The manuscript "Nitrate radical generation via continuous generation of dinitrogen pentoxide in a laminar flow reactor coupled to an oxidation flow reactor" by Lambe et al. presents a NO₃ oxidation flow reactors system for studying the reaction of organic compound with NO₃ reactions and secondary organic aerosol from NO₃ chemistry. The authors give a very comprehensive characterization of the system by lab study as well as the model simulation. The paper is well written and the results are presented in a clear way. The analysis are sound and the authors proposed ideas will be of interest and helpful to the community. I recommend this paper for publication subjects to some minor comments.

General comments.

1. Section 2.2.1, the section cited some references about the wall loss of NO₃ and N₂O₅ in Teflon/Pyrex tube, and the NO₃ and N₂O₅ wall loss in LFR and OFR is extrapolating or interpolation based on the reported results, which weak the results as the wall loss of NO₃ and N₂O₅ in the system is an important source of uncertainty. The lab quantification of the wall loss in LFR and OFR in the future works can further improve the value of this study. Additionally, page 5, line 9-10, the fixed condition of OFR is ambiguous, please clear it, at least add the simulation results in SI. Page 5, line 11-13, the Extrapolating results is confuse, the reference said 0.04 and 0.009 s⁻¹ corresponding to ID (4 and 7 cm), what is corresponding parameter of the $k_{wall,LFR}$ of 0.07 and 0.03 s⁻¹ mentioned here?

Specific comments.

- 2. Page 2, line 29-34, this introduction of the LFR is confusing. The authors can use a schematic figure to show more details about the OFR-iN₂O₅ (rather than Figure 1 from references), which would increase the paper's readability.
- 3. Page 4, line 8. Romanini et al. (1997) is not the right reference of the IBBCEAS principle, I suggest the author replace it by e.g., Fiedler et al., 2003.(Fiedler, S. E., Hese, A., and Ruth, A. A.: Incoherent broad-band cavity-enhanced absorption spectroscopy, Chem Phys Lett, 371, 284-294, 2003.)
- 4. In Eq .2 the k_{NO2} or k_{NO3} should revised to $k_{NO2+RO2}$ or $k_{NO3+RO2}$, the similar change also applied in Eq. 3 and Eq. 4