



Use of filter radiometer measurements to derive local photolysis rates and for future monitoring network application

Hannah L. Walker^{1,2}, Mathew R. Heal¹, Christine F. Braban², Mhairi Coyle^{2,3}, Sarah R. Leeson², Ivan Simmons², Matthew R. Jones², Richard Kift⁴, and Marsailidh M. Twigg²

¹School of Chemistry, University of Edinburgh, David Brewster Road, Edinburgh, EH9 3FJ, UK

²UK Centre for Ecology & Hydrology, Bush Estate, Penicuik, Edinburgh, EH26 0QB, UK

³The James Hutton Institute, Craigiebuckler, Aberdeen, AB15 8QH, UK

⁴Department of Earth and Environmental Sciences, University of Manchester, Oxford Road, Manchester, M13 9PL, UK

Correspondence: H. L. Walker (hannah.walker@ed.ac.uk) and M. R. Heal (m.heal@ed.ac.uk)

Abstract. Production of hydroxyl (OH) radicals is frequently dominated by the photolysis of tropospheric ozone (O₃). However, photolysis of nocturnal radical reservoirs, such as nitrous acid (HONO) and nitryl chloride (ClNO₂), also produces radicals (OH and Cl atoms) that contribute to the oxidising capacity of the local atmosphere, and initiate many radical-chain reactions that lead to the formation of harmful secondary pollutants. Photolysis of nitric acid (HNO₃) is also a minor radical production mechanism. In this paper, locally representative photolysis rate constants (*j*-values) for these molecules are shown to be critical for quantifying and understanding the rate of radical production in a local atmosphere.

The first long-term 4- π filter radiometer dataset in the UK (21 November 2018-20 November 2019) available for direct atmospheric model validation is reported. Measurements were made at Auchencorth Moss, a Scottish rural background site, and *j*(NO₂) is used to generate a measurement-driven adjustment factor (MDAF) for calculated *j*-values that accounts for local changes in meteorological variables without significantly increasing computational cost.

Modelled clear-sky *j*-values and actinic flux for Auchencorth Moss were generated using the Tropospheric Ultraviolet and Visible radiation model (TUV; v.5.3.1). Applying the MDAF metric resulted in the calculated photolytic production rate of OH radicals, from all sources considered, being $\sim 40\%$ lower over the year. Photolysis of HONO resulted in an increased rate of OH production compared to that from O₃ in low-light conditions, such as sunrise and sunset (Solar Zenith Angle $>80^\circ$). Hydroxyl radical production from HONO photolysis exceeded that from O₃ consistently throughout the day during the winter and autumn (by a factor of 5 and 2.1, respectively). Radical production rates from HONO and ClNO₂ reached maximum values during the early morning hours of summer (06:00-09:00 UTC), with OH produced at a rate of 1.03×10^6 OH radicals cm⁻³ s⁻¹, and Cl radicals at 3.20×10^4 Cl radicals cm⁻³ s⁻¹, with the MDAF metric applied.

This first application of the MDAF *j*-values demonstrates an efficient measurement and computational approach to improve modelling of the local atmospheric photochemistry that drives NO₂, O₃ and PM pollution levels. The incorporation of local radiation measurements in measurement networks, and the consequent greater spatial resolution of locally-relevant photolysis coefficients in model photolysis parameterisations, will improve the accuracy of assessment of air pollution and policy-intervention impacts.



25 1 Introduction

Atmospheric chemistry is largely driven by solar radiation. The photolysis of nitrogen dioxide (NO_2) has significant impact on air quality, contributing to the formation of harmful pollutants like tropospheric ozone (O_3 ; Reactions R1 and R2).



30 Photodissociation of many trace gases produce highly reactive radical species such as hydroxyl (OH) and chlorine atoms (Cl), which dominate the oxidising capacity of the atmosphere. Since reactions with these radicals are often the rate-determining step in the chemical loss of many species emitted into the atmosphere, such as volatile organic compounds (VOCs), these radicals dictate the atmospheric lifetimes and concentrations of these other species (Atkinson and Arey, 2003). Products of these reactions include a variety of harmful secondary pollutants like tropospheric ozone (O_3), and atmospheric reservoirs of NO_2
 35 such as peroxyacetyl nitrate (PAN), that contribute to long-range and regional pollution (Spataro and Ianniello, 2014). Conversely, the high reactivity of these radicals means they have very short atmospheric lifetimes (e.g. <1 s for OH; Monks, 2005), so their concentrations and potential oxidation impacts are only relevant on a local scale.

The OH radical is the most dominant radical diurnally, known to be produced through the photolysis of ozone and subsequent reaction with H_2O (Reactions R3 and R4).



Photolysis of oxidised reactive nitrogen species (NO_y = sum of NO_x , HONO, HNO_3 , $2\text{N}_2\text{O}_5$, ClNO_2 and organic gases with NO_x groups) also play an important role in radical formation. Nitrous acid (HONO) has been attributed as a large source of OH radicals in multiple locations across the globe (Reaction R5; Alicke et al., 2003; Elshorbany et al., 2009; Villena et al.,
 45 2011; Ryan et al., 2018), while nitryl chloride (ClNO_2) is the dominant early morning source of chlorine atoms (Reaction R6; Young et al., 2012, 2014; Riedel et al., 2014).



Production of Cl radicals from measured levels of ClNO_2 were first estimated in coastal locations near Houston, Texas
 50 (Osthoff et al., 2008), where an abundance of aerosol chloride in the form of sea salt coexists with nitrogen oxides (NO_x). However, formation of ClNO_2 is not limited to chloride in sea spray. Its presence has been repeatedly measured at sites up to 1400 km from the nearest coastline (Thornton et al., 2010; Bannan et al., 2015; Osthoff et al., 2018; Sommariva et al., 2018), with the particulate chloride precursor generated from anthropogenic sources such as fossil fuel combustion and biomass burning (Ahern et al., 2018).



55 Cl radicals are extremely reactive with VOCs, with rate constants often an order of magnitude greater than equivalent reactions with OH (Monks, 2005; Young et al., 2012). Therefore, even at low concentrations, Cl has a significant effect on local tropospheric oxidation (Riedel et al., 2014; Bannan et al., 2015). As a result, quantifying local photolysis rates is as critical to understanding the local air quality as measuring trace gas concentrations.

Photolysis is a unimolecular reaction characterised by a simple rate equation, demonstrated by Eq. (1) for the photolysis of
 60 NO₂ in Reaction (R1).

$$-\frac{d[\text{NO}_2]}{dt} = j(\text{NO}_2) \times [\text{NO}_2] \quad (1)$$

The rate constant for this photolysis reaction, $j(\text{NO}_2)$, can be measured in absolute terms using a chemical actinometer. This estimates $j(\text{NO}_2)$ by measuring the concentration of NO following photodissociation of a known concentration of NO₂ inside a reaction chamber exposed to ambient light (Bahe et al., 1980; Shetter et al., 1992; Lantz et al., 1996; Bohn et al.,
 65 2005). Spectroradiometers measure actinic flux independent of the angle of incidence in a 2- π sr field of view, as a function of wavelength (Kraus and Hofzumahaus, 1998). Photolysis frequencies are determined from Eq. (2) for any molecule whose absorption cross-section (σ) falls within the measured wavelength (λ) range (Shetter and Müller, 1999). For instruments at ground level, the lower-bound wavelength limit (λ_1) is typically set to 290 nm, while the upper-bound is dependent on the molecule (e.g. $\lambda_2 = 420$ nm for $j(\text{NO}_2)$, and $\lambda_2 = 340$ nm for $j(\text{O}^1\text{D})$; Edwards and Monks, 2003).

$$70 \quad j(\text{NO}_2) = \int_{\lambda_1}^{\lambda_2} \sigma(\lambda, T) \phi(\lambda, T) F(\lambda) d\lambda \quad (2)$$

The actinic flux (F) is the quantity of light available to molecules at a given point in the atmosphere, that upon absorption results in photodissociation (Kylling et al., 2003). It is a spherical integration of solar radiant energy over a surface plane, differing by a cosine of the angle of incidence (Madronich, 1987). The absorption cross-section (σ) quantifies the ability of the molecule to absorb radiation, and the quantum yield (ϕ) is the probability that the photodissociation leads to the product channel
 75 under consideration, in this case O(³P) and NO. These are molecule-specific parameters that vary with respect to wavelength and temperature, the latter of which is not often assessed for molecules. Hence, uncertainty in these values propagates through to uncertainty in calculated photolysis frequencies (Shetter et al., 1996; Kraus and Hofzumahaus, 1998).

Filter radiometers use band-pass filters to measure broadband actinic flux, between $\lambda_2 - \lambda_1$ in Eq. (2), designed to simulate the $\sigma \times \phi$ product of the molecule of interest at its strongest absorbing region. A 4- π sr filter radiometer provides a 360° field of
 80 view of the surrounding environment by utilising two identical optical inlets, designed by Junkermann et al. (1989) and modified by Volz-Thomas et al. (1996). The two 2- π sr domes face in opposite directions, separating the total measured actinic flux into down- and up-welling components ($F \downarrow$ and $F \uparrow$, respectively). These instruments exhibit good linear detector responses, and long-term stability in calibration factors (Bohn et al., 2008), and measurements can be recorded with high time resolution (1 s). These instruments are also easy to deploy and maintain, making them excellent candidates for routine measurements.
 85 However they remain reliant on absolute calibrations to quantify $j(\text{NO}_2)$ from recorded voltages, and measurements are only applicable for specified reactions, with limited potential to estimate the photolysis frequencies of other atmospheric species.



Solar global irradiance (G) measurements are more widespread than $j(\text{NO}_2)$, as this is a common meteorological parameter included in multiple global monitoring networks, such as the World Meteorological Organisation (WMO) Global Atmospheric Watch (GAW) programme. Extensive efforts have been made to derive non-linear parameterisations that utilise existing measurements of G to quantify downwelling $j(\text{NO}_2)$, thereby improving spatial coverage of j -value estimations (Bahe et al., 1980; Brauers and Hofzumahaus, 1992; Webb et al., 2002; Trebs et al., 2009). G is typically measured using a pyranometer, which has a horizontal sensor measuring total downwelling radiation (direct + diffuse), weighted by a cosine response subject to the angle of incident light (Webb et al., 2002). Trebs et al. (2009) extensively discuss this relationship, and present good agreement between co-located G and $j(\text{NO}_2)\downarrow$ measurements for 7 ground-based field sites up to 800 m above sea level despite measurements being from different years, seasons and continents. The authors acknowledge that this parameterisation is not a substitute for measurements of $j(\text{NO}_2)$, but a viable alternative to radiation transfer calculations where input parameters (like cloud cover) are inadequately known.

Most atmospheric and radiative transfer models use Eq. (2) to estimate j -values, but often oversimplify local meteorological conditions due to the computational cost of their inclusion, or the absence of necessary measured input parameters (Shetter et al., 2003). A myriad of variables impact the photolysis frequency, such as cloud cover, aerosol optical depth (AOD; Wild et al., 2000), surface albedo and ozone column. All of these variables can redirect incident radiation by absorption and/or reflection. Palancar et al. (2013) demonstrate that when measured AOD and NO_2 concentrations are included in model inputs, actinic flux is reasonably well predicted and the main source of uncertainty is then attributable to σ and ϕ .

This study presents the first long-term filter radiometer dataset in the UK (Auchencorth Moss, SE Scotland) co-located with relevant ancillary measurements to quantify radical production rates from photolysis pathways. It aims to assess the temporal distribution of local OH and Cl radical formation at a single site, using the filter radiometer measurements applied to clear-sky model outputs to account for changes in local meteorological variables. The inclusion of local values of such variables would otherwise considerably increase model computational cost. The dataset provides the first accurate long-term UK point of comparison for validation of model photolysis coefficients. The scientific, and ultimately policy, value of incorporating filter radiometer measurements alongside existing long-term and spatially-resolved air quality monitoring networks is emphasised.

2 Method

2.1 Site description

Auchencorth Moss is a rural air quality monitoring station located in south-east Scotland, approximately 18 km SSW of Edinburgh (lat: $55^\circ 47' 36''$ N, lon: $3^\circ 14' 41''$ W, altitude: 260 m). The site operates as a Level II/III European Monitoring and Evaluation Programme (EMEP) "supersite", representative of the northern UK background atmosphere, at which atmospheric components are measured using standard methodology (Tørseth et al., 2012). Collectively, EMEP monitoring sites are used to understand long-range transport of air pollution around Europe, verify regional modelling approaches and advise governmental bodies on concentration and deposition of pollutants (UNECE, 2004; Aas et al., 2012).



Auchencorth Moss is also a site for a number of national and local monitoring networks (see Twigg et al., 2015), hosting an extensive array of instrumentation to measure trace species concentrations and meteorological parameters (Coyle et al., 2019). Long-term instrumentation deployed at Auchencorth Moss utilised in this study is detailed in Table 1. It is acknowledged that both the HONO and HNO₃ concentrations reported by the Monitor for AeRosols and Gases in Ambient Air (MARGA; Metrohm Applikon, NL) have potential interferences from other NO_y species. These interferences have been shown to result in both over- (Phillips et al., 2013) and underestimates (Rumsey et al., 2014) in reported concentration. Since these interferences have not yet been formally quantified, this study uses the reported values as they are with caution.

2.2 Filter radiometer

The 4- π filter radiometer (Metcon, Meteorologie Consult GmbH, DE) was mounted ~ 3 m above the ground, recording measurements at 1 s time resolution for a full year (21 November 2018–20 November 2019). The inlet optic of each dome is designed to have a near-uniform angular response through use of a quartz diffusor (Bohn et al., 2004). Each optical inlet is surrounded by a light shield to provide an "artificial horizon", restricting the field-of-view for each dome to one hemisphere (Volz-Thomas et al., 1996) and preventing overlap. Transmitted light is guided through a set of optical filters (2-mm UG3, 1-mm UG5, Schott GmbH) that restrict transmitted photons to wavelengths of interest prior to their detection by a Hamamatsu photodiode, which proportionally converts incident radiation into an output voltage.

Output signals from the filter radiometer were calibrated against a Bentham DTM300 scanning spectroradiometer between 13 and 25 June 2019. This mid-summer period was selected to provide calibration over the maximum range of ambient incident radiation. The direction of the filter radiometer was turned 180° mid-way through this period, in order to calibrate each dome separately. $j(\text{NO}_2)$ was calculated from the actinic flux spectra measured by the spectroradiometer using $\sigma(\text{NO}_2)$ from Mérienne et al. (1995) and $\phi(\text{O}^3\text{P})$ from Troe (2000). Calibration factors for each dome were quantified using data presented in Fig. 1.

The limits of detection (LOD; 3σ of background signal) were $9.40 \times 10^{-6} \text{ s}^{-1}$ and $1.15 \times 10^{-5} \text{ s}^{-1}$ for the down- and upwelling domes, respectively. Background signals were determined from averaged measurements made after sunset and before sunrise (solar zenith angle (SZA) $\geq 96^\circ$) and removed prior to data analysis. On a few occasions of peak solar irradiance (noon in summer), recorded voltage of the downwelling dome exceeded the range of the detector and was reported as ~ 10 V, observed during the calibration in Fig. 1. These incidences of high $j(\text{NO}_2)\downarrow$ (< 9 V) comprised 1.4% of all data collected, and were removed from the dataset before further analysis. As a consequence, maximum $j(\text{NO}_2)\downarrow$ presented in this study is an underestimate (7 – 22% based on calibration). During the calibration, measurements made by the filter radiometer were averaged to match the duration of each spectroradiometer scan (3 mins). The standard deviation associated with each of the averaged filter radiometer measurements were used to remove calibration data points where actinic flux was highly variable, and not comparable between the instruments during the 3 min scan.

The uncertainty of the filter radiometer measurements were estimated as a combination of instrumental error (e.g. non-ideal inlet optic responses; Larason and Cromer, 2001), error from calibration, and errors due to external factors (e.g. temperature stability). For six of the same filter radiometers, Shetter et al. (2003) quoted overall error as 9.6 – 11%, the range dependent



on whether conditions were clear or cloudy. To provide a conservative estimate of error for both domes of the filter radiometer used here, the upper bound (11%; cloudy conditions) is combined with the calibration error of each dome. Overall errors for
 155 down- and upwelling domes were 13% and 12%, respectively.

2.3 Model approach

Calculations of photolysis frequencies for clear-sky conditions were performed using the Tropospheric Ultraviolet and Visible radiation model (TUV v5.3; Madronich, 1987; NCAR, 2019), one of the most widely used radiation models (Lee-Taylor and Madronich, 2002; Bais et al., 2003; Wilson, 2015; Ghosh and Sarkar, 2016; Wang et al., 2017; Xu et al., 2019). Photolysis
 160 frequencies (j -values) were calculated for the photodissociation of NO_2 , O_3 , HONO , HNO_3 and ClNO_2 . The model was set up for the location and height of the filter radiometer at Auchencorth Moss. Variables were set to assume clear-sky conditions, and those not influenced by cloud cover were kept constant for all calculations. Ozone column was estimated as the mean of daily Dobson spectrophotometer measurements at Lerwick for the study duration (Defra, 2020b), while default TUV values for surface albedo and AOD at 500 nm were used.

165 The model output included both actinic flux by wavelength (290 – 420 nm), and the j -values calculated for each hour of the day. The latter were used to determine an hourly measurement-driven adjustment factor (MDAF), shown in Eq. (3), while the former was used to determine $j(\text{ClNO}_2)$ using Eq. (2) and updated $\sigma(\text{ClNO}_2)$ measurements made by Ghosh et al. (2012), now the preferred IUPAC values (as of June 2013; Atkinson et al., 2007). It should be noted that for $j(\text{NO}_2)$ calculations, the molecule-specific parameters used by the TUV model are the same as those used in the calibration of the filter radiometer.

170 2.4 Theoretical calculations

2.4.1 Adjustment factor

Measurements of actinic flux made using radiometers intrinsically capture variations in solar flux caused by local meteorology and other factors, such as the presence of aerosols. Since it is computationally intensive to replicate these variations in radiation, which cannot be well represented if input parameters are poorly known, it has been demonstrated that radiometer measurements
 175 could be used to adjust j -values in chemical transport models (Elshorbany et al., 2012; Sommariva et al., 2020). Equation (3) illustrates how this can be done by defining a measurement-driven adjustment factor (MDAF), using $j(\text{NO}_2)$ as a reference.

$$MDAF = \frac{\text{measured } j(\text{NO}_2)}{\text{modelled } j(\text{NO}_2)} \quad (3)$$

Correction factors like the MDAF have only occasionally been applied both spatially and temporally (Stone et al., 2012; Bannan et al., 2015), with measured and modelled environments often not co-located (e.g. Sommariva et al., 2020). The
 180 MDAF metric, or equivalent, can only be used for model validation where radiometer measurements to complement the model study can be obtained.



2.4.2 Production rate of OH radicals

The application of a locally-derived MDAF to model-calculated j -values is demonstrated here for calculations of the rate of production of OH radicals at Auchencorth Moss from O_3 and two NO_y species, HONO and HNO_3 . For both the latter, OH radicals are produced by a direct photodissociation as shown in Reaction (R5) for the case of HONO. The rates are calculated using Eq. (4), and an equivalent for the photolysis of HNO_3 .

$$p(OH)_{HONO} = j(HONO) \times [HONO] \quad (4)$$

These are compared with the production rate from O_3 photolysis (Reactions R3 and R4) as calculated by Eq. (5) and (6).

$$p(OH)_{O_3} = 2 \times f \times j(O^1D) \times [O_3] \quad (5)$$

$$f = \frac{k_{H_2O}[H_2O]}{k_{N_2}[N_2] + k_{O_2}[O_2] + k_{H_2O}[H_2O]} \quad (6)$$

In these equations, $j(O^1D)$ is the photolysis rate constant for O_3 , and f is the fraction of $O(^1D)$ atoms that react with water vapour to form OH, as opposed to their quenched removal by N_2 and O_2 molecules. Local relative humidity and temperature measurements (Coyle et al., 2020) were used to derive absolute H_2O concentrations, and hourly concentrations of O_3 , HONO and HNO_3 measured at Auchencorth Moss were downloaded directly from UK-AIR (Defra, 2020a). For the study period at Auchencorth Moss, the fraction f was $6.1 \pm 2\%$. It should be noted that while O_3 is reasonably well-mixed throughout the boundary layer, HONO has been reported to have a clear vertical profile where concentrations are strongly enhanced at ground level (Kleffmann et al., 2003; Young et al., 2012; Ryan et al., 2018). As HONO is measured at a height of 3.55 m at Auchencorth Moss (Twigg et al., 2015), $p(OH)_{HONO}$ presented here are applicable for ground level, and would be an overestimate for higher altitudes (>1 km).

2.4.3 Production rate of Cl atom radicals

The MDAF is also demonstrated for calculation of the production rate of Cl atoms from the photodissociation of $ClNO_2$, shown in Eq. (7), to the products in Reaction (R6).

$$p(Cl)_{ClNO_2} = j(ClNO_2) \times [ClNO_2] \quad (7)$$

Values of $\sigma(ClNO_2)$ used to determine $j(ClNO_2)$ were compared between those used in the TUV model (Sander et al., 2006) and the more recent IUPAC preferred values from Ghosh et al. (2012). Since the dataset presented by Ghosh et al. includes the temperature dependence of $\sigma(ClNO_2)$, values in this study were adjusted according to the daily mean temperature measured at Auchencorth Moss prior to further calculation. Hourly $j(ClNO_2)$ was then determined using Eq. (2) from the temperature adjusted $\sigma(ClNO_2)$ and hourly actinic flux from the TUV model. Henceforth, $j(ClNO_2)$ values calculated in this way are referred to as "updated $j(ClNO_2)$ ". The MDAF was applied to both the TUV and updated $j(ClNO_2)$ values, and the production rate of Cl radicals ($p(Cl)_{ClNO_2}$) using all four $j(ClNO_2)$ estimates are evaluated.

$ClNO_2$ is not routinely measured at Auchencorth Moss, so an estimate for its concentrations at this rural site was obtained from the literature. Sommariva et al. (2018) reported a median diurnal concentration of 21.5 ppt at Weybourne Atmospheric



Observatory, UK ($52^{\circ}47'$ N, lon: $1^{\circ}07'$ E) during the summer of 2015. This is a rural site on the coast of East England, whose concentrations of ClNO_2 are the best available measured proxy for Auchencorth Moss values. In the calculations of $p(\text{Cl})_{\text{ClNO}_2}$ presented here, 21.5 ppt ($0.074 \mu\text{g m}^{-3}$, at Auchencorth Moss annual average T and p) is used as both a constant concentration, and as the mean concentration of ClNO_2 for an approximate diurnal cycle that might be expected at Auchencorth Moss (Fig. 2). The shape of the diurnal profile follows that reported by Sommariva et al. (2018) at Weybourne.

3 Results and discussion

3.1 Concentrations of trace gases and meteorology

Figure 3 shows the time series of O_3 , HONO and HNO_3 concentrations, total solar global irradiance (G), temperature, relative humidity, wind speed and total (downwelling + upwelling) $j(\text{NO}_2)$ for the year of measurements. Summary statistics of measured trace gas concentrations during each season are presented in Table 2, and diurnal cycles of O_3 and HONO for each season are shown in Fig. 4. UK seasons are defined by meteorological convention: winter (December to February), spring (March to May), summer (June to August) and autumn (September to November). All times are expressed in UTC.

Seasonal median O_3 concentrations ranged from $51.2 \mu\text{g m}^{-3}$ in autumn, to $74.2 \mu\text{g m}^{-3}$ in spring. The O_3 diurnal cycle is generally weak but most evident in spring and summer, where concentrations increase after sunrise and peak mid-afternoon at 15:00. Diurnal variation is also observed in both autumn and winter, but is far less pronounced. This is in part, due to high wind speeds at this rural site ($0.26 - 18.8 \text{ m s}^{-1}$, median of 5.0 m s^{-1}) which enhances vertical mixing. This is particularly evident at night where the boundary layer is generally shallower. The largest O_3 peaks are observed in April, May and August, coincident with the spring-time annual maximum in O_3 in the UK and increases in ambient temperature.

Median HONO concentrations show far less diurnal and seasonal variation, but a greater interquartile range of concentrations for each hour in the summer. Larger concentration episodes were observed in the spring and summer months, particularly July, reflected by the largest seasonal mean and median concentrations occurring in summer (0.065 and $0.096 \mu\text{g m}^{-3}$, respectively). Mean and median HNO_3 concentrations were largest in the spring (0.16 and $0.10 \mu\text{g m}^{-3}$, respectively), due to a large peak in mid-April reaching a maximum of $1.45 \mu\text{g m}^{-3}$.

3.2 Filter radiometer measurements

A direct comparison between the co-located downwelling filter radiometer and G measurements for the duration of the study is presented in Fig. 5. The relationship between the two measurements is linear until $G \approx 500 \text{ W m}^{-2}$, above which a slight curvature is observed. A quadratic fit to this data yields predicted values of $j(\text{NO}_2)$ that lie between two previously published parameterisations for this relationship. Predicted $j(\text{NO}_2)$ values in this study are lower (for $G > 150 \text{ W m}^{-2}$) than those predicted by the expression of Trebs et al. (2009) shown in Eq. (8), derived from spectral and filter radiometer measurements co-located with G (in W m^{-2}) at 9 sites worldwide. The α factor is the site-dependent UV-A surface albedo, which here is assumed to be zero as only downwelling radiation is compared. Trebs et al. show differences of up to 50% between their



relationship and Eq. (9), a linear relationship proposed by Bahe et al. (1980) with units of $j(\text{NO}_2)$ and G as min^{-1} and
 245 $\text{cal cm}^{-2} \text{min}^{-1}$, respectively.

$$j(\text{NO}_2) = (1 + \alpha) \times (1.47 \times 10^{-5} \times G - 4.84 \times 10^{-9} \times G^2) \quad (8)$$

$$j(\text{NO}_2) = 0.008 + 0.371G \quad (9)$$

While the maximum difference between this study and Trebs et al. is smaller ($\sim 19\%$), predicted values are still larger than those Bahe et al. reported. Trebs et al. (2009) suggest that one possible reason for the lack of curvature observed by Bahe et al.
 250 was that they did not include measurements at SZAs $< 30^\circ$. However Auchencorth Moss is at a higher latitude than any of these sites, with a minimum SZA of $\sim 32^\circ$ in June. Curvature of the plot of $j(\text{NO}_2)$ against G is still observed, although there is still a moderate amount of scatter.

The time series of measured and modelled $j(\text{NO}_2)$ derived from this study is presented in Fig. 6. In general, model results are larger than the measured counterpart, which is expected as meteorological conditions at Auchencorth Moss are typically
 255 overcast. One clear discrepancy with this generalisation occurs between 30 January and 2 February 2019. This peak in measured data is not reflected in the solar irradiance time series in Fig. 3 as it was largely caused by snow cover at Auchencorth Moss substantially increasing the surface albedo.

The MDAF values calculated as per Eq. (3) from the measured and modelled $j(\text{NO}_2)$ was largest during the sunrise and sunset hours, where SZA exceeded 80° , due to the model predicting very small values of $j(\text{NO}_2)$. In general, for the rest of
 260 the year the adjustment factor was greatest in the morning and steadily decreased throughout the day until sunset. The range of values was greatest in the winter months, and the smallest in the summer.

3.3 Estimates of $j(\text{HONO})$ and $j(\text{ClNO}_2)$

Parameterisations of $j(\text{HONO})$ and $j(\text{ClNO}_2)$ based on other more-commonly available variables have been proposed before from both measurement intercomparisons and modelling studies. For example, Kraus and Hofzumahaus (1998) noted that
 265 molecules which photodissociate in a similar spectral region display a higher correlation between rate coefficients. The empirical parameterisation that they derived from spectroradiometer measurements to calculate $j(\text{HONO})$ from measured $j(\text{NO}_2)$ (Eq. (10)) has been implemented in further studies requiring $j(\text{HONO})$ estimates (Alicke et al., 2003; Kleffmann et al., 2005; Acker and Möller, 2007).

$$j(\text{HONO}) = 0.189 \times j(\text{NO}_2) + 8.482 \times 10^{-2} \times [j(\text{NO}_2)]^2 \quad (10)$$

270 A comparison between annual mean diurnal cycles of $j(\text{HONO})$ at Auchencorth Moss parameterised using Eq. (10) and calculated using the TUV model output are presented in the top panel of Fig. 7. Both estimates of $j(\text{HONO})$ have been adjusted using the MDAF metric. Over the entire year, this shows that Eq. (10) is approximately 20% greater than calculated by the TUV model, a result which is consistent over all seasons.

Multiple estimates of $j(\text{ClNO}_2)$ have been proposed, including two studies that used parameterisations to update the Master
 275 Chemical Mechanism (MCM; <http://mcm.leeds.ac.uk/MCM>). Riedel et al. (2014) used measurements made on a research



vessel near California during the CalNex campaign to provide evidence for scaling down $j(\text{NO}_2)$ measurements by a factor of 30, while Young et al. (2014) used a linear combination of measured $j(\text{NO}_2)$ and $j(\text{O}^1\text{D})$, as well as $\sigma(\text{ClNO}_2)$ from Ghosh et al. (2012). The MDAF was applied to both of these parameterisations of $j(\text{ClNO}_2)$, as well as those developed in this study (TUV model output and updated $j(\text{ClNO}_2)$ values; Sect. 2.4.3). The results are shown in Fig. 7.

Overall, the TUV model yields the greatest annual mean $j(\text{ClNO}_2)$ values (Fig. 7, bottom panel), exceeding the others by 11–26%, with the maximum difference at noon. While the parameterisations from Young et al. (2014) and Riedel et al. (2014) demonstrate reasonable agreement to the updated $j(\text{ClNO}_2)$ results, they are closest at different periods during the diurnal cycle. The Riedel et al. parameterisation matches best during peak solar hours (11:00–14:00 UTC), but overestimate by $\sim 15\%$ in the morning (06:00–09:00 UTC) and early evening (17:00–18:00 UTC). In contrast, results from the Young et al. (2014) parameterisation reveal a closer match to updated $j(\text{ClNO}_2)$ during the non-peak solar hours, and underestimate by $\sim 5\%$ at noon.

3.4 Production rate of OH radicals

Average seasonal production rates of OH radicals ($p(\text{OH})$) from the photodissociations of O_3 , HONO and HNO_3 are presented in Table 3, calculated using j -values directly from TUV model output, and with the MDAF applied. In general, the application of the MDAF metric results in $\sim 40\%$ decrease in $p(\text{OH})$ from all considered sources at Auchencorth Moss. In all seasons, $p(\text{OH})_{\text{HNO}_3}$ is negligible compared to $p(\text{OH})_{\text{O}_3}$ and $p(\text{OH})_{\text{HONO}}$, reaching a maximum of about $4.0 \times 10^3 \text{ radicals cm}^{-3} \text{ s}^{-1}$ in April, coincident with near-maximum concentrations of HNO_3 ($1.3 \mu\text{g m}^{-3}$) at the site for the year.

The diurnal variations of these seasonal averages are shown in Fig. 8, where the $\sim 40\%$ reduction due to the MDAF application is clear in daylight hours. The maximum MDAF decrease in hourly measurements is 71%, corresponding to summertime O_3 photolysis, resulting in an adjusted $p(\text{OH})_{\text{O}_3}$ of $1.90 \times 10^6 \text{ radicals cm}^{-3} \text{ s}^{-1}$. In comparison, $p(\text{OH})_{\text{HONO}}$ in the same month reaches a maximum of $1.03 \times 10^6 \text{ radicals cm}^{-3} \text{ s}^{-1}$ at 09:00 (UTC; ~ 5 hours after sunrise). Conversely, over the daylight hours in winter, $p(\text{OH})_{\text{HONO}}$ consistently exceeds $p(\text{OH})_{\text{O}_3}$ by a factor of ~ 5 . A similar pattern is observed in autumn, but at a lower magnitude (factor of ~ 2.1). This is likely to be a consequence of the generally overcast conditions typical to Scotland, resulting in the shorter wavelengths of light necessary for O_3 photolysis ($\leq 340 \text{ nm}$) readily scattering before reaching ground-level, *c.f.* longer wavelengths for HONO photolysis ($305 \leq \lambda \leq 420 \text{ nm}$). This is also observed in spring and summer (Fig. 8), where $p(\text{OH})_{\text{HONO}}$ is greater in the early morning ($\sim 04:00$ – $09:00$ UTC) and the evening ($\sim 16:00$ – $20:00$ UTC). This is surpassed by $p(\text{OH})_{\text{O}_3}$ in the middle of the day, closely following the diurnal cycle of light intensity and peaking when shorter wavelengths are more prevalent. The same diurnal pattern of $p(\text{OH})_{\text{O}_3}$ is observed in all seasons (Fig. 9), but a difference of ~ 25 is observed between peak $p(\text{OH})_{\text{O}_3}$ in summer and winter ($1.90 \times 10^6 \text{ radicals cm}^{-3} \text{ s}^{-1}$ and $7.4 \times 10^4 \text{ radicals cm}^{-3} \text{ s}^{-1}$, respectively) despite similar concentrations of O_3 ($61.0 \mu\text{g m}^{-3}$ and $57.1 \mu\text{g m}^{-3}$, respectively).

The photolytic production of OH radicals presented here at Auchencorth Moss is lower than those reported in Melbourne, Australia for March 2017 (Ryan et al., 2018). Compared to the summertime $p(\text{OH})$ rates presented here, peak values of $p(\text{OH})_{\text{HONO}}$ were approximately 13 times larger in Australia, while that of $p(\text{OH})_{\text{O}_3}$ were just over 50% greater. At both locations, $p(\text{OH})_{\text{O}_3}$ peaked at a similar time of day (12:00–14:00 local time), while $p(\text{OH})_{\text{HONO}}$ reached a diurnal maximum



much earlier at Auchencorth Moss (09:00 UTC *c.f.* 13:00). The latter is likely due to the application of the MDAF metric, as without the adjustment factor the maximum production rate occurs at noon. Similarly when the MDAF is not applied in calculations of $p(\text{OH})_{\text{O}_3}$, values are comparable at both sites due to equivalent concentrations of O_3 ($\sim 54 \mu\text{g m}^{-3}$ average in Australia). The same is not observed in $p(\text{OH})_{\text{HONO}}$, as ~ 5 times larger concentrations of HONO in Australia are a considerable driver towards ~ 8 times larger $p(\text{OH})_{\text{HONO}}$ rates.

3.5 Production rate of Cl atom radicals

Figure 10 presents simulated seasonally-averaged diurnal production rates of Cl radicals, $p(\text{Cl})$, from ClNO_2 photolysis. Plots in the top and bottom panels respectively assume constant concentrations of ClNO_2 ($0.074 \mu\text{g m}^{-3}$), and the diurnally varying ClNO_2 concentration profile shown in Fig. 2. When using the former to estimate $p(\text{Cl})$, those calculated using $j(\text{ClNO}_2)$ directly from TUV model output provide the largest estimation for all seasons. When the updated $j(\text{ClNO}_2)$ values accounting for ambient temperature are used (by means of updated $\sigma(\text{ClNO}_2)$; Ghosh et al., 2012), total $p(\text{Cl})$ decreased by $\sim 30\%$ compared to the TUV output. As with calculations of $p(\text{OH})$, application of the MDAF metric resulted in $\sim 40\%$ reduction of $p(\text{Cl})$ when using both TUV and updated $j(\text{ClNO}_2)$. An exception is in winter, where during the lowest rates of Cl production, the adjustment factor and updated $j(\text{ClNO}_2)$ decrease $p(\text{Cl})$ calculated from the TUV output to an approximately equal quantity in the morning hours. This could be due to cooler temperatures in winter ($3.99 \pm 3.6^\circ\text{C}$) leading to decreased $\sigma(\text{ClNO}_2)$ *c.f.* other seasons ($6.76 \pm 4.3^\circ\text{C}$, $13.5 \pm 3.7^\circ\text{C}$ and $7.47 \pm 4.3^\circ\text{C}$ in spring, summer and autumn, respectively), so that it becomes comparable to TUV results with the MDAF applied.

When the assumed diurnal cycle of ClNO_2 concentrations are used to derive $p(\text{Cl})$ (bottom panels of Fig. 10), the maximum rate of Cl radical production decreases by 59% to $7.98 \times 10^4 \text{ radicals cm}^{-3} \text{ s}^{-1}$ compared with when the diurnal cycle of ClNO_2 is unaccounted for. When ClNO_2 concentrations drop close to zero (first occurring between 10:00-11:00 UTC), there is obviously little photolytic production of Cl. Early morning peaks in $p(\text{Cl})$ in all seasons are staggered, due to the anti-correlated relationship between falling ClNO_2 concentrations and increasing actinic flux with the rising sun at this time of day. In spring and summer evenings from 18:00 onwards, non-zero $p(\text{Cl})$ rates are observed, suggesting that as ClNO_2 begins to form around sunset, the remaining light ($\lambda \leq 475 \text{ nm}$) quickly results in its photolysis and consequently produces Cl atoms. However, caution should be exercised here as the time of sunset differs between Auchencorth Moss and Weybourne Atmospheric Observatory by around 15-30 minutes (from the start of spring to the summer solstice). This would impact both the photolysis and formation of ClNO_2 , anticipated to lead to ClNO_2 forming later in the diurnal cycle than is estimated in Fig. 2, and thus fewer molecules to undergo photolysis. This discrepancy may mean this apparent late evening $p(\text{Cl})$ is an artefact of the proxy ClNO_2 diurnal cycle.

Similar differences in the magnitude of $p(\text{Cl})$ across the seasons are observed for the $j(\text{ClNO}_2)$ and updated values used in calculation. However, when the MDAF metric is applied, morning hours have a faster rate of Cl production, particularly in spring and summer. Compared to $p(\text{OH})$ with the MDAF applied, the average maximum $p(\text{OH})$ exceed equivalent $p(\text{Cl})$ by 20 – 25 times when using $j(\text{ClNO}_2)$ output from the TUV model, and 30 – 35 times when using the updated $j(\text{ClNO}_2)$ values. When considering the diurnal profile of ClNO_2 concentrations, the maximum values of $p(\text{Cl})$ occur earlier in the day ($\sim 06:00$



UTC in spring and summer), than $p(\text{OH})_{\text{HONO}}$ ($\sim 09:00\text{--}12:00$ UTC). This early maximum in $p(\text{Cl})$ is likely driven by the
345 diurnal cycle of ClNO_2 concentrations in Fig. 2, where concentrations fall after 06:00 UTC. As rate coefficients for reactions
between Cl and VOCs can be up to an order of magnitude larger than comparable rate coefficients with OH, Cl atoms have
potential significant impacts on the radical budget at Auchencorth Moss.

3.6 Implications of the MDAF metric

Previous sections describe how application of the MDAF metric at Auchencorth Moss leads to a $\sim 40\%$ reduction in the
350 $j(\text{HONO})$, $j(\text{ClNO}_2)$ and $j(\text{O}^1\text{D})$ clear-sky values calculated by the TUV model, and consequently also in the radical produc-
tion rates from these photolyses. In contrast, Sommariva et al. (2020) report measurements of $j(\text{NO}_2)$ and $j(\text{O}^1\text{D})$ made in
Boulder, Colorado that were 25 – 30% greater than model estimates using the Master Chemical Mechanism (MCM) param-
eterisations. The difference was attributed to discrepancies between the measurement site and model output location (the latter
is > 1 km lower in altitude, and further north by a latitude of 5°) and to temporal variation in measurements (different seasons
355 and years). Taken together, however, these authors' findings combined with this study highlight the importance of matched
measurement and model location to generate a MDAF to derive molecule specific j -values.

Most local air quality modelling, e.g. for assessment of compliance and mitigation measures, include only simple chemical
schemes with highly parameterised photochemistry (e.g. ADMS). In contrast to model outputted gas and particle concentra-
tions, the values of the model photochemical variables often go unverified. In more detailed instances, atmospheric models
360 typically use separate radiation models, such as the TUV model used in this study, Fast-JX, or less explicit photolysis schemes
like the Generic Reaction Set (GRS), to approximate j -values within the model domain. Radiometer measurements can be used
to validate these model predictions of individual j -values. However as these measurements are not typically available, except
as part of specific measurement campaigns, modelled j -values are often used unverified.

The implication of the MDAF approach illustrated in this study is that long-term multi-site radiometer measurements could
365 fill a major gap in measurement and model knowledge. Implementation of such measurements into existing atmospheric trace
gas monitoring networks (e.g. AURN in the UK or EMEP in Europe), would provide invaluable data for understanding tro-
pospheric photochemistry and radical production rates. Accurate j -values are integral to accurate assessment of air quality, in
particular, to photolytic production of secondary pollutants which negatively impact both human and environmental health, in-
cluding tropospheric O_3 and particulate matter. Radiometer measurements will consequently have immense value in supporting
370 atmospheric chemistry measurement and modelling.

4 Conclusions

This study presents the first year of a long-term $4\text{-}\pi$ filter radiometer $j(\text{NO}_2)$ measurement dataset at Auchencorth Moss, co-
located with relevant ancillary measurements required to determine local radical production rates. Through illustration with the
TUV model, it is demonstrated that long-term filter radiometer measurements can be used (1) to evaluate calculated j -values
375 in current radiation models and parameterisations; and (2) to generate a straightforward measurement-driven adjustment factor



(MDAF) for correcting clear-sky modelled j -values for variations in local meteorology (e.g. cloud, aerosol and surface albedo) that would otherwise be both computationally intensive and costly to model.

When the $4\text{-}\pi$ filter radiometer measurement is used to capture these changes in solar flux through the MDAF at Auchencorth Moss, it was found that clear-sky modelled j -values at Auchencorth Moss were reduced by $\sim 40\%$ throughout the year.

380 Quantified production rates of OH radicals in this study differ by approximately one order of magnitude between summer and winter, with the maximum rate in summer reaching $1.90 \times 10^6 \text{ radicals cm}^{-3} \text{ s}^{-1}$ (Auchencorth Moss; 55° N). At this rural background site, rate of OH radical production from photolysis of HONO exceeds that from photolysis of O_3 during low-light hours, particularly during the sunrise and sunset hours of spring and summer ($\text{SZA} > 80^\circ$), and through the full diurnal cycle in winter and autumn (by a factor of 5 and 2.1, respectively). The enhanced contribution from HONO in the colder months is

385 likely due to Scotland's considerably more overcast conditions in these seasons reducing the shorter wavelengths contributing to O_3 photolysis relatively more than the longer wavelengths contributing to HONO photolysis.

Implementing radiometer measurements within existing long-term monitoring networks would provide a higher spatial density of model-measurement j -value comparison points in the UK. The consequent more accurate estimations of radical production rates would improve the quantification of subsequent radical-driven chemistry and secondary pollutant generation,

390 and of all their associated impacts on human health, crops, ecosystems and radiative forcing.

Data availability. Filter radiometer measurements for the study year are available in the CEDA Archive (Walker et al., 2020). Concentrations of O_3 , HONO and HNO_3 used were downloaded from UK-AIR (Defra, 2020a). Meteorological measurements used were provided by M. Coyle *et al.* and are available for previous years in the CEDA Archive (Coyle et al., 2020). The TUV model (NCAR, 2019) source code is available for download at: <https://www2.acom.ucar.edu/modeling/tuv-download>. All figures were created using the *ggplot2* package for R

395 (Wickham, 2016).

Author contributions. HLW, MRH, CFB and MMT devised the study. HLW and IS set up the filter radiometer. SRL, IS, and MRJ collected concentration and other ancillary measurements at Auchencorth Moss. RK ran the spectrophotometer used for calibration. MC provided meteorology data. HLW performed all TUV model runs and analysed the data, with help from MRH, CFB and MMT. The paper was written by HLW with contributions from all co-authors.

400 *Competing interests.* The authors declare that they have no conflict of interest.

Acknowledgements. HLW acknowledges studentship funding from the UK Centre for Ecology & Hydrology (CEH) co-funding of Environment Agency Contract grant NEC05967 and the University of Edinburgh School of Chemistry. The authors acknowledge Dr Carole Helfter for the Phenocam image of Auchencorth Moss, and the UK Department for Environment, Food and Rural Affairs (Defra) and the Devolved



Administrations for maintaining instrument operations (ECM48524). The operation of the $j(\text{NO}_2)$ filter radiometer was supported by the UK
405 Natural Environment Research Council award number NE/R016429/1 as part of the UK-SCAPE programme delivering National Capability.
AURN data from <https://uk-air.defra.gov.uk> are Crown copyright and licensed under the terms of the Open Government Licence.



References

- Aas, W., Tsyro, S., Bieber, E., Bergström, R., Ceburnis, D., Ellermann, T., Fagerli, H., Frölich, M., Gehrig, R., Spindler, G., Vana, M., and Yttri, K. E.: Lessons learnt from the first EMEP intensive measurement periods, *Atmos. Chem. Phys.*, 12, 8073–8094, <https://doi.org/10.5194/acp-12-8073-2012>, 2012.
- Acker, K. and Möller, D.: Atmospheric variation of nitrous acid at different sites in Europe, *Environ. Chem.*, 4, 242–255, <https://doi.org/10.1071/EN07023>, 2007.
- Ahern, A. T., Goldberger, L., Jahl, L., Thornton, J., and Sullivan, R. C.: Production of N₂O₅ and ClNO₂ through Nocturnal Processing of Biomass-Burning Aerosol, *Environ. Sci. Technol.*, 52, 550–559, <https://doi.org/10.1021/acs.est.7b04386>, 2018.
- Alicke, B., Geyer, A., Hofzumahaus, A., Holland, F., Konrad, S., Pätz, H. W., Schäfer, J., Stutz, J., Volz-Thomas, A., and Platt, U.: OH formation by HONO photolysis during the BERLIOZ experiment, *J. Geophys. Res.*, 108, 8247, <https://doi.org/10.1029/2001JD000579>, 2003.
- Atkinson, R. and Arey, J.: Atmospheric Degradation of Volatile Organic Compounds, *Chem. Rev.*, 103, 4605–4638, <https://doi.org/10.1021/cr0206420>, 2003.
- Atkinson, R., Baulch, D. L., Cox, R. A., Crowley, J. N., Hampson, R. F., Hynes, R. G., Jenkin, M. E., Rossi, M. J., and Troe, J.: Evaluated kinetic and photochemical data for atmospheric chemistry: Volume III - gas phase reactions of inorganic halogens, *Atmos. Chem. Phys.*, 7, 981–1191, <https://doi.org/10.5194/acpd-6-2281-2006>, 2007.
- Bahe, F. C., Schurath, U., and Becker, K. H.: The frequency of NO₂ photolysis at ground level, as recorded by a continuous actinometer, *Atmos. Environ.*, 14, 711–718, 1980.
- Bais, A. F., Madronich, S., Crawford, J., Hall, S. R., Mayer, B., van Weele, M., Lenoble, J., Calvert, J. G., Cantrell, C. A., Shetter, R. E., Hofzumahaus, A., Koepke, P., Monks, P. S., Frost, G., McKenzie, R., Krotkov, N., Kylling, A., Swartz, W. H., Lloyd, S., Pfister, G., Martin, T. J., Roeth, E.-P., Griffioen, E., Ruggaber, A., Krol, M., Kraus, A., Edwards, G. D., Mueller, M., Lefer, B. L., Johnston, P., Schwander, H., Flittner, D., Gardiner, B. G., Barrick, J., and Schmitt, R.: International Photolysis Frequency Measurement and Model Intercomparison (IPMMI): Spectral actinic solar flux measurements and modeling, *J. Geophys. Res.: Atmos.*, 108, 8543, <https://doi.org/10.1029/2002JD002891>, 2003.
- Bannan, T. J., Booth, A. M., Bacak, A., Muller, J. B. A., Leather, K. E., Le Breton, M., Jones, B., Young, D., Coe, H., Allan, J., Visser, S., Slowik, J. G., Furger, M., Prévôt, A. S. H., Lee, J., Dunmore, R. E., Hopkins, J. R., Hamilton, J. F., Lewis, A. C., Whalley, L. K., Sharp, T., Stone, D., Heard, D. E., Fleming, Z. L., Leigh, R., Shallcross, D. E., and Percival, C. J.: The first UK measurements of nitryl chloride using a chemical ionization mass spectrometer in central London in the summer of 2012, and an investigation of the role of Cl atom oxidation, *J. Geophys. Res.: Atmos.*, 120, 5638–5657, <https://doi.org/10.1002/2014JD022629>, 2015.
- Bohn, B., Kraus, A., Müller, M., and Hofzumahaus, A.: Measurement of atmospheric O₃ → O(¹D) photolysis frequencies using filterradiometry, *J. Geophys. Res.: Atmos.*, 109, D10S90, <https://doi.org/10.1029/2003JD004319>, 2004.
- Bohn, B., Rohrer, F., Brauers, T., and Wahner, A.: Actinometric measurements of NO₂ photolysis frequencies in the atmosphere simulation chamber SAPHIR, *Atmos. Chem. Phys.*, 5, 493–503, <https://doi.org/10.5194/acpd-4-8141-2004>, 2005.
- Bohn, B., Corlett, G. K., Gillmann, M., Sanghavi, S., Stange, G., Tensing, E., Vrekoussis, M., Bloss, W. J., Clapp, L. J., Kortner, M., Dorn, H. P., Monks, P. S., Platt, U., Plass-Dülmer, C., Mihalopoulos, N., Heard, D. E., Clemitshaw, K. C., Meixner, F. X., Prevot, A. S. H., and Schmitt, R.: Photolysis frequency measurement techniques: Results of a comparison within the ACCENT project, *Atmos. Chem. Phys.*, 8, 5373–5391, <https://doi.org/10.5194/acp-8-5373-2008>, 2008.



- Brauers, T. and Hofzumahaus, A.: Latitudinal Variation of Measured NO₂ Photolysis Frequencies over the Atlantic Ocean between 50° N and 30° S, *J. Atmos. Chem.*, 15, 269–282, <https://doi.org/10.1007/BF00115398>, 1992.
- Coyle, M., Cape, J. N., Flechard, C., Fowler, D., Helfter, C., Jones, M., Kentisbeer, J., Leeson, S. R., Leith, I. D., Mullinger, N., Nemitz, E., Roberts, E., Simmons, I., Storeton-West, R., Twigg, M., and van Dijk, N.: Meteorological measurements at Auchencorth Moss from 1995 to 2016, *Geosci. Data J.*, 6, 16–29, <https://doi.org/10.1002/gdj3.63>, 2019.
- Coyle, M., Roberts, E., Jones, M., Leeson, S., Mullinger, N., Simmons, I., Van Dijk, N., Kentisbeer, J., Leith, I., Storton-West, R., Flechard, C., Twigg, M., Helfter, C., Fowler, D., and Nemitz, E.: Auchencorth Moss Atmospheric Observatory (AU): Annual half-hourly meteorology since 1995, Near Edinburgh, UK., <https://catalogue.ceda.ac.uk/uuid/8e6cbb111cfd41a19c92aadcb2d040fd>, last access: 26 March 2020, 2020.
- Defra: Automatic Urban and Rural Network (AURN), <https://uk-air.defra.gov.uk/networks/network-info?view=aurn>, last access: 1 March 2019, 2019.
- Defra: UK AIR: Data Selector, https://uk-air.defra.gov.uk/data/data_selector, last access: 20 March 2020, 2020a.
- Defra: Stratospheric Ozone Data, <https://uk-air.defra.gov.uk/data/ozone-data>, last access: 1 March 2020, 2020b.
- Edwards, G. D. and Monks, P. S.: Performance of a single-monochromator diode array spectroradiometer for the determination of actinic flux and atmospheric photolysis frequencies, *J. Geophys. Res.: Atmos.*, 108, 8546–8558, <https://doi.org/10.1029/2002JD002844>, 2003.
- Elshorbany, Y. F., Kurtenbach, R., Wiesen, P., Lissi, E., Rubio, M., Villena, G., Gramsch, E., Rickard, A. R., Pilling, M. J., and Kleffmann, J.: Oxidation capacity of the city air of Santiago, Chile, *Atmos. Chem. Phys.*, 9, 2257–2273, <https://doi.org/10.5194/acp-9-2257-2009>, 2009.
- Elshorbany, Y. F., Kleffmann, J., Hofzumahaus, A., Kurtenbach, R., Wiesen, P., Brauers, T., Bohn, B., Dorn, H.-P., Fuchs, H., Holland, F., Rohrer, F., Tillmann, R., Wegener, R., Wahner, A., Kanaya, Y., Yoshino, A., Nishida, S., Kajii, Y., Martinez, M., Kubistin, D., Harder, H., Lelieveld, J., Elste, T., Plass-Dülmer, C., Stange, G., Berresheim, H., and Schurath, U.: HO_x budgets during HO_xComp: A case study of HO_x chemistry under NO_x-limited conditions, *J. Geophys. Res.: Atmos.*, 117, D03 307, <https://doi.org/10.1029/2011JD017008>, 2012.
- EuroPhen: European Phenology Camera Network, <http://european-webcam-network.net/>, last access: 1 April 2020, 2020.
- Ghosh, B., Papanastasiou, D. K., Talukdar, R. K., Roberts, J. M., and Burkholder, J. B.: Nitryl Chloride (ClNO₂): UV/Vis Absorption Spectrum between 210 and 296 K and O(³P) Quantum Yield at 193 and 248 nm, *J. Phys. Chem. A*, 116, 5796–5805, <https://doi.org/10.1021/jp207389y>, 2012.
- Ghosh, D. and Sarkar, U.: Analysis of the photochemical production of ozone using Tropospheric Ultraviolet-Visible (TUV) Radiation Model in an Asian megacity, *Air Qual., Atmos. Health*, 9, 367–377, <https://doi.org/10.1007/s11869-015-0346-3>, 2016.
- Junkermann, W., Platt, U., and Volz-Thomas, A.: A photoelectric detector for the measurement of photolysis frequencies of ozone and other atmospheric molecules, *J. Atmos. Chem.*, 8, 203–227, <https://doi.org/10.1007/BF00051494>, 1989.
- Kleffmann, J., Kurtenbach, R., Lörzer, J., Wiesen, P., Kalthoff, N., Vogel, B., and Vogel, H.: Measured and simulated vertical profiles of nitrous acid – Part I: Field measurements, *Atmos. Environ.*, 37, 2949–2955, [https://doi.org/10.1016/S1352-2310\(03\)00242-5](https://doi.org/10.1016/S1352-2310(03)00242-5), 2003.
- Kleffmann, J., Gavriloaiei, T., Hofzumahaus, A., Holland, F., Koppmann, R., Rupp, L., Schlosser, E., Siese, M., and Wahner, A.: Daytime formation of nitrous acid: A major source of OH radicals in a forest, *Geophys. Res. Lett.*, 32, L05 818, <https://doi.org/10.1029/2005GL022524>, 2005.
- Kraus, A. and Hofzumahaus, A.: Field Measurements of Atmospheric Photolysis Frequencies for O₃, NO₂, HCHO, CH₃CHO, H₂O₂ and HONO by UV Spectroradiometry, *J. Atmos. Chem.*, 31, 161–180, <https://doi.org/10.1016/j.mce.2010.01.026>, 1998.



- 480 Kylling, A., Webb, A. R., Bais, A. F., Blumthaler, M., Schmitt, R., Thiel, S., Kazantzidis, A., Kift, R., Misslbeck, M., Schallhart, B., Schreder, J., Topaloglou, C., Kazadzis, S., and Rimmer, J.: Actinic flux determination from measurements of irradiance, *J. Geophys. Res.: Atmos.*, 108, 4506–4516, <https://doi.org/10.1029/2002jd003236>, 2003.
- Lantz, K. O., Shetter, R. E., Cantrell, C. A., Flocke, S. J., Calvert, J. G., and Madronich, S.: Theoretical, actinometric, and radiometric determinations of the photolysis rate coefficient of NO₂ during the Mauna Loa Observatory Photochemistry Experiment 2, *J. Geophys. Res.: Atmos.*, 101, 14 613–14 629, 1996.
- 485 Larason, T. C. and Cromer, C. L.: Sources of error in UV radiation measurements, *J. Res. Natl. Inst. Stand. Technol.*, 106, 649–656, <https://doi.org/10.6028/jres.106.030>, 2001.
- Lee-Taylor, J. and Madronich, S.: Calculation of actinic fluxes with a coupled atmosphere-snow radiative transfer model, *J. Geophys. Res.: Atmos.*, 107, <https://doi.org/10.1029/2002JD002084>, 2002.
- 490 Madronich, S.: Photodissociation in the Atmosphere: 1. Actinic Flux and the Effects of Ground Reflections and Clouds, *J. Geophys. Res.*, 92, 9740–9752, 1987.
- Mérieu, M. F., Jenouvrier, A., and Coquart, B.: The NO₂ Absorption Spectrum. I: Absorption Cross-Sections at Ambient Temperature in the 300–500 nm Region, *J. Atmos. Chem.*, 20, 281–297, <https://doi.org/10.1007/BF00694498>, 1995.
- Monks, P. S.: Gas-phase radical chemistry in the troposphere, *Chem. Soc. Rev.*, 34, 376–395, <https://doi.org/10.1039/b307982c>, 2005.
- 495 NCAR: Tropospheric Ultraviolet and Visible (TUV) Radiation Model v5.3, <http://cprm.acd.ucar.edu/Models/TUV/>, last access: 22 February 2019, 2019.
- Osthoff, H. D., Roberts, J. M., Ravishankara, A. R., Williams, E. J., Lerner, B. M., Sommariva, R., Bates, T. S., Coffman, D., Quinn, P. K., Dibb, J. E., Stark, H., Burkholder, J. B., Talukdar, R. K., Meagher, J., Fehsenfeld, F. C., and Brown, S. S.: High levels of nitryl chloride in the polluted subtropical marine boundary layer, *Nat. Geosci.*, 1, 324–328, <https://doi.org/10.1038/ngeo177>, 2008.
- 500 Osthoff, H. D., Odame-Ankrah, C. A., Taha, Y. M., Tokarek, T. W., Schiller, C. L., Haga, D., Jones, K., and Vingarzan, R.: Low levels of nitryl chloride at ground level: nocturnal nitrogen in the Lower Fraser Valley of British Columbia, *Atmos. Chem. Phys.*, 18, 6293–6315, 2018.
- Palancar, G. G., Lefer, B. L., Hall, S. R., Shaw, W. J., Corr, C. A., Herndon, S. C., Slusser, J. R., and Madronich, S.: Effect of aerosols and NO₂ concentration on ultraviolet actinic flux near Mexico City during MILAGRO: measurements and model calculations, *Atmos. Chem. Phys.*, 13, 1011–1022, <https://doi.org/10.5194/acp-13-1011-2013>, 2013.
- 505 Phillips, G. J., Makkonen, U., Schuster, G., Sobanski, N., Hakola, H., and Crowley, J. N.: The detection of nocturnal N₂O₅ as HNO₃ by alkali- and aqueous-denuder techniques, *Atmos. Meas. Tech.*, 6, 231–237, <https://doi.org/10.5194/amt-6-231-2013>, 2013.
- Riedel, T. P., Wolfe, G. M., Danas, K. T., Gilman, J. B., Kuster, W. C., Bon, D. M., Vlasenko, A., Li, S.-M., Williams, E. J., Lerner, B. M., Veres, P. R., Roberts, J. M., Holloway, J. S., Lefer, B., Brown, S. S., and Thornton, J. A.: An MCM modeling study of nitryl chloride (ClNO₂) impacts on oxidation, ozone production and nitrogen oxide partitioning in polluted continental outflow, *Atmos. Chem. Phys.*, 14, 3789–3800, <https://doi.org/10.5194/acp-14-3789-2014>, 2014.
- 510 Rumsey, I. C., Cowen, K. A., Walker, J. T., Kelly, T. J., Hanft, E. A., Mishoe, K., Rogers, C., Proost, R., Beachley, G. M., Lear, G., Frelink, T., and Otjes, R. P.: An assessment of the performance of the Monitor for Aerosols and Gases in ambient air (MARGA): a semi-continuous method for soluble compounds, *Atmos. Chem. Phys.*, 14, 5639–5658, <https://doi.org/10.5194/acp-14-5639-2014>, 2014.
- 515 Ryan, R. G., Rhodes, S., Tully, M., Wilson, S., Jones, N., Frieß, U., and Schofield, R.: Daytime HONO, NO₂ and aerosol distributions from MAX-DOAS observations in Melbourne, *Atmos. Chem. Phys.*, 18, 13 969–13 985, <https://doi.org/10.5194/acp-2018-409>, 2018.



- Sander, S. P., Friedl, R. R., Golden, D. M., Kurylo, M. J., Moortgat, G. K., Wine, P. H., Ravishankara, A. R., Kolb, C. E., Molina, M. J., Finlayson-Pitts, B. J., Huie, R. E., and Orkin, V. L.: Chemical Kinetics and Photochemical Data for Use in Atmospheric Studies, Evaluation Number 15, JPL Publication 06-2, Jet Propulsion Laboratory, Pasadena, <http://jpldataeval.jpl.nasa.gov>, 2006.
- 520 Shetter, R. E. and Müller, M.: Photolysis frequency measurements using actinic flux spectroradiometry during the PEM-Tropics mission: Instrumentation description and some results, *J. Geophys. Res.: Atmos.*, 104, 5647–5661, <https://doi.org/10.1029/98JD01381>, 1999.
- Shetter, R. E., McDaniel, A. H., Cantrell, C. A., Madronich, S., and Calvert, J. G.: Actinometer and Eppley Radiometer Measurements of the NO₂ Photolysis Rate Coefficient During the Mauna Loa Observatory Photochemistry Experiment, *J. Geophys. Res.*, 97, 10 349–10 359, <https://doi.org/10.1029/91JD02289>, 1992.
- 525 Shetter, R. E., Cantrell, C. A., Lantz, K. O., Flocke, S. J., Orlando, J. J., Tyndall, G. S., Gilpin, T. M., Fischer, C. A., Madronich, S., Calvert, J. G., and Junkermann, W.: Actinometric and radiometric measurement and modeling of the photolysis rate coefficient of ozone to O(¹D) during Mauna Loa Observatory Photochemistry Experiment 2, *J. Geophys. Res.: Atmos.*, 101, 14,631–14,641, <https://doi.org/10.1029/96JD00211>, 1996.
- Shetter, R. E., Junkermann, W., Swartz, W. H., Frost, G. J., Crawford, J. H., Lefer, B. L., Barrick, J. D., Hall, S. R., Hofzumahaus, A., Bais, A., Calvert, J. G., Cantrell, C. A., Madronich, S., Müller, M., Kraus, A., Monks, P. S., Edwards, G. D., McKenzie, R., Johnston, P., Schmitt, R., Griffioen, E., Krol, M., Kylling, A., Dickerson, R. R., Lloyd, S. A., Martin, T., Gardiner, B., Mayer, B., Pfister, G., Röth, E. P., Koepke, P., Ruggaber, A., Schwander, H., and van Weele, M.: Photolysis frequency of NO₂: Measurement and modeling during the International Photolysis Frequency Measurement and Modeling Intercomparison (IPMMI), *J. Geophys. Res.: Atmos.*, 108, 8544, <https://doi.org/10.1029/2002JD002932>, 2003.
- 530 Sommariva, R., Hollis, L. D. J., Sherwen, T., Baker, A. R., Ball, S. M., Bandy, B. J., Bell, T. G., Chowdhury, M. N., Cordell, R. L., Evans, M. J., Lee, J. D., Reed, C., Reeves, C. E., Roberts, J. M., Yang, M., and Monks, P. S.: Seasonal and geographical variability of nitryl chloride and its precursors in Northern Europe, *Atmos. Sci. Lett.*, 19, 1–10, <https://doi.org/10.1002/asl.844>, 2018.
- Sommariva, R., Cox, S., Martin, C., Borońska, K., Young, J., Jimack, P. K., Pilling, M. J., Matthaios, V. N., Nelson, B. S., Newland, M. J., Panagi, M., Bloss, W. J., Monks, P. S., and Rickard, A. R.: AtChem (version 1), an open-source box model for the Master Chemical Mechanism, *Geosci. Model Dev.*, 13, 169–183, <https://doi.org/10.5194/gmd-13-169-2020>, 2020.
- 540 Spataro, F. and Ianniello, A.: Sources of atmospheric nitrous acid: State of the science, current research needs, and future prospects, *J. Air Waste Manage. Assoc.*, 64, 1232–1250, <https://doi.org/10.1080/10962247.2014.952846>, 2014.
- Stone, D., Whalley, L. K., and Heard, D. E.: Tropospheric OH and HO₂ radicals: Field measurements and model comparisons, *Chem. Soc. Rev.*, 41, 6348–6404, <https://doi.org/10.1039/c2cs35140d>, 2012.
- 545 Thornton, J. A., Kercher, J. P., Riedel, T. P., Wagner, N. L., Cozic, J., Holloway, J. S., Dubé, W. P., Wolfe, G. M., Quinn, P. K., Middlebrook, A. M., Alexander, B., and Brown, S. S.: A large atomic chlorine source inferred from mid-continental reactive nitrogen chemistry, *Nature*, 464, 271–274, <https://doi.org/10.1038/nature08905>, 2010.
- Tørseth, K., Aas, W., Breivik, K., Fjaeraa, A. M., Fiebig, M., Hjellbrekke, A. G., Myhre, C. L., Solberg, S., and Yttri, K. E.: Introduction to the European Monitoring and Evaluation Programme (EMEP) and observed atmospheric composition change during 1972–2009, *Atmos. Chem. Phys.*, 12, 5447–5481, <https://doi.org/10.5194/acp-12-5447-2012>, 2012.
- 550 Trebs, I., Bohn, B., Ammann, C., Rummel, U., Blumthaler, M., Königstedt, R., Meixner, F. X., Fan, S., and Andreae, M. O.: Relationship between the NO₂ photolysis frequency and the solar global irradiance, *Atmos. Meas. Tech.*, 2, 725–739, <https://doi.org/10.5194/amt-2-725-2009>, 2009.



- Troe, J.: Are Primary Quantum Yields of NO₂ Photolysis at $\lambda \leq 398$ nm Smaller than Unity?, *Z. Phys. Chem.*, 214, 573–581, <https://doi.org/10.1524/zpch.2000.214.5.573>, 2000.
- Twigg, M. M., Di Marco, C. F., Leeson, S., Van Dijk, N., Jones, M. R., Leith, I. D., Morrison, E., Coyle, M., Proost, R., Peeters, A. N. M., Lemon, E., Frelink, T., Braban, C. F., Nemitz, E., and Cape, J. N.: Water soluble aerosols and gases at a UK background site – Part 1: Controls of PM_{2.5} and PM₁₀ aerosol composition, *Atmos. Chem. Phys.*, 15, 8131–8145, <https://doi.org/10.5194/acp-15-8131-2015>, 2015.
- UNECE: The EMEP Monitoring Strategy and measurement programme 2004–2009, EB.AIR/GE.1/2004/5, available at: http://www.unece.org/fileadmin/DAM/env/lrtap/emep/MonitoringStrategy_full.pdf, 2004.
- Villena, G., Wiesen, P., Cantrell, C. A., Flocke, F., Fried, A., Hall, S. R., Hornbrook, R. S., Knapp, D., Kosciuch, E., Mauldin III, R. L., McGrath, J. A., Montzka, D., Richter, D., Ullmann, K., Walega, J., Weibring, P., Weinheimer, A., Staebler, R. M., Liao, J., Huey, L. G., and Kleffmann, J.: Nitrous acid (HONO) during polar spring in Barrow, Alaska: A net source of OH radicals?, *J. Geophys. Res.: Atmos.*, 116, D00R07, <https://doi.org/10.1029/2011JD016643>, 2011.
- Volz-Thomas, A., Lerner, A., Pätz, H.-W., Schultz, M., Mckenna, D. S., Schmitt, R., Madronich, S., and Röth, E. P.: Airborne measurements of the photolysis frequency of NO₂, *J. Geophys. Res.: Atmos.*, 101, 18 613–18 627, <https://doi.org/10.1029/96JD01375>, 1996.
- Walker, H., Heal, M., Braban, C., Coyle, M., Leeson, S., Simmons, I., Jones, M., and Twigg, M.: Auchencorth Moss Atmospheric Observatory (AU): Hourly averaged 4-pi filter radiometer measurements (21/11/2018 - 20/11/2019) near Edinburgh (UK)., <http://dx.doi.org/10.5285/f9a0eaf6cfb8479f89a62ecbc091ec7d>, last access: 15 June 2020, 2020.
- Wang, J., Zhang, X., Guo, J., Wang, Z., and Zhang, M.: Observation of nitrous acid (HONO) in Beijing, China: Seasonal variation, nocturnal formation and daytime budget, *Sci. Total Environ.*, 587–588, 350–359, <https://doi.org/10.1016/j.scitotenv.2017.02.159>, 2017.
- Webb, A. R., Kift, R., Thiel, S., and Blumthaler, M.: An empirical method for the conversion of spectral UV irradiance measurements to actinic flux data, *Atmos. Environ.*, 36, 4397–4404, [https://doi.org/10.1016/S1352-2310\(02\)00319-9](https://doi.org/10.1016/S1352-2310(02)00319-9), 2002.
- Wickham, H.: ggplot2: Elegant Graphics for Data Analysis, Springer-Verlag New York, <https://ggplot2.tidyverse.org>, 2016.
- Wild, O., Zhu, X., and Prather, M. J.: Fast-J: Accurate Simulation of In- and Below-Cloud Photolysis in Tropospheric Chemical Models, *J. Atmos. Chem.*, 37, 245–282, <https://doi.org/10.1023/A:1006415919030>, 2000.
- Wilson, S. R.: Characterisation of J(O¹D) at Cape Grim 2000–2005, *Atmos. Chem. Phys.*, 15, 7337–7349, <https://doi.org/10.5194/acp-15-7337-2015>, 2015.
- Xu, W., Kuang, Y., Zhao, C., Tao, J., Zhao, G., Bian, Y., Yang, W., Yu, Y., Shen, C., Liang, L., Zhang, G., Lin, W., and Xu, X.: NH₃-promoted hydrolysis of NO₂ induces explosive growth in HONO, *Atmos. Chem. Phys.*, 19, 10 557–10 570, 2019.
- Young, C. J., Washenfelder, R. A., Roberts, J. M., Mielke, L. H., Osthoff, H. D., Tsai, C., Pikelnaya, O., Stutz, J., Veres, P. R., Cochran, A. K., Vandenboer, T. C., Flynn, J., Grossberg, N., Haman, C. L., Lefer, B., Stark, H., Graus, M., De Gouw, J., Gilman, J. B., Kuster, W. C., and Brown, S. S.: Vertically resolved measurements of nighttime radical reservoirs in Los Angeles and their contribution to the urban radical budget, *Environ. Sci. Technol.*, 46, 10 965–10 973, <https://doi.org/10.1021/es302206a>, 2012.
- Young, C. J., Washenfelder, R. A., Edwards, P. M., Parrish, D. D., Gilman, J. B., Kuster, W. C., Mielke, L. H., Osthoff, H. D., Tsai, C., Pikelnaya, O., Stutz, J., Veres, P. R., Roberts, J. M., Griffith, S., Dusanter, S., Stevens, P. S., Flynn, J., Grossberg, N., Lefer, B., Holloway, J. S., Peischl, J., Ryerson, T. B., Atlas, E. L., Blake, D. R., and Brown, S. S.: Chlorine as a primary radical: evaluation of methods to understand its role in initiation of oxidative cycles, *Atmos. Chem. Phys.*, 14, 3427–3440, <https://doi.org/10.5194/acp-14-3427-2014>, 2014.



Table 1. Long-term measurement instrumentation at Auchencorth Moss used in this study.

Variable	Instrument	Reference
Photolysis rate constant, $j(\text{NO}_2)$	Metcon 4- π filter radiometer	(Volz-Thomas et al., 1996)
HONO, HNO_3	MARGA ^a	(Twigg et al., 2015; Rumsey et al., 2014)
O_3	Teledyne API 400E ^b	(Defra, 2019)
Relative humidity (RH)	Rotronics HC2S3	(Coyle et al., 2019)
Temperature (T)	Rotronics HC2S3	(Coyle et al., 2019)
Pressure (p)	Druck RPT 410F	(Coyle et al., 2019)
Global solar irradiance (G)	Skye SKS 1110	(Coyle et al., 2019)

^aMonitor for AeRosols and Gases in Ambient Air (Metrohm Applikon, NL).

^bUV-absorption.



Table 2. Summary statistics of hourly trace gas concentrations measured at Auchencorth Moss, for each meteorological season (winter, DJF; spring, MAM; summer, JJA; autumn, SON). Values are mean (μ_A), median (μ_M), minimum, maximum, arithmetic standard deviation (σ_A) and number of measurements (n).

Trace gas	Season	$\mu_A / \mu\text{g m}^{-3}$	$\mu_M / \mu\text{g m}^{-3}$	Min / $\mu\text{g m}^{-3}$	Max / $\mu\text{g m}^{-3}$	$\sigma_A / \mu\text{g m}^{-3}$	n
O ₃	Winter	57.1	58.6	8.2	91.4	12.5	2156
	Spring	73.9	74.2	24.0	147.8	18.7	2206
	Summer	61.0	61.3	11.6	149.7	16.3	2205
	Autumn	50.3	51.2	2.2	76.2	12.2	1963
HONO	Winter	0.076	0.061	0.020	0.54	0.053	1567
	Spring	0.076	0.059	0.003	0.53	0.058	1916
	Summer	0.091	0.065	0.016	0.57	0.074	2036
	Autumn	0.069	0.059	0.007	0.70	0.045	1769
HNO ₃	Winter	0.096	0.056	0.001	1.55	0.14	1567
	Spring	0.16	0.10	0.003	1.46	0.20	1916
	Summer	0.10	0.084	0.001	0.67	0.081	2036
	Autumn	0.067	0.052	0.001	0.56	0.062	1769



Table 3. Mean production rates of OH from the photolysis of O₃, HONO and HNO₃ during each meteorological season, using *j*-values directly output from the TUV model and those adjusted using the MDAF metric. Means are calculated using only daytime measurements, and are reported with corresponding confidence intervals (95%). The % decrement in production rate for the MDAF data is the average decrement across the season.

Season	$p(\text{OH})_{\text{O}_3} /$ $\times 10^4 \text{ radicals cm}^{-3} \text{ s}^{-1}$			$p(\text{OH})_{\text{HONO}} /$ $\times 10^4 \text{ radicals cm}^{-3} \text{ s}^{-1}$			$p(\text{OH})_{\text{HNO}_3} /$ $\text{radicals cm}^{-3} \text{ s}^{-1}$		
	TUV	MDAF	$\Delta / \%$	TUV	MDAF	$\Delta / \%$	TUV	MDAF	$\Delta / \%$
Winter (DJF)	5.08 ± 0.55	3.16 ± 0.35	24	24.3 ± 2.5	18.1 ± 2.2	24	27.3 ± 4.7	19.9 ± 3.6	24
Spring (MAM)	68.2 ± 4.4	37.1 ± 2.8	41	69.4 ± 4.5	41.1 ± 3.2	40	321 ± 34	204 ± 26	40
Summer (JJA)	110 ± 6.1	60.0 ± 3.7	46	78.1 ± 4.6	46.8 ± 3.0	47	214 ± 15	122 ± 10	47
Autumn (SON)	24.6 ± 2.1	13.1 ± 1.3	41	42.6 ± 4.1	22.5 ± 1.8	40	47.4 ± 5.4	27.5 ± 4.0	40

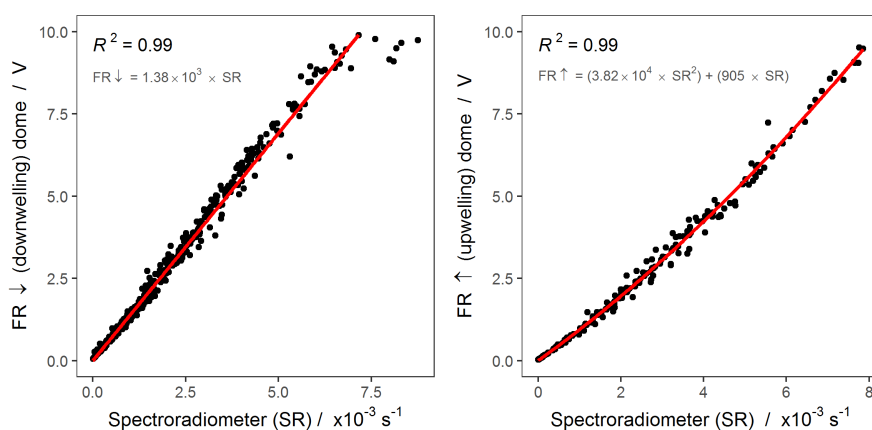


Figure 1. Calibration of both filter radiometer optical inlet domes against a Bentham DTM300 spectroradiometer from 13–25 June 2019. Filter radiometer measurements (1 s) are averaged to equal the scan duration of spectroradiometer (approx. 3 mins). Relationships used for subsequent conversion of filter radiometer measurements are presented in red.

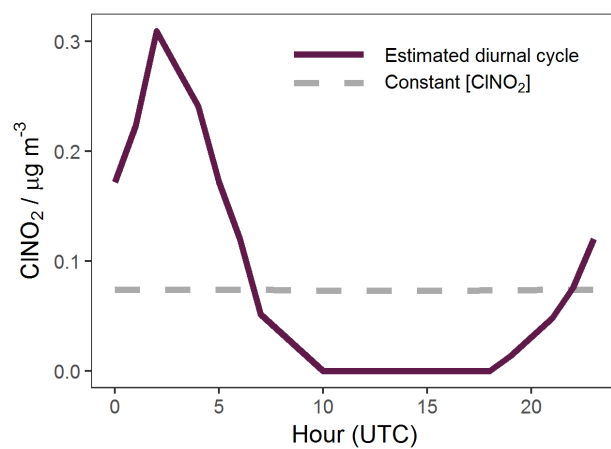


Figure 2. Diurnal cycle of ClNO_2 concentration estimates used to calculate the production rate of Cl radicals from ClNO_2 photolysis. Diurnal cycle based on summer Weybourne measurements from Sommariva et al. (2018).

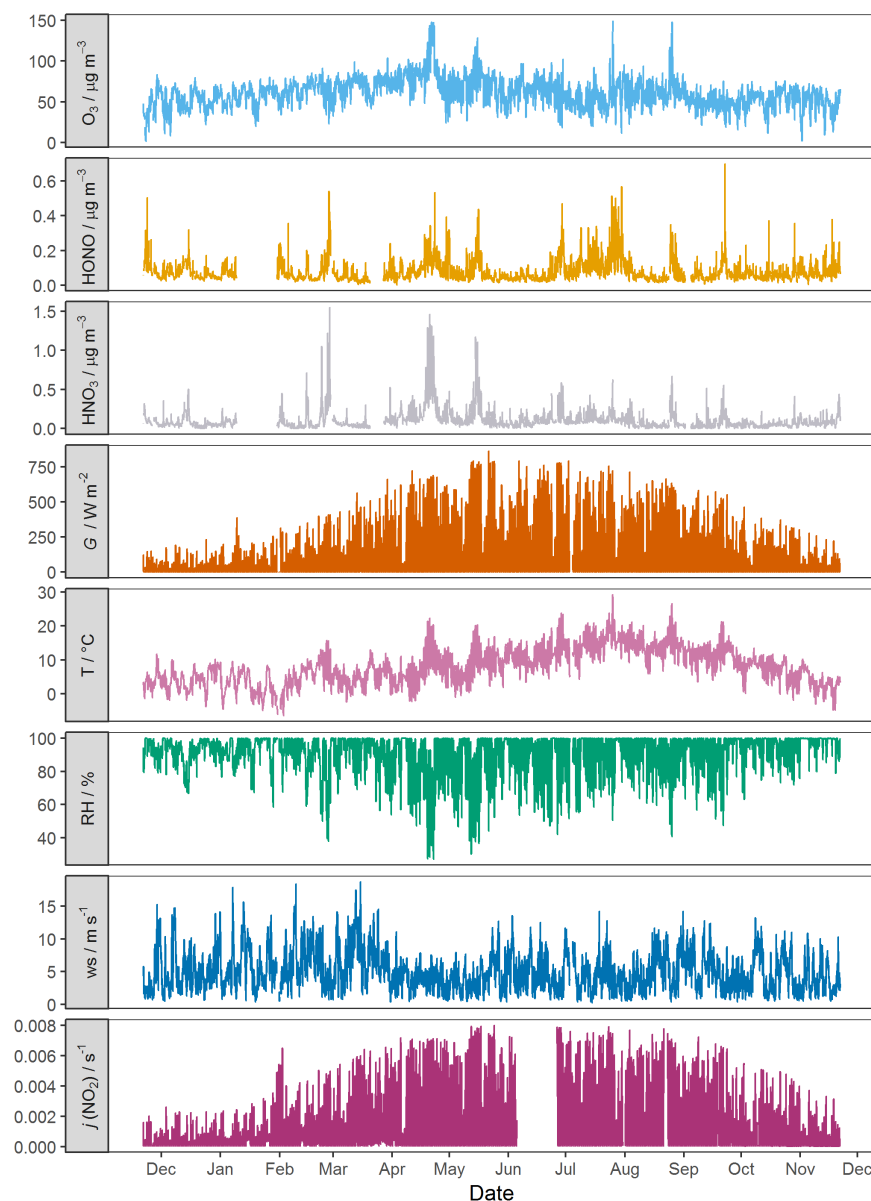


Figure 3. Time series of mean hourly trace gas concentrations (O_3 , HONO, HNO_3) and meteorological parameters (solar global irradiance (G), air temperature, relative humidity and wind speed) for the year of data included in this study. The $j(\text{NO}_2)$ measurements reported are the hourly mean of summed down- and up-welling components.

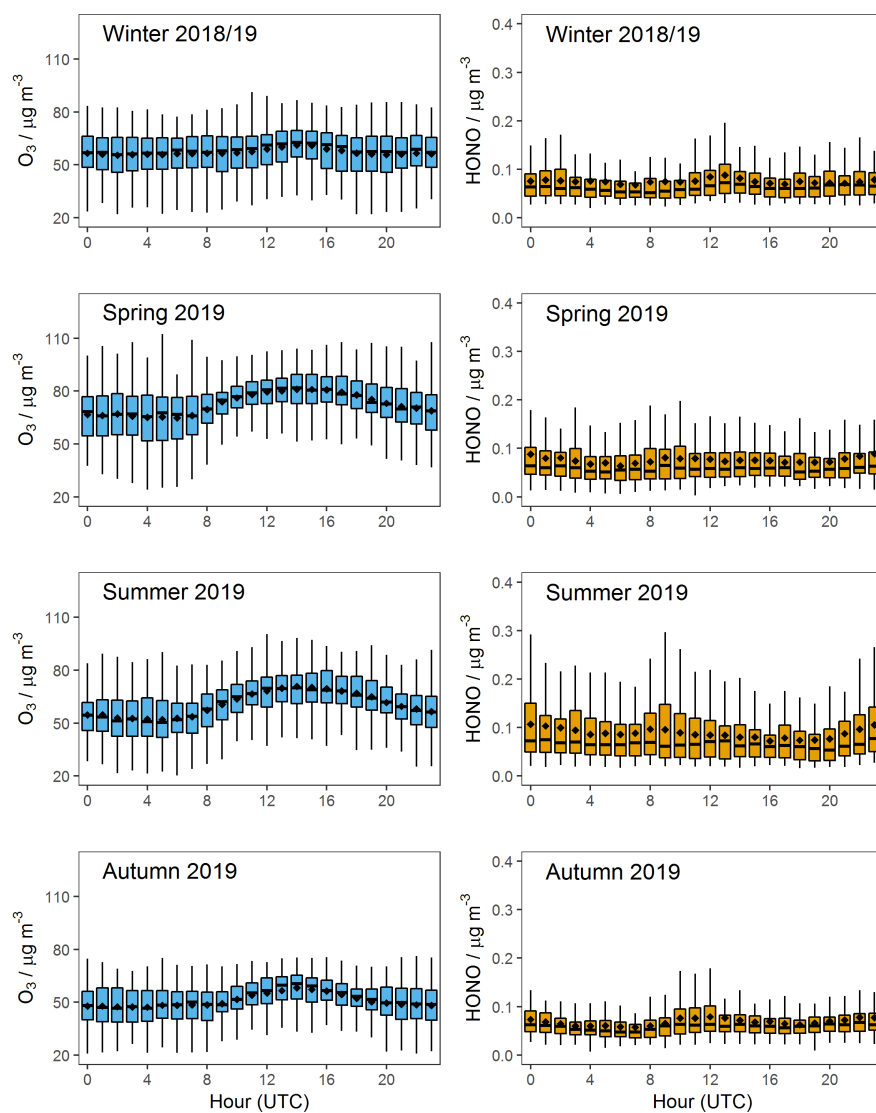


Figure 4. Median hourly diurnal trace gas concentrations of O_3 and HONO in each meteorological season. Mean values are represented by diamond points. Boxes show the upper and lower quartiles, and the whiskers present the 5-95% range. Time in all seasons refer to UTC.

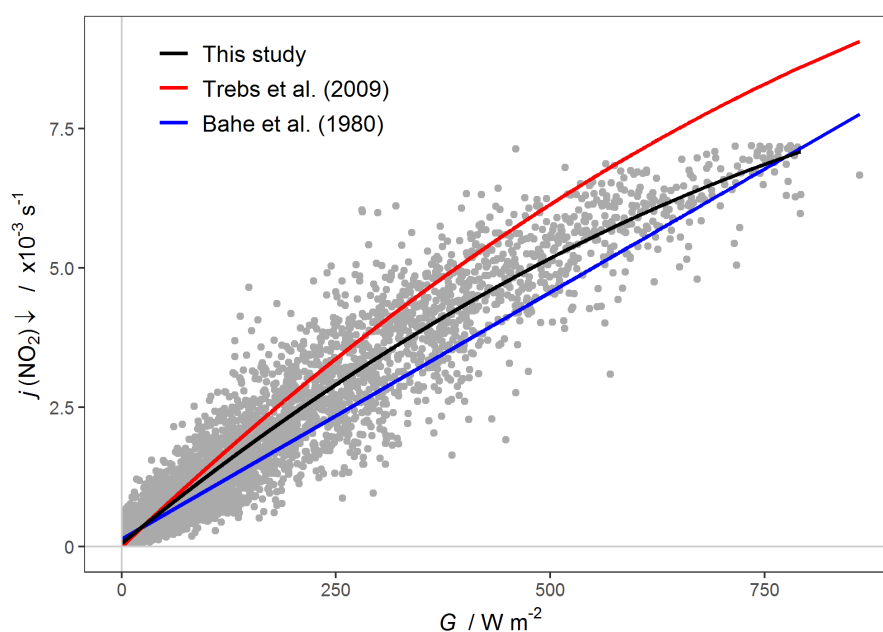


Figure 5. Scatter plot of hourly averaged down-welling $j(\text{NO}_2)$ measured by the upward facing dome of the filter radiometer, and the solar global irradiance (G) for the year of data collected. The second-order polynomial fit to these data is shown in black. The parameterisations between $j(\text{NO}_2)$ and G from Bahe et al. (1980) and Trebs et al. (2009) are also shown for comparison.

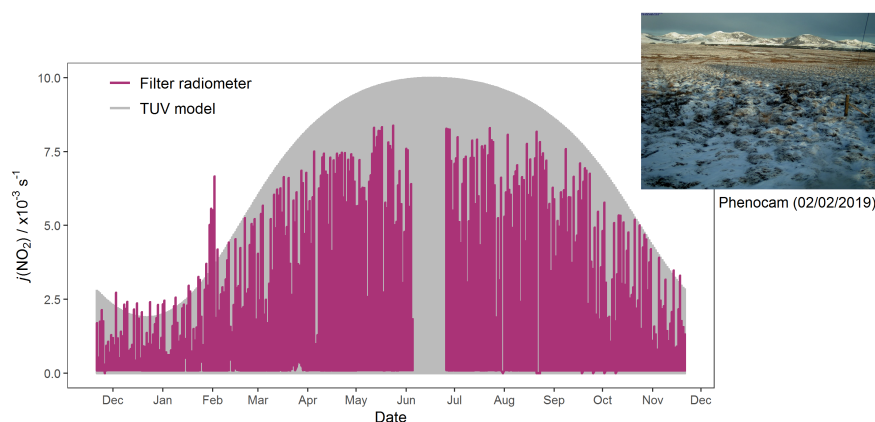


Figure 6. Comparison of hourly TUV modelled $j(\text{NO}_2)$ at Auchencorth Moss, and total down- and up-welling $j(\text{NO}_2)$ as measured by the filter radiometer at Auchencorth Moss for the year of study. Inset shows a photo of the site south of the filter radiometer taken on 02/02/2019 by an automatic Phenocam (EuroPhen, 2020)

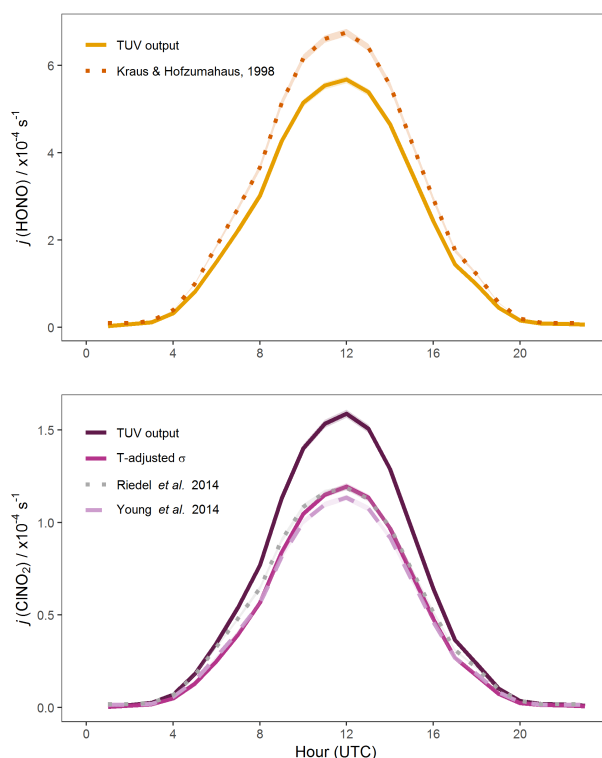


Figure 7. Top: Annual average diurnal cycle of $j(\text{HONO})$ at Auchencorth Moss calculated by the TUV model and using the Kraus and Hofzumahaus (1998) parameterisation. **Bottom:** Annual average diurnal cycle of $j(\text{ClNO}_2)$ estimates for Auchencorth Moss, using the direct TUV model output and updated $j(\text{ClNO}_2)$ values (accounting for daily average temperature), and parameterisations presented in Riedel et al. (2014) and Young et al. (2014). All values of $j(\text{HONO})$ and $j(\text{ClNO}_2)$ are the annual mean of each hour with the MDAF metric applied, to account for local influence on solar fluxes. 95% confidence intervals shaded.

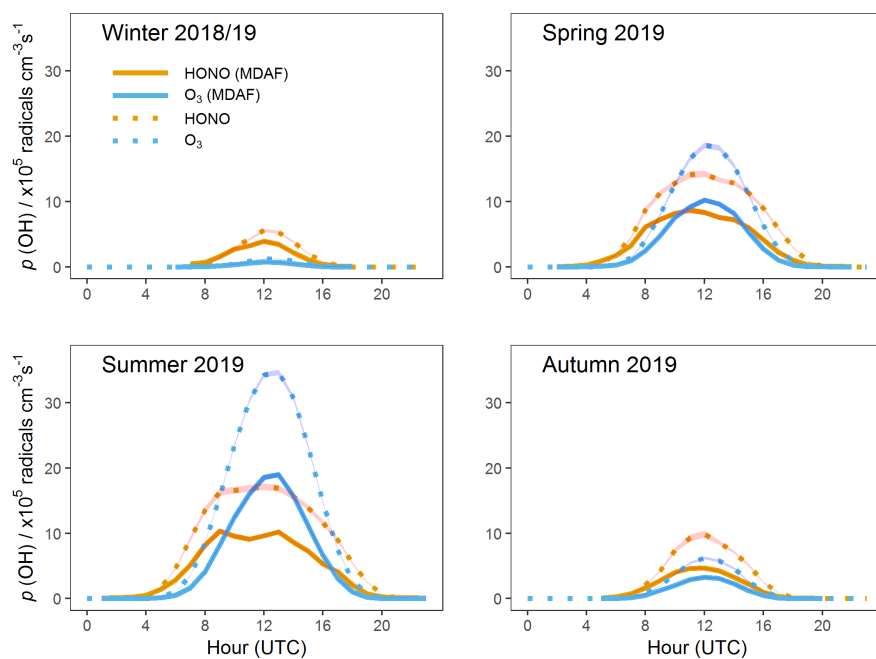


Figure 8. Diurnal variations in seasonally-averaged hourly mean $p(\text{OH})$ rates from O_3 and HONO photolysis. Dotted lines show rates calculated using j -values directly from the TUV model, while solid lines show rates where j -values were first corrected by the MDAF metric, as described in the text. Shading represents the 95% confidence intervals.

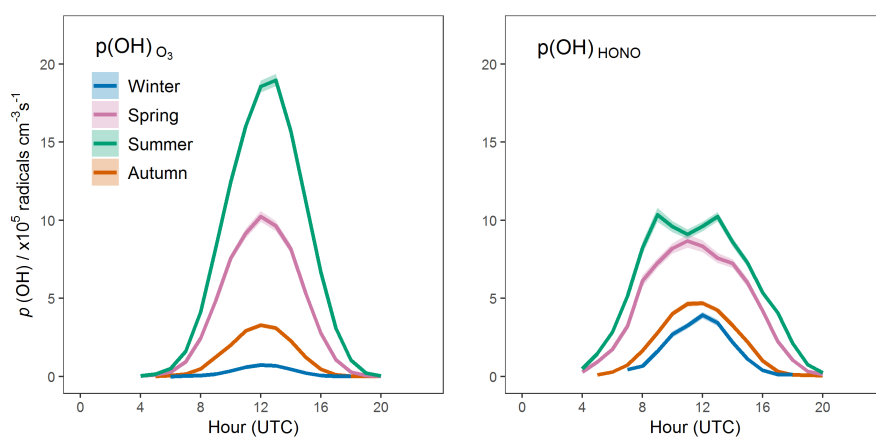


Figure 9. Diurnal variations in seasonally-averaged hourly mean $p(\text{OH})$ rates from O_3 and HONO photolysis, where j -values used are corrected by the MDAF. Shading represents 95% confidence interval.

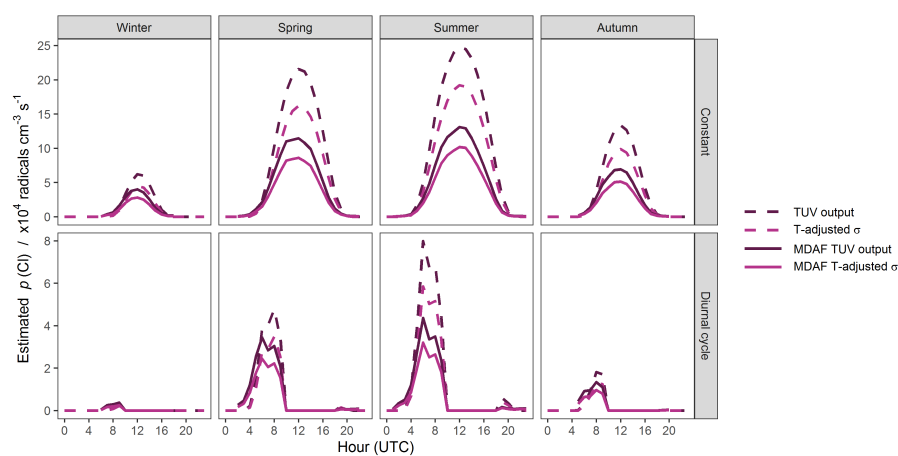


Figure 10. Diurnal variations in seasonally-averaged hourly production of chlorine atom radicals, $p(\text{Cl})$, from nitryl chloride, using a constant ClNO_2 concentration of 21.5 ppt (upper panel), and the diurnal cycle of ClNO_2 presented in Fig. 2 (lower panel).