Responses to reviewer #2 comments

Overview:

This study is a thorough analysis of several factors influencing the calibration of low-cost sensors: concentration range, calibration duration, and time averaging, and provides recommendations for each. The study utilizes the MAS sensors, which have a built-in in 'dynamic baseline tracking' feature that promises to eliminate (reduce?) the effects of environmental conditions such as temperature and humidity.

[Response]: We wish to thank the reviewer for the positive and valuable comments. We have carefully considered all suggestions when preparing this response and revising the manuscript. The incorporation of the reviewer's suggestions has led to a substantially improved manuscript, for which we are grateful. Below we provide a point-by-point response (in blue text) to the reviewer's comments (black text) and summarize the changes that have been made in the revised track-changed manuscript. The quoted texts from the revised manuscript with the change tracked are in purple text.

On the evidence of elimination or reduction of the environmental impact, we have given a detailed response in Comment 3. In general, we agree with the reviewer that 'elimination' is a strong claim and the current evidence from our long-term tests has only shown PDF-enabled sensors have very significant improvement in mitigating the environmental impact in the range of test conditions, while other more diverse and extreme conditions may be further investigated in the future to verify the performance of PDF sensors.

General comments:

[Comment]: 1. *The dynamic baseline tracking method described in section 2.2 is fascinating, and more detail in the explanation would help the reader understand it better. How does the PDF work, and can any information be shared on its accuracy at filtering out gases? Is it more or less accurate for any specific gases?*

[Response]: Thank you for your careful review. We have further clarified the details of the dynamic baseline tracking method as the reviewer has suggested. The method is made possible by the pair differential filter (PDF) technology from the manufacturer and we also gave more details in the responses to explain the working principle of PDF sensors. Our prior laboratory evaluations have demonstrated the effectiveness of the PDF sensor in filtering out target gases while maintaining the free passage of water molecules. For instance, when target gases such as CO, NO, NO₂, O₃, and SO₂ at varying concentrations are injected into the PDF sensors, no sensor response signal is observed showing the effectiveness of the method. The revised manuscript is as below:

Line 135-145:

The sensor device (MAS, Sapiens) has deployed a novel gas sensing technology that enables the isolation of the concentration signal from environmental variables of temperature and RH through a patented dynamic baseline tracking method by the manufacturer, which operates by differentiating the varying environment and target pollutant induced sensor signals using a dual-sensor module. Figure 2 shows the conceptual diagram of MAS sensor module and general working principle of the dynamic baseline tracking method. This gas sensor system comprises a primary sensor – that is directly exposed to the air, capturing the original signal (designated as ORG) influenced by varying pollutants, temperature, and RH - and a proprietary pair differential filter sensor (designated as PDF) to track the dynamic baseline signal driven only by temperature and RH. The PDF sensor is equipped with water molecule permeable membrane that allows the water vapor to penetrate through while filtering out the target gas modules from entering the sensor head. The differential signal (measured in volts) between the ORG and PDF sensors decouples the temperature and humidity effects, yielding a pure signal that reflects target gas concentrations.

[Comment]: 2. *Figure 2 is helpful for understanding this, but in the upper right plot, is there really zero difference between ORG and PDF in the lab? If not, a similar figure showing real data from the lab is important for readers understand how perfect or imperfect the method is, even if in the supplemental. It is not clear from the figure what "laboratory conditions" in the right panel means – were temperature, pressure, or humidity held constant, or were all fluctuating?*

[Response]: We appreciate your feedback regarding Figure 2. In response to your comment, we have reorganized the figure content and expanded the figure caption to provide more detailed descriptions. As the figure serves as a conceptual diagram illustrating the method, it does not specify the laboratory conditions in detail. To clarify, the tests were conducted under varying temperature and RH cycles, while pressure was not considered in the tests. We have included the test data from laboratory tests in the supplementary materials, along with an explanation of the tests provided in Lines 253-259. The updated manuscript includes a detailed explanation as follows:

Line 165-174:

Figure 2. A conceptual diagram of the PDF-enabled MAS sensor device. In laboratory tests, standard gas with constant concentrations is periodically injected into the PDF and ORG sensors throughout varying temperature and RH cycles to investigate their effects on the sensor performance. The PDF tracks the baseline signal driven only by temperature and RH, while the ORG sensor captures the concentration profile influenced by both the target gas module and environmental conditions. The differential signal between the ORG and PDF sensors decouples the baseline signal induced by temperature and RH, producing a pure signal that reflects the target gas concentrations. This concept is also applicable to ambient conditions, where the differential signal between the paired ORG and PDF sensors demonstrates the accuracy and robustness of PDF technology for ambient air monitoring.

Line 253-259:

Additionally, laboratory tests in environmental chambers assessed the MAS NO sensor (Figure S3), exposing it to broad temperature (0^oC to 30^oC) and RH (10% to 90%) ranges. Despite these fluctuations, MAS sensors maintained consistent and stable readings after applying the dynamic baseline tracking method, as shown in Figure S3(b), with concentration steps from 50 to 300 ppb. The outcomes from both field and laboratory tests confirm that the dynamic baseline tracking method effectively neutralizes temperature and RH effects, primarily for NO₂, NO, and O₃ sensors, achieving desired performance while focusing primarily on concentration factors for subsequent analysis. Similar pre-tests were also conducted with the MAS units in Macau and Shanghai to assess the effectiveness of the dynamic baseline tracking method.

Figure S3. (a) Laboratory environmental chamber setup and (b) the response of 3 MASs' NO sensors under multiple point concentrations in laboratory temperature and humidity test.

[Comment]: 3. *The authors later state that, "the influence of temperature and RH on sensor signals has been eliminated". Can you prove to the reader with real data that this is entirely eliminated, or to a certain extent eliminated? In theory, Figure 3 could help answer this for the ambient data, but it is hard to read. Figure 3 panel F is the only one I can somewhat make out the difference between solid and dotted lines for. Moving the black reference data to the back of these plots might help make the other colors and lines more visible, but additional edits might be necessary for readability.*

[Response]: Thank you for your valuable suggestion. To enhance the visibility of Figure 3, we have updated it in Section 3.1 to focus on one of the four tested MAS sensors as a representative example of PDF technology's robustness from our test results. The previous version has been moved to the supplementary material as a consistency overview across all four MAS performances. The new Figure 3 separately illustrates the outputs from the PDF sensor, the ORG sensor, and the differential output between the paired ORG-PDF sensors. This addition, in conjunction with the conceptual diagram in Figure 2, aims to clarify how the PDF technology facilitates dynamic baseline tracking method.

Additionally, we have included more detailed analysis of the three separate outputs. From the PDF sensor output shown in Figure 3, we observed that the outputs for all gas pollutants did not exhibit a linear relationship with temperature or RH profiles. Different sensor types demonstrated distinct response patterns to variations in temperature and RH. These findings highlight the complex non-linear characteristics of the regular electrochemical sensors in relation to baseline dependence on these environmental factors, while the PDF enabled sensor output reveals a clear gas concentration profile that aligns closely with reference data. The significantly higher R² values and lower RMSE for the ORG-PDF sensor output compared to the ORG sensor output indicate that the influence of temperature and RH on sensor signals has been effectively mitigated. Additionally, Section 3.1 includes data from laboratory tests conducted in environmental chambers (see Figure S3). It also provides an overview of the longterm (1-year) co-location performance data, as shown in Figures S4 and S5. The outcomes from both field and laboratory tests confirm that the dynamic baseline tracking method effectively neutralizes temperature and RH effects, primarily for $NO₂$, NO, and $O₃$ sensors, achieving desired performance while focusing primarily on concentration factors for subsequent analysis. We agree with the reviewer that 'elimination' is a strong claim and the current evidence from our long terms tests have only shown PDF enabled sensors have very significant improvement in mitigating the environmental impact in the range of test conditions, while other more diverse and extreme conditions may be further investigated in the future to verify the performance of PDF sensors. We have revised the manuscript and use 'significantly mitigated' instead of 'eliminated' to be in alignment with the tests in the study.

To better convey the focus of this research and avoid any potential misunderstandings, we have revised also the title of the manuscript to: "Performance Validation and Calibration Conditions for Novel Dynamic Baseline Tracking Air Sensors in Long-term Field Monitoring." We revised the relevant statement in the revised manuscript as follows:

Line 229-252:

We tested four MAS units and presented findings from this one MAS as an example to evaluate the robustness of the PDF technology. During the 15-day pre-test in the summer (June 1-15, 2019), temperatures varied between 28 ˚C and 42 ˚C, with RH levels from 45% to 87%. The outputs from the PDF sensor, the ORG sensor, and the differential output between the paired ORG - PDF sensor are illustrated separately in Figure $3(a)$ -(d). The voltage signals from the PDF and ORG sensors were converted into concentration outputs using coefficients derived from Eq. (1). As shown in the figure, even during the typical ambient concentration ranges, the accuracy of the ORG sensor outputs for gases other than CO was notably poor, primarily due to significant influences from field temperature and RH. It was observed that the PDF sensor outputs for all gas pollutants did not exhibit a linear relationship with temperature or RH profiles. Different sensor types demonstrated distinct response patterns to variations in temperature and RH, highlighting the complex non-linear characteristics of electrochemical sensors in relation to baseline dependence on these environmental factors.

With the PDF enabled sensors, the physical separation of the climatic driven baseline and target gas driven sensitivity is demonstrated to be feasible and effective. By subtracting the output of the PDF sensor from that of the ORG sensor, the resulting ORG – PDF output reveals a clear gas concentration profile that aligns closely with reference measurements. This relationship is illustrated in the scatter plots presented in Figure 3(f)-(i). For NO_2 , the $ORG - PDF$ sensors showed stronger performance, with a high R^2 (0.99) and low RMSE (0.94), compared to the lower R^2 (0.44) and higher RMSE (5.80) for the ORG sensors without the PDF module. For NO and O3, the ORG – PDF sensors also demonstrated stronger performance compared to the ORG sensors without the PDF module. Specifically, the ORG – PDF sensors had strong R^2 $(0.97$ for both NO and O_3) and low RMSEs $(1.72$ for NO, 1.05 for O_3), while the ORG sensors without the PDF module had weaker R^2 (0.73 for NO, 0.59 for O₃) and higher RMSEs (5.37 for NO, 4.18 for O₃). For CO, the sensors exhibited comparable performance, with R^2 around 0.93-0.94 and RMSE values between 16.70-19.00, regardless of the PDF module. We tested four MASs and the other PDF enabled sensors were shown in Figure S2. Their data quality performance has been consistent with the findings reported data here. These significant discrepancies between the ORG sensor output and ORG – PDF sensor output, especially for NO, NO2, and O3, highlight the importance of the dynamic baseline tracking method in improving the accuracy and reliability of measurements, notably under low concentration conditions influenced by temperature and RH.

Figure 3. (a-d) Performance validation of the MAS's ORG and PDF sensors for detecting NO₂, NO, CO, and O₃ under field conditions in 2019. (e) Displays the temperature and RH measured inside the MAS gas sensor modules. (f-i) Compares the readings from the ORG sensor and the MAS PDF-enabled sensor with reference measurements.

For details, please refer to section 3.1 in the revised manuscript.

[Comment]: 4. *The analyses of concentration range, calibration period, and time averaging are thoughtful and well-explained. The R2 and RMSE of the validation data are reported. Would these results change if you applied the best model from one location and applied it to another location? As the authors state, it is important to replicate the conditions of the deployment to the best extent possible during the colocation, but lack of availability of reference instruments in certain locations can make this difficult. The analysis shown here could be made more practical by showing examples of how these trends may deviate as lowcost sensor users frequently have to adhere to non-ideal constraints. The recommendations provided for each of these are well thought out for the best case scenario of being able to colocate exactly where the deployment will take place, but I am left wondering if these recommendations would still apply in a real-world scenario.*

[Response]: Thank you for your insightful and detailed comments. The reviewer raises a valid point regarding whether these results vary across diverse locations. The strategically deployed eight sensor devices in three different locations allow us to verify whether optimized calibration conditions differ under varying environmental settings. Previously, we combined these sensors together for analysis, and there was limited discussion on how these results varied across diverse sensors. In response to this comment, while retaining the previous results, we have incorporated separate analyses of each MAS sensor results within our analysis of the three impact factors.

Firstly, regarding calibration period optimization, the previous conclusions about the 5-7 day collocations were based on an average pattern derived from the combined data of all sensors. We have now expanded our discussion to address the differences in calibration periods observed among various regions, as illustrated in Figure S6. The updated content is as follows:

Line 296-307:

The aforementioned results are based on an average pattern derived from the combined data of all sensors. Figure S6 presents the separate performance of all eight MAS sensors over varying calibration periods. The NO₂, NO, CO, and O₃ sensors in MAS1-4 in Hong Kong and MAS5-6 in Macau exhibited trends consistent with those shown in Figure 4. A noteworthy observation in Figure $S6(a)$ -(b) is that the $NO₂$ and NO sensors in MAS7 and MAS8 of Shanghai campaign showed consistent performance over all calibration periods, lacking the trends observed in Figure 4. Considering that the NO and NO₂ concentrations in the site of Shanghai are significantly higher than those in the other two cities, it is hypothesized that the elevated pollutant concentrations at the Shanghai port provided a more favorable calibration condition, thereby diminishing the contribution of the calibration period. Thus, we conclude that for calibration condition with a narrower concentration range, a calibration period of at least 5 to 7 days is necessary, whereas more polluted ambient environments are more conducive to sensor concentration calibration. Despite the short calibration duration of 1–3 days, the extensive concentration range assessed contributed to more precise calibration coefficients and improved validation performance, as will be discussed in the next section.

Secondly, in the concentration range analysis, we discussed two distinct concentration scenarios: MASs 1-6 in Hong Kong and Macau were evaluated together in Figure 5 under a lower concentration range, with 90% of NO₂ and NO measurements falling below 40 ppb and 50 ppb, respectively. MAS7 and MAS8 deployed in Shanghai were assessed in Figure S7 under higher concentration ranges, where 90% of the readings for both gases exceeded these thresholds. The analysis of the lower concentration range reveals that the recommended concentration ranges are 40 ppb for $NO₂$, 10 ppb for NO , 500 ppb for CO , and 20 ppb for $O₃$. The higher concentration range analysis in Shanghai shows that increasing the concentration range beyond 40 ppb for NO₂ and 50 ppb for NO does not enhance validation \mathbb{R}^2 values. The overarching finding emphasizes the importance of ensuring an adequate concentration range during the calibration period, but beyond a certain threshold, further increases in the calibration range do not yield additional improvements in calibration results. Given that it already includes several separate analyses, we have chosen not to add further discussion in section 3.3.

Finally, regarding the discussion on time averaging, only the reference data from Hong Kong was obtained at a one-minute temporal resolution, which limited our evaluation of time averaging to data of MAS1-4. Our previous results utilized only one MAS as an example to demonstrate that applying a time averaging of 5 minutes or longer enhances sensor performance, bringing the calibration coefficients closer to optimal values. To address this, we have included results from additional sensors in Figure 6 as well as in Figures S8-S10, and we have expanded the discussion of results from different sensors in the main text, as detailed below:

Line 363-386:

As indicated in Table 1, only the reference data from Hong Kong was obtained at a one-minute temporal resolution. Thus, only the data from MAS1 - 4 will be used for time averaging evaluation. The time averaging process aims to enhance the accuracy of calibration coefficients while ensuring a substantial data volume for a reliable calibration process.

Figure 6 presents results from two different perspectives: (a)-(c) focus on the time averaging analysis and the consistency of results across different sensors, while (d)-(f) emphasize the patterns observed under varying calibration periods. Figure 6(a)-(c) show the performance of the NO₂ sensors from MAS1 to MAS4 across different time intervals, ranging from one minute to three hours. To eliminate the influence of the calibration period and adhere to the principles of single variable analysis, we utilized only 500 calibration samples from each MAS with a fixed calibration period of one day. The sensor and reference data for each calibration sample underwent time averaging across intervals of 1/3/5/7/9/11/30/60/120/180 minute(s). Subsequent calibration and validation led to the determination of the calibration slope, R^2 of the validation set, and RMSE for these time-averaged intervals. The results reveal a clear trend of improvement across all three metrics with increasing time averaging intervals, particularly notable between the 1-minute and 5-minute intervals. All four MAS NO₂ sensors exhibit a consistent trend in this regard.

These findings are based on a calibration period of 1 day, and we extended the analysis to other calibration periods. Using MAS1 as an illustrative case, Figure 6(d)-(f) display the trends across different time averaging under various calibration periods. We derived the median values under each category. Analysis of Figure 6(e)'s vertical axis reveals that, for a one-day calibration period, R^2 values improved post hourly ($R^2 = 0.68$) and 5-minute averaging ($R^2 =$ 0.66) compared to the baseline 1-minute data $(R^2 = 0.59)$, with a corresponding reduction in RMSE. For periods exceeding a day, median R^2 values exhibited a modest rise from 0.64-0.66 for 1-minute data to 0.68-0.70 for hourly data, suggesting the shorter the calibration period, the more pronounced the benefit of longer time averaging. Hence, calibrating with minute-level data over short periods of 1-3 days may lead to suboptimal validation performance. Similar trends were observed for NO and CO, as shown in Figures S8-S9; however, the trend for O₃ shown in Figure S10 was less pronounced, with only the calibration slope exhibiting a similar pattern. This may be attributed to the unique characteristics of $O₃$ calculations (Eq. 2), where the influence of cross-interference from NO₂ affects the results, thereby masking the impact of time averaging.

Figure 6. (a)-(c) The potential range of calibration slope, the *R2* **, and the RMSE of the validation set for MASs 1-4 NO₂ sensors, under various time averaging with a calibration period of 1 day. Different colored lines represent the results of different MAS units. The vertical error bar is the 25%–75% distribution of the results under different categories. (e)-(f) The calibration slope median, the** *R2* **median, and the RMSE median of the validation set for MAS1 NO₂ sensors across all calibration periods, with different colors denoting time averages ranging from one minute to three hours.**

[Comment]: 5. *The limitations and future works could be expanded into their own section instead of lumped into the conclusion, since these topics have not been discussed earlier in the paper. Acknowledgement of the practicalities of variation in sensor co-location vs deployment locations could also be expanded upon here.*

[Response]: We appreciate your suggestion regarding the limitations part. We would like to clarify that the limitations we discussed here are intended to highlight important considerations for the application of our findings in various scenarios. Given the diversity of sensor types, commercial sensor packages from different manufacturers, and even varying air sampling methods, the corresponding calibration protocols can differ significantly. We believe it is crucial to clearly state that optimal calibration conditions may vary depending on the specific features of the sensor and the calibration methods employed. The primary objective of this study is to provide methodological insights that can serve as a valuable reference for calibrating various sensor types. The developed dynamic baseline tracking method, along with the established optimal calibration period, concentration range thresholds, and time averaging period, can inform and guide future research and calibration efforts for a wide range of sensors used in air quality monitoring.

In light of this comment, we have incorporated an example into the limitations discussion and also expanded some limitation statement in the results section, as detailed below:

Line 464-467:

Optimal calibration conditions may vary depending on the sensor's specific features and the calibration methods employed. For instance, regarding the optimized calibration period, a duration of at least 5 to 7 days is necessary for conditions with a narrower concentration range. In contrast, in locations with more polluted ambient environments, a shorter calibration duration of 1 to 3 days may be sufficient for effective sensor concentration calibration.

Line 348-353:

It is important to acknowledge certain limitations in this section. The range of environmental concentrations tested was limited and may not encompass all possible calibration scenarios. Consequently, we lack sufficient data to support similar conclusions for environments with either significantly larger concentration ranges—such as those where NO , $NO₂$, and $O₃$ concentrations exceed 150 ppb—or those with consistently lower concentrations, where values remain below 10 ppb for extended periods. While our findings are applicable to most similar or closely related concentration environments, further investigation is needed to validate these conclusions across a broader spectrum of calibration conditions.