

~~Application of a new UAV measurement methodology to the quantification of CO₂ and CH₄ emissions from a major coking plant~~ Development of a continuous UAV-mounted air sampler and application to the quantification of CO₂ and CH₄ emissions from a major coking plant

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Abstract. The development in ~~unmanned-crewed~~ aerial vehicle (UAV) technologies over the past decade has led to a plethora of platforms that can potentially enable greenhouse gas ~~emission quantification measurements over the 3-dimensional space~~. Here, we report the development of a new air sampler, consisting of a pumped stainless ~~coiled~~ tube of 150 m in length with controlled time-stamping, and its deployment from an industrial UAV to quantify CO₂ and CH₄ emissions from the main coking plant stacks of a major steel maker in eastern China. ~~During flights, the air sampler starts sampling as soon as the UAV takes off, and stops sampling after landing. The air sample is immediately analyzed upon retrieval with a CRDS gas analyzer for CO₂ and CH₄ mixing ratios.~~ Laboratory tests show that the time series of CO₂ and CH₄ measured using the sampling system is smoothed when compared to online measurement by the ~~cavity ring-down spectrometer (CRDS)~~ analyzer. Further analyses show that the smoothing is akin to a convolution of the true time series signals with a heavy-tailed digital filter. For field test, the air sampler was mounted on the UAV and flown virtual boxes around two stacks in the coking plant at Shagang Steel Group. ~~Mole fraction~~ Mixing ratios of CO₂ and CH₄ in air and meteorological parameters were measured from the UAV during the test flight. A mass-balance computational algorithm was used on the data to estimate the CO₂ and CH₄ emission rates from the stacks. Using this algorithm, the emission rates for the two stacks from the coking plant were calculated to be $0.12 \pm 0.014 \text{ t h}^{-1}$ for CH₄ and $110 \pm 18 \text{ t h}^{-1}$ for CO₂, the

28 latter being in excellent agreement with material balance based estimates. A Gaussian plume inversion approach was also
29 used to derive the emission rates and the results were compared with those derived using the mass-balance algorithm,
30 showing a good agreement between the two methods.

31 **1 Introduction**

32 Atmospheric carbon dioxide (CO₂) and methane (CH₄) are the two major anthropogenic greenhouse gases (GHGs). Both
33 CO₂ and CH₄ in the atmosphere have been increasing since the industrial revolution, particularly rapidly over the past ten
34 years. Global networks consistently show that the globally averaged annual mean CO₂ molar fraction in the atmosphere
35 increased by 5.0 % from 2011 to 2019, reaching 409.9 ± 0.4 ppm in 2019. Likewise, the globally averaged surface
36 atmospheric molar fraction of CH₄ in 2019 was 1866.3 ± 3.3 ppb, 3.5 % higher than in 2011 (Gulev et al., 2021). CH₄ is
37 a stronger absorber of Earth's thermal infrared radiation than CO₂, with its global warming potential (GWP) 32 times
38 greater than that of CO₂ over a 100-year horizon (Saunois et al., 2020). Although its molar fractions in the atmosphere
39 are about 200 times lower than those of CO₂, the total radiative forcing of ~ 1.0 W m⁻² for CH₄ is about half of that of CO₂
40 (~ 2 W m⁻²) (Arias et al., 2021), contributed by its direct radiative forcing of (0.6 ± 0.1) W m⁻² and indirect forcing of 0.4
41 W m⁻² resulting from chemical reactions producing other GHGs including CO₂, O₃, and stratospheric water (Turner et al.,
42 2019). Furthermore, although global anthropogenic CH₄ emissions are estimated to be only 3 % of the global
43 anthropogenic CO₂ emissions in units of carbon mass flux, the increase in atmospheric CH₄ is responsible for about 20 %
44 of the warming induced by long-lived greenhouse gases since pre-industrial times (Etminan et al., 2016). Both CO₂ and
45 CH₄ are produced and released into the atmosphere from a variety of natural and anthropogenic sources. Natural emission
46 sources include vegetation, oceans, volcanoes and naturally occurring wildfires, but most of the increases in atmospheric
47 CO₂ and CH₄ are considered to have resulted from anthropogenic emissions, from sources including fossil fuel production
48 and uses, agricultural activities, land use and industrial processes (Canadell et al., 2021).

49 Quantification of CO₂ and CH₄ emissions from sources requires continuous measurements of their ~~mole~~
50 ~~fraction~~mixing ratios as well as meteorological parameters using a variety of stationary and mobile platforms, including
51 ground-based vehicles (Rella et al., 2015; Brantley et al., 2014), towers (Helfter et al., 2016; Takano and Ueyama, 2021),
52 aircrafts (Li et al., 2017; Ligio et al., 2019) and satellites (Miller et al., 2013; Turner et al., 2015). Small ~~unmanned~~
53 uncrewed aerial vehicles (UAVs) have become emerging platforms due to the recent rapid technological developments.
54 They are flexible, versatile and relatively inexpensive. Most importantly, a UAV platform ~~could fill~~s the sampling space
55 between the ground and altitudes of up to hundreds of meters above ground, in which other mobile platforms have been

56 unable to operate (Shaw et al., 2021). Due to their relatively low flying speeds, UAV platforms ~~could~~ offer a high
57 spatiotemporal resolution for sampling and thus enabling accurate plume mapping. On the other hand, UAVs have limited
58 endurance, being constrained by battery capacities and payloads, making them more suitable for small facility flux
59 quantification.

60 UAV platforms have been used to quantify CH₄ emissions in several studies, mainly focused on facility-scale
61 emission sources including landfills (Allen et al., 2019; Bel Hadj Ali et al., 2020), coal mines (Andersen et al., 2021),
62 dairy farms (Vinkovic et al., 2022), wastewater treatment plants (~~Gålfalk~~~~Gålfalk~~ et al., 2021) and oil and gas facilities
63 (Golston et al., 2018; Li et al., 2020; Nathan et al., 2015; Shah et al., 2020; Tuzson et al., 2021). UAV-based CH₄
64 measurements are generally made with three different methods: collecting on-board samples for subsequent analysis,
65 tethered sampling to a sensor on the ground and on-line measurements (Shaw et al., 2021). Gas samples could be stored
66 onboard a UAV for subsequent analyses on the ground after landing, using air bags (Brownlow et al., 2016) or sampling
67 canisters (Chang et al., 2016). Andersen et al. developed a UAV-based active ~~AirCore~~~~aireore~~ system, consisting of a long
68 coiled stainless-steel tubing, a small pinhole orifice, and a pump that drags air through the tube (Andersen et al., 2018),
69 which allows for a higher spatiotemporal resolution in the measurements. Direct comparisons between a quantum cascade
70 laser absorption spectrometer (QCLAS) and the active ~~AirCore~~~~aireore~~ measurements show that the active ~~AirCore~~~~aireore~~
71 measurements are smoothed by 20 s and had an average time lag of 7 s. The active ~~AirCore~~~~aireore~~ measurements also
72 stretch linearly with time at an average rate of 0.06 s for every second of QCLAS measurement (Morales et al., 2022).
73 The advances in active ~~AirCore~~~~aireore~~ sampling have made UAV measurements for CH₄ emissions feasible, even if still
74 with rooms for improvement. ~~To the best of our knowledge, there have been no reports~~ Studies of using UAVs for CO₂
75 plume detection and mapping to determine CO₂ emission rates from anthropogenic sources have also been reported
76 (Reuter et al., 2021; Liu et al., 2022; Leitner et al., 2023; Chiba et al., 2019). Reuter et al. presented the development of
77 a UAV platform to quantify the CO₂ emissions of anthropogenic point sources by deployment of an NDIR (non-dispersive
78 infrared) detector and a 2-D ultrasonic acoustic resonance anemometer on the platform (Reuter et al., 2021).

79 In this study, we developed a new active air sampling system for deployment from a UAV ~~for three dimensional~~
80 ~~measurements of on a trajectory in the three-dimensional space to measure~~ CO₂ and CH₄. The complete sampler plus
81 UAV system was deployed to quantify CO₂ and CH₄ emissions from the stacks of the main coking plant of Shagang, the
82 largest private steel maker in China. The top-down emission rate retrieval algorithm (TERRA) (Gordon et al., 2015) was
83 applied to the UAV data to determine stack CH₄ and CO₂ emissions rates. The iron and steel industry is one of the largest
84 contributing industries to global GHG emissions, accounting for around 7% of global total GHG emissions (Hasanbeigi,

85 2022). Coke production is one major process of iron and steel making that generate emissions of CO₂ and CH₄. During
86 coke production, coking coal is used to manufacture metallurgical coke that is subsequently used as the reducing agent in
87 the production of iron and steel (U.S. Environmental Protection Agency, 2016). Coke oven gas is the main sources of
88 CO₂ and CH₄ emissions during coke production (Angeli et al., 2021; IPCC, 2006). China is the largest coke producer in
89 the world, with a coke production of 4.72 billion tons in 2020. The GHG emissions from coke production in China are
90 reported based on the Tier 1 methodology of IPCC Guidelines, which multiplies generic default emission factors with the
91 tonnage of coke produced (Ministry of Ecology and Environment of China, 2018). Tier 1 methodologies are the simplest
92 and least complex requiring less resources on collection the necessary data and producing GHG emission estimates. The
93 present UAV measurement-based emission results can be compared with material balance based emission estimates and
94 the emissions based on the Tier 1 emission factors and coke production at the plant, and to shed light on the uncertainties
95 related to Tier 1 emission factors in the case of CH₄ emissions.

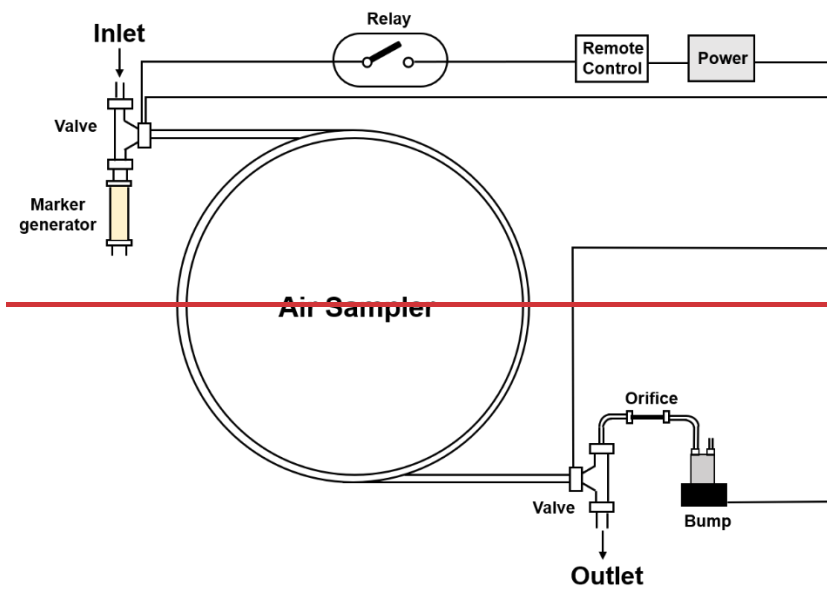
96 **2.Method**

97 **2.1 The air sampling system**

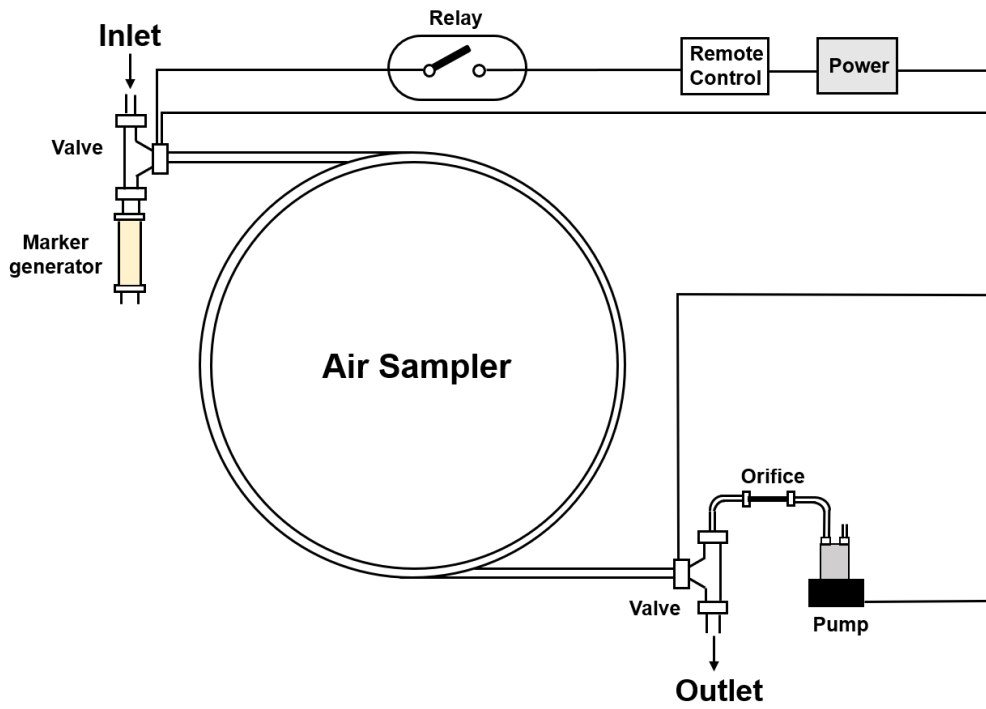
98 To realize GHG emission quantification by UAV measurement, a new compact air sampling system was developed
99 based on a variation of the active ~~AirCore~~ AirCore method. The AirCore system contains a 150-m-long stainless steel tube,
100 open at one end and closed at the other, that relies on positive changes in ambient pressure for passive sampling of the
101 atmosphere (Karion et al., 2010). Figure 1 shows an overview of the patent-pending design for this sampler. It consists
102 of a 150 m long thin-walled 1/8 inch outside diameter stainless-steel tubing, a pump, a micro-orifice, a CO₂ marker
103 generator, two three-way solenoid valves and electric relays, with all electrical devices powered by a 12V battery. The
104 tubing is wound into a multilayer coil, in whose center the other components of the system are mounted. The system is
105 housed in the highly compact patent-pending carbon fiber assembly design of 280 mm diameter and 98 mm height, that
106 can be quickly mounted at and dismantled from the bottom of an UAV. The sampler weighs about 5.9 kg and allows for
107 continuous sampling up to 35 minutes.

108 The sampler air intake is mounted at 70 cm above the center of gravity of the UAV, placed nearby a sonic
109 anemometer (below) for ensuring sampling the same air mass where wind speed is measured. The time stamp of the ~~mole~~
110 ~~fraction~~ mixing ratio observation was corrected for the short time lag of 4 seconds between sampling at the air intake and
111 the thin-walled stainless-steel tubing attributable to the length of the Teflon inlet tube. Shortly before every flight, the

112 pump is remotely turned on to sample the CO₂ marker for 5 seconds ~~to mark the beginning of the flight,~~ and then to collect
113 air samples. ~~The CO₂ marker helps~~ markers help to identify the starting point ~~of~~ and specific times subsequently during
114 the UAV air sampling during later in data extraction and analysis. During flight, the pump would alternatively sample the
115 marker and the ambient air on a preset timing schedule. The sampling flow rate remains at 18 ccm during the entire flight,
116 controlled with the micro-orifice which is placed between the pump and the coiled tubing. After landing, the pump is
117 remotely turned off and the air sample in the sampling tubing is immediately analyzed with a cavity ring-down
118 spectrometer (CRDS) (Picarro, Inc., CA, USA, model G2401) for CO₂ and CH₄ mixing ratios in the sampled air. Waiting
119 longer would lead to unwanted mixing of the samples in the tubing. The air sample enter the tubing from the air inlet
120 during sampling and leave the tubing from a different air outlet during later analysis. As a result, the samples at the
121 beginning of the flight spend the same amount of time within the tubing as those at the end of the flight. Using the
122 embedded CO₂ marker data, the CO₂ and CH₄ data series can be mapped to the sampling times and GPS locations during
123 flight.



124



125

126 **Figure 1.** Design of the air sampler.

127 **2.2 The 3D sonic anemometer**

128 Previous studies that applied UAV platforms for GHG monitoring generally relied wind data from nearby ground weather
 129 stations (Morales et al., 2022; Allen et al., 2019). However, Gålfalk et al. shows that wind speeds were inconsistent
 130 between a ground weather station at a 1.5 m height and an anemometer mounted on their UAV, especially when altitude
 131 increases, showing the need to have an on-board weather station for accurate flux calculations (GålfalkGålfalk et al.,
 132 2021). In the present study, in order to obtain meteorological data along the flight track, a 3D sonic anemometer (Geotech
 133 Inc, Denver, US, model Trisonica Mini) is attached on the top of the UAV via a 450 mm carbon fiber pole. The
 134 anemometer measures three-component wind speed (U_x, U_y, w) and temperature (T). The three-component wind
 135 speed measured data were further translated transformed into actual wind speeds speeds and wind direction based on the
 136 GPS information, directions after corrections for UAV attitude data (pitch, yaw, roll) changes and the accounting for its
 137 airspeed measured, as well as the perturbations caused by the UAV rotor propellers using a correction algorithm (Yang
 138 et al. 2023). The GPS information, airspeed, and attitude data (pitch, yaw, and roll) were extracted from the UAV data
 139 transmitted to the ground control station. The anemometer measures wind speeds within the range of 0 to 50 m s⁻¹, with
 140 an accuracy of ± 0.1 m s⁻¹ below the wind speed of 10 m s⁻¹. The accuracy for wind direction measurement is ± 1°. For
 141 temperature measurement, the operating range for the anemometer is between -40 °C to 85 °C and the accuracy is ± 2 °C.

142 For anemometers mounted on multi-rotor UAVs, how to correct for the effects of the translational and rotational
143 movements of the UAVs as well as the flows induced by the rotors to obtain accurate wind data is an on-going research
144 topic (GålfalkGalfalk et al., 2021; Wolf et al., 2017; De Boisblanc et al., 2014; Palomaki et al., 2017; Zhou et al., 2018;
145 Yang, 2023). During flight, rotary wing UAVs create thrust by drawing air from above the rotors and expelling it
146 downwards at a higher velocity. Such flows may extend to the anemometer position in addition to true atmospheric air
147 flows, masking the true wind signals in the data from the anemometer (Wolf et al., 2017). Previous studies have conducted
148 laboratory testing (Wolf et al., 2017; De Boisblanc et al., 2014; Palomaki et al., 2017) or flow field simulation (Zhou et
149 al., 2018) to determine the appropriate distance to place anemometers onto multi-rotor UAVs to minimize the impact
150 from the rotor-induced air flows. The anemometer in this research is mounted at an upward distance of 70 cm from the
151 center of gravity of the UAV. A full digital model of the UAV, the anemometer and its mounting frame, and the air
152 sampler was created. Using this digital model, computational fluid dynamics (CFD) simulations were performed to
153 quantify wind speed disturbances caused by the UAV's rotor propellers on the anemometer during flight under a vast
154 array of different wind conditions. An overall correction algorithm was developed in which parameters for propeller
155 disturbances determined based on the CFD simulations were included along with correction schemes for false signals
156 resulting from translational motions and changes in UAV pitch, roll and yaw. The correction algorithm was verified with
157 real-world UAV flight-meteorological tower measurement intercomparisons (Yang et al., 2023)~~The anemometer in this~~
158 ~~research is mounted at an upward distance of 7045 cm from the center of gravity of the UAV to minimize this interference~~
159 ~~based on the results from flow field simulations for combinations of the UAV flight envelope and true winds, and verified~~
160 ~~with UAV flight meteorological tower measurement intercomparisons(Yang, 2023). During the flight, the meteorological~~
161 ~~data including wind speed, wind direction and temperature were transmitted and collected on ground. The data are post-~~
162 ~~flight corrected for the rotor induced air flows, the true air speeds of the UAV, and the UAV attitude changes during~~
163 ~~flight to derive accurate wind speed and direction results(Yang, 2023) along the flight track.~~

164 2.3 The UAV

165 The air sampler and the anemometer are mounted on a hexacopter UAV (KWT-X6L-15). The UAV has a maximum flight
166 time of ~30 minutes at a maximum payload of 15 kg, or longer with a lighter payload. Such flight endurance and carrying
167 capacity meet our needs for loading the air sampler and the anemometer onto the UAV to realize emission quantification.
168 The UAV is capable of flying at winds up to 14.4 m s⁻¹ to an altitude of about 4000 m and has a maximum horizontal

169 flying speed of 18 m s^{-1} , a maximum ascending speed of 4 m s^{-1} and a maximum descending speed of 3 m s^{-1} . The
170 horizontal [hovering](#) precision of the GPS on the UAV is $\pm 2 \text{ m}$ and the vertical [hovering](#) precision is $\pm 1.5 \text{ m}$.

171 **2.4 Air sample analysis**

172 After landing, the air sample collected in the tubing is immediately analyzed with the CRDS analyzer. The withdrawal
173 flow rate of the air from the sample tubing during analysis is an important parameter in optimizing the results. High
174 withdrawal rates lead to unwanted mixing in the cavity of the analyzer. However, direct withdrawal of air from the sample
175 tubing by the analyzer at a flow rate as low as the sampling flow rate of 18 sccm results in smoothing of concentrations
176 from the [inner wall surface drag and desorption laminar flow](#) inside the tubing. We optimized the flow rate of the air from
177 the sample tubing into the CRDS analyzer at $\sim 54 \text{ sccm}$, 3 times the sampling flow rate, by diluting the air sample with
178 zero air, with two mass flow controllers separately controlling the flow rate of zero air and the withdrawal rate of the air
179 sample (Fig. 2b).

180 **2.5 Mass balance approaches for determining emission rates**

181 [The UAV-based measurements were coupled with the mass-balance approach TERRA to determine the emission rates of](#)
182 [the measured pollutants using their measured mixing ratios and the meteorological data \(three-component wind speed](#)
183 [\(\$U_x, U_y, w\$ \) and temperature \(\$T\$ \)\) collected on board the UAV during the flight. TERRA computes integrated mass fluxes](#)
184 [through airborne virtual box/screen measurements including those made from aircraft and in this case UAVs. TERRA](#)
185 [has been used successfully and extensively for emission rate determination of tens of volatile organic compounds \(Li et](#)
186 [al., 2017\), \$\text{CO}_2\$ \(Liggio et al., 2019\), \$\text{CH}_4\$ \(Baray et al., 2018\), oxidized sulphur and nitrogen \(Hayden et al., 2021\), black](#)
187 [carbon \(Cheng et al., 2020\) and secondary organic aerosol \(Liggio et al., 2016\) using aircraft measurements. To run](#)
188 [TERRA based on a virtual box flight, the first step is to map the \$\text{CH}_4\$ and \$\text{CO}_2\$ mixing ratio data measured along the level](#)
189 [flight tracks encircling a facility to the two-dimensional virtual walls of the virtual box, created from stacking the level](#)
190 [flight tracks, that surrounds the facility. The two-dimensional virtual walls \(or screens\) are derived from the unwrapping](#)
191 [of the virtual box, to assist the presentation of the \$\text{CH}_4\$ and \$\text{CO}_2\$ plumes along the flight tracks, with the horizontal path](#)
192 [length \(i.e., the ground line projection of the fitted flight track\) and altitude as the two dimensions. The start of the](#)
193 [horizontal path is typically defined as the south-east corner of the virtual box, but the selection of this starting position](#)
194 [has no effect on the emission rate computation, and the horizontal path distance increases in a counter clockwise direction.](#)
195 [This procedure results in a translation of each flight position point from a three-dimensional position of latitude \(\$\varphi\$ \),](#)

196 longitude (x), and altitude (z , above mean sea-level) to a two-dimensional screen position of horizontal path distance $s =$
197 $f(x,y)$. Subsequently, TERRA applies the Simple Kriging algorithm to interpolate the data and produces a mesh on the
198 two-dimensional virtual box walls whose resolution can be set depending on applications, e.g., at 20 m (vertical) by 40
199 m (horizontal) for aircraft measurements. For the modified version of TERRA applied to the UAV measurements in this
200 study, the mesh has been adjusted to a size of 1 m (vertical) by 2 m (horizontal), as UAVs fly significantly shorter
201 distances compared to piloted aircraft. The kriging weights were obtained with an isotropic spherical semivariogram
202 model. In TERRA, nugget, sill, and range can all be modified to fit the semivariogram model. The mixing ratios of both
203 CH₄ and CO₂ are extrapolated from the lowest flight altitudes to the ground digital elevation using one of several methods
204 or a combination thereof, namely (1) assuming a constant (2) linear extrapolation between a constant and background (3)
205 a background value below flight altitudes (4) linear fit between the lowest flight altitude and zero at the ground and (5)
206 exponential fit from the lower flight altitudes (Gordon et al., 2015). Concurrently measured wind speed from the UAV
207 (Yang, 2023) is decomposed into northerly and easterly components ($U_E(s, z), U_N(s, z)$) based on the wind direction and
208 similarly interpolated onto the 1 m x 2 m mesh. The decomposed wind speeds are further extrapolated to the ground
209 digital elevation using a log profile fit (Gordon et al., 2015). Based on the interpolated/extrapolated CH₄ and CO₂ mixing
210 ratio, temperature, pressure (calculated using barometric height formula), and wind speeds, TERRA computes the fluxes
211 of CH₄ and CO₂ through the virtual walls and finally their facility emission rates by integrating the fluxes.

212 To summarize, in TERRA the mass-balance in computing the emissions within a control box for a given inert
213 pollutant such as CH₄ or CO₂ is presented by:

$$214 \quad E_C = E_{C,H} + E_{C,V} - E_{C,M}, \quad (1)$$

215 where E_C is the emission rate, $E_{C,H}$ is the horizontal advective transfer rate through the box walls, $E_{C,V}$ is the advective
216 transfer rate through the box top and $E_{C,M}$ is the increase in mass within the volume due to a change in air density. Other
217 terms listed in the Gordon et al. computation algorithm that were used to solve for the total emission rate were often
218 neglected as they contribute little to the total emission rates (Gordon et al., 2015). Each term from Eq. (1) is estimated as:

$$219 \quad E_{C,H} = M_R \iint X_C \rho_{air} U_1 ds dz, \quad (2)$$

$$220 \quad E_{C,V} = M_R X_{C,Top} \iint \rho_{air} \omega dx dz, \quad (3)$$

$$221 \quad E_{C,M} = M_R \iiint X_C \frac{d\rho_{air}}{dt} dx dy dz, \quad (4)$$

222 where M_R is the ratio of the compound molar mass to the molar mass of air, $X_c(s, z)$ is the mixing ratio of the compound
223 in question, $\rho_{air}(s, z)$ is the air density, w is the vertical wind velocity at the box top, $X_{c,Top}$ is the mixing ratio at the top
224 of the box, and $U_{\perp}(s, z)$ is the horizontal normal wind vector to the flight track calculated from the northely and easterly
225 components ($U_E(s, z), U_N(s, z)$):

$$226 \quad U_{\perp}(s, z) = \frac{U_N(s, z) ds/dx - U_E(s, z) ds/dy}{\sqrt{(ds/dx)^2 + (ds/dy)^2}}, \quad (5)$$

227 The vertical transfer rate term $E_{C,V}$ is estimated by computing the air mass vertical transfer rate, determined from air
228 mass balance within the box, and multiplying it with the CO₂ or CH₄ mixing ratios at the box top. This term is normally
229 negligible in other top-down emission estimate approaches since it is typically miniscule compared to horizontal fluxes,
230 but can affect the computed emission rates when vertical air movement becomes more significant such as under unstable
231 atmospheric conditions. $E_{C,M}$ is often ignored in other mass-balance approaches; in TERRA it is estimated by taking the
232 time derivative of the ideal gas law in temperature and pressure during the flight time, and typically it does not change
233 significantly over the duration of 30 minutes or so for the UAV flight.

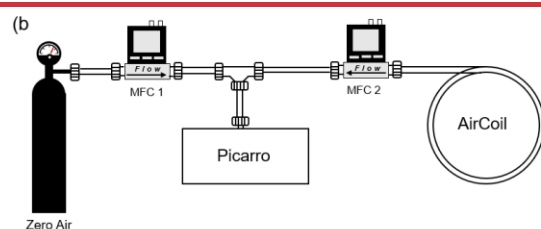
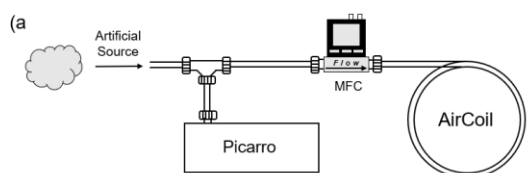
234 To suit the UAV measurements, the following modifications to the TERRA algorithm were made: (1) A much higher
235 interpolation resolution for the kriging mesh was implemented for application to the UAV measurements in this study,
236 with the interpolation mesh size adjusted to 1 m (vertical) by 2 m (horizontal), as UAVs fly significantly shorter distances
237 compared to applications to piloted aircraft for which the interpolation resolution was 20 m (vertical) by 40 m (horizontal);
238 (2) The modified TERRA now applies an embedded routine to automatically fit flight tracks using least squares, while
239 this procedure was previously conducted manually offline through geographic information system when using TERRA.
240 (3) The modified version of TERRA has added an algorithm for correcting negative weights during Kriging interpolation
241 following Deutsch (Deutsch, 1995). ~~TERRA has been updated at Peking University now with an embedded routine to~~
242 ~~automatically fit flight tracks, a critical first step in the computation and a procedure previously conducted offline through~~
243 ~~geographic information system (GIS) when using TERRA. TERRA has been updated at Peking University~~The updated
244 algorithm is now recoded using the Python language and runs under a browser-server environment with a new GUI and
245 new interactive data flow.

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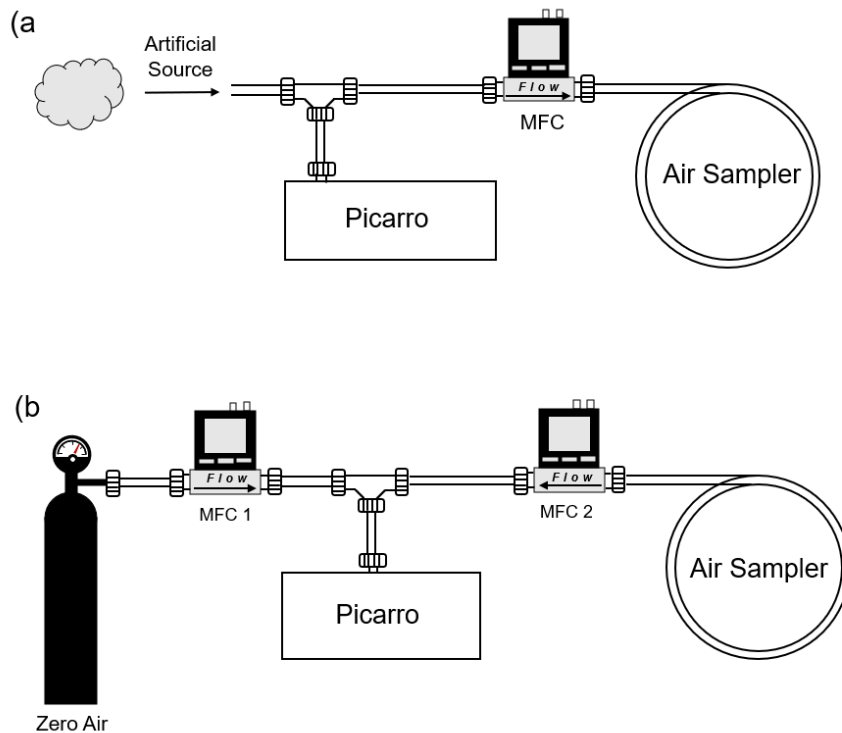
247 **3 Validation of the air sampler: Laboratory tests**

248 **3.1 Validation of the air sampler**

249 Prior to flights in the field, we validated the air sampler in laboratory experiments by first sampling an artificial air
250 while making simultaneous online measurements of the artificial air with the CRDS analyzer, a mixture of lab
251 air/standards of CO₂, CH₄ and then analyzing the sampled artificial air was with the same CRDS analyzer and
252 comparing the results from the air sampling/CRDS analysis sampler with those from simultaneous online measurement
253 of the same lab air mixture with the CRDS analyzer to the online measurements. An experimental apparatus was
254 constructed for the simultaneous sampling of the same lab air mixture artificial air with the air sampler and the online
255 measurement CRDS analyzer through a tee junction (Fig 2(a)), and subsequent air sample analysis using the same
256 CRDS analyzer (Fig. 2(b)). The In the artificial air, CH₄ and CO₂ standards were control-released into the lab air from
257 an 8 L gas cylinder filled with a gas mixture of 5 ppm CH₄, 2 ppm CO and 600 ppm CO₂ to generate the artificial air
258 source. The outlet of the standard gas cylinder was held at artificially different varying distances to the tee junction
259 over time to yield a time series of different CH₄ and CO₂ mixing ratios in the mixed lab air/standards, which was
260 designed to mimic plumes expected in the real atmosphere. During analysis, the flow rate through the zero air (Mass
261 Flow Controller 1) is adjusted to make sure that the flow rate through the air sampler (Mass Flow Controller MFC 2)
262 is stable and consistent at 54 sccm (Sec. 2.4).



263



264

265 **Figure 2.** Diagram of the air sampler testing setup in the laboratory. (a) simultaneous sampling by the air sampler and the Picarro
 266 CRDS analyzer. (b) subsequent air sample analysis using the picarro CRDS analyzer.

267 Figure 3-4(a) illustrates the ~~mole fraction mixing ratios~~ of CO₂ and CH₄ ~~measured by time series obtained from the~~
 268 air sampler ~~and followed online measurements~~ by the CRDS ~~analysis analyzer~~. It can be seen that the measured ~~results~~
 269 ~~from the air sampler samples~~ and the ~~online online CRDS measurements measurements analyzer~~ are in good agreement
 270 throughout the tests, ~~and the correlation coefficient is estimated to be 0.89 and 0.73 for CH₄ and CO₂ (Fig. 4(c) and (f)).~~
 271 For the measurements with the air sampler, short term variations and noises ~~in the CH₄ and CO₂ mixing ratios~~, that were
 272 fully captured by the ~~online measurement CRDS analyzer during the online measurements~~, were smoothed out, while the
 273 main features and ~~tendency tendencies~~ were preserved. In fact, the air sampler measurement ~~result results~~ should be a
 274 smoothed version of the ~~online measurement CRDS analyzer online measurements~~, due to mixing in the analyzer cavity,
 275 molecular diffusion during sample storage in the sampler, inner wall surface drag and desorption during its withdrawal
 276 from the tubing during analysis, as well as Taylor dispersion during sampling and analysis. (Karion et al., 2010). Dilution
 277 with zero air during ~~later~~ CRDS analysis also contributes to the smoothing.

278 **34.2 Data deconvolution to achieve high time resolution**

279 While it is impractical to delineate the individual smoothing effects when the air sample passes through the coupled
 280 system of the sampler plus the analysis setup as described above, the measured concentration $y(t)$ can be treated as a

281 result of the convolution of the air concentration before sampling $x(t)$ and a smoothing kernel $g(i)$ consisting of a series
 282 of weights, which are inherently determined by factors including the sampler properties (tubing length, inner diameter,
 283 temperature, absorptive properties, flow rates), storage time, dilution, and mixing in the cavity of the instrument. The
 284 smoothing can be described as:

$$285 \quad y(t) = \sum_{i=r}^s g(i)x(t-i) + n(t), t = s, s+1, \dots, n-1+r. \quad (6)$$

286 Or, expressed as a convolution of the form:

$$287 \quad y(t) = g(t) * x(t) + n(t). \quad (7a)$$

288 where $y(t)$ is the measured concentration at time t , $x(t)$ the air concentration, and $n(t)$ the unknown noise, assumed
 289 to be independent of $x(t)$. The kernel $g(i)$ contains $s-r+1$ non-zero kernel weight terms ($0 < g(i) < 1$). When all
 290 four terms in Eq. (2a7a) undergo Fourier transform, Eq. (2a7a) can be expressed in the frequency domain:

$$291 \quad Y(f) = G(f)X(f) + N(f). \quad (7b)$$

292 In order to characterize the kernel weights $g(i)$, a second lab experiment was conducted during which the sampler
 293 first sampled zero air for some time, and then sampled the CO₂ and CH₄ standards for one second, before returning to
 294 sampling zero air again, creating an original concentration pulse signal in the $x(t)$:

$$295 \quad x(t) = \begin{cases} C, & t = j \\ 0, & t \neq j \end{cases} \quad (8)$$

296 where $j = j^{\text{th}}$ second when the sampler collected the standard of a known concentration C . This air sample was then
 297 analyzed with the CRDS as described above. After sampling, storing and analyzing, smoothing of the original
 298 concentration pulse leads to the concentration signal output $Y(t)$ as follows:

$$299 \quad y(t) = \begin{cases} \sum_{i=r}^s g(i)x(t-i) + n(t) = g(t-j)C + n(t), & t-i = j \text{ and } i = r, r+1, \dots, s \\ n(t), & t-i \neq j \end{cases} \quad (9)$$

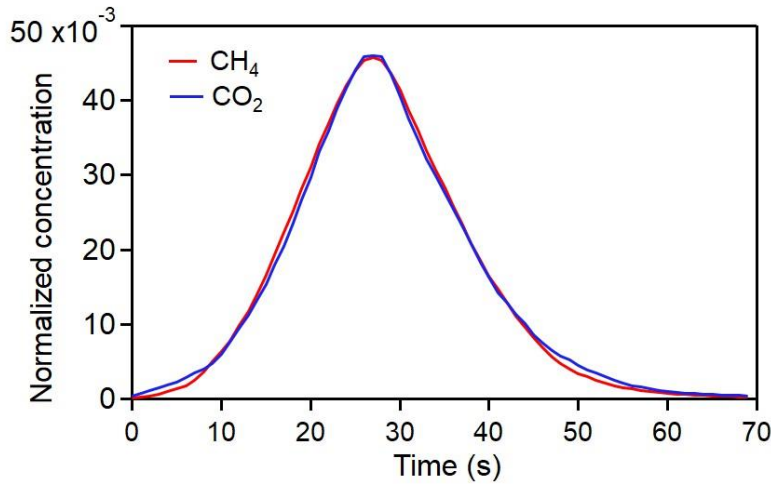
300 where $y(t)$ is the measured concentrations from the air sampler after sampling the concentration pulse and is non-zero
 301 when $t-i = j$, with the index i taking the values from r to s . The noise $n(t)$ term is zero for $t-i \neq j$ and can be
 302 assumed to have similar behavior for $t-i = j$. Therefore,

$$303 \quad g(i) = g(t-j) = \frac{1}{C}y(t) - \frac{1}{C}n(t), t = i+j \text{ and } i = r, r+1, \dots, s. \quad (10)$$

304 The second lab experiment showed that $y(t)$, and therefore the kernel $g(t)$, consists of 70 non-zero values. To remove
 305 the noise $n(t)$, $g(t)$ is further smoothed using a box-car running mean of 5 terms:

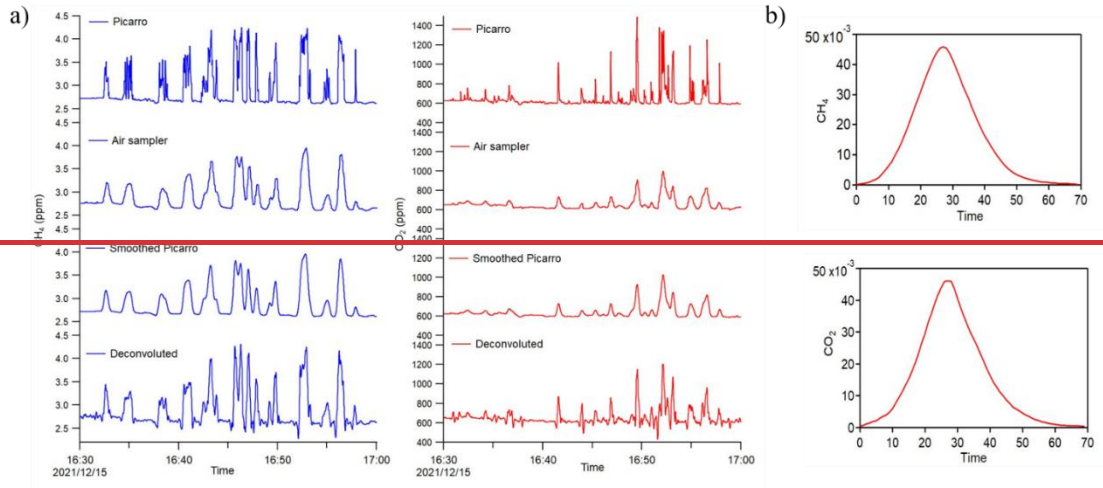
$$306 \quad \hat{g}(t) = \frac{1}{5} \sum_{k=t-2}^{k=t+2} g(k) \approx \frac{1}{C}y(t), t = i+j \text{ and } i = r, r+1, \dots, s. \quad (11)$$

307 It could be seen from Fig. 3(b) that $\hat{g}(t)$ has an asymmetrical distribution with a right trailing tail and a half-height width
308 of approximately 20 seconds for CO₂ and 21 seconds for CH₄, indicating that the smoothing had significantly reduced the
309 sampling/analysis method time resolution to about 20 second from the 1 second resolution of the original pulse in the air
310 concentration. The kernel shows that the influence from the neighboring points have on a given point decreases with
311 increases in the gap between the two points.

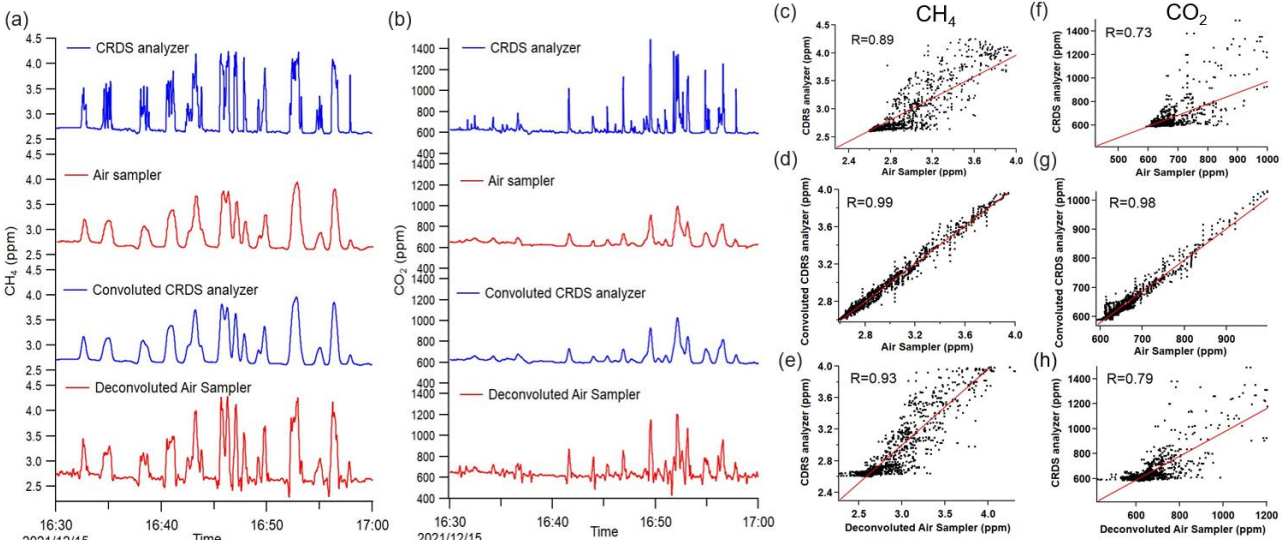


312
313 **Figure 3.** The output of the one-second signal after sampling, storing and analyzing using the air sampler for CO₂ and CH₄, normalized
314 by their respective concentrations in the standard. As shown in the text, these curves are the actual kernel weights of $\hat{g}(t)$.

315 To test whether the kernel weights $\hat{g}(t)$ can smooth the online measured concentrations from the first lab experiment
316 (top ~~data-seriesline~~ in Fig. 34(a), ~~left and (b)~~), the weights $\hat{g}(t)$ were used to convolute with the data from the online
317 measurements (i.e., $x(t)$), resulting in an estimated $\hat{y}(t)$ (Fig. 34(a) ~~and (b)~~, third ~~curve~~) that is in excellent agreement
318 with the measurements from the ~~sampler/analysis process~~air sampler, with the correlation coefficients increased to 0.99
319 ~~and 0.98 for CH₄ and CO₂ (Fig. 4 (d) and (g))~~(the second curve in Fig. 3(a)).



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Figure 34. (a) and (b) Mole fraction Mixing ratios of CO₂ and CH₄ measurements by online measurements with CRDS (the first line) and sampling/analysis the air sampler (the second line) in laboratory tests. The third line represents the smoothed CRDS data after convolution with the kernel $\hat{g}(t)$ and the fourth line represents the deconvoluted series after Wiener deconvolution. The signals of the same color represent the original signals and the corresponding signals after convolution or deconvolution (b) The output of the one-second signal after sampling, storing and analyzing using the air sampler for CO₂ and CH₄, normalized by their respective concentrations in the standard. As shown in the text, these curves are the actual kernel weights of $\hat{g}(t)$. (c)-(e) Correlation plots of CH₄ (f)-(h) Correlation plots of CO₂.

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The ultimate goal of determining $\hat{g}(t)$ in Fig. 3(b) is to deconvolve $y(t)$ from the air samplersampling/analysis process to obtain the original concentration series $x(t)$ using a number of deconvolution techniques. In the present study, we used the deconvolution method based on the Wiener theorem (Lin and Jin, 2013). The theorem provides the Wiener convolution filter $h(t)$ so that $x(t)$ can be estimated as follows:

$$\hat{x}(t) = \sum_{i=-\infty}^{\infty} h(i)y(t-i) = h(t) * y(t) \tag{12}$$

335 where $y(t)$ is the measured concentration, and $\hat{x}(t)$ an estimate of $x(t)$. In the frequency domain, Eq. (712) may be
 336 rewritten as a product of two scalars:

$$337 \quad \hat{X}(f) = H(f)Y(f) \tag{13}$$

338 where $\hat{X}(f)$, $H(f)$, and $Y(f)$ are the Fourier transforms of $\hat{x}(t)$, $h(t)$, and $y(t)$, respectively. The Wiener convolution
 339 filter $h(t)$ is derived from the minimization of the mean square error:

$$340 \quad \epsilon(f) = E|X(f) - \hat{X}(f)|^2 \tag{14}$$

341 with E denoting the expectation. When Eq. (2b7b) and Eq. (813) are substituted into Eq. (914) and the quadratic is
 342 expanded, the mean square error $\epsilon(f)$ can be differentiated with respect to $H(f)$ and the derivative $\frac{d\epsilon(f)}{dH(f)}$ is set to zero
 343 to achieve the minimization; under the assumption that the noise $N(f)$ is independent of $X(f)$, $H(f)$ is derived as

$$344 \quad H(f) = \frac{G(f)S(f)}{|G(f)|^2 S(f) + N(f)} \tag{15}$$

345 where $G(f)$ is the Fourier transform of $\hat{g}(t)$ derived from the second lab experiment described above, $S(f) = E|X(f)|^2$
 346 and $N(f) = E|N(f)|^2$ are the mean power spectral densities of the original concentration series $x(t)$ and the noise $n(t)$,
 347 respectively. Equation (4015) could be rewritten as:

$$348 \quad H(f) = \frac{1}{G(f)} \left[\frac{|G(f)|^2}{|G(f)|^2 + N(f)/S(f)} \right] = \frac{1}{G(f)} \left[\frac{|G(f)|^2}{|G(f)|^2 + 1/SNR(f)} \right] \tag{16}$$

349 where $SNR(f) = S(f)/N(f)$ is the signal-to-noise ratio.

350 Substituting Eq. (4116) into Eq. (813), $\hat{X}(f)$, the Fourier transforms of $\hat{x}(t)$, is derived. The deconvolution is
 351 completed with the inverse Fourier transform of $\hat{X}(f)$ to give $\hat{x}(t)$, the estimated air concentrations. The deconvolved
 352 series of CH₄ and CO₂ restored with the Wiener convolution filter are shown in Fig. 34(a) and (b), and the correlation
 353 coefficient between the deconvolved results and the online measurements with the CRDS analyzer are 0.93 and 0.79 for
 354 CH₄ and CO₂ (Fig. 4 (e) and (h)), higher than that between the original air sampler measurement and the CRDS analyzer.
 355 These results indicating indicates the effectiveness of the Wiener theorem to deconvolve a smoothed series to a much
 356 higher time resolution while accounting for noise. The restored series is improved in terms of time resolution, from about
 357 20 seconds mentioned above to about 3-4 seconds after the deconvolution. The lab test data from the online measurements
 358 contain strong high-frequency components, artificially manipulated to provide an extreme case for testing the
 359 deconvolution algorithm. Such high frequencies lead to some residual noise in the deconvolved results, primarily as a
 360 result of choosing the cutoff frequencies for the mean power spectral densities $S(f)$ and $N(f)$. Nevertheless, such a
 361 situation will be improved for sampling in the real atmosphere where sub-second high-frequency variations are not
 362 common.

363 **5.4. Field application**

364 To apply the UAV-based measurement system described above to atmospheric measurements of CO₂ and CH₄,
365 flights were made at the Shagang Group located in Jiangsu, China on 28 December 2021. Shagang Group is a major iron
366 and steel company on the south shore of the Yangtze River (31.9704° N, 120.6443° E). The company produces over 40
367 million tons of steel each year, making it one of China's top-five steel producers. Onsite coke making for iron production
368 is located in the western part of the Shagang Steel complex. The coke making process is to dry distill coal in a coking
369 oven at ~1000 °C temperature to boil off volatile components to form coke (metallic coal). During coke production,
370 combustion of coking oven gas, blast furnace gas from steel making, and coal tar plus light oil for heating the coking
371 oven is the main CO₂ and CH₄ emission source.

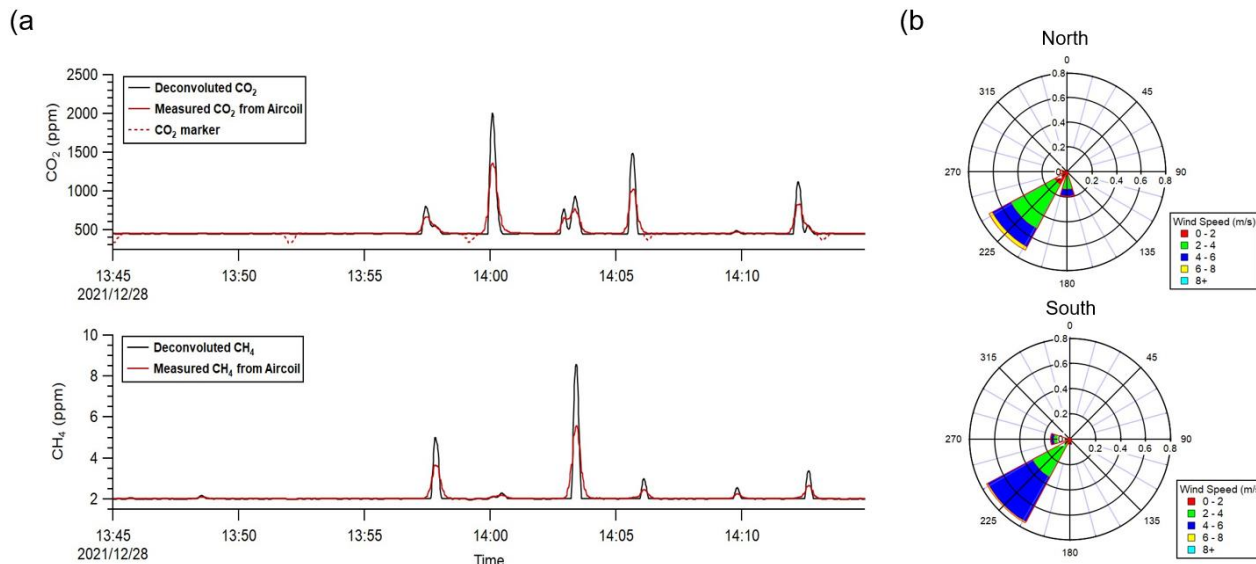
372 Two coking plant stacks were chosen as the target emission source for the field UAV flight. During flight, the UAV
373 was flown in a rectangle pattern (200m×500m) that encloses the two stacks, with repeated flight tracks at 9 altitude levels
374 that, when stacked, created a virtual box and intercepted the emitted CO₂ and CH₄ plumes on the downwind side of the
375 box. The UAV ascended from ~~15 m a.s.l. the ground~~ to ~~150-135~~ m a.s.g.l. and started the box flight at this altitude,
376 ascending 15 m every level and reaching a maximum altitude of ~~270-255~~ m a.s.g.l. before landing. The UAV maintained
377 a constant horizontal speed of 8 m s⁻¹ during flight. ~~The flight lasted for approximately 30 minutes. It's assumed that the~~
378 ~~plume remains steady during the time of measurement.~~ After landing, the air sample collected in the sampler was
379 immediately analyzed with the CRDS analyzer as per the procedure described above in Fig. 2(b).

380 ~~CH₄ and CO₂ emission rates from both stacks were determined using a modified version of the Top-down Emission~~
381 ~~Rate Retrieval Algorithm (TERRA)(Gordon et al., 2015) using their measured mixing ratios and the meteorological data~~
382 ~~collected on board the UAV during the flight. TERRA is a mass balance algorithm, where pollutant emission rates are~~
383 ~~estimated based on the divergence theorem which equates the change in mass within a control volume with the integrated~~
384 ~~mass flux through the walls of the control volume plus the emission rates. It has been used successfully and extensively~~
385 ~~for emission rate determination of tens of volatile organic compounds(Li et al., 2017), CO₂(Liggio et al., 2019),~~
386 ~~CH₄(Baray et al., 2018), oxidized sulphur and nitrogen(Hayden et al., 2021), black carbon(Cheng et al., 2020), and~~
387 ~~secondary organic aerosol(Liggio et al., 2016) using aircraft measurements. In this study, the original TERRA is further~~
388 ~~modified and tailored to make use of the high resolution UAV based measurements.~~

389 **65. Result and discussion**

390 **65.1 CH₄ and CO₂ mixing ratio enhancement from the coking plant**

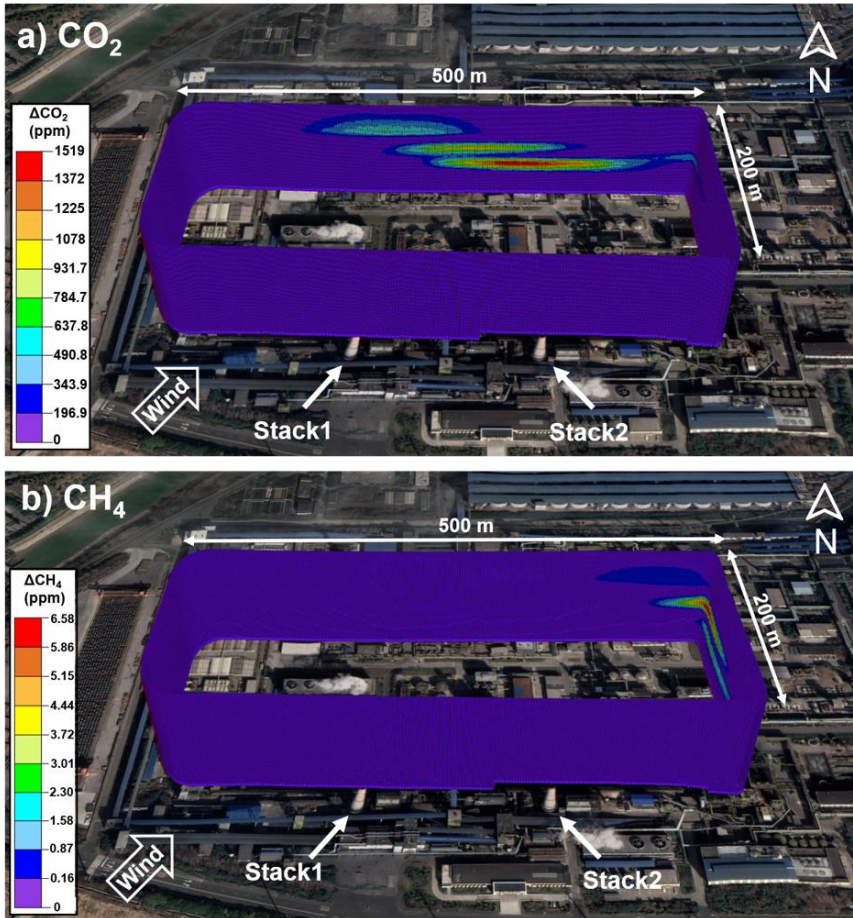
391 Figure 45(a) shows the time series of CH₄ and CO₂ mole fraction mixing ratios measured with the air sampler at the coking
392 plant during the flight (red line). The air sampler sampled for a total of 30 minutes during the flight. After landing, the air
393 sample was analyzed for 10 minutes, as the analysis flow rate triples the sampling flow rate (54.0 sccm vs. 18.0 sccm).
394 The time scales of instrument readings were then stretched three times to restore the original time scales. The CH₄ and
395 CO₂ time series were then deconvolved using the convolution kernel obtained from laboratory test (Section 3.2) to restore
396 the mixing ratio time series in air (black line). The meteorological parameters during the time of flight were measured by
397 the 3D anemometer, showing consistent southwesterly winds with a mean wind speed of 3.0 m s⁻¹ (Fig. 5(b)). The average
398 windspeed is 4.7±4.9 m/s and the average wind direction is 216.4±38.4° during the time of flight. Consistency of wind
399 measurements can be seen from the two wind rose plots for the northern wall and the southern wall respectively. During
400 the flight, the maximum mixing ratio measured was 5.6 ppm for CH₄ and 1356 ppm for CO₂. During the 30-minute flight,
401 a total of 5 CO₂ makers were generated during the 30 minutes of sampling (Fig. 5(a)), and the decreases in the marker
402 concentrations are corrected with a Gaussian form function.



403
404 **Figure 45.** (a) Red line represents CH₄ and CO₂ mole fraction mixing ratios measured from the air samples collected with the air
405 sampler during the flight at the coking plant. Black line represents the deconvolved CH₄ and CO₂ time series and red dashed line
406 sections represent the original marker CO₂ concentrations every 7 minutes. (b) Wind rose plot for the northern and southern wall based
407 on the onboard meteorological measurements during the flight.

408 **6.5.2 Emission estimation**

409 The CO₂ and CH₄ emission rates for the stacks from coking plant were estimated by applying a version of the computation
410 algorithm TERRA specifically modified to suit UAV measurements. The deconvolved mixing ratio time series of CO₂
411 and CH₄ were used in the TERRA algorithm. The algorithm first maps the mixing ratios to the walls of the virtual box,
412 then applies a kriging scheme to interpolate the data and produces a 2 m (vertical) by 1 m (horizontal) mesh on the virtual
413 box walls (200m×500m) (Fig. 56). ~~The semivariogram of the flight points was fitted with a spherical model (range=300,
414 sill=3, nugget=0).~~ Wind speed and wind direction are first decomposed into northly and easterly components, then further
415 converted to vectors that are normal to and parallel to the walls of the virtual box before kriging. ~~Background CH₄ and
416 CO₂ were determined using upwind measurements. The background between upwind data was linearly interpolated and
417 box-car smoothed within a 3-4 minute moving window to derive a variable baseline CH₄ and CO₂ for the entire 30-minute
418 flight.~~ As shown in Fig. 56, the CH₄ and CO₂ plumes can be seen at different locations on the downwind side of the box
419 wall, which indicates that the CH₄ plume and the CO₂ plume probably came from different sources within the box. Using
420 the modified version of TERRA, the emission rates for the two stacks in the coking plant were calculated to be $0.12 \pm$
421 0.01 t h^{-1} for CH₄ and $110 \pm 20 \text{ t h}^{-1}$ for CO₂. The uncertainties for the estimates were derived from detailed analyses of
422 each uncertainty source including measurement error in mixing ratio and wind speed, the near-surface wind extrapolation,
423 the near-surface mixing ratio extrapolation, box-top mixing ratio, box-top height and deconvolution. ~~For cases that uses
424 the Air Sampler system instead of online measuring instruments, as the CO₂ and CH₄ time series measured from the Air
425 Sampler were deconvoluted to restore the unsmoothed time series before putting into the TERRA algorithm, it is necessary
426 to account for the uncertainty that comes from deconvolution. Time series before and after deconvolution were applied
427 to the TERRA algorithm to obtain the total emission rates, calculation shows that emission rates before and after
428 deconvolution vary within 1%.~~



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Figure 56. Virtual flight box for monitoring CO₂ (a) and CH₄ (b) during the flight. The CO₂ and CH₄ plumes were captured on the north and east wall respectively. The wind came from the southwestern direction. [SatelliteSatellite](#)-imagery © Google Earth 2019.

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5.3 Uncertainty Analysis

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To determine the overall uncertainty in the emission rates, each source of uncertainty contributing to the overall uncertainty needs to be identified and quantified. For the emission rate quantification from UAV measurement, the sources of uncertainties include: measurement uncertainties in the mixing ratios and wind speeds (δ_M), the near-surface wind extrapolation (δ_{Wind}), the near-surface mixing ratio extrapolation (δ_{Ex}), box-top mixing ratio (δ_{Top}), box-top height (δ_{BH}) and uncertainties due to data deconvolution as shown in the main text (δ_{Deconv}). Each uncertainty is treated as an independent estimate, and all uncertainties are propagated in quadrature to determine the overall uncertainty in the estimated emission rate:

440

$$\delta^2 = \delta_M^2 + \delta_{Wind}^2 + \delta_{Ex}^2 + \delta_{Top}^2 + \delta_{BH}^2 + \delta_{Deconv}^2 \quad (17)$$

441

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The accuracy of the mixing ratio measurements from the Picarro CRDS analyzer is 50 ppb and 1 ppb for CO₂ and CH₄, respectively. By adding variations in the measured mixing ratios based on the measurement accuracies and re-

443 applying TERRA, the derived emission rates varied within 1% for both CO₂ and CH₄. Thus, the uncertainties in the
 444 emission rates due to mixing ratio measurements (δ_M) were estimated at 1% for both CH₄ and CO₂.

445 The anemometer measures wind speeds with an accuracy of ± 0.1 m s⁻¹ at wind speeds < 10 m/s and wind directions
 446 with an accuracy of $\pm 1^\circ$. The uncertainty of the wind measurements (δ_{Wind}) was estimated using error propagation in
 447 the normal wind $U_\perp(s, z)$, as it is calculated from the northerly and easterly wind components, thus from wind speed and
 448 wind direction:

$$449 \delta_{U_\perp} = \sqrt{\delta_{easterly}^2 + \delta_{northerly}^2 + 2\sigma_{easterly-northerly}} \quad (18)$$

$$450 \delta_{easterly} = |WScos(WD)\sigma_{WD}| \quad (19)$$

$$451 \delta_{northerly} = |WScin(WD)\sigma_{WD}| \quad (20)$$

452 Using this calculation, the uncertainty of the normal wind $\delta_{U_\perp}(s, z)$ was derived at each location. The uncertainty
 453 contributed to the total emission rates to the overall computed emission rate was examined by setting the normal wind to
 454 its upper and lower bounds defined by its uncertainty range, followed by computing the emission rates using TERRA.
 455 The derived CH₄ and CO₂ emission rates varied by 1.5% and 1.9% respectively. Hence the uncertainties from wind speed
 456 measurements (δ_{Wind}) were conservatively estimated to be 2% for both CH₄ and CO₂.

457 Due to a lack of near-surface measurements along the box walls, extrapolation of CH₄ and mixing ratios from the
 458 lowest flight path (~ 150 m above ground level) to the ground level has been shown to be a source of potentially large
 459 uncertainty within TERRA. The magnitude of the uncertainty depends on the nature of the emissions; for example, surface
 460 emissions which may not be fully captured by the flight altitude range have higher uncertainties at $\approx 20\%$, whereas elevated
 461 stack emissions which are fully captured by flight altitude range lead to much smaller uncertainties of <4% in the emission
 462 estimates (Gordon et al., 2015). In the present study, to estimate uncertainties due to extrapolating mixing ratios from the
 463 lowest flight track to the ground (δ_{Ex}), results from all extrapolation techniques (i.e., linear to the ground, constant value
 464 to the ground, linear to background value, or some combination of methods) were derived and compared with the result
 465 using a background value below flight altitudes. Therefore, this term of uncertainty was evaluated at 2% and 6% for CH₄
 466 and CO₂ respectively.

467 **Table1.** Emission rates derived using different extrapolation techniques

	<u>All</u>	<u>Constant</u>	<u>Linear</u>		
<u>Extrapolation</u>	<u>background</u>	<u>value from</u>	<u>between</u>		
<u>techniques</u>	<u>below flight</u>	<u>lowest flight</u>	<u>constant and</u>	<u>linear</u>	<u>exponential</u>
	<u>altitude (this</u>	<u>altitude to</u>	<u>background</u>		
	<u>study)</u>	<u>surface</u>	<u>at surface</u>		

<u>CH₄</u>					
<u>Emissions(kg/hr)</u>	<u>115.7</u>	<u>113.9</u>	<u>116.9</u>	<u>113.9</u>	<u>113.6</u>
<u>CO₂</u>					
<u>Emissions(kg/hr)</u>	<u>110100</u>	<u>109970</u>	<u>109400</u>	<u>109970</u>	<u>103960</u>

468 Additional components contributing to uncertainties in the computed emission rates specific to the box approach
469 include box-top mixing ratio (δ_{Top}) and box-top height (δ_{BH}). The TERRA box approach assumes a constant mixing ratio
470 at the box top ($X_{C,Top}$) by averaging the measured value at the top level. The term δ_{Top} is determined from the 95%
471 confidence interval ($2\sigma/\sqrt{n}$) of the interpolated measurements. The calculated confidence interval of the mixing ratio at
472 the box top is 0.01 ± 0.13 ppm for CH₄ and 70.1 ± 89.1 ppm for CO₂. A top average mixing ratio of 0.14 ppm for CH₄
473 and 159.2 ppm for CO₂ are set as input parameters to derive resulting uncertainties in the emissions rates. Thus, 106.6
474 kg/hr for CH₄ and 93760 kg/hr for CO₂ were derived. Then, this uncertainty term is conservatively taken as 8% and 16%
475 for CH₄ and CO₂.

476 The uncertainty due to the choice of box height, δ_{BH} , within TERRA is estimated by recomputing the emission rate
477 with a reduced box height (z) of 100 m. The recalculated emission rate after reducing the box height of 100m is 106.4
478 kg/hr for CH₄ and 113500 kg/hr for CO₂, thus δ_{BH} is estimated as 8% for CH₄ and 3% for CO₂.

479 For cases that use the air sampling system instead of online measuring instruments, as the CH₄ and CO₂ time series
480 measured from the air sampler were deconvoluted to restore the unsmoothed time series before being input into the
481 TERRA algorithm, it is necessary to account for the uncertainty that comes from such deconvolution as outlined in the
482 main text. Time series before and after deconvolution were applied to the TERRA algorithm to obtain the total emission
483 rates. The computations show that emission rates before and after deconvolution vary within 1%, which was taken as the
484 uncertainty δ_{deconv} . The assessment of uncertainties for the TERRA-computed emission rates from the coking plant are
485 listed in Table 2.

486 **Table 2.** Assessment of percent uncertainties for CH₄ and CO₂ emission rate estimations

	<u>CH₄ (%)</u>	<u>CO₂ (%)</u>
δ_M	<u>1</u>	<u>1</u>
δ_{Wind}	<u>2</u>	<u>2</u>
δ_{Ex}	<u>2</u>	<u>6</u>
δ_{Top}	<u>8</u>	<u>16</u>
δ_{BH}	<u>8</u>	<u>3</u>
δ_{Deconv}	<u>1</u>	<u>1</u>
δ	<u>12</u>	<u>18</u>

487 **6.5.4 Comparison with Gaussian Inversion Approach**

488 The TERRA computation results can be further evaluated. Of the multiple CH₄ plumes that were captured on the north
489 and east walls of the virtual box, the largest CH₄ one resembles a nearly perfect Gaussian plume distribution and is clearly
490 associated with the east stack of the two, for which the emission rate may be recalculated using the Gaussian plume model.
491 The Gaussian plume model makes basic assumptions that the plume is emitted from a point source and that the
492 atmospheric turbulence is constant in space and time (Visscher, 2014). In this study, the captured plume was completely
493 elevated and thus not constrained by boundaries. In the absence of boundaries, the equation for pollutant mixing ratios in
494 Gaussian plumes is as follows:

$$495 \quad c = \frac{Q}{2\pi\bar{u}\sigma_y\sigma_z} \exp\left(-\frac{y^2}{2\sigma_y^2}\right) \exp\left(-\frac{(z-h)^2}{2\sigma_z^2}\right) \quad (21)$$

496 where c is the concentration at a given position x , y and z (g m^{-3}), Q is the emission rate (g s^{-1}), \bar{u} is the mean wind speed
497 (m s^{-1}), h is the effective source height (m) and σ_y and σ_z are dispersion parameters in the horizontal (lateral) and vertical
498 directions respectively (m).

499 The dispersion parameters σ_y and σ_z were obtained by fitting the spatial distribution of CH₄ mixing ratios on the
500 measurement screen into a Gaussian function. As the wall intercepting the plume is not perpendicular to the wind direction,
501 the plume was projected to a different virtual wall perpendicular to the wind direction before fitting the Gaussian function.
502 By calculating the standard deviations of the Gaussian distributions in the y and z directions, σ_z is estimated to be $6.3 \pm$
503 0.3 m and σ_y is 15.7 ± 0.4 m. The downwind measurement plane is examined to find the point with the highest CH₄
504 mixing ratio of 6.575 ppm and its location ($s = 160$ m, $z = 217$ m). For the separate CH₄ plume, the Gaussian plume model
505 gives an emission rate of 49.40 ± 6.824 kg h⁻¹. The uncertainty is quantified by considering the accuracy of mixing ratio
506 measurement, the variation of wind speed and the confidence interval for the dispersion parameters given by Gaussian
507 function fitting. CH₄ measurement uncertainties from the instrument is <1%. The uncertainty contributed by the mean
508 wind speed estimation was examined by varying the average wind speed by the standard deviation of the wind data around
509 the plume (3.8 ± 0.6 m/s), followed by input into gaussian plume model. This mean wind speed sensitivity analysis resulted
510 in CH₄ emission rates that varied by 16%. The same sensitivity analysis was done with σ_y (15.7 ± 0.4 m) and σ_z ($6.3 \pm$
511 0.3 m), which resulted in CH₄ emission rates that varied by 4% and 3% respectively. Thus, the total uncertainty is added
512 in quadrature to be 17%. The TERRA algorithm is able to obtain the emission rate for a selected section through a certain
513 area of the screen. For this isolated CH₄ plume, the TERRA algorithm computed an emission rate of 65 ± 8 kg h⁻¹, which
514 is comparable to the emission rate estimation from the Gaussian plume model. The number is comparable to the emission

515 ~~rate estimation from the Gaussian plume model, showing the reliability of top-down emission estimation approaches of~~
516 ~~both TERRA and the Gaussian plume model analyses of the UAV measurements.~~

517 **6.5.5 Validation of UAV-based Emissions and Comparison with IPCC-based Emissions**

518 Coking process is one of the highest energy-consuming operations during iron and steel production that tends to emit
519 large amounts of CO₂ and CH₄. According to the Chinese national GHG inventory report, CO₂ and CH₄ emissions from
520 coke production in iron and steel production processes were calculated using the Tier 1 method in the IPCC Guidelines
521 (Ministry of Ecology and Environment of of China, 2018). In the Tier 1 method, default emission factors for coke
522 production are used to estimate the CO₂ and CH₄ emissions without considering local variations, respectively,

$$523 E_{CO_2} = P_{coke} \times EF_{CO_2} \text{ and } E_{CH_4} = P_{coke} \times EF_{CH_4} \text{ .} \quad (22)$$

524 where E_{CO_2} and E_{CH_4} represents the CO₂ and CH₄ emission rates from coke production, P_{coke} represents coke production,
525 EF_{CO_2} and EF_{CH_4} are the IPCC default emission factors for CO₂ and CH₄, which are 0.56 t CO₂/t of coke and 0.1 g CH₄/t
526 of coke, respectively. The measured Shagang coking plant consists of two coke oven batteries, each with its own stack.
527 Each battery produced 127.8 t coke h⁻¹, thus totalling 255.6 t coke h⁻¹ (P_{coke}) between the two batteries during the UAV
528 measurement period with a coke yield of 78.5%. A material balance analysis revealed that CO₂ emitted from the stacks
529 during the full coking process was 103±32 t CO₂ h⁻¹ (SI-Section S3). In comparison, the UAV measurement-based
530 emission rate obtained in this study is 110±18 t CO₂ h⁻¹, which is consistent with the CO₂ emissions based on the material
531 balance analysis. For comparison, multiplying the IPCC default emission factor with the coke production at the Shagang
532 coking plant yields an emission rate from coking of 143 t CO₂ h⁻¹, higher than either the material balance based result by
533 about 39% or the UAV-based result by 30%. This suggests that the IPCC default emission factor is too high for this
534 particular coking plant.

535 On the other hand, the UAV-measurement based emission of 0.12±0.014 t h⁻¹ for CH₄ is four orders of magnitude
536 higher than 1.28×10⁻⁵ t h⁻¹ emissions for CH₄ estimated using the IPCC Tier 1 emission factor EF_{CH_4} . The IPCC emission
537 factor for coke production is derived by averaging plant-specific CH₄ emissions data for 11 European coke plants reported
538 in the IPPC I&S BAT Document (European IPPC Bureau, 2001), but information about the data collection method such
539 as sampling methods, analysis methods, time intervals, computation methods and reference conditions is not available
540 according the report. It is important to note that the present UAV measurement represents a one-time measurement where
541 there was only one flight conducted in this campaign. The result clearly serves the purpose for validating the overall
542 methodology from air sampling and analysis, computing the emission rates, to estimating the associated errors. The

543 fundamental assumption in the mass balance approach is that plumes and emissions remain constant throughout the
544 measurement period. Given the short duration of the flight and the good comparison between the present emission result
545 and the material balance emission estimate, such an assumption appears to be valid. However, a hypothesis of a constant
546 emission rate over time remains to be tested. Conducting multiple flights over time, computing emission rates and
547 assessing their uncertainties will allow for statistical sampling of the probability distribution of the emission rates and
548 hence deriving the mathematical expectation of the emission rate. Only then the derived emission factors can be used for
549 inventory preparation and/or comparison with existing ones with statistical confidence. Given the limited circumstance
550 of having only one flight in this study, it becomes clear such purpose cannot be achieved. Consequently, the emission
551 values of CH₄ derived from measurements in this section are only suitable for qualitative comparisons with those used
552 published emission factors. The comparison results indicate that real-world emission factors may significantly differ from
553 the default emission factors but more work is needed.~~While the present UAV measurement represent a one-time~~
554 ~~measurement and it is difficult to determine the representativeness of this emission rate, the fact of the large discrepancies~~
555 ~~suggests that real world emission factors can be significantly different from the default emission factors.~~ The additional
556 CH₄ may come from ~~taps the leakage or door leakage in addition to the conventional combustion process during coke~~
557 ~~production of the coke oven gas when it is recycled as fuel in firing the coke oven (SI).~~ Both reasons point to a need for
558 further emission measurements to determine the local emission factors and a further validation of the CH₄ emission factors
559 of coke production.

560 **7.6 Conclusions**

561 In this paper, we present the development of a UAV measurement system for quantifying GHG emissions at facility levels.
562 The key element of this system is a newly designed air sampler, consisting of a 150-meter-long thin-walled stainless steel
563 tube with remote-controlled time stamping. Through laboratory testing, we found that the air sampler generated smoothed
564 time series data compared to online measurement by the CRDS analyzer. To addressing the smoothing effect, we
565 developed a deconvolution algorithm to restore the resolution of the time series obtained by the air sampler. For field
566 validation, the new UAV measurement system was deployed at the Shagang Steel to obtain CO₂ and CH₄ emissions from
567 the main coking plant at Shagang Steel. ~~Mole fraction~~Mixing ratios of CO₂ and CH₄ together with meteorological
568 parameters were measured during the test flight. The mass-balance algorithm TERRA was used to estimate the coking
569 plant CO₂ and CH₄ emission rates based on the UAV-measured data. For further analysis, we compared these emission
570 results with those derived using Gaussian plume inversion approach and carbon material balance methods, demonstrating

571 good consistency among different approaches. In addition, when compared the top-down UAV-based measurement
572 results to that derived from the bottom-up emission inventory method, the present findings indicated that the ~~use of IPCC~~
573 emission factors ~~can be significantly different from the actual emission factors for emission calculations can lead to~~
574 ~~overestimation.~~

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577 *Data availability.* Data are available upon request by the corresponding author.

578 *Author contribution.* TH, CX, YL and, SML conducted the fieldwork with the support by XG, XZ, and FB. TH and CX
579 conducted laboratory experiments with the guidance by SML. TH performed the primary data analysis, and wrote the
580 initial draft of the manuscript. YH provided expertise in model analysis. Algorithm programming was provided by YL.
581 YY and YZ did the wind data correction. SML reviewd and edited the manuscript, and ensured the accuracy and integrity
582 of the study.

583 *Competing interests.* The authors declare that they have no conflict of interest.

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