



# Intercomparison of nitrous acid (HONO) measurement techniques in a megacity (Beijing)

Leigh R. Crilley<sup>1</sup>, Louisa J. Kramer<sup>1</sup>, Bin Ouyang<sup>2</sup>, Jun Duan<sup>3</sup>, Wenqian Zhang<sup>4</sup>, Shengrui Tong<sup>4</sup>, Maofa Ge<sup>4</sup>, Ke Tang<sup>3</sup>, Min Qin<sup>3</sup>, Pinhua Xie<sup>3</sup>, Marvin D. Shaw<sup>5,6</sup>, Alastair C. Lewis<sup>5,6</sup> Archit Mehra<sup>7</sup>, Thomas J. Bannan<sup>7</sup>, Stephen D. Worrall<sup>7a</sup>, Michael Priestley<sup>7 b</sup>, Asan Bacak<sup>7</sup>, Hugh Coe<sup>7</sup>, James Allan<sup>7,6</sup> Carl J.

Percival<sup>7</sup><sup>c</sup>, Olalekan A. M. Popoola<sup>8</sup>, Roderic L. Jones<sup>8</sup> and William J. Bloss<sup>1</sup>.

<sup>1</sup>School of Geography, Earth and Environmental Science, University of Birmingham, Edgbaston, Birmingham, B15 2TT, UK <sup>2</sup>Lancaster Environment Centre, Lancaster University, LA1 4YQ, UK

10 <sup>3</sup>Key Laboratory of Environment Optics and Technology, Anhui Institute of Optics and Fine Mechanics, Chinese Academy of f Sciences, Hefei, 230031, China <sup>4</sup>Beijing National Laboratory for Molecular Sciences (BNLMS). State Key Laboratory for Structural Chemistry of Unstable

<sup>4</sup>Beijing National Laboratory for Molecular Sciences (BNLMS), State Key Laboratory for Structural Chemistry of Unstable and Stable Species, CAS Research/Education Center for Excellence in Molecular Sciences, Institute of Chemistry, Chinese Academy of Sciences, Beijing 100190, China

<sup>5</sup>Wolfson Atmospheric Chemistry Laboratories, University of York, Heslington, York, YO10 5DD, UK
 <sup>6</sup>National Centre for Atmospheric Science, UK
 <sup>7</sup>Centre for Atmospheric Science, School of Earth and Environmental Sciences, University of Manchester, Manchester, M13
 9PL, UK
 <sup>8</sup>Department of Chemistry, University of Cambridge, Cambridge, CB2 1EW, UK

<sup>a</sup> now at Chemical Engineering and Applied Chemistry, School of Engineering and Applied Sciences, Aston University, Birmingham, B4 7ET, UK
 <sup>b</sup> now at: Atmospheric Science, Department of Chemistry and Molecular Biology, University of Gothenburg, Gothenburg

Sweden <sup>c</sup> now at Jet Propulsion Laboratory, 4800 Oak Grove Drive, Pasadena, CA 91109, USA

25

5

Correspondence to: William J. Bloss (w.j.bloss@bham.ac.uk)

**Abstract.** Nitrous acid (HONO) is a key determinant of the daytime radical budget in the daytime boundary layer, with quantitative measurement required to understand OH radical abundance. Accurate and precise measurements of HONO are therefore needed; however HONO is a challenging compound to measure in the field, in particular in a chemically complex

- 30 and highly polluted environment. Here we report an inter-comparison exercise between HONO measurements performed by two wet chemical techniques (the commercially available LOPAP and a custom-built instrument) and two Broadband Cavity Enhanced Absorption Spectrophotometer (BBCEAS) instruments at an urban location in Beijing. In addition, we report a comparison of HONO measurements performed by Time of Flight Chemical Ionization Mass Spectrometer (ToF-CIMS) and Syft Proton Transfer Reaction Mass Spectrometer (PTR-MS) to the more established techniques (wet chemical and BBCEAS).
- 35 The key finding from the current work was that all instruments agree on the temporal trends / variability in HONO ( $r^2 > 0.97$ ), yet displayed some divergence in absolute concentrations, with the wet chemical methods consistently higher than the BBCEAS systems by between 12 and 39%. We found no evidence for any systematic bias in any of the instruments, with the exception of measurements near instrument detection limits. The causes of the divergence in absolute HONO concentrations were unclear, and may in part have been due to spatial variability, i.e. differences in instrument location / inlet position.





#### **1** Introduction

15

Nitrous acid (HONO) is one of the key daytime sources of radicals in the boundary layer, as it readily undergoes photolysis to form the OH radical, contributing up to 40% of the OH budget in London (Lee et al., 2016). The OH radical is the primary

- 5 oxidant in the troposphere that drives chemical processing, principally the oxidation of volatile organic compounds (VOC) that lead to the formation of ozone and secondary organic aerosols. There are a number of known sources of HONO including direct emissions, heterogeneous reactions, homogenous gas-phase reactions, biological processes and surface photolysis (see reviews by (Kleffmann, 2007;Spataro and Ianniello, 2014). Across urban areas, high spatial heterogeneity in HONO concentration can be observed depending on the proximity to emission sources (Crilley et al., 2016;Lee et al., 2013). Despite,
- 10 the importance of HONO to the overall radical budget, the contributions of these different sources, particularly in the urban environment, are poorly understood (See e.g. (Lee et al., 2016;Michoud et al., 2014)).

As a result of the significance of HONO to tropospheric photochemistry, accurate and precise concentration measurements are required, but are challenging due to a number of known potential artefacts in the available approaches: Positive artefacts can occur in inlet lines, as HONO is easily formed through heterogeneous reactions on wet surfaces (Zhou et al., 2003).

- Furthermore, the highly reactive nature of HONO means that within inlet lines wall interactions could also lead to a negative artefact, unless inert materials are employed (Pinto et al., 2014). Other challenges include interferences from species such as NO<sub>2</sub>. There are a number of approaches to measure HONO that can be classed as either wet chemical, spectroscopic techniques or off-line methods (Spataro and Ianniello, 2014). Some of the main instrumentation used recently to measure ambient HONO
- 20 in the literature include differential optical absorption spectroscopy (DOAS, e.g. (Perner and Platt, 1979)), wet chemical techniques (e.g. LOng Path Absorption Photometer (LOPAP), (Heland et al., 2001)), broadband cavity enhanced absorption spectroscopy (e.g. (Duan et al., 2018)), soft chemical ionization mass spectrometry (CIMS, e.g. (Veres et al., 2015)), on-line ion chromatography (e.g. (Stutz et al., 2010;Cheng et al., 2013)) and wet denuder (e.g. (Acker et al., 2004)). In order to compare reported measurements across studies, it is necessary to understand how the different approaches/techniques compare relative
- 25 to each other, under actual ambient (field) conditions.

There have been a number of studies reporting intercomparisons between HONO instrumentation (e.g. (Stutz et al., 2010;Ródenas et al., 2013;Pinto et al., 2014;Kleffmann et al., 2006)). Generally, HONO measurements by DOAS systems are used as a reference during inter-comparison studies, as optical methods are artefact free with respect to sampling method,

30 though impurities in the HONO and NO<sub>2</sub> reference spectra can affect retrievals (Stutz et al., 2010;Kleffmann et al., 2006). A further complication with using DOAS systems as a reference is that the spatial averaging inherent in the system means that comparison with point measurements may be subject to bias due to spatial heterogeneities in HONO concentrations. Typically,





most in-situ instruments report higher concentrations during the day compared to simultaneous measurements by a DOAS system, thought to be due to the positive interferences in the in-situ techniques (see e.g. (Febo et al., 1996;Appel et al., 1990;Stutz et al., 2010;Spindler et al., 2003)). An exception is the work by Kleffmann et al. (2006), who reported excellent agreement between a LOPAP and DOAS system in both chamber based and field measurements of HONO under both day and night conditions. The reason for the better performance of the LOPAP is the two channel stripping coil employed in the LOPAP

5 night conditions. The reason for the better performance of the LOPAP is the two channel stripping coil employed in the LOPAP successfully corrects for positive artefacts and chemical interferents during measurement, as demonstrated by Kleffmann and Weisen (2008).

Recently, there have been multi-instrumental inter-comparison studies performed in ambient air and in simulation chambers.
These include the Formal Intercomparison of Observations of Nitrous Acid (FIONA) project, which involved 18 instruments measuring within the European Photoreactor (EUPHORE) chamber over a wide range of HONO concentrations (Ródenas et al., 2013). While in general, good agreement was observed during the different experiments of FIONA, at high concentrations (>15 ppb) divergence was observed between some instruments possibly due to some instruments experiencing saturation (Ródenas et al., 2013). These high HONO concentrations however are not typical even in highly polluted locations like Beijing.

- 15 Pinto et al. (2014) described an inter-comparison of field measurements performed in Houston using a number of instruments for measuring HONO. The instruments tested included a long-path DOAS, a number of wet chemical (including a LOPAP), on-line ion-chromatography and a Time of Flight Chemical Ionisation Mass Spectrometer (ToF-CIMS) using iodide as a reagent ion CIMS. Overall, while good agreement between all the instruments was observed in terms of the temporal trends, the absolute concentrations varied. Pinto and co-workers were unable to pinpoint the cause of the disagreement in absolute
- 20 concentrations but speculated it might have been due to chemical interference in the in-situ techniques, and the effect of heterogeneous surface reactions due to the distance between some inlets.

Here, we report an inter-comparison exercise of co-located wet chemical and broadband cavity enhanced absorption spectrophotometer (BBCEAS) instruments for measuring HONO in an urban location within Beijing. Ambient concentrations of HONO can vary by several orders of magnitude in Beijing, up to 9 ppb during haze events with a typical concentration of 1.44±1.33 ppb (Wang et al., 2017), making it a challenging location for field measurements. In addition, we report a comparison of HONO retrievals by a ToF-CIMS and Syft PTR-MS to the more established techniques (wet chemical and BBCEAS) for measuring HONO. Based on the inter-comparison findings, the factors that may have influenced the measured

concentrations are investigated.





## 2 Method

5

## 2.1 Site description

Measurements formed part of the Air Pollution and Human Health in a Chinese megacity (APHH-Beijing, <u>www.aphh.org.uk</u>) program and of the 'An integrated study of air pollution processes in Beijing' (AIRPRO) project, which aimed to understand atmospheric processes affecting air pollution in Beijing. An overview of the APHH-Beijing project is provided in Shi et al. (2018). Measurements were performed at the Chinese Academy of Sciences' Institute of Atmospheric Physics (IAP) tower campus, an urban site located near the 4<sup>th</sup> ring road in the northern suburbs of Beijing. There were two field campaigns, the

## 2.2 Instrument descriptions

10 An overview of all the instruments that measured ground-level HONO at IAP is provided in Table 1. As instruments of the same type were used in this study, throughout this paper the instruments will be referred to by their institution name, as per Table 1. A brief description of each instrument follows.

first took place during Nov-Dec 2016 (referred to as winter) and second during May-June 2017 (referred to as summer).

#### 2.2.1 University of Birmingham LOPAP

The University of Birmingham operated a Long Path Absorption Photometer (LOPAP-03, QUMA Elektronik & Analytik GmbH) at IAP, referred to as the LOPAP throughout. The LOPAP is a wet chemical technique and has been described in detail in Heland et al. (2001) and Kleffmann et al. (2002). Briefly, a stripping coil is used to sample gas-phase HONO into an acidic solution where it is derivatized into an azo dye. The light absorption of the azo dye, principally at 550 nm (though higher wavelengths can also be used), is then measured with a spectrometer using an optical path length of 2.4 m. The LOPAP was operated and calibrated according to the standard procedures described in Kleffmann and Wiesen (2008). The time resolution

20 of the LOPAP was 5 minutes and baseline measurements were taken at frequent intervals (8hrs). The operationally defined detection limit (2σ) of the LOPAP was calculated to be 35 and 5 ppt for winter and summer, respectively and varied due to changes in purity of reagents and zero air used. The LOPAP was housed within a temperature controlled shipping container and sampled at a height of 3m above ground level.

#### 2.2.2 Institute of Chemistry, Chinese Academy of Sciences wet chemical HONO analyser

25 ICCAS applied a custom-built instrument, described in detail elsewhere (Hou et al., 2016). It is a wet chemical technique similar in principle to the LOPAP. Gas-phase HONO is almost completely absorbed by an absorption solution into a two-channel stripping coil, where it forms an azo dye, detected by absorption spectroscopy at a wavelength of 550 nm with an optical path length of 0.5 m. The instrument has a detection limit ( $2\sigma$ ) of 134 ppt for a response time of 5 min.





## 2.2.3 University of Cambridge BBCEAS

The University of Cambridge ran a three-channel BBCEAS instrument during the campaign, with one channel measuring  $NO_2$  and HONO simultaneously in the UV (362-374 nm) wavelength region. Reference absorption cross-sections of HONO (Stutz et al., 2000) and NO2 (Voigt et al., 2002), were fitted to the absorption coefficient to retrieve HONO and NO2 concentrations

- 5 Details of the instrument can be found in Kennedy et al. (2011). Two mirrors (Layertec 109053) with peak reflectivity of ~99.95% at 365 nm were used to form the cavity. Given an inter-mirror distance of 92 cm, the effective absorption pathlength in the case of an empty cavity was around 1.8 km. The limit of detection (1σ) was determined to be 25 ppt for a 60 s time response.
- 10 Both instrument inlet and the optical cavity were made of perfluoroalkoxy alkane (PFA) which is well known for its chemical inertness. During the winter phase of the campaign, instrument inlet was placed at the top of the container and was about 3 m from ground. During the summer phase however, the instrument was moved to an adjacent container, also housing the AIOFM instrument, and the height of instrument inlet was changed accordingly from ~3 to ~2 m.
- 15 To allow a more stable cavity throughput (i.e. to minimise flow turbulence effect on the optical signal), the sampling flow was set to 2 LPM for the HONO/NO<sub>2</sub> measurement channel. This was close to the very low end of the operational range of the flow controller (0-50 LPM), leading to the actual flow rate potentially differing from that set. Post-campaign analysis identified that this affected both the dilution factor (dilution of the sample flow by the two mirror purge lines) and the length the sample gas occupied in the cavity. A post-campaign calibration was therefore performed by injecting a known amount of NO<sub>2</sub> into the
- 20 cavity under otherwise identical operating conditions, and a scaling factor of  $\sim 1.27$  was found to be necessary to account for these two factors, and was then applied to the measured NO<sub>2</sub> and HONO concentrations.

## 2.2.4 Anhui Institute of Optics and Fine Mechanics BBCEAS

The custom-built BBCEAS instrument from the Anhui Institute of Optics and Fine Mechanics (AIOFM), Chinese Academy of Sciences has been described in detail in Duan et al. (2018); therefore only a brief description is given here. Light is emitted by a single light emitting diode with peak wavelength of 365 nm, FWHM of 13 nm and is introduced into the resonant cavity, consisting of a pair of high reflective (HR) mirrors with reflectivity of about 0.99985 at 368 nm, separated by 55.0 cm. The emergent light intensity passing through the cavity exit mirror is received by an Ocean Optics QE65000 spectrometer through an optical fibre with 600 µm diameter, and a 0.22 numerical aperture.

30 In order to avoid the drift of the center wavelength of the LED, the temperature of the LED was controlled to be approximately  $20 \pm 0.05$  °C by using a TEC unit. In order to prevent particulate matter from entering the cavity and reducing the effect of particulate matter on the effective absorption path, a 1 µm PTFE filter membrane (Tisch Scientific) was used in the front end





of the sampling port. The time resolution of the IBBCEAS instrument was 1 min, and the  $2\sigma$  detection limit of HONO was about 120 pptv. The fitting wavelength range was selected as 359–387 nm, with the same reference cross-sections used in the retrieval of NO<sub>2</sub> and HONO as for the University of Cambridge instrument. Sample loss and secondary formation of HONO were both considered in this instrument and the measurement error of HONO was estimated to be approximately 9%.

## 5 2.2.5 University of Manchester ToF-CIMS

A time of fight chemical ionisation mass spectrometer (ToF-CIMS) (Lee et al. 2014), using an iodide ionisation system was coupled with a filter inlet for gases and aerosols (FIGAERO) originally developed by Lopez-Hilfiker et al. (2014) and recently described and characterised by Bannan et al. (2019). The detailed setup during this campaign can be found in Zhou et al. (2018). The FIGAERO enabled near simultaneous, real-time measurements of both the gas and particle phase composition.

10 Only gas phase data is presented here, so of every 75 minutes of continuous data 35 minutes (particle phase mode) are omitted. The gas phase inlet consisted of 5 m <sup>1</sup>/<sub>4</sub>" I.D. PFA tubing connected to a fast inlet pump with a total flow rate of 13 standard litres per minute (slm) from which the ToF-CIMS sub-sampled 2 slm.

Methyl iodide gas mixtures (CH<sub>3</sub>I) in N<sub>2</sub> were made up in the field using a custom-made manifold (Bannan et al. 2014). 20
standard cubic centimetres per minute (sccm) of the CH<sub>3</sub>I mixture was diluted in 4 slm N<sub>2</sub> and ionised by flowing through a Tofwerk X-ray ionisation source. This flow enters an ion molecule region (IMR) which was maintained at a pressure of 400 mbar using an SSH-112 pump fitted with an Aerodyne pressure control box to account for changes in ambient pressure. A short segmented quadrupole (SSQ) was positioned behind the IMR and was held at a pressure of 2 mbar using a Tri scroll 600 pump.

During the campaign, gas phase backgrounds were established through regularly overflowing the inlet with dry  $N_2$  for five continuous minutes every 45 minutes (Priestley et al. 2018) and were applied consecutively. The overflowing of dry  $N_2$  will have a small effect on the sensitivity of the instrument for those compounds whose detection is water dependent. Here we find that due to the very low instrumental background for HONO, the absolute error remains small (<33 ppt) and is an acceptable limitation in order to measure the wide range of different compounds that can be detected using the iodide CIMS.

- 25 limitation in order to measure the wide range of different compounds that can be detected using the iodide CIMS. Field calibrations were regularly carried out using known concentration formic acid gas mixtures made in the manifold. The instrument was calibrated for a range of other species after the campaign, and relative calibration factors were derived using the measured formic acid sensitivity as has been performed previously (Le Breton et al., 2014;Le Breton et al., 2017;Bannan et al., 2014;Bannan et al., 2015). HONO was measured at m/z 174 as I.HNO<sub>2</sub><sup>-</sup> during the period of 27<sup>th</sup> May – 17<sup>th</sup> June 2017.
- 30 A stable and pure gas phase source of HONO was generated for calibrations using the method described by Ren et al. (2010) and Febo et al. (1995) and a sensitivity of 0.28 cps/ppt was applied to the data with a LOD of 33 ppt. Data analysis is performed using the "Tofware" package (version 2.5.11) running in the Igor Pro (WaveMetrics, OR, USA) environment. The mass axis was calibrated using I<sup>-</sup>, I<sub>2</sub><sup>-</sup> and I<sub>3</sub><sup>-</sup>. Extracted high resolution time series were then normalised to the iodide reagent ion trace.

<sup>20</sup> 





5

## 2.2.6 University of York Syft PTR-MS

The data presented in this paper were measured using a Voice200 Selected ion flow tube mass spectrometer (SIFT-MS, Syft Technologies, Christchurch, New Zealand). This instrument consists of a switchable reagent ion source capable of rapidly switching between multiple reagent ions. Reagent ions are generated in a microwave discharge ion source region, acting on an air/water mix at a pressure of approximately 440 mTorr to generate  $H_3O^+$ ,  $NO^+$ , and  $O_2^+$ . These ions are extracted into the upstream quadrupole chamber maintained at a pressure of approx. 5 x10<sup>-4</sup> torr, using a 70 L/s turbo-molecular pump. The

- reagent ions pass through an array of electrostatic lenses and the upstream quadrupole mass filter, and those not rejected by the mass filter are passed into the flow tube where they are carried along in a stream of nitrogen and selectively ionise target analytes. Gas phase data presented herein was determined using the  $H_3O^+$  reagent ion only. Sampling was carried out at a
- 10 height of approx. 3 m using a gas phase inlet consisting of 3.5 m ¼" I.D. PFA tubing connected to a diaphragm inlet pump (KNF) at a total flowrate of 5 standard litres per minute (slpm), from which the SIFT-MS sampled approximately 2 slpm through an in house built pressure controlled inlet maintaining a consistent absolute inlet pressure of 0.5 bar. The flow tube is pumped by a 35 m³/h scroll-type dry pump (Edwards) resulting in a mass flow controlled gas flow of 25 sccm for the nitrogen carrier gas (research grade, BOC) and a sample flow of 100 sccm taken from the pressure controlled inlet system. These flows
- 15 result in a continuous total flow tube pressure of approximately 460 mTorr and a reaction time of approx. 8 ms (Hera et al., 2017). During the campaign, gas phase backgrounds were established through regularly overflowing the sample inlet with dry nitrogen for five continuous minutes every hour.

Nitrous acid (HONO) was measured at product ion m/z 48 (H<sub>2</sub>ONO<sup>+</sup>) following protonation using the H<sub>3</sub>O<sup>+</sup> reagent ion. As a
HONO calibration source was not available, a relative transmission curve calibration (between 19-107 m/z) was carried out daily as describe by Taipale and colleagues (Taipale et al., 2008) using a 2 ppm gas standard containing ethylene, Isobutane, benzene, toluene, ethylbenzene, tetraflourobenzene, hexafluorobenzere and octafluoro benzene.

## 2.3 Formal inter-comparison

The formal inter-comparison measurements took place during the 9<sup>th</sup>-14<sup>th</sup> November 2016, and comprised the four established techniques for measuring HONO (2 wet chemical and 2 BBCEAS). All instruments had a sampling height of 3m during the inter-comparison, and inlets were located as close as possible to each other. The BHAM and ICCAS instrument were housed within the same shipping container, with their respective inlet heads located beside each other on the roof. The CAM and AIOFM BBCEAS instruments were housed in separate containers, with inlets located approximately 5 and 10 m, respectively, from the two wet chemical inlet heads. At the completion of the formal winter intercomparison, the inlet locations changed for some of the instruments.





There was no formal inter-comparison between all four instruments in the summer campaign. The BHAM, CAM and ICCAS inlets were located in the same position as per the winter intercomparison at the start (22nd May - 30<sup>th</sup> June 2017). Therefore, further analysis was performed between these three instruments for this period to examine for any changes in their relationships compared to the winter measurements. The AIOFM instrument was housed within the same container as per the winter however, the inlet was located approximately 3 m further away from other instruments in the summer. On the 30<sup>th</sup> May, the

5 however, the inlet was located approximately 3 m further away from other instruments in the summer. On the 30<sup>th</sup> May, the CAM instrument was moved to the same container as the AIOFM, with the inlets located approx. 3 m from each other. The ToF-CIMS and Syft PTR-MS were not initially set up to measure HONO at IAP, but were able to provide some useful data during the summer measurements and are therefore compared to the more established techniques.

#### 2.4 Data Analysis

10 The BHAM and ICCAS instruments were operated with a time response of 5 min, and as this was the longest (Table 1), 5 min averages were used for all instruments in the inter-comparison analyses. For each instrument, their normal quality control procedures were applied and only data that passed the quality control was used for subsequent analysis. Data analysis was performed in R (v 3.5.1) using the openair package (Carslaw and Ropkins, 2012) and the Imodel2 package for Reduced Major Axis (RMA) regression analysis.

#### 15 3 Results

## 3.1 Winter formal intercomparison

## 3.1.1 Time series

The time series (Figure 1) demonstrates that while all instruments captured the same temporal trends, the absolute concentrations differed. The correlation coefficients from regression analyses show that there is little scatter between 20 measurements from the different instruments with r values being consistently between 0.96 and 0.98 (Table 2). Overall, the BHAM LOPAP measurements were consistently the highest, followed by ICCAS, AIOFM and CAM. The slopes from the RMA analysis demonstrated that none of the instruments were in agreement (Table 2) within their stated error (Table 1) during the formal inter-comparison exercise. Therefore, in the following sections we investigate possible reasons to account for the lack of agreement between instruments.

#### 25 3.1.2 Analysis of Co-efficient of Variance

The co-efficient of variance (CV) is defined as the standard deviation divided by the mean and is used to compare the relative degree of variation between datasets. Therefore, the CV is a measure of precision, and as a guide a CV of 0.1 is considered as acceptable by the US EPA for PM instruments (Sousan et al., 2016). From Fig 2, the CV was fairly consistent throughout the winter inter-comparison, at an average of  $0.28\pm0.07$ . The CV was however observed to increase at the end of the inter-





comparison, coinciding with period of the lowest mean HONO concentration (< 1 ppb, Fig 2) suggesting that as the concentrations approached the detection limit (DL) of some instruments (Table 1), there was a decrease in the level of agreement. NO<sub>2</sub> is a potential interferent for the measurement of HONO for both wet chemical and BBCEAS instruments (Heland et al., 2001). For the BBCEAS instruments, the fitting algorithm used should isolate HONO and NO<sub>2</sub> very well in

- 5 theory; however, the applied reference NO<sub>2</sub> spectrum might contain absorption signatures from HONO. This would result in a higher NO<sub>2</sub>/HONO ratio retrieved from the BBCEAS compared to ambient air, and consequently reporting a low HONO mixing ratio (Kleffmann et al., 2006). However, the Voigt et al (2002) NO<sub>2</sub> cross-section, used by both BBCEAS instruments, has previously been shown to have negligible HONO absorption structures (Veitel, 2002). We also note that HONO reference spectrum should contain little structure from NO<sub>2</sub>. Fig 2, however, demonstrates no apparent relationship between the CV and
- 10 NO<sub>2</sub>. This likely reflects the efforts taken during processing and measurement to reduce the influence of interference from NO<sub>2</sub> in all instrument types.

#### 3.1.3 Normalised difference analysis

The systematic error for each instrument can be calculated by normalised sequential difference (NSD) according to Eqn 1 (Arnold et al., 2007):

15 
$$NSD = \frac{(Conc_t - Conc_{t+1})}{(Conc_t \times Conc_{t+1})^{0.5}}$$
(1)

The results are shown in Fig S1 (Supporting Information), and as each instrument showed a symmetrical and Gaussian distribution it suggests there was no internal systematic bias for any given instrument. We then examined the normalized difference (ND) between pairs of instruments to explore inter-instrument variability, calculated according to Eqn 2 (Pinto et al., 2014):

20 
$$ND_{ij} = \frac{(c_i - c_j)}{(c_i + c_j)}$$
 (2)

where  $C_i$  and  $C_j$  denote HONO levels measured by any pair of instruments (BHAM, ICCAS, AIOFM or CAM). We also calculated the co-efficient of divergence (CD), which is a normalized measure of the similarity between two measurement time series, derived via Eqn 3 (See Pinto et al. (2014) and references therein):

$$CD_{ij} = \sqrt{\left(\frac{1}{p}\right) \times \sum \left(ND_{ij}\right)^2}$$
(3)

25 Where p is the number of observations and ND<sub>ij</sub> is defined in Eqn 2. A CD of one means the time series are completely different, while of CD of zero indicates that they are identical. The calculated CD for each instrument pair is shown in Table 3 and demonstrate that each of the two overall approaches - wet chemical (BHAM and ICCAS) and BBCEAS (AIOFM and CAM) - agreed well internally. The ICCAS and AIOFM also agreed well, but CAM and BHAM had a higher CD with AIOFM and ICCAS (Table 3).





5

10

If there is no difference between a pair of instruments, then the calculated ND should be scattered around zero, and from Fig 3 this was not observed for any instrument pair, pointing to differences between instruments. The ND was evaluated as function of wind direction and measured HONO concentration (Fig 3), to explore if ambient concentration or spatial heterogeneity could explain the disagreements. From Figure 3, for all instrument pairs the highest ND, and therefore largest relative difference, between instruments was at low HONO mixing ratios (ca. <1 ppb) and was also associated with a westerly direction. At high wind speeds, the ND was also high between all instrument pairs (Fig S2, Supporting Information). As we observed high ND at relatively high wind speeds, it would suggest that spatial variability in ambient HONO concentrations did not affect the intercomparison as high wind speeds typically homogenise ambient concentrations from point and local sources. Overall from Figs 3 and S2, the periods of low HONO concentration, high wind speeds and westerly winds all coincided during the formal winter intercomparison making it difficult to disentangle the influence of these factors on the observed ND.

#### 3.1.4 Instrument agreement at low concentrations

There was evidence from the CV (Fig 2) and ND (Fig 3) analyses that the level of agreement between instruments decreased at low HONO mixing ratios. Therefore we applied RMA correlation analysis for periods when the HONO level was below 2 ppb (as measured by CAM), and the results are shown in Table 4. From Table 4, the observed slopes between the BHAM-

15 ICCAS-AIOFM at low concentrations (<2 ppb) were similar to those for the whole winter inter-comparison dataset (Table 2). This suggests that the difference in measured concentrations between these instruments (as indicated by the slope) was not related to concentration. There was however a notable decrease in the slope for the low concentrations between CAM and the other three instruments compared to whole inter-comparison (Tables 2 and 4, respectively), potentially pointing to changes in the CAM readings at lower concentrations. This change may be related to differences in instrument sensitivity (Table 1).</p>

#### 20 3.2 Summer Measurements

While there was no formal intercomparison during the summer measurements, at the start of the summer measurements the BHAM, ICCAS and CAM instrument inlets were co-located as per the winter formal intercomparison. The relationship between instruments for this period is shown in Table 5. The agreement (gradient) between BHAM and ICCAS improved in the summer to 0.91 compared to winter (0.77) but with slight changes in intercept (Tables 2 and 5). A change was also observed

25 between CAM and ICCAS with a lower slope observed for the start of summer (Table 5) compared to winter intercomparison period (Table 2).

The AIOFM instrument started measuring halfway through the summer campaign, and while the instrument was housed in the same container the inlet location was a few meters further from the other three instruments than in the winter inter-comparison.

30 As a result, we compared the instrument readings for one week after the AIOFM instrument started measuring (7<sup>th</sup>-14<sup>th</sup> June 2017), with the results shown in Table 6. From Table 6, the agreement between instruments of the same type were within their stated uncertainties for the summer. However, when comparing the between the two different instrument types (wet chemical





and BBCEAS, e.g. AIOFM-BHAM, etc), the agreement was notably worse compared to the winter (Tables 2 and 6). The exception was that the agreement between the BHAM and CAM, which was similar in the winter and summer (Tables 2 and 5), despite the CAM inlet moving further away from BHAM inlet on the 30<sup>th</sup> May 2017 (Table 6).

- 5 Generally, the level of agreement between instruments varied between the summer and winter and this may reflect spatial variability in HONO concentrations as some of the instrument inlet locations varied from summer to winter. In the summer, the CAM inlet moved closer to the AIOFM inlet and the agreement between the two BBCEAS improved to be within uncertainty (Table 6). However, we also note that the BHAM and ICCAS inlets were in the same location winter and summer and yet the agreement between instruments changed considerably between the two measurement periods. We re-calculated the
- 10 ND for two inter-comparison periods analysed in the summer (Tables 4 and 5) and found no relationship between the ND and wind direction (Figures S4 and S5, Supporting Information). This suggests that during the summer measurements the wind direction may have exerted less influence on the spatial variability of the HONO levels or that observed relationship between wind direction and ND in winter was associative not causal.

## 3.2.1 Performance of MANC ToF-CIMS

- 15 Measurements from the Manchester ToF-CIMS are compared to the BHAM and CAM instruments for the summer campaign as these instruments had the best data coverage for periods when the MANC instrument was measuring, as well as representing the typical upper and lower measurements (Fig 1). In general the MANC instrument captured the temporal trends (r > 0.84) but recorded higher HONO concentrations than the other instruments (Table 7). Similar distributions were observed between the BHAM and MANC datasets, with the exception of a number of outliers for MANC (Figure 4). We note that MANC was
- 20 not co-located with either BHAM or CAM instrument and while this will likely have affected the inter-comparison, the results do point to the MANC instrument capturing the temporal trends but at a higher concentration than the other instruments (157-239%, Table 7).

#### 3.2.2 Performance of the YORK Syft PTR-MS

- The York Syft PTR-MS was primarily used for measuring VOC fluxes and so did not typically measure at ground level. To enable an inter-comparison with the other techniques, the YORK instrument measured at ground level for an extended single continuous period, from 1800 30/05/2017 until 0900 31/05/2017. The results are shown in Fig 5 and while the short time period and spatial distance between inlets (approx. 50m) limits the conclusions that can be drawn, it is clear that the YORK instrument captured the temporal trends (r of 0.9-0.96 compared to other techniques) and gave comparable concentrations to the BHAM instrument (slope of 0.78). This suggests that the YORK instrument was measuring HONO at reasonable concentrations. Furthermore, we note that a co-located PTR-MS (PTR-TOR 1000, Ionicon) was unable to see a HONO signal
- despite both instruments using  $H_3O^+$  to detect HONO.





## **4** Discussion

From the literature, the recent inter-comparison of ambient field measurements of HONO concentrations described by Pinto et al. (2014) is the most relevant to the current work. Overall, in their study Pinto and co-workers found that in general the level of disagreement between instruments was greater than the stated uncertainties for each instrument. While there was some

- 5 evidence for a chemical interference (but Pinto and co-workers could not identify the compounds responsible definitively), there were additional factors that also appeared to affect the inter-comparison. The best agreement in Pinto et al. (2014) was found for instruments with co-located inlets compared to instruments with inlets several meters apart and so points to spatial heterogeneity in HONO concentrations (possibly due a source on the roof surface) affecting the inter-comparison. Overall, the results from the current work are similar to that observed previously (Pinto et al. 2014), as there was a separation of up to 13
- 10 meters between some instrument inlets and this may have affected the results for the inter-comparison in the current work. With respect to photolysis, the lifetime of HONO at midday ranged from 17-300 and 9-33 min for winter and summer, respectively (depending upon weather / cloud cover / aerosol loading) and may have contributed to spatial heterogeneity in HONO concentrations.
- 15 Duan et al. (2018) presented results of an inter-comparison at a rural site in China between a BBCEAS and a LOPAP, with good agreement observed (slope of 0.94 and r<sup>2</sup> 0.89). The slope appeared to deviate from linearity above approximately 2 ppb suggesting that at higher concentrations the relationship was changing, as observed here between CAM and wet chemical techniques (BHAM and ICCAS) (Tables 2 and 4). A divergence in the measured concentrations at high concentrations was also observed for all instruments as part of the FIONA inter-comparison (Ródenas et al., 2013), but at much higher
- 20 concentrations (>15 ppb) than most of those encountered here. However, we did not observe such a change in relationship at high and low concentrations between the AIOFM-BHAM/ICCAS, suggesting that this result was not related to instrument type. Furthermore, as the measurements from Duan et al. (2018) were performed in a rural site, conditions may also be more homogenous mix compared to an urban location and this may explain why there was better agreement between the LOPAP and BBCEAS in their study compared to the current work.

25

Throughout this work, the wet chemical techniques generally measured higher concentrations than the spectroscopic techniques, in agreement with previous studies (e.g. Stutz et al., 2010; Pinto et al., 2014). Pinto et al. (2014) suggested the possible cause may be a positive chemical interference in the wet chemical instruments. The observed dependence of the normalized difference between each instrument pair on wind direction may reflect changes in composition affecting the

30 instrument readings. We note that the two-channel stripping coil used in the sampling inlet for both the BHAM and ICCAS instruments should account for any chemical interferences, particularly in the gas-phase (Kleffmann et al., 2002). Particle-phase nitrite is also expected to be corrected by the two-channel system, and we also note that based on the effective Henry's law constant for HONO and the typically acidic nature of Beijing aerosol that little particle-phase nitrite would be expected





(Kleffmann et al. 2006). Therefore, we do not expect there to be significant chemical interference on the BHAM and ICCAS wet chemical instruments.

In the current work, differences were observed between measurements from instruments of the same type (BBCEAS and wet chemical). The cause of this disagreement was difficult to pinpoint, but may reflect differences in calibration and corrections applied by each group. In particular the BHAM and ICCAS instruments inlets were next to each other during the intercomparison (<0.5m), and thus the differences likely reflect more differences in the operating conditions of the BHAM and ICCAS instruments. Both instruments used the same nitrite standard for calibration. Notably, there is a significant difference in DL between the instruments (Table 1) likely due to the different methodologies for determining baseline correction. For example, the BHAM instrument used zero air sampled at the inlet to determine the baseline, whereas ICCAS used water introduced into the wet chemical side of the instrument. Tests have shown that water results in a lower baseline measurement

for the LOPAP (approx. 80-100 ppt), however when comparing the two wet chemical methods this would only result in a change in the intercept, not the slope.

- The scaling factor to correct for the discrepancy in flow rate applied to the CAM instrument after the campaigns is unlikely to be the cause of the disagreement between the two BBCEAS. The two BBCEAS systems agree to within  $\pm 10\%$  for NO<sub>2</sub> measurements, and the larger disagreement for HONO (13%, Table 2) likely reflects higher spatial variability of ambient HONO compared to NO<sub>2</sub>, as the CAM and AIOFM inlets were the furthest apart during the formal winter inter-comparison. The agreement between the two BBCEAS decreased at lower concentrations and this may reflect differences in DL (Table 1).
- 20 The two BBCEAS instruments were found to be in better agreement in the summer compared to the winter. This may reflect the inlets being closer in summer compared to the winter, however there was still a distance of 3m between inlets. We do not know the reasons why the agreement between the AIOFM and CAM instruments changed in the summer compared to winter. Whilst there were some variability in path length and purge flows between the two BBCEAS systems, these are not thought to account for the discrepancy as they did not vary winter to summer. Furthermore, another factor may be losses/production of
- 25 HONO and NO<sub>2</sub> on the inlet filter and/or tubing losses/production that may also influence the intercomparison results as these were naturally different across systems due to different residence times (~3 s and ~0.5 s for the CAM and AIOFM BBCEAS, respectively). Laboratory tests using the same tubing material (Teflon) have however shown that neither the production nor the loss of HONO were significant in the CAM instrument (to less than a few percent) even at considerably longer inlet and cavity residence times, suggesting this was insignificant.
- 30

Generally, the agreement between instrument pairs varied from winter to summer, with the exception of CAM and BHAM instruments. As all instruments were operated and calibrated according to the same procedures in winter and summer, there were no changes in instrument operation that can explain these changes and as such, the cause is unclear. The concentrations





observed during summer (mean of  $1.2\pm0.9$  ppb (1 $\sigma$ )) were typically lower compared to the winter (mean of  $2.35\pm1.9$  ppb (1 $\sigma$ )), and this may have affected the results.

## **5** Conclusions

- Overall, from the winter inter-comparison all instruments were found to agree on the temporal trends and variability in HONO (r >0.97), yet displayed some divergence in absolute concentrations (slopes of 0.61-0.88), with the wet chemical methods consistently somewhat higher than the BBCEAS systems. We found no evidence for any systematic bias in any of the instruments, with the exception of measurements near instrument detection limits. There was evidence that the relationship between some instruments varied for the different measurement periods (e.g. winter/summer) however the reason for this change was unclear. When considering the mass spectrometric methods (MANC ToF-CIMS and YORK PTR-MS), these 0 cantured the temporal trends in HONO concentrations but were found to differ in absolute concentration relative to the other
- 10 captured the temporal trends in HONO concentrations but were found to differ in absolute concentration relative to the other instrumentation.

There was no evidence for a definitive cause of systematic bias between the four instruments during the formal HONO intercomparison, which might justify scaling or excluding results from one or more instruments. As a result, we could not say with

15 confidence, which instrument (if any) provided the 'correct' measurement of HONO concentration. Therefore, to meet the needs of the wider APHH-Beijing program for a single ground level HONO measurement, a merged HONO dataset was produced using the mean and range concentration of the four instruments that participated in the formal winter intercomparison (two wet chemical and two BBCEAS). This merged dataset will be used for future ground level analyses (e.g. model evaluation) across the APHH-Beijing program.

## 20 Authors contributions

Study conceived by BO and LC. Measurements were performed by LC, LK, BO, JD, WZ, MS, AM, TB, SW, JA and AB. Formal analysis performed by LC, LK and BO. All co-authors contributed to data curation. LC prepared the manuscript with contributions from all co-authors.

## Acknowledgements

25 This work was funded by the UK Natural Environment Research Council (NERC), Medical Research Council and Natural Science Foundation of China under the framework of Newton Innovation Fund (NE/N007190/1 and NE/N007077/1). WJB, LJK and LRC acknowledge additional support by the UK NERC through the project Sources of Nitrous Acid in the Atmospheric Boundary Layer (SNAABL, NE/M013405/1). Authors also acknowledge the support of staff at Institute of





Atmospheric Physics in Beijing, in particular Prof Pingqing Fu, during the measurement campaigns. Original data are available on request from the authors, and have been deposited in the (open access) CEDA repository, available for public download following the project embargo period.

## References

 Acker, K., Spindler, G., and Brüggemann, E.: Nitrous and nitric acid measurements during the INTERCOMP2000 campaign in Melpitz, Atmospheric Environment, 38, 6497-6505, https://doi.org/10.1016/j.atmosenv.2004.08.030, 2004.
 Appel, B., Winer, A., Tokiwa, Y., and Biermann, H. J. A. E. P. A. G. T.: Comparison of atmospheric nitrous acid measurements by annular denuder and differential optical absorption systems, 24, 611-616, 1990.

Arnold, J. R., Hartsell, B. E., Luke, W. T., Rahmat Ullah, S. M., Dasgupta, P. K., Greg Huey, L., and Tate, P.: Field test of
four methods for gas-phase ambient nitric acid, Atmospheric Environment, 41, 4210-4226, https://doi.org/10.1016/j.atmosenv.2006.07.058, 2007.

Bannan, T. J., Bacak, A., Muller, J. B. A., Booth, A. M., Jones, B., Le Breton, M., Leather, K. E., Ghalaieny, M., Xiao, P., Shallcross, D. E., and Percival, C. J.: Importance of direct anthropogenic emissions of formic acid measured by a chemical ionisation mass spectrometer (CIMS) during the Winter ClearfLo Campaign in London, January 2012, Atmospheric
15 Environment, 83, 301-310, https://doi.org/10.1016/j.atmosenv.2013.10.029, 2014.

- Bannan, T. J., Booth, A. M., Bacak, A., Muller, J. B. A., Leather, K. E., Le Breton, M., Jones, B., Young, D., Coe, H., Allan, J., Visser, S., Slowik, J. G., Furger, M., Prévôt, A. S. H., Lee, J., Dunmore, R. E., Hopkins, J. R., Hamilton, J. F., Lewis, A. C., Whalley, L. K., Sharp, T., Stone, D., Heard, D. E., Fleming, Z. L., Leigh, R., Shallcross, D. E., and Percival, C. J.: The first UK measurements of nitryl chloride using a chemical ionization mass spectrometer in central London in the summer of
- 20 2012, and an investigation of the role of Cl atom oxidation, 120, 5638-5657, doi:10.1002/2014JD022629, 2015. Bannan, T. J., Le Breton, M., Priestley, M., Worrall, S. D., Bacak, A., Marsden, N. A., Mehra, A., Hammes, J., Hallquist, M., Alfarra, M. R., Krieger, U. K., Reid, J. P., Jayne, J., Robinson, W., McFiggans, G., Coe, H., Percival, C. J., and Topping, D.: A method for extracting calibrated volatility information from the FIGAERO-HR-ToF-CIMS and its experimental application, Atmos. Meas. Tech., 12, 1429-1439, 10.5194/amt-12-1429-2019, 2019.
- Carslaw, D. C., and Ropkins, K.: openair An R package for air quality data analysis, Environmental Modelling & Software, 27–28, 52-61, http://dx.doi.org/10.1016/j.envsoft.2011.09.008, 2012.
   Cheng, P., Cheng, Y., Lu, K., Su, H., Yang, Q., Zou, Y., Zhao, Y., Dong, H., Zeng, L., and Zhang, Y.: An online monitoring

system for atmospheric nitrous acid (HONO) based on stripping coil and ion chromatography, Journal of Environmental Sciences, 25, 895-907, https://doi.org/10.1016/S1001-0742(12)60251-4, 2013.

30 Crilley, L. R., Kramer, L., Pope, F. D., Whalley, L. K., Cryer, D. R., Heard, D. E., Lee, J. D., Reed, C., and Bloss, W. J.: On the interpretation of in situ HONO observations via photochemical steady state, Faraday Discussions, 189, 191-212, 10.1039/C5FD00224A, 2016.

acid (HONO), Environ Sci Pollut Res, 48-54, 2002.





5

Duan, J., Qin, M., Ouyang, B., Fang, W., Li, X., Lu, K., Tang, K., Liang, S., Meng, F., Hu, Z., Xie, P., Liu, W., and Häsler, R.: Development of an incoherent broadband cavity-enhanced absorption spectrometer for in situ measurements of HONO and NO2, Atmos. Meas. Tech., 11, 4531-4543, 10.5194/amt-11-4531-2018, 2018.

Febo, A., Perrino, C., Gherardi, M., Sparapani, R. J. E. s., and technology: Evaluation of a high-purity and high-stability continuous generation system for nitrous acid, 29, 2390-2395, 1995.

Febo, A., Perrino, C., and Allegrini, I. J. A. E.: Measurement of nitrous acid in Milan, Italy, by DOAS and diffusion denuders, 30, 3599-3609, 1996.

Heland, J., Kleffmann, J., Kurtenbach, R., and Wiesen, P.: A New Instrument To Measure Gaseous Nitrous Acid (HONO) in the Atmosphere, Environmental Science & Technology, 35, 3207-3212, 10.1021/es000303t, 2001.

10 Hera, D., Langford, V. S., McEwan, M. J., McKellar, T. I., and Milligan, D. B.: Negative Reagent Ions for Real Time Detection Using SIFT-MS, 4, 16, 2017.

Hou, S. Q., Tong, S. R., Ge, M. F., and An, J. L.: Comparison of atmospheric nitrous acid during severe haze and clean periods in Beijing, China, Atmos Environ, 124, 199-206, 10.1016/j.atmosenv.2015.06.023, 2016.

Kennedy, O. J., Ouyang, B., Langridge, J. M., Daniels, M. J. S., Bauguitte, S., Freshwater, R., McLeod, M. W., Ironmonger,

- 15 C., Sendall, J., Norris, O., Nightingale, R., Ball, S. M., and Jones, R. L.: An aircraft based three channel broadband cavity enhanced absorption spectrometer for simultaneous measurements of NO<sub>3</sub>, N<sub>2</sub>O<sub>5</sub> and NO<sub>2</sub>, Atmos. Meas. Tech., 4, 1759-1776, 10.5194/amt-4-1759-2011, 2011. Kleffmann, J., Heland, J., Kurtenbach, R., Lorzer, J., and Wiesen, P.: A new instrument (LOPAP) for the detection of nitrous
- Kleffmann, J., Lörzer, J., Wiesen, P., Kern, C., Trick, S., Volkamer, R., Rodenas, M., and Wirtz, K. J. A. E.: Intercomparison of the DOAS and LOPAP techniques for the detection of nitrous acid (HONO), 40, 3640-3652, 2006.
   Kleffmann, J.: Daytime Sources of Nitrous Acid (HONO) in the Atmospheric Boundary Layer, ChemPhysChem, 8, 1137-1144, 10.1002/cphc.200700016, 2007.

Kleffmann, J., Wiesen, P. J. A. C., and Physics: Quantification of interferences of wet chemical HONO LOPAP measurements under simulated polar conditions, 8, 6813-6822, 2008.

Le Breton, M., Bacak, A., Muller, J. B., Bannan, T. J., Kennedy, O., Ouyang, B., Xiao, P., Bauguitte, S. J.-B., Shallcross, D. E., and Jones, R. L. J. A. M.: The first airborne comparison of N 2 O 5 measurements over the UK using a CIMS and BBCEAS during the RONOCO campaign, 6, 9731-9743, 2014.

Le Breton, M., Bannan, T. J., Shallcross, D. E., Khan, M. A., Evans, M. J., Lee, J., Lidster, R., Andrews, S., Carpenter, L. J.,

30 Schmidt, J., Jacob, D., Harris, N. R. P., Bauguitte, S., Gallagher, M., Bacak, A., Leather, K. E., and Percival, C. J.: Enhanced ozone loss by active inorganic bromine chemistry in the tropical troposphere, Atmospheric Environment, 155, 21-28, https://doi.org/10.1016/j.atmosenv.2017.02.003, 2017.





Lee, B. H., Wood, E. C., Herndon, S. C., Lefer, B. L., Luke, W. T., Brune, W. H., Nelson, D. D., Zahniser, M. S., and Munger, J. W.: Urban measurements of atmospheric nitrous acid: A caveat on the interpretation of the HONO photostationary state, Journal of Geophysical Research: Atmospheres, 118, 2013JD020341, 10.1002/2013jd020341, 2013. Lee, J., Whalley, L., Heard, D., Stone, D., Dunmore, R., Hamilton, J., Young, D., Allan, J., Laufs, S., Kleffmann, J. J. A. C.,

- 5 and Physics: Detailed budget analysis of HONO in central London reveals a missing daytime source, 16, 2747-2764, 2016. Lopez-Hilfiker, F. D., Mohr, C., Ehn, M., Rubach, F., Kleist, E., Wildt, J., Mentel, T. F., Lutz, A., Hallquist, M., Worsnop, D., and Thornton, J. A.: A novel method for online analysis of gas and particle composition: description and evaluation of a Filter Inlet for Gases and AEROsols (FIGAERO), Atmos. Meas. Tech., 7, 983-1001, 10.5194/amt-7-983-2014, 2014. Michoud, V., Colomb, A., Borbon, A., Miet, K., Beekmann, M., Camredon, M., Aumont, B., Perrier, S., Zapf, P., Siour, G.,
- 10 Ait-Helal, W., Afif, C., Kukui, A., Furger, M., Dupont, J. C., Haeffelin, M., and Doussin, J. F.: Study of the unknown HONO daytime source at a European suburban site during the MEGAPOLI summer and winter field campaigns, Atmos. Chem. Phys., 14, 2805-2822, 10.5194/acp-14-2805-2014, 2014.

Perner, D., and Platt, U.: Detection of nitrous acid in the atmosphere by differential optical absorption, 6, 917-920, doi:10.1029/GL006i012p00917, 1979.

15 Pinto, J. P., Dibb, J., Lee, B. H., Rappenglück, B., Wood, E. C., Levy, M., Zhang, R.-Y., Lefer, B., Ren, X.-R., Stutz, J., Tsai, C., Ackermann, L., Golovko, J., Herndon, S. C., Oakes, M., Meng, Q.-Y., Munger, J. W., Zahniser, M., and Zheng, J.: Intercomparison of field measurements of nitrous acid (HONO) during the SHARP campaign, 119, 5583-5601, doi:10.1002/2013JD020287, 2014.

Priestley, M., le Breton, M., Bannan, T. J., Worrall, S. D., Bacak, A., Smedley, A. R. D., Reyes-Villegas, E., Mehra, A., Allan,

J., Webb, A. R., Shallcross, D. E., Coe, H., and Percival, C. J.: Observations of organic and inorganic chlorinated compounds and their contribution to chlorine radical concentrations in an urban environment in northern Europe during the wintertime, Atmos. Chem. Phys., 18, 13481-13493, 10.5194/acp-18-13481-2018, 2018.
 Ben X. Coo, H. Zhen, X. Craunes, L. Wennberg, B. Breume, E. LeFrenchi, B. Cohen, B. McKey, M. Coldstein, A. L.A.

Ren, X., Gao, H., Zhou, X., Crounse, J., Wennberg, P., Browne, E., LaFranchi, B., Cohen, R., McKay, M., Goldstein, A. J. A.C., and Physics: Measurement of atmospheric nitrous acid at Bodgett Forest during BEARPEX2007, 10, 6283-6294, 2010.

25 Ródenas, M., Muñoz, A., Alacreu, F., Brauers, T., Dorn, H.-P., Kleffmann, J., and Bloss, W.: Assessment of HONO measurements: The FIONA campaign at EUPHORE, in: Disposal of Dangerous Chemicals in Urban Areas and Mega Cities, Springer, 45-58, 2013.

Shi, Z., Vu, T., Kotthaus, S., Grimmond, S., Harrison, R. M., Yue, S., Zhu, T., Lee, J., Han, Y., Demuzere, M., Dunmore, R. E., Ren, L., Liu, D., Wang, Y., Wild, O., Allan, J., Barlow, J., Beddows, D., Bloss, W. J., Carruthers, D., Carslaw, D. C.,

30 Chatzidiakou, L., Crilley, L., Coe, H., Dai, T., Doherty, R., Duan, F., Fu, P., Ge, B., Ge, M., Guan, D., Hamilton, J. F., He, K., Heal, M., Heard, D., Hewitt, C. N., Hu, M., Ji, D., Jiang, X., Jones, R., Kalberer, M., Kelly, F. J., Kramer, L., Langford, B., Lin, C., Lewis, A. C., Li, J., Li, W., Liu, H., Loh, M., Lu, K., Mann, G., McFiggans, G., Miller, M., Mills, G., Monk, P., Nemitz, E., O'Connor, F., Ouyang, B., Palmer, P. I., Percival, C., Popoola, O., Reeves, C., Rickard, A. R., Shao, L., Shi, G., Spracklen, D., Stevenson, D., Sun, Y., Sun, Z., Tao, S., Tong, S., Wang, Q., Wang, W., Wang, X., Wang, Z., Whalley, L., Wu,





X., Wu, Z., Xie, P., Yang, F., Zhang, Q., Zhang, Y., Zhang, Y., and Zheng, M.: Introduction to Special Issue – In-depth study of air pollution sources and processes within Beijing and its surrounding region (APHH-Beijing), Atmos. Chem. Phys. Discuss., 2018, 1-62, 10.5194/acp-2018-922, 2018.

Sousan, S., Koehler, K., Thomas, G., Park, J. H., Hillman, M., Halterman, A., and Peters, T. M.: Inter-comparison of low-cost

5 sensors for measuring the mass concentration of occupational aerosols, Aerosol Science and Technology, 50, 462-473, 10.1080/02786826.2016.1162901, 2016.

Spataro, F., and Ianniello, A.: Sources of atmospheric nitrous acid: State of the science, current research needs, and future prospects, Journal of the Air & Waste Management Association, 64, 1232-1250, 10.1080/10962247.2014.952846, 2014.

Spindler, G., Hesper, J., Brüggemann, E., Dubois, R., Müller, T., and Herrmann, H. J. A. E.: Wet annular denuder measurements of nitrous acid: laboratory study of the artefact reaction of NO2 with S (IV) in aqueous solution and comparison with field measurements, 37, 2643-2662, 2003.

Stutz, J., Kim, E. S., Platt, U., Bruno, P., Perrino, C., and Febo, A.: UV-visible absorption cross sections of nitrous acid, 105, 14585-14592, 10.1029/2000jd900003, 2000.

Stutz, J., Oh, H.-J., Whitlow, S. I., Anderson, C., Dibb, J. E., Flynn, J. H., Rappenglück, B., and Lefer, B.: Simultaneous DOAS

- 15 and mist-chamber IC measurements of HONO in Houston, TX, Atmospheric Environment, 44, 4090-4098, https://doi.org/10.1016/j.atmosenv.2009.02.003, 2010. Taipale, R., Ruuskanen, T. M., Rinne, J., Kajos, M. K., Hakola, H., Pohja, T., and Kulmala, M.: Technical Note: Quantitative long-term measurements of VOC concentrations by PTR-MS – measurement, calibration, and volume mixing ratio calculation methods, Atmos. Chem. Phys., 8, 6681-6698, 10.5194/acp-8-6681-2008, 2008.
- Veitel, H.: Vertical profiles of NO 2 and HONO in the planetary boundary layer, 2002.
   Veres, P. R., Roberts, J. M., Wild, R. J., Edwards, P. M., Brown, S. S., Bates, T. S., Quinn, P. K., Johnson, J. E., Zamora, R. J., and de Gouw, J.: Peroxynitric acid (HO<sub>2</sub>NO<sub>2</sub>) measurements during the UBWOS 2013 and 2014 studies using iodide ion chemical ionization mass spectrometry, Atmos. Chem. Phys., 15, 8101-8114, 10.5194/acp-15-8101-2015, 2015.
- Voigt, S., Orphal, J., and Burrows, J. P.: The temperature and pressure dependence of the absorption cross-sections of NO2 in the 250–800 nm region measured by Fourier-transform spectroscopy, Journal of Photochemistry and Photobiology A: Chemistry, 149, 1-7, https://doi.org/10.1016/S1010-6030(01)00650-5, 2002. Wang, J., Zhang, X., Guo, J., Wang, Z., and Zhang, M.: Observation of nitrous acid (HONO) in Beijing, China: Seasonal variation, nocturnal formation and daytime budget, Science of The Total Environment, 587-588, 350-359,
- https://doi.org/10.1016/j.scitotenv.2017.02.159, 2017.
  Zhou, W., Zhao, J., Ouyang, B., Mehra, A., Xu, W., Wang, Y., Bannan, T. J., Worrall, S. D., Priestley, M., Bacak, A. J. A. C., and Physics: Production of N 2 O 5 and CINO 2 in summer in urban Beijing, China, 18, 11581-11597, 2018.





Zhou, X., Gao, H., He, Y., Huang, G., Bertman, S. B., Civerolo, K., and Schwab, J.: Nitric acid photolysis on surfaces in low-NOx environments: Significant atmospheric implications, Geophysical Research Letters, 30, n/a-n/a, 10.1029/2003GL018620, 2003.





## Table 1: Instrumentation measuring HONO at IAP.

Institution	Instrument	Manufacturer	DL (ppt)	Error	Time	Reference
					resolution	
Birmingham	Wet	QUMA	Winter: 35	10%	5 min	(Heland et al., 2001)
(BHAM)	Chemical		Summer: 5			
	(LOPAP.03)		(2o, 30s)			
ICCAS	Wet	Custom built	134 (2o, 30s)	10%	5 min	(Hou et al. 2016)
	Chemical					
Cambridge	BBCEAS	Custom built	25	9%	5 s	(Kennedy et al., 2011)
(CAM)			(1 <del>0</del> , 60s))			
AIOFM	BBCEAS	Custom built	120 (2σ, 60s)	9%	1 min	(Duan et al. 2018)
Manchester	ToF-CIMS	Aerodyne	33 (2o, 60s)	19 %	1 Hz	(Priestley et al., 2018)
(MANC)		Research Inc /				
		Tofwerk.				
York	Syft PTR-MS	Syft	130 (2o, 60s)	22%	19 secs	(Hera et al., 2017)
	Voice ultra	Technologies			(1min	
	200				averaged)	





Table 2: Results of the reduced major axis regression analysis with 95% confidence intervals during the formal winter intercomparison. Variability shown is the 95% confidence interval of the slope and intercepts.

INSTRUMENTS	INTERCEPT	SLOPE	R	Ν	
BHAM-ICCAS	$0.09 \pm 0.04$	0.77±0.01	0.97	865	
<b>BHAM-AIOFM</b>	-0.18±0.03	$0.71 \pm 0.01$	0.98	1070	
BHAM-CAM	-0.23±0.03	$0.61\pm0.01$	0.98	1125	
ICCAS-AIOFM	-0.20±0.03	$0.88 \pm 0.01$	0.98	954	
ICCAS-CAM	-0.38±0.04	$0.82 \pm 0.01$	0.97	991	
AIOFM-CAM	-0.09±0.03	0.87±0.01	0.96	1206	





Table 3: Calculated CD values for each instrument pair during the winter inter-comparison.

	ICCAS	AIOFM	CAM
BHAM	0.11	0.22	0.32
ICCAS	-	0.12	0.21
AIOFM	-	-	0.11





Table 4: RMA regression analysis (with 95% confidence intervals) for times when the abundance of HONO was less than 2 ppb as measured by CAM during the formal winter intercomparison period. Variability shown is the 95% confidence interval of the slope and intercepts.

INTERCEPT	SLOPE	R	Ν
-0.01±0.06	0.82±0.02	0.96	437
-0.25±0.03	$0.76 \pm 0.01$	0.98	529
-0.06±0.03	0.53±0.01	0.95	613
-0.23±0.05	0.91±0.03	0.95	478
-0.12±0.05	$0.68 \pm 0.02$	0.91	556
0.07±0.03	$0.72 \pm 0.02$	0.92	655
	INTERCEPT -0.01±0.06 -0.25±0.03 -0.06±0.03 -0.23±0.05 -0.12±0.05 0.07±0.03	INTERCEPT         SLOPE           -0.01±0.06         0.82±0.02           -0.25±0.03         0.76±0.01           -0.06±0.03         0.53±0.01           -0.23±0.05         0.91±0.03           -0.12±0.05         0.68±0.02           0.07±0.03         0.72±0.02	INTERCEPT         SLOPE         R           -0.01±0.06         0.82±0.02         0.96           -0.25±0.03         0.76±0.01         0.98           -0.06±0.03         0.53±0.01         0.95           -0.23±0.05         0.91±0.03         0.95           -0.12±0.05         0.68±0.02         0.91           0.07±0.03         0.72±0.02         0.92





Table 5: RMA regression relationships of HONO measured by BHAM-ICCAS-CAM during the co-located measurements at the start of the summer campaign (22<sup>nd</sup> May-30<sup>th</sup> May 2017). All three inlet locations were the same as the formal winter intercomparison. Variability shown is the 95% confidence interval of the slope and intercepts.

INSTRUMENTS	INTERCEPT	SLOPE	R	Ν
BHAM-ICCAS	-0.24±0.02	0.91±0.01	0.97	2061
BHAM-CAM	-0.34±0.02	$0.61 \pm 0.01$	0.90	1233
ICCAS-CAM	0.21±0.02	$0.69 \pm 0.01$	0.85	1346





Table 6: RMA regression relationships (with 95% confidence intervals) of HONO measured by all instruments in the middle of the summer campaign (7<sup>th</sup>-14<sup>th</sup> June 2017). Note that BHAM and ICCAS inlets were in same location for this period. The CAM instrument had moved to the same container as AIOFM, whose inlets were 3 meters apart. Variability shown is the 95% confidence interval of the slope and intercepts.

INSTRUMENTS	INTERCEPT	SLOPE	R	Ν
BHAM-ICCAS	0.17±0.04	0.93±0.03	0.90	900
BHAM-AIOFM	-0.18±0.03	$0.61 \pm 0.02$	0.81	1377
BHAM-CAM	$0.02 \pm 0.03$	$0.65 \pm 0.02$	0.86	1395
ICCAS-AIOFM	$-0.08 \pm 0.02$	0.43±0.01	0.81	1153
ICCAS-CAM	$0.07 \pm 0.03$	$0.53 \pm 0.02$	0.82	1167
AIOFM-CAM	0.20±0.01	$1.07 \pm 0.02$	0.92	1982





Table 7: RMA regression relationships of HONO measured by BHAM-CAMB-MANC for the whole summer period. Variability shown is the 95% confidence interval of the slope and intercepts.

INSTRUMENTS	INTERCEPT	SLOPE	R	Ν
BHAM-MANC	-0.35±0.06	1.57±0.04	0.84	1896
BHAM-CAM	$0.00 \pm 0.02$	0.63±0.01	0.84	4106
CAM-MANC	-0.30±0.05	2.39±0.05	0.88	2372







Figure 1: Time series of the measured mixing ratios during the formal winter inter-comparison for each instrument







Figure 2: Time series of the co-efficient of variance (CV), mean mixing ratio of HONO (top) and NO<sub>2</sub> (bottom) during the winter inter-comparison. Note only when all four instruments were measuring were the mean HONO and CV calculated.







Figure 3: Normalised Differences (ND) for each instrument pair as a function of wind direction coloured by measured HONO concentration (ppb) for the winter intercomparison. Note the different scales for the y-axes and HONO abundance colour key.







Figure 4: Histogram of measured summer concentrations (only for periods when all three instruments were measuring).







Figure 5: Time series for the period when the YORK instrument measured at ground level during the summer campaign.