

# Continuous methane concentration measurements at the Greenland Ice Sheet-atmosphere interface using a low-cost low-power metal oxide sensor system

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**Abstract.** In this paper, the performance of a low-cost and low-power methane (CH<sub>4</sub>) sensing system prototype based on a metal oxide sensor (MOS) sensitive to CH<sub>4</sub> is tested in a natural CH<sub>4</sub> emitting environment at the Greenland Ice sheet (GrIS). We investigate if the MOS could be used as a supplementary measurement technique for monitoring CH<sub>4</sub> emissions from the GrIS with the scope of setting up a CH<sub>4</sub> monitoring network along the GrIS. The performance of the MOS is evaluated on  
15 basis of simultaneous measurements using a cavity ringdown spectroscopy (CRDS) reference instrument for CH<sub>4</sub> over a field calibration period of approximately 100 h. Results from the field calibration period show that CH<sub>4</sub> concentrations measured with the MOS is in very good agreement with the reference CRDS. The absolute concentration difference between the MOS and the CRDS reference values within the measured concentration range of approximately 2-100 ppm CH<sub>4</sub> were generally lower than 5 ppm CH<sub>4</sub>, while the relative concentration deviations between the MOS and the CRDS were generally below 10  
20 %. Calculated root mean square error (RMSE) for the entire field calibration period was 1.69 ppm (n = 37,140). The results confirms that low-cost and low-power MOS can be effectively used for atmospheric CH<sub>4</sub> measurements under stable water vapor conditions. The primary scientific importance of the study is that it provides a clear example on how the application of low cost technology can enhance our future understanding on the climatic feedbacks from the cryosphere to the atmosphere.

## 1. Introduction

25 Constraining the various sources and sinks in the global methane (CH<sub>4</sub>) budget is becoming an increasingly important parameter in mitigating climate change (Saunois et al., 2016). The Arctic is generally considered a major global emitter of CH<sub>4</sub> to the atmosphere, but significant uncertainty exists to the seasonal dynamics and strength of both CH<sub>4</sub> sources and CH<sub>4</sub> sinks from both terrestrial and marine environments, as well as the cryosphere (Callaghan et al., 2011; Emmerton et al., 2014; Juncher Jørgensen et al., 2015; Pirk et al., 2017; Zona et al., 2016). A previously unknown source of CH<sub>4</sub> emission to the  
30 atmosphere was recently identified where CH<sub>4</sub> is emitted from meltwater originating in the subglacial domain of the Greenland Ice Sheet (GrIS) (Christiansen and Jørgensen, 2018; Lamarche-Gagnon et al., 2018; Wadham et al., 2019). The spatiotemporal

coverage of the new CH<sub>4</sub> source is yet to be determined and the overall climatic importance of this new component in the Arctic CH<sub>4</sub> budget is still unknown. Future studies are needed in order to assess the overall climatic significance of this source of CH<sub>4</sub> emission from the cryosphere to the atmosphere. The current state of knowledge on the CH<sub>4</sub> exchange from Greenland is inherently limited by the remoteness of many field sites with following high expedition cost and limitations to the spatial coverage and temporal duration of field measurements. Adding to this is the financial and logistical challenges of bringing high precision analyzers into the field, keeping them powered, running and shielded in the harsh environments often encountered in the Arctic. Thus, there is substantial potential and need to develop low-power techniques and measurement systems that can perform reliable autonomous CH<sub>4</sub> concentration measurements. The emergence of low-cost/low-power sensor technology in recent years provides an opportunity to overcome many of current restraints on obtaining continuous field measurements from a wide range of natural CH<sub>4</sub> emitting systems (wetlands, ice sheets, marine gas seeps, lakes, permafrost) and expand the network of continuous measurements in remote areas maximizing our understanding of these systems and minimizing the risk of losing valuable analytical equipment.

Low-cost metal oxide gas sensors (MOS) have been widely used for sensing various gases under atmospheric conditions (Wang et al., 2010). However, MOS sensors have significant obstacles to their direct use as air quality monitors as their output signal is influenced by the concentrations of both the target and interfering gases, as well as the temperature and humidity effects (Masson et al., 2015; Sohn et al., 2008). Other known challenges to the use of MOS are baseline drift over time, caused by either changes in the heat output of the sensing element or due to poisoning of the sensor surface (Peterson et al., 2017).

In recent years, investigations into the performance of CH<sub>4</sub> sensitive MOS sensors for the measurement of atmospheric CH<sub>4</sub> have been made under both natural and controlled conditions (van den Bossche et al., 2017; Eugster and Kling, 2012; Penza et al., 2015). These studies have been prompted by an increased interest in finding effective methods to quantify CH<sub>4</sub> emissions to the atmosphere from both natural systems and man-made systems such as landfills or biogas production plants. Using sensor specific post-processing to compensate for variations in relative humidity and air temperature, the previous studies have demonstrated a high potential for the low-cost and low-power monitoring of CH<sub>4</sub> concentrations above the atmospheric background level for various applications and in sensor networks. In the current study, we have tested the in situ performance of a CH<sub>4</sub> sensitive MOS (Figaro TGS2611-E00) against a state-of-the-art cavity ring-down spectrometer (CRDS) for CH<sub>4</sub> (Ultra-portable Greenhouse Gas Analyzer, Los Gatos Research Inc.) to measure CH<sub>4</sub> concentrations in the air expelled from a subglacial meltwater outlet at GrIS. This was done to assess the MOS's potential for serving as a sensing element for future studies of CH<sub>4</sub> emissions from the subglacial domain under the Greenland Ice Sheet.

## 2. Materials and methods

### 65 2.1 Field site and instrumentation

The study site is located on the southern flank of the Isunnguata Sermia Glacier at the western margin of the GrIS (67°09'16.40''N 50°04'08.48''W) at an elevation of 450 meter above sea level (Fig. 1). At a small subglacial meltwater discharge outlet in this area, we performed measurements of CH<sub>4</sub> concentrations in the subglacial air expelled from naturally occurring caves carved out by meltwater below the ice sheet. The measurements were done in the period between June 22<sup>nd</sup> and 26<sup>th</sup> 2018. A more detailed description of the study site at the GrIS is given in (Christiansen and Jørgensen, 2018).

To sample the subglacial air the sampling tube was attached to an aluminum pole inserted approximately 5 meters into an ice cave (Fig. 2a). The inlet of the sampling tube was connected to a 120 ml water trap. Humidity and temperature of the subglacial air were measured every 10 sec using a combined sensor (S-THB-M008, Onset, USA; resolution 0.02 °C, 0.1 % RH) mounted at the tip of the aluminium pole inserted into the cave. The data were recorded using a datalogger (U30, Onset, USA).

Real-time reference concentration (ppm) measurements of CH<sub>4</sub>, carbon dioxide (CO<sub>2</sub>) and water vapor (H<sub>2</sub>O) was obtained using a CRDS (Ultraportable Greenhouse Gas Analyzer, Los Gatos Research, USA). The inlet port of the CRDS was connected to the subglacial sampling point via a sampling tube (50 m length, inner diameter of 4 mm and total volume of 630 mL) which was zip-tied to the aluminium pole. Flow of sample gas from the subglacial sampling point to the measurement cell in the CRDS was obtained via the analyzer's internal diaphragm pump (800 mL min<sup>-1</sup>). The outlet port of the CRDS was connected in series via a 1 m plastic tube to a metal can enclosure (400 mL), where the lid had been removed (Fig. 2b). The prototype CH<sub>4</sub> sensing system (MOS) was placed in the metal enclosure, where the short serial tube connector ensured a rapid flushing of the headspace in which the CH<sub>4</sub> measurements with the MOS were made. Due to the non-destructive sampling principle of the CRDS and the rapid flushing of the headspace volume in the enclosure with the MOS system (2 times per minute), the concentration of CH<sub>4</sub> is estimated to be virtually identical at the same time step for the MOS and the CRDS during the entire field calibration period (22<sup>nd</sup> to 26<sup>th</sup> July 2018).

Following the field calibration test of approximately 100 h, the MOS system was left in the field as an autonomous monitoring system. For this autonomous measurement period, the CRDS was replaced by a 12 volt diaphragm pump (Thomas pumps, 1410VD DC) with a constant air-flow of approximately 3 L min<sup>-1</sup> attached to the common sample tube with similar connection of the pump inlet and outlet as the CRDS ports. During this period the MOS system was powered by 12V LiFePO<sub>4</sub> batteries connected to solar panels and a voltage regulator, placed in a water-proof case and buried under a pile of rocks to minimize the impact of sunlight induced temperature variations of the sensor system

## 2.2 The MOS system

The MOS system (Fig. 2c) consists of a microcontroller (Arduino Uno) and datalogger shield (DeekRobot data logging shield V1.0) holding the board-mounted metal oxide CH<sub>4</sub> sensor (Figaro TGS 2611-E00) and an additional temperature/relative humidity micro sensor (GY-21 HTU21). The final prototype was assembled in the laboratory at Aarhus University. Logging frequency of the CRDS and MOS was 1 and 10 seconds, respectively. The CH<sub>4</sub> sensitive MOS consists of a tin(IV)oxide (SnO<sub>2</sub>) semiconductor which has low conductivity in clean air. In the presence of CH<sub>4</sub>, the sensor's conductivity increases depending on the gas concentration in the air (Kumar et al., 2009). A simple electrical circuit converts the changes in conductivity at the sensing element as the gas concentrations vary to a change in output voltage across the voltage divider (see Fig. 3). Both the heater and the sensing circuit of the MOS were powered by the 5 volt regulated output of the Arduino Uno. The analogue output of the MOS was connected to the 10-bit analogue input on the Arduino Uno using a 10 kOhm precision load-resistor in the voltage divider.

## 2.3 Laboratory calibration of the MOS sensor

In preparing for the field test of the CH<sub>4</sub> sensing system prototype, the MOS was performance tested and calibrated in a controlled laboratory environment to evaluate both the response time to variations in methane concentration in the concentration range 0-100 ppm CH<sub>4</sub> at three different levels of relative humidity (37±2 %, 55±3 % and 76±3 %). Synthetic air (80 % N<sub>2</sub> and 20 % O<sub>2</sub>) was used as zero gas for the laboratory test to which various concentrations of a CH<sub>4</sub> containing span gas was mixed in using a HovaCAL calibration gas generator (IAS GmbH, Germany). After mixing of the zero gas and span gas, the calibration gas was humidified using a water filled impinger (van den Bossche et al., 2017). At each humidity level, the output voltage from the MOS was logged using a Campbell CR1000 datalogger at a 2 second sampling frequency. A pre-programmed calibration step sequence was used for all three humidity levels, consisting of time steps of each 10 minutes in which the sensor was exposed to either zero gas or a calibration gas mixture in the applied the concentration range in an alternating step pattern (Fig. 4). The temperature in the laboratory, zero gas, mixed calibration gas and water in impinger was kept constant around 22 °C throughout the laboratory calibration test.

The sensor resistance ( $R_0$ ) at exposure to the CH<sub>4</sub> free reference gas can be calculated at each of the three different humidity levels according to Eq. (1):

$$R_0 = \frac{V_C * R_L}{V_{OUT}} - R_L, \quad (1)$$

125 where  $V_C$  is the circuit voltage (i.e. 5 volt DC),  $R_L$  is the load resistance (10 kOhm) and  $V_{OUT}$  is the measured output voltage  
(see also Eugster and Kling (2012) for further description).

The sensor resistance at various calibration gas concentrations ( $R_S$ ) at different concentration steps in the calibration sequence  
can also be calculated using equation 1 for each of the three humidity levels (i.e.  $R_S$  replaces  $R_0$  in equation 1). For the tested  
130 type of MOS, the sensor resistance ratio ( $R_S/R_0$ ) between the sensor resistance at a given concentration level ( $R_S$ ) and the  
sensor resistance at the reference level ( $R_0$ ) are inversely proportional to the absolute  $CH_4$  concentration and can be modelled  
using e.g. a power fit (Fig. 5).

## 2.4 Field-calibration of the MOS

135 Field calibration of the MOS was done at the meltwater outlet at the Greenland Ice Sheet by simultaneous measurements of  
the same air mass by the MOS sensor system and a state-of-the-art CRDS in the configuration described above (section 2.1).  
For the calculation of the average ambient sensor resistance ( $R_{0*}$ ) using Eq. 1, the atmospheric background concentration of  
 $CH_4$  of the air (approximately 1.9 ppm) close to the ice sheet was used, in the absence of a controlled and humidified zero gas.  
Exact measurements of temperature and humidity of this air mass are not available, but T and RH are estimated to fall within  
140 the range of 1-4 °C and 90-100 % RH. The output value of the MOS under these conditions was then used to establish the  
resistance ratio ( $R_S/R_{0*}$ ) vs.  $CH_4$  concentration field calibration function for the MOS (Fig. 6).

## 2.5 Data processing

145 In order to compensate for potential effects of micro-turbulent mixing of subglacial air with atmospheric air observed with the  
CRDS (Christiansen and Jørgensen, 2018) and which occur at a faster frequency than the 10 s sampling interval of the MOS  
(see also section 3.3), the measured raw time series data from the MOS were smoothed using simple exponential smoothing  
according to Eq. (2):

$$150 \quad s_t = \alpha x_t + (1-\alpha)s_{t-1} \quad \text{for } t > 0 \quad (2)$$

where  $s_t$  is the smoothed  $CH_4$  concentration value (ppm),  $\alpha$  is the smoothing factor and  $s_{t-1}$  is the previous smoothed  $CH_4$   
concentration value (ppm). At time zero ( $t=0$ ), the  $s_t$  is equal to the first unsmoothed raw  $CH_4$  value of the MOS. The  
optimum value for  $\alpha$  was determined using Microsoft Excel solver, by minimizing the total average RMSE between the raw  
155 data from the MOS and the simultaneous concentration measurement of the CRDS. Results show an optimal value of 0.042,  
which was used for both the CRDS and MOS data series (Fig. 7).

### 3. Results and Discussion

#### 3.1 Laboratory calibration test of the MOS

160 Fig. 3 shows the relationship between the resistance ratio ( $R_S/R_0$ ) for the step test at three regulated humidity levels, where  $R_0$  is calculated for each humidity levels based on the average voltage output of the sensor in the time steps where it was exposed to the  $\text{CH}_4$ -free synthetic air. It is observed that a near identical response function can be obtained across the three different water vapor concentrations in the air, as long as the water concentration of the zero gas is the same as in the span gas. Based on existing knowledge of the expected air temperature variations at the in situ sampling point at the GrIS (Christiansen and  
165 Jørgensen, 2018), the humidity calibration was only carried out at a single temperature in this study. However, variations in the ambient air temperature is also expected to have a linear scaling effect for the type MOS system tested in this study (Bastviken et al., 2020; van den Bossche et al., 2017).

#### 3.2 Field-calibration of the metal oxide sensor

170 The measured  $R_S/R_0^*$  ratios per time step over the field calibration period were converted into absolute  $\text{CH}_4$  concentrations using the regression statistics of the applied power model (Fig. 6). A total of 37,140 data points are included in the regression model for converting the  $R_S/R_0^*$  ratios to  $\text{CH}_4$  concentrations. Inclusion of data points from the micro-turbulent periods produces a noisy visualization of the calibrations data at higher  $\text{CH}_4$  concentration levels (Fig. 6). However, this apparent noise is primarily a visual artefact that does not have significance for the underlying calibration statistics, which shows excellent  
175 statistical agreement between the independent and dependent variables ( $R^2 = 0.98$ ; p-value: 0.001). While the same regression model is used for both the laboratory calibration and field calibration, significant deviation in the model parameters are observed between the laboratory calibration as a group and the field calibration. The reason for this difference is unknown, but a possible explanation could be the potential difference in input heater voltage for the MOS sensor (i.e. pin 1 and 4 in Fig. 1), since variations in the input heater voltage have been reported to affect the  $\text{CH}_4$  concentration measurements (van den Bossche  
180 et al., 2017). In the laboratory test, the heater circuit of the MOS was supplied by the 5 volt regulated output from the CR1000 datalogger, whereas the heater circuit was supplied from the Arduino Uno's 5 volt regulated output. Future test should aim to investigate if the differences between the results from the laboratory and field calibration can be minimized by using the same type of datalogger and identical power supply (fx. Rechargeable lithium ion battery pack) both in the laboratory and in the field. Results from this type of test could reveal if field calibration for each individual MOS system is needed similar to the  
185 approach in Bastviken et al. (2020), or if batch calibrations of several identical MOS-system can be performed in the laboratory without the need for time-consuming field calibration.

### 3.3 Time-series plot of CH<sub>4</sub> concentration from reference CRDS and MOS

Due to the dynamic environment at the margin of the GrIS, the physical configuration of the sampling point will vary both  
190 over the melt-season as well as on an inter-annual basis. In our previous study, high frequency variations in CH<sub>4</sub> concentrations  
in the subglacial air were observed in a downward draping curve style where a high concentration plateau was interrupted by  
rapid decreases in CH<sub>4</sub> concentration (Christiansen and Jørgensen, 2018). This pattern was interpreted to be an effect of micro-  
turbulent and wind driven dilution of the sample gas in the ice cave by atmospheric air with a CH<sub>4</sub> concentration of  
approximately 1.9 ppm. In the current study, exponential smoothing of the raw values is used to compensate for the potential  
195 effects of physical disturbance of the sample gas caused by wind driven turbulent mixing of atmospheric background air at the  
subglacial sample point. Also, temporal smoothing can compensate for some of the sensor specific variation in response time  
improving the pairwise measurement comparability between the CRDS and the MOS. According to the manufacturer, the  
CRDS is specified to have a response time of less than 1 hz, while the response time of the MOS is expected to be slower.

200 The time series plot of the raw and exponentially smoothed CH<sub>4</sub> data from the CRDS (CRDS<sub>smooth</sub>) is shown together with the  
pairwise error between the raw data and the smoothed data (Fig. 7a). Similarly, the time series plot of the raw and exponentially  
smoothed CH<sub>4</sub> data from the MOS (MOS<sub>smooth</sub>) is shown together with the pairwise error between the raw data and the  
smoothed data (Fig. 7b). It is generally observed that over the first 4 days of the calibration test, very low differences are  
observed between the raw data CH<sub>4</sub> concentration and the smoothed CH<sub>4</sub> concentrations for both CRDS<sub>smooth</sub> and MOS<sub>smooth</sub>,  
205 with absolute errors below 5 ppm (Fig. 7a & 7b). At the end of the field calibrations, higher errors are observed following the  
larger spread in CH<sub>4</sub> concentration measurements of both the CRDS and the MOS. CRDS analyzers across different brands  
and manufacturers generally perform very consistently and have a highly linear measurement response across the effective  
concentration range without any tendencies for increasing analytical error with increasing gas concentrations (Brannon et al.,  
2016). Fluctuations in CH<sub>4</sub> concentrations in the subglacial air were also observed in (Christiansen and Jørgensen, 2018) using  
210 a CRDS from another manufacturer (G4301 GasScouter, Picarro Inc.). These variations were attributed to the dynamic and  
micro-turbulent environment in the subglacial cavity where the gas concentrations were measured and are likely produced by  
both air movement generated by the shear stress of the running meltwater as well as turbulent intrusion of atmospheric air  
generated by shifting winds speeds at the measurement location at the ice margin.

215 According to the field notes for the current study, a shift in overall wind regime took place during the 25<sup>th</sup> of June 2018, where  
the weather shifted from calm and sunny conditions to more windy conditions dominated by strong catabatic easterly winds  
coming off the GrIS. A best estimate of the overall time period where more windy conditions occurred during the field  
calibration period is indicated with grey background in Fig. 7. Unfortunately, no direct measurements of wind movement were  
made during the fieldwork period at the location. Measurements of air temperature at the sample inlet point in the subglacial  
220 cavity (Fig. 7c) show that an initial period with diurnal temperature variations of approximately 0.1 to 0.2 °C was followed by

a period with more fluctuating temperature variations of up to + 0.6 °C. The period with higher variability corresponds to the period where higher winds speeds predominate and the deviations between the raw and smoothed CH<sub>4</sub> are the greatest. The higher variability in air temperature measurements during the more windy weather is interpreted as being a product of more turbulent wind conditions right at the margin of the GrIS and opening to the subglacial cavity by which higher amounts of warmer atmospheric air with an approximate CH<sub>4</sub> concentration of approximately 1.9 ppm is introduced into the subglacial cavity. The introduction of these atmospheric air masses results in both short-term temperature increases as well as dilution of the subglacial CH<sub>4</sub> concentration in the cavity producing the more variable CH<sub>4</sub> concentration patterns observed in both the CRDS and MOS raw data. In the absence of direct measurements of wind speed and micro-turbulence at the margin of the ice, rapid variations in air temperature at the sample inlet point with an amplitude greater than the 0.2 °C appear as a reasonable indicator or proxy for micro-turbulent dilution and physical disturbance of the source signal, which can effectively be filtered out by the application of exponential smoothing.

The relative error between each MOS<sub>smooth</sub> and CRDS<sub>smooth</sub> measurement pair can be expressed as the percentage that the difference constitutes compared to the reference CRDS concentration (i.e  $\text{MOS}_{\text{smooth}} - \text{CRDS}_{\text{smooth}} / \text{CRDS}_{\text{smooth}} \times 100$ ). It is seen that the pairwise relative error between the MOS<sub>smooth</sub> and CRDS<sub>smooth</sub> shows similar non-systematic variations in both the calm weather and windy time period with relative errors typically below  $\pm 10\%$  (Fig. 7d). This result show both that the accuracy of the CH<sub>4</sub> concentration measured by the MOS are in close agreement with the reference CRDS and that the exponential smoothing effectively compensates for short term physical disturbances at the measurement point. The result also indicate that no systematic drift or over/underestimation is apparent when comparing the MOS<sub>smooth</sub> to the CRDS<sub>smooth</sub> over the 100 h field calibration period (Fig. 7b). When considering the magnitude of the absolute errors between the raw and smoothed CH<sub>4</sub> concentration for both the CRDS and the MOS, together with the temporal pattern in the development of the relative error, it shows that the high frequency concentration fluctuations measured with the MOS are most likely the product of physical disturbances at the measurement point (primary sampling error), and not by an analytical error introduced by the MOS itself (secondary sampling error).

As supplement to the pairwise error comparison, average time-series performance statistics for the difference between the MOS<sub>smooth</sub> and CRDS<sub>smooth</sub> time series can be calculated for both the full field calibration period, as well for the non-turbulent time period with limited observed physical disturbance at the sampling point (Table 1). RMSE the non-turbulent and full time series are approximately 1.3 to 1.7 ppm CH<sub>4</sub> respectively.

### 3.4 Post-correction and cross-interference evaluation

One of the main obstacles previously reported concerning the use of MOS's for monitoring of gases in ambient air is the possible effect of variations in air temperature and humidity in the sampling environment (Bastviken et al., 2020; Eugster et



al., 2019; Masson et al., 2015; Sohn et al., 2008). Different approaches exist to compensate for this potential measurement  
255 error and are related to post-correct for variations in temperature and humidity, based on either generic temperature and  
humidity dependency curves supplied by the sensor company (Eugster and Kling, 2012). This is achieved by performing sensor  
calibrations under controlled levels of temperature and humidity in the laboratory (van den Bossche et al., 2017) or by field  
calibration (Bastviken et al., 2020).

260 Measurements from the air-filled cavity under the GrIS document a very stable sampling environment with a relative humidity  
throughout the sampling period of close to 100 % RH (data not shown) and only minor air temperature variations between  
approximately 0.05 °C during the night and 0.25 °C during mid-day (Fig. 7d). Because of these stable and well-buffered  
environmental conditions, no post-corrections due to variations in temperature and relative humidity are evaluated to be  
necessary for this particular sampling environment.

265 Observed variations in maximum air temperature in the subglacial cavity correspond to field observation of the time of the day  
when maximum meltwater discharge occurs. We assume that the observed temperature pattern reflects the impact of thermal  
heat diffusion from this running meltwater to the air immediately above, but would need direct measurements of the daily  
variations in meltwater temperature to verify this assumption.

270 The emitted CH<sub>4</sub> may originate from both thermogenic and/or biogenic sources below the GrIS. If the primary source of CH<sub>4</sub>  
is thermogenic, the emission may also be accompanied by more complex hydrocarbons, including ethane (C<sub>2</sub>H<sub>6</sub>), while this  
will not be the case if the source is biogenic (Hopkins et al., 2016). Since the MOS used in the study is non-selective to CH<sub>4</sub>  
due to its basic principle of operation (Eugster and Kling, 2012; Wang et al., 2010), the presence of other hydrocarbons such  
275 as ethanol (C<sub>2</sub>H<sub>6</sub>O), isobutene (C<sub>4</sub>H<sub>10</sub>) and potentially also other low molecular weight alkanes could potentially cause cross-  
interference with the CH<sub>4</sub> measurement. It follows, that if the source of the CH<sub>4</sub> that is emitted for the subglacial domain  
originates in thermogenic natural gas reservoirs under the GrIS, the other non CH<sub>4</sub>-hydrocarbons could potentially affect the  
measurements performed by the MOS, while being undetected by the CRDS. However, since the magnitudes and temporal  
patterns in CH<sub>4</sub> concentrations are similar in both the CRDS and MOS it is assumed that the gases emitted from the subglacial  
280 domain are primarily CH<sub>4</sub> and CO<sub>2</sub> with very limited potential for cross-interference from other hydrocarbon gases. Also,  
isotopic analysis of the emitted CH<sub>4</sub> and CO<sub>2</sub> in (Lamarche-Gagnon et al., 2018) as well as unpublished data from this study,  
have shown that the emitted CH<sub>4</sub> is dominantly of microbial origin and has isotopic similarity to CH<sub>4</sub> produced by anaerobic  
decomposition of organic carbon in wetlands. We therefore assume that there is no need for any post correction of the CH<sub>4</sub>  
concentrations measured by the MOS in this type of environment due to lack of cross-interference from other hydrocarbon  
285 gases.

### 3.5 Autonomous CH<sub>4</sub> monitoring using MOS system

The combined time period in which CH<sub>4</sub> concentrations were measured can be divided into three separate periods depending on the analytical devices used, namely period 1 corresponding to the field calibration period where both the CRDS and MOS were in operation (approximately 100 h), period 2 where only the CRDS was in operation (approximately 24 h) and period 3 where only the MOS was in operation (Fig. 8). Continuous CH<sub>4</sub> data from period 3 exist for the period 27<sup>th</sup> June to 15<sup>th</sup> July 2018. When comparing the combined CH<sub>4</sub> concentration curves from all three periods it is observed that the CRDS<sub>smooth</sub> and MOS<sub>smooth</sub> follow each other as described above (Fig. 7). CRDS<sub>smooth</sub> data for period 2 fills the data gap between the MOS measurement of period 1 and 3, where the start concentration data of the MOS<sub>smooth</sub> are similar to the concentration level where the CRDS<sub>smooth</sub> measurements end. Due to the nature of the study design and difficult access to the remote field site at the GrIS, the accuracy and precision of the MOS<sub>smooth</sub> cannot be evaluated for the period 3 where only the MOS system was operating. However, the pattern in which subglacial CH<sub>4</sub> concentrations varied and the estimated minimum and maximum values appear similar to the values of the calibration period. When comparing the complete time period of this study to e.g. Eugster and Kling (2012), no significant sensor drift is expected over the monitoring time period. Additional and extended field work at the GrIS with repeated calibration at the end of the field deployment period is needed to quantify the potential sensor drift, as well as stability range over longer time scales (Eugster et al., 2019). Nonetheless, the observed performance of the MOS during the calibration period with ppm-level accuracies and subsequent trouble-free operation running as an autonomous unit shows that this type of low-cost and low-power CH<sub>4</sub> sensing system holds a great potential for the further development and refinement of a greater sensor network at representative meltwater outlets at the Greenland ice Sheet.

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The next steps and lessons learned from this study deals with the further development of the low-power system for actual CH<sub>4</sub> emission measurements, which involves measurements of air volume and meltwater discharge as well as continuous measurements of the dissolved CH<sub>4</sub> in the meltwater, similar to (Lamarche-Gagnon et al., 2018). Also, optimizing the positioning of gas sensing equipment at the measurement point should be done to reduce the potential physical disturbances due to micro-turbulence and intrusion of atmospheric air in the subglacial cavity. Furthermore, an improved adjustment scheme should be developed to account for the dynamic melt back of the ice margin over the melt season, which requires either manual or automated sample point relocation to keep the sampling point at an optimal physical location. Finally, more work is needed to test what modification to the systems are needed to establish a universal calibration curve in the laboratory, so that the need for field calibration with the reference CRDS can be eliminated (see also Bastviken et al., 2020; Eugster et al., 2019).

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#### 4. Conclusions

Recent discoveries at the Greenland Ice Sheet (GrIS) have revealed a so far overlooked cryospheric source of CH<sub>4</sub> from the subglacial domain under the ice to the atmosphere. Development of low-power CH<sub>4</sub> monitoring systems based on low cost metal oxide sensors (MOS) could enable the development of a sensor network at representative meltwater outlets at the GrIS which could significantly improve the fundamental understanding of the phenomena's climatic importance. In the current study, the performance of a metal oxide sensor sensitive to CH<sub>4</sub> was tested in an air-filled cavity at the edge of the Greenland Ice Sheet over an initial field calibration period of approximately 100 h. Simultaneous measurements by both cavity ring-down spectroscopy (CRDS) and a low-cost metal oxide sensor (MOS) using a common inlet show good agreement between the MOS and the CRDS over time under the stable environmental conditions under the ice. Exponential smoothing of the raw data from both the CRDS and MOS effectively remove high frequency concentration variations induced by physical disturbance of the air in the subglacial cavity under more turbulent wind conditions at the margin of the ice sheet. Based on concentration values of the smoothed CRDS and MOS data, the pairwise measurement errors were generally below  $\pm 5$  ppm CH<sub>4</sub> between the MOS and the CRDS reference value. Pairwise relative errors were generally below  $\pm 10$  % between the MOS and the CRDS reference value. The RMSE for the entire field calibration period was  $\pm 1.69$  ppm CH<sub>4</sub>. If only data for the non-turbulent time period was evaluated, the RMSE was reduced to  $\pm 1.35$  ppm CH<sub>4</sub>. Due to the ice buffered sampling environment in the air-filled cavity under the Greenland Ice Sheet, no post-corrections for variations in air temperature, humidity or cross interference from other hydrocarbon gases were needed for the MOS measurements. Combined with measurement of airflow and meltwater discharge, the measurement of CH<sub>4</sub> concentrations can be used for determination of the mass flux of CH<sub>4</sub> to the atmosphere. The study demonstrates a clear potential for expanded monitoring of spatial and temporal variation in CH<sub>4</sub> emissions from the subglacial domain of the Greenland Ice Sheet using low-cost and low-power MOS.

#### 5. Author contribution

CJJ and JRC designed and carried out the field experiments. CJJ, JMO and KFU planned and carried out the laboratory calibrations. CJJ and JRC prepared the manuscript with contributions from all co-authors.

#### 6. Acknowledgements

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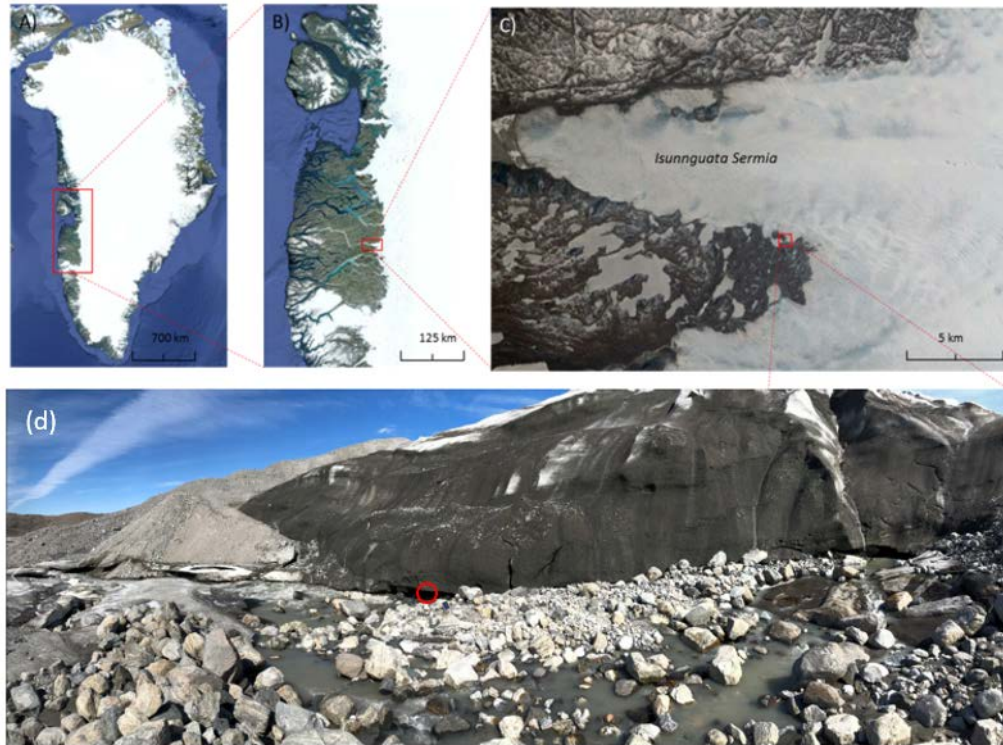
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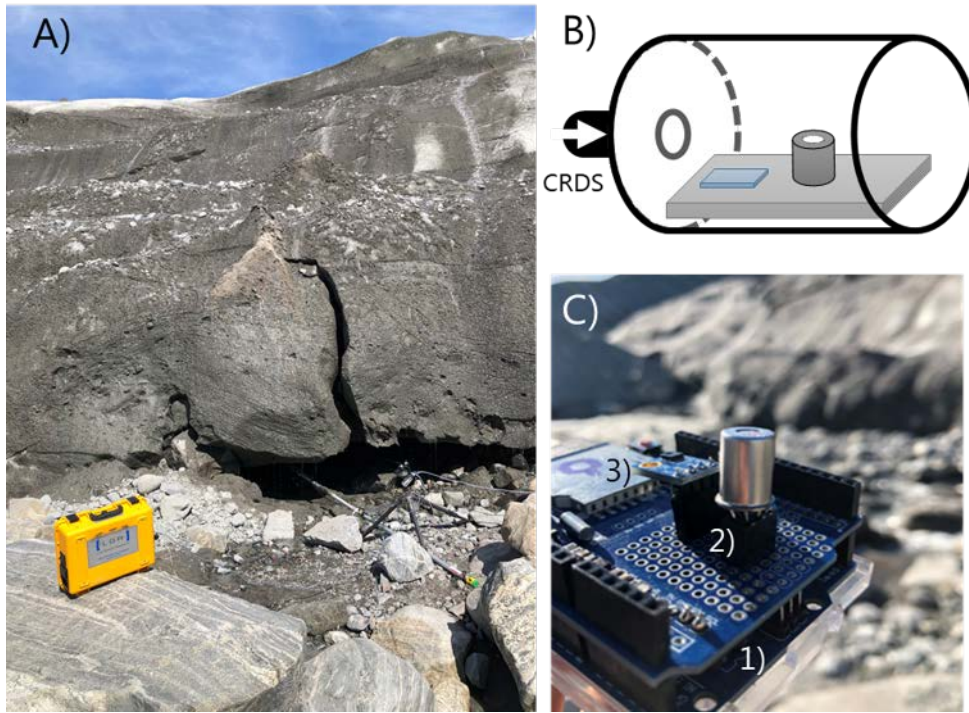
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## Figure captions

- 415 Figure 1: Overview of the sampling location at Isunnguata Sermia Glacier at the western margin of the Greenland Ice sheet during June 2018. (a) Location of sampling region the island of Greenland, (b) regional location of the outlet glacier, (c) location of the meltwater outlet at Isunnguata Sermia and (d) local sample location with investigated subglacial cavity marked with red circle. Source of (a), (b) and (c): Google Earth, [earth.google.com/web/](http://earth.google.com/web/).
- 420 Figure 2: (a) Close-up of air-filled cavity below the Greenland Ice Sheet next to the lateral meltwater outlet. The aluminum pole extends approximately 5 meters into the cavity and holds the common inlet tube and the temperature and humidity smart sensor. (b) Conceptual diagram of the MOS system which was connected in series to the outlet port of the CRDS analyzer (c) Close-up of the board mounted MOS and temperature/humidity micro sensor. The MOS system consisted of 1) a microcontroller, 2) Datalogger shield holding metal oxide CH<sub>4</sub> sensor and 3) an additional temperature/relative humidity micro sensor.
- 425 Figure 3: Simplified schematic of the metal oxide sensor (MOS) system consisting of a TGS2611-E00 with pin 3 and 4 connected to the 5-volt output of the Arduino Uno, pin 1 connected to ground and pin 2 connected to the analogue input of the Arduino Uno and a 10kOhm load resistor, which also connects to ground.
- 430 Figure 4: Outlet voltages of the MOS during laboratory step calibration at stabilized levels of relative humidity ( $37 \pm 2 \%$ ,  $55 \pm 3 \%$  and  $76 \pm 3 \%$ ) in both the zero and span gas at alternating concentration of CH<sub>4</sub> in the calibration gas between 10 and 100 ppm CH<sub>4</sub>. Each time step lasted 10 min and sequences with grey shadings show time periods where the sensor was exposed to CH<sub>4</sub> free zero gas.
- 435 Figure 5: Resistance ratio of MOS as three levels of relative humidity at CH<sub>4</sub> concentration levels between 10 to 100 ppm CH<sub>4</sub> in humidified synthetic air.
- 440 Figure 6: Regression plot of calculated MOS resistance ratio  $R_s/R_{0^*}$  vs. the reference in situ CH<sub>4</sub> concentrations from the CRDS (n = 37,140).
- 445 Figure 7: (a) Grey dots show raw CH<sub>4</sub> concentration from the cavity ringdown spectrometer (CRDS). Black line show CRDS CH<sub>4</sub> concentration values following exponential smoothing. Black bars show absolute error between raw and smoothed values. (b) Grey dots show calculated raw CH<sub>4</sub> concentration from metal oxide sensor (MOS). Black line show MOS CH<sub>4</sub> concentration values following exponential smoothing. Black bars show absolute error between raw and smoothed values. (c) Black dots show temperature of air in subglacial cavity. (d) Black bars show the relative error in percentage between the  $MOS_{smooth}$  and  $CRDS_{smooth}$  divided by the  $CRDS_{smooth}$  concentration. Grey background shading indicates period with higher observed turbulence at the margin of the GrIS. Temporal resolution is 10 seconds.
- 450 Figure 8: Smoothed time series measurements of CH<sub>4</sub> at the Greenland Ice Sheet using both the cavity ring-down spectroscopy (CRDS) reference monitor and the metal oxide sensor (MOS). Temporal resolution is 10 seconds.
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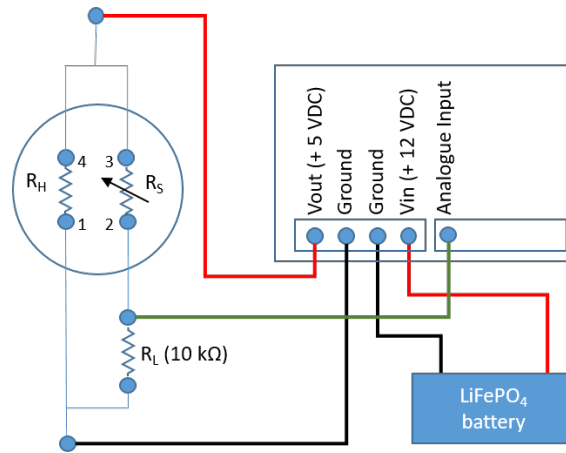


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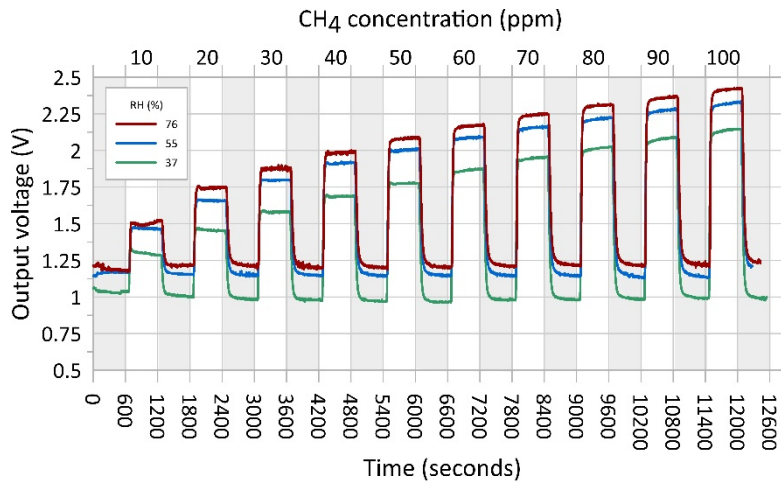


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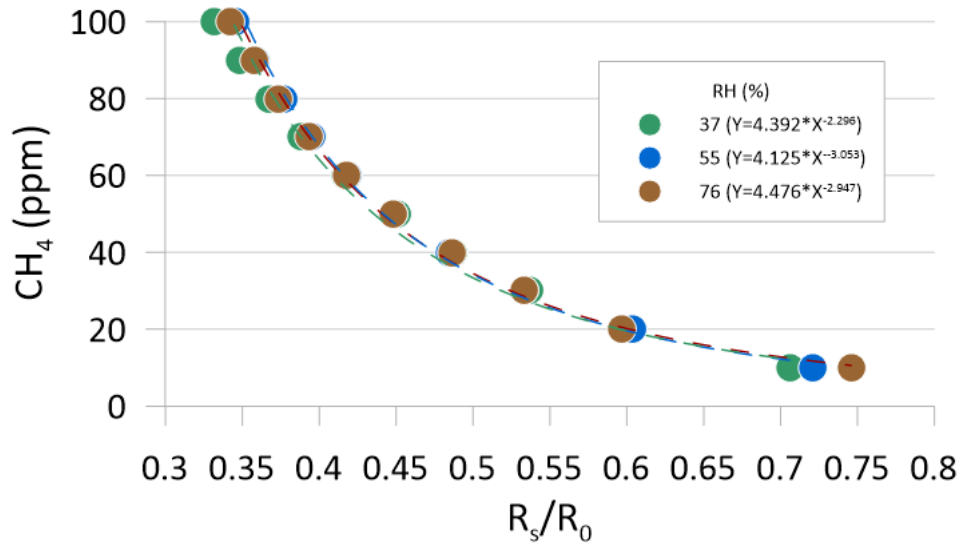




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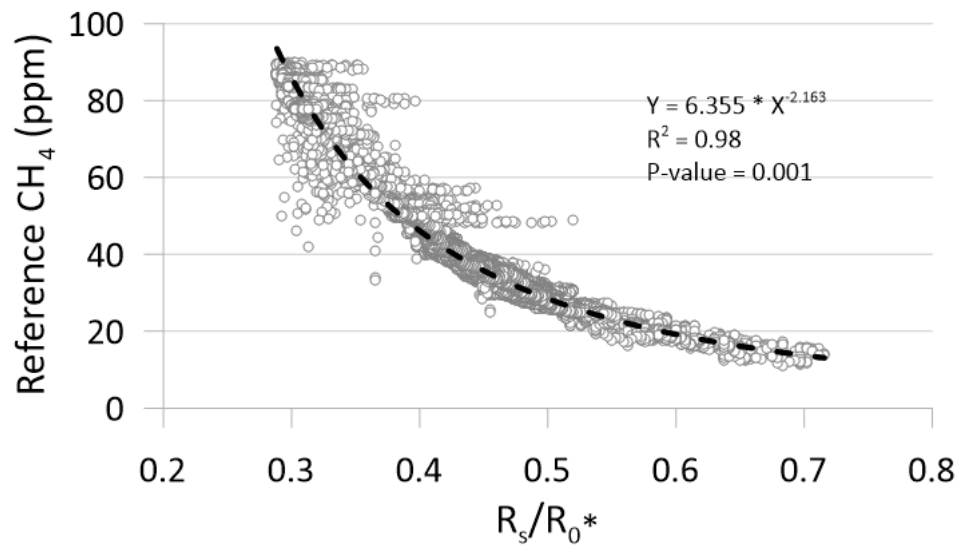


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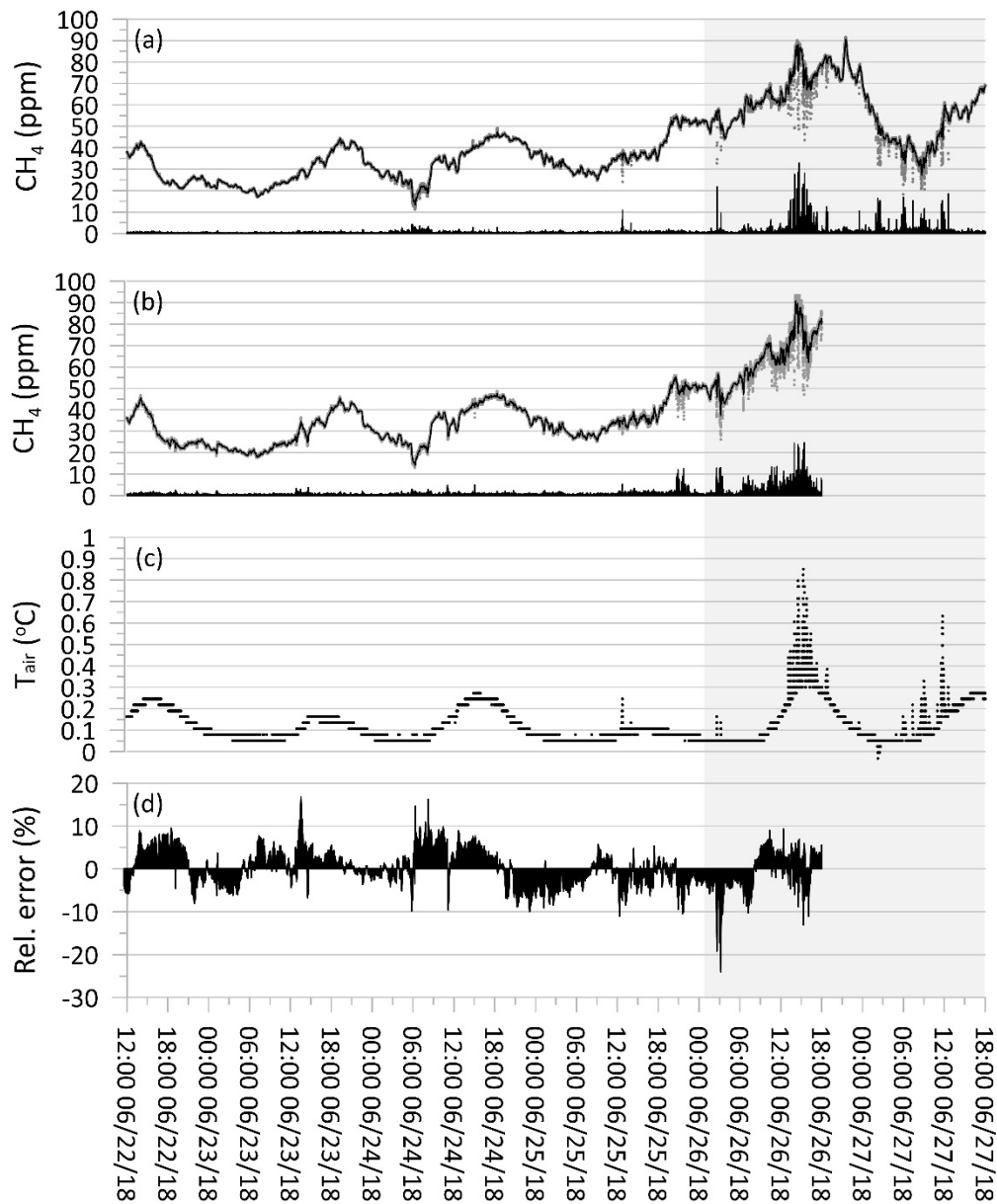
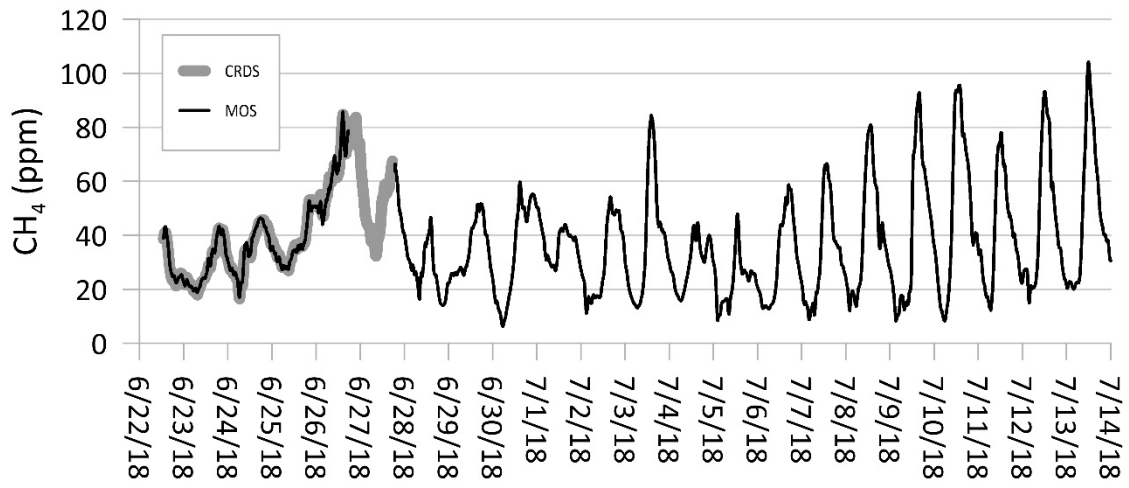


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**Tables**

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Table 1. Statistics for the calculated differences between the smoothed MOS and smoothed CDRS data series in both the non-turbulent time period and full field calibration period. The unit for error and difference values is ppm.

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Statistics: $MOS_{smooth} - CDRS_{smooth}$	Non-Turbulent	Full series
Mean bias error (MBE)	0.09	-0.05
Mean absolute error (MAE)	1.08	1.29
Root mean square error (RMSE)	1.35	1.69
Maximum negative difference	-3.96	-11.83
Maximum positive difference	5.04	5.91
Observations	28501	37140