

# Practical guidelines for reproducible N<sub>2</sub>O flux chamber measurements in nutrient-poor ecosystems

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**Abstract.** The atmospheric concentration of nitrous oxide (N<sub>2</sub>O) has increased significantly since 1800, mainly due to agricultural activities. However, due to their large area, nutrient-poor natural soils, including those in the (sub-) Arctic, also play a crucial role in N<sub>2</sub>O emissions and consumption. Despite their importance, these soils have been understudied due to methodological limitations in detecting low fluxes. Our study addresses this knowledge gap by testing a fast-responding, portable gas analyser (PGA; Aeris MIRA Ultra N<sub>2</sub>O/CO<sub>2</sub>) combined with manual chambers (height and diameter: 25 cm) for measuring N<sub>2</sub>O fluxes from a nutrient-poor, sub-Arctic peatland. Our results show that this setup can detect and quantify low N<sub>2</sub>O flux rates, with a mean and standard error of  $-0.61 \pm 0.08 \mu\text{g N}_2\text{O-N m}^{-2}\text{h}^{-1}$  for a 5-min closure time, as observed in our study. More than 70% of the measured N<sub>2</sub>O fluxes exceeded the minimum detectable flux ( $0.027 \pm 0.0002 \mu\text{mol m}^{-2}\text{h}^{-1}$ ), which varied according to chamber closure time. Our study highlights the importance of using fast-responding analysers to measure low N<sub>2</sub>O fluxes and improve our understanding of diverse N<sub>2</sub>O flux dynamics. For nutrient-poor soils, we recommend a chamber closure time of approximately 5 minutes. We also found that a non-linear flux calculation model yielded better results and was broadly applicable, including cases where data were linearly distributed. Overall, our study demonstrates the potential of fast-responding analysers to improve our understanding of N<sub>2</sub>O flux dynamics in nutrient-poor soils.

## 1 Introduction

Nitrous oxide (N<sub>2</sub>O) is the third most important greenhouse gas (GHG) with a global warming potential almost 300 times stronger than carbon dioxide (CO<sub>2</sub>) over a period of 100 years (IPCC, 2023). It stays in the atmosphere for around 110 years, and its atmospheric concentration has increased from 273 ppb to 336 ppb since 1800 (Thoning et al., 2022). As most of this increase can be attributed to human activities, particularly the use of nitrogen (N) fertilisers in agriculture, research has focused on N<sub>2</sub>O emissions from managed, agricultural soils (De Klein et al., 2020) that hold the potential for high N<sub>2</sub>O emissions.

20 This is because the input of N increases the readily available mineral N needed for plant growth and thus increases harvest, but, simultaneously, can also result in higher N<sub>2</sub>O emissions (Myhre et al., 2013).

Until about 15 years ago, few studies investigated N<sub>2</sub>O fluxes in the (sub-) Arctic, where soils often have a very low availability of reactive N (Virkkala et al., 2024) and thus are not expected to emit amounts of N<sub>2</sub>O relevant for the global climate (Voigt et al., 2020; Christensen et al., 1999; Grogan et al., 2004; Martikainen et al., 1993). In these low N ecosystems, N<sub>2</sub>O uptake could be expected, but has, so far, not been confirmed in field studies (Buchen et al., 2019; Schlesinger, 2013). Since 2009, multiple studies have reported N<sub>2</sub>O emissions similar to agricultural soils from organic-rich ecosystems in the Arctic (Repo et al., 2009; Marushchak et al., 2011; Elberling et al., 2010), shifting the focus to only selected, high-nutrient areas within the (sub) Arctic. Nevertheless, reporting near-zero N<sub>2</sub>O fluxes is crucial to avoid overestimating emissions caused by biased site selection favouring high-emitting areas (Voigt et al., 2020).

30 In most studies, N<sub>2</sub>O concentrations were sampled repeatedly with a syringe from the head space of a closed flux chamber and measured with a gas chromatograph (GC) in the laboratory (Hensen et al., 2013; Denmead, 2008; Pavelka et al., 2018). With this approach, typically between 4 and 6 discrete air samples are taken to measure the trend in N<sub>2</sub>O mixing ratios during chamber closure time and calculate the fluxes. The sensitivity of GCs varies, but with only few samples drawn from a fluctuating time series that may not necessarily display a linear trend, differences in low concentrations are hard to capture and highly dependent on single data points (Hübschmann, 2015). Additionally, in previous studies, opaque chambers have been mostly used because temperature inside the chamber would increase less above the ambient temperature compared to transparent chambers (Clough et al., 2020). As a result, there are only few studies investigating N<sub>2</sub>O fluxes under different light conditions (Stewart et al., 2012). Since this was the only available method for *in situ* N<sub>2</sub>O measurements in the field, our knowledge on (sub-) Arctic N<sub>2</sub>O fluxes is rather limited and makes it challenging to establish accurate baseline estimates, which are essential for detecting changes in fluxes.

Recent advances in laser spectroscopy led to novel, portable (< 15 kg) and fast-responding (1 Hz, *i.e.*, sampling every second) GHG analysers, offering new possibilities to measure low N<sub>2</sub>O concentrations in nutrient-poor ecosystems (Subke et al., 2021). These portable gas analysers (PGAs) allow near-continuous monitoring of concentration changes, providing higher precision and lower detection limits than GC-based methods (Hensen et al., 2013). While differences in flux estimates between PGAs and GCs have been well-documented for CH<sub>4</sub> and CO<sub>2</sub>, few studies have focused on N<sub>2</sub>O (Christiansen et al., 2015; Brümmer et al., 2017). The detection limit was a significant constraint, as many reported N<sub>2</sub>O fluxes were below the threshold of the GC method, limiting the ability to accurately assess their magnitude and trends. The availability of PGAs for *in situ* N<sub>2</sub>O flux measurements raises new methodological questions. Unlike CH<sub>4</sub> and CO<sub>2</sub>, where, under a fixed chamber height, approximately 3 minutes are well-accepted for reliable measurement, the minimal chamber closure time for N<sub>2</sub>O fluxes in nutrient-poor ecosystems is unclear. This is because N<sub>2</sub>O concentrations are very low and take longer to accumulate in the head space to accurately detect trends (Fiedler et al., 2022). Few studies have investigated the chamber closure time with portable N<sub>2</sub>O analysers, and the reported recommendations range between 3 and 10 min and originate from ecosystems with higher N<sub>2</sub>O flux rates (Fiedler et al., 2022; Brümmer et al., 2017).

The chamber community has been discussing the use of linear (LM) or non-linear (HM) models to calculate flux rates for decades (Pumpanen et al., 2004). The critique on the linear models is that they underestimate the flux rates due to the assumption that GHG concentrations keep increasing within the chamber head space (Fiedler et al., 2022; Hüppi et al., 2018). However, it is clear from the theory of molecular diffusion that the rate of concentration change within the chamber declines over time (Hutchinson and Mosier, 1981; Kutzbach et al., 2007; Kroon et al., 2008). As a result, there have been great efforts to implement non-linear flux calculations as alternatives to LM, for example, through software packages (Pedersen et al., 2010; Hüppi et al., 2018). However, non-linear flux calculations are not commonly used in the chamber community, likely because they are more complex to implement. For N<sub>2</sub>O, there is a lack of data sets from nutrient-poor ecosystems to evaluate the effect of LM and HM models.

The main aim of this paper is to present a mobile flux chamber method capable of quantifying (very) low N<sub>2</sub>O fluxes in nutrient-poor ecosystems. Using a novel portable N<sub>2</sub>O analyser (Aeris MIRA Ultra N<sub>2</sub>O/CO<sub>2</sub>) and our custom-made transparent and opaque flux chambers, we provide the first extensive data set of low N<sub>2</sub>O fluxes from the (sub-) Arctic. We tested the performance of our PGA in the laboratory and in the field across various land cover types from a thawing permafrost peatland in sub-Arctic Sweden. We compared N<sub>2</sub>O flux rates calculated from different chamber closure times (3 min- 10 min) and evaluated differences between linear and non-linear calculation methods. Additionally, we compared flux rates based on high-frequency *in situ* observations against an approach that randomly draws discrete data points from the full time series, mimicking a GC-based approach. Finally, we aim to provide guidance on measuring N<sub>2</sub>O fluxes in nutrient-poor ecosystems, such as the Arctic. Ideally, this will encourage researchers to measure low fluxes in (sub-) Arctic regions, get a better process understanding of N<sub>2</sub>O fluxes, and determine how the N cycle in nutrient-poor ecosystems will respond to Arctic warming.

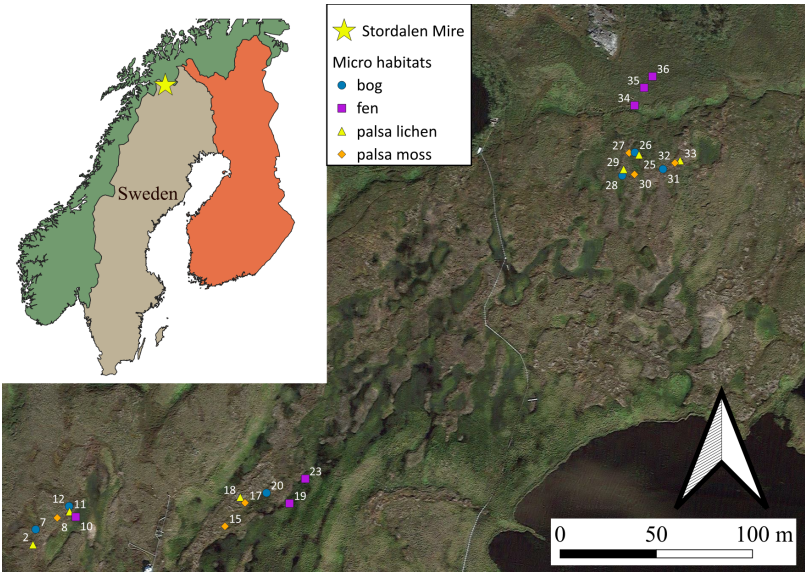
## 2 Methods

To facilitate the reader's understanding, we use the terminology proposed by Fiedler et al. (2022), with location describing the area where sampling occurs ("Stordalen mire"), site describing a vegetation unit within the location ("palsa lichen", "palsa moss", "bog", "fen"), and chamber base position (*i.e.*, plot) for the exact spot where N<sub>2</sub>O was measured. With "chamber closure time", we specify the time frame a chamber was closed onto the soil; one of these periods is then called "measurement period".

### 2.1 Study location and sampling sites

All data were collected at the Stordalen mire, a complex palsa mire underlain by sporadic permafrost located in subarctic Sweden (68° 20.0' N, 19° 30.0' E), 10 km east of Abisko (Ábeskovvu in Northern Sámi language). Permafrost has been rapidly thawing at this location over the last decades, and only remains in the dry uplifted areas on the peatland (palsas) (Sjögersten et al., 2023). For our study, we randomly selected 24 chamber base positions in three transects on a dry-to-wet thawing gradient from palsa to bog to fen, with 6 replicates for each land cover type: palsa lichen, palsa moss, bog, and fen (Fig. 1). Transects 1 and 2 each contain 6 chamber base positions and are located in the northern centre of the mire, within the

85 footprint of an Integrated Carbon Observation System (ICOS, SE-Sto) eddy covariance tower which has been operating since 2014 (Lundin et al., 2024), Fig. 1). Transect 3 lies in the most north-eastern part of the palsa.



**Figure 1.** Three transects with chamber base positions in Stordalen overlaid on satellite image from ©Google Maps. The location of the Stordalen mire is marked with a star. Here, micro habitats are represented with different colours and symbols for clarity. The spatial data of each country can be found at <https://simplemaps.com>, last access: 17/09/2024)

Vegetation on the palsa is mainly dominated by lichen (*Cladonia spp.*), shrubs (*Empetrum hermaphroditum*, *Betula nana*, *Vaccinium uliginosum*, *Vaccinium vitis-idaea*, *Rubus chamaemorus*) and some mosses (*Dicranum elongatum*, *Sphagnum fus-*  
 90 *cum*). Both bogs and fens contain peat-forming mosses (*Sphagnum balticum*, *Sphagnum lindbergii*, *Sphagnum riparium*), with the dominant vascular plants on fens being cotton grass (*Eriophorum vaginatum*, *Eriophorum angustifolium*) and in bogs  
 sedges (*Carex rotundata*, *Carex rostrata*). The soils in the area are classified as organic histosols or, if permafrost occurs within  
 2 m of cryoturbation activities, as cryosols (Siewert, 2018). Research at the Stordalen mire has been conducted for over a  
 century (Jonasson et al., 2012; Callaghan et al., 2013), and a vast amount of data on CH<sub>4</sub> and CO<sub>2</sub> fluxes has been published  
 (Łakomiec et al., 2021; Varner et al., 2022). The mean annual temperature at the Stordalen mire is -0.6°C and the annual  
 95 precipitation 304 mm (Malmer et al., 2005).

The data presented in this study were collected during three separate campaigns covering different seasons: spring (between  
 23 - 30 May 2023), summer (20 - 27 July 2023), and autumn (3 - 22 September 2023). PVC collars with an inner diameter of  
 245.1 mm, a height of 150 mm, and a wall width of 4.9 mm were inserted into the soil on 29 August 2022. We inserted them  
 as deeply as possible, between 100- 140 mm, to ensure a proper seal between the chamber head space and the atmosphere  
 100 even during strong wind conditions, and in the palsa where the top peat was dry and highly porous. Between the collar and  
 the chamber, we placed a custom-made sealing ring to avoid ambient air entering the chamber during our measurements (Fig.

2, S2). The sealing ring has an inner and outer diameter 235 and 265 mm, respectively, a height of 30 mm and is build from a metal ring wrapped in foam material (50 mm on each side). Tests confirmed that it sealed the chamber and the collar even under high wind conditions with up to  $18\text{ m s}^{-1}$  wind gusts.

## 105 2.2 Chamber and portable gas analyser (PGA) setup and protocol

For our measurements, we used a custom-built static, non-steady state, non-flow-through chamber (Livingston and Hutchinson, 1995) made from acrylic glass (Göli GmbH) with a height of 250 mm and a diameter of 250 mm (Fig. 2, S1). We placed a fan (SUNON Maglev, 80 mm x 80 mm x 25 mm, 2000 RPM) inside the chamber to ensure well mixed conditions within the chamber during the measurements. Additionally, we installed a relative humidity (RH) and temperature probe (EE08, E+E  
110 Elektronik, Germany) and a pressure sensor (61402V, RM Young) for measuring essential parameters to calculate the fluxes. We equipped the chamber with quick-release connectors on top to connect the inlet and outlet tubing to the portable gas analysers. As complementary variables, we measured soil temperature at 15 cm (PT100 4-wire sensors, JUMO GmbH & Co. KG) at each quadrant outside of the plot, soil moisture at 12 cm and 30 cm (CS655-DS and CS650-DS, Campbell Scientific), and photosynthetically active radiation (PAR) (PQS1, Kipp and Zonen). More detailed information on our chamber setup can  
115 be found in the supplementary information.



**Figure 2.** Chamber setup during measurement period, with soil moisture and soil temperature sensors installed in the soil, and all inlets connected. Photo: Fabio Cian, “Ubiquitous Anomaly”, CC BY-NC-ND 4.0

To measure  $\text{N}_2\text{O}$  concentrations, we used the Aeris MIRA Ultra  $\text{N}_2\text{O}/\text{CO}_2$  (from now onward: Aeris- $\text{N}_2\text{O}$ ) analyser (Aeris Technologies; sensitivity: 0.2 ppb/s for  $\text{CO}_2$  and  $\text{N}_2\text{O}$ , frequency: 1 Hz). As most PGAs, the Aeris- $\text{N}_2\text{O}$  provides dry mole fractions of the target gas. We performed several laboratory tests to assess the signal stability (*i.e.*, drifts and stabilisation time), uncertainties, noise level, and water interference of the Aeris- $\text{N}_2\text{O}$ . Furthermore, we tested the impact of the humidity on the  
120 Aeris- $\text{N}_2\text{O}$  analyser using a portable dew point generator (LI-610, Licor USA). By adjusting the dew point temperature, we examined four different humidity levels: 28, 45, 60, and 83 %. A calibration gas tank with a known  $\text{N}_2\text{O}$  concentration of

333.2 ppb was first connected to the dew point generator. The humidified gas was then connected to the Aeris-N<sub>2</sub>O analyser inlet and each humidity level was sampled for about 20 minutes. Nevertheless, only a 10-minute window was used for further analysis due to relatively long time (about 10 min) required for the humidity levels to stabilise (see Fig. A2).

125 In the field, we attached a custom made external battery box with two LiFePO<sub>4</sub> batteries (LiFePO<sub>4</sub> 12.8V 20Ah, Green Cell) to the Aeris-N<sub>2</sub>O, which could be switched whilst the analyser was running. In this study, one LiFePO<sub>4</sub> 12.8V 20Ah battery lasted for the whole day of field measurements (8h - max. 12h). A data logger (CR1000X, Campbell Scientific) was used to log all the sensor data including greenhouse gas concentrations which was placed inside a Pelican-case (Fig. S3 and S4). All GHGs and explanatory variables were logged with a frequency of 1 Hz. With this setup, all necessary information for the analyses  
130 was synchronised in a single data file, rather than many individual files from individual sensors and loggers.

Before we started a measurement period (*i. e.*, time when chamber is closed), we attached the tubes from the PGA to the chamber, ventilated the chamber for at least one minute, and gently closed it onto the sealing ring. The default chamber closure time for all measurement periods was 10 minutes. For dark measurements, a custom-made, light-impermeable tarp was placed over the chamber to prevent light from entering and minimise temperature fluctuations. We refer to these measurements as  
135 'opaque' for clarity.

All data were processed in R (version 4.3.3; R Core Team, 2024) and version controlled in GitLab (the scripts are publicly available from <https://git.bgc-jena.mpg.de/ntriches/data-analysis/-/tags/2024-12-12-triches-amtsubmission-n2oadvances>). A filter script was applied to the data to ensure it met certain quality control standards, including:

- 140 – Removing data from the beginning of each measurement period to account for the time lag of gases moving through the tubes to reach the laser cell
- Removing implausible values (*e. g.*, -9999) of chamber pressure, chamber temperature, chamber relative humidity, soil temperature, volumetric water content, and PAR
- Replacing negative PAR values with 0
- averaging soil temperature readings gained from four sensors
- 145 – Detecting and removing flat lines indicating instrument failure (see SI 1.3)
- Pre-processing data for concentrations of N<sub>2</sub>O, CO<sub>2</sub>, and H<sub>2</sub>O, as well as other relevant parameters.

## 2.3 Flux calculations

In our study, we calculated N<sub>2</sub>O fluxes using **all data points** from one measurement period. We removed 8 seconds in the start of the measurement period to account for the time delay until the concentration from the chamber reached the cell of the gas  
150 analyser. An extra 7 s were removed for opaque measurements, since we needed more time in the field to cover the chamber with the reflective tarp. To calculate the fluxes with both linear (LM) and non-linear (HM) methods in a reproducible way, we used the R package goFlux (v0.2.0, (Rheault et al., 2024)). We selected goFlux for several reasons: (i) it was specifically written

to process data gathered with PGAs, (ii) it uses measured temperature and pressure inside the chamber for flux calculation, (iii) it corrects for the dissolved gases in the water vapour inside the chamber, and (iv) it calculates fluxes using both LM and HM  
155 flux calculation methods. It also enables the comparison of results from LM and HM models using various statistical methods and flags, such as the minimal detectable flux (MDF, Eq. 4). Additionally, it generates plots for visualisation. For the flux calculation in LM, goFlux applies the commonly used linear equation as follows in Eq. 1:

$$F(t) = \frac{dC(t)}{dt} \frac{V}{A} \quad (1)$$

where  $F(t)$  is the gas flux rate at a given location during the chamber closure time ( $t$ ),  $\frac{dC(t)}{dt}$  is the mass concentration change  
160 with time,  $V$  is the volume of the chamber, and  $A$  the area of the soil covered by the collar (Subke et al., 2021). To report our flux rates, we used the atmospheric sign convention, *i. e.*, negative signs for an uptake of  $N_2O$  into the soil, and positive signs for emissions.

The HM model approach is based on the Hutchinson and Mosier (1981) approach as given in Eq. 2:

$$C(t) = \varphi + (C_0 - \varphi)e^{-\kappa t} \quad (2)$$

165 here,  $\varphi$  is the assumed constant gas concentration of the source within the soil (Pedersen et al., 2010),  $C_0$  is the gas concentration in the chamber at the moment of chamber closure, and  $\kappa$  is the model parameter. To limit  $\kappa$  with a maximum threshold  $\kappa_{\max}$ , Eq. 3 was adapted from Hüppi et al. (2018).

$$\kappa_{\max} = \frac{F(t)}{\text{MDF } t} \quad (3)$$

Here, the MDF is a theoretical threshold that represents the instrument's detection limit, based on its precision ( $\eta$ ) provided  
170 by the manufacturer. However, it does not account for potential errors in the model or chamber artefacts, but reflects the instrument's inherent uncertainty. The MDF can be calculated using Eq. 4.

$$\text{MDF} = \frac{\eta}{t} \theta \quad (4)$$

where,  $\theta$  is a flux term that corrects for the water vapour inside the chamber and converts the flux unit to  $\mu\text{mol m}^{-2} \text{s}^{-1}$  and  $t$   
175 is the measurement time, *i.e.*, the number of measurement points during the measurement period. This was calculated as given in Eq. 5

$$\theta = \frac{(1 - C_{H_2O})V P}{A R T} \quad (5)$$

where  $C_{H_2O}$  is the water vapour concentration in  $\text{mol mol}^{-1}$ ,  $P$  is the pressure inside the chamber in kPa,  $R$  is the universal gas constant in  $\text{L kPa K}^{-1} \text{mol}^{-1}$ , and  $T$  is the air temperature inside the chamber in K.

In the goFlux package, the fluxes that are below the detection limit are flagged. Note that owing to this function, all our flux estimates have their own MDF value. The package further implements the so called g-factor ( $g_f$ ) (Hüppi et al., 2018) to restrict large curvatures of the non-linear flux models as follows (Eq. 6):

$$g_f = \frac{HM_F}{LM_F} \quad (6)$$

Here,  $HM_F$  and  $LM_F$  are the flux values that are calculated by HM and LM models, respectively. In this study, we used a  $g_f$  of 4, meaning that the flux calculated by the HM model can be max. 4 times higher than the flux calculated by the LM to avoid a large overestimation of fluxes (Eq. 6). We used this factor because, upon visual assessment, it fit our data best, and has been previously used (Leiber-Sauheitl et al., 2014). We did not use the mean absolute error nor the root mean square error to define if the HM or LM model performed better, since they were very similar among all measurement periods. We did also not use  $R^2$  as a filter criteria since low and non-linear fluxes inherently results in low  $R^2$  values (Kutzbach et al., 2007).

## 2.4 Data processing and simulations

We used our openly available script to simulate differences between chamber closure times and GC sampling, and the associated sensitivity analysis. We first calculated all fluxes using the original 10-minute chamber closure time ( $\text{prec} = 0.2$ ,  $\text{g.limit} = 4$ ). To see how different closure times affect  $N_2O$  fluxes, we shortened the closure time by 1 minute at a time, starting from 9 minutes, and recalculated the fluxes for each new time (e.g., 9 minutes = 540 seconds, 8 minutes = 480 seconds, etc.). We compared how chamber closure time affects flux rates during transparent and opaque measurements, and identified the number of fluxes above the minimum detectable flux based on the goFlux output. While calculating our fluxes, we became aware of one chamber base position acting as a hot spot, i. e., showing much higher flux rates than the other chamber base positions. Since we wanted to focus our analyses on low fluxes, we removed this hot spot from all analyses.

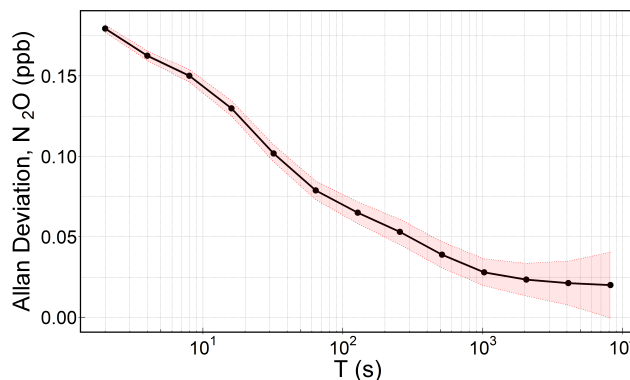
The simulations of GC measurements are based on drawing discrete sub-samples from the continuous *in situ* time series from the PGA, mimicking a sampling scheme where information on the increase of gas concentrations during chamber closure time is limited to a few snapshots in time. We simulated four scenarios: 3 GC samples (taken at 5 min, 7 min, and 10 min), 4 GC samples (3 min, 5 min, 7 min, and 10 min), 5 GC samples (1 min, 3 min, 5 min, 7 min, and 10 min), and 6 GC samples (1 min, 3 min, 5 min, 7 min, 8 min, and 10 min). For simulating the GC concentration, we picked the time stamp as defined above and took the average of 10 sec of  $N_2O$  concentrations measured by our PGA, i. e., 5 seconds before and after the defined time stamp, as single GC point measurement (Fig. 9). We then calculated the resulting fluxes with goFlux ( $\text{prec} = 1.9$ ,  $\text{g.limit} = 4$ ) using a precision of 1.9 ppb according to sensitivity tests on an instrument at our laboratory (Agilent Technologies, 7890 B GC System, mean  $N_2O$  concentration 395.746 ppb with a SD of 1.875 ppb over ten repetitions), before we plotted the simulated versus original flux concentration measurements. To evaluate the performance of each sampling scheme, we compared the slopes, p-values, and  $R^2$  values between the simulated and original data. To get an estimate on the uncertainties associated

with this GC simulation, we conducted a sensitivity analysis, where we did 21 simulations with a randomised selection of the  
 210 4 min sampling time (4 samples at 3,5,7,9 min), allowing a window of 60 sec around each selected GC point.

### 3 Results and discussion

#### 3.1 Laboratory tests with the Aeris-N<sub>2</sub>O

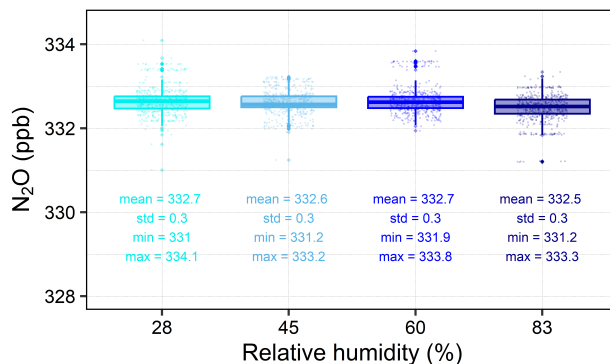
From the 15-hour ambient air sampling in our closed laboratory, we observed that the water vapour concentration in the ambient  
 air dropped from approximately 2500 ppm to about 800 ppm within the first 30 min. It continued to decrease progressively  
 215 throughout the sampling period; however, after 5 hours, the water vapour concentration somewhat stabilised, with a mean H<sub>2</sub>O  
 concentration of 476.9 ppm with a standard deviation of 18.7 ppm for the rest of the sampling. Note that these observed changes  
 in water vapour might be partly due to the analysers warm-up period. We therefore discarded the initial 5 hours of data and  
 used the remaining data to assess the signal stability and noise characteristics of the Aeris-N<sub>2</sub>O. The Aeris-N<sub>2</sub>O demonstrated  
 a very stable signal with no apparent drift for about 10 hours of sampling period, having a low standard deviation of 0.290  
 220 ppb. Using Allan deviation plots (Allan, 1987), we evaluated the analyser's noise characteristics and found that the Aeris-N<sub>2</sub>O  
 showed low instrument noise, approximately 0.18 ppb at 2-second averaging (see Fig. 3).



**Figure 3.** Computed Allan deviation plot based on 10 hours of continuous sampling, following a 5-hr spin up period during which the water vapour mole fraction was not stable. Here, T is the sampling time in log-scale and shaded region represents the 95% confidence interval.

Because PGAs are known to be sensitive to fluctuations in water vapour concentrations, we tested the Aeris-N<sub>2</sub>O against  
 different relative humidity (RH) conditions. Our tests with the four RH values (approx. 28%, 45%, 60% and 83%, respectively)  
 showed that the water interference of the Aeris-N<sub>2</sub>O was very small; we observed only slight differences in the mean N<sub>2</sub>O  
 225 concentrations for each humidity level (see Fig. 4), with mean N<sub>2</sub>O concentrations of 332.7 ppb, 332.6 ppb, 332.7 ppb, and  
 332.5 ppb for RH values of 28%, 45%, 60%, and 83%, respectively. Furthermore, we observed the same standard deviation  
 of about 0.3 ppb for each humidity level. Overall, our conducted laboratory tests indicated that the Aeris-N<sub>2</sub>O was a suitable  
 instrument for measuring low N<sub>2</sub>O fluxes, showing low noise and water interference, along with negligible signal drift after

the laser warms up. Nevertheless, the long warm-up period (approximately 5 h) of the analyser needs to be taken into account, as this can be a limiting factor for certain applications. To mitigate this, the Aeris-N<sub>2</sub>O remained powered on throughout the whole field campaign.



**Figure 4.** Measured N<sub>2</sub>O concentrations for different relative humidity (RH) values, with basic statistics of each RH values summarised under each box plots. Each humidity level was sampled for approximately 20-mins; however, only a 10-min window was used for further calculations. For our tests, we used a standard gas with a mean of  $333.16 \pm 0.16$  ppb as input. Jittered points are overlaid on the boxplots to visually separate overlapping data points, illustrating the distribution and density of the data.

### 3.2 Impact of chamber closure time on N<sub>2</sub>O fluxes

At our site, we commonly observed net N<sub>2</sub>O consumption, suggesting an atmospheric sink, with a mean flux of  $-0.469 \mu\text{g N}_2\text{O-N m}^{-2}\text{h}^{-1}$  and a 95% confidence interval (CI) of  $(-0.60, -0.3)$  during a chamber closure time of 10 min. Our calculated mean flux during transparent measurements was  $0.361 \mu\text{g N}_2\text{O-N m}^{-2}\text{h}^{-1}$ , with a 95% CI of  $(0.24, 0.48)$  during a chamber closure time of 10 min (Table 1). For opaque measurements, our calculated flux was  $-1.29 \mu\text{g N}_2\text{O-N m}^{-2}\text{h}^{-1}$ , with a 95% CI of  $(-1.45, -1.13)$ , indicating that our opaque measurements represent a real biochemical process, rather than an experimental artefact, in the (sub-) Arctic ecosystem. Nevertheless, the impact of environmental drivers on N<sub>2</sub>O fluxes, including the transparent and opaque measurements, is beyond the scope of this study. Overall, we collected 338 samples, with 60-90 % of N<sub>2</sub>O fluxes above the detectable limit. We therefore also acknowledge the possibility of unknown chamber artefacts that may remain undiscovered and could affect the interpretation of our data.

While chamber measurements are essential for understanding GHG emissions, they can alter soil-air conditions and introduce observational artefacts. These include potential impacts such as pushing atmospheric air into the soil when closing the chamber, flushing soil gas into the chamber head space, and changing conditions during closure, *e.g.*, increasing temperature and humidity inside the chamber due to soil and plant evaporation (Subke et al., 2021; Rochette and Eriksen-Hamel, 2008). As a result, the N<sub>2</sub>O concentration gradient between the soil and the chamber head space and potentially also the functioning of plants and soil microbes are altered and may cause a bias in the flux estimates (Davidson et al., 2002). At our site, condensation within the chamber during a measurement period increased drastically with time, especially in the colder months. Although

**Table 1.** Comparison of chamber closure times for both transparent and opaque measurements, with SE = Standard Error, CI = confidence interval, and % = percentage difference between the absolute mean flux of chamber closure time x compared to chamber closure time 10 min. CI is the margin of error calculated at 1.96 \* SE; the lower CI is calculated as mean flux - CI, the upper CI as mean flux + CI.

Chamber closure time	Transparent / opaque	Mean flux	SD flux	n	SE	CI	lower CI	upper CI	%
3	light	-0.00	1.54	169	0.12	0.23	-0.24	0.23	98.97
3	dark	-1.49	1.80	170	0.14	0.27	-1.76	-1.22	-15.16
4	light	0.20	1.26	169	0.10	0.19	0.01	0.39	43.58
4	dark	-1.51	1.53	170	0.12	0.23	-1.74	-1.28	-16.92
5	light	0.26	1.06	169	0.08	0.16	0.10	0.42	27.52
5	dark	-1.47	1.39	170	0.11	0.21	-1.68	-1.26	-14.18
6	light	0.30	0.94	169	0.07	0.14	0.16	0.45	16.21
6	dark	-1.43	1.29	170	0.10	0.19	-1.62	-1.24	-10.75
7	light	0.33	0.89	168	0.07	0.14	0.19	0.46	9.41
7	dark	-1.39	1.24	170	0.10	0.19	-1.58	-1.21	-8.05
8	light	0.35	0.83	168	0.06	0.13	0.23	0.48	2.35
8	dark	-1.35	1.21	170	0.09	0.18	-1.54	-1.17	-4.92
9	light	0.36	0.81	168	0.06	0.12	0.24	0.48	1.00
9	dark	-1.34	1.16	170	0.09	0.17	-1.51	-1.17	-3.79
10	light	0.36	0.80	168	0.06	0.12	0.24	0.48	0.00
10	dark	-1.29	1.07	170	0.08	0.16	-1.45	-1.13	0.00

our laboratory tests showed that for the Aeris-N<sub>2</sub>O, increased water vapour does not interfere with N<sub>2</sub>O concentrations, all  
250 laser cells are sensitive to water vapour. Too high water vapour contents can, even with a filter assembly (1 micrometers pore size) within the tube, reach the analyser cell and lead to an abrupt end of a field campaign (Fiedler et al., 2022). It is, therefore, crucial to know how water vapour concentrations vary over time during chamber closure. At our study site, H<sub>2</sub>O concentrations during transparent measurements increased, on average, from below 10000 ppm up to >16000 ppm, depending on the season. When we look at the rate of change over each minute, *i.e.*, 0-1 min, 1-2 min, 2-3 min etc., we can see that this rate of change  
255 drastically decreased within the first 2 min, and then exponentially decreased with increasing chamber closure time (Fig. S5). In other words, H<sub>2</sub>O concentrations rose drastically in the first 2 min (> 1300 ppm; data not shown), after which they levelled out until around 8 min, before they started increasing again (Fig. S5). For opaque measurements, the impact followed the same pattern, but was of much smaller magnitude (approx. 600 ppm rise within the first 2 min; data not shown). Because the increase of H<sub>2</sub>O concentrations did not directly affect N<sub>2</sub>O concentrations in our study, and was further considered when calculating  
260 the fluxes with goFlux (Rheault et al., 2024), we did not introduce further measures.

When using transparent chambers, temperature control is an additional constraint (Rochette and Hutchinson, 2015). Ideally, the chamber temperature remains stable throughout the measurement period. However, in the field, especially during sunny conditions, this is challenging to achieve without active cooling, as the chamber’s transparency creates a greenhouse effect

(Rochette and Hutchinson, 2015). An increase in temperature can enhance microbial activity, leading to either N<sub>2</sub>O produc-  
265 tion or consumption and potentially causing biased N<sub>2</sub>O concentrations (Rochette and Eriksen-Hamel, 2008; Clough et al.,  
2020). In our study, chamber temperatures during transparent measurements changed similarly to H<sub>2</sub>O concentrations, with  
the strongest decrease occurring within the first minutes of the measurement period (Fig. S5 and S6). At our site, temperature  
increased by around 0.7°C within the first two minutes, which slowed down to 0.3°C after 5 min (data not shown). During  
opaque measurements, temperature within the chamber decreased slightly by max. 0.2°C in the first two minutes, and below  
270 0.1°C after three minutes (data not shown). It is likely that during transparent measurements, the abrupt temperature increase in  
the first two minutes may have impacted N<sub>2</sub>O concentrations (Parkin and Venterea, 2010). However, we refrained from using  
cooling systems such as heat exchangers or ice packs since they also have drawbacks, *e.g.*, causing additional condensation  
within the chamber (Clough et al., 2020; Fiedler et al., 2022). As a result, we cannot exclude potential temperature effects on  
our N<sub>2</sub>O concentrations during our measurement periods. Even though temperature changes are considered in the final flux  
275 calculation (Rheault et al., 2024), we argue that temperature changes inside chambers call for shorter closure times.

To minimise disturbances to the soil gas-atmosphere gradient and obtain flux estimates close to pre-deployment levels, several  
researchers have recommended using short chamber closure times of around 5 minutes (Fiedler et al., 2022; Pavelka et al.,  
2018; Venterea and Baker, 2008). This is because during chamber closure, mean flux rates vary as N<sub>2</sub>O builds up or decreases  
in the chamber head space (Rochette and Hutchinson, 2015). As closure time increases, the concentration gradient between  
280 soil and chamber head space decreases, reducing the diffusive flux. Within a closed system, gases can reach a temporary state  
of equilibrium over time, where the rate of N<sub>2</sub>O production in the soil balances with the rate of N<sub>2</sub>O release into the chamber  
head space (Fiedler et al., 2022). Our results suggest that for transparent measurements, a chamber closure time of 3 min is too  
short and may result in significantly lower flux rates than longer chamber closure times (ANOVA,  $p < 0.05$  for 3 min compared  
to 8, 9, and 10 min; Fig. 5). This discrepancy may be attributed to low microbial activity, or the possibility that N<sub>2</sub>O production  
285 is countered by its rapid uptake or dissolution in the water present in the soil matrix, a phenomenon previously observed for  
CO<sub>2</sub> (Widén and Lindroth, 2003). From 4 min onward, our calculated mean N<sub>2</sub>O flux rates are not significantly different from  
one another (Table 1). The proportion of transparent fluxes above the minimum detectable flux (MDF) increased from 62.7%  
to 78.6 % as closure time increased from 3 to 10 minutes (Fig. 5 a) ). While longer closure times reduce uncertainty and the  
amount of fluxes below the MDF, the gain is small after 4 minutes. We thus recommend chamber closure times of more than  
290 4 minutes for reliable N<sub>2</sub>O flux estimates, as this balances the need for accurate transparent measurements while minimising  
soil disturbance, as well as the impact of increasing temperature on N<sub>2</sub>O concentrations within the chamber.

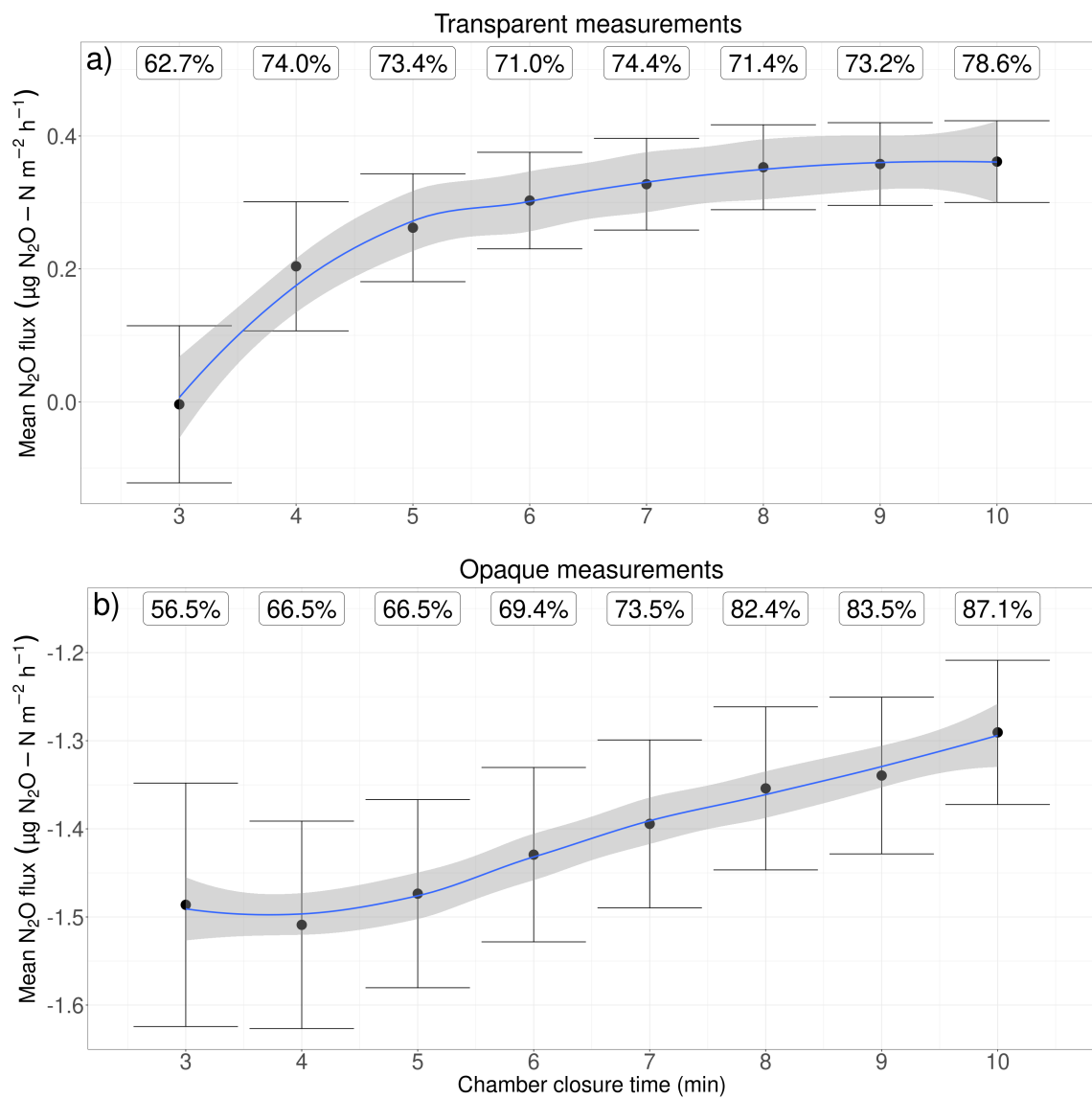
For opaque measurements, we find that our calculated fluxes show higher N<sub>2</sub>O uptake from shorter chamber closure times,  
with flux rates around 15% lower at 3 - 5 min than at 10 min, respectively (Table 1). At 6 min, the differences in our calculated  
N<sub>2</sub>O uptake was still 10% higher than at 10 min, decreasing to below 8% between 7 and 9 min. At the same time, the  
295 MDF increased from 56.5% to 87.1% between 3 min and 10 min (Fig. 5 b) ). Nevertheless, none of the flux rates across  
different closure times were significantly different from one another (Kruskal-Wallis,  $p = 0.99$ ). Especially for N<sub>2</sub>O uptake,  
it is essential to keep the chamber closure time as short as possible. This is because N<sub>2</sub>O availability through soil diffusion  
is often the limiting factor for microbial consumption, *i.e.*, atmospheric N<sub>2</sub>O consumption by N<sub>2</sub>O-reducing microbes (Liu

et al., 2022). When the chamber is closed, the N<sub>2</sub>O concentration in the head space decreases as it diffuses into the soil,  
300 driven by the concentration gradient. As a result, the uptake rate also decreases, since N<sub>2</sub>O reduction may become substrate  
limited. Consequently, long chamber closure times may underestimate the uptake of atmospheric N<sub>2</sub>O. Our analysis of the  
chamber closure time confirms this: during opaque measurement, we found that the uptake rate at 3-5 minutes were greatest,  
and decreased with every minute of added chamber closure time (Fig. 5). For opaque measurements, we therefore suggest to  
keep the chamber closure time between 3-5 min, unless very few data points are available, when aiming for fluxes above the  
305 MDF becomes more important.

With the goFlux output, we obtain an individual MDF for each measurement period, allowing us to determine on a case-  
by-case basis if a flux is above the MDF. For both transparent and opaque measurements, the MDF in our study was, on  
average,  $0.03 \mu\text{mol m}^{-2}\text{h}^{-1}$ . This is lower than the reported  $0.18 \mu\text{mol m}^{-2}\text{h}^{-1}$  MDF rates in other nutrient-poor ecosystems  
by Christiansen et al. (2015) (Table 1). However, as mentioned above, more than 40% of the fluxes were below the MDF at  
310 3-minute closure time. This confirms that very short closure times can lead to higher uncertainties of flux estimates because  
the concentration changes are too small to be accurately detected (Fiedler et al., 2022). It is, therefore, essential to consider  
the precision of the instruments used in the field to identify the best-suited chamber closure time. With the Aeris-N<sub>2</sub>O, we  
recommend chamber closure times above 4 min for transparent N<sub>2</sub>O measurements in low nutrient ecosystems. The optimal  
closure time depends on factors such as effective chamber height, micro habitat, and the duration of the field campaign.  
315 Shorter closure times allow for more repetitions, increasing the number of observation per chamber base position or adding  
more replicates, which is essential for the accurate representation of spatial variability (Jungkunst et al., 2018). For opaque  
measurements, we suggest shorter chamber closure times of 3-5 min. These findings are in line with other studies (Cowan  
et al., 2014; Kroon et al., 2008; Christiansen et al., 2015) but confirm, for the first time, that these recommendations are  
applicable to (very) nutrient-poor ecosystems.

### 320 3.3 Impact of linear and non-linear flux calculation approaches on N<sub>2</sub>O fluxes

To facilitate understanding of how N<sub>2</sub>O concentration build-up or reduction in the chamber head space can result in different  
flux estimates, goFlux automatically produces scatter plots with defined criteria (Fig. 6). These outputs allow for visual control  
of all measurement periods; additionally, csv outputs with the pre-defined quality checks are automatically produced. Our anal-  
ysis using the goFlux package (Rheault et al., 2024) revealed that 59% (n = 2560) of all N<sub>2</sub>O fluxes during different chamber  
325 closure times were best described by the HM model, indicating non-linear concentration changes over time. In contrast, 41%  
(n = 1728) of the fluxes were better explained by the LM model, showing a linear concentration change over time. However, all  
of the 41% fluxes calculated with the LM model had no HM flux, meaning that the software did not calculate a non-linear flux  
because the HM model gave the same results as the LM model, and therefore favoured the LM model. In other words, all fluxes  
were either calculated using the HM model or resulted in the same values as the LM model. This has two possible reasons:  
330 the first is that the linearity might be an outcome of short measurement time and low flux, so that the non-linear model was  
reduced to a linear model; the second is an overestimation of the flux (Kutzbach et al., 2007). Linear and non-linear models  
for concentration data during chamber closure may typically be seen as alternatives, not complementary approaches. However,



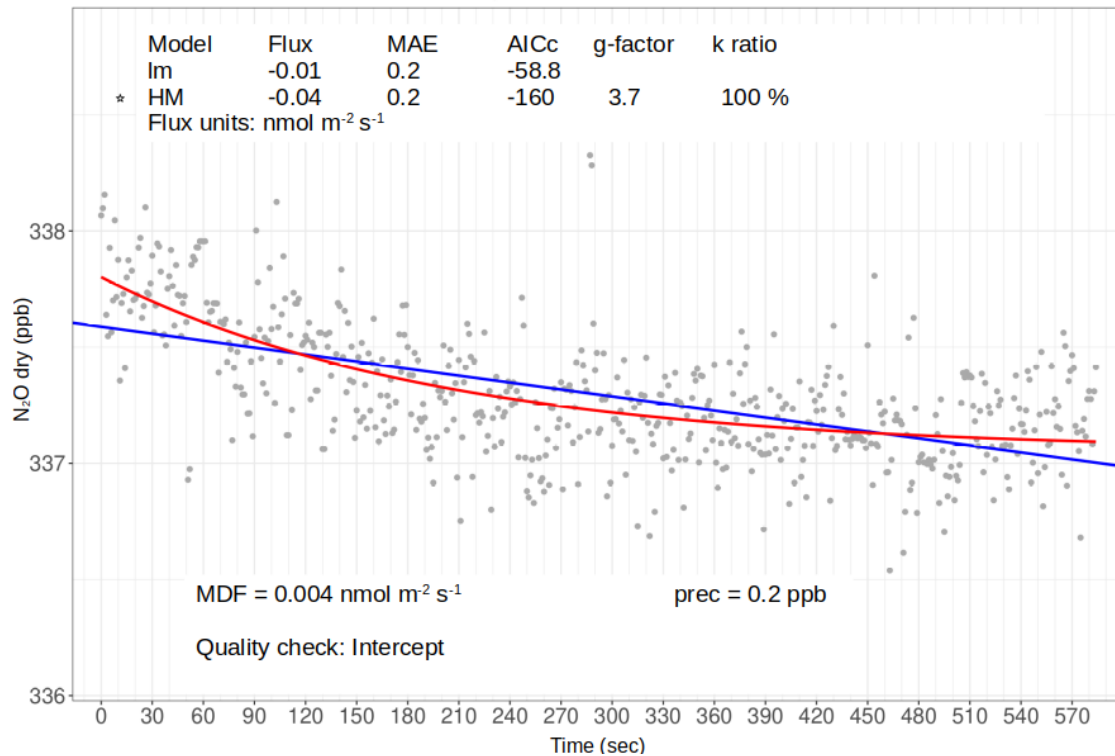
**Figure 5.** Mean  $\text{N}_2\text{O}$  *fluxes* (note: not concentrations) for transparent and opaque measurements, with number of measurement periods above the minimal detectable flux (MDF, %). Note the different y-axes for the upper and lower plots. The range indicates the upper and lower limit of the 95% confidence interval.

the non-linear fitting includes the linear fitting as a special case. When using a generic exponential function  $ae^{-bt}$  to fit data, where  $a$  and  $b$  are positive constants to be fitted, it can be approximated to a linear function if the data points are distributed linearly. This is because the exponential function can be expanded as a series, and when the rate constant  $b$  is small, the linear function dominates. Namely, the first three terms of the serial expansion are  $a(1 - bt + (bt)^2/2)$ , but when  $b$  is very small, *i.e.*,  $b \ll 1$ , it is reduced to  $a(1 - bt)$ , which is the linear form. The slope of the linear term is  $-ab$ ; if we take the time derivative of the original exponential function to calculate the slope, it gives  $-abe^{-bt}$ . When we expand it as a series and only take the first order term as  $b \ll 1$ , we again obtain the simplified  $-ab$  as slope. This means that if the data points are linear, the exponential fitting will automatically reduce to a linear fitting with the same slope. With our results, we show that for  $N_2O$  fluxes, indeed, flux estimates were reduced to the linear model and yielded identical results as the non-linear model. The second reason for favouring linear models in goFlux is that an unexplained nonlinearity can occur in curvature, *i.e.*, non-exponential curvatures, which can result in an overestimation of the flux estimates (Kutzbach et al., 2007). In goFlux, the curvature of the non-linear model can be restricted with  $g_f$  (see section 2.3). If the curvature was too large, leading to flux estimates over four times higher (with a  $g_f = 4$ ) than those from the linear model, the linear model was favoured. When we used a lower  $g_f$  of 1.25 for comparison, *i.e.*, allowing HM fluxes to be max. 25% higher than LM, we found that about one-quarter of the fluxes estimated by the non-linear model would have been excluded due to significant overestimation.

There is a tendency to favour LM over HM models in literature, primarily due to their simplicity. It is also generally assumed that concentration changes are linear during short chamber closure times, keeping uncertainties low (Hüppi et al., 2018; Kroon et al., 2008). However, because GHG concentrations naturally follow a non-linear behaviour within a closed system due to diffusion theory (Fick's first law) and leakage (Anthony et al., 1995; Kroon et al., 2008), LM models may introduce a bias, differing from HM model estimates by up to 60% (Hüppi et al., 2018; Kroon et al., 2008), resulting in less accurate flux estimates and GHG budgets. This has been thoroughly investigated for  $N_2O$  fluxes by Kroon et al. (2008), who found a large underestimation of  $N_2O$  fluxes by the LM model in their study. With our results, we suggest that all future  $N_2O$  flux chamber calculations should be done using HM models, which can be filtered for overestimation of fluxes when flux rates are larger. Novel software packages such as goFlux (Rheault et al., 2024) offer the possibility to easily integrate both LM and HM models, and report the flux rates in a reproducible way. We believe that these approaches will be crucial to facilitate the use of both LM and HM models, and, as a consequence, enable the chamber community to standardise their flux calculation methods. We further emphasise the importance of using all available data points for flux estimates, rather than selecting a subset of linear data. This is because our approach, which involves filtering out unrealistic values and visually verifying measurement periods after flux calculation, enhances the reproducibility and consistency of  $N_2O$  flux estimates.

### 3.4 Simulated differences in $N_2O$ flux rates between GC and PGA

Our results indicate that for transparent measurements, the  $N_2O$  fluxes we calculated using three simulated GC samples were, on average, 21.7% lower than the PGA fluxes ( $0.085 \mu\text{g } N_2O\text{-N m}^{-2}, \text{h}^{-1}$ ; data not shown). Specifically, positive values, *i.e.*, efflux, were generally underestimated ( $p < 0.001$ ,  $R_{\text{adj}}^2 = 0.37$ , Fig. 7). When we calculated fluxes using four simulated GC samples, negative  $N_2O$  fluxes appeared to be nearly identical with the  $N_2O$  uptake we calculated from the PGA (600 data



**Figure 6.** Example of scatterplot output from goFlux, showing  $\text{N}_2\text{O}$  concentrations in ppb from one measurement period, with information on linear (LM, blue line) and non-linear (HM, red line) flux calculations. For every measurement period, the chosen model is marked with a star (here, HM) according to the pre-defined quality check, indicated on the bottom of the graph (here, intercept). Flux values, mean absolute error (MAE), g-factor and k ratio are shown on top of the figure; here, a g-factor of 3.7 indicates that the HM flux value is 3.7 times higher than the one obtained from LM. Source: goFlux package (Rheault et al., 2024), font sizes modified.

points); however, efflux was still underestimated by 3% ( $p < 0.001$ ,  $R_{\text{adj}}^2 = 0.65$ ). Interestingly, by increasing the sample size to five or six, our calculated fluxes seemed to underestimate  $\text{N}_2\text{O}$  uptake, whilst efflux was overestimated ( $p < 0.001$ ,  $R_{\text{adj}}^2 = 0.85$ , Fig. 7). Overall, the  $\text{N}_2\text{O}$  fluxes we calculated using five samples underestimated  $\text{N}_2\text{O}$  fluxes by 2.2%, while calculations with

370

For opaque measurements, all  $\text{N}_2\text{O}$  flux estimates we calculated from GC simulations were lower than the fluxes estimates we calculated from the PGA, with an underestimation of 6.6%, 2.3%, 7.9%, and 8.1% for three, four, five, and six sample points (data not shown). With three samples, the calculated uptake rates were generally overestimated, while efflux was underestimated. When we calculated fluxes using four simulated GC- samples,  $\text{N}_2\text{O}$  uptake was still overestimated, while with

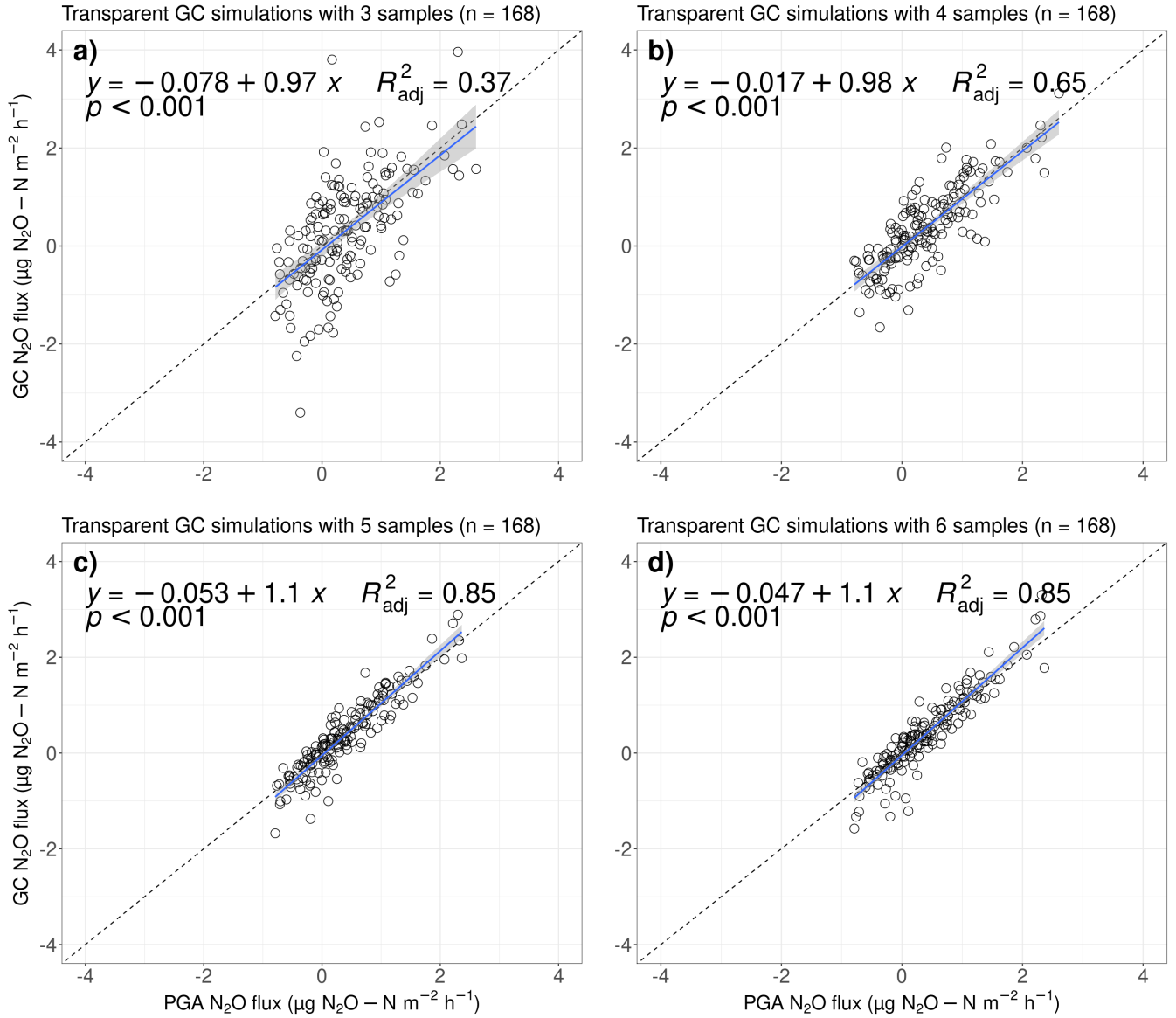
375 five and six sample points, it was slightly underestimated. However, compared to transparent measurements, the  $R_{\text{adj}}^2$  values were low (Fig. 8,  $R_{\text{adj}}^2 = 0.21, 0.51, 0.69$ , and  $0.68$ , respectively). The underestimation of GC fluxes may occur as a result of a smoothed out curve: when only few data points are available, variations in curves are naturally reduced. Furthermore, the

precision of our GC was 1.9 ppb compared to 0.2 ppb of the Aeris-N<sub>2</sub>O, resulting in less accurate measurements of the N<sub>2</sub>O concentrations. This may lead to a loss of detail in the curve, particularly in the peak values of the N<sub>2</sub>O concentrations, which  
380 can result in underestimation of the flux.

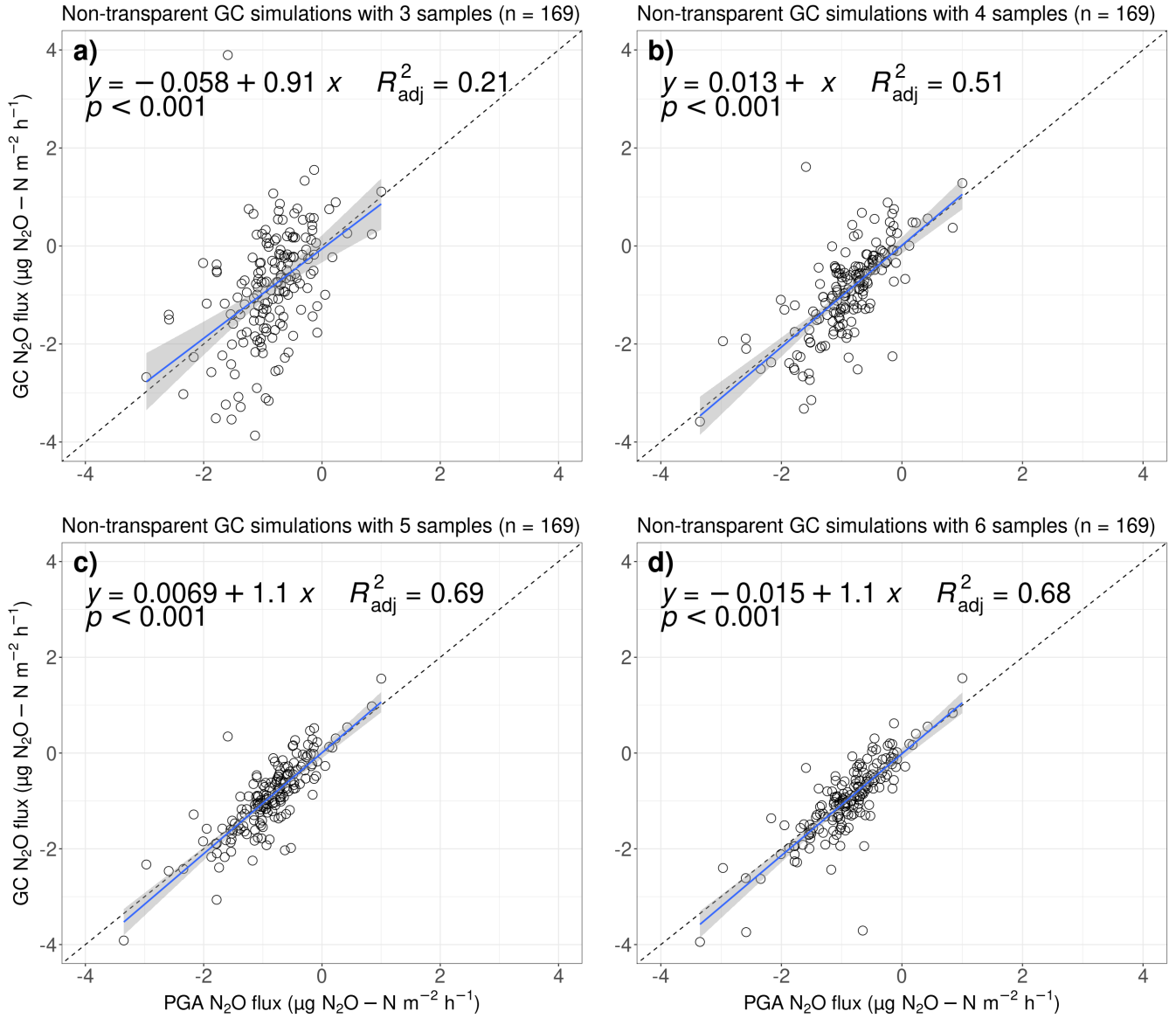
It is important to note that our comparison was made between our PGA and simulated GC measurements (Figure 9). For the GC simulations, we adjusted the instrument precision during the flux calculation, but no actual air samples were analysed by any GC instrument. Furthermore, our chamber closure time was considerably shorter than for most GC studies because of the condensation and temperature changes within the chamber. During prolonged chamber closure times, significant changes in  
385 the concentration gradient and chamber conditions can take place (see above), which are unlikely to be replicated in our GC-simulation. This difference in experimental design may actually be beneficial, as it allows us to isolate and study the effects of shorter closure times on N<sub>2</sub>O flux measurements. Furthermore, our sensitivity analysis with 4 simulated GC samples showed that even when we changed the sample times  $\pm 60$  sec compared to the original time stamp, flux rates differed less than 10%, with  $R^2_{\text{adj}}$  values between 92 and 98 (data not shown). We believe that the underestimation of N<sub>2</sub>O flux rates we calculated is,  
390 therefore, not a result of an inadequate simulation, but needs to be verified by future studies actually measuring N<sub>2</sub>O samples from nutrient-poor ecosystems in a GC.

Our results are consistent with previous studies that compared N<sub>2</sub>O flux rates between GCs and PGAs (Fig. 9), which concluded that GCs were suitable for measuring N<sub>2</sub>O fluxes under certain conditions (Christiansen et al., 2015; Brümmer et al., 2017). Christiansen et al. (2015) investigated the differences between a fast-responding analyser (a cavity ring-down  
395 spectroscopy analyser (Fleck et al., 2013)) and a high-precision GC in agricultural fields in Vancouver (Canada) by taking five GC samples at 0, 3, 10, 20, and 30 min chamber closure time. They found that N<sub>2</sub>O fluxes were very similar and did not differ significantly, with average N<sub>2</sub>O fluxes of  $47.6 \pm 8.4 \mu\text{g N}_2\text{O-N m}^{-2}\text{h}^{-1}$  from the fast-responding analyser and  $61.6 \pm 11.2 \mu\text{g N}_2\text{O-N m}^{-2}\text{h}^{-1}$  from GC, respectively. With a similar setup, Brümmer et al. (2017) compared N<sub>2</sub>O fluxes measured by a fast-responding analyser similar to the Aeris-N<sub>2</sub>O (a quantum cascade laser) and a GC from an low-flux  
400 agricultural grassland in Braunschweig (Germany). Their four GC samples taken at 0, 20, 40, and 60 min were highly scattered and rarely showed a distinctive trend, introducing a wide range of N<sub>2</sub>O fluxes between  $-26$  to  $39 \mu\text{g N}_2\text{O-N m}^{-2}\text{h}^{-1}$  with a standard error between 1 and  $44 \mu\text{g N}_2\text{O-N m}^{-2}\text{h}^{-1}$ . In contrast, the N<sub>2</sub>O fluxes measured by the fast-responding analyser only varied between 4 and  $32 \mu\text{g N}_2\text{O-N m}^{-2}\text{h}^{-1}$ , with standard errors below  $1.2 \mu\text{g N}_2\text{O-N m}^{-2}\text{h}^{-1}$ . This highlights three critical aspects: first, despite claiming low-flux environments, flux rates from agricultural fields are much higher than from a  
405 sub-Arctic peatland or other nutrient-poor ecosystems (Savage et al., 2014; Cowan et al., 2014), where capturing N<sub>2</sub>O fluxes is even more challenging. Second, low N<sub>2</sub>O fluxes tend to be very scattered, with large noise in comparison to the actual trend, *i.e.*, the change in concentration during chamber closure. This makes it very challenging to find trends when calculating fluxes if only few samples are available, let alone showing N<sub>2</sub>O uptake without high uncertainties (Cowan et al., 2014). Finally, it is crucial to know and test the precision of the instruments used in the field to get reliable flux estimates and minimal detectable  
410 fluxes (Kutzbach et al., 2007; Christiansen et al., 2015).

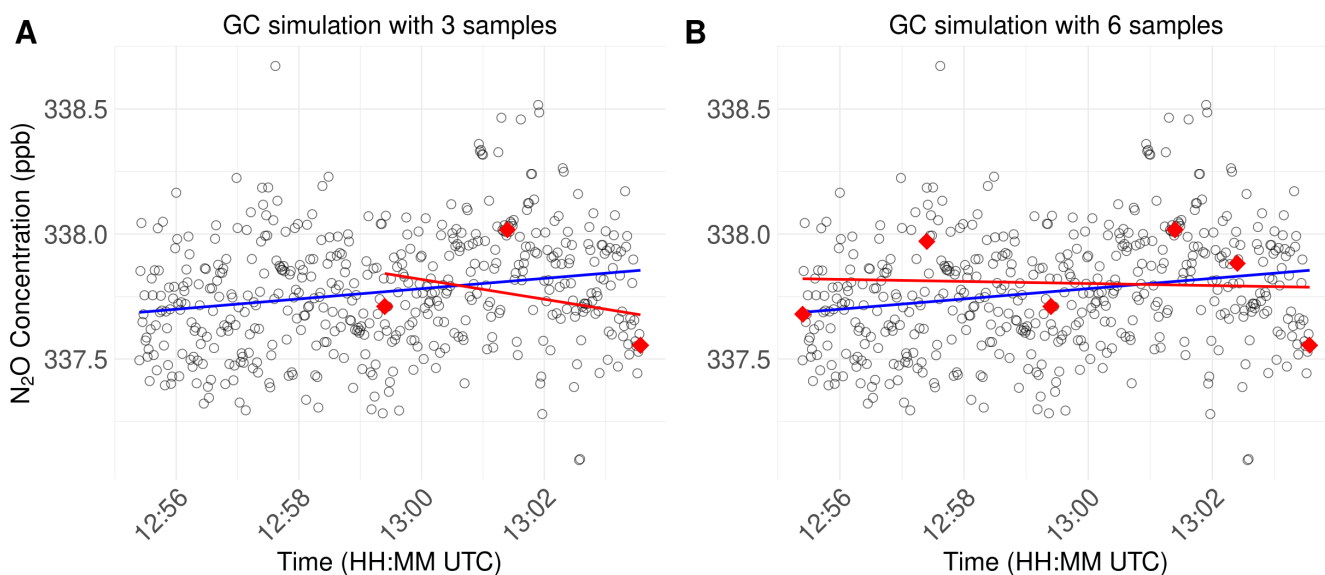
Our findings suggest that calculating N<sub>2</sub>O fluxes from three GC samples is likely to lead to an underestimation of the "real" flux (Kutzbach et al., 2007). We therefore strongly advise against using only three samples, as flux calculations may have to



**Figure 7.** Simulated GC fluxes from light measurements with a) 3, b) 4, c) 5, and d) 6 samples compared to flux measurements taken with our PGA for 10 min (n = 168). All fluxes were calculated with the goFlux package; results from "best.Flux" are shown. The blue trend line fits a generalised linear model, with the shaded area representing the 95% confidence interval. The shown equations and  $R^2_{\text{adj}}$  values follow the  $y \sim x$  equation.



**Figure 8.** Simulated GC fluxes from opaque measurements with a) 3, b) 4, c) 5, and d) 6 samples compared to flux measurements taken with our PGA for 10 min (n = 338). All fluxes were calculated with the goFlux package; results from "best.Flux" are shown. The blue trend line fits a generalised linear model, with the shaded area representing the 95% confidence interval. The shown equations and  $R^2_{\text{adj}}$  values follow the  $y \sim x$  equation.



**Figure 9.** Examples visualising the comparison of regression slopes obtained using 600 data points from the portable gas analyser (blue line) vs 3 (A) or 6 (B) virtual samples mimicking manually defined sample times for subsequent analysis in a gas chromatograph (red line), respectively.

be discarded if only one sample is erroneous. In contrast, using four to six samples yield very comparable results to those obtained with a PGA, depending on the precision of the GC method. However, the small sample size restricts our ability to confidently identify trends in  $\text{N}_2\text{O}$  fluxes. Undetected errors can bias flux estimates due to the high impact of each data point on the regression slope. To compare previous  $\text{N}_2\text{O}$  flux measurements to novel data sets measured with a PGA, it is crucial to investigate differences between these methods. To achieve this, novel instruments have to be tested on their precision, noise, and accuracy, as well as potential interference with water vapour (Grace et al., 2020; Ahmed et al., 2024).

We suggest that measuring  $\text{N}_2\text{O}$  fluxes with fast-responding analysers in nutrient-poor ecosystems should be considered for all future studies. PGAs, for example, have two main advantages over the GC method: they collect a large amount of sample points, and the quality of these can be checked *in situ* during the measurement period. With more samples, there are more data points, which then results in the option to reduce chamber closure times and further reduce artefacts caused by sealing off a part of a soil profile by a closed chamber (Brümmer et al., 2017). If errors happen in the field, *e.g.*, leakage and pressure change (Rochette and Hutchinson, 2015), they are visible in the online interface of the PGA. This real-time *in situ* control of  $\text{N}_2\text{O}$  concentrations allows for direct optimisation in the field and increases the quality of flux measurements (Fiedler et al., 2022). A practical result of that is that measurement periods can be interrupted and repeated in the field at any time, ensuring high quality of the flux measurements, as well as an optimal use of time in the field, particularly since chamber closure times with PGAs are shorter than with GCs. However, PGAs have some drawbacks: weighing 10-20 kg (including batteries), they are heavier than GC vials. Additionally, their power consumption requires regular backups, and factors such as heavy vibrations,

430 particles, water, and sudden pressure changes can contaminate the laser cell (Fiedler et al., 2022). With good planning and care, it is, however, easily possible to deal with these disadvantages.

## 4 Conclusions

In our study, we established a manual flux chamber method using a portable gas analyser (PGA) capable of quantifying low  $\text{N}_2\text{O}$  fluxes in nutrient-poor ecosystems and, based on our extensive experience with the system, provide detailed practical  
435 suggestions on how to collect high-quality measurements in low-flux ecosystems (see SI). To our knowledge, our study represents the first extensive analysis of  $\text{N}_2\text{O}$  fluxes measured with manual flux chambers in a (very) nutrient-poor, (sub-) Arctic ecosystem. Our laboratory tests confirmed that our PGA (Aeris MIRA Ultra  $\text{N}_2\text{O}$  /  $\text{CO}_2$ ) is well suited for measuring low  $\text{N}_2\text{O}$  fluxes, with low noise, minimal water interference, and negligible signal drift. With our PGA- chamber system, we are able to report very low  $\text{N}_2\text{O}$  flux values with positive and negative signs, indicating both  $\text{N}_2\text{O}$  efflux and uptake. Because PGAs  
440 allow for near-continuous monitoring of concentration changes with high precision and low detection limits, we recommend, with a chamber height of 25 cm, chamber closure times of 3-5 min for opaque and >4 min for transparent measurements to minimise the impacts of the ecosystem due to the measurement setup (*e.g.*, changes in temperature and humidity). To strike a balance between detection sensitivity and measurement efficiency, we suggest using a standard 5-minute closure time for all measurements with smaller chambers, which enables us to detect around 70% of fluxes. This allows for most  $\text{N}_2\text{O}$  fluxes to be  
445 detected in this setup; however, the sensitivity depends on the effective chamber height.

We further recommend using non-linear models for  $\text{N}_2\text{O}$  flux calculations (HM; (Hutchinson and Mosier, 1981)), with filters to address overestimation at higher flux rates. Novel software packages such as goFlux (Rheault et al., 2024) simplify the integration of both linear and non-linear models, and report flux rates in a reproducible way; such approaches are key to standardising flux calculations across the chamber community. Using non-linear models, as well as standardised and well-  
450 documented calculation and quality control, allows to consider the entire time series of measurement periods, and does not require (subjective) expert knowledge to first restrict the datasets to a suitable section, and only afterwards calculate flux estimates. We stress this importance of using all available data points for flux estimates to improve the reproducibility and consistency of  $\text{N}_2\text{O}$  flux estimates.

We recommend using PGAs in future  $\text{N}_2\text{O}$  studies in nutrient-poor ecosystems whenever feasible. PGAs collect more data  
455 points (typically about 1 sample / second) and allow for real-time quality control in the field, while GC measurements may be limited by low flux rates and thus fall below the detection limits or lack clear trends. To get the most out of PGAs, it is essential to determine their precision and use a suitable chamber closure time. In that way, most  $\text{N}_2\text{O}$  fluxes can be detected, which is especially important in nutrient-poor ecosystems, where  $\text{N}_2\text{O}$  fluxes are often very low. Future studies using other chamber designs may benefit from re-evaluating chamber closure times and flux calculation methods to find optimum customised setups.

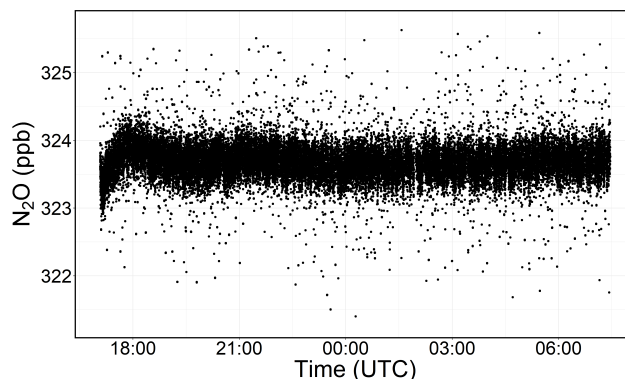
460 While this study concentrates on the methodological aspects of quantifying  $\text{N}_2\text{O}$  fluxes in a nutrient-poor ecosystem, a follow-up study will investigate the environmental drivers of  $\text{N}_2\text{O}$  fluxes. Because most studies in the (sub-) Arctic have reported  $\text{N}_2\text{O}$  from opaque measurements only, there is a lack of data on how soil  $\text{N}_2\text{O}$  fluxes differ in various light conditions,

especially in Arctic ecosystems (Stewart et al., 2012). Our results demonstrate notable differences between transparent and opaque conditions that require further investigation. This fills an important gap in N<sub>2</sub>O studies from the Arctic, where negative  
465 fluxes have been observed, but could not be investigated due to measurement accuracy not being high enough (Voigt et al., 2020). Further, this novel finding highlights the need for future research on N<sub>2</sub>O fluxes in sub-Arctic ecosystems and other nutrient-poor ecosystems, and their potential response to global warming.

*Code availability.* The scripts for processing and analysing the data are publicly available at <https://git.bgc-jena.mpg.de/ntriches/data-analysis/-/tags/2024-12-12-triches-amts submission-n2oadvances> under the terms of the GNU General Public License version 3.

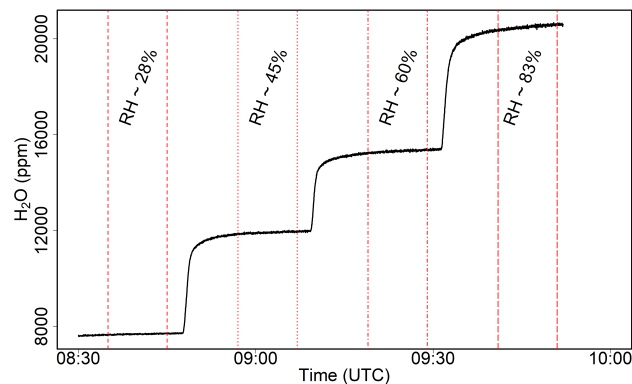
## 470 Appendix A

Sampling ambient air using Aeris-N<sub>2</sub>O analyser, Fig. A1 shows the entire 15 hours long run without removing the first 5 hours (1 Hz), where the water vapour mole fractions were not stable.



**Figure A1.** Time series of the N<sub>2</sub>O-Analyser of 15 hrs long run.

Changes of H<sub>2</sub>O mole fractions during the water interference test (see Fig. A2). Corresponding relative humidity values were noted and the data used for the comparison were designated with red vertical lines.



**Figure A2.** H<sub>2</sub>O mole fractions measured by Aeris-N<sub>2</sub>O analyser during the water interference test. The corresponding relative humidity (RH) values were given for each step. Red lines demonstrate the data used to assess the water interference at each RH steps.

475 *Author contributions.* NYT designed the experiment, collected, and processed the data, and did the laboratory test together with AB. AB further analysed and reported the laboratory data, and created the QGIS figures. NYT and JE developed and implemented the scripts used for data processing, quality checks, analysis, and GC simulations. TV wrote the mathematical explanation of non-linear and linear models. NYT wrote the first draft of the manuscript, and AB, AMV, MEM, MG, MH and TV provided valuable comments that helped improving it. NYT was supervised by MG, MH, TV, AMV, and MEM. MH and MG were responsible for funding acquisition.

480 *Competing interests.* The authors declare that they have no conflict of interest.

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