



1 **Comparison of Ozone Measurement Methods in Biomass Burning Smoke: An evaluation under** 2 **field and laboratory conditions**

3 Russell W. Long¹, Andrew Whitehill¹, Andrew Habel², Shawn Urbanski³, Hannah Halliday¹, Maribel Colón¹, Surender
4 Kaushik¹, Matthew S. Landis¹

5 ¹Center for Environmental Measurement and Modeling, Office of Research and Development, United States Environmental
6 Protection Agency, Research Triangle Park, North Carolina, United States of America

7 ²Jacobs Technology Inc., Research Triangle Park, North Carolina, United States of America

8 ³U.S. Forest Service, Rocky Mountain Research Station, Missoula, MT, United States of America

9

10 *Correspondence to:* Russell W. Long (long.russell@epa.gov; (919) 541-7744)

11 **Abstract**

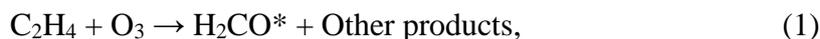
12 In recent years wildland fires in the United States have had significant impacts on local and regional air
13 quality and negative human health outcomes. Although the primary health concerns from wildland fires
14 come from fine particulate matter (PM_{2.5}), large increases in ozone (O₃) are also observed downwind of
15 wildland fire plumes. Conditions generated in and around wildland fire plumes, including the presence of
16 interfering chemical species, can make the accurate measurement of O₃ concentrations using the
17 ultraviolet (UV) photometric method challenging if not impossible. UV photometric method instruments
18 are prone to interferences by volatile organic compounds (VOCs) that are present at high concentrations
19 in wildland fire smoke. Four different O₃ measurement methodologies were deployed in a mobile
20 sampling platform downwind of active prescribed grassland fire lines in Kansas and Oregon and during
21 controlled chamber burns at the United States Forest Service, Rocky Mountain Research Station Fire
22 Sciences Laboratory in Missoula, Montana. We demonstrate that the Federal Reference Method (FRM)
23 nitric oxide (NO) chemiluminescence monitors and Federal Equivalent Method (FEM) gas-phase (NO)
24 chemical scrubber UV photometric O₃ monitors are relatively interference-free, even in near-field
25 combustion plumes. In contrast, FEM UV photometric O₃ monitors using solid-phase catalytic scrubbers
26 show positive artifacts that are positively correlated with carbon monoxide (CO) and total gas phase
27 hydrocarbons (THC), two indicator species of biomass burning. Of the two catalytic scrubber UV
28 photometric methods evaluated, the instruments that included a Nafion® tube dryer in the sample
29 introduction system had artifacts an order of magnitude smaller than the instrument with no humidity



30 correction. We hypothesize that Nafion[®]--permeable VOCs (such as aromatic hydrocarbons) could be a
31 significant source of interference for catalytic scrubber UV photometric O₃ monitors, and that the
32 inclusion of a Nafion[®] tube dryer assists with the mitigation of these interferences. The interference-free
33 chemiluminescence FRM method is highly recommended for accurate measurements of O₃ in wildland
34 fire plume studies and at regulatory ambient monitoring sites frequently impacted by wildland fire smoke.

35 **1 Introduction**

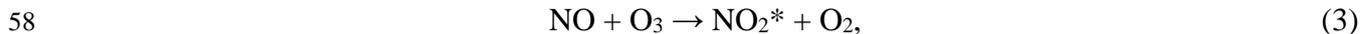
36 Ground-level ozone (O₃) is a secondary air pollutant generated from the photochemical interactions of
37 nitrogen oxides (NO_x) and volatile organic compounds (VOCs). The most robust methods for
38 interference-free O₃ measurements are based on chemiluminescence reactions with ethylene (ET-CL, for
39 ethylene chemiluminescence) or nitric oxide (NO-CL, for nitric oxide chemiluminescence) (Long et al.,
40 2014). The overall reaction mechanism for ET-CL generally proceeds as detailed in Eqs. (1-2):



43
44
45 The reaction generates electronically-activated formaldehyde (H₂CO*) which luminesces in the high
46 ultraviolet (UV) to visible portion of the spectrum (380 nm - 550 nm) and vibrationally-activated
47 hydroxide ions which luminesce in the visible light to the low infrared (IR) portion of the spectrum (550
48 nm - 800 nm). The number of photons emitted during the reaction is directly proportional to the amount
49 of O₃ present and are counted by a photomultiplier tube (PMT), with its response centered at 440 nm,
50 then the count is converted to O₃ concentration. The ET-CL method requires a constant supply of ethylene
51 for continuous operation. NO-chemiluminescence analyzers measure O₃ concentrations using the
52 principle that the dry, gas-phase reaction between NO and O₃ generates nitrogen dioxide in an activated
53 state (NO₂*), and oxygen (O₂) (Ollison et. al., 2013; Boylan et.al., 2014). As each unstable, NO₂*
54 molecule returns to a lower energy state (NO₂), it emits a photon (hν). The reaction causes luminescence
55 in a broadband spectrum ranging from visible light to infrared light (approximately 590 nm – 2800 nm).
56 The two-step gas-phase reaction proceeds as detailed in Eqs. (3-4):



57



60

61 The ET-CL method is no longer used nor produced commercially and has been replaced by the NO-CL
62 method. Similar to the ET-CL method, the NO-CL method requires a constant supply of gas, in this case
63 NO, for continuous operation. The ET-CL method was promulgated as the Federal Reference Method
64 (FRM) for measuring O₃ in the atmosphere in 1971 and the NO-CL method promulgated as the FRM in
65 2015 (U.S. EPA, 2015).

66

67 While the chemiluminescence method for measuring O₃ is technically robust and free of analytical
68 artifacts (Long et al., 2014), it is not widely used in the United States. Instead, Federal Equivalent Methods
69 (FEM) based upon UV photometry are employed at the majority of O₃ regulatory monitoring locations.
70 According to July 2020 data from the United States Environmental Protection Agency (EPA) Air Quality
71 System (AQS) database, the UV photometric method represents 99% of the roughly 1200 instruments
72 deployed in network monitoring for O₃ National Ambient Air Quality Standard (NAAQS) attainment.
73 UV photometric methods for O₃ are generally considered easier to deploy, operate, and in most cases do
74 not require external compressed gasses for operation. UV photometric analyzers determine O₃
75 concentrations by quantitatively measuring the attenuation of light due to absorption by O₃ present in an
76 absorption cell at the specific wavelength of 254 nm (Parrish et al., 2000; Williams et al., 2006). The O₃
77 concentration is determined through a two-step process in which the light intensity passing through the
78 sample air (I) is compared with the light intensity passing through similar sample air from which all O₃
79 is first removed (I₀). The ratio of these two light intensity values (I/I₀) provides the measure of the light
80 absorbed at 254 nm, and the O₃ concentration in the sample is then determined through the use of the
81 Beer-Lambert Law as given in Eq. (5):

82

83
$$I/I_0 = e^{-KLC} \quad (C = 1/KL \ln [I/I_0]); \quad (5)$$

84



85 where L is the length of the absorption cell (cm), C is the O_3 concentration (ppm), and K is the absorption
86 cross section of O_3 at 254 nm at standard atmospheric temperature and pressure conditions ($308 \text{ atm}^{-1} \text{ cm}^{-1}$)
87 ¹). Photometric monitors generally use mercury vapor lamps as the UV light source, with optical filters
88 to attenuate lamp output at other than the 254 nm wavelength.

89
90 Air for the reference cell measurement (I_0) is typically obtained by passing the ambient air sample stream
91 through a catalytic scrubber containing manganese dioxide (MnO_2), hopcalite (a mixture of Cu, Mn, and
92 Ag oxides), heated silver wool, or another solid state material to ‘scrub’ only O_3 from the sample air while
93 preserving all other substances in the sample air that potentially absorb at 254 nm (e.g., elemental gaseous
94 mercury [Hg^0], hydrogen, sulfide [H_2S], carbon dioxide [CO_2], VOCs) so that their effects are cancelled
95 in the differential I/I_0 measurement. The integrity of the O_3 reference scrubber is critical and may allow
96 measurement interferences if it does not perform adequately. Similarly, any tendency of the scrubber to
97 fail to effectively remove all O_3 from the reference sample will result in a measurement bias. In addition
98 to O_3 , catalytic scrubbers have been shown to remove UV-active VOCs (Kleindienst et al., 1993), creating
99 the potential for positive artifacts in O_3 measurements when the efficiency of this VOC removal is
100 impacted.

101
102 Although FEM designated UV photometric instruments are accurate under most ambient conditions,
103 locations with high VOC concentrations can produce significant analytical artifacts. Smoke plume
104 impacted locations and measurements downwind from wildland fires are a particular concern; O_3
105 measurements of up to 320 ppb were observed in a smoke plume in western Oregon using a Dasibi
106 1003AH UV photometric O_3 monitor (Huntzicker and Johnson, 1979), which also showed a correlation
107 between apparent O_3 and aerosol concentrations (b_{scat} , a combustion plume indicator in this case). O_3
108 measurements from UV photometric instruments exceeding 1500 ppb at night (22:00-05:00) were
109 observed in Fort McMurray, Alberta during smoke impacts from the 2016 Horse River Fire, which were
110 positively correlated with NO and non-methane hydrocarbons (Landis et al., 2018). Follow-up pyrolysis
111 experiments demonstrated that ET-CL instruments do not show a similar response to biomass burning
112 smoke (Huntzicker and Johnson, 1979). Photochemical chamber experiments comparing the O_3 response



113 of UV (Dasibi Model 1003AH, Dasibi Model 1008AH, and Thermo Model 49) and ET-CL (Bendix
114 Model 8002 and Monitor Labs Model 8410) mixtures show negligible differences for irradiated
115 paraffin/NO_x and olefin/NO_x mixtures, but do show a positive UV interference in mixtures with toluene
116 and other aromatics present (Kleindienst et al., 1993). Laboratory studies comparing the response of UV
117 (Thermo Model 49, Horiba APOA-370, and 2B Tech Model 202) and ET-CL (Bendix) instruments
118 showed a positive interference for o-nitrophenol, naphthalene, and p-tolualdehyde for the UV instruments
119 but not the ET-CL instruments (Grosjean and Harrison, 1985; Spicer et al., 2010). Additionally, during
120 the Mexico City Metropolitan Area field campaign (MCMA-2003) a mobile laboratory using an FEM
121 designated UV photometric O₃ monitor (unheated MnO₂ scrubber, Thermo 49 series) showed a large
122 positive O₃ interference (~400 ppb) associated with PM_{2.5} and polyaromatic hydrocarbons (PAHs) when
123 following some diesel vehicles (Dunlea et al., 2006). Although not compared to a chemiluminescence
124 instrument, those high O₃ values are unlikely real considering the high concurrent NO concentrations (in
125 some cases, >1000 ppb). The authors of this study attributed the interference to fine particles, based on
126 the correlation with PM_{2.5} and the lack of a correlation with gas-phase organic species measured by the
127 proton transfer reaction-mass spectrometer (PTR-MS, Dunlea et al., 2006).

128

129 In addition to interferences from the presence of aromatic VOCs and semi-volatile PAHs, water vapor
130 (relative humidity) issues have also been observed with older generation FRM and FEM designated
131 chemiluminescence and UV photometric O₃ instruments, respectively (Kleindienst et al., 1993; Leston et
132 al., 2005; Wilson and Birks, 2006). As such, Nafion[®] tube dryers are regularly incorporated into some
133 newer generation chemiluminescence and UV photometric O₃ monitors in an attempt to mitigate the
134 humidity related measurement artifacts.

135

136 A recently introduced variation of the UV photometric method, known as the “scrubberless” UV
137 photometric (SL-UV) method (Ollison et al., 2013), specifies removal of O₃ from the sample air for the
138 reference by a gas-phase reaction with NO rather than using a conventional solid state catalytic scrubber.
139 The NO gas phase chemical scrubber reacts with O₃ much faster and more selectively than with other
140 potential interfering compounds and is very effective at removing the O₃ without affecting other



141 interfering compounds that may be present in ambient air. The differential UV measurement can then
142 effectively reduce interferences to an insignificant level. Similar to NO-CL, the SL-UV method requires
143 a supply of NO gas for continuous operation.

144

145 In this study, we investigate UV photometric FEM instrument O₃ measurement interferences in fresh
146 biomass burning smoke plumes from prescribed grassland fires and during controlled burn experiments
147 in a large scale combustion chamber. We directly compare NO-CL FRM O₃ measurements to several
148 FEM designated UV photometric technologies, including a gas-phase scrubber and catalytic scrubbers
149 with and without Nafion[®] tube dryer systems. Based on the results from the measurements, we assess the
150 magnitude of the observed artifacts for different technologies and under various smoke conditions and
151 provide suggestions for potential mitigation of the interferences.

152

153 **2. Methods**

154 **2.1 Overview of Methods Evaluated**

155 In this study we compared the measurement results from six different commercially available FRM/FEM
156 designated O₃ instruments operated in ambient or chamber generated biomass burning smoke. All
157 instruments were operated according to their FRM or FEM designation. The six instruments differed by
158 measurement principle (chemiluminescence *versus* UV photometric), and by sample treatment
159 configuration (scrubber material, presence of dryer, etc.). For interference free O₃ measurements we
160 utilized the newly designated FRM NO-CL method (U.S. EPA, 2015). For the UV photometric methods,
161 we compared both catalytic scrubber and “scrubberless” (gas phase chemical scrubber) technologies, with
162 the “scrubberless” monitor using a NO chemical scrubber. Finally, within the catalytic scrubber UV
163 photometric category, we compared instruments with and without Nafion tube dryer systems. The
164 operation principle and designations (FRM vs FEM) for the analyzers under investigation are summarized
165 in Table 1 and described in Sections 2.1.1-2.1.4. These analyzers were operated immediately downwind



166 of fresh biomass burning plumes during eight days of prescribed fires in grassland ecosystems in Oregon
 167 and Kansas and during laboratory-based studies at the U.S. Forest Service’s (USFS) combustion facility
 168 at the Fire Sciences Laboratory (FSL) in Missoula, Montana. The grassland fire fuels consisted primarily
 169 of mixed native prairie tall grass of varying moisture content. Seven of the eight days of prescribed
 170 burning were conducted in the Tallgrass Prairie ecosystem of central Kansas (four days in March of 2017
 171 and three days in November of 2017). The additional burn day was conducted at the Sycan Marsh in
 172 central Oregon (October of 2017). Laboratory based chamber burns at the FSL were conducted during
 173 April 2018 and again during April 2019. Fuels for the laboratory based chamber burns consisted of
 174 ponderosa pine needles and fine woody debris. Details of the individual studies are provided in Sections
 175 2.2-2.6.

176

177 **Table 1: Ozone measurement methods investigated.**

Name	Manufacturer	Model	Method	Scrubber	Cells	Humidity Correction	Deployment ^a
U.S. EPA Federal Reference Methods (FRM)							
NO-CL	Teledyne API	T-265	CL (NO)	N/A	1	Nafion [®] -based (dryer)	K1, S, K2, T, M1, M2
U.S. EPA Federal equivalent methods (FEM)							
UV-C	Thermo Scientific	49i	UV (254 nm)	Catalyst (MnO ₂)	2	None	K1, S, K2, T, M1, M2
UV-C-H	2B Technologies	205	UV (254 nm)	Catalyst (Hopcalite)	2	Nafion [®] -based (equilibration)	K1, S, K2, T, M1
SL-UV	2B Technologies	211	UV (254 nm)	Gas chemical (NO)	2	Nafion [®] -based (equilibration)	K1, M1, M2
UV-G	2B Technologies	211-G	UV (254 nm)	Heated graphite	2	Nafion [®] -based (equilibration)	M2

178 ^aK1-Konza Prairie March 2017; S-Sycan Marsh, October 2017; K2-Konza Prairie November 2017; T-Tallgrass Prairie
 179 November 2017; M1-Missoula chamber April 2018; M2-Missoula chamber April 2019.

180

181 2.1.1 NO Chemiluminescence

182 The FRM O₃ measurement method was the interference-free Teledyne API (San Diego, CA, USA) Model
 183 T265 Chemiluminescence Monitor (TAPI T265), which utilizes a NO-CL measurement principle. These
 184 NO-CL O₃ analyzers have been shown to be free of interferences (Long et al. 2014) , and have been used
 185 as a reference method in other O₃ comparison studies (Williams et al., 2006; Landis et al., 2020). Although



186 there is a known water vapor interference with chemiluminescence technology (Kleindienst et al., 1993),
187 the TAPI T265 uses a Nafion[®] tube dryer system to remove water vapor from the air prior to making the
188 measurement, thus eliminating any humidity-related effects. Like the ET-CL technologies (Kleindienst
189 et al., 1993), the NO-CL analyzers have no documented VOC interferences.

190

191 **2.1.2 Catalytic Scrubber UV Photometric**

192 For this study the UV photometric method with no humidity correction was represented by the Thermo
193 Scientific (Franklin, MA, USA) Model 49i (Thermo 49i), which is a dual cell instrument with a
194 manganese oxide (MnO₂) catalytic scrubber, referred to as UV-C. Nafion[®]-based humidity systems or
195 dryers have been employed within photometric O₃ monitors with catalytic scrubbers before the
196 measurement cell, offering a reduction in relative humidity interferences and artifacts (Wilson and Birks,
197 2006).

198

199 The UV photometric with a Nafion[®] humidity conditioning system was represented in this study by a 2B
200 Technologies (Boulder, CO, USA) Model 205 (2B 205) O₃ monitor. The 2B 205 utilized a dual-cell
201 design where sample air and scrubbed air are measured simultaneously. The 2B 205 uses a Hopcalite
202 (CuO/MnO₂) catalytic scrubber to remove O₃ from the reference stream. This instrument will be referred
203 to as UV-C-H.

204

205 **2.1.3 Scrubberless UV Photometric**

206 For comparison with the NO-CL, UV-C and UV-C-H methodologies, a “scrubberless” UV (SL-UV)
207 photometric analyzer with a gas-phase (NO) chemical scrubber was employed (Ollison et al., 2013;
208 Johnson et al., 2014). The addition of NO gas to the reference stream selectively scrubs O₃ while not
209 significantly affecting interfering VOC species, resulting in an interference free O₃ determination.
210 Inclusion of this instrument into the study allows evaluation of the impact of the UV method in general
211 (as compared with chemiluminescence) versus the influence of specific scrubber technologies. The SL-



212 UV method is represented by the 2B Technologies Model 211 “Scrubberless” Ozone Monitor (2B 211).
213 The Model 2B 211 requires a continuous supply of compressed NO or nitrous oxide (N₂O) (which the
214 instrument converts to NO). The SL-UV method also utilizes a Nafion[®]-based sample humidity
215 conditioning system to eliminate any humidity effects. The SL-UV instrument was not used in the October
216 or November 2017 burns due to the lack of the necessary reagent gas (nitrous oxide, N₂O) needed to run
217 the instrument.

218

219 **2.1.4 Heated Graphite Scrubber UV Photometric**

220 During the final phase of laboratory-based burning a 2B Technologies Model 211-G UV photometric
221 analyzer (2B 211-G) was operated for comparison to the monitors detailed in Sections 2.1.1-2.1.3. The
222 2B 211-G differs from the 2B 211 in that it employs a heated graphite scrubber to remove O₃ from the
223 reference sample stream (I₀) (Turnipseed et al., 2017). The 2B 211-G utilizes the same Nafion[®]-based
224 sample humidity conditioning system as employed in the 2B 211. For the purposes of this manuscript the
225 UV photometric method employing the heated graphite scrubber be referred to as UV-G.

226

227 **2.2 Prescribed Fire Burn Mobile Sampling Platform**

228 During the prescribed fire grass burns, all study instrumentation (analyzers, data acquisition systems, and
229 peripheral systems) were mounted in portable instrument racks and installed inside an enclosed EPA 4x4
230 vehicle (Whitehill et al., 2019). During initial set up and in between burning periods instruments were
231 powered via land line AC power routed through two onboard mounted Tripp Lite (Chicago, IL, USA)
232 Model SMART3000RM2U 3000VA line interactive sine wave uninterruptible power supplies (UPS) each
233 running off its own 20A circuit to ensure that the instruments had a stable supply of clean power.
234 Immediately prior to deploying to the burns, the instruments were then switched to a bank of six 12V
235 100Ah batteries run through an AIMS Power (Reno, NV, USA) model PICOGLF30W12V120V 3000W
236 pure sine wave global low frequency inverter and charger located in a trailer towed behind the EPA
237 vehicle. An on-board 6500W Honda generator was used to maintain charge on the batteries. The



238 instruments were connected via perfluoroalkoxy alkane (PFA) Teflon[®] tubing (0.64 cm diameter) to PFA
239 Teflon[®] filter packs loaded with 47 mm, 5 micron pore size pressure drop equivalent Millipore
240 (Burlington, MA, USA) Omnipore[®] polytetrafluoroethylene (PTFE) filter membranes which were (i)
241 mounted to a rooftop sampling platform during Spring 2017 sampling, or (ii) connected to a cross-linked
242 Teflon[®]-coated high flow manifold mounted on the inside roof of the truck compartment during Fall 2017
243 sampling. The truck was positioned downwind of active biomass burning plumes, usually within meters
244 to hundreds of meters of the active fire line, and positioned so that the trailer was downwind of the sample
245 inlets (to avoid interferences from generator exhaust). In addition to the O₃ analyzers under investigation,
246 additional monitors were also operated for the determination of carbon monoxide (CO), NO, NO₂, total
247 oxides of nitrogen (NO_x=NO+NO₂), and THC. The operation principle and designation (FRM vs FEM)
248 information for the additional analyzers deployed in this study are summarized in Table 2. Data from all
249 instruments was recorded on a Envivas Ultimate data acquisition system.

250

251 **Table 2: Additional measurement methods operated during the present study.**

Pollutant	Manufacturer	Model	Method	FRM/FEM	Deployment ^f
CO	Teledyne API	48C	NDIR ^a	FRM	K1, S, K2, T, M1, M2
NO ₂	Teledyne API	T500U	CAPS ^b	FEM	K1, S, K2, T, M1, M2
NO, NO ₂ , NO _x	Thermo Scientific	42C	CL (O ₃) ^c	FRM	K1, K2, T, M1
NO, NO ₂ , NO _x	Teledyne API	T200/T201 ^e	CL (O ₃)	FRM	M1, M2
THC	Thermo Scientific	51i	FID ^d	NA	K2, T, M1, M2

252 ^aNon-Dispersive Infrared Absorption

253 ^bCavity Attenuated Phase Shift

254 ^cOzone Chemiluminescence

255 ^dFlame Ionization Detector

256 ^eThe Teledyne API Model T201 is not a designated FRM or FEM however it employs the same operating principle as the FRM
257 designated model T200

258 ^fK1-Konza Prairie March 2017; S-Sycan Marsh October 2017; K2-Konza Prairie November 2017; T-Tallgrass Prairie
259 November 2017; M1-Missoula chamber April 2018; M2-Missoula chamber April 2019.

260

261 All instruments were calibrated with multipoint calibrations before and after each sampling day. All pre-
262 and post-calibrations met our quality performance objectives of +/- 10% and linearity of $r^2 \geq 0.99$. For
263 the O₃ analyzers under investigation, field and laboratory calibrations were performed using a Teledyne
264 API Model T700U Dynamic Dilution Calibrator with a NIST traceable O₃ photometer and O₃ generation
265 system. Zero air for the calibrator was supplied by a Teledyne API Model T701H Zero Air Generator.
266 Calibrations for NO, NO₂, NO_x and CO were performed using the same calibrator and zero air generator



267 utilizing a certified EPA protocol tri-blend (CO, NO, SO₂) gas cylinder (Airgas). Calibrations for THC
268 were performed using the T700U calibrator and a certified EPA methane/propane gas cylinder (Airgas).
269 In addition, for THC calibrations, the T701H zero air generator was replaced with scientific grade zero
270 air compressed gas cylinders (Airgas).

271

272 **2.3 Kansas Prescribed Burns, March 2017**

273 Biomass burning plumes were sampled during four days of prescribed burns (March 15-17, 2017 and
274 March 20, 2017) on the Konza Prairie Long Term Ecological Research (LTER) site outside of Manhattan,
275 Kansas. The fuels for this series of burns consisted of mixed native prairie tall grass of varying moisture
276 content. Over the four-day period, a total of 13 burns were conducted and sampled.

277

278 **2.4 Oregon Prescribed Burns, October 2017**

279 A single 10-hour day of prescribed grassland burning was measured at the Sycan Marsh Preserve in
280 central Oregon on October 11, 2017. Fuels for the Sycan Marsh burn consisted of mixed native prairie
281 tall grass of varying moisture content.

282

283 **2.5 Kansas Prescribed Burns, November 2017**

284 Biomass burning plumes were sampled during a single day of prescribed burning (November 10, 2017)
285 on the Konza Prairie LTER site outside of Manhattan, Kansas and on two additional days (November 13,
286 2017 and November 15, 2017) at the Tall Grass Prairie National Preserve outside Strong City, Kansas.
287 Fuels for the November 2017 burns consisted of mixed native prairie tall grass of varying moisture
288 content. During the November 10 sampling, two separate burns were conducted. Three burns were
289 conducted over the two days at Tallgrass Prairie.

290



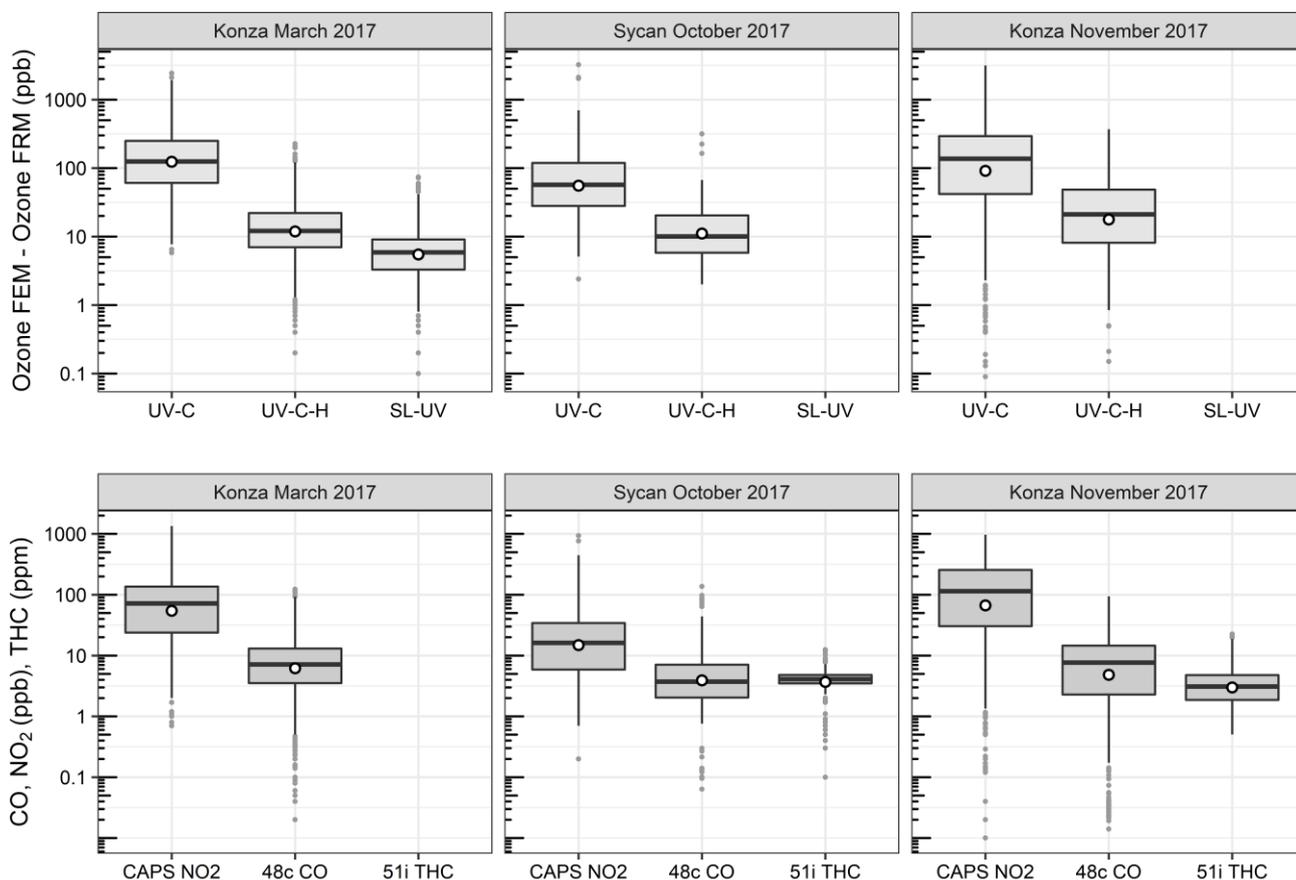
291 **2.6 USFS Missoula Burn Chamber Burns 2018, 2019**

292 Laboratory-based studies were performed at the U.S. Forest Service's combustion testing facility at the
293 FSL in Missoula, Montana by EPA and USFS personnel. These static chamber burns occurred in the
294 spring of 2018 (April 16-24, 2018; 33 burns; Landis et al., 2020) and again in the spring of 2019 (April
295 15-26, 2019; 31 burns). The main combustion chamber is a square room with internal dimensions 12.4 x
296 12.4 x 19.6 m high and a total volume of 3000 m³ and has been described previously (Bertschi et al.,
297 2003; Christian et al., 2004; Yokelson et al., 1996; Landis et al., 2020). During the combustion chamber
298 studies, the facility was fitted with identical instrumentation racks, calibration systems, systems for
299 sampling of test atmosphere, and data acquisition systems, as those described in Section 2.2. All
300 instrumentation were housed in an observation room immediately adjacent to the combustion chamber
301 with PFA inlet lines extending through the wall into the chamber. All inlet lines contained an identical
302 filter pack/filter assembly described in Section 2.2 to protect inlet lines and the analyzers from particulate
303 contamination. Fuel beds consisting of ponderosa pine needles and mixed woody debris were prepared
304 and placed in the middle of chamber. The amount and moisture content of the fuels were varied to generate
305 different flaming/smoldering conditions during the burns. During the chamber burns the combustion room
306 was sealed and the fuel bed was ignited. Two large circulations fans on the chamber walls and one on the
307 ceiling facilitated mixing and assured homogeneous conditions during the burn periods (Landis et al.,
308 2020).

309 **3 Results and Discussion**

310 **3.1 Results from Ozone Measurements in Prescribed Grassland Fire Plumes**

311 O₃ measurement results from the Oregon and Kansas prescribed grassland fires studies are shown as the
312 difference between the FEM and FRM in Fig. 1 and 1-minute average time series plots for the studies are
313 presented in Supplementary Figs. S1-S3. There were significant differences in the measurement results
314 obtained from the different O₃ monitors operated during the prescribed fires. The UV-C instrument
315 (Thermo 49i) consistently showed large increases in O₃ concentration readings in fresh biomass burning
316 plumes, with measurements exceeding the FRM measurement by 2-3 ppm. The O₃ exceedances were



317

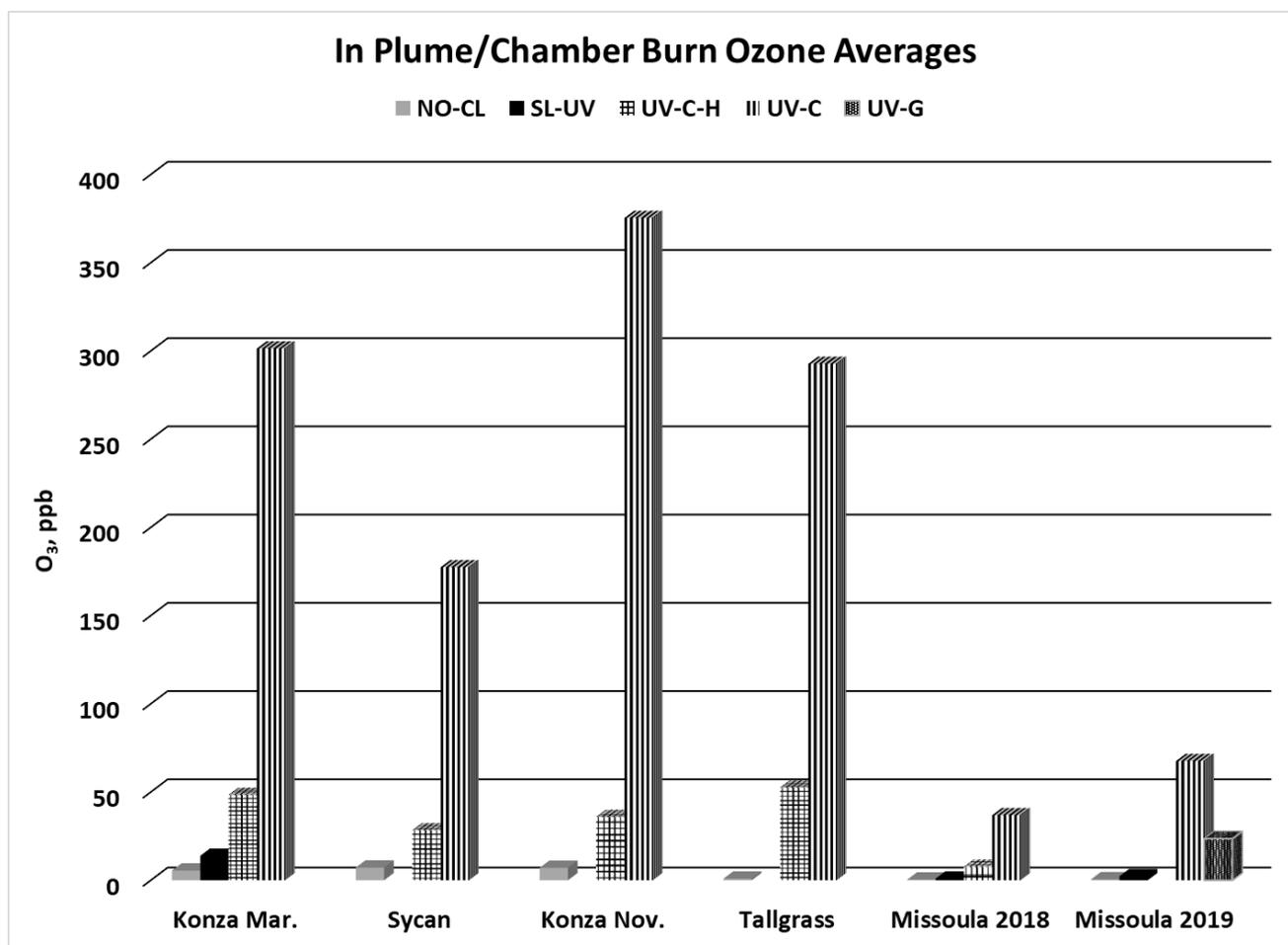
318 **Figure 1.** Ozone concentration differences between FEM instruments and the FRM instrument (FEM-
319 FRM), and the measured NO₂, CO, and total hydrocarbons (THCs) during the three 2017 wildfire
320 deployments. All measurements included are within-smoke only measurements, and show both the
321 elevated smoke tracers (NO₂, CO, THC), and the persistent elevation of the FEM O₃ measurements. The
322 box and whisker plots indicate the 25th, 50th, and 75th quartiles, with the whiskers extending to 1.5 times
323 the inner quartile range. The open dots indicate the mean values for each instrument within smoke.

324

325 generally correlated in time with CO and THC (biomass burning indicators) and NO₂. These correlations
326 will be discussed separately. The UV-C-H instrument (2B 205) also showed increased readings in smoke
327 plumes (also correlated with CO, THC, and NO₂), but with absolute measurements roughly an order of
328 magnitude smaller than the UV-C instruments. The NO-CL (T265) instrument results showed the
329 opposite behavior, with reductions in O₃ readings inversely correlated with increases in NO₂



330 concentrations, as expected from general O₃ titration by NO (NO + O₃ → NO₂ + O₂). For the March 2017
331 measurements the SL-UV instrument (2B 211) produced readings roughly comparable with the NO-CL
332 monitor, but with substantially more noise on a minute-to-minute timescale. The “in plume” average O₃
333 concentrations from the four prescribed grassland burning periods are shown in Fig. 2. For the purposes
334 of this comparison, CO measurements were used as an indicator of when sampling occurred “in plume.”



335

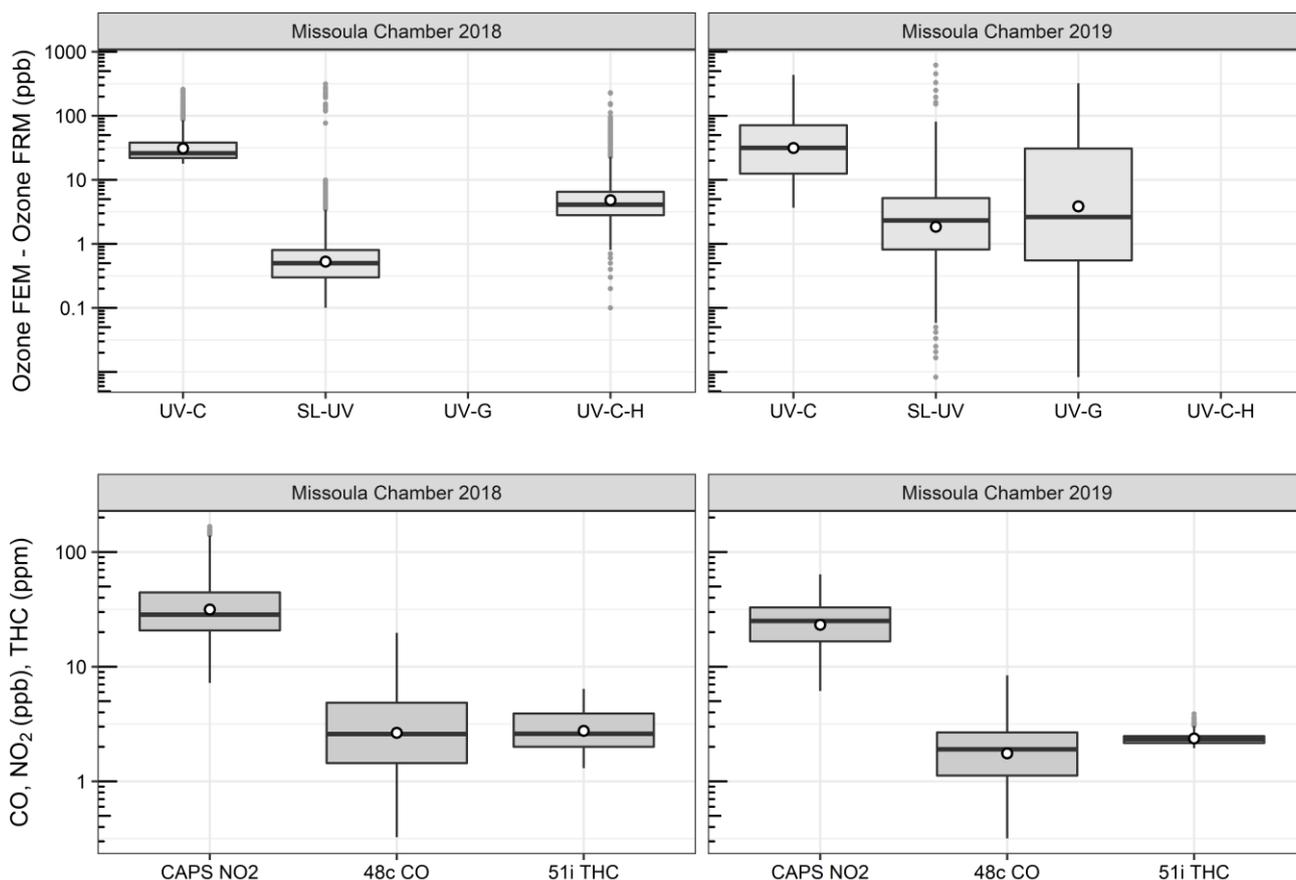
336 **Figure 2.** In plume O₃ concentration averages from the 2017 prescribed grassland burns and the 2018 and
337 2019 Missoula chamber burns. For the 2017 grassland burning periods, CO concentration results (≥ 1
338 ppm) were used as an indicator of when “in-smoke” sampling was occurring.

339



340 3.2 Results from Ozone Measurements in USFS Chamber Burns

341 O₃ measurement results from the 2018 and 2019 USFS chamber burn studies are shown in Fig. 3. Time

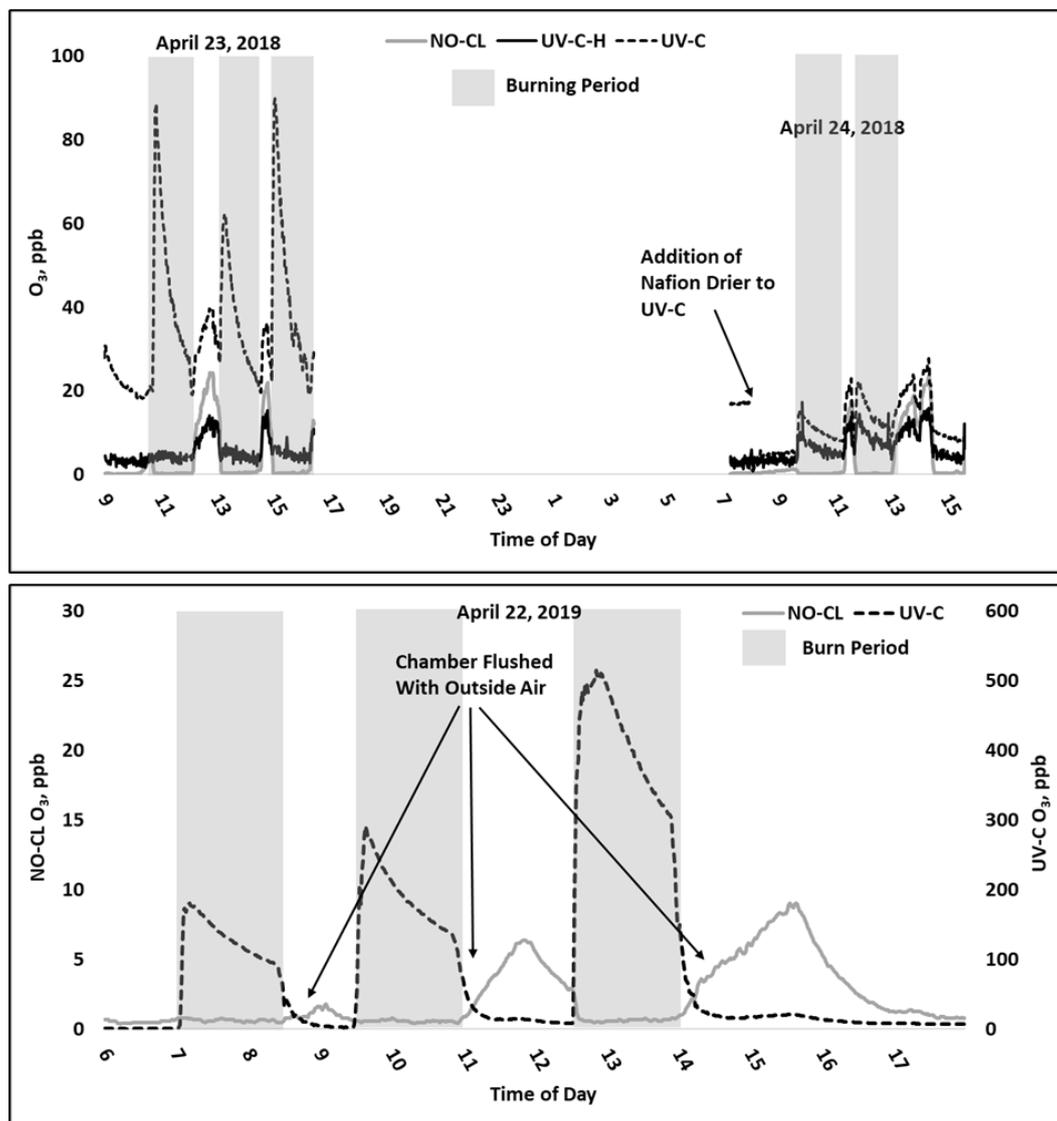


342

343 **Figure 3.** Differences between the FEM and FRM instrument concentrations (FEM-FRM), and NO₂,
344 CO, and total hydrocarbons (THCs) concentrations during the 2018 and 2019 Missoula chamber studies.
345 All measurements included are within-smoke only measurements, and show both the elevated smoke
346 tracers (NO₂, CO, THC), and the persistent elevation of the FEM O₃ measurements compared to the FRM.
347 The box and whisker plots indicate the 25th, 50th, and 75th quartiles, with the whiskers extending to 1.5
348 times the inner quartile range. The open dots indicate the mean values for each instrument within smoke.

349

350 series plots of the chamber Study data are included in Supplementary Figs. S4 and S5. Figure 4 gives a
351 more detailed view of UV-C and NO-CL O₃ results (two days from 2018 and one day from 2019) during



352

353 **Figure 4.** Time series example of USFS chamber burn O₃ results from the NO-CL, UV-C, and UV-C-H
354 (2018 only) from April 23-24, 2018 (top) and April 22, 2019 (bottom).

355

356 the chamber burns. In contrast to the prescribed grassland burns, the Missoula chamber burns employed
357 differing fuel types (ponderosa pine needles and fine woody debris) that are more typical of fuels
358 consumed during western U.S. forest fires. In addition, the concentrations of pollutants generated and
359 observed during the chamber studies were approximately an order of magnitude smaller than those



360 observed during the prescribed grassland fires. For reference, maximum $\text{PM}_{2.5}$ concentrations observed
361 during the prescribed fires were in excess of 50 mg m^{-3} while maximum chamber $\text{PM}_{2.5}$ concentrations
362 were less than 2 mg m^{-3} range. Regardless of these differences, there were still significant (order of
363 magnitude or more) differences in the measurement results between the different FEM O_3 instruments
364 operated during both the 2018 and 2019 chamber studies. The NO-CL method showed identical trends to
365 those observed during the grassland burns in that its measured O_3 concentrations dropped to near zero
366 during the active burning periods as indicated in Fig. 4 (active burning periods shaded in grey). The only
367 periods when significant O_3 concentrations were measured by the NO-CL method was when outside air
368 was brought in to flush the chamber in between burns. As in the grassland fire plumes, the UV-C method
369 showed increased O_3 concentration (positive analytical artifact) readings that were correlated in time with
370 CO and NO_2 ; See Supplementary Figs. S9 and S10. Similarly, the UV-C-H instrument also showed
371 increased positive analytical artifacts during the chamber burns, but with absolute measurement values
372 about an order of magnitude smaller than the UV-C instruments. The SL-UV method gave similar results
373 to the NO-CL method during both the 2018 and 2019 chamber burns. Newly added during the 2019 burns,
374 the UV-G method (2B 211-G) gave mixed results: at times it provided similar results compared to the
375 NO-CL and SL-UV methods, and at others it provided results in line with those provided by the UV-C
376 method. See Supplementary Fig. S5 for the 2019 chamber burn time series plot. The burn average O_3
377 concentrations from the 2018 and 2019 chamber burns are presented in Fig. 2.

378 During the 2018 chamber burns the UV-C results were biased high by 15-20 ppb even during non-burn
379 (i.e., overnight) periods as evident in Fig. 4 (top panel) and Fig. S4. The initial hypothesis was that the
380 bias was associated with high chamber backgrounds of interfering species due to years of heavy burning
381 in the chamber. However, it was later discovered during a subsequent summer/fall 2018 ambient air study
382 in North Carolina in the absence of smoke, that sampling heavy smoke plumes during the spring and fall
383 2017 prescribed grassland burns irreversibly damaged the MnO_2 scrubber in the UV-C instrument. The
384 effect of the bias was observed mainly when sampling ambient air and not readily observed during routine
385 calibration checks (zeroes and spans) except for an increase in the time required to obtain stable zero and
386 span values. During the summer/fall 2018 North Carolina study and prior to the start of the 2019 chamber



387 burns, a new MnO₂ scrubber was installed and resulted in a significant and immediate reduction of the
388 observed high bias, shown in Fig. 4 (bottom panel) and Fig. S5.

389 **3.3 Methodological Influence on Ozone Measurements in Biomass Burning Smoke**

390 As discussed in Sections 3.1 and 3.2, there are large (order of magnitude level) differences in O₃
391 concentration measurement results obtained from the FRM (NO-CL) and the FEM UV photometric with
392 catalytic scrubber (UV-C) O₃ methods. The extremely low O₃ concentrations measured by the NO-CL
393 instrument is consistent with O₃ depletion in the presence of high NO_x concentrations (up to ppm levels)
394 observed in the grass burning plumes and during chamber burns. The reaction between NO and O₃ is
395 rapid and occurs on the timescales of seconds to minutes. As a result, high NO in the fresh biomass
396 combustion plumes will efficiently titrate out O₃ leading to near-field depletion within the plumes relative
397 to background concentrations. There was no sign of a positive interference in the NO-CL monitors, and
398 it remains the most robust and accurate routine method for O₃ measurement in fresh and downwind
399 biomass burning plumes.

400

401 In contrast with the NO-CL FRM instrument results, the UV-C FEM results showed substantial increases
402 in reported O₃ concentrations in the fresh biomass burning plumes. There is no known pathway for direct
403 O₃ emission from biomass burning, and the proximity (meters to hundreds of meters) and timescales
404 (seconds to minutes travel time from the combustion source to measurement) involved are too short for
405 the usual NO_x – VOC photochemistry to produce secondary O₃. Further, since the FSL chamber interior
406 is not exposed to sunlight, photochemistry would not have been active in the Missoula laboratory burns.
407 For the purposes of this work, the positive analytical artifact in the UV-C method, termed $\Delta O_{3(\text{UV-C})}$, is
408 estimated using Eq. (6) as the difference between UV-C and the NO-CL O₃ concentration measurement
409 results for the same time period:

410

411

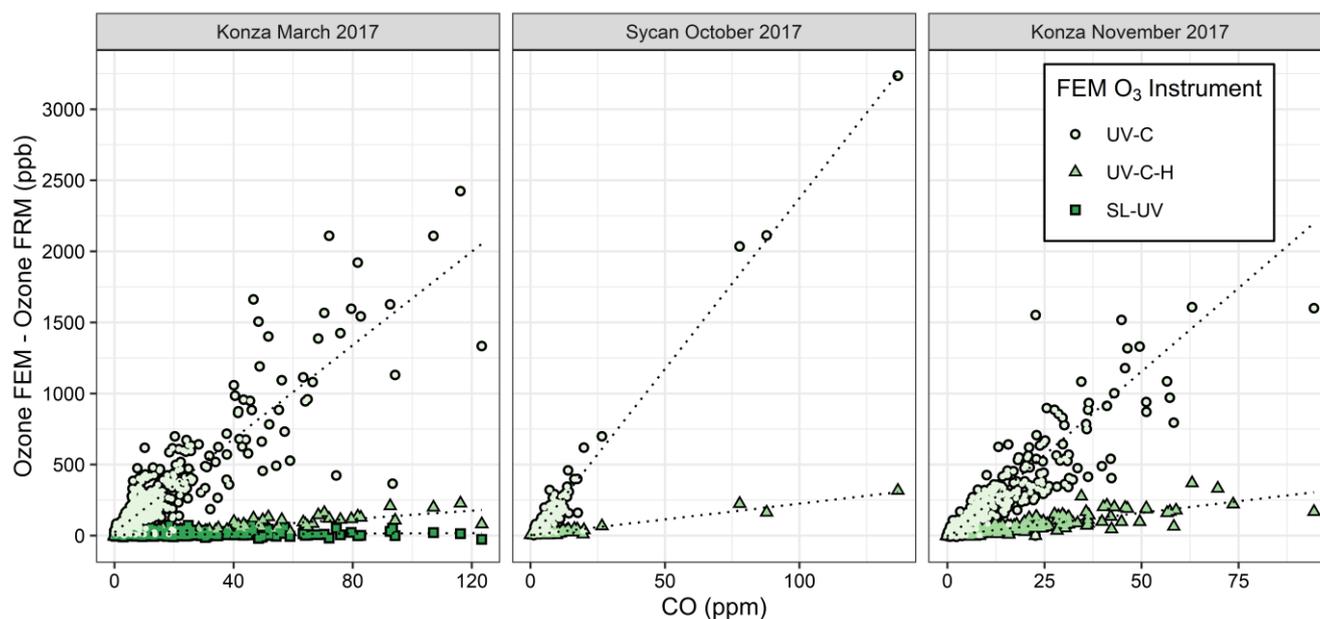
412

$$\Delta O_{3(\text{UV-C})} = \text{UV-C} - \text{NO-CL} \quad (6)$$



413 Figure 5 shows “in plume” regressions between $\Delta O_{3(\text{UV-C})}$ and the FRM measurement and CO for the
414 three measured prescribed grassland burns in 2017 (Supplementary Fig. S6 shows the time series of
415 $\Delta O_{3(\text{UV-C})}$ and CO). Figure 5 and Supplementary Fig. S6 show good correlations within the smoke plumes.
416 The average and maximum values of $\Delta O_{3(\text{UV-C})}$ are summarized in Table 3. It is hypothesized that the
417 large “O₃” measurement observed in the UV-C method results from a positive interference or artifact,
418 likely linked to VOC emissions in the grassland burn plumes. VOCs are emitted in higher concentrations
419 from the smoldering phase of combustion, which is also characterized by large CO emissions (Yokelson
420 et al., 1996; Yokelson et al., 1997), so a correlation between CO and O₃ artifact would support the
421 hypothesis of a VOC-linked interference for the UV-C instruments. This is also consistent with observed
422 VOC interferences in previous studies (Grosjean and Harrison, 1985; Kleindienst et al., 1993; Spicer et
423 al., 2010) and observed following fireworks (Fiedrich et al., 2017; Xu et al., 2018).

424



425

426 **Figure 5.** Scatter plots between FEM and FRM O₃ differences and the CO measurements within the
427 grassland fires smoke plumes. The FEM measurements are differentiated by color and shape. The SL-UV
428 method was only run during the Konza March 2017 measurements.

429



430 **Table 3: Ozone artifact (ΔO_3) averages, maximum values, and CO, NO₂, and THC averages from**
 431 **the prescribed fire and USFS chamber burns as measured by the UV-C, UV-C-H, and UV-G**
 432 **instruments.**

Study	ΔO_3 avg. (ppb)	ΔO_3 max (ppb)	CO avg. (ppm)	NO ₂ avg. (ppb)	THC avg. (ppm)
ΔO_3(UV-C)					
Mar. 2017 Konza Prairie (KS)	295.8	2423.7	15.8	147.3	-
Oct. 2017 Sycan Marsh (OR)	170.2	3235.5	8.5	60.5	2.7
Nov. 2017 Konza & Tallgrass Prairies (KS)	330.0	3156	14.1	156.9	4.0
Apr. 2018 USFS Chamber (MT)	36.5	309.6	3.8	35.6	2.8
Apr. 2019 USFS Chamber (MT)	66.9	530.9	2.1	31.7	4.8
ΔO_3(UV-C-H)					
Mar. 2017 Konza Prairie (KS)	42.8	227.1	15.8	147.3	-
Oct. 2017 Sycan Marsh (OR)	21.1	316.4	8.5	60.5	2.7
Nov. 2017 Konza & Tallgrass Prairies (KS)	40.2	369.0	14.1	156.9	4.0
Apr. 2018 USFS Chamber (MT)	7.2	136.8	3.8	35.6	2.8
ΔO_3(UV-G)					
Apr. 2019 USFS Chamber (MT)	22.9	376.8	2.1	31.7	4.8
ΔO_3(SL-UV)					
Mar. 2017 Konza Prairie (KS)	8.3	74.2	15.8	147.3	-
Apr. 2018 USFS Chamber (MT)	0.5	11.5	3.8	35.6	2.8
Apr. 2019 USFS Chamber (MT)	1.7	32.1	2.1	31.7	4.8

433
 434 The presence of a Nafion[®]-based humidity conditioning system (Nafion[®] tube dryer) significantly
 435 reduced the magnitude of the observed artifact as evident by comparing the UV-C and UV-C-H results
 436 shown in Figs. 1-3 and Supplementary Figs. S1 – S5. As with the UV-C method, the artifact in the UV-
 437 C-H method, $\Delta O_{3(\text{UV-C-H})}$, is calculated using Eq. (7) as the difference between UV-C-H and the NO-CL
 438 O₃ concentration measurement results for the same time period:

$$\Delta O_{3(\text{UV-C-H})} = \text{UV-C-H} - \text{NO-CL} \quad (7)$$

439
 440
 441
 442 The addition of the Nafion[®]-based humidity conditioning system reduces the magnitude of the $\Delta O_{3(\text{UV-C-}}$
 443 $\text{H})$ artifact by approximately an order of magnitude as compared with the UV-C method. This is further
 444 illustrated in the 2018 chamber burns, where prior to beginning the final burn day on April 24, 2018, a



445 Nafion[®] tube dryer (PermaPure, MD Monotube Dryer Series) was installed on the UV-C method (Thermo
446 49i) in effect, converting it to a UV-C-H method. As shown in Fig. 4 and Supplementary Fig. S4, the
447 addition of the Nafion[®] tube dryer significantly reduced the $\Delta O_{3(UV-C-H)}$ artifact to a point comparable with
448 that observed in the UV-C-H method (2B 205). A possible explanation for this effect is presented and
449 discussed in Section 3.5. In addition, the previously described bias related to the damaged MnO₂ scrubber
450 was also reduced upon addition of the Nafion[®] dryer to the UV-C method.

451

452 For the March 2017 Konza Prairie study (Fig. 1) and the 2018 and 2019 USFS chamber studies (Fig. 3)
453 the SL-UV instrument concentration results were comparable to, although noisier and slightly higher than,
454 the NO-CL reference instrument. On numerous occasions during the prescribed and chamber burns, the
455 SL-UV instrument shows short (i.e. one-minute data point) positive or negative excursions that are not
456 also observed in the NO-CL results. In addition, these excursions are not correlated with changes in CO
457 concentrations. Because the SL-UV is a dual cell instrument that measures O₃ by comparing the
458 absorbance of two cells, it is critical in highly dynamic environments (such as during this study) that both
459 cells be measuring the same air at the same time. A slight difference in flow rates or residence times
460 between the two pathways (or a delay in one pathway relative to the other) will cause short term variability
461 in the difference between the two cells. Although this does not pose an issue for longer time averaging
462 (i.e. hourly data) under stable conditions, the dynamic nature of biomass burning plumes (i.e. changing
463 on the order of seconds) and short time averages (i.e. minute) can create issues (noise) for the SL-UV
464 method.

465

466 Significant analytical artifacts were observed for FEM UV photometric O₃ instruments with (UV-C-H)
467 and without (UV-C) Nafion[®]-based humidity conditioning system, where it appears that the dual effect
468 of ambient humidity fluctuations and VOC interferences caused large positive over-measurement of “in-
469 smoke” O₃ concentrations. Chemiluminescence monitors are highly specific to O₃ and have long been
470 known to be free of VOC interferences (Long et al., 2014; U.S. EPA, 2015). However, studies have shown
471 that the chemiluminescence method can be impacted by changes in relative humidity (Kleindienst et al.,
472 1993). As such, upon promulgation in 2015, the new NO-CL FRM regulatory text requires a humidity



473 correction/dryer system to eliminate the potential water vapor interference. As configured from the
474 manufacturer, the NO-CL based Teledyne-API Model T265 instrument operated during this comparative
475 study employs Nafion[®] drying technologies to reduce or eliminate the water vapor interferences. The use
476 of a chemical (NO) scrubber for UV photometric instruments (such as the 2B Technologies Model 211)
477 is very specific to O₃ and shows a much better response than the catalytic scrubber instruments,
478 performing almost as well as the NO-CL FRM, and has significant potential as a low-interference O₃
479 method. Of the catalytic scrubber photometric instruments those with Nafion[®]-based humidity
480 equilibration (2B Technologies Model 205) perform significantly better than those without (Thermo 49
481 series).

482

483 In areas highly impacted by smoke or for studies focusing on biomass burning plumes, the use of a NO-
484 CL FRM instrument is highly recommended as it was found to be essentially interference-free. These
485 instruments are anchored to absolute O₃ concentrations through the use of certified O₃ calibration sources,
486 many of which are based on UV photometry. The newest generation of commercially-available NO-CL
487 FRM instruments, including that used here (the Teledyne T265), have a built-in drying system to correct
488 for the humidity artifacts that affected earlier generation chemiluminescence instruments (Kleindienst et
489 al., 1993), making remaining interferences negligible compared to other technologies.

490

491 The gas-phase chemical scrubber UV instrument (2B 211), did not perform as well as the FRM under the
492 prescribed grassland burns or chamber experimental conditions tested here, with the high time resolution
493 (1-minute) data showing a much higher degree of variability than the NO-CL FRM instrument. We
494 hypothesize that the main factor driving this divergence between this method and the NO-CL FRM is the
495 dual-cell differential configuration of the instrument, which is not conducive to rapidly changing
496 concentrations in O₃ or other absorbing gases, such as VOCs.

497

498 In smoke-impacted monitoring situations where the use of a UV photometric instrument is still preferred
499 or required, the choice of a monitor with humidity equilibration provides a significant analytical
500 improvement over those monitors without the humidity corrections. In the absence of an instrument with



501 a Nafion[®] tube dryer and in non-regulatory applications, a dryer can be installed before the inlet or
502 measurement cells to reduce the interference as was demonstrated on the final day of the 2018 Missoula
503 chamber burns. This will have the added benefit of reducing positive biases from humidity and reducing
504 equilibration time for calibrations (especially when switching from high humidity ambient air to dry
505 calibration gases).

506 **3.4 Magnitude of Ozone Artifact in Fresh Biomass Burning Plumes Relative to Markers of** 507 **Combustion**

508 It is difficult to estimate an absolute magnitude or correct for the observed O₃ analytical artifact since
509 primary emissions from biomass combustion are highly variable and depend upon the fuel loading, fuel
510 type and condition, phase of the fire, and the burn conditions (Yokelson et al., 1996; Yokelson et al.,
511 1997). However, assuming the interference is driven primarily by VOCs, the artifact should be correlated
512 with the excess CO ($\Delta\text{CO} = \text{CO}_{\text{plume}} - \text{CO}_{\text{background}}$). Because CO_{background} during the prescribed grassland
513 burns was below 200 ppb (relative to typical conditions of >2 ppm in the plume), ΔCO is estimated as the
514 total measured CO concentration. A simplified view of biomass combustion assumes an approximate
515 linear combination of two dominant emission phases, flaming combustion (characterized by emission of
516 highly oxidized compounds, such as CO₂, NO_x, and SO₂), and smoldering combustion (characterized by
517 emission of reduced or mixed oxidation state compounds, such as CO, CH₄, NH₃, H₂S, and most VOCs)
518 (Yokelson et al., 1996; Yokelson et al., 1997). Because the majority of VOCs are in a reduced or mixed
519 oxidation state, they tend to be co-emitting with CO during smoldering combustion, and the VOC
520 concentrations tend to be highly correlated with CO in fresh biomass burning plumes (Yokelson et al.,
521 1996). Scatterplots comparing the FEM instrument artifacts ($\Delta\text{O}_{3(\text{UV-C})}$) and CO for the three prescribed
522 grassland burning periods are shown in Fig. 5. Regression statistics of the comparison of $\Delta\text{O}_{3(\text{UV-C})}$ and
523 $\Delta\text{O}_{3(\text{UV-C-H})}$ with CO and THC for grassland burns are given in Table 4. The magnitude of the artifact
524 (estimated by the slope of the regression line of the CO vs ΔO_3 comparison), in ppb apparent O₃ per ppm
525 CO, ranges between 16 - 24 ppb ppm⁻¹ for the UV-C instrument, and 1-3.5 ppb ppm⁻¹ for the instrument
526 with humidity correction (UV-C-H). It is important to point out that CO, in and of itself, is not considered
527 to be an interfering species in the UV photometric determination of O₃ in that CO absorbs in the infrared



528 (IR). Similar to CO, THC_s and NO₂ are indicative of combustion processes and are correlated with ΔO₃
 529 as given in Table 4 and Supplementary Figs. S7 and S8. In terms of THC, the magnitude of the artifact,
 530 in ppb apparent O₃ per ppm THC, is significantly higher at ~88 ppb ppm⁻¹ for the UV-C instrument and
 531 ~13 ppb ppm⁻¹ for the UV-C-H instrument. Both the prescribed grassland and Missoula chamber burns
 532 resulted in what would be considered high PM concentrations (2-50 mg m⁻³). These high PM
 533 concentrations however, are not considered to be interfering due to the presence of the inline particle
 534 filter assemblies described in Sections 2.2 and 2.6.

535

536 **Table 4: Regression statistics for the ozone artifact (ΔO₃) versus CO and THC for UV photometric**
 537 **instruments without (UV-C) and with (UV-C-H) a Nafion®-based humidity equilibration system**
 538 **during the 2017 prescribed grassland burns.**

Study	Slope (ppb/ppm)	Intercept (ppb)	r ²	n
ΔO₃(UV-C) vs CO				
Mar. 2017 Konza Prairie (KS)	16.46	22.9	0.79	679
Oct. 2017 Sycan Marsh (OR)	24.02	-28.05	0.96	295
Nov. 2017 Konza & Tallgrass Prairies (KS)	23.51	-20.8	0.74	461
ΔO₃(UV-C) vs THC				
Nov. 2017 Konza & Tallgrass Prairies (KS)	87.14	-85.36	0.59	461
ΔO₃(UV-C-H) vs CO				
Mar. 2017 Konza Prairie (KS)	1.46	0.07	0.80	163
Oct. 2017 Sycan Marsh (OR)	2.21	3.44	0.88	296
Nov. 2017 Konza & Tallgrass Prairies (KS)	3.24	-1.17	0.77	461
ΔO₃(UV-C-H) vs THC				
Nov. 2017 Konza & Tallgrass Prairies (KS)	13.27	-14.53	0.75	461
CO vs THC				
Nov. 2017 Konza & Tallgrass Prairies (KS)	0.21	1.55	0.79	461

539

540 Since the CO concentrations (from upwind fires) observed at most stationary sites from fire plumes are
 541 usually on the order of one ppm to greater than 10 ppm (Landis et al., 2018), it is reasonable to assume
 542 that O₃ artifacts in the range of 15 ppb to greater than 250 ppb can be observed when employing a UV-C
 543 method. Similarly, O₃ artifacts in the range of 1.5 to above 30 ppb might be observed at smoke-impacted
 544 sites monitoring with UV-C-H methods. As such, Nafion®-based humidity conditioning systems are



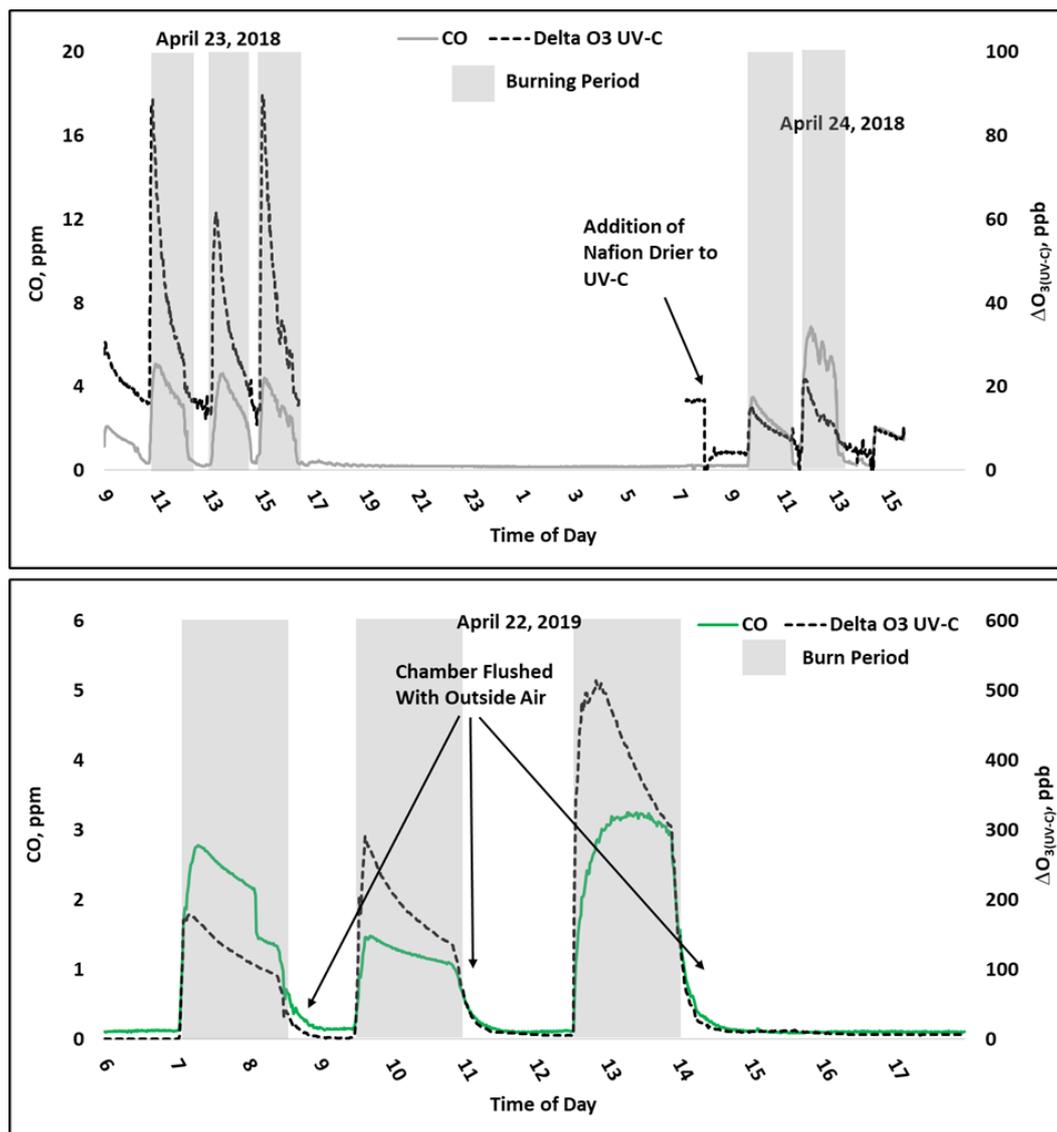
545 highly recommended for use if employing UV photometric methodology for monitoring O₃ in areas
546 impacted by wildfires or prescribed burns. As stated previously and as seen in Fig. 3 and Table 3, O₃
547 artifacts were observed during the Missoula chamber 2018 and 2019 burns in both the UV-C and UV-C-
548 H methods, although reduced compared to the prescribed grassland burns. The presence and magnitude
549 of the O₃ artifact strongly suggests that smoke generated from fuels typical of forests in the western United
550 States also result in a measurement interference in UV photometric methods. Since downwind O₃
551 production in biomass burning plumes is a significant issue in fire impacted regions, having reliable,
552 interference-free methods is critical for assessing the contribution of wildland fires to ambient O₃ levels.
553 Figure 6 gives a detailed time series view of $\Delta\text{O}_{3(\text{UV-C})}$ and CO from two burn days from 2018 and a single
554 day during 2019. As indicated, $\Delta\text{O}_{3(\text{UV-C})}$ and CO appear to be correlated in time but when performing
555 linear regression comparisons of $\Delta\text{O}_{3(\text{UV-C})}$ and CO during each years chamber burns as a whole,
556 correlations tend to be poor. We suspect the positive O₃ bias is driven by one or more VOCs (likely
557 oxygenated VOCs). In fresh smoke the excess concentrations of individual VOCs (ΔX), and VOC sums
558 (ΔVOC), tend to be highly correlated with ΔCO (Yokelson et al., 1999; Gilman et al. 2015). The emission
559 ratios of individual VOCs to CO ($\Delta\text{X}/\Delta\text{CO}$) can vary considerably with combustion conditions such as
560 fuel type and condition (e.g. moisture content and decay state), fuel bed properties, such as bulk density,
561 and the relative mix of flaming and smoldering combustion (Gilman et al. 2015; Koss et al., 2017).
562 Additionally, the response of $\Delta\text{X}/\Delta\text{CO}$ to burn conditions varies among VOCs. When each burn is
563 considered individually or in groups with similar conditions, the correlations between ΔO_3 , CO, and THC
564 are enhanced. An example of this behavior is shown in Supplementary Fig. S10. The lack of a consistent
565 relationship between the O₃ artifact and ΔCO across all FSL chamber burns, while observing a good
566 correlation for individual burns, likely reflects the variable response of artifact producing emission(s) to
567 the different combustion conditions of the burns.

568

569 One interesting observation from the data obtained from both the prescribed grassland and chamber burns
570 is the order of magnitude difference in the average and maximum O₃ artifact between the UV-C and the
571 UV-C-H instruments as shown in Table 3. The size of the difference (as large as hundreds of ppb) cannot
572 be explained purely by the previously observed relative humidity effects on measurements (Leston et al.,



573



574

575 **Figure 6.** Time series example of USFS chamber burn $\Delta O_3(UV-C)$ and CO concentration results from
576 April 23-24, 2018 (top) and April 22, 2019 (bottom).

577

578 2005; Wilson et al., 2006), suggesting that the Nafion[®] dryer is directly impacting the concentrations of
579 other interferences in the sample stream.

580



581 **3.5 Potential Reason for Lower Artifacts with Methods Employing Nafion[®]-based Humidity** 582 **Equilibration**

583 Nafion[®] is a sulfonated tetrafluoroethylene polymer that is highly permeable to water but shows little
584 permeability to many other organic and inorganic species (Mauritz et al., 2004). As a result, Nafion[®]-
585 based drying systems are often used as part of sample preparation or conditioning systems in analytical
586 chemistry to remove water vapor from sample streams prior to sample analysis. Nafion[®] membranes were
587 introduced to some O₃ monitors as a method to address humidity effects observed in UV-C O₃ monitors,
588 particularly when there are rapid changes in relative humidity level (Wilson and Birks, 2006). Humidity
589 can affect the transmission of the UV light through the detection cell and catalytic O₃ scrubbers can
590 modulate the water vapor in the scrubbed channel by acting as a temporary reservoir, resulting in
591 significant positive or negative O₃ interferences during rapid swings in relative humidity Wilson et al.,
592 2006). Adding a Nafion[®]-based equilibration dryer immediately prior to the measurement cells reduces
593 this water vapor interference without affecting O₃ concentrations, and thus significantly reduces the
594 humidity artifacts in UV photometric O₃ instruments.

595

596 Despite the high selectivity of Nafion[®] to water vapor, it does demonstrate partial to complete
597 permeability to various VOC or semivolatile organic compounds. Nafion[®] membranes are highly
598 permeable to alcohols, amines, ketones, and some water-soluble ethers (Baker, 1974), as well as some
599 biogenic oxygenated compounds (Burns et al., 1983). In addition, Nafion[®] membranes have been shown
600 to catalyze the decomposition and rearrangement of monoterpene compounds (Burns et al., 1983).
601 Systematic study of Nafion[®] permeability and reactivity for polar and oxygenated compounds has been
602 limited, with most users of Nafion[®] membranes basing their use on operational testing and confirmation
603 for the targeted use.

604

605 The significant (order of magnitude) reduction in the O₃ artifact with the addition of a Nafion[®]-based
606 dryer to the UV-C suggests that the Nafion[®] dryer is directly impacting the major interfering species
607 which was hypothesized to be VOCs emitted during combustion processes. The species that are
608 responsible for most of the O₃ artifact in UV-C O₃ instruments would have to be permeable through



609 Nafion[®] membranes or reactive with Nafion[®] membranes, be scrubbed by solid-phase, catalytic O₃
610 scrubbers (such as MnO₂ or hopcalite), and would have a significant absorption cross section around 254
611 nm. The absorption cross-section of O₃ around 254 nm is on the order of 10⁻¹⁷ cm² molecule⁻¹ (Molina
612 and Molina, 1986), which means species with absorptions around 10⁻¹⁷ cm² molecule⁻¹ at 254 nm would
613 be potential interfering species. As a class, aromatic VOCs and specifically oxygenated aromatic species
614 (and other polar derivatized species) absorb strongly in this region of the UV spectrum, and their potential
615 permeability through Nafion[®] membranes result in them being likely compounds to interfere in UV-C
616 instruments. As an example, aromatic aldehydes such as o-tolualdehyde and p-tolualdehyde absorbances
617 around 5x10⁻¹⁸ cm² molecule⁻¹ and 4x10⁻¹⁸ cm² molecule⁻¹, respectively (Etzkorn et al., 1999). Both 2,4-
618 dimethylbenzaldehyde and 2,6-dimethylbenzaldehyde have absorption cross sections above 10⁻¹⁷ cm²
619 molecule⁻¹ at 254 nm (El Dib et al., 2008). Baker (1974) found 75% of benzaldehyde was removed by a
620 Nafion[®] membrane, meaning that the Nafion[®] permeability of tolualdehydes and dimethylbenzaldehydes
621 is also likely to be high. In addition, benzaldehyde was almost quantitatively removed by several
622 commercial catalytic O₃ scrubbers, including the Thermo 49i MnO₂ catalytic scrubber (Kleindienst et al.,
623 1993), so similar aldehydes are likely to behave in a similar manner. Therefore, substituted aromatic
624 aldehyde species are one class of compounds that fit the necessary criteria for causing the interference on
625 the UV-C while having a reduced interference on the UV-C-H instrument. Future work examining the
626 potential interferences from different species (or classes of species) on a species or class specific basis
627 are required to confirm this potential mechanism and suggest others.

628 **4 Conclusions**

629 In this study, we compare two different O₃ measurement methods (chemiluminescence and UV
630 photometry) in fresh biomass burning plumes from prescribed grassland fires and during controlled
631 chamber burns. Within the UV photometry category, we look at two different technologies, one using a
632 gas-phase chemical scrubber (NO) and the second using solid phase catalysts to scrub O₃ from analytical
633 reference channels. Among the UV photometric instruments employing solid phase catalytic scrubbers,
634 we evaluated and compared methods that include a Nafion[®]-based humidity equilibration system with
635 those that do not.



636

637 The NO-CL method, recently promulgated as the O₃ FRM, performed well, even in fresh plumes, whereas
638 the UV photometric methods displayed varying degrees of positive measurement artifacts. The UV
639 photometric method employing the dynamic NO gas phase scrubber performed comparably with the NO-
640 CL method but was not well suited to the rapidly varying concentrations of VOCs in the smoke plumes.
641 The catalytic scrubber photometric methods demonstrated positive analytical artifacts that were correlated
642 with CO and THC concentrations (both biomass burning plume indicators). There was a significant
643 difference between the catalytic scrubber UV instruments with and without Nafion[®]-based humidity
644 correction, with the dryer system reducing the positive O₃ artifact by an order of magnitude as compared
645 with the UV photometric method employing no humidity correction. The observed reduction in artifacts
646 cannot be attributed only to elimination of the relative humidity/water vapor interferences and likely result
647 from post-scrubber equilibration or reaction of Nafion[®]-permeable VOCs prior to the measurement cell.
648 The results of this study strongly suggest that careful consideration be given to employed measurement
649 methods when monitoring O₃ concentrations in regions where impacts from biomass burning routinely
650 occur due to the significant impact of potential measurement interferences. In addition to consideration
651 of operating methods containing Nafion[®]-based humidity condition systems, attention should be focused
652 on the scrubbers employed by UV photometric methods and the adverse effects that operation in smoke
653 may have on those scrubbers. Further research is being conducted to evaluate the magnitude of the artifact
654 in the UV photometric method at routine monitoring sites that are often impacted by wildland fire smoke
655 events under the EPA Mobile Ambient Smoke Investigation Capability (MASIC) program (U.S. EPA
656 2019).

657

658 **Data Availability**

659 Datasets related to this manuscript can be found at <https://catalog.data.gov/dataset/epa-sciencehub>.

660

661 **Author Contributions**

662 Russell W. Long served as principal investigator and prepared the manuscript with contributions from all
663 co-authors. Russell W. Long, Andrew Whitehill, Andrew Habel, Maribel Colón, Shawn Urbanski, and



664 Matthew S. Landis performed the prescribed grassland fire and FSL chamber-based data collection and/or
665 analysis. Surender Kaushik performed supervisory review of this research effort and corresponding
666 manuscript.

667

668 **Competing Interests**

669 The authors declare that they have no conflict of interest.

670

671 **Disclaimer**

672 The views expressed in this paper are those of the authors and do not necessarily reflect the views or
673 policies of EPA. It has been subjected to Agency review and approved for publication. Mention of trade
674 names or commercial products do not constitute an endorsement or recommendation for use.

675

676 **Acknowledgements**

677 The EPA through its Office of Research and Development (ORD) funded and conducted this research.
678 We thank Kansas State University, The Nature Conservancy, Konza Prairie Biological Station staff,
679 Sycan March Preserve staff, Tallgrass Prairie National Preserve staff, numerous burn crews, Brian Gullet
680 (EPA), Cortina Johnson (EPA), Melinda Beaver (EPA), Libby Nessley (EPA), and Kyle Digby (Jacobs).

681 **References**

682 Baker, B. B., Measuring trace impurities in air by infrared spectroscopy at 20 meters path and 10
683 atmospheres pressure, *Am. Ind. Hyg. Assoc. J.*, 35, 735-740, 1974.

684

685 Bertschi, I.; Yokelson, R.J.; Ward, D.E.; Babbitt, R.E.; Susott, R.A.; Goode, J.G.; Hao, W.M., Trace gas
686 and particle emissions from fires in large diameter and belowground biomass fuels, *J. Geophys. Res.*, 108
687 (D13):8472, 2003.

688

689 Christian, T. J., Kleiss, B., Yokelson, R. J., Holzinger, R., Crutzen, P. J., Hao, W. M., W.M., Saharjo,
690 B.H., Ward, D.E., Comprehensive laboratory measurements of biomass-burning emissions: 2. First



- 691 intercomparison of open-path FTIR, PTR-MS, and GC- MS/FID/ECD. *J. Geophys. Res-Atmos.*, 109(D2),
692 2004.
- 693
- 694 Boylan, P., Helmig, D., and Park, J.H., Characterization and mitigation of water vapor effects in the
695 measurement of ozone by chemiluminescence with nitric oxide, *Atmos. Meas. Tech.* 7, 1231-1244, 2014.
- 696
- 697 Burns, W. F., Tingey, D. T., Evans, R. C., and Bates, E. H., Problems with a Nafion® membrane dryer
698 for drying chromatographic samples, *J. Chromatogr. A*, 269, 1-9, 1983.
- 699
- 700 Dunlea, E., Herndon, S., Nelson, D., Volkamer, R., Lamb, B., Allwine, E., Grutter, M., Ramos Villegas,
701 C., Marquez, C., and Blanco, S., Evaluation of standard ultraviolet absorption ozone monitors in a
702 polluted urban environment, *Atmos. Chem. Phys.*, 6, 3163-3180, 2006.
- 703
- 704 El Dib, G., Chakir, A., and Mellouki, A., UV absorption cross-sections of a series of
705 dimethylbenzaldehydes, *J. Phys. Chem. A*, 112, 8731-8736, 2008.
- 706
- 707 Etzkorn, T., Klotz, B., Sørensen, S., Patroescu, I. V., Barnes, I., Becker, K. H., and Platt, U., Gas-phase
708 absorption cross sections of 24 monocyclic aromatic hydrocarbons in the UV and IR spectral ranges,
709 *Atmos. Environ.* 33, 525-540, 1999.
- 710
- 711 Fiedrich, M., Kurtenbach, R., Wiesen, P., and Kleffmann, J., Artificial O₃ formation during fireworks,
712 *Atmos. Environ.* 165, 57-61, 2017.
- 713
- 714 Gilman, J. B., Lerner, B. M., Kuster, W. C., Goldan, P. D., Warneke, C., Veres, P. R., et al., Biomass
715 burning emissions and potential air quality impacts of volatile organic compounds and other trace gases
716 from fuels common in the US. *Atmos. Chem. Phys.*, 15(24), 13915-13938, 2015.
- 717
- 718 Grosjean, D., and Harrison, J., Response of chemiluminescence NO_x analyzers and ultraviolet ozone
719 analyzers to organic air pollutants, *Environ. Sci. Tech.*, 19, 862-865, 1985.
- 720
- 721 Huntzicker, J. J., and Johnson, R. L., Investigation of an ambient interference in the measurement of
722 ozone by ultraviolet absorption photometry, *Environ. Sci. Tech.*, 13, 1414-1416, 1979.
- 723
- 724 Johnson, T., Capel, J., Ollison, W., Measurement of microenvironmental ozone concentrations in
725 Durham, North Carolina, using a 2B Technologies 205 Federal Equivalent Method monitor and
726 interference-free 2B Technologies 211 monitor, *J. Air Waste Manage.*, 64, 360-371, 2014.
- 727
- 728 Kleindienst, T. E., Hudgens, E. E., Smith, D. F., McElroy, F. F., and Bufalini, J. J., Comparison of
729 chemiluminescence and ultraviolet ozone monitor responses in the presence of humidity and
730 photochemical pollutants, *Air Waste*, 43, 213-222, 1993.
- 731



- 732 Koss, A. R., Sekimoto, K., Gilman, J. B., Selimovic, V., Coggon, M. M., Zarzana, K. J., et al., Non-
733 methane organic gas emissions from biomass burning: identification, quantification, and emission factors
734 from PTR-ToF during the FIREX 2016 laboratory experiment. *Atmos. Chem. Phys.*, 18(5), 3299-3319,
735 2018.
- 736
- 737 Landis, M.S., Edgerton, E.S., White, E.M., Wentworth, G.R., Sullivan, A.P., Dillner, A.M., The impact
738 of the 2016 Fort McMurray Horse River Wildfire on ambient air pollution levels in the Athabasca Oil
739 Sands Region, Alberta, Canada. *Sci. Total Environ.*, 618:1665-1676, 2018.
- 740
- 741 Landis, M.S., Long, R.W., Krug, J., Colon, M., Vanderpool, R., Habel, A., Urbanski, S., The U.S. EPA
742 Wildland Fire Sensor Challenge: Performance and evaluation of Solver Submitted Multi-Pollutant Sensor
743 Systems. *Atmos. Environ.* In review.
- 744
- 745 Leston, A. R., Ollison, W. M., Spicer, C. W., and Satola, J., Potential interference bias in ozone standard
746 compliance monitoring, *J. Air Waste Manage.*, 55, 1464-1472, 2005.
- 747
- 748 Long, R.W., Hall, E., Beaver, M., Duvall, R., Kaushik, S., Kronmiller, K., Wheeler, M., Garvey, S.,
749 Drake, Z., McElroy, F., Performance of the Proposed New Federal Reference Methods for Measuring
750 Ozone Concentrations in Ambient Air, EPA/600/R-14/432, 2014.
- 751
- 752 Mauritz, K. A., Moore, R.B., State of Understanding of Nafion, *Chem. Rev.*, 104, 10, 4535–4586, 2004.
- 753
- 754 Molina, L.T., Molina, M.J., Absolute Absorption Cross Sections of Ozone in the 185- to 350-nm
755 Wavelength Range, *J. Geophys. Res.; Atmos.*, 91 (D13):4719, 1986.
- 756
- 757 Ollison, W. M., Crow, W., Spicer, C. W., Field testing of new-technology ambient air ozone monitors, *J.*
758 *Air Waste Manage.*, 63, 855-863, 2013.
- 759
- 760 Parrish, D.D., Fehsenfeld, F.C., Methods for gas-phase measurements of ozone, ozone precursors and
761 aerosol precursors, *Atmos. Environ.*, 34, 1921-1957, 2000.
- 762
- 763 Spicer, C. W., Joseph, D. W., and Ollison, W. M., A re-examination of ambient air ozone monitor
764 interferences, *J. Air Waste Manage.*, 60, 1353-1364, 2010.
- 765
- 766 Turnipseed, A.A., Andersen, P., Williford, C., Ennis, C., Birks, J., Use of a heated graphite scrubber as a
767 means of reducing interferences in UV-absorbance measurements of atmospheric ozone, *Atmos. Meas.*
768 *Tech.*, 10, 2253–2269, 2017.
- 769
- 770 U.S. Environmental Protection Agency (EPA), National Ambient Air Quality Standards for Ozone,
771 *Federal Register*, 80, 206, October 26, 2015.
- 772



- 773 U.S. Environmental Protection Agency (EPA), Studies Advance Air Monitoring During Wildfires and
774 Improve Forecasting of Smoke, [https://www.epa.gov/sciencematters/studies-advance-air-monitoring-](https://www.epa.gov/sciencematters/studies-advance-air-monitoring-during-wildfires-and-improve-forecasting-smoke)
775 [during-wildfires-and-improve-forecasting-smoke](https://www.epa.gov/sciencematters/studies-advance-air-monitoring-during-wildfires-and-improve-forecasting-smoke), July 30, 2019.
776
- 777 Whitehill, A.; George, I.; Long, R.; Baker, K.R.; Landis, M.S., Volatile organic compound emissions
778 from prescribed burning in tallgrass prairie ecosystems. *Atmosphere*, 10, 464, 2019.
779
- 780 Williams, E. J., Fehsenfeld, F. C., Jobson, B. T., Kuster, W. C., Goldan, P. D., Stutz, J., and McClenny,
781 W. A., Comparison of ultraviolet absorbance, chemiluminescence, and DOAS instruments for ambient
782 ozone monitoring, *Environ. Sci. Technol.*, 40, 5755-5762, 2006.
783
- 784 Wilson, K. L., and Birks, J. W., Mechanism and elimination of a water vapor interference in the
785 measurement of ozone by UV absorbance, *Environ. Sci. Technol.*, 40, 6361-6367, 2006.
786
- 787 Yokelson, R.J.; Griffith, D.W.T.; Ward, D.E., Open-path Fourier transform infrared studies of large-scale
788 laboratory biomass fires. *J. Geophys. Res-Atmos.*, 101(D15):21067-21080, 1996.
789
- 790 Yokelson, R.J., R Susott, R., Ward, D.E., Reardon, J., Griffith D.W.T., Emissions from smoldering
791 combustion of biomass measured by open-path Fourier transform infrared spectroscopy
792 *J. Geophys. Res-Atmos.*, 102 (D15), 18865-18877, 1997.
793
- 794 Yokelson, R. J., Goode, J. G., Ward, D. E., Susott, R. A., Babbitt, R. E., Wade, D. D., et al., Emissions
795 of formaldehyde, acetic acid, methanol, and other trace gases from biomass fires in North Carolina
796 measured by airborne Fourier transform infrared spectroscopy. *J. Geophys. Res-Atmos.*, 104(D23),
797 30109-30125, 1999.
798
- 799 Xu, Z., Nie, W., Chi, X., Huang, X., Zheng, L., Xu, Z., Wang, J., Xie, Y., Qi, X., and Wang, X., Ozone
800 from fireworks: Chemical processes or measurement interference?, *Sci. Total Environ.*, 633, 1007-1011,
801 2018.