

1 **Comparison of Ozone Measurement Methods in Biomass Burning Smoke: An evaluation under** 2 **field and laboratory conditions**

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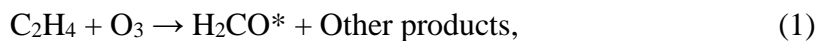
11 **Abstract**

12 In recent years wildland fires in the United States have had significant impacts on local and regional air
13 quality and negative human health outcomes. Although the primary health concerns from wildland fires
14 come from fine particulate matter (PM_{2.5}), large increases in ozone (O₃) have been observed downwind
15 of wildland fire plumes (DeBell et al., 2004; Bytnerowicz et al., 2010; Preisler et al., 2010; Jaffe et al.,
16 2012; Bytnerowicz et al., 2013; Jaffe et al., 2013; Lu et al., 2016; Lindaas et al., 2017; McClure and
17 Jaffe, 2018; Liu et al 2018; Baylon et al., 2018; Buysse et al. 2019). Conditions generated in and around
18 wildland fire plumes, including the presence of interfering chemical species, can make the accurate
19 measurement of O₃ concentrations using the ultraviolet (UV) photometric method challenging if not
20 impossible. UV photometric method instruments are prone to interferences by volatile organic
21 compounds (VOCs) that are present at high concentrations in wildland fire smoke. Four different O₃
22 measurement methodologies were deployed in a mobile sampling platform downwind of active prescribed
23 grassland fire lines in Kansas and Oregon and during controlled chamber burns at the United States Forest
24 Service, Rocky Mountain Research Station Fire Sciences Laboratory in Missoula, Montana. We
25 demonstrate that the Federal Reference Method (FRM) nitric oxide (NO) chemiluminescence monitors
26 and Federal Equivalent Method (FEM) gas-phase (NO) chemical scrubber UV photometric O₃ monitors
27 are relatively interference-free, even in near-field combustion plumes. In contrast, FEM UV photometric
28 O₃ monitors using solid-phase catalytic scrubbers show positive artifacts that are positively correlated
29 with carbon monoxide (CO) and total gas phase hydrocarbons (THC), two indicator species of biomass

30 burning. Of the two catalytic scrubber UV photometric methods evaluated, the instruments that included
31 a Nafion® tube dryer in the sample introduction system had artifacts an order of magnitude smaller than
32 the instrument with no humidity correction. We hypothesize that Nafion®--permeable VOCs (such as
33 aromatic hydrocarbons) could be a significant source of interference for catalytic scrubber UV
34 photometric O₃ monitors, and that the inclusion of a Nafion® tube dryer assists with the mitigation of
35 these interferences. The chemiluminescence FRM method is highly recommended for accurate
36 measurements of O₃ in wildland fire plume studies and at regulatory ambient monitoring sites frequently
37 impacted by wildland fire smoke.

38 **1 Introduction**

39 Ground-level ozone (O₃) is a secondary air pollutant generated from the photochemical interactions of
40 nitrogen oxides (NO_x) and volatile organic compounds (VOCs). The most robust methods for O₃
41 measurements are based on chemiluminescence reactions with ethylene (ET-CL, for ethylene
42 chemiluminescence) or nitric oxide (NO-CL, for nitric oxide chemiluminescence) (Long et al., 2014).
43 The overall reaction mechanism for ET-CL generally proceeds as detailed in Eqs. (1-2):



46
47
48 The reaction generates electronically-activated formaldehyde (H₂CO*) which luminesces in the high
49 ultraviolet (UV) to visible portion of the spectrum (380 nm - 550 nm) and vibrationally-activated
50 hydroxide ions which luminesce in the visible light to the low infrared (IR) portion of the spectrum (550
51 nm - 800 nm). The number of photons emitted during the reaction is directly proportional to the amount
52 of O₃ present and are counted by a photomultiplier tube (PMT), with its response centered at 440 nm,
53 then the count is converted to O₃ concentration. The ET-CL method requires a constant supply of ethylene
54 for continuous operation. NO-chemiluminescence analyzers measure O₃ concentrations using the
55 principle that the dry, gas-phase reaction between NO and O₃ generates nitrogen dioxide in an
56 electronically excited state (NO₂*), and oxygen (O₂) (Ollison et. al., 2013; Boylan et.al., 2014). As each

57 unstable, NO_2^* molecule returns to a lower energy state (NO_2), it emits a photon ($h\nu$). The reaction causes
58 luminescence in a broadband spectrum ranging from visible light to infrared light (approximately 590 nm
59 – 2800 nm). The two-step gas-phase reaction proceeds as detailed in Eqs. (3-4):



64 The ET-CL method is no longer used nor produced commercially and has been replaced by the NO-CL
65 method. Similar to the ET-CL method, the NO-CL method requires a constant supply of gas, in this case
66 NO, for continuous operation. Both the ET-CL and NO-CL methods are subject to slight interferences by
67 water vapor. However, these potential interferences can be eliminated through the use of Nafion based drier
68 or equivalent sample water vapor treatment system. The ET-CL method was promulgated as the Federal
69 Reference Method (FRM) for measuring O_3 in the atmosphere in 1971 and the NO-CL method
70 promulgated as the FRM in 2015 (U.S. EPA, 2015).

71
72 While the chemiluminescence method for measuring O_3 is technically robust and free of analytical
73 artifacts (Long et al., 2014), it is not widely used in the United States. Instead, Federal Equivalent Methods
74 (FEM) based upon UV photometry are employed at the majority of O_3 regulatory monitoring locations.
75 According to July 2020 data from the United States Environmental Protection Agency (EPA) Air Quality
76 System (AQS) database, the UV photometric method represents 99% of the roughly 1200 instruments
77 deployed in network monitoring for O_3 National Ambient Air Quality Standard (NAAQS) attainment.
78 UV photometric methods for O_3 are generally considered easier to deploy, operate, and in most cases do
79 not require external compressed gases for operation. UV photometric analyzers determine O_3
80 concentrations by quantitatively measuring the attenuation of light due to absorption by O_3 present in an
81 absorption cell at the specific wavelength of 254 nm (Parrish et al., 2000; Williams et al., 2006). The O_3
82 concentration is determined through a two-step process in which the light intensity passing through the
83 sample air (I) is compared with the light intensity passing through similar sample air from which all O_3
84 is first removed (I_0). The ratio of these two light intensity values (I/I_0) provides the measure of the light

85 absorbed at 254 nm, and the O₃ concentration in the sample is then determined through the use of the
86 Beer-Lambert Law as given in Eq. (5):

$$I/I_0 = e^{-KLC} \quad (C = 1/KL \ln [I/I_0]); \quad (5)$$

87
88
89 where L is the length of the absorption cell (cm), C is the O₃ concentration (ppm), and K is the absorption
90 cross section of O₃ at 254 nm at standard atmospheric temperature and pressure conditions (308 atm⁻¹ cm⁻¹
91 ¹). Photometric monitors generally use mercury vapor lamps as the UV light source, with optical filters
92 to attenuate lamp output at other than the 254 nm wavelength.
93

94
95 Air for the reference cell measurement (I₀) is typically obtained by passing the ambient air sample stream
96 through a catalytic scrubber containing manganese dioxide (MnO₂), hopcalite (a mixture of Cu, Mn, and
97 Ag oxides), heated silver wool, or another solid state material to ‘scrub’ only O₃ from the sample air while
98 preserving all other substances in the sample air that potentially absorb at 254 nm (e.g., elemental gaseous
99 mercury [Hg⁰], hydrogen, sulfide [H₂S], VOCs) so that their effects are cancelled in the differential I/I₀
100 measurement. The integrity of the O₃ reference scrubber is critical and may allow measurement
101 interferences if it does not perform adequately. Similarly, any tendency of the scrubber to fail to
102 effectively remove all O₃ from the reference sample will result in a measurement bias. In addition to O₃,
103 catalytic scrubbers have been shown to remove UV-active VOCs (Kleindienst et al., 1993), creating the
104 potential for positive artifacts in O₃ measurements when the efficiency of this VOC removal is impacted.
105

106 Although FEM designated UV photometric instruments are accurate under most ambient conditions,
107 locations with high VOC concentrations can produce significant analytical artifacts. Smoke plume
108 impacted locations and measurements downwind from wildland fires are a particular concern; O₃
109 measurements of up to 320 ppb were observed in a smoke plume in western Oregon using a Dasibi
110 1003AH UV photometric O₃ monitor (Huntzicker and Johnson, 1979), which also showed a correlation
111 between apparent O₃ and aerosol concentrations (**b**_{scat}, a combustion plume indicator in this case). O₃
112 measurements from UV photometric instruments exceeding 1500 ppb at night (22:00-05:00) were

113 observed in Fort McMurray, Alberta during smoke impacts from the 2016 Horse River Fire, which were
114 positively correlated with NO and non-methane hydrocarbons (Landis et al., 2018). Follow-up pyrolysis
115 experiments demonstrated that ET-CL instruments do not show a similar response to biomass burning
116 smoke (Huntzicker and Johnson, 1979). Photochemical chamber experiments comparing the O₃ response
117 of UV (Dasibi Model 1003AH, Dasibi Model 1008AH, and Thermo Model 49) and ET-CL (Bendix
118 Model 8002 and Monitor Labs Model 8410) mixtures show negligible differences for irradiated
119 paraffin/NO_x and olefin/NO_x mixtures, but do show a positive UV interference in mixtures with toluene
120 and other aromatics present (Kleindienst et al., 1993). Laboratory studies comparing the response of UV
121 (Thermo Model 49, Horiba APOA-370, and 2B Tech Model 202) and ET-CL (Bendix) instruments
122 showed a positive interference for o-nitrophenol, naphthalene, and p-tolualdehyde for the UV instruments
123 but not the ET-CL instruments (Grosjean and Harrison, 1985; Spicer et al., 2010). Additionally, during
124 the Mexico City Metropolitan Area field campaign (MCMA-2003) a mobile laboratory using an FEM
125 designated UV photometric O₃ monitor (unheated MnO₂ scrubber, Thermo 49 series) showed a large
126 positive O₃ interference (~400 ppb) associated with PM_{2.5} and polyaromatic hydrocarbons (PAHs) when
127 following some diesel vehicles (Dunlea et al., 2006). Although not compared to a chemiluminescence
128 instrument, those high O₃ values are unlikely real considering the high concurrent NO concentrations (in
129 some cases, >1000 ppb). The authors of this study attributed the interference to fine particles, based on
130 the correlation with PM_{2.5} and the lack of a correlation with gas-phase organic species measured by the
131 proton transfer reaction-mass spectrometer (PTR-MS, Dunlea et al., 2006).

132

133 In addition to interferences from the presence of aromatic VOCs and semi-volatile PAHs, water vapor
134 (relative humidity) issues have also been observed with older generation FRM and FEM designated
135 chemiluminescence and UV photometric O₃ instruments, respectively (Kleindienst et al., 1993; Leston et
136 al., 2005; Wilson and Birks, 2006). As such, Nafion[®] tube dryers are regularly incorporated into some
137 newer generation chemiluminescence and UV photometric O₃ monitors in an attempt to mitigate the
138 humidity related measurement artifacts.

139

140 A recently introduced variation of the UV photometric method, known as the “scrubberless” UV
141 photometric (SL-UV) method (Ollison et al., 2013), specifies removal of O₃ from the sample air for the
142 reference by a gas-phase reaction with NO rather than using a conventional solid state catalytic scrubber.
143 The NO gas phase chemical scrubber reacts with O₃ much faster and more selectively than with other
144 potential interfering compounds and is very effective at removing the O₃ without affecting other
145 interfering compounds that may be present in ambient air. The differential UV measurement can then
146 effectively reduce interferences to an insignificant level. Similar to NO-CL, the SL-UV method requires
147 a continuous supply of compressed NO or nitrous oxide (N₂O) (which the instrument converts to NO) to
148 serve as the scrubber gas.

149

150 In this study, we investigate UV photometric FEM instrument O₃ measurement interferences in fresh
151 biomass burning smoke plumes from prescribed grassland fires and during controlled burn experiments
152 in a large scale combustion chamber. We directly compare NO-CL FRM O₃ measurements to several
153 FEM designated UV photometric technologies, including a gas-phase scrubber and catalytic scrubbers
154 with and without Nafion[®] tube dryer systems. Based on the results from the measurements, we assess the
155 magnitude of the observed artifacts for different technologies and under various smoke conditions and
156 provide suggestions for potential mitigation of the interferences.

157

158 **2. Methods**

159 **2.1 Overview of Methods Evaluated**

160 In this study we compared the measurement results from six different commercially available FRM/FEM
161 designated O₃ instruments operated in ambient or chamber generated biomass burning smoke. All
162 instruments were operated according to their FRM or FEM designation. The six instruments differed by
163 measurement principle (chemiluminescence *versus* UV photometric), and by sample treatment
164 configuration (scrubber material, presence of dryer, etc.). For interference free O₃ measurements we
165 utilized the newly designated FRM NO-CL method (U.S. EPA, 2015). For the UV photometric methods,

166 we compared both catalytic scrubber and “scrubberless” (gas phase chemical scrubber) technologies, with
 167 the “scrubberless” monitor using a NO chemical scrubber. Finally, within the catalytic scrubber UV
 168 photometric category, we compared instruments with and without Nafion tube dryer systems. The
 169 operation principle and designations (FRM vs FEM) for the analyzers under investigation are summarized
 170 in Table 1 and described in Sections 2.1.1-2.1.4. These analyzers were operated immediately downwind
 171 of fresh biomass burning plumes during eight days of prescribed fires in grassland ecosystems in Oregon
 172 and Kansas and during laboratory-based studies at the U.S. Forest Service’s (USFS) combustion facility
 173 at the Fire Sciences Laboratory (FSL) in Missoula, Montana. The grassland fire fuels consisted primarily
 174 of mixed native prairie tall grass of varying moisture content. Seven of the eight days of prescribed
 175 burning were conducted in the Tallgrass Prairie ecosystem of central Kansas (four days in March of 2017
 176 and three days in November of 2017). The additional burn day was conducted at the Sycan Marsh in
 177 central Oregon (October of 2017). Laboratory based chamber burns at the FSL were conducted during
 178 April 2018 and again during April 2019. Fuels for the laboratory based chamber burns consisted of
 179 ponderosa pine needles and fine woody debris. Details of the individual studies are provided in Sections
 180 2.2-2.6.

181

182 **Table 1: Ozone measurement methods investigated.**

Name	Manufacturer	Model	Method	Scrubber	Cells	Humidity Correction	Deployment ^a
U.S. EPA Federal Reference Methods (FRM)							
NO-CL	Teledyne API	T-265	CL (NO)	N/A	1	Nafion [®] -based (dryer)	K1, S, K2, T, M1, M2
U.S. EPA Federal equivalent methods (FEM)							
UV-C	Thermo Scientific	49i	UV (254 nm)	Catalyst (MnO ₂)	2	None	K1, S, K2, T, M1, M2
UV-C-H	2B Technologies	205	UV (254 nm)	Catalyst (Hopcalite)	2	Nafion [®] -based (equilibration)	K1, S, K2, T, M1
SL-UV	2B Technologies	211	UV (254 nm)	Gas chemical (NO)	2	Nafion [®] -based (equilibration)	K1, M1, M2
UV-G	2B Technologies	211-G	UV (254 nm)	Heated graphite	2	Nafion [®] -based (equilibration)	M2

183 ^aK1-Konza Prairie March 2017; S-Sycan Marsh, October 2017; K2-Konza Prairie November 2017; T-Tallgrass Prairie
 184 November 2017; M1-Missoula chamber April 2018; M2-Missoula chamber April 2019.

185

186 **2.1.1 NO Chemiluminescence**

187 The FRM O₃ measurement method was the Teledyne API (San Diego, CA, USA) Model T265
188 Chemiluminescence Monitor (TAPI T265), which utilizes a NO-CL measurement principle. These NO-
189 CL O₃ analyzers have been shown to be free of interferences (Long et al. 2014) , and have been used as
190 a reference method in other O₃ comparison studies (Williams et al., 2006; Landis et al., 2020). Although
191 there is a known water vapor interference with chemiluminescence technology (Kleindienst et al., 1993),
192 the TAPI T265 uses a Nafion[®] tube dryer system to remove water vapor from the air prior to making the
193 measurement, thus eliminating any humidity-related effects. Like the ET-CL technologies (Kleindienst
194 et al., 1993), the NO-CL analyzers have no documented VOC interferences. Manufacturer provided
195 performance specifications for the NO-CL based TAPI T265 are given in Table S1.

196

197 **2.1.2 Catalytic Scrubber UV Photometric**

198 For this study the UV photometric method with no humidity correction was represented by the Thermo
199 Scientific (Franklin, MA, USA) Model 49i (Thermo 49i), which is a dual cell instrument with a
200 manganese oxide (MnO₂) catalytic scrubber, referred to as UV-C. Nafion[®]-based humidity systems or
201 dryers have been employed within photometric O₃ monitors with catalytic scrubbers before the
202 measurement cell, offering a reduction in relative humidity interferences and artifacts (Wilson and Birks,
203 2006). Manufacturer provided performance specifications for the UV-C based Thermo 49i are given in
204 Table S1.

205

206 The UV photometric with a Nafion[®] humidity conditioning system was represented in this study by a 2B
207 Technologies (Boulder, CO, USA) Model 205 (2B 205) O₃ monitor. The 2B 205 utilized a dual-cell
208 design where sample air and scrubbed air are measured simultaneously. The 2B 205 uses a Hopcalite
209 (CuO/MnO₂) catalytic scrubber to remove O₃ from the reference stream. This instrument will be referred
210 to as UV-C-H. Manufacturer provided performance specifications for the UV-C-H based 2B 205 are
211 given in Table S1.

212

213 **2.1.3 Scrubberless UV Photometric**

214 For comparison with the NO-CL, UV-C and UV-C-H methodologies, a “scrubberless” UV (SL-UV)
215 photometric analyzer with a gas-phase (NO) chemical scrubber was employed (Ollison et al., 2013;
216 Johnson et al., 2014). The addition of NO gas to the reference stream selectively scrubs O₃ while not
217 significantly affecting interfering VOC species, resulting in an interference free O₃ determination.
218 Inclusion of this instrument into the study allows evaluation of the impact of the UV method in general
219 (as compared with chemiluminescence) versus the influence of specific scrubber technologies. The SL-
220 UV method is represented by the 2B Technologies Model 211 “Scrubberless” Ozone Monitor (2B 211).
221 The Model 2B 211 requires a continuous supply of compressed NO or nitrous oxide (N₂O) (which the
222 instrument converts to NO). The SL-UV method also utilizes a Nafion[®]-based sample humidity
223 conditioning system to eliminate any humidity effects. The SL-UV instrument was not used in the October
224 or November 2017 burns due to the lack of the necessary reagent gas (nitrous oxide, N₂O) needed to run
225 the instrument. Manufacturer provided performance specifications for the S1-UV based 2B 211 are given
226 in Table S1.

227

228 **2.1.4 Heated Graphite Scrubber UV Photometric**

229 During the final phase of laboratory-based burning a 2B Technologies Model 211-G UV photometric
230 analyzer (2B 211-G) was operated for comparison to the monitors detailed in Sections 2.1.1-2.1.3. The
231 2B 211-G differs from the 2B 211 in that it employs a heated graphite scrubber to remove O₃ from the
232 reference sample stream (I₀) (Turnipseed et al., 2017). The 2B 211-G utilizes the same Nafion[®]-based
233 sample humidity conditioning system as employed in the 2B 211. For the purposes of this manuscript the
234 UV photometric method employing the heated graphite scrubber be referred to as UV-G. Manufacturer
235 provided performance specifications for the UV-G based 2B 211-G are given in Table S1.

236

237 2.2 Prescribed Fire Burn Mobile Sampling Platform

238 During the prescribed fire grass burns, all study instrumentation (analyzers, data acquisition systems, and
239 peripheral systems) were mounted in portable instrument racks and installed inside an enclosed EPA 4x4
240 vehicle (Whitehill et al., 2019). The instruments were connected via perfluoroalkoxy alkane (PFA)
241 Teflon[®] tubing (0.64 cm diameter) to PFA Teflon[®] filter packs loaded with 47 mm, 5 micron pore size
242 pressure drop equivalent Millipore (Burlington, MA, USA) Omnipore[®] polytetrafluoroethylene (PTFE)
243 filter membranes which were (i) mounted to a rooftop sampling platform during Spring 2017 sampling,
244 or (ii) connected to a cross-linked Teflon[®]-coated high flow manifold mounted on the inside roof of the
245 truck compartment during Fall 2017 sampling. The truck was positioned downwind of active biomass
246 burning plumes, usually within meters to hundreds of meters of the active fire line, and positioned so that
247 the trailer was downwind of the sample inlets (to avoid interferences from generator exhaust). In addition
248 to the O₃ analyzers under investigation, additional monitors were also operated for the determination of
249 carbon monoxide (CO), NO, NO₂, total oxides of nitrogen (NO_x=NO+NO₂), and total hydrocarbons
250 (THC, to approximate VOC concentrations). The operation principle and designation (FRM vs FEM)
251 information for the additional analyzers deployed in this study are summarized in Table 2. Data from all
252 instruments was recorded on a Envidas Ultimate data acquisition system.

253

254 **Table 2: Additional measurement methods operated during the present study.**

Pollutant	Manufacturer	Model	Method	FRM/FEM	Deployment^f
CO	Teledyne API	48C	NDIR ^a	FRM	K1, S, K2, T, M1, M2
NO ₂	Teledyne API	T500U	CAPS ^b	FEM	K1, S, K2, T, M1, M2
NO, NO ₂ , NO _x	Thermo Scientific	42C	CL (O ₃) ^c	FRM	K1, K2, T, M1
NO, NO ₂ , NO _x	Teledyne API	T200/T201 ^e	CL (O ₃)	FRM	M1, M2
THC	Thermo Scientific	51i	FID ^d	NA	K2, T, M1, M2

255 ^aNon-Dispersive Infrared Absorption

256 ^bCavity Attenuated Phase Shift

257 ^cOzone Chemiluminescence

258 ^dFlame Ionization Detector

259 ^eThe Teledyne API Model T201 is not a designated FRM or FEM however it employs the same operating principle as the FRM
260 designated model T200

261 ^fK1-Konza Prairie March 2017; S-Sycan Marsh October 2017; K2-Konza Prairie November 2017; T-Tallgrass Prairie
262 November 2017; M1-Missoula chamber April 2018; M2-Missoula chamber April 2019.

263

264 All instruments were calibrated with multipoint calibrations before and after each sampling day. All pre-
265 and post-calibrations met our quality performance objectives of +/- 10% and linearity of $r^2 \geq 0.99$. For
266 the O₃ analyzers under investigation, field and laboratory calibrations were performed using a Teledyne
267 API Model T700U Dynamic Dilution Calibrator with a NIST traceable O₃ photometer and O₃ generation
268 system. Zero air for the calibrator was supplied by a Teledyne API Model T701H Zero Air Generator.
269 Calibrations for NO, NO₂, NO_x and CO were performed using the same calibrator and zero air generator
270 utilizing a certified EPA protocol tri-blend (CO, NO, SO₂) gas cylinder (Airgas). Per the manufacturer
271 provided operators manual, calibrations for THC were performed using the T700U calibrator and a
272 certified EPA methane/propane gas cylinder (Airgas). FID response factors for organic compounds can
273 vary significantly based upon factors such as carbon number and compound class (Tong and Karasek
274 1984). The carbon numbers for methane and propane vary by a factor of three and the FID response
275 factors for those compounds may also vary by a similar amount. In addition, the complex mixture of
276 hydrocarbons found in smoke will have large variations in carbon number and FID response factors. As
277 such, the results obtained with the THC analyzer are an approximation of THC (and VOC) concentrations
278 in smoke. In addition, for THC calibrations, the T701H zero air generator was replaced with scientific
279 grade zero air compressed gas cylinders (Airgas).

280

281 **2.3 Kansas Prescribed Burns, March 2017**

282 Biomass burning plumes were sampled during four days of prescribed burns (March 15-17, 2017 and
283 March 20, 2017) on the Konza Prairie Long Term Ecological Research (LTER) site outside of Manhattan,
284 Kansas. The fuels for this series of burns consisted of mixed native prairie tall grass of varying moisture
285 content. Over the four-day period, a total of 13 burns were conducted and sampled.

286

287 **2.4 Oregon Prescribed Burns, October 2017**

288 A single 10-hour day of prescribed grassland burning was measured at the Sycan Marsh Preserve in
289 central Oregon on October 11, 2017. Fuels for the Sycan Marsh burn consisted of mixed native prairie
290 tall grass of varying moisture content.

291

292 **2.5 Kansas Prescribed Burns, November 2017**

293 Biomass burning plumes were sampled during a single day of prescribed burning (November 10, 2017)
294 on the Konza Prairie LTER site outside of Manhattan, Kansas and on two additional days (November 13,
295 2017 and November 15, 2017) at the Tall Grass Prairie National Preserve outside Strong City, Kansas.
296 Fuels for the November 2017 burns consisted of mixed native prairie tall grass of varying moisture
297 content. During the November 10 sampling, two separate burns were conducted. Three burns were
298 conducted over the two days at Tallgrass Prairie.

299

300 **2.6 USFS Missoula Burn Chamber Burns 2018, 2019**

301 Laboratory-based studies were performed at the U.S. Forest Service's combustion testing facility at the
302 FSL in Missoula, Montana by EPA and USFS personnel. These static chamber burns occurred in the
303 spring of 2018 (April 16-24, 2018; 33 burns; Landis et al., 2020) and again in the spring of 2019 (April
304 15-26, 2019; 31 burns). The main combustion chamber is a square room with internal dimensions 12.4 x
305 12.4 x 19.6 m high and a total volume of 3000 m³ and has been described previously (Bertschi et al.,
306 2003; Christian et al., 2004; Yokelson et al., 1996; Landis et al., 2020). During the combustion chamber
307 studies, the facility was fitted with identical instrumentation racks, calibration systems, systems for
308 sampling of test atmosphere, and data acquisition systems, as those described in Section 2.2. All
309 instrumentation were housed in an observation room immediately adjacent to the combustion chamber
310 with PFA inlet lines extending through the wall into the chamber. All inlet lines contained an identical
311 filter pack/filter assembly described in Section 2.2 to protect inlet lines and the analyzers from particulate
312 contamination. Fuel beds consisting of ponderosa pine needles and mixed woody debris were prepared

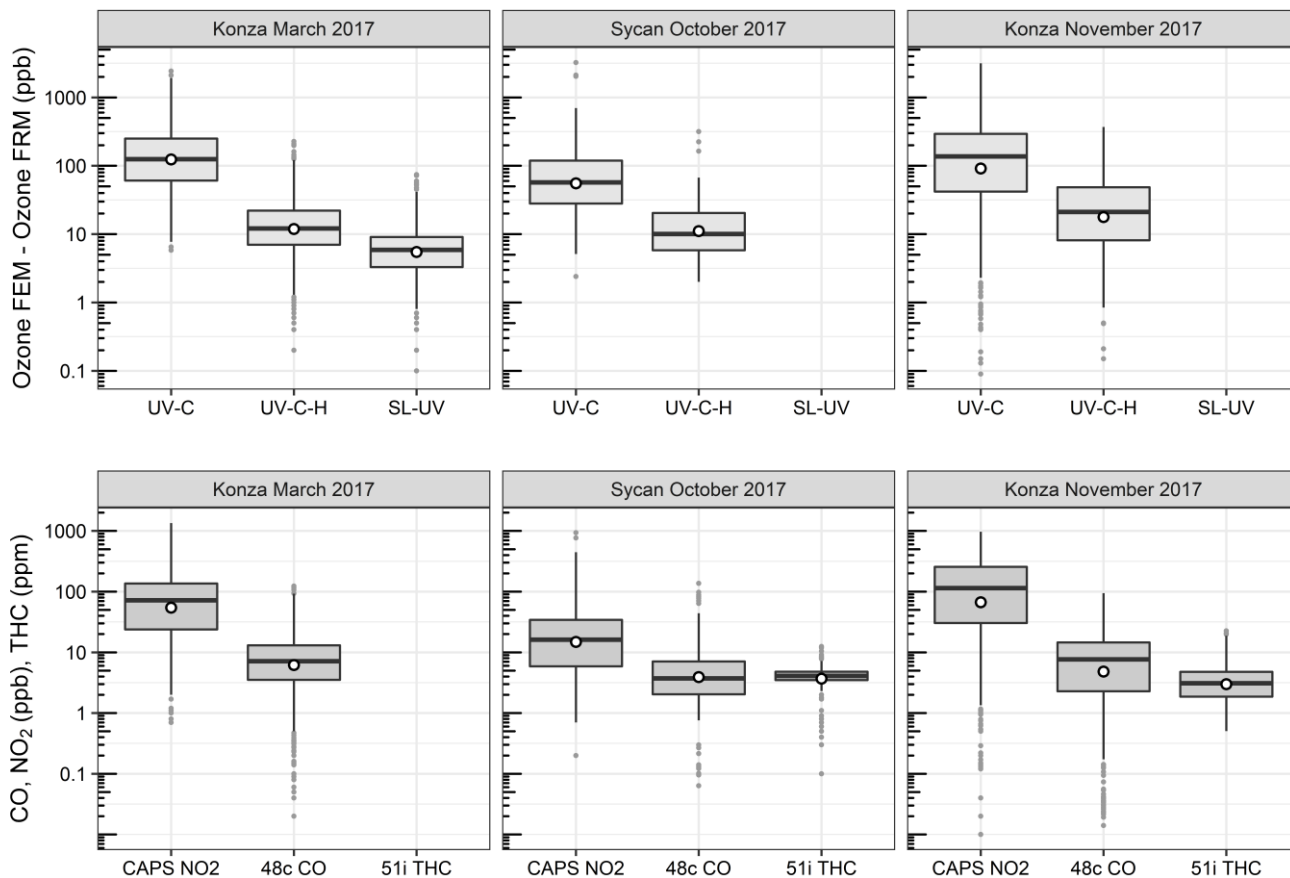
313 and placed in the middle of chamber. The amount and moisture content of the fuels were varied to generate
314 different flaming/smoldering conditions during the burns. During the chamber burns the combustion room
315 was sealed and the fuel bed was ignited. Two large circulations fans on the chamber walls and one on the
316 ceiling facilitated mixing and assured homogeneous conditions during the burn periods (Landis et al.,
317 2020). In general, chamber RH values were below 50% facilitating dry burning condition.

318 **3 Results and Discussion**

319 **3.1 Results from Ozone Measurements in Prescribed Grassland Fire Plumes**

320 O₃ measurement results from the Oregon and Kansas prescribed grassland fires studies are shown as the
321 difference between the FEM and FRM in Fig. 1 and 1-minute average time series plots for the studies are
322 presented in Supplementary Figs. S1-S3. There were significant differences in the measurement results
323 obtained from the different O₃ monitors operated during the prescribed fires. The UV-C instrument
324 (Thermo 49i) consistently showed large increases in O₃ concentration readings in fresh biomass burning
325 plumes, with measurements exceeding the FRM measurement by 2-3 ppm. The O₃ exceedances were
326 generally correlated in time with CO and THC (biomass burning indicators) and NO₂. These correlations
327 will be discussed separately. The UV-C-H instrument (2B 205) also showed increased readings in smoke
328 plumes (also correlated with CO, THC, and NO₂), but with absolute measurements roughly an order of
329 magnitude smaller than the UV-C instruments. The NO-CL (T265) instrument results showed the
330 opposite behavior, with reductions in O₃ readings inversely correlated with increases in NO₂
331 concentrations, as expected from general O₃ titration by NO ($\text{NO} + \text{O}_3 \rightarrow \text{NO}_2 + \text{O}_2$). For the March 2017
332 measurements the SL-UV instrument (2B 211) produced readings roughly comparable with the NO-CL
333 monitor, but with substantially more noise on a minute-to-minute timescale. The “in plume” average O₃
334 concentrations from the four prescribed grassland burning periods are shown in Fig. 2. For the purposes
335 of this comparison, CO measurements were used as an indicator of when sampling occurred “in plume.”
336 In addition, ambient RH values were generally below 50% indicating that the spring and fall 2017
337 prescribed burns were conducted under dry conditions.

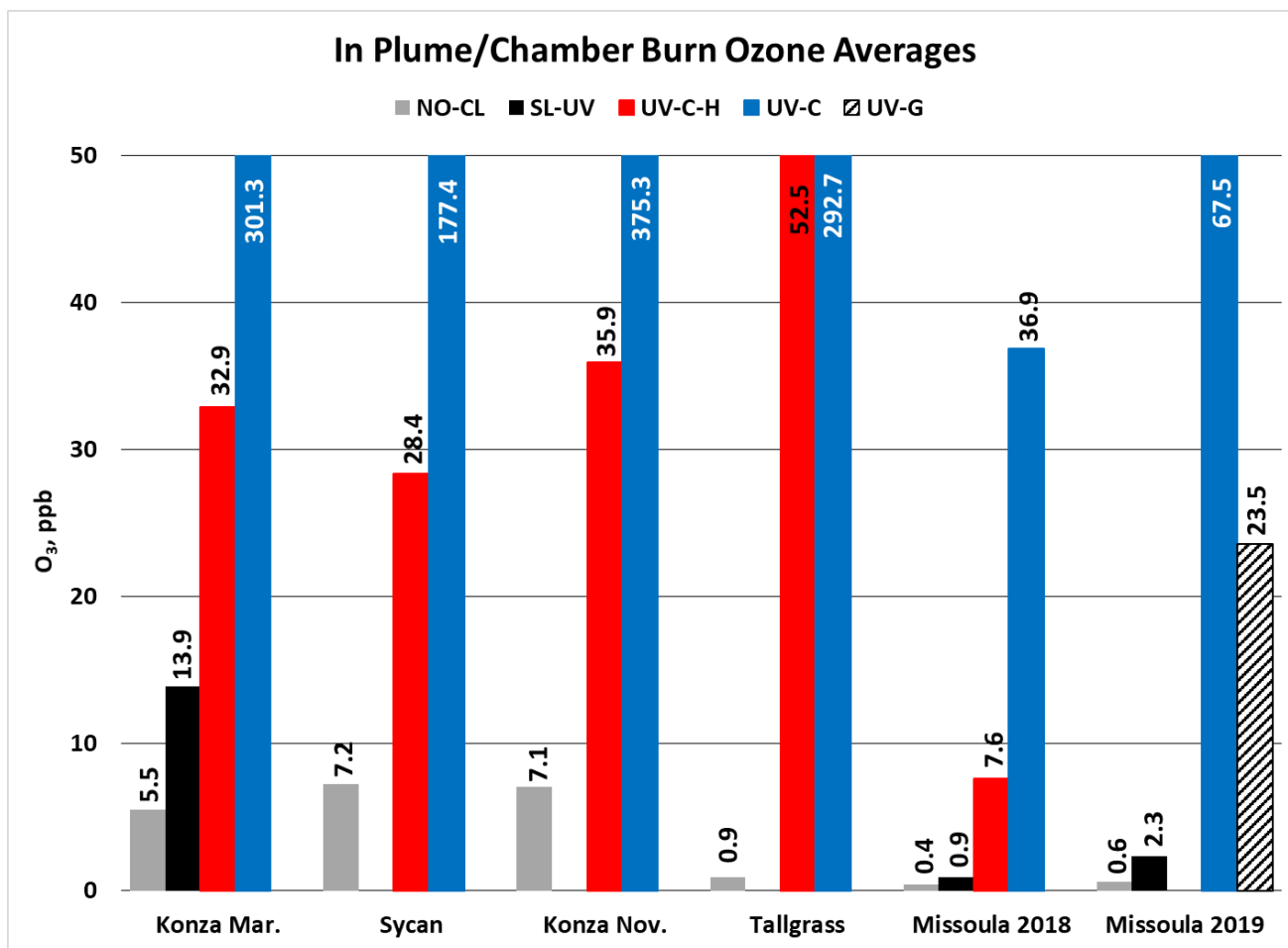
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339

340 **Figure 1.** Ozone concentration differences between FEM instruments and the FRM instrument (FEM-
 341 FRM), and the measured NO₂, CO, and total hydrocarbons (THCs) during the three 2017 wildfire
 342 deployments. All measurements included are within-smoke only measurements, and show both the
 343 elevated smoke tracers (NO₂, CO, THC), and the persistent elevation of the FEM O₃ measurements. The
 344 box and whisker plots indicate the 25th, 50th, and 75th quartiles, with the whiskers extending to 1.5 times
 345 the inner quartile range. The open dots indicate the mean values for each instrument within smoke.

346



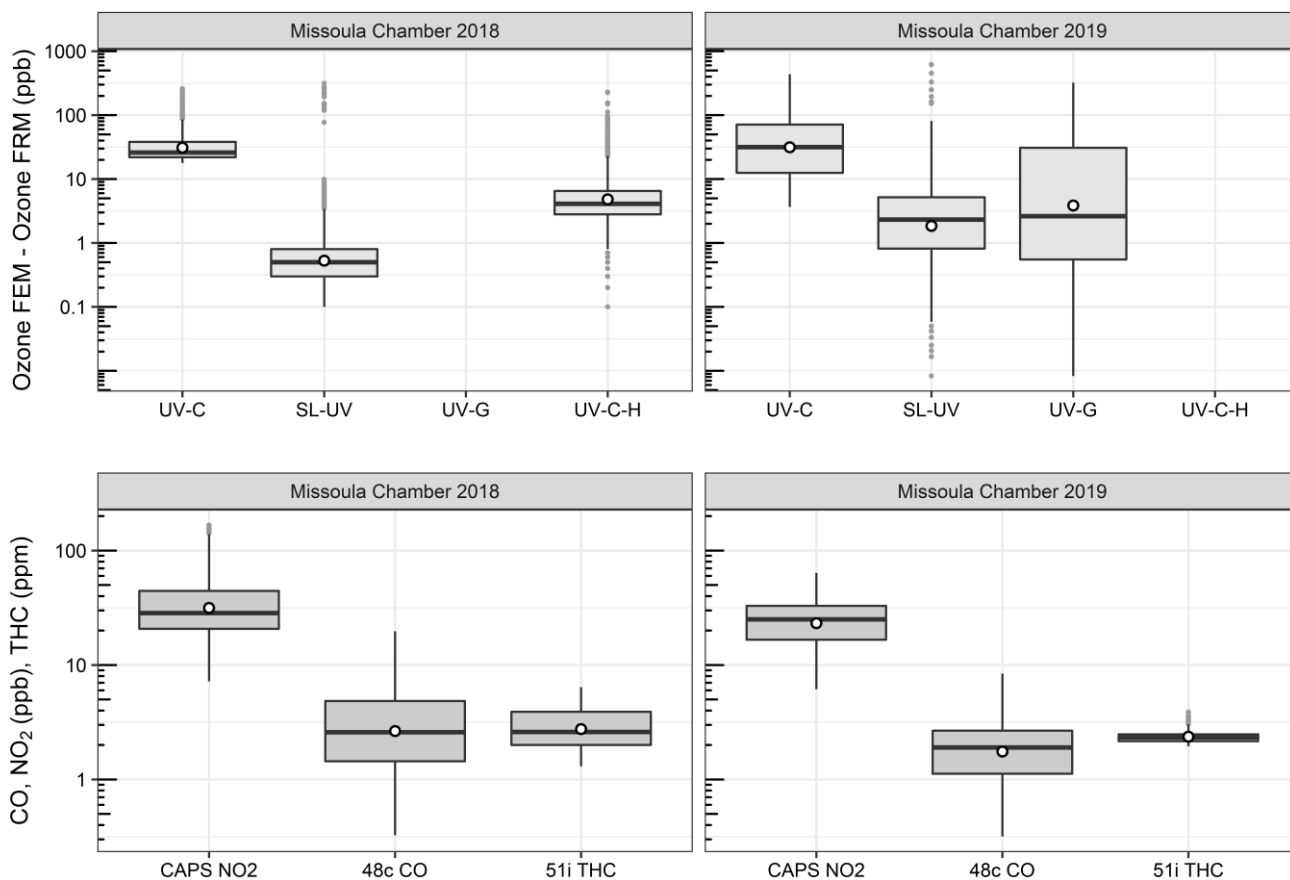
347

348 **Figure 2.** In plume O₃ concentration averages from the 2017 prescribed grassland burns and the 2018 and
 349 2019 Missoula chamber burns. For the 2017 grassland burning periods, CO concentration results (≥ 1
 350 ppm) were used as an indicator of when “in-smoke” sampling was occurring.

351

352 3.2 Results from Ozone Measurements in USFS Chamber Burns

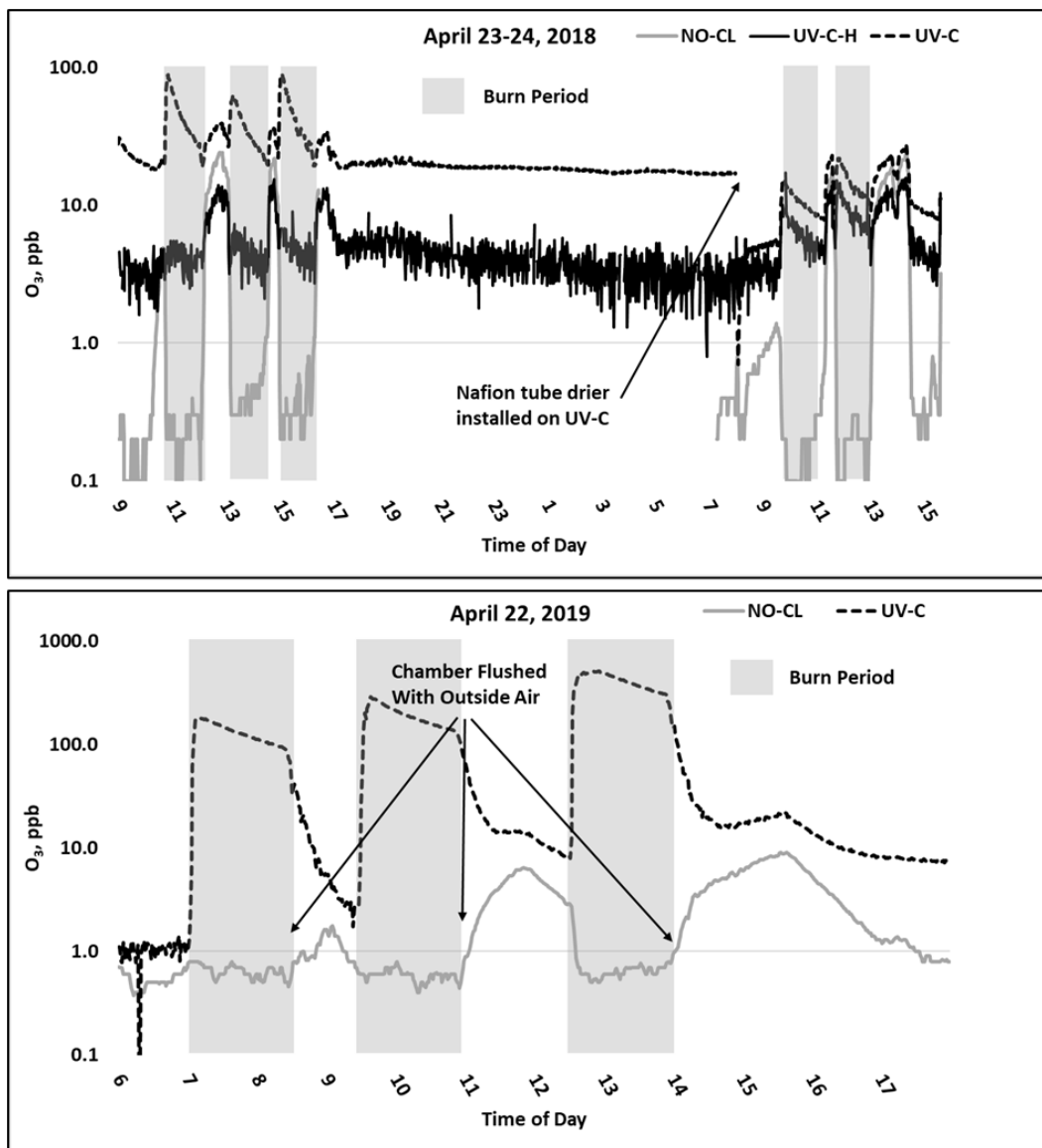
353 O₃ measurement results from the 2018 and 2019 USFS chamber burn studies are shown in Fig. 3. Time
 354 series plots of the chamber Study data are included in Supplementary Figs. S4 and S5. Figure 4 gives a
 355 more detailed view of UV-C and NO-CL O₃ results (two days from 2018 and one day from 2019) during
 356 the chamber burns. In contrast to the prescribed grassland burns, the Missoula chamber burns employed
 357 differing fuel types (ponderosa pine needles and fine woody debris) that are more typical of fuels



358

359 **Figure 3.** Differences between the FEM and FRM instrument concentrations (FEM-FRM), and NO₂,
 360 CO, and total hydrocarbons (THCs) concentrations during the 2018 and 2019 Missoula chamber studies.
 361 All measurements included are within-smoke only measurements, and show both the elevated smoke
 362 tracers (NO₂, CO, THC), and the persistent elevation of the FEM O₃ measurements compared to the FRM.
 363 The box and whisker plots indicate the 25th, 50th, and 75th quartiles, with the whiskers extending to 1.5
 364 times the inner quartile range. The open dots indicate the mean values for each instrument within smoke.

365 consumed during western U.S. forest fires. In addition, the concentrations of pollutants generated and
 366 observed during the chamber studies were approximately an order of magnitude smaller than those
 367 observed during the prescribed grassland fires. For reference, maximum PM_{2.5} concentrations observed
 368 during the prescribed fires were in excess of 50 mg m⁻³ while maximum chamber PM_{2.5} concentrations



369

370 **Figure 4.** Time series example of USFS chamber burn O₃ results from the NO-CL, UV-C, and UV-C-H
 371 (2018 only) from April 23-24, 2018 (top) and April 22, 2019 (bottom). O₃ concentrations are displayed
 372 in a logarithmic scale. The post burn calibration checks on April 23, 2018 revealed a +8 % bias in the NO-
 373 CL method and a -2 % bias in the UV-C-H method. These biases were evident during the chamber flush
 374 periods on that day. Each analyzer was re-zeroed and spanned resulting in the elimination of the bias
 375 between the two methods as observed in the results from the subsequent day (April 24, 2018).

376 were less than 2 mg m⁻³ range. Regardless of these differences, there were still significant (order of
 377 magnitude or more) differences in the measurement results between the different FEM O₃ instruments

378 operated during both the 2018 and 2019 chamber studies. The NO-CL method showed identical trends to
379 those observed during the grassland burns in that its measured O₃ concentrations dropped to near zero
380 during the active burning periods as indicated in Fig. 4 (active burning periods shaded in grey). The only
381 periods when significant O₃ concentrations were measured by the NO-CL method was when outside air
382 was brought in to flush the chamber in between burns. The post burn calibration checks on April 23, 2018
383 revealed a +8 % bias in the NO-CL method and a -2 % bias in the UV-C-H method. These biases were
384 evident during the chamber flush periods on that day. Each analyzer was re-zeroed and spanned resulting
385 in the elimination of the bias between the two methods as observed in the results from the subsequen day
386 (April 24, 2018).” No other calibration corrections werer made during the 2018 and 2019 chamber
387 studies. As in the grassland fire plumes, the UV-C method showed increased O₃ concentration (positive
388 analytical artifact) readings that were correlated in time with CO and NO₂; See Supplementary Figs. S9
389 and S10. Similarly, the UV-C-H instrument also showed increased positive analytical artifacts during the
390 chamber burns, but with absolute measurement values about an order of magnitude smaller than the UV-
391 C instruments. The SL-UV method gave similar results to the NO-CL method during both the 2018 and
392 2019 chamber burns. Newly added during the 2019 burns, the UV-G method (2B 211-G) gave mixed
393 results: at times it provided similar results compared to the NO-CL and SL-UV methods, and at others it
394 provided results in line with those provided by the UV-C method. See Supplementary Fig. S5 for the 2019
395 chamber burn time series plot. The burn average O₃ concentrations from the 2018 and 2019 chamber
396 burns are presented in Fig. 2.

397 During the 2018 chamber burns the UV-C results were biased high by 15-20 ppb even during non-burn
398 (i.e., overnight) periods as evident in Fig. 4 (top panel) and Fig. S4. The initial hypothesis was that the
399 bias was associated with high chamber backgrounds of interfering species due to years of heavy burning
400 in the chamber. However, it was later discovered during a subsequent summer/fall 2018 ambient air study
401 in North Carolina in the absence of smoke, that sampling heavy smoke plumes during the fall 2017
402 prescribed grassland burns followed by subsequent storage of the UV-C analyzer, irreversibly damaged
403 the MnO₂ scrubber in the UV-C instrument. It is hypothesized that the damage resulted in the scrubber
404 removing some of the interfering species in additon to ozone, preventing them from being subtracted off
405 as background in the reference measurment, and subsequent detection as ozone (positive bias) during the

406 measurement cycle. The effect of the bias was observed mainly when sampling ambient/chamber air and
407 not readily observed during routine calibration checks (zeroes and spans) except for an increase in the
408 time required to obtain stable zero and span values. The bias was not observed during any of the 2017
409 prescribed grassland burns. During the summer/fall 2018 North Carolina study and prior to the start of
410 the 2019 chamber burns, a new MnO₂ scrubber was installed and resulted in a significant and immediate
411 reduction of the observed high bias, shown in Fig. 4 (bottom panel) and Fig. S5.

412 **3.3 Methodological Influence on Ozone Measurements in Biomass Burning Smoke**

413 As discussed in Sections 3.1 and 3.2, there are large (order of magnitude level) differences in O₃
414 concentration measurement results obtained from the FRM (NO-CL) and the FEM UV photometric with
415 catalytic scrubber (UV-C) O₃ methods. The extremely low O₃ concentrations measured by the NO-CL
416 instrument is consistent with O₃ depletion in the presence of high NO_x concentrations (up to ppm levels)
417 observed in the grass burning plumes and during chamber burns. The reaction between NO and O₃ is
418 rapid and occurs on the timescales of seconds to minutes. As a result, high NO in the fresh biomass
419 combustion plumes will efficiently titrate out O₃ leading to near-field depletion within the plumes relative
420 to background concentrations. There was no sign of a positive interference in the NO-CL monitors, and
421 it remains the most robust and accurate routine method for O₃ measurement in fresh and downwind
422 biomass burning plumes.

423

424 In contrast with the NO-CL FRM instrument results, the UV-C FEM results showed substantial increases
425 in reported O₃ concentrations in the fresh biomass burning plumes. There is no known pathway for direct
426 O₃ emission from biomass burning, and the proximity (meters to hundreds of meters) and timescales
427 (seconds to minutes travel time from the combustion source to measurement) involved are too short for
428 the usual NO_x – VOC photochemistry to produce secondary O₃. Further, since the FSL chamber interior
429 is not exposed to sunlight, photochemistry would not have been active in the Missoula laboratory burns.
430 For the purposes of this work, the positive analytical artifact in the UV-C method, termed $\Delta O_{3(UV-C)}$, is
431 estimated using Eq. (6) as the difference between UV-C and the NO-CL O₃ concentration measurement
432 results for the same time period:

433

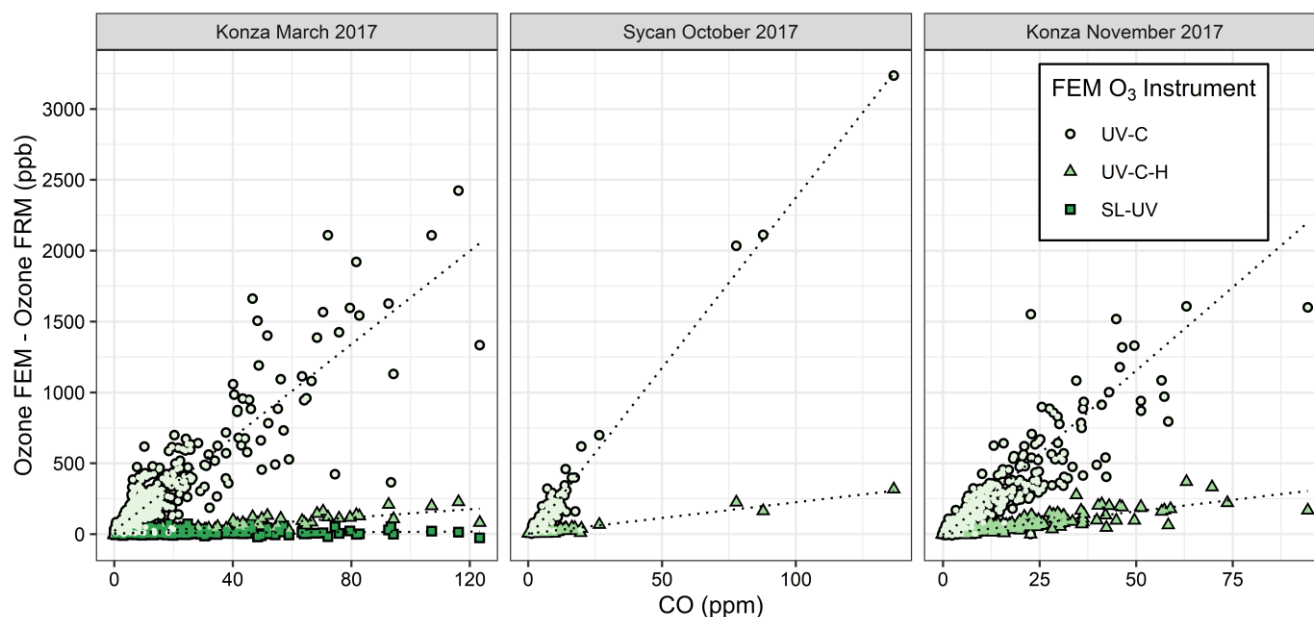
434

$$\Delta O_{3(UV-C)} = UV-C - NO-CL \quad (6)$$

435

436 Figure 5 shows “in plume” regressions between $\Delta O_{3(UV-C)}$ and the FRM measurement and CO for the
 437 three measured prescribed grassland burns in 2017 (Supplementary Fig. S6 shows the time series of

438



439

440 **Figure 5.** Scatter plots between FEM and FRM O_3 differences and the CO measurements within the
 441 grassland fires smoke plumes. The FEM measurements are differentiated by color and shape. The SL-UV
 442 method was only run during the Konza March 2017 measurements.

443 $\Delta O_{3(UV-C)}$ and CO). Figure 5 and Supplementary Fig. S6 show good correlations within the smoke plumes.

444 The average and maximum values of $\Delta O_{3(UV-C)}$ are summarized in Table 3. It is hypothesized that the

445 large “ O_3 ” measurement observed in the UV-C method results from a positive interference or artifact,

446 likely linked to VOC emissions in the grassland burn plumes. VOCs are emitted in higher concentrations

447 from the smoldering phase of combustion, which is also characterized by large CO emissions (Yokelson

448 et al., 1996; Yokelson et al., 1997), so a correlation between CO and O_3 artifact would support the

449 hypothesis of a VOC-linked interference for the UV-C instruments. This is also consistent with observed

450

451 **Table 3: Ozone artifact (ΔO_3) averages, maximum values, and CO, NO₂, and THC averages from**
 452 **the prescribed fire and USFS chamber burns as measured by the UV-C, UV-C-H, and UV-G**
 453 **instruments.**

Study	ΔO_3 avg. (ppb)	ΔO_3 max (ppb)	CO avg. (ppm)	NO ₂ avg. (ppb)	THC avg. (ppm)
ΔO_3(UV-C)					
Mar. 2017 Konza Prairie (KS)	295.8	2423.7	15.8	147.3	-
Oct. 2017 Sycan Marsh (OR)	170.2	3235.5	8.5	60.5	2.7
Nov. 2017 Konza & Tallgrass Prairies (KS)	330.0	3156	14.1	156.9	4.0
Apr. 2018 USFS Chamber (MT)	36.5	309.6	3.8	35.6	2.8
Apr. 2019 USFS Chamber (MT)	66.9	530.9	2.1	31.7	4.8
ΔO_3(UV-C-H)					
Mar. 2017 Konza Prairie (KS)	42.8	227.1	15.8	147.3	-
Oct. 2017 Sycan Marsh (OR)	21.1	316.4	8.5	60.5	2.7
Nov. 2017 Konza & Tallgrass Prairies (KS)	40.2	369.0	14.1	156.9	4.0
Apr. 2018 USFS Chamber (MT)	7.2	136.8	3.8	35.6	2.8
ΔO_3(UV-G)					
Apr. 2019 USFS Chamber (MT)	22.9	376.8	2.1	31.7	4.8
ΔO_3(SL-UV)					
Mar. 2017 Konza Prairie (KS)	8.3	74.2	15.8	147.3	-
Apr. 2018 USFS Chamber (MT)	0.5	11.5	3.8	35.6	2.8
Apr. 2019 USFS Chamber (MT)	1.7	32.1	2.1	31.7	4.8

454

455 VOC interferences in previous studies (Grosjean and Harrison, 1985; Kleindienst et al., 1993; Spicer et
 456 al., 2010) and observed following fireworks (Fiedrich et al., 2017; Xu et al., 2018).

457

458 The presence of a Nafion[®]-based humidity conditioning system (Nafion[®] tube dryer) significantly
 459 reduced the magnitude of the observed artifact as evident by comparing the UV-C and UV-C-H results
 460 shown in Figs. 1-3 and Supplementary Figs. S1 – S5. As with the UV-C method, the artifact in the UV-
 461 C-H method, ΔO_3 (UV-C-H), is calculated using Eq. (7) as the difference between UV-C-H and the NO-CL
 462 O₃ concentration measurement results for the same time period:

463

464

$$\Delta O_3(\text{UV-C-H}) = \text{UV-C-H} - \text{NO-CL} \quad (7)$$

465

466 The addition of the Nafion[®]-based humidity conditioning system reduces the magnitude of the $\Delta O_{3(UV-C-H)}$ artifact by approximately an order of magnitude as compared with the UV-C method. This is further
467 H) artifact by approximately an order of magnitude as compared with the UV-C method. This is further
468 illustrated in the 2018 chamber burns, where prior to beginning the final burn day on April 24, 2018, a
469 Nafion[®] tube dryer (PermaPure, MD Monotube Dryer Series) was installed on the UV-C method (Thermo
470 49i) in effect, converting it to a UV-C-H method. As shown in Fig. 4 and Supplementary Fig. S4, the
471 addition of the Nafion[®] tube dryer significantly reduced the $\Delta O_{3(UV-C)}$ artifact to a point comparable with
472 that observed in the UV-C-H method (2B 205). A possible explanation for this effect is presented and
473 discussed in Section 3.5. In addition, the previously described bias related to the damaged MnO₂ scrubber
474 was also reduced upon addition of the Nafion[®] dryer to the UV-C method.

475

476 For the March 2017 Konza Prairie study (Fig. 1) and the 2018 and 2019 USFS chamber studies (Fig. 3)
477 the SL-UV instrument concentration results were comparable to, although noisier and slightly higher than,
478 the NO-CL reference instrument. On numerous occasions during the prescribed and chamber burns, the
479 SL-UV instrument shows short (i.e. one-minute data point) positive or negative excursions that are not
480 also observed in the NO-CL results. In addition, these excursions are not correlated with changes in CO
481 concentrations. Because the SL-UV is a dual cell instrument that measures O₃ by comparing the
482 absorbance of two cells, it is critical in highly dynamic environments (such as during this study) that both
483 cells be measuring the same air at the same time. A slight difference in flow rates or residence times
484 between the two pathways (or a delay in one pathway relative to the other) will cause short term variability
485 in the difference between the two cells. Although this does not pose an issue for longer time averaging
486 (i.e. hourly data) under stable conditions, the dynamic nature of biomass burning plumes (i.e. changing
487 on the order of seconds) and short time averages (i.e. minute) can create issues (noise) for the SL-UV
488 method.

489

490 Significant analytical artifacts were observed for FEM UV photometric O₃ instruments with (UV-C-H)
491 and without (UV-C) Nafion[®]-based humidity conditioning system, where it appears that the dual effect
492 of ambient humidity fluctuations and VOC interferences caused large positive over-measurement of “in-
493 smoke” O₃ concentrations. Chemiluminescence monitors are highly specific to O₃ and have long been

494 known to be free of VOC interferences (Long et al., 2014; U.S. EPA, 2015). However, studies have shown
495 that the chemiluminescence method can be impacted by changes in relative humidity (Kleindienst et al.,
496 1993). As such, upon promulgation in 2015, the new NO-CL FRM regulatory text requires a humidity
497 correction/dryer system to eliminate the potential water vapor interference. As configured from the
498 manufacturer, the NO-CL based Teledyne-API Model T265 instrument operated during this comparative
499 study employs Nafion[®] drying technologies to reduce or eliminate the water vapor interferences. The use
500 of a chemical (NO) scrubber for UV photometric instruments (such as the 2B Technologies Model 211)
501 is very specific to O₃ and shows a much better response than the catalytic scrubber instruments,
502 performing almost as well as the NO-CL FRM, and has significant potential as a low-interference O₃
503 method. Of the catalytic scrubber photometric instruments those with Nafion[®]-based humidity
504 equilibration (2B Technologies Model 205) perform significantly better than those without (Thermo 49
505 series).

506

507 In areas highly impacted by smoke or for studies focusing on biomass burning plumes, the use of a NO-
508 CL FRM instrument is highly recommended as it was found to be essentially interference-free. These
509 instruments are anchored to absolute O₃ concentrations through the use of certified O₃ calibration sources,
510 many of which are based on UV photometry. The newest generation of commercially-available NO-CL
511 FRM instruments, including that used here (the Teledyne T265), have a built-in drying system to correct
512 for the humidity artifacts that affected earlier generation chemiluminescence instruments (Kleindienst et
513 al., 1993), making remaining interferences negligible compared to other technologies.

514

515 The gas-phase chemical scrubber UV instrument (2B 211), did not perform as well as the FRM under the
516 prescribed grassland burns or chamber experimental conditions tested here, with the high time resolution
517 (1-minute) data showing a much higher degree of variability than the NO-CL FRM instrument. We
518 hypothesize that the main factor driving this divergence between this method and the NO-CL FRM is the
519 dual-cell differential configuration of the instrument, which is not conducive to rapidly changing
520 concentrations in O₃ or other absorbing gases, such as VOCs.

521

522 In smoke-impacted monitoring situations where the use of a UV photometric instrument is still preferred
523 or required, the choice of a monitor with humidity equilibration provides a significant analytical
524 improvement over those monitors without the humidity corrections. In the absence of an instrument with
525 a Nafion[®] tube dryer and in non-regulatory applications, a dryer can be installed before the inlet or
526 measurement cells to reduce the interference as was demonstrated on the final day of the 2018 Missoula
527 chamber burns. This will have the added benefit of reducing positive biases from humidity and reducing
528 equilibration time for calibrations (especially when switching from high humidity ambient air to dry
529 calibration gases).

530 **3.4 Magnitude of Ozone Artifact in Fresh Biomass Burning Plumes Relative to Markers of** 531 **Combustion**

532 It is difficult to estimate an absolute magnitude or correct for the observed O₃ analytical artifact since
533 primary emissions from biomass combustion are highly variable and depend upon the fuel loading, fuel
534 type and condition, phase of the fire, and the burn conditions (Yokelson et al., 1996; Yokelson et al.,
535 1997). However, assuming the interference is driven primarily by VOCs, the artifact should be correlated
536 with the excess CO ($\Delta\text{CO} = \text{CO}_{\text{plume}} - \text{CO}_{\text{background}}$). Because CO_{background} during the prescribed grassland
537 burns was below 200 ppb (relative to typical conditions of >2 ppm in the plume), ΔCO is estimated as the
538 total measured CO concentration. A simplified view of biomass combustion assumes an approximate
539 linear combination of two dominant emission phases, flaming combustion (characterized by emission of
540 highly oxidized compounds, such as CO₂, NO_x, and SO₂), and smoldering combustion (characterized by
541 emission of reduced or mixed oxidation state compounds, such as CO, CH₄, NH₃, H₂S, and most VOCs)
542 (Yokelson et al., 1996; Yokelson et al., 1997). Because the majority of VOCs are in a reduced or mixed
543 oxidation state, they tend to be co-emitting with CO during smoldering combustion, and the VOC
544 concentrations tend to be highly correlated with CO in fresh biomass burning plumes (Yokelson et al.,
545 1996). Scatterplots comparing the FEM instrument artifacts ($\Delta\text{O}_{3(\text{UV-C})}$) and CO for the three prescribed
546 grassland burning periods are shown in Fig. 5. Regression statistics of the comparison of $\Delta\text{O}_{3(\text{UV-C})}$ and
547 $\Delta\text{O}_{3(\text{UV-C-H})}$ with CO and THC for grassland burns are given in Table 4. The magnitude of the artifact
548 (estimated by the slope of the regression line of the CO vs ΔO_3 comparison), in ppb apparent O₃ per ppm

549 CO, ranges between 16 - 24 ppb ppm⁻¹ for the UV-C instrument, and 1.5-3 ppb ppm⁻¹ for the instrument
550 with humidity correction (UV-C-H). It is important to point out that CO, in and of itself, is not considered
551 to be an interfering species in the UV photometric determination of O₃ in that CO absorbs in the infrared
552 (IR). The slight differences in the magnitude of the artifacts (fitted regression slopes) along with the low
553 uncertainty (standard errors) values indicate that the magnitude of the artifact may be influenced by local
554 conditions that make each burn unique. Such conditions might include meteorological conditions, fuel
555 composition, fuel moisture content, and times spent in combustion phase (flaming vs smoldering). Similar
556 to CO, THC_s and NO₂ are indicative of combustion processes and are correlated with ΔO₃ as given in
557 Table 4 and Supplementary Figs. S7 and S8. In terms of THC, the magnitude of the artifact, in ppb
558 apparent O₃ per ppm THC, is significantly higher at ~88 ppb ppm⁻¹ for the UV-C instrument and ~13 ppb
559 ppm⁻¹ for the UV-C-H instrument. Both the prescribed grassland and Missoula chamber burns resulted in
560 what would be considered high PM concentrations (2-50 mg m⁻³). These high PM concentrations
561 however, are not considered to be interfering due to the presence of the inline particle filter assemblies
562 described in Sections 2.2 and 2.6.

563

564 Since the CO concentrations (from upwind fires) observed at most stationary sites from fire plumes are
565 usually on the order of one ppm to greater than 10 ppm (Landis et al., 2018), it is reasonable to assume
566 that O₃ artifacts in the range of 15 ppb to greater than 250 ppb can be observed when employing a UV-C
567 method. Similarly, O₃ artifacts in the range of 1.5 to above 30 ppb might be observed at smoke-impacted
568 sites monitoring with UV-C-H methods. As such, Nafion[®]-based humidity conditioning systems are
569 highly recommended for use if employing UV photometric methodology for monitoring O₃ in areas
570 impacted by wildfires or prescribed burns. As stated previously and as seen in Fig. 3 and Table 3, O₃
571 artifacts were observed during the Missoula chamber 2018 and 2019 burns in both the UV-C and UV-C-
572 H methods, although reduced compared to the prescribed grassland burns. The presence and magnitude
573 of the O₃ artifact strongly suggests that smoke generated from fuels typical of forests in the western United
574 States also result in a measurement interference in UV photometric methods. Since downwind O₃
575 production in biomass burning plumes is a significant issue in fire impacted regions, having reliable,
576 interference-free methods is critical for assessing the contribution of wildland fires to ambient O₃ levels.

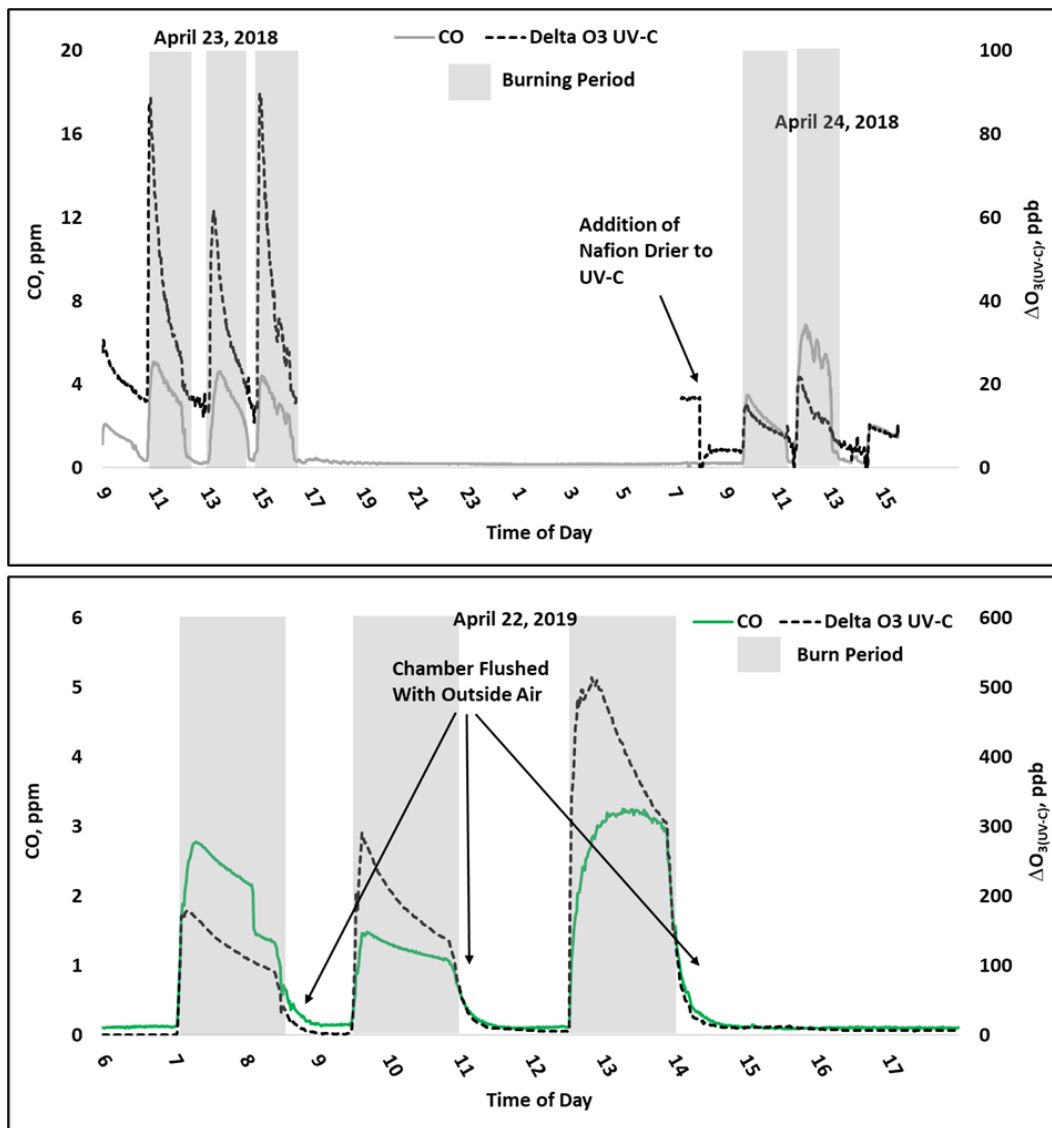
577 **Table 4: Regression statistics for the ozone artifact (ΔO_3) versus CO and THC for UV photometric**
 578 **instruments without (UV-C) and with (UV-C-H) a Nafion®-based humidity equilibration system**
 579 **during the 2017 prescribed grassland burns.**

Study	Slope (ppb/ppm)	Intercept (ppb)	r ²	n
$\Delta O_{3(UV-C)}$ vs CO				
Mar. 2017 Konza Prairie (KS)	16.46(±0.34) ^a	18.53(±6.72) ^b	0.79	653
Oct. 2017 Sycan Marsh (OR)	24.02(±0.25)	-28.05(±2.73)	0.96	295
Nov. 2017 Konza & Tallgrass Prairies (KS)	23.51(±0.73)	-20.8(±13.03)	0.74	461
$\Delta O_{3(UV-C)}$ vs THC				
Nov. 2017 Konza & Tallgrass Prairies (KS)	87.14(±3.74)	-85.36(±18.63)	0.59	461
$\Delta O_{3(UV-C-H)}$ vs CO				
Mar. 2017 Konza Prairie (KS)	1.46(±0.04)	0.87(±1.03)	0.80	163
Oct. 2017 Sycan Marsh (OR)	2.21(±0.05)	3.44(±0.54)	0.88	296
Nov. 2017 Konza & Tallgrass Prairies (KS)	3.24(±0.09)	-1.17(±1.67)	0.77	461
$\Delta O_{3(UV-C-H)}$ vs THC				
Nov. 2017 Konza & Tallgrass Prairies (KS)	13.27(±0.39)	-14.53(±1.92)	0.75	461
THC vs CO				
Nov. 2017 Konza & Tallgrass Prairies (KS)	0.21(±0.004)	1.55(±0.08)	0.79	461

580 ^aStandard error or uncertainty of the linear regression slope in ppb/ppm

581 ^bStandard error or uncertainty of the linear regression intercept in ppb

582 Figure 6 gives a detailed time series view of $\Delta O_{3(UV-C)}$ and CO from two burn days from 2018 and a single
 583 day during 2019. As indicated, $\Delta O_{3(UV-C)}$ and CO appear to be correlated in time but when performing
 584 linear regression comparisons of $\Delta O_{3(UV-C)}$ and CO during each years chamber burns as a whole,
 585 correlations tend to be poor. We suspect the positive O₃ bias is driven by one or more VOCs (likely
 586 oxygenated VOCs). In fresh smoke the excess concentrations of individual VOCs (ΔX), and VOC sums
 587 (ΔVOC), tend to be highly correlated with ΔCO (Yokelson et al., 1999; Gilman et al. 2015). The emission
 588 ratios of individual VOCs to CO ($\Delta X/\Delta CO$) can vary considerably with combustion conditions such as
 589 fuel type and condition (e.g. moisture content and decay state), fuel bed properties, such as bulk density,
 590 and the relative mix of flaming and smoldering combustion (Gilman et al. 2015; Koss et al., 2017).
 591 Additionally, the response of $\Delta X/\Delta CO$ to burn conditions varies among VOCs. When each burn is
 592 considered individually or in groups with similar conditions, the correlations between ΔO_3 , CO, and THC
 593 are enhanced. An example of this behavior is shown in Supplementary Fig. S10. For the chamber burns



595

596 **Figure 6.** Time series example of USFS chamber burn $\Delta O_3(UV-C)$ and CO concentration results from
 597 April 23-24, 2018 (top) and April 22, 2019 (bottom).

598

599 the magnitude of the ozone artifacts in ppb apparent O_3 per ppm CO, ranges between 6 - 210 ppb ppm⁻¹
 600 for the individual burns. R^2 and standard error values were consistent with those observed during the
 601 prescribed burns (see Table 4). The lack of a consistent relationship between the O_3 artifact and ΔCO

602 across all FSL chamber burns, while observing a good correlation for individual burns, likely reflects the
603 variable response of artifact producing emission(s) to the different combustion conditions of the burns.

604

605 One interesting observation from the data obtained from both the prescribed grassland and chamber burns
606 is the order of magnitude difference in the average and maximum O₃ artifact between the UV-C and the
607 UV-C-H instruments as shown in Table 3. Considering that the prescribed grassland and chamber burns
608 were conducted under dry (RH < 50%) conditions, the size of the difference (as large as hundreds of ppb)
609 cannot be explained purely by the previously observed relative humidity effects on measurements (Leston
610 et al., 2005; Wilson et al., 2006), suggesting that the Nafion[®] dryer is directly impacting the concentrations
611 of other interferents in the sample stream.

612

613 **3.5 Potential Reason for Lower Artifacts with Methods Employing Nafion[®]-based Humidity** 614 **Equilibration**

615 Nafion[®] is a sulfonated tetrafluoroethylene polymer that is highly permeable to water but shows little
616 permeability to many other organic and inorganic species (Mauritz et al., 2004). As a result, Nafion[®]-
617 based drying systems are often used as part of sample preparation or conditioning systems in analytical
618 chemistry to remove water vapor from sample streams prior to sample analysis. Nafion[®] membranes were
619 introduced to some O₃ monitors as a method to address humidity effects observed in UV-C O₃ monitors,
620 particularly when there are rapid changes in relative humidity level (Wilson and Birks, 2006). Humidity
621 can affect the transmission of the UV light through the detection cell and catalytic O₃ scrubbers can
622 modulate the water vapor in the scrubbed channel by acting as a temporary reservoir, resulting in
623 significant positive or negative O₃ interferences during rapid swings in relative humidity Wilson et al.,
624 2006). Adding a Nafion[®]-based equilibration dryer immediately prior to the measurement cells reduces
625 this water vapor interference without affecting O₃ concentrations, and thus significantly reduces the
626 humidity artifacts in UV photometric O₃ instruments.

627

628 Despite the high selectivity of Nafion[®] to water vapor, it does demonstrate partial to complete
629 permeability to various VOC or semivolatile organic compounds. Nafion[®] membranes are highly

630 permeable to alcohols, amines, ketones, and some water-soluble ethers (Baker, 1974), as well as some
631 biogenic oxygenated compounds (Burns et al., 1983). In addition, Nafion[®] membranes have been shown
632 to catalyze the decomposition and rearrangement of monoterpene compounds (Burns et al., 1983).
633 Systematic study of Nafion[®] permeability and reactivity for polar and oxygenated compounds has been
634 limited, with most users of Nafion[®] membranes basing their use on operational testing and confirmation
635 for the targeted use.

636

637 The significant (order of magnitude) reduction in the O₃ artifact with the addition of a Nafion[®]-based
638 dryer to the UV-C suggests that the Nafion[®] dryer is directly impacting the major interfering species
639 which was hypothesized to be VOCs emitted during combustion processes. The species that are
640 responsible for most of the O₃ artifact in UV-C O₃ instruments would have to be permeable through
641 Nafion[®] membranes or reactive with Nafion[®] membranes, be scrubbed by solid-phase, catalytic O₃
642 scrubbers (such as MnO₂ or hopcalite), and would have a significant absorption cross section around 254
643 nm. The absorption cross-section of O₃ around 254 nm is on the order of 10⁻¹⁷ cm² molecule⁻¹ (Molina
644 and Molina, 1986), which means species with absorptions around 10⁻¹⁷ cm² molecule⁻¹ at 254 nm would
645 be potential interfering species. As a class, aromatic VOCs and specifically oxygenated aromatic species
646 (and other polar derivatized species) absorb strongly in this region of the UV spectrum, and their potential
647 permeability through Nafion[®] membranes result in them being likely compounds to interfere in UV-C
648 instruments. As an example, aromatic aldehydes such as o-tolualdehyde and p-tolualdehyde absorbances
649 around 5x10⁻¹⁸ cm² molecule⁻¹ and 4x10⁻¹⁸ cm² molecule⁻¹, respectively (Etzkorn et al., 1999). Both 2,4-
650 dimethylbenzaldehyde and 2,6-dimethylbenzaldehyde have absorption cross sections above 10⁻¹⁷ cm²
651 molecule⁻¹ at 254 nm (El Dib et al., 2008). Baker (1974) found 75% of benzaldehyde was removed by a
652 Nafion[®] membrane, meaning that the Nafion[®] permeability of tolualdehydes and dimethylbenzaldehydes
653 is also likely to be high. In addition, benzaldehyde was almost quantitatively removed by several
654 commercial catalytic O₃ scrubbers, including the Thermo 49i MnO₂ catalytic scrubber (Kleindienst et al.,
655 1993), so similar aldehydes are likely to behave in a similar manner. Therefore, substituted aromatic
656 aldehyde species are one class of compounds that fit the necessary criteria for causing the interference on
657 the UV-C while having a reduced interference on the UV-C-H instrument. Future work examining the

658 potential interferences from different species (or classes of species) on a species or class specific basis
659 are required to confirm this potential mechanism and suggest others.

660 **4 Implications**

661 Wildland fires (wildfires and prescribed fires) emit significant amounts of VOCs and NO_x, two important
662 precursors in the photochemical formation of tropospheric O₃. Therefore, it is not surprising that large
663 increases in O₃ are routinely reported at ambient monitoring sites downwind from wildland fires (DeBell
664 et al., 2004; Bytnerowicz et al., 2010; Preisler et al., 2010; Jaffe et al., 2012; Bytnerowicz et al., 2013;
665 Jaffe et al., 2013; Lu et al., 2016; Lindaas et al., 2017; Baylon et al., 2018; Liu et al 2018; McClure and
666 Jaffe, 2018). For example, Buysse et al. (2019) examined regulatory air monitoring data from 18 cities
667 over a five period, and found that July – September exceedances of NAAQS for O₃ were far more common
668 on days with known wildland fire smoke impacts (4.6%) than those without (<0.1%). However, the results
669 of this study suggest caution when interpreting UV photometric method O₃ measurements under
670 conditions of wildfire smoke impact due to the significant positive artifacts that were observed. The
671 analytical artifacts were also shown to be positively correlated with widely used markers of combustion
672 such as CO and THC suggesting that the artifacts arise from photometric measurement interferences by
673 VOCs and further complicating the interpretation of smoke impacted UV photometric O₃ data. As
674 described in section 3.4, it reasonable to assume that O₃ artifacts in the range of a few ppb to greater than
675 250 ppb in addition to actual photochemically formed O₃ can be observed when employing UV
676 photometric methods at sites downwind from fires.

677

678 A detailed example of observed artifacts in the UV photometric method occurred during the 2016 Fort
679 McMurray Horse River wildfire in Alberta, Canada, where elevated “O₃” concentrations were observed
680 at multiple community based air monitoring sites utilizing UV-C instruments in the vicinity of the fire
681 (Landis et al., 2018). Reported “O₃” concentrations reached maximum hourly concentrations in excess of
682 1500 ppb using UV-C methods at night (between 10:00 PM and 5:00 AM local) in the absence of
683 photochemistry and were positively correlated with the combustion markers NO and non-methane

684 hydrocarbon (NMHC). Peaks in O₃ concentration are expected to be negatively correlated with peaks in
685 NO concentration as it rapidly titrates O₃ to NO₂, and the authors hypothesized that UV photometric
686 measurement artifacts may have been responsible for the unexpected observations.

687

688 The findings from this research effort and the observations from ambient studies (Landis et al., 2018)
689 raise concerns that routine regulatory monitoring and wildland fire research study O₃ measurements
690 utilizing UV photometric FEM instruments may be reporting positive measurement artifacts as O₃ during
691 smoke impacted events. Some studies have hypothesized that rapid photochemical processing was
692 responsible for reported elevated O₃ concentrations reported in downwind wildfire plumes (Liu et al.,
693 2017). Since downwind O₃ production in biomass burning plumes is a significant issue in fire impacted
694 regions, having reliable, interference-free methods is critical for assessing the contribution of wildland
695 fires to ambient O₃ levels and developing/validating accurate deterministic air quality models. Air quality
696 researchers and environmental regulators are strongly encouraged to utilize NO-CL FRM O₃ instruments
697 in areas routinely impacted by wildland fire smoke.

698 **5 Conclusions**

699 In this study, we compare two different O₃ measurement methods (chemiluminescence and UV
700 photometry) in fresh biomass burning plumes from prescribed grassland fires and during controlled
701 chamber burns. Within the UV photometry category, we look at two different technologies, one using a
702 gas-phase chemical scrubber (NO) and the second using solid phase catalysts to scrub O₃ from analytical
703 reference channels. Among the UV photometric instruments employing solid phase catalytic scrubbers,
704 we evaluated and compared methods that include a Nafion[®]-based humidity equilibration system with
705 those that do not.

706

707 The NO-CL method, recently promulgated as the O₃ FRM, performed well, even in fresh plumes, whereas
708 the UV photometric methods displayed varying degrees of positive measurement artifacts. The UV
709 photometric method employing the dynamic NO gas phase scrubber performed comparably with the NO-

710 CL method but was not well suited to the rapidly varying concentrations of VOCs in the smoke plumes.
711 The catalytic scrubber photometric methods demonstrated positive analytical artifacts that were correlated
712 with CO and THC concentrations (both biomass burning plume indicators). There was a significant
713 difference between the catalytic scrubber UV instruments with and without Nafion[®]-based humidity
714 correction, with the dryer system reducing the positive O₃ artifact by an order of magnitude as compared
715 with the UV photometric method employing no humidity correction. The observed reduction in artifacts
716 cannot be attributed only to elimination of the relative humidity/water vapor interferences and likely result
717 from post-scrubber equilibration or reaction of Nafion[®]-permeable VOCs prior to the measurement cell.
718 The results of this study strongly suggest that careful consideration be given to employed measurement
719 methods when monitoring O₃ concentrations in regions where impacts from biomass burning routinely
720 occur due to the significant impact of potential measurement interferences. In addition to consideration
721 of operating methods containing Nafion[®]-based humidity condition systems, attention should be focused
722 on the scrubbers employed by UV photometric methods and the adverse effects that operation in smoke
723 may have on those scrubbers. Further research is being conducted to evaluate the magnitude of the artifact
724 in the UV photometric method at routine monitoring sites that are often impacted by wildland fire smoke
725 events under the EPA Mobile Ambient Smoke Investigation Capability (MASIC) program (U.S. EPA
726 2019).

727

728 **Data Availability**

729 Datasets related to this manuscript can be found at <https://catalog.data.gov/dataset/epa-sciencehub>.

730

731 **Author Contributions**

732 Russell W. Long served as principal investigator and prepared the manuscript with contributions from all
733 co-authors. Russell W. Long, Andrew Whitehill, Andrew Habel, Maribel Colón, Shawn Urbanski, and
734 Matthew S. Landis performed the prescribed grassland fire and FSL chamber-based data collection and/or
735 analysis. Surender Kaushik performed supervisory review of this research effort and corresponding
736 manuscript.

737

738 **Competing Interests**

739 The authors declare that they have no conflict of interest.

740

741 **Disclaimer**

742 The views expressed in this paper are those of the authors and do not necessarily reflect the views or
743 policies of EPA. It has been subjected to Agency review and approved for publication. Mention of trade
744 names or commercial products do not constitute an endorsement or recommendation for use.

745

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751 **References**

752 Akagi, S. K., Craven, J. S., Taylor, J. W., McMeeking, G. R., Yokelson, R. J., Burling, I. R., et al.,
753 Evolution of Trace Gases and Particles Emitted by a Chaparral Fire in California. *Atmosph. Chem. Phys.*,
754 *12*: 1397-1421, 2012.

755

756 Baker, B. B., Measuring trace impurities in air by infrared spectroscopy at 20 meters path and 10
757 atmospheres pressure, *Am. Ind. Hyg. Assoc. J.*, *35*, 735-740, 1974.

758

759 Baylon, P., Jaffe, D. A., Hall, S. R., Ullmann, K., Alvarado, M. J., & Lefer, B. L., Impact of Biomass
760 Burning Plumes on Photolysis Rates and Ozone Formation at the Mount Bachelor Observatory, *J.*
761 *Geophys. Res.; Atmos.*, *123*: 2272–2284, 2018.

762

763 Bertschi, I.; Yokelson, R.J.; Ward, D.E.; Babbitt, R.E.; Susott, R.A.; Goode, J.G.; Hao, W.M., Trace gas
764 and particle emissions from fires in large diameter and belowground biomass fuels, *J. Geophys. Res.*, *108*
765 (D13):8472, 2003.

766

767 Boylan, P., Helmig, D., and Park, J.H., Characterization and mitigation of water vapor effects in the
768 measurement of ozone by chemiluminescence with nitric oxide, *Atmos. Meas. Tech.* 7, 1231-1244, 2014.
769

770 Burns, W. F., Tingey, D. T., Evans, R. C., and Bates, E. H., Problems with a Nafion® membrane dryer
771 for drying chromatographic samples, *J. Chromatogr. A*, 269, 1-9, 1983.
772

773 Buysse, C. E., Kaulfus, A., Nair, U., & Jaffe, D. A., Relationships Between Particulate Matter, Ozone,
774 and Nitrogen Oxides During Urban Smoke Events in the Western US. *Environ. Sci. Technol.*, 53: 12519-
775 12528, 2019.
776

777 Bytnerowicz, A., Cayan, D., Riggan, P., Schilling, S., Dawson, P., Tyree, M., Wolden, L., Tissell, R.,
778 Preisler, H., Analysis of the Effects of Combustion Emissions and Santa Ana Winds on Ambient Ozone
779 During the October 2007 Southern California Wildfires, *Atmos. Environ.*, 44: 678-687, 2010.
780

781 Bytnerowicz, A., Burley, J.D., Cisneros, R., Preisler, H.K., Schilling, S., Schweizer, D., Ray, J., Dulen,
782 D., Beck, C., Auble, B., Surface Ozone at the Devils Postpile National Monument Receptor Site during
783 Low and High Wildland Fire Years, *Atmos. Environ.*, 65: 129-141, 2013.
784

785 Christian, T. J., Kleiss, B., Yokelson, R. J., Holzinger, R., Crutzen, P. J., Hao, W. M., W.M., Saharjo,
786 B.H., Ward, D.E., Comprehensive laboratory measurements of biomass-burning emissions: 2. First
787 intercomparison of open-path FTIR, PTR-MS, and GC- MS/FID/ECD, *J. Geophys. Res.; Atmos.*,
788 109(D2), 2004.
789

790 DeBell, L. J., Talbot, R. W., Dibb, J. E., Munger, J. W., Fischer, E. V., & Frohking, S. E., A Major
791 Regional Air Pollution Event in the Northeastern United States Caused by Extensive Forest Fires in
792 Quebec, Canada, *J. Geophys. Res.; Atmos.*, 109(D19): 2004.
793

794 Dunlea, E., Herndon, S., Nelson, D., Volkamer, R., Lamb, B., Allwine, E., Grutter, M., Ramos Villegas,
795 C., Marquez, C., and Blanco, S., Evaluation of standard ultraviolet absorption ozone monitors in a
796 polluted urban environment, *Atmos. Chem. Phys.*, 6, 3163-3180, 2006.
797

798 El Dib, G., Chakir, A., and Mellouki, A., UV absorption cross-sections of a series of
799 dimethylbenzaldehydes, *J. Phys. Chem. A*, 112, 8731-8736, 2008.
800

801 Etzkorn, T., Klotz, B., Sørensen, S., Patroescu, I. V., Barnes, I., Becker, K. H., and Platt, U., Gas-phase
802 absorption cross sections of 24 monocyclic aromatic hydrocarbons in the UV and IR spectral ranges,
803 *Atmos. Environ.* 33, 525-540, 1999.
804

805 Fiedrich, M., Kurtenbach, R., Wiesen, P., and Kleffmann, J., Artificial O₃ formation during fireworks,
806 *Atmos. Environ.* 165, 57-61, 2017.
807

808 Gilman, J. B., Lerner, B. M., Kuster, W. C., Goldan, P. D., Warneke, C., Veres, P. R., et al., Biomass
809 burning emissions and potential air quality impacts of volatile organic compounds and other trace gases
810 from fuels common in the US. *Atmos. Chem. Phys.*, 15(24), 13915-13938, 2015.
811

812 Grosjean, D., and Harrison, J., Response of chemiluminescence NO_x analyzers and ultraviolet ozone
813 analyzers to organic air pollutants, *Environ. Sci. Tech.*, 19, 862-865, 1985.
814

815 Huntzicker, J. J., and Johnson, R. L., Investigation of an ambient interference in the measurement of
816 ozone by ultraviolet absorption photometry, *Environ. Sci. Tech.*, 13, 1414-1416, 1979.
817

818 Jaffe, D.A., Wigder, N.L., Ozone Production from Wildfires: A Critical Review, *Atmos. Environ.*, 51: 1-
819 10, 2012.
820

821 Jaffe, D.A., Wigder, N., Downey, N., Pfister, G., Boynard, A., Reid, S.B., Impact of Wildfires on Ozone
822 Exceptional Events in the Western U.S., *Environ. Sci. Technol.*, 47,11065–11072, 2013.
823

824 Johnson, T., Capel, J., Ollison, W., Measurement of microenvironmental ozone concentrations in
825 Durham, North Carolina, using a 2B Technologies 205 Federal Equivalent Method monitor and
826 interference-free 2B Technologies 211 monitor, *J. Air Waste Manage.*, 64, 360-371, 2014.
827

828 Kleindienst, T. E., Hudgens, E. E., Smith, D. F., McElroy, F. F., and Bufalini, J. J., Comparison of
829 chemiluminescence and ultraviolet ozone monitor responses in the presence of humidity and
830 photochemical pollutants, *Air Waste*, 43, 213-222, 1993.
831

832 Koss, A. R., Sekimoto, K., Gilman, J. B., Selimovic, V., Coggon, M. M., Zarzana, K. J., et al., Non-
833 methane organic gas emissions from biomass burning: identification, quantification, and emission factors
834 from PTR-ToF during the FIREX 2016 laboratory experiment. *Atmos. Chem. Phys.*, 18(5), 3299-3319,
835 2018.
836

837 Landis, M.S., Edgerton, E.S., White, E.M., Wentworth, G.R., Sullivan, A.P., Dillner, A.M., The impact
838 of the 2016 Fort McMurray Horse River Wildfire on ambient air pollution levels in the Athabasca Oil
839 Sands Region, Alberta, Canada. *Sci. Total Environ.*, 618:1665-1676, 2018.
840

841 Landis, M.S., Long, R.W., Krug, J., Colon, M., Vanderpool, R., Habel, A., Urbanski, S., The U.S. EPA
842 Wildland Fire Sensor Challenge: Performance and evaluation of Solver Submitted Multi-Pollutant Sensor
843 Systems. *Atmos. Environ.* In Press.
844

845 Leston, A. R., Ollison, W. M., Spicer, C. W., and Satola, J., Potential interference bias in ozone standard
846 compliance monitoring, *J. Air Waste Manage.*, 55, 1464-1472, 2005.
847

848 Lindaas, J., Farmer, D. K., Pollack, I. B., Abeleira, A., Flocke, F., Roscioli, R., et al., Changes in ozone
849 and precursors during two aged wildfire smoke events in the Colorado Front Range in summer 2015.
850 *Atmos. Chem. Phys.*, 17(17): 10691-10707, 2017.

851

852 Liu, X., Huey, L.G., Yokelson, R.J., Selimovic, V., Simpson, I.J., Müller, M., Jimenez, J.L.,
853 Campuzano-Jost, P., Beyersdorf, A.J., Blake, D.R., Butterfield, Z., Choi, Y., Crouse, J.D.,
854 Day, D.A., Diskin, G.S., Dubey, M.K., Fortner, E., Hanisco, T.F., Hu, W., King, L.E.,
855 Kleinman, L., Meinardi, S., Mikoviny, T., Onasch, T.B., Palm, B.B., Peischl, J., Pollack,
856 I.B., Ryerson, T.B., Sachse, G.W., Sedlacek, A.J., Shilling, J.E., Springston, S., St. Clair,
857 J.M., Tanner, D.J., Teng, A.P., Wennberg, P.O., Wisthaler, A., Wolfe, G.M., Airborne
858 Measurements of Western U.S. Wildfire Emissions: Comparison with Prescribed Burning
859 and Air Quality Implications. *J. Geophys. Res.; Atmos.* 122:6108–6129, 2017.

860

861 Liu, Z., Liu, Y., Murphy, J.P., Maghirang, R., Contributions of Kansas Rangeland Burning to Ambient
862 O₃: Analysis of data from 2001 to 2016, *Sci. Total Environ.*, 618: 1024–1031, 2018.

863

864 Long, R.W., Hall, E., Beaver, M., Duvall, R., Kaushik, S., Kronmiller, K., Wheeler, M., Garvey, S.,
865 Drake, Z., McElroy, F., Performance of the Proposed New Federal Reference Methods for Measuring
866 Ozone Concentrations in Ambient Air, EPA/600/R-14/432, 2014.

867

868 Lu, X., Zhang, L., Yue, X., Zhang, J., Jaffe, D.A., Stoh, A., Zhao, Y., Shao, J., Wildfire Influences on the
869 Variability and Trend of Summer Surface Ozone in the Mountainous Western United States, *Atmos.*
870 *Chem. Phys.*, 16, 14687–14702, 2016.

871

872 Mauritz, K. A., Moore, R.B., State of Understanding of Nafion, *Chem. Rev.*, 104, 10, 4535–4586, 2004.

873

874 McClure, C. D., & Jaffe, D. A., Investigation of High Ozone Events due to Wildfire Smoke in an Urban
875 Area. *Atmos. Environ.*, 194; 146-157, 2018.

876

877 Molina, L.T., Molina, M.J., Absolute Absorption Cross Sections of Ozone in the 185- to 350-nm
878 Wavelength Range, *J. Geophys. Res.; Atmos.*, 91 (D13):4719, 1986.

879

880 Ollison, W. M., Crow, W., Spicer, C. W., Field testing of new-technology ambient air ozone monitors, *J.*
881 *Air Waste Manage.*, 63, 855-863, 2013.

882

883 Parrish, D.D., Fehsenfeld, F.C., Methods for gas-phase measurements of ozone, ozone precursors and
884 aerosol precursors, *Atmos. Environ.*, 34, 1921-1957, 2000.

885

886 Preisler, H.K., Zhong, S., Esperanza, A., Brown, T.J., Bytnerowicz, A., Tarna, L., Estimating Contribution
887 of Wildland Fires to Ambient Ozone Levels in National Parks in the Sierra Nevada, California, *Environ.*
888 *Pollut.*, 158: 778-787, 2010.

889

890 Spicer, C. W., Joseph, D. W., and Ollison, W. M., A re-examination of ambient air ozone monitor
891 interferences, *J. Air Waste Manage.*, 60, 1353-1364, 2010.

892

893 Tong, H. Y., Karasek, F.W., Flame ionization detector response factors for compound classes in
894 quantitative analysis of complex organic mixtures. *Anal. Chem.*, 56, 2124–2128, 1984.

895

896 Turnipseed, A.A., Andersen, P., Williford, C., Ennis, C., Birks, J., Use of a heated graphite scrubber as a
897 means of reducing interferences in UV-absorbance measurements of atmospheric ozone, *Atmos. Meas.*
898 *Tech.*, 10, 2253–2269, 2017.

899

900 U.S. Environmental Protection Agency (EPA), National Ambient Air Quality Standards for Ozone,
901 *Federal Register*, 80, 206, October 26, 2015.

902

903 U.S. Environmental Protection Agency (EPA), Studies Advance Air Monitoring During Wildfires and
904 Improve Forecasting of Smoke, [https://www.epa.gov/sciencematters/studies-advance-air-monitoring-](https://www.epa.gov/sciencematters/studies-advance-air-monitoring-during-wildfires-and-improve-forecasting-smoke)
905 [during-wildfires-and-improve-forecasting-smoke](https://www.epa.gov/sciencematters/studies-advance-air-monitoring-during-wildfires-and-improve-forecasting-smoke), July 30, 2019.

906

907 Whitehill, A.; George, I.; Long, R.; Baker, K.R.; Landis, M.S., Volatile organic compound emissions
908 from prescribed burning in tallgrass prairie ecosystems. *Atmosphere*, 10, 464, 2019.

909

910 Williams, E. J., Fehsenfeld, F. C., Jobson, B. T., Kuster, W. C., Goldan, P. D., Stutz, J., and McClenny,
911 W. A., Comparison of ultraviolet absorbance, chemiluminescence, and DOAS instruments for ambient
912 ozone monitoring, *Environ. Sci. Technol.*, 40, 5755-5762, 2006.

913

914 Wilson, K. L., and Birks, J. W., Mechanism and elimination of a water vapor interference in the
915 measurement of ozone by UV absorbance, *Environ. Sci. Technol.*, 40, 6361-6367, 2006.

916

917 Yokelson, R.J.; Griffith, D.W.T.; Ward, D.E., Open-path Fourier transform infrared studies of large-scale
918 laboratory biomass fires. *J. Geophys. Res.; Atmos.*, 101(D15):21067-21080, 1996.

919

920 Yokelson, R.J., R Susott, R., Ward, D.E., Reardon, J., Griffith D.W.T., Emissions from smoldering
921 combustion of biomass measured by open-path Fourier transform infrared spectroscopy
922 *J. Geophys. Res.; Atmos.*, 102 (D15), 18865-18877, 1997.

923

924 Yokelson, R. J., Goode, J. G., Ward, D. E., Susott, R. A., Babbitt, R. E., Wade, D. D., et al., Emissions
925 of formaldehyde, acetic acid, methanol, and other trace gases from biomass fires in North Carolina
926 measured by airborne Fourier transform infrared spectroscopy. *J. Geophys. Res.; Atmos.*, 104(D23),
927 30109-30125, 1999.

928

929 Xu, Z., Nie, W., Chi, X., Huang, X., Zheng, L., Xu, Z., Wang, J., Xie, Y., Qi, X., and Wang, X., Ozone
930 from fireworks: Chemical processes or measurement interference?, *Sci. Total Environ.*, 633, 1007-1011,
931 2018.