General comments.

This paper uses measurements in highly concentrated fire plumes (within 100m of wildland grass fires, and in controlled burns at the Missoula Fire Lab) to assess interferences in UV absorption measurements of ozone at 254 nm.

This paper is motivated by the ozone measurements of UV absorption instruments, and the health impacts from that ozone. Large increases in ozone may be observed after precursor NOx and VOC have had time to react. The time scale to produce that ozone is highly dependent on plume dilution, which itself is highly variable in time, but typically takes place over hours since emission. A fundamental question: for direct emissions, the interfering species will also be diluted, such that the lowest interferences may be expected at the highest levels of plume ozone. Secondary production of UV-active hydrocarbons, e.g., production of nitroaromatics following oxidation in the presence of NO2, may dominate the ozone interference downwind. What balance of directly emitted vs. secondary species are conjectured to lead to interferences in ambient ozone measurements? Regardless of the source of the interference (primary vs. secondary), given the lack of consistency from fire to fire (or even between different implementations of the UV absorption technique) in the level of interferences measured, can the authors say what level of "fire impact" causes a non-negligible interference? Is 1 ppm of ozone acceptable? 10 ppm of ozone?

Regardless, despite the experimental detail in this paper, it is not clear what ozone monitoring locations are expected to suffer from significant interferences as a result of wildfires or prescribed burns. Lacking these considerations, the paper’s conclusions are qualitative at best, and by implication condemn a much larger portion of the U.S. ozone monitoring network during the fire season than I suspect is warranted. For a given UV absorption monitor, can they recommend what data to retain, and what data to eliminate because of fire impacts? Some additional clarity in the real-world effects of fire smoke on ozone monitoring is needed for this to make a novel and useful contribution to the literature.

The paper is overly long and can be shortened by removing extraneous details, repetitive text, and tables that do not provide any usefully generalizable data as suggested below. Earlier literature is not well cited, and additional references are also suggested below.

Response: The authors appreciate the time required to provide the review and feel that the suggestions provided by the reviewer will result in an improved manuscript for resubmission.

The authors do agree that a more detailed look at data collected at sites being impacted by aged smoke (ex. State and local regulatory monitoring sites being impacted by nearby wildfires and long range transport of photochemically aged smoke plumes) and are currently collecting this data as part of the EPA MASiC study in Boise, ID; Missoula, MT; and Reno, NV. This additional data collection will aid in linking these research chamber and near field prescribed grassland burn measurements back to real world
regulatory monitoring situations. We will address these issues in a new “implications” section prior to the manuscript conclusion.

The authors will, as suggested by the reviewer, attempt to shorten the length of the manuscript by removing text that is tangential to the scope of the paper. In addition, the authors will review earlier literature and cite as appropriate including those references suggested by the reviewer.

**Manuscript Revision:** The authors, as suggested by the reviewer removed text from the manuscript to shorten the length of the document. An additional “Implications” section was added prior to the conclusions section to tie the results of the research detailed in this manuscript to real world monitoring applications. Included in this section will be a review of data from monitoring sites downwind of fires to show the impact of the measurement artifacts described in this manuscript.

**Specific comments.**

**line 14:** "... large increases in ozone are also observed downwind ..." Is this always true?

**Response:** The authors did not imply that large increases in ozone are always observed downwind of wildfire events. To clarify this and prevent assumptions that these increases in ozone “always occur”, the text will be rewritten to include a statement like "... large increases in ozone have been observed downwind ..."

**Manuscript Revision:** text changed to read "... large increases in ozone have been observed downwind ...

**line 32 (and lines 38 and 182 and elsewhere):** The NO-induced chemiluminescence measurement of ozone is repeatedly described as "interference-free", which is misleading - it has a known dependency on water vapor, which can lead to sensitivity variations of up to 8% if not accounted for. Please rephrase.

**Response:** The authors will rephrase these statements to emphasize that sample treatment steps, including the use of a drier, must be taken prior to analysis to remove the effects of water vapor.

**Manuscript Revision:** Removed “interference free” from line 32. Removed “interference free” from line 38. Clarifying statement inserted in line 62 “Both the ET-CL and NO-CL methods are subject to slight interferences by water vapor. However, these potential interferences can be eliminated through the use of Nafion based drier or equivalent sample water vapor treatment system.” Removed “interference free” from line 182.

**line 52:** for clarity, please change to "... generates nitrogen dioxide in an electronically excited state..."

The original citation is Clough and Thrush, 1966, Chemical Communications, 728, pp. 783-784.

**Response:** The authors agree with this suggestion and will change the text accordingly.

**Manuscript Revision:** Text changed to read "... generates nitrogen dioxide in an electronically excited state..."
line 93: please remove CO2, as its absorption is negligible at 254 nm.
Response: The authors agree with this suggestion and will remove CO2 from the text accordingly.

Manuscript Revision: CO2 removed from the text

line 142: "...a supply of NO gas..." is not always needed - line 213 refers to one implementation of the "scrubberless" UV absorption method uses a supply of N2O gas and produces NO by photolysis.
Response: The authors agree with this suggestion and to clarify will rewrite the sentence to read: “The SL-UV method requires a continuous supply of compressed NO or nitrous oxide (N2O) (which the instrument converts to NO) to serve as the scrubber gas.

Manuscript Revision: Sentence rewritten to read “Similar to NO-CL, the SL-UV method requires a continuous supply of compressed NO or nitrous oxide (N2O) (which the instrument converts to NO) to serve as the scrubber gas.

lines 230 - 237: Details of power, generator, charger, and batteries are tangential to the performance of the analyzers and could be eliminated to shorten the text.
Response: The authors agree with this suggestion and will review the manuscript and eliminate non-relevant text that will shorten the document.

Manuscript Revision: Details of power, generator, charger, and batteries and other non-relevant material removed from the manuscript text.

lines 267-8: "...calibrations for THC were performed using... a methane/propane gas cylinder..." This work eventually concludes that VOCs are "likely to interfere with UV absorption measurements of O3", no surprise there. What is surprising is the rudimentary approach to quantifying those VOCs in this manuscript. FID response factors vary with carbon number (for example, by up to a factor of 3 between methane and propane!), between aliphatic, aromatic, and cyclic structures, and with heteroatomic functionality. A sentence noting the uncertainty introduced in their measurement of VOCs (here called THC) by using only methane and propane to determine FID sensitivity would be appropriate here.
Response: The authors agree with this comment and will add a sentence to address the uncertainty associated with our use of the THC method and its calibration procedure to approximate VOC concentrations.

Manuscript Revision: The THC calibration text was rewritten as follows to emphasize that the THC results are an approximation of THC concentration in smoke “Per the manufacturer provided operators manual, calibrations for THC were performed using the T700U calibrator and a certified EPA methane/propane gas cylinder (Airgas). FID response factors for organic compounds can vary significantly based upon factors such as carbon number and compound class (Tong and Karasek 1984). The carbon numbers for methane and propane vary by a factor of three and the FID response factors for those compounds may also vary by a similar amount. In addition, the complex mixture of hydrocarbons found in smoke will have large variations in carbon number and FID response factors. As such, the results obtained with the THC analyzer are an approximation of THC (and VOC) concentrations in smoke.

In
addition, for THC calibrations, the T701H zero air generator was replaced with scientific grade zero air compressed gas cylinders (Airgas).”

Figure 2: This is not a good graphic. There is absolutely no information conveyed by the third dimension of this graph; please turn this into a 2D bar graph and improve the legibility of the different hatches. The high level of interference from the UV-C and UVC-H techniques overwhelms any useful information on the other techniques - suggest plotting only to 50 ppb and annotating the UV-C maxima with text. These data are presented as O3 in ppb - what is the correct, or expectation value? The NO-CL data are lost in this presentation and should be emphasized as the correct value.

Response: The authors agree the reviewers comment. The figure will be reformatted into 2D and assuming that AMT allows colored figures will include a color scheme to improve clarity and viewability. In addition, the y axis scale will be reduced to 50 ppb and the average values for all methods will be included in the figure as text. The figure caption will be revised to reflect these changes.

Manuscript Revision: Figure 2 was reformatted into 2D and a color scheme added to improve viewability. The y-axis scale was capped at 50 ppb and the average values for all methods and study periods were included as text in the figure.

Figure 4: The NO-CL reference trace in the upper figure is the hardest to see; these figures could use some work for legibility. The text refers to positive artifacts for the UV methods during burning periods, ascribed to interferences from VOCs and PM2.5. Another problematic feature is the negative artifact when the chamber is flushed with outside air, where the UV-C method falls below the NO-CL method (bottom panel). Why is that? Did I miss the explanation?

Response: The authors will work on this time series as well as others to make the figures more legible including looking into using a different scale on the y-axis. The post burn calibration checks on April 23, 2018 revealed a +8 % bias in the NO-CL method and a -2 % bias in the UV-C-H method. These biases were evident during the chamber flush periods on that day. Each analyzer was re-zeroed and spanned resulting in the elimination of the bias between the two methods as observed in the results from the subsequent day (April 24, 2018). This will be addressed in the figure caption.

Manuscript Revision: Figure 4 was reformatted to include a logarithmic scale for O3 concentrations making comparisons between the different methods more clear. The following text was added to the figure caption to address the bias observed during the chamber flush periods “The post burn calibration checks on April 23, 2018 revealed a +8 % bias in the NO-CL method and a -2 % bias in the UV-C-H method. These biases were evident during the chamber flush periods on that day. Each analyzer was re-zeroed and spanned resulting in the elimination of the bias between the two methods as observed in the results from the subsequent day (April 24, 2018).”

Lines 378-388: I could not follow the confusing thread discussing how and when the MnO2 scrubber failed in these experiments - for clarity I’d recommend deleting this section and removing all data taken with an inoperative scrubber.
Response: The scope of this paper is a comparison/evaluation of ozone monitoring methods in smoke and the damage to the converter occurred while operating the UV-C analyzer in heavy smoke, the authors feel that this potential measurement issue is very important to those utilizing these instruments and should at a minimum be mentioned in this manuscript. The converter issue is important in that the effect continuous long after the smoke exposure is over and is not obvious when conducting typical QA/QC reviews (e.g., zero/span calibrations and checks). The authors will add/remove text to clarify when the damage occurred and the impact that the damaged converter had on the results obtained with the UV-C method.

Manuscript Revision: The authors clarified some text in this section but feel the section is well explained as to when the damage occurred and the overall impact. The section now reads “During the 2018 chamber burns the UV-C results were biased high by 15-20 ppb even during non-burn (i.e., overnight) periods as evident in Fig. 4 (top panel) and Fig. S4. The initial hypothesis was that the bias was associated with high chamber backgrounds of interfering species due to years of heavy burning in the chamber. However, it was later discovered during a subsequent summer/fall 2018 ambient air study in North Carolina in the absence of smoke, that sampling heavy smoke plumes during the fall 2017 prescribed grassland burns irreversibly damaged the MnO2 scrubber in the UV-C instrument. The effect of the bias was observed mainly when sampling ambient air and not readily observed during routine calibration checks (zeroes and spans) except for an increase in the time required to obtain stable zero and span values. During the summer/fall 2018 North Carolina study and prior to the start of the 2019 chamber burns, a new MnO2 scrubber was installed and resulted in a significant and immediate reduction of the observed high bias, shown in Fig. 4 (bottom panel) and Fig. S5.”

Table 3: Since there appears to be very large fire-to-fire and technique-to-technique variability in the interferences, with no consistent dependence on any of the variables measured, quantifying their precise values in a table seems not very useful. I’m not sure what information this table provides; what quantitative use is it? Recommend deleting.

Response: The authors disagree with this comment. Regardless of the burning conditions or techniques used, artifacts in the UV photometric methods were observed and are presented in this table. The authors intend to include Table 3 in the manuscript.

Manuscript Revision: None

line 498: This section recommends using Nafion dryers to minimize smoke interferences in UV absorption ozone measurements. This begs the question - under what range of conditions does the use of a Nafion dryer allow EPA to actually accept an ozone measurement by the UV absorption measurement? Please discuss.

Response: This comment goes beyond the scope of this paper which is primarily focused evaluation/comparison of ozone monitoring methods in smoke plumes. However, the authors intend to include an additional implication section that will discuss the potential impact of our findings on real world monitoring application at sites that might be impacted by nearby wildfire smoke plumes.
Manuscript Revision: An implication section was added immediately preceding the conclusion section that discusses the potential impact of our findings on real world monitoring application at sites that might be impacted by nearby wildfire smoke plumes.

Table 4: Same comment as for Table 3, above: "Since there appears to be very large fire-to-fire and technique-to-technique variability in the interferences, with no consistent dependence on any of the variables measured, quantifying their precise values in a table seems not very useful. I’m not sure what information this table provides; what quantitative use is it? Recommend deleting."

Response: The authors disagree with this comment. Regardless of the burning conditions or techniques used, artifacts in the UV photometric method were observed and those artifacts are correlated with makers of combustion as illustrated in this table. The authors intend to include Table 4 in the manuscript.

Manuscript Revision: None

Line 581: I would suggest the authors review and cite the use of perfluorosulfonate membrane tubing to remove UV-active hydrocarbons, e.g., in SO2 pulsed fluorescence instruments (Luke, W., 1997, JGR, 102, 16,255-16,265).

Response: The authors will review the suggested manuscript and if appropriate cite in the text as a possible solution in mitigating interferences by wildfire generated UV-active hydrocarbons as suggested by the reviewer.

Manuscript Revision: None. The authors reviewed the suggested manuscript and choose not to cite it in this manuscript. The authors could not find mention of perfluorosulfonate membrane in the manuscript which is similar to the make up of Nafion but did notice several instances of the proprietary “kicker” that may or may not remove interfering hydrocarbons.

Reviewer 2 Comments – Responses and Manuscript Revisions

General Comments: This study compares O3 measurement techniques in fresh, concentrate smoke plumes. The authors sample smoke plumes from both prescribed prairie grass burns and controlled chamber burns using a NO chemiluminescence measurement as the interference-free standard with which to compare several iterations of UV absorption-based measurements. This study is motivated by the prevalence of UV-based O3 analyzers at EPA air quality monitoring stations and the increasing impact of fire emissions on local and regional air quality. Although these comparisons provide insight into the potential for UV-active VOCs in smoke plumes to generate positive artifacts in the UV-based O3 measurements, a more quantitative assessment is limited by the lack of detailed VOC measurements and the inability to quantitatively disentangle the various CO-O3 regimes. The authors also suggest the role of Nafion in mitigating potential artifacts, but do not provide enough information on the relative humidity conditions during the various sampling periods or the potential for interactions between water vapor and VOC. Further, the analysis emphasizes the effects of VOC interreferences in near-fire smoke plumes but does
not provide much discussion on how the potential for interference diminishes with plume age and dispersion. For example, how quickly do VOC react/diffuse to the point where their levels are no longer of concern? How many ozone monitoring sites would be practically affected by these interferences?

Response: The authors appreciate the time required to provide the review and feel that the suggestions provided by the reviewer will result in an improved manuscript for resubmission.

During both the prescribed and chamber burns, data were obtained for RH values and water vapor concentration and is included in the data associated with this paper that will be provided through the EPA Science Hub Web site (https://catalog.data.gov/dataset/epa-sciencehub) following the acceptance of this paper. However, the correlations between RH and the magnitude of the ozone artifact were not significant and therefore not included in the manuscript. In general, both the prescribed fire and chamber burns were conducted under dry conditions with RH ≤ 50%. Past studies, which are now referenced in the updated manuscript indicate that at those RH values humidity effects are expected to have little to no impact. It is the intention of the authors to add an additional section to this manuscript discussing implications of this research on real world ozone monitoring such as that occurs at State and local monitoring sites. The authors intend to review data from sites downwind of wildfires that potentially show the artifact in the UV-C O3 method and how it is correlated with markers of combustion processes. As stated in the text of the manuscript, the authors plan future studies to dig deeper into the hypothesized VOC caused artifact and which will include, as the reviewer suggest looking into interaction between VOCs and water vapor and the capabilities of Nafion in removing certain VOCs.

Manuscript Revision: An additional “Implications” section was added prior to the conclusion to tie the results of the research detailed in this manuscript to real world monitoring applications. Included in this section will be a review of data from monitoring sites downwind of fires to show the impact of the measurement artifacts described in this manuscript.

Specific Comments:

L243-244: Is there any dependence of the artifact magnitude on distance from the active fire line? How quickly do the VOC react/diffuse to the point where their levels are no longer detectable as a positive artifact? All the measurements presented are taken within ~100 m from the fires, but any data collected from aged smoke would be a useful counterpoint.

Response: The authors did not look at the dependencies of the artifact magnitude on distance from the active fire line. However, the authors do agree that a more detailed look at data collected at sites being impacted by aged smoke (ex. State and local monitoring sites being impacted by nearby wildfires). This would aid in tying these measurements made in or near plume back to real world monitoring situations. Most likely this will be done by adding an implications section prior to the manuscript conclusion.

Manuscript Revision: An implications section was added prior to the conclusion to address some of reviewer 2 comments.
L262: The authors mention a +/- 10% performance objective between analyzers. Do the calibrations reveal any systematic offset between the CL and UV analyzers? In describing the prescribed and chamber burns, the authors mention varying moisture content in the burn material. Did the authors observe whether the wetter grasses produced more VOC (lower combustion efficiency) in any systematic way?

Response: The calibrations only revealed a significant offset during one period during this study. The post burn calibration checks on April 23, 2018 revealed a +8 % bias in the NO-CL method and a -2 % bias in the UV-C-H method. These biases were evident during the chamber flush periods on that day. Each analyzer was re-zeroed and spanned resulting in the elimination of the bias between the two methods as observed in the results from the subsequent day (April 24, 2018). All other calibrations did not reveal any systematic offsets or biases between the different analyzers and we will clarify this in the updated version of the manuscript. At present the authors have not investigated the relationship between fuel moisture content and VOC production. In order to simulate a range of natural burning conditions, the chamber burns manipulated the moisture content, fuel type (pine needles, pine needles + fine woody debris), and bulk density of the fuelbeds. These fuelbed properties influence the relative mix of flaming and smoldering combustion and the chamber burns covered a range of combustion efficiencies (modified combustion efficiencies of 0.85 – 0.97). The authors will investigate further and address these findings in a future manuscript.

Manuscript Revision: The following text was added to the figure caption to address the bias observed during the chamber flush periods “The post burn calibration checks on April 23, 2018 revealed a +8 % bias in the NO-CL method and a -2 % bias in the UV-C-H method. These biases were evident during the chamber flush periods on that day. Each analyzer was re-zeroed and spanned resulting in the elimination of the bias between the two methods as observed in the results from the subsequent day (April 24, 2018).” The following text was also added to section 3.2 “The post burn calibration checks on April 23, 2018 revealed a +8 % bias in the NO-CL method and a -2 % bias in the UV-C-H method. These biases were evident during the chamber flush periods on that day. Each analyzer was re-zeroed and spanned resulting in the elimination of the bias between the two methods as observed in the results from the subsequent day (April 24, 2018).” No other calibration corrections were made during the 2018 and 2019 chamber studies.

Figure 4: In general, the scale mismatch on the O3 timeseries makes immediate comparison between methods difficult. The authors should perhaps switch to a log-scale on the y-axis that can effectively compare low and high concentrations and offsets in both smoke plumes and background air. The authors attempt to explain the positive offset of the UV-C method outside of the burning period, but there is also a significant negative offset in the UV-C-H method that is not discussed. Could the authors provide more insight on why the UV-C-H and NO-CL techniques disagree in background air?

Response: The authors will work on this time series plot as well as others to make the figures more legible including looking into using a different scale on the y-axis. As suggested by the reviewer, the authors will provide more insight into why the UV-C-H and NO-CL techniques disagree in background air.
Manuscript Revision: Figure 4 was reformatted to include a logarithmic scale for O₃ concentrations making comparisons between the different methods more clear.

L378+: If the damaged MnO₂ scrubber ineffectively removed O₃, I would expect the UV-C measurement to be biased low in background air rather than high. Please elaborate on the mechanism of MnO₂ damage resulting in a significant positive offset. Also, it's unclear when the scrubber damage became an issue. Did it affect data from the 2017 prescribed burns?

Response: In order for the scrubber to work correctly, it must remove O₃ and only O₃. Based upon the data, the damage most likely resulted in the scrubber also removing significant amounts of interfering species during the reference measurement which would then be detected as ozone during the sample measurement resulting in the positive artifact. The data collected during the 2017 prescribed burns indicate that the scrubber was functioning properly in that there was excellent agreement between the UV-C and NO-CL methods when sampling out of the smoke plume.

Manuscript Revision: To clarify the section describing the bias observed during the 2018 chamber studies was re-written as follows: “During the 2018 chamber burns the UV-C results were biased high by 15-20 ppb even during non-burn (i.e., overnight) periods as evident in Fig. 4 (top panel) and Fig. S4. The initial hypothesis was that the bias was associated with high chamber backgrounds of interfering species due to years of heavy burning in the chamber. However, it was later discovered during a subsequent summer/fall 2018 ambient air study in North Carolina in the absence of smoke, that sampling heavy smoke plumes during the fall 2017 prescribed grassland burns followed by subsequent storage of the UV-C analyzer, irreversibly damaged the MnO₂ scrubber in the UV-C instrument. It is hypothesized that the damage resulted in the scrubber removing some of the interfering species in addition to ozone, preventing them from being removed in the reference measurement, and subsequent detection as ozone (positive bias) during the measurement cycle. The effect of the bias was observed mainly when sampling ambient/chamber air and not readily observed during routine calibration checks (zeroes and spans) except for an increase in the time required to obtain stable zero and span values. The bias was not observed during any of the 2017 prescribed grassland burns. During the summer/fall 2018 North Carolina study and prior to the start of the 2019 chamber burns, a new MnO₂ scrubber was installed and resulted in a significant and immediate reduction of the observed high bias, shown in Fig. 4 (bottom panel) and Fig. S5.”

Figure S9 indicates there is potential artifact even <1-2 ppm CO. Do these plots just use data from the burn periods or include points when chamber is flushed with outside air?

Response: Figure S9 includes data from the burn periods only. In the figure caption it describes it as “in-plume”. The authors will add clarifying text similar to the following, “…and THC for all in-plume (burn period only) measurements…”.

Manuscript Revision: The figure caption was re-written as follows: “Scatter plots between FRM and FEM O₃ differences and CO, NO₂, and THC for all in-plume (burn period only) measurements made during the 2018 and 2019 Missoula Fire Chamber studies. Observation points have been colored by the O₃ instrument. Over all observations there is little correlation between the O₃ instrument differences,
but straight line structures within the overall scatters indicate that individual burn events measured in
the chamber have good correlations with distinct ratios.”

**L459-461:** How does the residence time and sample rate vary for each instrument?

**Response:** Sampling rates and hence residence times are going to be similar for all instruments as they
all operate with similar flow rates. The authors will address this comment by either adding analyzer flow
rate to Table 1 or by inserting text in the Methods section under each corresponding analyzer type.
Generally, UV photometric type analyzers require a greater flow rate becaust the flow is split between
the two cells (reference and measurement). The NO-CL method has only a single cell and requires a much
smaller flow rate to achieve a similar residence time.

**Manuscript Revision:** The flow rates of each method along with manufacturer reported performance
specifications were included in Table S1 which was added to the supplementary materials document. In
the text describing each method, a sentence similar to the following was added “Manufacturer provided
performance specifications for the NO-CL based TAPI T265 are given in Table S1.”

**Table 4:** The slope and intercept uncertainties should be included with the fit parameters. How different
are the range of fitted slope values statistically? In general, there is lack of uncertainty treatment in the
paper. How do the uncertainties compare for each measurement technique? This information should be
included in the manuscript.

**Response:** The authors agree with this comment and will work to include uncertainties (both in tables
and in the text) of measurement methods and in fit parameters associated with regression statistics.

**Manuscript Revision:** Data for the Konza March 2017 were re-analyzed and new values included fro
slope, intercept R2 and n. The previous analysis included a few values that were associated with CO
levels that were below 1 ppm (our threshold of sampling in plume). Standard errors for the regression
slope and intercept were included in Table 4. In addition, the following text was added to discuss the
results of the regression analysis between markers of combustion CO and THC and the magnitude of the
ozone artifact: “The slight differences in the magnitude of the artifacts (fitted regression slopes) along
with the low uncertainty (standard errors) values indicate that the magnitude of the artifact may be
influenced by local conditions that make each burn unique. Such conditions might include meteorological
conditions, fuel composition, fuel moisture content, and times spent in combustion phase (flaming vs
smoldering).”

**L550-552:** See question 1 above. How close to the plume do you have to be for interferences to matter?
Is this relevant for air quality monitoring stations not located int he immediate vicinity of the fire line?

**Response:** The authors focused on determining if significant ozone measurement artifacts do occur in
near-field smoke events and did not look at the dependencies of the artifact magnitude as a function of
distance from the active fire line. However the authors do agree that a more detailed look at data collected
at sites being impacted by aged smoke (ex. State and local monitoring sites being impacted by nearby
wildfires) and are currently collecting this data as part of the EPA MASIC study in Boise, ID; Missoula,
MT; and Reno, NV. This additional data collection will aid in linking these research chamber and near
field prescribed grassland burn measurements back to real world regulatory monitoring situations. We will address these issues in a new “implications” section prior to the manuscript conclusion.

**Manuscript Revision:** An implications section was added to the manuscript prior to the conclusion to address this and other comments provided by reviewer 2.

**L554:** What is estimated CO-ΔO3 correlation for the chamber studies? It would still be worthwhile to include this information in the supplement.

**Response:** Regarding the correlation between ΔO3 and CO from the chamber based burns, the authors refer the reviewer to the original manuscript text:

“As indicated, ΔO3(UV-C) and CO appear to be correlated in time but when performing linear regression comparisons of ΔO3(UV-C) and CO during each years chamber burns as a whole, correlations tend to be poor. We suspect the positive O3 bias is driven by one or more VOCs (likely oxygenated VOCs). In fresh smoke the excess concentrations of individual VOCs (ΔX), and VOC sums (ΔVOC), tend to be highly correlated with ΔCO (Yokelson et al., 1999; Gilman et al. 2015). The emission ratios of individual VOCs to CO (ΔX/ΔCO) can vary considerably with combustion conditions such as fuel type and condition (e.g. moisture content and decay state), fuel bed properties, such as bulk density, and the relative mix of flaming and smoldering combustion (Gilman et al. 2015; Koss et al., 2017). Additionally, the response of ΔX/ΔCO to burn conditions varies among VOCs. When each burn is considered individually or in groups with similar conditions, the correlations between ΔO3, CO, and THC are enhanced. An example of this behavior is shown in Supplementary Fig. S10.”

With that being stated, the authors will consider adding the CO-ΔO3 correlation (both for the entire chamber study period and also a subset of individual burns) either in Table 4 or in the body of the text give evidence to the above statement. Visual representations of the correlations are given in Figures S9 and S10.

**Manuscript Revision:** The following text was added to section 3.4 to address this comment: “For the chamber burns the magnitude of the ozone artifacts in ppb apparent O3 per ppm CO, ranges between 6 - 210 ppb ppm⁻¹ for the individual burns. R² and standard error values were consistent with those observed during the prescribed burns (see Table 4).” In addition, the requested information is provided visually in figures S9 and S10.

**Figures S9 and S10:** Can you demonstrably separate CO-ΔO3 regimes based on “burn condition”? The authors allude to this in the text (L563) and show an individual burn in Fig S10, but a more in-depth analysis of the contributing burn condition factors would provide a more quantitative and perhaps predictive assessment of how CO links to O3 artifacts under the varied burn conditions. The authors also perform separate regressions for NO2 and THC, but a separate correlation with humidity might be illustrative (if the data exists).

**Response:** The authors will consider elaborating further per the reviewers suggestion on CO-ΔO3 regimes based on burn conditions (i.e., individual burns or burns grouped by similar burn conditions). The authors previously attempted to establish a correlation between ΔO3 and humidity (water vapor
concentration) but those correlation were extremely poor. As such the authors chose not to include this
analysis.

**Manuscript Revision:** The following text was added to section 3.4 to elaborate on the lack of correlation
between ΔO3 and CO when considered as a whole but showing improvements when considering
individual burns: “For the chamber burns the magnitude of the ozone artifacts in ppb apparent O3 per ppm
CO, ranges between 6 - 210 ppb ppm⁻¹ for the individual burns. R² and standard error values were
consistent with those observed during the prescribed burns (see Table 4).”

**L571:** Is it possible that interactions between water vapor and VOC somehow compound the VOC effect?
In other studies (e.g., Spicer et al. 2010, Turnipseed et al. 2017), Nafion alone seems to play little role in
mitigating VOC artifacts but does significantly reduce water vapor artifacts. In drier environments, does
adding Nafion affect the positive artifact magnitude? This would be more conclusive evidence that Nafion
does in fact remove certain permeable VOC species.

**Response:** Both the 2017 prescribed fire and 2018-2019 chamber based burns were conducted under dry
conditions (RH ≤ 50%) and humidity interferences are expected to be minimal. As stated in the previous
comment, the correlation between in plume water vapor concentration and ΔO3 was not significant. In
addition, there is no significant correlation between the magnitude of the artifact and RH. In both the
prescribed grassland and chamber burns there was a UV instrument with a Nafion drier and a UV
instrument without the drier and they were operated simultaneously. The magnitude of the artifact (both
average and maximum) was greatly reduced in the method using the Nafion drier. This is evident in
comparing the magnitude of the UV-C artifact with that of the UV-C-H (UV method employing a Nafion
based drying system. In all cases, the UV-C artifact was nearly an order of magnitude greater than that of
the UV-C-H. This is also became further evident when the Nafion drier was added to the UV-C method
on the final day of burning during the 2018 chamber studies, thus reducing the magnitude of the UV-C
artifact to a point comparable to that of the UV-C-H method. The effect of Nafion on the magnitude of
the artifact is detailed in section 3.3. In section 3.5 of the manuscript, the authors will attempt to clarify
that in addition to our hypothesis of certain VOCs being removed by the Nafion, there may also be
interactions between water vapor and VOCs that may be confounding the observed artifact.

**Manuscript Revision:** The authors feel that text and discussion provided in section 3.3 already provide
a response to the reviewer 2’s comment suggestion. As stated in the response listed above, during this
study humidity effects are expected to be at a minimum due to the low RH values that existed during all
study periods. As such and to clarify, the following text was inserted in section 2.6: “In general, chamber
RH values were below 50% facilitating dry burning condition.” And section 3.1: “In addition, ambient
RH values were generally below 50% indicating that the spring and fall 2017 prescribed burns were
conducted under dry conditions.”

The last sentence of section 3.4 was re-written to read “Considering that the prescribed grassland and
chamber burns were conducted under dry conditions, the size of the difference (as large as hundreds of
ppb) cannot be explained purely by the previously observed relative humidity effects on measurements
(Leston et al., 2005; Wilson et al., 2006), suggesting that the Nafion® dryer is directly impacting the
concentrations of other interferents in the sample stream.”
Could this also be confounded by the faulty MnO2 scrubber?

Response: We do know that during the 2018 chamber studies the damaged scrubber did cause an approximate +10-15 ppb bias in the UV-C method which was present even in the absence of smoke. At the end of the 2018 chamber studies, the authors added a Nafion drier to the UV-C method as indicated in Figure 4. The addition of the Nafion to the UV-C method reduced the magnitude of the artifact by a factor of three making it compatible to the artifact observed for the UV C-U method. The addition of the nafion did result in a slight reduction in the bias that we attributed to damaged scrubber but not on the order of 3X. W suspect that the addition of the drier would reduce or remove many of the VOC species prior to also being removed by the faulty scrubber thus resulting in a reduction of the bias but not completely eliminating it. The authors will add clarifying text in the body of the manuscript explaining the damage to the MnO2 scrubber and its hypothesized effect on the observed bias. The reviewers comment would only apply to the 2018 chamber study as the MnO2 scrubber in the UV-C method was functioning properly during all other studies.

Manuscript Revision: Clarifying text was added in section 3.2 to explain the effect that the damaged scrubber had on the UV-C ozone results (positive bias).

Technical Corrections:

Table 1: Add uncertainty associated with each measurement technique. Sample rate would also be useful.

Response: The authors will address this comment by either adding analyzer flow rate and uncertainties to table 1 or by inserting text in the Methods section under each corresponding analyzer type.

Manuscript Revision: An additional table (Table S1) was added to the supplemental materials document containing manufacturer provided performance specifications for each analyzer to address this comment from reviewer 2. In the text describing each method, a sentence similar to the following was added: “Manufacturer provided performance specifications for the NO-CL based TAPI T265 are given in Table S1.”

Figure S1 and other timeseries in general: It’s difficult to compare NO-CL and UV measurements of plumes and background air given the large mis-match in scale. Some other way of presenting this material (e.g., semi-log) might help the visual comparison. The lines are also not very easy to distinguish. Using different colors instead of just patterns would help.

Response: The authors agree with this comment and will take steps to improve the time series plots, including looking into different scales (e.g. semi-log) and also using colored lines in the figures.

Manuscript Revision: Figures 4 and S1-5 were reformatted adding logarithmic scales where appropriate and color schemes to improve readability.

Figure 2: Does not need to be in 3D and could use a color scheme instead of patterns.
Response: The authors agree with the reviewers comment. The figure will be reformatted into 2D and assuming that AMT allows colored figures will include a color scheme to improve clarity and viewability. In addition, the y axis scale will be reduced to 50 ppb and the average values for all methods will be included in the figure as text. The figure caption will be revised to reflect these changes.

Manuscript Revision: Figure 2 was reformatted into 2D and a color scheme added to improve viewability. The y-axis scale was capped at 50 ppb and the average values for all methods and study periods were included as text in the figure.
Comparison of Ozone Measurement Methods in Biomass Burning Smoke: An evaluation under field and laboratory conditions

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Abstract

In recent years wildland fires in the United States have had significant impacts on local and regional air quality and negative human health outcomes. Although the primary health concerns from wildland fires come from fine particulate matter (PM₂.₅), large increases in ozone (O₃) have been observed downwind of wildland fire plumes (DeBell et al., 2004; Bytnerowicz et al., 2010; Preisler et al., 2010; Jaffe et al., 2012; Bytnerowicz et al., 2013; Jaffe et al., 2013; Lu et al., 2016; Lindaas et al., 2017; McClure and Jaffe, 2018; Liu et al, 2018; Baylon et al., 2018; Buysse et al. 2019). Conditions generated in and around wildland fire plumes, including the presence of interfering chemical species, can make the accurate measurement of O₃ concentrations using the ultraviolet (UV) photometric method challenging if not impossible. UV photometric method instruments are prone to interferences by volatile organic compounds (VOCs) that are present at high concentrations in wildland fire smoke. Four different O₃ measurement methodologies were deployed in a mobile sampling platform downwind of active prescribed grassland fire lines in Kansas and Oregon and during controlled chamber burns at the United States Forest Service, Rocky Mountain Research Station Fire Sciences Laboratory in Missoula, Montana. We demonstrate that the Federal Reference Method (FRM) nitric oxide (NO) chemiluminescence monitors and Federal Equivalent Method (FEM) gas-phase (NO) chemical scrubber UV photometric O₃ monitors are relatively interference-free, even in near-field combustion plumes. In contrast, FEM UV photometric O₃ monitors using solid-phase catalytic scrubbers show positive artifacts that are positively correlated with carbon monoxide (CO) and total gas phase hydrocarbons (THC), two indicator species of biomass
burning. Of the two catalytic scrubber UV photometric methods evaluated, the instruments that included a Nafion® tube dryer in the sample introduction system had artifacts an order of magnitude smaller than the instrument with no humidity correction. We hypothesize that Nafion®-permeable VOCs (such as aromatic hydrocarbons) could be a significant source of interference for catalytic scrubber UV photometric O₃ monitors, and that the inclusion of a Nafion® tube dryer assists with the mitigation of these interferences. The interference-free chemiluminescence FRM method is highly recommended for accurate measurements of O₃ in wildland fire plume studies and at regulatory ambient monitoring sites frequently impacted by wildland fire smoke.

1 Introduction

Ground-level ozone (O₃) is a secondary air pollutant generated from the photochemical interactions of nitrogen oxides (NOₓ) and volatile organic compounds (VOCs). The most robust methods for interference-free O₃ measurements are based on chemiluminescence reactions with ethylene (ET-CL, for ethylene chemiluminescence) or nitric oxide (NO-CL, for nitric oxide chemiluminescence) (Long et al., 2014). The overall reaction mechanism for ET-CL generally proceeds as detailed in Eqs. (1-2):

C₂H₄ + O₃ → H₂CO* + Other products, (1)

H₂CO* → H₂CO + hv (2)

The reaction generates electronically-activated formaldehyde (H₂CO*) which luminesces in the high ultraviolet (UV) to visible portion of the spectrum (380 nm - 550 nm) and vibrationally-activated hydroxide ions which luminesce in the visible light to the low infrared (IR) portion of the spectrum (550 nm - 800 nm). The number of photons emitted during the reaction is directly proportional to the amount of O₃ present and are counted by a photomultiplier tube (PMT), with its response centered at 440 nm, then the count is converted to O₃ concentration. The ET-CL method requires a constant supply of ethylene for continuous operation. NO-chemiluminescence analyzers measure O₃ concentrations using the principle that the dry, gas-phase reaction between NO and O₃ generates nitrogen dioxide in an electronically excited state (NO₂*), and oxygen (O₂)
(Ollison et. al., 2013; Boylan et al., 2014). As each unstable, NO\(_2^*\) molecule returns to a lower energy state (NO\(_2\)), it emits a photon (hv). The reaction causes luminescence in a broadband spectrum ranging from visible light to infrared light (approximately 590 nm – 2800 nm). The two-step gas-phase reaction proceeds as detailed in Eqs. (3-4):

\[
\text{NO} + \text{O}_3 \rightarrow \text{NO}_2^* + \text{O}_2, \quad (3)
\]

\[
\text{NO}_2^* \rightarrow \text{NO}_2 + \text{hv}. \quad (4)
\]

The ET-CL method is no longer used nor produced commercially and has been replaced by the NO-CL method. Similar to the ET-CL method, the NO-CL method requires a constant supply of gas, in this case NO, for continuous operation. Both the ET-CL and NO-CL methods are subject to slight interferences by water vapor. However, these potential interferences can be eliminated through the use of Nafion based drier or equivalent sample water vapor treatment system. The ET-CL method was promulgated as the Federal Reference Method (FRM) for measuring O\(_3\) in the atmosphere in 1971 and the NO-CL method promulgated as the FRM in 2015 (U.S. EPA, 2015).

While the chemiluminescence method for measuring O\(_3\) is technically robust and free of analytical artifacts (Long et al., 2014), it is not widely used in the United States. Instead, Federal Equivalent Methods (FEM) based upon UV photometry are employed at the majority of O\(_3\) regulatory monitoring locations. According to July 2020 data from the United States Environmental Protection Agency (EPA) Air Quality System (AQS) database, the UV photometric method represents 99% of the roughly 1200 instruments deployed in network monitoring for O\(_3\) National Ambient Air Quality Standard (NAAQS) attainment. UV photometric methods for O\(_3\) are generally considered easier to deploy, operate, and in most cases do not require external compressed gasses for operation. UV photometric analyzers determine O\(_3\) concentrations by quantitatively measuring the attenuation of light due to absorption by O\(_3\) present in an absorption cell at the specific wavelength of 254 nm (Parrish et al., 2000; Williams et al., 2006). The O\(_3\) concentration is determined through a two-step process in which the light intensity passing through the sample air (I) is compared with the light intensity passing through similar sample air from which all O\(_3\)
is first removed \((I_0)\). The ratio of these two light intensity values \((I/I_0)\) provides the measure of the light absorbed at 254 nm, and the \(O_3\) concentration in the sample is then determined through the use of the Beer-Lambert Law as given in Eq. (5):

\[
\frac{I}{I_0} = e^{-KLC} (C = 1/KL \ln (I/I_0));
\]

(5)

where \(L\) is the length of the absorption cell (cm), \(C\) is the \(O_3\) concentration (ppm), and \(K\) is the absorption cross section of \(O_3\) at 254 nm at standard atmospheric temperature and pressure conditions \((308\ atm\cdot cm^{-1})\). Photometric monitors generally use mercury vapor lamps as the UV light source, with optical filters to attenuate lamp output at other than the 254 nm wavelength.

Air for the reference cell measurement \((I_0)\) is typically obtained by passing the ambient air sample stream through a catalytic scrubber containing manganese dioxide \((MnO_2)\), hopcalite \((a\ mixture\ of\ Cu,\ Mn,\ and\ Ag\ oxides)\), heated silver wool, or another solid state material to ‘scrub’ only \(O_3\) from the sample air while preserving all other substances in the sample air that potentially absorb at 254 nm \((e.g.,\ elemental\ gaseous\ mercury\ [Hg^0],\ hydrogen,\ sulfide\ [H_2S],\ carbon\ dioxide\ [CO_2],\ VOCs)\) so that their effects are cancelled in the differential \(I/I_0\) measurement. The integrity of the \(O_3\) reference scrubber is critical and may allow measurement interferences if it does not perform adequately. Similarly, any tendency of the scrubber to fail to effectively remove all \(O_3\) from the reference sample will result in a measurement bias. In addition to \(O_3\), catalytic scrubbers have been shown to remove UV-active VOCs \((\text{Kleindienst et al., 1993})\), creating the potential for positive artifacts in \(O_3\) measurements when the efficiency of this VOC removal is impacted.

Although FEM designated UV photometric instruments are accurate under most ambient conditions, locations with high VOC concentrations can produce significant analytical artifacts. Smoke plume impacted locations and measurements downwind from wildland fires are a particular concern; \(O_3\) measurements of up to 320 ppb were observed in a smoke plume in western Oregon using a Dasibi 1003AH UV photometric \(O_3\) monitor \((\text{Huntzicker and Johnson, 1979})\), which also showed a correlation...
between apparent O$_3$ and aerosol concentrations ($b_{scat}$, a combustion plume indicator in this case). O$_3$ measurements from UV photometric instruments exceeding 1500 ppb at night (22:00-05:00) were observed in Fort McMurray, Alberta during smoke impacts from the 2016 Horse River Fire, which were positively correlated with NO and non-methane hydrocarbons (Landis et al., 2018). Follow-up pyrolysis experiments demonstrated that ET-CL instruments do not show a similar response to biomass burning smoke (Huntzicker and Johnson, 1979). Photochemical chamber experiments comparing the O$_3$ response of UV (Dasibi Model 1003AH, Dasibi Model 1008AH, and Thermo Model 49) and ET-CL (Bendix Model 8002 and Monitor Labs Model 8410) mixtures show negligible differences for irradiated paraffin/NOx and olefin/NOx mixtures, but do show a positive UV interference in mixtures with toluene and other aromatics present (Kleindienst et al., 1993). Additionally, during the Mexico City Metropolitan Area field campaign (MCMA-2003) a mobile laboratory using a FEM designated UV photometric O$_3$ monitor (unheated MnO$_2$ scrubber, Thermo 49 series) showed a large positive O$_3$ interference (~400 ppb) associated with PM$_{2.5}$ and polyaromatic hydrocarbons (PAHs) when following some diesel vehicles (Dunlea et al., 2006). Although not compared to a chemiluminescence instrument, those high O$_3$ values are unlikely real considering the high concurrent NO concentrations (in some cases, >1000 ppb). The authors of this study attributed the interference to fine particles, based on the correlation with PM$_{2.5}$ and the lack of a correlation with gas-phase organic species measured by the proton transfer reaction-mass spectrometer (PTR-MS, Dunlea et al., 2006).

In addition to interferences from the presence of aromatic VOCs and semi-volatile PAHs, water vapor (relative humidity) issues have also been observed with older generation FRM and FEM designated chemiluminescence and UV photometric O$_3$ instruments, respectively (Kleindienst et al., 1993; Leston et al., 2005; Wilson and Birks, 2006). As such, Nafion® tube dryers are regularly incorporated into some newer generation chemiluminescence and UV photometric O$_3$ monitors in an attempt to mitigate the humidity related measurement artifacts.
A recently introduced variation of the UV photometric method, known as the “scrubberless” UV photometric (SL-UV) method (Ollison et al., 2013), specifies removal of $O_3$ from the sample air for the reference by a gas-phase reaction with NO rather than using a conventional solid state catalytic scrubber. The NO gas phase chemical scrubber reacts with $O_3$ much faster and more selectively than with other potential interfering compounds and is very effective at removing the $O_3$ without affecting other interfering compounds that may be present in ambient air. The differential UV measurement can then effectively reduce interferences to an insignificant level. Similar to NO-CL, the SL-UV method requires a continuous supply of compressed NO or nitrous oxide ($N_2O$) (which the instrument converts to NO) to serve as the scrubber gas. Similar to NO-CL, the SL-UV method requires a supply of NO gas for continuous operation.

In this study, we investigate UV photometric FEM instrument $O_3$ measurement interferences in fresh biomass burning smoke plumes from prescribed grassland fires and during controlled burn experiments in a large scale combustion chamber. We directly compare NO-CL FRM $O_3$ measurements to several FEM designated UV photometric technologies, including a gas-phase scrubber and catalytic scrubbers with and without Nafion® tube dryer systems. Based on the results from the measurements, we assess the magnitude of the observed artifacts for different technologies and under various smoke conditions and provide suggestions for potential mitigation of the interferences.

2. Methods

2.1 Overview of Methods Evaluated

In this study we compared the measurement results from six different commercially available FRM/FEM designated $O_3$ instruments operated in ambient or chamber generated biomass burning smoke. All instruments were operated according to their FRM or FEM designation. The six instruments differed by measurement principle (chemiluminescence versus UV photometric), and by sample treatment
configuration (scrubber material, presence of dryer, etc.). For interference free O₃ measurements we utilized the newly designated FRM NO-CL method (U.S. EPA, 2015). For the UV photometric methods, we compared both catalytic scrubber and “scrubberless” (gas phase chemical scrubber) technologies, with the “scrubberless” monitor using a NO chemical scrubber. Finally, within the catalytic scrubber UV photometric category, we compared instruments with and without Nafion tube dryer systems. The operation principle and designations (FRM vs FEM) for the analyzers under investigation are summarized in Table 1 and described in Sections 2.1.1-2.1.4. These analyzers were operated immediately downwind of fresh biomass burning plumes during eight days of prescribed fires in grassland ecosystems in Oregon and Kansas and during laboratory-based studies at the U.S. Forest Service’s (USFS) combustion facility at the Fire Sciences Laboratory (FSL) in Missoula, Montana. The grassland fire fuels consisted primarily of mixed native prairie tall grass of varying moisture content. Seven of the eight days of prescribed burning were conducted in the Tallgrass Prairie ecosystem of central Kansas (four days in March of 2017 and three days in November of 2017). The additional burn day was conducted at the Sycan Marsh in central Oregon (October of 2017). Laboratory based chamber burns at the FSL were conducted during April 2018 and again during April 2019. Fuels for the laboratory based chamber burns consisted of ponderosa pine needles and fine woody debris. Details of the individual studies are provided in Sections 2.2-2.6.

Table 1: Ozone measurement methods investigated.

<table>
<thead>
<tr>
<th>Name</th>
<th>Manufacturer</th>
<th>Model</th>
<th>Method</th>
<th>Scrubber</th>
<th>Cells</th>
<th>Humidity Correction</th>
<th>Deployment*</th>
</tr>
</thead>
<tbody>
<tr>
<td>U.S. EPA Federal Reference Methods (FRM)</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>NO-CL</td>
<td>Teledyne API</td>
<td>T-265</td>
<td>CL (NO)</td>
<td>N/A</td>
<td>1</td>
<td>Nafion®-based (dryer)</td>
<td>K1, S, K2, T, M1, M2</td>
</tr>
<tr>
<td>U.S. EPA Federal equivalent methods (FEM)</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>UV-C</td>
<td>Thermo Scientific</td>
<td>49i</td>
<td>UV (254 nm)</td>
<td>Catalyst (MnO₂)</td>
<td>2</td>
<td>None</td>
<td>K1, S, K2, T, M1, M2</td>
</tr>
<tr>
<td>UV-C-H</td>
<td>2B Technologies</td>
<td>205</td>
<td>UV (254 nm)</td>
<td>Catalyst (Hopcalite)</td>
<td>2</td>
<td>Nafion®-based (equilibration)</td>
<td>K1, S, K2, T, M1, M2</td>
</tr>
<tr>
<td>SL-UV</td>
<td>2B Technologies</td>
<td>211</td>
<td>UV (254 nm)</td>
<td>Gas chemical (NO)</td>
<td>2</td>
<td>Nafion®-based (equilibration)</td>
<td>K1, M1, M2</td>
</tr>
<tr>
<td>UV-G</td>
<td>2B Technologies</td>
<td>211-G</td>
<td>UV (254 nm)</td>
<td>Heated graphite</td>
<td>2</td>
<td>Nafion®-based (equilibration)</td>
<td>M2</td>
</tr>
</tbody>
</table>

*K1-Konza Prairie March 2017; S-Sycan Marsh, October 2017; K2-Konza Prairie November 2017; T-Tallgrass Prairie November 2017; M1-Missoula chamber April 2018; M2-Missoula chamber April 2019.
2.1.1 NO Chemiluminescence

The FRM O₃ measurement method was the interference-free Teledyne API (San Diego, CA, USA) Model T265 Chemiluminescence Monitor (TAPI T265), which utilizes a NO-CL measurement principle. These NO-CL O₃ analyzers have been shown to be free of interferences (Long et al. 2014), and have been used as a reference method in other O₃ comparison studies (Williams et al., 2006; Landis et al., 2020). Although there is a known water vapor interference with chemiluminescence technology (Kleindienst et al., 1993), the TAPI T265 uses a Nafion® tube dryer system to remove water vapor from the air prior to making the measurement, thus eliminating any humidity-related effects. Like the ET-CL technologies (Kleindienst et al., 1993), the NO-CL analyzers have no documented VOC interferences. Manufacturer provided performance specifications for the NO-CL based TAPI T265 are given in Table S1.

2.1.2 Catalytic Scrubber UV Photometric

For this study the UV photometric method with no humidity correction was represented by the Thermo Scientific (Franklin, MA, USA) Model 49i (Thermo 49i), which is a dual cell instrument with a manganese oxide (MnO₂) catalytic scrubber, referred to as UV-C. Nafion®-based humidity systems or dryers have been employed within photometric O₃ monitors with catalytic scrubbers before the measurement cell, offering a reduction in relative humidity interferences and artifacts (Wilson and Birks, 2006). Manufacturer provided performance specifications for the UV-C based Thermo 49i are given in Table S1.

The UV photometric with a Nafion® humidity conditioning system was represented in this study by a 2B Technologies (Boulder, CO, USA) Model 205 (2B 205) O₃ monitor. The 2B 205 utilized a dual-cell design where sample air and scrubbed air are measured simultaneously. The 2B 205 uses a Hopcalite (CuO/MnO₂) catalytic scrubber to remove O₃ from the reference stream. This instrument will be referred
to as UV-C-H. Manufacturer provided performance specifications for the UV-C-H based 2B 205 are given in Table S1.

2.1.3 Scrubberless UV Photometric

For comparison with the NO-CL, UV-C and UV-C-H methodologies, a “scrubberless” UV (SL-UV) photometric analyzer with a gas-phase (NO) chemical scrubber was employed (Ollison et al., 2013; Johnson et al., 2014). The addition of NO gas to the reference stream selectively scrubs O$_3$ while not significantly affecting interfering VOC species, resulting in an interference free O$_3$ determination. Inclusion of this instrument into the study allows evaluation of the impact of the UV method in general (as compared with chemiluminescence) versus the influence of specific scrubber technologies. The SL-UV method is represented by the 2B Technologies Model 211 “Scrubberless” Ozone Monitor (2B 211). The Model 2B 211 requires a continuous supply of compressed NO or nitrous oxide (N$_2$O) (which the instrument converts to NO). The SL-UV method also utilizes a Nafion®-based sample humidity conditioning system to eliminate any humidity effects. The SL-UV instrument was not used in the October or November 2017 burns due to the lack of the necessary reagent gas (nitrous oxide, N$_2$O) needed to run the instrument. Manufacturer provided performance specifications for the SL-UV based 2B 211 are given in Table S1.

2.1.4 Heated Graphite Scrubber UV Photometric

During the final phase of laboratory-based burning a 2B Technologies Model 211-G UV photometric analyzer (2B 211-G) was operated for comparison to the monitors detailed in Sections 2.1.1-2.1.3. The 2B 211-G differs from the 2B 211 in that it employs a heated graphite scrubber to remove O$_3$ from the reference sample stream (I$_0$) (Turnipseed et al., 2017). The 2B 211-G utilizes the same Nafion®-based sample humidity conditioning system as employed in the 2B 211. For the purposes of this manuscript the UV photometric method employing the heated graphite scrubber be refered to as UV-G. Manufacturer provided performance specifications for the UV-G based 2B 211-G are given in Table S1.
2.2 Prescribed Fire Burn Mobile Sampling Platform

During the prescribed fire grass burns, all study instrumentation (analyzers, data acquisition systems, and peripheral systems) were mounted in portable instrument racks and installed inside an enclosed EPA 4x4 vehicle (Whitehill et al., 2019). During initial set up and in between burning periods instruments were powered via land line AC power routed through two onboard mounted Tripp Lite (Chicago, IL, USA) Model SMART3000RM2U 3000VA line interactive sine wave uninterrupted power supplies (UPS) each running off its own 20A circuit to ensure that the instruments had a stable supply of clean power. Immediately prior to deploying to the burns, the instruments were then switched to a bank of six 12V 100Ah batteries run through an AIMS Power (Reno, NV, USA) model PICOGLF30W12V120V 3000W pure sine wave global low frequency inverter and charger located in a trailer towed behind the EPA vehicle. An on-board 6500W Honda generator was used to maintain charge on the batteries. The instruments were connected via perfluoroalkoxy alkane (PFA) Teflon® tubing (0.64 cm diameter) to PFA Teflon® filter packs loaded with 47 mm, 5 micron pore size pressure drop equivalent Millipore (Burlington, MA, USA) Omnipore® polytetrafluoroethylene (PTFE) filter membranes which were (i) mounted to a rooftop sampling platform during Spring 2017 sampling, or (ii) connected to a cross-linked Teflon®-coated high flow manifold mounted on the inside roof of the truck compartment during Fall 2017 sampling. The truck was positioned downwind of active biomass burning plumes, usually within meters to hundreds of meters of the active fire line, and positioned so that the trailer was downwind of the sample inlets (to avoid interferences from generator exhaust). In addition to the O₃ analyzers under investigation, additional monitors were also operated for the determination of carbon monoxide (CO), NO, NO₂, total oxides of nitrogen (NOₓ=NO+NO₂), and total hydrocarbons (THC, to approximate VOC concentrations). The operation principle and designation (FRM vs FEM) information for the additional analyzers deployed in this study are summarized in Table 2. Data from all instruments was recorded on a Envidas Ultimate data acquisition system.

<table>
<thead>
<tr>
<th>Pollutant</th>
<th>Manufacturer</th>
<th>Model</th>
<th>Method</th>
<th>FRM/FEM</th>
<th>Deployment</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
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</table>

Table 2: Additional measurement methods operated during the present study.
All instruments were calibrated with multipoint calibrations before and after each sampling day. All pre- and post-calibrations met our quality performance objectives of +/- 10% and linearity of $r^2 \geq 0.99$. For the O$_3$ analyzers under investigation, field and laboratory calibrations were performed using a Teledyne API Model T700U Dynamic Dilution Calibrator with a NIST traceable O$_3$ photometer and O$_3$ generation system. Zero air for the calibrator was supplied by a Teledyne API Model T701H Zero Air Generator. Calibrations for NO, NO$_2$, NOx and CO were performed using the same calibrator and zero air generator utilizing a certified EPA protocol tri-blend (CO, NO, SO$_2$) gas cylinder (Airgas). Per the manufacturer provided operators manual, calibrations for THC were performed using the T700U calibrator and a certified EPA methane/propane gas cylinder (Airgas). FID response factors for organic compounds can vary significantly based upon factors such as carbon number and compound class (Tong and Karasek 1984). The carbon numbers for methane and propane vary by a factor of three and the FID response factors for those compounds may also vary by as similar amount. In addition, the complex mixture of hydrocarbons found in smoke will have large variations in carbon number and FID response factors. As such, the results obtained with the THC analyzer are an approximation of THC (and VOC) concentrations in smoke. In addition, for THC calibrations, the T701H zero air generator was replaced with scientific grade zero air compressed gas cylinders (Airgas).
2.3 Kansas Prescribed Burns, March 2017

Biomass burning plumes were sampled during four days of prescribed burns (March 15-17, 2017 and March 20, 2017) on the Konza Prairie Long Term Ecological Research (LTER) site outside of Manhattan, Kansas. The fuels for this series of burns consisted of mixed native prairie tall grass of varying moisture content. Over the four-day period, a total of 13 burns were conducted and sampled.

2.4 Oregon Prescribed Burns, October 2017

A single 10-hour day of prescribed grassland burning was measured at the Sycan Marsh Preserve in central Oregon on October 11, 2017. Fuels for the Sycan Marsh burn consisted of mixed native prairie tall grass of varying moisture content.

2.5 Kansas Prescribed Burns, November 2017

Biomass burning plumes were sampled during a single day of prescribed burning (November 10, 2017) on the Konza Prairie LTER site outside of Manhattan, Kansas and on two additional days (November 13, 2017 and November 15, 2017) at the Tall Grass Prairie National Preserve outside Strong City, Kansas. Fuels for the November 2017 burns consisted of mixed native prairie tall grass of varying moisture content. During the November 10 sampling, two separate burns were conducted. Three burns were conducted over the two days at Tallgrass Prairie.

2.6 USFS Missoula Burn Chamber Burns 2018, 2019

Laboratory-based studies were performed at the U.S. Forest Service’s combustion testing facility at the FSL in Missoula, Montana by EPA and USFS personnel. These static chamber burns occurred in the spring of 2018 (April 16-24, 2018; 33 burns; Landis et al., 2020) and again in the spring of 2019 (April 15-26, 2019; 31 burns). The main combustion chamber is a square room with internal dimensions 12.4 x 12.4 x 19.6 m high and a total volume of 3000 m$^3$ and has been described previously (Bertschi et al.,...
During the combustion chamber studies, the facility was fitted with identical instrumentation racks, calibration systems, systems for sampling of test atmosphere, and data acquisition systems, as those described in Section 2.2. All instrumentation were housed in an observation room immediately adjacent to the combustion chamber with PFA inlet lines extending through the wall into the chamber. All inlet lines contained an identical filter pack/filter assembly described in Section 2.2 to protect inlet lines and the analyzers from particulate contamination. Fuel beds consisting of ponderosa pine needles and mixed woody debris were prepared and placed in the middle of chamber. The amount and moisture content of the fuels were varied to generate different flaming/smoldering conditions during the burns. During the chamber burns the combustion room was sealed and the fuel bed was ignited. Two large circulation fans on the chamber walls and one on the ceiling facilitated mixing and assured homogeneous conditions during the burn periods (Landis et al., 2020). In general, chamber RH values were below 50% facilitating dry burning condition.

3 Results and Discussion

3.1 Results from Ozone Measurements in Prescribed Grassland Fire Plumes

O3 measurement results from the Oregon and Kansas prescribed grassland fires studies are shown as the difference between the FEM and FRM in Fig. 1 and 1-minute average time series plots for the studies are presented in Supplementary Figs. S1-S3. There were significant differences in the measurement results obtained from the different O3 monitors operated during the prescribed fires. The UV-C instrument (Thermo 49i) consistently showed large increases in O3 concentration readings in fresh biomass burning plumes, with measurements exceeding the FRM measurement by 2-3 ppm. The O3 exceedances were
Figure 1. Ozone concentration differences between FEM instruments and the FRM instrument (FEM-FRM), and the measured NO$_2$, CO, and total hydrocarbons (THCs) during the three 2017 wildfire deployments. All measurements included are within-smoke only measurements, and show both the elevated smoke tracers (NO$_2$, CO, THC), and the persistent elevation of the FEM O$_3$ measurements. The box and whisker plots indicate the 25$^{\text{th}}$, 50$^{\text{th}}$, and 75$^{\text{th}}$ quartiles, with the whiskers extending to 1.5 times the inner quartile range. The open dots indicate the mean values for each instrument within smoke.

generally correlated in time with CO and THC (biomass burning indicators) and NO$_2$. These correlations will be discussed separately. The UV-C-H instrument (2B 205) also showed increased readings in smoke plumes (also correlated with CO, THC, and NO$_2$), but with absolute measurements roughly an order of magnitude smaller than the UV-C instruments. The NO-CL (T265) instrument results showed the opposite behavior, with reductions in O$_3$ readings inversely correlated with increases in NO$_2$.
concentrations, as expected from general $O_3$ titration by NO (NO + $O_3 \rightarrow NO_2 + O_2$). For the March 2017 measurements the SL-UV instrument (2B 211) produced readings roughly comparable with the NO-CL monitor, but with substantially more noise on a minute-to-minute timescale. The “in plume” average $O_3$ concentrations from the four prescribed grassland burning periods are shown in Fig. 2. For the purposes of this comparison, CO measurements were used as an indicator of when sampling occurred “in plume.” In addition, ambient RH values were generally below 50% indicating that the spring and fall 2017 prescribed burns were conducted under dry conditions.

**Figure 2.** In plume $O_3$ concentration averages from the 2017 prescribed grassland burns and the 2018 and 2019 Missoula chamber burns. For the 2017 grassland burning periods, CO concentration results ($\geq$1 ppm) were used as an indicator of when “in-smoke” sampling was occurring.
3.2 Results from Ozone Measurements in USFS Chamber Burns

O$_3$ measurement results from the 2018 and 2019 USFS chamber burn studies are shown in Fig. 3. Time

Figure 3. Differences between the FEM and FRM instrument concentrations (FEM-FRM), and NO$_2$, CO, and total hydrocarbons (THCs) concentrations during the 2018 and 2019 Missoula chamber studies. All measurements included are within-smoke only measurements, and show both the elevated smoke tracers (NO$_2$, CO, THC), and the persistent elevation of the FEM O$_3$ measurements compared to the FRM. The box and whisker plots indicate the 25th, 50th, and 75th quartiles, with the whiskers extending to 1.5 times the inner quartile range. The open dots indicate the mean values for each instrument within smoke.
series plots of the chamber Study data are included in Supplementary Figs. S4 and S5. Figure 4 gives a more detailed view of UV-C and NO-CL O₃ results (two days from 2018 and one day from 2019) during

![Time series example of USFS chamber burn O₃ results from the NO-CL, UV-C, and UV-C-H (2018 only) from April 23-24, 2018 (top) and April 22, 2019 (bottom). O₃ concentrations are displayed in a logarithmic scale. The post burn calibration checks on April 23, 2018 revealed a +8 % bias in the NO-CL method and a -2 % bias in the UV-C-H method. These biases were evident during the chamber flush.](image)

**Figure 4.**
periods on that day. Each analyzer was re-zeroed and spanned resulting in the elimination of the bias between the two methods as observed in the results from the subsequent day (April 24, 2018).

In contrast to the prescribed grassland burns, the Missoula chamber burns employed differing fuel types (ponderosa pine needles and fine woody debris) that are more typical of fuels consumed during western U.S. forest fires. In addition, the concentrations of pollutants generated and observed during the chamber studies were approximately an order of magnitude smaller than those observed during the prescribed grassland fires. For reference, maximum PM$_{2.5}$ concentrations observed during the prescribed fires were in excess of 50 mg m$^{-3}$ while maximum chamber PM$_{2.5}$ concentrations were less than 2 mg m$^{-3}$ range. Regardless of these differences, there were still significant (order of magnitude or more) differences in the measurement results between the different FEM O$_3$ instruments operated during both the 2018 and 2019 chamber studies. The NO-CL method showed identical trends to those observed during the grassland burns in that its measured O$_3$ concentrations dropped to near zero during the active burning periods as indicated in Fig. 4 (active burning periods shaded in grey). The only periods when significant O$_3$ concentrations were measured by the NO-CL method was when outside air was brought in to flush the chamber in between burns. The post burn calibration checks on April 23, 2018 revealed a +8 % bias in the NO-CL method and a -2 % bias in the UV-C-H method. These biases were evident during the chamber flush periods on that day. Each analyzer was re-zeroed and spanned resulting in the elimination of the bias between the two methods as observed in the results from the subsequent day (April 24, 2018).” No other calibration corrections were made during the 2018 and 2019 chamber studies. As in the grassland fire plumes, the UV-C method showed increased O$_3$ concentration (positive analytical artifact) readings that were correlated in time with CO and NO$_2$; See Supplementary Figs. S9 and S10. Similarly, the UV-C-H instrument also showed increased positive analytical artifacts during the chamber burns, but with absolute measurement values about an order of magnitude smaller than the UV-C instruments. The SL-UV method gave similar results to the NO-CL method during both the 2018 and 2019 chamber burns. Newly added during the 2019 burns, the UV-G method (2B 211-G) gave mixed results: at times it provided similar results compared to the NO-CL and SL-UV methods, and at others it provided results in line with those provided by the UV-C method. See Supplementary Fig. S5 for the 2019
chamber burn time series plot. The burn average O$_3$ concentrations from the 2018 and 2019 chamber burns are presented in Fig. 2.

During the 2018 chamber burns the UV-C results were biased high by 15-20 ppb even during non-burn (i.e., overnight) periods as evident in Fig. 4 (top panel) and Fig. S4. The initial hypothesis was that the bias was associated with high chamber backgrounds of interfering species due to years of heavy burning in the chamber. However, it was later discovered during a subsequent summer/fall 2018 ambient air study in North Carolina in the absence of smoke, that sampling heavy smoke plumes during the spring and fall 2017 prescribed grassland burns followed by subsequent storage of the UV-C analyzer, irreversibly damaged the MnO$_2$ scrubber in the UV-C instrument. It is hypothesized that the damage resulted in the scrubber removing some of the interfering species in addition to ozone, preventing them from being subtracted off as background in the reference measurement, and subsequent detection as ozone (positive bias) during the measurement cycle. The effect of the bias was observed mainly when sampling ambient/chamber air and not readily observed during routine calibration checks (zeroes and spans) except for an increase in the time required to obtain stable zero and span values. The bias was not observed during any of the 2017 prescribed grassland burns. During the summer/fall 2018 North Carolina study and prior to the start of the 2019 chamber burns, a new MnO$_2$ scrubber was installed and resulted in a significant and immediate reduction of the observed high bias, shown in Fig. 4 (bottom panel) and Fig. S5.

3.3 Methodological Influence on Ozone Measurements in Biomass Burning Smoke

As discussed in Sections 3.1 and 3.2, there are large (order of magnitude level) differences in O$_3$ concentration measurement results obtained from the FRM (NO-CL) and the FEM UV photometric with catalytic scrubber (UV-C) O$_3$ methods. The extremely low O$_3$ concentrations measured by the NO-CL instrument is consistent with O$_3$ depletion in the presence of high NO$_x$ concentrations (up to ppm levels) observed in the grass burning plumes and during chamber burns. The reaction between NO and O$_3$ is rapid and occurs on the timescales of seconds to minutes. As a result, high NO in the fresh biomass combustion plumes will efficiently titrate out O$_3$ leading to near-field depletion within the plumes relative to background concentrations. There was no sign of a positive interference in the NO-CL monitors, and
it remains the most robust and accurate routine method for \( O_3 \) measurement in fresh and downwind biomass burning plumes.

In contrast with the NO-CL FRM instrument results, the UV-C FEM results showed substantial increases in reported \( O_3 \) concentrations in the fresh biomass burning plumes. There is no known pathway for direct \( O_3 \) emission from biomass burning, and the proximity (meters to hundreds of meters) and timescales (seconds to minutes travel time from the combustion source to measurement) involved are too short for the usual \( \text{NO}_x - \text{VOC} \) photochemistry to produce secondary \( O_3 \). Further, since the FSL chamber interior is not exposed to sunlight, photochemistry would not have been active in the Missoula laboratory burns.

For the purposes of this work, the positive analytical artifact in the UV-C method, termed \( \Delta O_3(\text{UV-C}) \), is estimated using Eq. (6) as the difference between UV-C and the NO-CL \( O_3 \) concentration measurement results for the same time period:

\[
\Delta O_3(\text{UV-C}) = \text{UV-C} - \text{NO-CL}
\]  

Figure 5 shows “in plume” regressions between \( \Delta O_3(\text{UV-C}) \) and the FRM measurement and CO for the three measured prescribed grassland burns in 2017 (Supplementary Fig. S6 shows the time series of \( \Delta O_3(\text{UV-C}) \) and CO). Figure 5 and Supplementary Fig. S6 show good correlations within the smoke plumes.

The average and maximum values of \( \Delta O_3(\text{UV-C}) \) are summarized in Table 3. It is hypothesized that the large “\( O_3 \)” measurement observed in the UV-C method results from a positive interference or artifact, likely linked to VOC emissions in the grassland burn plumes. VOCs are emitted in higher concentrations from the smoldering phase of combustion, which is also characterized by large CO emissions (Yokelson et al., 1996; Yokelson et al., 1997), so a correlation between CO and \( O_3 \) artifact would support the hypothesis of a VOC-linked interference for the UV-C instruments. This is also consistent with observed VOC interferences in previous studies (Grosjean and Harrison, 1985; Kleindienst et al., 1993; Spicer et al., 2010) and observed following fireworks (Fiedrich et al., 2017; Xu et al., 2018).
**Figure 5.** Scatter plots between FEM and FRM O\(_3\) differences and the CO measurements within the grassland fires smoke plumes. The FEM measurements are differentiated by color and shape. The SL-UV method was only run during the Konza March 2017 measurements.

<table>
<thead>
<tr>
<th>Study</th>
<th>(\Delta O_3) avg. (ppb)</th>
<th>(\Delta O_3) max (ppb)</th>
<th>CO avg. (ppm)</th>
<th>NO(_2) avg. (ppb)</th>
<th>THC avg. (ppm)</th>
</tr>
</thead>
<tbody>
<tr>
<td>(\Delta O_3) (UV-C)</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Mar. 2017 Konza Prairie (KS)</td>
<td>295.8</td>
<td>2423.7</td>
<td>15.8</td>
<td>147.3</td>
<td>-</td>
</tr>
<tr>
<td>Oct. 2017 Sycan Marsh (OR)</td>
<td>170.2</td>
<td>3235.5</td>
<td>8.5</td>
<td>60.5</td>
<td>2.7</td>
</tr>
<tr>
<td>Nov. 2017 Konza &amp; Tallgrass Prairies (KS)</td>
<td>330.0</td>
<td>3156</td>
<td>14.1</td>
<td>156.9</td>
<td>4.0</td>
</tr>
<tr>
<td>Apr. 2018 USFS Chamber (MT)</td>
<td>36.5</td>
<td>309.6</td>
<td>3.8</td>
<td>35.6</td>
<td>2.8</td>
</tr>
<tr>
<td>Apr. 2019 USFS Chamber (MT)</td>
<td>66.9</td>
<td>530.9</td>
<td>2.1</td>
<td>31.7</td>
<td>4.8</td>
</tr>
<tr>
<td>(\Delta O_3) (UV-C-H)</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Mar. 2017 Konza Prairie (KS)</td>
<td>42.8</td>
<td>227.1</td>
<td>15.8</td>
<td>147.3</td>
<td>-</td>
</tr>
<tr>
<td>Oct. 2017 Sycan Marsh (OR)</td>
<td>21.1</td>
<td>316.4</td>
<td>8.5</td>
<td>60.5</td>
<td>2.7</td>
</tr>
<tr>
<td>Nov. 2017 Konza &amp; Tallgrass Prairies (KS)</td>
<td>40.2</td>
<td>369.0</td>
<td>14.1</td>
<td>156.9</td>
<td>4.0</td>
</tr>
<tr>
<td>Apr. 2018 USFS Chamber (MT)</td>
<td>7.2</td>
<td>136.8</td>
<td>3.8</td>
<td>35.6</td>
<td>2.8</td>
</tr>
<tr>
<td>(\Delta O_3) (UV-G)</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Apr. 2019 USFS Chamber (MT)</td>
<td>22.9</td>
<td>376.8</td>
<td>2.1</td>
<td>31.7</td>
<td>4.8</td>
</tr>
</tbody>
</table>
The presence of a Nafion®-based humidity conditioning system (Nafion® tube dryer) significantly reduced the magnitude of the observed artifact as evident by comparing the UV-C and UV-C-H results shown in Figs. 1-3 and Supplementary Figs. S1 – S5. As with the UV-C method, the artifact in the UV-C-H method, ΔO₃(UV-C-H), is calculated using Eq. (7) as the difference between UV-C-H and the NO-CL O₃ concentration measurement results for the same time period:

$$\Delta O₃(\text{UV-C-H}) = \text{UV-C-H} - \text{NO-CL}$$ (7)

The addition of the Nafion®-based humidity conditioning system reduces the magnitude of the ΔO₃(UV-C-H) artifact by approximately an order of magnitude as compared with the UV-C method. This is further illustrated in the 2018 chamber burns, where prior to beginning the final burn day on April 24, 2018, a Nafion® tube dryer (PermaPure, MD Monotube Dryer Series) was installed on the UV-C method (Thermo 49i) in effect, converting it to a UV-C-H method. As shown in Fig. 4 and Supplementary Fig. S4, the addition of the Nafion® tube dryer significantly reduced the ΔO₃(UV-C-H) artifact to a point comparable with that observed in the UV-C-H method (2B 205). A possible explanation for this effect is presented and discussed in Section 3.5. In addition, the previously described bias related to the damaged MnO₂ scrubber was also reduced upon addition of the Nafion® dryer to the UV-C method.

For the March 2017 Konza Prairie study (Fig. 1) and the 2018 and 2019 USFS chamber studies (Fig. 3) the SL-UV instrument concentration results were comparable to, although noisier and slightly higher than, the NO-CL reference instrument. On numerous occasions during the prescribed and chamber burns, the SL-UV instrument shows short (i.e. one-minute data point) positive or negative excursions that are not also observed in the NO-CL results. In addition, these excursions are not correlated with changes in CO concentrations. Because the SL-UV is a dual cell instrument that measures O₃ by comparing the

<table>
<thead>
<tr>
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</thead>
<tbody>
<tr>
<td>Value</td>
<td>8.3</td>
<td>0.5</td>
<td>1.7</td>
</tr>
<tr>
<td></td>
<td>74.2</td>
<td>11.5</td>
<td>32.1</td>
</tr>
<tr>
<td></td>
<td>15.8</td>
<td>3.8</td>
<td>2.1</td>
</tr>
<tr>
<td></td>
<td>147.3</td>
<td>35.6</td>
<td>31.7</td>
</tr>
<tr>
<td></td>
<td>-</td>
<td>2.8</td>
<td>4.8</td>
</tr>
</tbody>
</table>
absorbance of two cells, it is critical in highly dynamic environments (such as during this study) that both cells be measuring the same air at the same time. A slight difference in flow rates or residence times between the two pathways (or a delay in one pathway relative to the other) will cause short term variability in the difference between the two cells. Although this does not pose an issue for longer time averaging (i.e. hourly data) under stable conditions, the dynamic nature of biomass burning plumes (i.e. changing on the order of seconds) and short time averages (i.e. minute) can create issues (noise) for the SL-UV method.

Significant analytical artifacts were observed for FEM UV photometric O₃ instruments with (UV-C-H) and without (UV-C) Nafion®-based humidity conditioning system, where it appears that the dual effect of ambient humidity fluctuations and VOC interferences caused large positive over-measurement of “in-smoke” O₃ concentrations. Chemiluminescence monitors are highly specific to O₃ and have long been known to be free of VOC interferences (Long et al., 2014; U.S. EPA, 2015). However, studies have shown that the chemiluminescence method can be impacted by changes in relative humidity (Kleindienst et al., 1993). As such, upon promulgation in 2015, the new NO-CL FRM regulatory text requires a humidity correction/dryer system to eliminate the potential water vapor interference. As configured from the manufacturer, the NO-CL based Teledyne-API Model T265 instrument operated during this comparative study employs Nafion® drying technologies to reduce or eliminate the water vapor interferences. The use of a chemical (NO) scrubber for UV photometric instruments (such as the 2B Technologies Model 211) is very specific to O₃ and shows a much better response than the catalytic scrubber instruments, performing almost as well as the NO-CL FRM, and has significant potential as a low-interference O₃ method. Of the catalytic scrubber photometric instruments those with Nafion®-based humidity equilibration (2B Technologies Model 205) perform significantly better than those without (Thermo 49 series).

In areas highly impacted by smoke or for studies focusing on biomass burning plumes, the use of a NO-CL FRM instrument is highly recommended as it was found to be essentially interference-free. These instruments are anchored to absolute O₃ concentrations through the use of certified O₃ calibration sources,
many of which are based on UV photometry. The newest generation of commercially-available NO-CL FRM instruments, including that used here (the Teledyne T265), have a built-in drying system to correct for the humidity artifacts that affected earlier generation chemiluminescence instruments (Kleindienst et al., 1993), making remaining interferences negligible compared to other technologies.

The gas-phase chemical scrubber UV instrument (2B 211), did not perform as well as the FRM under the prescribed grassland burns or chamber experimental conditions tested here, with the high time resolution (1-minute) data showing a much higher degree of variability than the NO-CL FRM instrument. We hypothesize that the main factor driving this divergence between this method and the NO-CL FRM is the dual-cell differential configuration of the instrument, which is not conducive to rapidly changing concentrations in O₃ or other absorbing gases, such as VOCs.

In smoke-impacted monitoring situations where the use of a UV photometric instrument is still preferred or required, the choice of a monitor with humidity equilibration provides a significant analytical improvement over those monitors without the humidity corrections. In the absence of an instrument with a Nafion® tube dryer and in non-regulatory applications, a dryer can be installed before the inlet or measurement cells to reduce the interference as was demonstrated on the final day of the 2018 Missoula chamber burns. This will have the added benefit of reducing positive biases from humidity and reducing equilibration time for calibrations (especially when switching from high humidity ambient air to dry calibration gases).

### 3.4 Magnitude of Ozone Artifact in Fresh Biomass Burning Plumes Relative to Markers of Combustion

It is difficult to estimate an absolute magnitude or correct for the observed O₃ analytical artifact since primary emissions from biomass combustion are highly variable and depend upon the fuel loading, fuel type and condition, phase of the fire, and the burn conditions (Yokelson et al., 1996; Yokelson et al., 1997). However, assuming the interference is driven primarily by VOCs, the artifact should be correlated with the excess CO (ΔCO = CO_{plume} – CO_{background}). Because CO_{background} during the prescribed grassland burns was below 200 ppb (relative to typical conditions of >2 ppm in the plume), ΔCO is estimated as the
total measured CO concentration. A simplified view of biomass combustion assumes an approximate linear combination of two dominant emission phases, flaming combustion (characterized by emission of highly oxidized compounds, such as CO\(_2\), NO\(_x\), and SO\(_2\)), and smoldering combustion (characterized by emission of reduced or mixed oxidation state compounds, such as CO, CH\(_4\), NH\(_3\), H\(_2\)S, and most VOCs) (Yokelson et al., 1996; Yokelson et al., 1997). Because the majority of VOCs are in a reduced or mixed oxidation state, they tend to be co-emitting with CO during smoldering combustion, and the VOC concentrations tend to be highly correlated with CO in fresh biomass burning plumes (Yokelson et al., 1996). Scatterplots comparing the FEM instrument artifacts (\(\Delta O_3\) (UV-C)) and CO for the three prescribed grassland burning periods are shown in Fig. 5. Regression statistics of the comparison of \(\Delta O_3\) (UV-C) and \(\Delta O_3\) (UV-C-H) with CO and THC for grassland burns are given in Table 4. The magnitude of the artifact (estimated by the slope of the regression line of the CO vs \(\Delta O_3\) comparison), in ppb apparent O\(_3\) per ppm CO, ranges between 16 - 24 ppb ppm\(^{-1}\) for the UV-C instrument, and 1.5 - 3.5 ppb ppm\(^{-1}\) for the instrument with humidity correction (UV-C-H). It is important to point out that CO, in and of itself, is not considered to be an interfering species in the UV photometric determination of O\(_3\) in that CO absorbs in the infrared (IR). The slight differences in the magnitude of the artifacts (fitted regression slopes) along with the low uncertainty (standard errors) values indicate that the magnitude of the artifact may be influenced by local conditions that make each burn unique. Such conditions might include meteorological conditions, fuel composition, fuel moisture content, and times spent in combustion phase (flaming vs smoldering). Similar to CO, THCs and NO\(_2\) are indicative of combustion processes and are correlated with \(\Delta O_3\) as given in Table 4 and Supplementary Figs. S7 and S8. In terms of THC, the magnitude of the artifact, in ppb apparent O\(_3\) per ppm THC, is significantly higher at ~88 ppb ppm\(^{-1}\) for the UV-C instrument and ~13 ppb ppm\(^{-1}\) for the UV-C-H instrument. Both the prescribed grassland and Missoula chamber burns resulted in what would be considered high PM concentrations (2-50 mg m\(^{-3}\)). These high PM concentrations however, are not considered to be interfering due to the presence of the inline particle filter assemblies described in Sections 2.2 and 2.6.
Table 4: Regression statistics for the ozone artifact ($\Delta O_3$) versus CO and THC for UV photometric instruments without (UV-C) and with (UV-C-H) a Nafion®-based humidity equilibration system during the 2017 prescribed grassland burns.

<table>
<thead>
<tr>
<th>Study</th>
<th>$\Delta O_3$ (ppb/ppm)</th>
<th>Intercept (ppb)</th>
<th>$r^2$</th>
<th>n</th>
</tr>
</thead>
<tbody>
<tr>
<td>$\Delta O_3$ vs CO</td>
<td>Mar. 2017 Konza Prairie (KS)</td>
<td>16.46 $\pm$0.34$^a$</td>
<td>22.91(8.53$\pm$6.72)$^a$</td>
<td>0.79</td>
</tr>
<tr>
<td>Oct. 2017 Sycan Marsh (OR)</td>
<td>24.02 $\pm$0.25</td>
<td>-28.05 $\pm$2.73</td>
<td>0.96</td>
<td>295</td>
</tr>
<tr>
<td>Nov. 2017 Konza &amp; Tallgrass Prairies (KS)</td>
<td>23.51 $\pm$0.73</td>
<td>-20.8 $\pm$13.03</td>
<td>0.74</td>
<td>461</td>
</tr>
<tr>
<td>$\Delta O_3$ vs THC</td>
<td>Nov. 2017 Konza &amp; Tallgrass Prairies (KS)</td>
<td>87.14 $\pm$3.74</td>
<td>-85.36 $\pm$18.63</td>
<td>0.59</td>
</tr>
<tr>
<td>$\Delta O_3$ vs CO</td>
<td>Mar. 2017 Konza Prairie (KS)</td>
<td>1.46 $\pm$0.04</td>
<td>0.87 $\pm$1.03</td>
<td>0.80</td>
</tr>
<tr>
<td>Oct. 2017 Sycan Marsh (OR)</td>
<td>2.21 $\pm$0.05</td>
<td>3.44 $\pm$0.54</td>
<td>0.88</td>
<td>296</td>
</tr>
<tr>
<td>Nov. 2017 Konza &amp; Tallgrass Prairies (KS)</td>
<td>3.24 $\pm$0.09</td>
<td>-1.17 $\pm$1.67</td>
<td>0.77</td>
<td>461</td>
</tr>
<tr>
<td>$\Delta O_3$ vs THC</td>
<td>Nov. 2017 Konza &amp; Tallgrass Prairies (KS)</td>
<td>13.27 $\pm$0.39</td>
<td>-14.53 $\pm$1.92</td>
<td>0.75</td>
</tr>
</tbody>
</table>

Since the CO concentrations (from upwind fires) observed at most stationary sites from fire plumes are usually on the order of one ppm to greater than 10 ppm (Landis et al., 2018), it is reasonable to assume that O$_3$ artifacts in the range of 15 ppb to greater than 250 ppb can be observed when employing a UV-C method. Similarly, O$_3$ artifacts in the range of 1.5 to above 30 ppb might be observed at smoke-impacted sites monitoring with UV-C-H methods. As such, Nafion®, based humidity conditioning systems are highly recommended for use if employing UV photometric methodology for monitoring O$_3$ in areas impacted by wildfires or prescribed burns. As stated previously and as seen in Fig. 3 and Table 3, O$_3$ artifacts were observed during the Missoula chamber 2018 and 2019 burns in both the UV-C and UV-C-H methods, although reduced compared to the prescribed grassland burns. The presence and magnitude of the O$_3$ artifact strongly suggests that smoke generated from fuels typical of forests in the western United States also result in a measurement interference in UV photometric methods. Since downwind O$_3$ production in biomass burning plumes is a significant issue in fire impacted regions, having reliable, interference-free methods is critical for assessing the contribution of wildland fires to ambient O$_3$ levels.
Figure 6 gives a detailed time series view of $\Delta O_3(\text{UV-C})$ and CO from two burn days from 2018 and a single day during 2019. As indicated, $\Delta O_3(\text{UV-C})$ and CO appear to be correlated in time but when performing linear regression comparisons of $\Delta O_3(\text{UV-C})$ and CO during each years chamber burns as a whole, correlations tend to be poor. We suspect the positive O$_3$ bias is driven by one or more VOCs (likely oxygenated VOCs). In fresh smoke the excess concentrations of individual VOCs ($\Delta X$), and VOC sums ($\Delta \text{VOC}$), tend to be highly correlated with $\Delta \text{CO}$ (Yokelson et al., 1999; Gilman et al. 2015). The emission ratios of individual VOCs to CO ($\Delta X/\Delta \text{CO}$) can vary considerably with combustion conditions such as fuel type and condition (e.g. moisture content and decay state), fuel bed properties, such as bulk density, and the relative mix of flaming and smoldering combustion (Gilman et al. 2015; Koss et al., 2017).

Additionally, the response of $\Delta X/\Delta \text{CO}$ to burn conditions varies among VOCs. When each burn is considered individually or in groups with similar conditions, the correlations between $\Delta O_3$, CO, and THC are enhanced. An example of this behavior is shown in Supplementary Fig. S10. For the chamber burns the magnitude of the ozone artifacts in ppb apparent O$_3$ per ppm CO, ranges between 6 - 210 ppb ppm$^{-1}$ for the individual burns. $R^2$ and standard error values were consistent with those observed during the prescribed burns (see Table 4). The lack of a consistent relationship between the O$_3$ artifact and $\Delta \text{CO}$ across all FSL chamber burns, while observing a good correlation for individual burns, likely reflects the variable response of artifact producing emission(s) to the different combustion conditions of the burns.

One interesting observation from the data obtained from both the prescribed grassland and chamber burns is the order of magnitude difference in the average and maximum O$_3$ artifact between the UV-C and the UV-C-H instruments as shown in Table 3. Considering that the prescribed grassland and chamber burns were conducted under dry (RH < 50%) conditions, the size of the difference (as large as hundreds of ppb) cannot be explained purely by the previously observed relative humidity effects on measurements (Leston et al.,
**Figure 6.** Time series example of USFS chamber burn $\Delta$O$_3$(UV-C) and CO concentration results from April 23-24, 2018 (top) and April 22, 2019 (bottom).

2005; Wilson et al., 2006), suggesting that the Nafion® dryer is directly impacting the concentrations of other interferents in the sample stream.
3.5 Potential Reason for Lower Artifacts with Methods Employing Nafion®-based Humidity Equilibration

Nafion® is a sulfonated tetrafluoroethylene polymer that is highly permeable to water but shows little permeability to many other organic and inorganic species (Mauritz et al., 2004). As a result, Nafion®-based drying systems are often used as part of sample preparation or conditioning systems in analytical chemistry to remove water vapor from sample streams prior to sample analysis. Nafion® membranes were introduced to some O₃ monitors as a method to address humidity effects observed in UV-C O₃ monitors, particularly when there are rapid changes in relative humidity level (Wilson and Birks, 2006). Humidity can affect the transmission of the UV light through the detection cell and catalytic O₃ scrubbers can modulate the water vapor in the scrubbed channel by acting as a temporary reservoir, resulting in significant positive or negative O₃ interferences during rapid swings in relative humidity Wilson et al., 2006). Adding a Nafion®-based equilibration dryer immediately prior to the measurement cells reduces this water vapor interference without affecting O₃ concentrations, and thus significantly reduces the humidity artifacts in UV photometric O₃ instruments.

Despite the high selectivity of Nafion® to water vapor, it does demonstrate partial to complete permeability to various VOC or semivolatile organic compounds. Nafion® membranes are highly permeable to alcohols, amines, ketones, and some water-soluble ethers (Baker, 1974), as well as some biogenic oxygenated compounds (Burns et al., 1983). In addition, Nafion® membranes have been shown to catalyze the decomposition and rearrangement of monoterpene compounds (Burns et al., 1983). Systematic study of Nafion® permeability and reactivity for polar and oxygenated compounds has been limited, with most users of Nafion® membranes basing their use on operational testing and confirmation for the targeted use.

The significant (order of magnitude) reduction in the O₃ artifact with the addition of a Nafion®-based dryer to the UV-C suggests that the Nafion® dryer is directly impacting the major interfering species which was hypothesized to be VOCs emitted during combustion processes. The species that are responsible for most of the O₃ artifact in UV-C O₃ instruments would have to be permeable through
Nafion® membranes or reactive with Nafion® membranes, be scrubbed by solid-phase, catalytic O₃ scrubbers (such as MnO₂ or hopcalite), and would have a significant absorption cross section around 254 nm. The absorption cross-section of O₃ around 254 nm is on the order of 10⁻¹⁷ cm² molecule⁻¹ (Molina and Molina, 1986), which means species with absorptions around 10⁻¹⁷ cm² molecule⁻¹ at 254 nm would be potential interfering species. As a class, aromatic VOCs and specifically oxygenated aromatic species (and other polar derivatized species) absorb strongly in this region of the UV spectrum, and their potential permeability through Nafion® membranes result in them being likely compounds to interfere in UV-C instruments. As an example, aromatic aldehydes such as o-tolualdehyde and p-tolualdehyde absorbances around 5x10⁻¹⁸ cm² molecule⁻¹ and 4x10⁻¹⁸ cm² molecule⁻¹, respectively (Etzkorn et al., 1999). Both 2,4-dimethylbenzaldehyde and 2,6-dimethylbenzaldehyde have absorption cross sections above 10⁻¹⁷ cm² molecule⁻¹ at 254 nm (El Dib et al., 2008). Baker (1974) found 75% of benzaldehyde was removed by a Nafion® membrane, meaning that the Nafion® permeability of tolualdehydes and dimethylbenzaldehydes is also likely to be high. In addition, benzaldehyde was almost quantitatively removed by several commercial catalytic O₃ scrubbers, including the Thermo 49i MnO₂ catalytic scrubber (Kleindienst et al., 1993), so similar aldehydes are likely to behave in a similar manner. Therefore, substituted aromatic aldehyde species are one class of compounds that fit the necessary criteria for causing the interference on the UV-C while having a reduced interference on the UV-C-H instrument. Future work examining the potential interferences from different species (or classes of species) on a species or class specific basis are required to confirm this potential mechanism and suggest others.

4 Implications

Wildland fires (wildfires and prescribed fires) emit significant amounts of VOCs and NOx, two important precursors in the photochemical formation of tropospheric O₃. Therefore, it is not surprising that large increases in O₃ are routinely reported at ambient monitoring sites downwind from wildland fires (DeBell et al., 2004; Bytnerowicz et al., 2010; Preisler et al., 2010; Jaffe et al., 2012; Bytnerowicz et al., 2013; Jaffe et al., 2013; Lu et al., 2016; Lindaas et al., 2017; Baylon et al., 2018; Liu et al 2018; McClure and Jaffe, 2018). For example, Buysse et al. (2019) examined regulatory air monitoring data from 18 cities over a five period, and found that July – September exceedances of NAAQS for O₃ were far more common.
on days with known wildland fire smoke impacts (4.6%) than those without (<0.1%). However, the results of this study suggests caution when interpreting UV photometric method \( \text{O}_3 \) measurements under conditions of wildfire smoke impact due to the significant positive artifacts that were observed. The analytical artifacts were also shown to be positively correlated with widely used markers of combustion such as CO and THC suggesting that the artifacts arise from photometric measurement interferences by VOCs and further complicating the interpretation of smoke impacted UV photometric \( \text{O}_3 \) data. As described in section 3.4, it reasonable to assume that \( \text{O}_3 \) artifacts in the range of a few ppb to greater than 250 ppb in addition to actual photochemically formed \( \text{O}_3 \) can be observed when employing UV photometric methods at sites downwind from fires.

A detailed example of observed artifacts in the UV photometric method occurred during the 2016 Fort McMurray Horse River wildfire in Alberta, Canada, where elevated “\( \text{O}_3 \)” concentrations were observed at multiple community based air monitoring sites utilizing UV-C instruments in the vicinity of the fire (Landis et al., 2018). Reported “\( \text{O}_3 \)” concentrations reached maximum hourly concentrations in excess of 1500 ppb using UV-C methods at night (between 10:00 PM and 5:00 AM local) in the absence of photochemistry and were positively correlated with the combustion markers NO and non-methane hydrocarbon (NMHC). Peaks in \( \text{O}_3 \) concentration are expected to be negatively correlated with peaks in NO concentration as it rapidly titrates \( \text{O}_3 \) to \( \text{NO}_2 \), and the authors hypothesized that UV photometric measurement artifacts may have been responsible for the unexpected observations.

The findings from this research effort and the observations from ambient studies (Landis et al., 2018) raise concerns that routine regulatory monitoring and wildland fire research study \( \text{O}_3 \) measurements utilizing UV photometric FEM instruments may be reporting positive measurement artifacts as \( \text{O}_3 \) during smoke impacted events. Some studies have hypothesized that rapid photochemical processing was responsible for reported elevated \( \text{O}_3 \) concentrations reported in downwind wildfire plumes (Liu et al., 2017). Since downwind \( \text{O}_3 \) production in biomass burning plumes is a significant issue in fire impacted regions, having reliable, interference-free methods is critical for assessing the contribution of wildland
fires to ambient \( O_3 \) levels and developing/validating accurate deterministic air quality models. Air quality researchers and environmental regulators are strongly encouraged to utilize NO-CL FRM \( O_3 \) instruments in areas routinely impacted by wildland fire smoke.

### 4.5 Conclusions

In this study, we compare two different \( O_3 \) measurement methods (chemiluminescence and UV photometry) in fresh biomass burning plumes from prescribed grassland fires and during controlled chamber burns. Within the UV photometry category, we look at two different technologies, one using a gas-phase chemical scrubber (NO) and the second using solid phase catalysts to scrub \( O_3 \) from analytical reference channels. Among the UV photometric instruments employing solid phase catalytic scrubbers, we evaluated and compared methods that include a Nafion\textsuperscript{®}-based humidity equilibration system with those that do not.

The NO-CL method, recently promulgated as the \( O_3 \) FRM, performed well, even in fresh plumes, whereas the UV photometric method employing the dynamic NO gas phase scrubber performed comparably with the NO-CL method but was not well suited to the rapidly varying concentrations of VOCs in the smoke plumes. The catalytic scrubber photometric methods demonstrated positive analytical artifacts that were correlated with CO and THC concentrations (both biomass burning plume indicators). There was a significant difference between the catalytic scrubber UV instruments with and without Nafion\textsuperscript{®}-based humidity correction, with the dryer system reducing the positive \( O_3 \) artifact by an order of magnitude as compared with the UV photometric method employing no humidity correction. The observed reduction in artifacts cannot be attributed only to elimination of the relative humidity/water vapor interferences and likely result from post-scrubber equilibration or reaction of Nafion\textsuperscript{®}-permeable VOCs prior to the measurement cell. The results of this study strongly suggest that careful consideration be given to employed measurement methods when monitoring \( O_3 \) concentrations in regions where impacts from biomass burning routinely occur due to the significant impact of potential measurement interferences. In addition to consideration of operating methods containing Nafion\textsuperscript{®}-based humidity condition systems, attention should be focused
on the scrubbers employed by UV photometric methods and the adverse effects that operation in smoke may have on those scrubbers. Further research is being conducted to evaluate the magnitude of the artifact in the UV photometric method at routine monitoring sites that are often impacted by wildland fire smoke events under the EPA Mobile Ambient Smoke Investigation Capability (MASIC) program (U.S. EPA 2019).

**Data Availability**

Datasets related to this manuscript can be found at https://catalog.data.gov/dataset/epa-sciencehub.

**Author Contributions**

Russell W. Long served as principal investigator and prepared the manuscript with contributions from all co-authors. Russell W. Long, Andrew Whitehill, Andrew Habel, Maribel Colón, Shawn Urbanski, and Matthew S. Landis performed the prescribed grassland fire and FSL chamber-based data collection and/or analysis. Surender Kaushik performed supervisory review of this research effort and corresponding manuscript.

**Competing Interests**

The authors declare that they have no conflict of interest.

**Disclaimer**

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References


U.S. Environmental Protection Agency (EPA), National Ambient Air Quality Standards for Ozone, Federal Register, 80, 206, October 26, 2015.


