Estimation of particulate organic nitrates from thermodenuderaerosol mass spectrometer measurements in North China Plain

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Abstract. Particulate organic nitrates (pON) are an important component of secondary organic aerosol in biogenic emission dominant environments, and play a critical role in NO_x cycles. However, estimation of pON has been a challenge in polluted environments, e.g., North China Plain with high concentrations of inorganic nitrate and NO_x. Here we developed a method for estimation of pON from the measurements of high-resolution aerosol mass spectrometer coupled with a thermodenuder based on the volatility differences between inorganic nitrate and pON. The results generally correlated well with those estimated from positive matrix factorization of combined organic and inorganic mass spectra and from the ratio of NO⁺ to NO₂⁺ (NO_x⁺ ratio), yet had improvements in reducing negative values due to the influences of high concentration of inorganic nitrate and constant NO_x⁺ ratio of organic nitrates (R_{ON}). By applying this approach to the measurements at an urban (Beijing) and a rural site (Gucheng) in summer and winter in North China Plain, we estimated that the average mass concentrations of NO_{3,org} (1.8 μg m⁻³ vs.1.0 μg m⁻³) and pON to OA (27.5% vs. 14.8%) were higher in summer than in winter in Beijing, indicating more pON formation in biogenically and anthropogenically mixed environments. In addition, the average NO_{3,org} loading in Gucheng was 1.9 μg m⁻³, and the pON at the rural site also showed higher contribution to OA than that in Beijing during wintertime due to higher primary emissions and gaseous precursors in Gucheng. In addition, R_{ON} was determined and showed considerable differences between day-night and clean-polluted periods, highlighting the complexity of pON compounds from different chemical pathways (e.g., OH and NO₃ oxidation) and sources.

Introduction

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Particulate organic nitrates (pON), accounting for 5-77% of organic aerosol (OA) (Ng et al., 2017), are important atmospheric constituents as they affect NO_x recycling, ozone, and secondary organic aerosol (SOA) production (Perring et al., 2013; Present et al., 2020). Most previous studies on pON quantification are based on filter measurements using gas chromatography (GC) (O'Brien et al., 1995) and Fourier transformed infrared (FTIR) spectroscopy (Blando et al., 1998). On-line measurement techniques, e.g., thermal dissociation laser induced fluorescence (TD-LIF) (Day et al., 2002), cavity ring-down spectroscopy (TD-CRDS) (Thieser et al., 2016) and cavity attenuated phase shift spectroscopy (TD-CAPS) (Sadanaga et al., 2016) were also used to characterize pON in the atmosphere. The principle is to convert pON to NO₂ first and then to subtract ambient NO₂ from the total NO₂. However, the pON may not be accurately quantified under ambient NO₂ dominated conditions. High-resolution time-of-flight chemical ionization mass spectrometer (HR-ToF-CIMS) using iodide-adduct ionization (Lee et al., 2014) with a filter inlet for gases and aerosols (FIGAERO) (Lopez-Hilfiker et al., 2014) (FIGAERO-CIMS) allows for simultaneous characterization of molecular compositions of gas and particle phases of ON. However, the lack of authentic ON standards and different instrument sensitivity for various ON compounds increase the challenges in quantification of pON.

Recently, pON were also quantified using high-resolution aerosol mass spectrometer (HR-AMS) mainly based on the differences of fragmentation patterns between pON and inorganic nitrate (Xu et al., 2015b; Farmer et al., 2010; Kortelainen et al., 2017; Ng et al., 2017). "PMF method" and "NO_x method" are two most commonly used method for pON quantification (Fry et al., 2013; Xu et al., 2015b; Yu et al., 2019; Dai et al., 2019). The "PMF method" by combining organic and inorganic fragments in positive matrix factorization (PMF) analysis was able to identify organic aerosol factors that showed largely different NO⁺/NO₂⁺ ratio from pure ammonium nitrate (AN). The NO_x fragments are then assumed to be mainly from pON (Xu et al., 2015b; Zhang et al., 2016). However, the separation of inorganic and organic NO_x^+ may be limited by the bilinear model depending on the correlations between inorganic nitrate and pON. In addition, some primary organic aerosol (POA) factors from "PMF method" also contained much higher NO+ or NO2+ compared to directly measured mass spectrum of primary emissions (Xu et al., 2020b) (Fig. S1), increasing the uncertainties in quantification of pON. The "NO_x method" is based on the upper and lower bounds of R_{ON} (ratio of NO⁺ to NO₂⁺ of pON) that were determined from oxidations of isoprene and monoterpenes (β-pinene) with nitrate (NO₃) radicals (Fry et al., 2009; Bruns et al., 2010; Boyd et al., 2015). One of the major uncertainties is that R_{ON} varies largely depending on reaction conditions and precursors in ambient rather than being constant values. Nevertheless, previous field studies failed to explore the temporal variation of Ron due to the limitation of methods and the diversity of pON. In some environments, the measured ratio of NO⁺ to NO₂⁺ in ambient is even lower than the average ratio of NO⁺ to NO₂⁺ from ammonium nitrate (R_{AN}), resulting in negative values of pON estimated by the "NO_x method", especially for the wintertime in China when aerosol particles are dominated by inorganic nitrate (Yu et al., 2019; Zhu et al., 2016). Despite the uncertainties above, extensive studies have been conducted to characterize spatial and seasonal

variations of pON in low NO_x environments with high biogenic emissions, e.g., the southeastern United States and Europe (Kiendler-Scharr et al., 2016; Lee et al., 2016; Xu et al., 2015b; Ng et al., 2017). Similar work, however, is limited in China, especially in northern China with high emissions of NO_x and anthropogenic volatile organic compounds (VOCs). In recent years, some studies explored the concentrations of pON at two background sites in spring (Zhu et al., 2016) and during biomass burning and coal combustion periods in Beijing (Zhang et al., 2016). Yu et al. (2019) found a good correlation between pON and fresh SOA at night at an urban site in southern China based on "NO_x method" and "PMF method". In contrast, Qiao et al. (2020) reported that pON were more correlated with the emission from biomass burning but poorly correlated with less-oxidized oxygenated OA (LO-OOA) at a rural site (Xianghe) in summer. Despite these results, our understanding of the levels, sources and processes of pON in North China Plain (NCP) remains poor.

In this study, we demonstrate the applications of thermodenuder (TD)-HR-AMS in characterization of pON (referred to as "TD-AMS method") during wintertime and summertime in NCP (including an urban site and a rural site) and compare pON mass loadings from three different methods, i.e., "NO_x method", "PMF method" and "TD-AMS method". The temporal variation of R_{ON} during three campaigns are elucidated. The mass loadings and diurnal variations of pON during one summer and two winter campaigns are elucidated and the impacts of relative humidity and temperature on pON are discussed.

2 Experimental methods

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2.1 Experimental set-up

This study was performed at the Institute of Atmospheric Physics, an urban site in Beijing, and Gucheng, a rural site in Hebei province. The field observations at the urban site were conducted in summer from 20 May to 23 June and winter from 20 November to 25 December in 2018, and at the rural site from 10 December 2019 to 13 January 2020. The set up and operations of TD and HR-AMS were similar to previous studies (Xu et al., 2019). The HR-AMS that was operated in V mode with a time resolution of 3 min was placed downstream of the TD. TD alternated between the bypass line and TD line in a cycle of 30 min. The TD heating temperature was set as 50, 70, 90, 109, 120, 127 and 250 °C in summer and 50, 70, 90, 109, 127, 150 and 250 °C in winter in Beijing, while in winter of 2019 in Gucheng, the setting TD temperature ramped linearly from 50 °C to 250 °C with a sampling cycle of 150 min. The average particle loss in the TD was corrected according to the calibration by using NaCl and was found to be 10% (Huffman et al., 2008).

2.2 Data analysis

The HR-AMS data analysis was performed using PIKA (V 1.62F). The HR-AMS was calibrated for ionization efficiency (IE) using pure ammonium nitrate by following the standard protocols (Jayne et al., 2000). Here, we apply three independent

methods of quantifying pON. Note that the default relative ionization efficiencies (RIE) of 1.1 and 1.4 were applied for inorganic nitrate and NO_{3,org}, respectively, during three campaigns. Considering the high mass fraction of ammonium nitrate, the composition-dependent collection efficiency (CDCE) was applied to the bypass data (Middlebrook et al., 2012). In contrast, a constant CE of 0.5 was used for those from TD, consistent with the data processing in previous studies (Xu et al., 2016; Huffman et al., 2009). The elemental composition of OA was calculated using "Improved-Ambient (I-A)" method (Canagaratna et al., 2015).

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The first method is "NO_x method" which is based on the differences in NO⁺/NO₂⁺ ratio between organic and inorganic nitrates (Farmer et al., 2010; Xu et al., 2015b). The R_{AN} was determined from ionization efficiency (IE) calibrations as 3.7 in summer and 3.4 in winter in both Beijing and Gucheng. However, we found that the winter R_{AN} was higher than those from ambient observations. Therefore, we chose a period with high NO₃ loadings (assuming predominant inorganic nitrate (NO_{3,inorg}) contribution) to determine R_{AN} which was 2.8 and 2.2, respectively in Beijing and Gucheng in winter. Note that NO_{3,org} loadings calculated by "NO_x method" were slightly underestimated in winter in this case due to the organic nitrate contribution even in a period with high NO₃ loadings. In addition, the large uncertainties in pON quantification were found from the "NO_x method" under high PM levels during wintertime due to the assumption for R_{AN}. To reduce the effect of variation in R_{ON} value among different instruments, we determined R_{ON} by multiplying R_{ON}/R_{AN} with R_{AN} in using "NO_x method", assuming that R_{ON}/R_{AN} is independent of instrument (Fry et al., 2013). The R_{ON}/R_{AN} ratios of 2.08 and 4.17 from isoprene and β -pinene oxidation experiments were used as upper and lower bounds of pON in this study (Xu et al., 2015b). Hence, R_{ON} of 7 and 15, 5 and 12, 4 and 10 obtained in summer and winter in Beijing and Gucheng during wintertime, respectively, were used to estimate the upper and lower bounds of pON loadings in this study.

The second method is PMF analysis of combined organic and inorganic aerosol (referred to "PMF method") (Sun et al., 2012; Yu et al., 2019; Xu et al., 2015b). By including NO⁺ and NO₂⁺ ions, PMF analysis was able to separate organic from inorganic nitrates. For instance, Sun et al. (2012) performed factor analysis on combined organic and inorganic aerosol mass spectra and found that the NO⁺ and NO₂⁺ ions in the nitrate factor were dominantly from inorganics, and the ratio of NO⁺/NO₂⁺ was close to that of pure ammonium nitrate, while those in OA factors with high NO⁺/NO₂⁺ were generally assumed as organic nitrates in summer in New York City. In this study, the concentrations of NO_{3,Org} are calculated by summing up NO⁺ and NO₂⁺ from both POA and SOA factors. The time series and mass spectra of PMF factors resolved from PMF_{Org+NO3} were shown in Figs. S2-S3. The mass spectral profiles of all PMF factors resolved from PMF_{Org+NO3} are similar to those previously resolved in ambient (Yu et al., 2019; Xu et al., 2015b). The mass concentrations of PMF factors resolved from PMF_{Org+NO3} overall were well correlated with that from PMF_{Org} (Table.S2) and also resembled their related tracers. However, some discrepancies in mass concentrations were observed. For example, the concentrations of MO-OOA and LO-OOA resolved

from PMF_{Org} were lower than that from PMF_{Org+NO3}, which was caused by the signals of NO_x⁺ in mass spectra of SOA in winter in Beijing. The OA factors representing different sources were identified by PMF and the detailed PMF analysis during three campaigns have been given in (Xu et al., 2020a) and Chen et al. (2021).

The third method denotes the "TD-AMS method" which is based on the different volatility between organic and inorganic nitrates. Briefly, NO⁺ and NO₂⁺ (hereafter NO_x⁺) signals in HR-AMS mass spectra can be from fragmentation of both particulate organic and inorganic nitrates, whereas $C_x H_y N_z^+$ and $C_x H_y O_z N_p^+$ are completely from particulate organic nitragen compounds including ON and other nitrogen-containing organic compounds. The $C_xH_vN_z^+$ and $C_xH_vO_zN_p^+$ dominated the N mass for non-ON organic nitrogen compounds (Fig. S4), while NO⁺ and NO₂⁺ were the main N-containing ions for ON (Farmer et al., 2010; Boyd et al., 2015; Xu et al., 2015a). Note that the $C_xH_yN_z^+$ and $C_xH_yO_zN_p^+$ families refer to the sum of N-containing ions at m/z < 60 due to the limited mass resolution of V-mode (Xu et al., 2017b). Owing to higher volatility of NO_{3,inorg} than ON (Huffman et al., 2008; Ng et al., 2017), the remaining mass loadings of nitrate fragments from pON (denoted as NO x +, Org) is expected to be much higher than that of nitrate fragments from inorganic nitrates (denoted as NO x + inorg) under high TD temperatures (Huffman et al., 2008; Berkemeier et al., 2020). As shown in Fig. 1, NO⁺ and NO₂⁺ decreased significantly from 25 °C to 90 °C, whereas the decreases slowed down substantially for T > 90 °C. Hence, we assumed that the NO_{3,inorg} evaporated completely at T = 90 °C. In addition, the N mass from NO⁺ and NO₂⁺ dominated the total N at T > 90 °C (Fig. 1 (a3-c3)), indicating that pON dominated the total particulate organic nitrogen compounds in NCP in both summer and winter. The NO_x⁺ measured by HR-AMS for T > 90 °C was the remaining organic nitrates compounds upon heating, consistent with previous studies reporting a complete evaporation of pure AN at ~60 °C (Huffman et al., 2008). The slightly different evaporation temperature of AN between this study and Huffman et al. (2008) could be due to the different aerosol mixing states in different environments. For example, Li et al. (2021) and Liu et al. (2021) found that that secondary inorganic aerosol particles were frequently coated with organics in NCP, which may affect the evaporation temperature of AN.

Assuming that the volatility of pON can be represented by total organic nitrogen compounds, the mass concentrations of pON in ambient can be determined as:

$$[NO_{3,org}] = \frac{[NO_3]_{T=90^{\circ}C}}{MFR_{CHN^+ + CHON^+ T=90^{\circ}C}}$$
(1)

The ratio of NO⁺ to NO₂⁺ of pON (denoted as R_{ON,calc}) can be determined as

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$$[NO_{3,inorg}] = [NO_3] - [NO_{3,org}]$$
(2)

$$\left[NO_{\text{inorg}}\right] = \frac{R_{AN}}{1 + R_{AN}} \times \left[NO_{3,inorg}\right] \tag{3}$$

$$[NO_{2,\text{inorg}}] = \frac{1}{1 + R_{AN}} \times [NO_{3,inorg}]$$
(4)

$$\left[NO_{org}\right] = \left[NO\right] - \left[NO_{\text{inorg}}\right]$$
(5)

$$[NO_{2,org}] = [NO_2] - [NO_{2,inorg}]$$

$$(6)$$

$$R_{ON,\text{calc}} = \frac{[NO_{org}]}{[NO_{2,org}]} \tag{7}$$

The subscript "T = 90 °C" denotes the mass concentrations and mass fraction remaining (MFR) of fragments at T = 90 °C. The subscripts "org" and "inorg" denote the mass concentration of fragments of particulate organic and inorganic nitrates. NO, NO₂ and NO₃ are the measured mass concentrations of total nitrate fragments. R_{AN} is the average ratio of NO⁺ to NO₂⁺ from ammonium nitrate. The sensitive tests of NO_{3,Org} and R_{ON} with the variation of MFR and R_{AN} are shown in Fig. S5, demonstrating the impact of accurate determination of MFR and R_{AN} on R_{ON} in "TD-AMS method".

Note that the pON/OA in this study refers to the contribution of pON to total OA including the contribution of nitrate functional groups (i.e., total OA = $Org + NO_{3,Org}$). The conversion method from $NO_{3,Org}$ /Org into pON/OA is illustrated in Takeuchi and Ng (2019).

3 Results and discussion

3.1 Intercomparisons

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Figure 2 shows the NO_{3,org} mass loadings estimated from three different methods during 3 campaigns. The NO_{3,org} loadings calculated by "TD-AMS method" correlated well with those of "NO_x method" during summertime (R^2 =0.62), but different slopes were found in different periods of the measurements. For example, the slope was 1.1 during 15-17 June (R^2 =0.94), implying higher NO_{3,org} loadings calculated by "NO_x method" with R_{ON} =15 than that using "TD-AMS method". Comparatively, the slope was 0.35 during 11-12 June (R^2 =0.62), suggesting lower loadings from the "NO_x method". Such differences were likely due to different pON species that were formed under different precursors and meteorological conditions

(e.g., temperature, NO_x levels and RH). Also, the slope differences indicate the variations of pON compounds leading to the variation of R_{ON} ($R_{ON,calc}$ =5.0 during 11-12 June vs. 13.6 during 15-17 June), highlighting the importance of determining the time-dependent R_{ON} . On average, the mass concentration of $NO_{3,org}$ calculated by "TD-AMS method" in summer was 1.8 μ g m⁻³, which was slightly higher than that from "NO_x method" (1.0-1.5 μ g m⁻³). Negative $NO_{3,org}$ mass loadings from the "NO_x method" were observed, accounting for 8.5% and 12.8% of the time during wintertime in Beijing and Gucheng, respectively. Such phenomenon was also observed in previous studies in China (Zhu et al., 2016; Yu et al., 2019), particularly in winter and was likely due to (1) high emissions of anthropogenic volatile organic compounds (AVOC) and high NO_x condition resulting in similar R_{ON} and R_{AN} , and (2) much higher inorganic nitrate loading than pON. As shown in Fig.2, the $NO_{3,org}$ from the "TD-AMS method" correlated well with that from the "NO_x method", yet the mass loadings were deviated from the lower and upper bounds from the "NO_x method" during some periods likely due to different pON compounds. The $NO_{3,org}$ loadings from the "NO_x method" ranged from 0.9 to 1.3 μ g m⁻³ and from 0.8 to 1.6 μ g m⁻³ in Beijing and Gucheng during wintertime, respectively, which is overall comparable to that of "TD-AMS method" in Beijing (1.0 μ g m⁻³) and slightly lower than that in Gucheng (1.9 μ g m⁻³).

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The average mass concentration of NO_{3,org} calculated from the "PMF method" (0.28 μg m⁻³) is much lower than those from the other two methods in summer in Beijing, in agreement with the results in Southeastern U.S. (Xu et al., 2015b). In contrast, the average NO_{3,org} (2.5 µg m⁻³) in winter in Beijing was higher, and that (1.8 µg m⁻³) in winter in Gucheng was comparable to that from the other two methods. Such differences during three campaigns indicate the uncertainties of "PMF method" in estimation of NO_{3,org} can originate from, for example, (1) the incorrect assignment of NO⁺ and NO₂⁺ to POA factors, (2) the different ratio of NO⁺ to NO₂⁺ in nitrate inorganic aerosol (NIA) factor from pure ammonium nitrate, and (3) the contribution of organic ions to NIA factor. As shown in Fig. S2, the nitrogen-containing fragments (C_xH_vN⁺, accounting for 5% of NIA) were unexpectedly assigned to NIA factor, while the NO_x⁺ in MO-OOA, a factor often related to aqueous-phase processing (Xu et al., 2017a) were negligible, resulting in the minor response of NO_{3,org} loadings as a function of RH. This is also supported by the fact that the concentrations of NO_{3,org} from the "PMF method" were much lower than those from "TD-AMS method" under high RH levels (10-12 June, RH = 70%). The NO_x^+ signals assigned to POA in winter in Beijing were much higher than those in source emission experiments. For example, f_{NO+} (3.5%) and f_{NO2+} (1.5%, fractions of NO⁺ or NO₂⁺ in total mass spectrum) of biomass burning OA in winter in Beijing (Fig.S3) were higher than that from burning of different biofuels (f_{NO+} =0.9% and f_{NO2+} =0.65%) (Xu et al., 2020b). Further analysis suggested that the episodes dominated by POA also resulted in higher pON loadings from the "PMF method" in winter in Beijing. Considering the uncertainties of "PMF method", the following discussions were mainly based on the results from the "TD-AMS method" and "NO_x method".

The average mass concentration of NO_{3,org} in summer was higher than that in winter in Beijing, and also the contribution of

NO_{3,org} to NO₃ (24.1% vs. 9.8%). This result is consistent with the seasonal variation of NO_{3,org} loadings observed in the Southeastern U.S. (Xu et al., 2015b). One explanation is stronger secondary formation during summertime than winter, as indicated by the higher fraction of SOA and SIA (Zhou et al., 2020). The dominant biogenic volatile organic compounds (BVOC) in summer and AVOC in winter (Liu et al., 2020) would be another reason for the differences in pON loadings in two seasons. The contribution of NO_{3,org} to NO₃ (9.8% vs. 11.6%) is comparable between Beijing and Gucheng during wintertime, yet the average NO_{3,org} in Gucheng was higher than that in Beijing during wintertime. This can be attributed to much higher anthropogenic emissions (e.g. coal combustion and biomass burning) at the rural site than urban site (Sun et al., 2020; Xu et al., 2020a), resulting in different pON compounds as suggested by the different R_{ON} (R_{ON,calc}=2.2 in Beijing and 4.3 in Gucheng on average). Also, the variations of N mass contributions from different N-containing ions as a function of TD temperature supported the differences in pON compounds (Fig. 1). Note that the contribution of pON (assuming MW_{ON}=200 g mol⁻¹) to OA in NCP (14.8%-27.5%) was in the range of previous studies (Ng et al., 2017; Yu et al., 2019), and the seasonal variations of pON/OA in Beijing (14.8% in winter vs. 27.5% in summer, Table S1) were also similar to previous studies (Xu et al., 2015b). The higher contribution of pON to OA in Gucheng than Beijing (24.0% vs. 14.8%) indicated more important role of pON in OA at the rural site in NCP during wintertime.

The negligible correlation between NO_{3,org} with POA in summer in Beijing (*R*²=0.05) is in contradiction with the results in Xianghe (a rural site in NCP), which shows that pON formation might be highly associated with primary emissions (Qiao et al., 2020). Such differences were caused by the different primary emission in different environments. For example, BBOA is the dominant POA in Xianghe (Qiao et al., 2020), but not in Beijing during summertime. The higher correlation between NO_{3,org} with POA was found in winter in NCP (*R*²=0.49-0.52), suggesting that primary emissions can be an important source of pON in winter in NCP, which may be related to NO₃ oxidation of AVOC emitted from anthropogenic activities (Li et al., 2019; Lee et al., 2019). Our results are consistent with the considerable contributions of alkanes (a typical AVOC) to pON production in previous field (Lee et al., 2015), model (Jordan et al., 2008) and chamber studies (Lim and Ziemann, 2009). In fact, there is a growing consensus about the importance of AVOC (e.g. styrene (Yu et al., 2019), toluene (Ramasamy et al., 2019) and long-chain aliphatics (Matsunaga and Ziemann, 2010)) in pON formation under high NO_x condition (Lee et al., 2015; Li et al., 2019; Lee et al., 2019).

3.2 Diurnal variation

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As shown in Fig.3, the NO_{3,org} loadings and contributions from the "TD-AMS method" and "NO_x method" showed similar diurnal profiles in summer in Beijing, which were characterized by decreased concentration during daytime, in agreement with the behaviors of total NO₃ in Beijing (Sun et al., 2013) and NO_{3,org} in Shenzhen (Yu et al., 2019). The elevated planetary

boundary layer (PBL) and temperature-dependent gas–particle partitioning could be the major reason. We noticed an elevated fraction of $NO_{3,org}$ in NO_3 during daytime in summer, likely suggesting that photochemical processing played an important role in the formation of pON in summer in Beijing. This is consistent with the moderate correlation of O_x with $NO_{3,org}$ during 12:00-18:00, and tight correlation between LO-OOA, an OA factor related to photochemical production (Xu et al., 2017a) and $NO_{3,org}$ from 6:00-16:00 during summertime (Fig. S6). Note that $NO_{3,org}$ loadings, the contributions of pON to OA, and $NO_{3,org}/NO_3$ showed apparent increases starting at 20:00 in summer, highlighting the important role of NO_3 radical in the nocturnal pON formation (Fry et al., 2013; Zhang et al., 2016; Ng et al., 2017). Indeed, we calculated the ratio of NO^+ to NO_2^+ of ON during daytime and nighttime and found higher R_{ON} at night ($R_{ON,calc}=5.8\pm2.8$) than daytime ($R_{ON,calc}=5.2\pm3.0$) in summer (Fig.S7, the t value of 1.79 between day and night was larger than that 90% confidence level (1.66)), suggesting that the pON compounds from OH-initiated and NO_3 -initiated chemical reactions were different. This higher R_{ON} for pON formed via NO_3 oxidation than those via OH oxidation is consistent with the result of chamber experiments (Takeuchi and Ng, 2019).

The diurnal variations of NO_{3,org} loadings in winter in Beijing and Gucheng were characterized by significant increases during daytime, indicating that photochemical production played an important role in pON formation. However, the fraction of pON in OA remained relatively stable throughout the day, suggesting that photochemical processing also produced other SOA besides pON. NO_{3,org} from the two different methods showed some different trend during some periods in winter (Fig.3). For example, the NO_{3,org} from the "NO_x method" dropped at 8:00, while that from "TD-AMS method" peaked at 8:00 in winter in Beijing. The differences in NO_{3,org} from the two different methods also appeared in the morning in Gucheng, where NO_{3,org} loadings from the "TD-AMS method" showed an obvious enhancement at ~6:00, yet not in "NO_x method". Also, the fraction of pON from the "TD-AMS method" in OA (22.8%) was higher than that from "NO_x method" (5.7-11.0%). Such differences in the morning during wintertime could be attributed to the differences in organic nitrogen composition with different volatility and also different R_{ON} between morning and afternoon. In Beijing in winter, the peak of NO_{3,org} from the "TD-AMS method" was partly caused by the changes in AVOC emissions (e.g., acetone, benzene, c8-aromatics and toluene) (Sheng et al., 2019). In Gucheng in winter, photochemical aqueous-phase processing was one of the reasons for those differences in the morning due to the frequent occurrence of radiation fog that dissipated after sunrise (Kuang et al., 2020). Overall, our results suggest that the use of a fixed R_{ON} value in "NO_x method" was limited in representing diurnal variation characteristics of pON in winter in NCP.

3.3 Dependence of pON on NR-PM₁ loading

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As shown in Fig.4, NO_{3,org} showed an overall increasing trend as the increase of PM in both summer and winter. However, the variations of pON in OA were different between summer and winter. The pON fraction in OA were comparable (~14%)

between summer and winter in Beijing in clean days (NR-PM₁ < 20 μ g m⁻³), yet the R_{ON} in summer (R_{ON,calc}=7.0) was much higher than that in winter (R_{ON,calc}=2.8), emphasizing the seasonal differences in pON composition. Comparatively, the pON fraction in OA at the rural site was higher than that in Beijing in winter, and the R_{ON} was also different (R_{ON,calc}=6.0 in Gucheng vs. R_{ON,calc}=2.8 in Beijing) highlighting the very different pON compounds between rural and urban sites during clean days. The fraction of pON in OA at high PM levels shows significant differences. For example, the fraction of pON in OA showed a clear increase as a function of PM and reached ~30% at high PM level (> 50 μ g m⁻³) in summer in Beijing, indicating the increased contribution of pON to OA during polluted episodes. Comparatively, the contribution of pON to OA remained small under different PM levels in winter in Beijing and Gucheng. By linking the variations of pON with OA factors, we found that LO-OOA (a factor related to the photochemical processing (Xu et al., 2017a)) shows obvious increase in polluted conditions in summer, yet not in winter in Beijing (Fig. S8), indicating that photochemical processing played an increasing role in polluted conditions during summer yet limited in winter in Beijing. In addition, the largely enhanced primary emissions at high PM levels in winter could be another reason for the lower contribution of pON to OA. Different from the variations of pON/OA, the fraction of NO_{3,org} in total NO₃ showed overall decreasing trends as a function of PM levels, indicating that the increases of inorganic nitrate were more rapid than NO_{3,org}. We found that NO_{3,org} was more important than inorganic nitrate during clean days in summer by contribution $^{\circ}$ 0% although the contribution decreased rapidly to 20% in polluted episodes.

Table 1 summarizes the average $R_{ON,calc}$ in clean and polluted days during three campaigns. $R_{ON,calc}$ showed an lower value in polluted days than that in clean days, indicating the changes of ON compounds in different PM levels. Note that the $R_{ON,calc}$ of 1.2 in polluted days in winter in Beijing was lower than the R_{AN} of 2.8. Besides the intense primary emission, complicated precursors in winter, the mixture effect from AVOC and BVOC oxidations may have played a part in this low $R_{ON,calc.}$, which warrants further studies. In addition, the uncertainties of R_{AN} from a period with high NO₃ loadings could be another possible reason.

3.4 Effects of meteorological conditions

Figure 5 shows the variations of NO_{3,org} mass loadings as a function of RH and temperature during daytime (8:00-20:00) and nighttime (20:00-8:00) during three campaigns. In general, NO_{3,org} showed an increasing trend as function of RH below 80% in both daytime and nighttime. Previous studies in NCP showed ubiquitously increased PM levels as a function of RH. In addition to the increased aqueous-phase processing, the stagnant meteorological conditions under high RH levels were often found to be the major reason. Therefore, the increases of NO_{3,org} as a function of RH could be mainly associated with the increased PM levels. As RH increased above 90%, we observed some decreases in NO_{3,org}, possibly indicating the impact of hydrolysis of pON (Takeuchi and Ng, 2019; Rindelaub et al., 2015; Liu et al., 2012). However, since OA concentration also exhibited a decreasing trend above 80-90% RH as with NO_{3,org}, the decrease in NO_{3,org} may have been also affected by

partitioning. The day and night differences in NO_{3,org} under low and high RH levels were also shown in Fig.5. For example, we observed higher NO_{3,org} loadings during daytime than nighttime at RH<60%, indicating that photochemical formation of pON was more important than nocturnal production, which is further supported by the elevated fraction of NO_{3,Org} in NO₃ in daytime (Fig. 3). However, higher NO_{3,Org} loadings at nighttime than daytime were observed at RH > 60% in Gucheng, likely suggesting the importance of nocturnal reactions associated with enhanced primary emissions at high RH level. Overall, the day-night differences highlight the impacts of multiple factors on ON formation, including precursors, e.g., biogenic and anthropogenic VOCs, and oxidants, e.g., OH and NO₃ radicals, which needs to be further investigated by analyzing molecular compositions of ON.

The NO_{3,org} loadings showed a slightly decreasing trend as function of T in summer due to the evaporative loss under high temperature. This result was consistent with the chamber experiments showing decreasing monoterpenes pON as temperature increased from ~25 to ~40 °C (Boyd et al., 2017; Berkemeier et al., 2020). In contrast, the NO_{3,org} loadings in both daytime and nighttime showed increasing trends as a function of T in winter in Beijing suggesting the importance of photochemical production. In contrast, the NO_{3,org} loadings decreased as temperature elevated in Gucheng during wintertime, and then remained relatively stable at T > 0 °C. One reason was due to much enhanced primary emissions from residential heating under low temperature.

4 Limitations and Conclusions

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In this study, we demonstrated the capability of combining HR-AMS and TD measurements to quantify and characterize pON in NCP in summer and winter. However, this approach has several limitations. Firstly, we assumed that $NO_{3,inorg}$ evaporated completely at $T = \sim 90$ °C, which might not be the case in some environments and could overestimate evaporative loss of NO^+ and NO_2^+ from ON. Secondly, the average mass loss of $C_xH_yN_z^+$ and $C_xH_yO_zN_p^+$ at T = 90 °C were used as the proportion of mass loss of pON, however, the volatilities of pON vary among different pON species. Indeed, it is challenging to quantify due to the extremely complex pON in ambient air, and also the absence of pON standards. In addition, separating nitrogen ion fragments is a challenge due to the limited resolution of HR-AMS, leading to the uncertainties in calculating mass losses of $C_xH_yN_z^+$ and $C_xH_yO_zN_p^+$. Thirdly, using a default value of $RIE_{ON} = 1.4$, and a fixed molar mass ($MW_{ON} = 200$ g mol⁻¹) for estimation of pON would introduce additional uncertainties considering the complexity of pON compounds. Fourthly, the low volatility nitrogen-containing compounds (e.g. metal nitrates) with high NO_x^+ ratio cannot be separated resulting in an overestimation of pON. Lastly, HR-AMS only measures submicron non-refractory pON, which might also underestimate pON in ambient air.

Overall, the pON estimated from the "TD-AMS method" correlated well with those from PMF and NO_x methods, yet has advantages in estimating pON in anthropogenically dominant environment. Particularly, we found that R_{ON} varied differently between day and night, clean and polluted episodes, highlighting different pON compounds from different chemical pathways and sources. Our estimation showed higher mass concentrations of $NO_{3,org}$ in summer than that in winter (1.8 vs. 1.0 $\mu g m^{-3}$), and also the contribution of pON to OA (27.5% vs. 14.8%), indicating more pON formation in biogenically and anthropogenically mixed environment. In addition, the $NO_{3,org}$ loadings at the rural site (1.9 $\mu g m^{-3}$) was higher than that in Beijing due to higher primary emissions and gaseous precursors during wintertime.

Data availability. The data in this study are available from the authors upon request (sunyele@mail.iap.ac.cn).

Author contributions. YS and WX designed the research. WX, CC, YQ, CX, WanX and NM conducted the measurements.
 WX, MT, CC and YQ analyzed the data. MT, DW and NLN reviewed and commented on the paper. WX and YS wrote the paper.

Competing interests. The authors declare that they have no conflict of interest.

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References

15

25

- Berkemeier, T., Takeuchi, M., Eris, G., and Ng, N. L.: Kinetic modelling of formation and evaporation of SOA from NO3 oxidation of pure and mixed monoterpenes, Atmos. Chem. Phys. Discuss., 2020, 1-31, 10.5194/acp-2020-55, 2020.
- Blando, J. D., Porcja, R. J., Li, T. H., Bowman, D., Lioy, P. J., and Turpin, B. J.: Secondary formation and the Smoky Mountain organic aerosol: An examination of aerosol polarity and functional group composition during SEAVS, Environ. Sci. Technol., 32, 604-613, 1998.
 - Boyd, C. M., Sanchez, J., Xu, L., Eugene, A. J., Nah, T., Tuet, W. Y., Guzman, M. I., and Ng, N. L.: Secondary organic aerosol formation from the β-pinene+NO₃ system: effect of humidity and peroxy radical fate, Atmos. Chem. Phys., 15, 7497-7522, 10.5194/acp-15-7497-2015, 2015.

- Boyd, C. M., Nah, T., Xu, L., Berkemeier, T., and Ng, N. L.: Secondary Organic Aerosol (SOA) from Nitrate Radical Oxidation of Monoterpenes: Effects of Temperature, Dilution, and Humidity on Aerosol Formation, Mixing, and Evaporation, Environ. Sci. Technol., 51, 7831-7841, 10.1021/acs.est.7b01460, 2017.
- Bruns, E. A., Perraud, V. r., Zelenyuk, A., Ezell, M. J., Johnson, S. N., Yu, Y., Imre, D., Finlayson-Pitts, B. J., and Alexander, M. L.: Comparison of FTIR and Particle Mass Spectrometry for the Measurement of Particulate Organic Nitrates, Environ. Sci. Technol., 44, 1056-1061, 10.1021/es9029864, 2010.

10

15

25

- Canagaratna, M. R., Jimenez, J. L., Kroll, J. H., Chen, Q., Kessler, S. H., Massoli, P., Hildebrandt Ruiz, L., Fortner, E., Williams, L. R., Wilson, K. R., Surratt, J. D., Donahue, N. M., Jayne, J. T., and Worsnop, D. R.: Elemental ratio measurements of organic compounds using aerosol mass spectrometry: characterization, improved calibration, and implications, Atmos. Chem. Phys., 15, 253-272, 10.5194/acp-15-253-2015, 2015.
- Chen, C., Li, Z., Sun, J., Xu, W., Qiu, Y., Xie, C., He, Y., Li, Y., Li, J., Fu, P., Wang, Z., Worsnop, D. R., and Sun, Y.: Characterization of submicron aerosol in North China Plain (NCP), in preparation, 2021.
- Dai, Q., Schulze, B. C., Bi, X., Bui, A. A. T., Guo, F., Wallace, H. W., Sanchez, N. P., Flynn, J. H., Lefer, B. L., Feng, Y., and Griffin, R. J.: Seasonal differences in formation processes of oxidized organic aerosol near Houston, TX, Atmos. Chem. Phys., 19, 9641-9661, 10.5194/acp-19-9641-2019, 2019.
- Day, D. A., Wooldridge, P. J., Dillon, M. B., Thornton, J. A., and Cohen, R. C.: A thermal dissociation laser-induced fluorescence instrument for in situ detection of NO2, peroxy nitrates, alkyl nitrates, and HNO3, Journal of Geophysical Research: Atmospheres, 107, ACH 4-1-ACH 4-14, 10.1029/2001jd000779, 2002.
- Farmer, D. K., Matsunaga, A., Docherty, K. S., Surratt, J. D., Seinfeld, J. H., Ziemann, P. J., and Jimenez, J. L.: Response of an aerosol mass spectrometer to organonitrates and organosulfates and implications for atmospheric chemistry, Proc. Natl. Acad. Sci. U.S.A., 107, 6670-6675, doi:10.1073/pnas.0912340107, 2010.
 - Fry, J. L., Kiendler-Scharr, A., Rollins, A. W., Wooldridge, P. J., Brown, S. S., Fuchs, H., Dubé, W., Mensah, A., dal Maso, M., Tillmann, R., Dorn, H. P., Brauers, T., and Cohen, R. C.: Organic nitrate and secondary organic aerosol yield from NO3 oxidation of β-pinene evaluated using a gas-phase kinetics/aerosol partitioning model, Atmos. Chem. Phys., 9, 1431-1449, 10.5194/acp-9-1431-2009, 2009.
 - Fry, J. L., Draper, D. C., Zarzana, K. J., Campuzano-Jost, P., Day, D. A., Jimenez, J. L., Brown, S. S., Cohen, R. C., Kaser, L., Hansel, A., Cappellin, L., Karl, T., Hodzic Roux, A., Turnipseed, A., Cantrell, C., Lefer, B. L., and Grossberg, N.: Observations of gas- and aerosol-phase organic nitrates at BEACHON-RoMBAS 2011, Atmos. Chem. Phys., 13, 8585-8605, 10.5194/acp-13-8585-2013, 2013.
- Huffman, J. A., Ziemann, P. J., Jayne, J. T., Worsnop, D. R., and Jimenez, J. L.: Development and Characterization of a Fast-Stepping/Scanning Thermodenuder for Chemically-Resolved Aerosol Volatility Measurements, Aerosol Sci. Tech., 42, 395 407, 2008.
 - Huffman, J. A., Docherty, K. S., Aiken, A. C., Cubison, M. J., Ulbrich, I. M., DeCarlo, P. F., Sueper, D., Jayne, J. T., Worsnop, D. R., Ziemann, P. J., and Jimenez, J. L.: Chemically-resolved aerosol volatility measurements from two megacity field studies, Atmos. Chem. Phys., 9, 7161-7182, 2009.
 - Jayne, J. T., Leard, D. C., Zhang, X., Davidovits, P., Smith, K. A., Kolb, C. E., and Worsnop, D. R.: Development of an aerosol

- mass spectrometer for size and composition analysis of submicron particles, Aerosol Sci. Tech., 33, 49-70, 2000.
- Jordan, C., Ziemann, P., Griffin, R., Lim, Y., Atkinson, R., and Arey, J.: Modeling SOA formation from OH reactions with C8–C17 n-alkanes, Atmos. Environ., 42, 8015-8026, 2008.
- Kiendler-Scharr, A., Mensah, A. A., Friese, E., Topping, D., Nemitz, E., Prevot, A. S. H., Aijala, M., Allan, J., Canonaco, F., Canagaratna, M., Carbone, S., Crippa, M., Dall Osto, M., Day, D. A., De Carlo, P., Di Marco, C. F., Elbern, H., Eriksson, A., Freney, E., Hao, L., Herrmann, H., Hildebrandt, L., Hillamo, R., Jimenez, J. L., Laaksonen, A., McFiggans, G., Mohr, C., O'Dowd, C., Otjes, R., Ovadnevaite, J., Pandis, S. N., Poulain, L., Schlag, P., Sellegri, K., Swietlicki, E., Tiitta, P., Vermeulen, A., Wahner, A., Worsnop, D., and Wu, H. C.: Ubiquity of organic nitrates from nighttime chemistry in the European submicron aerosol, Geophys. Res. Lett., 43, 7735-7744, 10.1002/2016gl069239, 2016.

15

25

- 10 Kortelainen, A., Hao, L. Q., Tiitta, P., Jaatinen, A., Miettinen, P., Kulmala, M., Smith, J. N., Laaksonen, A., Worsnop, D. R., and Virtanen, A.: Sources of particulate organic nitrates in the boreal forest in Finland, Boreal Environment Research, 22, 13-26, 2017.
 - Kuang, Y., He, Y., Xu, W., Yuan, B., Zhang, G., Ma, Z., Wu, C., Wang, C., Wang, S., Zhang, S., Tao, J., Ma, N., Su, H., Cheng, Y., Shao, M., and Sun, Y.: Photochemical Aqueous-Phase Reactions Induce Rapid Daytime Formation of Oxygenated Organic Aerosol on the North China Plain, Environ. Sci. Technol., 10.1021/acs.est.9b06836, 2020.
 - Lee, A. K. Y., Adam, M. G., Liggio, J., Li, S. M., Li, K., Willis, M. D., Abbatt, J. P. D., Tokarek, T. W., Odame-Ankrah, C. A., Osthoff, H. D., Strawbridge, K., and Brook, J. R.: A large contribution of anthropogenic organo-nitrates to secondary organic aerosol in the Alberta oil sands, Atmos. Chem. Phys., 19, 12209-12219, 10.5194/acp-19-12209-2019, 2019.
- Lee, B. H., Lopez-Hilfiker, F. D., Mohr, C., Kurten, T., Worsnop, D. R., and Thornton, J. A.: An iodide-adduct high-resolution time-of-flight chemical-ionization mass spectrometer: application to atmospheric inorganic and organic compounds, Environ. Sci. Technol., 48, 6309-6317, 10.1021/es500362a, 2014.
 - Lee, B. H., Mohr, C., Lopez-Hilfiker, F. D., Lutz, A., Hallquist, M., Lee, L., Romer, P., Cohen, R. C., Iyer, S., Kurtén, T., Hu, W., Day, D. A., Campuzano-Jost, P., Jimenez, J. L., Xu, L., Ng, N. L., Guo, H., Weber, R. J., Wild, R. J., Brown, S. S., Koss, A., de Gouw, J., Olson, K., Goldstein, A. H., Seco, R., Kim, S., McAvey, K., Shepson, P. B., Starn, T., Baumann, K., Edgerton, E. S., Liu, J., Shilling, J. E., Miller, D. O., Brune, W., Schobesberger, S., D'Ambro, E. L., and Thornton, J. A.: Highly functionalized organic nitrates in the southeast United States: Contribution to secondary organic aerosol and reactive nitrogen budgets, Proc. Natl. Acad. Sci. U.S.A., 113, 1516-1521, 10.1073/pnas.1508108113, 2016.
 - Lee, L., Wooldridge, P. J., deGouw, J., Brown, S. S., Bates, T. S., Quinn, P. K., and Cohen, R. C.: Particulate organic nitrates observed in an oil and natural gas production region during wintertime, Atmos. Chem. Phys., 15, 9313-9325, 10.5194/acp-15-9313-2015, 2015.
 - Li, K., Liggio, J., Han, C., Liu, Q. F., Moussa, S. G., Lee, P., and Li, S. M.: Understanding the Impact of High-NOx Conditions on the Formation of Secondary Organic Aerosol in the Photooxidation of Oil Sand-Related Precursors, Environ. Sci. Technol., 53, 14420-14429, 10.1021/acs.est.9b05404, 2019.
- Li, W., Liu, L., Zhang, J., Xu, L., Wang, Y., Sun, Y., and Shi, Z.: Microscopic Evidence for Phase Separation of Organic Species and Inorganic Salts in Fine Ambient Aerosol Particles, Environ. Sci. Technol., 55, 2234-2242, 10.1021/acs.est.0c02333, 2021.

- Lim, Y. B., and Ziemann, P. J.: Chemistry of secondary organic aerosol formation from OH radical-initiated reactions of linear, branched, and cyclic alkanes in the presence of NO x, Aerosol Sci. Tech., 43, 604-619, 2009.
- Liu, L., Zhang, J., Zhang, Y., Wang, Y., Xu, L., Yuan, Q., Liu, D., Sun, Y., Fu, P., Shi, Z., and Li, W.: Persistent residential burning-related primary organic particles during wintertime hazes in North China: insights into their aging and optical changes, Atmos. Chem. Phys., 21, 2251-2265, 10.5194/acp-21-2251-2021, 2021.

10

- Liu, S., Shilling, J. E., Song, C., Hiranuma, N., Zaveri, R. A., and Russell, L. M.: Hydrolysis of Organonitrate Functional Groups in Aerosol Particles, Aerosol Sci. Tech., 46, 1359-1369, 10.1080/02786826.2012.716175, 2012.
- Liu, Y., Song, M., Liu, X., Zhang, Y., Hui, L., Kong, L., Zhang, Y., Zhang, C., Qu, Y., An, J., Ma, D., Tan, Q., and Feng, M.: Characterization and sources of volatile organic compounds (VOCs) and their related changes during ozone pollution days in 2016 in Beijing, China, Environ. Pollut., 257, 113599, https://doi.org/10.1016/j.envpol.2019.113599, 2020.
- Lopez-Hilfiker, F. D., Mohr, C., Ehn, M., Rubach, F., Kleist, E., Wildt, J., Mentel, T. F., Lutz, A., Hallquist, M., Worsnop, D., and Thornton, J. A.: A novel method for online analysis of gas and particle composition: description and evaluation of a Filter Inlet for Gases and AEROsols (FIGAERO), Atmospheric Measurement Techniques, 7, 983-1001, 10.5194/amt-7-983-2014, 2014.
- Matsunaga, A., and Ziemann, P. J.: Yields of β-hydroxynitrates, dihydroxynitrates, and trihydroxynitrates formed from OH radical-initiated reactions of 2-methyl-1-alkenes, Proc. Natl. Acad. Sci. U.S.A., 107, 6664-6669, 2010.
 - Middlebrook, A. M., Bahreini, R., Jimenez, J. L., and Canagaratna, M. R.: Evaluation of Composition-Dependent Collection Efficiencies for the Aerodyne Aerosol Mass Spectrometer using Field Data, Aerosol Sci. Tech., 46, 258-271, 10.1080/02786826.2011.620041, 2012.
- Ng, N. L., Brown, S. S., Archibald, A. T., Atlas, E., Cohen, R. C., Crowley, J. N., Day, D. A., Donahue, N. M., Fry, J. L., Fuchs, H., Griffin, R. J., Guzman, M. I., Herrmann, H., Hodzic, A., Iinuma, Y., Jimenez, J. L., Kiendler-Scharr, A., Lee, B. H., Luecken, D. J., Mao, J., McLaren, R., Mutzel, A., Osthoff, H. D., Ouyang, B., Picquet-Varrault, B., Platt, U., Pye, H. O. T., Rudich, Y., Schwantes, R. H., Shiraiwa, M., Stutz, J., Thornton, J. A., Tilgner, A., Williams, B. J., and Zaveri, R. A.: Nitrate radicals and biogenic volatile organic compounds: oxidation, mechanisms, and organic aerosol, Atmos. Chem. Phys., 17, 2103-2162, 10.5194/acp-17-2103-2017, 2017.
 - O'Brien, J. M., Shepson, P. B., Muthuramu, K., Hao, C., Niki, H., Hastie, D. R., Taylor, R., and Roussel, P. B.: Measurements of alkyl and multifunctional organic nitrates at a rural site in Ontario, Journal of Geophysical Research: Atmospheres, 100, 22795-22804, 10.1029/94jd03247, 1995.
 - Perring, A., Pusede, S., and Cohen, R.: An observational perspective on the atmospheric impacts of alkyl and multifunctional nitrates on ozone and secondary organic aerosol, Chem Rev, 113, 5848-5870, 2013.
 - Present, P. S. R., Zare, A., and Cohen, R. C.: The changing role of organic nitrates in the removal and transport of NOx, Atmos. Chem. Phys., 20, 267-279, 10.5194/acp-20-267-2020, 2020.
 - Qiao, Z., Li-Ming, C., Meng-Xue, T., Xiao-Feng, H., Eri, S., and He, L. Y.: Characterization of organic aerosol at a rural site in North China Plain: sources, volatility and organonitrates, 2020.
- Ramasamy, S., Nakayama, T., Imamura, T., Morino, Y., Kajii, Y., and Sato, K.: Investigation of dark condition nitrate radicaland ozone-initiated aging of toluene secondary organic aerosol: Importance of nitrate radical reactions with phenolic

products, Atmos. Environ., 219, 9, 10.1016/j.atmosenv.2019.117049, 2019.

5

15

20

25

- Rindelaub, J. D., McAvey, K. M., and Shepson, P. B.: The photochemical production of organic nitrates from alpha-pinene and loss via acid-dependent particle phase hydrolysis, Atmos. Environ., 100, 193-201, 10.1016/j.atmosenv.2014.11.010, 2015.
- Sadanaga, Y., Takaji, R., Ishiyama, A., Nakajima, K., Matsuki, A., and Bandow, H.: Thermal dissociation cavity attenuated phase shift spectroscopy for continuous measurement of total peroxy and organic nitrates in the clean atmosphere, Rev. Sci. Instrum., 87, 074102, 10.1063/1.4958167, 2016.
- Sheng, J., Wang, F., Li, X., Ma, Z., Liu, Q., Zhao, D., Huang, M., and Ding, D.: The temporal variation and photochemical characters of VOCs in summer and winter of Beijing based on PTR-TOF-MS, Huanjing Huaxue-Environmental Chemistry, 38, 1590-1599, 10.7524/j.issn.0254-6108.2018092601, 2019.
- Sun, Y., He, Y., Kuang, Y., Xu, W., Song, S., Ma, N., Tao, J., Cheng, P., Wu, C., Su, H., Cheng, Y., Xie, C., Chen, C., Lei, L., Qiu, Y., Fu, P., Croteau, P., and Worsnop, D. R.: Chemical Differences Between PM1 and PM2.5 in Highly Polluted Environment and Implications in Air Pollution Studies, Geophys. Res. Lett., 47, e2019GL086288, 10.1029/2019gl086288, 2020.
 - Sun, Y. L., Zhang, Q., Schwab, J. J., Yang, T., Ng, N. L., and Demerjian, K. L.: Factor analysis of combined organic and inorganic aerosol mass spectra from high resolution aerosol mass spectrometer measurements, Atmos. Chem. Phys., 12, 8537-8551, 10.5194/acp-12-8537-2012, 2012.
 - Sun, Y. L., Wang, Z. F., Fu, P. Q., Yang, T., Jiang, Q., Dong, H. B., Li, J., and Jia, J. J.: Aerosol composition, sources and processes during wintertime in Beijing, China, Atmos. Chem. Phys., 13, 4577-4592, 10.5194/acp-13-4577-2013, 2013.
 - Takeuchi, M., and Ng, N. L.: Chemical composition and hydrolysis of organic nitrate aerosol formed from hydroxyl and nitrate radical oxidation of α-pinene and β-pinene, Atmos. Chem. Phys., 19, 12749-12766, 10.5194/acp-19-12749-2019, 2019.
 - Thieser, J., Schuster, G., Schuladen, J., Phillips, G. J., Reiffs, A., Parchatka, U., Pöhler, D., Lelieveld, J., and Crowley, J. N.: A two-channel thermal dissociation cavity ring-down spectrometer for the detection of ambient NO2, RO2NO2 and RONO2, Atmos. Meas. Tech., 9, 553-576, 10.5194/amt-9-553-2016, 2016.
 - Xu, L., Guo, H. Y., Boyd, C. M., Klein, M., Bougiatioti, A., Cerully, K. M., Hite, J. R., Isaacman-VanWertz, G., Kreisberg, N. M., Knote, C., Olson, K., Koss, A., Goldstein, A. H., Hering, S. V., de Gouw, J., Baumann, K., Lee, S. H., Nenes, A., Weber, R. J., and Ng, N. L.: Effects of anthropogenic emissions on aerosol formation from isoprene and monoterpenes in the southeastern United States Proc. Natl. Acad. Sci. U. S. A., 112, E4509-E4509, 10.1073/pnas.1512279112, 2015a.
 - Xu, L., Suresh, S., Guo, H., Weber, R. J., and Ng, N. L.: Aerosol characterization over the southeastern United States using high-resolution aerosol mass spectrometry: spatial and seasonal variation of aerosol composition and sources with a focus on organic nitrates, Atmos. Chem. Phys., 15, 7307-7336, 10.5194/acp-15-7307-2015, 2015b.
 - Xu, L., Williams, L. R., Young, D. E., Allan, J. D., Coe, H., Massoli, P., Fortner, E., Chhabra, P., Herndon, S., Brooks, W. A., Jayne, J. T., Worsnop, D. R., Aiken, A. C., Liu, S., Gorkowski, K., Dubey, M. K., Fleming, Z. L., Visser, S., Prevot, A. S. H., and Ng, N. L.: Wintertime aerosol chemical composition, volatility, and spatial variability in the greater London area, Atmos. Chem. Phys., 16, 1139-1160, 10.5194/acp-16-1139-2016, 2016.
- 35 Xu, W., Han, T., Du, W., Wang, Q., Chen, C., Zhao, J., Zhang, Y., Li, J., Fu, P., Wang, Z., Worsnop, D. R., and Sun, Y.: Effects of Aqueous-Phase and Photochemical Processing on Secondary Organic Aerosol Formation and Evolution in Beijing,

China, Environ. Sci. Technol., 51, 762-770, 10.1021/acs.est.6b04498, 2017a.

10

15

- Xu, W., Sun, Y., Wang, Q., Du, W., Zhao, J., Ge, X., Han, T., Zhang, Y., Zhou, W., Li, J., Fu, P., Wang, Z., and Worsnop, D. R.: Seasonal Characterization of Organic Nitrogen in Atmospheric Aerosols Using High Resolution Aerosol Mass Spectrometry in Beijing, China, ACS Earth and Space Chemistry, 10.1021/acsearthspacechem.7b00106, 2017b.
- 5 Xu, W., Chen, C., Qiu, Y., Li, Y., Zhang, Z., Karnezi, E., Pandis, S. N., Xie, C., Li, Z., Sun, J., Ma, N., Xu, W., Fu, P., Wang, Z., Zhu, J., Worsnop, D. R., Ng, N. L., and Sun, Y.: Organic aerosol volatility and viscosity in North China Plain: Contrast between summer and winter, Atmos. Chem. Phys. Discuss., 2020, 1-22, 10.5194/acp-2020-1105, 2020a.
 - Xu, W., He, Y., Qiu, Y., Chen, C., Xie, C., Lei, L., Li, Z., Sun, J., Li, J., Fu, P., Wang, Z., Worsnop, D. R., and Sun, Y.: Mass spectral characterization of primary emissions and implications in source apportionment of organic aerosol, Atmos. Meas. Tech., 13, 3205-3219, 10.5194/amt-13-3205-2020, 2020b.
 - Xu, W. Q., Xie, C. H., Karnezi, E., Zhang, Q., Wang, J. F., Pandis, S. N., Ge, X. L., Zhang, J. W., An, J. L., Wang, Q. Q., Zhao, J., Du, W., Qiu, Y. M., Zhou, W., He, Y., Li, Y., Li, J., Fu, P. Q., Wang, Z. F., Worsnop, D. R., and Sun, Y. L.: Summertime aerosol volatility measurements in Beijing, China, Atmos. Chem. Phys., 19, 12, 10.5194/acp-19-10205-2019, 2019.
 - Yu, K., Zhu, Q., Du, K., and Huang, X. F.: Characterization of nighttime formation of particulate organic nitrates based on high-resolution aerosol mass spectrometry in an urban atmosphere in China, Atmos. Chem. Phys., 19, 5235-5249, 10.5194/acp-19-5235-2019, 2019.
 - Zhang, J. K., Cheng, M. T., Ji, D. S., Liu, Z. R., Hu, B., Sun, Y., and Wang, Y. S.: Characterization of submicron particles during biomass burning and coal combustion periods in Beijing, China, Sci. Total Environ., 562, 812-821, 10.1016/j.scitotenv.2016.04.015, 2016.
- Zhou, W., Xu, W., Kim, H., Zhang, Q., Fu, P., Worsnop, D. R., and Sun, Y.: A review of aerosol chemistry in Asia: insights from aerosol mass spectrometer measurements, Environmental science. Processes & impacts, 22, 1616-1653, 10.1039/d0em00212g, 2020.
 - Zhu, Q., He, L.-Y., Huang, X.-F., Cao, L.-M., Gong, Z.-H., Wang, C., Zhuang, X., and Hu, M.: Atmospheric aerosol compositions and sources at two national background sites in northern and southern China, Atmos. Chem. Phys., 16, 10283-10297, 10.5194/acp-16-10283-2016, 2016.

Table 1. A summary of average RoN,cale during three campaigns

	Beijing _{Summer}	Beijingwinter	Guchengwinter
NR-PM ₁ < 20 μ g m ⁻³	7.0	2.8	6.0
$NR\text{-}PM_1 > 50~\mu g~m^{-3}$	3.4	1.2	3.5
Average	5.7	2.2	4.3

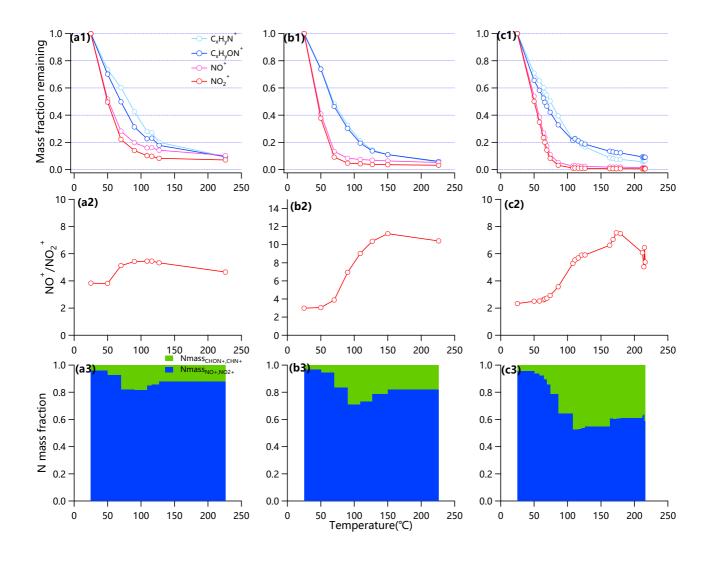


Figure 1. Thermograms of $C_xH_yN^+$, $C_xH_yON^+$, NO^+ and NO_2^+ in (a) summer in Beijing, (b) winter in Beijing and (c) winter in Gucheng. Variations of fraction of N mass from N-containing ions, ratio of NO^+ and NO_2^+ as a function of temperature are also shown.

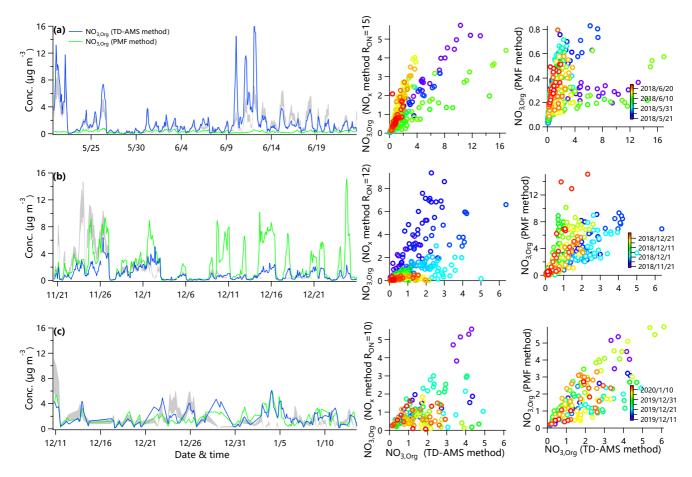


Figure 2. Intercomparisons of $NO_{3,org}$ mass loadings using three different methods in (a) summer in Beijing, (b) winter in Beijing and (c) Gucheng. The top and bottom of shaded areas in time series are the upper and lower bounds of pON loadings from "NO_x method".

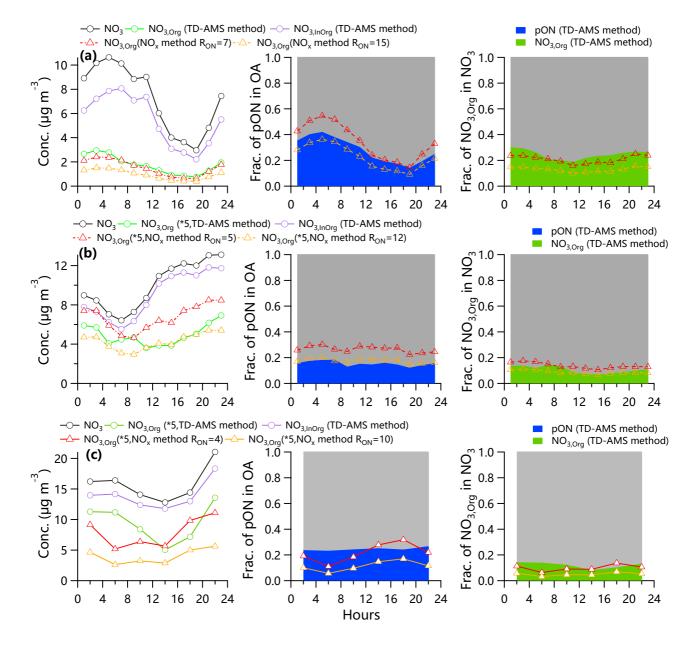


Figure 3. Diurnal profiles of NO_{3,org} loadings, fraction of pON in OA, and fraction of NO_{3,org} in NO₃ calculated by "TD-AMS method" and "NO_x method" in (a) summer in Beijing, (b) winter in Beijing and (c) Gucheng. Because there are fewer data in Gucheng due to more TD temperature settings, the diurnal profile shows 4-hour averaged data.

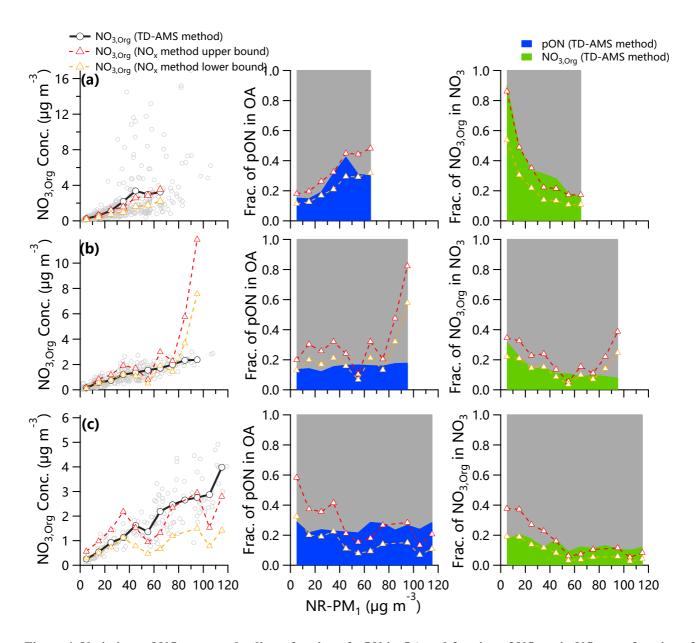


Figure 4. Variations of $NO_{3,org}$ mass loadings, fraction of pON in OA and fraction of $NO_{3,org}$ in NO_3 as a function of particulate matter (PM) loadings in (a) summer in Beijing, and (b) winter in Beijing and (c) Gucheng. Because there are fewer data points in Gucheng due to more TD temperature settings, the chemically resolved PM pollution is 20 μg m⁻³ average.

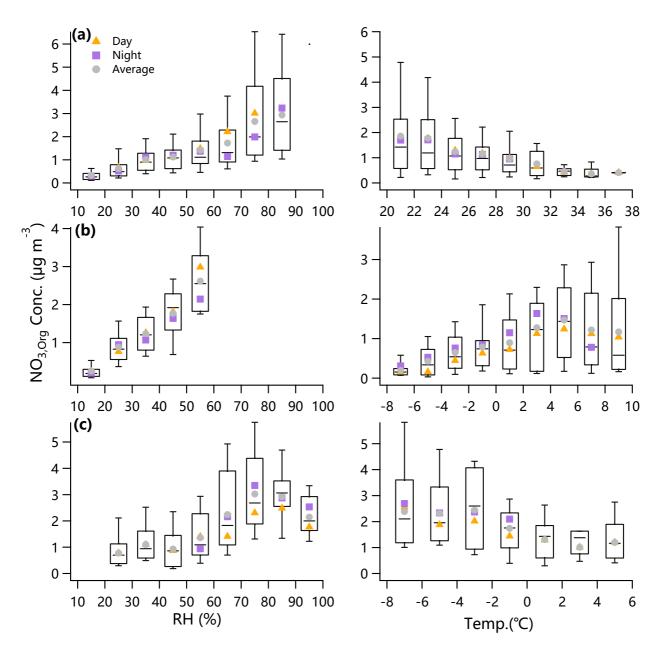


Figure 5. Variations of NO_{3,org} mass loadings as a function of RH and temperature in (a) summer in Beijing, (b)winter in Beijing and (c) Gucheng calculated by "TD-AMS method". The mid-point line, lower and upper boxes, lower and upper whiskers refer to median, 25th percentiles, 75th percentiles, 10th percentiles, and 90th percentiles, respectively for the entire study.