



1 **Development of an Automatic Linear Calibration Method for High**
2 **Resolution Single Particle Mass Spectrometry: Improved Chemical**
3 **Species Identification for Atmospheric Aerosols**

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19 **Atmospheric Measurement Techniques Discussions**

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21 March 8th, 2020

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33 **Abstract**

34

35 The mass resolution of laser desorption ionization (LDI) single particle aerosol mass
36 spectrometry (SPAMS) is usually low (~ 500), which has been greatly improved by recent
37 development of delayed ion extraction technique. However, due to large fluctuations
38 among LDI processes during each laser shot, accurate calibration of mass-to-charge ratio
39 for high resolution SPAMS spectra is challenging. Here we developed an automatic linear
40 calibration method to improve the accuracy of mass-to-charge (m/z) measurement for
41 single atmospheric aerosol particles. Laboratory generated sea spray aerosol and
42 atmospheric ambient aerosol were tested. After the calibration, the fluctuation ranges of
43 the reference ions (e.g. Pb^+ and SO_4^+) m/z reaches ± 0.018 for sea spray aerosol and \pm
44 0.024 for ambient aerosol in average mass spectra. With such m/z accuracy, the HR-
45 SPAMS spectra of sea spray aerosol can easily identify elemental compositions of organic
46 peaks, such as C_x , C_xH_y and $C_xH_yO_z$. While the chemical compositions of ambient aerosols
47 are more complicated, C_xH_y , $C_xH_yO_z$ and CNO peaks can also be identified based on their
48 accurate mass. With the improved resolution, the time series of peaks with small m/z
49 differences can be separated and measured. In addition, it is also found that applying high
50 resolution data with enhanced mass calibration can significantly affect particle
51 classification (identification) using the ART-2a algorithm, which classify particles based on
52 similarities among single particle mass spectra.

53

54



55 1. Introduction

56

57 Atmospheric aerosols significantly impact radiative forcing, cloud formation and human
58 health(Ackerman et al., 2004; Zhang and Kin-Fai, 2012). They originate from various
59 sources and undergo many atmospheric aging processes, resulting in an extremely
60 complicated mixture of particles with a large range of sizes and chemical compositions.
61 The mixture is usually referred as “mixing state”. Measurement of aerosol mixing state
62 requires single particle characterization techniques. Utilizing laser ablation/ionization of
63 single aerosol particle, Single Particle Aerosol Mass Spectrometer (SPAMS) has been
64 widely used to measure chemical compositions and sizes of aerosols in real-time(Moffet
65 and Prather, 2009; Murphy, 2010; Sullivan and Prather, 2005). Based on this technique,
66 ART-2a and other algorithms had been developed to classify the ambient particles based
67 on their mass spectra and identify their sources and aging paths (Reinard et al., 2007;
68 Zelenyuk and Imre, 2009).

69

70 However, SPAMS with LDI has several serious limitations(Manuel et al., 2006; Wenzel et
71 al., 2003). A major issue is that the mass resolution of the SPAMS is relatively low (~500)
72 and the accuracy of m/z (mass to charge ratio) is usually at integer level, resulting in
73 uncertainties about the identification of chemical species (Nash et al., 2006; Pratt and
74 Prather, 2012; Qin et al., 2006). Due to the low mass resolution, many organic and
75 inorganic peaks cannot be separated, such as $K^+ / C_3H_3^+$ with the integer m/z at 39, Al^+
76 $/ C_2H_3^+$ with the integer m/z at 27 and $CN^- / C_2H_2^-$ with the integer m/z at -26 (Li et al.,
77 2018). To identify these chemical species, A SPAMS with higher mass resolution and better
78 m/z accuracy is needed.

79

80 Recently, Li et al., significantly increased SPAMS’s mass resolution to ~2000 by applying
81 delayed ion extraction technique, which combined a standard rectangular extraction pulse
82 with an exponential pulse (Li et al., 2018). This updated SPAMS is called high resolution
83 (HR)-SPAMS. Unfortunately, in spite of resolution enhancement with this new technique,
84 ion peak position was still very sensitive to initial ion coordinate and speed (Chudinov et
85 al., 2019), i.e. the calibration parameter for each mass spectrum is significantly different.



86 Therefore, in order to get accurate m/z , Chudinov et al. used several peaks with known m/z
87 to calibrate every SPAMS spectrum for $\text{Pb}(\text{NO}_3)_2$ and NaI particles produced from an
88 atomizer.

89

90 However, atmospheric particles are extremely complicated with a wide range of chemical
91 compositions and sizes (Zhang et al., 2013), which bring much greater challenge to
92 properly calibrating each SPAMS mass spectra and obtaining accurate m/z measurement.
93 We need to develop a new MS calibration method for atmospheric aerosols and evaluate
94 its performance comprehensively.

95

96 In this study, we report a calibration method for single particle high resolution mass spectra
97 data. Its performance had been evaluated in detail for both laboratory-generated sea spray
98 aerosol and ambient aerosol. In addition, the impact of using high resolution SPAMS data
99 on particle classification by ART-2a algorithm was accessed.

100

101 **2. Experimental Section**

102 **2.1 High Resolution Single Particle Aerosol Mass Spectrometer (HR-SPAMS)**

103 The detailed description of HR-SPAMS (Hexin Analytical Instrument Co., Ltd., China) can
104 be found elsewhere (Li et al., 2018). Briefly, a HR-SPAMS consists of an aerodynamic lens
105 as its particle inlet, two laser beams system for particle sizing, a UV laser for LDI and a
106 bipolar time-of-flight mass analyzer for the detection of positive and negative ions. Positive
107 and negative ions are detected by two z-shape bipolar TOF reflectron mass analyzers. The
108 size detection range of HR-SPAMS is 200-2000 nm. As introduced before, this HR-SPAMS
109 used delayed ion extraction technique.

110

111 **2.2 Laboratory generated sea spray aerosol**

112 Sea spray aerosol was produced by water jet method. In a sea spray aerosol production
113 tank, a seawater jet was hitting seawater surface and producing bubbles, which would rise
114 to the surface and burst. Bubble bursting process produces sea spray aerosols. Seawater
115 was collected at Fengxian, Shanghai (30°92'N and 121°47'E) on March 30th 2019 (Fig.S1).

116



117 **2.3 Ambient aerosol sampling**

118 Ambient aerosol sampling was conducted at Fudan university, Shanghai (31°20'N and
119 121°30'E) on May 29th 2019 (Fig.S1). The ambient particles were dried a diffusional dryer
120 before being sampled by the HR-SPAMS.

121

122 **3. Development of Calibration Methods**

123 **3.1 Automatic linear calibration method**

124 To improve the accuracy of m/z for HR-SPAMS spectra, an automatic linear calibration
125 method has been developed. Noticeably, due to the technical limitation of data acquisition,
126 the whole HR-SPAMS spectrum is not continuous but divided by a large number of m/z
127 bins, which are described in Fig.S2. Here we denote “m/z bin value” as the median m/z
128 value of each bin.

129

130 The linear calibration method is described as the following steps:

131

132 Step 0: The SPAMS data was coarsely-calibrated by the traditional method, which usually
133 selected a few particles with distinct ion patterns, i.e. the molecular composition of some
134 distinct peaks in the mass spectra can be easily identified. Then, the time of flight of these
135 peaks and the m/z of the corresponding ions were used to calculate a set of calibration
136 parameters for both positive and negative spectra. The parameters were then applied to the
137 whole mass spectra dataset, and the coarsely-calibrated was completed.

138

139 Step 1: a pool of ion peaks in the single particle mass spectra were selected as the potential
140 m/z calibration reference ions. The selection criteria are (1) these peaks should be present
141 in most of the spectra; (2) the identification of these ion peaks should not be significantly
142 affected by other adjacent peaks. For example, 27[Al]⁺ was not selected, as its adjacent
143 peak 27[C₂H₃]⁺ may affect the peak shape and identification of 27[Al]⁺.

144

145 According to the previous research, possible peak assignments for the m/z of reference ions
146 for sea spray aerosol and ambient aerosol were listed on Table 1(Bertram et al., 2018;
147 Collins et al., 2014; Tsunogai et al., 1972; Wang et al., 2016; Wang et al., 2019). For sea



148 spray aerosol, according to several studies (Bertram et al., 2018; Collins et al., 2014;
149 Tsunogai et al., 1972), the reference ions with m/z 23 24 39 -35 -37 were $23[\text{Na}]^+$ $24[\text{Mg}]^+$
150 $39[\text{K}]^+$ $-35[\text{Cl}]^-$ and $-37[\text{Cl}]^-$ respectively. And Collins et al. shows that the reference ions
151 with m/z 81, 83, -26, -42, -58, -129, and -131 were $81[\text{Na}_2\text{Cl}]^+$, $83[\text{Na}_2\text{Cl}]^+$, $-26[\text{CN}]^-$, -
152 $42[\text{CNO}]^-$, $-58[\text{NaCl}]^-$, $-129[\text{MgCl}_3]^-$, and $-131[\text{MgCl}_3]^-$, respectively (Collins et al., 2014).
153 Due to the fact that Na, Mg and K were abundant in sea spray aerosol, the reference ions
154 with m/z 113 and 115 should be $113[\text{K}_2\text{Cl}]^+$ and $115[\text{K}_2\text{Cl}]^+$. Thus, in this study, we select
155 $23[\text{Na}]^+$, $24[\text{Mg}]^+$, $39[\text{K}]^+$, $81[\text{Na}_2\text{Cl}]^+$, $83[\text{Na}_2\text{Cl}]^+$, $113[\text{K}_2\text{Cl}]^+$, $115[\text{K}_2\text{Cl}]^+$, $-35[\text{Cl}]^-$, -
156 $37[\text{Cl}]^-$, $-26[\text{CN}]^-$, $-42[\text{CNO}]^-$, $-129[\text{MgCl}_3]^-$, $-131[\text{MgCl}_3]^-$, $-58[\text{NaCl}]^-$ as the potential
157 reference ions for sea spray aerosols.

158

159 For the ambient aerosol, according to the previous ambient SPAMS measurements (Wang
160 et al., 2016; Wang et al., 2019), the reference ions with m/z 12, 23, 36, 39, 56, 207, 208,
161 and 209 were assigned to $12[\text{C}]^+$, $23[\text{Na}]^+$, $36[\text{C}_3]^+$, $39[\text{K}]^+$, $56[\text{Fe}]^+$, $207[\text{Pb}]^+$, $208[\text{Pb}]^+$
162 and $209[\text{Pb}]^+$, the reference ions with m/z -26, -35, -46, -62, -96, and -97 were assigned to
163 $-26[\text{CN}]^-$, $-35[\text{Cl}]^-$, $-46[\text{CNO}]^-$, $-62[\text{NO}_2]^-$, $-96[\text{SO}_4]^-$ and $-97[\text{HSO}_4]^-$, respectively.
164 Therefore, in this study, we select the $12[\text{C}]^+$, $23[\text{Na}]^+$, $39[\text{K}]^+$, $36[\text{C}_3]^+$, $56[\text{Fe}]^+$, $208[\text{Pb}]^+$,
165 $206[\text{Pb}]^+$, $207[\text{Pb}]^+$, $-62[\text{NO}_3]^-$, $-26[\text{CN}]^-$, $-35[\text{Cl}]^-$, $-96[\text{SO}_4]^-$, $-46[\text{NO}_2]^-$, $-97[\text{HSO}_4]^-$ as the
166 potential reference ions for ambient aerosols.

167

168 Step 2: a set of reference ions was chosen from the potential reference ion pool for each
169 spectrum. The selection was based on the absolute ion intensity of the reference ions in this
170 spectrum. They must be greater than a threshold, e.g. we set 15 a.u. for ambient aerosol
171 and 8 a.u. for sea spray aerosol, respectively. A particle was discarded from the spectra
172 database if it did not have enough reference ions (the minimum number of reference ions
173 was set to be 5) in either positive or negative mass spectrum.

174

175 Step 3: the reference ions were used to calibrate m/z for mass spectra of each particle. As
176 introduced before, a HR-SPAMS spectrum consists of a number of bins. The measured m/z
177 bin values of the reference ions were calibrated based on their theoretic (or true) m/z bin
178 values. A linear regression between the two set of variables (measured vs. theoretic m/z bin



179 values) was conducted, and two calibration parameters (a slope and an intersect) were
180 obtained. After applied calibration parameters, the m/z of the whole spectrum had been
181 corrected. However, the corrected m/z may not be equal to the m/z bin value. Thus, we
182 assigned a m/z bin value to each corrected m/z based on proximity principle. Finally, mass
183 spectra with well calibrated bin value can be obtained for each single particle.

184

185 A GUI program for this automatic linear calibration method had been developed for the
186 sake of easy use (Fig. 1). The MATLAB codes for this GUI and the automatic linear
187 calibration method are open access and available at <https://github.com/zhuxiaoqiang-fdu/zhuxiaoqiang-fdu>.
188

189

190 **3.2 Evaluation of the calibration method**

191 In this study, a total of 5,130 sea spray aerosol particles and 5,007 ambient aerosol particles
192 were analyzed. And 4,624 sea spray particles and 1,409 ambient particles were successfully
193 calibrated. Figure 2 shows that the calibration curves for a random selected sea spray
194 aerosol particle and ambient aerosol particle. The adj- R^2 coefficients of both calibration
195 curves are equal to ~ 1 , demonstrating that this calibration method is effective and accurate.
196 All the slopes and intercepts of the linear calibration can be found in the Fig.S3 and Fig.S4.
197 In addition, Figure 3a and 3b report a comparison of m/z distributions of reference ions
198 between before and after automatic linear calibration. The results show that the fluctuations
199 of the reference ions m/z were significantly reduced after automatic linear calibration. The
200 average m/z deviation of the reference ions was reduced from ~ 0.04 to ~ 0.001 for sea spray
201 aerosol, and from ~ 0.035 to ~ 0.006 for ambient aerosol, respectively.

202

203 **3.3 Automatic linear calibration method with a larger reference ion pool**

204 It is important to note that a large number of ambient particles were filtered because their
205 spectra did not have 5 or more reference peaks to conduct calibrations. Especially, only
206 $\sim 29.0\%$ of total ambient particles had sufficient number of reference ions in their positive
207 spectra. To solve this problem, extra reference ions, including $67[\text{VO}]^+$, $67[\text{C}_5\text{H}_7]^+$,
208 $89[\text{C}_7\text{H}_5]^+$, $89[\text{Na}_2\text{BO}_2]^+$, $102[\text{C}_8\text{H}_6]^+$, $102[\text{CaNO}_3]^+$, were added into the original positive
209 reference ion pool. Obviously, these ions share the same integer m/z value with other ions.



210 We needed to identify them using additional information other than their integer m/z values.

211

212 The specific reference ions were determined by their coarsely-calibrated m/z . Table S2
213 shows that the m/z deviation ranges of the reference positive ambient ions in coarsely-
214 calibrated spectra before automatic linear calibration were around 0.011~0.048, while the
215 m/z differences between $67[\text{VO}]^+$ and $67[\text{C}_5\text{H}_7]^+$, $89[\text{C}_7\text{H}_5]^+$ and $89[\text{Na}_2\text{BO}_2]^+$, $102[\text{C}_8\text{H}_6]^+$
216 and $102[\text{CaNO}_3]^+$ were 0.1213, 0.0622, and 0.083, respectively, which were larger than the
217 m/z deviations of these reference ions in coarsely-calibrated spectra. Therefore, the
218 coarsely-calibrated spectra can be used to determine these specific reference ions. With
219 these additional potential reference ions, a total of 2490 ambient particles were calibrated,
220 much more than the previous analysis (1,409 ambient particles). The deviations from
221 theoretical m/z for applying this expanded ion pool are summarized in Table S3. The
222 average m/z deviation of the reference ions is ~ 0.0068 .

223

224 **4. Application to atmospheric aerosols measurement**

225 **4.1 HR-SPAMS measurement of sea spray aerosol**

226 SPAMS data usually contains a large number of individual mass spectra. It is impossible
227 to manually analyze every spectrum from a large dataset. Averaging a number of mass
228 spectra is often preferred. However, to obtain averaged high-resolution spectrum, each
229 spectrum must be well calibrated. Therefore, it would be very interesting to see what new
230 information can be obtained from HR-SPAMS measurement of aerosols with the automatic
231 calibration method. Figure 4 reports the average positive and negative mass spectra for the
232 laboratory generated sea spray aerosols. Similar to the low-resolution sea spray aerosol
233 mass spectra, they contain major peaks of Na^+ , Mg^{2+} , K^+ , Na_2Cl^+ , CN^- , Cl^- , CNO^- , NaCl^- ,
234 NaCl_2^- and MgCl_3^- , as well as many smaller peaks, such as Ca^+ , $\text{SiO}_2^-/\text{SiO}_3^-$, and KCl_2^- .
235 With the improved m/z measurement, many peaks, which cannot be determined by integer
236 resolution mass spectra, now can be clearly identified (Table 2). For example, the ion with
237 m/z at 27.0267 is C_2H_3^+ rather than Al^+ . The ion with m/z at 76.9336 is CaCl^+ rather than
238 C_6H_5^+ . And some sulfur containing organic ions, such as CS^+ , can be determined.
239 Surprisingly, we can identify the presence of HCO_2^- and CaCO_3^- , demonstrating that carbon
240 hydrates were contained in sea spray aerosols.



241

242 **4.2 HR-SPAMS measurement of atmospheric aerosol**

243 Laboratory generated sea spray aerosol can be viewed as a relatively simple aerosol system,
244 while the chemical compositions of ambient aerosols are much more complicated. Figure
245 5 shows the averaged HR-SPAMS mass spectra of the ambient aerosols sampled at Fudan
246 University Jiangwan Campus on May 29th, 2019. With the improved m/z measurement,
247 many organic ions, such as C_x , C_xH_y , and $C_xH_yO_z$ can be directly identified (Table 3). Also,
248 we can separate the organic and inorganic species more directly with the high mass
249 resolution. For instance, $C_6H_8^+$ can be clearly distinguished from possible interference of
250 Ca_2^- , TiO_2^+ and $NaKO^+$. $C_{10}H^+$ can also be identified from possible assignment of $NaSO_4^-$
251 etc.

252

253 **4.3 Time variation of HR-SPAMS measurement**

254 With the high mass resolution of HR-SPAMS and enhanced m/z calibration, we were able
255 to obtain an average mass spectrum from many particles. The accurate m/z values in the
256 average mass spectrum can be used to separate peaks with close m/z and track their
257 intensity variations. Here we conducted a time variation measurement for ambient aerosol
258 from 11:00 on May 29th to 11:00 on May 30th 2019. We selected first 500 particles collected
259 by SPAMS during every hour for elemental analysis. Figure 6 shows that the peak at m/z
260 41 has a bimodal structure, whose m/z were at 40.9546 ± 0.0105 and 41.0194 ± 0.0105 ,
261 respectively. Thus the peak with the smaller m/z is an isotope of K^+ (theoretical m/z value
262 of $^{41}K^+ = 40.96182$, theoretical m/z bin value of $^{41}K^+ = 40.9667$; this peak also follows the
263 isotopic pattern of K) and the other peak should be $C_3H_5^+$ (theoretical m/z value 41.03913,
264 theoretical m/z bin value 41.03). Figure 6 shows that HR-SPAMS was able to separately
265 measure the time series of these two peaks with a small m/z difference. In contrast, it is
266 impossible for a LR-SPAMS to provide such detailed time variation measurement of these
267 peaks.

268

269 **4.4 Particle classification by ART-2a**

270 Adaptive resonance theory neural network (ART-2a) is a widely-used method to classify
271 particles based on the similarity among their mass spectra (Song et al., 1999). Here we



272 make a comparison of ART-2a classification between the HR-SPAMS data and traditional
273 low resolution (LR)-SPAMS data. Particles with the positive and negative spectra were
274 analyzed by ART-2a with a learning rate of 0.05, a vigilance factor of 0.7, and an iteration
275 number of 20. The previous ambient aerosol SPAMS dataset was used. The LR-SPAMS
276 data, whose m/z was at integer level, was generated by summing high resolution SPAMS
277 peaks in each integer m/z bin. The classification results show that the HR-SPAMS data was
278 grouped to 93 categories and the top 45 categories accounted for 96 percent of all particles.
279 The particle number of the first eight categories was 122, 101, 99, 86, 82, 70, 68 and 60
280 respectively. In contrast, the LR-SPAMS data was only grouped to 33 categories in total
281 and the top 20 categories accounted for the 96 percent of all particles. The particle number
282 of the first eight categories was 170, 118, 107, 107, 106, 92, 90 and 88 respectively. The
283 detailed results can be found in the Fig.S5-S6. Obviously, ART-2a classification of high
284 resolution SPAMS data generated more particle categories. This is mainly because HR-
285 SPAMS mass spectra can differentiate peaks with close m/z , which may be viewed as one
286 peak in LR-SPAMS data.

287

288 The ART-2a classification of the HR-SPAMS results (Fig.S6) show that the signal at
289 $23[\text{Na}^+]$ in the second particle type was stronger than first particle type while the signals at
290 $26[\text{CN}^-]$ and $42[\text{CNO}^-]$ were weaker in first particle type. In contrast, these two first particle
291 types were lumped together into the first particle type in the LR-SPAMS classification
292 results (Fig.S5). The third, fourth and fifth particle types from the HR-SPAMS
293 classification results contain strong signals at $26[\text{CN}^-]$, $42[\text{CNO}^-]$, $46[\text{NO}_2^-]$, $62[\text{NO}_3^-]$ and
294 $97[\text{HSO}_4^-]$, while these three particle types were lumped together as the second particle
295 type in the LR-SPAMS classification results. In general, ART-2a classification of HR-
296 SPAMS would generate more particle types than that of LR-SPAMS. Given HR-SPAMS
297 spectra have much more detailed chemical information about particles, we would propose
298 that the ART-2a classification of HR-SPAMS might be more accurate.

299

300 5. Conclusion

301 An automatic linear calibration method had been developed for data analysis of high-
302 resolution SPAMS data. This technique can significantly improve the m/z accuracy of



303 SPAMS spectra for atmospheric aerosol samples. The analysis of HR-SPAMS data for
304 laboratory generated sea spray aerosols shows many details of its chemical compositions.
305 For example, many organic ions, such as $C_2H_3^+$ and CS^+ , can be directly determined. The
306 chemical compositions of ambient aerosols are much more complicated. It is found that,
307 besides major ions (e.g. Na^+ , K^+ , Ca^+ , Fe^+ , Cl^- , CN^- , NO_3^- and HSO_4^-), C_xH_y , $C_xH_yO_z$ and
308 CNO^- can be identified. With this method, HR-SPAMS can also separately measure the
309 time series of peaks, whose m/z are close to each other (e.g. $^{41}K^+$ with the theoretical m/z
310 value at 40.96182 and $C_3H_5^+$ with the theoretical m/z value at 41.03913). Moreover, the
311 ART-2a classification of HR-SPAMS data would generate more particle types compared to
312 the classification results of LR-SPAMS data. All the codes for the automatic linear
313 calibration method are open access and can be found at [https://github.com/zhuxiaoqiang-
314 fdu/zhuxiaoqiang-fdu](https://github.com/zhuxiaoqiang-fdu/zhuxiaoqiang-fdu).

315

316 **Author Contribution**

317 Y.X and X.W. supervised this study. X.W. and S.Z. designed the calibration and data
318 analysis methods. S. Wang. and S.Z. performance the sea spray aerosol and ambient aerosol
319 experiment. S.Z. wrote the open source code for calibration and data analysis of the single
320 particle mass spectra and made the GUI program with suggestions from X.W. and X.Y..
321 X.W. and S.Z. prepared the manuscript with contributions from all co-authors.

322

323 **Acknowledgments**

324 This work was partially supported by the National Natural Science Foundation of China
325 (Nos. 41827804, 41775150, 21906024, 91544224) and Shanghai Natural Science
326 Foundation (No. 19ZR1404000). The authors thank Hexin Analytical Instrument Co., Ltd.,
327 China for providing HR-SPAMS.

328

329 **Competing interests**

330 The authors declare that they have no conflict of interest.

331

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396 **Figure Captions**

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398 **Figure 1.** The GUI program for HR-SPAMS calibration

399 **Figure 2.** Linear calibration with reference ion peaks

400 **Figure 3.** Probability distributions of the marker peak locations before and after Automatic
401 Linear Calibration (AL-Cal) for (a) sea spray aerosol and (b) ambient aerosol

402 **Figure 4.** Averaged positive and negative mass spectra of sea spray aerosols

403 **Figure 5.** Averaged positive and negative mass spectra of ambient aerosols

404 **Figure 6.** Time series of peak intensities at m/z 40.95 and m/z 41.01

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406 Table 1. Possible peak assignments for the m/z of reference ions for sea spray aerosol and
 407 ambient aerosol

Unit resolution	mass m/z	Possible species (Sea spray aerosol)	Unit resolution	mass m/z	Possible species (Ambient aerosol)
+24		Mg ⁺ C ₂ ⁺	+39		K ⁺ C ₃ H ₃ ⁺
+39		K ⁺ C ₃ H ₃ ⁺	+56		Fe ⁺ Si ₂ ⁺ CaO ⁺ KOH ⁺
+81		Na ₂ Cl ⁺ Br ⁺ C ₆ H ₉ ⁺	-26		CN ⁻ BO ⁻ C ₂ H ₂ ⁻
+113		K ₂ Cl ⁺ C ₉ H ₅ ⁺	-62		NO ₃ ⁻ C ₃ H ₂ ⁻
+115		K ₂ Cl ⁺ C ₉ H ₇ ⁺	-96		SO ₄ ⁻ BrOH ⁻
-26		CN ⁻ BO ⁻ C ₂ H ₂ ⁻	-97		HSO ₄ ⁻ C ₈ H ⁻ BrO ⁻ NaCl ₂ ⁻ H ₂ PO ₄ ⁻
-37		Cl ⁻ C ₃ H ⁻			
-42		BO ₂ ⁻ CNO ⁻			
-129		MgCl ₃ ⁻ C ₁₀ H ₉ ⁻ (C ₃ H ₇) ₂ C ₂ H ₅ ⁻ CaCl ₂ OH ⁻			
-131		MgCl ₃ ⁻			

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415 Table 2. Peak identification of important chemical species in sea spray aerosols. The first
 416 column is the measured m/z for peaks. The second and third columns shows the theoretical
 417 m/z bin value and theoretical m/z value of most possible specie for each peak

Measurement m/z(positive)	Possible species (theoretical m/z bin value)	Possible species (theoretical m/z value)	Measurement m/z(negative)	Possible species (theoretical m/z bin value)	Possible species (theoretical m/z value)
22.993	Na ⁺ (22.993)	Na ⁺ (22.98977)	15.0344	CH ₃ ⁻ (15.0216)	CH ₃ ⁻ (15.02348)
23.9829	Mg ⁺ (23.9829)	Mg ⁺ (23.98505)	34.9641	Cl ⁻ (34.9641)	Cl ⁻ (34.96885)
27.0267	C ₂ H ₃ ⁺ (27.0267)	C ₂ H ₃ ⁺ (27.02348)	41.9864	CNO ⁻ (41.9971)	CNO ⁻ (41.99799)
38.9672	K ⁺ (38.9672)	K ⁺ (38.96371)	25.0163	C ₂ H ⁻ (25.0081)	C ₂ H ⁻ (25.00783)
39.9711	Ca ⁺ (39.9607)	Ca ⁺ (39.96259)	38.0024	C ₃ H ₂ ⁻ (38.0126)	C ₃ H ₂ ⁻ (38.01565)
43.9614	CS ⁺ (43.9723)	CS ⁺ (43.9721)	44.9883	HCO ₂ ⁻ (44.9994)	HCO ₂ ⁻ (44.99767)
45.983	Na ₂ ⁺ (45.983)	Na ₂ ⁺ (45.97954)	49.002	C ₄ H ⁻ (49.0135)	C ₄ H ⁻ (49.00783)
59.9569	SiO ₂ ⁺ (59.9696)	SiO ₂ ⁺ (59.96677)	57.9574	NaCl(57.9574)	NaCl(57.95865)
71.9872	C ₆ ⁺ (72.0012)	C ₆ ⁺ (72)			
80.9438	Na ₂ Cl ⁺ (80.9438)	Na ₂ Cl ⁺ (80.94839)	63.9574	SO ₂ ⁻ (63.9574)	SO ₂ ⁻ (63.96191)
112.898	K ₂ Cl ⁺ (112.898)	K ₂ Cl ⁺ (112.89627)	75.9498	SiO ₃ ⁻ (75.9642)	SiO ₃ ⁻ (75.96196)
138.89	Na ₃ Cl ₂ ⁺ (138.907)	Na ₃ Cl ₂ ⁺ (138.90702)			
140.897	Na ₃ Cl ₂ ⁺ (140.897)	Na ₃ Cl ₂ ⁺ (140.90407)	79.9547	SO ₃ ⁻ (79.9547)	SO ₃ ⁻ (79.95683)
			80.9015	Br ⁻ (80.9164)	Br ⁻ (80.91629)
			85.9484	NaPO ₂ ⁻ (85.9484)	NaPO ₂ ⁻ (85.95337)
			99.9499	CaCO ₃ ⁻ (99.9499)	CaCO ₃ ⁻ (99.94735)
			109.917	CaCl ₂ ⁻ (109.9)	CaCl ₂ ⁻ (109.9003)
			128.901	MgCl ₃ ⁻ (128.883)	MgCl ₃ ⁻ (128.89161)

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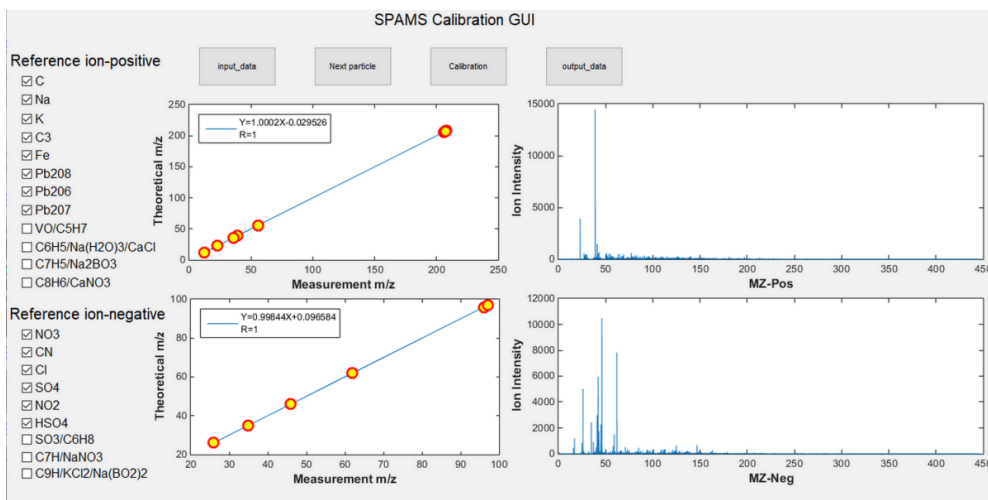
425 Table 3. Peak identification of important chemical species in ambient aerosols. The first
 426 column is the measured m/z for each peak. The second and third columns shows the
 427 theoretical m/z bin value and theoretical m/z value of most possible specie for each peak

Measurement m/z(positive)	Possible species(theoretical m/z bin value)	Possible species(theoretical m/z value)	Measurement m/z(negative)	Possible species(theoretical m/z bin value)	Possible species(theoretical m/z value)
22.993	Na ⁺ (22.993)	Na ⁺ (22.98977)	15.0408	CH ₃ ⁻ (15.0216)	CH ₃ ⁻ (15.02348)
23.9991	C ₂ ⁺ (23.9991)	C ₂ ⁺ (24)	16.0091	O ⁻ (15.9959)	O ⁻ (15.99492)
25.002	C ₂ H ⁺ (25.0103)	C ₂ H ⁺ (25.00783)	17.0145	OH ⁻ (17.0009)	OH ⁻ (17.00275)
26.0087	C ₂ H ₂ ⁺ (26.0171)	C ₂ H ₂ ⁺ (26.01565)	26.0078	CN ⁻ (25.9994)	CN ⁻ (26.00307)
30.0171	NO ⁺ (29.999)	NO ⁺ (29.99799)	31.987	O ₂ ⁻ (31.987)	O ₂ ⁻ (31.98984)
35.9925	C ₃ ⁺ (36.0023)	C ₃ ⁺ (36)	34.9641	Cl ⁻ (34.9641)	Cl ⁻ (34.96885)
36.9976	C ₃ H ⁺ (37.0076)	C ₃ H ⁺ (37.00783)	41.9971	CNO ⁻ (41.9971)	CNO ⁻ (41.99799)
38.0065	C ₃ H ₂ ⁺ (38.0166)	C ₃ H ₂ ⁺ (38.01565)	45.9897	NO ₂ ⁻ (45.9897)	NO ₂ ⁻ (45.99291)
38.9672	K ⁺ (38.9672)	K ⁺ (38.96371)	47.9911	C ₄ ⁻ (47.993)	C ₄ ⁻ (48)
47.993	C ₄ ⁺ (47.993)	C ₄ ⁺ (48)	61.9808	NO ₃ ⁻ (61.9938)	NO ₃ ⁻ (61.98783)
48.9911	C ₄ H ⁺ (49.0026)	C ₄ H ⁺ (49.00783)	71.0014	C ₃ H ₃ O ₂ ⁻ (71.0153)	C ₃ H ₃ O ₂ ⁻ (71.01332)
49.9994	C ₄ H ₂ ⁺ (50.0111)	C ₄ H ₂ ⁺ (50.01565)	78.9548	PO ₃ ⁻ (78.9548)	PO ₃ ⁻ (78.95852)
55.9443	Fe ⁺ (55.932)	Fe ⁺ (55.93494)	79.94	SO ₃ ⁻ (79.9547)	SO ₃ ⁻ (79.95683)
59.9951	C ₅ ⁺ (59.9951)	C ₅ ⁺ (60)	80.946	HSO ₃ ⁻ (80.9609)	HSO ₃ ⁻ (80.96466)
60.9946	C ₅ H ⁺ (61.0074)	C ₅ H ⁺ (61.00783)	95.9825	SO ₄ ⁻ (95.9502)	SO ₄ ⁻ (95.95175)
62.0023	C ₅ H ₂ ⁺ (62.0152)	C ₅ H ₂ ⁺ (62.01565)	96.9546	HSO ₄ ⁻ (96.9546)	HSO ₄ ⁻ (96.95958)
72.0012	C ₆ ⁺ (72.0012)	C ₆ ⁺ (72)	121.01	C ₁₀ H ⁻ (121.01)	C ₁₀ H ⁻ (121.00783)
84.0108	C ₇ ⁺ (83.9957)	C ₇ ⁺ (84)	122.01	C ₁₀ H ₂ ⁻ (122.01)	C ₁₀ H ₂ ⁻ (122.01565)
207.976	Pb ⁺ (207.967)	Pb ⁺ (207.97664)	134.008	C ₁₁ H ₂ ⁻ (134.008)	C ₁₁ H ₂ ⁻ (134.01565)

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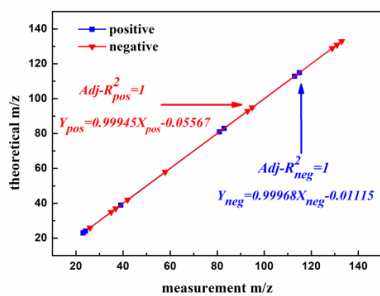
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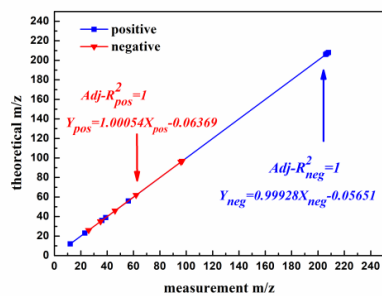
Figure 1. The GUI program for HR-SPAMS calibration



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(a) sea spray aerosol



(b) ambient aerosol

Figure 2. Linear calibration with reference ion peaks

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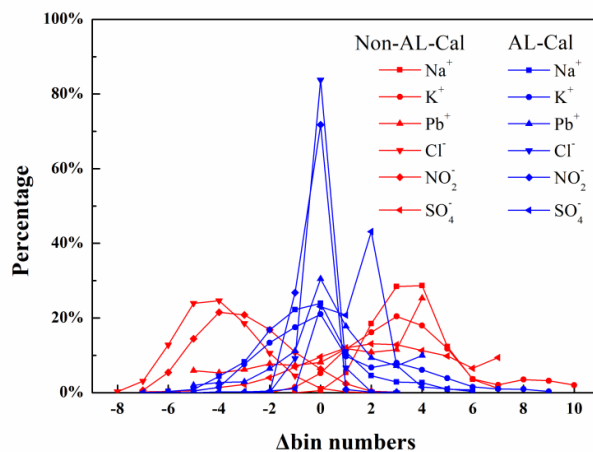
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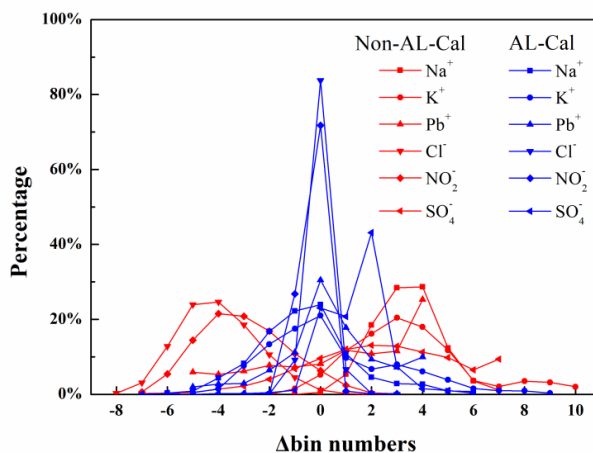


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455 b.



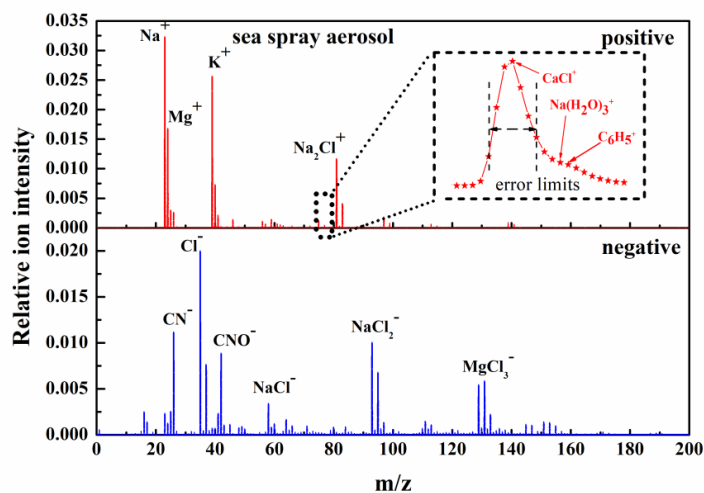
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457 Figure 3. Probability distributions of the marker peak locations before and after Automatic
458 Linear Calibration (AL-Cal) for (a) sea spray aerosol and (b) ambient aerosol

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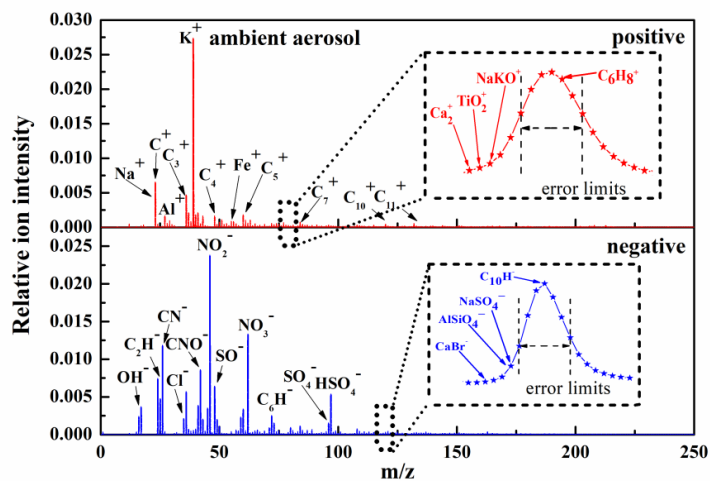
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Figure 4. Averaged positive and negative mass spectra of sea spray aerosols



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Figure 5. Averaged positive and negative mass spectra of ambient aerosols

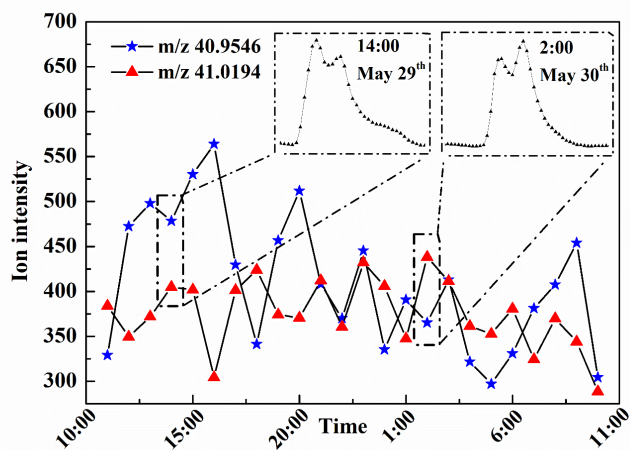
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Figure 6. Time series of peak intensities at m/z 40.95 and m/z 41.01