

1 **Development of a balloon-borne instrument**
2 **for CO₂ vertical profile observations in the troposphere**

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24 **Abstract**

25 A novel, practical observation system for measuring tropospheric carbon dioxide (CO₂)
26 concentrations using a non-dispersive infrared analyzer carried by a small helium-filled balloon (CO₂
27 sonde), has been developed for the first time. Vertical profiles of atmospheric CO₂ can be measured
28 with a 240-400 m altitude resolution through regular onboard calibrations using two different CO₂
29 standard gases. The standard deviations (1σ) of the measured mole fractions in the laboratory
30 experiments using a vacuum chamber at a temperature of 298 K were approximately 0.6 ppm at 1010
31 hPa and 1.2 ppm at 250 hPa. Two CO₂ vertical profile data obtained using the CO₂ sondes, which were
32 launched on January 31st and February 3rd, 2011 at Moriya, were compared with the chartered aircraft
33 data on the same days and the commercial aircraft data obtained by the Comprehensive Observation
34 Network for TRace gases by Airliner (COTRAIL) program on the same day (January 31rd) and one
35 day before (February 2nd). The difference between the CO₂ sonde data and these four sets of *in-situ*
36 aircraft data (over the range of each balloon altitude ± 100 m) up to the altitude of 7 km was 0.6 ± 1.2
37 ppm (average $\pm 1\sigma$). In field experiments, the CO₂ sonde detected an increase in CO₂ concentration in
38 an urban area and a decrease in a forested area near the surface. The CO₂ sonde was shown to be a
39 useful instrument for observing and monitoring the vertical profiles of CO₂ concentration in the
40 troposphere.

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42

43 **1. Introduction**

44 Atmospheric carbon dioxide (CO₂) is one of the most important anthropogenic greenhouse gases
45 for global warming. Certain human activities, such as fossil fuel combustion, cement production, and
46 deforestation are the major cause of atmospheric CO₂, making the global average concentration of
47 atmospheric CO₂ to increase from 280 ppm before the Industrial Revolution to 400.0 ppm in 2015
48 (World Meteorological Organization, WMO 2016). Over the last 10 years, the average rate of
49 atmospheric CO₂ increase is measured at 2.21 ppm yr⁻¹ (WMO 2016). Atmospheric CO₂ is measured
50 by ground-based stations and ships using the flask sampling and continuous instrument methods such
51 as non-dispersive infrared absorption (NDIR) (Tanaka et al. 1983, Hodgkinson et al. 2013) and cavity
52 ring-down spectroscopy (CRDS) (Winderlich et al. 2010). A network of ground-based Fourier
53 transforms spectrometers (FTS), that record the direct solar spectra in the near-infrared spectral region
54 (Total Carbon Column Observing Network, TCCON), is used to observe the column-averaged mole
55 fraction of CO₂ in dry air (total column XCO₂) (Wunch et al. 2011). These observations have provided
56 extensive information, regarding the distribution and temporal variation of CO₂ in the atmosphere
57 (Pales and Keeling, 1965; Conway et al. 1988; Komhyr et al. 1989; Tans et al. 1989; Conway et al.
58 1994). Moreover, atmospheric CO₂ measurements data are useful for estimating CO₂ fluxes at the
59 surface through inverse modeling (Gurney et al. 2004; Baker et al. 2006). Due to the limited number
60 of observation sites and the limitations of their altitudinal range, a large degree of uncertainty in the
61 current estimates of the regional CO₂ sources and sinks is noted (Gurney et al. 2002). More
62 atmospheric CO₂ measurements are needed to reduce the uncertainties in CO₂ fluxes estimation using
63 an inverse modeling.

64 To address the issues with insufficient CO₂ observational data, satellite remote sensing techniques
65 have been used to investigate the CO₂ distribution on a global scale (Chédin et al. 2002; Crevoisier et
66 al. 2004; Dils et al. 2006). The Greenhouse Gases Observing SATellite (GOSAT), which measures the
67 short wavelength infrared (SWIR) spectra of sunlight reflected by the earth's surface with a Fourier

68 transform spectrometer and obtains the total column XCO₂, has been in operation since early 2009
69 (Yokota et al. 2009; Yoshida et al. 2011; Morino et al. 2011). Since 2014, the Orbiting Carbon
70 Observatory-2 (OCO-2) satellite has also measured the IR spectra of the surface reflected sunlight
71 with a diffraction grating spectrometer and obtains total column XCO₂ (Eldering et al. 2017). However,
72 these satellite observations provide only nadir total column XCO₂, and do not measure the vertical
73 distributions of CO₂ concentrations, as the observed spectra of the surface-reflected sunlight do not
74 provide enough information to determine the vertical distributions. Furthermore, the satellites overpass
75 a specific earth-based target once several days only at about noon in the solar time because of their
76 sun-synchronous orbits.

77 The altitude distributions of CO₂ concentrations have been measured using other techniques. For
78 instance, tall towers measure vertical profiles of CO₂ near the ground (Bakwin et al. 1992, Inoue and
79 Matsueda, 2001; Andrews et al. 2014). CO₂ vertical profiles up to 10 km near the airports have been
80 observed by the equipment installed by the commercial airlines, such as the Comprehensive
81 Observation Network for TRace gases by Airliner (CONTRAIL program) (Machida et al. 2008;
82 Matsueda et al. 2008). Measurements by equipment installed on chartered aircrafts have also been
83 undertaken, which include the High-performance Instrumented Airborne Platform for Environmental
84 Research (HIAPER), Pole-to-Pole Observations (HIPPO) program up to 14 km in the altitude spanning
85 the Pacific from 85° N to 67° S (Wofsy et al. 2011), the NIES/JAXA (National Institute of
86 Environmental Studies and Japan Aerospace eXploration Agency) program at an altitude from 2 to 7
87 km (Tanaka et al. 2012), and the NOAA/ESRL Global Greenhouse Gas Reference Network Aircraft
88 Program (Sweeney et al. 2015). Although these aircraft measurements provided the vertical profiles
89 of CO₂ concentrations, vertical profile measurements using the commercial airlines are limited around
90 the large airports and frequency of the measurements using chartered airplane is often limited by their
91 relatively high cost. The continuation and expansion of airborne measurement programs for CO₂ and
92 related tracers are expected to enhance the estimation of the global carbon cycling greatly (Stephens

93 et al., 2007).

94 Atmospheric CO₂ observations using balloons, to select specific locations unless prohibited or
95 restricted by aircraft flight paths, are useful for solving the issues associated with the sparseness of
96 CO₂ vertical data. Balloon-borne observations of stratospheric CO₂ are previously conducted by other
97 studies. For instance, stratospheric air sampling was conducted using cryogenic sampler onboard
98 balloons once a year from 1985 to 1995 over the northern part of Japan (Nakazawa et al. 1995).
99 Balloon-borne near-infrared tunable diode laser spectrometers have been developed to provide in situ
100 data for CO₂ in the stratospheric atmosphere (Durry et al. 2004; Joly et al. 2007, Ghysels et al. 2012).
101 Furthermore, two in situ CO₂ analyzers adopting the NDIR technique, using a modified commercial
102 detector for stratospheric measurements, have been developed for deployment on the NASA ER-2
103 aircraft and on a balloon (Daube et al. 2002). These balloon borne instruments described above were
104 specially designed to measure CO₂ concentrations in the stratosphere.

105 Observation of the CO₂ vertical distribution in the troposphere is essential because the uncertainties
106 in the estimated fluxes, using the inverse method, can be attributed to the inaccurate representations of
107 the atmospheric processes in transport models. Misrepresentation of vertical mixing by the transport
108 models, particularly inside of the boundary layer, which is the layer closest to the ground where CO₂
109 is taken up and released, is one of the dominant causes of the uncertainty in CO₂ flux estimation
110 (Stephens et al. 2007; Ahmadov et al. 2009). Recently, the observation of tropospheric CO₂ was
111 conducted, using a lightweight unmanned aerial vehicle, such as a kite plane, with a commercial NDIR
112 instrument. CO₂ profiles were observed in and above the planetary boundary layer up to 2 km to
113 investigate the temporal and spatial variations of CO₂ (Watai et al. 2006). A passive air sampling system
114 for atmospheric CO₂ measurements, using a 150 m long stainless-steel tube called an AirCore was
115 developed (Karion et al. 2010). The AirCore mounted on an airplane or a balloon ascends with
116 evacuating inside of the tube to a high altitude of 30 km at flight maximum, then, collecting ambient air
117 by pressure changes along a decrease in altitude. The sampled air in the tube is analyzed with the

118 precision of 0.07 ppm for CO₂ indicated as one standard deviation in the laboratory and the vertical
119 profile of CO₂ is obtained.

120 In the present study, we have developed a practical CO₂ sonde system that can measure in situ CO₂
121 vertical profiles in the atmosphere from the ground to altitudes up to about 10 km with a 240-400 m
122 altitude resolution by using a small-sized balloon. Although the sonde system is thrown away after
123 every flight due to the difficulties associated with recovery, the sonde systems are easily prepared with
124 a relatively low cost. We have tested the sonde flight experiments more than 20 times in Japan. The
125 CO₂ sonde developed has the following advantages, compared with other measurement techniques
126 described above: (1) its cost of operation is low and the flight permission is easy to obtain from the
127 authorities as compared with the aircraft observations; (2) the CO₂ sonde can be easily carried to the
128 launch sites since the instrument is light; (3) a limited amount of power is required for the operation;
129 (4) it can generally be launched at any time; and (5) the meteorological data are obtained
130 simultaneously with CO₂ profile data. In this study, the design of our novel CO₂ sonde and the results
131 of the comparison experiments with aircraft measurements are described. The target accuracy and
132 precision in the measurements with the CO₂ sonde are below about 1 ppm CO₂ mole fraction in the
133 atmosphere of 400 ppm CO₂, preferable for carbon cycle studies (e.g. Maksyutov et al. 2008). The
134 developed CO₂ sonde system attained virtually all the targets from the ground to an altitude of about
135 10 km.

136 Inai et al. (2018) measured vertical profiles of CO₂ mole fraction in the equatorial eastern and
137 western Pacific in February 2012 and February–March 2015, respectively, by using our novel CO₂
138 sondes which are described in this report. They found that the 1–10 km vertically averaged CO₂ mole
139 fractions lie between the background surface values in the Northern Hemisphere (NH) and those in the
140 Southern Hemisphere (SH) monitored at ground-based sites during these periods. Their study showed
141 that the combination of CO₂ sonde measurements and trajectory analysis, taking account of convective
142 mixing, was a useful tool in investigating CO₂ transport processes.

143

144 **2. Materials and methods**

145 **a. Design of the CO₂ sonde**

146 Many severe restrictions are noted for the operation of balloon-borne CO₂ sondes. First, the weight
147 of the CO₂ sonde package should be less than about 2 kg, based on the legal restriction by the US FAA
148 (Federal Aviation Administration) and by the Japanese aviation laws for the payload weight of 2.721
149 kg for unmanned free balloons. Balloon systems heavier than the above regulation weight are not
150 useful for the frequent flights because the flight permission from the authorities is much more difficult
151 to obtain, and the additional safety requirements are more expensive. The balloon system is thrown
152 away in the ocean after each flight due to a long-distance transportation (100 km or more to the east)
153 by strong westerly winds in the upper atmosphere of mid-latitude area. This is done to avoid the
154 accidents associated with a falling onto the urban areas, resulting in high recovery costs. Therefore,
155 the cost of the CO₂ sonde system should be low for frequent observations. The non-recovery system
156 implies that every instrument should perform consistently.

157 In this study the NDIR technique was adopted for a detection of CO₂ concentrations. The NDIR
158 CO₂ measurement techniques have been widely used in many places such as WMO/GAW (Global
159 Atmosphere Watch) stations. Our target instrumental accuracy and precision of approximately 1 ppm
160 are less stringent than those of the ground-based instruments (± 0.1 ppm) used at the WMO/GAW
161 stations (WMO, 2016). However, the surrounding conditions for the instrument are substantially
162 severe during the flight experiments, as the pressure changes from 1,000 to 250 hPa and the
163 surrounding temperature changes from 300 to 220 K during flights from the surface to an altitude of
164 10 km in about 60 min.

165 In the NDIR technique for CO₂ measurements, the IR emission from a broadband wavelength source
166 is passed through an optical cell and two filters, and then the light intensities are detected by two IR
167 detectors. The one optical filter covers the whole absorption band of CO₂ around 4.3 μm , while the

168 other covers a neighboring non-absorbed region around 4.0 μm . provided that the chosen active and
169 reference channel filters do not significantly overlap with the absorption bands of other gas species
170 present in the application. (Hodgkinson et al., 2013).

171 The Beer–Lambert Law is expressed by Eq. (1), defining the light intensity in the absence of CO_2
172 in the cell as I_0 and the light intensity in the presence of CO_2 in the cell as I ,

$$173 \quad \frac{I}{I_0} = \exp(-\varepsilon C L) \quad (1),$$

174 where C is the CO_2 concentration in molecules cm^{-3} , L is the optical path length in cm, and ε is
175 the absorption cross-section in $\text{cm}^2 \text{ molecule}^{-1}$. Using the relationship of $C = XP (k_B T)^{-1}$, where X
176 is the CO_2 mole fraction and P is the pressure of dehumidified ambient air, and the approximation
177 of $\exp(-\varepsilon C L) = 1 - \varepsilon C L$, under the condition of $\varepsilon C L \ll 1$, Eq. (1) is rewritten as:

$$178 \quad \frac{(I_0 - I)}{P} = X \frac{I_0 \varepsilon L}{k_B T_c} \quad (2),$$

179 where T_c is the sample air temperature in the sensor cell and k_B is the Boltzmann constant. **The eq. (1)**
180 **and (2) hold for monochromatic light only and that eq. (2) only holds for small absorptions.** Although
181 the NDIR analyzer exhibits non-linear absorption due to the saturation of strong absorption lines, the
182 NDIR analyzer is known to have a good linearity within a certain concentration range (Galais et al.
183 1985). In our analyses of the balloon data, eq. (2) was used only for the interpolation between the low
184 and high mole fractions of the in-flight calibration gases to obtain the ambient CO_2 mole fractions.
185 With a 120 mm long absorption cell, the absorption intensity is approximately 3% at 400 ppm CO_2
186 with our CO_2 NDIR system, i.e., $\varepsilon C L \approx 0.03$ and the approximation of $\exp(-\varepsilon C L) = 1 - \varepsilon C L$ are
187 well fitted. The values of $[I(4.0) - I(4.3)]$ were used instead of $(I_0 - I)$ to obtain the CO_2 mole
188 fraction values in the NDIR measurements, where $I(4.0)$ and $I(4.3)$ were the signal intensities at
189 the 4.0 μm wavelength for background measurements and the 4.3 μm wavelength for CO_2 absorption
190 measurements, respectively. Thus, the value of $[I(4.0) - I(4.3)]/P$ is thus proportional to the CO_2

191 mole fraction X in the optical cell. The proportionality constant is usually determined by the
192 measurements of the standard gases. In the NDIR measurements at the ground WMO/GAW stations,
193 carbon dioxide mole fractions are referenced to a high working standard and a low working standard
194 and are determined by the interpolations of the signals with the two standards, and the calibration with
195 the two standard gases are carried out every 12 h (Fang et al., 2014).

196

197 **b. System configuration of the CO₂ sonde system**

198 A schematic diagram and photograph of the CO₂ measurement instrument are shown in Fig. 1. The
199 CO₂ sonde has three inlets installed for ambient air and two calibration gases with mesh filters (EMD
200 Millipore, Millex-HA, 0.45 μm pore size) to remove the atmospheric particles. Three solenoid valves
201 (Koganei, G010LE1-21) were used to switch the gas flow to the CO₂ sensor. A constant-volume piston
202 pump with a flow rate of 300 $\text{cm}^3 \text{min}^{-1}$ (Meisei Electric co., Ltd.), which is originally used for
203 ozonesonde instruments, directed the gas flows from the inlets through the solenoid valves into a
204 dehumidifier, a flow meter, and a CO₂ sensor. The absolute STP (standard temperature and pressure)
205 flowrate decreased with a decrease in pressure. Since the exit port of the CO₂ sensor was opened to
206 the ambient air, the pressure of dehumidified outside air and calibration gases in the absorption cell
207 were equal to the ambient pressure during the flight. Next to the pump, the gases were introduced to a
208 glass tube filled with the magnesium perchlorate grains (dehumidifier) installed upstream to the CO₂
209 sensor to remove the water vapor. Fabric filters were installed on both ends of the dehumidifier, and a
210 mesh filter was installed downstream of the dehumidifier to prevent the CO₂ sensor from the incursion
211 of magnesium perchlorate grains to the optical cell.

212 The infrared absorption cell consisted of a gold-coated glass tube, a light source, and a photodetector.
213 The light source (Helioworks, EP3963) consisted of a tungsten filament with a spectral peak intensity
214 wavelength of approximately 4 μm . The light from the source passed through a gold-coated glass tube
215 (length 120 mm, and inside diameter 9.0 mm). The commercial CO₂ NDIR photodetector (Perkin-

216 Elmer TPS2734) had two thermopile elements, one of which was equipped with a band-pass filter at a
217 wavelength of 4.3 μm for the measurement of the CO_2 absorption signal, whereas the other was
218 equipped with a band-pass filter at a wavelength of 4.0 μm for the measurement of the background
219 signal. The signals from the sensors were amplified by an operational amplifier and converted to 16
220 bit digital values by an A/D convertor. The signal intensities of the detectors at 4.0 and 4.3 μm without
221 CO_2 gas were set to the equal levels by adjusting the amplification factors in the laboratory. The electric
222 power for the CO_2 sensor, pump, and valves was supplied through a control board using three 9 V
223 lithium batteries, lasted for more than 3 h during the flight. The control board connected to the
224 components regulated the measurement procedures, such as switching the solenoid valves and
225 processing the signal. As shown in Fig. 1, the measurement system has an expanded polystyrene box
226 molded specially to settle the optical absorption cell, electronic board, pump, battery and other
227 components.

228

229 **c. Calibration gas package**

230 Under the wide ranges of temperature and pressure conditions, the CO_2 sensor signal was unstable,
231 and the calibration of the CO_2 sensor only on the ground before launch was insufficient to obtain the
232 precise values of the CO_2 concentrations. To solve this problem, an in-flight calibration system was
233 incorporated into the CO_2 sonde. A calibration gas package was attached to the CO_2 sonde for the in-
234 flight calibration, as shown in Fig. 2. The calibration gas package consisted of two aluminum coated
235 with polytetrafluoroethylene (PTFE) bags (maximum volume: 20 L), containing reference gases with
236 low (~ 370 ppm) and high (~ 400 ppm) CO_2 concentrations. In each bag, ~ 8 L (STP) of the reference
237 gas was introduced from standard CO_2 gas cylinders just before launch. Since the gas bags were soft,
238 their inner pressures were equal with the ambient air pressures during the balloon flight. The gas
239 volumes in the bags increased with the altitude during the ascent of the balloon due to a decrease in
240 the ambient pressure, while the reference gases were consumed during the calibration procedures. The

241 optimum amounts of gas in the bags were determined by both the ascending speed of the balloon and
242 the consumption rate to avoid the bursting of the bags and exhaustion of the gases. The CO₂
243 concentrations of the reference gases in the bags were checked by the NDIR instrument (LICOR, LI-
244 840) before launching. Thereafter, approximately 6 L of the reference gas was left in each bag for a
245 subsequent in-flight calibration. The change in the CO₂ mole fraction in the bags was less than 1 ppm
246 over a 3 days period, which was negligible over the observations time during the balloon flight. All
247 measurements were reported as dry-air mole fractions relative to the internally consistent standard
248 scales maintained at Tohoku University (Tanaka et al. 1987; Nakazawa et al. 1992).

249 Since the gas exit port of the optical absorption cell was opened to the ambient air, the cell pressure
250 was equalized with the ambient pressure for measuring both the ambient air and two standard gases.
251 During the balloon-borne flights, the temperatures inside the CO₂ sonde package were measured with
252 thermistors. The temperature inside the CO₂ sonde package gradually decreased by approximately 5
253 K, from 298 K on the ground to 293 K at an altitude of 10 km during the flights. Probably due to the
254 polystyrene box, and the heat produced by the NDIR lamp, pump and solenoid valves, temperature
255 inside the sonde package remained virtually constant in spite of low ambient temperatures at high
256 altitudes (~220 K). Within one measurement cycle time (160 s) with the standard gases, the
257 temperature change was less than 0.4 K in the sonde package. The temperatures of the sample gas in
258 the tube just before the inlet of the CO₂ NDIR cell were also measured using a thin wire thermistor,
259 commonly used for ambient temperature measurement in GPS sonde equipment with a quick response
260 time (shorter than 2 s). The gas temperature change was negligible at the valve change timings between
261 the standard gas and ambient air (< 0.1 K). The result indicated that the gas temperatures were
262 relatively constant after passing through the valves, pump, dehumidifier cell, and piping for both the
263 standard gases and ambient air.

264 The performances of the CO₂ sonde instruments were checked before the balloon launching since
265 the CO₂ sonde systems were not recovered after the launch experiments were performed. For about 60

266 min. before the launch, the values of $[I(4.0) - I(4.3)]/P$ were measured with the valve cycles (each
267 step 40 s, total 160 s) for two standard gas packages (~370 ppm and ~400 ppm) for calibration and one
268 intermediate concentration gas package (~385 ppm) as a simulated ambient gas sample.

269

270 **d. Total sonde system**

271 The CO₂ sonde was equipped with a GPS radiosonde (Meisei Electric co., Ltd., RS-06G). The
272 balloon carried the instrument packages in the altitude with measuring CO₂ and meteorological data
273 (GPS position and time, temperature, pressure, and humidity). The CO₂ sonde transmitted those data
274 to a ground receiver (Meisei Electric co., Ltd., RD-08AC) at 1 s intervals, thus it was unnecessary to
275 recover the CO₂ sonde after the balloon burst. Figure 2 showed an overall view of the CO₂ sonde
276 developed in this study, which consisted of a CO₂ measurement package, a calibration gas package, a
277 GPS radiosonde, a balloon, and a parachute. The total weight of the CO₂ sonde was 1700 g, including
278 the GPS radiosonde (150 g), CO₂ measurement package (1000 g), and calibration gas package (550 g).
279 The dimensions of the CO₂ measurement package were width (W) 280 mm × height (H) 150 mm ×
280 depth (D) 280 mm. The size of the calibration gas package was W 400 mm × H 420 mm × D 490 mm.

281 The CO₂ sonde system was flown by a 1200 g rubber balloon (Totex). The ascending speed was
282 around 4 m / s by controlling the helium gas amount in the rubber balloon and checking the buoyancy
283 force. In practice, it was difficult to precisely control the ascending speed of the balloon, and the actual
284 resulting speeds were in the range of 3 - 5 m s⁻¹. This corresponds to the height resolution of
285 approximately 240–400 m for the measurements of the CO₂ vertical profiles.

286 Ascending speed slower than 3 m s⁻¹ can lead to a collision with a nearby tree and building, result
287 in equipment falling in the urban areas. With faster ascending speeds, the altitude resolution of the
288 measurements decreased and the gas standard bag became full and the pressure inside the gas bags
289 became higher than the ambient pressure because of the lower ambient pressures at higher altitudes.
290 The high pressure inside the gas bag resulted in the fast flow speed in the optical absorption cell of

291 NDIR, which shifted the signal values for the pressurized gas sample. Since pressure relief valves for
292 the bags did not work at low pressures at high altitudes, we did not use the pressure relief valve for the
293 standard gas bags. When the ascending speed was low, the standard gas bags became empty since they
294 were consumed by the in-flight calibration procedures during the long ascending time. Since the
295 measurements after the over-pressurization or the exhaustion of the reference gas bag are useless, this
296 technical problem determines the upper limit (10 km) of altitude for the measurements in this study.
297 Based on our experiences, this problem generally occurred at an altitude above approximately 10 km.
298 A prototype of the CO₂ sonde is available from Meisei Co. Ltd. (Isesaki, Japan) with about \$4,500.

299

300 **e. Data processing procedures**

301 Since the surrounding conditions of the sonde change significantly during the ascending period,
302 the NDIR measurement system is calibrated with the two standard gases at every altitudes. However,
303 since the balloon-borne instrument is only equipped with one NDIR absorption cell and the balloon
304 ascends continuously, it is not possible to measure the ambient air sample and the two standard gases
305 at the same time and at the same altitude. Therefore, the measurement cycle during the flights consisted
306 of the following steps: (1) low concentration standard gas, (2) ambient air, (3) high concentration
307 standard gas, and (4) ambient air. The measurement time for each step was 40 s. At switching timings
308 of the valve cycles, the signal became stable within 10 s, and the averages of residual 30-s period
309 signals were used for the calculation of the CO₂ mole fractions. Since the gas exit port of the NDIR
310 optical absorption cell was opened to the ambient air, the cell pressure was equalized with the ambient
311 pressure. During the period of the 40 s gas change, the pressure would change about 2 % when the
312 ascending speed of the balloon was 4 m s⁻¹. The temperature of the ambient air and standard gas
313 samples at the inlet port of the optical cells was measured and found to be constant during each cycle
314 of the calibration procedure.

315 Figure 3 shows an example of the raw data obtained from the CO₂ sonde experiment. Figure 3

316 presents the plots of the values of $[I(4.0) - I(4.3)]/P$ against the altitude, where $I(4.0)$ and $I(4.3)$
317 are the signal intensities at the wavelength of 4.0 μm for background measurements and the 4.3 μm
318 wavelength for CO_2 absorption measurements, as obtained by the NDIR CO_2 sensor on the balloon,
319 and P is the ambient atmospheric pressure obtained by the GPS sonde data and pressure
320 measurements on the ground.

321 The values of $[I(4.0) - I(4.3)]/P$ are proportional to the CO_2 mole fraction X according to the
322 Beer-Lambert law as expressed by Eq. (2). By using the values of $[I(4.0) - I(4.3)]/P$, we can
323 compensate for the pressure change to determine the CO_2 concentration. As shown in Fig. 3, the
324 differences in the $[I(4.0) - I(4.3)]/P$ values between the low and high standard gases remained
325 relatively constant while ascending to the higher altitudes. However, the $[I(4.0) - I(4.3)]/P$ values
326 for the each standard gas did not change linearly but sometimes displayed some curvatures as shown
327 in Fig. 3. This may be due to the differences between the baseline drift of the two sensors at 4.3 μm
328 and 4.0 μm in the NDIR detector. Since the measurements were performed alternately for the standard
329 gases and the ambient air with the NDIR cell and are not performed simultaneously, the values for the
330 standard gas signals at the time of the ambient air measurement was estimated. Therefore, the cubic
331 spline fitting curves for the observation points of the 30 s average values (red circles in Fig. 3) of the
332 same standard gas were used to obtain the low and high calibration points for the calculation of the
333 mole fractions in the ambient air. In Fig. 3, the cubic spline fitting curves are represented by the red
334 curves, and the estimated values for the standard gases at the ambient gas measuring time are
335 represented by the small black dots on the cubic spline curves, which are used for the interpolation to
336 determine the ambient air concentrations. Linear line fitting between the standard gas values did not
337 work well because the connection lines of the values sometimes displayed curvatures as shown in Fig.
338 3. Since there were in-phase fluctuations in the $I(4.0)$ and $I(4.3)$ signals during the flights, the
339 subtraction of $[I(4.0) - I(4.3)]$ could partly improve the signal-to-noise ratios by canceling in-phase
340 fluctuations with each other.

341

342 **3. Results and discussion**

343 **a. Laboratory tests**

344 Since the linear interpolation method for the $[I(4.0) - I(4.3)]/P$ values was used to determine the
345 ambient air CO₂ mole fractions in the balloon-borne experiments, the deviations from the linear
346 interpolation process were also investigated. The measurements of various mole fractions gas samples
347 in the laboratory indicated that the linear interpolation error with the two standard gas packages (~370
348 ppm and ~400 ppm) was less than 0.2 ppm in the range between 360 and 410 ppm. Figure 4 shows the
349 measurement results of the NDIR cell developed in this study at various CO₂ mole fractions. The outlet
350 port of the NDIR system was connected to the commercial CO₂ instrument (LICOR, LI-840A) as a
351 standard device, and the two instruments simultaneously measure the sample gas at 1010 hPa. The
352 standard gases of 365 and 402 ppm were used for the calibration, and the mixtures of the standard
353 gases were used for the samples. This indicated the values of $[I(4.0) - I(4.3)]/P$ of the system were
354 proportional to the mole fraction of CO₂. This type of experiment could not be performed at low
355 pressures, since we did not have a standard device which can be operated under low pressures.

356 Figure 5 shows the results of an experiment using a vacuum chamber in the laboratory, where the
357 flight pressure conditions were simulated and the performances of the CO₂ sonde instruments was
358 evaluated. The temperature inside the chamber was not controlled and was about 298 K. In the actual
359 flights, the temperature inside the sonde package did not change more than 5 K. The CO₂ sonde system
360 and two standard gas packages were placed in the vacuum chamber. The chamber was filled with the
361 mole fraction sample gas of 377.3 ppm before the pumping. The pressure of the chamber was gradually
362 and continuously decreased using a mechanical pump from 1010 hPa (ground surface pressure) to 250
363 hPa (about 10 km altitude pressure) over 60 min, corresponded to a balloon ascending speed of 3 m /s
364 in actual flights, whereas the sample gas was slowly and continuously supplied to the chamber. The
365 values $[I(4.0) - I(4.3)]/P$ were measured for the two standard gas packages, and the sample gas with

366 the valve cycles (each step 40 s, total 160 s) as described in the previous section. The mole fractions
367 of the sample gas in the chamber were calculated by the interpolation of the signals for the two standard
368 gases. The 30 s signals 10 s after the valve changes were used for the interpolation calculations to
369 avoid the incomplete gas exchanges in the NDIR optical cell. The black circle in Fig. 5 indicates the
370 sample gas mole fraction obtained from the linearly interpolated standard gas signals in each
371 calibration cycle. The vertical error bar in Fig. 5 indicates the square-root of the sum of squares for the
372 standard deviations of the sample and standard gas signals at each step. The errors in the CO₂ mole
373 fractions were estimated to be 0.6 ppm at 1010 hPa and 1.2 ppm at 250 hPa using the calibration cycles.
374 The results in Fig. 5 indicated that the determination of the sample gas concentration using the linear
375 interpolation with the standard gases was appropriate within the error, when the pressure continuously
376 decreased from 1000 to 250 ppm over 60 min.

377 When the CO₂ sonde instrument was inclined and vibrated in the laboratory, the fluctuations in the
378 signals were observed. The quantitative correlation between the signal fluctuation intensities and
379 acceleration speed, measured by a 3-dimensional acceleration sensor, was investigated, but no distinct
380 correlation was detected. However, the in-flight calibration system partly solved this problem by taking
381 the signal difference of $[I(4.0) - I(4.3)]$ and also by measuring alternately the two standard gases
382 every 40 s during the balloon flights.

383 The temperature characteristics of the CO₂ sensor were also investigated by changing the sensor cell
384 block temperature from 273 to 323 K at the pressure of ~1010 hPa, using a heater in the laboratory.
385 The laboratory experiment related to the temperature dependence suggested that the measurement error
386 is less than 0.2 ppm due to the temperature change during one valve cycle (160 s) in the balloon-borne
387 experiments.

388 In principle, the absorption intensities $(I_0 - I)$ in the NDIR measurements are proportional to the
389 absolute CO₂ concentrations in the sample air in the absorption cell. Therefore, at higher altitudes
390 where the pressures were lower, the values of $[I(4.0) - I(4.3)]$ were smaller and the signal-to-noise

391 ratios of $[I(4.0) - I(4.3)]/P$ decreased. The error of the CO₂ mole fractions of 1.2 ppm at 250 hPa
392 corresponds to an absolute CO₂ concentration of 3.2×10^{13} molecule cm⁻³. The equivalent altitude for
393 this value was 90 km with a CO₂ molar fraction of 400 ppm. As described previously, the purpose of
394 CO₂ balloon observations is to measure the CO₂ mole fraction within 1 ppm errors in the atmospheres
395 around 400 ppm CO₂. The upper limit of the altitude for the observations with the developed CO₂
396 sonde is considered to be ~10 km. Furthermore, as described in section 2d, the problems of the vacancy
397 or over-pressure in the standard gas bags took place around 10 km altitudes, which resulted in large
398 errors. This also practically determines the upper altitude limit for CO₂ sonde observations.

399

400 **b. Comparison with aircraft data**

401 Two types of aircraft measurement data, the NIES/JAXA chartered aircraft and the CONTRAIL
402 data, were used for comparison with the CO₂ sonde measurement data. The NIES/JAXA chartered
403 aircraft measurements were conducted on the same days as the CO₂ sonde observations (January 31st,
404 2011 and February 3rd, 2011). The chartered aircraft observations were performed as a part of the
405 campaign for validating the GOSAT data and calibrating the TCCON FTS data at Tsukuba (36.05°N,
406 140.12°E) (Tanaka et al., 2012). The chartered aircraft data were obtained using an NDIR instrument
407 (LICOR LI-840) that had a control system of constant pressure and had the uncertainty of 0.2 ppm.
408 On both January 31st and February 3rd, the chartered aircraft measured the CO₂ mole fractions during
409 descent spirals over Tsukuba and Kumagaya (Fig. 6). Because the air traffic was strictly regulated near
410 the Haneda and Narita international airports, the aircraft observations at altitudes above 2 km over
411 Tsukuba were prohibited. Therefore, the descent spiral observations were conducted over Kumagaya
412 at altitudes of 7–2 km and over Tsukuba at altitudes of 2–0.5 km. Tsukuba is located approximately 20
413 km northeast of Moriya, whereas Kumagaya is located approximately 70 km northwest of Moriya.

414 Seven profiles based on the CONTRAIL measurements, obtained during the ascent and descent of
415 aircrafts over Narita airport and had passage times close to the CO₂ sonde observations, were available

416 within two days after or before the dates of the CO₂ sonde measurements (Table 1). The CO₂ sonde
417 observations were conducted on January 31st and February 3rd, 2011 from Moriya. One set of
418 CONTRAIL data, obtained from the flight from Hong Kong to Narita (data set name: 11_060d), was
419 available on January 31st, but no CONTRAIL data were available for February 3rd. Therefore, the
420 CONTRAIL data, obtained from the flight from Hong Kong to Narita on February 2nd (data set name:
421 11_062d), were used for comparison with the February 3rd CO₂ sonde data. Figure 6 also shows the
422 CONTRAIL 11_060d and 11_062d flight paths and the CO₂ sonde launched at Moriya on January 31st
423 and February 3rd, 2011. On January 31st, the flight time of the CONTRAIL 11_060d over the Narita
424 airport and the launch time of the CO₂ sonde at Moriya were relatively close to one another. The flight
425 path of the CONTRAIL 11_062d data on February 2nd, 2011 was close to that of the CO₂ sonde on
426 February 3rd, 2011 and both observations were conducted in the early afternoon. The CONTRAIL
427 data referred in the present study was obtained using the Continuous CO₂ Measuring Equipment
428 (CME) located onboard commercial airliners (Machida et al. 2008; Matsueda et al. 2008). The typical
429 measurement uncertainty (1σ) of the CME has been reported as 0.2 ppm (Machida et al. 2008).

430 Figure 7 shows the vertical profiles of CO₂ observed by the CO₂ sonde at Moriya, the chartered
431 aircraft at Kumagaya and Tsukuba, and the CONTRAIL over the Narita airport on January 31st, 2011.
432 The overall vertical distribution of the CO₂ sonde data resembled with those of the chartered aircraft.
433 The vertical profiles of the CONTRAIL 11_060d flight on January 31st at the 5.3–6.8 km altitude
434 range consisted of the missing data because of the CME calibration period.

435 Figure 8 shows the comparison of the CO₂ vertical profiles obtained by the CO₂ sonde over Moriya,
436 NIES/JAXA chartered aircraft over Kumagaya and Tsukuba on February 3rd, 2011, and the
437 CONTRAIL on February 2nd, 2011 over Narita. The shape of the vertical profile obtained by the
438 chartered aircraft on February 3rd resembled that obtained by the CO₂ sonde, although the profile from
439 the chartered aircraft was shifted to the lower CO₂ concentration side compared to that of the CO₂
440 sonde.

441 Table 2 lists the comparisons of the CO₂ mole fractions measured by the balloon CO₂ sonde and
442 NIES/JAXA chartered aircraft on January 31st and February 3rd, 2011. The averaged values of the
443 aircraft measurement over the range of each balloon altitude ± 100 m are listed in Table 2, since the
444 altitude resolution of the aircraft measurements is higher than that of the CO₂ sonde. From the February
445 3rd measurements, the height of the boundary layer around an altitude of 1 km was different between
446 the CO₂ sonde and the NIES/JAXA aircraft measurements as shown in Fig. 8. Therefore, the data
447 below 1 km on February 3rd are not included in Table 2. From the data on January 31st, the averaged
448 value of the differences between the CO₂ sonde and the NIES/JAXA aircraft was relatively small (0.42
449 ppm), which corresponded to the bias of the measurements. The standard deviation of the differences
450 was 1.24 ppm. From the February 3rd data, the bias was large (1.41 ppm), whereas the standard
451 deviation of the differences was not so large (1.00 ppm), which corresponded to the similar but shifted
452 vertical profiles in shapes between the CO₂ sonde and aircraft measurements as shown in Fig. 8. The
453 difference between the CO₂ sonde data and the NIES/JAXA chartered aircraft data on February 3rd is
454 nearly equal to the difference between CONTRAIL data on February 2nd and the NIES/JAXA
455 chartered aircraft data on February 3rd.

456 Table 3 lists the comparisons of the CO₂ mole fractions measured by the balloon CO₂ sonde and
457 CONTRAIL aircraft, 11_060d on January 31st and 11_062d on February 2nd, 2011 up to the altitude
458 of 7,000 m. The averaged values of the aircraft measurements over the range of each balloon altitude
459 ± 100 m are listed in Table 3. The biases between the CO₂ sonde and the CONTRAIL aircraft results
460 were relatively small, 0.33 and 0.35 ppm, and the standard deviations of the differences were 1.16 and
461 1.30 ppm for the results on January 31st and February 3rd, respectively.

462 From the comparison between the CO₂ sonde data and the aircrafts (NIES/JAXA and CONTRAIL)
463 data, it was found that the CO₂ sonde observation was larger than those of aircrafts by about 0.6 ppm
464 on average. The standard deviation of the difference from the CO₂ sonde and aircraft observations was
465 1.2 ppm (1σ). If the 4 sets of aircraft measurement data obtained by the NIES/JAXA and CONTRAIL

466 observations were accurate within the published uncertainties, ignoring the differences in the flight
467 time and geographical routes, the measurement error of the CO₂ sonde system was estimated from the
468 standard deviations of all the difference values in Tables 2 and 3. The estimated error value up to an
469 altitude of 7 km was 0.6 ± 1.2 ppm for the CO₂ sonde observation with a 240 m altitude resolution and
470 3 m s^{-1} ascending speed. The root mean square value (1.3 ppm) from all the difference value in Table
471 2 and 3 indicated that the CO₂ sonde could measure the CO₂ vertical profiles within 1.3 ppm on average
472 compared to the aircraft observations. It is noted that, although error estimation was conducted for the
473 data up to an altitude of 7 km due to the availability of the chartered aircraft data, the CO₂ sonde data
474 above 7 km up to about 10 km. The measurement errors for the data above 7 km are expected to be
475 larger than the above estimation.

476

477 **c. CO₂ sonde observations over a forested area**

478 Figure 9 shows the vertical profiles of the CO₂ mole fraction, temperature, and relative humidity
479 obtained from the balloon-borne experiments of the CO₂ sonde at Moshiri (44.4°N, 142.3°E) on
480 August 26, 2009. The launch site is in a rural area of Hokkaido, Japan and is surrounded by forests.
481 The CO₂ sonde was launched at 13:29 LST and ascended with a mean vertical speed of approximately
482 3 m s^{-1} . The CO₂ sonde reached an altitude of 10 km after 56 min. The wind horizontally transported
483 the CO₂ sonde distances of 10 km and 21 km northeast when the CO₂ sonde reached the altitudes of 5
484 km and 8 km, respectively. The CO₂ sonde rapidly moved 52 km southeast at an altitude of 16 km.
485 Finally, the CO₂ sonde reached an altitude of 28 km before the balloon burst and the subsequent fall
486 of the sonde was directed by the parachute into the Sea of Okhotsk located 80 km east of the launch
487 site. The error bars for the CO₂ mole fraction in Fig. 9a were calculated from the deviation of the signal
488 intensities from the CO₂ sensor during the 40 s measurement periods for the ambient air and the two
489 standard gases.

490 The vertical temperature profile in Fig. 9b indicated the existence of three inversion layers of the

491 altitudes of approximately 2.0, 3.2, and 4.3 km. The relative humidity from the ground to the first
492 inversion layer at 2.0 km and between the second and third inversion layers from 3.2 to 4.3 km were
493 higher compared with those observed from 2.0 to 3.2 km and from 4.3 to 7.5 km. The CO₂ mole
494 fraction was the lowest near the ground (~373 ppm) and increased to approximately 384 ppm at an
495 altitude of 4–5 km around the third inversion layer before reaching a value of 387 ppm in the upper
496 troposphere (5–9 km). Significant decreases in the CO₂ mole fractions were observed in the two lower
497 layers from the ground to 3.2 km. Considering the clear weather on the day of the balloon experiment,
498 these results are explained by the uptake of CO₂ near the surface by plants in the forests through
499 photosynthesis processes in the daytime hours, and the diffusion and advection of the air mass
500 containing low CO₂ concentrations in the upper altitudes.

501 Because the CO₂ mole fraction for the vertical profiles near the surface is critically important to
502 estimating the flux around the observation point, the vertical profile data taken by our CO₂ sonde is
503 useful.

504

505 **d. CO₂ sonde observations over an urban area**

506 Figure 10 shows the vertical profiles of the CO₂ mole fraction, temperature, and relative humidity
507 obtained by the CO₂ sonde at Moriya (35.93°N, 140.00°E) on February 3rd, 2011. The launching time
508 was 13:10 LST and the sonde ascended with a mean vertical speed of approximately 2.9 m s⁻¹. Moriya
509 is located in the Kanto region and is 40 km northeast of the Tokyo metropolitan area. The launching
510 site was surrounded by the heavy traffic roads and residential areas. As seen in Fig. 10a, high CO₂
511 mole fractions were observed from the ground up to an altitude of 1 km. The average CO₂ volume
512 mole fraction in this layer was higher than that measured in the free troposphere approximately above
513 15 ppm. A small temperature inversion layer appeared at approximately 1 km, and the maximum
514 relative humidity was observed just below this inversion layer (Figs. 10b and c). These results
515 suggested that the CO₂ emitted from anthropogenic sources in and/or around the Tokyo metropolitan

516 area accumulated in the boundary layer at altitudes below 1 km.

517 An analysis of Figs. 9 and 10 indicated that there were a clear local consumption and emission of
518 CO₂ from the comparison of the levels of CO₂ concentration in the free troposphere, which suggested
519 a decoupling with the boundary-layer and synoptic inversion layers (Mayfield and Fochesatto, 2013).
520 When a small increase in a column XCO₂ value is observed by a satellite, it is difficult to estimate
521 which part of the atmosphere is responsible for the increase in XCO₂, the boundary layer with strong
522 CO₂ emission in the nearby area, or the free troposphere. Considering this fact, the vertical profile data
523 obtained by the CO₂ sonde around urban areas should provide more useful information than the column
524 averaged observations obtained by the satellites and FTS measurements to estimate the flux of
525 anthropogenic CO₂ emitted in and/or around the urban areas.

526

527 **4. Conclusion**

528 The CO₂ sonde is shown to be a feasible instrument for CO₂ measurements in the troposphere. The
529 laboratory test with a vacuum chamber has shown the precision of the CO₂ sonde at ~1010 hPa for 0.6
530 ppm and at ~250 hPa for 1.2 ppm. Comparisons of the CO₂ vertical profiles obtained by the CO₂ sonde
531 with two types of aircraft observations, the CONTRAIL and the NIES/JAXA chartered aircraft, were
532 carried out. The CO₂ sonde and CONTRAIL data were consistent. The CO₂ sonde data on January
533 31st, 2011 was in good agreement with the chartered aircraft data on the same day, but the CO₂ sonde
534 data observed on February 3rd, 2011 was larger by approximately 1.4 ppm, as compared with the
535 chartered aircraft data obtained on the same day from the ground to an altitude of 7 km. The
536 measurement errors of the CO₂ sonde system up to an altitude of 7 km were estimated to be 1.4 ppm
537 for a single point of 80 s period measurements with a vertical height resolution of 240–400 m. We
538 conducted the field CO₂ sonde observations more than 20 times in Japan and successfully obtained
539 CO₂ vertical profiles from the ground up to altitudes of approximately 10 km.

540 Our results showed that low-cost CO₂ sondes could potentially be used for frequent measurements

541 of vertical profiles of CO₂ in many parts of the world providing as useful information to understand
542 the global and regional carbon budgets by replenishing the present sparse observation coverage. The
543 CO₂ sondes can detect the local and regional transport evidence by determining CO₂ concentrations in
544 the air layer trapped between elevated inversion layers. Also, the CO₂ sonde observation data could
545 help improve the inter-comparison exercise for inverse models and for the partial validation of satellite
546 column integral data. In future, the CO₂ sonde data will be used for the validation of satellites and the
547 calibration of ground-based observations of sunlight spectroscopic measurements for column values
548 of CO₂ concentration.

549

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561

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698

699 **Table 1.** CONTAIL flight data near to the CO₂ sonde measurements on 31 January and 3 February

700 2011.

701

702

Data set name	Date	Time (LST) ^a
11_057a	CONTRAIL (29 January)	19:01
11_058d	CONTRAIL (30 January)	15:06
11_059a	CONTRAIL (30 January)	18:46
11_060d	CONTRAIL (31 January)	15:07
11_061a	CONTRAIL (1 February)	18:46
11_062d	CONTRAIL (2 February)	14:58
11_063a	CONTRAIL (4 February)	18:58
	CO ₂ sonde (31 January)	13:06
	CO ₂ sonde (3 February)	13:10

703

704 ^a Time for the CONTRAIL data represents the flight time in Japan Standard Time at an altitude of 1
705 km over the Narita airport. Time for the CO₂ sonde data represents the launching time at Moriya.

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708

709 **Table 2.** Comparisons of the CO₂ mole fractions between the balloon CO₂ sonde and NIES/JAXA
 710 chartered aircraft measurements on 31st January and 3rd February 2011.
 711

JAXA-NIES Chartered Aircraft (31 January 2011)				JAXA-NIES Chartered Aircraft (3 February 2011)			
Altitude (m) ^a	Balloon CO ₂ (ppm)	Aircraft CO ₂ (ppm) ^b	Difference (ppm) ^c	Altitude (m) ^a	Balloon CO ₂ (ppm)	Aircraft CO ₂ (ppm) ^b	Difference (ppm) ^c
849	399.05	397.62	1.43	1324	396.60	394.45	2.15
1202	398.16	397.53	0.63	1612	394.65	393.03	1.62
1610	398.00	397.17	0.83	1917	394.86	394.10	0.76
2038	396.50	396.95	-0.45	2223	395.77	393.54	2.23
2291	398.03	396.04	1.99	2539	395.41	393.95	1.45
2463	396.54	395.65	0.89	2867	394.71	395.11	-0.40
2844	393.44	395.24	-1.79	3215	394.99	392.99	2.00
3329	395.45	394.15	1.30	3543	393.59	393.07	0.52
3732	393.51	393.63	-0.12	3764	393.69	393.40	0.28
4161	395.47	393.54	1.93	3938	395.15	393.11	2.04
4575	394.62	392.94	1.68	4169	393.83	392.68	1.15
4918	393.24	393.64	-0.41	4458	396.57	393.51	3.06
5273	392.41	393.25	-0.84	4750	394.88	393.69	1.19
5654	393.02	393.47	-0.45	5047	396.53	394.01	2.53
6083	391.87	392.91	-1.04	5214	395.91	393.45	2.46
6510	392.76	391.65	1.11	5383	396.78	393.58	3.20
		Average =	0.42	5565	395.83	393.67	2.15
		Std Dev ^d =	1.16	5781	395.18	393.39	1.80
		RMS ^e =	1.20	6092	391.75	392.83	-1.09
				6287	392.44	392.42	0.02
				6467	393.67	392.23	1.44
				6639	395.07	392.42	2.65
				6815	394.00	393.00	1.00
						Average =	1.41
						Std Dev ^d =	1.00
						RMS ^e =	1.62

- 712 a. Altitudes of the balloon-borne experiments using the in-flight calibration with 40-s time intervals.
 713 b. Averaged values of the aircraft measurement results over the range of the balloon altitudes ± 100 m.
 714 c. Difference values of [balloon CO₂] - [Aircraft CO₂]
 715 d. Standard deviation of the differences (1σ).
 716 e. Root mean square values.

717

718 **Table 3.** Comparisons of the CO₂ mole fractions between the balloon CO₂ sonde measurements on
 719 31 January and CONTRAIL aircraft CME on 31 January (11_060d) and between the CO₂ sonde on 3
 720 February and CONTRAIL on 2 February (11_062d) up to the altitude of 7 km. The annotations are
 721 same as Table 2.
 722

CONTRAIL 11_060d (31 January 2011)				CONTRAIL 11_062d (2 February 2011)			
Altitude (m)	Balloon CO ₂ (ppm)	Aircraft CO ₂ (ppm)	Difference (ppm)	Altitude (m)	Balloon CO ₂ (ppm)	Aircraft CO ₂ (ppm)	Difference (ppm)
849	399.05	398.21	0.84	1917	394.86	396.59	-1.73
1202	398.16	399.56	-1.40	2223	395.77	396.45	-0.68
1610	398.00	398.77	-0.76	2539	395.41	395.71	-0.30
2038	396.50	397.07	-0.57	2867	394.71	394.67	0.04
2291	398.03	395.97	2.06	3215	394.99	393.34	1.65
2463	396.54	394.55	1.99	3543	393.59	394.25	-0.66
2844	393.44	393.41	0.04	3764	393.69	394.33	-0.64
3329	395.45	394.25	1.20	3938	395.15	394.69	0.46
3732	393.51	393.58	-0.07	4458	396.57	394.09	2.48
4161	395.47	393.86	1.61	4750	394.88	395.02	-0.14
4575	394.62	393.18	1.44	5047	396.53	396.55	-0.01
4918	393.24	393.62	-0.38	5214	395.91	396.01	-0.10
5273	392.41	392.76	-0.35	5383	396.78	394.78	2.00
6866	392.31	393.26	-0.96	5565	395.83	393.69	2.14
		Average =	0.33	5781	395.18	393.79	1.39
		Std Dev =	1.16	6092	391.75	393.57	-1.82
		RMS =	1.17	6287	392.44	393.32	-0.88
				6467	393.67	392.89	0.78
				6639	395.07	392.84	2.23
				6815	394.00	393.11	0.90
						Average =	0.35
						Std Dev =	1.30
						RMS =	1.31

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 724

725 **Figure captions**

726 **Figure 1.** Left: Schematic diagram of the CO₂ measurement package, where F1 and F2 represent the
727 band-pass filters at wavelengths of 4.0 μm and 4.3 μm, respectively. The outlet port of the CO₂ sensor
728 is opened to ambient air. Details of the system are described in the text. Right: Photograph of the inside
729 of the CO₂ sonde package. The components were placed in a specially modeled expanded polystyrene
730 box.

731 **Figure 2.** Photograph of the CO₂ sonde developed in this study before launching. a. CO₂
732 measurement package is shown in Fig. 1, b. GPS sonde, and c. Calibration gas package.

733 **Figure 3.** Raw data obtained by the CO₂ sonde launched on September 26, 2011 at Moriya, Japan. The
734 vertical axis is the difference between the 4.0 μm and 4.3 μm signal intensities divided by the ambient
735 pressure. The black line indicates the observation results during the balloon flight with calibration
736 cycles. The red circle indicates the 30 s average values in each step of the calibration. Red curve
737 indicates the cubic spline fitting curves for the observation points of the 30 s average values of the
738 same standard gas. The small black dots on the cubic spline curves indicate the estimated values for
739 the standard gases at the ambient gas measuring timing, which were is used for the interpolation to
740 determine the ambient air concentrations.

741 **Figure 4.** $[I(4.0) - I(4.3)]/P$ values versus CO₂ mole fraction, where $I(4.0)$ and $I(4.3)$ are the
742 signal intensities at the 4.0 μm wavelength for background measurements and the 4.3 μm wavelength
743 for CO₂ absorption measurements, obtained by the NDIR CO₂ sensor, and P is the ambient
744 atmospheric pressure. CO₂ mole fractions were measured with a standard NDIR instrument (LICOR,
745 LI-840A) connected to the balloon sensor in series. The pressure while carrying out the
746 measurements was constant at 1010 hPa.

747 **Figure 5.** Results of a chamber experiment of the CO₂ sonde. Pressure in the chamber was reduced
748 from 1010 hPa (ground level pressure) to 250 hPa (about 10 km altitude pressure) at a temperature of
749 about 298 K. The black circles indicate the value of the CO₂ mole fraction of the sample air in the

750 chamber, which was obtained from the interpolation of the standard gas values in each calibration
751 cycle. Vertical error bars indicate the square-root of sum of squares for the standard deviations of
752 the sample and standard gas signals at each step in the calibration cycle. The black dashed line shows
753 an average of all the values obtained for the sample gas. See the text for more details.

754 **Figure 6.** Flight paths of the CO₂ sonde observations launched at Moriya on January 31st (blue solid
755 line) and February 3rd (red solid line), 2011, the CONTRAIL 11_060d data on January 31st, 2011
756 (black solid line) and 11_062d data on February 2nd, 2011 (black dashed line) from Hong Kong to
757 Narita, and the NIES/JAXA chartered aircraft experiment on January 31st (green solid line) and
758 February 3rd (purple dotted line). The altitudes of the flight paths are also indicated.

759 **Figure 7.** The CO₂ vertical profiles obtained by the CO₂ sonde (circles connected with blue lines),
760 NIES/JAXA chartered aircraft data (dots connected with green lines), and the CONTRAIL data
761 (diamonds connected with black lines) on January 31st, 2011.

762 **Figure 8.** The CO₂ vertical profiles obtained by the CO₂ sonde (circles connected with red lines),
763 NIES/JAXA chartered aircraft data (dots connected with purple lines) on February 3rd, and
764 CONTRAIL data (diamonds connected with black lines) on February 2nd, 2011.

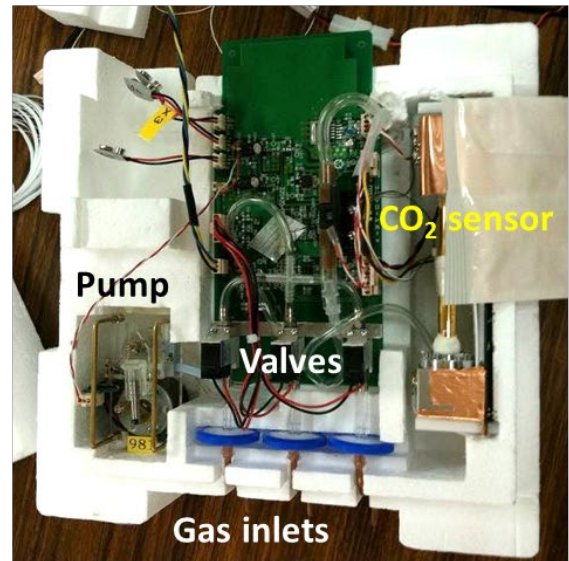
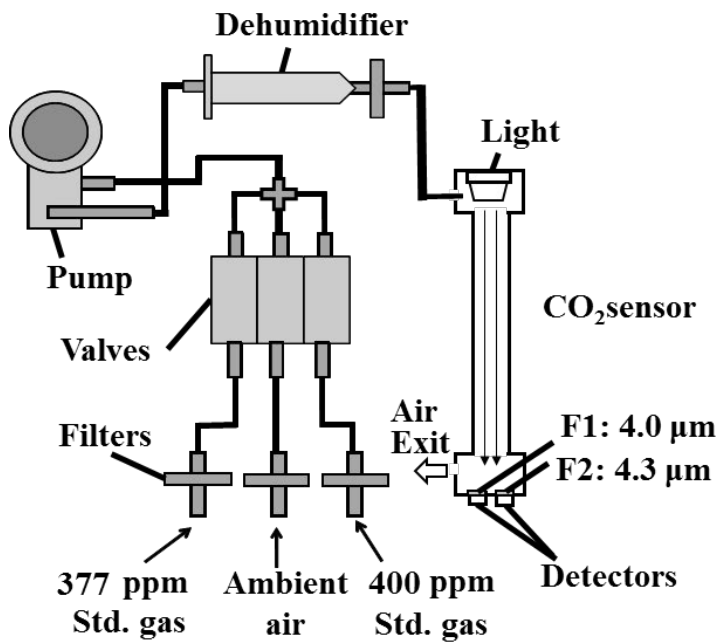
765 **Figure 9.** Profiles of (a) CO₂ mole fraction, (b) temperature (solid line) and potential temperature
766 (dotted line), and (c) relative humidity observed over a forest area, Moshiri in Hokkaido, Japan by
767 the balloon launched on August 26, 2009 at 13:30 (LST). The black circles with error bars in panel
768 (a) represent the data obtained by the CO₂ sonde.

769 **Figure 10.** Profiles of (a) CO₂ mole fraction, (b) temperature (solid line) and potential temperature
770 (dotted line), and (c) relative humidity observed over an urban area, Moriya near Tokyo on February
771 3rd, 2011 at 13:10 (LST).

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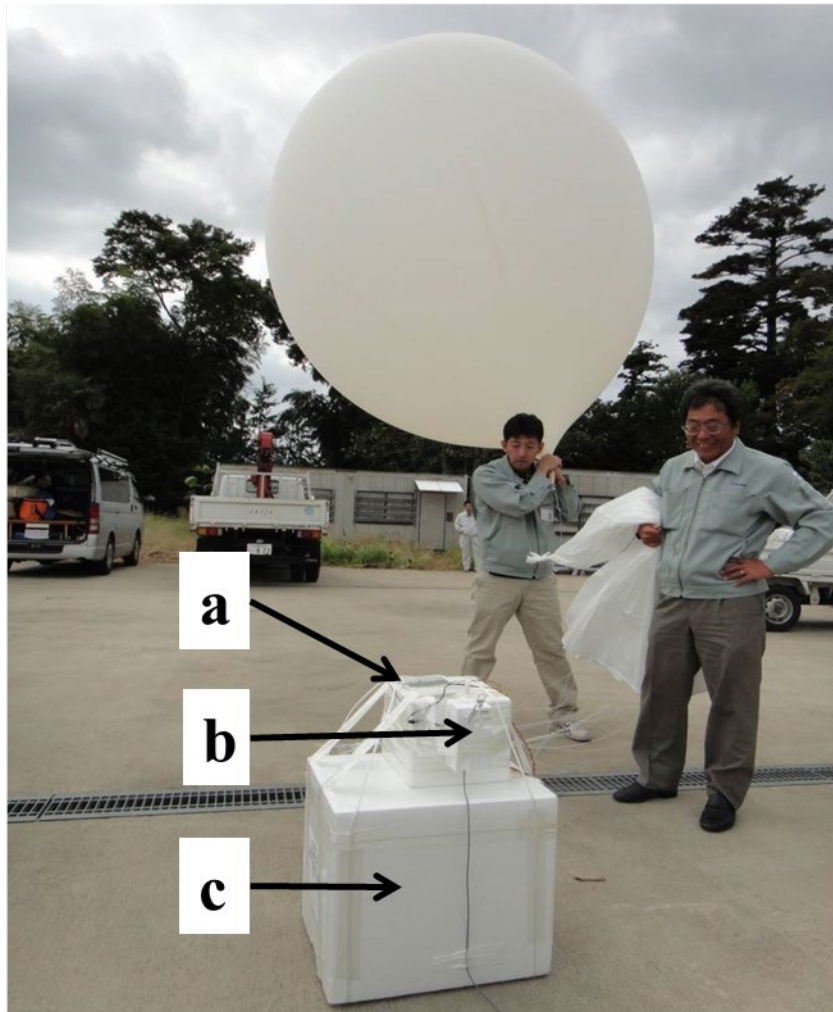
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776 **Figure 1.** Left: Schematic diagram of the CO₂ measurement package, where F1 and F2 represent the
777 band-pass filters at wavelengths of 4.0 μm and 4.3 μm, respectively. The outlet port of the CO₂ sensor
778 is opened to ambient air. Details of the system are described in the text. Right: Photograph of the inside
779 of the CO₂ sonde package. The components were placed in a specially modeled expanded polystyrene
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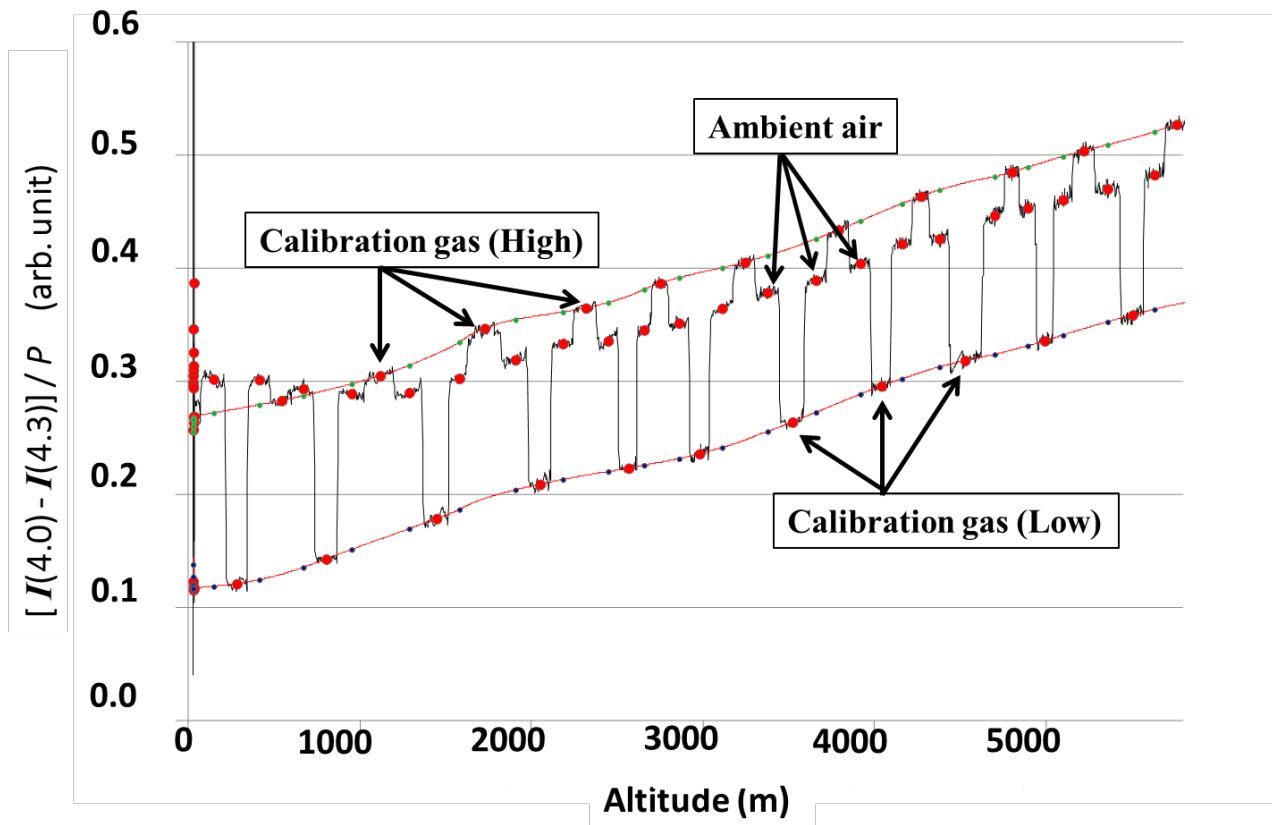
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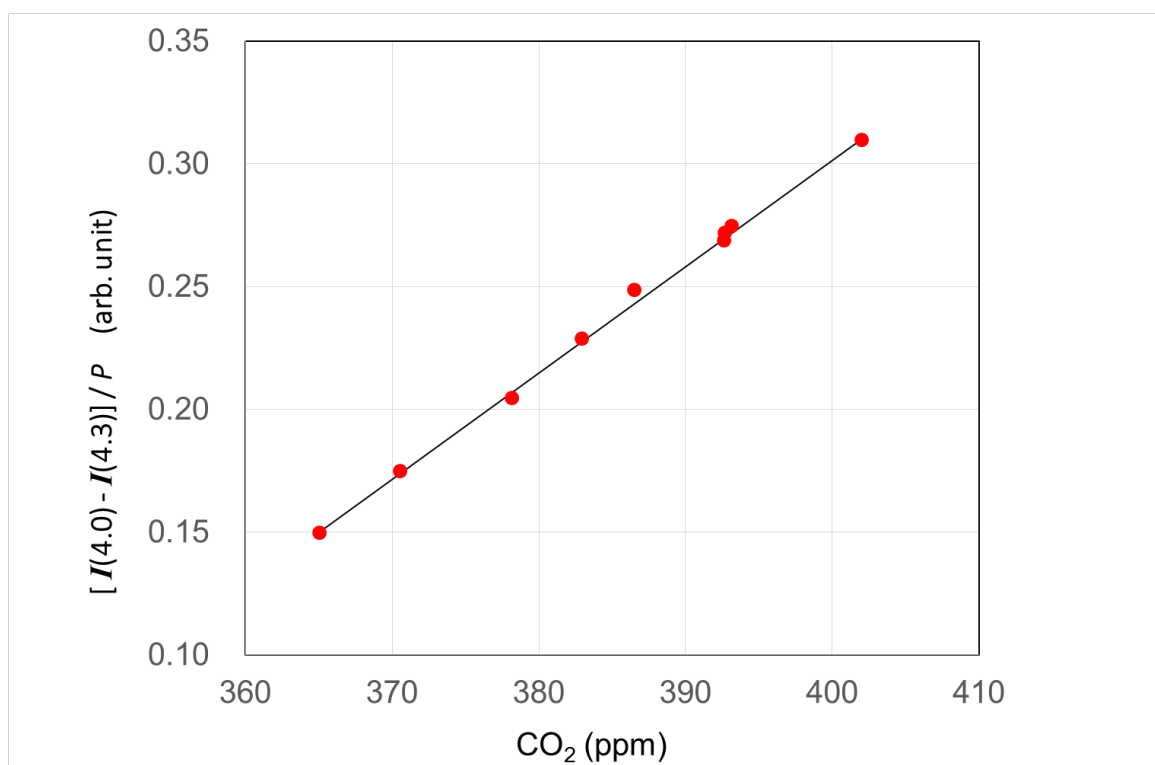
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784 **Figure 2.** Photograph of the CO₂ sonde developed in this study before launching. a. CO₂
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787 **Figure 3.** Raw data obtained by the CO₂ sonde launched on September 26, 2011 at Moriya, Japan.
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 790 calibration cycles. The red circle indicates the 30 s average values in each step of the calibration. Red
 791 curve indicates the cubic spline fitting curves for the observation points of the 30 s average values of
 792 the same standard gas. The small black dots on the cubic spline curves indicate the estimated values
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 794 to determine the ambient air concentrations.
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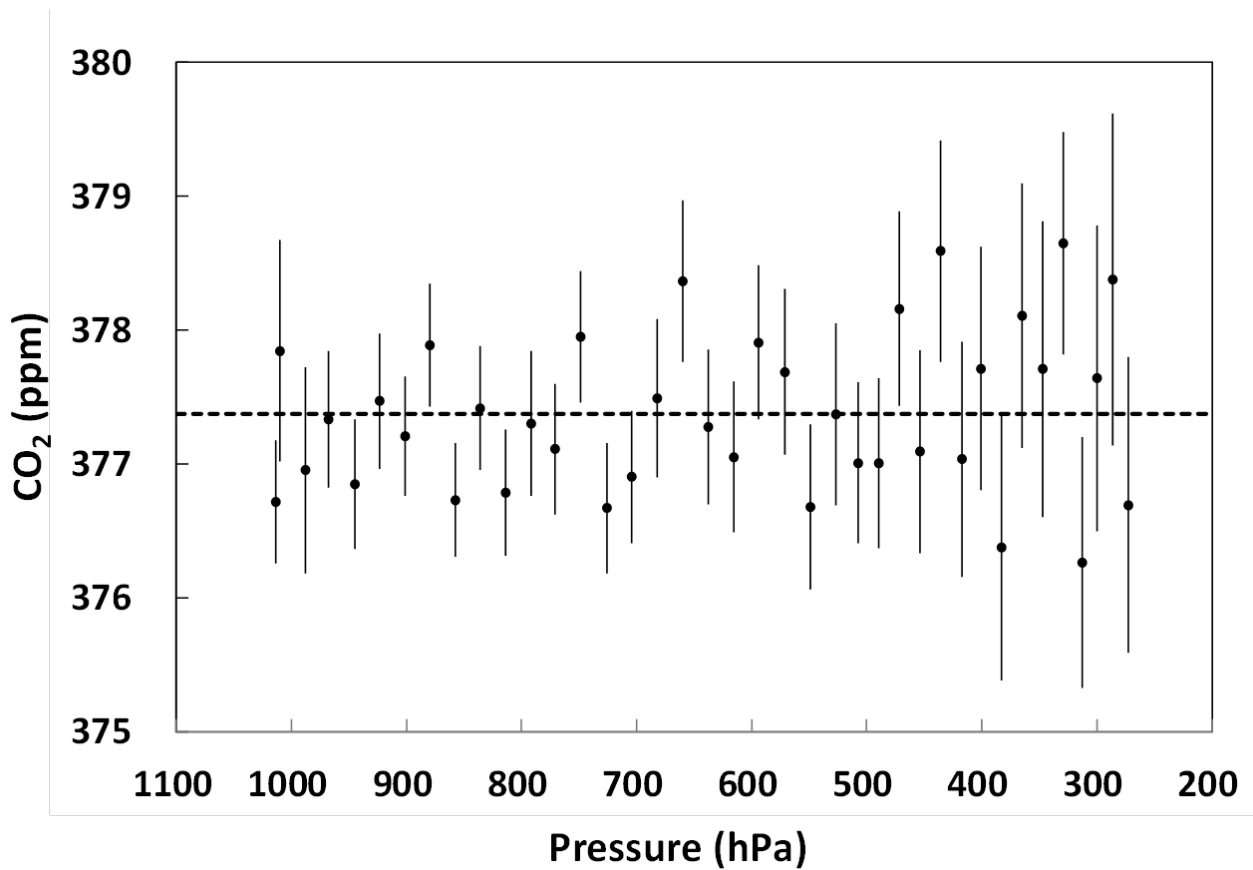


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799 **Figure 4.** $[I(4.0) - I(4.3)] / P$ values versus CO₂ mole fraction, where $I(4.0)$ and $I(4.3)$ are the
800 signal intensities at the 4.0 μm wavelength for background measurements and the 4.3 μm wavelength
801 for CO₂ absorption measurements, obtained by the NDIR CO₂ sensor, and P is the ambient
802 atmospheric pressure. CO₂ mole fractions were measured with a standard NDIR instrument (LICOR,
803 LI-840A) connected to the balloon sensor in series. The pressure while carrying out the
804 measurements was constant at 1010 hPa.

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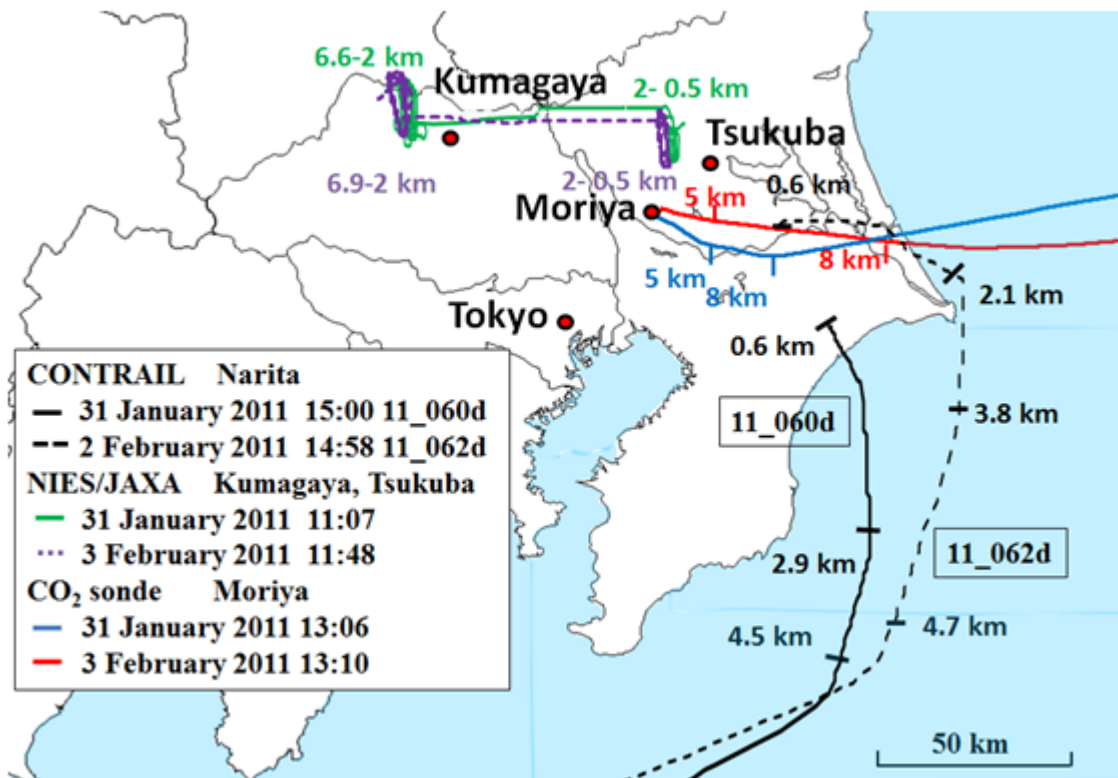
809 **Figure 5.** Results of a chamber experiment of the CO₂ sonde. Pressure in the chamber was reduced
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 812 chamber, which was obtained from the interpolation of the standard gas values in each calibration
 813 cycle. Vertical error bars indicate the square-root of sum of squares for the standard deviations of
 814 the sample and standard gas signals at each step in the calibration cycle. The black dashed line shows
 815 an average of all the values obtained for the sample gas. See the text for more details.

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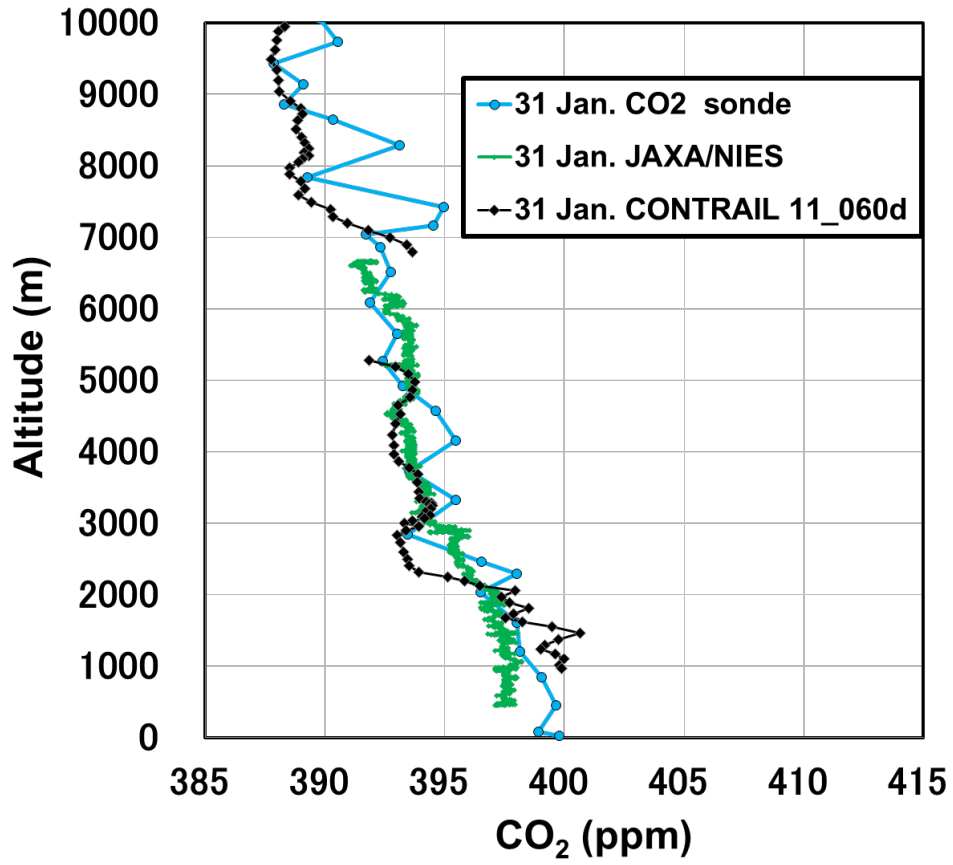


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 824 (black solid line) and 11_062d data on February 2nd, 2011 (black dashed line) from Hong Kong to
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 826 February 3rd (purple dotted line). The altitudes of the flight paths are also indicated.

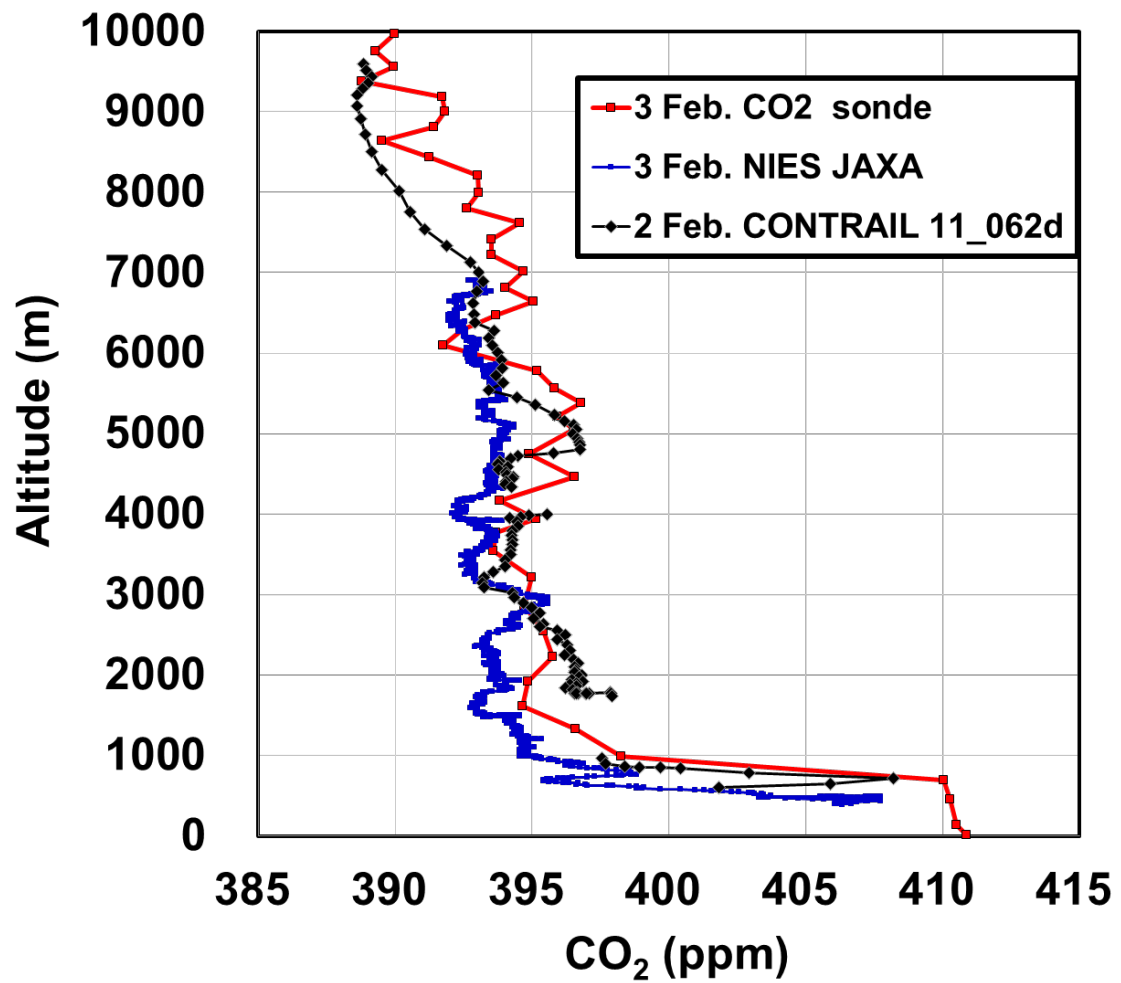
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829 **Figure 7.** The CO₂ vertical profiles obtained by the CO₂ sonde (circles connected with blue lines),
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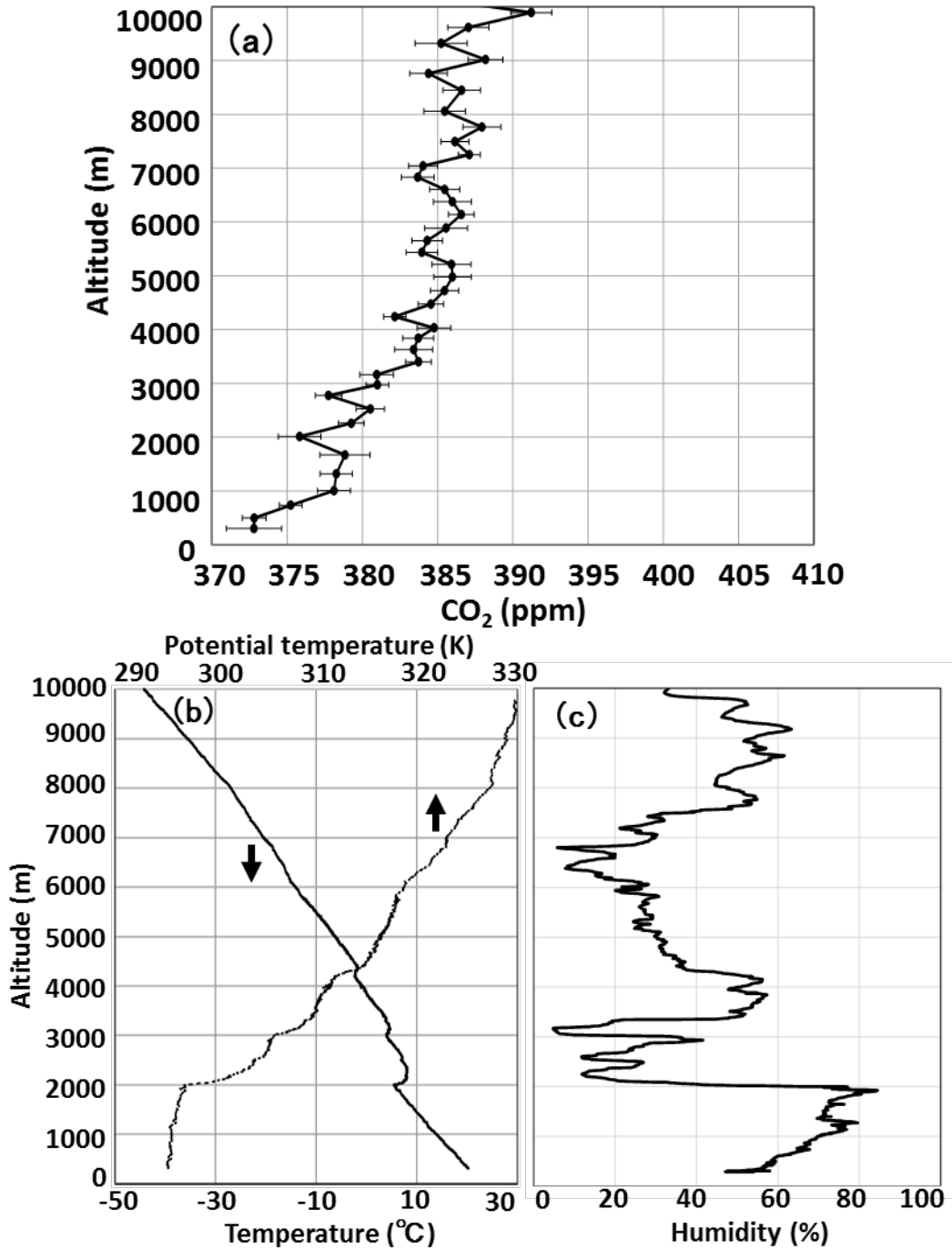
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835 **Figure 8.** The CO₂ vertical profiles obtained by the CO₂ sonde (circles connected with red lines),

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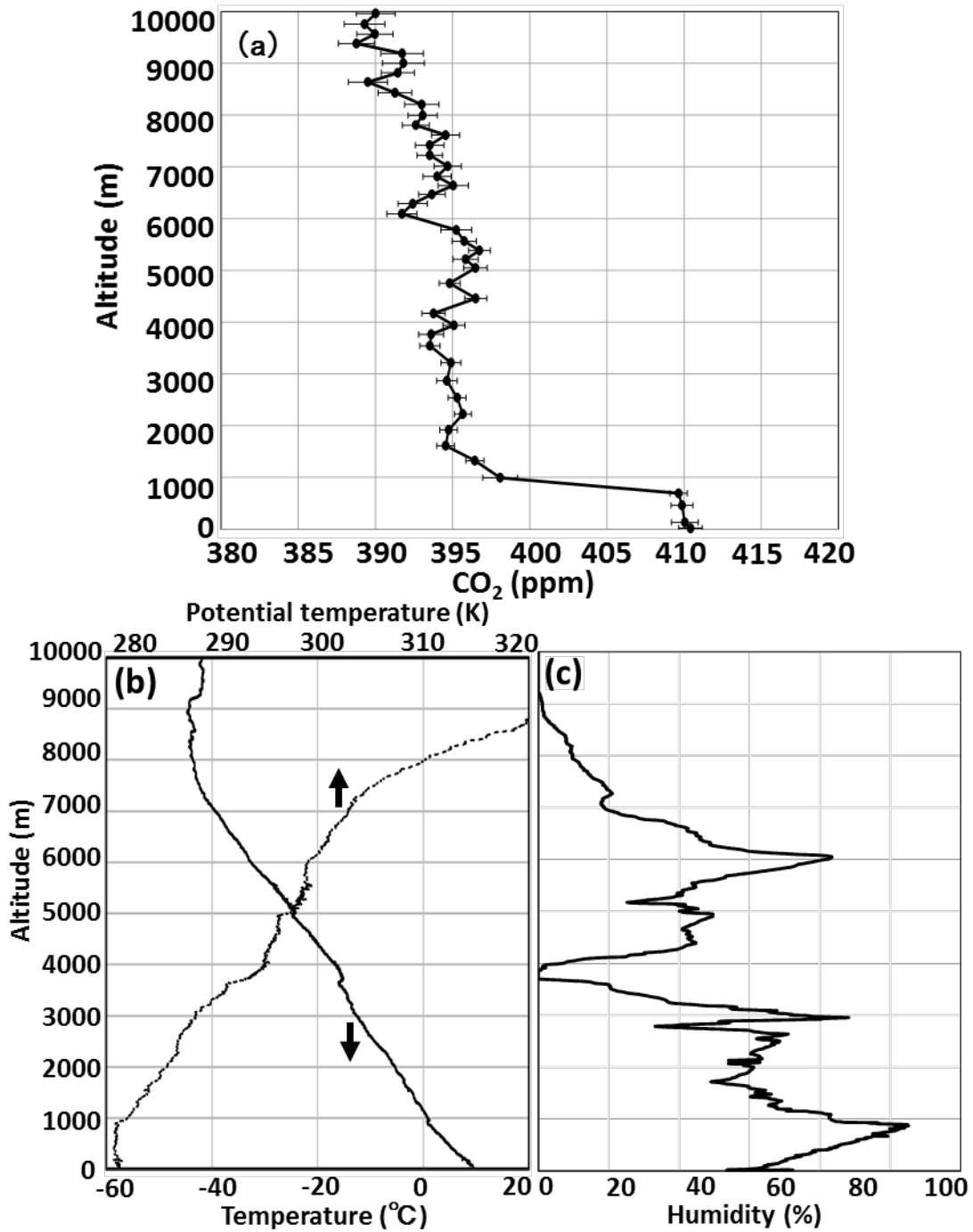
837 CONTRAIL data (diamonds connected with black lines) on February 2nd, 2011.



839

840 **Figure 9.** Profiles of (a) CO₂ mole fraction, (b) temperature (solid line) and potential temperature
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 848