

Responses to Reviewers

Reviewer 1

The authors are presenting an innovative approach to address two well-known recycling issues. This starts from improving the reactor, particularly, to prevent UV photon leaks and introduce excess NO.

The manuscript is easy to follow and has merits to the atmospheric chemistry community particularly who intends applying the CRM method for their field observations or using the dataset for the research. I would recommend the publication of this manuscript after addressing following points.

Reply: We would like to thank the reviewer for the insightful comments, which helped us tremendously in improving the quality of our work. Please find the response to individual comments below.

1. The reaction of CO with OH will produce HO₂, which ends up recycling OH in the excess NO environment. Please include the discussion in the manuscript.

Reply: Many thanks, we have included the discussion in the manuscript.

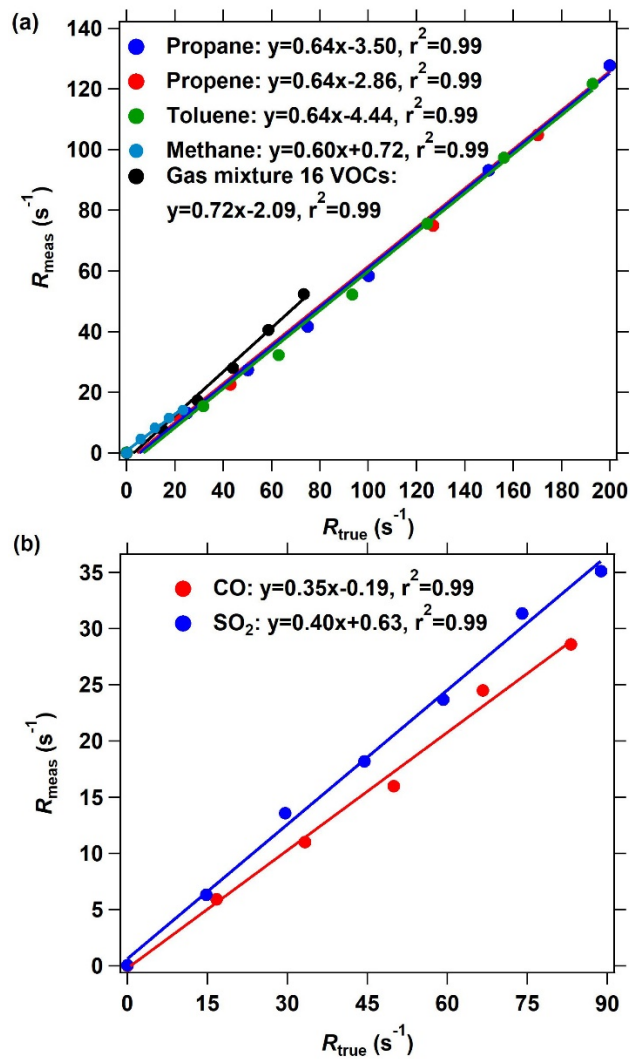
We added a sentence in the revised manuscript (Line 338-340): **Similarly, the produced HO₂ from the reactions of CO and SO₂ with OH will end up recycling OH in the excess NO environments and thus reduce the fitting slopes.**

2. It is certainly a good start to test the ICRM reactor with three VOCs but likely VOCs with various function groups required to evaluate. If the authors could suggest the list of compounds to be tested and provide justification, that would trigger follow up studies from other research groups.

Reply: We appreciate the reviewer for the comment. In addition to the four individual VOC species, we also calibrate using a mixed gas standard with 16 VOC species, namely acetaldehyde, methanol, ethanol, isoprene, acetone, acetonitrile, methyl vinyl ketone, methyl ethyl ketone, benzene, toluene, o-xylene, α -pinene, 1,2,4-trimethylbenzene, phenol, m-cresol, naphthalene. The calibration slope is close to those

30 of the three individual VOC species, indicating that the $\text{RO}_2 + \text{NO}$ reactions for these
31 investigated VOCs should be similar. In addition, we also calibrated methane in the
32 revised manuscript and the results were added in Figure 4. Nevertheless, we agree with
33 the reviewer that it is necessary to calibrate more VOC species in the future, especially
34 considering that different VOCs species dominate in different environment, such as
35 forest areas and various emission sources. For example, isoprene and terpenes have
36 high reactivity contribution in forests and rural sites. Therefore, isoprene, α -Pinene and
37 β -Pinene is suggested to be calibrated in the following study. Typical branched olefin,
38 other aromatics (such as ethylbenzene) and important oxygenated VOCs (such as
39 formaldehyde and glyoxal) should also be calibrated in the future.

40 We added the following sentences in the revised manuscript (Line 404-414): **In**
41 **this study, we calibrated four individual representative VOC species (methane,**
42 **propane, propene, toluene). In addition, we also calibrated the mixed standard**
43 **gases with 16 VOC species including representative oxygenated VOCs**
44 **(acetaldehyde, methanol, ethanol, acetone, acetonitrile, methyl vinyl ketone,**
45 **methyl ethyl ketone), biogenic VOCs (isoprene, α -pinene), typical aromatics**
46 **(benzene, toluene, o-xylene, 1,2,4-trimethylbenzene, naphthalene, phenol, m-**
47 **cresol). The calibration slope is close to those of the four individual VOC species,**
48 **indicating that the $\text{RO}_2 + \text{NO}$ reactions for these investigated VOCs should be**
49 **similar. Nevertheless, given that there are different VOCs compositions in**
50 **different environment such as forest, urban area and emission sources,**
51 **calibrations for more individual VOCs species might be also needed.**



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53 Figure 4. The OH reactivity calibration of the improved CRM system using different
 54 standard gases. (a) The calibrating results of organic species including methane,
 55 propane, propene, toluene, and a mixture of 16 VOC species through arm C. (b) The
 56 calibrating results of inorganic species including CO and SO₂. The measured OH
 57 reactivity was calculated based on the C2 mode shown in Fig. 2 in the ICRM system.

