# Comparison of airborne measurements of NO, NO $_2$ , HONO, NO $_y$ and CO during FIREX-AQ

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#### **Abstract**

1 We present a comparison of fast-response instruments installed onboard the NASA DC-8 2 aircraft that measured nitrogen oxides (NO and NO<sub>2</sub>), nitrous acid (HONO), total reactive odd nitrogen (measured both as the total (NO<sub>v</sub>) and from the sum of individually measured 3 4 species  $(\Sigma NO_v)$ ) and carbon monoxide (CO) in the troposphere during the 2019 Fire Influence on Regional to Global Environments and Air Quality (FIREX-AQ) campaign. By 5 targeting smoke from summertime wildfires, prescribed fires and agricultural burns across 6 the continental United States, FIREX-AQ provided a unique opportunity to investigate 7 8 measurement accuracy in concentrated plumes where hundreds of species coexist. Here, we 9 compare NO measurements by chemiluminescence (CL) and laser induced fluorescence (LIF); NO<sub>2</sub> measurements by CL, LIF and cavity enhanced spectroscopy (CES); HONO 10 11 measurements by CES and iodide-adduct chemical ionization mass spectrometry (CIMS); 12 and CO measurements by tunable diode laser absorption spectrometry (TDLAS) and 13 integrated cavity output spectroscopy (ICOS). Additionally, total NO<sub>v</sub> measurements using the CL instrument were compared with  $\Sigma NO_v$  (= NO + NO<sub>2</sub> + HONO + nitric acid (HNO<sub>3</sub>) + 14 acyl peroxy nitrates (APNs) + submicron particulate nitrate (pNO<sub>3</sub>)). Other NO<sub>v</sub> species were 15 not included in  $\Sigma NO_v$  as they either contributed minimally to it (e.g.,  $C_1$ – $C_5$  alkyl nitrates, 16 nitryl chloride (ClNO<sub>2</sub>), dinitrogen pentoxide (N<sub>2</sub>O<sub>5</sub>)) or were not measured during FIREX-17 18 AQ (e.g., higher oxidized alkyl nitrates, nitrate (NO<sub>3</sub>), non-acyl peroxynitrates, coarse mode 19 aerosol nitrate). The aircraft instrument intercomparisons demonstrate the following: 1) NO 20 measurements by CL and LIF agreed well within instrument uncertainties, but with 21 potentially reduced time response for the CL instrument; 2) NO<sub>2</sub> measurements by LIF and 22 CES agreed well within instrument uncertainties, but CL NO<sub>2</sub> was on average 10% higher; 3) 23 CES and CIMS HONO measurements were highly correlated in each fire plume transect, but 24 the correlation slope of CES vs. CIMS for all 1 Hz data during FIREX-AQ was 1.8, which 25 we attribute to a reduction in the CIMS sensitivity to HONO in high temperature 26 environments; 4) NO<sub>v</sub> budget closure was demonstrated for all flights within the combined 27 instrument uncertainties of 25%. However, we used a fluid dynamic flow model to estimate 28 that average pNO<sub>3</sub> sampling fraction through the NO<sub>v</sub> inlet in smoke was variable from one 29 flight to another and ranged between 0.36 and 0.99, meaning that approximately 0–24% on 30 average of the total measured NO<sub>v</sub> in smoke may have been unaccounted for and may be due 31 to unmeasured species such as organic nitrates; 5) CO measurements by ICOS and TDLAS 32 agreed well within combined instrument uncertainties, but with a systematic offset that 33 averaged 2.87 ppby; and 6) integrating smoke plumes followed by fitting the integrated 34 values of each plume improved the correlation between independent measurements.

#### 1. Introduction

36 Biomass burning (BB) can take multiple forms (e.g., wildfires, prescribed fires, agricultural 37 burns, grass fires, peat fires) and accounts for a large fraction of global carbon emissions with consequences for climate (Bowman et al., 2009; van der Werf et al., 2010, 2017) and 38 39 biogeochemical cycles (Crutzen & Andreae, 2016). BB also contributes substantially to the 40 atmospheric burden of trace gases and aerosols (Andreae, 2019), causing poor air quality on 41 regional to continental scales (Jaffe et al., 2020; O'Dell et al., 2019; Wotawa, 2000) and 42 posing a major threat to public health (Johnston et al., 2012, 2021). In the United States (US), wildfires mainly occur in the western states and in Alaska and burned over 4.5 million acres 43 44 in 2019 (US National Interagency Fire Center, https://www.nifc.gov/fire-information). Wildfires frequency and severity are predicted to increase in response to a warmer, drier 45 climate (Burke et al., 2021; Westerling, 2016) and also to increasing human-caused ignition 46 47 (Balch et al., 2017). In comparison, prescribed fires, which are common practice in the 48 southeastern US, burned an estimated 10 million acres in 2019, to which agricultural burns 49 added another 2-3 million acres (Melvin, 2020). While agricultural burns are usually smaller 50 and less intense than wildfires or prescribed fires, they occur more frequently and throughout 51 the whole year, and can significantly impact local air quality (Dennis et al., 2002; McCarty, 52 2011).

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Rising interest in the impact of fires on climate and air quality over the past decades has resulted in a series of laboratory studies of BB emissions in the US such as the FLAME-4 experiment in 2012 (e.g., Stockwell et al., 2014) and the FIRELAB study in 2016 (e.g., Selimovic et al., 2018). Recent, large-scale field studies such as AMMA (e.g., Liousse et al., 2010), BBOP (e.g., Collier et al., 2016) and WE-CAN (e.g., Juncosa Calahorrano et al. (2021)) have been dedicated to sampling and characterizing emissions and atmospheric chemistry from fires. The focus of the joint National Oceanic and Atmospheric Administration (NOAA) / National Aeronautics and Space Administration (NASA) Fire Influence on Regional to Global Environments and Air Quality (FIREX-AQ) airborne campaign was to provide comprehensive observations to investigate the impact of summer time wildfires, prescribed fires and agricultural burns on air quality and climate across the

65 conterminous US (Warneke et al., 2022).

> Accurate measurements facilitate understanding of fire emissions, processing and impacts. In situ, fast-response measurements of trace gases in the atmosphere conducted from airborne platforms provide unique data sets that enhance our understanding of atmospheric composition and chemistry. One method for evaluating measurement accuracy is by comparison of independent measurements using different techniques. A relatively small body of literature reported comparisons of methods for in flight detection of tropospheric carbon monoxide (CO) and reactive odd nitrogen species measured both as the total (NO<sub>v</sub>) and from the sum of individually measured species ( $\Sigma NO_v$ ), and these studies have shown that such comparisons are valuable for identifying instrument artifacts and quantifying measurement uncertainties (Eisele et al., 2003; Gregory et al., 1990; Hoell et al., 1987; Hoell et al., 1987; Sparks et al., 2019). During FIREX-AQ, a large suite of airborne instruments, detailed in the following sections, performed independent in situ tropospheric measurements of one or more

fire-science relevant reactive nitrogen species and CO aboard the NASA DC-8 aircraft (Table

1). Additionally, FIREX-AQ provides a unique opportunity to investigate measurement accuracy in concentrated smoke plumes where hundreds of species coexist.

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98 99 Nitric oxide (NO) and nitrogen dioxide (NO<sub>2</sub>) are among the largest components of the reactive nitrogen budget emitted by biomass burning and are produced by the oxidation of reduced nitrogen species present in the fuel in the flaming stage of combustion (Roberts et al., 2020). NO<sub>x</sub>, defined as the sum of NO and NO<sub>2</sub>, directly affects atmospheric oxidation rates and ozone (O<sub>3</sub>) production within fire plumes (Robinson et al., 2021; Xu et al., 2021). It also contributes to the formation of secondary aerosols and N transport and deposition to ecosystems downwind (Galloway et al., 2003; Kroll & Seinfeld, 2008; Ziemann & Atkinson, 2012). Therefore, two independent NO and three independent NO<sub>2</sub> measurements were part of FIREX-AQ to provide continuous in situ observations, as described in section 2 below. Nitrous acid (HONO) is emitted directly to the atmosphere through various combustion processes including BB. The rapid production of OH from HONO at the early stage of smoke plume formation (Peng et al., 2020) results in rapid initiation of photochemistry, with a strong influence on downwind chemical evolution of smoke plumes (Bourgeois et al., 2021; Robinson et al., 2021; Theys et al., 2020). Total NO<sub>v</sub> can be measured through conversion of individual species to NO (Fahey et al., 1985). It is a more conserved tracer for NO<sub>x</sub> emissions than NO<sub>x</sub> itself since it accounts for NO<sub>x</sub> oxidation products, and it provides a mean to assess from a mass-balance approach the accuracy of  $\Sigma NO_v$  budget closure (Bollinger et al., 1983; Fahey et al., 1986; Williams et al., 1997). Fahey et al. (1986) define  $\Sigma NO_v$  as the sum of

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 $\Sigma NO_y = NO + NO_2 + nitric acid (HNO_3) + HONO + peroxynitric acid (HO_2NO_2) + nitrate (NO_3) + dinitrogen pentoxide (2×N_2O_5) + peroxyacetyl nitrate (PAN) + particulate nitrate (pNO_3) + ... (Eq. 1)$ 

important nitrogen species as illustrated by Eq. 1:

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Other nitrogen compounds that can contribute to  $\Sigma NO_y$  include alkyl nitrates (Day et al., 2003), acyl peroxynitrates (APNs; Juncosa Calahorrano et al., 2021), non-acyl peroxynitrates (RO<sub>2</sub>NO<sub>2</sub>; Murphy et al., 2004), nitryl chloride (ClNO<sub>2</sub>; Kenagy et al., 2018), nitro compounds and nitroaromatics (Decker et al., 2021).

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Carbon monoxide (CO) is emitted from incomplete combustion in fires and other sources, and is especially important for characterizing the combustion stage of fires (i.e., flaming vs. smoldering) through the use of the modified combustion efficiency (Yokelson et al., 1996). Due to its relatively long chemical lifetime, CO is commonly used as a conserved tracer to account for dilution with ambient air as smoke plumes are transported downwind, and accurate CO measurements are necessary to better constrain emission factors (EFs) used in emission inventories.

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This study builds on past airborne instrument comparisons and extends these analyses to a new species (HONO), new measurement techniques (first airborne deployment of the NOAA NO-LIF (laser induced fluorescence) and the NOAA CO-ICOS (integrated cavity output spectroscopy) instruments) and new environments (concentrated fire smoke). In this paper we present a comparison of NO, NO<sub>2</sub>, HONO, NO<sub>y</sub> and CO measurements, which are compounds of major interest for fire-related science, air quality and climate. In the first part of this paper, we describe the FIREX-AQ campaign, the deployed instruments and the methodology used to perform the comparisons. In the second part, we provide a detailed instrument comparison for each species.

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## 2. FIREX-AQ overview and instruments

## 2.1 FIREX-AQ airborne mission

The FIREX-AQ campaign (<a href="https://www-air.larc.nasa.gov/missions/firex-aq/">https://www.esrl.noaa.gov/csl/projects/firex-aq/</a>) took place from July to September 2019. FIREX-AQ included the deployment of multiple aircraft and mobile platforms over the course of the campaign, however this study focuses on the heavily instrumented NASA DC-8 aircraft. The NASA DC-8 portion of the project achieved two flights over the Los Angeles (LA) Basin and the Central Valley in California, 13 flights originating from Boise, Idaho, and 7 flights based out of Salina, Kansas. The flights from Boise were conducted over the Western US to sample smoke from wildfires, while the flights from Salina focused on agricultural and prescribed burns (hereafter referred to as eastern fires) in the Southeastern US.

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Most wildfire flights were designed to sample background mixing ratios, fresh emissions, and aged smoke, whereas the eastern fire flights typically transected numerous fresh smoke plumes several times each. For wildfires, the NASA DC-8 first flew upwind of the fire to characterize ambient conditions unaffected by targeted fire emissions. Subsequent cross-wind plume transects were conducted as close as possible to the fire to sample the emissions with the least possible atmospheric ageing. Plume transects were designed to be perpendicular to the wind direction and through the center of the vertical extent of the plume, terrain permitting. The vertical structure of the plume was systematically assessed using a differential absorption lidar during a lengthwise overpass above the plume from end to start. The aircraft transected the smoke plume successively further downwind, at approximately 15–40 km intervals, to characterize smoke evolution in a "lawnmower" pattern (Figure 1a). For several wildfires, the DC-8 also executed flight transects along the plume axis, both toward and away from the fire source. Most eastern fires sampled during FIREX-AQ did not produce plumes large enough to enable regularly spaced plume transects. Most smoke plumes were therefore sampled repetitively at the same location, sometimes with varying altitude and/or approach angle (Figure 1b).

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## 2.2 Instruments

## 2.2.1 Chemiluminescence (NO/NO<sub>2</sub>/NO<sub>v</sub>)

The NOAA CL instrument has been frequently used for both ground-based and airborne measurements of NO, NO<sub>2</sub> and NO<sub>y</sub> and uses the CL detection of NO with O<sub>3</sub> added as reagent gas (Fontijn et al., 1970; Ridley & Grahek, 1990; Ridley & Howlett, 1974; Ryerson et al., 1999, 2000). NO, NO<sub>2</sub> and NO<sub>y</sub> are measured on three independent channels of the instrument. The NO channel measures NO, the NO<sub>2</sub> channel measures the sum of NO and

166 photolyzed NO<sub>2</sub> as NO, and the NO<sub>v</sub> channel measures the total reactive nitrogen oxides species reduced to NO. NO2 is determined from the difference between signals from the NO 167 and NO<sub>2</sub> channels. Ambient air is continuously sampled from a pressure-building ducted 168 169 aircraft inlet to the instrument at a typical flow of  $1045.1 \pm 0.2$ ,  $1030.2 \pm 0.2$  and  $1029.5 \pm$ 170 0.2 standard cubic centimeters per minute (sccm) in flight for NO, NO<sub>2</sub>, and NO<sub>y</sub>, 171 respectively. In the NO<sub>2</sub> channel, NO<sub>2</sub> is photolyzed to NO with a  $40 \pm 1$  % conversion efficiency using two ultraviolet (UV) LEDs (Hamamatsu, model L11921) at 385 nm in a 45 172 173 cm long quartz cell (inner diameter of 1.2 cm) pressure-controlled at 209.8  $\pm$  0.3 Torr. In the 174 NO channel, a similar quartz cell wrapped in aluminum foil to avoid NO<sub>2</sub> photolysis and pressure controlled at 209.7  $\pm$  0.3 Torr, ensures similar residence time of sampled air in both 175 176 channels. In the NO<sub>v</sub> channel, reactive odd nitrogen species are first sampled through an inlet 177 heated at  $90.0 \pm 0.1$  °C then catalytically reduced to NO on a gold tube surface heated at  $300.0 \pm 0.2$ °C in the presence of added pure CO flowing at  $3.19 \pm 0.01$  sccm. Approximately 178 179 5%  $O_3$  in oxygen is produced by corona discharge, delivered at  $73.80 \pm 0.02$  (NO channel),  $74.11 \pm 0.03$  (NO<sub>2</sub> channel), and  $74.60 \pm 0.04$  (NO<sub>2</sub> channel) sccm, and mixed with sampled 180 air in a pressure  $(8.65 \pm 0.02, 8.79 \pm 0.02, 8.56 \pm 0.02)$  Torr for NO, NO<sub>2</sub>, and NO<sub>y</sub> 181 182 respectively) and temperature  $(25.0 \pm 0.2, 25.1 \pm 0.2 \text{ and } 25.1 \pm 0.2 ^{\circ}\text{C} \text{ for NO}, \text{NO}_2 \text{ and NO}_v,$ 183 respectively) controlled reaction vessel. O<sub>3</sub>-induced CL is detected with a red-sensitive 184 photomultiplier tube controlled at -78°C with dry ice, and the amplified digitized signal is 185 recorded using an 80 MHz counter. Pulse coincidence at high count rates was calculated after the mission by fitting an inverse function to the curve between observed and theoretical count 186 187 rates for known NO mixing ratios ranging from ppbv to ppmv levels. Instrument calibrations were routinely performed both on the ground and during flight by standard addition of NO 188 189 from a gravimetrically determined NO in N<sub>2</sub> mixture (1.38  $\pm$  0.03 ppmv) delivered at 4.04  $\pm$ 190 0.02 (NO channel),  $4.84 \pm 0.02$  (NO<sub>2</sub> channel), and  $4.96 \pm 0.02$  (NO<sub>3</sub> channel) sccm. All 191 measurements were taken at a temporal resolution of 0.1 second (s), averaged to 1 s, and 192 corrected for the dependence of instrument sensitivity on ambient water vapor content (Ridley et al., 1992). Finally, NO<sub>2</sub> data were further corrected for a HONO interference (5% 193 194 of the HONO mixing ratios) due to HONO photolysis at 385 nm quantified from theoretical 195 calculation and confirmed in the laboratory using a HONO source described in Lao et al. 196 (2020). Under these conditions the total estimated 1Hz uncertainty at sea level was  $\pm$  (4 % + 197 6 pptv),  $\pm$  (7 % + 20 pptv), and  $\pm$  (12 % + 15 pptv) for NO, NO<sub>2</sub>, and NO<sub>v</sub>, respectively.

## 2.2.2 Laser Induced Florescence (NO)

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The NOAA NO-LIF measurements were performed using a custom-built laser-induced fluorescence instrument as detailed in Rollins et al. (2020). Air was continuously sampled from outside the aircraft through an optical cell in the DC-8 cabin held to near 90 hPa. The instrument utilizes a fiber laser system with a narrow-band laser tuned to a rotationally resolved NO spectral feature near 215 nm. Rapid dithering on and off of this resonance achieves 0.1 s measurements with a continuously monitored background to reduce uncertainty in the instrument zero. The laser induced excitation of NO is followed by redshifted fluorescence which is detected by a photomultiplier tube operated in single-photon counting mode. The laser is directed through both a sampling and reference cell in a single pass for continuous monitoring of any changes in the instrument sensitivity due to changes in

the laser spectrum, or pressure of the optical cells. 500 ppbv of NO in air was flown at 50 sccm through the reference cell to ensure that measurements are occurring with the laser tuned to the peak online wavelength. A constant flow of approximately 2500 sccm is maintained within the sampling cell through the use of a custom inlet valve (Gao et al., 1999) and the exhaust of both cells are tied together allowing for any changes in sensitivity due to pressure fluctuations to be accounted for during data reduction. Hourly calibrations were performed during each flight in which 2–10 sccm of 5 ppmv NO in N<sub>2</sub> mixture was added to the sample flow resulting in mixing ratios of 4–20 ppbv. As discussed in Rollins et al (2020), given the sensitivity typically observed during FIREX, nonlinearity associated with saturation of the LIF instrument is not problematic until mixing ratios well above 100 ppbv are encountered. The sensitivity of the instrument was determined using the in-flight calibrations to be typically 10 counts per second (CPS) pptv<sup>-1</sup> with 10 CPS background achieving a detection limit of 1 pptv for 1 s integration. The uncertainty of the instrument sensitivity is  $\pm$ 6–9%. The effect of water vapor, which reduces the sensitivity by quenching of the electronically excited NO, was accounted for during data reduction using water vapor measurements provided by an ICOS instrument on the DC-8. The NO measurement uncertainty is estimated to be  $\pm$  (8% + 1 pptv).

2.2.3 Laser Induced Fluorescence (NO<sub>2</sub>)

The NASA Compact Airborne  $NO_2$  Experiment (CANOE) measured  $NO_2$  using non-resonant LIF. The instrument is a modified version of a formaldehyde (HCHO) instrument (St. Clair et al., 2019) with the excitation wavelength changed to 532 nm. The technique utilizes the pulsed (80 kHz) output of a fixed wavelength, 2W, 532 nm laser to excite  $NO_2$  molecules and detects the resulting fluorescence with two identical detection axes consisting of a photomultiplier tube (PMT) and optical filters that transmit > 695 nm. Delayed time gate PMT counts are recorded at 10 Hz and a laboratory calibration, along with an intercept determined by preflight zeroing, are used to provide 1Hz  $NO_2$  data. The  $NO_2$  measurement uncertainty is estimated to be  $\pm$  (10% + 100 pptv).

During FIREX-AQ, ambient air was sampled using a shared inlet that provided a large (10–25 standard liter per minute (slpm)) bypass flow to the instrument rack. The inlet tube is a 45 cm length of 0.94 cm inner diameter Silcosteel (Restek) coated with FluoroPel (Cazorla et al., 2015). The CANOE instrument pulled its 750 sccm sample flow from a shared manifold (with another four instruments) at the instrument rack. An inline particle filter on the sample line prevented laser scatter by fine aerosol that were not removed by the particle-rejecting inlet. A manual three-way valve outside the instrument was used to sample from a scrubber (Drierite/molecular sieve) and provides a zero before and periodically during the flight. Pressure in the CANOE detection cell was maintained at 40 Torr by a pressure controller that precedes the cell in the flow path.

#### 2.2.4 Cavity Enhanced Spectroscopy (NO<sub>2</sub>/HONO)

NO<sub>2</sub> and HONO were also measured by the NOAA airborne cavity enhanced spectroscopy (ACES) instrument. This technique is based on incoherent broadband cavity enhanced

spectroscopy (CES, Fiedler et al., 2003). The CES instrument is described in full detail by Min et al. (2016) with only minor changes for FIREX-AQ. Briefly, the system consists of two parallel 45 cm optical cavities capped by highly reflective mirrors, with reflectivity curves centered at 365 nm (R = 0.99987) and 455 nm (R = 0.99992). Each cavity is illuminated by a broadband LED light source (centered at 365 and 455 nm respectively) collimated by an off-axis parabola, and passively coupled into the cavity. The light makes many passes before exiting the cavity into a fiber optic cable, which transmits the light to a grating spectrometer spanning 350–475 nm. The LEDs are modulated on for 0.4 s and off for 0.08 s for charged-couple device (CCD) readout, giving a total integration time of 0.48 s per light intensity spectrum. An absorption spectrum of the ambient air sample is determined using the procedure presented by Washenfelder et al. (2008). The procedure requires comparing the measured light intensity spectrum to a background spectrum of the cavity filled with cylinder zero air, which is determined here every 10 minutes. The mirror reflectivity is measured every hour using the Rayleigh scattering difference between helium and zero air, and the spectrometer dark counts and wavelength calibration are measured every two hours. A small flow from a mixture of 25 ppm NO<sub>2</sub> in air is diluted into the cavity every hour, resulting in NO<sub>2</sub> concentrations between 50 and 100 ppbv, to assess the NO<sub>2</sub> spectral retrieval features on the spectrometer. The absolute concentration was not used for calibration of the NO<sub>2</sub> response, but rather for providing a reference NO<sub>2</sub> spectrum. Glyoxal reference spectra was obtained by bubbling zero air through a Teflon bubbler with 40% glyoxal in water as in Min et al. (2016).

Ambient air is pulled through the inlet into the two optical cavities at a flow rate of 5.4 volumetric liters per minute per cavity by a scroll pump. The air passes through two 1 µm pore size Teflon filters (changed before each flight) before entering the instrument to remove any aerosol particles. Mirror cleanliness is maintained by flowing 150 sccm cylinder zero air over each mirror to prevent condensation of semi-volatile species. A pressure controller consisting of a Teflon orifice and a variable flow to a bypass maintains the internal pressure at one of two pressure set points: 400 mbar when the aircraft was below 7.3 km, and 150 mbar above 7.3 km. The residence time of the air inside the optical cavities is estimated to be 0.5 s.

The measured absorption spectrum is fit to a linear combination of literature or reference spectra of absorbing gas-phase species and a polynomial to account for drifts in the cavity stability or light source intensity, as detailed by Min et al. (2016), using a Levenberg-Marquardt least-squares fitting algorithm. For the 365 nm channel, those species are NO<sub>2</sub>, HONO, O<sub>4</sub>, and a 4<sup>th</sup> order polynomial. For the 455 nm channel, those species are NO<sub>2</sub>, glyoxal, methylglyoxal, H<sub>2</sub>O, and O<sub>4</sub>, as well as a 0<sup>th</sup> order polynomial, though only NO<sub>2</sub> is presented here. The algorithm uses reference spectra for NO<sub>2</sub> and glyoxal, as measured in the field, scaled linearly to the literature spectra of Vandaele et al. (1998) at 296 K and Volkamer et al. (2005) at 294 K, respectively. The literature spectra from Stutz et al. (2000), Meller et al. (1991), Harder & Brault, (1997), and Keller-Rudek et al. (2013) are used for HONO, methylglyoxal, H<sub>2</sub>O, and O<sub>4</sub>, respectively. The fitting range was 438 – 467 nm for the 455 nm channel, and 362 – 387 nm for the 365 nm channel. No structure was observed in the fit

residuals. Because the 455 nm channel has higher precision, only those NO<sub>2</sub> data are presented here, although the two channels agree to within 3%. The data are averaged to 1 s. The reported uncertainties are  $\pm$  (9% + 0.6 ppbv) for HONO and  $\pm$  (5% + 0.26 ppbv) for NO<sub>2</sub>, representing the accuracy and 2-sigma precision in 1 second.

## 2.2.5 Iodide-Adduct Chemical Ionization Mass Spectrometry (HONO)

HONO was measured using a modified commercial time of flight chemical ionization mass spectrometer (TOF CIMS, Aerodyne Research, Inc.; Lee et al., 2014; Veres et al., 2020). Trace gases are ionized by mixing ambient air with reagent ions made in flight, and the resulting product ions are detected. Ions are separated by mass-to-charge ratio (m/z) using a time-of-flight mass spectrometer with a resolving power of 5000 m/Δm and a range of mass to charge ratio up to 494 m/z. Spectra were obtained at a 25 kHz repetition rate, and then averaged to 1 s. High resolution peak fitting was performed on the spectra, using over 500 known masses. Reagent ions were formed by flowing 1 slpm N<sub>2</sub> through a temperature controlled CH<sub>3</sub>I permeation tube followed by a 20 mCi <sup>210</sup>Po radioactive source. Two reagent ions are generated: Iodide ions (I·) are formed in the radioactive source, and iodide-water clusters (I·H<sub>2</sub>O) are formed when I-reacts with water in the ion-molecule reactor (IMR). In the IMR, the reagent ions cluster with analyte gases to form stable iodide adducts. The IMR was controlled at 40 mbar pressure to reduce the effects of secondary ion chemistry that increase at higher pressures.

Ambient air was sampled through a mass flow controlled (6 slpm) heated perfluoroalkoxy (PFA) inlet (70 cm length, 0.64 cm inner diameter). A pressure control region upstream of a critical orifice at the entrance to the IMR was maintained at 140 mbar, so that a constant flow of 1.2 slpm ambient air entered the IMR to mix with the 1 slpm ion source flow. A small nitrogen flow of about 20 sccm containing water vapor was added directly into the IMR region and controlled to maintain a measured I- $\Phi$ I-cluster ratio of 50  $\pm$  2%, in order to maintain constant detection sensitivity. The reagent ion signals during FIREX-AQ were typically 2 MHz for I-•H<sub>2</sub>O and 4 MHz for I-, and they were stable as a function of aircraft altitude. In the most concentrated fire plumes with CO over 7 ppm, the abundance of reactants reduced the reagent ion signals by up to 15%. The product cluster ions were normalized by the iodide (I-H<sub>2</sub>O) signals to account for changes in reagent ions. The instrument background signal was determined inflight by overflowing the inlet with scrubbed ambient air for 30 seconds every 10 minutes through a port located 2 cm downstream of the inlet entrance. Calibrations with Cl<sub>2</sub> and HNO<sub>3</sub> permeation sources were performed hourly in flight to diagnose the stability of instrument sensitivity. The standard deviation of in flight calibrations was typically 10%.

HONO was detected as a cluster with I- that has a mass to charge ratio of 173.90575 m/z. Contributions from the <sup>13</sup>C isotope of formic acid at 173.91342 m/z are not completely mass resolved but are accounted for using high resolution peak fitting and isotope ratios based on the formic acid signal at its most abundant isotope. We know of no other contributions to the

signal at the mass used for HONO detection, consistent with previous studies (Neuman et al., 2016). The background HONO signals were typically equivalent to a mixing ratio of 40 ppt. and these were subtracted from the total signal to determine ambient HONO. Sensitivity to HONO was determined in the laboratory, using a tunable, calibrated HONO source that uses HCl reactions on humid NaNO<sub>2</sub> to generate HONO (Lao et al., 2020). The output was calibrated spectroscopically using the NOAA ACES instrument (Min et al., 2016). The absolute sensitivity to HONO was 3.4 ion counts/s/pptv for typical conditions. Sensitivities normalized by the reagent ions are used to determine mixing ratios from the normalized product ion signals. The HONO measurement uncertainty at fixed temperature was  $\pm$  (15% + 3 pptv), where the first term was from the laboratory calibrations and the second was the variability of the in-flight background determinations. The HONO measurement precision was  $\pm$  2 pptv for 1-second data. Calibrations and fieldwork conducted subsequent to FIREX-AQ identified a temperature dependence to the CIMS calibration. Section 3.3 below describes this sensitivity in more detail.

 $2.2.6 \Sigma NO_{\rm v}$ 

To determine the extent of budget closure for reactive odd nitrogen species during FIREX-AQ, we compare measured NO<sub>y</sub> (see section 2.2.1) with a simplified definition of  $\Sigma$ NO<sub>y</sub> as illustrated in the following equation:

 $\Sigma NO_v \approx NO_x + HONO + HNO_3 + pNO_3 + APN_S$  (Eq. 2)

Other nitrogen oxides were also measured during FIREX-AQ but were not included in this equation as they contributed on average less than 7% to the NO<sub>y</sub> budget (see section 3.4). Further, including these measurements would have decreased data availability for comparison with the total NO<sub>y</sub> measurement by more than 60%. These minor NO<sub>y</sub> species are alkene hydroxy nitrates, nitromethane (CH<sub>3</sub>NO<sub>2</sub>), N<sub>2</sub>O<sub>5</sub>, ClNO<sub>2</sub>, and C<sub>1</sub>–C<sub>5</sub> alkyl nitrates (Figures S1 and S2). Measurements used in Eq. 2 are CL NO<sub>x</sub>, CIMS HONO, CIMS HNO<sub>3</sub>, HR-AMS pNO<sub>3</sub> and CIMS APNs. These measurements were primarily used because they had better precision. Using LIF NO, CES NO<sub>2</sub> and CES HONO as primary measurements changed the correlation slope between  $\Sigma$ NO<sub>y</sub> and measured NO<sub>y</sub> by -2%, -6% and 6%, respectively (Table S1). In smoke, using LIF NO, CES NO<sub>2</sub> and CES HONO as primary measurements changed the correlation slope between  $\Sigma$ NO<sub>y</sub> and measured NO<sub>y</sub> by -1%, -8% and 9%, respectively (Table S1).

• Observations of HNO<sub>3</sub>, HCN, and hydroxyl nitrates produced from the oxidation of ethane, propene, butane, and isoprene were made by the California Institute of Technology Chemical Ionization Mass Spectrometer (CIT-CIMS) compact time-of-flight (cToF, TofWerk/Caltech) sensor using CF<sub>3</sub>O<sup>-</sup> ion chemistry (Crounse et al., 2006). In short, a large flow of ambient air (about 40 m<sup>3</sup> s<sup>-1</sup>) was rapidly brought into the aircraft through a Teflon coated glass inlet (warmed slightly above ambient temperature), where it was subsampled, diluted with dry N<sub>2</sub>, reacted with CF<sub>3</sub>O<sup>-</sup>, and underwent subsequent product ion analysis by time-of-flight mass spectrometry. The

HF•NO<sub>3</sub><sup>-</sup> (m/z 82) product ion is used to quantify HNO<sub>3</sub>. The HCN and hydroxy nitrates are detected as cluster ions. Laboratory-generated, T-dependent and water-dependent calibration curves were performed to produce ambient mixing ratios from raw signals for HNO<sub>3</sub> and hydroxy nitrates The HCN sensitivity is tracked in situ based on the continuous addition of isotopically labeled H<sup>13</sup>C<sup>15</sup>N into the instrument from a custom-made gravimetrically based compressed gas cylinder. In-flight instrumental zeros were performed every ~15 minutes using dry N<sub>2</sub> and ambient air passed through NaHCO<sub>3</sub>-coated nylon wool. Continuous data, with the exception of zero and calibration periods, are reported with 1Hz frequency. The uncertainties for HNO<sub>3</sub>, HCN, and hydroxy nitrates are  $\pm$  (30% + 50 pptv),  $\pm$  (25% + 70 pptv), and  $\pm$  (25% + 3 pptv), respectively.

Particulate nitrate (pNO<sub>3</sub>) was measured with a high-resolution time-of-flight AMS (HR-AMS, Aerodyne Research, Inc., Billerica, MA, USA). The HR-AMS measured submicron (PM<sub>0.9</sub>; calibrated in the field as described in Guo et al., 2021) aerosol composition at high time resolution (0.1–1 s) by flash vaporization of the aerosol, 70 eV electron ionization of the volatilized gas phase and subsequent analysis by mass spectrometry (Canagaratna et al., 2007; DeCarlo et al., 2006). pNO<sub>3</sub> is detected in the HR-AMS as the sum of H<sub>x</sub>NO<sub>y</sub><sup>+</sup> ions (mostly NO<sup>+</sup> and NO<sub>2</sub><sup>+</sup>). Typical 1 s detection limits for pNO<sub>3</sub> were about 90 ng sm<sup>-3</sup> (30 pptv) for urban/background conditions. Given the size cut in the HR-AMS instrument, pNO<sub>3</sub> does not include coarse nitrate from the reaction of HNO<sub>3</sub> with sea salt or dust aerosol. It does include particulate organic nitrates (pRONO<sub>2</sub>; Day et al., 2021; Farmer et al., 2010), which are speciated using the algorithm described in Fry et al. (2013) and Day et al. (2021). Likewise, particulate aryl nitrates such as nitrocatechol also contribute to the total pNO<sub>3</sub> signal (Guo et al., 2020). Nitrocatechol was also characterized by extractive electrospray ionization time-of-flight mass spectrometry (EESI-MS; Pagonis et al., 2021) and positive matrix factorization and tracer analysis suggests that total aryl nitrates could be 3–7 times the concentration of nitrocatechol.

• APNs were measured using a thermal dissociation – chemical ionization mass spectrometer (TD-CIMS) method. The CIMS instrument used during the FIREX-AQ campaign was similar to that described in Slusher et al. (2004) and Lee et al. (2020). Briefly, ambient air is sampled into the TD-CIMS through heated Teflon tubing at a temperature of approximately 150°C to thermally dissociate APNs. The thermal dissociation region was maintained at a constant pressure of 60 torr using a commercial pressure controller (MKS 640) to minimize negative interference due to NO, NO<sub>2</sub> and radical-radical reactions. In-flight calibrations were performed by continuous addition of isotopically labeled peroxyacetyl nitrate (PAN) standard quantified as acetate ion (61 m/z; C<sup>13</sup>H<sub>3</sub>C<sup>13</sup>(O)O<sup>-</sup>) in the TD-CIMS. NO was periodically added to the inlet (~10 ppm) to react away peroxyacyl radicals and thus to measure the instrument background signal. APNs species measured during FIREX-AQ include PAN, acryloyl peroxynitrate (APAN), propionyl peroxynitrate (PPN), and

peroxybutyryl nitrate (PBN) with an uncertainty of 20%, 30%, 30% and 30%, respectively.

• Nitromethane (CH<sub>3</sub>NO<sub>2</sub>), along with other volatile organic compounds (VOCs), was measured by proton-transfer-reaction time-of-flight mass spectrometry (PTR-ToF-MS; Gkatzelis et al., in prep). The PTR-ToF-MS sampled VOCs at 5Hz through short (1 m) heated inlet. Periodically, instrument backgrounds were determined by passing ambient air through a platinum catalyst heated to 350°C. The instrument response to VOCs was calibrated by gravimetrically prepared standards or by liquid calibration, as described by Gkatzelis et al. (2021). CH<sub>3</sub>NO<sub>2</sub> mixing ratios were determined by liquid calibration with an uncertainty of 30%.

•  $N_2O_5$  was detected as a cluster with  $\Gamma$  at mass 234.88574 m/z. Sensitivity was determined by standard addition laboratory calibrations, with  $N_2O_5$  generated by reacting a  $NO_2$  calibration standard with  $O_3$  (Bertram et al., 2009), and quantified using cavity ring down  $NO_y$  measurements (Womack et al., 2017). For typical operating conditions during FIREX-AQ,  $N_2O_5$  sensitivity was 70 ion counts/s/ppt.  $N_2O_5$  was measured with  $\pm$  (15% + 2 pptv) accuracy and 0.1 pptv precision for 1 second data. Iodide ions cluster with a DMS oxidation product, hydroperoxymethyl thioformate (HPMTF), that has a mass only 0.0074 amu greater than  $N_2O_5$ , and these two molecules cannot be completely resolved spectrometrically with the resolution (m/ $\Delta$ m = 5000) of this instrument (Veres et al., 2020). For these measurements over the continent, the contribution from HPMTF to the signal at the iodide  $N_2O_5$  cluster is assumed to be negligible.

• C<sub>1</sub>–C<sub>5</sub> alkyl nitrates were measured by the NOAA integrated whole air sampling system with off-line analysis by gas chromatography-mass spectrometry (iWAS/GC-MS as described in Lerner et al. (2017)). There were 142 iWAS samples collected over the LA Basin with an average fill time of 5.2 ± 0.7 seconds. There were 897 wildfire samples and 467 eastern fire samples with average fill times of 7.6 ± 1.1 and 4.5 ± 0.8 seconds, respectively. Due to the relatively fast fill times and targeted, ondemand sampling capabilities of the iWAS, 88% and 74% were "full smoke" samples for wildfire and eastern fire samples, respectively. All samples were analyzed in the NOAA Chemical Science Laboratory within 213 hours of sample collection with an average sample age of 87 ± 34 hours between sample collection and sample analysis for FIREX-AQ.

# 2.2.7 Integrated Cavity Output Spectroscopy (CO)

CO was measured using a modified commercial off-axis ICOS instrument (Los Gatos Research (LGR)  $N_2O/CO$ -30-EP; Arévalo-Martínez et al., 2013; Baer et al., 2002) at 4.566  $\mu$ m. The commercial instrument has two flow paths, a slow flow path with cavity pressure controlled by an internal proportional valve, and a parallel high flow path with a needle valve to control pressure. The instrument was modified to use only the high flow path, but with an

automatic cavity pressure controller. The needle valve was removed from the flow path in favor of a Piezo proportional valve (Horiba Stec UR-Z732M) located near the inlet.

Air was sampled from a ram-air intake inlet through 0.64 cm (outside diameter) stainless steel tubing. Cavity pressure was maintained at  $85.0 \pm 0.2$  Torr in flight. Immediately inside the fuselage, two CO (and  $N_2O$ ) calibration gas standards known to within  $\pm 0.4$  ppb CO were regularly delivered to the inlet line during flight to evaluate instrument sensitivity between 58.4 and 993.3 ppb CO (both ICOS-CO and TDLAS-CO mixing ratios are reported as dry air mole fractions). The calibration standards were added to displace ambient air and overflow the inlet, and were calibrated before and after the project using standard tanks tied to the World Meteorological Organization CO\_X2014A scale from the NOAA Global Monitoring Laboratory (Hall et al., 2007; Novelli et al., 1991). The 1-sigma variability of the slope and intercept of all in-flight calibrations was 0.6% and 0.9 ppb, respectively. A third calibration standard, referred to as a "target" (Peischl et al., 2010), was regularly introduced to the inlet between calibrations and treated as an unknown to evaluate long-term instrument performance. The retrieved value of 109 in-flight targets during FIREX-AQ was 301.6  $\pm$  1.0 ppb CO compared with the calibrated value of 301.1  $\pm$  0.4 ppb. The 1-Hz precision of the measurement in flight is estimated to be 0.4 ppb.

After the campaign, the  $H_2O$  measurement was calibrated using a MBW 373LX chilled-mirror hygrometer (MBW Calibration AG; Rollins et al., 2020). The  $H_2O$  measurement is estimated to have an uncertainty of  $\pm$  (50 ppmv + 4%), and was used to convert the CO measurement to a dry air mole fraction. The uncertainty of the dry air mole fraction of CO is estimated to be  $\pm$  (2.0 ppb + 2%) for mixing ratios below 1 ppm.

# 2.2.8 Tunable Diode Laser Absorption Spectroscopy (CO)

Carbon Monoxide (CO) was measured by tunable diode laser absorption spectroscopy (TDLAS) using the DACOM (Differential Absorption Carbon monOxide Measurement) instrument (Sachse et al., 1987). The TDLAS instrument configuration used during FIREX-AQ also included channels for measurements of methane (CH<sub>4</sub>) and carbon dioxide isotopes ( $^{12}$ CO<sub>2</sub> and  $^{13}$ CO<sub>2</sub>). This instrument utilizes three single-mode tunable diode lasers, with CO measured using a quantum cascade laser (QCL) at approximately 4.7 μm. The three individual mid-infrared laser beams were combined by the use of dichroic filters and directed through a small volume (0.3 liter) Herriott cell enclosing a 36-meter optical path. After exiting the Herriott cell, the beams were spectrally separated and directed to individual HgCdTe (MCT) detectors.

The lasers were operated in a wavelength-modulated mode, each at an independent frequency, and line-locked to the centers of the species' selected absorption lines. Lines were selected to provide both good sensitivity and good isolation from any potential spectral interferences. Detector signals were demodulated at twice the lasers' modulation frequencies (2F detection), and normalized by average detected laser intensity.

Ambient air was sampled through an inlet probe, compressed, and passed through a permeable membrane dryer to remove water vapor prior to being introduced into the Herriott cell. Due to the need for very fast time response during FIREX-AQ, the instrument was operated with a flow of approximately 14 slpm with the Herriott cell at a pressure of approximately 67 mbar. The resulting time response, verified with a fast-acting valve, was faster than 0.2 s. Data were reported at both 0.2 s and 1 s timesteps.

The TDLAS instrument was calibrated using the same gas standards as for the ICOS instrument, nominally with a 4-minute period, but often advanced or delayed in time to avoid calibrating during fire plume encounters. Calibrations provided both slope and intercept values tying signals to species concentrations. The very large CO concentrations encountered necessitated post-campaign correction calibrations to account for response nonlinearity.

Post-campaign analysis of the TDLAS CO data indicated that measurement precision ( $1\sigma$ ) was approximately 0.1% at 1 s and 0.14% at 0.2 s. Accuracy was dependent on CO mixing ratio, and varied from 2% to 7%.

## $2.2.9 H_2O$

 $H_2O$  was measured using the NASA diode laser hygrometer, an open-path infrared absorption spectrometer that uses a laser locked to one of three water vapor absorption features near 1.395  $\mu$ m, depending on the abundance of water vapor (Diskin et al., 2002; Podolske et al., 2003).  $H_2O$  mixing ratios were determined with an uncertainty of 5%.

#### 2.2.10 Smoke age

The age of smoke from emission to sampling by the aircraft was determined from an ensemble of upwind trajectories from the aircraft (Holmes et al., 2020). Trajectories were computed with HYSPLIT (Stein et al., 2015) using three meteorological datasets (HRRR, NAM CONUS Nest, and GFS  $0.25^{\circ}$ ). In each of the three trajectories, the advection time was determined from the point where the trajectory most closely approached the source fire. The age also includes plume rise time from the surface to trajectory altitude, which was estimated with a mean rise time of  $7 \pm 4$  m s<sup>-1</sup> (Lareau et al., 2018). Trajectories and ages that were grossly inconsistent with smoke transport patterns seen in geostationary satellite images were excluded from further analysis. The ensemble of age estimates was then averaged to provide a best estimate of smoke age. The median uncertainty in smoke age is about 27%, as determined by the sum in quadrature of the spread among the ensemble of estimates, the uncertainties in the updraft speed, the fire location and the wind speed, and uncertainties in the model.

#### 2.3 Methodology

This study focuses on comparing the different techniques used for the measurements of one or several reactive nitrogen species as well as CO during FIREX-AQ. Here we compare both archived 1 s data (<a href="https://www-air.larc.nasa.gov/missions/firex-aq/index.html">https://www-air.larc.nasa.gov/missions/firex-aq/index.html</a>) and the plume-integrated data. Plume-integrated data are obtained from integrating the 1Hz data of a given measurement over a smoke plume transect. A smoke plume transect was identified using the

time period between a CO and/or black carbon (BC) increase above a local background value (beginning of the plume transect) and the CO and/or BC decrease back to a background value (end of the plume transect). Background values on either side of a plume were different for some fires in spatially heterogeneous source regions. Note that any 10 s period of background air, even if experienced during a single smoke plume transect, was sufficient to mark the end of one transect and the start of the next. All 1Hz data were time-aligned prior to comparison by synchronizing features in the time series of each species. Time shifts were typically less than 4 seconds. Some disagreement between measurement techniques is expected due to the rapid variations sampled during FIREX-AQ, particularly when those variations occur faster than the measurement period and/or with greater spatial heterogeneity than the distance between the sampling locations on a large aircraft that can reach 25m in some cases. Comparisons in this manuscript are not blind as all PIs had access to other instruments measurements throughout the campaign.

We first calculated the slope of the linear least-squares (LLS) orthogonal distance regression (ODR; Boggs et al., 1987) to characterize the percent difference between measurements of a pair of instruments weighted by the inverse of the instrument precision. Here, we used a mixing ratio-independent instrument precision that corresponded to the 1 $\sigma$  precision in clean air. Weighting the fit by this term, rather than a more accurate but labor-intensive mixingratio-dependent precision, tend to overweight the highest measured mixing ratios. The slope and intercept resulting from the ODR regression analysis provide a measure of systematic or species-dependent instrumental biases. Additionally, we calculated the difference between a given pair of measurements. The difference, noted  $\Delta Y_{XI-X2}$  where X1 and X2 are the two measurement techniques for detection of the Y species, provides an understanding of the temporal evolution and environmental dependency of instrumental discrepancies. Note that the regression analysis yields slightly different information than the calculation of the difference: while the former is weighted more by fire plumes, where mixing ratios were greatest, the latter is weighted more by background conditions, where most of the measurements took place. Unless specified otherwise, all data available (i.e., both background and fire smoke data) were included in the following comparisons. We also calculated the fractional error (FE =  $\Delta Y_{XI-X2}/Y_{avg}$  where  $Y_{avg} = (Y_{XI} + Y_{X2})/2$ ) between pair of instruments using specifically fire smoke data to minimize measurements below instrument detection limits (Figures S3 and S4). In the following sections, combined instrument uncertainties were calculated by adding in quadrature individual instrument

## 3 Flight data comparisons

3.1 NO

uncertainties.

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## 3.1.1 Campaign-wide comparison

The 1Hz data comparison between the CL and LIF instruments is shown in Figure 2. The overall comparison slope ( $\pm$  combined instrument uncertainties) is  $0.98 \pm 0.08$  ( $R^2 = 0.93$ ) with an intercept of  $-2 \pm 0$  pptv (Figure 2a). Figures 3a and 4a show the two instruments' response in smoke from a wildfire and an eastern fire, respectively. While the NO signals track each other remarkably well, there is a difference in time response that is typical of the

entire campaign. Figure S5 shows an expanded view of 10Hz NO and CO measurements in a partial smoke plume transect, including the transition from smoke to background air sampling. The NO signal in the CL instrument exhibits less structure than in the LIF instrument and a tail following the plume-to-ambient air transition. These tails were commonly observed during this transition. This effect in the CL instrument may partly explain the elevated scatter below the 1:1 line in Figure 2b. Integrating the NO signal across plume passes reduces the scatter due to different instrument time response: the regression analysis of smoke plume-integrated NO mixing ratios yields a slope of 0.99 (R<sup>2</sup> = 0.95) for the whole dataset (Figure 2c).

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A histogram of the absolute difference between LIF and CL ( $\Delta$ NO<sub>LIF-CL</sub>) is shown in Figure 5a. 90% of the values were between –44 and 43 pptv, and the whole dataset is normally distributed around 0 ± 0 pptv (central value of the Gaussian fit and standard deviation).  $\Delta$ NO<sub>LIF-CL</sub> exhibits no significant correlation with NO and H<sub>2</sub>O mixing ratios, which suggests that there was no systematic bias between the two instruments over a wide range of NO mixing ratios and environmental conditions (Figures S6a and 6a). Similar slopes and intercepts were obtained when separately comparing NO measurements during the wildfire, eastern fire, and LA Basin sampling periods (Figures 2b and S7).

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## 3.1.2 Literature aircraft NO measurement comparisons

Overall, the comparison between the two NO instruments shows an agreement within stated uncertainties. While the single-photon LIF detection of NO is a new technique that was evaluated for the first time during FIREX-AQ (Rollins et al., 2020), there are several studies that compared CL detection of NO to other measurement techniques during airborne field campaigns. The Global Tropospheric Experiment Chemical Instrumentation Test and Evaluation (GTE-CITE) was designed in the 1990's to intercompare airborne measurement techniques for trace species including NO, NO<sub>2</sub> and CO. Comparison of two CL instruments and a two-photon LIF instrument showed agreement when NO mixing ratios were higher than 50 pptv, but pointed out periods of disagreement when NO mixing ratios were lower than 20 pptv (Gregory et al., 1990; Hoell et al., 1987). The Deep Convective Clouds & Chemistry (DC3) experiment in 2012 allowed for side-by-side comparison of instruments aboard two aircrafts at two level flight legs (7 and 12 km) for flight periods spanning 20–30 minutes. Pollack et al. (2016) showed that these NO measurements from two CL instruments agreed within 2% for NO mixing ratios up to 1 ppbv. More recently, Sparks et al. (2019) reported an intercomparison of several NO<sub>v</sub> species measurements, including NO, from the Wintertime Investigation of Transport, Emissions, and Reactivity (WINTER) airborne experiment over the Northeast US in 2015. During WINTER, NO measured by CRDS and CL differed on average by 16 % across all flights, which is outside of the combined instrument uncertainties. CL measurements were more consistent with an independent calculation of NO based on a photostationary state assumption.

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3.2 <u>NO</u><sub>2</sub>

3.2.1 Campaign-wide comparison

644 Three instruments measured NO<sub>2</sub> mixing ratios during FIREX-AQ using CL, CES and LIF detection techniques. The 1Hz data comparison between all three instruments is shown in 645 Figure 7. We find that the LIF and CES overall comparison yields a slope (± combined 646 instrument uncertainties) of  $1.03 \pm 0.08$  ( $R^2 = 0.98$ ), well within the combined instrument 647 uncertainties of 8% (Figure 7c). However, we find that comparing either the LIF or CES 648 649 instruments to the CL instrument results in correlation slopes (± combined instrument uncertainties) ranging from  $0.88 \pm 0.12$  to  $0.90 \pm 0.11$  (R<sup>2</sup> = 0.97), on the upper limit of the 650 8–11% combined uncertainties for each pair of instruments (Figures 7a and b). The higher 651 NO<sub>2</sub> mixing ratios measured by the CL instrument are further illustrated in the time series in 652 Figures 3b and 4b, and is consistent with a calibration error in one or all instruments, or an 653 654 interference from another species in the CL instrument. However, it is unlikely that the 655 difference between CL NO2 and other NO2 measurements was due to a calibration issue. If so, the CL NO measurement, which was calibrated using the same standard as for the CL 656 657 NO<sub>2</sub> measurement, would also have been 10-12% higher than the NO LIF measurement (which was calibrated using an independent standard). This was not the case during FIREX-658 659 AQ (see section 3.1). HONO is a known source of interference in measured NO<sub>2</sub> by 660 instruments that use photolysis in the near-UV region (Pollack et al., 2010). However, this 661 interference was determined to be low (less than 5% of HONO concentration; typical HONO 662 to NO<sub>2</sub> ratios ranged between 0.2–0.4 during FIREX-AQ) following laboratory tests using a HONO calibration source (Lao et al., 2020), and the NO<sub>2</sub> measurement by CL was corrected 663 for it. Additionally, we did not find a correlation between either  $\Delta NO_{2CES-CL}$  or  $\Delta NO_{2LIF-CL}$ 664 and HONO mixing ratios. There was better agreement between the CL and the other two 665 instruments when sampling the wildfires (slopes of 0.91) than the eastern fires (slopes of 0.75 666 and 0.87 for the LIF and CES, respectively) (Figures 7d and e). Similarly, the agreement 667 between the CES and the LIF instruments was near perfect during the first period (slope of 668 1.00), but worse during the latter period (slope of 1.13; Figure 7f). Note that the LIF 669 670 instrument did not report data for three flights out of seven during the eastern fires sampling 671 period. The increased difference may be caused by the physical distance between instrument 672 inlets combined with higher spatial heterogeneity of trace gases in the smaller and thinner 673 eastern fire plumes, although higher mixing ratios of a potential interferent may still exist. 674 Non-acyl peroxynitrate species such as pernitric acid (HO<sub>2</sub>NO<sub>2</sub>) and methyl peroxy nitrate (MPN) can be abundant in smoke plumes and interfere with NO<sub>2</sub> measurements (Browne et 675 al., 2011; Nault et al., 2015). This interference is the result of the thermal dissociation of 676 677 HO<sub>2</sub>NO<sub>2</sub> and MPN in heated inlets and sampling lines, and impact differently each instrument depending on their flush time. During FIREX-AQ, the CES and CL instruments 678 679 had similar flush time of about 750ms meaning that the thermal decomposition of non-acyl 680 peroxynitrates is unlikely to explain the 10–12% higher NO<sub>2</sub> signal in the CL instrument. Further,  $\Delta NO_{2CES-CL}$  or  $\Delta NO_{2LIF-CL}$  did not depend on altitude or outside temperature, which 681 682 also suggests little influence from thermally labile species. Nitrated phenolic compounds can be abundant in aged smoke (Decker et al., 2021), and have large UV cross sections (Chen 683 et al., 2011). They are unlikely to contribute to the interference as their NO<sub>2</sub> photolysis 684 quantum yields are very low. Nevertheless, further laboratory work on the NO<sub>2</sub> interference 685 686 of such species in photolytic converters is of interest. The agreement between all three

instruments for individual flights was generally within combined instrument uncertainties, but with some variability (Figures S8–S10).

Histograms of the absolute difference between CES, LIF and CL ( $\Delta NO_{2LIF-CL}$ ,  $\Delta NO_{2CES-CL}$  and  $\Delta NO_{2CES-LIF}$ ) are shown in Figures 5b–d. 90% of  $\Delta NO_{2LIF-CL}$ ,  $\Delta NO_{2CES-CL}$  and  $\Delta NO_{2CES-LIF}$  values were between –298 and 338 pptv, –469 and 302, and –576 and 393 pptv, respectively, and all are normally distributed around the central value of the Gaussian fit of  $0.038 \pm 0.001$ ,  $-0.052 \pm 0.001$ , and  $-0.071 \pm 0.001$ , respectively.  $\Delta NO_{2LIF-CL}$ ,  $\Delta NO_{2CES-CL}$  and  $\Delta NO_{2CES-LIF}$  exhibit no significant trend with H<sub>2</sub>O mixing ratios (Figures 6b–d), yet  $\Delta NO_{2LIF-CL}$  and  $\Delta NO_{2CES-CL}$  were weakly (R<sup>2</sup> = 0.36 and 0.31, respectively) correlated with the absolute NO<sub>2</sub> mixing ratio (Figures S6b and d).

#### 3.2.2 Literature aircraft NO<sub>2</sub> measurement comparisons

Previous comparisons of NO<sub>2</sub> airborne measurements often show periods of disagreement between instruments, although there were some occasions where instruments agreed within stated uncertainties. During the GTE-CITE experiment, the comparison of NO<sub>2</sub> measurements using a two-photon NO LIF system with laser photolysis of NO<sub>2</sub> to NO with a CL detector equipped with a xenon arc lamp for NO<sub>2</sub> photolysis into NO showed agreement within 30–40% (Gregory et al., 1990). Pollack et al. (2016) showed that two NO<sub>2</sub> measurements, both using CL but each in a different aircraft, agreed within 28% during the DC3 campaign. During WINTER, NO<sub>2</sub> measurements by CRDS and LIF agreed with an average proportional bias of 2% across all flights – well within combined uncertainties (Sparks et al., 2019). During SENEX, three techniques were used to measure NO<sub>2</sub>: a CRDS instrument, a CES instrument and a CL instrument. The agreement between CRDS and CES measurements with the CL technique was on average 6 and 10% (Warneke et al., 2016).

## **3.3 HONO**

# 3.3.1 Campaign-wide comparison

The 1Hz data comparison between the CES and the CIMS instruments is shown in Figure 8, and timeseries of HONO measurements in wildfires and eastern fires are shown in Figures 3c and 4c, respectively. The correlation between the CES and CIMS was very high in each plume transect (Figures 3c and 4c), but the overall comparison yielded a slope (± combined instrument uncertainties) of  $1.80 \pm 0.16$  ( $R^2 = 0.77$ ) and an intercept of  $-0.12 \pm 1.10$  ppby (Figure 8a). Integrating across plume transects yielded a slope of  $1.34 \pm 0.16$  (Figure 8c). The CIMS consistently reported less HONO than the CES in smoke plumes, and the average slope between the two measurements was considerably greater during the eastern fires compared to the wildfires (Figures 8b and S9). However, flight averages of the absolute difference between the two measurements (ΔHONO<sub>CES-CIMS</sub>) ranged between -332 and 245 pptv throughout the campaign and were similarly scattered around zero during the two different time periods (Figure S11). A histogram of ΔHONO<sub>CES-CIMS</sub> is shown in Figure 5e. 90% of the values were between –965 and 880 pptv, and the whole dataset is normally distributed around the central value of the Gaussian fit ( $\pm$  standard deviation) of  $-119 \pm 2$ pptv.  $\Delta HONO_{CES-CIMS}$  exhibits no significant slope with HONO (Figure S6e). While the deployment out of Salina was operated under noticeably more humid conditions (H<sub>2</sub>O ranged

from 0.002 to 2.944%) than out of Boise ( $H_2O$  ranged from 0.004 to 1.479%), we find no significant correlation between  $\Delta HONO_{CES-CIMS}$  and  $H_2O$  mixing ratios (Figure 6e).

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However, further laboratory studies, field measurements, and examination of this comparison has revealed that the CIMS sensitivity to HONO is reduced when the instrument reaches temperatures greater than 30°C (Figure S12). This sensitivity dependence on temperature does not affect all compounds measured by the CIMS, and the sensitivity to Cl<sub>2</sub> and HNO<sub>3</sub> used for in-flight calibrations was independent of instrument temperature. The aircraft cabin temperature was greatest during the eastern agricultural flights, when the CIMS instrument temperatures were often 40°C and far greater than the typical 25°C instrument temperatures in the laboratory when the CIMS HONO sensitivity was determined. As a consequence, the reported CIMS HONO values were spuriously low, especially during the eastern fires, and particularly later in flights when the aircraft temperatures were greatest. This intercomparison has yielded new insights into the CIMS HONO detection sensitivity, and future work will identify and implement appropriate corrections to this measurement (Robinson et al. 2022).

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3.3.2 Literature aircraft and ground HONO measurement comparisons The interpretation of literature suggest that HONO measurements are notoriously difficult due to the potential for artifacts associated with inlet surfaces as well as interferences associated with some methods (e.g., Kleffmann et al., 2006; Xu et al., 2019). Past groundbased intercomparisons often revealed significant discrepancies in HONO measurements. For example, six ground-based HONO measurement techniques including a CIMS instrument were compared during the Study of Houston Atmospheric Radical Precursors (SHARP) campaign in 2009 (Pinto et al., 2014). While three out of six of these techniques agreed within 20%, larger deviations were found when the other three instruments were considered and attributed to the physical separation of these instruments. Three different techniques, including a CIMS instrument, were used to measure HONO in the urban area of Shanghai, China (Bernard et al., 2016). The percent difference between these measurements ranged from 27 to 46%. In 2019, six HONO measurement techniques were again compared in a Chinese urban area, this time in Beijing, and included a CIMS instrument as well as two broadband cavity enhanced absorption spectrometers (BBCEAS) (Crilley et al., 2019). Percent differences up to 39% were observed during this intercomparison and again attributed to the physical distance separating inlets coupled to high spatial heterogeneity of HONO mixing ratios. Airborne measurements of HONO by CIMS and CES were made during the Southeast Nexus Experiment (SENEX), and the CES instrument was approximatively 25% higher than the CIMS instrument (Neuman et al., 2016).

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# 3.4 NO<sub>v</sub>

## 3.4.1 Campaign-wide comparison

The 1Hz data comparison between the total  $NO_y$  measurement by CL and  $\Sigma NO_y$  is shown in Figure 9.  $\Sigma NO_y$  definition is given by Eq. 2 (see section 2.2.6).  $C_1$ – $C_5$  alkyl nitrates and other minor  $NO_y$  species (including  $CINO_2$ ,  $N_2O_5$ ,  $CH_3NO_2$ , and alkene hydroxy nitrates) contributed less than 7% of the  $NO_y$  budget on average and were not included in  $\Sigma NO_y$ 

774 (Figure 10). Based on comparisons of HR-AMS pNO<sub>3</sub> with on-board filters collecting 775 aerosols with a size cut around 4um (Brock et al., 2019; Dibb et al., 2002), coarse mode 776 particulate nitrate did not significantly contribute to the total NO<sub>v</sub> budget during FIREX-AQ. 777 Additionally, coarse mode particulate nitrate was not measured by either the HR-AMS or the 778 NO<sub>v</sub> inlet in the CL instrument and therefore does not contribute to the intercomparison 779 presented here. The overall comparison yielded a slope (± combined instrument uncertainties) of  $1.00 \pm 0.25$  (R<sup>2</sup> =0.98) and an intercept of  $-0.52 \pm 0.01$  ppbv (Figure 9a). 780 The regression analysis of smoke plume-integrated NO<sub>v</sub> mixing ratios yields a slope of 1.00 781  $(R^2 = 0.99)$  for the whole dataset (Figure 9c). Comparison  $\Sigma NO_v$  to CL  $NO_v$  in fresh (<1h 782 783 since emission) and aged (>1h since emission) smoke during the wildfires sampling period 784 showed similar agreement (slopes of 0.98 and 1.05, respectively) despite the chemical 785 evolution of NO<sub>v</sub> species, highlighted by the different proportion of those species to the NO<sub>v</sub> 786 balance (Figure S13). Including minor NO<sub>v</sub> species (i.e., ClNO<sub>2</sub>, N<sub>2</sub>O<sub>5</sub>, CH<sub>3</sub>NO<sub>2</sub>, and alkene hydroxy nitrates) in the  $\Sigma NO_v$  had little effect on the correlation between  $\Sigma NO_v$  and CL  $NO_v$ 787 and resulted in a slope of  $1.02 \pm 0.25$  (R<sup>2</sup> =0.94) and an intercept of  $-0.68 \pm 0.01$  ppbv (Figure 788 789 S14). 790

Despite this correlation, two modes are apparent in the overall distribution of the absolute difference ( $\Delta NO_{yCL-Sum}$ ) between  $\Sigma NO_y$  and the total  $NO_y$  measurement (Figure 5f). The first mode is distributed around  $-0.068 \pm 0.001$  ppbv (central value of the first mode of the Gaussian fit), while the second is distributed around an average value of  $0.158 \pm 0.009$  ppbv (central value of the second mode of the Gaussian fit). Separating the comparison into three time periods reveals that this two-mode distribution of  $\Delta NO_{yCL-Sum}$  comes from the eastern fires sampling period as well as from the LA Basin flights whereas during the wildfires sampling period  $\Delta NO_{yCL-Sum}$  distribution is unimodal (Figure 11).

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Higher  $\Sigma NO_v$  compared to  $NO_v$  (first mode) could be explained by (i) a lower conversion efficiency of one or more NO<sub>v</sub> species in the CL instrument than estimated in the laboratory, (ii) sampling loss of pNO<sub>3</sub> through the NO<sub>y</sub> inlet, and (iii) inaccuracy in one or more of the individual NO<sub>v</sub> species measurement techniques. Here, we further investigated the sampling loss of pNO<sub>3</sub> through the CL instrument NO<sub>v</sub> inlet using a multistage flow model following the template of the Particle Loss Calculator (von der Weiden et al., 2009). The model calculates aerodynamic losses at each stage of the NO<sub>v</sub> inlet and provides the resulting total pNO<sub>3</sub> sampling efficiency (See Section S1 and Figure SA). We find that the main aerosol sampling loss occurs at the NO<sub>v</sub> inlet tip orifice (1.0 mm in diameter) due to the inlet orientation (perpendicular to the aircraft flight direction). Additional loss was calculated to be negligible once pNO<sub>3</sub> penetrated the NO<sub>v</sub> inlet, meaning that pNO<sub>3</sub> is fully volatilized into NO inside the heated gold catalyst (Miyazaki et al., 2005; see Section S1 and Figure SA). Particle sampling through the NO<sub>v</sub> inlet is highly dependent on altitude, air speed (see section S1 and Figure SB) and pNO<sub>3</sub> mass size distribution (Figure 12a). Figure 12b shows the average modeled particle sampling fraction through the NO<sub>v</sub> inlet, given as a ratio where a value of 1 means the total pNO<sub>3</sub> is sampled, for each flight during FIREX-AQ. Particle sampling fraction was calculated for three different air speeds for each flight: 40%, 65%, and

- 817 100% of the aircraft speed. An assumed sampled air speed of 65% that of the aircraft
- improved the correlation between  $\Delta NO_{vCL-Sum}$  and the modeled pNO<sub>3</sub> loss in the inlet (see
- Section S1 and Figure SB). At that speed, the calculated average particle sampling fraction
- varied between 0.36 and 0.99 for each flight (Figure 12b). The variability in the  $\Sigma NO_v$  to  $NO_v$
- correlation slope between aged and fresh smoke (Figure S13a) likely illustrates the non-
- quantitative sampling of pNO<sub>3</sub> in the NO<sub>y</sub> instrument. Indeed, higher  $\Sigma$ NO<sub>y</sub> than measured
- NO<sub>v</sub> in aged smoke (slope of 1.05), where pNO<sub>3</sub> is one of the main components of  $\Sigma$ NO<sub>v</sub>
- 824 (Figure S13b), may be explained by the non-quantitative sampling of pNO<sub>3</sub> in the NO<sub>4</sub>
- instrument. In fresh smoke, pNO<sub>3</sub> is a smaller component of NO<sub>y</sub>, and non-quantitative
- sampling of pNO<sub>3</sub> in the CL instrument may have less impact on the comparison (slope of
- 827 0.98)
- We calculated the fraction of measured NO<sub>y</sub> in smoke initially attributed to pNO<sub>3</sub> that may
- result from other reactive nitrogen species than those included in the  $\Sigma NO_v$  according to
- equation 3:
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- Missing  $NO_y$  fraction =  $((1 particle sampling fraction) \times pNO_3) / NO_y$  (Eq. 3)
- Where particle sampling fraction corresponds to the modeled pNO<sub>3</sub> sampling fraction in the
- NO<sub>v</sub> inlet. We found that missing NO<sub>v</sub> accounted for 0–24% of the measured NO<sub>v</sub> in smoke
- 835 (assuming a sampled air speed 65% that of the aircraft; Figure 12b). This additional
- contribution has a large uncertainty because the model may underestimate pNO<sub>3</sub> sampling
- through the NO<sub>y</sub> inlet due to the large uncertainty when the losses are calculated at high air
- speed (see Section S1). Further, we used bulk aerosol volume size distributions measured
- with a Laser Aerosol Spectrometer (LAS; Moore et al., 2021) to derive pNO<sub>3</sub> sampling
- fractions in Figure 12b as pNO<sub>3</sub> mass size distribution measurements were not available for
- all flights during FIREX-AQ. At a typical FIREX-AQ sampling altitude of 5 km, the LAS
- and HR-AMS size distributions can differ by about 10% (See Section S1 and Figure SC),
- which adds to the uncertainty of the pNO<sub>3</sub> sampling fraction through the NO<sub>3</sub> inlet.
- Correcting for particle sampling through the NO<sub>v</sub> inlet still yields an agreement between
- measured  $NO_v$  and  $\Sigma NO_v$  that is within the combined instrument uncertainties of 25%.
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- On the other hand, the positive  $\Delta NO_{vCL-Sum}$  mode (second mode) may indicate either an
- inaccuracy in one of the individual NO<sub>v</sub> species measurement techniques or an NO<sub>v</sub> species
- not measured. Further, we find that positive  $\Delta NO_{vCL-Sum}$  occurred both in smoke (Figure 11d)
- and in background air (Figure 11c) when sampling the eastern fires and that  $\Delta NO_{vCL-Sum}$
- exponentially decreased with altitude, a pattern also observed during the LA Basin flights but
- not during the wildfires sampling period (Figure 13b). Note that flight altitude when
- sampling the wildfires was 4.6 km on average, higher than the altitude average of 0.6 and 1.1
- km during the eastern fires and the LA Basin flights, respectively. Both water vapor and C<sub>1</sub>-
- 855  $C_5$  alkyl nitrates (not included in  $\Sigma NO_v$  thus far) were enhanced at lower altitude and may be
- possible causes for the positive  $\Delta NO_{vCL-Sum}$  mode. Alkyl nitrates have been shown to account
- for a significant fraction of the NO<sub>v</sub> budget in past studies (e.g., Fisher et al., 2016; Hayden et
- al., 2003; Horii et al., 2005). However, we find only a weak correlation between  $\Delta NO_{vCL-Sum}$

and  $C_1$ – $C_5$  alkyl nitrates during both the wildfires ( $R^2$  = 0.07) and eastern fires ( $R^2$  = 0.08) sampling periods (Figure 13c). The correlation is stronger ( $R^2$  = 0.44) during the LA Basin flights (Figure 13c). Further, we find that  $C_1$ – $C_5$  alkyl nitrates contributed similarly to the NO<sub>y</sub> budget when smoke from the wildfires (1.1% on average) and the eastern fires (0.8% on average) was sampled (Figure 10a), while the positive mode in the  $\Delta$ NO<sub>yCL–Sum</sub> distribution is present in the latter period only. H<sub>2</sub>O is a known source of interference in most instruments, and its impact on measurements is minimized when an accurate correction can be applied. Increasing  $\Delta$ NO<sub>yCL–Sum</sub> is associated with increasing H<sub>2</sub>O mixing ratios in the eastern fires, although the correlation is weak ( $R^2$  = 0.05) due to the elevated scatter of the data (Figure 13a). Similar slopes and intercepts were obtained when separately comparing NO<sub>y</sub> measurements in smoke from the wildfires and eastern fires (Figures 9b and S13). The slope of 0.81 during the LA Basin flights, may be caused by the lower precision of  $\Sigma$ NO<sub>y</sub> than that of the CL NO<sub>y</sub> (Figure 9b).

## 3.4.2 Discussion and other NO<sub>v</sub> measurement comparisons

Overall, the agreement between the total  $NO_v$  measured by the CL instrument and the  $\Sigma NO_v$ is within instrument uncertainties. Budget closure implies that the historical definition of NO<sub>v</sub> (i.e., NO<sub>x</sub> and its oxidation products, excluding reduced nitrogen species such as NH<sub>3</sub> and HCN) is adequate even in extremely reactive conditions that foster rapid changes in NO<sub>v</sub> speciation. Reduced nitrogen species such as hydrogen cyanide (HCN) or ammonia (NH<sub>3</sub>) represent a large fraction of the total nitrogen emission from biomass burning (Roberts et al., 2020) and have been shown to cause a small interference in CL instruments in dry air (Fahey et al., 1985, 1986). This interference is often neglected because of either the low atmospheric abundance of these species or sampling in humid air where such interference is thought to be negligible. Here, we find no evidence for a potential interference of HCN or NH<sub>3</sub>, despite their high abundance (tens of ppbv) in smoke plumes (Figure S16). The NO<sub>v</sub> to CO ratio was approximately conserved with smoke age, but showed both increasing and decreasing trends with different fires, likely as a result of variability in the NO<sub>x</sub> to CO emission ratio during the course of a day with changing fire conditions. Altogether, our findings show that the NO<sub>v</sub> instrument provides an accurate and conservative measurement of total reactive nitrogen species, although further work is needed to empirically characterize pNO<sub>3</sub> sampling through the NO<sub>v</sub> inlet.

There are a few studies that recently examined the  $NO_y$  budget closure from aircraft measurements. Juncosa Calahorrano et al. (2021) presented reactive odd nitrogen partitioning during the Western wildfire Experiment for Cloud chemistry, Aerosol absorption and Nitrogen (WE-CAN) that sampled western American wildfires during the summer 2018. The authors found significant (15–26%) contribution of organic N species other than APNs and alkyl nitrates to  $\Sigma NO_y$ . However, there was no total  $NO_y$  measurement during WE-CAN, and the conclusion is based on summed individual reactive nitrogen species. The FIREX-AQ comparison of  $\Sigma NO_y$  to total  $NO_y$  finds 2-13% of the total  $NO_y$  unaccounted for, smaller than the estimate of a 15-26% contribution from multifunctional organic nitrates from WE-CAN. While the FIREX-AQ  $NO_y$  difference suggests a smaller contribution from organic nitrates, the WE-CAN estimate is within the uncertainty of the FIREX-AQ analysis. During the

WINTER campaign, budget closure of NO<sub>z</sub> (=NO<sub>v</sub> – NO<sub>x</sub>) was demonstrated to occur within 20% for all flights following the comparison of  $\Sigma NO_z$  with total  $NO_z$  from three different measurement techniques, including a CL instrument (Sparks et al., 2019). A recent groundbased study in New York State in the US found that the sum of the individual reactive odd nitrogen species accounted for 95% of the total NO<sub>v</sub>, well within measurement uncertainties (Ninneman et al., 2021). These recent studies contrast with somewhat older literature that often reported a significant shortfall in the NO<sub>v</sub> balance, where measured NO<sub>v</sub> was higher than  $\Sigma NO_v$  (Hayden et al., 2003; Horii et al., 2005; Williams et al., 1997; Zhang et al., 2008). This shortfall has often been attributed to unmeasured organic N species and more specifically alkyl nitrates (Day et al., 2003; Horii et al., 2005). During FIREX-AQ, C<sub>1</sub>-C<sub>5</sub> alkyl nitrates accounted for less than 7% on average of the NO<sub>v</sub> budget (Figure 10), consistent with findings from other regions in the US (Benedict et al., 2018; Russo et al., 2010). However, FIREX-AQ did not include a measurement of total alkyl nitrates. A recent analysis of the California Rim Fire during the 2013 NASA Studies of Emissions, Atmospheric Composition, Clouds and Climate Coupling by Regional Surveys (SEAC<sup>4</sup>RS) mission report that total alkyl nitrates measured by TD-LIF accounted for ~10% of the NO<sub>v</sub> budget (Wolfe et al., 2022).

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### 3.5.1 Campaign-wide comparison

The 1Hz data comparison between the ICOS and the TDLAS instruments is shown in Figure 14. The overall comparison yielded a slope ( $\pm$  combined instrument uncertainties) of 0.98  $\pm$  $0.03 \text{ (R}^2 = 0.99)$  and an intercept of  $-1.06 \pm 0.01$  ppbv (Figure 14a). The regression analysis of smoke plume-integrated CO mixing ratios yields a slope of 0.99 ( $R^2 = 1$ ) for the whole dataset (Figure 14c). A histogram of the absolute difference between CO measurements  $(\Delta CO_{ICOS-TDLAS})$  is shown in Figure 5g. 90% of the values were between -6.05 and 2.35 ppbv, and the whole dataset is normally distributed around the central value of the Gaussian fit of –  $2.87 \pm 0.02$  ppbv. This is indicative of an offset between the two CO instruments, with the TDLAS systematically higher than the ICOS instrument. This average 2.87 ppbv offset was consistent throughout the campaign regardless of the type of fires that were sampled. Therefore, it cannot explain the significantly lower agreement of the instruments during the eastern fires compared to the wildfires sampling period (Figures 14b and S12). During the first period, the overall slope was 0.99 and ranged from 0.97 to 1.02 (average of 0.99) for individual flights, well within the combined instrument uncertainties of 3% (Figures 14b and S12). However, all individual flight measurements during the eastern fires sampling period exhibit slopes reduced by about 10% (range = 0.86–0.91 with an average of 0.89) and largely positive intercepts (range 6.75–19.04 with an average of 11.51) (Figure S17). As observed for other species, the second period proved to be a more challenging environment for CO measurements. This may be attributed to a spectral issue with one or the other of these two instruments, although we could not identify the source of the discrepancy.  $\Delta CO_{ICOS-TDLAS}$ exhibit no significant slope with CO (Figure S6g) and H<sub>2</sub>O (Figure 6g) mixing ratios.

3.5.2 Literature aircraft CO measurement comparisons

Overall, the comparison between the two CO instruments shows an agreement well within stated uncertainties. We find that the agreement between the two CO instruments used during FIREX-AQ is well in line with past intercomparisons. During the GTE-CITE experiment, the comparison of a TDLAS technique with two grab sample/gas chromatograph methods for detection of CO showed agreement across the instruments – within the combined instrument uncertainties and strong correlations (R<sup>2</sup> = 0.85–0.98) for CO ranging from 60 to 140 ppbv (Hoell et al., 1987). During the North Atlantic Regional Experiment (NARE 97) CO was measured by TDLAS and vacuum ultra-violet fluorescence with agreement to within 11% and systematic offsets of less than 1ppbv (Holloway et al., 2000). CO was also more recently measured by TDLAS and vacuum ultra-violet fluorescence during the side-by-side comparison of instruments aboard two aircraft during the DC3 experiment. There, CO measurements agreed within 5% during flight periods typically ranging from 20 to 30 minutes (Pollack et al., 2016).

#### 4 Conclusion

In this study, we compare airborne measurements of NO, NO<sub>2</sub>, HONO, NO<sub>y</sub> and CO conducted during the FIREX-AQ campaign in the summer 2019. This dataset offers the opportunity to assess the accuracy of a large suite of detection techniques in a challenging environment where species mixing ratios increased by tens of ppbv in seconds between background air and fire smoke. For NO, NO<sub>2</sub> (CES and LIF), NO<sub>y</sub> and CO, correlations agree better than the combined instrument uncertainties, indicating that the stated individual uncertainties are conservative estimates. For NO<sub>2</sub> (CL) and HONO, the percent difference between measurements is higher than the combined instrument uncertainties, indicating potential interferences or calibration inaccuracies that are not identified at this time. Based on the analysis above, we make the following recommendations, which are specific to the FIREX-AQ campaign.

1) Comparison of NO measurements by LIF and CL showed an overall agreement well within instrument uncertainties. Flight-to-flight agreement was generally more variable during the eastern fires sampling period than during the wildfires sampling period, which was attributed to the heterogeneous nature of smoke plumes combined with the physical separation of inlets. Both measurements are considered reliable for FIREX-AQ, although the LIF instrument has better 1Hz precision (1 pptv) than the CL instrument (6 pptv), and the CL instrument exhibited slower time response.

2) Comparison of NO<sub>2</sub> measurements by LIF and CES showed an overall agreement well within the stated instrument uncertainties. However, NO<sub>2</sub> measured by CL is on average 10% higher than that measured by the other two techniques. The agreement worsens for all instruments when comparing NO<sub>2</sub> measured during the eastern fires sampling period, likely for similar reasons as indicated for the NO measurements.

3) The CES and CIMS HONO measurements were highly correlated in each fire plume transect, but the correlation slope of CES vs. CIMS for all 1 Hz data from the entire campaign was 1.80. The HONO measured by CIMS was on average 74% of that measured by

CES during the wildfires sampling period, and on average 40% of CES during the eastern fires sampling period. The higher precision data from the CIMS are most useful for analysis of HONO when mixing ratios are lower. The redundancy of HONO measurements during FIREX-AQ led to the discovery that the CIMS sensitivity to HONO was reduced in a high temperature environment. This intercomparison has initiated further studies of the CIMS sensitivity to HONO and other compounds.

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4) Closure of the NO<sub>v</sub> budget between the total NO<sub>v</sub> measurement by CL and  $\Sigma$ NO<sub>v</sub> was achieved for all flights and correlation slopes were usually much better than the combined instrument uncertainties of 25%. NO<sub>x</sub>, HNO<sub>3</sub>, HONO, APNs and pNO<sub>3</sub> are the main contributors to the NO<sub>v</sub> budget, with the other reactive N species contributing less than 10% on average. We find that the modeled pNO<sub>3</sub> sampling fraction through the NO<sub>y</sub> inlet is highly dependent on altitude, air speed and pNO<sub>3</sub> mass size distribution, and varied on average between 0.36 and 0.99 during FIREX-AQ. Therefore, approximately 0–24% on average of the total measured NO<sub>v</sub> by CL may be unaccounted for and possibly explained by other species such as multifunctional organic nitrates. The reason for the secondary positive mode of 0.4 ppbv in the  $\Delta NO_{yCL-Sum}$  distribution in the eastern fires and LA Basin flights could not be clearly identified. Potential explanations include the contribution of gas-phase organic nitrates, not included in the  $\Sigma NO_v$ , and/or a water vapor interference in one or more instruments. Regardless, we conclude that the total NO<sub>v</sub> measurement by CL provides a robust quantification of the reactive nitrogen species in background air as well as in smoke plumes, and that the total NO<sub>v</sub> measurement is not sensitive to interference from reduced nitrogen species in fire plumes. Further laboratory and field work will be needed to fully characterize pNO<sub>3</sub> sampling through the NO<sub>v</sub> inlet.

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5) Comparison of CO measurements by TDLAS and ICOS showed an agreement well within the combined instrument uncertainties. An offset of ~2 ppbv between the two instruments was identified but has little impact on the correlation. There was a clear difference in the agreement between the wildfires sampling period and the eastern fires sampling period, where the correlation slopes were about 10% lower.

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6) Integrating data across smoke plume transects generally improved the correlation between independent measurements and may be necessary for fire-science related analyses, especially for smaller plumes with greater spatial heterogeneity compared to the distance between the sampling locations on a large aircraft.

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#### Data availability

All data used in this manuscript are archived online and available at https://www-air.larc.nasa.gov/cgi-bin/ArcView/firexaq.

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## **Author contribution**

- 1030 I.B. and T.B.R. designed research. All authors performed FIREX-AQ measurements. P.C.-J.,
- H. G., and J.L.J performed the flow modelling analysis. All authors analyzed data. I.B., J.P.,
- J.A.N., and S.S.B. wrote the original draft and all authors edited and revised the paper.

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- 1034 Competing interests
- The authors declare they have no conflict of interest.

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- 1049 References
- 1050 Andreae, M. O. (2019). Emission of trace gases and aerosols from biomass burning an
- 1051 updated assessment. Atmospheric Chemistry and Physics, 19(13), 8523–8546.
- 1052 https://doi.org/10.5194/acp-19-8523-2019
- 1053 Arévalo-Martínez, D. L., Beyer, M., Krumbholz, M., Piller, I., Kock, A., Steinhoff, T., et al.
- 1054 (2013). A new method for continuous measurements of oceanic and atmospheric  $N_2O$ , CO
- and CO<sub>2</sub>: performance of off-axis integrated cavity output spectroscopy (OA-ICOS) coupled
- to non-dispersive infrared detection (NDIR). Ocean Science, 9(6), 1071–1087.
- 1057 https://doi.org/10.5194/os-9-1071-2013
- Baer, D. S., Paul, J. B., Gupta, M., & O'Keefe, A. (2002). Sensitive absorption measurements
- in the near-infrared region using off-axis integrated-cavity-output spectroscopy. Applied
- 1060 Physics B, 75(2), 261–265. https://doi.org/10.1007/s00340-002-0971-z
- 1061 Balch, J. K., Bradley, B. A., Abatzoglou, J. T., Nagy, R. C., Fusco, E. J., & Mahood, A. L.
- 1062 (2017). Human-started wildfires expand the fire niche across the United States. Proceedings
- 1063 *of the National Academy of Sciences*, 114(11), 2946–2951.
- 1064 https://doi.org/10.1073/pnas.1617394114
- Benedict, K. B., Prenni, A. J., Sullivan, A. P., Evanoski-Cole, A. R., Fischer, E. V., Callahan,
- 1066 S., et al. (2018). Impact of Front Range sources on reactive nitrogen concentrations and
- 1067 deposition in Rocky Mountain National Park. PeerJ, 6, e4759.
- 1068 https://doi.org/10.7717/peerj.4759
- 1069 Bernard, F., Cazaunau, M., Grosselin, B., Zhou, B., Zheng, J., Liang, P., et al. (2016).
- 1070 Measurements of nitrous acid (HONO) in urban area of Shanghai, China. Environmental
- 1071 Science and Pollution Research, 23(6), 5818–5829. https://doi.org/10.1007/s11356-015-
- 1072 5797-4

- 1073 Bertram, T. H., Thornton, J. A., & Riedel, T. P. (2009). An experimental technique for the
- direct measurement of  $N_2O_5$  reactivity on ambient particles. Atmospheric Measurement
- 1075 Techniques, 2(1), 231–242. https://doi.org/10.5194/amt-2-231-2009
- 1076 Boggs, P. T., Byrd, R. H., & Schnabel, R. B. (1987). A Stable and Efficient Algorithm for
- Nonlinear Orthogonal Distance Regression. SIAM Journal on Scientific and Statistical
- 1078 Computing, 8(6), 1052–1078. https://doi.org/10.1137/0908085
- 1079 Bollinger, M. J., Sievers, R. E., Fahey, D. W., & Fehsenfeld, F. C. (1983). Conversion of
- 1080 nitrogen dioxide, nitric acid, and n-propyl nitrate to nitric oxide by a gold-catalyzed
- reduction with carbon monoxide. Analytical Chemistry, 55(12), 1980–1986.
- 1082 https://doi.org/10.1021/ac00262a034
- Bourgeois, I., Peischl, J., Neuman, J. A., Brown, S. S., Thompson, C. R., Aikin, K. C., et al.
- 1084 (2021). Large contribution of biomass burning emissions to ozone throughout the global
- remote troposphere. Proceedings of the National Academy of Sciences, 118(52),
- 1086 *e2109628118. https://doi.org/10.1073/pnas.2109628118*
- Bowman, D. M. J. S., Balch, J. K., Artaxo, P., Bond, W. J., Carlson, J. M., Cochrane, M. A.,
- 1088 et al. (2009). Fire in the Earth System. Science, 324(5926), 481–484.
- 1089 https://doi.org/10.1126/science.1163886
- 1090 Brock, C. A., Williamson, C., Kupc, A., Froyd, K. D., Erdesz, F., Wagner, N., et al. (2019).
- 1091 Aerosol size distributions during the Atmospheric Tomography Mission (ATom): methods,
- 1092 uncertainties, and data products. Atmospheric Measurement Techniques, 12(6), 3081–3099.
- 1093 https://doi.org/10.5194/amt-12-3081-2019
- Browne, E. C., Perring, A. E., Wooldridge, P. J., Apel, E., Hall, S. R., Huey, L. G., et al.
- 1095 (2011). Global and regional effects of the photochemistry of  $CH_3O_2NO_2$ : evidence from
- 1096 ARCTAS. Atmospheric Chemistry and Physics, 11(9), 4209–4219.
- 1097 https://doi.org/10.5194/acp-11-4209-2011
- 1098 Burke, M., Driscoll, A., Heft-Neal, S., Xue, J., Burney, J., & Wara, M. (2021). The changing
- 1099 risk and burden of wildfire in the United States. Proceedings of the National Academy of
- 1100 Sciences, 118(2). https://doi.org/10.1073/pnas.2011048118
- Canagaratna, M. R., Jayne, J. T., Jimenez, J. L., Allan, J. D., Alfarra, M. R., Zhang, Q., et al.
- 1102 (2007). Chemical and microphysical characterization of ambient aerosols with the aerodyne
- aerosol mass spectrometer. Mass Spectrometry Reviews, 26(2), 185–222.
- 1104 https://doi.org/10.1002/mas.20115
- 1105 Cazorla, M., Wolfe, G. M., Bailey, S. A., Swanson, A. K., Arkinson, H. L., & Hanisco, T. F.
- 1106 (2015). A new airborne laser-induced fluorescence instrument for in situ detection of
- 1107 formaldehyde throughout the troposphere and lower stratosphere. Atmospheric Measurement
- 1108 Techniques, 8(2), 541–552. https://doi.org/10.5194/amt-8-541-2015
- 1109 Chen, J., Wenger, J. C., & Venables, D. S. (2011). Near-Ultraviolet Absorption Cross
- 1110 Sections of Nitrophenols and Their Potential Influence on Tropospheric Oxidation Capacity.
- 1111 *The Journal of Physical Chemistry A, 115(44), 12235–12242.*
- 1112 https://doi.org/10.1021/jp206929r
- 1113 Collier, S., Zhou, S., Onasch, T. B., Jaffe, D. A., Kleinman, L., Sedlacek, A. J., et al. (2016).
- 1114 Regional Influence of Aerosol Emissions from Wildfires Driven by Combustion Efficiency:
- 1115 Insights from the BBOP Campaign. Environmental Science & Technology, 50(16), 8613–
- 1116 *8622. https://doi.org/10.1021/acs.est.6b01617*

- 1117 Crilley, L. R., Kramer, L. J., Ouyang, B., Duan, J., Zhang, W., Tong, S., et al. (2019).
- 1118 Intercomparison of nitrous acid (HONO) measurement techniques in a megacity (Beijing).
- 1119 Atmospheric Measurement Techniques, 12(12), 6449–6463. https://doi.org/10.5194/amt-12-
- 1120 *6449-2019*
- 1121 Crounse, J. D., McKinney, K. A., Kwan, A. J., & Wennberg, P. O. (2006). Measurement of
- 1122 Gas-Phase Hydroperoxides by Chemical Ionization Mass Spectrometry. Analytical
- 1123 Chemistry, 78(19), 6726–6732. https://doi.org/10.1021/ac0604235
- 1124 Crutzen, P. J., & Andreae, M. O. (2016). Biomass Burning in the Tropics: Impact on
- 1125 Atmospheric Chemistry and Biogeochemical Cycles. In P. J. Crutzen & H. G. Brauch (Eds.),
- 1126 Paul J. Crutzen: A Pioneer on Atmospheric Chemistry and Climate Change in the
- 1127 Anthropocene (pp. 165–188). Cham: Springer International Publishing.
- 1128 https://doi.org/10.1007/978-3-319-27460-7 7
- 1129 Day, D. A., Dillon, M. B., Wooldridge, P. J., Thornton, J. A., Rosen, R. S., Wood, E. C., &
- 1130 Cohen, R. C. (2003). On alkyl nitrates, O3, and the "missing NOy." Journal of Geophysical
- 1131 Research: Atmospheres, 108(D16). https://doi.org/10.1029/2003JD003685
- 1132 Day, D. A., Campuzano-Jost, P., Nault, B. A., Palm, B. B., Hu, W., Guo, H., et al. (2021). A
- 1133 Systematic Re-evaluation of Methods for Quantification of Bulk Particle-phase Organic
- 1134 Nitrates Using Real-time Aerosol Mass Spectrometry. Atmospheric Measurement Techniques
- 1135 Discussions, 1–35. https://doi.org/10.5194/amt-2021-263
- 1136 DeCarlo, P. F., Kimmel, J. R., Trimborn, A., Northway, M. J., Jayne, J. T., Aiken, A. C., et al.
- 1137 (2006). Field-Deployable, High-Resolution, Time-of-Flight Aerosol Mass Spectrometer.
- 1138 Analytical Chemistry, 78(24), 8281–8289. https://doi.org/10.1021/ac061249n
- 1139 Decker, Z. C. J., Robinson, M. A., Barsanti, K. C., Bourgeois, I., Coggon, M. M., DiGangi, J.
- 1140 P., et al. (2021). Nighttime and daytime dark oxidation chemistry in wildfire plumes: an
- observation and model analysis of FIREX-AQ aircraft data. Atmospheric Chemistry and
- 1142 Physics, 21(21), 16293–16317. https://doi.org/10.5194/acp-21-16293-2021
- 1143 Dennis, A., Fraser, M., Anderson, S., & Allen, D. (2002). Air pollutant emissions associated
- with forest, grassland, and agricultural burning in Texas. Atmospheric Environment, 36(23),
- 1145 *3779–3792. https://doi.org/10.1016/S1352-2310(02)00219-4*
- 1146 Dibb, J. E., Talbot, R. W., Seid, G., Jordan, C., Scheuer, E., Atlas, E., et al. (2002). Airborne
- sampling of aerosol particles: Comparison between surface sampling at Christmas Island
- and P-3 sampling during PEM-Tropics B. Journal of Geophysical Research: Atmospheres,
- 1149 107(D2), PEM 2-1-PEM 2-17. https://doi.org/10.1029/2001JD000408
- 1150 Diskin, G. S., Podolske, J. R., Sachse, G. W., & Slate, T. A. (2002). Open-path airborne
- tunable diode laser hygrometer. In Diode Lasers and Applications in Atmospheric Sensing
- 1152 (Vol. 4817, pp. 196–204). International Society for Optics and Photonics.
- 1153 https://doi.org/10.1117/12.453736
- Eisele, F. L., Mauldin, L., Cantrell, C., Zondlo, M., Apel, E., Fried, A., et al. (2003).
- 1155 Summary of measurement intercomparisons during TRACE-P. Journal of Geophysical
- 1156 Research: Atmospheres, 108(D20). https://doi.org/10.1029/2002JD003167
- 1157 Fahey, D. W., Eubank, C. S., Hübler, G., & Fehsenfeld, F. C. (1985). Evaluation of a
- catalytic reduction technique for the measurement of total reactive odd-nitrogen NO y in the
- atmosphere. Journal of Atmospheric Chemistry, 3(4), 435–468.
- 1160 Fahey, D. W., Hübler, G., Parrish, D. D., Williams, E. J., Norton, R. B., Ridley, B. A., et al.
- 1161 (1986). Reactive nitrogen species in the troposphere: Measurements of NO, NO2, HNO3,

- particulate nitrate, peroxyacetyl nitrate (PAN), O3, and total reactive odd nitrogen (NO y ) at
- Niwot Ridge, Colorado. Journal of Geophysical Research: Atmospheres, 91(D9), 9781–9793.
- 1164 https://doi.org/10.1029/JD091iD09p09781
- 1165 Farmer, D. K., Matsunaga, A., Docherty, K. S., Surratt, J. D., Seinfeld, J. H., Ziemann, P. J.,
- 1166 & Jimenez, J. L. (2010). Response of an aerosol mass spectrometer to organonitrates and
- organosulfates and implications for atmospheric chemistry. Proceedings of the National
- 1168 Academy of Sciences, 107(15), 6670–6675. https://doi.org/10.1073/pnas.0912340107
- 1169 Fiedler, S. E., Hese, A., & Ruth, A. A. (2003). Incoherent broad-band cavity-enhanced
- absorption spectroscopy. Chemical Physics Letters, 371(3), 284–294.
- 1171 https://doi.org/10.1016/S0009-2614(03)00263-X
- Fisher, J. A., Jacob, D. J., Travis, K. R., Kim, P. S., Marais, E. A., Chan Miller, C., et al.
- 1173 (2016). Organic nitrate chemistry and its implications for nitrogen budgets in an isoprene-
- and monoterpene-rich atmosphere: constraints from aircraft (SEAC<sup>4</sup>RS) and ground-based
- 1175 (SOAS) observations in the Southeast US. Atmospheric Chemistry and Physics, 16(9), 5969–
- 1176 *5991. https://doi.org/10.5194/acp-16-5969-2016*
- 1177 Fontijn, Arthur., Sabadell, A. J., & Ronco, R. J. (1970). Homogeneous chemiluminescent
- measurement of nitric oxide with ozone. Implications for continuous selective monitoring of
- gaseous air pollutants. Analytical Chemistry, 42(6), 575–579.
- 1180 https://doi.org/10.1021/ac60288a034
- 1181 Fry, J. L., Draper, D. C., Zarzana, K. J., Campuzano-Jost, P., Day, D. A., Jimenez, J. L., et
- al. (2013). Observations of gas- and aerosol-phase organic nitrates at BEACHON-RoMBAS
- 1183 2011. Atmospheric Chemistry and Physics, 13(17), 8585–8605. https://doi.org/10.5194/acp-
- 1184 *13-8585-2013*
- Galloway, J. N., Aber, J. D., Erisman, J. W., Seitzinger, S. P., Howarth, R. W., Cowling, E.
- 1186 B., & Cosby, B. J. (2003). The Nitrogen Cascade. BioScience, 53(4), 341.
- 1187 https://doi.org/10.1641/0006-3568(2003)053[0341:TNC]2.0.CO;2
- 1188 Gao, R. S., McLaughlin, R. J., Schein, M. E., Neuman, J. A., Ciciora, S. J., Holecek, J. C., &
- 1189 Fahey, D. W. (1999). Computer-controlled Teflon flow control valve. Review of Scientific
- 1190 Instruments, 70(12), 4732–4733. https://doi.org/10.1063/1.1150137
- 1191 Gregory, G. L., Hoell, J. M., Torres, A. L., Carroll, M. A., Ridley, B. A., Rodgers, M. O., et
- al. (1990). An intercomparison of airborne nitric oxide measurements: A second opportunity.
- Journal of Geophysical Research: Atmospheres, 95(D7), 10129–10138.
- 1194 https://doi.org/10.1029/JD095iD07p10129
- 1195 Gregory, G. L., Hoell, J. M., Carroll, M. A., Ridley, B. A., Davis, D. D., Bradshaw, J., et al.
- 1196 (1990). An intercomparison of airborne nitrogen dioxide instruments. Journal of Geophysical
- 1197 Research: Atmospheres, 95(D7), 10103–10127. https://doi.org/10.1029/JD095iD07p10103
- Guo, H., Campuzano-Jost, P., Pagonis, D., Schueneman, M., Day, D. A., Nault, B. A., et al.
- 1199 (2020). Submicron Particle Composition and Acidity in Fire Plumes during FIREX-AQ
- 1200 aircraft study, 2020, A232-10. Presented at the AGU Fall Meeting Abstracts.
- 1201 Guo, Hongyu, Campuzano-Jost, P., Nault, B. A., Day, D. A., Schroder, J. C., Kim, D., et al.
- 1202 (2021). The importance of size ranges in aerosol instrument intercomparisons: a case study
- 1203 for the Atmospheric Tomography Mission. Atmospheric Measurement Techniques, 14(5),

- 1205 Hall, B. D., Dutton, G. S., & Elkins, J. W. (2007). The NOAA nitrous oxide standard scale for
- 1206 atmospheric observations. Journal of Geophysical Research: Atmospheres, 112(D9).
- 1207 https://doi.org/10.1029/2006JD007954
- 1208 Harder, J. W., & Brault, J. W. (1997). Atmospheric measurements of water vapor in the 442-
- 1209 nm region. Journal of Geophysical Research: Atmospheres, 102(D5), 6245–6252.
- 1210 https://doi.org/10.1029/96JD01730
- 1211 Hayden, K. L., Anlauf, K. G., Hastie, D. R., & Bottenheim, J. W. (2003). Partitioning of
- reactive atmospheric nitrogen oxides at an elevated site in southern Quebec, Canada.
- 1213 Journal of Geophysical Research: Atmospheres, 108(D19).
- 1214 https://doi.org/10.1029/2002JD003188
- 1215 Hoell, J. M., Gregory, G. L., McDougal, D. S., Sachse, G. W., Hill, G. F., Condon, E. P., &
- 1216 Rasmussen, R. A. (1987). Airborne intercomparison of carbon monoxide measurement
- techniques. Journal of Geophysical Research: Atmospheres, 92(D2), 2009–2019.
- 1218 https://doi.org/10.1029/JD092iD02p02009
- 1219 Hoell, J. M., Gregory, G. L., McDougal, D. S., Torres, A. L., Davis, D. D., Bradshaw, J., et
- 1220 al. (1987). Airborne intercomparison of nitric oxide measurement techniques. Journal of
- 1221 Geophysical Research: Atmospheres, 92(D2), 1995–2008.
- 1222 https://doi.org/10.1029/JD092iD02p01995
- 1223 Holloway, J. S., Jakoubek, R. O., Parrish, D. D., Gerbig, C., Volz Thomas, A., Schmitgen,
- 1224 S., et al. (2000). Airborne intercomparison of vacuum ultraviolet fluorescence and tunable
- diode laser absorption measurements of tropospheric carbon monoxide. Journal of
- 1226 Geophysical Research: Atmospheres, 105(D19), 24251–24261.
- 1227 https://doi.org/10.1029/2000JD900237
- 1228 Holmes, C. D., Fite, C., Agastra, A., Schwarz, J. P., Yokelson, R. J., Bui, T. V., & Peterson,
- 1229 D. A. (2020). Critical evaluation of smoke age inferred from different methods during
- 1230 FIREX-AQ, 2020, A225-0010. Presented at the AGU Fall Meeting Abstracts.
- 1231 Jaffe, D. A., O'Neill, S. M., Larkin, N. K., Holder, A. L., Peterson, D. L., Halofsky, J. E., &
- 1232 Rappold, A. G. (2020). Wildfire and prescribed burning impacts on air quality in the United
- 1233 States. Journal of the Air & Waste Management Association, 70(6), 583–615.
- 1234 https://doi.org/10.1080/10962247.2020.1749731
- Johnston, F. H., Henderson Sarah B., Chen Yang, Randerson James T., Marlier Miriam,
- 1236 DeFries Ruth S., et al. (2012). Estimated Global Mortality Attributable to Smoke from
- 1237 Landscape Fires. Environmental Health Perspectives, 120(5), 695–701.
- 1238 https://doi.org/10.1289/ehp.1104422
- 1239 Johnston, F. H., Borchers-Arriagada, N., Morgan, G. G., Jalaludin, B., Palmer, A. J.,
- Williamson, G. J., & Bowman, D. M. J. S. (2021). Unprecedented health costs of smoke-
- related PM2.5 from the 2019–20 Australian megafires. Nature Sustainability, 4(1), 42–47.
- 1242 https://doi.org/10.1038/s41893-020-00610-5
- Juncosa Calahorrano, J. F., Lindaas, J., O'Dell, K., Palm, B. B., Peng, Q., Flocke, F., et al.
- 1244 (2021). Daytime Oxidized Reactive Nitrogen Partitioning in Western U.S. Wildfire Smoke
- 1245 Plumes. Journal of Geophysical Research: Atmospheres, 126, e2020JD033484.
- 1246 https://doi.org/10.1029/2020JD033484
- 1247 Keller-Rudek, H., Moortgat, G. K., Sander, R., & Sörensen, R. (2013). The MPI-Mainz
- 1248 UV/VIS Spectral Atlas of Gaseous Molecules of Atmospheric Interest. Earth System Science
- 1249 Data, 5(2), 365–373. https://doi.org/10.5194/essd-5-365-2013

- 1250 Kenagy, H. S., Sparks, T. L., Ebben, C. J., Wooldrige, P. J., Lopez-Hilfiker, F. D., Lee, B. H.,
- 1251 et al. (2018). NOx Lifetime and NOy Partitioning During WINTER. Journal of Geophysical
- 1252 Research: Atmospheres, 123(17), 9813–9827. https://doi.org/10.1029/2018JD028736
- 1253 Kleffmann, J., Lörzer, J. C., Wiesen, P., Kern, C., Trick, S., Volkamer, R., et al. (2006).
- 1254 Intercomparison of the DOAS and LOPAP techniques for the detection of nitrous acid
- 1255 (HONO). Atmospheric Environment, 40(20), 3640–3652.
- 1256 https://doi.org/10.1016/j.atmosenv.2006.03.027
- 1257 Kleffmann, Jörg. (2007). Daytime Sources of Nitrous Acid (HONO) in the Atmospheric
- 1258 Boundary Layer. ChemPhysChem, 8(8), 1137–1144. https://doi.org/10.1002/cphc.200700016
- 1259 Kroll, J. H., & Seinfeld, J. H. (2008). Chemistry of secondary organic aerosol: Formation
- and evolution of low-volatility organics in the atmosphere. Atmospheric Environment,
- 1261 42(16), 3593–3624. https://doi.org/10.1016/j.atmosenv.2008.01.003
- 1262 Lao, M., Crilley, L. R., Salehpoor, L., Furlani, T. C., Bourgeois, I., Neuman, J. A., et al.
- 1263 (2020). A portable, robust, stable, and tunable calibration source for gas-phase nitrous acid
- 1264 (HONO). Atmospheric Measurement Techniques, 13(11), 5873–5890.
- 1265 https://doi.org/10.5194/amt-13-5873-2020
- 1266 Lareau, N. P., Nauslar, N. J., & Abatzoglou, J. T. (2018). The Carr Fire Vortex: A Case of
- 1267 Pyrotornadogenesis? Geophysical Research Letters, 45(23), 13,107-13,115.
- 1268 https://doi.org/10.1029/2018GL080667
- 1269 Lee, B. H., Lopez-Hilfiker, F. D., Mohr, C., Kurtén, T., Worsnop, D. R., & Thornton, J. A.
- 1270 (2014). An Iodide-Adduct High-Resolution Time-of-Flight Chemical-Ionization Mass
- 1271 Spectrometer: Application to Atmospheric Inorganic and Organic Compounds.
- 1272 Environmental Science & Technology, 48(11), 6309–6317.
- 1273 https://doi.org/10.1021/es500362a
- 1274 Lee, Y. R., Ji, Y., Tanner, D. J., & Huey, L. G. (2020). A low-activity ion source for
- measurement of atmospheric gases by chemical ionization mass spectrometry. Atmospheric
- 1276 Measurement Techniques, 13(5), 2473–2480. https://doi.org/10.5194/amt-13-2473-2020
- 1277 Lerner, B. M., Gilman, J. B., Aikin, K. C., Atlas, E. L., Goldan, P. D., Graus, M., et al.
- 1278 (2017). An improved, automated whole air sampler and gas chromatography mass
- spectrometry analysis system for volatile organic compounds in the atmosphere. Atmospheric
- 1280 Measurement Techniques, 10(1), 291–313. https://doi.org/10.5194/amt-10-291-2017
- 1281 Liousse, C., Guillaume, B., Grégoire, J. M., Mallet, M., Galy, C., Pont, V., et al. (2010).
- 1282 Updated African biomass burning emission inventories in the framework of the AMMA-IDAF
- program, with an evaluation of combustion aerosols. Atmospheric Chemistry and Physics,
- 1284 10(19), 9631–9646. https://doi.org/10.5194/acp-10-9631-2010
- 1285 McCarty, J. L. (2011). Remote Sensing-Based Estimates of Annual and Seasonal Emissions
- 1286 from Crop Residue Burning in the Contiguous United States. Journal of the Air & Waste
- 1287 Management Association, 61(1), 22–34. https://doi.org/10.3155/1047-3289.61.1.22
- 1288 Meller, R., Raber, W., Crowley, J. N., Jenkin, M. E., & Moortgat, G. K. (1991). The UV-
- 1289 visible absorption spectrum of methylglyoxal. Journal of Photochemistry and Photobiology
- 1290 A: Chemistry, 62(2), 163–171. https://doi.org/10.1016/1010-6030(91)87017-P
- Melvin, M. A. (2020). 2020 National Precribed Fire Use Report. National Association of
- 1292 State Foresters. Retrieved from https://www.stateforesters.org/wp-
- 1293 content/uploads/2020/12/2020-Prescribed-Fire-Use-Report.pdf

- 1294 Min, K.-E., Washenfelder, R. A., Dubé, W. P., Langford, A. O., Edwards, P. M., Zarzana, K.
- 1295 J., et al. (2016). A broadband cavity enhanced absorption spectrometer for aircraft
- measurements of glyoxal, methylglyoxal, nitrous acid, nitrogen dioxide, and water vapor.
- 1297 Atmospheric Measurement Techniques, 9(2), 423–440. https://doi.org/10.5194/amt-9-423-
- 1298 *2016*
- 1299 Miyazaki, Y., Kondo, Y., Takegawa, N., Weber, R. J., Koike, M., Kita, K., et al. (2005).
- 1300 Contribution of particulate nitrate to airborne measurements of total reactive nitrogen.
- 1301 Journal of Geophysical Research: Atmospheres, 110(D15).
- 1302 https://doi.org/10.1029/2004JD005502
- 1303 Moore, R. H., Wiggins, E. B., Ahern, A. T., Zimmerman, S., Montgomery, L., Campuzano
- 1304 Jost, P., et al. (2021). Sizing response of the Ultra-High Sensitivity Aerosol Spectrometer
- 1305 (UHSAS) and Laser Aerosol Spectrometer (LAS) to changes in submicron aerosol
- 1306 composition and refractive index. Atmospheric Measurement Techniques, 14(6), 4517–4542.
- 1307 https://doi.org/10.5194/amt-14-4517-2021
- 1308 Murphy, J. G., Thornton, J. A., Wooldridge, P. J., Day, D. A., Rosen, R. S., Cantrell, C., et al.
- 1309 (2004). Measurements of the sum of  $HO_2NO_2$  and  $CH_3O_2NO_2$  in the remote troposphere.
- 1310 Atmospheric Chemistry and Physics, 4(2), 377–384. https://doi.org/10.5194/acp-4-377-2004
- Nault, B. A., Garland, C., Pusede, S. E., Wooldridge, P. J., Ullmann, K., Hall, S. R., &
- Cohen, R. C. (2015). Measurements of  $CH_3O_2NO_2$  in the upper troposphere. Atmospheric
- 1313 Measurement Techniques, 8(2), 987–997. https://doi.org/10.5194/amt-8-987-2015
- Neuman, J. A., Trainer, M., Brown, S. S., Min, K.-E., Nowak, J. B., Parrish, D. D., et al.
- 1315 (2016). HONO emission and production determined from airborne measurements over the
- 1316 Southeast U.S. Journal of Geophysical Research: Atmospheres, 121(15), 2016JD025197.
- 1317 https://doi.org/10.1002/2016JD025197
- Ninneman, M., Marto, J., Shaw, S., Edgerton, E., Blanchard, C., & Schwab, J. (2021).
- 1319 Reactive oxidized nitrogen speciation and partitioning in urban and rural New York State.
- 1320 Journal of the Air & Waste Management Association, 71(3), 348–365.
- 1321 https://doi.org/10.1080/10962247.2020.1837289
- Novelli, P. C., Elkins, J. W., & Steele, L. P. (1991). The development and evaluation of a
- 1323 gravimetric reference scale for measurements of atmospheric carbon monoxide. Journal of
- 1324 Geophysical Research: Atmospheres, 96(D7), 13109–13121.
- 1325 https://doi.org/10.1029/91JD01108
- 1326 O'Dell, K., Ford, B., Fischer, E. V., & Pierce, J. R. (2019). Contribution of Wildland-Fire
- 1327 Smoke to US PM2.5 and Its Influence on Recent Trends. Environmental Science &
- 1328 Technology, 53(4), 1797–1804. https://doi.org/10.1021/acs.est.8b05430
- 1329 Pagonis, D., Campuzano-Jost, P., Guo, H., Day, D. A., Schueneman, M. K., Brown, W. L., et
- al. (2021). Airborne extractive electrospray mass spectrometry measurements of the chemical
- 1331 composition of organic aerosol. Atmospheric Measurement Techniques, 14(2), 1545–1559.
- 1332 https://doi.org/10.5194/amt-14-1545-2021
- 1333 Peischl, J., Ryerson, T. B., Holloway, J. S., Parrish, D. D., Trainer, M., Frost, G. J., et al.
- 1334 (2010). A top-down analysis of emissions from selected Texas power plants during TexAQS
- 1335 2000 and 2006. Journal of Geophysical Research: Atmospheres, 115(D16).
- 1336 https://doi.org/10.1029/2009JD013527
- 1337 Peng, O., Palm, B. B., Melander, K. E., Lee, B. H., Hall, S. R., Ullmann, K., et al. (2020).
- 1338 HONO Emissions from Western U.S. Wildfires Provide Dominant Radical Source in Fresh

- 1339 Wildfire Smoke. Environmental Science & Technology.
- 1340 https://doi.org/10.1021/acs.est.0c00126
- 1341 Pinto, J. P., Dibb, J., Lee, B. H., Rappenglück, B., Wood, E. C., Levy, M., et al. (2014).
- 1342 Intercomparison of field measurements of nitrous acid (HONO) during the SHARP
- campaign. Journal of Geophysical Research: Atmospheres, 119(9), 5583–5601.
- 1344 https://doi.org/10.1002/2013JD020287
- 1345 Platt, U., Perner, D., Harris, G. W., Winer, A. M., & Pitts, J. N. (1980). Observations of
- nitrous acid in an urban atmosphere by differential optical absorption. Nature, 285(5763),
- 1347 *312–314. https://doi.org/10.1038/285312a0*
- 1348 Podolske, J. R., Sachse, G. W., & Diskin, G. S. (2003). Calibration and data retrieval
- 1349 algorithms for the NASA Langley/Ames Diode Laser Hygrometer for the NASA Transport
- and Chemical Evolution Over the Pacific (TRACE-P) mission. Journal of Geophysical
- 1351 Research: Atmospheres, 108(D20). https://doi.org/10.1029/2002JD003156
- 1352 Pollack, I. B., Homeyer, C. R., Ryerson, T. B., Aikin, K. C., Peischl, J., Apel, E. C., et al.
- 1353 (2016). Airborne quantification of upper tropospheric NOx production from lightning in deep
- 1354 convective storms over the United States Great Plains. Journal of Geophysical Research:
- 1355 Atmospheres, 121(4), 2002–2028. https://doi.org/10.1002/2015JD023941
- 1356 Pollack, Ilana B., Lerner, B. M., & Ryerson, T. B. (2010). Evaluation of ultraviolet light-
- emitting diodes for detection of atmospheric NO2 by photolysis chemiluminescence. Journal
- 1358 of Atmospheric Chemistry, 65(2–3), 111–125. https://doi.org/10.1007/s10874-011-9184-3
- 1359 Ridley, B. A., & Grahek, F. E. (1990). A Small, Low Flow, High Sensitivity Reaction Vessel
- 1360 for NO Chemiluminescence Detectors. Journal of Atmospheric and Oceanic Technology,
- 1361 7(2), 307–311. https://doi.org/10.1175/1520-0426(1990)007<0307:ASLFHS>2.0.CO;2
- 1362 Ridley, B. A., & Howlett, L. C. (1974). An instrument for nitric oxide measurements in the
- stratosphere. Review of Scientific Instruments, 45(6), 742–746.
- 1364 https://doi.org/10.1063/1.1686726
- 1365 Ridley, B. A., Grahek, F. E., & Walega, J. G. (1992). A Small High-Sensitivity, Medium-
- 1366 Response Ozone Detector Suitable for Measurements from Light Aircraft. Journal of
- 1367 Atmospheric and Oceanic Technology, 9(2), 142–148. https://doi.org/10.1175/1520-
- 1368 0426(1992)009<0142:ASHSMR>2.0.CO;2
- 1369 Roberts, J. M., Stockwell, C. E., Yokelson, R. J., de Gouw, J., Liu, Y., Selimovic, V., et al.
- 1370 (2020). The nitrogen budget of laboratory-simulated western US wildfires during the FIREX
- 1371 *2016 Fire Lab study. Atmospheric Chemistry and Physics*, *20(14)*, *8807–8826*.
- 1372 https://doi.org/10.5194/acp-20-8807-2020
- 1373 Robinson, M. A., Decker, Z. C. J., Barsanti, K. C., Coggon, M. M., Flocke, F. M., Franchin,
- 1374 A., et al. (2021). Variability and Time of Day Dependence of Ozone Photochemistry in
- 1375 Western Wildfire Plumes. Environmental Science & Technology, 55(15), 10280–10290.
- 1376 https://doi.org/10.1021/acs.est.1c01963
- 1377 Rollins, A. W., Rickly, P. S., Gao, R.-S., Ryerson, T. B., Brown, S. S., Peischl, J., &
- 1378 Bourgeois, I. (2020). Single-photon laser-induced fluorescence detection of nitric oxide at
- 1379 sub-parts-per-trillion mixing ratios. Atmospheric Measurement Techniques, 13(5), 2425–
- 1380 2439. https://doi.org/10.5194/amt-13-2425-2020
- 1381 Russo, R. S., Zhou, Y., Haase, K. B., Wingenter, O. W., Frinak, E. K., Mao, H., et al. (2010).
- 1382 Temporal variability, sources, and sinks of  $C_1$ - $C_5$  alkyl nitrates in coastal New England.

- 1383 Atmospheric Chemistry and Physics, 10(4), 1865–1883. https://doi.org/10.5194/acp-10-
- 1384 *1865-2010*
- 1385 Ryerson, T. B., Huey, L. G., Knapp, K., Neuman, J. A., Parrish, D. D., Sueper, D. T., &
- 1386 Fehsenfeld, F. C. (1999). Design and initial characterization of an inlet for gas-phase NOy
- measurements from aircraft. Journal of Geophysical Research: Atmospheres, 104(D5),
- 1388 *5483–5492. https://doi.org/10.1029/1998JD100087*
- Ryerson, T. B., Williams, E. J., & Fehsenfeld, F. C. (2000). An efficient photolysis system for
- 1390 fast-response NO2 measurements. Journal of Geophysical Research: Atmospheres,
- 1391 105(D21), 26447–26461. https://doi.org/10.1029/2000JD900389
- 1392 Sachse, G. W., Hill, G. F., Wade, L. O., & Perry, M. G. (1987). Fast-response, high-precision
- carbon monoxide sensor using a tunable diode laser absorption technique. Journal of
- 1394 Geophysical Research: Atmospheres, 92(D2), 2071–2081.
- 1395 https://doi.org/10.1029/JD092iD02p02071
- 1396 Selimovic, V., Yokelson, R. J., Warneke, C., Roberts, J. M., de Gouw, J., Reardon, J., &
- 1397 Griffith, D. W. T. (2018). Aerosol optical properties and trace gas emissions by PAX and OP-
- 1398 FTIR for laboratory-simulated western US wildfires during FIREX. Atmospheric Chemistry
- 1399 and Physics, 18(4), 2929–2948. https://doi.org/10.5194/acp-18-2929-2018
- 1400 Slusher, D. L., Huey, L. G., Tanner, D. J., Flocke, F. M., & Roberts, J. M. (2004). A thermal
- 1401 dissociation—chemical ionization mass spectrometry (TD-CIMS) technique for the
- simultaneous measurement of peroxyacyl nitrates and dinitrogen pentoxide. Journal of
- 1403 Geophysical Research: Atmospheres, 109(D19). https://doi.org/10.1029/2004JD004670
- 1404 Sparks, T. L., Ebben, C. J., Wooldridge, P. J., Lopez Hilfiker, F. D., Lee, B. H., Thornton, J.
- 1405 A., et al. (2019). Comparison of Airborne Reactive Nitrogen Measurements During WINTER.
- 1406 Journal of Geophysical Research: Atmospheres, 124(19), 10483–10502.
- 1407 https://doi.org/10.1029/2019JD030700
- 1408 St. Clair, J. M., Swanson, A. K., Bailey, S. A., & Hanisco, T. F. (2019). CAFE: a new,
- improved nonresonant laser-induced fluorescence instrument for airborne in situ
- measurement of formaldehyde. Atmospheric Measurement Techniques, 12(8), 4581–4590.
- 1411 https://doi.org/10.5194/amt-12-4581-2019
- 1412 Stein, A. F., Draxler, R. R., Rolph, G. D., Stunder, B. J. B., Cohen, M. D., & Ngan, F. (2015).
- 1413 NOAA's HYSPLIT Atmospheric Transport and Dispersion Modeling System. Bulletin of the
- 1414 American Meteorological Society, 96(12), 2059–2077. https://doi.org/10.1175/BAMS-D-14-
- 1415 00110.1
- 1416 Stockwell, C. E., Yokelson, R. J., Kreidenweis, S. M., Robinson, A. L., DeMott, P. J., Sullivan,
- 1417 R. C., et al. (2014). Trace gas emissions from combustion of peat, crop residue, domestic
- biofuels, grasses, and other fuels: configuration and Fourier transform infrared (FTIR)
- 1419 component of the fourth Fire Lab at Missoula Experiment (FLAME-4). Atmospheric
- 1420 Chemistry and Physics, 14(18), 9727–9754. https://doi.org/10.5194/acp-14-9727-2014
- 1421 Stutz, J., Kim, E. S., Platt, U., Bruno, P., Perrino, C., & Febo, A. (2000). UV-visible
- absorption cross sections of nitrous acid. Journal of Geophysical Research: Atmospheres,
- 1423 105(D11), 14585–14592. https://doi.org/10.1029/2000JD900003
- 1424 Theys, N., Volkamer, R., Müller, J.-F., Zarzana, K. J., Kille, N., Clarisse, L., et al. (2020).
- 1425 Global nitrous acid emissions and levels of regional oxidants enhanced by wildfires. Nature
- 1426 Geoscience, 13(10), 681–686. https://doi.org/10.1038/s41561-020-0637-7

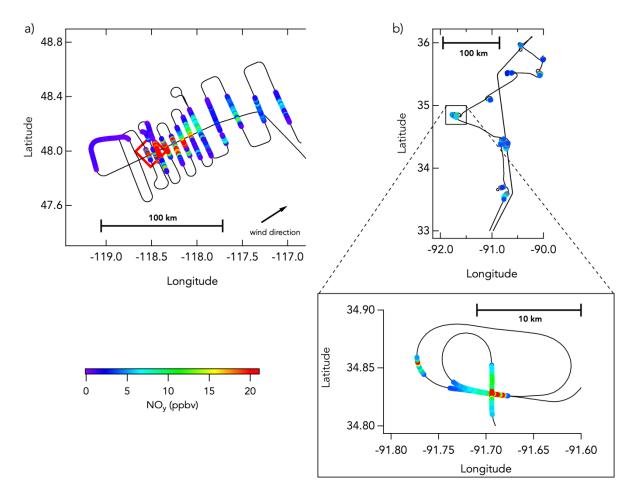
- 1427 Vandaele, A. C., Hermans, C., Simon, P. C., Carleer, M., Colin, R., Fally, S., et al. (1998).
- 1428 Measurements of the NO2 absorption cross-section from 42 000 cm-1 to 10 000 cm-1 (238–
- 1429 1000 nm) at 220 K and 294 K. Journal of Quantitative Spectroscopy and Radiative Transfer,
- 1430 *59(3), 171–184. https://doi.org/10.1016/S0022-4073(97)00168-4*
- 1431 Veres, P. R., Neuman, J. A., Bertram, T. H., Assaf, E., Wolfe, G. M., Williamson, C. J., et al.
- 1432 (2020). Global airborne sampling reveals a previously unobserved dimethyl sulfide oxidation
- mechanism in the marine atmosphere. Proceedings of the National Academy of Sciences,
- 1434 117(9), 4505–4510. https://doi.org/10.1073/pnas.1919344117
- 1435 *Volkamer, R., Molina, L. T., Molina, M. J., Shirley, T., & Brune, W. H. (2005). DOAS*
- 1436 measurement of glyoxal as an indicator for fast VOC chemistry in urban air. Geophysical
- 1437 Research Letters, 32(8). https://doi.org/10.1029/2005GL022616
- 1438 Volpe Horii, C., William Munger, J., Wofsy, S. C., Zahniser, M., Nelson, D., & Barry
- 1439 McManus, J. (2005). Atmospheric reactive nitrogen concentration and flux budgets at a
- Northeastern U.S. forest site. Agricultural and Forest Meteorology, 133(1), 210–225.
- 1441 https://doi.org/10.1016/j.agrformet.2004.08.009
- Warneke, C., Trainer, M., Gouw, J. A. de, Parrish, D. D., Fahey, D. W., Ravishankara, A. R.,
- 1443 et al. (2016). Instrumentation and measurement strategy for the NOAA SENEX aircraft
- campaign as part of the Southeast Atmosphere Study 2013. Atmospheric Measurement
- 1445 Techniques, 9(7), 3063–3093. https://doi.org/10.5194/amt-9-3063-2016
- Washenfelder, R. A., Langford, A. O., Fuchs, H., & Brown, S. S. (2008). Measurement of
- 1447 *glyoxal using an incoherent broadband cavity enhanced absorption spectrometer.*
- 1448 Atmospheric Chemistry and Physics, 8(24), 7779–7793. https://doi.org/10.5194/acp-8-7779-
- 1449 2008
- 1450 von der Weiden, S.-L., Drewnick, F., & Borrmann, S. (2009). Particle Loss Calculator a
- 1451 *new software tool for the assessment of the performance of aerosol inlet systems.*
- 1452 Atmospheric Measurement Techniques, 2(2), 479–494. https://doi.org/10.5194/amt-2-479-
- 1453 *2009*
- 1454 van der Werf, G. R., Randerson, J. T., Giglio, L., Collatz, G. J., Mu, M., Kasibhatla, P. S., et
- 1455 al. (2010). Global fire emissions and the contribution of deforestation, savanna, forest,
- 1456 agricultural, and peat fires (1997–2009). Atmospheric Chemistry and Physics, 10(23),
- 1457 11707–11735. https://doi.org/10.5194/acp-10-11707-2010
- 1458 van der Werf, G. R., Randerson, J. T., Giglio, L., van Leeuwen, T. T., Chen, Y., Rogers, B. M.,
- et al. (2017). Global fire emissions estimates during 1997–2016. Earth System Science Data,
- 1460 9(2), 697–720. https://doi.org/10.5194/essd-9-697-2017
- Warneke, C., Schwarz, J. P., Dibb, J., Kalashnikova, O., Frost, G., Seidel, F., Al-Saadi, J.,
- 1462 Brown, S. S., Washenfelder, R., Brewer, A., Moore, R. H., Anderson, B. E., Yacovitch, T.,
- 1463 Herndon, S., Liu, S., Jaffe, D., Johnston, N., Selimovic, V., Yokelson, B., Giles, D., Holben,
- 1464 B., Goloub, P., Popovici, I., Trainer, M., Pierce, B., Fahey, D., Roberts, J., Soja, A.,
- 1465 Peterson, D., Saide, P. E., Holmes, C., Wang, S., Coggon, M. M., Decker, Z. C. J., Ye, X.,
- 1466 Stockwell, C., Xu, L., Gkatzelis, G., Lefer, B., Crawford, J. Fire Influence on Regional to
- 1467 Global Environments and Air Quality (FIREX-AQ). To be submitted. 2022
- 1468 Westerling, A. L. (2016). Increasing western US forest wildfire activity: sensitivity to changes
- in the timing of spring. Philosophical Transactions of the Royal Society B: Biological
- 1470 Sciences, 371(1696), 20150178. https://doi.org/10.1098/rstb.2015.0178

- Williams, E. J., Roberts, J. M., Baumann, K., Bertman, S. B., Buhr, S., Norton, R. B., &
- 1472 Fehsenfeld, F. C. (1997). Variations in NOy composition at Idaho Hill, Colorado. Journal of
- 1473 Geophysical Research: Atmospheres, 102(D5), 6297–6314.
- 1474 https://doi.org/10.1029/96JD03252
- 1475 Wolfe, G. M., Hanisco, T. F., Arkinson, H. L., Blake, D. R., Wisthaler, A., Mikoviny, T., et al.
- 1476 (2022). Photochemical evolution of the 2013 California Rim Fire: synergistic impacts of
- 1477 reactive hydrocarbons and enhanced oxidants. Atmospheric Chemistry and Physics, 22(6),
- 1478 *4253–4275.* https://doi.org/10.5194/acp-22-4253-2022
- 1479 Womack, C. C., Neuman, J. A., Veres, P. R., Eilerman, S. J., Brock, C. A., Decker, Z. C. J., et
- 1480 al. (2017). Evaluation of the accuracy of thermal dissociation CRDS and LIF techniques for
- 1481 atmospheric measurement of reactive nitrogen species. Atmos. Meas. Tech., 10(5), 1911–
- Wotawa, G. (2000). The Influence of Canadian Forest Fires on Pollutant Concentrations in
- 1484 the United States. Science, 288(5464), 324–328.
- 1485 https://doi.org/10.1126/science.288.5464.324
- 1486 Xu, L., Crounse, J. D., Vasquez, K., Allen, H. M., Wennberg, P. O., Bourgeois, I., et al.
- 1487 (2021). Ozone chemistry in Western U.S. wildfire plumes. Science Advances, in press.
- 1488 Xu, Z., Liu, Y., Nie, W., Sun, P., Chi, X., & Ding, A. (2019). Evaluating the measurement
- interference of wet rotating-denuder—ion chromatography in measuring atmospheric HONO
- in a highly polluted area. Atmospheric Measurement Techniques, 12(12), 6737–6748.
- 1491 https://doi.org/10.5194/amt-12-6737-2019
- 1492 Yokelson, R. J., Griffith, D. W. T., & Ward, D. E. (1996). Open-path Fourier transform
- infrared studies of large-scale laboratory biomass fires. Journal of Geophysical Research:
- 1494 Atmospheres, 101(D15), 21067–21080. https://doi.org/10.1029/96JD01800
- 1495 Zhang, L., Wiebe, A., Vet, R., Mihele, C., O'Brien, J. M., Iqbal, S., & Liang, Z. (2008).
- 1496 Measurements of reactive oxidized nitrogen at eight Canadian rural sites. Atmospheric
- 1497 Environment, 42(34), 8065–8078. https://doi.org/10.1016/j.atmosenv.2008.06.034
- 1498 Ziemann, P. J., & Atkinson, R. (2012). Kinetics, products, and mechanisms of secondary
- organic aerosol formation. Chemical Society Reviews, 41(19), 6582–6605.
- 1500 https://doi.org/10.1039/C2CS35122F

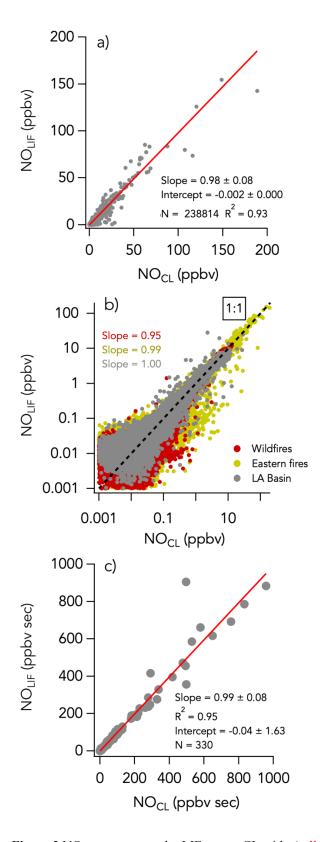
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SPECIES	<b>INSTRUMENT</b>	UNCERTAINTY
NO	CL	$\pm (4 \% + 6 \text{ pptv})$
	LIF	$\pm (8 \% + 1 \text{ pptv})$
NO <sub>2</sub>	CL	$\pm (7 \% + 20 \text{ pptv})$
	CES	$\pm (5\% + 0.26 \text{ ppbv})$
	LIF	$\pm (10\% + 100 \text{ pptv})$
HONO	CIMS	$\pm (15\% + 3 \text{ pptv})$
	CES	$\pm (9\% + 0.6 \text{ ppbv})$
NO <sub>y</sub>	CL	$\pm (12 \% + 15 \text{ pptv})$
	Sum	~ 25%
CO	TD-LAS	2–7%
	ICOS	$\pm (2.0 \text{ ppb} + 2\%)$

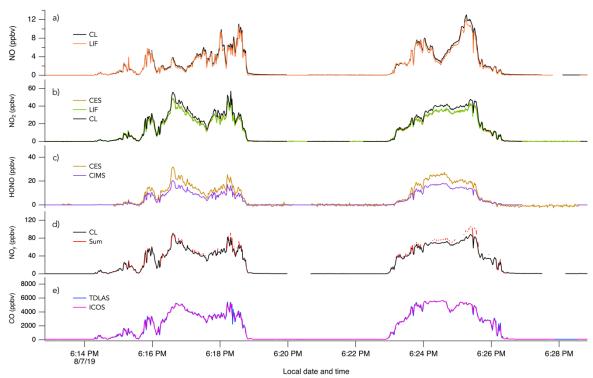
**Table 1** List of measured species and instruments, including the corresponding uncertainties, during FIREX-AQ. For uncertainties given as  $\pm$  (x %  $\pm$  y pptv), x represents the accuracy, and y represents the 2-sigma precision in 1 s.



**Figure 1** Examples DC-8 flight tracks from western wildfires and eastern agricultural fires. Panel a) shows the DC-8 flight track (black line) during the sampling of the Williams Flat fire (03/08/2019) smoke plume, colored by NO<sub>y</sub> mixing ratios (only data in smoke are colored here). Panel b) shows the DC-8 flight track during the sampling of multiple agricultural burns (21/08/2019), also colored by NO<sub>y</sub> mixing ratios (only data in smoke are colored here).



**Figure 2** NO measurements by LIF versus CL with a) all 1 s data on a linear scale, b) all 1 s data on a log scale, and c) integrals of 330 crosswind smoke plume transects. N is the number of independent 1 s observations or smoke plume transects that are compared. In panel b, the three sampling periods are shown in different colors with the wildfires sampling period in red, the eastern fires sampling period in mustard, and the Los Angeles (LA) Basin flights in grey. The red lines indicate the fit of the data. The dotted black line is the 1:1 line.



**Figure 3** 1 s measurements of a) NO, b) NO<sub>2</sub>, c) HONO, d) NO<sub>y</sub>, and e) CO during two crosswind plume transects of smoke from the Williams Flat fire on 07/08/2019. The plume transects were chosen due to the significant enhancement of all species at that time. Note that in panel b) the NO<sub>2</sub> trace from the CES instrument is hidden behind the NO<sub>2</sub> trace from the LIF instrument.

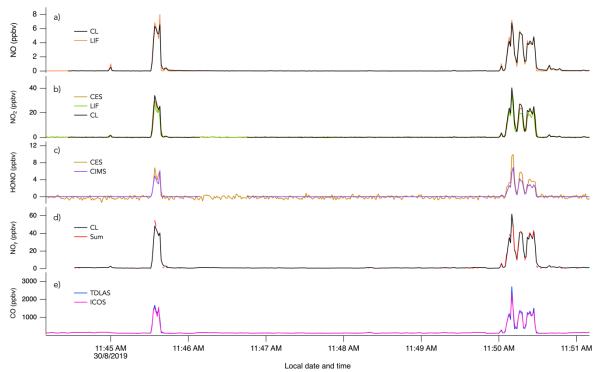
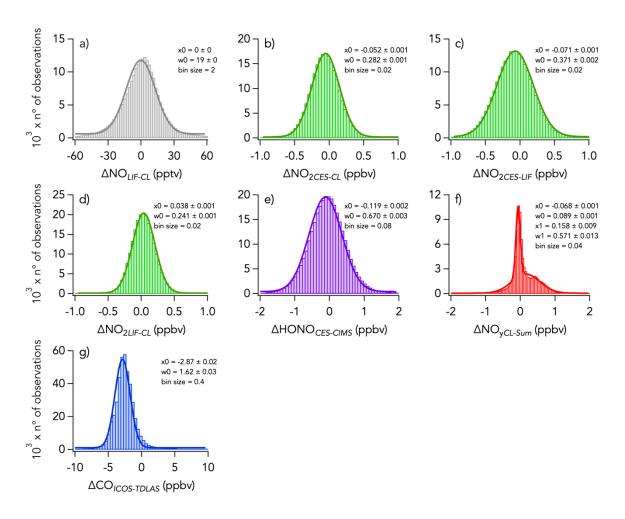


Figure 4 1 s measurements of a) NO, b) NO<sub>2</sub>, c) HONO, d) NO<sub>y</sub>, and e) CO during crosswind plume transects of smoke from crop burning in southeastern US on 30/08/2019.



**Figure 5** Histograms of the absolute difference of 1 s measurements of a) NO, b)–d) NO<sub>2</sub>, e) HONO, f) NO<sub>y</sub>, g) CO for the entire campaign. Parameters of the gaussian fit to the histogram is indicated in each panel with x0 and w0 being the central value and the width of the fit, respectively. Note that in panel f) a double gaussian was fitted to the histogram and that the parameters for the second mode are given by x1 and w1.

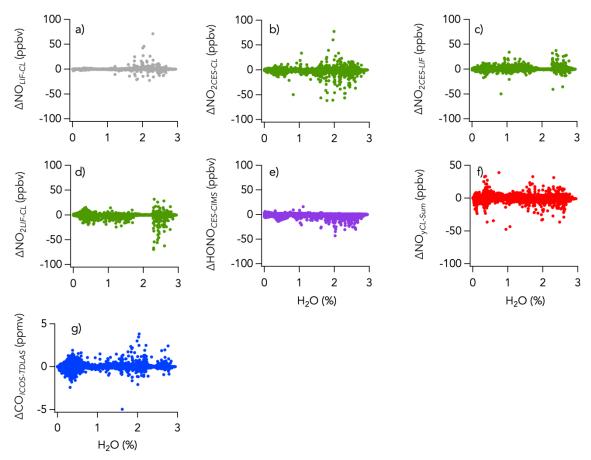


Figure 6 Measurement difference (1 s data) of a) NO, b)–d) NO<sub>2</sub>, e) HONO, f) NO<sub>y</sub>, g) CO as a function of water vapor for the entire campaign.

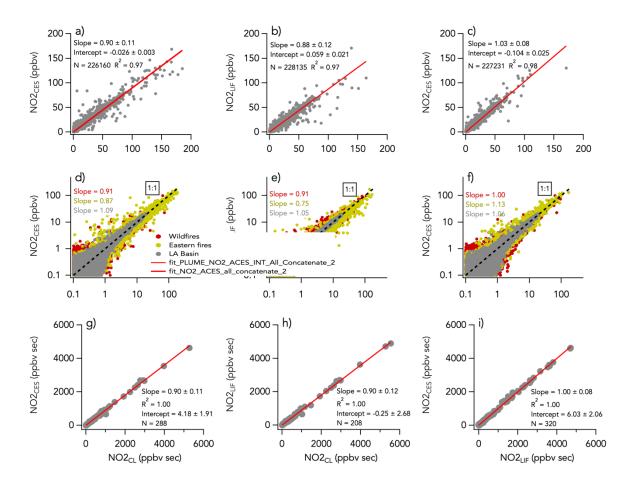
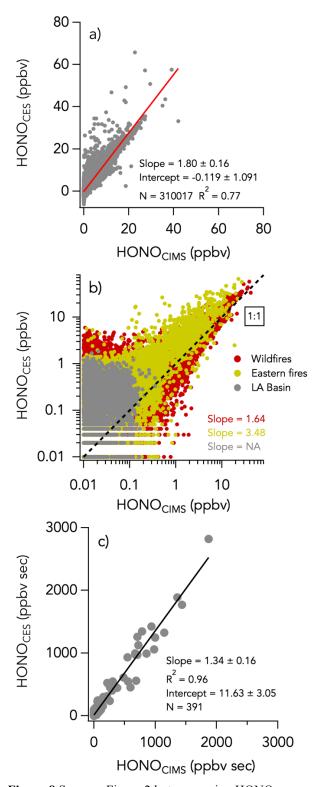
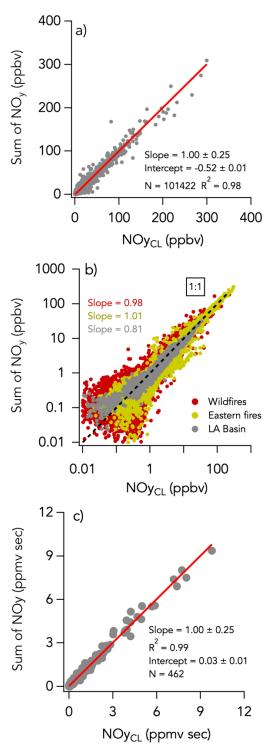


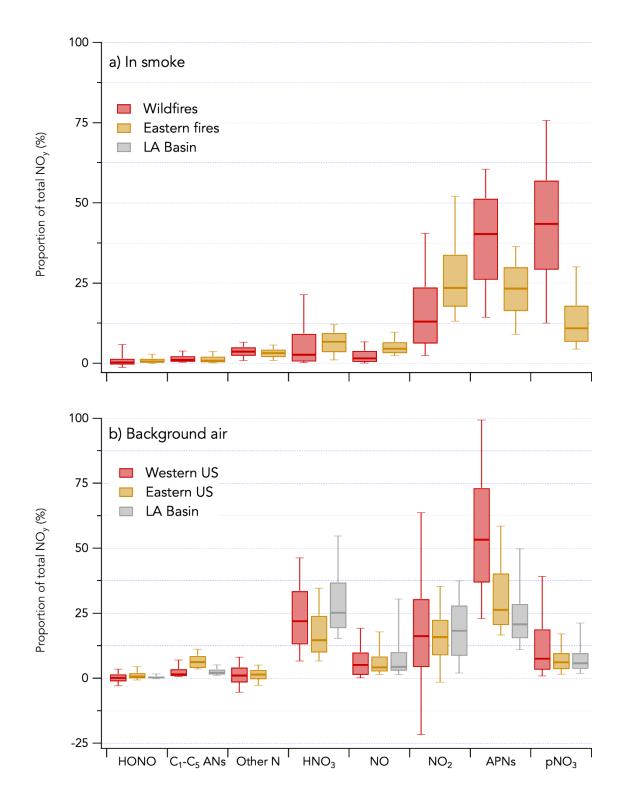
Figure 7 NO<sub>2</sub> measurements by LIF, CES and CL with a)–c) all 1 s data on a linear scale, d)–f) all 1 s data on a log scale, and g)–i) integrals of 208–320 crosswind smoke plume transects. N is the number of independent 1 s observations or smoke plume transects that are compared. In the panels d)–f), the three sampling periods are shown in different colors with the wildfires sampling period in red, the eastern fires sampling period in mustard, and the Los Angeles (LA) Basin flights in grey. The red lines indicate the fit of the data. The dotted black lines are the 1:1 line.



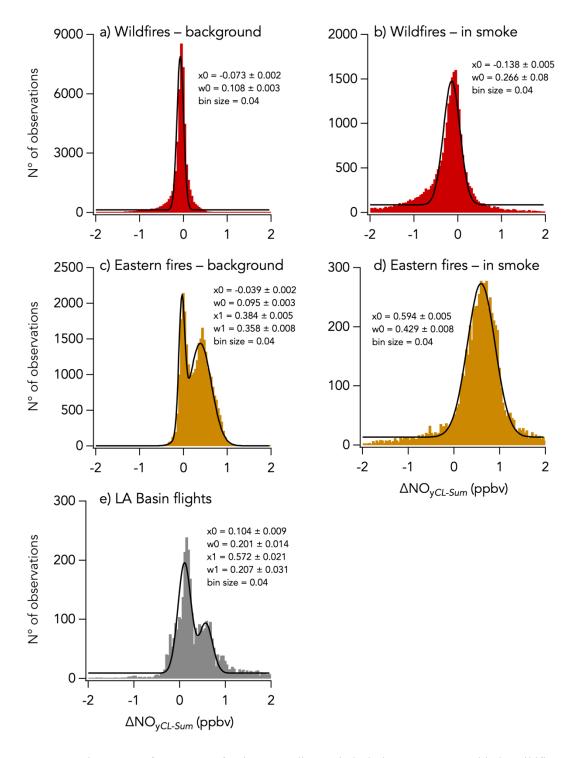
**Figure 8** Same as Figure 2 but comparing HONO measurements by CES and CIMS. No slope is given for the Los Angeles (LA) flights in panel as most of the HONO signal at that time was below the instruments' detection limits. Data from the entire campaign are presented in panels a) and b).



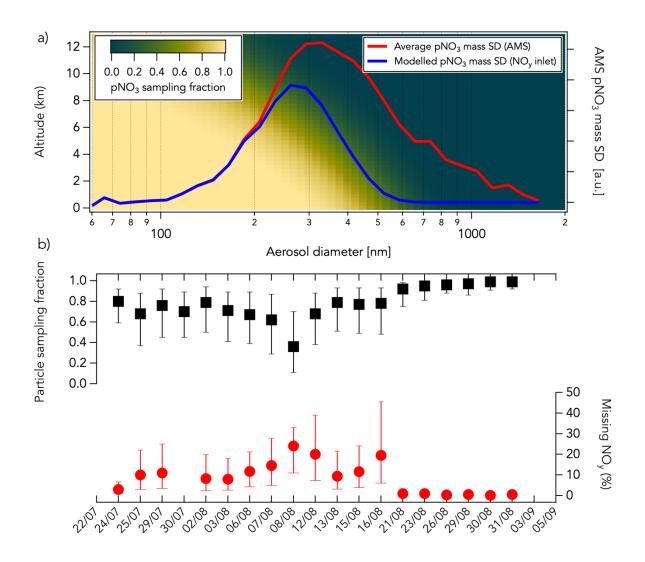
**Figure 9** Same as Figure 2 but comparing the sum of individually measured  $NO_y$  species (=  $NO_x$  + HONO +  $HNO_3$  + APNs +  $pNO_3$ ) with the total  $NO_y$  measurement by CL. Data from the entire campaign are presented in panels a) and b).



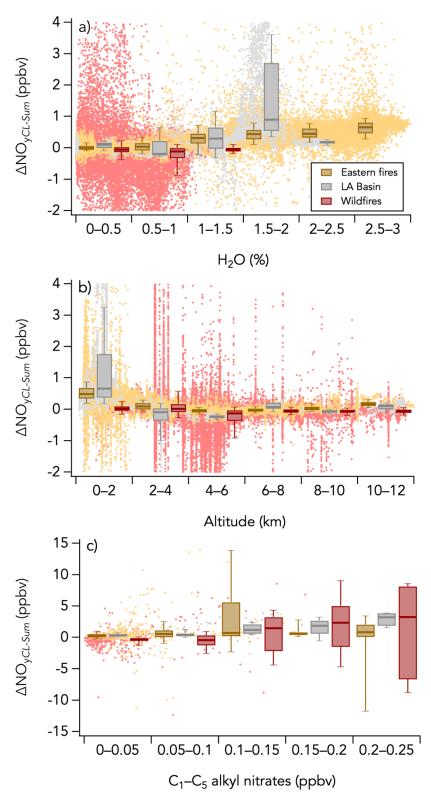
**Figure 10** Contribution of individually measured reactive odd nitrogen species to the total  $NO_y$  budget during FIREX-AQ. The campaign is separated in three periods (wildfires sampling period in red, eastern fires sampling period in yellow, and Los Angeles (LA) Basin flights in grey). The panel a) show the  $NO_y$  budget in smoke plumes, while the panel b) shows that in background air.  $C_1$ – $C_5$  alkyl nitrates are referred to as  $C_1$ – $C_5$  ANs. Other nitrogen species include  $N_2O_5$ ,  $CH_3NO_2$ , and alkene hydroxy nitrates. The box and whisker plots show the  $10^{th}$ ,  $25^{th}$ ,  $50^{th}$ ,  $75^{th}$ , and  $90^{th}$  percentiles.



**Figure 11** Histograms of  $\Delta NO_{yCL-Sum}$  for three sampling periods during FIREX-AQ with the wildfires sampling period in red, the eastern fires sampling period in yellow, and the Los Angeles (LA) Basin flights in grey. Further separation was made between in smoke measurements (panels b and d) and background air measurements (panels a, c, and e). Parameters of the gaussian fit to the histogram is indicated in each panel with x0 and w0 being the central value and the width of the fit, respectively. Note that in the panels c) and e) a double gaussian was fitted to the histogram and that the parameters for the second mode are given by x1 and w1.



**Figure 12** Panel a): The modeled pNO $_3$  sampling fraction through the NO $_y$  inlet as a function of altitude and pNO $_3$  mass size distribution (SD) is shown with a gradient of color from green (low sampling fraction) to yellow (high sampling fraction). The average pNO $_3$  mass size distribution measured in the Williams Flat fire smoke on 07/08/2019 by HR-AMS is shown in red. The modeled pNO $_3$  size distribution sampled in the NO $_y$  inlet assuming an altitude of 5km and a sampled air speed 65% that of the aircraft is shown in blue. In this example case, the sampled pNO $_3$  mass fraction is ~50%. Panel b): The average modeled particle sampling fraction in the NO $_y$  inlet (in black) and the corresponding percentage of measured NO $_y$  that may be unaccounted for (in red) are shown for each flight assuming a sampled air speed of 40% (bottom bars), 65% (markers) and 100% (top bars) that of the aircraft speed. The sampling fractions were calculated using bulk aerosol volume distributions measured by a Laser Aerosol Spectrometer (see Section S1 and Figure SC). The missing NO $_y$  corresponds here to the percentage of measured NO $_y$  that pNO $_3$  not sampled through the NO $_y$  inlet represents. Data shown in the panel b) are from air in smoke only.



**Figure 13** Scatterplots of a)  $\Delta NO_{yCL-Sum}$  vs  $H_2O$ , b)  $\Delta NO_{yCL-Sum}$  vs altitude and c)  $\Delta NO_{yCL-Sum}$  vs  $C_1-C_5$  alkyl nitrates measured by the iWAS instrument for three sampling periods during FIREX-AQ (wildfires sampling period in red, eastern fires sampling period in yellow, and Los Angeles (LA) Basin flights in grey). The box and whisker plots show the  $10^{th}$ ,  $25^{th}$ ,  $50^{th}$ ,  $75^{th}$ , and  $90^{th}$  percentiles of  $\Delta NO_{yCL-Sum}$  distributions in each bin. The dots are the 1Hz data in panels a) and b), and 1Hz data averaged to match the iWAS sampling time in panel c).

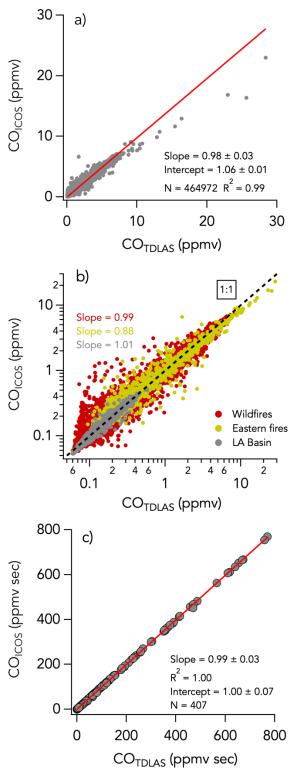


Figure 14 Same as Figure 2 but comparing CO measurements by TDLAS and ICOS.