

1 **Development of an Automatic Linear Calibration Method for High**
2 **Resolution Single Particle Mass Spectrometry: Improved Chemical**
3 **Species Identification for Atmospheric Aerosols**

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32

33 **Abstract**

34

35 The mass resolution of laser desorption ionization (LDI) single particle aerosol mass
36 spectrometry (SPAMS) is usually low (~ 500), which has been greatly improved by recent
37 development of delayed ion extraction technique. However, due to large fluctuations
38 among LDI processes during each laser shot, accurate calibration of mass-to-charge ratio
39 for high resolution SPAMS spectra is challenging. Here we developed an automatic linear
40 calibration method to improve the accuracy of mass-to-charge (m/z) measurement for
41 single atmospheric aerosol particles. Laboratory generated sea spray aerosol and
42 atmospheric ambient aerosol were tested. After the calibration, the fluctuation ranges of
43 the reference ions (e.g. Pb^+ and SO_4^+) m/z reaches ± 0.018 for sea spray aerosol and \pm
44 0.024 for ambient aerosol in average mass spectra. With such m/z accuracy, the HR-
45 SPAMS spectra of sea spray aerosol can easily identify elemental compositions of organic
46 peaks, such as C_x , C_xH_y and $\text{C}_x\text{H}_y\text{O}_z$. While the chemical compositions of ambient aerosols
47 are more complicated, C_xH_y , $\text{C}_x\text{H}_y\text{O}_z$ and CNO peaks can also be identified based on their
48 accurate mass. With the improved resolution, the time series of peaks with small m/z
49 differences can be separated and measured. In addition, it is also found that applying high
50 resolution data with enhanced mass calibration can significantly affect particle
51 classification (identification) using the ART-2a algorithm, which classify particles based on
52 similarities among single particle mass spectra.

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54

55 **1. Introduction**

56

57 Atmospheric aerosols can significantly impact radiative forcing, cloud formation and
58 human health(Ackerman et al., 2004; Zhang and Kin-Fai, 2012). They originate from
59 various sources and undergo many atmospheric aging processes, resulting in an extremely
60 complicated mixture of particles with a large range of sizes and chemical compositions.
61 This mixture is usually referred as “mixing state”. Measurement of aerosol mixing state
62 requires single particle characterization techniques. Utilizing laser ablation/ionization of
63 single aerosol particle, Single Particle Aerosol Mass Spectrometer (SPAMS) has been
64 widely used to measure chemical compositions, sizes and refractory index of aerosols in
65 real-time(Moffet and Prather, 2009; Murphy, 2010; Sullivan and Prather, 2005). Based on
66 this technique, ART-2a and other algorithms had been developed to classify the ambient
67 particles based on their mass spectra and identify their sources (Reinard et al., 2007;
68 Zelenyuk and Imre, 2009).

69

70

71 However, SPAMS with laser desorption/ ionization (LDI) method has several serious
72 limitations (Manuel et al., 2006; Wenzel et al., 2003). A major issue is that the mass
73 resolution of the SPAMS is relatively low (~500) and the accuracy of m/z (mass to charge
74 ratio) is usually at integer level, resulting in uncertainties about the identification of
75 chemical species (Nash et al., 2006; Pratt and Prather, 2012; Qin et al., 2006). Due to the
76 low mass resolution, many organic and inorganic peaks cannot be separated, such as K^+
77 $/C_3H_3^+$ from the m/z peak of 39, $Al^+ C_2H_3^+$ from the m/z peak of 27 and $CN^- /C_2H_2^-$ from
78 the m/z peak of -26 (Li et al., 2018). To better identify these particulate chemical species,
79 A higher mass resolution version of the SPAMS with better m/z accuracy is needed.

80

81

82 Recently, Li et al., significantly increased SPAMS’s mass resolution to ~2000 by applying
83 delayed ion extraction technique, which combined a standard rectangular extraction pulse
84 with an exponential pulse (Li et al., 2018). This new SPAMS is called high resolution (HR)-
85 SPAMS. Unfortunately, in spite of resolution enhancement with this new technique, ion

86 peak position was still very sensitive to initial ion coordinate and speed. Chudinov et al.,
87 has demonstrated that the ion peak shifts of $208[\text{Pb}^+]$ and $147[\text{Na}(\text{NO}_3)_2^-]$ could be varied
88 in the range of ± 10 ns and the ion start position could be varied in the range of $\pm 150 \mu\text{m}$.
89 As a result, substantial peak jittering is observed when switching between mass spectra of
90 each individual particle. This peak jittering leads to a fact that isotopic pattern identification
91 becomes more difficult by averaged mass spectrum (Chudinov et al., 2019). Furthermore,
92 the peak jittering is different in each single particle mass spectrum. In other words, the
93 calibration parameter for each mass spectrum should be significantly different and
94 calibration is required for each particle. Therefore, in order to get accurate m/z , Chudinov
95 et al. used several peaks with known m/z to calibrate every SPAMS spectrum for $\text{Pb}(\text{NO}_3)_2$
96 and NaI particles produced from an atomizer.

97

98

99 In this study, we report a calibration method for single particle high resolution mass spectra
100 data. Based on the assumption that the sea spray aerosol has relatively simple chemical
101 composition while the ambient aerosol has more complex chemical composition, the
102 performance of the calibration method had been evaluated in detail for these two aerosol
103 systems with different complexity. In addition, the impact of using high resolution SPAMS
104 data on particle classification by ART-2a algorithm was assessed. An open source code
105 specific for HR-SPAMS was made and we proposed the principle of this calibration method
106 can be applied into some similar instruments, such as single particle mode Aerosol Mass
107 Spectrometer (AMS).

108

109

110 However, atmospheric particles are extremely complicated with a wide range of chemical
111 compositions and sizes (Zhang et al., 2013), which brings much greater challenge to
112 properly calibrating each SPAMS mass spectra and obtaining accurate m/z measurement.
113 We need to develop a new MS calibration method for atmospheric aerosols and evaluate
114 its performance comprehensively.

115

116

117 **2. Experimental Section**

118 **2.1 High Resolution Single Particle Aerosol Mass Spectrometer (HR-SPAMS)**

119 The detailed description of HR-SPAMS (Hexin Analytical Instrument Co., Ltd., China) can
120 be found elsewhere(Li et al., 2018). Briefly, a HR-SPAMS consists of an aerodynamic lens
121 as its particle inlet, two laser beams system for particle sizing, a UV laser for LDI and a
122 bipolar time-of-flight mass analyzer for the detection of positive and negative ions. Positive
123 and negative ions are detected by two z-shape bipolar TOF reflectron mass analyzers. The
124 size detection range of HR-SPAMS is 200-2000 nm. As introduced before, this HR-SPAMS
125 used delayed ion extraction technique to enhance its mass resolution.

126

127 **2.2 Laboratory generated sea spray aerosol**

128 Sea spray aerosol was produced by water jet method. In a sea spray aerosol production
129 tank, a seawater jet was hitting seawater surface and producing bubbles, which would rise
130 to the surface and burst. Bubble bursting process produces sea spray aerosols. Seawater
131 was collected at Fengxian, Shanghai (30°92'N and 121°47'E) on March 30st (Fig.S1).

132

133 **2.3 Ambient aerosol sampling**

134 Ambient aerosol sampling was conducted at Fudan university, Shanghai (31°20'N and
135 121°30'E) on May 29th 2019 (Fig.S1). The ambient particles were dried by a diffusional
136 dryer before being sampled by the HR-SPAMS.

137

138 **3. Development of Calibration Methods**

139 **3.1 Automatic linear calibration method**

140 To improve the accuracy of m/z for HR-SPAMS spectra, an automatic linear calibration
141 method has been developed. Noticeably, due to the technical limitation of data acquisition,
142 the whole HR-SPAMS spectrum is not continuous but divided by a large number of m/z
143 bins, which are described in Fig.S2(a partial enlarged detail in the single particle mass
144 spectra) and can be viewed as the probability density histogram of the m/z . Here we denote
145 “ m/z bin value” as the median m/z value of each bin.

146

147 The linear calibration method is described as the following steps:

148

149 Step 0: The SPAMS data was coarsely-calibrated by the traditional method, which usually
150 selected a few particles with distinct ion patterns, i.e. the molecular composition of some
151 distinct peaks in the mass spectra can be easily identified. Then, the time of flight of these
152 peaks and the true m/z of the corresponding ions were used to calculate a set of calibration
153 parameters for positive and negative spectra. The parameters were finally applied to the
154 whole mass spectra dataset, and coarsely-calibrated was completed.

155

156 Step 1: a pool of ion peaks in the single particle mass spectra were selected as the potential
157 m/z calibration reference ions. The selection criteria are (1) these peaks should be present
158 in most of the spectra; (2) the identification of these ion peaks should not be significantly
159 affected by other adjacent peaks. For example, 27[Al]⁺ was not selected, as its adjacent
160 peak 27[C₂H₃] may affect the peak shape and identification of 27[Al] .

161

162 According to the previous research, possible peak assignments for the m/z of reference ions
163 for sea spray aerosol and ambient aerosol were listed on Table 1(Bertram et al., 2018;
164 Collins et al., 2014; Tsunogai et al., 1972; Wang et al., 2016; Wang et al., 2019). For sea
165 spray aerosol, according to several studies (Bertram et al., 2018; Collins et al., 2014;
166 Tsunogai et al., 1972), the reference ions with m/z 23 24 39 -35 -37 were 23[Na]⁺ 24[Mg]⁺
167 39[K]⁺ -35[Cl]⁻ and -37[Cl]⁻ respectively. And Collins et al. shows that the reference ions
168 with m/z 81, 83, -26, -42, -58, -129, and -131 were 81[Na₂Cl] , 83[Na₂Cl] , -26[CN] , -
169 42[CNO]⁻, -58[NaCl]⁻, -129[MgCl₃]⁻, and -131[MgCl₃]⁻, respectively (Collins et al., 2014).
170 Due to the fact that Na, Mg and K were abundant in sea spray aerosol, the reference ions
171 with m/z 113 and 115 should be 113[K₂Cl]⁺ and 115[K₂Cl]⁺. Thus, in this study, we select
172 23[Na]⁺, 24[Mg]⁺, 39[K]⁺, 81[Na₂Cl]⁺, 83[Na₂Cl]⁺, 113[K₂Cl]⁺, 115[K₂Cl]⁺, -35[Cl]⁻, -
173 37[Cl]⁻, -26[CN]⁻, -42[CNO]⁻, -129[MgCl₃]⁻, -131[MgCl₃]⁻, -58[NaCl]⁻ as the potential
174 reference ions for sea spray aerosols.

175

176 For the ambient aerosol, according to the previous ambient SPAMS measurements(Wang
177 et al., 2016; Wang et al., 2019), the reference ions with m/z 12, 23, 36, 39, 56, 207, 208,
178 and 209 were assigned to 12[C]⁺, 23[Na]⁺, 36[C₃]⁺, 39[K]⁺, 56[Fe]⁺, 207[Pb]⁺, 208[Pb]⁺

179 and $209[\text{Pb}]^+$, the reference ions with m/z -26, -35, -46, -62, -96, and -97 were assigned to
180 $-26[\text{CN}]^-$, $-35[\text{Cl}]^-$, $-46[\text{CNO}]^-$, $-62[\text{NO}_2]^-$, $-96[\text{SO}_4]^-$ and $-97[\text{HSO}_4]^-$ respectively. So in
181 this study, we select the $12[\text{C}]^+$, $23[\text{Na}]^+$, $39[\text{K}]^+$, $36[\text{C}_3]^+$, $56[\text{Fe}]^+$, $208[\text{Pb}]^+$, $206[\text{Pb}]^+$,
182 $207[\text{Pb}]^+$, $-62[\text{NO}_3]^-$, $-26[\text{CN}]^-$, $-35[\text{Cl}]^-$, $-96[\text{SO}_4]^-$, $-46[\text{NO}_2]^-$, $-97[\text{HSO}_4]^-$ as the
183 potential reference ions for ambient aerosols.

184

185 Step 2: a set of reference ions was chosen from the potential reference ion pool for each
186 spectrum. The selection was based on the absolute ion intensity of the reference ions in this
187 spectrum. They must be greater than a threshold, e.g. we set 15 a.u. for ambient aerosol
188 and 8 a.u. for sea spray aerosol, respectively. A particle was discarded from the spectra
189 database if it did not have enough reference ions (the minimum number of reference ions
190 was set to be 5) in either positive or negative mass spectrum.

191

192 Step 3: the reference ions were used to calibrate m/z for mass spectra of each particle. As
193 introduced before, a HR-SPAMS spectrum consists of a number of bins. The measured m/z
194 bin values of the reference ions mentioned in the Step0 were calibrated based on their
195 theoretic (or true) m/z bin values. A linear regression between the two set of variables
196 (measured vs. theoretic m/z bin values) was conducted, and two calibration parameters (a
197 slope and an intersect) can be obtained. Then we used these parameters to make the
198 calibration for every bin value in this mass spectra. Finally, the m/z of the whole spectrum
199 had been corrected. Thus, we assigned a m/z bin value to each corrected m/z based on
200 proximity principle. Finally, mass spectra with well calibrated bin value can be obtained
201 for each single particle.

202

203 A GUI program for this automatic linear calibration method had been developed for the
204 sake of easy use (Fig. 1). The MATLAB codes for this GUI and the automatic linear
205 calibration method are open access and available at [https://github.com/zhuxiaoqiang-](https://github.com/zhuxiaoqiang-fdu/zhuxiaoqiang-fdu)
206 [fdu/zhuxiaoqiang-fdu](https://github.com/zhuxiaoqiang-fdu/zhuxiaoqiang-fdu).

207

208 **3.2 Evaluation of the calibration method**

209 In this study, a total of 5,130 sea spray aerosol particles and 5,007 ambient aerosol particles

210 were analyzed. And 4,624 sea spray particles and 1,409 ambient particles were successfully
211 calibrated. As some fraction of particles had been filtered because their mass spectrum did
212 not have 5 or more reference peaks to conduct the calibrations. To mitigate this problem,
213 we proposed some adjustments in the next section. Figure 2 shows that the calibration
214 curves for a random selected sea spray aerosol particle and ambient aerosol particle. The
215 adj-R² coefficients of both calibration curves are equal to ~1, demonstrating that this
216 calibration method is effective and accurate. All the slopes and intercepts of the linear
217 calibration can be found in the Fig.S3 and Fig.S4. In addition, Figure 3a and 3b report a
218 comparison of m/z distributions of reference ions between before and after automatic linear
219 calibration. The results show that the fluctuations of the reference ions m/z were
220 significantly reduced after automatic linear calibration. The average m/z deviation of the
221 reference ions was reduced from ~0.04 to ~0.001 for sea spray aerosol, and from ~0.035
222 to ~0.006 for ambient aerosol, respectively.

223

224 **3.3 Automatic linear calibration method with a larger reference ion pool**

225 It is important to note that a large number of ambient particles were filtered because their
226 spectra did not have 5 or more reference peaks to conduct calibrations. Especially, only
227 ~29.0% of total ambient particles had sufficient number of reference ions in their positive
228 spectra. To solve this problem, extra reference ions, including 67[VO]⁺, 67[C₅H₇]⁺,
229 89[C₇H₅]⁺, 89[Na₂BO₂]⁺, 102[C₈H₆]⁺, 102[CaNO₃]⁺, were added into the original positive
230 reference ion pool. Obviously, these ions share the same integer m/z value with other ions.
231 We needed to identify them using additional information other than their integer m/z values.

232

233 The specific reference ions was determined by their coarsely-calibrated m/z. Table S2
234 shows that the m/z deviation ranges of the reference positive ambient ions in coarsely-
235 calibrated spectra before automatic linear calibration were around 0.011~0.048, while the
236 m/z differences between 67[VO]⁺ and 67[C₅H₇]⁺, 89[C₇H₅]⁺ and 89[Na₂BO₂]⁺, 102[C₈H₆]⁺
237 and 102[CaNO₃]⁺ were 0.1213, 0.0622, and 0.083, respectively, which were larger than the
238 m/z deviations of these reference ions in coarsely-calibrated spectra. Therefore, the
239 coarsely-calibrated spectra can be used to determine these specific reference ions. With
240 these additional potential reference ions, a total of 2490 ambient particles were calibrated,

241 much more than the previous analysis (1,409 ambient particles). The deviations from
242 theoretical m/z for applying this expanded ion pool are summarized in Table S3. The
243 average m/z deviation of the reference ions is ~ 0.0068 .

244

245 **4. Application to atmospheric aerosols measurement**

246 **4.1 HR-SPAMS measurement of sea spray aerosol**

247 SPAMS data usually contains a large number of individual mass spectra. It is impossible
248 to manually analyze every spectrum from a large dataset. Averaging a number of mass
249 spectra is often preferred. However, to obtain averaged high-resolution spectrum, each
250 spectrum must be well calibrated. Therefore, it would be very interesting to see what new
251 information can be obtained from HR-SPAMS measurement of aerosols with the automatic
252 calibration method. Figure 4 reports the average positive and negative mass spectra for the
253 laboratory generated sea spray aerosols and the error limits mean the concluded accepted
254 error range. Similar to the low-resolution sea spray aerosol mass spectra, they contain
255 major peaks of Na^+ , Mg^+ , K^+ , Na_2Cl^+ , CN^- , Cl^- , CNO^- , NaCl^- , NaCl_2^- and MgCl_3^- , as well
256 as many smaller peaks, such as Ca^+ , $\text{SiO}_2^-/\text{SiO}_3^-$, and KCl_2^- . With the improved m/z
257 measurement, many peaks, which cannot be determined by integer resolution mass spectra,
258 now can be clearly identified (Table 2). For example, the ion with m/z at 27.0267 is C_2H_3^+
259 rather than Al^+ . The ion with m/z at 76.9336 is CaCl^+ rather than C_6H_5^+ . And some sulfur
260 containing organic ions, such as CS^+ , can be determined. Surprisingly, we can identify the
261 presence of HCO_2^- and CaCO_3^- , demonstrating that carbon hydrates are contained in sea
262 spray aerosols.

263

264 **4.2 HR-SPAMS measurement of atmospheric aerosol**

265 Laboratory generated sea spray aerosol can be viewed as a relatively simple aerosol system,
266 while the chemical compositions of ambient aerosols are much more complicated. Figure
267 5 shows the averaged HR-SPAMS mass spectra of the ambient aerosols sampled at Fudan
268 University Jiangwan Campus on May 29th, 2019. With the improved m/z measurement,
269 many organic ions, such as C_x , C_xH_y , and $\text{C}_x\text{H}_y\text{O}_z$ can be directly identified (Table 3). Also,
270 we can separate the organic and inorganic species more directly with the high mass
271 resolution. For instance, C_6H_8^+ can be clearly distinguished from possible interference of

272 Ca_2^- , TiO_2^+ and NaKO^+ . C_{10}H^- can also be identified from possible assignment of NaSO_4^-
273 etc. More importantly, $139[\text{C}_2\text{H}_3\text{O}_5\text{S}^-]$ (the theoretical m/z value: -138.97) can be clearly
274 distinguished from other possible assignments, such as $139[\text{C}_{11}\text{H}_7^-]$ with the m/z value of
275 -139.55 and $139[\text{AsO}_4^-]$ with the m/z value of -138.90. Moreover, $153[\text{C}_3\text{H}_5\text{O}_5\text{S}^-]$ with
276 theoretical m/z value of -152.986 can be distinguished from other possible assignments,
277 such as $153[\text{C}_{12}\text{H}_9^-]$ with m/z value of -153.070 and $153[\text{Na}_2\text{Cl}_3^-]$ with m/z value of -
278 152.883. These two important organic ion peaks have been suggested to be the
279 characteristic ion peaks for the organosulfates in secondary organic aerosols (Surratt et al.,
280 2010; Surratt et al., 2007).

281

282

283 **4.3 Time variation of HR-SPAMS measurement**

284 With the high mass resolution of HR-SPAMS and enhanced m/z calibration, we were able
285 to obtain an average mass spectrum from many particles. The accurate m/z values in the
286 average mass spectrum can be used to separate peaks with close m/z and track their
287 intensity variations. Here we conducted a time variation measurement for ambient aerosol
288 from 11:00 on May 29th to 11:00 on May 30th 2019. We selected first 500 particles collected
289 by SPAMS during every hour for elemental analysis. Figure 6 shows that the peak at m/z
290 41 has a bimodal structure, whose m/z were at 40.9546 ± 0.0105 and 41.0194 ± 0.0105 ,
291 respectively. Thus the peak with the smaller m/z is an isotope of K^+ (theoretical m/z value
292 of $^{41}\text{K}^+ = 40.96182$, theoretical m/z bin value of $^{41}\text{K}^+ = 40.9667$; this peak also follows the
293 isotopic pattern of K) and the other peak should be C_3H_5^+ (theoretical m/z value 41.03913,
294 theoretical m/z bin value 41.03). Figure 6 shows that HR-SPAMS was able to separately
295 measure the time series of these two peaks with a small m/z difference. In contrast, it is
296 impossible for a low resolution (LR)-SPAMS to provide such detailed time variation
297 measurement of these peaks.

298

299 **4.4 Particle classification by ART-2a**

300 Adaptive resonance theory neural network (ART-2a) is a widely-used method to classify
301 particles based on the similarity among their mass spectra (Song et al., 1999). Here we
302 make a comparison of ART-2a classification between the HR-SPAMS data and traditional

303 low resolution (LR)-SPAMS data. Particles with the positive and negative spectra were
304 analyzed by ART-2a with a learning rate of 0.05, a vigilance factor of 0.7, and an iteration
305 number of 20. The previous ambient aerosol SPAMS dataset (1,400 particles) was used for
306 the matrix size of the ART-2a is around 7×10^7 . The LR-SPAMS data, whose m/z was at
307 integer level, was generated by summing high resolution SPAMS peaks in each integer m/z
308 bin. The classification results show that the HR-SPAMS data was grouped to 93 categories
309 and the top 45 categories accounted for 96 percent of all particles. The particle number of
310 the first eight categories was 122, 101, 99, 86, 82, 70, 68 and 60 respectively. In contrast,
311 the LR-SPAMS data was only grouped to 33 categories in total and the top 20 categories
312 accounted for the 96 percent of all particles. The particle number of the first eight categories
313 was 170, 118, 107, 107, 106, 92, 90 and 88 respectively. The detailed results can be found
314 in the Fig.S5-S6. Obviously, ART-2a classification of high resolution SPAMS data
315 generated more particle categories. This is mainly because HR-SPAMS mass spectra can
316 differentiate peaks with close m/z, which may be viewed as one peak in LR-SPAMS data.

317

318

319 The ART-2a classification of the HR-SPAMS results (Fig.S6) show that the signal at
320 $23[\text{Na}^+]$ in Type 2HR was stronger than Type 1HR while the signals at $26[\text{CN}^-]$ and
321 $42[\text{CNO}^-]$ were weaker in Type 1 HR. Meanwhile the averaged mass spectra of the Type
322 2HR showed the presence of $206[\text{Pb}^+]$, $207[\text{Pb}^+]$ and $208[\text{Pb}^+]$, which are known to be
323 harmful to human health (Das et al., 2018; Peng et al., 2020). Furthermore, particles of
324 Type 2HR containing abundant secondary inorganic components like $[\text{NO}_2^-]$, $[\text{NO}_3^-]$ and
325 $[\text{SO}_4^-]$, which originated from the aerosol aging processes (Dall'Osto and Harrison, 2012;
326 Ma et al., 2016). In contrast, these two first particle types were lumped together into Type
327 1LR in the LR-SPAMS classification results (Fig.S5). Due to the merge of these two
328 particle types, $[\text{Pb}^+]$ ions were not significant. Meanwhile, Type 3,4,5HR classification
329 results contain strong signals at $26[\text{CN}^-]$, $42[\text{CNO}^-]$, $46[\text{NO}_2^-]$, $62[\text{NO}_3^-]$ and $97[\text{HSO}_4^-]$,
330 suggesting that these three types were from biomass burning or residential cooking burning.
331 $[\text{K}^+]$ is also another feature of this type particle emission (Bi et al., 2011; Hudson et al.,
332 2004). There were obvious relative ion intensity differences at $26[\text{CN}^-]$, $42[\text{CNO}^-]$,
333 $46[\text{NO}_2^-]$ and $62[\text{NO}_3^-]$ among these types, which implied that these three particle types

334 might be from different burning sources or experienced different levels of aging (Luo et
335 al., 2020). While these three particle types were lumped together as Type 2LR. This critical
336 information which could be potentially used to distinguish particle sources and aging
337 processes was lost. Additionally, Type 7HR can be assigned as ECOC type, based on its
338 strong signals at $[C_x^+]$, $[C_xH_y^+]$ and $[C_xH_yO_z^+]$. Particles of this type may come from the
339 primary emission sources, and the emitted black carbon particles would also form this type
340 particles after absorbing some low volatile organic compounds in the atmosphere
341 (Sodeman et al., 2005). $97[HSO_4^-]$ can be observed to have a weaker signal than $62[NO_3^-]$
342 and $46[NO_2^-]$, which implied that the secondary reaction of SO_2 as the precursor of HSO_4^-
343 was not significant in the particle surface for ECOC type particles in this study
344 (SULLIVAN and PRATHER, 2007). In contrast, the classification results of the LR-
345 SPAMS were not so clear and generated less particle types. Given HR-SPAMS spectra
346 have much more detailed chemical information about particles, we would propose that the
347 ART-2a classification of HR-SPAMS might be more accurate.

348

349

350 **5. Conclusion**

351 An automatic linear calibration method had been developed for data analysis of high-
352 resolution SPAMS data. This technique can significantly improve the m/z accuracy of
353 SPAMS spectra for atmospheric aerosol samples. The analysis of HR-SPAMS data for
354 laboratory generated sea spray aerosols shows many details of its chemical compositions.
355 For example, many organic ions, such as $C_2H_3^+$ and CS^+ , can be directly determined. The
356 chemical compositions of ambient aerosols are much more complicated. It is found that,
357 besides major ions (e.g. Na^+ , K^+ , Ca^+ , Fe^+ , Cl^- , CN^- , NO_3^- and HSO_4^-), C_xH_y , $C_xH_yO_z$ and
358 CNO^- can be identified. With this method, HR-SPAMS can also determine the time series
359 of organic and inorganic peaks, whose m/z are very close to each other (e.g. $41K^+$ with the
360 theoretical m/z value at 40.96182 and $C_3H_5^+$ with the theoretical m/z value at 41.03913).
361 Important organic ion peaks, such as tracer peaks for secondary organic matter like
362 $139[C_2H_3O_5S^-]$ and $153[C_3H_5O_5S^-]$, can be identified. More importantly, our ART-2a
363 classification from HR-SPAMS dataset clearly showed a particle type containing heavy
364 metals like Pb^+ , which was obviously ignored in the ART-2a classification from the LR-

365 SPAMS dataset. More particle types were generated by the ART-2a classification of HR-
366 SPAMS data compared to that of LR-SPAMS data, as the original biomass burning particle
367 type can be divided into three more detailed types based on the different signals of 26[CN⁻],
368 42[CNO⁻], 46[NO₂⁻] and other organic species like C_xH_y and C_xH_yO_z, implying different
369 aerosol aging processes or burning conditions. Such detailed information may be critical
370 to study the aging processes and source appointment of atmospheric aerosols. There is a
371 deficiency of this HR-SPAMS calibration method, which has been showed that some
372 fraction of particles cannot be calibrated due to the presence of weak signals of the marker
373 ions. It can be mitigated by applying some additional marker ions. All the automatic linear
374 calibration method codes specific for HR-SPAMS are open access and can be found at
375 <https://github.com/zhuxiaoqiang-fdu/zhuxiaoqiang-fdu>. And we proposed the principle of
376 this calibration method can be adopted in other aerosol mass spectrometers.

377

378

379 **Author Contribution**

380 Y.X and X.W. supervised this study. X.W. and S.Z. designed the calibration and data
381 analysis methods. S. Wang. and S.Z. performance the sea spray aerosol and ambient aerosol
382 experiment. S.Z. wrote the open source code for calibration and data analysis of the single
383 particle mass spectra and made the GUI program with suggestions from X.W. and X.Y..
384 X.W. and S.Z. prepared the manuscript with contributions from all co-authors.

385

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392 **Competing interests**

393 The authors declare that they have no conflict of interest.

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491 **Figure Captions**

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493 **Figure 1.** The GUI program for HR-SPAMS calibration

494 **Figure 2.** Linear calibration with reference ion peaks

495 **Figure 3.** Probability distributions of the marker peak locations before and after Automatic
496 Linear Calibration (AL-Cal) for (a) sea spray aerosol and (b) ambient aerosol

497 **Figure 4.** Averaged positive and negative mass spectra of sea spray aerosols

498 **Figure 5.** Averaged positive and negative mass spectra of ambient aerosols

499 **Figure 6.** Time series of peak intensities at m/z 40.95 and m/z 41.01

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501 Table 1. Possible peak assignments for the m/z of reference ions for sea spray aerosol and
 502 ambient aerosol

Unit resolution	mass m/z	Possible species (Sea spray aerosol)	Unit resolution	mass m/z	Possible species (Ambient aerosol)
+24		Mg ⁺ C ₂ ⁺	+39		K ⁺ C ₃ H ₃ ⁺
+39		K ⁺ C ₃ H ₃ ⁺	+56		Fe ⁺ Si ₂ ⁺ CaO ⁺ KOH ⁺
+81		Na ₂ Cl ⁺ Br ⁺ C ₆ H ₉ ⁺	-26		CN ⁻ BO ⁻ C ₂ H ₂ ⁻
+113		K ₂ Cl ⁺ C ₉ H ₅ ⁺	-62		NO ₃ ⁻ C ₅ H ₂ ⁻
+115		K ₂ Cl ⁺ C ₉ H ₇ ⁺	-96		SO ₄ ⁻ BrOH ⁻
-26		CN ⁻ BO ⁻ C ₂ H ₂ ⁻	-97		HSO ₄ ⁻ C ₈ H ⁻ BrO ⁻ NaCl ₂ ⁻ H ₂ PO ₄ ⁻
-37		Cl ⁻ C ₃ H ⁻			
-42		BO ₂ ⁻ CNO ⁻			
-129		MgCl ₃ ⁻ C ₁₀ H ₉ ⁻ (C ₃ H ₇) ₂ C ₂ H ₅ ⁻ CaCl ₂ OH ⁻			
-131		MgCl ₃ ⁻			

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510 Table 2. Peak identification of important chemical species in sea spray aerosols. The first
 511 column is the measured m/z for peaks. The second and third columns shows the theoretical
 512 m/z bin value and theoretical m/z value of most possible species for each peak

Measurement m/z(positive)	Possible species (theoretical m/z bin value)	Possible species (theoretical m/z value)	Measurement m/z(negative)	Possible species (theoretical m/z bin value)	Possible species (theoretical m/z value)
22.993	Na ⁺ (22.993)	Na ⁺ (22.98977)	15.0344	CH ₃ ⁻ (15.0216)	CH ₃ ⁻ (15.02348)
23.9829	Mg ⁺ (23.9829)	Mg ⁺ (23.98505)	34.9641	Cl ⁻ (34.9641)	Cl ⁻ (34.96885)
27.0267	C ₂ H ₃ ⁺ (27.0267)	C ₂ H ₃ ⁺ (27.02348)	41.9864	CNO ⁻ (41.9971)	CNO ⁻ (41.99799)
38.9672	K ⁺ (38.9672)	K ⁺ (38.96371)	25.0163	C ₂ H ⁻ (25.0081)	C ₂ H ⁻ (25.00783)
39.9711	Ca ⁺ (39.9607)	Ca ⁺ (39.96259)	38.0024	C ₃ H ₂ ⁻ (38.0126)	C ₃ H ₂ ⁻ (38.01565)
43.9614	CS ⁺ (43.9723)	CS ⁺ (43.9721)	44.9883	HCO ₂ ⁻ (44.9994)	HCO ₂ ⁻ (44.99767)
45.983	Na ₂ ⁺ (45.983)	Na ₂ ⁺ (45.97954)	49.002	C ₄ H ⁻ (49.0135)	C ₄ H ⁻ (49.00783)
59.9569	SiO ₂ ⁺ (59.9696)	SiO ₂ ⁺ (59.96677)	57.9574	NaCl ⁻ (57.9574)	NaCl ⁻ (57.95865)
71.9872	C ₆ ⁺ (72.0012)	C ₆ ⁺ (72)			
80.9438	Na ₂ Cl ⁺ (80.9438)	Na ₂ Cl ⁺ (80.94839)	63.9574	SO ₂ ⁻ (63.9574)	SO ₂ ⁻ (63.96191)
112.898	K ₂ Cl ⁺ (112.898)	K ₂ Cl ⁺ (112.89627)	75.9498	SiO ₃ ⁻ (75.9642)	SiO ₃ ⁻ (75.96196)
138.89	Na ₃ Cl ₂ ⁺ (138.907)	Na ₃ Cl ₂ ⁺ (138.90702)			
140.897	Na ₃ Cl ₂ ⁺ (140.897)	Na ₃ Cl ₂ ⁺ (140.90407)	79.9547	SO ₃ ⁻ (79.9547)	SO ₃ ⁻ (79.95683)
			80.9015	Br ⁻ (80.9164)	Br ⁻ (80.91629)
			85.9484	NaPO ₂ ⁻ (85.9484)	NaPO ₂ ⁻ (85.95337)
			99.9499	CaCO ₃ ⁻ (99.9499)	CaCO ₃ ⁻ (99.94735)
			109.917	CaCl ₂ ⁻ (109.9)	CaCl ₂ ⁻ (109.9003)
			128.901	MgCl ₃ ⁻ (128.883)	MgCl ₃ ⁻ (128.89161)

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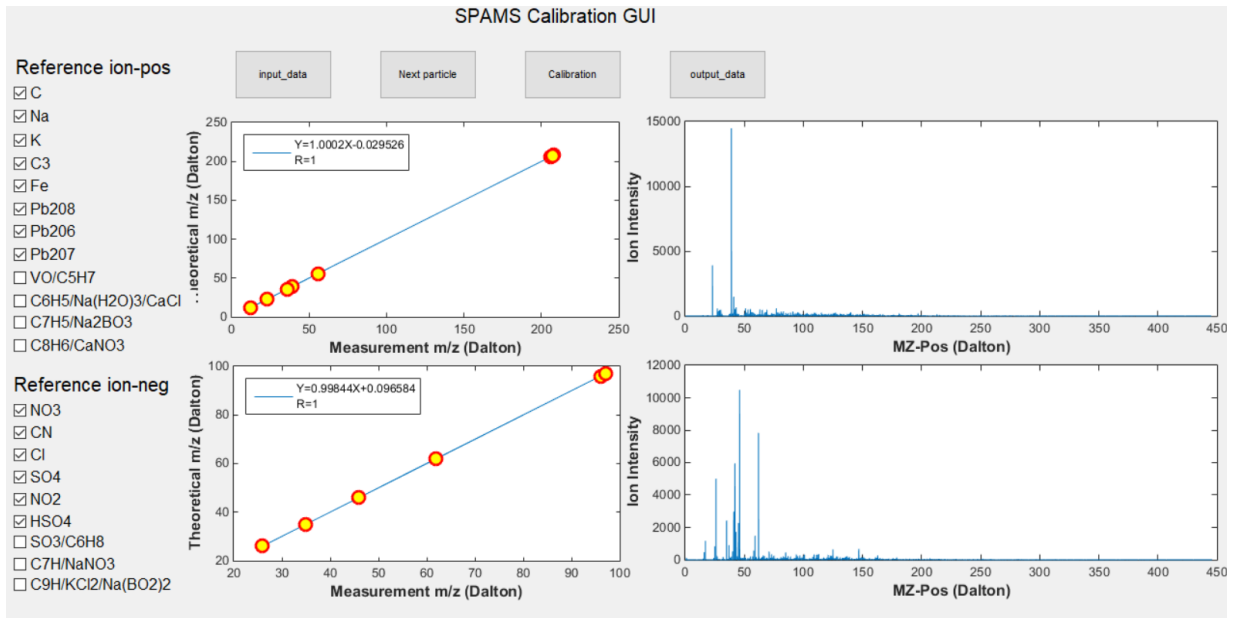
520 Table 3. Peak identification of important chemical species in ambient aerosols. The first
 521 column is the measured m/z for each peak. The second and third columns shows the
 522 theoretical m/z bin value and theoretical m/z value of most possible specie for each peak

Measurement m/z(positive)	Possible species(theoretical m/z bin value)	Possible species(theoretical m/z value)	Measurement m/z(negative)	Possible species(theoretical m/z bin value)	Possible species(theoretical m/z value)
22.993	Na ⁺ (22.993)	Na ⁺ (22.98977)	15.0408	CH ₃ ⁻ (15.0216)	CH ₃ ⁻ (15.02348)
23.9991	C ₂ ⁺ (23.9991)	C ₂ ⁺ (24)	16.0091	O ⁻ (15.9959)	O ⁻ (15.99492)
25.002	C ₂ H ⁺ (25.0103)	C ₂ H ⁺ (25.00783)	17.0145	OH ⁻ (17.0009)	OH ⁻ (17.00275)
26.0087	C ₂ H ₂ ⁺ (26.0171)	C ₂ H ₂ ⁺ (26.01565)	26.0078	CN ⁻ (25.9994)	CN ⁻ (26.00307)
30.0171	NO ⁺ (29.999)	NO ⁺ (29.99799)	31.987	O ₂ ⁻ (31.987)	O ₂ ⁻ (31.98984)
35.9925	C ₃ ⁺ (36.0023)	C ₃ ⁺ (36)	34.9641	Cl ⁻ (34.9641)	Cl ⁻ (34.96885)
36.9976	C ₃ H ⁺ (37.0076)	C ₃ H ⁺ (37.00783)	41.9971	CNO ⁻ (41.9971)	CNO ⁻ (41.99799)
38.0065	C ₃ H ₂ ⁺ (38.0166)	C ₃ H ₂ ⁺ (38.01565)	45.9897	NO ₂ ⁻ (45.9897)	NO ₂ ⁻ (45.99291)
38.9672	K ⁺ (38.9672)	K ⁺ (38.96371)	47.9911	C ₄ ⁻ (47.993)	C ₄ ⁻ (48)
47.993	C ₄ ⁺ (47.993)	C ₄ ⁺ (48)	61.9808	NO ₃ ⁻ (61.9938)	NO ₃ ⁻ (61.98783)
48.9911	C ₄ H ⁺ (49.0026)	C ₄ H ⁺ (49.00783)	71.0014	C ₃ H ₃ O ₂ ⁻ (71.0153)	C ₃ H ₃ O ₂ ⁻ (71.01332)
49.9994	C ₄ H ₂ ⁺ (50.0111)	C ₄ H ₂ ⁺ (50.01565)	78.9548	PO ₃ ⁻ (78.9548)	PO ₃ ⁻ (78.95852)
55.9443	Fe ⁺ (55.932)	Fe ⁺ (55.93494)	79.94	SO ₃ ⁻ (79.9547)	SO ₃ ⁻ (79.95683)
59.9951	C ₅ ⁺ (59.9951)	C ₅ ⁺ (60)	80.946	HSO ₃ ⁻ (80.9609)	HSO ₃ ⁻ (80.96466)
60.9946	C ₅ H ⁺ (61.0074)	C ₅ H ⁺ (61.00783)	95.9825	SO ₄ ⁻ (95.9502)	SO ₄ ⁻ (95.95175)
62.0023	C ₅ H ₂ ⁺ (62.0152)	C ₅ H ₂ ⁺ (62.01565)	96.9546	HSO ₄ ⁻ (96.9546)	HSO ₄ ⁻ (96.95958)
72.0012	C ₆ ⁺ (72.0012)	C ₆ ⁺ (72)	121.01	C ₁₀ H ⁻ (121.01)	C ₁₀ H ⁻ (121.00783)
84.0108	C ₇ ⁺ (83.9957)	C ₇ ⁺ (84)	122.01	C ₁₀ H ₂ ⁻ (122.01)	C ₁₀ H ₂ ⁻ (122.01565)
207.976	Pb ⁺ (207.967)	Pb ⁺ (207.97664)	134.008	C ₁₁ H ₂ ⁻ (134.008)	C ₁₁ H ₂ ⁻ (134.01565)

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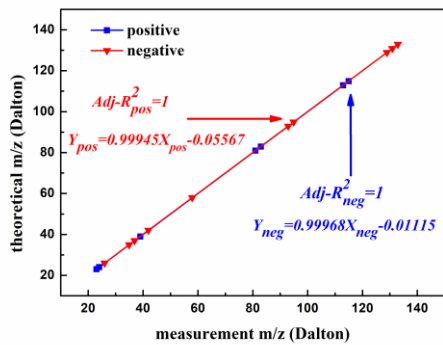
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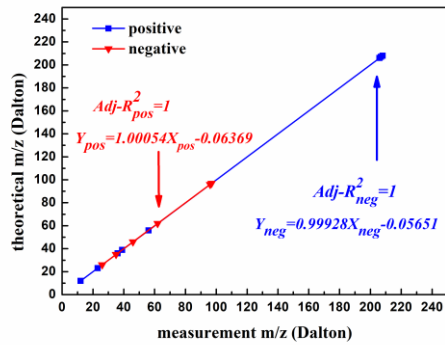
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Figure 1. The GUI program for HR-SPAMS calibration

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(a) sea spray aerosol



(b) ambient aerosol

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Figure 2. Linear calibration with reference ion peaks

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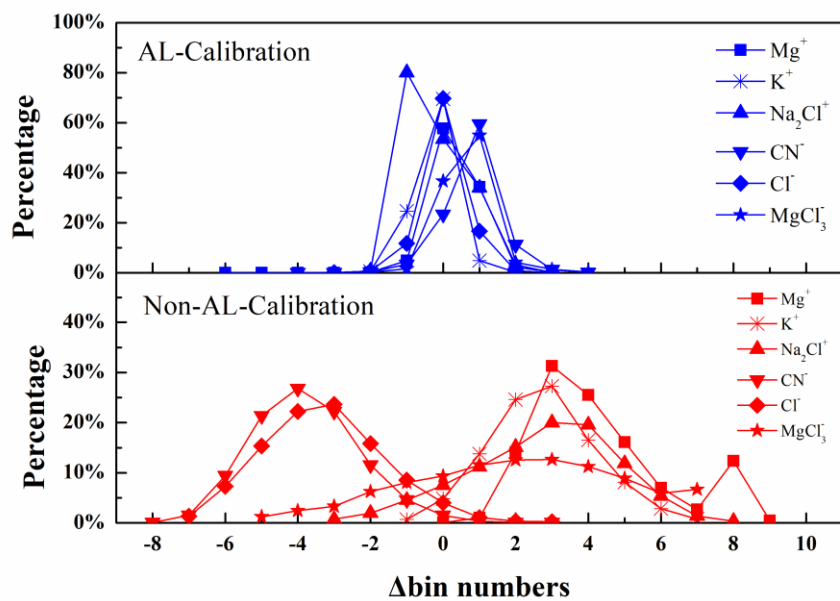
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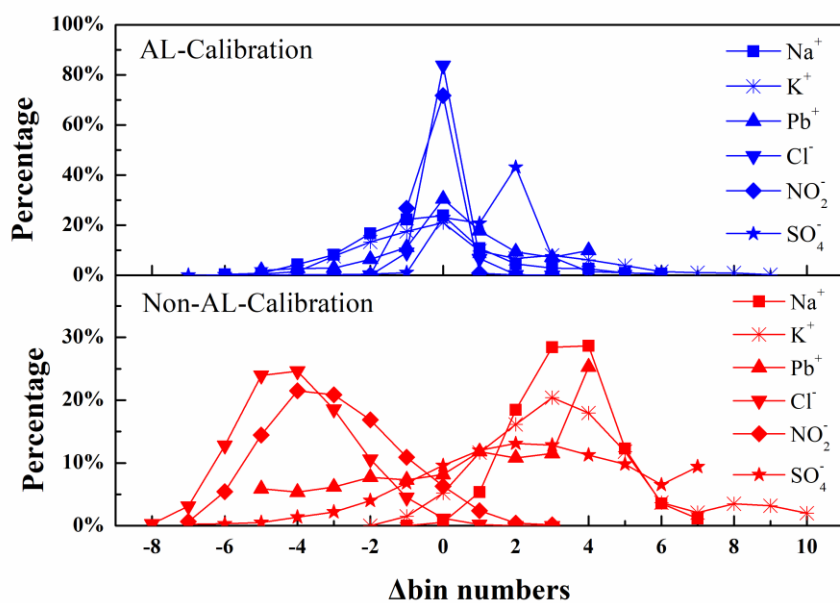
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548 a.



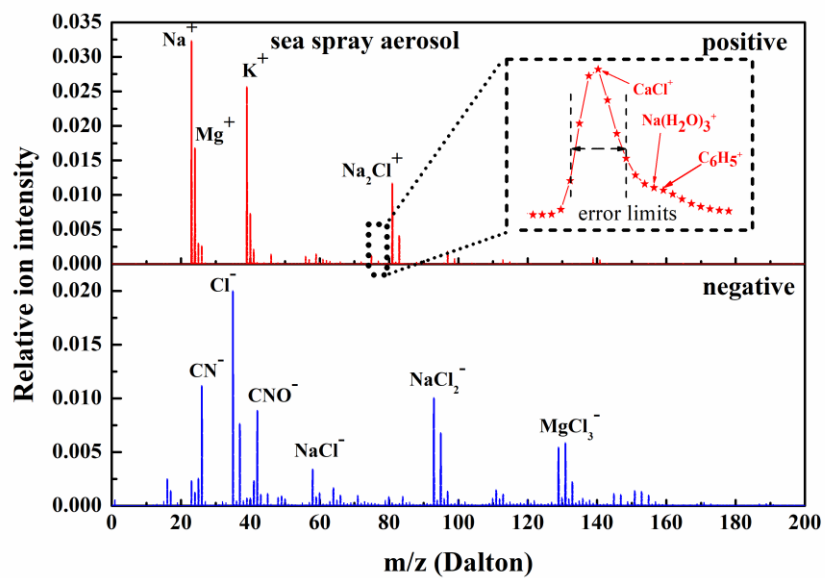
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552 Figure 3. Probability distributions of the marker peak locations before and after Automatic
553 Linear Calibration (AL-Cal) for (a) sea spray aerosol and (b) ambient aerosol

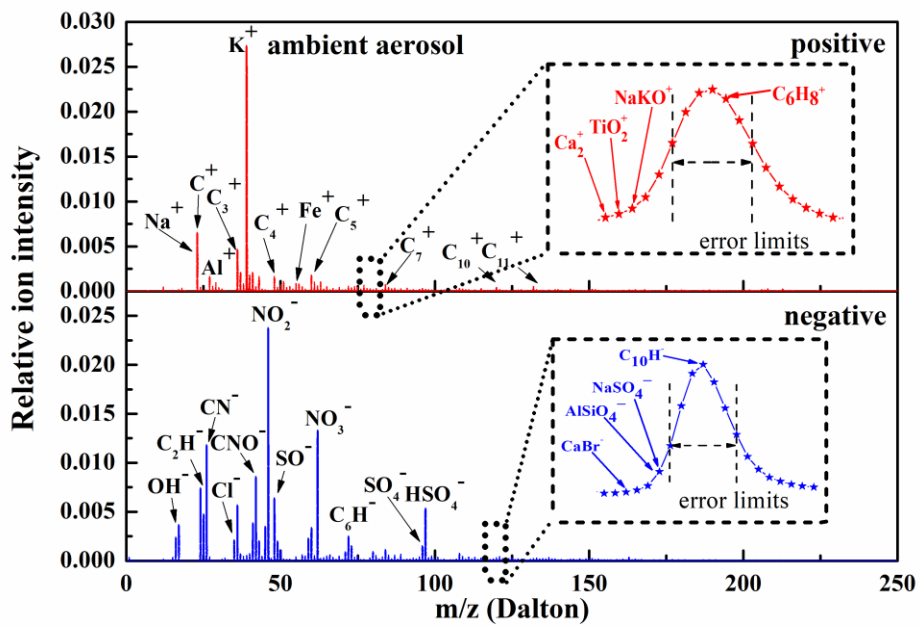


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Figure 4. Averaged positive and negative mass spectra of sea spray aerosols

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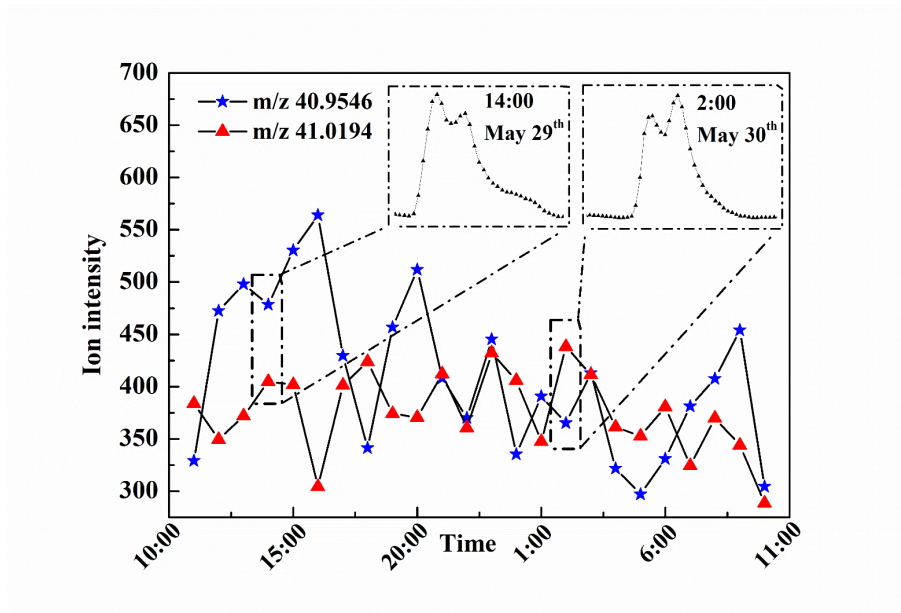
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Figure 5. Averaged positive and negative mass spectra of ambient aerosols

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Figure 6. Time series of peak intensities at m/z 40.95 and m/z 41.01

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