 0.8 ml min^{-1} . (b) Zoom in
 91 of chromatogram in (a).

92



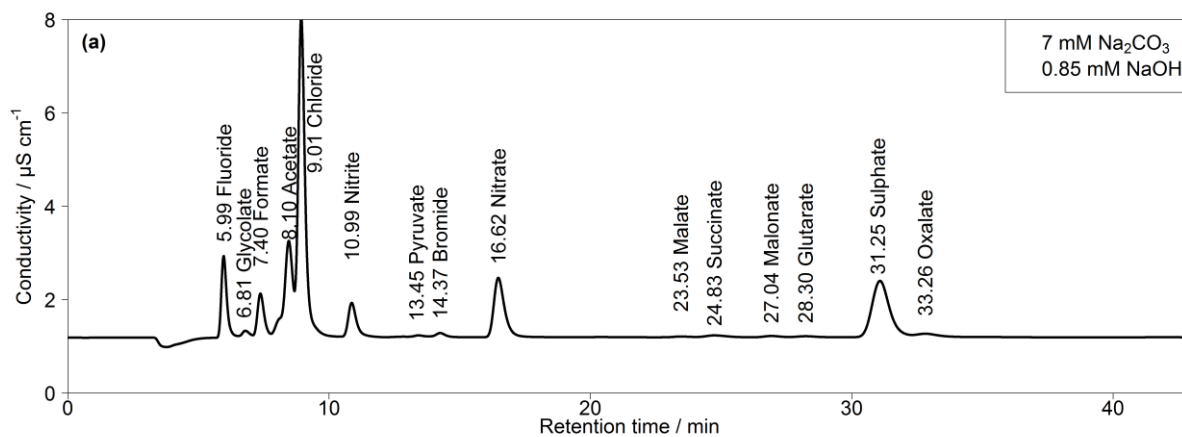
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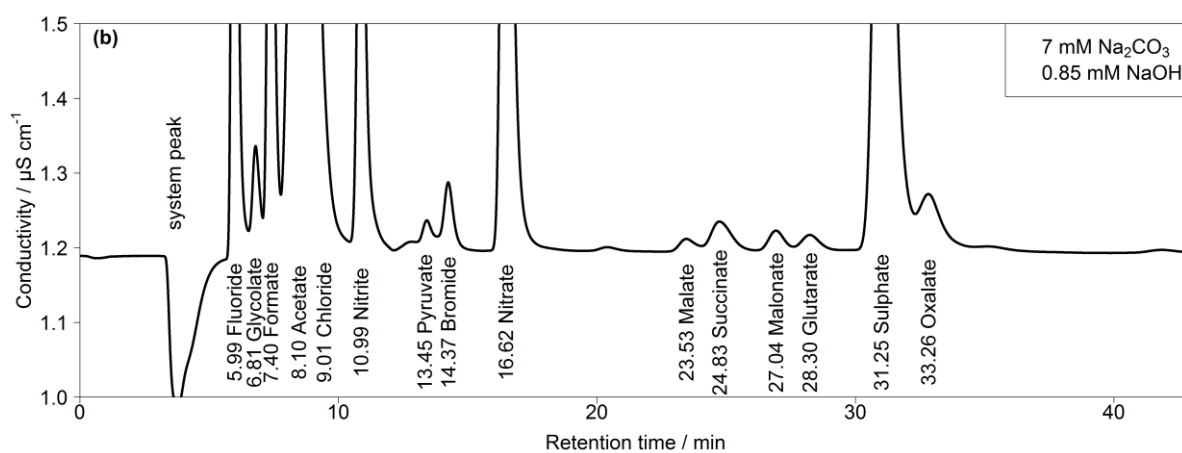
94

95 **Figure S13.** (a) Different eluent concentration of 7 mM Na₂CO₃ and 0.8 mM NaOH.
 96 Chromatogram of a standard solution with aqueous concentrations of 10 μg l⁻¹ for Cl⁻, NO₃⁻,
 97 SO₄²⁻, 5 μg l⁻¹ for NO₂⁻ and 1 μg l⁻¹ for F⁻, Br⁻ as well as all organic acids. Numbers in front of
 98 the ion names are the retention times. T = 65 °C and eluent flow of 0.8 ml min⁻¹. (b) Zoom in
 99 of chromatogram in (a).

100



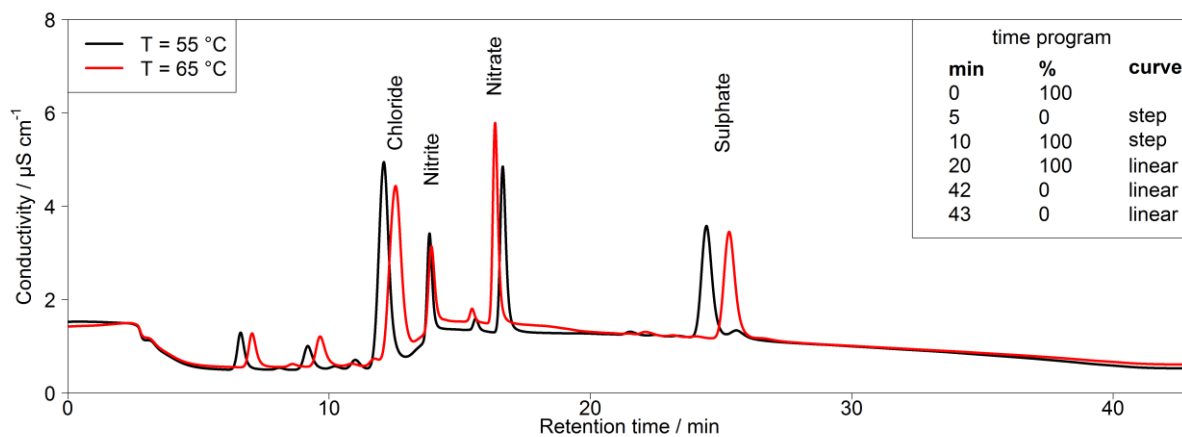
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102

103 **Figure S14.** (a) Different eluent concentration of 7 mM Na₂CO₃ and 0.85 mM NaOH.
 104 Chromatogram of a standard solution with aqueous concentrations of 10 μg l⁻¹ for Cl⁻, NO₃⁻,
 105 SO₄²⁻, 5 μg l⁻¹ for NO₂⁻ and 1 μg l⁻¹ for F⁻, Br⁻ as well as all organic acids. Numbers in front of
 106 the ion names are the retention times. T = 65 °C and eluent flow of 0.8 ml min⁻¹. (b) Zoom in
 107 of chromatogram in (a).

108

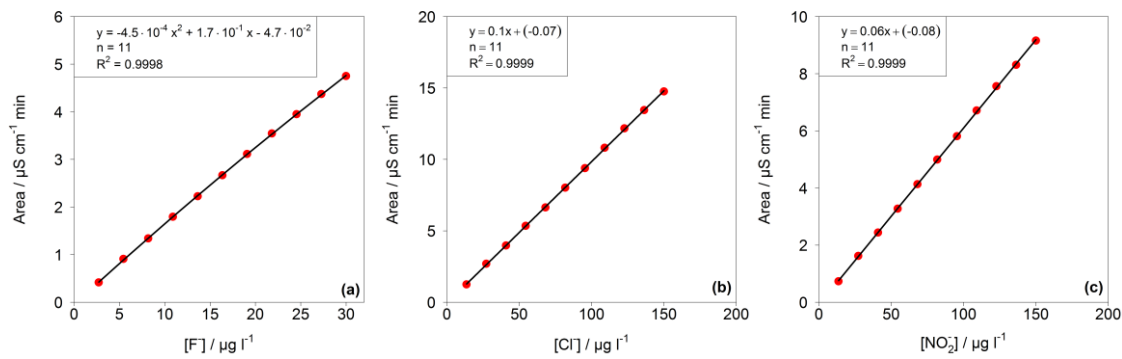


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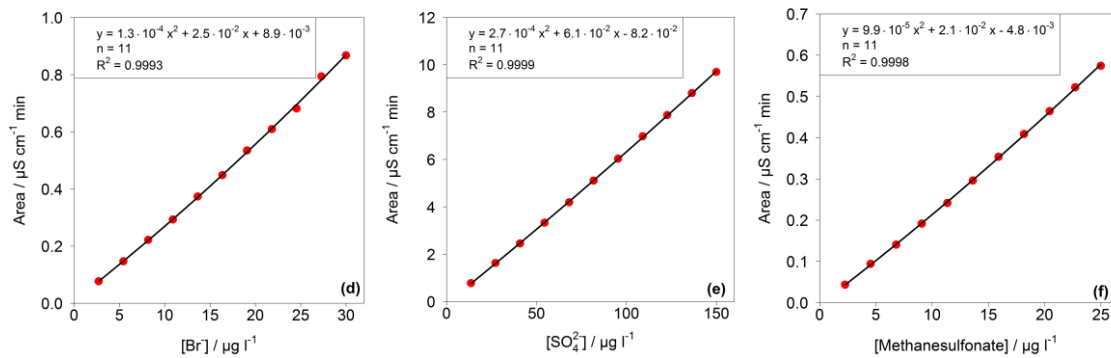
110 **Figure S15.** Temperature variation of the column oven for 55 °C (black) and 65 °C (red).
 111 Eluent concentration of 1-14 mM Na_2CO_3 and 0.75 mM NaOH. Chromatogram of a standard
 112 solution with aqueous concentrations of $50 \mu\text{g l}^{-1}$ for Cl^- , NO_3^- , SO_4^{2-} , $25 \mu\text{g l}^{-1}$ for NO_2^- and
 113 $3 \mu\text{g l}^{-1}$ for F^- , Br^- as well as all organic acids. Numbers in front of the ion names are the
 114 retention times. Eluent flow of 1.0 ml min^{-1} .

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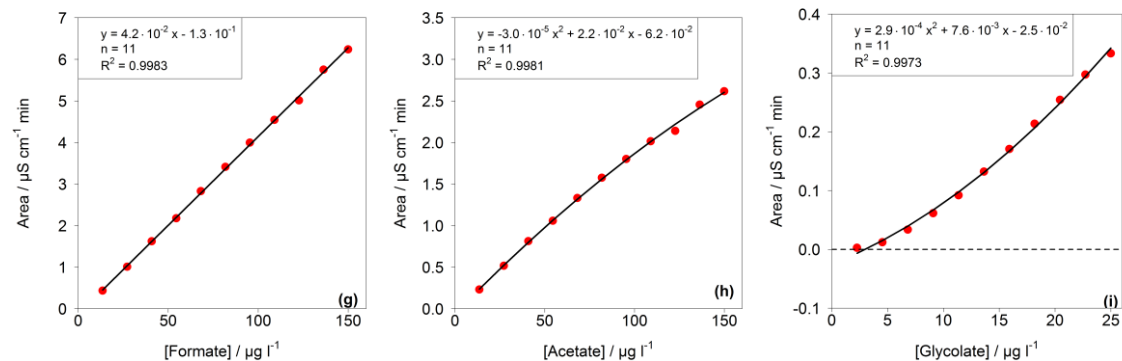
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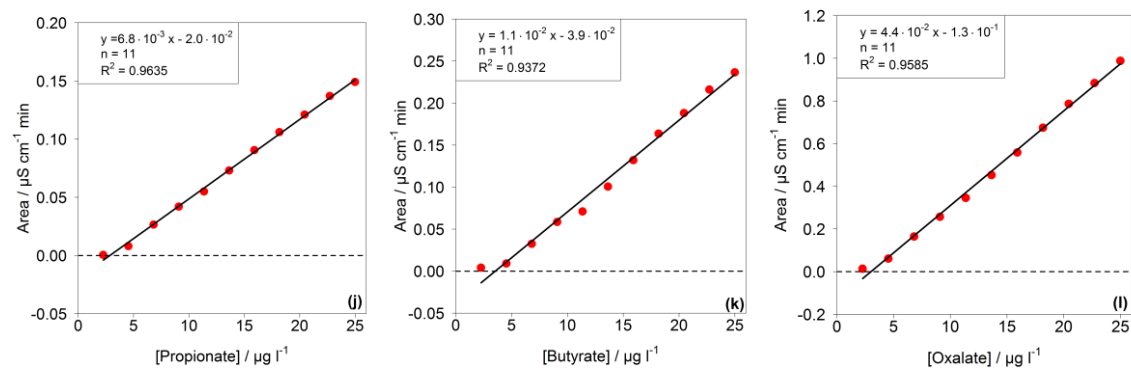
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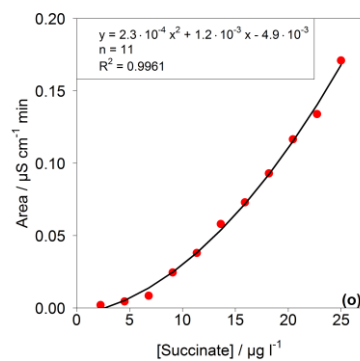
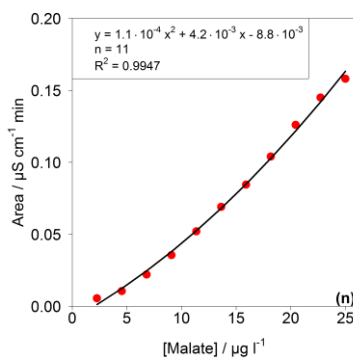
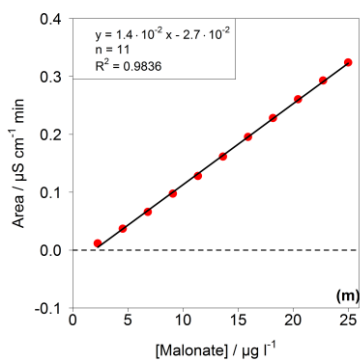


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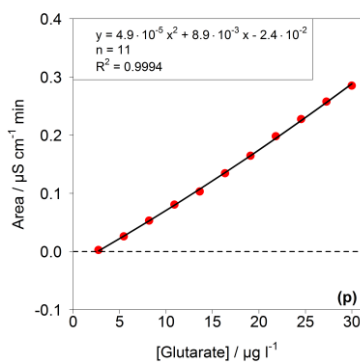


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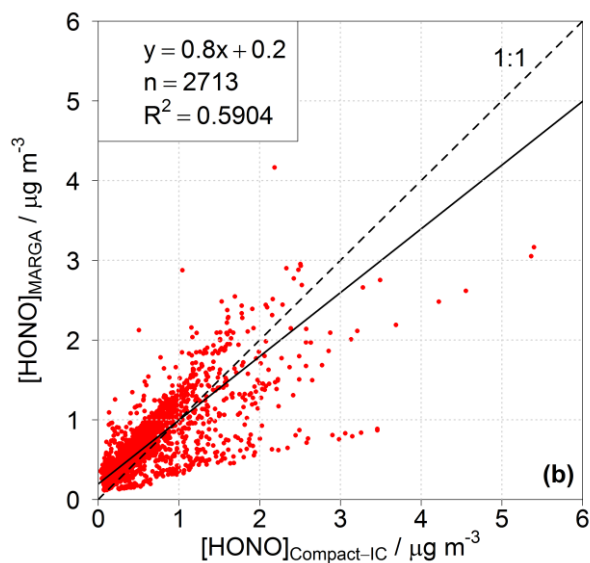
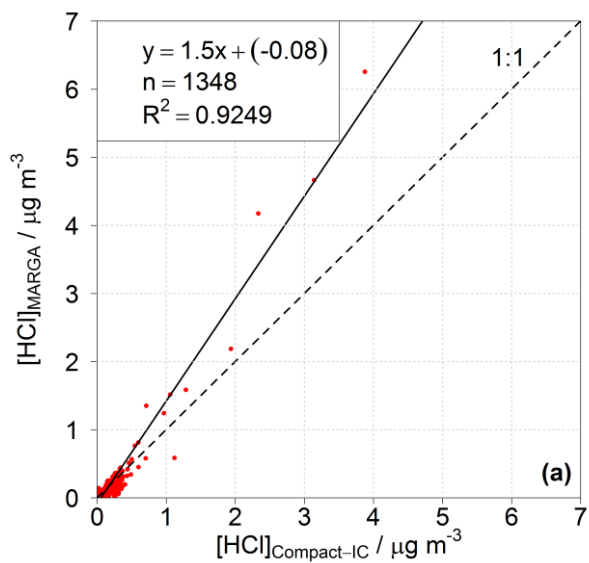
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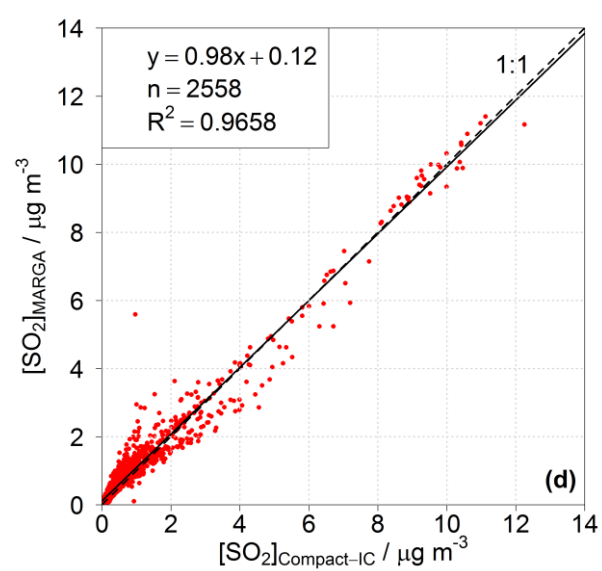
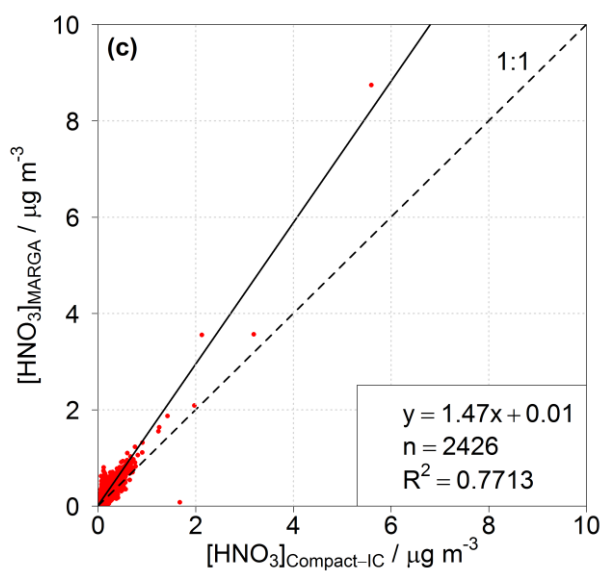
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122 **Figure S16.** Calibration functions of (a) F⁻, (b) Cl⁻, (c) NO₃⁻, (d) Br⁻, (e) SO₄²⁻, (f)
 123 methanesulfonate, (g) formate, (h) acetate, (i) glycolate, (j) propionate, (k) butyrate, (l) oxalate,
 124 (m) malonate, (n) malate, (o) succinate and (p) glutarate.

125

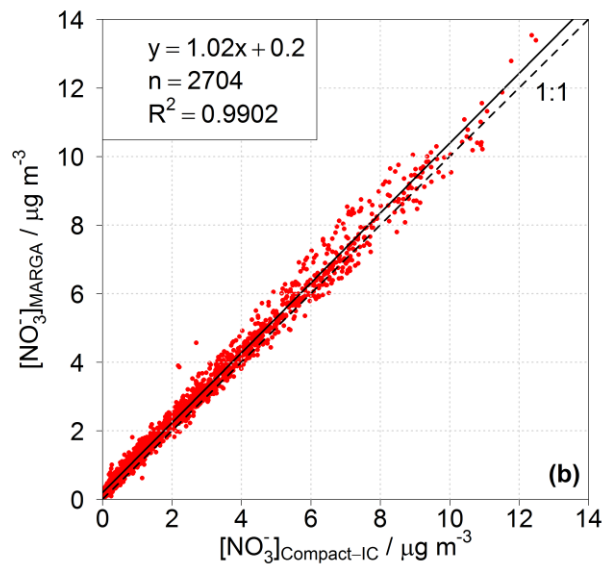
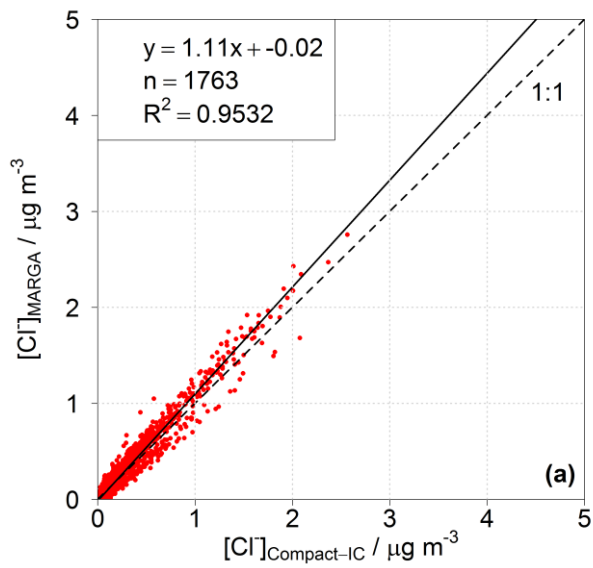


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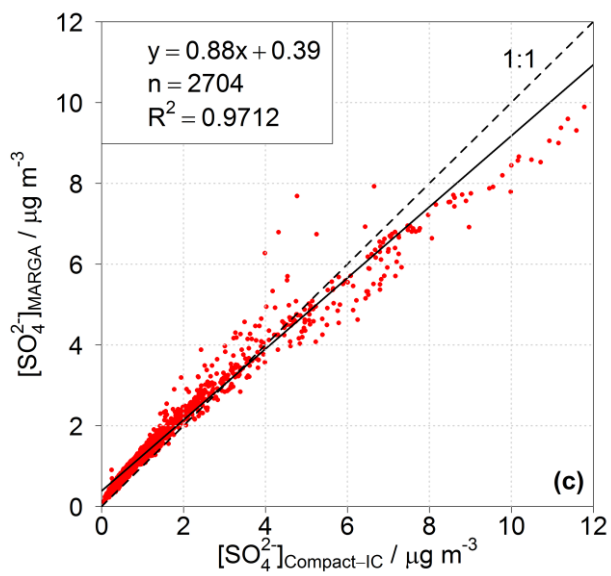


127

128 **Figure S17.** Scatter plots of (a) HCl, (b) HONO, (c) HNO₃ and (d) SO₂ for MARGA and
 129 Compact-IC measurements in Melpitz during one-year measurement campaign.



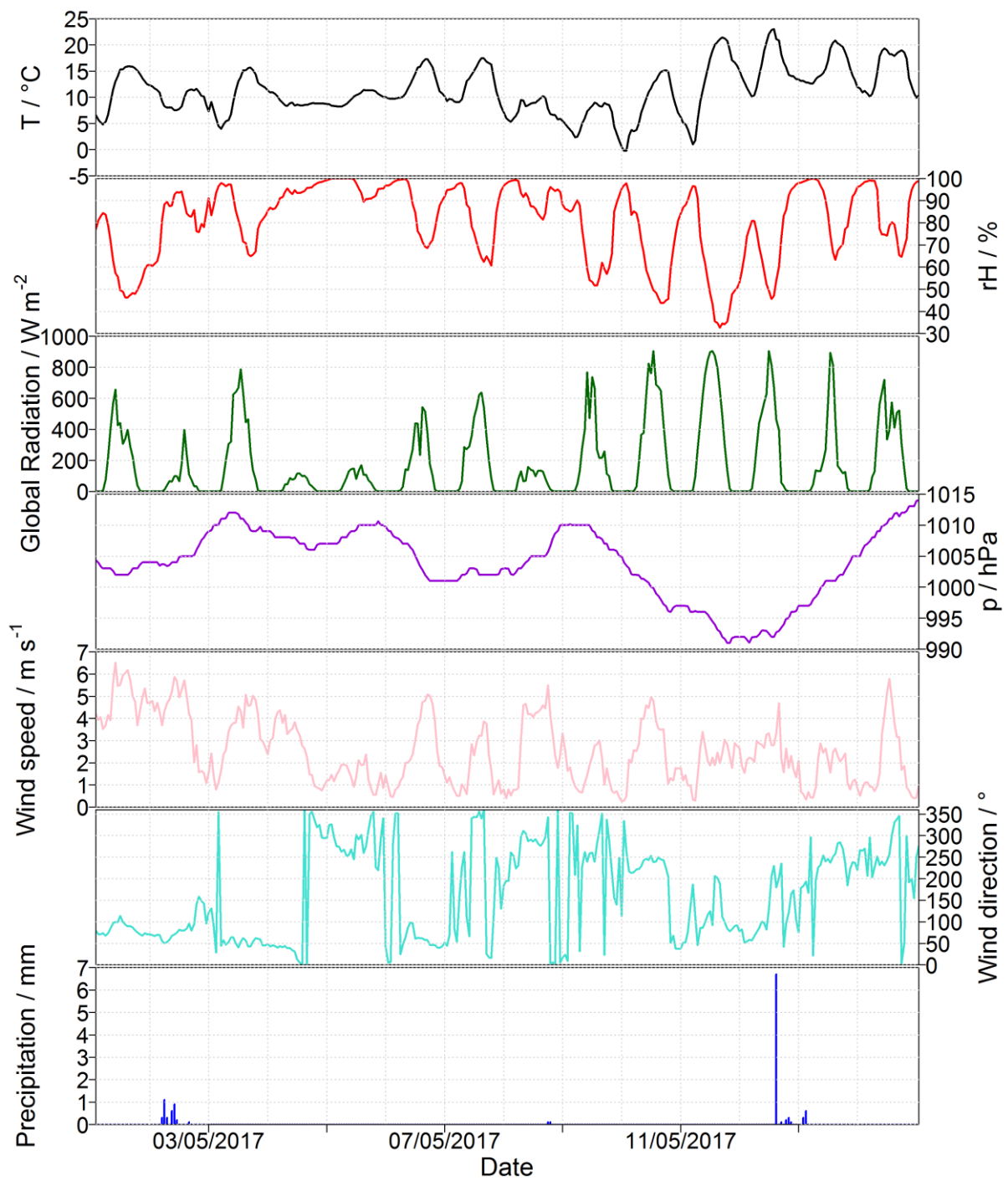
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131

132 **Figure S18.** Scatter plots of (a) Cl^- , (b) NO_3^- and (c) SO_4^{2-} for MARGA and Compact-IC
 133 measurements in Melpitz during the one-year measurement campaign.

134



135

136 **Figure S19.** Meteorological parameters during the example application from 1st May until 14th
 137 May 2017.

138

139 **Table S1.** Linearity test for the calibration and resulting test values (TV) for each ion.

Ion	TV	Linearity
F ⁻	56.9	quadratic
Cl ⁻	0.2	linear
NO ₂ ⁻	2.4	linear
Br ⁻	16.4	quadratic
NO ₃ ⁻	2.6	linear
SO ₄ ²⁻	26.4	quadratic
Methanesulfonate	36.8	quadratic
Formate	1.6	linear
Acetate	24.6	quadratic
Glycolate	34.9	quadratic
Propionate	0.3	linear
Butyrate	9.1	linear
Pyruvate	267.8	quadratic
Oxalate	2.8	linear
Malonate	10.7	linear
Malate	23.7	quadratic
Succinate	88.8	quadratic
Glutarate	16.8	quadratic

140

141