- Characterization of a commercial lower-cost medium precision NDIR sensor for atmospheric CO<sub>2</sub>
   monitoring in urban areas.
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- 12 Abstract:
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14 CO<sub>2</sub> emission estimates from urban areas can be obtained with a network of in-situ instruments 15 measuring atmospheric CO<sub>2</sub> combined with high-resolution (inverse) transport modeling. Because 16 the distribution of CO<sub>2</sub> emissions is highly heterogeneous in space and variable in time in urban 17 areas, gradients of atmospheric CO<sub>2</sub> (here, dry air mole fractions) need to be measured by 18 numerous instruments placed at multiple locations around and possibly within these urban areas. 19 This calls for the development of lower-cost medium precision sensors to allow a deployment at

20 required densities.

21 Medium precision is here set to be a random error (uncertainty) on hourly measurements of  $\pm 1$ ppm or less, a precision requirement based on previous studies of network design in urban areas. 22 23 Here we present tests of newly developed NDIR sensors manufactured by Senseair AB performed 24 in the laboratory and at actual field stations, the latter for  $CO_2$  dry air mole fractions in the Paris area. The lower-cost medium precision sensors are shown to be sensitive to atmospheric pressure 25 and temperature conditions. The sensors respond linearly to CO<sub>2</sub> when measuring calibration 26 tanks, but the regression slope between measured and assigned CO<sub>2</sub> differs between individual 27 sensors and changes with time. In addition to pressure and temperature variations, humidity 28 29 impacts the measurement of CO<sub>2</sub>, all of these factors resulting in systematic errors. In the field, 30 an empirical calibration strategy is proposed based on parallel measurements with the lower-cost medium precision sensors and a high-precision instrument cavity ring-down instrument during 6 31 months. The empirical calibration method consists of using a multivariable regression approach, 32 33 based on predictors of air temperature, pressure and humidity. This error model shows good 34 performances to explain the observed drifts of the lower-cost medium precision sensors on time 35 scales of up to 1-2 months when trained against 1-2 weeks of high-precision instrument time series. Residual errors are contained within the ±1 ppm target. showing the feasibility to use 36 networks of HPP3 instruments for urban CO<sub>2</sub> networks, Provided that they could be regularly 37 38 calibrated against one anchor reference high-precision instrument these sensors could thus provide  $CO_2$  (dry air) mole fraction data required as for top-down  $CO_2$  flux estimates. 39

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#### 41 **1. Introduction**

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43 Urban areas cover only a small portion (< 3 %) of the land surface but account for about 70% of 44 fossil fuel  $CO_2$  emissions (Liu et al. 2014, Seto et al. 2014). Uncertainties of fossil fuel  $CO_2$ 45 emissions from inventories based on statistics of fuel amounts and/or energy consumption are on 46 the order of 5% for OECD countries and up to 20% in other countries (Andres et al. 2014) but they

47 are larger in the case of cities (Breon et al. 2015, Wu et al. 2016). Further, in many cities of the

48 world, there are no emission inventories available. The need for more reliable information on 49 emissions and emission trends has prompted research projects seeking to provide estimates of GHG budget cities, power plants and industrial sites. These are often based on in situ 50 measurements made at surface stations (Staufer et al. 2016, Lauvaux et al. 2016, Verhulstet al. 51 52 2017), aircraft campaigns around emitting locations (Mays et al. 2009, Cambaliza et al. 2014) and satellite imagery (Broquet et al. 2017, Nassar et al. 2017). Although sampling strategies and 53 measurement accuracies differ between these approaches, the commonly used principle is to 54 measure atmospheric CO<sub>2</sub> dry air mole fraction gradients at stations between the upwind and 55 downwind vicinity of an emitting area and infer the emissions that are consistent with those CO<sub>2</sub> 56 57 gradients and their uncertainties, using an atmospheric transport model. This approach is known 58 as atmospheric CO<sub>2</sub> inversion or as a "top-down" estimate.

Inversion studies from Paris, France attempting to constrain CO<sub>2</sub> emissions from measurements of CO<sub>2</sub> dry air mole fractions at stations located around the city along the dominant wind direction, have pointed out that the fast mixing by the atmosphere and the complex structure of urban CO<sub>2</sub> emissions requires high resolution atmospheric transport models, and continuous measurements of the atmosphere to select gradients induced by emission plumes (Broquet et al. 2015, Wu et al. 2016) that can be captured at the scale of the model.

With the existing three stations, the CO<sub>2</sub> emissions from the Paris megacity could be retrieved with an accuracy of  $\approx 20\%$  on monthly budgets (Staufer et al. 2016). A denser network of stations would help to obtain more information on the spatial details of CO<sub>2</sub> emissions. A network design study by Wu et al. 2016 for the retrieval of CO<sub>2</sub> emissions per sector for the Paris Megacity has shown that with 10 stations measuring CO<sub>2</sub> with 1 ppm accuracy on hourly time-steps, the error of the annual emission budget could be reduced down to a 10% uncertainty. Wu et al. 2016

- furthermore found that for a more detailed separation of emissions into different sectors, more stations were needed, on the order of 70 stations to be able to separate road transport from
- residential CO<sub>2</sub> emissions. This inversion based on pseudo-data allowed estimating total CO<sub>2</sub>
- emissions with a better accuracy than 10% and emissions of most major source sectors (building,
- road energy) with an accuracy better than 20%. Another urban network design study over the San
   Francisco Bay area reached a similar conclusion, i.e. that in-situ CO<sub>2</sub> measurements from 34
   stations with 1 ppm accuracy at an hourly resolution could estimate weekly CO<sub>2</sub> emissions from
- the city area with less than 5 % error (Turner et al. 2016).

In the studies from Wu et al. 2016 and Turner et al. 2016, the additional number of atmospheric CO<sub>2</sub> measurement stations rather than the individual accuracy of each measurement helped to constrain emissions, provided that CO<sub>2</sub> observation errors have random errors of less than 1 ppm on hourly measurements, uncorrelated in time and in space between stations. Therefore, we will adopt here a 1 ppm uncertainty on hourly CO<sub>2</sub> data as the target performance for new urban lower-

84 cost medium precision  $CO_2$  sensors.

Today, the continuous CO<sub>2</sub> gas analyzers used for continental scale observing systems like ICOS 85 (https://www.esrl.noaa.gov/gmd/) 86 (https://www.icos-ri.eu/), NOAA or ECCC 87 (https://www.canada.ca/en/environment-climate-change.html) follow the WMO/GAW guidelines and are at least ten times more precise than our target of 1 ppm, but are also guite expensive. For 88 urban inversion-based flux estimates for Paris, Wu et al. 2016 found that the number of 89 90 instruments is more important than their individual precision. Furthermore, Turner et al. (2016) reported that weekly urban CO<sub>2</sub> fluxes in the Bay Area (California, USA) can be estimated at a 91 92 precision of 5% when deploying a dense network of sensors (ca. every 2km) with an assumed mismatch error of 1ppm. This underlines that significant expansion of urban networks is desirable 93 94 and could be achieved at an acceptable cost if low-cost sensors could be produced with the 95 specifications of 1 ppm random error (i.e. bias free long-term repeatability). . 96 Recently, inexpensive sensors, measuring trace gases, particulate matter, as well as traditional

96 Recently, mexpensive sensors, measuring trace gases, particulate matter, as well as traditional 97 meteorological variables, using various technologies and accuracy have become commercially

98 available. Evaluation and implementation of these sensors is quite promising (Eugster and Kling

2012, Holstius et al. 2014, Piedrahita et al. 2014, Young et al. 2014, Wang et al. 2015). With the advent of low cost mid-IR light sources and detectors, different non-dispersive infrared (NDIR)
CO<sub>2</sub> sensors have become commercially available and were tested for their suitability for CO<sub>2</sub> monitoring (e.g. Martin et al. 2017, Kunz et al. 2017) or for CO<sub>2</sub> in combination with air pollutants (e.g. Shusterman et al. 2016, Zimmerman et al. 2018).

104 In this study, we present the development and stability tests of a low-cost sensor (HPP3, SenseAir 105 AB, Sweden) to measure the mole fraction of  $CO_2$  of ambient air (Hummelgard et al. 2015). Throughout the manuscript we will use {CO<sub>2</sub>} to signify the mole fraction and/or dry air mole 106 107 fraction of CO<sub>2</sub> in air. To improve performance and eventually derive dry air mole fractions, 108 additional parameters are measured in ambient air and the sensor is integrated into a platform, 109 which we will refer to as instrument. . Then, the instrument linearity is evaluated against a suite of CO<sub>2</sub> reference gases with CO<sub>2</sub> dry air mole fractions from 330 to 1000 ppm. The instrument's 110 sensitivities to ambient air temperature, pressure and water vapor content are assessed in 111 laboratory experiments and climate chambers tests. The calibrated low-cost medium precision 112 (LCMP) instruments are then compared to highly precise CRDS instruments (G2401, Picarro Inc, 113 114 Santa Clara, USA).

Lastly, we present the time series of ambient air  $CO_2$  measurements in the Paris region. The time series are compared to measurements by co-located cavity ring down spectroscopy (CRDS) analyzers, and an empirical corrections and calibration scheme to the HPP3-based instrument are proposed based on measured  $CO_2$  dry mole air fractions and meteorological variables. These corrections and calibrations are established during a period of 1 or 2 weeks are used to estimate the drift of the HPP3-instrument on time scales of up to a month and a half.

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#### 122 **2. Sensor integration**

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#### 124 **2.1. HPP3 sensor**

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The HPP (High-Performance Platform) NDIR (Non-Dispersive InfraRed) CO<sub>2</sub> sensor from 126 SenseAir AB (Delsbo, Sweden) is a commercially available lower-cost system (Hummelgard et al. 127 2015). The main components of this sensor are an infrared source (lamp), a sample chamber (ca. 128 129 1 m optical path length), a light filter and an infrared detector. The gas in the sample chamber causes absorption of specific wavelengths (Hummelgard et al. 2015) according to the Beer-130 Lambert law, and the attenuation of light at these wavelengths is measured by a detector to 131 determine the gas mole fraction. The detector has an optical filter in front of it that eliminates all 132 light except the wavelength that the selected gas molecules can absorb. The HPP has a factory 133 134 pre-calibrated CO<sub>2</sub> measurement range of 0 to 1000 ppm. The HPP sensor itself uses ca. 0.6 W and requires an operating voltage of 12 V direct currentand has a life expectancy superior to 15 135 136 vears according to the manufacturer.

Three generation of HPP sensors were built by SenseAir AB (Delsbo, Sweden). In this manuscript we only report on the tests carried out on the latest generation (HPP3) being the most performant among the HPP sensors family. Previous HPP generations were used for more short-term airborne measurements, for example in the COCAP system (Kunz et al. 2017) and were found to have an accuracy of 1-1.2 ppm during short-term mobile campaigns. A number of technical improvements have been made for the new HPP3 generation described here:

- Simple interface through USB connection and the development of a new software made data transfer easier, quicker and more efficient
- Improved temperature stability due to 6 independent heaters dispatched inside the unit.

• To reduce long-term drift the sensor is equipped with new electronics and the IR sources were preconditioned prior to shipment.

- The improved second version of HPP3 (HPP3.2) sensors was equipped with a pressure 148 sensor (LPS331AP - ST Microelectronics, Switzerland) to allow real-time corrections. The 149 high resolution mode of the LPS331AP has a pressure range of 263 to 1277 hPa and a 150 Root Mean Square (RMS) of 0.02 hPa can be achieved with a low power consumption (i.e. 151 152 30 µA).
- The impact of leaks on the measurements are reduced since the third generation sensor 153 ٠ 154 works in a high pressure mode. A pump is thus needed upstream of the sensor inlet in order to create a high pressure in the measurement cell. 155
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157 Different sensors from two versions of HPP were tested and used in this study, that is, three sensors from a first version (HPP3.1) named S1.1, S1.2 and S1.3, and three others from the 158 159 second version (HPP3.2) named S2.1, S2.2 and S2.3. For the HPP3.1 sensors, an internal 160 pressure compensation does not exist, but the HPP3.2 series includes a pressure sensor together with a compensation algorithm, which normalizes measured CO<sub>2</sub> dry air mole fractions according 161 to ambient pressure (Gavnullin et al. 2016). 162

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#### 165 2.2. Portable integrated instrument

166 167 The HPP3 sensors were integrated into a custom-built portable unit, which we will refer to as 168 instrument. This instrument should be suitable to perform in-situ CO<sub>2</sub> measurements on ambient 169 air. The instrument is composed of the HPP3  $CO_2$  sensor, temperature (T) and relative humidity (RH) sensors. To be able to continuously flush the measurement cell a diaphragm micro pump 170 with a built-in potentiometer (GardnerDenverThomas, USA, Model 1410VD/1.5/E/BLDC/12V) was 171 added upstream of the HPP3's optical cell. Temperature and RH were measured at the exterior 172 of the optical cell were gas is released into the surrounding enclosure. . For these humidity and 173 174 temperature, a DHT22 sensor kit (Adafruit, USA) was added and connected through an I2C 175 interface. The accuracy of the sensor is ±2-5% RH and ±0.5°C. Its range is 0-100 %RH and -40 to +80°C, respectively. The response time for all sensors was less than one minute (which is the 176 time-step to which data was integrated). 177

178 A Raspberry Pi3 (RPi3) (Raspberry Pi Foundation, 2015) is used to collect the data of all sensors. The RPi3 is a small (85x56 mm<sup>2</sup>) single-bard computer running Rasbian OS, an open-source 179 GNU/Linux distribution. The HPP3 sensors were connected via USB.A 7" touch screen monitor is 180 181 connected via an adapter board which handles power and signal conversion. The package is powered by a switching power supply providing 12V, but can also be run on a 12V battery pack. 182 183 An image of the components of the portable instrument package is available in Figure 1.







#### 188 **3. Methods**

189 190 NDIR sensors are sensitive to IR light absorption by CO<sub>2</sub> molecules in the air contained in their optical cell, but the retrieval of  $CO_2$  dry air mole fraction to the desired uncertainty of 1 ppm is 191 192 made difficult by sensitivities to temperature, pressure and humidity. Therefore, these parameters 193 should be controlled as much as possible, and their sensitivities characterized, to correct and calibrate reported {CO<sub>2</sub>}. A series of tests were carried out to characterize the HPP3.1 (S1.1, S1.2, 194 195 S1.3) and HPP3.2 (S2.1, S2.2, S2.3) performances and sensitivities to {CO<sub>2</sub>}, T, P and RH. Firstly, temperature, pressure and {CO<sub>2</sub>} sensitivities were determined in laboratory experiments. Then, 196 field measurements were conducted with an accurate CRDS instrument (Picarro, USA, G2401) 197 measuring the same air as the HPP3 sensors. The CRDS short-term repeatability is estimated to 198 be below 0.02 ppm and the long-term repeatability to be below 0.03 ppm (Yver-Kwok et al. 2015). 199 Table 1 summarizes all laboratory tests and field test measurements, which are presented in this 200 201 section. 202

Name	Purpose	Location	Air measured	Parameter	Range of T (°C) and P (atm)	Range of [CO2] in ambient air (ppm)	Range of [CO2] in Cal. Cylinders (ppm)	Duratio n (days)	Sensors tested	S
PT1	Correlation between [CO2] and P,T	Laboratory (Saclay)	Calibration cylinders	Т, р	16-32 and 0.965-1.025	N/A	420 to 450	3	S1.1, S1.3	S1.2,
PT2	Correlation between [CO2] and P,T	PIT climate chamber (Guyancourt)	Calibration cylinders	Т, р	-2 to 35 and 0.75 to 0.95	N/A	420 to 450	5	S2.1, S2.3	S2,2,

DA1	Test calibration frequency	Laboratory (Saclay)	Calibration cylinders and dried ambient air	T,P, RH, {CO2} from CRDS	24 to 31 and 0.996 to 1.010	417 to 575	330 to 1000	48	S1.1, S1.2, S1.3
WA2-1	Test calibration frequency	Field station (Saclay)	Ambient air	Ambient T, p, RH, 25 air {CO2} from ar CRDS 0.1		389 to 508	N/A	45	S2.2, S2.3
WA2-2	Test calibration frequency	Field station (Jussieu)	Ambient air	T, p, RH, {CO2} from CRDS	29 to 31 and 0.993 to 1.021	393 to 521	N/A	60	S2.1

203 Table 1: Summary of all laboratory tests.

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#### 205 3.1. Laboratory tests

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207 All laboratory tests used the same fundamental setup shown in Figure 2 with only slight modifications. A diaphragm pump (KNF Lab, Germany, Model N86KN.18) was used to pump air 208 from either an ambient air line or calibration cylinders to a Nafion dryer (PermaPure, USA, MD-209 070 series) was used to eliminate  $H_2O$  traces in the gas line. A flow controller (Bronkhorst, France, 210 El-Flow series) was used to regulate the airflow distributed with a manifold to the HPP3 211 212 instruments at 500 mL min<sup>-1</sup> to ensure stable experimental conditions while a CRDS instrument could also be connected through a gas split to measure the same air. 213



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#### 3.1.1. Sensitivity to temperature and pressure variations 219

221 assess the linearity of the response of each sensor to  $\{CO_2\}$  for different pressure and То temperature conditions, two series of temperature and pressure sensitivity tests (PT1, PT2) were 222 223 realized in a closed chamber with controlled T and p for the HPP3.1 and HPP3.2 sensors No drier was necessary as dried air from high-pressure cylinders was used. The CRDS instrument (Picarro,
 G2401, serial number 2125) was not connected during these tests

In test PT1 (table1), three HPP3.1 sensors were put in a simple plastic chamber and exposed to pressure changes ranging from 977.8 to 1038.6 hPa, and temperature ranges of 16 to 32 °C, while measuring gas from a calibration cylinder. Pressure and temperature were measured by a highprecision pressure sensor (Keller, Germany, Series 33x, 0.2hPa and 0.05K precision).

230 In test PT2, to test wider ranges for pressure and temperature, that might be experienced during 231 field measurements, three HPP3.2 sensors were placed inside a dedicated temperatures and 232 pressure chamber at the Plateforme d'Integration et de Tests (PIT) at OVSQ Guyancourt, France 233 where a much larger range of T and p variations could be applied. During each T, p test, four calibration cylinders with dry air CO<sub>2</sub> dry air mole fractions from 420 to 450 ppm were measured 234 by all the HPP3.2 sensors for a period of approximately 120 hours for each cylinder. In the PIT 235 236 chamber, temperature was varied from -2 to 35 °C with a constant rate of change of 1 °C/h keeping 237 pressure constant at a value of 1013.25 hPa. During pressure tests the chamber pressure was 238 varied from 1013.25to 759.94 hPa with a decrement of 50.66 hPa, being regulated with a primary 239 pump, with temperature fixed at 15°C. 240

# **3.1.2.** Correction and calibration of CO<sub>2</sub> measurements for dry and wet air

These experiments were performed to evaluate the response of HPP3 sensors to  $\{CO_2\}$  changes in ambient air. Corrections were established to allow compensating for unintended instrument behavior and sensitivities, while calibrations are applied to translate the instruments readings to an official  $CO_2$  scale (here,  $XCO_2$  2007). Both steps are combined into one procedure. Two modes of operation for the HPP3 sensors have been tested i.e. using a dried or an undried gas stream, as those are two common modes of operation in greenhouse gas measurements in different local, regional and global networks (GAW report 242).

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#### 251 **Dry air experiments**

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Water vapor is known to interfere with  $CO_2$  measurements, in particular for NDIR sensors. It is thus important to determine the response of the sensors to  $\{CO_2\}$  under the best possible conditions, that is, dry air. The experimental setup shown in Figure 2 was used. In test DA1 (Table 2) different HPP3 sensors were flushed with the same dry ambient air, passed through a Nafion dryer. CRDS measurement were used to monitor and confirm that H<sub>2</sub>O was reduced to trace amounts, i.e. 0.05 ±0.05 % H<sub>2</sub>O. HPP3.1 sensors S1.1, S1.2, S1.3 were tested extensively during 45 days, and HPP3.2 sensors S2.1, S2.2 were tested during 12 days.

Additionally, for a period of 45 days during spring 2016, S1.1, S1.2 and S1.3 measurements of 260 dry ambient air in parallel with a co-deployed CRDS instrument (Picarro, USA, G2401) were 261 262 conducted at the Saclay field site (see section 3.3.1) There ambient air was pumped from a 263 sampling line fixed on the roof of the building (ca. 4 m above ground level) to flush the setup described in Fig. 2. Four dry-air calibration cylinders (330 ppm, 375 ppm, 445 ppm and 1000 ppm 264 265 of  $\{CO_2\}$  were measured once every 13 hours they were sampled successively each for 30 minutes (Figure 3). As the HPP3 responses can be slow and in order to remove memory effects, 266 only the last 15 minutes of each measurement period were used. 267

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# 269 Undried (wet) air experiments270

- As drying is impractical for some applications, we also measured the HPP3's sensitivities to water
- vapor in undried ambient air and calibration cylinders. If these sensitivities were stable over time,
- they could be used to correct reported  $\{CO_2\}$  for the H<sub>2</sub>O interference. For WA2-1 and WA2-2

tests, the Nafion dryer was removed from the setup. The only modification of the experiment was

the removal of the Nafion drier.



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**Figure 3:** (a)  $CO_2$  dry air mole fractions measured by S1.1 (blue) and the Picarro CRDS analyzer (black). (b) Calibrated dry air mole fractions of S1.1 (red) compared to the raw values (blue). (c)Four reference gases (assigned values are 367 ppm, 413 ppm, 487 ppm and 997 ppm of {CO<sub>2</sub>}), are used for the calibration. No saturation effects are observed within our CO<sub>2</sub> dry air mole fraction range.

## 282283 **3.2. Instrument correction and calibration procedure**

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In order to correct the reported  $\{CO_2\}$  we have to define a function that allows to correct for unintended instrument sensitivities i.e. to p, T, H<sub>2</sub>O as well as to correct  $\{CO_2\}$  measurements to an official scale should they show any offset or non-linear behavior.

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#### 289 Linearity of instrument response

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For dry air measurements in test DA1, a linear calibration curve was found to be appropriate. panel C of Figure 3 shows that the response of the HPP3 instruments to  $CO_2$  dry air mole fraction is linear ( $R^2 = 0.95$ ) from 330 to 1000 ppm. No saturation effects are observed within this  $CO_2$  dry air mole fraction range since residuals are included in the ±1 ppm range. Therefore, a linear response to { $CO_2$ } is assumed further on.

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#### 297 Multivariable correction and calibration

Due to the high correlation of air temperature and water vapor content, which were both found to be linear (see section 4), we suggest a multivariable regression method, which includes all environmental variables. Indeed, if variables are corrected one at a time, an overcorrection of one of the correlated variables may occur.

Multivariable regression is a generalization of linear regression by considering more than one variable. We used a multivariable linear regression of the form:

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305  $\{CO_2\}_{calibrated, corrected} = b + a_{CO2}\{CO_2\}_{HPP3} + a_pp + a_TT + a_wW + a_dd$  (1)

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 $\{CO_2\}_{calibrated, corrected}$  corresponds to the measured  $\{CO_2\}$  by the reference instrument (CRDS) calibrated on the WMO CO<sub>2</sub> X2007 scale.. *C* is the  $\{CO_2\}_{HPP3}$  reported by the HPP3 instrument, with additional factors to capture the influence of the pressure *p*, the temperature *T*, the water mixing ratio *W* (as calculated from our *T*, *p* and *RH* measurements), the baseline drift *d* and a baseline offset *b*. All instrument specific coefficients for the multivariable linear regression are determined using measurements of the parameters during several days.

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#### **3.3. Field tests with urban air measurements**

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316 To assess their real-world performance, we conducted additional tests for the HPP3 sensors 317 measuring ambient air at two field sites under typical conditions for urban air monitoring. After the 318 sensors were fully integrated into instruments as described in section 2.2. Three HPP3.1 319 instruments (S1.1, S1.2, S1.3) and two HPP3.2 instruments (S2.2, S2.3) were installed Saclay 320 field site (48.7120N, 2.1462E) to measure ambient air on top of the building roof. Saclay is located 321 20 kilometers south of the center of Paris in a low-urbanized area. In addition, one HPP3.2 instrument (S2.1) was installed to measure air at the Jussie field site on the Jussieu University 322 campus in the center of Paris (48.8464N, 2.3566E). 323 324

#### 325 3.3.1. Saclay field Site

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327 The sampling line, a 5 meter Dekabon tube with an inner tube diameter of 0.6 cm, was fixed on 328 the rooftop of the building at about 4 meters above the ground, which was connected to a setup 329 that was a copy of the laboratory tests. However, up to five HPP3 instruments were connected 330 and a pump of the same built as in the previous experiments was used to regulate the air-flow distributed with a manifold to the HPP3 instruments at 500 mL min<sup>-1</sup> to ensure stable experimental 331 conditions. The field site is equipped with a cooling/heating unit that was turned off most of the 332 time so that room temperature varied between 24 and 31 °C. During the test of HPP3.2 were 333 tested for 45 days. Four reference gas cylinders (330ppm, 375ppm, 445ppm and 1000 ppm of 334 CO<sub>2</sub>) were used and each HPP3 was flushed every 12 hours for 30 minutes per cylinder during 335 the dry air experiment. No calibration cylinders were used during the undried air experiment, since 336 the calibration was based on the co-located high precision measurement with the CRDS analyzer. 337 The mean dry air mole fraction of ambient CO<sub>2</sub> was 420 ppm and varied between 388 ppm and 338 575 ppm during dry air experiments and a mean of 409 ppm and variations between 389 ppm and 339 340 509 ppm were found during the undried air experiments.

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#### 342 3.3.2. Jussieu field site

The measurements were conducted at the OCAPI (Observatoire de la Composition de l'Air de Paris a l'IPSL) field station. The measurements from the HPP3.2 instrument (S2.1) in Jussieu were compared with those of a co-located CRDS analyzer (Picarro, USA, G2401). Two independent sampling lines (about 5 meter Dekabon tube with an inner tube of 0.6 cm) were used for the CRDS and the S2.1 instrument. The air-flow into S2.1 was regulated by the micro pump (see section 2.2) and set to 500 mL min<sup>-1</sup> using a potentiometer. At this location neither calibration cylinders nor a drying system were deployed for S2.1, but used to calibrate the CRDS. The measurement period was 60 days and the mean ambient  $CO_2$  dry air mole fraction was 410 ppm and minute averages varied between 393 ppm and 521 ppm. Room temperature varied between 28 and 31 °C during the observation period.

- 353
- 354 **4. Results**
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## 6 **4.1. Sensitivity to temperature and pressure variations using dried air**

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### 4.1.1. HPP3.1 instruments tested in the simple chamber (PT1)

Linear relationships are observed between reported  $CO_2$  dry air mole fractions and p, T (R<sup>2</sup> =0.99 with p and R<sup>2</sup> = 0.92 with T) in the simple chamber (see Figure 4 and 5). Due to the limitation of experimental conditions in these simple plastic chambers, only a narrow pressure range of 977.78 to 1033.52hPa and a temperature range of 16 °C to 32 °C could be tested for these instruments. Different slopes and intercepts are found for each instrument as reported in Table 2. This indicates that there is no single universal p and T calibration curve that could be determined for one instrument and used for others.

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#### 368 4.1.2. HPP3.2 instruments tested in the PIT chamber (PT2)

369 The PT2 tests results with pressure changes from 1013.25 to 759.94 hPa with an increment of 370 371 50.66 hPa are shown in Figure 4. The top panel of the Figure 4 shows the variations of  $CO_2$  dry air mole fractions due to p changes (from 0.0049 to 0.0177 ppm per hPa). Despite the built-in 372 pressure compensation algorithm developed for HPP3.2, reported {CO<sub>2</sub>} and p can still co-vary 373 374 with a positive (S2.1 and S2.2) or a negative (S2.3) correlation, indicating that an additional correction is required when aiming to achieve the best possible results (see also Figure S1. 375 Consequently, we applied a linear fit between  $\Delta$ {CO<sub>2</sub>} (differences between the assigned dry air 376 377 mole fraction in the cylinder and the dry air mole fraction reported by HPP3.2 instruments) and pressure (Figure 6). The slope and intercept obtained are then used to determine the offset due 378 379 to p variations that has to be added on  $\{CO_2\}$  reported by the HPP3.2 instruments. The corrected {CO<sub>2</sub>} values have a root mean square deviation from the assigned dry air mole fraction in the 380 381 calibration cylinder (428.6 ppm) of less than 0.02 ppm for all three HPP3.2 (see also Figure S2). 382

Figure 5 shows the effect of temperature variations in the PIT chamber going from -2 to 35 °C (see section 3.1) on reported  $CO_2$  dry air mole fraction of the HPP3.2 instruments. For the three HPP3.2, { $CO_2$ } is negatively correlated to T. As for the tests in the simple chamber with the HPP3.1 instruments, different linear T slopes and intercepts are observed for each HPP3.1 instrument (Figure 5) in the PIT chamber. After correction for temperature variations, we obtain corrected { $CO_2$ } values with a root mean square deviation which does not exceed 0.01 ppm from the assigned value of the cylinder (444 ppm) for the three HPP3.2 instruments (see also Figure S3).





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Figure 4: Linear relationship experimentally found between reported {CO<sub>2</sub>} and p for S1.1, S1.2,
 S1.3 (la) and for the instruments S2.1, S2.2, S2.3 (b). Note the different p range, going from 972.7
 to 1030hPa for the HPP3.1 instruments in the simple plastic chamber and 759.9 to 1013.25hPa

397 for the HPP3.2 instruments in the PIT chamber.



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**Figure 5:** Linear relationships between reported {CO<sub>2</sub>} for S1.1 S1.1, S1.2 and S1.3 (la) at temperature values going from 17 to 30 °C in plastic chamber, and for S2.1, S2.2 and S2.3 at temperature values going from 5 to 35 °C in the PIT chamber (b).

403 Table 2 summarizes the results of the pressure and temperature tests for all instruments. These tests results show a sensor-specific response to p and T. A large difference of reported {CO<sub>2</sub>} 404 sensitivity to pressure variations is observed between the two HPP3 versions. A sensitivity of 405 0.564 to 0.744 ppm/hPa is found for the HPP3.1 sensors, whereas this sensitivity ranges from -406 407 0.0045 up to 0.0174 ppm/hPa for the newer HPP3.2 sensors. The lower sensitivity among HPP3.2 408 prototypes is due to the pressure compensation algorithm, which is included in this model. Since 409 the pressure compensation algorithm does still not fully correct the reported {CO<sub>2</sub>} variations due to pressure changes, we found that it is necessary to apply a correction for pressure, and that this 410 correction should be sensor specific. The {CO<sub>2</sub>} sensitivity to temperature variations are found to 411 412 be in similar range for both sensor makes. Sensitivities of -0.3 to 0.1 ppm °C<sup>-1</sup> and -0.2 to -0.7 ppm °C<sup>-1</sup> are found for HPP3.1 and HPP3.2 instruments, respectively. 413

414

	Pressure		Temperature				
	Slope (ppm/hPa)	Intercept (ppm)	R <sup>2</sup>	Slope (ppm/°C)	Intercept (ppm)	R <sup>2</sup>	
S1.1	0.664±0.004	-297.7 ±4.3	0.94	-0.124 ±0.003	391.34 ±0.07	0.85	
S1.2	0.744 ±0.001	-363.3 ±1.1	0.95	-0.29 ±0.01	408.1 ±0.2	0.80	
S1.3	0.564±0.001	-189.5 ±1.4	0.94	0.107 ±0.004	381.2 ±0.1	0.63	
S2.1	0.0174 ±0.0002	394. ±0.2	0.95	-0.5854 ±0.0004	435.530 ±0.01	0.99	
S2.2	0.0164 ±0.0.0001	392.4 ±0.2	0.97	-0.716 ±0.001	427.31 ±0.02	0.99	
S2.3	-0.0045 ±0.0002	429.0 ±0.0	0.75	-0.2453 ±0.0004	442.16 ±0.01	0.99	

415 **Table 2:** Slopes and intercept calculated for  $CO_2$  correction due to temperature and pressure.

Sensor 1 to 3 are type HPP3.1, whereas Sensor 4 to 6 are HPP3.2.

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418 After applying our correction for temperature and pressure, no more correlations are observed 419 between corrected {CO<sub>2</sub>} and pressure and temperature. Corrected CO<sub>2</sub> mole fractions of HPP3.2 are stable and standard errors do not exceed 0.3 ppm and 0.2 ppm for pressure and temperature 420 corrections respectively, except for  $\{CO_2\}$  after temperature correction for S2.2 which reaches a 421 422 standard deviation (STD) of 0.5 ppm. However, we do not reach the same stability after pressure and temperature correction for HPP3.1 prototypes. Standard deviations of 0.9, 0.2 and 0.2 ppm 423 are calculated for S1.1, S1.2 and S1.3 respectively after pressure correction, and Standard 424 deviations of 1.3, 2.6 and 1.6 ppm are determined for S1.1, S1.2 and S1.3 respectively after 425 temperature corrections. These differences between the results of the two HPP3 versions can be 426 partly explained by the fact that HPP3.2 prototypes had the opportunity to be tested in a 427 sophisticated climatic chamber which respects precise temperature and pressure set points for 428

429 more longer-term measurements and in which only one of the two variables are modified one at a430 time.

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- 432 433

#### 4.2. Instrument calibration and stability during continuous measurements

Our instrument described in this study is intended for use in field campaign studies and longerterm monitoring, we assess their performance during continuous measurements. We also evaluate which calibration frequency is necessary to track the changes in the sensitivities to p and T found in section 4.1 and if the instruments can be calibrated when using an undried gas stream. Given that the instrument response to {CO<sub>2</sub>} is also affected by atmospheric water vapor, we present the results from dried and wet ambient air measurements separately.

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### 441 **4.2.1.** Measurements of dried ambient air (DA1)

- Four calibration cylinders were used in order to calibrate the three HPP3.1 instruments (see section 3.1). To assess the quality of this calibration, the mean and standard deviation (STD) of  $\Delta$ {CO<sub>2</sub>} (i.e. {CO<sub>2</sub>}<sub>HPP3</sub>-{CO<sub>2</sub>}<sub>CRDS</sub>) of 1-minute averaged data were calculated, and shown in Figure 6. Although calibration cylinders were measured each 12 hours, by ignoring some calibration data, we processed the time series to re-compute calibrated {CO<sub>2</sub>} assuming a range of different time
- intervals between two calibrations. The results shown in figure 6 are for calibrations intervals of
  0.5, 6, 12, 19, 25, 31, 38 and 45 days. Each point in this Figure corresponds to the values
  calculated for the instruments S1.1, S1.2, S1.3.
  We find that the 1 ppm repeatability threshold is pearly met when measuring dried air for calibration
- 451 We find that the 1 ppm repeatability threshold is nearly met when measuring dried air for calibration 452 intervals of 6 days. The STD  $\Delta$ {CO<sub>2</sub>} of the minute averages slowly increases with increasing calibration intervals but seems to stabilize between 3 and 4 ppm. We also see a marked difference 453 between the performances of each sensor: S1.1 shows the best performance, followed by S1.3 454 and S1.2. Besides an increased STD, we also see that the mean of  $\Delta$ {CO<sub>2</sub>} increases significantly 455 after not calibrating for 19 days. Surprisingly, one calibration each 45 days does not seem to 456 deteriorate the mean of  $\Delta$ {CO<sub>2</sub>} further. Infact, the mean  $\Delta$ {CO<sub>2</sub>} seems to decrease over longer 457 time periods. 458



460 **Figure 6:** STD (a) and mean (b) values of one minute average of mean  $\Delta$ {CO<sub>2</sub>}, during a 461 measurement period of 48 days depending of the calibration frequency.

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#### 464 **4.2.2. Saclay ambient air measurements (WA2-1)**

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During this test (section 3.3, Saclay field site), all atmospheric variables in wet ambient air which 466 affect the performance of the instruments, i.e. pressure, temperature, water vapor content and 467 {CO<sub>2</sub>}, were measured from July 20<sup>th</sup> until August 8<sup>th</sup>. Our previous measurements already 468 indicated that regular recalibration of the HPP3s is required because of sensitivities to T, p and 469 470 water vapor that are instrument-specific and time dependent. We call the period during which the six calibration coefficients of Eq. (1) are calculated by using the CRDS {CO<sub>2</sub>} time series, the 471 calibration period. Attempting to determine those calibration coefficients during a short calibration 472 period e.g. of one week, leads to high mean  $\Delta \{CO_2\}$ , as can be seen in Figure 7. A calibration 473 period of two weeks leads to significantly better results. We benchmark the instrument 474 475 performance for both minute averages, the instruments typical temporal resolution, and hourly averages, as those are widely used in modelling studies and data assimilation systems. 476

We also compared different calibration periods of the same length. As an example, considering a 45 days experiment, we chose three different calibration periods of successive 15 days. We also tested the approach of using the first and last weeks of a 45 days period to create a nonsuccessive two weeks calibration period.

481 Figure 7 (a,b) shows the STD and mean  $\Delta$ {CO<sub>2</sub>} values considering 3 calibration periods (C1, C2, 482 C3) of 15 days each. The regression coefficients of the multivariable model of Eq. (1) for C1, C2 483 and C3 are calculated using the first, second and third consecutive 15 days of the experimental 484 period. These coefficients are then used to predict corrected {CO<sub>2</sub>}<sub>HPP3 2</sub> for the three crossvalidation periods of 15, 30 and 45 days. Also, calibration coefficients (W1, W6) were calculated 485 486 using the first and sixth week of the 45 days period for calibration. Unsurprisingly, using C1 coefficients gives the best results for the first 15 days used for training, and lead a higher bias for 487 488 the last 15 days. Using C2 coefficients to correct adjacent 15 days from the calibration period gives comparable results (Considering the last calibration period, C3 coefficients show a mean 489 bias of -2.5 ppm when calibration is from the first 15 days. One reason that can explain this 490 491 behavior is the greater variability of CO<sub>2</sub> dry air mole fraction during the last 15 days of the experiment. The interquartile range of  $CO_2$  dry air mole fraction is 10, 15 and 25 ppm respectively 492 493 for the first, second and third period. The CO<sub>2</sub> dry air mole fraction correction is accomplished 494 mostly by correcting T, P, H<sub>2</sub>O and the instrument offset. A small variation of sensitivities may lead to a less appropriate correction for periods of smaller variability. Another reason for this difference 495 496 is the drift component of the correction in Eq. 1. The linear drift of the instrument also varies with time. One method to better correct for the slow linear drift of the instrument is to combine the first 497 498 and last week of the experiment into a calibration period instead of using two consecutive weeks. 499 Figure 8 shows corrected  $\Delta$ {CO<sub>2</sub>} of S2.2 during 45 days when using this approach. When using the first week (W1) and the last week (W6) for calibration, the instrument drift is not properly 500 corrected and a residual slope of 0.14 and 0.28 ppm/week is shown in the black (W1) and the red 501 502 (W6) curves of the figure, respectively. Nearly no drift (0.01 ppm/week) is observed when 503 considering both W1 and W6 for the training (blue curve). On Figure 10, magenta stars show STD  $\Delta$ {CO<sub>2</sub>} and mean  $\Delta$ {CO<sub>2</sub>} values of the whole 45 day time series considering both W1 and W6 504 505 as calibration periods. With this coefficient determination method, mean  $\Delta CO_2$  bias can be reduced to nearly 0 ppm. A comparison of corrected minute averages of  $\Delta$ {CO<sub>2</sub>} when using only one week 506 for calibration (week 1 or week6) or the combined first and last week (week1 and week 6) can be 507 508 seen in Figure S4. Finally, we should note that averaging the 1-minute data to hourly averages 509 can further improve STD  $\Delta$ {CO<sub>2</sub>} values up to 28%. As expected, mean values do not change for 510 hourly averages.



513

**Figure 7:** STD  $\triangle$ {CO<sub>2</sub>} (a) and mean  $\triangle$ {CO<sub>2</sub>} (b) values considering 3 calibration periods of 15 514 consecutive days for calibration each with C1, C2 and C3 correspond to first, second and third 15 515 516 consecutive days of measurements at the Saclay field site, respectively. W1W6 corresponds to using the first and sixth week as calibration period. Mean  $\Delta CO_2$  calculated for the four calibration 517 periods choices. STD  $\Delta$ {CO<sub>2</sub>} (c) and mean  $\Delta$ {CO<sub>2</sub>} (d) values considering 4 calibration periods of 518 15 consecutive days for calibration each with C1, C2, C3, C4 correspond to first, second, third and 519 fourth 15 consecutive days of measurements at the Jussieu field site, respectively. W1W6 520 521 corresponds to using the first and eight week as calibration period. Hourly and minute values are represented in full and empty symbols respectively. 522

524 Furthermore, we can investigate which of the six term multivariable linear regression is most 525 important here. The offset and dry air mole fraction dependent corrections terms (b and 526  $a_{cx}$ {CO<sub>2</sub>}<sub>HPP3</sub>) are the most significant corrections among all 5 parameters and allow reducing the mean  $\triangle CO_2$  from 45 ppm to 0 ppm (see Table 3 and Figure S3). The other 4 parameters (pressure, 527 temperature, water vapor and drift corrections) further reduce the difference between CRDS and 528 HPP3.2 reducing the STD  $\Delta$ {CO<sub>2</sub>} of minute averages from 1.03 ppm to 0.67 ppm. Here, the 529 temperature correction (d) and the water vapor correction (e) provide a correction of similar 530 magnitude, keeping the same STD and improving mean  $\Delta$ {CO<sub>2</sub>} only from 0.16 to 0.13 ppm. This 531 532 is understandable since temperature and water vapor are correlated for this type of measurement. 533

	Reported (raw)	Offset correction	Pressure correction	Temperature correction	RH correction	Drift correction
STD ∆{CO₂} (ppm)	1.11	1.03	1.00	0.97	0.97	0.67
Mean ∆{CO₂} (ppm)	45.33	1.0 10-3	8.6 10-4	0.16	0.13	-0.08

534 **Table 3**: STD and mean values of one minute average  $\Delta$ {CO<sub>2</sub>} data for each correction step. Note 535 that corrections are cumulative from left to right.

## 536537 4.2.3. Urban site of Jussieu (WA2-2)

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539 To assess the further performance of the HPP3.2 instruments, additonal wet ambient air 540 measurements at second field site in Jussieu were carried out for 60 consecutive days using instrument S2.1 alongside a CRDS. Figure 7 (c,d) shows STD  $\Delta$ {CO<sub>2</sub>} and mean  $\Delta$ {CO<sub>2</sub>} values 541 calculated with four calibration periods of 15 consecutive days each and one calibration 542 543 considering both first and last week of the experiment. Calibration coefficients for C1, C2, C3 and C4 are calculated considering calibration periods of first, second, third and fourth 15 consecutive 544 days of the experiment respectively. W1W8 coefficients are calculated considering week one (W1) 545 546 and week eight (W8) of the experiment. The results are qualitatively very similar to the 547 measurements at the Saclay field site and combing the first and last week as calibration period 548 also results allow achieving our target of STD  $\Delta$ {CO<sub>2</sub>} >1ppm.

549 550

#### 551 **5. Conclusion and perspective**

552

553 We integrated HPP3.1 and HPP3.2 NDIR sensors into a portable low-cost instrument with 554 additional sensors and internal data acquisition. The laboratory tests reveal a strong sensitivity of 555 reported  $CO_2$  dry air mole fractions to ambient air pressure for the HPP3.1 series and a significantly decreased, yet noticeable, sensitivity to pressure, for the upgraded HPP3.2 sensors 556 equipped with the built-in manufacturer p-correction. To achieve the targeted stability (long-term 557 repeatability) for urban observations of 1 ppm or better, instruments have to be corrected at regular 558 559 intervals against data from a reference instrument (here: CRDS) to account for their crosssensitivities to T, p, W (H<sub>2</sub>O mixing ratio) changes and electronic drift, unless those parameters 560 could be controlled externally in the future. We found that commercially available p, T and RH 561

562 sensors that are compatible with the chosen Raspberry Pi3 platform are sufficiently precise to use 563 these parameters as predictors of the linear equation use to calibrate each HPP3 instrument against the reference instrument, which was calibrated to the official WMO CO<sub>2</sub> X2007 CO<sub>2</sub> scale. 564 565 Two common modes of operation have been successfully tested i.e. using the HPP3 instruments 566 for either dried or undried ambient air measurements. Our results indicate that using a dried gas stream does not improve measurement precision or stability compared to an undried gas streams 567 568 provided that a multivariable regression model is used for calibration, which accounts for all crosssensitivities including to H<sub>2</sub>O mixing ratio changes. 569

570 We furthermore find that sensor specific corrections are required and they should be considered 571 time-dependent, e.g. by including a linear drift that only becomes more apparent for longer-term observations. Different calibration windows were tested for both the Saclay field site and Jussieu 572 field site ambient air measurements and their results evaluated against CRDS data that were not 573 574 used for calibration. Those sites exhibit the typical {CO<sub>2</sub>} levels in urban GHG monitoring networks 575 were future low-cost medium precision instruments could be deployed. Regular (six weekly) recalibrations are found to be appropriate to capture sensor drifts and changes in relevant cross-576 sensitivities, while not increasing the burden of performing calibration too often. A dedicated set 577 578 of calibration gases was not necessary if the low cost instrument was calibrated against {CO2} from a CRDS using the same air. Calibration periods of one week with parallel CRDS 579 measurements before and after a 45 day deployment was sufficient for the STD  $\Delta$ {CO<sub>2</sub>} data to 580 581 be within 1 ppm of CRDS during that period (with near zero bias, i.e.  $\Delta$ {CO<sub>2</sub>} << 1 ppm). This calibration approach can thus be an alternative to permanently deploying calibration gases for 582 583 each individual sensor...

The field tests at the Saclay and Jussieu station are being continued to see if the instrument performance deteriorates over its lifetime. Since the start of the test in 2015 until now multiple HPP3.1 sensors have been in use for without significant performance loss. Other research groups have also started integrating HPP sensors into their low-cost GHG monitoring strategy (e.g. Carbosense, www.nano-tera.ch/projects/491.php, last access March 11<sup>th</sup>, 2019).

589 Future improvements for the LCMP instruments will include the addition of batteries to allow their 590 transport to the central calibration lab without power cut as well as using them in field campaigns, 591 e.g. landfills when connected to solar panels or small wind turbines. During future tests at sites 592 without reference instruments, small pressurized gas containers (12I, minican, Linde Gas) will be 593 used to regularly inject target gas to track the performance during a deployment period.

594 The overall operational cost of the new calibration scheme using a central laboratory and rotating 595 the LCMP systems can also only be assessed after more extensive field deployment has been 596 performed.

597

#### 598 **Code and data availability**

599 The python scripts for the data collection from the HPP3 and DHT22 for Raspberry Pi-3 and the 600 data from the experiments described here are available from the corresponding authors upon 601 request.

602

### 603 Author contributions

The shared first authors EA and FRV conducted the measurements reported in this study and planned the work with support of PC. AB and BG supported the development of the data logging routine and processing scripts. OL and MR supported measurements by providing access to the ICOS test laboratory equipment and supplied reference gases. The initial draft of the manuscript was developed by EA, FRV and PC. All authors have updated, contributed and edited the manuscript.

611

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