

General Response:

We thank the Referees for their time to provide comments and feedback on this manuscript. We have edited the manuscript to address and incorporate their suggestions. In many cases, we realize that the clarity of our writing could be improved to address comments where the Reviewers had to make assumptions about technical details of our technique, and its validation. The Introduction has been thoroughly streamlined, as have sections of the Results and Discussion. We think that the manuscript contents are now clear, reflect the necessary corrections, and that the work is improved to a state we think meets the expectations for publication in Atmospheric Measurement Techniques. We hope the Editor and Reviewers agree. Our responses to specific comments from the Reviewers below are highlighted in yellow, while additions and alterations to our manuscript or in cases where we are redirecting concerns to existing discussion are **highlighted, bolded, and underlined in green for additional clarity.**

Editor

This manuscript has been reviewed by two referees. They believe that the instrument you developed is of great interest and that the work is well conducted. Nevertheless, they both feel that the current manuscript contained a lot of redundant materials and should be substantially shortened. I agree with them very much; in addition, the structure of this manuscript should be further improved, as also pointed out by the two referees.

As a result, in addition to the revised manuscript and point-to-point reply, **could you please also provide a summary of major changes you make in the revision** (in order to make it easier for the referees and also the handling editor)? Thank you very much.

You - as the contact author - are requested to individually respond to all referee comments (RCs) by posting final author comments (ACs) on behalf of all co-authors no later than 28 March 2024 (final response phase).

We thank the Editor for their contribution to the discussion on changes required for this manuscript. We have collected the major points that required addressing from all parties to provide a succinct summary of the modifications to the revised manuscript. Listed below are the five major changes made to the manuscript:

1. The Introduction (Section 1.0) has been revised for clarity and substantially shortened to be better focused on the topic(s) of this manuscript. **The following are the major modifications:**
 - a. In the original manuscript, the Introduction was 2,261 words total and the revised Introduction section is now 1,783 words total, a reduction by over 1/5th.
 - b. Any discussion/mention of those chemical species (e.g., POPs, nitrates/sulfates, and ON) not specifically studied in this work has been moved and integrated succinctly into the new Conclusions and Future Directions (Section 4.0).
 - c. Small – yet impactful – modifications were made to the preexisting information within the last paragraph of the Introduction to better addresses comments made by

either Reviewer regarding the mixing of system validation and the scientific data analysis. The Authors believe that the location of these new modifications clarify and emphasize that sample preservation is based on established methodologies and that comparison to previously reported values in the literature is the primary metric for validation of the chemical measurements within this work.

2. A new subsection has been included within the Materials and Methods – more specifically, to the NL-BELT field site descriptions (Section 2.3).

Sample Preservation (Section 2.3.1) has been added to the manuscript to address Reviewer comments regarding topics such as sample preservation and method validation.

This is a re-working and expansion on the last paragraph in Section 2.3 of the original manuscript. This subsection details our use of mercuric chloride (HgCl_2) as a sample preservation technique and cites its established efficacy in previous studies. With the well-established and long-studied nature of this preservation method, and its use for over a decade within the aquatic and soil biogeochemical work at NL-BELT, the Authors felt that additional tests were not needed to verify the temporal preservation efficacy of collected samples.

We highlight in this new section the role of selective use of HgCl_2 within our replicate samplers as a method of internal sample validation (i.e. comparing samples in sterilized vs. unsterilized containers). We also highlight the versatility of our modular design as being open for use with many other established sample preservation methods, which ultimately allows it to become analyte specific without burdening researchers with laborious method re-development.

3. Within the Results and Discussion section comparing collected samples volumes (3.3), the paragraph detailing our intercomparison of ECCC and DAYMET measurements has been moved to the Supporting Information.

This comparison can now be found in Section S2, “Deposition Comparison: This Work, DAYMET, and ECCC.”

4. A rather minor, yet substantive, comment made regarding the paragraph in Section 3.3 discussing the presented triplicate sampler RSD values has been modified.

The variability observed in the samples is now briefly revised and better addressed, with the preexisting sentences were reordered for better flow.

5. The paragraph examining cation exchange capacity has been completely removed from the Results and Discussion section analyzing precipitation pH within the collected NL-BELT samples (3.4.1). The Authors agree that it serves only our own scientific objectives at NL-BELT was unnecessary and tangential to the manuscript contents.

Reviewer 1

Overview:

This manuscript describes the development of an autonomous precipitation collector (excluding snow) that can be deployed off-grid in remote areas and is suitable for measuring pH, conductivity, and DOC (Dissolved Organic Carbon) in wet deposition with a monthly resolution. The collector was deployed and tested over a period of 2 years (2015-2016 on the NL-BELT network) for open fall and throughfall depositions.

General comments:

The subject is scientifically interesting and is well within the scope of the journal. But I think the presenting quality and arguments for validation of this system should be substantially improved before it can be accepted for publication in AMT:

The text is very dense, overly detailed, and often confusing, even off-topic, making it challenging to understand and preventing a clear presentation of the study's results. Moreover, certain parts of the text resemble a promotional brochure (e.g., in the "general design advantages" section) or a technical manual (e.g., 3.2.1) that does not appear suitable for a scientific article. I have the impression that this is a study report that has simply been slightly reformatted for submission to AMT (without even sorting through the relevant or irrelevant information for the purpose of this paper). A scientific article about a new method/system should demonstrate that the developed tool meets the expectations and requirements set to address scientific questions, rather than being a comparison with commercial tools or a series of technical information. For example, in 3.2.1, it doesn't matter that commercial samplers consume 100 times more; what is relevant here is the autonomy of your device.

We appreciate the thorough and thoughtful comments of the Reviewer. Where applicable, we have modified the text to be less dense, more concise, and less confusing for the readership. Given that there are no affordable commercial solutions for this type of sampling equipment, the manuscript needs to necessarily argue the rationale for the component selection and design. These sections aim to convey the unique features and advantages of our design, which is essential for assessment and effective uptake by the scientific community, particularly with respect to modular components. This is a central thrust of making instrumentation Open Access and research tools globally accessible.

We respectfully disagree with the position of the Reviewer on this point, as there are many manuscripts in this journal, and others related to instrument development, that are similarly detailed where novel platforms have been developed. As such, comparison with commercially available tools is relevant to underscore the application of our device in a broader context. When it comes to autonomy of using these samplers off-grid, power requirements are surely a fair comparison for stating why current commercial solutions are not viable.

The "Results" section mixes the points about the system validation and the scientific data analysis. The results/discussion section could be divided into two parts for clarity:

- A section titled "Validation": I believe there would be a significant benefit in simplifying the argument and presenting, point by point, the performance parameters. The article aims to demonstrate that the developed collectors are suitable for autonomously measuring conductive rain deposition in remote areas, so the key points of this demonstration could be highlighted:
 1. Autonomous off-grid operations: low power consumption + opening conditions for the collection of wet deposition (time taken to open or close...)
 2. Representativeness of the conductive rain collector for deposition measurement: Define what constitutes conductive rain compared to "common" rain.
 3. Validation of measured chemical parameters: Verify the preservation of chemical species even after a month, including pH, conductivity, and DOC.

We thank the Reviewer for their comments. We must note a major oversight on the part of the Reviewer and in our emphasis of method validation before proceeding further with our responses. Validation throughout this manuscript was conducted by controlling and observing in two ways:

- i) Microbes will transform bioavailable molecules and so we microbially sterilized two out of three sample replicates with HgCl_2 , which is standard practice for biogeochemical handling of environmental water samples for chemical characterization. This technique has been in use for over a decade within our aquatic and soil biogeochemical work at NL-BELT (Ziegler et al., 2017; Bowering et al., 2020; Bowering et al., 2022; Bowering et al., 2023). We then compare those sterilized results against a replicate that was not administered HgCl_2 to ascertain whether microbial issues exist for measurement metrics (e.g., conductivity, pH, DOC). Generally, we did not find a difference between measurements that were microbially sterilized versus those that were not for the presented measures, but this is not a given for other molecules. Given the longstanding history of this approach, we do not feel that extensive restructuring of the manuscript into the Reviewer's three sections is warranted.

We also strongly emphasize that this approach may not work for other molecules of interest to the atmospheric deposition community (e.g., those that are volatile). Sample preservation approaches should be conducted by users of this new platform based on their scientific objectives and review of the literature. We are not in a position to detail all preservation techniques. Please see our detailed response to comment 3 below and the newly added Section 2.3.1 entitled "Sample Preservation."

- ii) We perform comparisons to the literature for validation of our measurements against longstanding measurement techniques used by academics and regulatory agencies globally. We do not see a scientific contribution that validates precipitation pH measurements as a worthwhile activity, given that the control experiments we have conducted to ensure our samples are preserved have been made in line with methodologies that have been in place since the 1960s.

We have addressed the numbered points offered by the Reviewer in detail as follows:

1. The time it takes for the system to open, based on several years of qualitative observation, has been added to the section entitled “Heated Precipitation Sensor” (2.1.2):

An output of 12 VDC is sent to the digital control board by the relay when rain is sensed, or 0 VDC in its absence, for signal processing and motor control (Figure S7). **When rain is sensed, the lid of each sampler in the array is simultaneously opened (<5 seconds) and is dependent on the rotational rate of the lid motor.** To increase the sensitivity of this sensor and to extend the sampling duration when conductive atmospheric constituents are completely washed out of the atmosphere, a sloped tin chute (e.g., Home Depot SKU# 1001110514) was added to extend the surface of the rain sensor.

2. For improved clarity, we will redefine what constitutes conductive rain compared to "common" rain in Section 3.1 (General Design Advantages):

The chute does this by accumulating water-soluble materials between rain events that require time to be completely washed off and through the release of ions from the material itself, which ages under environmental conditions. As the conductivity of the precipitation falls below the sensor threshold – **conductive precipitation being that which initially contains high solute levels that progress through trace level concentrations**, the added ions from the chute prolong the collection of rain past this time point. In rainfall events where extended atmospheric wash-out occurs, **where precipitation becomes ultrapure water**, the sampler lids will eventually close – preventing dilution of the sample while maintaining the collection of analytes of interest.

3. Section 2.3 (Continuous Monthly Collection of Remote Samples at NL-BELT) was revised to include a new ‘Sample Preservation’ subsection (Section 2.3.1). This subsection addresses our use of a well-established sample preservation technique and clarifies why additional tests were not done to verify the preservation of collected chemical species over time. We believe that this new subsection, in addition to the small modifications made to clarify the motivations and objective of our approach in the final paragraph of the Introduction addresses the comments made by both Reviewers regarding their initial confusion on the mixing of system validation and the scientific data analysis. Please see below for the modifications made to the Introduction as well as the new additions to Section 2.3.1 specifically pertaining to sample preservation and validation:

1.0. Introduction (final paragraph)

The materials used can be easily changed in order to optimize collection **and preservation** of a wide array of target analytes, such as DOC, **when using high density polyethylene and mercuric chloride (HgCl₂)**. We demonstrate that these platforms are capable of continuous operation off-grid for monthly wet deposition collection of precipitation across the Newfoundland and Labrador Boreal Ecosystem Latitudinal Transect (NL-BELT) during snow-free periods in 2015 and 2016.

The captured fraction compared to total volume deposited is used to gain insight into how these samplers can limit analyte dilution effects and improve method detection limits, **such as rejecting 50% of the total volume delivered as ultrapure precipitation leading to a factor of two improvement**. Chemical parameters of pH, conductivity, and DOC fluxes **collected according to established preservation protocols** were then **compared to prior measurements** to validate this proof-of-concept system. Measurement methods for pH and conductivity of rainwater are very well-established in the literature and serve as a baseline reference to ensure that the samples collected by the new devices presented in this work are consistent with what is expected in samples from a remote coastal environment, given the selective sampling strategy. We then move away from these well-established parameters to quantify DOC fluxes **using established biogeochemical preservation techniques for fresh water and groundwater** to demonstrate the potential of these samplers in application to automated collection of analytes of emerging importance and interest in the remote locations of our latitudinal transect.

2.3.1 Sample Preservation

Four of the six sample containers (two each of OF and TF) were biologically sterilized using 1 mL of a saturated aqueous solution of mercuric chloride (HgCl₂) to preserve against biological growth and loss of bioavailable nutrients over the collection periods. Unsterilized sample containers (without HgCl₂) were used for measurements of recalcitrant species and to assess any matrix effects exerted on target analyte quantitation. **The use of HgCl₂ as a sample preservation technique has been long-studied and well-established (Kirkwood, 1992; Kattner, 1999); thus, additional tests to verify the preservation of collected chemical species over time were not performed. The analysis of deposition collected in unsterilized and**

sterilized containers, however, serves as a method for internal sample validation - as does our evaluation of measurement outcomes in comparison to those reported within the literature.

Filtered samples were transferred to new clean HDPE containers and stored for up to two months at 4°C in a cold room until analysis. **The target analytes in this work are non-volatile and the described sample collection methods consider this analyte property, as well as their interactions with container materials. The versatility of the system design allows for the use of different collection materials, keeper solvents for volatile organics, etc., so that the experimental design can be analyte specific, depending on end user needs. Sample preservation approaches should thus be identified by users of this new platform based on their scientific objectives and review of the literature (Galloway and Likens, 1978; Peden et al., 1986; Dossett and Bowersox, 1999; Wetherbee et al., 2010). In addition to the internal validation approach described here, we aim to demonstrate that the precipitation samplers in this work are suitable for measuring conductive deposition on- and off-grid. Below we highlight autonomous off-grid operations, determine the fraction of conductive rainfall collected from the total volume of precipitation, and validate our measurements through comparison to the literature.**

A section titled "Application": to demonstrate the usefulness of the collector in identifying canopy effects, through a comparison of the obtained data with the literature and between OF and TF for pH, conductivity, and DOC.

We thank the Reviewer for their suggestion. We believe that our existing Sections 3.3 and 3.4 demonstrate the 'Applications' of our collector in identifying canopy effects as well as comparing our measurements to those reported within the relevant literature. The extent of this analysis serves as an example of how the samplers may be used, which will be exploited in our future work to explicitly consider canopy effects on deposited species. For brevity, as both Reviewers and the Editor have requested reduced content, we have opted not to separate and expand the discussion of canopy effects with respect to the observed OF and TF deposition.

Instead, we have tried to emphasize this in our Conclusions and Future Directions (Section 4.0):

Comparability and complementarity of our results to well-established and current measurements of interest like DOC, demonstrate their efficacy and potential application to the study of processes such as canopy-precipitation interactions through the collection of open fall and throughfall replicates.

The samples collected in this work from this new instrumentation, specifically, are expected to be used further in several upcoming complementary and novel environmental monitoring studies. Not only will this future work extend our biogeochemical analysis, but it will also assist in our studying of the transport of other anthropogenic pollutants of emerging interest which are beyond the scope of describing this new platform.

In addition to presenting quality, the developed arguments do not justify the scientific quality of the proposed measurements. The validation of the system's operation and measurement is somewhat fragmented and, as such, not very convincing. Specifically, three points are missing in the developed arguments:

1. This is a rain collection system where the opening/closing is controlled by a resistive sensor which is activated from the conductivity of the rain. However, the operational range of the sensor is not specified (it triggers at 1 mohm.cm, it is the only information). Thus, there is no definition of what is being truly measured.

We thank the Reviewer for highlighting this point, but it is also somewhat confusing. A threshold for generating a response would exist in all sensors, so the one we have provided represents the lower limit of the range. This is common practice for many instrumental techniques, where the detection limit is reported. We will note that all standard deposition samplers in government monitoring networks use the same sensing approach for precipitation (often without a design to increase their sensitivity; e.g., NADP, CAPMoN, etc.; (Canadian Air and Precipitation Monitoring Network, 1985b)). Without a specific reference to other programs' definitions, and some sort of example expectation from the Reviewer, is it hard to ascertain what might be most useful to address their concern. As such, we have retained our original lower limit conductivity definition and added the equivalent concentration in sodium chloride. We think this clarifies the work and hope the Reviewer agrees.

We have added the following to Section 2.1.2, "Heated Precipitation Sensor":

The detection of rain modulates the opening and closing of the collection units by an interdigitated resistive sensor (M152; Kemo Electronic GmbH, Geestland, Germany; Figures S6 to S8). **This approach is consistent with established precipitation detection techniques used by government monitoring programs (e.g., CAPMoN; Canadian Air and Precipitation Monitoring Network, 1985a, 1985b).** The rain sensor detects conductive deposition by the completion of a conductive circuit when electrolytes bridge the connection between the interdigitated gold electrodes. The sensor is supplied with 12 VDC from the power system to trigger a relay when precipitation conductance above 1 M Ω ·cm conductivity is detected

(determined experimentally, see Section S1). **This is equivalent to approximately 8 μ M sodium chloride. The sensor detection limit reflects an upper limit of precipitation ion loading because the design of the rain chute leads to an increase in surface area of more than a factor of 25 on which solutes can accumulate to enhance the ionic content of the deposited water.** An output of 12 VDC is sent to the digital control board by the relay when rain is sensed, or 0 VDC in its absence, for signal processing and motor control (Figure S7).

2. The time it takes for the system to open is also not mentioned. We know that the first seconds of rain contain strong concentrations, so if the opening occurs 30 seconds after the start of the rain, a significant amount of information will be lost.

The Reviewer mentioned this point in a prior comment, which has already been addressed with a technical addition to the paper. We thank the Reviewer for this comment, and we understand the need to communicate that the samplers open readily when rain is sensed, for the facts they mention here. The opening of the lid is fast (<5 seconds) and is dependent on the rotation rate of the motor selected. We typically use 2 to 6 rpm motors, depending on their availability from our suppliers. We hope the inclusion of this information in our revision is satisfactory.

3. The system is dedicated to measuring monthly fluxes of chemical species, but no information is provided on the preservation of the tested parameters (pH, conductivity, and DOC) between each sample retrieval."

We agree with the Reviewer, and we have now included a new subsection (Section 2.3.1) to address our use of a well-established sample preservation technique for microbially labile DOC, and hence, why additional tests were not done to verify the preservation of collected chemical species over time. We have also now clearly acknowledged that other analytes may require other considerations, depending on their chemical and physical properties (e.g., volatility).

We have also noted in our responses above that our comparison to long-standing reports of a parameter such as precipitation pH, using calibrated meters and standard collection/storage/analysis techniques from the literature (Dossett and Bowersox, 1999; Wetherbee et al., 2010) means that further evaluation of the preservation is not warranted, as the outcomes (i.e., the samples will be stable) are predictable. Given that both Reviewers have also indicated that the manuscript is too long, we do not feel that adding more results to this work are warranted. They have taken the time and attention to identify redundant components of the original work and we feel that it is important to emphasize where we find suggested additions to also be redundant. Especially results that fail to make substantive contributions to this field. We hope that our efforts to improve clarity in our justified approach, guidance for the community, and doing so concisely for the overall improvement of the manuscript are agreeable and meaningful changes to address the concerns of the Reviewer.

Specific comments:

Introduction

The introduction is very general and is organized into different paragraphs discussing the importance of monitoring atmospheric deposition of various chemical species (nitrates/sulfates (L83-91), POP (L119-129), ON (L157-165)) that are not subsequently addressed in the study, as only pH, conductivity, and DOC are studied here. If these species need to be detailed, it should be done in the conclusion to demonstrate that the developed system could also be used to study them. Please refocus the introduction on aspects related to your study.

We thank the Reviewer for this comment and Reviewer 2 raised similar concerns. The Introduction has been revised and the mention/discussion of studying the deposition of various chemical species (POPs, ON, etc.) not specifically studied in this work has been moved and integrated into the newly titled Section 4.0, "Conclusions and Future Directions" (see below). In the original manuscript, the Introduction was 2,261 words total, and the revised Introduction section is now 1,783 words total.

4.0 Conclusions and Future Directions (last paragraph):

For the broader deposition-motivated community, the instrument design also allows for easy cost-effective modification of the number of replicate samplers, the material composition of all surfaces the aqueous samples interact with, as well as preservation strategies - depending on the analyte of interest. For example, the lack of organic nitrogen measurements within universally established sampling and measurement procedures serves as a general example of the substantial knowledge gaps that may result when translating limited data sets to the wider global picture. This includes incomplete speciation and quantification across precipitation, aerosol, and gas phases. Monitoring systems that support U.S. deposition assessments (e.g., the NADP) only characterize the inorganic fraction of wet deposition. Additionally, modern emerging issues that require the continuation of existing deposition measurements or expansion of observation programs revolve around identifying and quantifying compound classes of concern, such as persistent organic pollutants. As reported in the literature, the deposition of these types of pollutants (e.g., polychlorinated biphenyls, polycyclic aromatic hydrocarbons, etc.) can be monitored using suitable collectors made of amber-coloured glass or stainless steel (Fingler et al., 1994; Amodio et al., 2014) - modifications which can be applied to the sample design detailed here. The samples collected in this work from this new instrumentation, specifically, are expected to be used further in several upcoming complementary and novel environmental monitoring studies. Not only will

this future work extend our biogeochemical analysis, but it will also assist in our studying of the transport of other anthropogenic pollutants of emerging interest which are beyond the scope of describing this new platform.

L148-151: The justification for developing this new collector is based on comparisons with commercial devices, mentioning the cost and difficulty of making measurements in remote areas. It would be interesting to mention that precipitation collectors have already been developed to address specific questions about atmospheric fluxes. Here are a few examples:

- **Laquer, F. C.:** Sequential precipitation samplers: A literature review, *Atmos. Environ. A.-Gen.*, 24, 2289–2297, [https://doi.org/10.1016/0960-1686\(90\)90322-E](https://doi.org/10.1016/0960-1686(90)90322-E), 1990.
- **Germer, S.,** Neill, C., Krusche, A. V., Neto, S. C. G., and Elsenbeer, H.: Seasonal and within-event dynamics of rainfall and throughfall chemistry in an open tropical rainforest in Rondônia, Brazil, *Biogeochemistry*, 86, 155–174, <https://doi.org/10.1007/s10533-007-9152-9>, 2007.
- **Laurent, B.,** Losno, R., Chevaillier, S., Vincent, J., Rouillet, P., Bon Nguyen, E., Ouboulmane, N., Triquet, S., Fornier, M., Raimbault, P., and Bergametti, G.: An automatic collector to monitor insoluble atmospheric deposition: application for mineral dust deposition, *Atmos. Meas. Tech.*, 8, 2801–2811, <https://doi.org/10.5194/amt-8-2801-2015>, 2015.
- **Brahney, J.,** Wetherbee, G., Sexstone, G. A., Youngbull, C., Strong, P., and Heindel, R. C.: A new sampler for the collection and retrieval of dry dust deposition, *Aeolian Research*, 45, 100600, <https://doi.org/10.1016/j.aeolia.2020.100600>, 2020.
- **Audoux, T.,** Laurent, B., Desboeufs, K., Noyalet, G., Maisonneuve, F., Lauret, O., and Chevaillier, S.: Intra-event evolution of elemental and ionic concentrations in wet deposition in an urban environment, *Atmos. Chem. Phys.*, 23, 13485–13503, <https://doi.org/10.5194/acp-23-13485-2023>, 2023.

We thank the Reviewer for directing our attention to these overlooked studies. We have added these in locations where we think they best fit in the manuscript.

1. We mention two of the five references listed above (Laquer, 1990; Germer, 2007), amongst others, within a modified Introduction paragraph to better address the preexisting precipitation collectors that have been more broadly developed for a variety of scientific objectives:

Over the past 60 years, the precipitation chemistry community has made advancements in deposition collectors to better understand atmospheric processes (Sikсна, 1959). While bulk deposition collection (i.e., a collection bucket or jug fitted with a funnel open at all times; Hall, 1985) is both a simple and economically feasible sampling method utilized by monitoring networks, it is subject to bias through collection of inputs other than atmospheric

deposition (e.g., bird droppings, insects, plant debris). As a result, bulk collectors can overestimate total deposition and underestimate wet deposition in a variety of locations (Lindberg et al., 1986; Richter and Lindberg, 1988; Stedman et al., 1990). **Sequential precipitation collection methods include manually segmenting samplers (requiring only a shelter, clean surface, and an operator), linked collection vessels (sample containers that are filled in sequence via gravitational flow), amongst others and have been developed to analyze rainwater composition and measure parameters such as pH and conductivity (Gatz et al., 1971; Reddy et al., 1985; Vermette and Drake, 1987; Laquer, 1990). Sequential sampler designs have also been adapted to collect precipitation in remote field sites (Germer et al., 2007; Sanei et al., 2010).** Although it is a more costly and time intensive method when compared to bulk deposition collection, the major appeal of measuring isolated wet deposition is **the ability to isolate this individual atmospheric process.**

2. We have briefly mentioned three references (Laurent et al., 2015; Brahney et al., 2020; Audoux et al., 2023), in addition to Germer et al., 2007 in Section 3.1 (General Design Advantages) rather than in the Introduction. The point of precipitation samplers being designed to address specific scientific objectives is a segue into the existing need for innovation geared towards more versatile sample collection:

While several precipitation collectors have been similarly developed to address specific scientific objectives – e.g., the quantitation of dust in wet and dry deposition (Laurent et al., 2015; Brahney et al., 2020), determination of ions and DOC in a tropical rainforest (Germer et al., 2007) and urban environments (Audoux et al., 2023), here we present a more general design for modular adaptability. When compared to other precipitation collection apparatuses, the automated precipitation sampler developed in this work has several advantages. Most notable is the ability to collect integrated samples at remote locations by exploiting its off-grid capabilities. ... In rainfall events where extended atmospheric wash-out occurs, where precipitation becomes ultrapure water, the sampler lids will eventually close – preventing dilution of the sample while maintaining the collection of analytes of interest. **A recent study has found that rainfall events could exhibit variability and the lower atmosphere can be supplied with aerosols due to specific sources, atmospheric dynamics, and meteorological conditions (Audoux et al., 2023). If this occurs, the automated lid will reopen to sample the polluted air masses.** In application to trace pollutants, this also reduces methodological sample preparation

time as it decreases the extent to which additional handling steps, like solid-phase extraction, are required prior to analytical determinations.

2.1.2 Heated Precipitation Sensor

Could you add information about time and delay between resistivity sensor activation and opening of the tip?

We thank the Reviewer for this suggestion, and we understand the need to include this information within the manuscript. As the Reviewer has mentioned this point in prior comments, it has been already addressed with a technical addition to the paper (see Section 2.1.2, “Heated Precipitation Sensor”).

Results and Discussion

3.0 Results and Discussion

The paragraph begins with a summary of the strategies used for validation. Wouldn't it be better to place these points at the beginning of each section to avoid redundancy?

We thank the Reviewer for their comment. Due to the detailed nature of Section 3.0, we thought including this summary would effectively preface each of the following subsections. Additionally, with these important details summarized at the beginning of the Results and Discussion section, this benefits those readers who would choose only to skim through the paper to points of particular interest to them. As such, we have opted to retain this section.

3.1 General Design Advantage

L518-543: This entire section mentions the advantages of the system without any results validating these claims. For example, regarding sample preservation, have you conducted tests on the preservation of reactive, volatile, or biologically transformed species (L538)? Similarly, you argue the possibility of detecting ultra-traces through the resistivity sensor systems (collection even at low conductivity and avoidance of dilution) and the types of materials used. I have no doubt that low-conductivity rains are collected, but what about the quality of the samples in the container? Could you provide information on field blanks, for example?

I believe that if you have a 'Validation' section, this part should be supported by your data.

We thank the Reviewer for their comment. This section was addressed in response to previously made Reviewer comments via a technical addition to the paper (see Section 2.3.1., Sample Preservation).

With respect to field blanks, we have an existing statement on how these were prepared in Section 2.4 ‘Cleaning and Preparation of Sample Containers’ and have added a minor clarification that

these were used to perform blank subtractions on measurements of all relevant metrics (e.g., conductivity and DOC here):

Field and method blanks were collected through the addition of DIW to cleaned containers, and/or sample handling devices, in order to quantify appropriate method detection limits and to identify any sources of systematic or random contamination. **Blank subtraction was applied to measurements, where appropriate.**

An example of an appropriate mention of how field blanks were used for considering conductivity (Section 3.4.2, first paragraph):

The conductivity of saturated HgCl_2 in water (at 0.1% vol/vol) was $13.6 \pm 0.4 \mu\text{S/cm}$, which is also comparable to **but statistically higher than** our field blanks ($p = 0.0015$; unpaired t test) and less than what was observed for our samples ($p < 2 \times 10^{-6}$ for each site considered separately and also across all sites; unpaired t-test).

3.3. Comparison of Sample Collection Volumes

L647: It is announced that the automatic collectors and total deposition collectors are colocated, but if I understand correctly, the values discussed in this paragraph only concern the volumes obtained with the total deposition collectors in OF?

The Reviewer understands correctly that this paragraph deals with the fraction of total deposition collected in the automated samplers. This paragraph demonstrates the extent to which atmospheric washout can potentially dilute bulk samples, making the determination of trace analytes challenging and often then subject to complex and error-prone sample concentration methods. We refer the Reviewer to the existing introductory text in the Methods (Section 2.3) describing the collocation of the samplers:

One array of six automated collection units (3 OF, 3 TF) were deployed within one forested experimental field site located in each of the four watershed regions of the NL-BELT (24 samplers in total) between 2015 and 2016. Additionally, between one to three total deposition samplers were located at each of the four field sites (Table 1, Figure S10).

In Section 3.3 itself we have clarified the comparators in the second paragraph, alongside expectations:

The wet deposition volumes collected for the snow free period using the automated precipitation samplers did not follow the trends in total deposition (Figure 4), as might be expected (**e.g., due to pollutant loading, rainfall quantity/rate, and scavenging processes**). For the 2015 collection period from June through October, the summed volumes of OF precipitation, from south to north across the NL-BELT, were 25.4, 10.9, 20.4, and 2.2 L, while in 2016 they were 17.3, 30.4, 13.5, and 5.1 L. **While the total and OF fractions would typically be much closer to unity in more polluted regions, it would be expected in these remote NL-BELT field sites for the differences to be driven by complex, non-linear processes that cannot be easily disentangled.** Here we present three reasons as to why the measured wet OF deposition volumes do not follow the total deposition trend across the transect.

L653 - 705: I don't understand your strategy. Why validate total deposition and then wet deposition, since the idea is to validate rain collectors? I question the need for an entire paragraph on the comparison of total deposition in the text with ECCC and DAYMET data. Comparing with model data at a km² scale is interesting to show the representativity of measured fluxes at a single point on a larger scale (e.g., watershed scale), but it does not demonstrate that your collectors effectively capture the totality of precipitation. This study does not seem relevant to the paper's topic. I believe it could be included as supplementary material.

We thank the Reviewer for their comment. We agree with the suggestion that this comparison (L673-701 in the original manuscript) be removed and we have relocated it to the Supporting Information. This significantly simplifies and focuses this section and we thank the Reviewer for encouraging us to do so with this comment.

The text in Section 3.3 of the manuscript now simply refers to this material, presenting only the most important results of the comparison:

The automated samplers were collocated with total deposition samplers and deployed across the experimental forests of four NL-BELT regions during the 2015 and 2016 growing seasons to observe deposition trends. In addition, we briefly compare these observations to the long-term climate normals reported by ECCC and estimated deposition at 1 km x 1 km resolution from the DAYMET reanalysis model (Table 1, **Section S2**). Three automated samplers were deployed in the open to collect incident precipitation (OF) and another three under the experimental forest site canopy (TF). ... The total deposition samplers were installed in HR in late 2014 and the automated samplers were then set up at PB. Despite this, there is good agreement between the trends in predicted deposition values by DAYMET with the measured values, although the absolute amounts

from these are systematically lower in all of our observations (**Section S2**) Regardless, by following the recommended siting criteria from the NADP and CAPMoN as best as possible, the very strong agreement of our temporal trends at both annual and monthly timescales with both comparators demonstrates the suitability of the total deposition samplers and, therefore, the automated samples for use in quantifying deposited chemical species of atmospheric interest into the experimental sites.

In my opinion, the comparison between total deposition and collector values (and their differences) is the best argument to show that the automatic collectors effectively capture all rainfall. The question is what is the representativity of the “conductive” rains collected here compared to the conventional rainfall collection?

We thank the Reviewer for this comment. We assume that the conventional rainfall referred to by the Reviewer is bucket-based bulk deposition, collected on an event basis, rather than integrated sampling. If this is the case, our conductive rain is the fraction containing solutes above the sensor detection threshold stated in the main manuscript (addressed previously, see above).

The representativeness is exactly that defined by the sensor threshold, excluding ultrapure water from precipitation during washout events, which serves to dilute analytes in a sample. In more polluted regions, for example, we would expect the total and open fall fractions to be much closer to unity (see response and manuscript additions made to the comment immediately before this one). At the remote locations of the NL-BELT experimental forests - knowing that the region is subject to intense rainfall - it is not surprising that the collected fractions are smaller than found in the total deposition samplers. The fact that we do not see any comparisons where open fall volume exceeds that of the total also satisfies the basic premise of our comparison.

L733-753: The RSD obtained on the triplicates of the rain collectors of 10 samples out of 33, or 1/3 of the samples, have an RSD greater than 40%. This indicates a very high variability in the collected volumes. It would be interesting to discuss the reasons for this variability, which is crucial to estimate the performance/artifacts of your device (it's unfortunate that the data are in supplementary material). No analysis is done on these high RSD values. Is it for the same period of the year, e.g., when the winds are strongest? Is it at a specific site? Did the replicates use the same sensor, or is each system independent of the other? Could the observed variability in the replicates be due to different closing or opening times? Has simultaneity been tested?

We thank the Reviewer for their comment, and we agree that the high RSD values should be addressed better. To do so, we will answer your questions posed individually as follows:

- i. Is it for the same period of the year, e.g., when the winds are strongest?

Winds are highly variable throughout the year. Storms can be accompanied by winds up to 140 km/hr in any month. Siting of samplers was conducted according to standard requirements of government sampling statements, so subject to the same potential artifacts. Government agencies

tend to only collect a single sample, so potential uncertainty in volume and analyte deposited quantities likely goes unreported (see addition to Section 3.3. below).

ii. Is it at a specific site?

No; it occurs at many of the sites and varies over the sampling periods.

iii. Did the replicates use the same sensor, or is each system independent of the other? Could the observed variability in the replicates be due to different closing or opening times? Has simultaneity been tested?

As described in Sections 2.1.2 and 2.3, each array of 6 samplers (3 OF, 3 TF) are controlled by the same sensor and the sampler lids all open within 5 seconds of detected conductive rainfall. Thus, after several years of qualitative observations, the observed variability is not a result of different sampler lid opening and closing times (i.e., a lack of simultaneity).

The following modifications have been made to Section 3.3 in the manuscript to communicate the information we have provided above for questions (i) and (ii). In addition to the new text underlined and bolded, some sentences from the original manuscript were reordered in this modified paragraph for better flow:

The automated OF wet deposition volumes collected each year have peak values that range from 1 to 4 L with an overall variability of $\pm 33\%$ for any triplicate of samples across the entire dataset. Across our 33 sample collection periods, our replicate relative standard deviations (RSDs) follow a log-normal distribution where volume reproducibility is typically within $\pm 12.5\%$ and almost always within $\pm 31.5\%$ (Figure S11). A few outliers with higher variability skew the overall view of volume precision. Out of 33 OF samples collected, **10** have RSDs greater than 40.5% with 2 of those **10** having RSDs greater than 100%. **Those values greater than 40.5% had no systematic relationship with site or time of year. Wind speeds were considered as a possible source of variability. The prevailing winds over Atlantic Canada are known to be southwesterly in the summer – intensifying during the autumn months – and westerly to northwesterly in the winter (Bowyer, 1995; Jacob, 1999; Randall, 2015). Strong wind speeds (i.e., >100 km/hr) could occur on an event basis during any time of the year and, thus, could contribute to the variability seen at each field site.** Wind is known to generate bias in gauge-based precipitation measurements where unshielded precipitation gauges can catch less than half of the amount of a shielded gauge (Colli et al., 2016). A windscreen design for obtaining rainfall rates – and thus, volumes – to be more reproducible could be considered in future deployments of our developed

samplers, similar to recently reported innovations for smaller rainfall rate devices (Kochendorfer et al., 2023). This would, however, increase costs and logistical considerations in deploying the developed devices which currently operate synonymously to deposition systems employed by government monitoring programs. **Our siting approach is consistent with these, which often deploy a single sampler without wind protection. Thus, by employing replicates, we are able to better ascertain the environmental variability.** In addition, collection of replicate samples allows our observations to span a wider physical area, reducing the impact of confounding variables such as wind speed in comparison to a more typical sample size of one for many field collections. Imperfect siting and lack of shielding is necessary where remote field sampling prevents the setup of such infrastructure. As a result, the deployment of triplicate samplers provides researchers with a better opportunity to implement quality control as they can reduce bias in the event of dynamic OF. While the effect of wind is reduced, additional factors can drive variability when the samplers are placed under a forest canopy for TF collection.

The comparison between total volumes and precipitation volumes shows a clear difference. The explanations given for these differences are not quantified, while they could provide information on the quality of the collection and, therefore, on the definition of conductive rains (and provide information on the dilution parameter). For example, is there a link between the total collected volume and the difference between total/conductive deposition? Is it related to the site? This type of analysis could be done with scatterplots between total volume and rain, with different colors for each site to highlight if biases are site-related, then with colors by season.

Similar to our response to a comment above, the difference between open fall and total volume is not simple (e.g., pollutant loading, rainfall quantity/rate, scavenging processes). It is generally true that in more polluted regions, we would expect the total and open fall fractions to be much closer to unity. However, at the remote locations of the NL-BELT experimental forests, we expect the differences (as shown in Figure 4) to be driven by complex, non-linear processes that cannot be easily disentangled.

To better communicate this point, we have made the following modifications in Section 3.3:

The wet deposition volumes collected for the snow free period using the automated precipitation samplers did not follow the trends in total deposition (Figure 4), as might be expected **(e.g., due to pollutant loading, rainfall quantity/rate, and scavenging processes)**. For the 2015 collection period from June through October, the summed volumes of OF precipitation, from south to north across the NL-BELT, were 25.4, 10.9, 20.4, and 2.2 L, while in 2016 they were 17.3, 30.4, 13.5, and 5.1 L. **While the total and OF fractions would typically be much closer to unity in more**

polluted regions, it would be expected in these remote NL-BELT field sites for the differences to be driven by complex, non-linear processes that cannot be easily disentangled. Here we present three reasons as to why the measured wet deposition volumes do not follow the total deposition trend across the transect.

Could you discuss these differences considering technical aspects? For example, the closing/opening of the collector lids is conditioned by the conductivity of the rainwater, assuming that concentrations in the rains decrease as the rain progresses. However, it has already been demonstrated that this is not necessarily the case (e.g. Audoux et al., 2023 see before). There can be refeeding of the lower layers with aerosols or mixtures of air masses that induce increases in conductivity during rain. The question is, what will be the behavior of the collector in this case? Is there a risk that the lid closes and reopens, or not? Are these phenomena that you observe from the datalogging?

We thank the Reviewer for this question. We acknowledge that the lower layers of the atmosphere can be supplied with aerosols, or that the arrival of contaminated air masses can increase the conductivity during a washout event. In that case, the lid of our samplers would re-open and sample the conductive precipitation, as the sensor is always active and the discriminating factor between sample collection (or not). We have witnessed this phenomenon firsthand in the field during deployment. To better address this topic, we have modified Section 3.1 to include the following underlined text:

In rainfall events where extended atmospheric wash-out occurs, where precipitation becomes ultrapure water, the sampler lids will eventually close – preventing dilution of the sample while maintaining the collection of analytes of interest. A recent study has found that rainfall events could exhibit variability and the lower atmosphere can be supplied with aerosols due to specific sources, atmospheric dynamics, and meteorological conditions (Audoux et al., 2023). If this occurs, the automated lid will reopen to sample the polluted air masses. In application to trace pollutants, this also reduces methodological sample preparation time as it decreases the extent to which additional handling steps, like solid-phase extraction, are required prior to analytical determinations.

It is known that the conductivity of a solution depends on temperature, and here the temperatures can vary greatly between seasons. The sensor is one of the key elements in the autonomy of your system. Have you tested how temperature affects the sensor's response? Could this have an influence on the differences observed between sites or with total depositions?

We thank the Reviewer for highlighting this point. The sensor surface is heated during detected precipitation events, which would mitigate the majority of ambient temperature effects. The operational temperature range provided by the manufacturer, based on a personal communication

with their support team, is -10 °C to +55 °C. Based on many years of qualitative observations, we have not noticed seasonal temperatures influencing the response of the sensor and know that similar considerations are not made by commercial systems on the market, or by government agency samplers. Lastly, the automated deposition samplers were decommissioned during the winter at the NL-BELT so we felt that this information was not relevant to include in the main text. It is noteworthy to mention that we have deployed these automated samplers year-round, since this initial study, in temperatures below -10 °C in Toronto and they continue to detect snow and ice deposition without issue. To do so, we have heated our chute to melt any deposited snow or ice, which again prevents any temperature-dependent conductivity detection issues. This will be featured in future studies reporting on winter deposition of pollutants of interest.

In addition, we have included the operating temperature range provided by the manufacturer of (-10°C to +55°C) in Table S2 within the Supporting Information.

L718: The authors rely on acquired data that are not collected simultaneously with their samples and justify the discrepancies due to the heterogeneity of precipitation. However, this heterogeneity is known. Why choose to use ECCC measurement sites that are not collocated with NL-BELT sites to do this validation work for the collectors since it is certain that there will be a discrepancy between the two?

We thank the Reviewer for highlighting this. We agree with their assessment and, as described above, have moved this comparison to the Supporting Information.

L733-776: This part pertains to the application of the collectors and no longer their validation; I think it should appear in a different section.

We thank the Reviewer for their comment. As there is now a separate “Sample Preservation” subsection (2.3.1), we believe that this particular analysis can remain where it is within the main text.

3.4. Characterizing Chemical Parameters from NL-BELT

I think the validation part of the chemical measurement should be in a separate section and thoroughly discuss:

Could you present sample preservation tests for pH, conductivity, and DOC? How can you ensure that concentrations are maintained over time despite all precautions taken? For example, have you taken a sample after rain and observed it after a month outdoors, considering temperature and sunlight variations (the choice of black color may lead to high temperatures inside the units during summer)? Could this have an impact on chemical parameters, such as the desorption of organic species from the surface of particles in the rain? The system is designed to limit evaporation, but is this really the case? Have you tested volumes pre- and post-sampling?

The Reviewer mentioned this point in a prior comment, which has already been addressed with a technical addition to the paper (Section 2.3.1, Sample Preservation). This subsection addresses our

use of a well-established sample preservation technique and hence, why additional tests were not done to verify the preservation of collected chemical species over time. We believe that this new subsection addresses all further comments made by either Reviewer regarding system validation, as do the references we have provided, should they have further interest in this topic.

L995: This paragraph should be positioned earlier (in the validation section) to show the agreement between the values measured here with this new system and the values expected from the literature.

We respectfully disagree with the Reviewer in this case. After the thorough restructuring of the manuscript, we believe this paragraph is most effective in its current position.

A significant portion of these (Section 3.4) results is applicable and pertains to the study of canopy effects on deposition fluxes and should be placed in a separate section and discussed accordingly.

We thank the Reviewer for this suggestion. For brevity, as previously mentioned, we have opted not to separate the discussion of canopy effects with respect to the observed OF and TF deposition. The measurements are meant to demonstrate the extended application of these samplers, not to study canopy-precipitation interactions in detail. This is the subject of a future manuscript and beyond the scope of this work, as we have noted in the revised Conclusions and Future Directions (Section 4.0) in a prior response above.

Reviewer 2

Overview:

The paper outlines the design of an automatic precipitation sampler for off-grid use. It is suitable for measuring pH, conductivity, and dissolved organic carbon. The new sampler was tested in the Newfoundland and Labrador Boreal Ecosystem Latitudinal Transect over a two-year period for open-fall and throughfall precipitation. A notable disadvantage to this sampler is the inability to collect snowfall, limiting year-round monitoring of precipitation.

We understand the Reviewer's concern regarding the inability to sample snowfall. We share the same opinion. In fact, since the first iteration, we have modified the funnels and rain sensor chute by installing heating cables to detect, melt, and therefore sample snowfall. This requires a constant power demand in excess of 4 A and, therefore, access to grid power. Since this alteration of the system is still in the testing/development phase, and is not applicable to off-grid sampling, we have decided to exclude the modifications from the current manuscript. This is a substantial engineering challenge and would also undermine our desire to communicate on how to obtain a widely accessible automated deposition sampling method (i.e., cheap).

General comments

This work is of great interest to atmospheric scientists and is within the scope of the journal. Overall, this work is appropriate for acceptance in AMT following the revisions outlined below.

The revisions mostly focus on reformatting, reducing lengthy descriptions, and improving clarity (particularly in the introduction and results/discussion sections).

We thank the Reviewer for the positive feedback and appreciate the time they have taken to provide these comments. Please see below where we address these points in our responses and manuscript alterations. Where appropriate, to retain concision, we refer the Reviewer to prior responses above where overlapping concerns with the other Reviewer occur.

Specific comments

Introduction:

Overall, the introduction is very lengthy. Finding ways to pare down this material would allow for clarity of the important topics related to this work.

We agree with the Reviewer. It is challenging to speak to a broad audience while also retaining a focused Introduction. We appreciate the thoughtfulness of the Reviewer in their suggestions that follow and have done our best to reduce the content of the Introduction. The other Reviewer made a similar suggestion and we have managed to reduce the word count of the Introduction from 2,261 words to 1,783 words in the revision. We hope that this is satisfactory.

Line 74-91: It seems like this paragraph is trying to introduce pH, conductivity, and DOC, however it is hard to separate the information of these three topics from each other. The discussion of pH, conductivity, and DOC are mixed together, making it hard for the reader to parse out the relevant information for each topic. This paragraph needs to be revised and simplified.

We agree with the Reviewer. Given the established measurements of pH and conductivity, we feel that removing the majority of this paragraph to better focus on the deposition collectors themselves would address this comment accordingly as well as the other comments pertaining to the dense and general nature of the Introduction.

Line 119-132: This paragraph discusses persistent organic pollutants (POPs), which the authors do not monitor during the testing period presented in this work. This paragraph could be removed from the introduction and incorporated later as a future use for the sampler.

Line 152-165: This paragraph discusses monitoring reactive nitrogen in atmospheric deposition, however this was not a focus of the precipitation characteristics (pH, conductivity, and DOC) that was highlighted in the results. If this does in fact connect with those three characteristics, those connections need to be made clearer, or this paragraph can be removed.

Response to comments pertaining to L119-132 and L152-165:

We thank the Reviewer for these comments and Reviewer 1 raised similar concerns. The Introduction has been refocused and the mention/discussion of studying the deposition of various chemical species (POPs, ON, etc.) not specifically studied in this work has been either removed

completely or integrated into a succinct addition in the revised Conclusions and Future Directions (Section 4.0).

Methods

2.1.2 Heated Precipitation Sensor

Line 305-306: Is the conductivity that triggers the sensor typically for that of precipitation (both in your geographic region and others)? I wonder about the variability of precipitation globally and if this would cause differences in sampling. In addition, what is the time delay for the lid opening once triggered?

We thank the Reviewer for this comment and Reviewer 1 raised similar concerns. While we acknowledge that conductivity in precipitation could vary depending on sampling locations, the threshold that we report represents the lower limit of the range and this excludes our modification to its design to increase the sensitivity of the sensor by adding the chute. We do not expect precipitation conductance to fall below our threshold unless there is a washout event in the atmosphere that effectively consists of ultrapure water. We note that established sensors on deposition samplers used by government monitoring programs that rely on conductivity would be similarly impacted. As such, we have retained our original lower limit conductivity definition and added the equivalent concentration in sodium chloride. We think this clarifies the operation of the samplers, their detection limits, and hope the Reviewer agrees.

We have added the following to Section 2.1.2 Heated Precipitation Sensor:

The detection of rain modulates the opening and closing of the collection units by an interdigitated resistive sensor (M152; Kemo Electronic GmbH, Geestland, Germany; Figures S6 to S8). **This approach is consistent with established precipitation detection techniques used by government monitoring programs (e.g., CAPMoN; Canadian Air and Precipitation Monitoring Network, 1985a, 1985b).** The rain sensor detects conductive deposition by the completion of a conductive circuit when electrolytes bridge the connection between the interdigitated gold electrodes. The sensor is supplied with 12 VDC from the power system to trigger a relay when precipitation conductance above 1 M Ω ·cm conductivity is detected (determined experimentally, see Section S1). **This is equivalent to approximately 8 μ M sodium chloride. The sensor detection limit reflects an upper limit of precipitation ion loading because the design of the rain chute leads to an increase in surface area of more than a factor of 25 on which solutes can accumulate to enhance the ionic content of the deposited water.**

An output of 12 VDC is sent to the digital control board by the relay when rain is sensed, or 0 VDC in its absence, for signal processing and motor control (Figure S7).

Additionally, Reviewer 1 raised the point regarding the opening and closing times of the lid. We thank the Reviewer for this comment, and we understand the need to communicate that the samplers open rapidly when rain is sensed, for the facts they mention here. The opening of the lid is fast (<5 seconds) and is dependent on the rotation rate of the motor selected. We typically use 2 to 6 rpm motors, depending on their availability from our suppliers. We hope the inclusion of this information in our revision is satisfactory.

An output of 12 VDC is sent to the digital control board by the relay when rain is sensed, or 0 VDC in its absence, for signal processing and motor control (Figure S7). **When rain is sensed, the lid of each sampler in the array is simultaneously opened (<5 seconds) and is dependent on the rotational rate of the lid motor.** To increase the sensitivity of this sensor and to extend the sampling duration when conductive atmospheric constituents are completely washed out of the atmosphere

Results and Discussion

3.1. General design advantage

Overall, I think the results and discussion could be reformatted. It was confusing to read and keep track of the sampler's validation results and how you were actually applying the sampler to gain new information (like TF and OF comparisons). I appreciate the "General Design Advantages" section to highlight the ability to use this sampler in remote locations and to collect replicates.

We thank the Reviewer for recognizing our rationale for including the "General Design Advantage" section in the manuscript. This is contrary to the comments made by Reviewer 1, as they felt that this section resembles a promotional brochure or technical document. Given the conflicting opinions of the Reviewers, we have elected to retain our preference of having this section in the manuscript.

Below, we have attempted to address the Reviewer's detailed comments on improving the Results and Discussion section of the manuscript. We hope they find these to be acceptable.

3.3. Comparison of Sample Collection Volumes

Line 681: If there is a lot of data missing from the ECCC monitoring site, why would you choose this dataset for your comparisons?

We thank the Reviewer for their comment. The Authors initially believed that although a lot of data was missing from the monitoring site in question, it was still better to include any available data for collocated sampling sites. Upon reflection, we now agree that this is not a suitable comparison to include within the main text but is instructive information to include in the

Supporting Information. In also addressing comments made by Reviewer 1, suggesting that our comparison of total deposition with ECCC and DAYMET data is tangential to the central topics of this paper, Section 3.3 was greatly simplified by moving the ECCC/DAYMET comparison to the Supporting Information (Section S2). We agree with both Reviewers that while this was instructive for our biogeochemistry and hydrology work at the NL-BELT, that it is not instructive for the wider scientific community.

Line 689-690: The R² values are presented in a confusing manner, especially the ones found in parentheses. Please make it more clear which value corresponds to which site.

We thank the Reviewer for their comment. As mentioned above, to address comments made by both Reviewers, Section 3.3. was greatly simplified by moving the ECCC/DAYMET comparison to the Supporting Information. Regarding the R² values, the way in which they are presented in the Supporting Information has been revised for simplicity as follows:

S2. Deposition Comparison: This Work, DAYMET, and ECCC

The DAYMET observations are representative of a larger spatial scale, where our discrete samplers could be subject to heterogeneity in deposition (e.g., orographic precipitation, driven by topography like steep slopes) or impacted by meteorological conditions not captured by the model (e.g., undercatch driven by local winds). The temporally resolved volume comparisons at sampling interval timescales better-demonstrates comparability, despite the systematic differences. **The month-to-month relationships between DAYMET and our observations, as well as between ECCC and our observations, all show strong correlations with linear regressions having R² values, from north to south, of 0.72, 0.99, 0.99, and 0.86, and N/A, 0.94, N/A, and 0.93, respectively (Table S3).** The discrepancy between DAYMET, ECCC, and our observations for total deposition were highest in the most northerly site, where the experimental site was located on a steep slope, with only 43 % of the predicted volume collected. At all of the sample collection sites on the island of Newfoundland, a consistent difference was observed with 65 ± 4 % of the estimated volume collected, except at GC where our measurements and those from ECCC are identical and starkly contrasting to DAYMET.

Caption of Table S3:

Table S3. Collected precipitation volumes from NL-BELT in bulk deposition samplers for rainwater were deployed for one to two months, while snow was collected as an integrated sample throughout the winter because sites were not accessible. The Environment and Climate Change Canada (ECCC) precipitation data was obtained for identical sampling intervals. The DAYMET model for North America (1 km x 1 km resolution) for precipitation was obtained for the identical sampling intervals, which utilizes interpolated

and extrapolated data from daily weather observations to predict inputs at the NL-BELT locations. Linear regression results for slope (m) and correlation coefficient (R^2) between observations and DAYMET (*italics*), and observations and ECCC (where possible; underlined), were calculated. For sampling periods where a measurement was compromised or not collected for a given interval in this work, these are reported as with '-' and an estimated volume from the regression relationship with ECCC is reported in parentheses when used to replace compromised samples.

Overall, Section 3.3 Comparison of Sample Collection Volumes is very long and feels repetitive. This section should be made more concise.

We thank the Reviewer for their comment and appreciate their suggestions. We have moved the ECCC/DAYMET comparison to the Supporting Information, resulting in a more succinct Section 3.3 (1,893 words now reduced to 1,675).

3.4. Characterizing Chemical Parameters from NL-BELT

Line 815-819: Mentioning and comparing the pH of the TF samples to the pH of the soil in that area seems like extraneous information. Why are you making this comparison? If it's truly needed, provide some justification or explanation as to why this is important.

We thank the Reviewer for their comment, and we agree that this is extraneous information to include here, as was the comparison in the prior section. Again, while there is value in this for our ongoing work at the NL-BELT, we agree that it was a misplaced idea on our part to include it in the main text of manuscript.

We have removed this material entirely from Section 3.4.1 and incorporated it into Table 1 as "footnote a" for contextual purposes:

^aSoil pH for the organic and mineral soil horizons determined by addition of 400 μL of aqueous 0.5 M CaCl_2 to a 50:50 w/w slurry of dried soil in deionised water. Note: the four remote NL-BELT sites are dominated by balsam fir trees underlain by humo-ferric podzol soil with pH ranging between 3.0 and 4.5.

Section 3.4.2 – Most other sections have a comparison of OF and TF samples, but this one does not. Was that intentional? A brief statement comparing these two would be great.

We thank the Reviewer for their comment and this brief statement has been included within Section 3.4.2:

In all the collected OF and TF precipitation samples, across all four NL-BELT sites, the average measured conductivity values ranged from 21 to 166 $\mu\text{S}/\text{cm}$ following no apparent seasonal or temporal trend (Figure 6). Additionally, the conductivity in both OF and TF also appear to vary across the field sites - only within the 2016 TF samples does the conductivity

appear to increase with decreasing latitude. Yet, with the typical conductivity of surface and drinking waters being between 1 to 1000 $\mu\text{S}/\text{cm}$ (Lin et al., 2017), and typically below 200 $\mu\text{S}/\text{cm}$ in stream water measurements within the watersheds of each of the NL-BELT sites, our observations are comparable and fall within the expected range. Our field blanks – encompassing a variety of materials and apparatuses, and our cleaning procedures, routinely produced conductivities of $9 \pm 5 \mu\text{S}/\text{cm}$.

Line 893-894: p-values to support your statements that the conductivity of HgCl_2 in water is comparable to field blanks and less than your samples would strengthen this argument.

The Reviewer is making a request that may not be as informative as they wish it to be, but we have decided to incorporate the results of a statistical comparison in line with their request. Our measurements of HgCl_2 in water at the saturation limit is an upper boundary. This is the measured value of $13.6 \pm 0.4 \mu\text{S}/\text{cm}$ is expected to be much higher than its impact on any sample where the small volume of such solutions (1 to 5 mL) is diluted by the total volume of collected precipitation. Thus, the comparison and resulting statistics are expected to be skewed towards similarity as they do not account for this physical reality.

In addition, our field blanks show less conductive contamination compared to this upper limit of saturated HgCl_2 statistically, and the samples are significantly higher ($p = 7.65 \times 10^{-8}$; unpaired t-test). Therefore, the results hold up in line with expectations of high-quality analytical performance, despite the above stated caveat. We have added brief statistical outcomes from unpaired t tests to the main manuscript:

Our field blanks – encompassing a variety of materials and apparatuses, and our cleaning procedures, routinely produced conductivities of $9 \pm 5 \mu\text{S}/\text{cm}$. The conductivity of saturated HgCl_2 in water (at 0.1% vol/vol) was $13.6 \pm 0.4 \mu\text{S}/\text{cm}$, which is also comparable to **but statistically higher than** our field blanks ($p = 0.0015$; unpaired t test) and less than what was observed for our samples ($p < 2 \times 10^{-6}$ for each site considered separately and also across all sites; unpaired t-test).

Line 958-963: You mention wildfire plumes being a potential cause of increased DOC levels in precipitation. Was there a wildfire event near your sites during your sampling period to warrant mentioning this possibility? Or would the possibility of increased DOC still be observed some amount of time after a wildfire event? In order to justify including this argument, I would like to see if you could make a potential correlation to an actual event that may explain this.

We thank the Reviewer for highlighting the very important link between wildfire plumes and increased DOC deposition. The province of Newfoundland and Labrador does experience wildfires, forest fire season is in effect from May to September each year. Based on 20 years of

wildfire data, there are on average 118 wildfires burning 22,993 hectares in Newfoundland and Labrador each year (Government of Newfoundland and Labrador, 2023). We strongly believe this argument is warranted since (i) there has been an increase in wildfire activity across North America and (ii) several studies have now linked increased organic and inorganic carbon deposition to wildfire events. (Wagner et al., 2021; Coward et al., 2022; Barton et al., 2024). Given the variability in atmospheric transport from these fire locations (typically in Labrador, but also reaching Quebec and further west), it is challenging to provide deep insight regarding the magnitude of impact in monthly integrated samples without detailed back trajectory analysis driven by known precipitation intercepting these airmasses. So, although we did not make a quantitative determination of wildfire plumes during the campaign, it would be unusual not to include wildfires as a possible reason for increased DOC deposition.

The discussion on DOC seems less than a validation of instrument performance, but rather a capability of the sampler. Unless you have DOC data from your sampling locations to compare to, this seems like new information for remote sampling sites.

We thank the Reviewer for this comment. We may have miscommunicated our approach and would now like to clarify it. In Section 3.4, we aimed to highlight the sampler's capability to accurately quantify precipitation pH, conductivity, and DOC fluxes, with a special interest in demonstrating the capacity to discern and investigate canopy effects. To validate our new DOC measurements, we compared our observed fluxes to other studies in forested regions. We strongly feel that this section is a validation of the instrument performance and re-emphasize that future applications for these samplers are stated in the Conclusions and Future Directions (Section 4.0), as further scientific inquiry is beyond the scope of this work.

To add clarity to the manuscript, we have added the following text in Section 3.4.3 (paragraph following Figure 7):

Additionally, we cannot rule out that the chemical speciation differs between OF, TF, and SF even if the DOC values are similar, but such insights require more selective instrumentation for chemical analysis (e.g., high resolution mass spectrometry).

The ability to accurately determine DOC in OF and TF precipitation demonstrates the capability of the automated deposition samplers. To validate our measurements, we compared our observed fluxes to other studies in different forest types. Mean annual DOC fluxes were generally similar to those reported in some other boreal forests (Table 3).

Line 1014: what evidence do you have that supports the same that the automated system better maintains the integrity of DOC in samples?

We thank the Reviewer for highlighting this and we welcome the opportunity to clarify our approach. Our rationale in stating that the automated system maintains the integrity of DOC in the collected samples is because we have followed the standard approach used by biogeochemists to study DOC in soil and freshwater samples, by microbially fixing them through the addition of

HgCl₂. In addition, we have outlined apparatus cleaning procedures to minimize contamination. Finally, the sampler design also prevents the intrusion of forest litter and other materials that could potentially influence the levels of DOC in forested regions if these were to be submerged in the collected aqueous precipitation sample, leading to bias due to leaching of organic compounds from these solid organic materials. This is particularly important if the objective is to selectively quantify DOC in wet deposition.

To better address this comment, we have made a small addition to the final paragraph of Section 3.4.3 which now reads:

The automated system better maintains the integrity of DOC in the samples. **This was achieved by following standard procedures for biogeochemical sample preservation (i.e., adding HgCl₂) (Argentino et al, 2023), employing a rigorous cleaning procedure, and preventative design against the intrusion of forest litter** which could result in a positive bias for DOC in the collected precipitation.

Additionally, we've added a new subsection "Sample Perseveration" (2.3.1) to address comments regarding topics such as sample preservation and our approach to method validation.

Technical Corrections Starting with section 3.4.3 – it is mentioned that flux (with the units: mg C m⁻² d⁻¹) is used. However, at several points in this discussion flux is given as mg C m⁻² a⁻¹. I'm providing a list locations I've found this mistake, but please check throughout the manuscript for places I may have missed:

Lines 50, 921, 925, 944, 955, 966, 997, 998, 100

We thank the Reviewer for this comment; however, we think there was some miscommunication with our approach and we would like the opportunity to clarify. The deposition fluxes were calculated daily; however, this was summed for each sampling period and reported as an equivalent annual flux with units of (mg m⁻² a⁻¹). We have checked the values and units throughout the manuscript and Supporting Information to confirm that they are correct in all locations.

To clarify, the text at the end of the first paragraph of Section 3.4.3 now reads:

The concentrations were converted to elemental fluxes using the volume of precipitation collected, the area of the funnel and the number of sampling days of each sampling period (Figure 7). The total flux for each sample period was summed **and reported as an equivalent annual flux with the following units: mg m⁻² a⁻¹. Annual fluxes** ranged from 600 to 4200 mg C m⁻² a⁻¹ across the study sites for the snow free period (Table S4).

References

Argentino, C., Kalenitchenko, D., Lindgren, M. and Panieri, G.: HgCl₂ addition to pore water samples from cold seeps can affect the geochemistry of dissolved inorganic carbon ([DIC], δ¹³CDIC). *Marine Chemistry*, 251, p.104236, 2023.

Audoux, T., Laurent, B., Desboeufs, K., Noyalet, G., Maisonneuve, F., Lauret, O., and Chevaillier, S.: Intra-event evolution of elemental and ionic concentrations in wet deposition in an urban environment, *Atmospheric Chem. Phys.*, 23, 13485–13503, <https://doi.org/10.5194/acp-23-13485-2023>, 2023.

Barton, R., Richardson, C.M., Pae, E., Montalvo, M.S., Redmond, M., Zimmer, M.A. and Wagner, S.: Hydrology, rather than wildfire burn extent, determines post-fire organic and black carbon export from mountain rivers in central coastal California. *Limnology and Oceanography Letters*, 2024.

Bowering, K., Edwards, K., Prestegard, K., Zhu, X. and S.E. Ziegler.: Dissolved organic carbon mobilized from organic horizons of mature and harvested black spruce plots in a mesic boreal region. *Biogeosciences* 17:581-595. <https://doi.org/10.5194/bg-17-581-2020>, 2020.

Bowyer, Peter J. (Ed.): *Where the Wind Blows: A Guide to Marine Weather in Atlantic Canada*, Breakwater Books Ltd., St John's, Newfoundland, 53 pp., 1995.

Brahney, J., Wetherbee, G., Sexstone, G. A., Youngbull, C., Strong, P., and Heindel, R. C.: A new sampler for the collection and retrieval of dry dust deposition, *Aeolian Res.*, 45, 100600, <https://doi.org/10.1016/j.aeolia.2020.100600>, 2020.

Canadian Air and Precipitation Monitoring Network: *Precipitation Sampling Instruments Operation and Maintenance Manual – Operators Edition*. Available at: https://publications.gc.ca/collections/collection_2020/eccc/en56/En56-264-1985-eng.pdf (Accessed: 22 February 2024), 1985b.

Coward, E. K., Seech, K., Carter, M. L., Flick, R. E., and Grassian, V. H.: Of Sea and Smoke: Evidence of Marine Dissolved Organic Matter Deposition from 2020 Western United States Wildfires, *Environ. Sci. Technol. Lett.*, 9, 869–876, <https://doi.org/10.1021/acs.estlett.2c00383>, 2022.

Dossett, S.R. and Bowersox, V.C.: *National Trends Network Site Operation Manual*. Available at: <https://www.arlis.org/docs/vol1/123915417/opman.pdf> (Accessed: 12 March 2024). 1999.

Galloway, J.N. and Likens, G.E.: The collection of precipitation for chemical analysis. *Tellus*, 30(1), pp.71-82, 1978.

Gatz, D. F., Selman, R. F., Langs, R. K., and Holtzman, R. B.: An Automatic Sequential Rain Sampler, *J. Appl. Meteorol.* 1962-1982, 10, 341–344, 1971.

Germer, S., Neill, C., Krusche, A. V., Neto, S. C. G., and Elsenbeer, H.: Seasonal and within-event dynamics of rainfall and throughfall chemistry in an open tropical rainforest in Rondônia, Brazil, *Biogeochemistry*, 86, 155–174, <https://doi.org/10.1007/s10533-007-9152-9>, 2007.

Government of Newfoundland and Labrador: Forest Fire Season, Fisheries, Forestry and Agriculture. Available at: <https://www.gov.nl.ca/ffa/public-education/forestry/forest-fires/> (Accessed: 02 February 2024). 2023

Jacob, Daniel J.: *Introduction to Atmospheric Chemistry*, Princeton University Press, Princeton, New Jersey, 49 pp., 1999.

Kattner, G.: Storage of dissolved inorganic nutrients in seawater: poisoning with mercuric chloride, *Mar. Chem.*, 67, 61–66, [https://doi.org/10.1016/S0304-4203\(99\)00049-3](https://doi.org/10.1016/S0304-4203(99)00049-3), 1999.

Kirkwood, D. S.: Stability of solutions of nutrient salts during storage, *Mar. Chem.*, 38, 151–164, [https://doi.org/10.1016/0304-4203\(92\)90032-6](https://doi.org/10.1016/0304-4203(92)90032-6), 1992.

Laquer, F. C.: Sequential precipitation samplers: A literature review, *Atmospheric Environ. Part Gen. Top.*, 24, 2289–2297, [https://doi.org/10.1016/0960-1686\(90\)90322-E](https://doi.org/10.1016/0960-1686(90)90322-E), 1990

Laurent, B., Losno, R., Chevaillier, S., Vincent, J., Rouillet, P., Bon Nguyen, E., Ouboulmane, N., Triquet, S., Fornier, M., Raimbault, P., and Bergametti, G.: An automatic collector to monitor insoluble atmospheric deposition: application for mineral dust deposition, *Atmospheric Meas. Tech.*, 8, 2801–2811, <https://doi.org/10.5194/amt-8-2801-2015>, 2015.

Peden, M.E., Bachman, S.R., Brennan, C.J., Demir, B., James, K.O., Kaiser, B.W., Lockard, J.M., Rothery, J.E., Sauer, J., Skowron, L.M. and Slater, M.J.: Methods for collection and analysis of precipitation. ISWS Contract Report CR 381, 1986.

Randall, David: *An Introduction to the Global Circulation of the Atmosphere*, Princeton University Press, Princeton, New Jersey, 43 pp., 2015.

Reddy, M. M., Liebermann, T. D., Jelinski, J. C., and Caine, N.: Variation in pH During Summer Storms Near the Continental Divide in Central Colorado, U.S.A.*, *Arct. Alp. Res.*, 17, 79–88, <https://doi.org/10.1080/00040851.1985.12004450>, 1985.

Sanei, H., Outridge, P. M., Goodarzi, F., Wang, F., Armstrong, D., Warren, K., and Fishback, L.: Wet deposition mercury fluxes in the Canadian sub-Arctic and southern Alberta, measured using an automated precipitation collector adapted to cold regions, *Atmos. Environ.*, 44, 1672–1681, <https://doi.org/10.1016/j.atmosenv.2010.01.030>, 2010.

Siksna, R.: The electrolytical conductivity of precipitation water as an aid to the chemical analysis. *Geofisica Pura e Applicata*, 42, 32–41 <https://doi.org/10.1007/BF02113385>, 1959.

Vermette, S. J. and Drake, J. J.: Simplified wet-only and sequential fraction rain collector, *Atmospheric Environ.* 1967, 21, 715–716, [https://doi.org/10.1016/0004-6981\(87\)90053-9](https://doi.org/10.1016/0004-6981(87)90053-9), 1987.

Wagner, S., Harvey, E., Baetge, N., McNair, H., Arrington, E. and Stubbins, A.: Investigating atmospheric inputs of dissolved black carbon to the Santa Barbara Channel during the Thomas Fire (California, USA). *Journal of Geophysical Research: Biogeosciences*, 126(8), p.e2021JG006442, 2021.

Wetherbee, G.A., Shaw, M.J., Latysh, N.E., Lehmann, C.M. and Rothert, J.E.: Comparison of precipitation chemistry measurements obtained by the Canadian Air and Precipitation Monitoring Network and National Atmospheric Deposition Program for the period 1995–2004. *Environmental monitoring and assessment*, 164(1), pp.111-132, 2010.

1 Cost Effective Off-Grid Automatic Precipitation Samplers for 2 Pollutant and Biogeochemical Atmospheric Deposition

3
4 Alessia A. Colussi¹, Daniel Persaud¹, Melodie Lao¹, Bryan K. Place^{2,‡}, Rachel F.
5 Hems^{2,§}, Susan E. Ziegler³, Kate A. Edwards^{4,‡}, Cora J. Young^{1,2}, and Trevor C.
6 VandenBoer^{1,3}

7
8
9 ¹ Department of Chemistry, York University, Toronto, ON

10 ² Department of Chemistry, Memorial University, St. John's, NL

11 ³ Department of Earth Science, Memorial University, St. John's, NL

12 ⁴ Canadian Forest Service, Natural Resources Canada, Corner Brook, NL

13
14 [‡] Now at: SciGlob Instruments & Services LLC, Columbia, MD, USA

15 [§] Now at: Department of Chemistry and Biochemistry, Oberlin College and Conservatory, OH, USA

16 [†] Now at: Climate Change Impacts and Adaptation Division, Lands and Minerals Sector, Natural Resources Canada, Ottawa, ON

17
18 **Correspondence:** Trevor VandenBoer (tvandenb@yorku.ca)

19 20 21 22 **Abstract**

23 An important transport process for particles and gases from the atmosphere to aquatic and
24 terrestrial environments is through dry and wet deposition. An open-source, modular, off-grid, and
25 affordable instrument that can automatically collect wet deposition samples allows for more
26 extensive deployment of deposition samplers in fieldwork and would enable more comprehensive
27 monitoring of remote locations. Precipitation events selectively sampled using a conductivity
28 sensor powered by a battery-based supply are central to off-grid capabilities. The prevalence of
29 conductive precipitation - that which initially contains high solute levels and progresses through
30 trace level concentrations to ultrapure water in full atmospheric washout, depends on the sampling
31 location but is ubiquitous. This property is exploited here to trigger an electric motor via limit
32 switches to open and close a lid resting over a funnel opening. The motors are operated via a
33 custom-built and modular digital logic control board, which have low energy demands. All
34 components, their design and rationale, and assembly are provided for community use. The
35 modularity of the control board allows operation of up to six independent wet deposition units,

36 such that replicate measurements (e.g., canopy throughfall) or different collection materials for
37 various targeted pollutants can be implemented as necessary.

38 We demonstrate that these platforms are capable of continuous operation off-grid for integrated
39 monthly and bimonthly collections performed across the Newfoundland and Labrador Boreal
40 Ecosystem Latitudinal Transect (47° to 53° N) during the growing seasons of 2015 and 2016.
41 System performance was assessed through measured power consumption from 115 volts of
42 alternating current (VAC; grid power) or 12 volts of direct current from battery supplies during
43 operation under both standby (40 or 230 mA, respectively) and in-use (78 or 300 mA, respectively)
44 conditions. In the field, one set of triplicate samplers was deployed in the open to collect incident
45 precipitation (open fall) while another set was deployed under the experimental forest canopy
46 (throughfall). The proof-of-concept systems were validated with basic measurements of rainwater
47 chemistry including: i) pH ranging from 4.14 to 5.71 in incident open fall rainwater; ii)
48 conductivity ranging from 21 to 166 $\mu\text{S}/\text{cm}$; and iii) dissolved organic carbon concentrations in
49 open fall and canopy throughfall of 16 ± 10 mg/L and 22 ± 12 mg/L, respectively; with incident
50 fluxes spanning 600 to 4200 mg C $\text{m}^{-2} \text{a}^{-1}$ across the transect. Ultimately, this demonstrates that
51 the customized precipitation sampling design of this new platform enables more universal
52 accessibility of deposition samples to the atmospheric observation community – for example, those
53 who have made community calls for targeting biogeochemical budgets and/or contaminants of
54 emerging concern in sensitive and remote regions.

55

56 **1.0 Introduction**

57 Atmospheric deposition is the central loss process for particles and gases to terrestrial and
58 aquatic surfaces (Pacyna, 2008). Particles and gases can be deposited by both dry and wet
59 deposition processes. Dry deposition is facilitated by the direct interaction of gases and particles
60 with boundary layer surfaces such as water, vegetation, and/or soil, while wet deposition involves
61 in-cloud scavenging and below-cloud interception of gases and aerosols by, e.g., rain droplets and
62 snow crystals (Fowler, 1980; Lovett and Kinsman, 1990). Dry and wet deposition are global
63 processes coupled to regional synoptic scale conditions, but their relative importance depends on
64 local sources and global transport of atmospheric analytes of interest. Dry deposition consists of a
65 variety of mechanisms for particles and gases, with fine mode particles and their chemical
66 constituents (compared to ultrafine and coarse mode particles) being more likely to undergo

67 atmospheric long-range transport prior to being deposited (Farmer et al., 2021). Wet deposition
68 occurs when such long-lived atmospheric particles and gases are included and/or scavenged into
69 cloud water and transported to the surface of the Earth in precipitation (e.g., snow and rain). With
70 the size and number of droplets in the atmosphere largely controlling the rate, wet deposition
71 depends on a variety of meteorological factors affecting precipitation, such as the size distribution
72 and concentration of ice and droplet nucleating particles, as well as the solubility, concentration,
73 and reactivity of gases (Lovett, 1994). Ultimately, deposition plays an important role in pollutant
74 distribution and biogeochemical cycling of long-studied major nutrients (e.g., nitrogen and sulfur
75 in acid rain) and those with increasing recognition of importance such as dissolved organic carbon
76 (DOC) (Meteorological Service of Canada, 2005; Vet et al., 2014; Safieddine and Heald, 2017;
77 United States Environmental Protection Agency, 2020).

78 Recognizing the significance of atmospheric trace chemical deposition has led to the
79 establishment of monitoring networks. For example, long-term wet deposition monitoring
80 networks, like the Canadian Air and Precipitation Monitoring Network (CAPMoN) and the
81 National Atmospheric Deposition Program (NADP), aim to provide critical data on the spatial and
82 temporal patterns of wet and dry deposition. As a result, this has allowed for the estimation of
83 regional and continental deposition rates of species regulated by national or international policies
84 (Lovett, 1994). Data from these networks have been critical to understanding the efficacy of policy
85 to reduce environmental issues like acid rain (Likens and Butler, 2020). In particular, the Oslo and
86 Geneva protocols have achieved an 80% decrease in both North American and European SO₂
87 emissions since 1980 (Grennfelt et al., 2020). Despite these successes, reduction in acid deposition
88 has had unexpectedly slow recovery in ecosystems leaving them sensitized – necessitating
89 continued deposition monitoring (Stoddard et al., 1999; Kuylenstierna et al., 2001).

90 **Over the past 60 years, the precipitation chemistry community has made**
91 **advancements in deposition collectors to better understand atmospheric processes (Siksnas,**
92 **1959).** While bulk deposition collection (i.e., a collection bucket or jug fitted with a funnel open
93 at all times; Hall, 1985) is both a simple and economically feasible sampling method utilized by
94 monitoring networks, it is subject to bias through collection of inputs other than atmospheric
95 deposition (e.g., bird droppings, insects, plant debris). As a result, bulk collectors can overestimate
96 total deposition and underestimate wet deposition in a variety of locations (Lindberg et al., 1986;
97 Richter and Lindberg, 1988; Stedman et al., 1990). **Sequential precipitation collection methods**

98 **include manually segmenting samplers (requiring only a shelter, clean surface, and an**
99 **operator), linked collection vessels (sample containers that are filled in sequence via**
100 **gravitational flow), amongst others and have been developed to analyze rainwater**
101 **composition and measure parameters such as pH and conductivity (Gatz et al., 1971; Reddy**
102 **et al., 1985; Vermette and Drake, 1987; Laquer, 1990). Sequential sampler designs have also**
103 **been adapted to collect precipitation in remote field sites (Germer et al., 2007; Sanei et al.,**
104 **2010).** Although it is a more costly and time intensive method when compared to bulk deposition
105 collection, the major appeal of measuring isolated wet deposition is **the ability to isolate this**
106 **individual atmospheric process.** Further innovation can reduce bias and improve the preservation
107 of samples, such as the use of sensors to automate isolation of collected precipitation or the
108 addition of polymeric mesh barriers to reduce debris input in windy environments (Lovett, 1994)
109 - yet commercial solutions often come at a substantial expense.

110 When targeting biogeochemically relevant species in deposition collectors, additional
111 standard practices have been developed to improve the representativeness of sample composition.
112 First, an appropriate monitoring site must be selected. Three categories of siting criteria,
113 established by organizations such as CAPMoN and the NADP, are of particular importance: (i)
114 site representativeness and physical characteristics, (ii) distance from potential pollution sources,
115 and (iii) operational requirements (Canadian Air and Precipitation Monitoring Network, 1985a;
116 National Atmospheric Deposition Program, 2009). This means that each site must be a location
117 that receives precipitation representative of the hydrologic area; is ideally not within 500 m of
118 local pollution sources, such as wood-burning stoves, garbage dumps, and vehicle parking lots;
119 and is accessible for daily collections, maintenance, and can be serviced by reliable 115 volts of
120 alternating current (VAC) electrical power (Canadian Air and Precipitation Monitoring Network,
121 1985a; National Atmospheric Deposition Program, 2009). Despite these guidelines, there are many
122 reasonable scenarios in which these siting conditions cannot be met. As an example, remote sample
123 collections are often required for global assessments on persistent contaminants or nutrients of
124 biogeochemical importance. Remote locations, however, can result in sampling sites with no
125 power provision, infrequent sample collection, and/or the infrastructure-bearing location itself
126 being a source of the targeted pollutants. As a result, innovation in collection strategies such as
127 time-integrated off-grid sampling, with modularity in the deployment of replicates, as well as

128 materials for quantitative collection of environmental targets, is still needed to expand and/or
129 modify networks to meet current and future monitoring and policy needs.

130 In biogeochemical cycles, for example, improvement of constraints in atmospheric carbon
131 linkages to terrestrial and aquatic processes is necessary. This would play a critical role in correctly
132 assessing climate feedbacks and reducing uncertainty in Earth system models. The measurement
133 of atmospheric DOC transport to surfaces has been limited and impedes landscape scale carbon
134 balance from being obtained (Casas-Ruiz et al., 2023). The pool of compounds from which DOC
135 is derived in the atmosphere has also been limited and is only recently seeing an increase in
136 research intensity. Reactive organic carbon (ROC) is defined as the sum of nonmethane organic
137 gases and primary and secondary organic aerosols (Safieddine and Heald, 2017). The major
138 removal mechanism of water-soluble organic compounds produced through oxidation from the
139 atmosphere is by dry deposition of particle-bound pollutants and scavenging by rainfall (Jurado et
140 al., 2004, 2005). When ROC is scavenged into rainfall, it becomes DOC and enters terrestrial and
141 aquatic systems. Deposition measurements of ROC compounds are needed since they play a
142 crucial role in the formation of secondary species such as ozone, particulate matter, and carbon
143 dioxide (CO₂) (Safieddine and Heald, 2017; Heald and Kroll, 2020).

144 There are several evolving drivers around studying atmospheric ROC; for example, light-
145 absorbing organic carbon that can affect global radiative balance and undergo photochemical
146 transformations in the condensed phase (Saleh, 2020; Wang et al., 2021; Washenfelder et al., 2022;
147 George, 2023). Reactive organic carbon can also influence cloud formation and contribute to
148 precipitation acidity (Avery et al., 2006; Ramanathan and Carmichael, 2008). Measurements of
149 speciated ROC are difficult due to the chemical complexity of emitted compounds and oxidation
150 products (Heald and Kroll, 2020). To circumvent this, monitoring and quantifying DOC can be
151 used as a proxy to estimate the total ROC in precipitation. However, quantitative measurements of
152 DOC in precipitation samples are sparse due to its relatively low concentration of 0.1 to 10 mg C
153 L⁻¹ (Iavorivska et al., 2016; Safieddine and Heald, 2017). Recently, calls for carbon closure on
154 atmospheric processing of ROC make this measurement of increasing importance (Kroll et al.,
155 2011; Heald et al., 2020; Barber and Kroll, 2021). Similarly, to obtain net landscape or watershed
156 carbon exchange, studies require effective methods for capturing and preserving atmospheric DOC
157 deposition to constrain biogeochemical linkages at global interfaces as outlined above.

158 In this work, we present the design of a custom-built automated array of precipitation
159 samplers that can be operated both on- and off-grid for wet deposition collection. The purpose of
160 these samplers is to enable cost-effective collection of integrated water-soluble conductive
161 atmospheric constituents deposited in remote environments without grid power or routine access.
162 A sensor interfaces with a custom-built motor control board capable of operating up to six
163 independent wet deposition units such that canopy throughfall (TF) and incident precipitation
164 (open fall, OF) measurements are possible to collect in replicate. **The materials used can be easily**
165 **changed in order to optimize collection and preservation** of a wide array of target analytes, such
166 **as DOC, when using high density polyethylene and mercuric chloride (HgCl₂).** We
167 demonstrate that these platforms are capable of continuous operation off-grid for monthly wet
168 deposition collection of precipitation across the Newfoundland and Labrador Boreal Ecosystem
169 Latitudinal Transect (NL-BELT) during snow-free periods in 2015 and 2016. Extremes in system
170 performance were evaluated by testing the power consumption of a sampling array from spring
171 through fall when paired with a solar top-up system, and during snow-free winter conditions using
172 only a battery. The two years of field samples were collected using an array of six collection units,
173 with triplicate collection of both incident precipitation and throughfall from rain passing through
174 a forest canopy. Samples were analyzed in terms of deposition volumes relative to total bulk
175 volumes, reproducibility of replicate samples, and to determine the fraction of conductive rainfall
176 within the total volume of precipitation at these remote sites. **The captured fraction compared to**
177 **total volume deposited is used to gain insight into how these samplers can limit analyte dilution**
178 **effects and improve method detection limits, such as rejecting 50% of the total volume**
179 **delivered as ultrapure precipitation leading to a factor of two improvement.** **Chemical**
180 **parameters of pH, conductivity, and DOC fluxes collected according to established preservation**
181 **protocols** were then **compared to prior measurements** to validate this proof-of-concept system.
182 Measurement methods for pH and conductivity of rainwater are very well-established in the
183 literature and serve as a baseline reference to ensure that the samples collected by the new devices
184 presented in this work are consistent with what is expected in samples from a remote coastal
185 environment, given the selective sampling strategy. **We then move away from these well-**
186 **established parameters to quantify DOC fluxes using established biogeochemical preservation**
187 **techniques for fresh water and groundwater** to demonstrate the potential of these samplers in

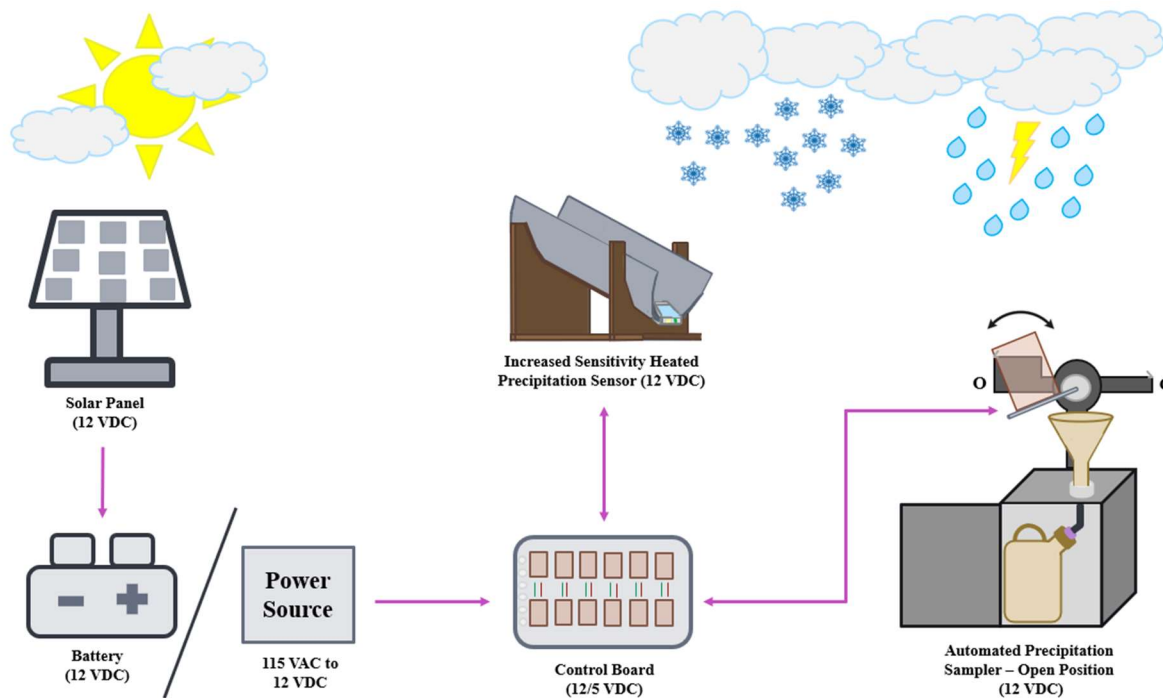
188 application to automated collection of analytes of emerging importance and interest in the remote
189 locations of our latitudinal transect.

190

191 **2.0 Materials and Methods**

192 **2.1 Precipitation Sampling Array Components**

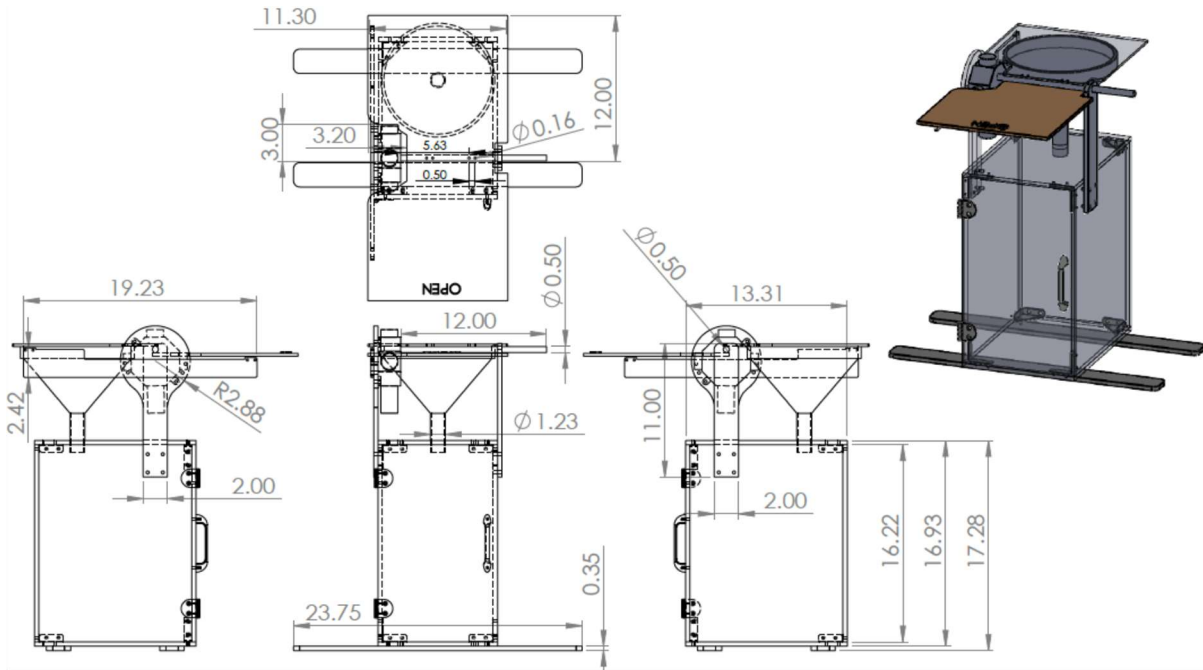
193 Each automated precipitation sampling setup can be operated as an array, here being used
194 in groups of up to six collection units (Figure 1). A collection unit is a simple opaque doored box.
195 The box protects the sample containers against exposure to direct sunlight and provides a mounting
196 location for the funnel and lid, while also facilitating easy exchange of sample containers. The
197 collection units can be fitted with stabilizing legs that allow them to be bolted to concrete or pinned
198 by retaining rods when on soil. In both cases, this prevents tipping and loss of sample during high
199 winds or wildlife-sampler interactions (e.g., Figures 2 and S1). The collection of precipitation is
200 facilitated by a funnel mounted through the top of the sampling unit. The funnel tip extends into
201 the opening of the sample collection container placed inside. The connection can be sealed to better
202 preserve volatile analytes with tubing that passes through a sealed grommet (P/N 9280K34,
203 McMaster-Carr) to enter the sample collection container and minimize evaporative losses.
204 Precipitation events are sampled selectively by modulating the position of a lid over the funnel
205 with an electric motor. The collection unit motors are operated by a digital control board, which
206 interfaces with a precipitation sensor and requires 12 volts of direct current (VDC) power supplied
207 to this system. Switches detecting the lid position ensure complete opening or closure of the funnel
208 mouth for each collection unit.



209
 210 **Figure 1.** Schematic of custom-built automated precipitation sampling array components for off-
 211 grid wet deposition collection. The pink arrows denote the direction of electrical signal and power
 212 exchanged between components. The curved black arrow indicates the rotation of a motorized lid
 213 to obtain open (O) or closed (C) sampler configurations.
 214

215 **2.1.1 Collection Units**

216 The collection unit materials to date have been made of both 3/8” plywood and black
 217 polyacrylate sheeting. The materials have demonstrated high durability on the order of four years
 218 under field conditions (Figures S1 and S2). Opaque materials were explicitly selected to minimize
 219 photochemical reactions and growth of photosynthetic microorganisms within the sample. The
 220 dimensions of the collection unit are detailed in Figure 2. Each can accommodate sample
 221 containers up to 20 L in volume for collection in locations with large monthly wet deposition
 222 volumes, such as in Newfoundland and Labrador (Table 1).
 223



224

225 **Figure 2.** Detailed collection unit schematic with all dimensions provided in inches. Further
 226 specifications for the lid dimensions can be found in Figure S3. The shaded 3D rendering depicts
 227 both open and closed states for the lid, positioning of legs to secure it to surfaces, placement of
 228 corner brackets, and the door handle and hinges.

229

230

231

232

233

234

235

236

237

238

239

240

241

242

243

244

The box panels can be joined using hardware inserts (P/N 1556A54 and 1088A31, McMaster-Carr, Aurora, OH), 3D printed corners (Figure S4), or along the box edges with screws if using wood. The door is attached with two hinges (P/N 1549A57, McMaster-Carr) and held closed with a magnetic contact (P/N1674A61; McMaster-Carr) or hooked latch. The electric motor controlling the lid is enclosed in a standard polyvinylchloride electrical junction box, which is attached to a short paddle mounted on one side of the collection unit. Here we used an electric worm-gear motor (12 VDC, 2 revolutions per minute; TS-32GZ370-1650; Tsiny Motor Industrial Co., Dong Guan, China) mounted inside the enclosure with matching hex bolts (P/N 91251A146, McMaster-Carr) that passed through the weather-tight cover while the drive shaft protrudes through a 3/8" hole drilled in the cover. The drive shaft has a flat edge to affix the lid rod using a short set screw (Figure S5) that is cemented semi-permanently in place with thread locking compound (P/N 91458A112; Loctite Threadlocker Blue 242; McMaster-Carr). The lid rod is 3/8" aluminum machined on one end to allow connection to the motor drive shaft (Figure S5) with four threaded holes along its length to affix the lid (Figure S3). The lid rod passes through a second mounting paddle on the box that keeps the lid level and capable of isolating the funnel from the

245 atmosphere in the absence of precipitation. The lids used here were made of 1/8" Lexan
246 polycarbonate sheet.

247 Selective precipitation sampling is performed using a logic-based assessment of sensor and
248 switch states (defined in Figure S6) by the control board quadNOR gate chip (Fairchild
249 Semiconductor P/N DM74LS02) which activates the H-bridge motor driver chipset (Figure S7).
250 A 12 VDC signal drives the clockwise or counterclockwise rotation of the motor, installed in a
251 suitable port of the junction box, via a cable from the control board, which passes through a
252 weather-tight compression fitting (e.g., Home Depot SKU# 1000116446). The motor rotation
253 signal is interrupted when the lid makes contact with one of two weather-tight limit switches (P/N
254 SW1257-ND; Omron, Digi-Key Electronics, Thief River Falls, MN) mounted on opposite ends of
255 a horizontal armature connected to the vertical motor mounting paddle (Figure 2). The switches
256 controlling the lid location ensure that the funnel is completely open or covered as necessary for
257 precipitation collection. The funnels used in this work are 20 cm in diameter and made from high-
258 density polyethylene (HDPE; Dynalon, P/N 71070-020, VWR International, Mississauga, ON). A
259 7" x 5" piece of filtration mesh (P/N 9265T49, McMaster-Carr) that was tied together as a fitted
260 cone insert with Nylon thread (e.g., fishing line) to prevent large debris entering the sampler
261 containers when used, for example, in the collection of TF precipitation under a forest canopy
262 when accompanying litterfall is also expected. The exit of the funnel directs the collected
263 precipitation into the narrow-mouth opening the container inside the collection unit, such as 20 L
264 HDPE jugs or 10 L HDPE jerricans (Bel-Art Products; P/N 11215-314, VWR International).

265

266 2.1.2 Heated Precipitation Sensor

267 The detection of rain modulates the opening and closing of the collection units by an
268 interdigitated resistive sensor (M152; Kemo Electronic GmbH, Geestland, Germany; Figures S6
269 to S8). **This approach is consistent with established precipitation detection techniques used
270 by government monitoring programs (e.g., CAPMoN; Canadian Air and Precipitation
271 Monitoring Network, 1985a, 1985b).** The rain sensor detects conductive deposition by the
272 completion of a conductive circuit when electrolytes bridge the connection between the
273 interdigitated gold electrodes. The sensor is supplied with 12 VDC from the power system to
274 trigger a relay when precipitation conductance above 1 M Ω ·cm conductivity is detected
275 (determined experimentally, see Section S1). **This is equivalent to approximately 8 μ M sodium**

276 **chloride. The sensor detection limit reflects an upper limit of precipitation ion loading**
277 **because the design of the rain chute leads to an increase in surface area of more than a factor**
278 **of 25 on which solutes can accumulate to enhance the ionic content of the deposited water.**

279 An output of 12 VDC is sent to the digital control board by the relay when rain is sensed, or 0
280 VDC in its absence, for signal processing and motor control (Figure S7). **When rain is sensed,**
281 **the lid of each sampler in the array is simultaneously opened (<5 seconds) and is dependent**
282 **on the rotational rate of the lid motor.** To increase the sensitivity of this sensor and to extend
283 the sampling duration when conductive atmospheric constituents are completely washed out of the
284 atmosphere, a sloped tin chute (e.g., Home Depot SKU# 1001110514) was added to extend the
285 surface of the rain sensor. The sensor was placed at the end of the chute and sealed in place with
286 caulking to allow water droplets to move easily from the chute onto the sensor.

287 The angle of the chute can be adjusted to control the momentum of collected droplets so
288 that they collect on the sensor surface and only flow off it when the rate of precipitation exceeds
289 the sensor evaporation capability. When soil is available, two bent rods can be used to hold the
290 chute at the optimized angle of 10° (Figure S2). They are inserted into the soil and the chute is
291 affixed to the tops of the rods with zip ties passed through small holes drilled in the sides of the
292 chute, which are subsequently sealed with caulking. When soil is unavailable, for example in urban
293 environments, we have created a mounting frame to hold the chute at the optimized angle of 10°
294 (Figure S8). When precipitation is detected the sensor surface draws current up to 1.0 ampere (A)
295 into a heater to actively evaporate water from its surface so that it accurately detects the active
296 period of rain events. The heated sensor has undergone preliminary field tests and is also capable
297 of detecting ice and snow, provided they contain electrolytes.

298

299 **2.1.3 Power Supply Systems**

300 Power for this system can be supplied from a battery at 12 VDC or using a 115 VAC to 12
301 VDC transformer power supply (P/N 285-1818-ND; TDK-Lambda Americas, Digi-Key
302 Electronics). Depending on the duration of sampling and the time of year, the battery capacity can
303 be changed to suit power needs (Section 3.2.2). To provide sufficient power density in this study,
304 over one to two month-long collection periods, the battery capacity was carefully matched; with
305 top-up options implemented when prolonged or high-frequency precipitation was expected.
306 Absorbent glass mat (AGM) marine deep cycle batteries can withstand discharge events down to

307 less than 60 % capacity and are robust under nearly all environmentally relevant temperatures (\leq
308 -20°C to 40°C). Additionally, these batteries interface easily with solar charging options as they
309 are able to accept high current input. Monthly collections in Newfoundland were powered with 76
310 amp-hour (Ah) AGM batteries (Motomaster Nautilus; Ultra XD Group 24 High-Performance
311 AGM Deep Cycle Battery, 12 VDC) topped up by a 40 W solar panel interfaced with a charge
312 controller to prevent overcharging (Coleman; Model # 51840, max current of 8 A at 14 VDC).

313 For collections made every second month in Labrador, a 120 Ah battery with the same
314 solar top-up strategy was used to ensure continuous operation. For either remote field deployment,
315 batteries and charge controllers were housed in a Pelican™ case (Model 1440, Ocean Case Co.
316 Ltd., Enfield, NS) fitted with weathertight bulkhead cord grips (P/N 7529K655, McMaster-Carr)
317 through which charging and power cables were passed (Belden, Coleman; S/N 7004608,
318 70875227, Allied Electronics, Inc., Ottawa, ON). Humidity in all weatherproof cases was
319 minimized by exchanging reusable desiccant packs (Ocean Case Co. Ltd.) when depleted batteries
320 were exchanged for fully charged replacements. Solar panels were repositioned monthly to
321 optimize orientation for solar power provision. Using either power source, the control board
322 converts and distributes the 12 VDC to the other components in the precipitation sampling array.

323

324 **2.1.4 Custom Control Board**

325 A custom control board to operate a six-collection unit array was designed based on prior
326 digital logic circuits for standalone collectors (VandenBoer, 2009). The 12 VDC battery or
327 transformer output is supplied directly to the rain sensor and relay, as well as to the motor drivers
328 for lid opening (Figure S9). Each collection unit is controlled independently to ensure lids are fully
329 opened or closed, thereby requiring six replicate motor driver control circuits that respond to their
330 independent switch signals. The remainder of the signaling and digital logic operates on 5 VDC
331 which is produced by on-board voltage regulators (Micro Commercial Co; P/N MC7805CT-BP,
332 Digi-Key Electronics). The lid switches are provided with 0 and 5 VDC to indicate collection unit
333 open or closed status (Omron Electronics; P/N D2FW-G271M(D), Digi-Key Electronics). The
334 signals from the sensor and switches connect to the board through four-conductor cable (Belden;
335 S/N 70003678, Allied Electronics Inc.) passed through weathertight bulkhead cord grips and
336 secured to screw terminals (Figure S9). The sensor and switch signal inputs interface with a quad
337 NOR GATE chipset (Texas Instruments; P/N 296-33594-5-ND, Digi-Key Electronics) to trigger

338 the motor driver (STMicroelectronics; P/N 497-1395-5-ND, Digi-Key Electronics) such that it
339 rotates or remains stationary. The additional resistors, capacitors, and diodes are necessary to
340 maintain stable signaling throughout the printed circuit board (Figure S9, Table S1).

341 The custom control board was housed in a Pelican™ case (1400 NF; Pelican Zone,
342 Mississauga, ON) fitted with cut-to-use foam inserts and a reusable desiccant pack that was also
343 exchanged alongside those for the battery cases. All collection units, sensors, and power supply
344 cables were passed through eight weathertight bulkhead cord grips and fixed to screw terminals
345 on the board. The opposing ends of the cables were fitted with weathertight Bulgin Buccaneer 400
346 or 4000 Series circular cable connectors (Table S2; Allied Electronics, Inc.) to allow easy field
347 installation with mated connectors on the cables originating from each of the previously mentioned
348 array components. Connected cables could then be buried in shallow soil trenches to reduce the
349 attention of gnawing animals, as well as potential entanglement hazards with other wildlife.
350 Precipitation events were logged from the control boards using a HOBO 4-channel analog data
351 logger (UX120-006M; Onset®, Bourne, MA) that records the sensor, switch, and motor voltages.
352 The fourth channel is reserved to monitor battery or power supply voltages over time (Section 3.2).

353

354 **2.2 Power Demand and Management Tests**

355 Power demand was calculated based on the cumulative component requirements prior to
356 the selection of batteries. This was to ensure adequate capacity to collect samples over one to two
357 month-long field deployments and are sufficient for an assumed worst-case scenario of one week
358 of constant rain without solar power charge restoration. Solar panel power production capacity
359 was determined based on the calculated energy required to recharge the battery. As a result, we
360 selected the 40 W panel which could complete charging at 14 VDC with a week of direct sunlight
361 at 8 hours per day. The power demand for a six-sampler array was measured in standby and during
362 operation with a digital power meter (Nashone PM90, Dalang Town, China) in real-time when
363 supplying 12 VDC with a transformer. Contrasting power demand tests were performed under
364 different environmental conditions and power management configurations. The first was
365 performed using the 76 Ah AGM battery with a solar top-up in an urban environment from July
366 through August 2018, while the other was performed using a 103 Ah AGM battery alone from
367 January through February 2019.

368

369 2.3 Continuous Monthly Collection of Remote Samples at NL-BELT

370 One array of six automated collection units (3 OF, 3 TF) were deployed within one forested
371 experimental field site located in each of the four watershed regions of the NL-BELT (24 samplers
372 in total) between 2015 and 2016. Additionally, between one to three total deposition samplers were
373 located at each of the four field sites (Table 1, Figure S10). The watersheds span 5.5° latitude from
374 the southernmost site Grand Codroy (GC), through the colocated Pynn's Brook (PB) and Humber
375 River Camp 10 (HR) sites, to Salmon River (SR) as the highest latitude site on the island of
376 Newfoundland. The northernmost forested watershed, Eagle River (ER), is located in southern
377 Labrador and extensive details characterizing each of the four sites can be found in Ziegler et al.
378 (2017). All sampling locations are far from anthropogenic pollutant point sources, except for the
379 ubiquitous presence of marine sea spray from the nearby marine coastlines. The total deposition
380 samplers were identical to the automatic collection units except that they were not fitted with a
381 motor arm and lid, so they did not require a source of power. Three of the six automated samplers
382 were deployed in the open at a distance from the forest stand, equal to or greater than the height
383 of the trees, in line with CAPMoN and NADP guidelines. The other three automated samplers
384 were placed under the canopy to collect TF precipitation within the forest sites. These samplers
385 actively collected wet deposition into integrated monthly (Newfoundland) or two-month
386 (Labrador) samples during snow-free periods (approximately June through November). The arrays
387 were collected and stored during the winter months while total deposition samplers remained in
388 field locations year-round. It is also important to note that during the growing season, sample
389 collections were made at the same time – that is, OF and TF deposition were collected on a single
390 day at each sampling site and within a few days of each other across the transect. Collected sample
391 volumes were compared between the automated samplers and total deposition collectors for each
392 collection interval as a check on proper function (i.e., less than or equal volumes in automated
393 samples). During each site visit, the slope of the sensor was confirmed to be correct, sample
394 containers were collected and replaced with clean units, the battery and desiccant packs replaced
395 with fully recharged devices, and the entire array confirmed operational.

396

397 **Table 1.** NL-BELT sampling site details provide locations and identifiers, alongside those from
398 long-term weather stations operated by Environment and Climate Change Canada (ECCC). Soil
399 pH was determined from samples collected at the same time as precipitation. Mean annual
400 temperature was derived from ECCC climate normals. Annual total deposition precipitation

401 volumes were either measured for the 2015-16 period (ECCC, This Work) or calculated by the
 402 Oak Ridge National Lab DAYMET archive.
 403

Sampling Site	Sampling Site Location	Station (Climate ID)	Station Location	Soil pH ^a	MAT (°C) ^b	Average Annual Precipitation (L)		
						ECCC ^c	DAYMET ^g	This Work
Grand Codroy (GC)	47°50'43.1"N 59°16'16.0"W	Stephenville A (8403801)	48°32'29.00" N 58°33'00" W	3 to 4	5.0 ^e	53.2	58.9	45.6 (+5.17)
Pynn's Brook (PB)	49° 05' 13.20"N 57° 32' 27.60" W	South Brook Pasadena (8403693)	49°01'00" N 57°37'00" W	3 to 4	4.6 ^e	21.4	54.3	38.6 ^h
Salmon River (SR)	51°15'21.6"N 56°08'16.8"W	Plum Point (40KE88)	51°04'00" N 56°53'00" W	3 to 4	2.4 ^e	47.1	45.4	32.3
Eagle River (ER)	53°33'00.0"N 56°59'13.2"W	Cartwright A (8501100)	53°42'30" N 57°02'06" W	3 to 4	0 ^d	- ^f	56.3	25.8

404
 405 ^aSoil pH for the organic and mineral soil horizons determined by addition of 400 µL of aqueous 0.5 M CaCl₂ to a 50:50 w/w slurry
 406 of dried soil in deionised water. **Note: the four remote NL-BELT sites are dominated by balsam fir trees underlain by humo-**
 407 **ferric podzol soil with pH ranging between 3.0 and 4.5.**

408 ^bEnvironment Canada: Canadian Climate Normals, 1981 to 2010, https://climate.weather.gc.ca/climate_normals/ (last accessed:
 409 14 July 2023).

410 ^cAt least 20 years of measurements.

411 ^dThe World Meteorological Organization's "3 and 5 rule" (i.e., no more than 3 consecutive and no more than 5 total missing for
 412 either temperature or precipitation).

413 ^eAnnual precipitation averages determined using ECCC daily precipitation reports.

414 ^fLarge quantity of missing data for this location from January 2015 to December 2016 prevents any reliable estimate.

415 ^gEstimated deposition rates converted to volume using DAYMET (Thornton et al., 1997, 2021, 2022).

416 ^hVolumes merged for 2015 and 2016 at PB and HR.

417
 418

419 **2.3.1 Sample Preservation**

420 Four of the six sample containers (two each of OF and TF) were biologically sterilized
 421 using 1 mL of a saturated aqueous solution of mercuric chloride (HgCl₂) to preserve against
 422 biological growth and loss of bioavailable nutrients over the collection periods. Unsterilized
 423 sample containers (without HgCl₂) were used for measurements of recalcitrant species and to
 424 assess any matrix effects exerted on target analyte quantitation. **The use of HgCl₂ as a sample**
 425 **preservation technique has been long-studied and well-established (Kirkwood, 1992;**
 426 **Kattner, 1999); thus, additional tests to verify the preservation of collected chemical species**
 427 **over time were not performed. The analysis of deposition collected in unsterilized and**
 428 **sterilized containers, however, serves as a method for internal sample validation - as does**

429 **our evaluation of measurement outcomes in comparison to those reported within the**
430 **literature.** Collected sample volumes were measured with a 1000 ± 10 mL graduated cylinder and
431 aliquots were collected for chemical analysis via transfer to precleaned 500- or 1000-mL HDPE
432 containers (Nalgene; VWR International). Samples were stored at 4°C before returning to the
433 laboratory where they were filtered with a 1000 mL Nalgene vacuum filtration system (P/N ZA-
434 06730-53; ThermoFisher Scientific, Waltham, MA), fitted with 0.45 μm polyethersulfone filters
435 (PES, P/N HPWP 04700, EMD Millipore), to remove suspended solids. Filtered samples were
436 transferred to new clean HDPE containers and stored for up to two months at 4°C in a cold room
437 until analysis. **The target analytes in this work are non-volatile and the described sample**
438 **collection methods consider this analyte property, as well as their interactions with container**
439 **materials. The versatility of the system design allows for the use of different collection**
440 **materials, keeper solvents for volatile organics, etc., so that the experimental design can be**
441 **analyte specific, depending on end user needs. Sample preservation approaches should thus**
442 **be identified by users of this new platform based on their scientific objectives and review of**
443 **the literature (Galloway and Likens, 1978; Peden et al., 1986; Dossett and Bowersox, 1999;**
444 **Wetherbee et al., 2010). In addition to the internal validation approach described here, we**
445 **aim to demonstrate that the precipitation samplers in this work are suitable for measuring**
446 **conductive deposition on- and off-grid. Below we highlight autonomous off-grid operations,**
447 **determine the fraction of conductive rainfall collected from the total volume of precipitation,**
448 **and validate our measurements through comparison to the literature.**

449

450 2.4 Cleaning and Preparation of Sample Containers

451 All sample collection and storage containers, as well as all sample handling apparatuses,
452 were made of HDPE or polypropylene for the quantitative analysis of target analytes. Prior to use
453 in handling samples, these were all acid-washed in 10 % v/v HCl (P/N BDH7417-1; VWR
454 International) followed by six sequential rinses with distilled water and ten rinses with 18.2
455 $\text{M}\Omega\cdot\text{cm}$ deionised water (DIW; EMD Millipore Corporation, Billerica, MA, USA). Containers
456 were dried by inversion on a clean benchtop protector overnight, or with protection from dust using
457 lint-free lab wipes over container openings when necessary. **Field and method blanks were**
458 **collected through the addition of DIW to cleaned containers, and/or sample handling devices, in**

459 order to quantify appropriate method detection limits and to identify any sources of systematic or
460 random contamination. **Blank subtraction was applied to measurements, where appropriate.**

461

462 **2.5 Measurements of pH and Conductivity**

463 The pH and conductivity of each sample was determined using a ThermoScientific™ Orion
464 Versa Star meter (ORIVSTAR52) interfaced with a pH electrode (Model: 8157BNUMD, Ultra
465 pH/ATC Triode, ROSS) and 4-electrode conductivity cell (Model: 013005MD, DuraProbe,
466 ROSS). Prior to use, the probes were calibrated daily with standard solutions specific for these
467 probes (ThermoScientific™ Orion™ conductivity standard 1413, and pH 4, 7, and 10 buffers) and
468 then stored between analyses according to manufacturer directions. Aliquots of 15 mL of
469 precipitation from archived samples were subsampled into 40 mL polypropylene Falcon tubes.
470 This was followed by immersion of a cleaned electrode for the conductivity measurement,
471 followed by the pH probe measurement to prevent conductivity bias due to potassium chloride
472 migration across the glass frit of the pH probe. Readings were recorded once signals had stabilized.

473

474 **2.6 Measurements of Dissolved Organic Carbon (DOC)**

475 Measurements of DOC were performed by catalytic combustion of samples in a platinum
476 bead-packed quartz furnace at 720°C to quantitatively produce CO₂, followed by non-dispersive
477 infrared absorption spectrophotometry using a Shimadzu Total Organic Carbon (model: TOC-V)
478 analyzer and an autosampler (model: ASI-V). Cleaning of materials prior to DOC determination
479 follows the same procedure as for the sample containers. Precipitation aliquots of at least 12 mL
480 were transferred to clean and combusted (500°C, 5 hours) 40 mL borosilicate glass vials, then
481 capped and stored at 4°C until analysis. Prior to analysis, vial caps were replaced with cleaned
482 polytetrafluoroethylene-lined septa. Inorganic dissolved carbon (e.g., H₂CO₃) was purged from
483 samples by acidification to pH 2 with HPLC grade H₃PO₄ (20 % v/v) and bubbling with an inert
484 carrier gas. Samples were analyzed in triplicate and quantified using calibrations spanning 0.1 to
485 10 or 10 to 100 ppm (mg C L⁻¹) with potassium hydrogen phthalate (KHP), depending on the
486 relative sample concentration range. Accuracy and precision were assessed using 1 and 10 ppm
487 KHP check standards analyzed every 10 injections, respectively. Calibrations were performed at
488 the beginning of every analysis day.

489

490 3.0 Results and Discussion

491 In addition to the general design advantages in the section that follows, we present the
492 results of various physical and chemical parameters to validate this new open source custom-built
493 modular system. The power consumption and snow-free performance testing are used to
494 demonstrate the off-grid capabilities of these samplers, as are the two-year datasets. The lower
495 power requirements are compared to existing commercial samplers and paired with solar top-up
496 to prolong the use and reduce the need to replace batteries on timescales shorter than planned
497 sampling duration (i.e., < 1 month). We then evaluate the automated wet deposition volumes, in
498 which the samplers prevent dilution during atmospheric washout events, compared to total
499 volumes collected from co-located samplers to depict the fractionation by volume as a function of
500 time. We also investigate the advantages of replicates collected across the four watersheds, using
501 deployments of triplicate samplers under field conditions. The ratio of collected TF to OF
502 replicates highlights the ability of these samplers to capture the dynamic nature of precipitation
503 interacting with forest canopies. Simple pH and conductivity measurements are then used as
504 benchmarks to situate the NL-BELT data within the established literature to emphasize the robust
505 operation of the samplers and impact of the selective sampling. Fluxes of DOC are also
506 interrogated across all four sampling sites as we demonstrate the potential of these samplers to
507 make measurements of more complex analyte pools that are of current interest to the atmospheric
508 measurement community.

509

510 3.1 General Design Advantages

511 **While several precipitation collectors have been similarly developed to address**
512 **specific scientific objectives – e.g., the quantitation of dust in wet and dry deposition (Laurent**
513 **et al., 2015; Brahney et al., 2020), determination of ions and DOC in a tropical rainforest**
514 **(Germer et al., 2007) and urban environments (Audoux et al., 2023), here we present a more**
515 **general design for modular adaptability.** When compared to other precipitation collection
516 apparatuses, the automated precipitation sampler developed in this work has several advantages.
517 Most notable is the ability to collect integrated samples at remote locations by exploiting its off-
518 grid capabilities. Our approach also maximizes the sensitivity of the rain sensor as long as
519 electrolytes remain in the water reaching it. The chute ensures that even if the precipitation contains
520 ultra-trace analyte quantities, they are still collected and quantified for an extended period when

521 high-purity water may be deposited during an atmospheric wash-out event. The chute does this by
522 accumulating water-soluble materials between rain events that require time to be completely
523 washed off and through the release of ions from the material itself, which ages under environmental
524 conditions. As the conductivity of the precipitation falls below the sensor threshold **– conductive**
525 **precipitation being that which initially contains high solute levels that progress through trace**
526 **level concentrations**, the added ions from the chute prolong the collection of rain past this time
527 point. In rainfall events where extended atmospheric wash-out occurs, **where precipitation**
528 **becomes ultrapure water**, the sampler lids will eventually close – preventing dilution of the
529 sample while maintaining the collection of analytes of interest. **A recent study has found that**
530 **rainfall events could exhibit variability and the lower atmosphere can be supplied with**
531 **aerosols due to specific sources, atmospheric dynamics, and meteorological conditions**
532 **(Audoux et al., 2023). If this occurs, the automated lid will reopen to sample the polluted air**
533 **masses**. In application to trace pollutants, this also reduces methodological sample preparation
534 time as it decreases the extent to which additional handling steps, like solid-phase extraction, are
535 required prior to analytical determinations.

536 The six replicate measurements used in each array provide a means of assessing sampling
537 reproducibility (e.g., canopy TF has expected heterogeneity) and for multiple analyte classes to be
538 targeted. Various analytes, with different chemical properties and/or contamination considerations,
539 can be targeted by changing the materials used for the components that encounter the sample (i.e.,
540 lids, funnels, and sample holding containers). Replicate collection can also allow for selective
541 sample preservation when quantifying deposited chemical species that may be reactive, volatile,
542 or biologically transformed. The modularity of the overall system design also allows the collection
543 units to be dismantled entirely and easily reassembled on-site, minimizing logistical issues and
544 costs for transport to remote regions. Lastly, these collection units are cost-effective. We were able
545 to produce four arrays, each consisting of six collection units, at a fraction of the cost of a single
546 equivalent commercial off-grid automated precipitation sampling unit.

547 With the majority of commercial precipitation samplers requiring a source of electricity,
548 on-grid sample collection necessitates high infrastructure costs and/or samplers being positioned
549 closer than desired to point sources of anthropogenic pollution. As a result, especially in remote
550 locations, site selection becomes heavily restricted and expensive when factoring in all the
551 standard criteria, particularly with respect to the need for an easily accessible power source. Thus,

552 the off-grid capabilities of our samplers lend dexterity to these systems and makes deposition
553 sampling that follows standard siting guidelines, like those of CAPMoN or NADP but without
554 power, more accessible to the global atmospheric research community (Vet et al., 2014). To further
555 highlight and validate their capabilities, a series of fundamental performance parameters were
556 collected and are discussed in detail in the sections that follow.

557

558 **3.2 Power Consumption and Performance Testing**

559 **3.2.1 Power Consumption of Instrumental Setup**

560 The simplicity of the automated precipitation samplers allows for low power consumption
561 during operation, which is particularly important for off-grid operation. The motors operating and
562 rain sensor heating during active precipitation are the most energy-intensive elements of the system
563 (Table 2). The integrated contribution of the motor over a month-long sampling period is however
564 negligible compared to other components, since it is operational for short periods of 5 to 10
565 seconds with a current usage of only 38 mA. The continuous need to provide 5 VDC to the digital
566 logic via step-down from 12 VDC is actually the largest power consuming component of the setup
567 in the absence of rain. When the samplers are in the closed position, under rain-free conditions,
568 the power consumption of the entire array is 4.66 Watts (W) and 2.86 W for transformed 115 VAC
569 and battery 12 VDC supplies, respectively. The provision of 12 VDC to the board with a
570 transformer for the 115 VAC application results in greater total power requirements. These values
571 increase to 10.00 W and 5.04 W with the detection of a conductive liquid on the precipitation
572 sensor as it heats the sensor surface to capture the active period of the event. Based on the measured
573 power consumption, a fully charged 103 Ah AGM battery would provide at most 447 hours (or 18
574 days) in standby mode under rain free conditions and 294 hours (or 12 days) if the heated surface
575 of the sensor is in continuous use (Table 2). The lower range limit is unlikely since the sensor only
576 operates for the duration of a rain event, after which the battery is available for solar top-up again.
577 In the fieldwork conducted here, battery life was extended through the addition of 40 W solar
578 panels to the systems. The entire array was confirmed to be operational at the end of monthly (SR,
579 PB, and GC) and two month (ER) integrated sampling periods on an ongoing basis, prior to
580 exchange with a new fully charged battery, for two years.

581

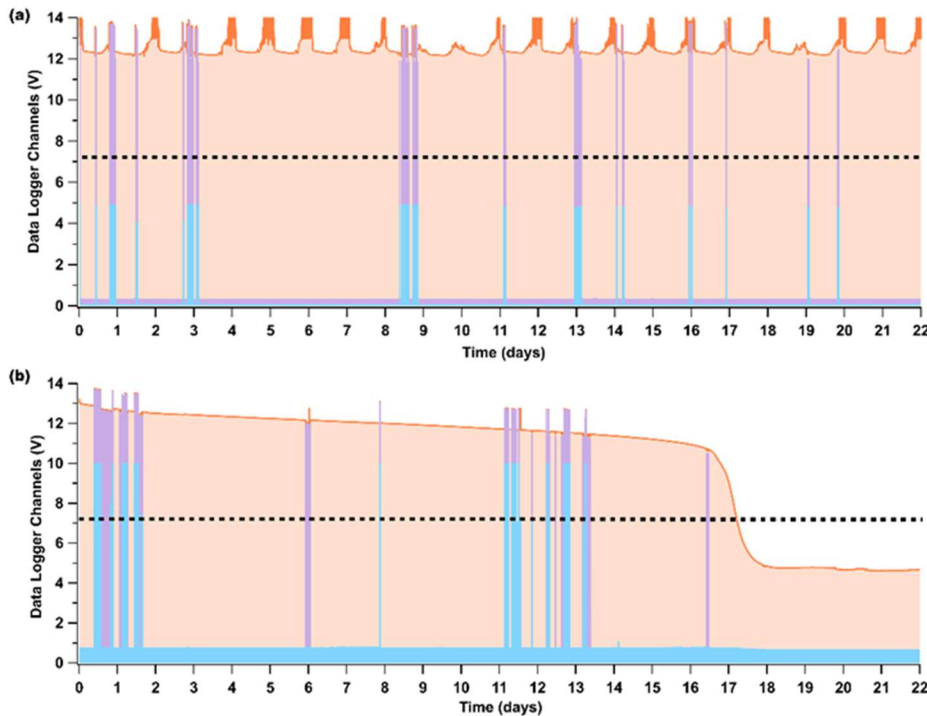
582 **Table 2.** Measured voltage, current, and power consumption of the rain sensor and circuitry in
 583 both the idle and maximally operational state when connected to a 12 VDC battery or transformed
 584 115 VAC. Total power demand was measured for wet and dry sensor scenarios.

Parameters	Rain Sensor		AC Outlet		DC Battery		Total			
	Idle	Active	Idle	Motors	Idle	Motors	AC Outlet		DC Battery	
			Board	In-Use	Board	in-Use	Dry	Wet	Dry	Wet
Voltage (V)	12 DC	12 DC	114 AC	110 AC	12 DC	12 DC	-	-	-	-
Current (A)	0.008	0.120	0.040	0.078	0.230	0.300	-	-	-	-
Power (W)	0.10	1.44	4.56	8.58	2.76	3.60	4.66	10.00	2.86	5.04

585
 586 In comparison to two commercial samplers used by national monitoring networks, the
 587 power requirements of our new samplers are substantially lower. The first commercial sampler we
 588 reviewed draws a maximum of 2 A, with a ceramic heater housed within the sampler case that
 589 draws 0.8 A constantly, resulting in an upper limit power demand of 230 W (at 115 VAC) and a
 590 lower limit of 92 W. The commercial sampler can be upgraded to utilize a thermostated space
 591 heater for winter operation, drawing an additional 4.2 A (480 W), resulting in a maximum power
 592 demand of about 800 W when using a 115 VAC power supply. A second commercial precipitation
 593 sampler reviewed is used by national monitoring networks and draws approximately 5 A, resulting
 594 in a power requirement of 575 W at 115 VAC. The commercial and standard precipitation samplers
 595 for deposition monitoring programs have much higher power requirements compared to those
 596 presented in this work. The commercial samplers utilize 80 to 100 times more power. With our
 597 lower power requirements, the new automated samplers prove to be advantageous in both on- and
 598 off-grid sampling yet are disadvantaged in being unable to collect snow in the winter.

599
 600 **3.2.2 Precipitation Sampler Performance Tests and Data Logging**

601 In addition to low power consumption during precipitation sampling, a supplied battery
 602 can obtain constant power renewal when outfitted with a solar top-up that is kept exposed to
 603 sunlight by proper orientation. At NL-BELT, adjustments were made for this during each site
 604 visit during sample collection. During the solar top-up tests below, voltages of the sensor and
 605 batteries were consistently monitored. Over a test period of 22 days, no appreciable decline in
 606 battery performance of a 76 Ah unit was observed despite the detection of more than 10 rain
 607 events during that period (Figure 3a).



609

610 **Figure 3.** Performance of off-grid precipitation samplers during sample collections from (a) 13
 611 July to 7 August 2018, using a 76 Ah battery and solar panel top-up and (b) 22 January to 13
 612 February 2019 with a 103 Ah battery and no solar panel. Battery voltage (shaded orange) is
 613 elevated above 12 VDC when charging, or decreases over time when no solar panel is used and
 614 precipitation is sensed/collected. The 12 VDC rain sensor relay signal (purple) and the open
 615 sampling lid switch voltage (blue) indicate active periods of detected precipitation. The black
 616 dashed line indicates the 60% efficiency cut off, 7.2 V, at which the battery should be recharged.
 617

618 In comparison, winter sampling with these devices is not recommended without substantial
 619 investment in a sufficient power density provided high-performance cold weather batteries. The
 620 lack of sunlight during winter at higher latitudes also negates the use of effective small scale solar
 621 top-up. Our tests show that when the samplers were deployed without a solar backup under snow-
 622 free winter conditions (temperatures ranging from -17.8 to 7°C), with a 103 Ah battery, the off-
 623 grid system only lasted for 17 days. At this point, the larger capacity battery was fully depleted by
 624 frequent snow and rainfall – probably due to the heated precipitation sensor requiring additional
 625 energy to phase change snow and ice to water and then to evaporate that water. This depletion
 626 occurred despite housing the battery in an insulated enclosure during the test. In addition, on days
 627 6 and 16, the precipitation sensor relay was activated but the lid did not rotate to the open position
 628 (Figure 3b, blue trace). This could have been because the precipitation event was not intense

629 enough for the lid to open fully and trigger the 5 V lid open switch or because of snow and ice
630 buildup around the lids resulting in them being unable to physically open. Overall, these samplers
631 may be possible to deploy during the winter if line power can be supplied. Such a deployment
632 would further necessitate that the sampling funnel be heated to render a liquid sample for collection
633 in the jugs in addition to the sensor chute to prevent snow and ice accumulation. A heated funnel
634 would also prevent snow or ice accumulation on top of the automated lids. Together, such power-
635 hungry requirements for winter operation exceed simple off-grid use with a battery package that
636 is easily transported into and out of remote field sites.

637

638 **3.3 Comparison of Sample Collection Volumes**

639 The automated samplers were colocated with total deposition samplers and deployed across
640 the experimental forests of four NL-BELT regions during the 2015 and 2016 growing seasons to
641 observe deposition trends. In addition, we compare these observations to the long-term climate
642 normals reported by ECCC and estimated deposition at 1 km x 1 km resolution from the DAYMET
643 reanalysis model (Table 1, **Section S2**). Three automated samplers were deployed in the open to
644 collect incident precipitation (OF) and another three under the experimental forest site canopy
645 (TF). The mean OF volumes of triplicate measurements from south to north were 1.42, 1.38, 1.31,
646 and 0.79 L, whereas the corresponding TF volumes were generally similar in magnitude at 0.96,
647 0.98, 1.02, and 1.13 L, for the 2015-16 sampling period (Figure 4). It is evident that the volume of
648 precipitation decreased as latitude increased for OF samples, whereas the opposite relationship
649 was observed in TF samplers, although the absolute volumes are more comparable in magnitude.
650 The total deposition volumes collected were as expected, decreasing from south to north in
651 agreement with the expectations from the long-term normals and comparable to the estimates from
652 the DAYMET model (Table 1), where the largest integrated volume of precipitation was collected
653 at the lowest latitude (GC) and a lower amount at the highest latitude (ER), with the intermediate
654 sites (HR and PB) having the lowest inputs overall during this observation period. Total annual
655 deposition volumes collected by our deployed samplers from south to north in 2015 were 39.5,
656 39.4, 31.9, and 17.5 L, while in 2016 they were 51.7, 37.8, 32.8, and 34.2 L. Total deposition
657 volume collected from HR was used for comparison to automated sample volumes collected at PB
658 in 2015, as they both share the same watershed. This approach had to be taken, as the HR site was
659 initially planned for full experimental use before becoming inaccessible in early 2015. The relative

660 error between the two sites for samples collected in 2016 was $\pm 15\%$ (24.6 L in PB and 32.2 L in
661 HR), comparable to the reproducibility we observe for replicates collected within a given site (see
662 below). The total deposition samplers were installed in HR in late 2014 and the automated samplers
663 were then set up at PB. Despite this, there is good agreement between the trends in predicted
664 deposition values by DAYMET with the measured values, although the absolute amounts from
665 these are systematically lower in all of our observations (Section S2). Regardless, by following
666 the recommended siting criteria from the NADP and CAPMoN as best as possible, the very strong
667 agreement of our temporal trends at both annual and monthly timescales with both comparators
668 demonstrates the suitability of the total deposition samplers and, therefore, the automated samplers
669 for use in quantifying deposited chemical species of atmospheric interest into the experimental
670 sites.

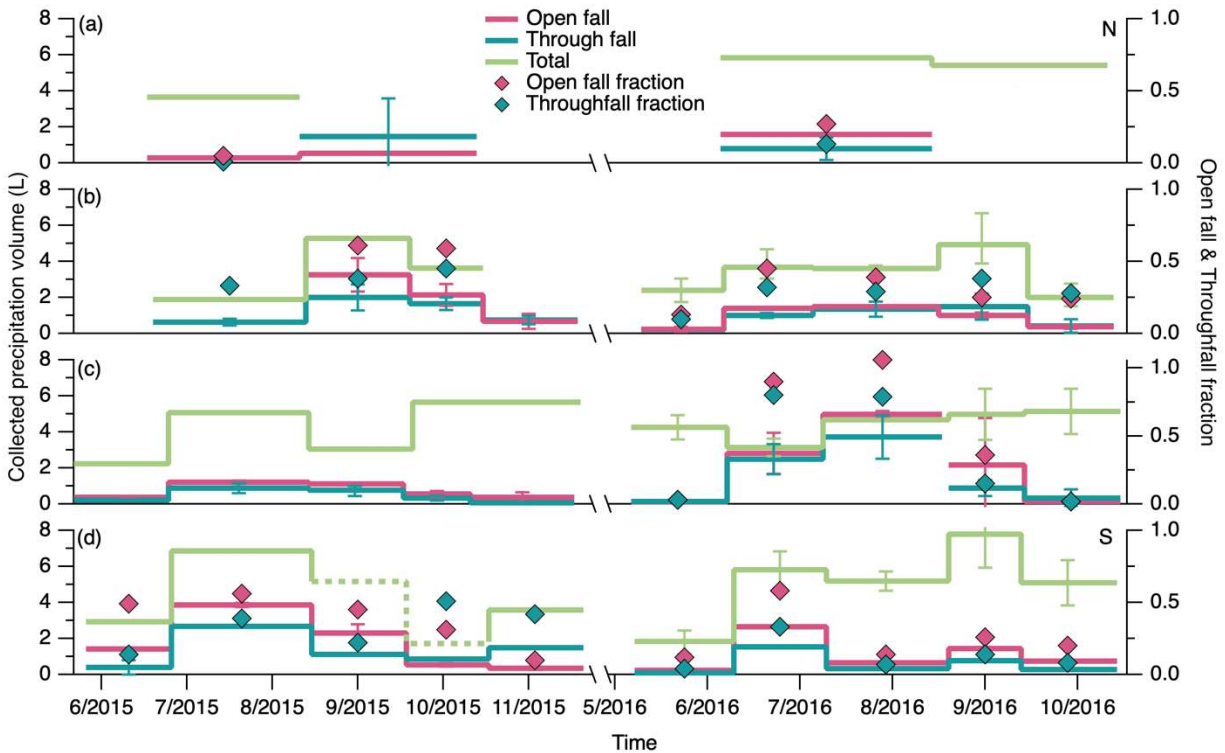
671 The wet deposition volumes collected for the snow free period using the automated
672 precipitation samplers did not follow the trends in total deposition (Figure 4), as might be expected
673 (e.g., due to pollutant loading, rainfall quantity/rate, and scavenging processes). For the 2015
674 collection period from June through October, the summed volumes of OF precipitation, from south
675 to north across the NL-BELT, were 25.4, 10.9, 20.4, and 2.2 L, while in 2016 they were 17.3, 30.4,
676 13.5, and 5.1 L. While the total and OF fractions would typically be much closer to unity in
677 more polluted regions, it would be expected in these remote NL-BELT field sites for the
678 differences to be driven by complex, non-linear processes that cannot be easily disentangled.
679 Here we present three reasons as to why the measured wet OF deposition volumes do not follow
680 the total deposition trend across the transect. First, these samplers are designed specifically to
681 collect only conductive precipitation (i.e., containing conductive atmospheric compounds) not
682 total/bulk precipitation. As a result, the OF wet deposition volume collected across the sites is
683 mostly below 50% of total volumes collected, while TF volumes are similar in magnitude or lower
684 than that of OF (Figure 4). The wet deposition fraction collected was variable within and between
685 regions, sometimes less than 10%, despite large volumes collected in total and presumably due to
686 intense atmospheric washout that this region is well-known for. Second, the NL-BELT total
687 deposition trend estimated using the ECCC long-term climate normals represents a 30-year period
688 (Bowering et al., 2022) while the automated volume measurements here represent two years of
689 targeted conductive precipitation collection. The combined summed volumes of targeted
690 conductive wet deposition across the 2015 and 2016 field seasons were 42.7, 41.3, 33.9, and 7.3

691 L, somewhat better reflect the expected precipitation trends within the transect (Table 1). Lastly,
692 our monthly automated wet deposition sample collection periods occurred from June through
693 November and so it is temporally incomplete with respect to the substantial amount of precipitation
694 volume deposited as snow delivered during the winter (Table S3). The discrepancies between the
695 long-term trends and our shorter-term observations therefore make sense as they are sensitive to
696 interannual changes in synoptic scale transport and rainwater solute loadings, as exemplified by
697 the volumes collected in SR in 2015 (Figure 4b) and PB in 2016 (Figure 4c). Overall, for the
698 automated sampler observations on a per-year basis, there is no consistent trend between site
699 latitude and the volume collected in either OF or TF. This is unsurprising as they are dependent
700 on the conditions that drive the rate of atmospheric wash-out and presence of conductive solutes.

701 The automated OF wet deposition volumes collected each year have peak values that range
702 from 1 to 4 L with an overall variability of $\pm 33\%$ for any triplicate of samples across the entire
703 dataset. Across our 33 sample collection periods, our replicate relative standard deviations (RSDs)
704 follow a log-normal distribution where volume reproducibility is typically within $\pm 12.5\%$ and
705 almost always within $\pm 31.5\%$ (Figure S11). A few outliers with higher variability skew the overall
706 view of volume precision. Out of 33 OF samples collected, **10** have RSDs greater than 40.5% with
707 2 of those **10** having RSDs greater than 100%. **Those values greater than 40.5% had no**
708 **systematic relationship with site or time of time. Wind speeds were considered as a possible**
709 **source of variability. The prevailing winds over Atlantic Canada are known to be**
710 **southwesterly in the summer – intensifying during the autumn months – and westerly to**
711 **northwesterly in the winter (Bowver, 1995; Jacob, 1999; Randall, 2015). Strong wind speeds**
712 **(i.e., >100 km/hr) could occur on an event basis during any time of the year and, thus, could**
713 **contribute to the variability seen at each field site.** Wind is known to generate bias in gauge-
714 based precipitation measurements where unshielded precipitation gauges can catch less than half
715 of the amount of a shielded gauge (Colli et al., 2016). A windscreen design for obtaining rainfall
716 rates – and thus, volumes – to be more reproducible could be considered in future deployments of
717 our developed samplers, similar to recently reported innovations for smaller rainfall rate devices
718 (Kochendorfer et al., 2023). This would, however, increase costs and logistical considerations in
719 deploying the developed devices which currently operate synonymously to deposition systems
720 employed by government monitoring programs. **Our siting approach is consistent with these,**
721 **which often deploy a single sampler without wind protection. Thus, by employing replicates,**

722 **we are able to ascertain the environmental variability.** In addition, collection of replicate
723 samples allows our observations to span a wider physical area, reducing the impact of confounding
724 variables such as wind speed in comparison to a more typical sample size of one for many field
725 collections. Imperfect siting and lack of shielding is necessary where remote field sampling
726 prevents the setup of such infrastructure. As a result, the deployment of triplicate samplers
727 provides researchers with a better opportunity to implement quality control as they can reduce bias
728 in the event of dynamic OF. While the effect of wind is reduced, additional factors can drive
729 variability when the samplers are placed under a forest canopy for TF collection.

730 To demonstrate canopy dynamics impacting interception volumes within the sampling
731 sites, the ratio of throughfall to open fall (TF/OF) volume was compared amongst our total pool
732 of 31 samples. This group of samples encompassed the monthly average TF/OF values for each
733 set of triplicate samplers, at all four sites, from 2015 to 2016. These measurements were then split
734 into two separate populations – samples that have a TF/OF less than one (n=24) and those that
735 have a TF/OF greater than one (n=7). The samplers were positioned identically between years and
736 no single sampler was reproducibly found in the second population. In the first population, the
737 fraction collected was $56 \pm 21\%$ (ranging from 19 to 88%), likely due to the known processes of
738 canopy and stem interception (Eaton et al., 1973; Howard et al., 2022). For example, in two young
739 balsam fir-white birch mixed forest stands, the amount of precipitation intercepted by the forest
740 canopy, in similar snow-free conditions, was estimated to be $11 \pm 5\%$ (Hadiwijaya et al., 2021).
741 In mature boreal forests, 9% to 55% of rainfall can be intercepted by the canopy (Pomeroy et al.,
742 1999). Relevant to deposition of atmospheric constituents, Pomeroy et al. (1999) also reported that
743 up to 70% of intercepted rainfall may evaporate directly from the canopy, which can leave behind
744 non-volatile rainfall solutes. Wet deposition that undergoes stemflow (SF) proceeds down the
745 branches, stems, and/or trunks of a plant, transferring precipitation and nutrients from the canopy
746 to the soil at the trunk or stem base (Ciruzzi and Loheide, 2021). These known mechanisms of
747 canopy interception ultimately reduce the amount of precipitation reaching the ground as TF, and
748 thus, this explains the smaller volumes found in our samplers compared to the OF measured
749 simultaneously. In contrast, the fractions that ranged from 108% to 424%, averaging 186%,
750 demonstrates a different aspect of the highly dynamic nature of canopies where they can sometimes
751 intercept rainfall like an impermeable surface to act as a funnel, guiding large volumes of
752 precipitation on to the ground, or in this case into the TF samplers (Metzger et al., 2019).



754
 755 **Figure 4.** Average volume collected from replicate automated samplers deployed from June 2015
 756 to October 2016, from north (N) to south (S), at the NL-BELT field sites: **(a)** ER, **(b)** SR, **(c)** PB,
 757 and **(d)** GC. The red trace represents open fall, teal for throughfall, and light green for total
 758 deposition (the sum of conductive and non-conductive precipitation). The total precipitation
 759 volumes depicted for PB, from July 2015 to November 2015, were collected at the nearby HR site
 760 in the same watershed since no total deposition measurements were in place at PB during this
 761 period. The missing volume for GC in 2015 was estimated from the determined ECCC station
 762 linear relationship and is presented as a broken line. The fraction of precipitation collected as open
 763 fall or throughfall, compared to the total deposition (right axis), are represented by diamonds of
 764 the corresponding color. Error bars represent the standard deviation of three measurements from
 765 replicate samples. The axis break spans the winter months when the off-grid automated samplers
 766 were stored.

767 768 3.4 Characterizing Chemical Parameters from NL-BELT

769 In addition to assessing physical parameters, chemical parameters were also evaluated in
 770 this work. Conductivity and pH are measurements commonly made on precipitation samples
 771 collected from the field and so incorporating them into our analysis is useful for instrumental
 772 validation. Additionally, with increasing recognition of their importance as a proxy for ROC
 773 estimation, and in biogeochemical carbon budget closure, DOC flux measurements were used to
 774 compare against a limited number of prior reports, each using different sampling or data
 775 interpretation strategies. These chemical measurements were also made in an underrepresented

776 part of the world in terms of atmospheric deposition sampling and are useful additions to the
777 overarching study of precipitation chemistry.

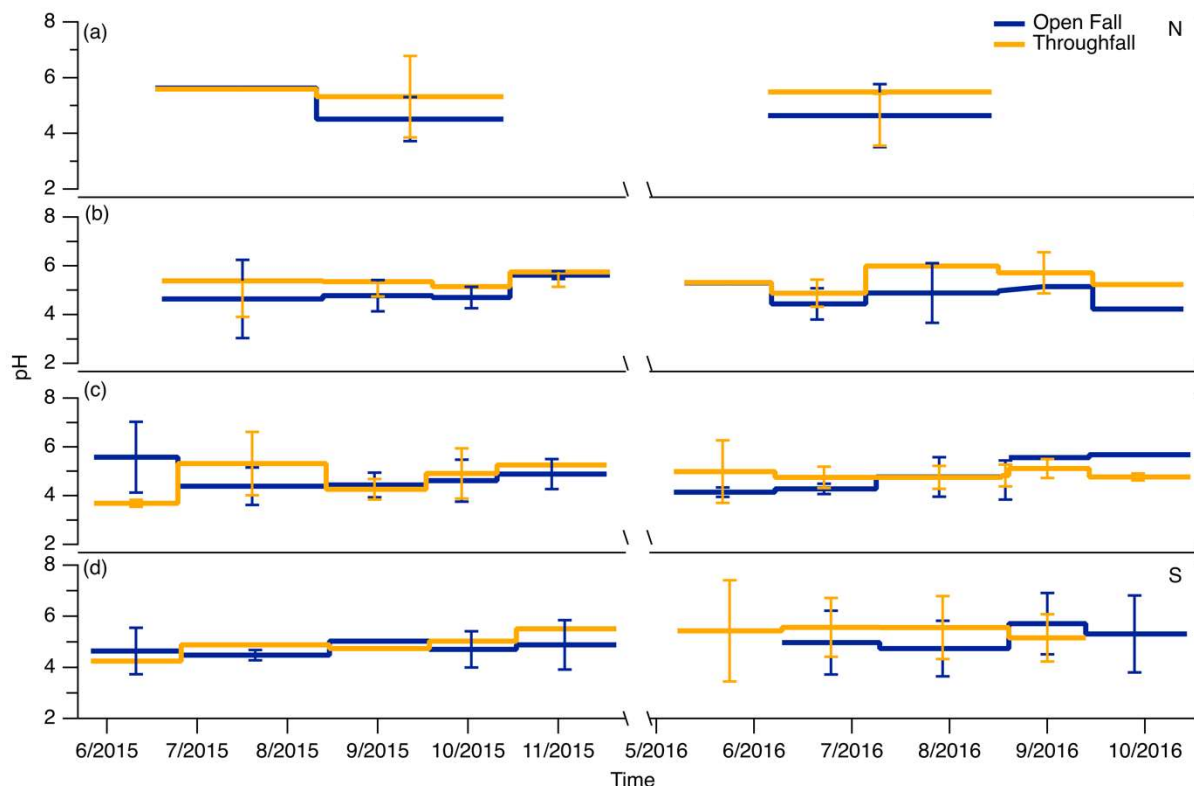
778

779 **3.4.1 Precipitation pH**

780 The deposition of atmospherically persistent pollutants and biogeochemically relevant species
781 to the Earth's surface, or even NO_3^- and SO_4^{2-} historically, can affect the environmental health of
782 soil, air, and water. With the pH range of natural rainwater in equilibrium with atmospheric CO_2
783 expected to be between 5.0 to 5.6, acid rain is defined by values lower than this (Han et al., 2019).
784 Traditionally, the extent of acidity depended on the intercepted atmospheric concentrations of
785 HNO_3 and H_2SO_4 . In any case, monitoring acidity and deposition is especially relevant in remote
786 regions, where major uncertainties and gaps in deposition measurements and global ion
787 concentrations exist (Escarré et al., 1999; Vet et al., 2014). A change in pH can modify the
788 chemical state of many pollutants, altering their transport, bioavailability, and solubility (Guinotte
789 and Fabry, 2008). For example, this can increase exposure and toxicity of metals and nutrients in
790 marine habitats which can go undetected for longer periods in remote areas.

791 Most TF samples were observed to have slightly higher pH than those from OF which had pH
792 values ranging from 4.14 to 5.71 (Figure 5, Table 1). The TF precipitation pH, on average, ranged
793 from 4.74 to 5.99 with rare exceptions falling outside of that range (e.g., July and September 2015
794 PB pH of 3.69 and 4.26, respectively, and the July 2015 GC with pH of 4.12). Excluding these
795 exceptions, there are no major variations observed spatially between the four sites, or temporally
796 between seasons or years (Figure 5). The pH values reported at each of the NL-BELT field sites
797 are comparable to recent OF measurements made at CAPMoN sites in Nova Scotia and
798 Newfoundland and Labrador, where the reported pH of precipitation ranged from 4.44 to 5.19
799 (Houle et al., 2022). The more basic TF overall is expected, as it has been found that up to 90% of
800 H_3O^+ in precipitation can be absorbed by leaves while passing through the canopy (Cappellato et
801 al., 1993). Foliar leaching, the release of ions from leaves, has been commonly reported for base
802 cations such as Mg^{2+} , K^+ , and Ca^{2+} while being minimally observed for other ions such as Cl^- and
803 SO_4^{2-} (Carlson et al., 2003). Mechanisms for foliar leaching include passive cation exchange of
804 H_3O^+ with, for example, cells in the interior of the leaf (Burkhardt and Drechsel, 1997).
805 Additionally, alkaline dust – deposited on the leaves of the canopy, can decrease the acidity of TF
806 precipitation. Such dust can accumulate on leaf surfaces as a result of anthropogenic (i.e., industrial

807 processes) or natural (i.e., wind erosion) sources (Csavina et al., 2012), so that precipitation
 808 passing through the canopy can interact with it (e.g., CaCO_3); thus, neutralizing acidic species and
 809 increasing the TF pH observed in our automated samplers.
 810



811
 812 **Figure 5.** Average pH values from replicate samples collected at each NL-BELT field site, from
 813 north (N) to south (S), at (a) ER, (b) SR, (c) PB, and (d) GC, from June 2015 to August 2016.
 814 Open fall collections are represented using the solid blue trace whereas the orange trace is the pH
 815 of the precipitation collected as throughfall under the balsam fir canopy.
 816

817 The pH of the collected precipitation appears to be similar in both TF and OF as a function of
 818 time – despite the potential for foliar leaching and dust dissolution in the canopy. The same
 819 chemical components may be setting the pH, as these measurements do not vary much seasonally,
 820 geographically, or temporally. As pH is a long-studied measurement, its purpose in this work was
 821 to validate the sample quality from our described collection approach, rather than drive any
 822 scientific objective. Nevertheless, while the NL-BELT measurements demonstrate a recovery
 823 compared to rainwater pH in 1980s eastern North America – prior to NO_x and SO_2 regulation (pH
 824 from 4.1 to 5.0; Barrie and Hales, 1984), the present-day pH remains lower than expected for
 825 natural rainwater (~5.6; Boyd, 2020). Keeping in mind the successful environmental policies
 826 limiting SO_2 and NO_x , leading to considerable decreases in atmospheric concentrations of H_2SO_4

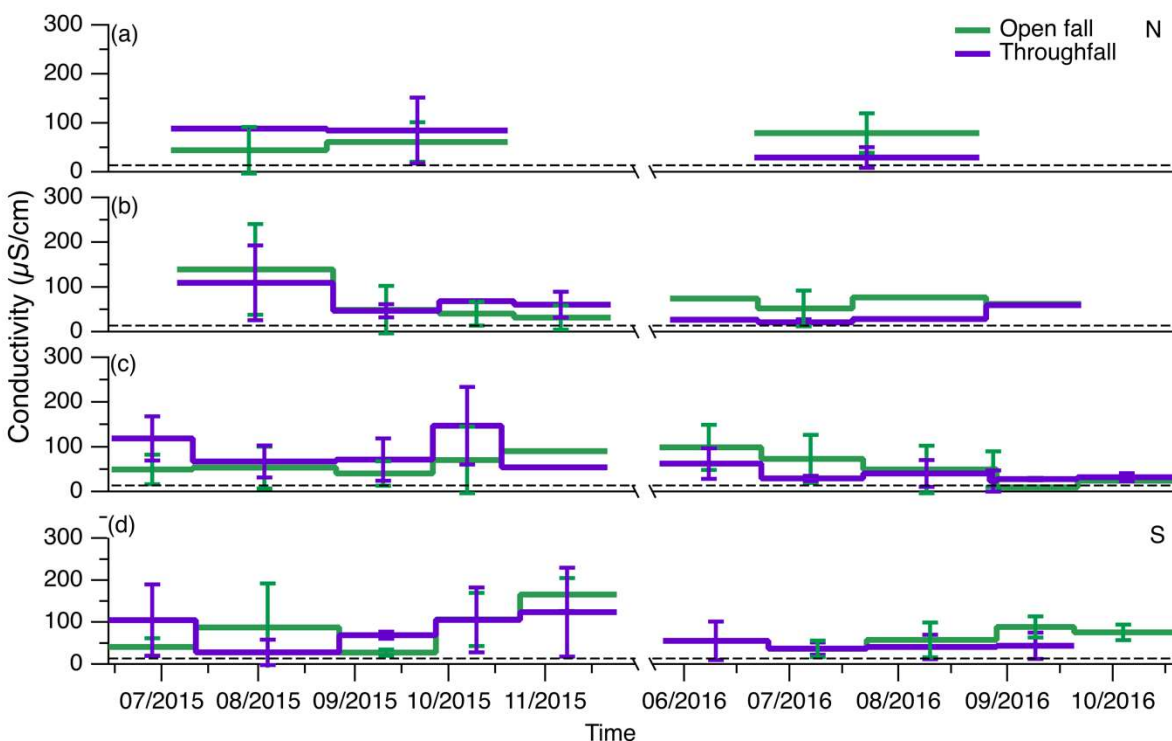
827 and HNO₃, a modern view on the trajectory of continental U.S. cloud water composition and pH
828 has recently been reported (Lawrence et al., 2023). Across the U.S. and eastern Canada,
829 measurements of anion molar charge equivalents have been lower than cations – a potential
830 explanation being an increase in the presence of weak organic acids which commonly have pKa
831 values near 4 (Feng et al., 2021), an outcome we have also observed in aerosol sample chemical
832 composition from Atlantic Canada (Di Lorenzo et al., 2018). With the frequency of acid rain
833 having a pH < 5 decreasing over the past 20 years, these recently reported measurements depict
834 deposition composition shifting away from a ‘linear’ chemical regime dominated by H₃O⁺ and
835 SO₄²⁻ towards a ‘non-linear’ regime designated by low acidity, moderate to high conductivity,
836 potentially weak acid-base buffer systems, and increasing base cation and TOC concentrations
837 (Lawrence et al., 2023). It would seem the evolving chemical contributors to global rainwater pH
838 remain an open line of investigation.

839

840 3.4.2 Precipitation Conductivity

841 In all the collected OF and TF precipitation samples, across all four NL-BELT sites, the
842 average measured conductivity values ranged from 21 to 166 µS/cm **following no apparent**
843 **seasonal or temporal trend** (Figure 6). **Additionally, the conductivity in both OF and TF also**
844 **appear to vary across the field sites - only within the 2016 TF samples does the conductivity**
845 **appear to increase with decreasing latitude.** Yet, with the typical conductivity of surface and
846 drinking waters being between 1 to 1000 µS/cm (Lin et al., 2017), and typically below 200 µS/cm
847 in stream water measurements within the watersheds of each of the NL-BELT sites, our
848 observations are comparable and fall within the expected range. Our field blanks – encompassing
849 a variety of materials and apparatuses, and our cleaning procedures, routinely produced
850 conductivities of 9 ± 5 µS/cm. **The conductivity of saturated HgCl₂ in water (at 0.1% vol/vol) was**
851 **13.6 ± 0.4 µS/cm, which is also comparable to but statistically higher than our field blanks (p =**
852 **0.0015; unpaired t test) and less than what was observed for our samples (p < 2 x 10⁻⁶ for each**
853 **site considered separately and also across all sites; unpaired t-test).** Even with this background
854 correction applied, the conductivity values presented here are expected to be similar to or higher
855 than what would typically be found in rainwater (4 to 150 µS/cm; Beverland et al., 1997) as the
856 rain sensor deliberately selects for precipitation containing ionic chemical components with
857 conductivity greater than 1.0 µS/cm, while excluding pure water during atmospheric washout,

858 which would dilute the dissolved solutes in the wet deposition sample and lower the resulting
 859 conductivity values. The overall comparability between our range and those previously reported,
 860 where the lower limit is slightly higher in our dataset, demonstrates that the principle of operation
 861 of our instrument is robust. It decisively collects precipitation with the property of conductance
 862 indicating dissolved ionic solutes of interest to atmospheric chemical processes.
 863



864
 865 **Figure 6.** Average conductivity measured from replicate automated samplers at the NL-BELT
 866 field sites, from north (N) to south (S), at (a) ER, (b) SR, (c) PB, and (d) GC, from June 2015 to
 867 October 2016. The green trace represents open fall samplers whereas the purple trace represents
 868 throughfall samples. The error bar represents the standard deviation between replicate
 869 measurements. The dashed black line represents the upper threshold of conductivity (13.6 $\mu\text{S}/\text{cm}$)
 870 that arises due when an addition of saturated aqueous HgCl_2 is made to microbially sterilize
 871 samples. Note that all samples have conductivities equivalent to or higher than 13.6 $\mu\text{S}/\text{cm}$.
 872

873 3.4.3. Wet Deposition of Dissolved Organic Carbon (DOC) at NL-BELT

874 The concentration of DOC in OF and TF precipitation, across all four sites, ranged from 3
 875 to 46 mg L^{-1} and 5 to 65 mg L^{-1} with averages of $16 \pm 10 \text{ mg L}^{-1}$ and $22 \pm 12 \text{ mg L}^{-1}$, respectively
 876 (Table 3). Concentrations are influenced by the volume collected and are not useful when
 877 discerning deposition trends and/or mechanisms. The concentrations were converted to elemental
 878 fluxes using the volume of precipitation collected, the area of the funnel and the number of

879 sampling days of each sampling period (Figure 7). **The total flux for each sample period was**
880 **summed and reported as an equivalent annual flux with the following units: mg m⁻² a⁻¹.**
881 **Annual fluxes** ranged from 600 to 4200 mg C m⁻² a⁻¹ across the study sites for the snow free period
882 (Table S4).

883 The TF DOC fluxes were enhanced compared to the corresponding OF samples as
884 precipitation was intercepted by the forest canopy, with fluxes higher in TF samples by 600, 400,
885 and 400 mg C m⁻² a⁻¹ at GC, SR, and ER, respectively (Table S5). The accumulation of water-
886 soluble organics on forest canopies that increases DOC detected in TF could originate in part from
887 organic carbon-containing compounds aged through oxidation reactions in the atmosphere, which
888 increases their water solubility and propensity for surface interactions. In periods without
889 substantial rain, these oxidized organics deposit effectively to the high surface area of forest
890 canopies, contributing to the elevated DOC measured in TF. Additionally, non-volatile organics
891 left behind from evaporated precipitation intercepted by the canopy could also contribute.
892 Conversely, other mechanisms within the forest could result in enhanced DOC in TF. Recently,
893 Cha et al. (2023) utilized a mass balance approach to determine whether DOC deposition is driven
894 by canopy leaching (i.e., soluble tree resin, leaf exudates, internal tissues and microbes) or
895 dissolution of dry deposited gases and PM_{2.5} on plant foliage into rainwater. It was found that
896 canopy leaching is the major contributor to TF DOC, accounting for ~83% of throughfall DOC.
897 Whereas, PM_{2.5} and rainwater only accounted for ~3 and 14%, respectively, while dry deposited
898 gases were not considered. This suggests that internal cycling of DOC within the forest could be
899 an important source of DOC to the throughfall soil interface (Cha et al., 2023). It is possible that a
900 similar mechanism may be responsible for the elevated levels of DOC in TF at the NL-BELT sites,
901 but we cannot explicitly distinguish between internal cycling versus external deposition in the
902 current study.

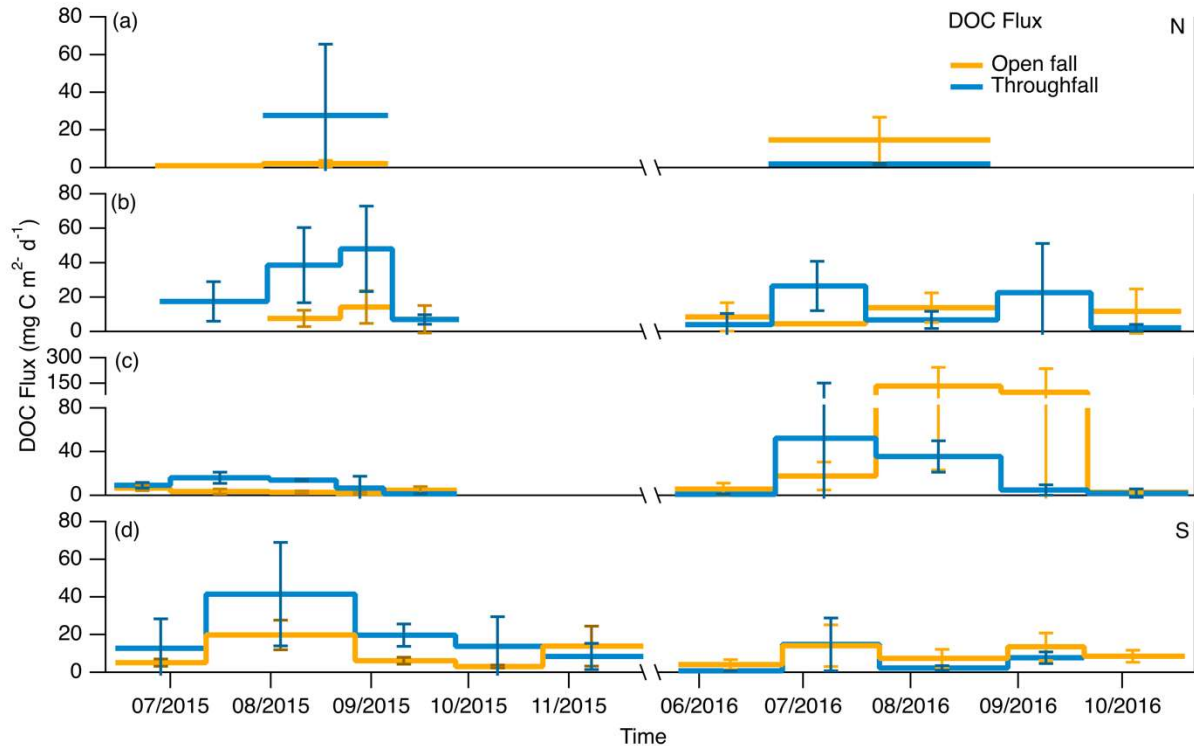
903 A notable exception was observed at PB, where the DOC fluxes in the open fall sample
904 were enhanced up to 1800 mg C m⁻² a⁻¹ when compared to the TF in 2016. This may be attributed
905 to a difference in forest type within this NL-BELT region being black spruce (*Picea mariana*)
906 instead of balsam fir (Bowering et al., 2023). Some studies have suggested that forest type could be
907 a major factor affecting DOC variability (Arisci et al., 2012; Sleutel et al., 2009). Specific
908 differences in canopy height, leaf area index, canopy structure and the shape of leaves and needles
909 could drive DOC differences between forest types (Smith, 1981; Erisman and Draaijers, 2003;

910 Sleutel et al., 2009). The elevated levels in OF samples relative to TF within PB are consistent
911 with idea of uptake and/or leaching of canopy DOC in the internal cycling of DOC, while the
912 enhanced TF at the rest of the sites is more difficult to observational constrain the participating
913 processes.

914 Episodic events, such as polluted air masses from wildfires could also result in elevated
915 deposition of DOC. It is estimated that $\sim 116 - 385 \text{ Tg C a}^{-1}$ is produced globally due to the
916 incomplete combustion of biomass during landscape fires (Santín et al., 2016; Coward et al., 2022).
917 Several studies have associated enhanced DOC levels with wildfires (Gao et al., 2003; Moore,
918 2003; Wonaschütz et al., 2011; Myers-Pigg et al., 2015). More recently, Coward et al. (2022)
919 measured DOC in Pacific surface waters along the California coastline and observed 100 to 400
920 % increases in DOC concentration, when compared to pre-wildfire conditions. It is possible that a
921 similar biomass burning plume that underwent atmospheric washout, could be responsible for the
922 enhancement in the observed DOC at NL-BELT, overlaid on a background more typical of
923 seasonal oxidation of biogenic DOC. This also coincides with the seasonal variability observed in
924 OF samples from the same summer where elevated levels of DOC were measured. For instance,
925 the DOC deposition at PB for August 2016 was 4800 mg C m^{-2} , whereas the total deposition for
926 the same year was $7800 \text{ mg C m}^{-2} \text{ a}^{-1}$. This single period accounts for 62% of the total DOC
927 deposition at this site. This underscores the pivotal role that episodic transport may play in
928 influencing the dynamics of DOC deposition, particularly with a warming future where wildfires
929 are more prevalent.

930 The deposition trend observed in the current study also highlights the complexity of the
931 varied drivers of atmospheric ROC, in which some months have more DOC in TF versus OF and
932 occasionally the opposite is observed. Generally, we observed similar fluxes in both samples –
933 suggesting that the amount of deposited carbon is comparable. Although the volume of
934 precipitation captured in TF samplers are generally lower when compared to the corresponding
935 OF samplers, the deposition flux of DOC is greater in TF samplers. With DOC enhanced in TF
936 samples, the values reported here could be an underestimation of the amount of carbon reaching
937 the forest floor during precipitation events due to competing processes within the canopy. One
938 such process is stemflow (SF), where a fraction of precipitation intercepted by the forest canopy
939 is funneled over the bark of the tree surface to the base of the tree stem (Oka et al., 2021). Although,
940 SF was not measured in the current study, several studies have demonstrated that DOC

941 concentrations are enhanced in SF when compared to the corresponding TF and bulk precipitation
 942 samples (Stubbins et al., 2017; Van Stan and Stubbins, 2018; Ryan et al., 2021). Additionally, we
 943 cannot rule out that the chemical speciation differs between OF, TF, and SF even if the DOC values
 944 are similar, but such insights require more selective instrumentation for chemical analysis (e.g.,
 945 high resolution mass spectrometry).
 946



947
 948 **Figure 7.** Average DOC fluxes ($\text{mg C m}^{-2} \text{d}^{-1}$) in replicate samples collected at the NL-BELT field sites, from north (N) to south (S), at (a) ER, (b) SR, (c) PB, and (d) GC, from June 2015 to August
 949 2016. The yellow trace represents samplers that were placed in the open without any obstruction
 950 whereas the blue trace represents samplers that were placed under the canopy. Error bars represent
 951 the standard deviation of three measurements from three independent samples.
 952
 953
 954

955 **The ability to accurately determine DOC in OF and TF precipitation demonstrates**
 956 **the capability of the automated deposition samplers. To validate our measurements, we**
 957 **compared our observed fluxes to other studies in different forest types.** Mean annual DOC

958 fluxes were generally similar to those reported in some other boreal forests (Table 3). These
 959 include Finland, with work in stands that consisted mainly of Scots pine (*Pinus sylvestris L.*) (mean
 960 OF 2.32; TF, 4.35 $\text{g C m}^{-2} \text{a}^{-1}$; Pumpanen et al., 2014), as well as in Mont St. Hilarie, Québec (mean
 961 OF 0.49; TF 2.05 $\text{g C m}^{-2} \text{a}^{-1}$; Dalva and Moore, 1991), which also consisted of a variety of tree

962 species such as yellow birch (*Betula allenganien*), red maple (*Acer rubrum*), and sugar maple
963 (*Acer saccharum*). Conversely, the annual fluxes were orders of magnitude lower than
964 measurements made at the University of Georgia (23 to 48 g C m⁻² a⁻¹) which has a subtropic
965 climate consisting mainly of southern live oak (*Quercus virginiana Mill.*) and eastern red cedar
966 (*Juniperus virginiana L.*) occasionally hosting dense epiphytes (Van Stan et al., 2017). This
967 highlights the potential variability to expect when measuring DOC in different forest systems, as
968 the annual DOC fluxes vary depending on factors such as climate, tree species composition, and
969 environmental conditions.

970 These results underscore the pivotal role the off-grid custom-built automated deposition
971 samplers can play in advancing scientific research, particularly in precipitation monitoring and
972 analysis. The automated system enabled long term continuous sample collection in remote
973 locations, which was previously challenging to attain due to the need for frequent human
974 intervention and resources required to regularly access these experimental forest stands. These
975 samplers also allowed us to compare DOC through replicate measurements in TF and OF samples
976 which sheds light on the potentially different DOC deposition chemistries within the NL-BELT
977 region. **The automated system better maintains the integrity of DOC in the samples. This was
978 achieved by following standard procedures for biogeochemical sample preservation (i.e.,
979 adding HgCl₂) (Argentino et al., 2023), employing a rigorous cleaning procedure, and
980 preventative design against the intrusion of forest litter which could result in a positive bias
981 for DOC in the collected precipitation.** The use of replicates also results in more robust scientific
982 conclusions and broader applicability of the results, and they can be obtained for a fraction of the
983 cost of a commercial equivalent, highlighting the contribution these automated systems are capable
984 of when applied to current precipitation monitoring. As a result, these samplers show promise in
985 the quantification of biogeochemical and anthropogenic chemical species of interest, which will
986 be visited in future manuscripts drawing from the samples presented in this dataset, and others
987 since obtained, but are beyond the scope of this manuscript in demonstrating the performance of
988 this new instrumentation.

989

990 **Table 3.** Concentrations (mg C L⁻¹) and annual fluxes (g C m⁻² a⁻¹) of DOC in precipitation (P),
991 open fall (OF), throughfall (TF), and stemflow (SF) collected at forested sites. Where volumes are
992 not available for other studies, fluxes are not possible to calculate. The values reported in the
993 current study are the estimated DOC flux for the wet deposition sampling period (~June through

994 October) for each year and therefore represents the lower limit of DOC deposition, as the dataset
 995 excludes snow.

Site	Type	Mean Concentration (mg C L ⁻¹)	Annual Flux (g C m ⁻² a ⁻¹)	References
Grand Codroy, NL, Canada (2015 to 2016)	OF	12.83	1.56	This study
	TF	23.40	2.20	
Pynn's Brook, NL, Canada (2015 to 2016)	OF	19.98	4.21	
	TF	21.24	2.44	
Salmon River, NL, Canada (2015 to 2016)	OF	16.14	1.33	
	TF	21.00	2.65	
Eagle River, NL, Canada (2015 to 2016)	OF	11.59	0.53	
	TF	28.26	0.86	
Mont St. Hilaire, QC, Canada (1987)	P	2.00	0.49	Dalva and Moore, 1991
	TF	12.13	2.05	
	SF	40.10	0.10	
Northern China (2007 to 2008)	P	2.4 to 3.9	1.4 to 2.7	Pan et al., 2010
Coulissenhieb, Northeast Bavaria (1995 to 1997)	P	2.70	-	Michalzik and Matzner, 1999
	TF	15.20	-	
Hobcaw Barony, South Carolina, USA (2014 to 2015)	P	1.20	-	Chen et al., 2019
	Pine TF	26.00	-	
	Oak TF	38.8	-	
University of Georgia, USA 2015 to 2016	TF Epiphyte Oak	17	23**	Van Stan et al., 2017
	TF Bare Cedar	20	32**	
	TF Epiphyte Cedar	54	48**	
SMEARII Site, Southern Finland (1998 to 2012)	P	3.24	2.32	Pumpanen et al., 2014
	TF	10.10	4.35	

** Estimated DOC yield for 2016 (g C m⁻² a⁻¹) where sampled storms values (g C event⁻¹) were scaled to an annual deposition value using meteorological data and a linear rainfall-DOC yield relationship.

996 4.0 Conclusions and Future Directions

997 This paper presents a cost-effective automated deposition sampler for continuous
 998 collection of precipitation. An open-source procedure and schematics for building these samplers
 999 is provided alongside the rationale for selecting the materials in the current study to target analytes
 1000 of scientific interest in wet deposition samples. These low-power systems are demonstrated in
 1001 being capable of continuous off-grid use for sample collection over two years at the NL-BELT
 1002 experimental sites, with replacement of battery power packs monthly or bimonthly, with on-grid

1003 performance also provided for comparison. The resulting systems enhance the accessibility of
1004 automated wet deposition samplers to scientists globally and this work highlights their robust
1005 performance in collecting and preserving rainwater conductivity and pH, alongside providing
1006 measurements of DOC from this understudied region that builds a broader picture of the
1007 atmosphere-surface exchange of this biogeochemical pool across the NL-BELT. **Comparability**
1008 **and complementariness of our results to well-established and current measurements of interest like**
1009 **DOC, demonstrate their efficacy and potential application to the study of processes such as**
1010 **canopy-precipitation interactions through the collection of open fall and throughfall**
1011 **replicates.** The capacity to autonomously collect wet deposition, in addition to traditional bulk
1012 deposition samples can shed light on competing wet and dry deposition processes. Should on-grid
1013 capacity suit scientific objectives, these samplers are anticipated to be possible for use year-round
1014 when paired with more power-intensive strategies to facilitate solid to liquid phase transfer for
1015 detected and collected precipitation in the winter.

1016 For the broader deposition-motivated community, the instrument design also allows for
1017 easy cost-effective modification of the number of replicate samplers, the material composition of
1018 all surfaces the aqueous samples interact with, as well as preservation strategies - depending on
1019 the analyte of interest. **For example, the lack of organic nitrogen measurements within**
1020 **universally established sampling and measurement procedures serves as a general example**
1021 **of the substantial knowledge gaps that may result when translating limited data sets to the**
1022 **wider global picture. This includes incomplete speciation and quantification across**
1023 **precipitation, aerosol, and gas phases. Monitoring systems that support U.S. deposition**
1024 **assessments (e.g., the NADP) only characterize the inorganic fraction of wet deposition.**
1025 **Additionally, modern emerging issues that require the continuation of existing deposition**
1026 **measurements or expansion of observation programs revolve around identifying and**
1027 **quantifying compound classes of concern, such as persistent organic pollutants. As reported**
1028 **in the literature, the deposition of these types of pollutants (e.g., polychlorinated biphenyls,**
1029 **polycyclic aromatic hydrocarbons, etc.) can be monitored using suitable collectors made of**
1030 **amber-coloured glass or stainless steel (Fingler et al., 1994; Amodio et al., 2014) -**
1031 **modifications which can be applied to the sample design detailed here. The samples collected**
1032 **in this work from this new instrumentation, specifically, are expected to be used further in**
1033 **several upcoming complementary and novel environmental monitoring studies. Not only will**

1034 **this future work extend our biogeochemical analysis, but it will also assist in our studying of**
1035 **the transport of other anthropogenic pollutants of emerging interest which are beyond the**
1036 **scope of describing this new platform.**

1037

1038 *Data availability.* The data are available from the corresponding author (TV) on request.

1039

1040 *Author contributions.* AC, DP, and ML performed the data analysis. AC and DP wrote the
1041 manuscript with contributions from all authors. Sampler design and construction were led by TV,
1042 with assistance from BP and RH for initial prototypes, DP and ML for the revised iteration, and
1043 AC for the final modular control boards. Sample collection and associated characterization
1044 measurements were performed by BP and TV. Conceptualization and conduct of the sampling
1045 experiments were made by TV, CY, KE, and SZ. All authors were involved in examining and
1046 reviewing the results. All authors were involved in editing the paper.

1047

1048 *Competing interests.* The contact author has declared that none of the authors has any competing
1049 interests.

1050

1051 **5.0 Acknowledgements**

1052 Funding for this work was provided by the Newfoundland and Labrador Department of
1053 Agrifoods and Forestry, Centre for Forestry Science and Innovation (Project 221269), and the
1054 Harris Centre at Memorial University. T. C. VandenBoer was supported for this work in-part
1055 through a Government of Canada Banting Postdoctoral Fellowship. Fieldwork sample collection
1056 by B. K. Place was supported by funding from Polar Knowledge Canada through the Northern
1057 Scientific Training Program. Additional financial support for full redesign of the samplers was
1058 provided through Environment and Climate Change Canada Grants & Contributions
1059 (GCXE20S009). A. A. Colussi acknowledges support for this work through a Natural Sciences
1060 and Engineering Research Council of Canada (NSERC) Graduate Scholarship – Master’s program
1061 (CGSM) and Ontario Graduate Scholarship (OGS). M. Lao acknowledges support for this work
1062 through a NSERC Undergraduate Student Research Award (USRA). We thank C. M. Laprise and
1063 C. Conlan for aid in the collection and organization of samples for analysis, supported in part by
1064 the Memorial University Career Experience Program (MUCEP) and V. Sitahai through a York
1065 University Dean’s Undergraduate Research Award (DURA). T. C. VandenBoer, C. J. Young and
1066 S. E. Ziegler were supported through the NSERC Discovery (RGPIN-2020-06166; RGPIN-2018-

1067 05990; RGPIN-2018-05383) and Strategic Partnerships (479224) Programs. The authors would
1068 also like to thank B. Hearn, D. Harris, A. Skinner, C. Young, J. J. MacInnis, J. Warren, and L.
1069 Souza for their invaluable assistance in sampling site access and set up, off-season storage of
1070 collection units, sample collection and analysis, and meteorological reanalysis. We thank H. Hung
1071 and C. Shunthirasingham for productive discussions on modular design and considerations for
1072 collection of persistent pollutants.

1073

1074 **6.0 References**

1075 Amodio, M., Catino, S., Dambruoso, P. R., de Gennaro, G., Di Gilio, A., Giungato, P., Laiola, E.,
1076 Marzocca, A., Mazzone, A., Sardaro, A., and Tutino, M.: Atmospheric Deposition: Sampling
1077 Procedures, Analytical Methods, and Main Recent Findings from the Scientific Literature, *Adv.*
1078 *Meteorol.*, 2014, 161730, <https://doi.org/10.1155/2014/161730>, 2014.

1079 Argentino, C., Kalenitchenko, D., Lindgren, M., and Panieri, G.: HgCl₂ addition to pore water
1080 samples from cold seeps can affect the geochemistry of dissolved inorganic carbon ([DIC],
1081 $\delta^{13}\text{CDIC}$), *Mar. Chem.*, 251, 104236, <https://doi.org/10.1016/j.marchem.2023.104236>, 2023.

1082 Arisci, S., Rogora, M., Marchetto, A., and Dichiaro, F.: The role of forest type in the variability of
1083 DOC in atmospheric deposition at forest plots in Italy, *Environ. Monit. Assess.*, 184, 3415–3425,
1084 <https://doi.org/10.1007/s10661-011-2196-2>, 2012.

1085 Audoux, T., Laurent, B., Desboeufs, K., Noyalet, G., Maisonneuve, F., Lauret, O., and Chevaillier,
1086 S.: Intra-event evolution of elemental and ionic concentrations in wet deposition in an urban
1087 environment, *Atmospheric Chem. Phys.*, 23, 13485–13503, [https://doi.org/10.5194/acp-23-](https://doi.org/10.5194/acp-23-13485-2023)
1088 13485-2023, 2023.

1089 Avery, G. B., Willey, J. D., and Kieber, R. J.: Carbon isotopic characterization of dissolved organic
1090 carbon in rainwater: Terrestrial and marine influences, *Atmos. Environ.*, 40, 7539–7545,
1091 <https://doi.org/10.1016/j.atmosenv.2006.07.014>, 2006.

1092 Barber, V. P. and Kroll, J. H.: Chemistry of Functionalized Reactive Organic Intermediates in the
1093 Earth's Atmosphere: Impact, Challenges, and Progress, *J. Phys. Chem. A*, 125, 10264–10279,
1094 <https://doi.org/10.1021/acs.jpca.1c08221>, 2021.

1095 Barrie, L. A. and Hales, J. M.: The spatial distributions of precipitation acidity and major ion wet
1096 deposition in North America during 1980, *Tellus B Chem. Phys. Meteorol.*, 36, 333–355,
1097 <https://doi.org/10.3402/tellusb.v36i5.14915>, 1984.

1098 Beverland, I. J., Heal, M. R., Crowther, J. M., and Srinivas, M. S. N.: Real-time measurement and
1099 interpretation of the conductivity and pH of precipitation samples, *Water. Air. Soil Pollut.*, 98,
1100 325–344, <https://doi.org/10.1007/BF02047042>, 1997.

1101 Bowering, K. L., Edwards, K. A., Wiersma, Y. F., Billings, S. A., Warren, J., Skinner, A., and
1102 Ziegler, S. E.: Dissolved Organic Carbon Mobilization Across a Climate Transect of Mesic Boreal

- 1103 Forests Is Explained by Air Temperature and Snowpack Duration, *Ecosystems*, 26, 55–71,
1104 <https://doi.org/10.1007/s10021-022-00741-0>, 2022.
- 1105 Bowering, K. L., Edwards, K. A., and Ziegler, S. E.: Seasonal controls override forest harvesting
1106 effects on the composition of dissolved organic matter mobilized from boreal forest soil organic
1107 horizons, *Biogeosciences*, 20, 2189–2206, <https://doi.org/10.5194/bg-20-2189-2023>, 2023.
- 1108 Bowyer, Peter J. (Ed.): *Where the Wind Blows: A Guide to Marine Weather in Atlantic Canada*,
1109 Breakwater Books Ltd., St John's, Newfoundland, 53 pp., 1995.
- 1110 Boyd, C. E.: Carbon Dioxide, pH, and Alkalinity, in: *Water Quality: An Introduction*, edited by:
1111 Boyd, C. E., Springer International Publishing, Cham, 177–203, [https://doi.org/10.1007/978-3-](https://doi.org/10.1007/978-3-030-23335-8_9)
1112 [030-23335-8_9](https://doi.org/10.1007/978-3-030-23335-8_9), 2020.
- 1113 Brahney, J., Wetherbee, G., Sexstone, G. A., Youngbull, C., Strong, P., and Heindel, R. C.: A new
1114 sampler for the collection and retrieval of dry dust deposition, *Aeolian Res.*, 45, 100600,
1115 <https://doi.org/10.1016/j.aeolia.2020.100600>, 2020.
- 1116 Burkhardt, J. and Drechsel, P.: The synergism between SO₂ oxidation and manganese leaching on
1117 spruce needles — A chamber experiment, *Environ. Pollut.*, 95, 1–11,
1118 [https://doi.org/10.1016/S0269-7491\(96\)00126-1](https://doi.org/10.1016/S0269-7491(96)00126-1), 1997.
- 1119 Canadian Air and Precipitation Monitoring Network: Inspector's Reference Manual -
1120 Precipitation, 1985.
- 1121 Cappellato, R., Peters, N. E., and Ragsdale, H. L.: Acidic atmospheric deposition and canopy
1122 interactions of adjacent deciduous and coniferous forests in the Georgia Piedmont, *Can. J. For.*
1123 *Res.*, 23, 1114–1124, <https://doi.org/10.1139/x93-142>, 1993.
- 1124 Carlson, J., Gough, W. A., Karagatzides, J. D., and Tsuji, L. J. S.: Canopy Interception of Acid
1125 Deposition in Southern Ontario, *Can. Field-Nat.*, 117, 523–530,
1126 <https://doi.org/10.22621/cfn.v117i4.799>, 2003.
- 1127 Casas-Ruiz, J. P., Bodmer, P., Bona, K. A., Butman, D., Couturier, M., Emilson, E. J. S., Finlay,
1128 K., Genet, H., Hayes, D., Karlsson, J., Paré, D., Peng, C., Striegl, R., Webb, J., Wei, X., Ziegler,
1129 S. E., and del Giorgio, P. A.: Integrating terrestrial and aquatic ecosystems to constrain estimates
1130 of land-atmosphere carbon exchange, *Nat. Commun.*, 14, 1571, [https://doi.org/10.1038/s41467-](https://doi.org/10.1038/s41467-023-37232-2)
1131 [023-37232-2](https://doi.org/10.1038/s41467-023-37232-2), 2023.
- 1132 Cha, J.-Y., Lee, S.-C., Lee, E.-J., Lee, K., Lee, H., Kim, H. S., Ahn, J., and Oh, N.-H.: Canopy
1133 Leaching Rather than Desorption of PM_{2.5} From Leaves Is the Dominant Source of Throughfall
1134 Dissolved Organic Carbon in Forest, *Geophys. Res. Lett.*, 50, e2023GL103731,
1135 <https://doi.org/10.1029/2023GL103731>, 2023.
- 1136 Chen, H., Tsai, K.-P., Su, Q., Chow, A. T., and Wang, J.-J.: Throughfall Dissolved Organic Matter
1137 as a Terrestrial Disinfection Byproduct Precursor, *ACS Earth Space Chem.*, 3, 1603–1613,
1138 <https://doi.org/10.1021/acsearthspacechem.9b00088>, 2019.

- 1139 Ciruzzi, D. M. and Loheide, S. P.: Monitoring Tree Sway as an Indicator of Interception Dynamics
1140 Before, During, and Following a Storm, *Geophys. Res. Lett.*, 48, e2021GL094980,
1141 <https://doi.org/10.1029/2021GL094980>, 2021.
- 1142 Colli, M., Lanza, L. G., Rasmussen, R., and Thériault, J. M.: The Collection Efficiency of Shielded
1143 and Unshielded Precipitation Gauges. Part II: Modeling Particle Trajectories, *J. Hydrometeorol.*,
1144 17, 245–255, <https://doi.org/10.1175/JHM-D-15-0011.1>, 2016.
- 1145 Coward, E. K., Seech, K., Carter, M. L., Flick, R. E., and Grassian, V. H.: Of Sea and Smoke:
1146 Evidence of Marine Dissolved Organic Matter Deposition from 2020 Western United States
1147 Wildfires, *Environ. Sci. Technol. Lett.*, 9, 869–876, <https://doi.org/10.1021/acs.estlett.2c00383>,
1148 2022.
- 1149 Csavina, J., Field, J., Taylor, M. P., Gao, S., Landázuri, A., Betterton, E. A., and Sáez, A. E.: A
1150 review on the importance of metals and metalloids in atmospheric dust and aerosol from mining
1151 operations, *Sci. Total Environ.*, 433, 58–73, <https://doi.org/10.1016/j.scitotenv.2012.06.013>, 2012.
- 1152 Dalva, M. and Moore, T. R.: Sources and sinks of dissolved organic carbon in a forested swamp
1153 catchment, *Biogeochemistry*, 15, 1–19, <https://doi.org/10.1007/BF00002806>, 1991.
- 1154 Di Lorenzo, R. A., Place, B. K., VandenBoer, T. C., and Young, C. J.: Composition of Size-
1155 Resolved Aged Boreal Fire Aerosols: Brown Carbon, Biomass Burning Tracers, and Reduced
1156 Nitrogen, *ACS Earth Space Chem.*, 2, 278–285,
1157 <https://doi.org/10.1021/acsearthspacechem.7b00137>, 2018.
- 1158 Dossett, S. R. and Bowersox, V. C.: National Trends Network Site Operation Manual, 1999.
- 1159 Eaton, J. S., Likens, G. E., and Bormann, F. H.: Throughfall and Stemflow Chemistry in a Northern
1160 Hardwood Forest, *J. Ecol.*, 61, 495–508, <https://doi.org/10.2307/2259041>, 1973.
- 1161 Erisman, J. W. and Draaijers, G.: Deposition to forests in Europe: most important factors
1162 influencing dry deposition and models used for generalisation, *Environ. Pollut.*, 124, 379–388,
1163 [https://doi.org/10.1016/S0269-7491\(03\)00049-6](https://doi.org/10.1016/S0269-7491(03)00049-6), 2003.
- 1164 Escarré, A., Carratalá, A., Àvila, A., Bellot, J., Piñol, J., and Milán, M.: Precipitation Chemistry
1165 and Air Pollution, in: *Ecology of Mediterranean Evergreen Oak Forests*, edited by: Rodà, F.,
1166 Retana, J., Gracia, C. A., and Bellot, J., Springer Berlin Heidelberg, Berlin, Heidelberg, 195–208,
1167 https://doi.org/10.1007/978-3-642-58618-7_14, 1999.
- 1168 Farmer, D. K., Boedicker, E. K., and DeBolt, H. M.: Dry Deposition of Atmospheric Aerosols:
1169 Approaches, Observations, and Mechanisms, *Annu. Rev. Phys. Chem.*, 72, 375–397,
1170 <https://doi.org/10.1146/annurev-physchem-090519-034936>, 2021.
- 1171 Feng, J., Vet, R., Cole, A., Zhang, L., Cheng, I., O'Brien, J., and Macdonald, A.-M.: Inorganic
1172 chemical components in precipitation in the eastern U.S. and Eastern Canada during 1989–2016:
1173 Temporal and regional trends of wet concentration and wet deposition from the NADP and
1174 CAPMoN measurements, *Atmos. Environ.*, 254, 118367,
1175 <https://doi.org/10.1016/j.atmosenv.2021.118367>, 2021.

- 1176 Fingler, S., Tkalčević, B., Fröbe, Z., and Drevenkar, V.: Analysis of polychlorinated biphenyls,
1177 organochlorine pesticides and chlorophenols in rain and snow, *Analyst*, 119, 1135–1140,
1178 <https://doi.org/10.1039/AN9941901135>, 1994.
- 1179 Fowler, D.: *Wet and Dry Deposition of Sulphur and Nitrogen Compounds from the Atmosphere*,
1180 1980.
- 1181 Galloway, J. N. and Likens, G. E.: The collection of precipitation for chemical analysis, *Tellus*,
1182 30, 71–82, <https://doi.org/10.3402/tellusa.v30i1.10318>, 1978.
- 1183 Gao, S., Hegg, D. A., Hobbs, P. V., Kirchstetter, T. W., Magi, B. I., and Sadilek, M.: Water-soluble
1184 organic components in aerosols associated with savanna fires in southern Africa: Identification,
1185 evolution, and distribution, *J. Geophys. Res. Atmospheres*, 108,
1186 <https://doi.org/10.1029/2002JD002324>, 2003.
- 1187 Gatz, D. F., Selman, R. F., Langs, R. K., and Holtzman, R. B.: An Automatic Sequential Rain
1188 Sampler, *J. Appl. Meteorol.* 1962-1982, 10, 341–344, 1971.
- 1189 George, C.: *Photosensitization is in the air and impacts the multiphase on oxidation capacity*, 2023.
- 1190 Germer, S., Neill, C., Krusche, A. V., Neto, S. C. G., and Elsenbeer, H.: Seasonal and within-event
1191 dynamics of rainfall and throughfall chemistry in an open tropical rainforest in Rondônia, Brazil,
1192 *Biogeochemistry*, 86, 155–174, <https://doi.org/10.1007/s10533-007-9152-9>, 2007.
- 1193 Grennfelt, P., Engleryd, A., Forsius, M., Hov, Ø., Rodhe, H., and Cowling, E.: Acid rain and air
1194 pollution: 50 years of progress in environmental science and policy, *Ambio*, 49, 849–864,
1195 <https://doi.org/10.1007/s13280-019-01244-4>, 2020.
- 1196 Guinotte, J. M. and Fabry, V. J.: Ocean Acidification and Its Potential Effects on Marine
1197 Ecosystems, *Ann. N. Y. Acad. Sci.*, 1134, 320–342, <https://doi.org/10.1196/annals.1439.013>,
1198 2008.
- 1199 Hadiwijaya, B., Isabelle, P.-E., Nadeau, D. F., and Pepin, S.: Observations of canopy storage
1200 capacity and wet canopy evaporation in a humid boreal forest, *Hydrol. Process.*, 35, e14021,
1201 <https://doi.org/10.1002/hyp.14021>, 2021.
- 1202 Hall, D. J.: *Precipitation collector for use in the Secondary National Acid Deposition Network*,
1203 United States, 1985.
- 1204 Han, G., Song, Z., Tang, Y., Wu, Q., and Wang, Z.: Ca and Sr isotope compositions of rainwater
1205 from Guiyang city, Southwest China: Implication for the sources of atmospheric aerosols and their
1206 seasonal variations, *Atmos. Environ.*, 214, 116854,
1207 <https://doi.org/10.1016/j.atmosenv.2019.116854>, 2019.
- 1208 Heald, C. L. and Kroll, J. H.: The fuel of atmospheric chemistry: Toward a complete description
1209 of reactive organic carbon, *Sci. Adv.*, 6, <https://doi.org/10.1126/sciadv.aay8967>, 2020.

- 1210 Heald, C. L., Gouw, J. de, Goldstein, A. H., Guenther, A. B., Hayes, P. L., Hu, W., Isaacman-
1211 VanWertz, G., Jimenez, J. L., Keutsch, F. N., Koss, A. R., Misztal, P. K., Rappenglück, B.,
1212 Roberts, J. M., Stevens, P. S., Washenfelder, R. A., Warneke, C., and Young, C. J.: Contrasting
1213 Reactive Organic Carbon Observations in the Southeast United States (SOAS) and Southern
1214 California (CalNex), *Environ. Sci. Technol.*, 54, 14923–14935,
1215 <https://doi.org/10.1021/acs.est.0c05027>, 2020.
- 1216 Houle, D., Augustin, F., and Couture, S.: Rapid improvement of lake acid–base status in Atlantic
1217 Canada following steep decline in precipitation acidity, *Can. J. Fish. Aquat. Sci.*, 79, 2126–2137,
1218 <https://doi.org/10.1139/cjfas-2021-0349>, 2022.
- 1219 Howard, M., Hathaway, J. M., Tirpak, R. A., Lisenbee, W. A., and Sims, S.: Quantifying urban
1220 tree canopy interception in the southeastern United States, *Urban For. Urban Green.*, 77, 127741,
1221 <https://doi.org/10.1016/j.ufug.2022.127741>, 2022.
- 1222 Iavorivska, L., Boyer, E. W., and DeWalle, D. R.: Atmospheric deposition of organic carbon via
1223 precipitation, *Acid Rain Its Environ. Eff. Recent Sci. Adv.*, 146, 153–163,
1224 <https://doi.org/10.1016/j.atmosenv.2016.06.006>, 2016.
- 1225 Jacob, Daniel J.: *Introduction to Atmospheric Chemistry*, Princeton University Press, Princeton,
1226 New Jersey, 49 pp., 1999.
- 1227 Jurado, E., Jaward, F. M., Lohmann, R., Jones, K. C., Simó, R., and Dachs, J.: Atmospheric Dry
1228 Deposition of Persistent Organic Pollutants to the Atlantic and Inferences for the Global Oceans,
1229 *Environ. Sci. Technol.*, 38, 5505–5513, <https://doi.org/10.1021/es049240v>, 2004.
- 1230 Jurado, E., Jaward, F., Lohmann, R., Jones, K. C., Simó, R., and Dachs, J.: Wet Deposition of
1231 Persistent Organic Pollutants to the Global Oceans, *Environ. Sci. Technol.*, 39, 2426–2435,
1232 <https://doi.org/10.1021/es048599g>, 2005.
- 1233 Kattner, G.: Storage of dissolved inorganic nutrients in seawater: poisoning with mercuric
1234 chloride, *Mar. Chem.*, 67, 61–66, [https://doi.org/10.1016/S0304-4203\(99\)00049-3](https://doi.org/10.1016/S0304-4203(99)00049-3), 1999.
- 1235 Kirkwood, D. S.: Stability of solutions of nutrient salts during storage, *Mar. Chem.*, 38, 151–164,
1236 [https://doi.org/10.1016/0304-4203\(92\)90032-6](https://doi.org/10.1016/0304-4203(92)90032-6), 1992.
- 1237 Kochendorfer, J., Meyers, T. P., Hall, M. E., Landolt, S. D., and Diamond, H. J.: A new reference-
1238 quality precipitation gauge wind shield, *Atmospheric Meas. Tech. Discuss.*, 2023, 1–17,
1239 <https://doi.org/10.5194/amt-2023-2>, 2023.
- 1240 Kroll, J. H., Donahue, N. M., Jimenez, J. L., Kessler, S. H., Canagaratna, M. R., Wilson, K. R.,
1241 Altieri, K. E., Mazzoleni, L. R., Wozniak, A. S., Bluhm, H., Mysak, E. R., Smith, J. D., Kolb, C.
1242 E., and Worsnop, D. R.: Carbon oxidation state as a metric for describing the chemistry of
1243 atmospheric organic aerosol, *Nat. Chem.*, 3, 133–139, <https://doi.org/10.1038/nchem.948>, 2011.
- 1244 Kuylenstierna, J. C., Rodhe, H., Cinderby, S., and Hicks, K.: Acidification in developing countries:
1245 ecosystem sensitivity and the critical load approach on a global scale., *Ambio*, 30, 20–28,
1246 <https://doi.org/10.1579/0044-7447-30.1.20>, 2001.

- 1247 Laquer, F. C.: Sequential precipitation samplers: A literature review, *Atmospheric Environ. Part*
1248 *Gen. Top.*, 24, 2289–2297, [https://doi.org/10.1016/0960-1686\(90\)90322-E](https://doi.org/10.1016/0960-1686(90)90322-E), 1990.
- 1249 Laurent, B., Losno, R., Chevaillier, S., Vincent, J., Rouillet, P., Bon Nguyen, E., Ouboulmane, N.,
1250 Triquet, S., Fornier, M., Raimbault, P., and Bergametti, G.: An automatic collector to monitor
1251 insoluble atmospheric deposition: application for mineral dust deposition, *Atmospheric Meas.*
1252 *Tech.*, 8, 2801–2811, <https://doi.org/10.5194/amt-8-2801-2015>, 2015.
- 1253 Lawrence, C. E., Casson, P., Brandt, R., Schwab, J. J., Dukett, J. E., Snyder, P., Yerger, E., Kelting,
1254 D., VandenBoer, T. C., and Lance, S.: Long-term monitoring of cloud water chemistry at
1255 Whiteface Mountain: the emergence of a new chemical regime, *Atmospheric Chem. Phys.*, 23,
1256 1619–1639, <https://doi.org/10.5194/acp-23-1619-2023>, 2023.
- 1257 Likens, G. E. and Butler, T. J.: Atmospheric Acid Deposition, in: *The Handbook of Natural*
1258 *Resources, Atmosphere and Climate*, vol. 6, edited by: Wang, Y., CRC Press, 2020.
- 1259 Lin, W.-C., Brondum, K., Monroe, C. W., and Burns, M. A.: Multifunctional Water Sensors for
1260 pH, ORP, and Conductivity Using Only Microfabricated Platinum Electrodes, *Sensors*, 17,
1261 <https://doi.org/10.3390/s17071655>, 2017.
- 1262 Lindberg, S. E., Lovett, G. M., Richter, D. D., and Johnson, D. W.: Atmospheric Deposition and
1263 Canopy Interactions of Major Ions in a Forest, *Science*, 231, 141–145,
1264 <https://doi.org/10.1126/science.231.4734.141>, 1986.
- 1265 Lovett, G. M.: Atmospheric Deposition of Nutrients and Pollutants in North America: An
1266 Ecological Perspective, *Ecol. Appl.*, 4, 629–650, <https://doi.org/10.2307/1941997>, 1994.
- 1267 Lovett, G. M. and Kinsman, J. D.: Atmospheric pollutant deposition to high-elevation ecosystems,
1268 *Atmospheric Environ. Part Gen. Top.*, 24, 2767–2786, [https://doi.org/10.1016/0960-](https://doi.org/10.1016/0960-1686(90)90164-I)
1269 [1686\(90\)90164-I](https://doi.org/10.1016/0960-1686(90)90164-I), 1990.
- 1270 Meteorological Service of Canada: 2004 Canadian Acid Deposition Science Assessment, , Library
1271 and Archives Canada, 2005.
- 1272 Metzger, J. C., Schumacher, J., Lange, M., and Hildebrandt, A.: Neighbourhood and stand
1273 structure affect stemflow generation in a heterogeneous deciduous temperate forest, *Hydrol. Earth*
1274 *Syst. Sci.*, 23, 4433–4452, <https://doi.org/10.5194/hess-23-4433-2019>, 2019.
- 1275 Michalzik, B. and Matzner, E.: Dynamics of dissolved organic nitrogen and carbon in a Central
1276 European Norway spruce ecosystem, *Eur. J. Soil Sci.*, 50, 579–590, [https://doi.org/10.1046/j.1365-](https://doi.org/10.1046/j.1365-2389.1999.00267.x)
1277 [2389.1999.00267.x](https://doi.org/10.1046/j.1365-2389.1999.00267.x), 1999.
- 1278 Moore, T. R.: Dissolved organic carbon in a northern boreal landscape, *Glob. Biogeochem. Cycles*,
1279 17, <https://doi.org/10.1029/2003GB002050>, 2003.
- 1280 Myers-Pigg, A. N., Louchouart, P., Amon, R. M. W., Prokushkin, A., Pierce, K., and Rubtsov,
1281 A.: Labile pyrogenic dissolved organic carbon in major Siberian Arctic rivers: Implications for

- 1282 wildfire-stream metabolic linkages, *Geophys. Res. Lett.*, 42, 377–385,
1283 <https://doi.org/10.1002/2014GL062762>, 2015.
- 1284 National Atmospheric Deposition Program: NADP Site Selection and Installation Manual, 2009.
- 1285 Oka, A., Takahashi, J., Endoh, Y., and Seino, T.: Bark Effects on Stemflow Chemistry in a
1286 Japanese Temperate Forest I. The Role of Bark Surface Morphology, *Front. For. Glob. Change*, 4,
1287 <https://doi.org/10.3389/ffgc.2021.654375>, 2021.
- 1288 Pacyna, J. M.: Ecological Processes: Atmospheric Deposition, in: *Encyclopedia of Ecology*, vol.
1289 1, edited by: Jorgensen, S. E. and Fath, B. D., Elsevier Science, 275–285, 2008.
- 1290 Pan, Y., Wang, Y., Xin, J., Tang, G., Song, T., Wang, Y., Li, X., and Wu, F.: Study on dissolved
1291 organic carbon in precipitation in Northern China, *Atmos. Environ.*, 44, 2350–2357,
1292 <https://doi.org/10.1016/j.atmosenv.2010.03.033>, 2010.
- 1293 Peden, M. E., Bachman, S. R., Brennan, C. J., Demir, B., James, K. O., Kaiser, B. W., Lockard, J.
1294 M., Rothery, J. E., Sauer, J., Skowron, L. M., and Slater, M. J.: Methods for collection and analysis
1295 of precipitation, 1986.
- 1296 Pomeroy, J. W., Granger, R., Pietroniro, J., Elliott, J., Toth, B., and Hedstrom, N.: Classification
1297 of the Boreal Forest for Hydrological Processes, in: *Proceedings of the Ninth International Boreal
1298 Forest Research Association Conference*, 1999.
- 1299 Pumpanen, J., Lindén, A., Miettinen, H., Kolari, P., Ilvesniemi, H., Mammarella, I., Hari, P.,
1300 Nikinmaa, E., Heinonsalo, J., Bäck, J., Ojala, A., Berninger, F., and Vesala, T.: Precipitation and
1301 net ecosystem exchange are the most important drivers of DOC flux in upland boreal catchments,
1302 *J. Geophys. Res. Biogeosciences*, 119, 1861–1878, <https://doi.org/10.1002/2014JG002705>, 2014.
- 1303 Ramanathan, V. and Carmichael, G.: Global and regional climate changes due to black carbon,
1304 *Nat. Geosci.*, 1, 221–227, <https://doi.org/10.1038/ngeo156>, 2008.
- 1305 Randall, David: *An Introduction to the Global Circulation of the Atmosphere*, Princeton University
1306 Press, Princeton, New Jersey, 43 pp., 2015.
- 1307 Reddy, M. M., Liebermann, T. D., Jelinski, J. C., and Caine, N.: Variation in pH During Summer
1308 Storms Near the Continental Divide in Central Colorado, U.S.A.*, *Arct. Alp. Res.*, 17, 79–88,
1309 <https://doi.org/10.1080/00040851.1985.12004450>, 1985.
- 1310 Richter, D. D. and Lindberg, S. E.: Wet Deposition Estimates from Long-Term Bulk and Event
1311 Wet-Only Samples of Incident Precipitation and Throughfall, *J. Environ. Qual.*, 17, 619–622,
1312 <https://doi.org/10.2134/jeq1988.00472425001700040017x>, 1988.
- 1313 Ryan, K. A., Adler, T., Chalmers, A., Perdrial, J., Shanley, J. B., and Stubbins, A.: Event Scale
1314 Relationships of DOC and TDN Fluxes in Throughfall and Stemflow Diverge From Stream
1315 Exports in a Forested Catchment, *J. Geophys. Res. Biogeosciences*, 126, e2021JG006281,
1316 <https://doi.org/10.1029/2021JG006281>, 2021.

- 1317 Safieddine, S. A. and Heald, C. L.: A Global Assessment of Dissolved Organic Carbon in
1318 Precipitation, *Geophys. Res. Lett.*, 44, 11,672-11,681, <https://doi.org/10.1002/2017GL075270>,
1319 2017.
- 1320 Saleh, R.: From Measurements to Models: Toward Accurate Representation of Brown Carbon in
1321 Climate Calculations, *Curr. Pollut. Rep.*, 6, 90–104, <https://doi.org/10.1007/s40726-020-00139-3>,
1322 2020.
- 1323 Sanei, H., Outridge, P. M., Goodarzi, F., Wang, F., Armstrong, D., Warren, K., and Fishback, L.:
1324 Wet deposition mercury fluxes in the Canadian sub-Arctic and southern Alberta, measured using
1325 an automated precipitation collector adapted to cold regions, *Atmos. Environ.*, 44, 1672–1681,
1326 <https://doi.org/10.1016/j.atmosenv.2010.01.030>, 2010.
- 1327 Santín, C., Doerr, S. H., Kane, E. S., Masiello, C. A., Ohlson, M., de la Rosa, J. M., Preston, C.
1328 M., and Dittmar, T.: Towards a global assessment of pyrogenic carbon from vegetation fires, *Glob.
1329 Change Biol.*, 22, 76–91, <https://doi.org/10.1111/gcb.12985>, 2016.
- 1330 Siksna, R.: The electrolytical conductivity of precipitation water as an aid to the chemical analysis,
1331 *Geofis. Pura E Appl.*, 42, 32–41, <https://doi.org/10.1007/BF02113385>, 1959.
- 1332 Sleutel, S., Vandenbruwane, J., De Schrijver, A., Wuyts, K., Moeskops, B., Verheyen, K., and De
1333 Neve, S.: Patterns of dissolved organic carbon and nitrogen fluxes in deciduous and coniferous
1334 forests under historic high nitrogen deposition, *Biogeosciences*, 6, 2743–2758,
1335 <https://doi.org/10.5194/bg-6-2743-2009>, 2009.
- 1336 Smith, W. H.: *Air Pollution and Forests: Interactions Between Air Contaminants and Forest
1337 Ecosystems*, 1st ed., Springer, New York, NY, 379 pp., 1981.
- 1338 Stedman, J. R., Heyes, C. J., and Irwin, J. G.: A comparison of bulk and wet-only precipitation
1339 collectors at rural sites in the United Kingdom, *Water. Air. Soil Pollut.*, 52, 377–395,
1340 <https://doi.org/10.1007/BF00229445>, 1990.
- 1341 Stoddard, J. L., Jeffries, D. S., Lükewille, A., Clair, T. A., Dillon, P. J., Driscoll, C. T., Forsius,
1342 M., Johannessen, M., Kahl, J. S., Kellogg, J. H., Kemp, A., Mannio, J., Monteith, D. T., Murdoch,
1343 P. S., Patrick, S., Rebsdorf, A., Skjelkvåle, B. L., Stainton, M. P., Traaen, T., van Dam, H.,
1344 Webster, K. E., Wieting, J., and Wilander, A.: Regional trends in aquatic recovery from
1345 acidification in North America and Europe, *Nature*, 401, 575–578, <https://doi.org/10.1038/44114>,
1346 1999.
- 1347 Stubbins, A., Silva, L. M., Dittmar, T., and Van Stan, J. T.: Molecular and Optical Properties of
1348 Tree-Derived Dissolved Organic Matter in Throughfall and Stemflow from Live Oaks and Eastern
1349 Red Cedar, *Front. Earth Sci.*, 5, <https://doi.org/10.3389/feart.2017.00022>, 2017.
- 1350 Thornton, M. M., Shrestha, R., Wei, Y., Thornton, P. E., Kao, S.-C., and Wilson, B. E.: Daymet:
1351 Monthly Climate Summaries on a 1-km Grid for North America, Version 4 R1,
1352 <https://doi.org/10.3334/ORNLDAAC/2131>, 2022.

- 1353 Thornton, P. E., Running, S. W., and White, M. A.: Generating surfaces of daily meteorological
1354 variables over large regions of complex terrain, *Aggreg. Descr. Land-Atmosphere Interact.*, 190,
1355 214–251, [https://doi.org/10.1016/S0022-1694\(96\)03128-9](https://doi.org/10.1016/S0022-1694(96)03128-9), 1997.
- 1356 Thornton, P. E., Shrestha, R., Thornton, M., Kao, S.-C., Wei, Y., and Wilson, B. E.: Gridded daily
1357 weather data for North America with comprehensive uncertainty quantification, *Sci. Data*, 8, 190,
1358 <https://doi.org/10.1038/s41597-021-00973-0>, 2021.
- 1359 United States Environmental Protection Agency: Integrated Science Assessment (ISA) for Oxides
1360 of Nitrogen, Oxides of Sulfur and Particulate Matter Ecological Criteria (Final Report, 2020), U.S.
1361 Environmental Protection Agency, Washington, DC, 2020.
- 1362 Van Stan, J. T. and Stubbins, A.: Tree-DOM: Dissolved organic matter in throughfall and
1363 stemflow, *Limnol. Oceanogr. Lett.*, 3, 199–214, <https://doi.org/10.1002/lol2.10059>, 2018.
- 1364 Van Stan, J. T., Wagner, S., Guillemette, F., Whitetree, A., Lewis, J., Silva, L., and Stubbins, A.:
1365 Temporal Dynamics in the Concentration, Flux, and Optical Properties of Tree-Derived Dissolved
1366 Organic Matter in an Epiphyte-Laden Oak-Cedar Forest, *J. Geophys. Res. Biogeosciences*, 122,
1367 2982–2997, <https://doi.org/10.1002/2017JG004111>, 2017.
- 1368 VandenBoer, T. C.: AIM-IC: Applications to Nitrous Acid (HONO) in the Ambient Atmosphere
1369 and Precipitation Monitoring, Masters of Science, University of Toronto, 2009.
- 1370 Vermette, S. J. and Drake, J. J.: Simplified wet-only and sequential fraction rain collector,
1371 *Atmospheric Environ.* 1967, 21, 715–716, [https://doi.org/10.1016/0004-6981\(87\)90053-9](https://doi.org/10.1016/0004-6981(87)90053-9), 1987.
- 1372 Vet, R., Artz, R. S., Carou, S., Shaw, M., Ro, C.-U., Aas, W., Baker, A., Bowersox, V. C.,
1373 Dentener, F., Galy-Lacaux, C., Hou, A., Pienaar, J. J., Gillett, R., Forti, M. C., Gromov, S., Hara,
1374 H., Khodzher, T., Mahowald, N. M., Nickovic, S., Rao, P. S. P., and Reid, N. W.: A global
1375 assessment of precipitation chemistry and deposition of sulfur, nitrogen, sea salt, base cations,
1376 organic acids, acidity and pH, and phosphorus, *Glob. Assess. Precip. Chem. Depos. Sulfur*
1377 *Nitrogen Sea Salt Base Cations Org. Acids Acidity PH Phosphorus*, 93, 3–100,
1378 <https://doi.org/10.1016/j.atmosenv.2013.10.060>, 2014.
- 1379 Wang, X., Gemayel, R., Baboosian, V. J., Li, K., Boreave, A., Dubois, C., Tomaz, S., Perrier, S.,
1380 Nizkorodov, S. A., and George, C.: Naphthalene-Derived Secondary Organic Aerosols Interfacial
1381 Photosensitizing Properties, *Geophys. Res. Lett.*, 48, e2021GL093465,
1382 <https://doi.org/10.1029/2021GL093465>, 2021.
- 1383 Washenfelder, R. A., Azzarello, L., Ball, K., Brown, S. S., Decker, Z. C. J., Franchin, A.,
1384 Fredrickson, C. D., Hayden, K., Holmes, C. D., Middlebrook, A. M., Palm, B. B., Pierce, R. B.,
1385 Price, D. J., Roberts, J. M., Robinson, M. A., Thornton, J. A., Womack, C. C., and Young, C. J.:
1386 Complexity in the Evolution, Composition, and Spectroscopy of Brown Carbon in Aircraft
1387 Measurements of Wildfire Plumes, *Geophys. Res. Lett.*, 49, e2022GL098951,
1388 <https://doi.org/10.1029/2022GL098951>, 2022.
- 1389 Wetherbee, G. A., Shaw, M. J., Latysh, N. E., Lehmann, C. M. B., and Rothert, J. E.: Comparison
1390 of precipitation chemistry measurements obtained by the Canadian Air and Precipitation

1391 Monitoring Network and National Atmospheric Deposition Program for the period 1995–2004,
1392 Environ. Monit. Assess., 164, 111–132, <https://doi.org/10.1007/s10661-009-0879-8>, 2010.

1393 Wonaschütz, A., Hersey, S. P., Sorooshian, A., Craven, J. S., Metcalf, A. R., Flagan, R. C., and
1394 Seinfeld, J. H.: Impact of a large wildfire on water-soluble organic aerosol in a major urban area:
1395 the 2009 Station Fire in Los Angeles County, Atmospher. Chem. Phys., 11, 8257–8270,
1396 <https://doi.org/10.5194/acp-11-8257-2011>, 2011.

1397 Ziegler, S. E., Benner, R., Billings, S. A., Edwards, K. A., Philben, M., Zhu, X., and Laganière, J.:
1398 Climate Warming Can Accelerate Carbon Fluxes without Changing Soil Carbon Stocks, Front.
1399 Earth Sci., 5, <https://doi.org/10.3389/feart.2017.00002>, 2017.

1400

Supporting Information for

**Cost Effective Off-Grid Automatic Precipitation Samplers for
Pollutant and Biogeochemical Atmospheric Deposition**

Alessia A. Colussi¹, Daniel Persaud¹, Melodie Lao¹, Bryan K. Place^{2,‡}, Rachel F. Hems^{2,§}, Susan E. Ziegler³, Kate A. Edwards^{4,‡}, Cora J. Young^{1,2}, and Trevor C. VandenBoer^{1,3}

¹ *Department of Chemistry, York University, Toronto, ON*

² *Department of Chemistry, Memorial University, St. John's, NL*

³ *Department of Earth Science, Memorial University, St. John's, NL*

⁴ *Canadian Forest Service, Natural Resources Canada, Corner Brook, NL*

[‡] *Now at: SciGlob Instruments & Services LLC, Columbia, MD, USA*

[§] *Now at: Department of Chemistry and Biochemistry, Oberlin College and Conservatory, OH, USA*

[†] *Now at: Climate Change Impacts and Adaptation Division, Lands and Minerals Sector, Natural Resources Canada, Ottawa, ON*

Correspondence: Trevor VandenBoer (tvandenb@yorku.ca)

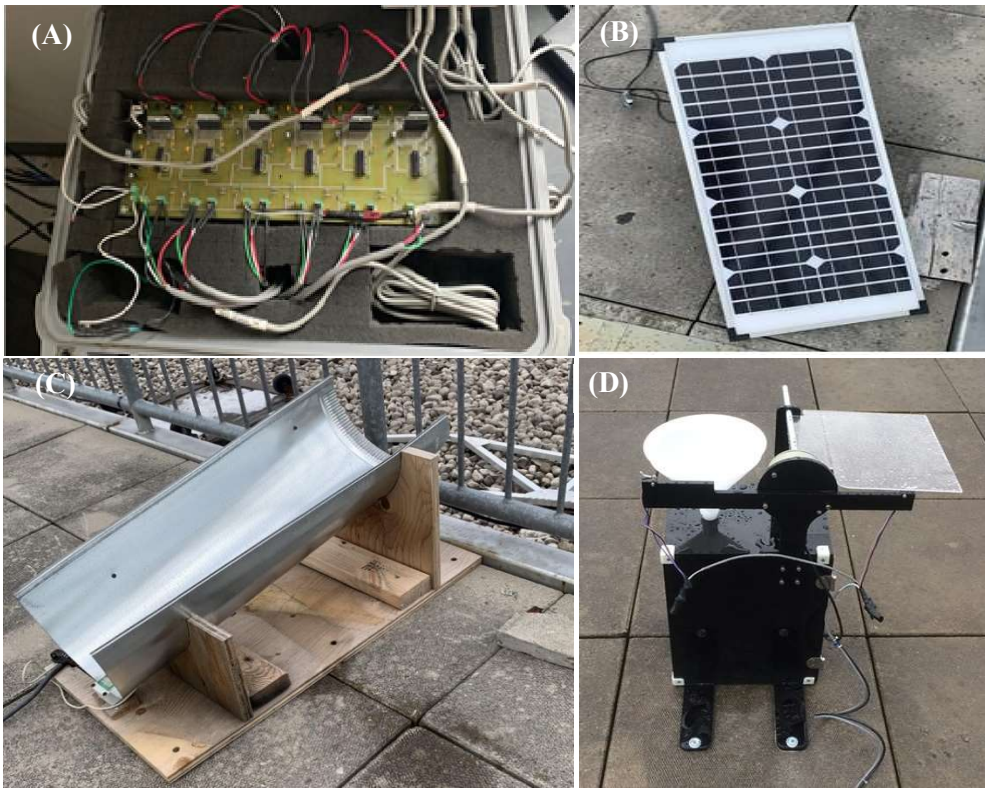


Figure S1. Automated precipitation array stationed at the Air Quality Research Station, York University, Toronto, consisting of: (A) weather-protected control board, (B) 40 W solar panel, (C) rain sensor, and enhanced sensitivity chute, and (D) an automated collection unit fixed to concrete with lag sleeves and bolts during a rain event.

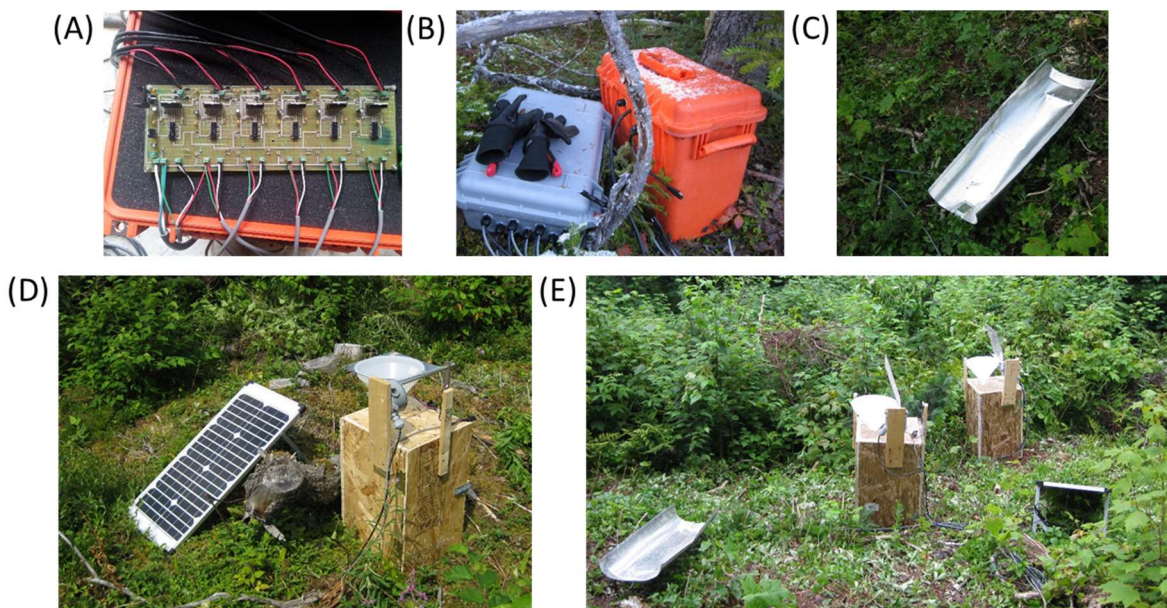


Figure S2. Automated precipitation samplers deployed at NL-BELT. The (A) weatherproof control board was powered by (B) an off-grid AGM battery. The (C) sensor chute modulated the

opening of the samplers for precipitation collection with **(D)** a 40 W solar panel to recharge the power package between **(E)** precipitation events when the lids open to collect wet deposition.

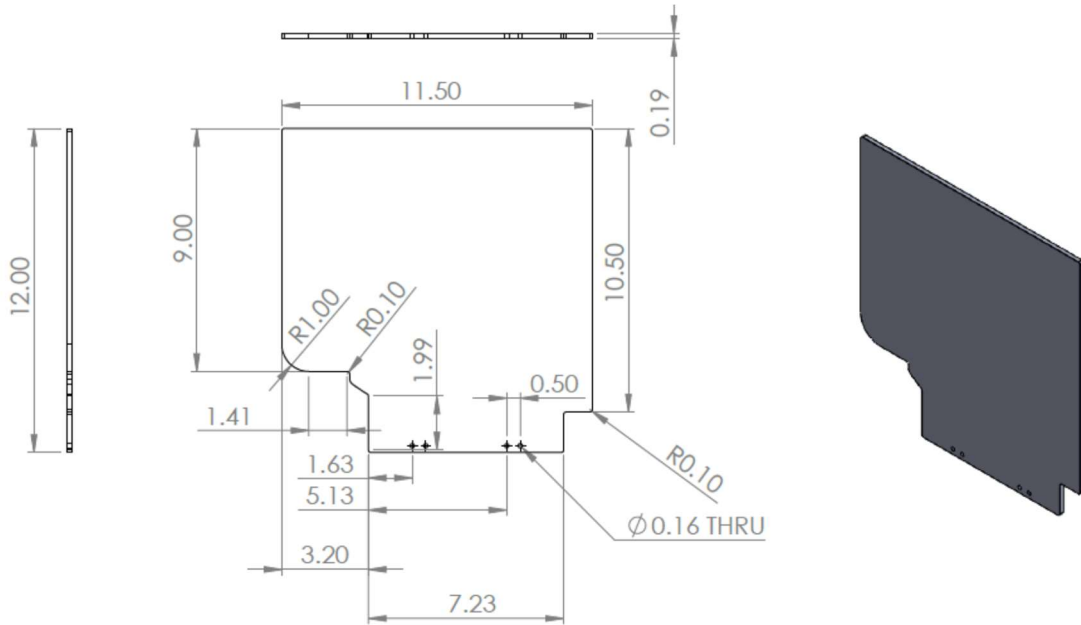


Figure S3. Dimensions (in inches) for automated collection unit lid, with mounting holes on the bottom edge to secure it to the aluminum lid rod.

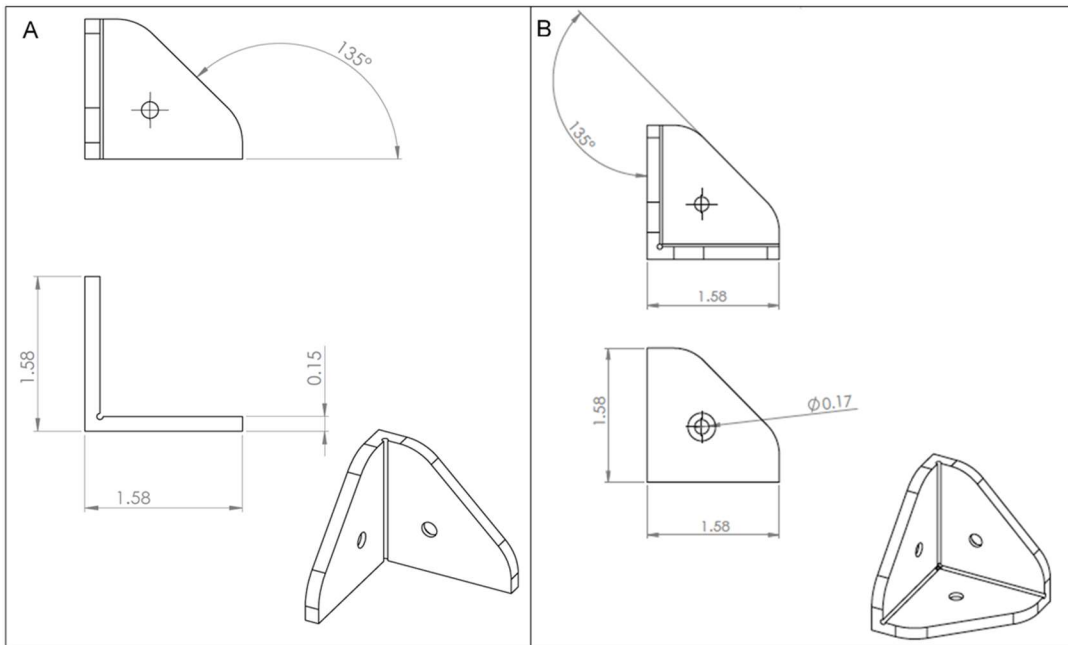


Figure S4. Dimensions (in millimeters) of 3D printed corner options: **(A)** double corner to affix panels adjoining the door and **(B)** multi corner used to secure three panels throughout the remainder of the collection unit.

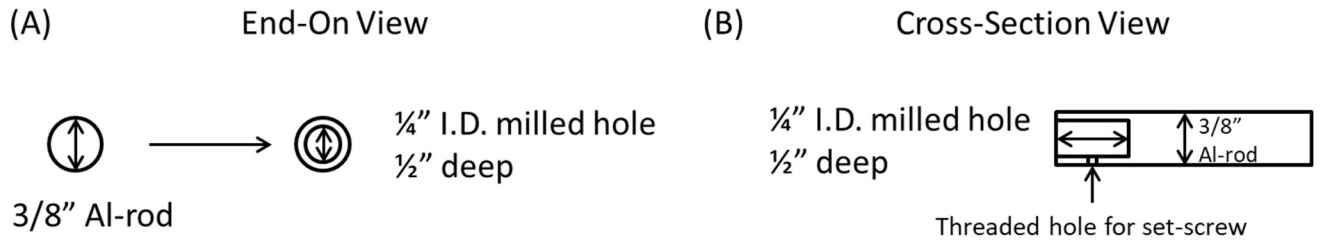


Figure S5. Aluminum rod milling schematics showing **(A)** end-on view for motor drive shaft and **(B)** cross section for situating the setscrew to hold the rod to the drive shaft of the motor.

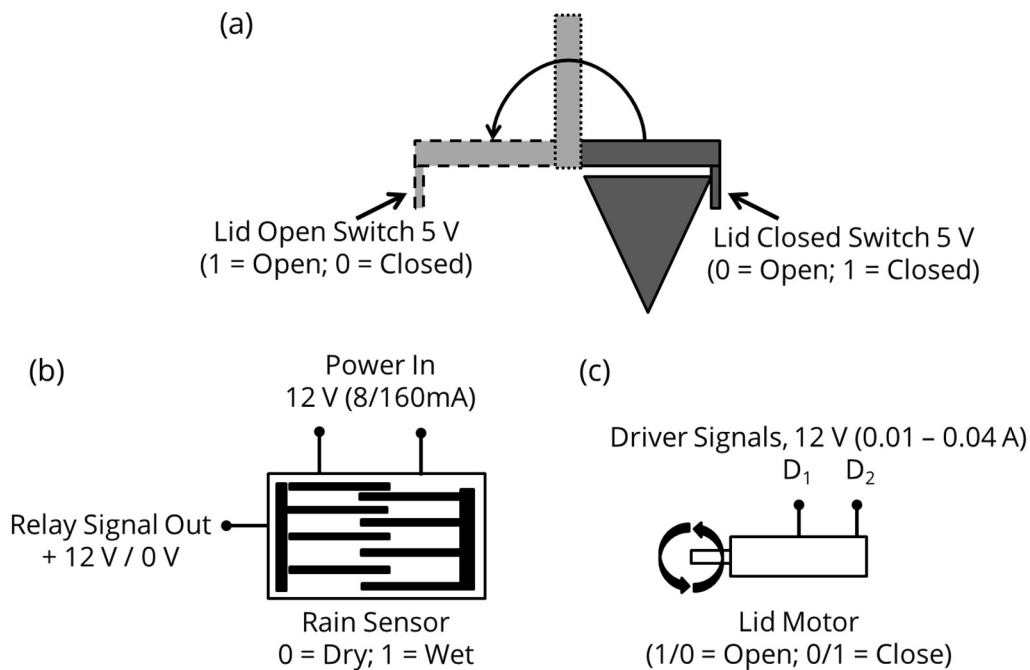
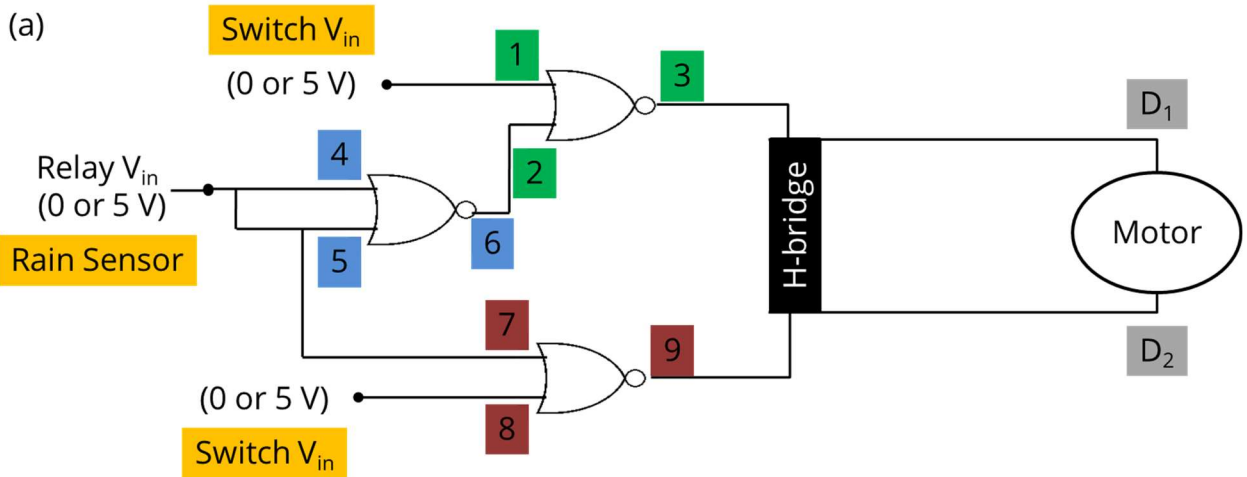


Figure S6. Voltage and logic states of the **(a)** lid switches, **(b)** precipitation sensor, and **(c)** lid motor used to drive digital decision making on custom control boards.



(b)

Switch	Switch	Rain	1	2	3	4	5	6	7	8	9	D1	D2
1	1	1	X	X	X	X	X	X	X	X	X	X	X
1	1	0	X	X	X	X	X	X	X	X	X	X	X
1	0	1	1	0	0	1	1	0	1	0	0	0	0
1	0	0	1	1	0	0	0	1	0	0	1	0	1
0	1	1	0	0	1	1	1	0	1	1	0	1	0
0	1	0	0	1	0	0	0	1	0	1	0	0	0
0	0	1	0	0	1	1	1	0	1	0	0	1	0
0	0	0	0	1	0	0	0	1	0	0	1	0	1

Figure S7. (a) Input layout for digital logic control of precipitation sampler using three NOR gates to operate a H-bridge motor driver and **(b)** corresponding truth table to input and output logic states for switches and rain sensor (yellow), first NOR gate (green), second NOR gate (blue), and third NOR gate (red) to control 12 VDC from the motor driver to the motor terminals (D₁ and D₂).

S1. Measuring Conductivity Threshold of Kemo Electronic Rain Sensor.

A simple procedure determined the conductivity threshold of the 12 VDC Kemo Electronic rain sensor. A 1000 $\mu\text{S}/\text{cm}$ stock standard was prepared by dissolving 0.225 g of sodium chloride in 500 mL deionized water, ensuring thorough mixing for homogeneity. A series of conductivity standards were then prepared by diluting the stock standard with deionized water (DIW) to different levels near the suspected threshold of the sensor (0.75, 0.8, 0.9, 1.0, 2.0, 3.0, 4.0, and 5.0 $\mu\text{S}/\text{cm}$). Prior to this testing, the rain sensor was flushed with copious amount of DIW to remove any conductive substance from the surface of the sensor - this was also repeated in between trials. The result from testing with these solutions determined that the conductivity threshold for the relay is 1.0 $\mu\text{S}/\text{cm}$, as the rain sensor was observed to be only partially activated at 0.9 $\mu\text{S}/\text{cm}$ and the result was not consistent.

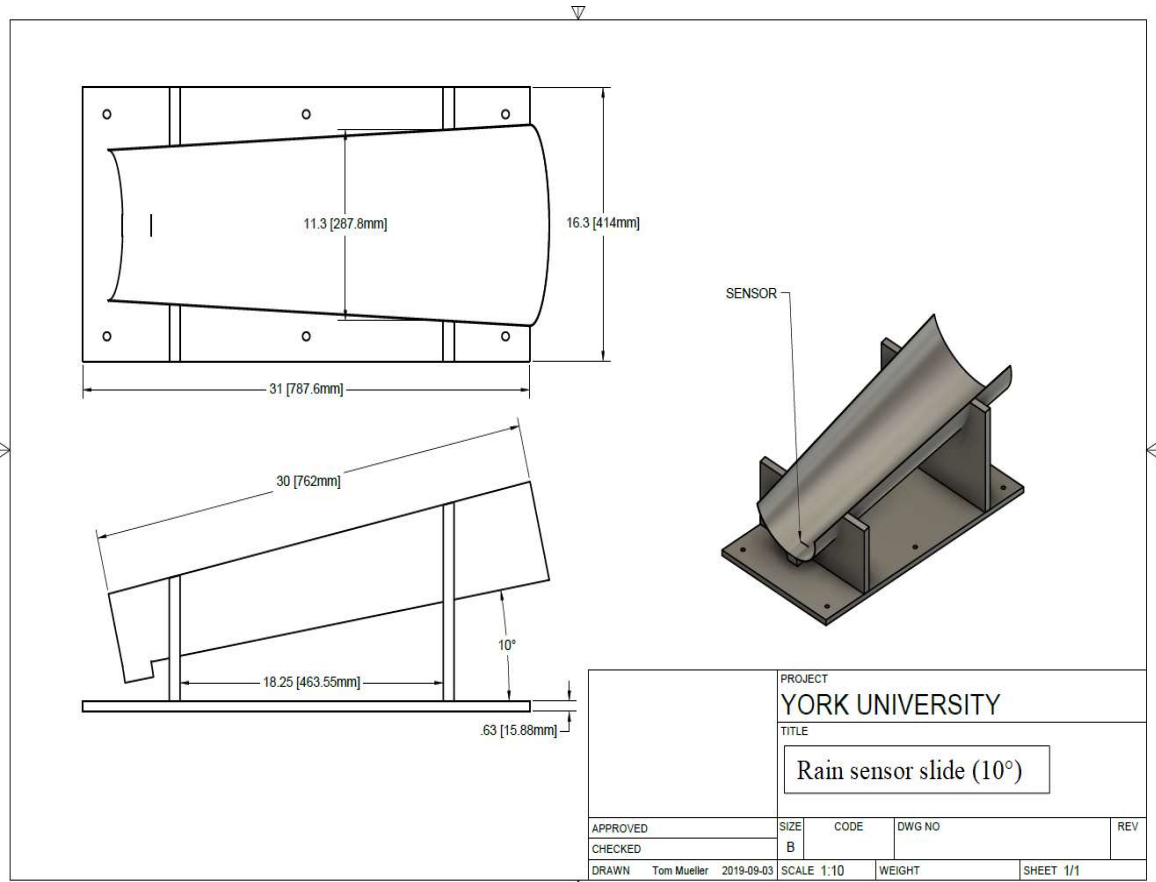


Figure S8. Design and dimensions (in mm) of the mounting chute for the rain sensor. The chute slope is fixed at an optimized angle of 10° , that maintains water on the surface of the sensor throughout a rain event but allows excess water to flow over the sensor when necessary. We also tested 15° and 20° through manual observation but found that rainwater would flow over the sensor too rapidly – limiting or even preventing detection of wet deposition events.

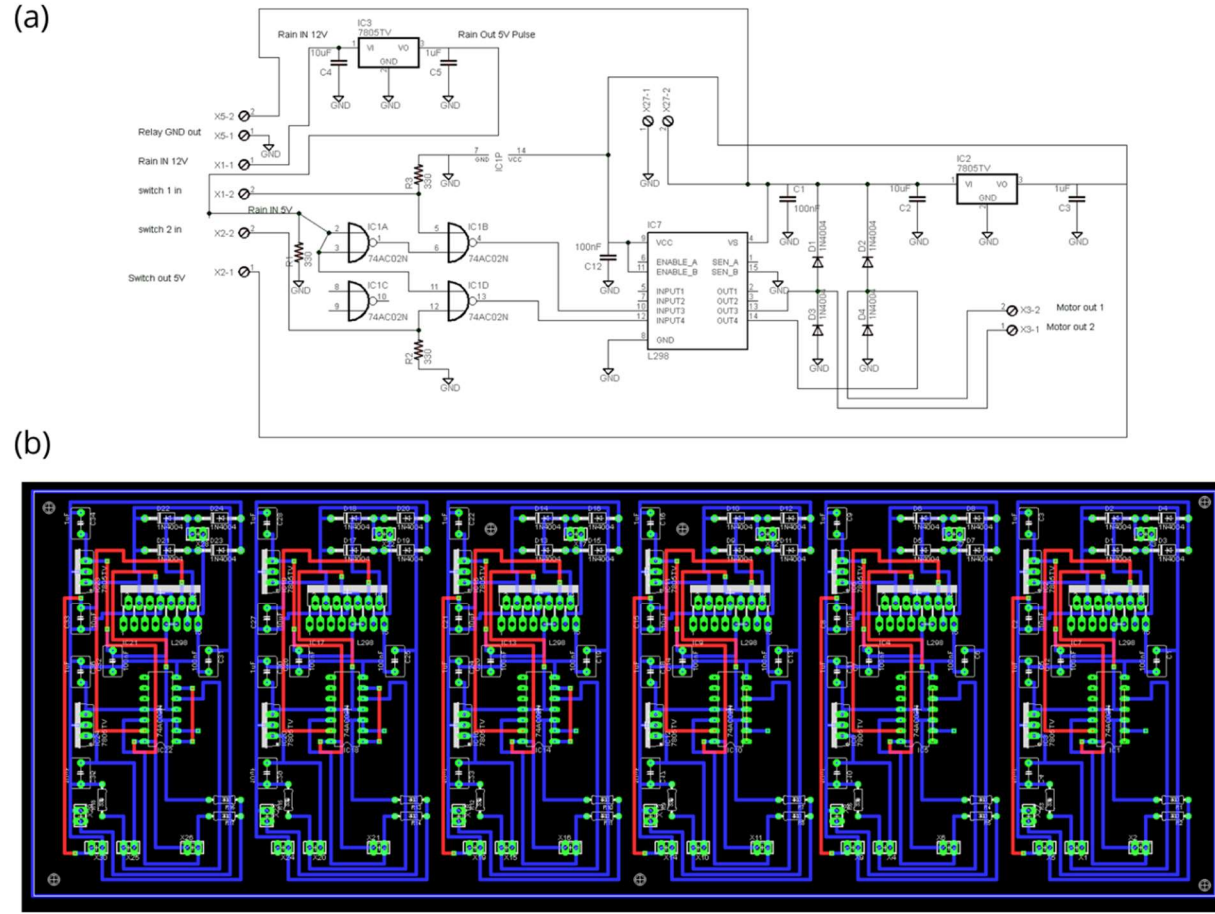


Figure S9. Technical schematics for all printed circuit board components required to control (a) one precipitation collector with all components and connections to chips indicated and (b) a complete six-unit array composed of replicate layouts.

Table S1. List of required components, part numbers, and quantities required to assemble a single printed circuit board.

Component	Part Identification	Description	Quantity	Unit Price (CAD)
Integrated Circuit 1 (IC1)	74AC02N = SN7402n	Four channels, two input NOR gate	1	3.52
Integrated Circuit 7(IC7)	L298N	Dual Full Bridge Driver	1	7.41
Integrated Circuit 2 (IC2), Integrated Circuit 3 (IC3)	7805TV = MC7805CT-BP	5V regulator, TO220	2	0.67
Diode 1, Diode 2, Diode 3, Diode 4	1N4004	Diode	4	0.14
Resistor 1, Resistor 2, Resistor 3	10k	resistor, %1, 1/4W	3	0.21
Capacitor 1, Capacitor 12	100nF cap	Capacitor, 0.1Uf / 50V	2	0.35
Capacitor 3, Capacitor 5	1Uf Cap	Capacitor, 1UF, 50V	2	0.84
Capacitor 2, Capacitor 4	10Uf	Capacitor, %10,10Uf,50V	2	1.84

Table S2. List of specific components, manufacturers, and specifications required to construct precipitation collection arrays.

Components	Description	Selection Rationale
Collection Units		
Boxes	<p>Material: 3/8" Acrylic or Plywood</p> <p>Dimensions: (33.78 W x 33.78 L x 42.94 H) cm</p> <p>See Figure 2 for more details</p>	Acrylic and plywood were chosen for their low cost and durability against a variety of environmental conditions. Both materials are opaque to minimize light intrusion.
Collection Container	<p>Material: High Density Polyethylene</p> <p>Brand: Bel-Art Products</p> <p>Dimensions: (19 H x 26 W x 36 H) cm; Volume: 10 L</p>	Large volume collection of precipitation and resistance of material to environmental degradation. Quantitatively transfers analytes such as inorganic ions and DOC.
Funnel	<p>Material: High Density Polyethylene</p> <p>Brand: Dynalon</p> <p>Dimensions: 9.4" W x 8.68" H; 20 cm diameter</p>	Guides conductive precipitation into the collection container. Material is matched to maintain quantitative analyte transfer.
Sampler Lids	<p>Materials: Lexan</p> <p>Brand: Kraloy, Carlon[®] Lamson & Sessions</p> <p>Dimensions: See Figure S3 for more details</p> <p>Voltage: 12 V</p>	Prevents foreign material entering the collection container when closed. Lexan was chosen due to its high flexibility and impact resistance. Acrylic was determined to be too fragile and not suitable for environmental conditions with elevated winds.
Geared Box DC Motor	<p>Brand: Tsiny Motor Industrial Co. Ltd. TS-32GZ370-1650</p> <p>Operating voltage range: 6-24 VDC</p> <p>Nominal voltage: 12 VDC Speed: 2-8 RPM</p>	The worm gear motor is connected to the aluminum rod and sampler lids to open and close them when conductive precipitation is detected by the rain sensor.

Table S2 (cont'd). List of specific components, manufacturers, and specifications required to construct precipitation collection arrays.

Components	Description	Selection Rationale
Collection Units (cont'd)		
Lid Rod	<p>Material: Aluminum</p> <p>Outer diameter: 3/8"</p> <p>Inner diameter: 1/4"</p> <p>Depth: 1/2" deep</p>	The aluminum rod is attached to the motor and facilitates the opening and closing of the Lexan lids. The rod is attached to the motor by a 8/32 threaded 1/4" set screw.
Lid Limit Switches	<p>Brand: Omron Electronics</p> <p>Part Number: D2FW-G271M(D)</p> <p>Max Current: 1 A</p>	Snap action switches are used to detect the open and closed position of the motorized sampler lids. Weatherproofing keeps them watertight.
Funnel Mesh	<p>Material: HDPE</p> <p>Brand: McMaster-Carr</p> <p>Part Number: 9265T49</p>	When the automated sampler is in the open position, these cone-shaped filters prevent debris from entering the collection jug.
Lag shields, lag bolts and washers	<p>Lag shields dimensions: 3/8" x 1-3/4"</p> <p>(Requires masonry drill bit: 5/8")</p> <p>Lag bolt and washer dimensions: 1/2"</p>	<p>These are utilized to secure precipitation collectors to concrete. This prevents unit tipping during storms or from animal interactions.</p> <p>Tent pegs or rebar may be alternatively used to secure the samplers into soils.</p>
Power System		
Solar Panel	<p>Brand: Coleman 40W Folding Panel</p> <p>Output Voltage: 17.1 V, 2.3 amps</p> <p>Dimensions: (79 L x 35.1 W x 1.8 H) cm</p>	The solar panel recharges the off-grid battery. This can be repositioned to optimize sunlight exposure and maximize recharging capabilities.

Table S2 (cont'd). List of specific components, manufacturers, and specifications required to construct precipitation collection arrays.

Components	Description	Selection Rationale
Power System (cont'd)		
Solar Charge Controller	Brand: Coleman Load Charge: 8.5 A Cut-out: 14.2 V Cut-in: 13 V	The ability of the solar panel to deliver charge to the battery or stop when it is fully charged is regulated. The controller prevents the battery from overcharging.
103 Ah off-grid battery	Brand: Motomaster Nautilus Ultra XD Group 31 High Performance AGM Deep Cycle Battery (103 Ah) Voltage: 12 V Dimensions: (33 L x 18 W x 25 H) cm	The 99.99% lead AGM (absorbed glass mat) was used for long lasting power and reliability for field testing without a solar panel.
76 Ah Off-grid battery	Brand: Motomaster Nautilus Ultra XD Group 24 High Performance AGM Deep Cycle Battery (76 Ah) Voltage: 12 V Dimensions: (27.6 L x 17.1 W x 22.2 H) cm	The 99.99% lead AGM with 76 Ah battery was used for field testing while being charged with a solar panel for long periods.
Transformer	Brand: TDK Lambda Americas Voltage: 115 VAC to 12 VDC	The transformer replaces an off-grid battery and converts 115 VAC to 12 VDC. This can be used when sampling in locations with grid power available.

Table S2 (cont'd). List of specific components, manufacturers, and specifications required to construct precipitation collection arrays.

Components	Description	Selection Rationale
Rain Detection		
Heated Precipitation Sensor	Brand: Kemo Electronic M152 Voltage: 12 V Dimensions: (6.5 x 4.5 x 3.6) cm	A waterproof conductive sensor that becomes activated and switches on a relay when in contact with conductive rain or snow. The sensor relay triggers the control board so that the sampler lids rotate open. The operating temperature range provided by the manufacturer is -10 to +55°C.
Sensor Chute	Dimensions: (see Figure S8)	Increases surface area of precipitation sensor to activate the relay. Delays lid closing by providing more conductive ions to the sensor surface when atmospheric wash out may occur.
Digital Control Board		
Control Boards	See Figure S9 for more details.	Controls communication between the motors, power supply, and rain sensor (12 VDC), as well as the two limit switches (5 VDC).
Control Board Protective Case	Brand: Pelican™ 1450 Case Material of body: polypropylene Exterior Dimensions: (41.8 L x 33 W x 17.3 D) cm Interior Dimensions: (37.2 L x 26 W x 15.5 D) cm Weight with foam (polyurethane): 2.9 kg	The 1450 case contains custom foam for cable accommodations with automatic pressure equalization valve to balance interior pressure. It is watertight, crushproof, and dustproof with double throw latches and a rubber handle.
Reusable Hydrosorbent Silica Gel Beads	Brand: Pelican Case Desiccant Canister Dimensions: (1.3 L x 5.1 W x 10.2 H) cm Mass: 40 g	The silica gel beads absorb moisture to prevent water damage to the control board; when clear, must reactivate in an oven for 3 hours at 300°F.

Table S2 (cont'd). List of specific components, manufacturers, and specifications required to construct precipitation collection arrays.

Components	Description	Selection Rationale
-------------------	--------------------	----------------------------

Miscellaneous

Data logger	Brand: ONSET® HOBO 4-channel analog data logger Part #: UX120-006M Program: HOBOWare	To collect and offload data from the control board using one of four channels. This work: CH1 – battery voltage, CH2 – motor driver voltage, CH3 – rain sensor voltage, CH4 – limit switch voltage.
Cable Connectors	Brand: Bulgin (400 or 4000 Series Buccaneer) IP68 Sealed Electrical Circular Cable Connectors PXP4010/03P/3540 and PSP4011/03S/3540 PSP4010/08P/3540 and PSP4011/08S/3540 SA3349, SA3350, SA3348, SA3347 Material: Polyamide UL94-V0 body, gold contacts Cable acceptance: 3 mm – 7 mm Contact insertion: crimp or solder	Waterproof and dustproof cable connectors provide electrical connections from the control board to the battery, solar panel, and sampler motors and switches. Different numbers of contacts represent different cable requirements within the system (2- or 3-contact = battery and solar panel cables, 6- or 8-contact = rain samplers).
Extension Cables	Any commercial outdoor-rated extension cord	To increase distance from VAC power source to sampling location.

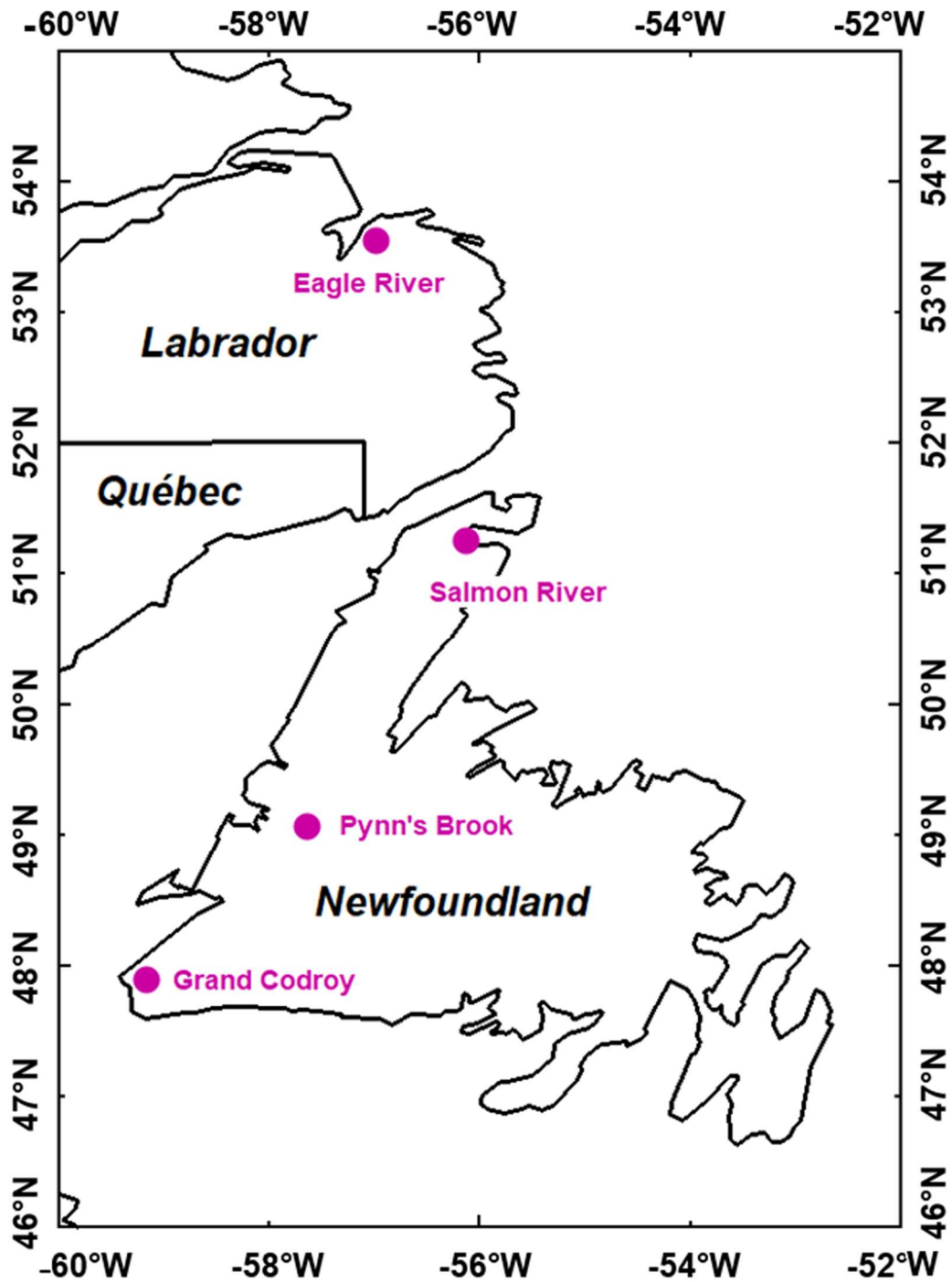


Figure S10. Map of the NL-BELT highlighting the four watershed regions where sampling occurred: Grand Codroy (GC), Pynn’s Brook (PB), Salmon River (SR), and Eagle River (ER).

S2. Deposition Comparison: This Work, DAYMET, and ECCC

Sufficiently continuous measurements from ECCC stations near each site are challenging to obtain for the 2015-16 period. When available with greater than 80% coverage, the ECCC datasets both agree and disagree with our observations in GC and SR, respectively, suggesting that there is substantial deposition volume heterogeneity at the scale of ~10 km in this region. In SR, the disagreement with our measurements is identical to the DAYMET model which uses ECCC observations as input data, while at GC the ECCC measurements are identical to ours (Table S3). Further, the discrepancy in the PB or ER average annual precipitation volume between ECCC and those of this work and DAYMET are not possible to interrogate due to the large quantity of data missing from the ECCC monitoring station (35.22% in ER and 39.65% in HR/PB; Table S3). The DAYMET observations are representative of a larger spatial scale, where our discrete samplers could be subject to heterogeneity in deposition (e.g., orographic precipitation, driven by topography like steep slopes) or impacted by meteorological conditions not captured by the model (e.g., undercatch driven by local winds). The temporally resolved volume comparisons at sampling interval timescales better-demonstrates comparability, despite the systematic differences. **The month-to-month relationships between DAYMET and our observations, as well as between ECCC and our observations, all show strong correlations with linear regressions having R² values, from north to south, of 0.72, 0.99, 0.99, and 0.86, and N/A, 0.94, N/A, and 0.93, respectively (Table S3).** The discrepancy between DAYMET, ECCC, and our observations for total deposition were highest in the most northerly site, where the experimental site was located on a steep slope, with only 43 % of the predicted volume collected. At all of the sample collection sites on the island of Newfoundland, a consistent difference was observed with 65 ± 4 % of the estimated volume collected, except at GC where our measurements and those from ECCC are identical and starkly contrasting to DAYMET. Overall, parsing these comparisons is difficult and demonstrates that there may be up to 55% additional uncertainty in deposited species, should given measurements of a species be scaled for a watershed like ER by concentration in total deposition samples. We propose, that by isolating only the deposited analytes and using analyte fluxes instead of concentrations in precipitation samples, that uncertainty issues in representing volumes, improves overall deposition budget certainties.

Table S3. Collected precipitation volumes from NL-BELT in bulk deposition samplers for rainwater were deployed for one to two months, while snow was collected as an integrated sample throughout the winter because sites were not accessible. The Environment and Climate Change Canada (ECCC) precipitation data was obtained for identical sampling intervals. The DAYMET model for North America (1 km x 1 km resolution) for precipitation was obtained for the identical sampling intervals, which utilizes interpolated and extrapolated data from daily weather observations to predict inputs at the NL-BELT locations. Linear regression results for slope (m) and correlation coefficient (R²) between observations and DAYMET (*italics*), and observations and ECCC (where possible; underlined), were calculated. For sampling periods where a measurement was compromised or not collected for a given interval in this work, these are reported as with ‘-’ and an estimated volume from the regression relationship with ECCC is reported in parentheses when used to replace compromised samples.

Date (mmm- yy)	Grand Codroy			Pynn’s Brook/ Humber River			Salmon River			Eagle River		
	This Work (L)	DAYMET (L)	ECCC (L)	This Work (L)	DAYMET (L)	ECCC (L)	This Work (L)	DAYMET (L)	ECCC (L)	This Work (L)	DAYMET (L)	ECCC (L)
Jun-15	22.68	40.98	23.80 ^b	23.44	36.94	11.47 ^b	-	-	-	-	-	-
Jul-15	2.92	3.38	2.06	2.24	2.93	1.04 ^b	21.12	31.13	30.94	13.85	40.92	15.09 ^b
Aug-15	6.86	8.08	7.20	5.06	6.86	-	1.89	4.71	9.52	3.66	6.96	2.64 ^c
Sep-15	(5.17) ^a	2.93	5.17	3.05	4.70	-	5.29	6.62	6.44	-	-	-
Oct-15	1.71	4.31	4.44	-	-	-	3.62	5.20	6.18	-	-	-
Nov-15	3.57	5.92	4.92	5.66	8.61	-	-	-	-	-	-	-
May-16	26.00	28.15	25.38	15.62	24.18	11.22 ^b	16.19	22.73	25.33	-	-	-
Jun-16	1.83	4.21	3.79	4.25	4.51	2.79	2.39	3.11	3.05 ^a	22.94	45.68	22.70 ^b
Jul-16	5.81	3.96	6.10	3.13	3.80	3.19	3.68	3.26	-	-	-	-
Aug-16	5.19	4.10	3.98	4.67	6.14	4.23	3.60	5.18	3.89 ^a	5.82	10.79	6.94 ^b
Sep-16	7.79	5.17	4.06	4.97	4.04	-	4.95	5.95	6.69	-	-	-
Oct-16	5.10	6.69	5.30	5.15	5.84	3.47 ^b	2.00	2.82	2.58	5.42	8.32	4.39 ^b
m, R ²	0.62, 0.859; 1.02, <u>0.934</u>			0.60, 0.990; -, =			0.68, 0.987; 0.67, <u>0.941</u>			0.43, 0.719; -, =		

^aSample compromised by wildlife-sampler interaction.

^bThese values result from periods where <80% of the observation days report collected data. Partial volumes from the available data are presented.

^cThese values result from periods where <50% of the observation days report collected data. Partial volumes from the available data are presented.

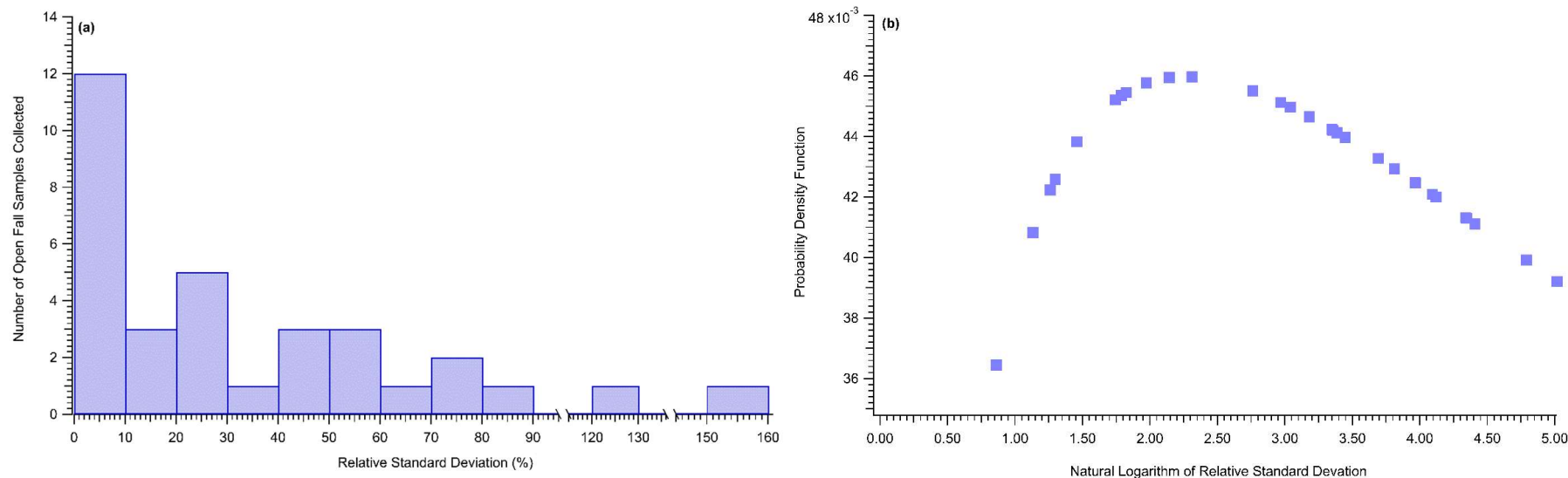


Figure S11. Thirty-three sets of triplicate open fall samples were collected across the NL-BELT between 2015 and 2016. **(a)** The replicate relative standard deviations (RSDs) appear to follow a log-normal distribution where reproducibility is typically within $\pm 12.5\%$ and almost always within $\pm 31.5\%$. **(b)** A variable ‘x’ is said to have a lognormal distribution if $y = \ln(x)$ is normally distributed. The RSDs were treated as variable ‘x’ and then log-transformed, resulting in a mean (μ) and standard deviation (σ) of 2.81 and 1.41, respectively. The probability density function for the log-normal, plotted against these log-transformed values, is defined by the two parameters μ and σ , where $x > 0$. As a result, one of the 33 RSD values was excluded from this plot as it became a negative value after log-transformed.

Table S4. Sum of DOC fluxes ($\text{mg C m}^{-2} \text{a}^{-1}$) in wet deposition samples for 2015 and 2016 at GC, PB, SR, and ER. The average DOC flux error ($\text{mg C m}^{-2} \text{a}^{-1}$) propagated by summing the errors across the sampling sites to obtain the annual values.

Site	Year	OF	TF
		Deposition Flux ($\text{mg C m}^{-2} \text{a}^{-1}$)	
GC	2015	1800 (± 800)	3700 (± 2400)
	2016	1400 (± 800)	700 (± 700)
PB	2015	600 (± 300)	1800 (± 600)
	2016	7800 (± 7700)	3000 (± 3600)
SR	2015	2000 (± 900)	3500 (± 2000)
	2016	4000 (± 1700)	1700 (± 1500)
ER	2015	100 (± 100)	1600 (± 2200)
	2016	1000 (± 800)	100 (± 30)

Table S5. Average DOC flux ($\text{mg C m}^{-2} \text{a}^{-1}$) for 2015 to 2016 at GC, PB, SR, and ER. The difference between calculated fluxes in TF and OF samples are also included.

Site	OF	TF	TF - OF
	Flux ($\text{mg C m}^{-2} \text{a}^{-1}$)		
GC	1600	2200	600
PB	4200	2400	-1800
SR	3000	2600	400
ER	500	900	400