Development of the DRoplet Ice Nuclei Counter Zürich (DRINCZ): Validation and application to field collected snow samples

Robert O. David[1,*], Maria Cascajo-Castresana[1,2], Killian P. Brennan[1], Michael Rösch[1], Nora Els[4], Julia Werz[3], Vera Weichlinger[1], Lin S. Boynton[3], Sophie Bogler[3], Nadine Borduas-Dedekind[1,3], Claudia Marcolli[1], Zamin A. Kanji[1]

[1]Institute of Atmospheric and Climate Science, ETH Zürich, Zürich, 8092, Switzerland
[2]CIC nanoGUNE Consolider, Donostia-San Sebastian, E-20018, Spain
[3]Institute for Biogeochemistry and Pollutant Dynamics, ETH Zürich, Zürich, 8092, Switzerland
[4]Institute of Ecology, University of Innsbruck, Innsbruck, 6020, Austria
*Now at Department of Geosciences, University of Oslo, Oslo, 0315, Norway

Correspondence to: Robert O. David (r.o.david@geo.uio.no) and Zamin A. Kanji (zamin.kanji@env.ethz.ch)

Abstract. Ice formation in the atmosphere is important for regulating cloud lifetime, Earth's radiative balance and initiating precipitation. Due to the difference in the saturation vapor pressure over ice and water, in mixed-phase clouds (MPCs), ice will grow at the expense of supercooled cloud droplets. As such, MPCs, which contain both supercooled liquid and ice, are particularly susceptible to ice formation. However, measuring and quantifying the concentration of ice nucleating particles (INPs) responsible for ice formation at temperatures associated with MPCs is challenging due to their very low concentrations in the atmosphere (~ 1 in 10^5 at -30 °C). Atmospheric INP concentrations vary over several orders of magnitude at a single temperature and strongly increase as temperature approaches the homogeneous freezing threshold of water. To further quantify the INP concentration in nature and perform systematic laboratory studies to increase the understanding of the properties responsible for ice nucleation, a new drop freezing instrument, the DRoplet Ice Nuclei Counter Zurich (DRINCZ) is developed. The instrument is based on the design of previous drop freezing assays and uses a USB camera to automatically detect freezing in a 96-well tray cooled in an ethanol chilled bath with a user friendly and fully automated analysis procedure. Based on an in-depth characterization of DRINCZ, we develop a new method for quantifying and correcting temperature biases across drop freezing assays. DRINCZ is further validated performing NX-illite experiments, which compare well with the literature. The temperature uncertainty in DRINCZ was determined to be ± 0.9 ˚C. Furthermore, we demonstrate the applicability of DRINCZ by measuring and analyzing field collected snow samples during an evolving synoptic situation in the Austrian Alps. The field samples fall within previously observed ranges for cumulative INP concentrations and show a dependence on air mass origin and upstream precipitation amount.

1 Introduction

In the atmosphere, ice plays an important role in initiating precipitation and affects the radiative properties of clouds. As much as 80% of land falling precipitation initiates through the ice phase (Mülmenstädt et al., 2015), making it essential to understand the pathways for ice formation in the atmosphere. The ratio of cloud droplets to ice crystals in a mixed-phase cloud (MPC) alters the radiative properties of the cloud and its lifetime (Lohmann and Feichter, 2005; Matus and L'Ecuyer, 2017; Tan et al., 2016). This ratio is important for future climate projections as warmer temperatures will lead to a decrease in ice content, ultimately increasing cloud lifetime and cloud albedo (Tan et al., 2016). Additionally, ice formation at temperatures above -38 ˚C in the atmosphere occurs primarily in MPCs through the freezing of cloud droplets (Ansmann et al., 2009; Boer et al., 2011; Westbrook and Illingworth, 2011). Therefore, understanding ice formation in conditions associated with MPCs is of the utmost importance.

When an ice nucleating particle (INP) gets immersed in a cloud droplet either by acting as cloud condensation nucleus or through scavenging by a cloud droplet, the INP can induce ice formation by reducing the energy barrier associated with the formation of an ice germ and thus freeze at warmer temperatures than homogeneous freezing (Vali et al., 2015). To reproduce the immersion freezing pathway in the laboratory, several methods are used. Single particle methods, such as continuous flow diffusion chambers (Rogers, 1988; Stetzer et al., 2008) operated at water supersaturated conditions (DeMott et al., 2015, 2017; Hiranuma et al., 2015), or with extended chambers that activate individual particles into cloud droplets before exposing them to supercooled conditions (Burkert-Kohn et al., 2017; Kohn et al., 2016; Lüönd et al., 2010) allow for the quantification of the number concentration of INPs as a function temperature. Larger laboratory based single particle methods for examining INPs in the immersion mode include expansion chambers where cloud droplets are first formed by adiabatic cooling due to the expansion of an air volume (Niemand et al., 2012) or experiments where droplets are initially activated and then subsequently cooled as they travel through a laminar flow tube (Hartmann et al., 2011). Aerosols introduced into such systems by dry dispersion or atomization of suspensions and solutions allow for a range of particulates to be examined. However, the single particle methods have detection limitations due to the background ice crystal concentration of the chamber and the optical methods for discriminating between ice and water. Due to the rarity of INPs at MPC conditions, single particle methods are typically unable to quantify INP concentrations within natural ambient samples at temperatures higher than approximately -22 ˚C in remote regions without the use of concentrators (Cziczo et al., 2017).

In contrast bulk methods such as, drop freezing assays (Hill et al., 2014; Stopelli et al., 2014; Vali, 1971), differential scanning calorimetry (Kaufmann et al., 2016; Pinti et al., 2012) and microfluidic devices (Reicher et al., 2018; Riechers et al., 2013; Stan et al., 2009; Tarn et al., 2018) immerse the samples in water and can be used to detect lower atmospheric INP concentrations. The majority of atmospheric INP concentrations at temperatures above -15 ˚C has been quantified using drop freezing assays. To retrieve the concentrations of INP from such bulk suspensions, Vali, (1971; 2019) showed that by dividing

a sample into several aliquots, it is possible to calculate the number of INPs present in the sample as a function of temperature. The probability for more than one INP in an aliquot that freezes at the same temperature can be predicted using Poisson's Law (Vali, 1971). Following Vali (1971), the cumulative number of INPs in a given sample for each temperature can be calculated as:

$$INP(T) = \frac{-ln(1-FF(T))}{V_a} \qquad (1)$$

where $FF(T)$ is the fraction of frozen aliquots at a given temperature, T, and V_a is the volume of an aliquot. As can be seen in Eq. 1, the only way to extend the range of measureable INPs across temperature scales is to change V_a. Due to instrumental limitations, it is often difficult to change V_a by significantly enough values for a change in $INP(T)$ within a single instrumental setup. Rather it is easier to dilute the initial sample thereby reducing the number of INPs in each aliquot. Alternatively, to explore freezing towards warmer temperatures, field samples (e.g. rain or snow samples) can be concentrated by evaporating a part of the water. To account for dilution, Eq. 1 can be rewritten as:

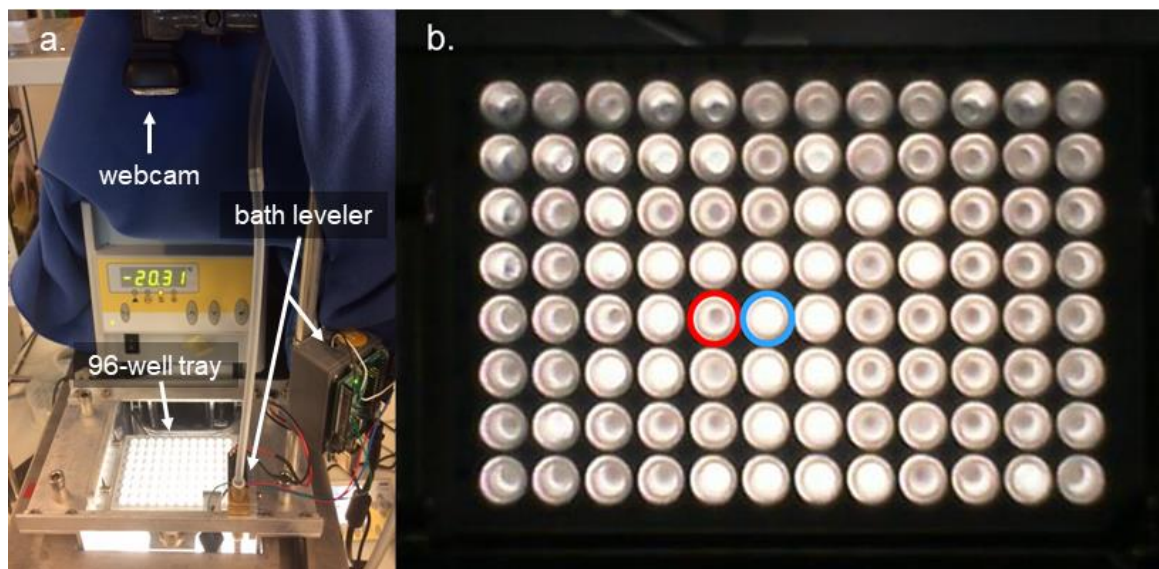
$$INP(T) = \frac{-ln(1-FF(T))DF}{V_a} \qquad (2)$$

where DF is the dilution factor of the initial sample. However, in some cases dilution alone cannot be used to observe the total number of $INP(T)$ due to the presence of impurities that act as INPs in the water used for dilution (Polen et al., 2018). Therefore, it is necessary to use different bulk techniques that measure aliquots with volumes that span several orders of magnitude, typically microliter to picoliter volumes (Harrison et al., 2018; Hill et al., 2014; Murray et al., 2010; Whale et al., 2015).

Studies have investigated the concentrations of INPs in the atmosphere over the last 50 years and show that the concentration in the atmosphere spans several orders of magnitude (Fletcher, 1962; Kanji et al., 2017; Petters and Wright, 2015; Welti et al., 2018). Some of the original studies investigated the INP concentrations in melted hail and snow samples e.g. (Vali, 1971). Since then, studies have diversified to sampling INPs directly from the air (Boose et al., 2016b; Creamean et al., 2013; DeMott et al., 2003; Lacher et al., 2017; Richardson et al., 2007; Welti et al., 2018), from precipitation (Christner et al., 2008; Hill et al., 2014; Petters and Wright, 2015; Stopelli et al., 2015) and investigated potential types of INPs in the laboratory from commercial and naturally occurring samples as well as field collected samples (Atkinson et al., 2013; Boose et al., 2016a; Broadley et al., 2012; Felgitsch et al., 2018; Hill et al., 2014; Hiranuma et al., 2015, 2019; Kaufmann et al., 2016; Murray et al., 2012; Pummer et al., 2012; Wex et al., 2015). Yet the atmospheric variability in INP concentrations remains unresolved (Hoose and Möhler, 2012; Kanji et al., 2017; Petters and Wright, 2015; Welti et al., 2018). In order to further quantify the variability of ambient INP concentration relevant for ice formation in MPCs and increase the understanding of the ice nucleation ability of laboratory and field collected samples, we developed and characterized the DRoplet Ice Nuclei Counter Zurich (DRINCZ). DRINCZ is a drop freezing instrument to investigate ice nucleation at temperature conditions between -25 ˚C and 0 ˚C, representative for MPCs. Furthermore, DRINCZ complements and extends the INP concentration measurement capabilities of the single particle and bulk methods employed at ETH Zürich e.g. (Kohn et al., 2016; Lacher et al., 2017; Lüönd

96   et al., 2010; Marcolli et al., 2007; Stetzer et al., 2008). The automation of DRINCZ and its portable design allows for the
97   acquisition of INP data in the field and laboratory, ultimately increasing the attainable information about the global distribution
98   of INPs.

99   **2 Instrument Design**

100   DRINCZ is based on the design of Stopelli et al. (2014) and Hill et al. (2014), which was initially suggested by (Vali and
101   Upper, 1995). It consists of a temperature controlled ethanol bath (Lauda ProLine RP 845, Lauda-Königshofen, Germany), a
102   home-built LED light consisting of several LED light strips enclosed in an ethanol proof housing, a home-built 96-well tray
103   holder and camera mount, a webcam (Microsoft Lifecam HD-3000) and a custom designed bath leveler, composed of a bath
104   level sensor and valve (see Section 2.2) (Fig. 1a). The working principle is similar to that of Stopelli et al. (2014), in that a
105   USB camera detects the light transmission through aliquots of sample. In DRINCZ, the aliquots are typically 50 μL and
106   dispensed into a 96-well polypropylene tray (732-2386, VWR, USA). To avoid contamination, the top of the 96-well tray is
107   sealed with a transparent non-permeable foil (Axgen, Platemax CyclerSeal Sealing Film, PCR-TS). The well tray is placed in
108   the tray holder (Fig. A1) and left to rest for 1 min at 0 °C before the cooling ramp is started. The webcam is programmed to
109   take a picture every 15 seconds, which corresponds to a picture taken approximately every 0.25 ˚C decrease when the bath is
110   cooled at a rate of 1 ˚C min^{-1}. Moreover, both the picture frequency and cooling rate are adjustable. Upon freezing, the light
111   transmission through an individual well decreases (red circled well in Fig. 1b) due to the polycrystallinity of the ice frozen in
112   the wells.

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114   **Figure 1: (a) Picture of DRINCZ. (b) Change in light transmission through the wells during an experiment with an example of an**
115   **unfrozen (blue circle) and frozen (red circle) well.**

The cooling cycle of the ethanol-based Lauda bath is controlled using LabVIEW® and the bath temperature is written to a text file that is then read in by MATLAB® during the analysis. In addition, MATLAB® is also used to take and save the pictures from the webcam. Both the LabVIEW® generated text file and pictures from the experiment are stored in the same folder for data handling. A suite of MATLAB® functions have been written to automatically analyze and store the data from each experiment, allowing for minimal user input (details of the code are provided in Appendix A) and rapid experiment throughput of approximately 30 minutes per experiment and 2 minutes to process the data for frozen fraction as a function of temperature.

2.1 Detection Method

Similar to Stopelli et al, (2014), the ice nucleation detection in DRINCZ is achieved by the attenuation of visible radiation due to a frozen well compared to transmission through a supercooled well. The images are analyzed by first detecting the pixels that correspond to each well of the 96-well tray and then calculating the change of the average well brightness during an experiment between one picture and the next. The well detection method is described in the following subsection, followed by the technique used to detect well freezing.

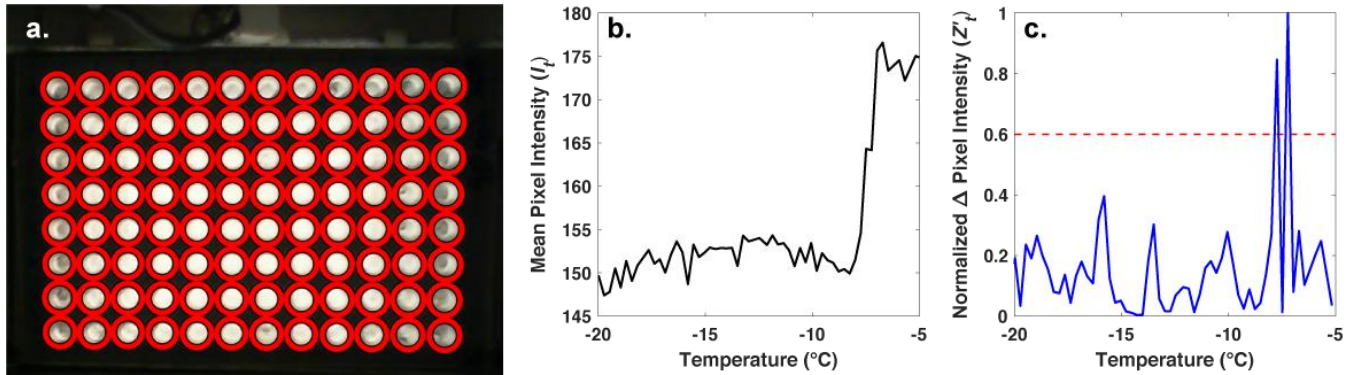
2.1.1 Circular Hough Transform for Well Detection

A fixed 96-well tray holder with an integrated webcam mount reduces variations in setting up the experiment. Nevertheless, small changes in the location of the webcam due to mechanical shock during transport or testing, can produce misidentified wells when algorithms rely on fixed well locations. Therefore, a freezing detection algorithm was developed to avoid errors arising from small changes in the location of the wells. To optimize contrast, the PCR tray holder was constructed out of aluminum so that light transmission only occurs through the wells (see Fig. A1). The high contrast between the illuminated wells and dark tray holder allows for the automatic detection of the wells using a Circular Hough Transform (CHT) (e.g. Atherton and Kerbyson, 1999). The CHT first identifies pixels along regions of large gradients in brightness, to identify the pixels at the edge of the well. To determine the center of each well, the algorithm draws circles of varying diameter (ranging between 7 and 15 pixels in radius, which corresponds to the previously observed diameters of a well in terms of pixel number) around these edge pixels and classifies the pixel intersecting the largest number of circles as the well center. The radius of the well is then given as the radius of the circles that led to the highest number of intersections. The pixels within a well are then identified as the ones encompassed by a circle drawn from a well center with the calculated radius as denoted by the red circles in Fig. 2a. Since the CHT identifies the well center locations in random order, they must be sorted based on their x and y coordinates using a pixel scale for spatial biases or refreezing results to be analyzed. The wells are sorted based on their center locations using the following equation:

$$C_i = \frac{y_i}{D} L_x + x_i \qquad (3)$$

where C_i is the value of the well center based on its pixel location in y and x coordinates, y_i and x_i, respectively, with the origin taken as the pixel in the upper left hand corner of the image. L_x is the pixel number across the well array in the x

147   coordinate and D is the diameter (pixel number) of the wells. All the C_i values are then sorted to ensure that the wells are

148   identified based on their location independent of the experiment.

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150 **Figure 2: (a) Automatic detection of the wells (red circles) using a CHT. (b) Light intensity or I_t of a single well as a function of**
151 **temperature as observed by the webcam and (c) the normalized change in pixel intensity, Z'_t, for the same well as in b between**
152 **subsequent pictures taken during an experiment, as a function of temperature. The most intense peak corresponds to the ice**
153 **nucleation temperature and the second most intense peak is due to the slow freezing of the solution after nucleation. The dashed red**
154 **line represents the 0.6 threshold required for a well to be classified as frozen.**

155

156 **2.1.2 Freezing Detection**

157   With the well locations identified, the intensity values of the pixels within each well are averaged for each image recorded

158   during an experiment (I_t). The change in I_t between subsequent images is used to identify the image where freezing occurred

159   and the corresponding temperature (Fig. 2b). However, due to the slow freezing process which is limited by the latent heat

160   release, the light transmission of a well continuously changes until the water is completely frozen as can be seen as two large

161   peaks in Fig. 2c. To correctly identify the point in time when ice nucleation and not just freezing within the well occurs, the

162   maximum change in I_t between subsequent images is normalized to 1 using the following procedure:

163   First, the Z-score (Z_t) of I_t is taken to level out differences in illumination within the 96-well tray:

164     $Z_t = \frac{I_t - \mu}{\sigma}$ (4)

165   where μ and σ are the mean and standard deviation of I_t for all images of a well, respectively. The absolute value of the time

166   derivative or the change in Z_t between subsequent images (dt) is given as:

167     $Z'_t = \left| \frac{Z_t}{dt} \right|$ (5)

168   Z'_t is then normalized to 1 by dividing by the maximum Z'_t of the well. The normalization ensures that a fixed threshold for the

169   identification of ice nucleation can be used rather than relying on a fixed change in light transmission through the well as done

170   by other drop freezing setups (Beall et al., 2017). This ensures that the initial freezing detection is independent of the absolute

171   change in light transmission through a well. Based on validation experiments, a threshold value of 0.6 $\left(\frac{Z'_t}{\max(Z'_t)} \geq 0.6 \right)$ was

172   found to be best for detecting the initial freezing and to avoid assigning subsequent changes in transparency as a nucleation

173   event due to slow freezing.

2.2 Bath Leveler

175   Due to the thermal contraction of the ethanol in the chilled bath between 0 and -30 ˚C, the ethanol level within the bath

176   decreases during an experiment, affecting the immersion level of the wells and thus the thermal contact. It has been shown that

177   large vertical gradients of up to 1.8 °C can exist between the bottom of a well and the air above it in block-based drop freezing

178   setups (Beall et al., 2017). We anticipate vertical gradients to be reduced in DRINCZ due to the direct contact between the

179   cooling medium (ethanol) and the well tray. Therefore, we incorporated a bath leveler composed of a level sensor and solenoid

180   valve to ensure that the ethanol level remains constant. The level sensor (Honeywell LLE 102101 liquid level sensor) detects

181   when the ethanol falls below a fixed level relative to the wells and triggers the solenoid valve (Kuhnke 64.025, 12 VDC valve)

182   to open, allowing additional ethanol to flow into the bath. The level sensor and solenoid are monitored and controlled using a

183   'sketch' written in Arduino (Arduino Uno Rev3 SMD). In order to minimize thermal gradients by adding warm ethanol to the

184   bath, the ethanol is precooled to 0 ˚C using an ice water bath and then added through a copper pipe that extends to the bottom

185   of the bath. Thus, the bath leveler ensures that the wells remain in good thermal contact due to a constant level of ethanol

186   during experiments, while minimizing temperature fluctuations within the bath. The resulting increased reproducibility of

187   experiments due to the bath leveler is discussed in section 3.4.

3 Validation

189   The validation of the instrument is presented in four sections, with the first discussing the temperature calibration followed by

190   discussing the observed bias in freezing, the quantification of instrumental uncertainty and lastly, the improved reproducibility

191   of DRINCZ due to the addition of the bath leveller.

3.1 Temperature Calibration

193   The temperature reported as the freezing temperature is based on the ethanol bath temperature measured by the Lauda chiller

194   (T_{lauda}). In order to correct for the difference between the temperatures of the sample in the wells (T_{well}) and T_{lauda}, a temperature

195   calibration was performed. The calibration was conducted by measuring the temperature (Type K thermocouple) within the

196   four corner wells and a center well of the 96-well tray (Fig. 3a). The same thermocouple was used for all the well temperature

197   measurements to avoid biases between different thermocouples. The wells were filled with 50 µL of ethanol instead of water

198   to extend the calibration across the entire experimental temperature range of DRINCZ without the interference of freezing.

199   The temperature bias between the wells and T_{lauda} was measured every 1 ˚C while the bath was cooled at the typical ramp rate

200   of 1 ˚C min^{-1}. The calibration was performed three times for each well (Fig 3b). Not surprisingly, we found that the ethanol

201   temperature in the bath was consistently lower than the temperature in the five calibration wells and the difference between

202   bath and well temperature increased linearly as the bath temperature decreased. Based on these results the linear function

203      $T_{corr} = 0.917 * T_{lauda} + 1.3$, with T_{lauda} in ˚C (black line in Fig. 3b) was derived to correct the well temperature. The

204   maximum standard deviation taken as the temperature difference between the temperature fit and the individual well

205   temperature was ±0.6 ˚C.

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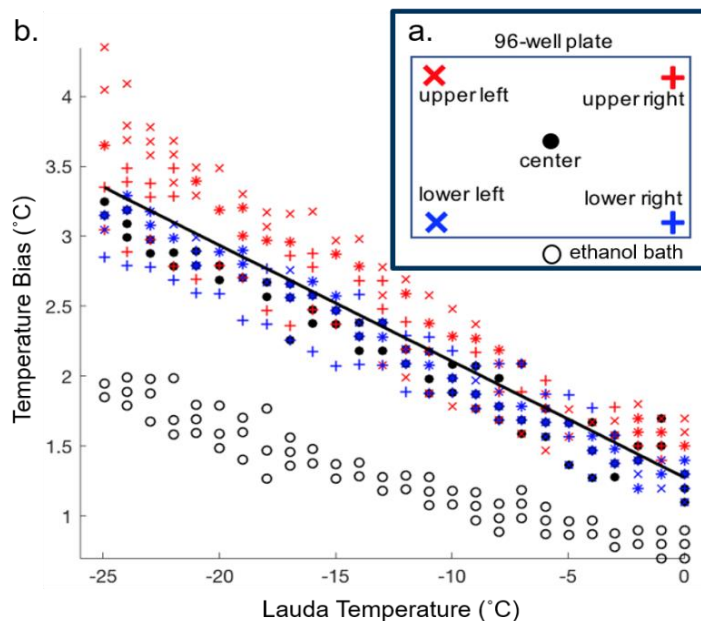
Figure 3: (a) Locations of the type-K thermocouples tested during the temperature calibration. Additionally, the temperature difference between the Lauda temperature and the ethanol bath was measured at the indicated location (black open circle). (b) The temperature bias between the wells and ethanol bath is displayed versus the Lauda bath temperature. The linear temperature correction is shown in black.

3.2 Freezing Bias across the 96-well Tray

212   The temperature calibration discussed above revealed potential variations in the well temperatures between the corner and the

213   center wells. We thus quantified the bias for individual wells, but conclude that it is within the instrument experimental error

214   as discussed below. To do so, 20 pure water (Molecular Biology Reagent, W4502 SigmaAldrich; hereafter referred to as SA

215   water) experiments were analyzed. SA water was chosen for this analysis due to its homogeneity and low freezing temperature,

216   where the observed spread in well temperature was maximized (see Fig. 3). For each well the median freezing temperature (or

217   temperature when frozen fraction $(FF) = 0.5$) $(\widetilde{w}_i)$ was compared to the median freezing temperature of the 4 corner wells

218      $(\widetilde{w}_{4ref})$ used for the temperature calibration (see Fig. 3a and Fig. A2 for the distribution in freezing temperatures of the wells).

219   The difference between $\widetilde{w}_{4ref}$ and $\widetilde{w}_i$ $(\widetilde{w}_{4ref} - \widetilde{w}_i)$ is shown in Fig. 4a. The red (blue) shading indicates a warm (cold) bias

220   and signifies that the solution in these wells are exposed to warmer (colder) temperatures than the average of the four reference

221   wells. The higher concentration of red shades in the middle of the tray suggests that the center of the tray is exposed to as much

222   as 1.5 ˚C warmer ethanol flow than the tray periphery. Indeed, the chilled ethanol circulates clockwise in the Lauda chiller and

223   thus the freezing appears to track the flow (arrows in Fig. 4). Thus, the ethanol circulation explains the observed bias in freezing

224   temperatures across the well plate. The same analysis procedure was applied to the same 20 samples separated by user (12 and

225   8 experiments) and a similar bias was observed (see Appendix Fig. A3). Therefore, the reported bias is instrumental,

226   reproducible and any potential user bias can be excluded. The bias was found to be statistically significant at the 95%

227   confidence interval for 30% of the wells and resulted in an overall bias of 0.23 ˚C (see Fig 4b and Appendix A). As such, a

228   well by well bias correction was developed and tested as described in Appendix A. Although the bias correction performed as

229   expected, the bias of 0.23 ˚C falls within the instrumental uncertainty as discussed in Section 3.3 and is therefore not applied

230   to DRINCZ measurements by default. Nevertheless, the potential benefits and impacts of a bias correction is discussed in the

231   following section.

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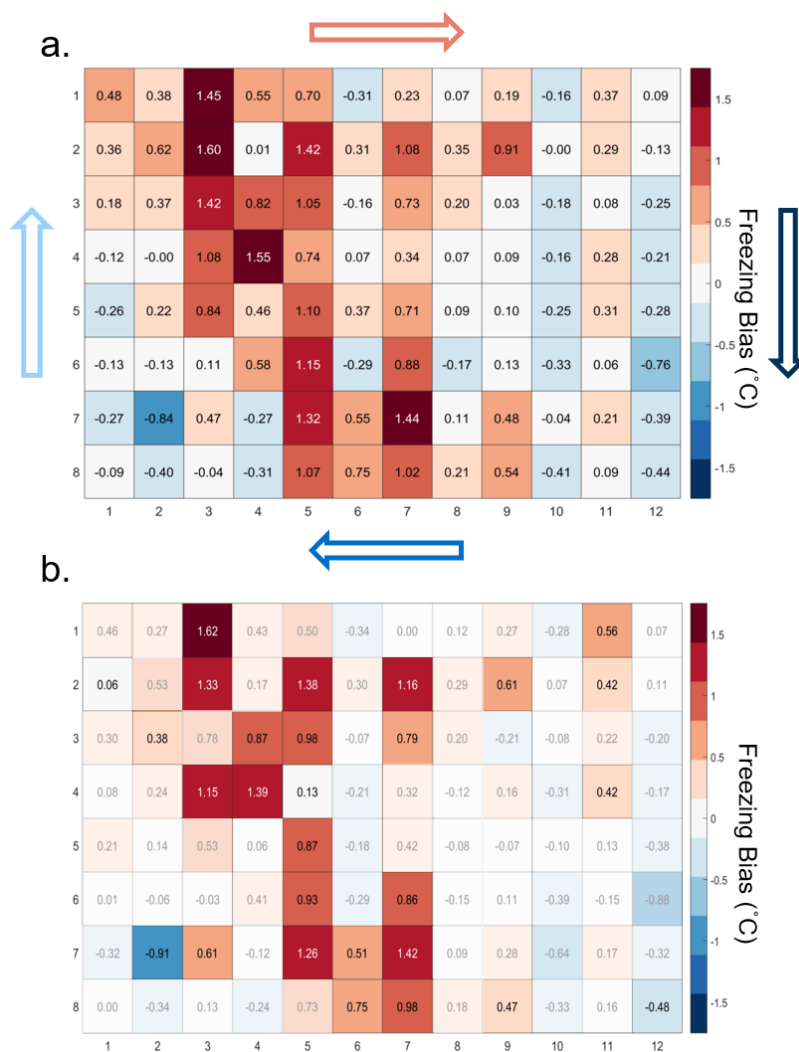
Figure 4: (a) Bias in the freezing of SA water ($\widetilde{w}_{4ref} - \widetilde{w}_i$ in ˚C) based on the median value of each well over 20 experiments relative to the median temperature of freezing for the 4 corner wells used during the temperature calibration. A positive (negative) bias indicates that the wells experience a warmer (colder) temperature than the four corner wells used for temperature calibration and therefore freeze at lower (higher) temperatures than reported. The arrows represent the ethanol circulation in the chiller and the color represents the temperature trend of the ethanol as it circulates in the bath with dark blue being the coldest and red the warmest. (b) Mean freezing bias of SA water between the four reference wells and each well ($\overline{w}_{4ref} - \overline{w}_i$). Positive (negative) values indicate, as denoted by shades of red (blue), wells that systematically freeze at colder (warmer) temperatures and therefore experience warmer (colder) temperatures than reported. Statistically insignificant biases as determined by a Welch's *t*-test (see Eq. A1) are depicted as greyed out.

3.2.1 Impact of Bias Correction on Frozen Fraction

By accounting for the bias in freezing temperature across the 96-well tray by first applying the temperature calibration and then the bias correction such that corrected well value ($\overline{\overline{w}}_i$) becomes:

$$\overline{\overline{w}}_i = \overline{w}_i + (\overline{w}_{4ref} - \overline{w}_i), \tag{6}$$

the slope of the *FF* curves steepens and becomes smoother, which is expected as the observed freezing temperatures become more constrained (see Fig. 5). Although the median freezing temperature with and without the bias correction only changes by 0.2 ˚C (consistent with the correction of the mean bias of 0.23 ˚C found above), the narrowing of the freezing temperature distribution is significant at the 95% significance level (Welch's *t*-test, see Eq. A1). This result shows that by using the spatial dependent freezing information of a well from optically based drop freezing instruments like DRINCZ, temperature can be better constrained. Such a bias correction should also be applicable to freezing methods that use block based cooling, where gradients across the block have been observed or modelled (Beall et al., 2017; Harrison et al., 2018).
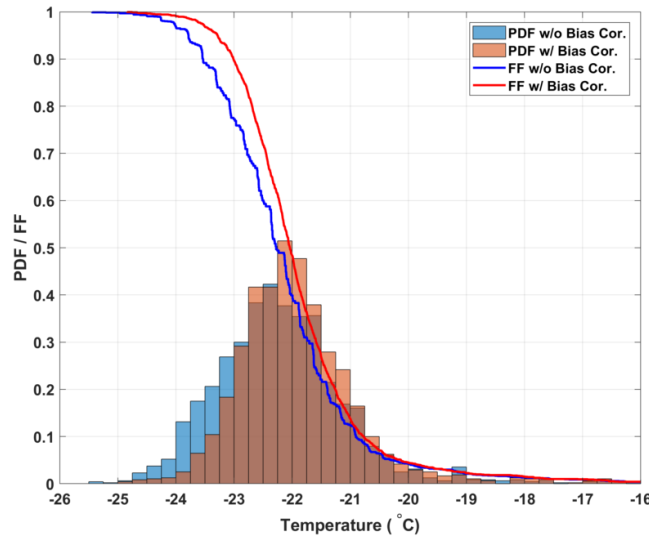
Figure 5: Histograms representing the probability distribution functions for freezing temperatures of the 20 SA water experiments without (blue bars) and with the bias correction (red bars). The calculated cumulative distribution functions, or frozen fraction curves without and with the bias correction are represented as the blue and red lines, respectively.

257

3.3 Instrument Uncertainty

The instrumental uncertainty for DRINCZ is assessed by using the standard deviation in the observed freezing temperatures of the SA water experiments across all wells in combination with the error in the temperature of the wells established during the temperature calibration. The standard deviation of the freezing temperature of the SA water is dependent on FF, with a minimum at 0.5 FF (Fig. 6a). This dependence is expected as the 0.5 FF corresponds to the most likely temperature for the SA water to freeze and therefore, should show the least variability across the 20 experiments used in the analysis. Furthermore, by using the 0.5 FF the influence of contamination and outliers is minimized. The standard deviation at each FF is the uncertainty due to the instrument as well as the variability in the freezing temperature of the SA water and represents the upper limit of the instrumental uncertainty. Given the contribution to the uncertainty due to the variability of the freezing temperature of the SA water, the standard deviation at $FF = 0.5$ can be used as the upper limit of the instrumental uncertainty across the entire FF range. Incorporating a bias correction results in a negligible average difference in the standard deviation (as shown by dashed lines in Fig. 6a). Thus, the upper limit of the instrumental precision is ± 0.3 ˚C (the mean of the standard deviation of freezing temperature over the entire freezing spectrum).

Although the instrumental precision indicates that DRINCZ is very reproducible (± 0.3 ˚C), the accuracy in the reported temperature must be accounted for. Based on the temperature calibration, the standard deviation of the well temperatures is temperature dependent. At the coldest temperatures of the freezing range of the SA water (~ -25 ˚C), the standard deviation of the well temperatures is largest, likely due to the increased gradient between the bath and air temperature and therefore, the importance of the ethanol circulation through the bath is increased. To account for this temperature dependence, the maximum standard deviation of ± 0.6 ˚C from the temperature calibration, corresponding to the lowest observable freezing temperature in DRINCZ (freezing temperature of SA water) is used. Therefore, when accounting for both the precision of the measurements and the accuracy of the temperature, the overall uncertainty of the reported freezing temperature of a well in DRINCZ is ± 0.9 ˚C. This value is comparable to other recently developed drop freezing techniques, which report uncertainties ranging between ± 0.9 ˚C (Harrison et al., 2018) and ± 2.2 ˚C (Beall et al., 2017).

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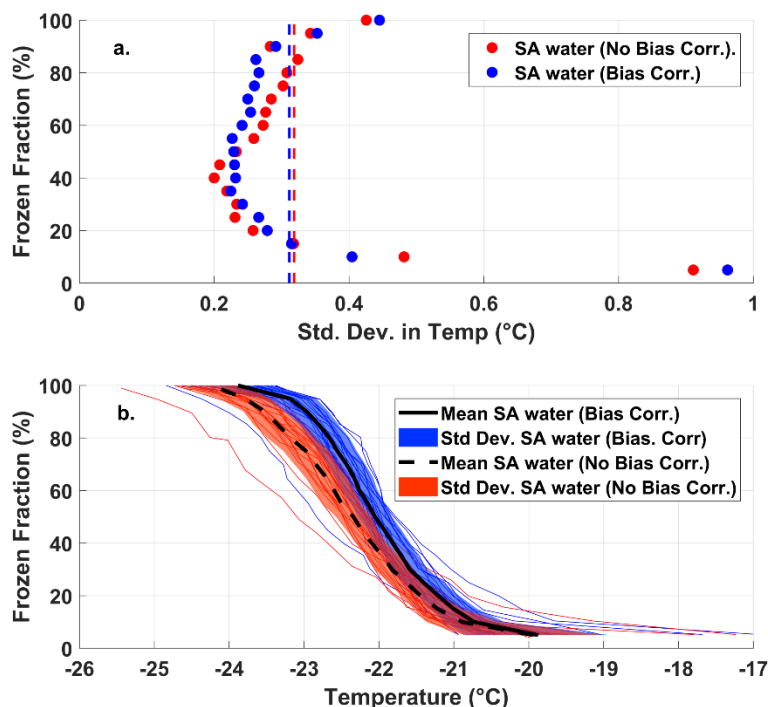


Figure 6: (a) *FF* **and the corresponding standard deviation of the freezing temperatures from the 20 SA experiments with and**
without the bias correction shown as blue and red dots, respectively. The red and blue dashed lines represent the standard deviations
in temperature averaged over all *FF* **values without and with the bias correction, respectively. (b)The** *FF* **of the 20 SA water**
experiments as a function of temperature with and without the bias correction (thin blue and red lines, respectively). The color fill
represents the standard deviations of the SA water from the mean freezing temperature with (solid black line) and without (the
dashed black line) the bias correction.

3.4 Importance of the Bath Leveler

To assess the impact of the decreasing ethanol level on experiments in DRINCZ, 32 experiments with SA water without a bath

leveler were compared to the 20 SA water with a bath level sensor, the same 20 SA water discussed in the previous section.

Figure 7a shows that the bath sensor reduces the spread in freezing temperatures observed. The decrease in the 0.5 *FF*

temperature without the bath leveler is due to a larger gradient between the aliquot and the bath temperatures, thus the well is

warmer than expected, requiring further cooling to observe freezing. The additional cooling in combination with the variable

starting level of the ethanol relative to the wells in the cases of no bath leveler is responsible for the longer freezing tail of the

FF curve (blue line) at higher *FF*s. Without the bath leveler, the initial height of ethanol relative to the wells is user dependent

and not reproducible, leading to both the higher and lower observed freezing temperatures.

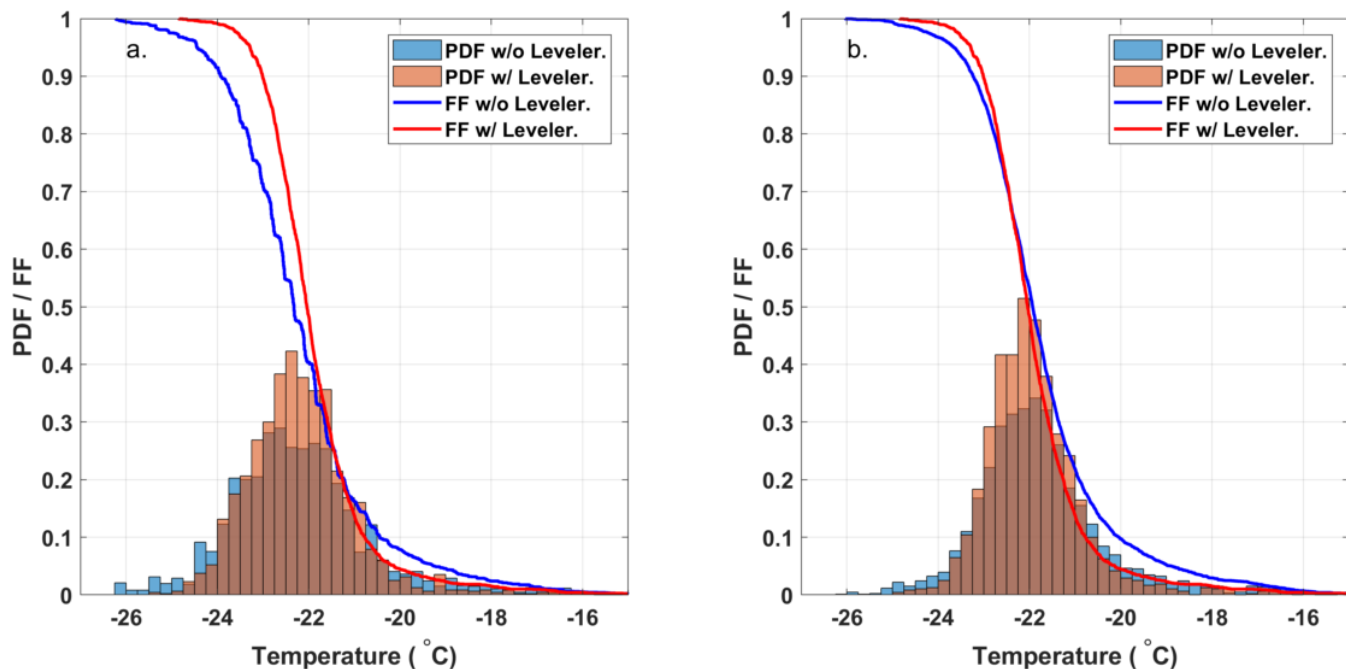



Figure 7: (a) Comparison of the freezing temperature of SA water without (32 experiments, blue) and with (20 experiments, red) the bath leveler. The histograms are normalized to represent the PDF of the freezing temperatures and the lines represent the mean FF curves of the SA water experiments. (b) Shows the same as panel a, except that a bias correction is applied to both sets of experiments.

Although the median freezing temperature (FF=0.5) only decreased by 0.25 ˚C without the bath leveler, the freezing curves steepen when the bath leveler is incorporated in DRINCZ, leading to a decrease of the standard deviation from ± 1.4 to ± 1.0 ˚C over the entire FF range. A bias correction applied following the procedure in Section 3.2 reduces the issues associated with a variable bath level as seen by the similar FF curves and histograms normalized using the probability density function (PDF) estimate in Fig. 7b for experiments with and without the bath leveler. The difference in mean freezing temperatures decrease to 0.05 ˚C at FF=0.5 and the standard deviation of the SA water freezing temperature without the leveler decreases from ± 1.4 to ± 1.2 ˚C over the entire FF range. This decrease is expected as the bias correction is designed to reduce the spread in freezing temperatures within the 96 aliquots. Although the bias correction reduces the need for a bath leveler in DRINCZ, the bias is instrument dependent and may be less pronounced in other drop freezing setups. Therefore, we recommend the use of a bath leveler in any bath-based drop freezing device.

4 Freezing Experiments

To verify the performance of DRINCZ in the context of other published drop freezing techniques, we use the SA water experiments to characterize the instrumental background (Section 4.1) and perform freezing experiments with NX-illite suspensions (Section 4.2). To demonstrate applicability of the instrument to analysis of field samples, the evolution of the ice nucleating ability of atmospheric aerosol particles collected in snow samples at the Sonnblick Observatory in the Hohe Tauern

318 region of Austria during a mid-latitude storm system is assessed in Section 4.3. Lastly, some uncertainties associated with
319 measuring INP in snow samples (Section 4.4) and further validation of DRINCZ through dilutions are discussed (Section 4.5).
320

4.1 Background of DRINCZ

322 The background freezing due to the experimental technique and the SA water used to suspend and dilute samples must be
323 known to discriminate freezing events due to the sample from freezing events due to the water used. Furthermore, an SA water
324 sample is run as a standard at the beginning of each measurement day to ensure the system is operating correctly. The 20 SA
325 water experiments are therefore used to assess the instrument background freezing. It is important to note that in cases where
326 solvents other than SA water are used or where contamination from a sampling technique (e.g. snow collection or impinger
327 measurements) is possible, a different background calculation must be used to accurately assess the freezing ability of a sample.
328 The background of DRINCZ when used with SA water, is calculated by fitting the 20 SA water experiments with a five
329 parameter Boltzmann fit. The five parameter version was chosen to account for asymmetry (Spiess et al., 2008) in the freezing
330 of the SA water but due to the minimum and maximum values of FF given as 0 and 1, respectively, the fit reduces to three
331 parameters and takes the form:

332 $$FF_{BGfit}\left(T_{frzBG}, a, b, c\right) = \frac{1}{\left(1 + e^{a\left(T_{frzBG} - b\right)}\right)^c}, \tag{8}$$

333 where FF_{BGfit} is the fitted FF of the SA water as a function of the observed freezing temperatures of the SA water, T_{frzBG},
334 and the fitting parameters, a, b, c represent the slope of the fit (a = 1.9651), the inflection point (b = -22.7134) and the
335 asymmetry factor (c = 0.6160), respectively. The value of 1 in the numerator represents the maximum FF. The fit and
336 associated coefficients (including 95% confidence range and r^2) are shown in Table 1 and Fig. 8 respectively.
337

338 **Table 1: Coefficients for the three parameter Boltzmann fit of the SA water freezing background and 95th percentile confidence**
339 **interval bound values.**

	a	b	c	r^2
Best	1.9651	-22.7134	0.6160	0.97
-95th %	1.7254	-22.8955	0.4683	N/A
+95th %	2.2049	-22.5312	0.7637	N/A

340
341 The fitted freezing background is used to correct for the contribution of SA water to the observed freezing of a sample. To
342 account for the presence of multiple ice nucleating particles coexisting in a single well, the background is removed by
343 subtracting the differential nucleus concentration of the background from that of the sample (Vali, 1971, 2019). The differential
344 nucleus concentration ($k(T)$) is initially defined in Vali (1971) as:

345 $$k(T) = -\frac{1}{V_a \Delta T} \cdot ln\left(1 - \frac{\Delta N}{N(T)}\right),$$ (9)

346 where $N(T)$ is the number of unfrozen aliquots at the beginning of a temperature step while ΔN is the number of aliquots that

347 freeze during the temperature step (between pictures) or ΔT.

348 The background corrected differential nucleus concentration ($k_{corr}(T)$) is obtained by:

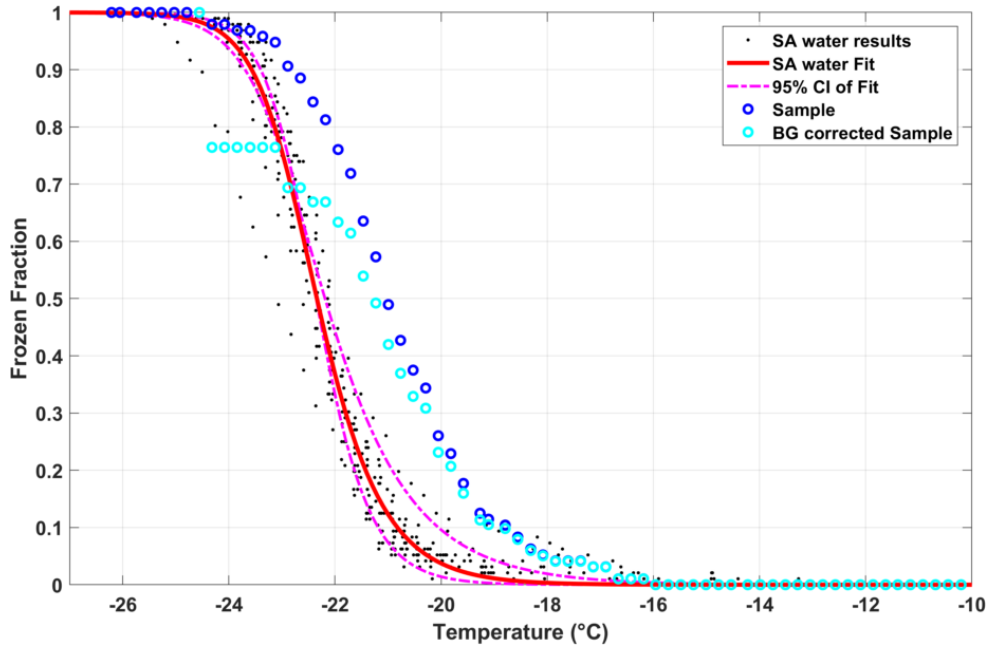
349 $$k_{corr}(T) = k_{sam}(T) - k_{bg}(T),$$ (10)

350 where $k_{sam}(T)$ and $k_{bg}(T)$ are the sample and background differential nucleus concentration, respectively. The background

351 corrected $FF_{cor}(T)$ is then achieved by inverting Eq. 9 and taking the cumulative sum of $k_{corr}(T)$:

352 $$FF_{cor}(T) = 1 - exp(-\sum[k_{corr}(T) \cdot \Delta T] \cdot V_a),$$ (11)

353 An example of the impact of the background correction on the FF of the diluted snow sample collected on Nov 30th 2017

354 (discussed in section 4.3) is shown in Fig. 8.

355

356 **Figure 8: SA water data (black dots) and corresponding fit (red line, Eq. 8) including the 95th percentile confidence interval (dashed-**
357 **dot magenta lines). The blue circles represents the diluted snow sample collected on Nov 30th 2017 which is then corrected for the**
358 **contributions of freezing from the SA water using the background correction ($FF_{cor}(T)$ as described in Eq. 11; cyan circles).**

359

4.2 Comparison of DRINCZ to other immersion freezing techniques

361 To validate the performance of DRINCZ, we use different wt. % NX-illite suspensions to compare the results from DRINCZ

362 to those summarized in Hiranuma et al. (2015), Beall et al. (2017) and Harrison et al. (2018). In the atmosphere, illite constitutes

363 up to ~40 % of the transported dust fraction (Broadley et al., 2012; Murray et al., 2012), making it an excellent surrogate for

364 atmospherically relevant dust. An initial stock suspension of 0.1 wt. % NX-illite was prepared with SA water and then diluted

365   to produce mass concentrations of NX-illite of 0.05 and 0.01 wt. %. The suspensions were manually shaken for 30 s, poured

366   into a dispensing tray and then immediately pipetted into the well plate. Triplicates of each suspension concentration were

367   investigated with DRINCZ (see Fig. A4 for *FF* curves) and then normalized to the number of active sites per BET-derived

368   surface area (n_{sBET}) using a variation of Eq. 2 as follows:

369     $$n_{sBET} = -\frac{\ln(1-FF)}{V_a * SA_{BET} * C_{NX}},\qquad\qquad(12)$$

370   where SA_{BET} is the BET surface area of the particles used (NX-illite) and C_{NX} is the mass concentration of NX-illite in an

371   experiment.

372

373   The n_{sBET} of NX-illite calculated using Eq. 12 from the measurements made with DRINCZ and background corrected (using

374   Eq. 11) falls within the results from Hiranuma et al. (2015), Beall et al. (2017) and Harrison et al. (2018) (Fig. 9). In theory,

375   n_{sBET} should be insensitive to concentration as the number of ice nucleating sites is normalized to the total surface area. Indeed,

376   the differing weight percent samples overlap to an extent (Fig. 9). Furthermore, the lower-weight-percentage samples extend

377   the observable n_{sBET} to higher values and colder temperatures. Similar to the observations of Harrison et al. (2018), the data

378   points from the 0.01 wt. % suspension appear as outliers at the warmest temperatures. However, it is not possible to determine

379   if these outliers are due to random freezing events that occur at high temperatures and therefore produce elevated cumulative

380   n_{sBET} values at lower temperatures or if they are due to an uneven distribution of the active sites in each aliquot that may

381   result from diluting a single stock suspension rather than preparing individual weight percent suspensions (Harrison et al.,

382   2018). Thus a spread equivalent to or less than the spread in the concentrations, up to an order of magnitude in this case, can

383   be expected. Furthermore, considering the ± 0.9 ˚C uncertainty, depicted by the horizontal error bars, the differences between

384   concentrations are not significant. They fall within the same range as the measurements of Beall et al. (2017) and between

385   BINARY and Leeds-NIPI and IR-NIPI at colder temperatures (Fig. 9). The overlap between the n_{sBET} measured with DRINCZ

386   and the NX-illite parameterization (Hiranuma et al., 2015) indicate that DRINCZ is capable of accurately measuring the

387   concentration of INPs and their active sites in the immersion freezing mode (Fig. 9).

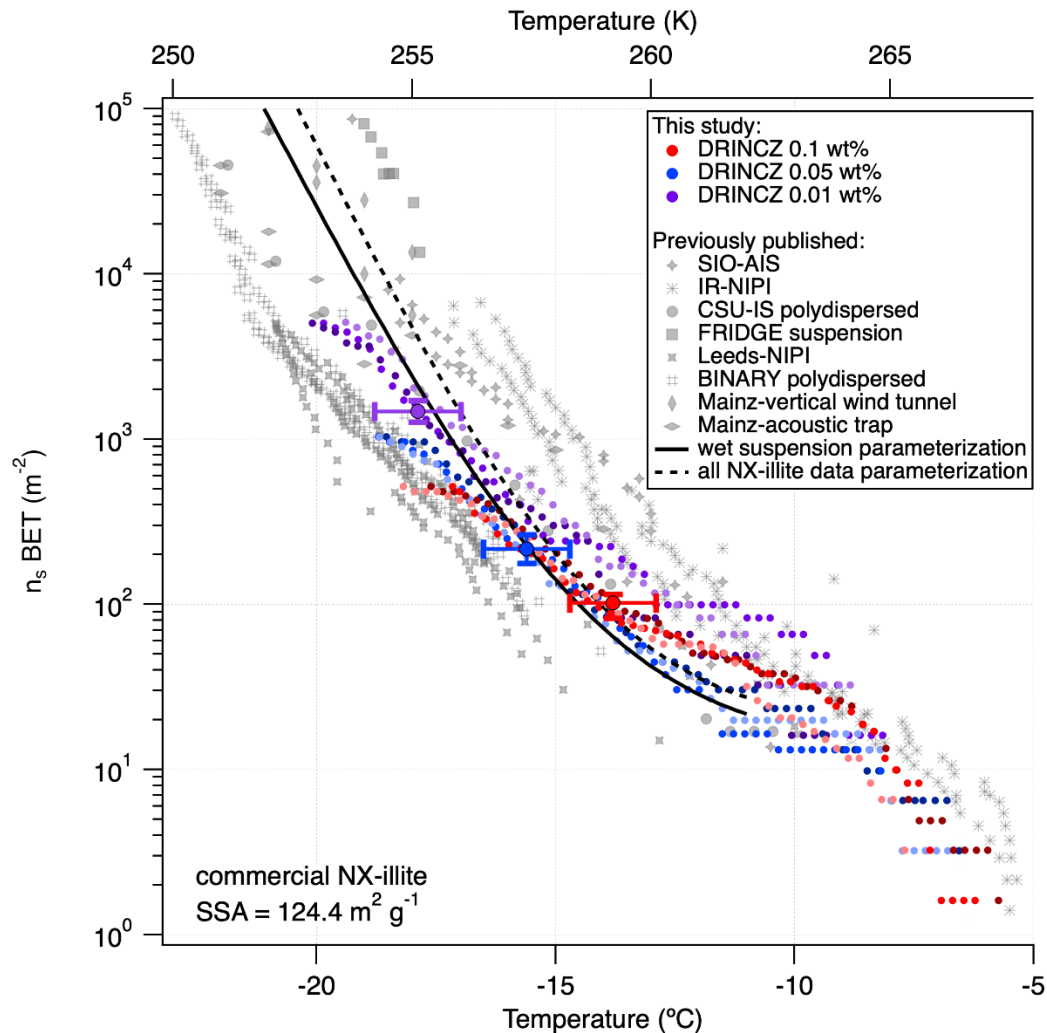



Figure 9: Triplicates of n_{sBET} (depicted by shading of the same color)as a function of temperature for three concentrations of NX-illite, 10^{-3} g ml^{-1} (red dots), 5×10^{-4} g ml^{-1} (blue dots) and 10^{-4} g ml^{-1} (purple dots),measured by DRINCZ. An example of the temperature uncertainty and the uncertainty due to the background correction are depicted for each weight percent as horizontal and vertical error bars, respectively. Literature values from Hiranuma et al, (2015), Beall et al, (2017) and Harrison et al, (2018) are shown for comparison. n_{sBET} was calculated using a BET surface area of 124.4 m^{2}g^{-1} (Hiranuma et al., 2015).

4.3 Ice Nucleating Particle Concentrations in Snow Samples from a Mountaintop Observatory in Austria

In order to demonstrate the performance of DRINCZ, snow samples collected between the 27th and 30th of November 2017 at the Sonnblick Observatory (SBO) were analyzed. The SBO is located at 3106 m on the summit of Mt. Sonnblick in the Hohe Tauern Region of Austria and has previously been used for cloud microphysical measurements (e.g. Beck et al., 2018; Puxbaum and Tscherwenka, 1998). Freshly fallen snow was collected from a wind-sheltered area where the snow could not drift. A stainless steel shovel (Roth) was conditioned with snow by turning (10 times) in the surface snow next to the sampling

site prior to sampling. The snow was then sampled into sterile NascoWhirlPaks (Roth) and then melted at room temperature (20 ˚C), immediately after which aliquots of snow-meltwater were filled into sterile centrifugation tubes (15 ml, Falcon tubes) and stored at -20 ˚C. The samples were shipped and stored frozen until processed with DRINCZ at the Atmospheric Physics Laboratory at ETH Zurich, to minimize any bacterial growth or changes due to liquid storage (Stopelli et al., 2014). The snowfall collected at SBO occurred during two snowfall events. The first event began on the 25th and ended overnight on the 26th (early hours of the 27th) while the second event (28th -30th) was associated with an intensifying upper level trough, a developing surface cyclone, a strong cold front and an associated secondary low (see Fig. A5 and A6).

The frozen fractions of five different snow samples were determined using DRINCZ and the cumulative concentration of active sites (or *INP(T)*, see Eq. 1) were normalized to per L of meltwater (n_{mw}) (Fig. 10). Overall, the n_{mw} of the snow samples fall within the range of previously reported values for precipitation samples (Petters and Wright, 2015) except for the November 30th sample. Within these samples, we identify (1) a particularly active snow sample (Nov 28th), (2) samples having intermediate IN activity (Nov 27, 29), and (3) a least active sample (Nov 30th). We attempt to compare these snow samples based on their air mass origin.

The snowfall sampled on the 28th had the highest n_{mw} of all collected samples (Fig. 10). The meteorological conditions and a comparison of back trajectories indicate that the air mass was associated with the warm sector of a synoptic system (Fig. A7) that originated from North America and the North Atlantic that then crossed France and Switzerland, before arriving at SBO (Fig. A8). In contrast, the arctic air mass responsible for the snowfall sampled on the 27th originated over Svalbard before crossing Iceland, the British Isles, Northern France and Germany (Fig. A8).

Even though the local conditions at SBO did not change significantly between the 28th and 29th, a decrease in n_{mw} was observed relative to the 28th and n_{mw} gradually decreased between the first and second sample on the 29th (Fig. 10). The back trajectories show that the origin of the air mass changed from North America and the North Atlantic on the 28th to exclusively originating over the North Atlantic on the 29th (Fig. A8). Additionally some of the back trajectories on the 29th show an increased interaction with the boundary layer over Europe (Fig. A8). Nevertheless, the decrease in n_{mw} suggests that if boundary layer aerosols from parts of Europe did reach the precipitating clouds at the SBO, they are less efficient INPs than the marine aerosols (Lacher et al., 2017, 2018) associated with the samples on the 27th and 28th.

Finally, the lowest n_{mw} observed were from meltwater collected on the 30th. The cold frontal passage and associated cold air advection caused the temperature to drop by 6 ˚C by noon on the 30th (Fig. A7) and the n_{mw} in the associated snowfall decreased substantially, exceeding the lower limit of previously reported n_{mw} values (Petters and Wright, 2015, Fig. 10). The decrease in n_{mw}, however, cannot be explained solely on the origin of the air mass as the arctic air mass on the 27th also crossed similar parts of the UK or had significant interaction with the marine boundary layer. Nevertheless, the concentration of INPs in the

sea surface microlayer is variable and the efficiency of emitting marine INP from the surface is wind speed dependent (DeMott et al., 2016; Irish et al., 2017; McCluskey et al., 2018; Wilson et al., 2015). Therefore, even though the trajectories on the 27th and 30th interacted with the marine boundary layer, they may contain different concentrations of INPs, yielding the observed differences in n_{mw}. In addition to air mass origin, it has been shown that precipitation efficiently removes INP and thus influences n_{mw} (Stopelli et al., 2015). Indeed, the most upstream precipitation (see Fig. A8) corresponds to the sample collected on the 30th, which has the lowest n_{mw}. Therefore, the most efficient INPs could have been removed in the upstream precipitation, contributing to the observed decrease in n_{mw}.

The differences in n_{mw} could not be rectified by a single metric in this study but rather a combination of factors likely led to the observed variability. In particular, as the warm sector of the cyclone approached the sampling site (28th), n_{mw} increased. Conversely, after cold frontal passage (30th) the n_{mw} decreased. Back trajectories indicate that the air mass source region and the amount of upstream precipitation differed between the two sectors of the cyclone. This result is consistent with previous studies that suggest that air mass origin (e.g. Ault et al., 2011; Creamean et al., 2013; Field et al., 2006; Lacher et al., 2017, 2018) and upstream precipitation (Stopelli et al., 2015) influences the INP concentration. Furthermore, the dependence on the long range air mass history to the observed variability in n_{mw} suggests that local sources are not responsible for the observed INPs.

4.4 Limitations of snow meltwater sample comparisons

One limitation when comparing snow samples collected at different times and locations is the unknown number of aerosols, INPs and ice crystals that contributed to the collected meltwater. Since n_{mw} depends on the number and mass of the ice crystals within a snow sample, the melt water volume or density of each snowflake influences n_{mw}. For example, snow to liquid ratios, which can be used as a proxy for snow flake density and melt water equivalent, can vary between 5 to 1 in heavy wet snow and 100 to 1 in powdery snow (Roebber et al., 2003). However, even when considering this variability in the required amount of snow to produce the same volume of ice crystal melt water, n_{mw} would only differ by a factor of 20. As can be seen in Fig. 10, n_{mw} varies by two orders of magnitude or more between the 28th and the 30th of November and the difference is therefore robust. Additionally, heavy wet snow has been found to occur in the warm core of a synoptic system while lighter, more powdery snow was found in the air mass after cold frontal passage, where air temperatures are colder (Roebber et al., 2003). As the n_{mw} on the 28th was collected in the warm sector and the sample on the 30th was post cold front, differences in snow density may lead to an underestimation in the difference between the n_{mw} of these two samples. Therefore, we recommend that future studies also consider the snow water equivalent when comparing the n_{mw} as this could influence n_{mw} by a factor of 20 or more.

Another uncertainty with using precipitation samples for analyzing INP concentrations is associated with aerosol scavenging and chemical ageing (e.g. Petters and Wright, 2015). As previously mentioned, the samples were stored frozen to avoid any

468   decrease in ice nucleating ability associated with storage (Stopelli et al., 2014) and therefore degradation is likely not an issue

469   in this study (Wex et al., 2019). The ability of a falling ice crystal to scavenge aerosols or rime cloud droplets depends on the

470   ice crystal habit, size, and the difference between the fall velocity of the crystal and the interstitial aerosol or cloud droplets.

471   With the exception of interstitial aerosol concentration which has been shown to influence n_{mw} by a factor of 2 (Petters and

472   Wright, 2015), these factors are all important when estimating snow density and thus make it difficult to disentangle their

473   effects on n_{mw}. Therefore, there is value in future studies of INP in MPCs to investigate the INP concentrations in cloud water,

474   interstitial aerosols and snow samples.

475   **4.5 Ice Nucleating Particle Concentrations in Diluted Snow Samples**

476   In order to extend the reported temperature range of DRINCZ, the snow samples were also diluted by a factor of 10 with SA

477   water (see Eq. 2). The dilutions (open symbols) overlay the pure samples except at the warmest temperatures where, as

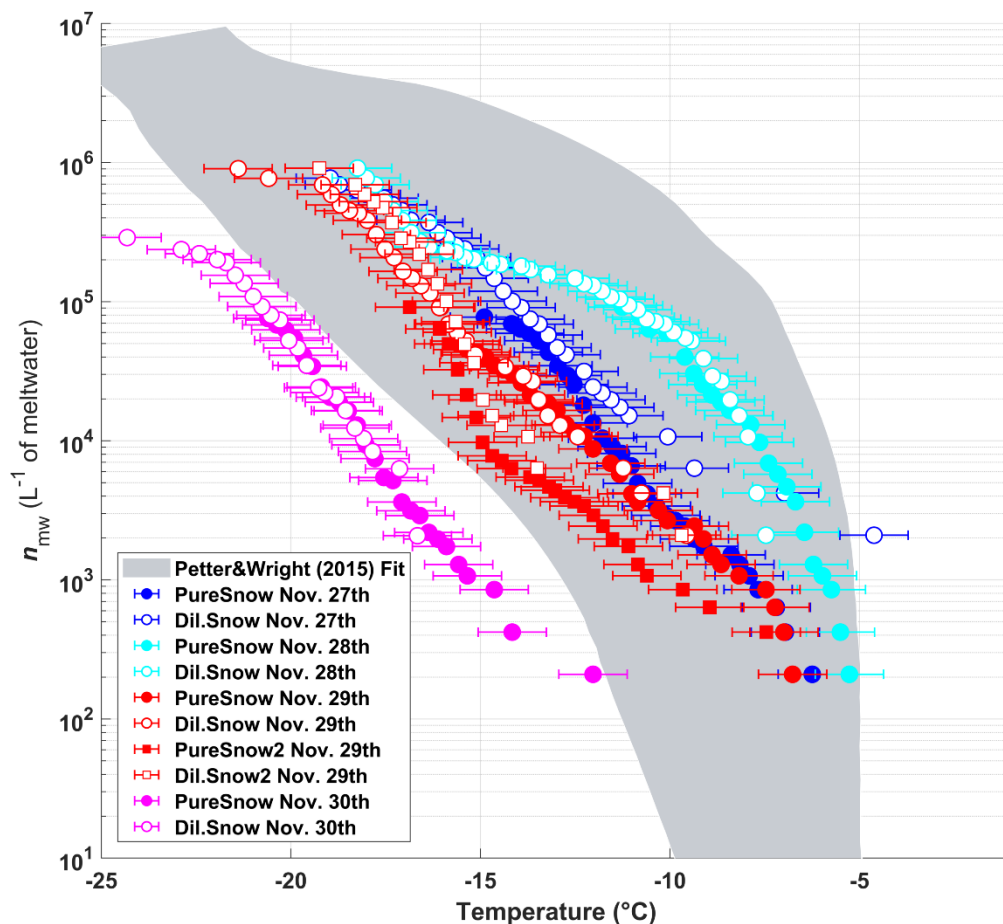
478   previously mentioned, a single freezing event can lead to an increase in n_{mw} of an order of magnitude relative to the undiluted

479   sample. This effect is especially evident on the 27th when the first few wells of the diluted sample (open blue circles) froze at

480   the same or higher temperatures than the undiluted sample (filled blue circles) and led to an increase in n_{mw} of up to an order

481   of magnitude. However, this issue has been previously observed when diluting from stock suspensions (Harrison et al., 2018)

482   which is similar to diluting a snow water sample. Therefore, the dilutions further validate DRINCZ as an INP measurement

483   technique.

484

485 **Figure 10: The cumulative number of active sites per L of meltwater (n_{mw}) of snow for undiluted snow (filled) and of snow samples**
486 **diluted by a factor of 10 (white-filled symbols) as a function of temperature. The colors represent the different sampling days. On**
487 **the 29th of Nov. two samples were taken and the second sample of the day is indicated by square symbols. The shaded area represents**
488 **the previously reported n_{mw} from precipitation events as described in Petters and Wright (2015). The error bars represent the**
489 **instrumental temperature uncertainty of ± 0.9 ˚C.**

490 **5 Conclusions**

491 We describe and characterize DRINCZ as a newly developed drop freezing instrument for quantifying the ability of aerosols

492 to act as ice nucleating particles in the immersion freezing mode. The instrument uncertainty is ± 0.9 ˚C, similar to previously

493 published drop freezing techniques. We show that thermal contraction of ethanol as a coolant used in bath-based drop freezing

494 techniques increases temperature variations within the sample. This issue can be corrected by incorporating a bath leveler

495 which ensures the coolant level in the bath remains constant during an experiment. Typical drop freezing methods report

496 temperature measured in the corner wells of a 96-well tray, at the edge of a cooling block or within the block itself (Beall et

497 al., 2017; Hill et al., 2014; Stopelli et al., 2014). Here we show that by making use of the freezing sequence of pure water

498 aliquots, the spatial pattern of temperature bias in the 96-well tray can be assessed. Although variations are within the

instrumental uncertainty of DRINCZ and are not used for DRINCZ data analysis, we present our detailed analysis of this potential bias and draw attention to this issue for other drop freezing techniques. The calculated bias correction increases the precision of drop freezing setups, and is an alternative to computationally expensive heat transfer simulations (Beall et al., 2017). Validation experiments conducted with NX-illite showed good agreement with data reported in the literature for this INP standard.

We exemplify the use of DRINCZ by measuring the concentration of INP in snow samples collected at the Sonnblick Observatory in Austria. The observed INP concentrations are within previously reported values as summarized in Petters and Wright, (2015) for the same temperature range as investigated here (-22 to 0 ˚C). Differences in INP concentration can be explained by differing sectors of a mid-latitude cyclone. As the warm sector of the cyclone approached the sampling site, the INP concentration increased while after the cold front passed the INP concentration decreased. Back trajectories indicate that the air mass source region and the amount of upstream precipitation differed between the two sectors of the cyclone. This result is consistent with previous studies that suggest that air mass origin (e.g. (Ault et al., 2011; Creamean et al., 2013; Field et al., 2006; Lacher et al., 2017) and upstream precipitation (Stopelli et al., 2015) influence the INP concentration. This suggests that INP in precipitation samples are likely transported from specific source regions rather than originate from local sources. Thus identifying the specific sources responsible for INP and their transport pathways are essential for accurately modelling the ice phase in clouds and ultimately, climate.

Competing Interests

The authors declare that they have no conflict of interest

Code Availability

The code for detecting the wells and determining when a freezing event occurs is written in MATLAB and available upon request from the authors.

Data Availability

The data presented in this publication are available at the following DOI: https://doi.org/10.3929/ethz-b-000369839

Author Contributions

DRINCZ was developed and designed by R.O.D with the assistance of M.C.C, M.R., L.S.B, K.P.B and N.B.D. The SA water experiments were conducted by M.C.C, K.P.B., L.S.B, V.W, J.W, S.B, and R.O.D. The temperature calibration and NX-illite experiments were conducted and analyzed by K.P.B and N.B.D. The snow samples were collected by N.E. and analyzed by R.O.D. The instrumental error, uncertainties and calibration were conducted by R.O.D. with contributions from Z.A.K and C.M. The automation and analysis software was developed by R.O.D. with contributions from K.P.B and M.C.C. The well plate holder was designed by M.R. and R.O.D. and manufactured by M.R. The manuscript was written by R.O.D with contributions from N.B.D, C.M. and Z.A.K. The project was supervised by Z.A.K.

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Appendix A

Freezing bias by user

The 20 SA water experiments were performed over a three month period by two users. The SA water was unaffected by aging over this period as it originated from varying bottles distributed by the manufacturer (Sigma Aldrich). The user bias was calculated the same way as the bias for all 20 experiments. The bias is relative to the median freezing temperature of the 4 corner wells obtained by the respective user. As can be seen in Fig. A3, the pattern of freezing bias is consistent regardless of the user. This similarity indicates that the reported bias is instrumental and not user specific.

Bias significance and correction

To ensure that the observed bias is statistically significant, a two-sample, two-tailed t-test was performed. In particular, a Welch's t-test was used due to the different number of samples between the combination of the 4 reference wells (20

552 experiments x 4 wells = 80 values) and each well (20 experiments x 1 well = 20 values) and the different variance of freezing

553 for each well (Derrick and White, 2016). In a Welch's *t*-test the location parameter of two independent data samples is assessed

554 as follows:

555
$$t = \frac{\bar{w}_{4ref} - \bar{w}_i}{\sqrt{\frac{s^2_{4ref}}{Nw_{4ref}} + \frac{s^2_i}{Nw_i}}}$$
(A1)

556 where $\bar{w}_{4ref}$ and $\bar{w}_i$ are the mean freezing temperature of the reference wells and an individual well, respectively. s^2_{4ref} and

557 s^2_i are the variances of freezing in the reference and the individual wells and Nw_{4ref} and Nw_i are the number of samples for

558 the reference wells and an individual well, respectively. The variance of the freezing temperature of SA water in each well is

559 shown as boxplots in the Appendix (Fig. A2). The temperature of approximately 30% of the wells was found to be statistically

560 different from the average freezing temperature of the 4 reference wells at the 95% confidence level, with a resultant mean

561 bias of 0.23 ˚C (Fig. 4b). Due to a fraction of wells with a statistically significant bias, a correction factor based on the mean

562 bias from the 20 SA water experiments is tested for all wells excluding the 4 corner wells used as the reference to avoid

563 overfitting the data. Of note, the reported bias is derived based on the freezing range of SA water from -16 to -26 °C. However,

564 based on the relatively constant spread in the temperature calibration data (see Fig. 3b), it is reasonable to assume that the bias

565 has a weak temperature dependence.

566 Although the freezing bias was shown to be representative when the SA water data was split in two (8 and 12 samples), it is

567 still necessary to validate its robustness on a larger sample size. In order to artificially increase the sample size of the

568 experiments, the bias was recalculated randomly such that only 90% or 18 of the experiments were used. The resultant bias

569 correction was then applied to the remaining 10% or 2 of the experiments and tested to see if the mean freezing temperature

570 of the bias corrected tray was closer to the reference freezing temperature of the 4 corner wells. This procedure was repeated

571 1000 times at random. The difference in the median freezing temperature (*FF*= 0.5) and 4 corner reference wells decreased

572 from 0.23 ˚C to 0.04 ˚C, while the standard deviation of the bias corrected data increased by 0.007 ˚C. Thus, the bias correction

573 performed as expected and reduced the bias in freezing temperature. Nonetheless, this improvement falls within the uncertainty

574 of the instrument, as discussed in Section 3.3 and is therefore not applied to DRINCZ measurements by default.

575

576 **Synoptic Summary Nov. 27th-30th**

577 The synoptic pattern over Europe on the 27th through 30th of November produced large variations in both temperature and air

578 mass origin at the SBO. As can be seen from the surface pressure maps shown in Fig. A5, an evolving cyclone tracked across

579 Northern Europe before occluding in the vicinity of Denmark. This cyclone produced strong warm advection at SBO on the

580 27th (see Fig. A7) in advance of the approaching cold front. As the cyclone began to fill over Southern Scandinavia, the cold

581 front stalled along the Alps and westerly flow continued at SBO from the 28th – 29th (Fig. A7). Farther west, the cold front

582 reached the Mediterranean where a secondary low developed along the remnant baroclinic zone (Fig. A6 panel c.). This

583   secondary low traversed Italy and rapidly intensified as it crossed the Adriatic Sea before entering the northern Balkans (Fig.

584   A6 panel d.). The secondary low and an amplifying ridge over the British Isles forced the cold front over SBO at 00Z on the

585   30th when cold air advection ensued over the SBO region (Fig. A7), as shown by the back trajectories (Fig. A8.e and f.).

586   **HYSPLIT back trajectories**

587   The Hybrid Single-Particle Lagrangian Integrated Trajectory model (HYSPLIT) (Stein et al., 2015) was run using the

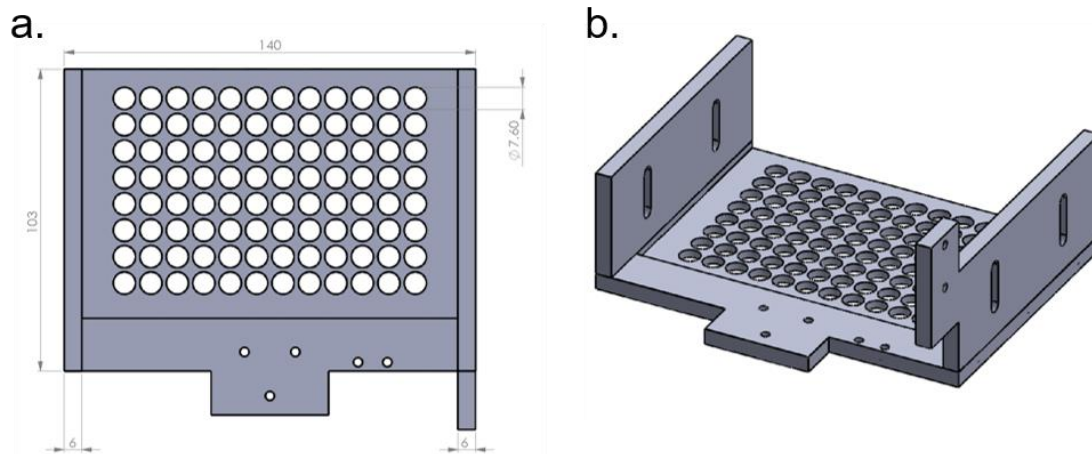
588   interactive web portal (Rolph et al., 2017). The trajectories were calculated using 0.5˚ resolution and the trajectories were

589   initialized 1000, 2000 and 3000 meters above the model terrain height. Although the majority of snow mass growth has been

590   shown to occur between mountaintop and 1 km above the surface (Lowenthal et al., 2016), these heights were chosen due to

591   the coarse resolution of the model terrain height and the observed sensitivity of the back trajectories with height. HYSPLIT

592   was initialized using the 0.5˚ hourly Global Data Assimilation System (GDAS) archived database and the vertical velocity was

593   model based rather than isentropic.

594

595   **Figure A1: Schematic of the 96-well tray holder from above (a) and the side (b), dimensions are in millimeters.**

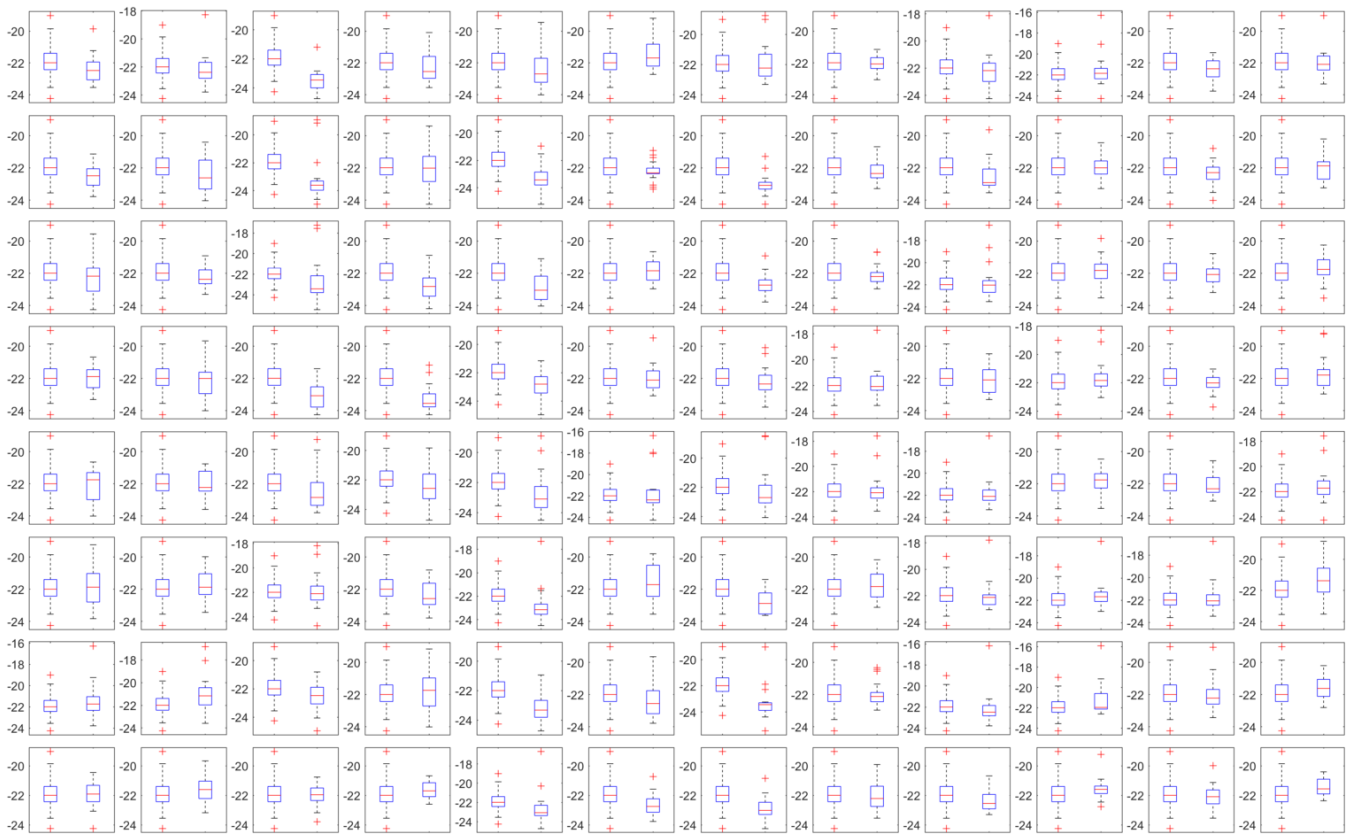



Figure A2. A side by side comparison of box plots for the freezing temperatures of the 20 SA water experiments of the reference wells (left box) and the well represented by the location (right box) of each subplot. The median (red line), inter-quartile range (blue box), extreme values not considered outliers (whiskers) and outliers (red crosses) are shown as a function of temperature in ˚C (y-axes).

602

603 **Figure A3: (a) Bias in the freezing of SA water (°C) based on the median value of each well over 12 experiments and (b) 8 experiments**
604 **relative to the median temperature of freezing for the 4 corner wells used during the temperature calibration. A positive (negative)**
605 **bias indicates that the wells experience a warmer (colder) temperature than the four corner wells used for temperature calibration**
606 **and therefore freeze at lower (higher) temperatures than reported.**

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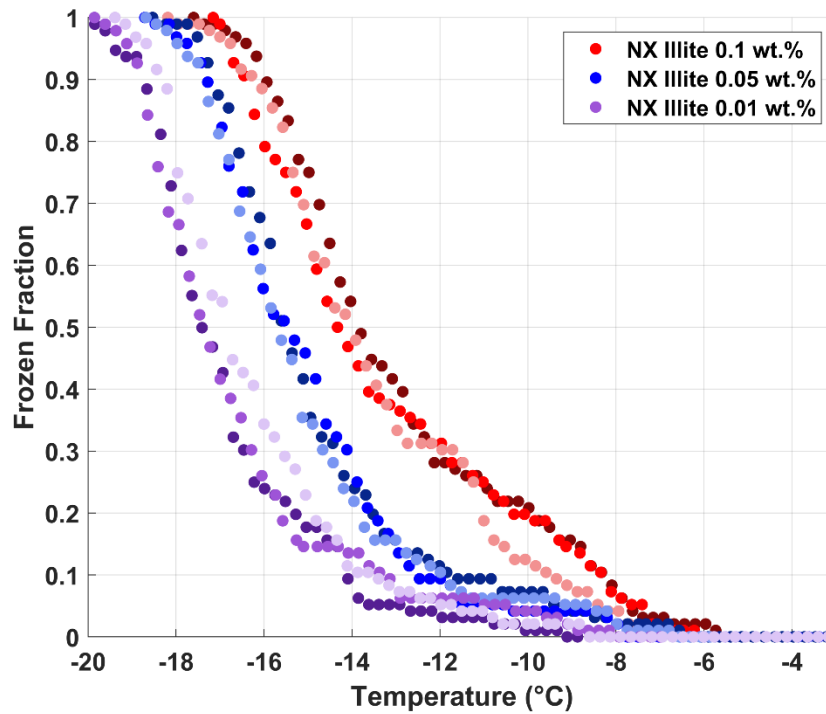
Figure A4. Frozen fraction curves of suspensions of 0.01 wt. % (magenta dots), 0.05 wt. % (red dots) and 0.01 wt. % (purple dots) of NX-illite run in triplicates as shown by shading.
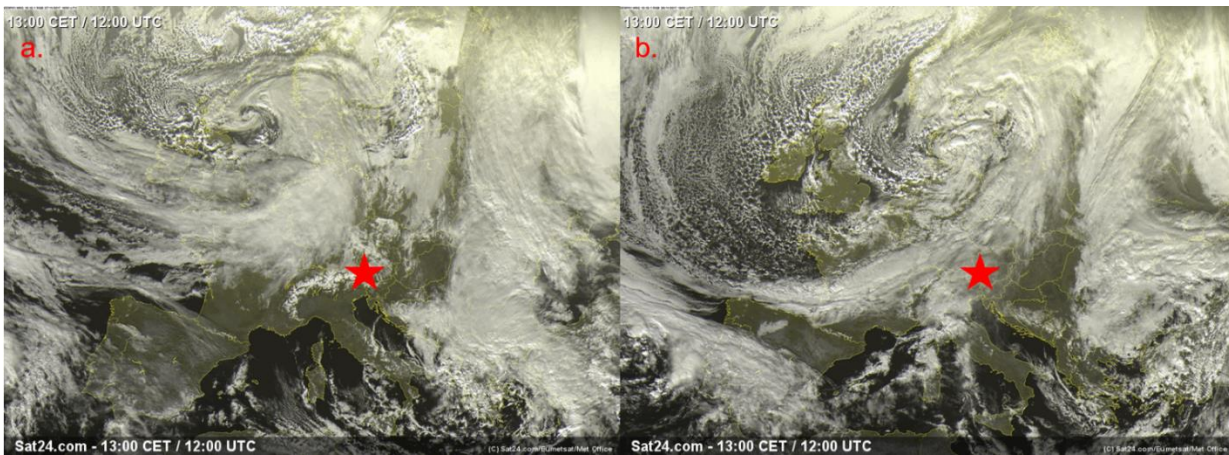
Figure A5: Visible satellite image of the storm system impacting the SBO (red star) taken at 1200UTC on (a) Nov. 27th and (b) 28th. Images courtesy of Sat24.com/Eumetsat/Met Office (http://www.sat24.com/history.aspx).

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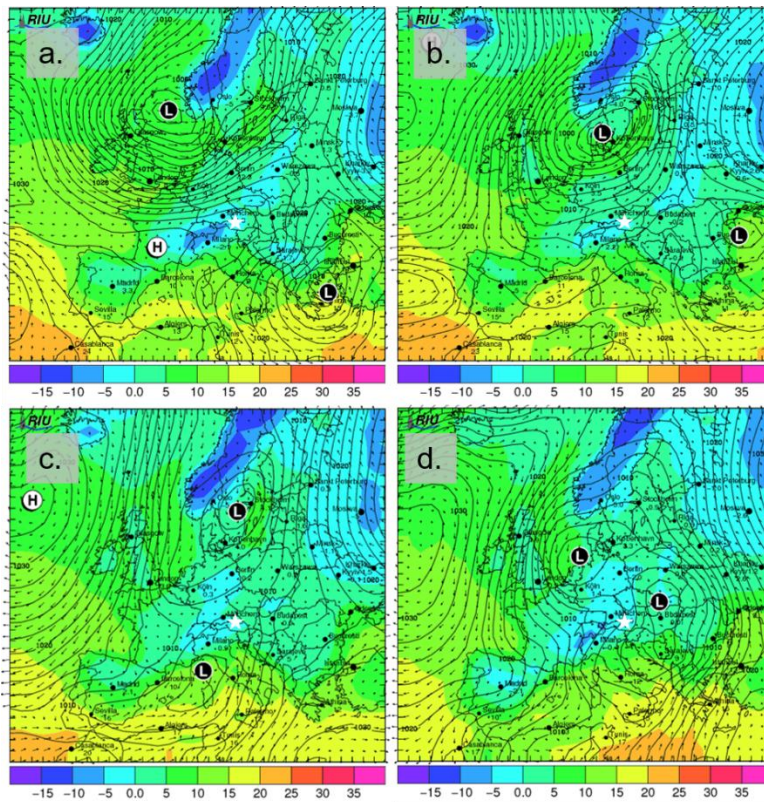
Figure A6: Forecasted surface pressure in hPa (black contours), 2 meter surface temperature in ˚C (color fill), and wind vectors in m/s (black arrows) for 12 UTC on (a) 27ᵗʰ , (b) 28ᵗʰ , (c) 29ᵗʰ and (d) 30ᵗʰ. Forecasts are based on model runs initialized on 00 UTC of the day of interest (12 hours before shown values). Surface low and high pressure centers are indicated with L and H, respectively. The location of SBO is shown by the white star. Images are taken and adapted from the Rhenish Institute for Environmental Research at the University of Cologne (http://www.uni-koeln.de/math-nat-fak/geomet/eurad/index_e.html).

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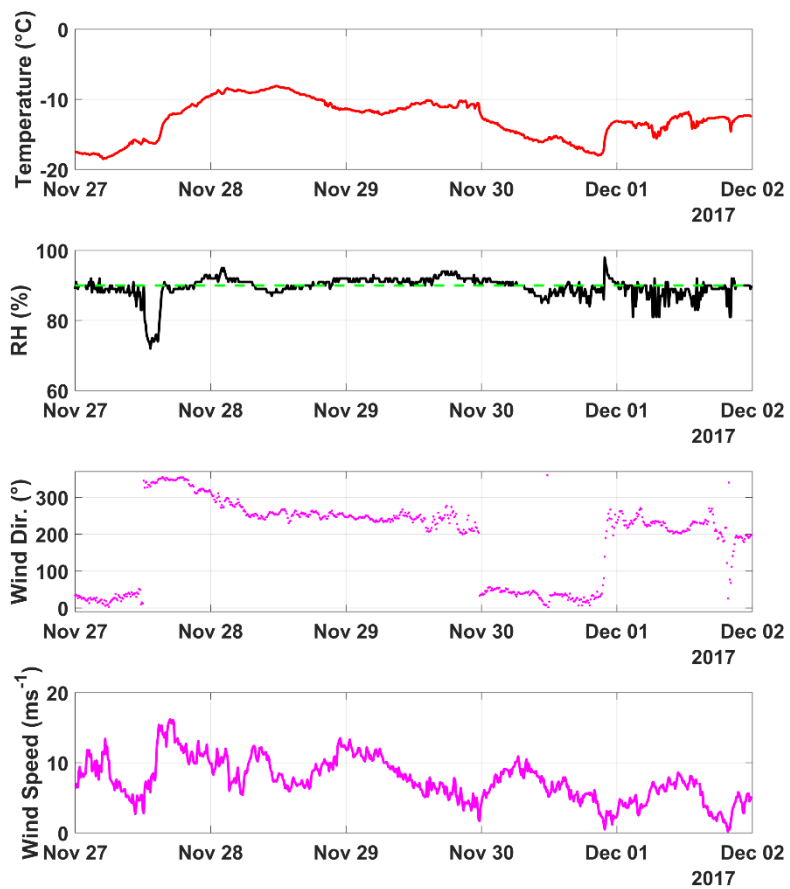
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625 **Figure A7: (top panel) Temperature (°C), (top-middle panel) humidity (%), (bottom-middle panel) wind direction (°) and (bottom**
626 **panel) wind speed (ms⁻¹) as a function of date spanning from the 27th of November to the 2nd of December (in UTC). The humidity**
627 **when cloud is present at SBO (90%) is shown (dashed green line).**

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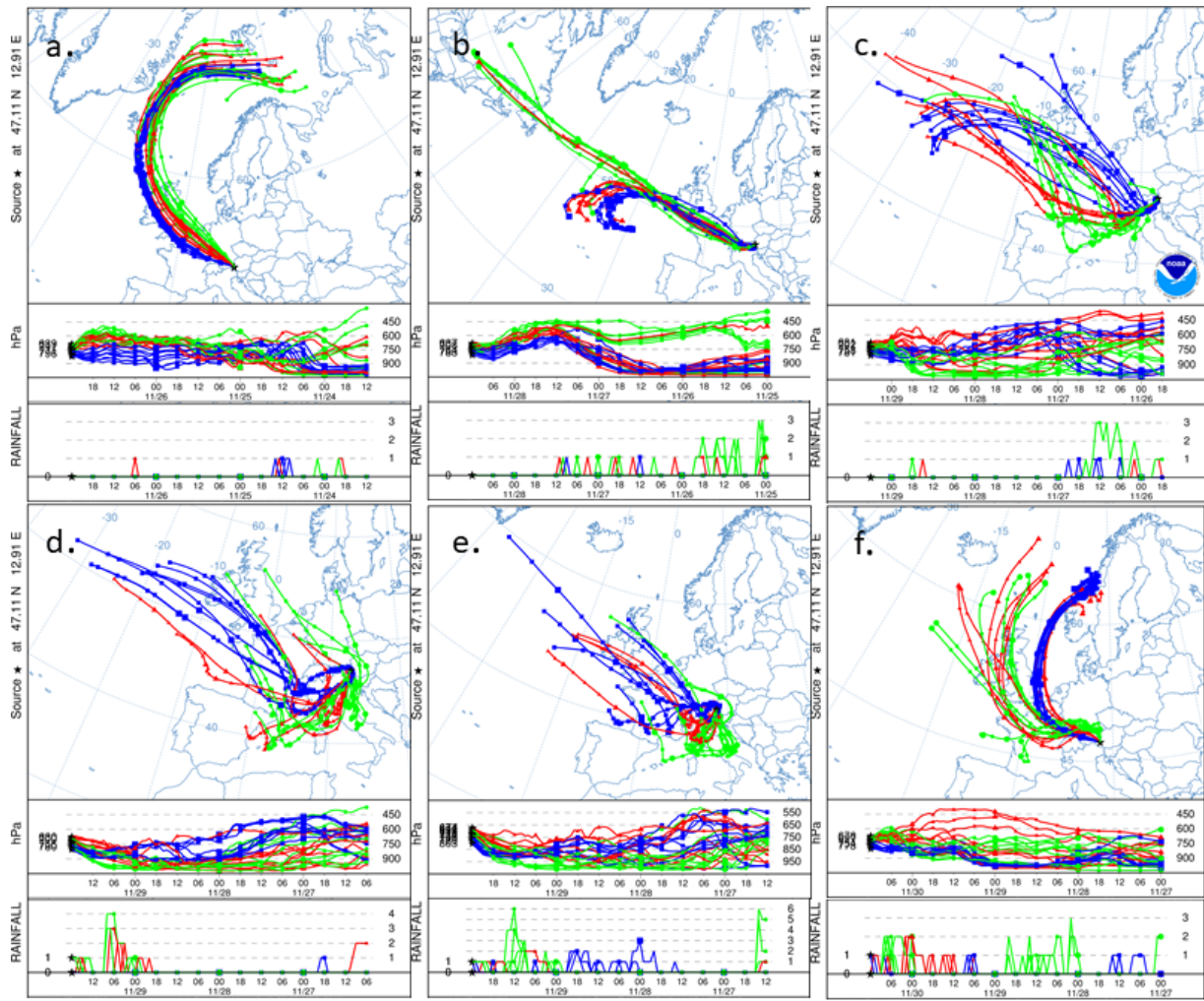
629 **Figure A8: (a) 84-hour HYSPLIT back trajectories from the Sonnblick Observatory initialized on 00 UTC on the 27th, (b) 12 UTC**
630 **on the 28th, (c) 06 UTC and (d) 18 UTC on the 29th, and (e) 00 UTC and (f) 12 UTC on the 30th of November. The blue, green and**
631 **red lines represent 8 ensemble back trajectories initialized 1000 m, 2000 m and 3000 m above the model terrain height, respectively.**
632 **The two lower panels in each subplot show the back trajectory height in units of pressure (hPa) and rainfall (mm) as a function of**
633 **time (in 6 hourly intervals) as a function of pressure (in hPa).**

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